

## Supporting Information

### Asymmetric Nanochannel-Ionchannel Hybrid for Ultrasensitive and Label-free Detection of Copper Ions in Blood

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**Scheme S1.** The chemical structural formula of PGA.

**Figure S1.** ICR ratio for different ionic transport process.

**Figure S2.** Cu<sup>2+</sup>-PGA recognition kinetics.

**Figure S3.** *I-V* curve of PGA-free PAA with and without Cu<sup>2+</sup> (10 μM).

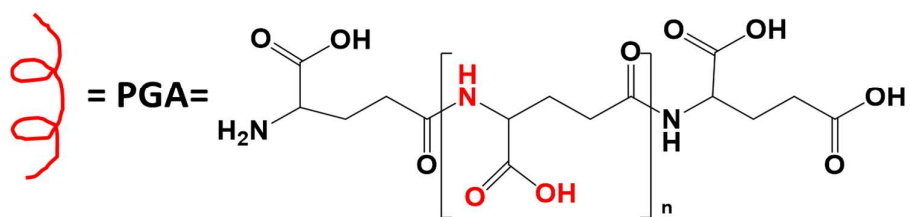
**Figure S4.** *I-V* curve of PGA- modified symmetric PAA membranes in the present of different Cu<sup>2+</sup> (the amplify inset).

**Figure S5.** The *I-V* profiles of the PGA modified nanochannel-ionchannel hybrid for detection of Cu<sup>2+</sup> using different methods.

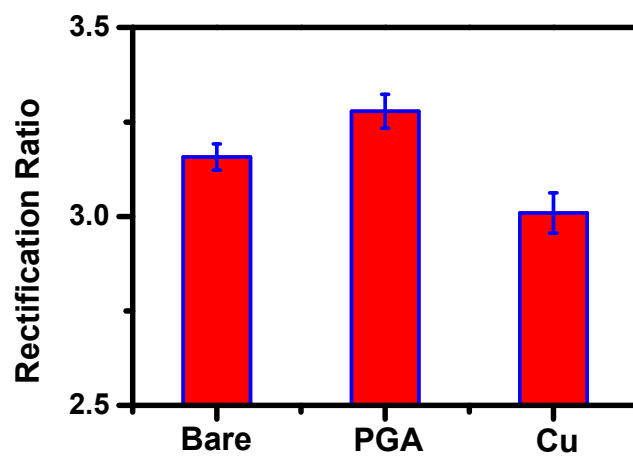
**Figure S6.** Cu<sup>2+</sup>-PGA recognition kinetics for different systems.

**Figure S7.** The current values versus of Cu<sup>2+</sup> concentrations.

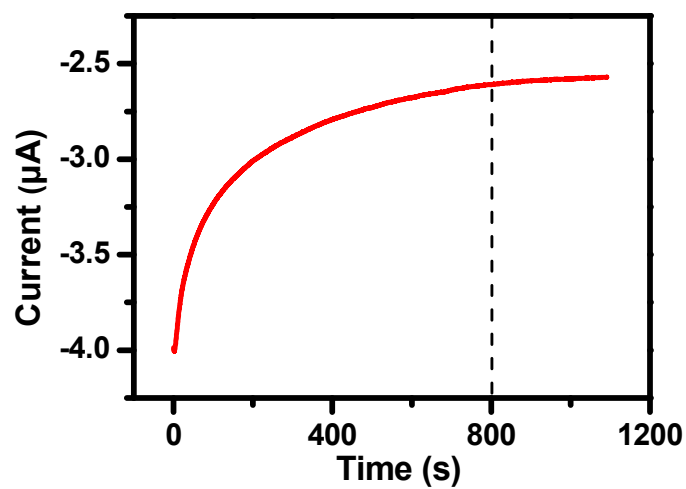
**Figure S8.** The *I-V* profiles of different concentrations of Cu<sup>2+</sup> in tap water samples.



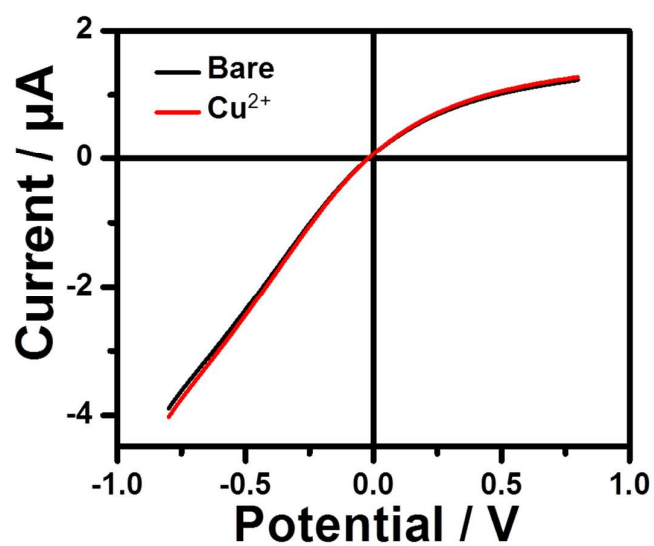
**Scheme S1.** The chemical structural formula of PGA.



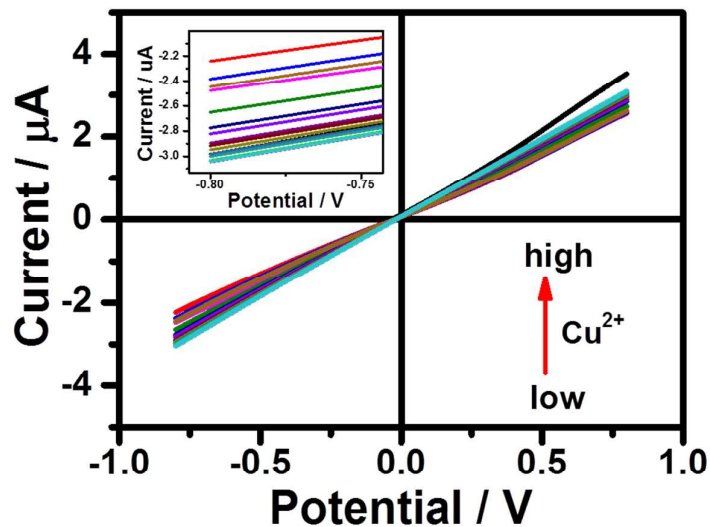
**Figure S1.** ICR ratio for different processes.



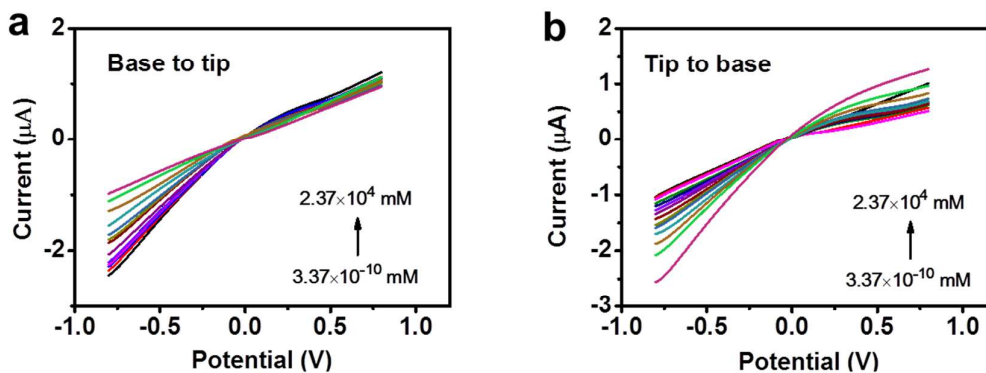
**Figure S2.**  $\text{Cu}^{2+}$ -PGA recognition kinetics. The dashed line means the time needed for PGA- $\text{Cu}^{2+}$  recognition equilibrium. The  $\text{Cu}^{2+}$  concentration is  $3.37 \times 10^{-10} \mu\text{M}$ .



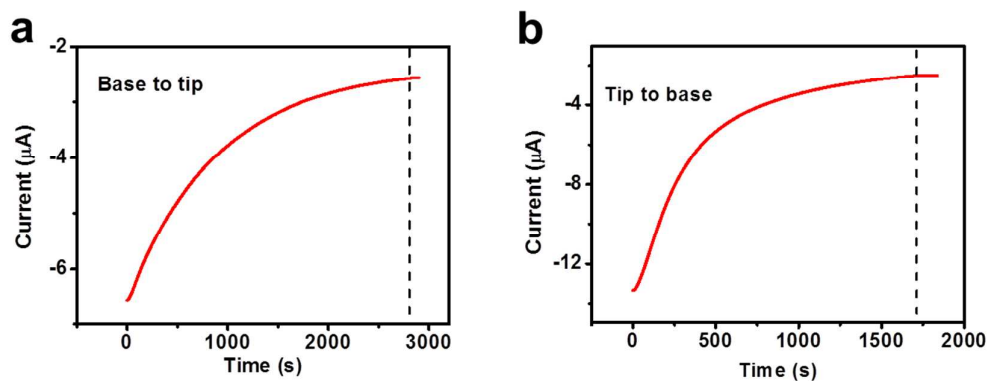
**Figure S3.**  $I$ - $V$  curve of PGA-free PAA with and without  $\text{Cu}^{2+}$  ( $10 \mu\text{M}$ ).



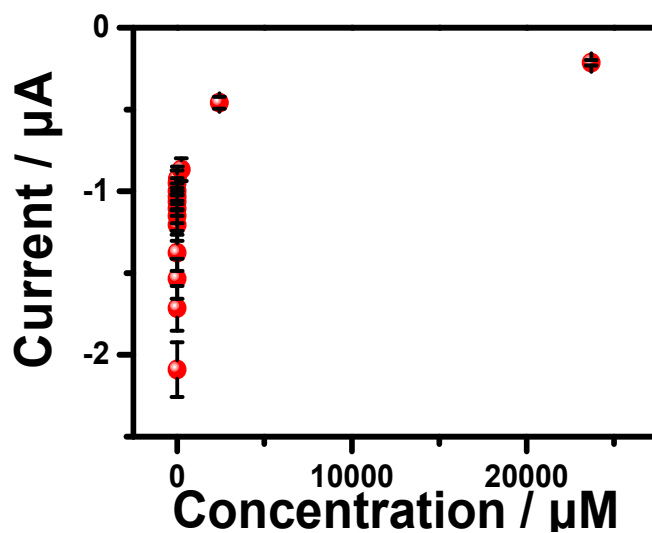
**Figure S4.** *I-V* curve of PGA- modified symmetric PAA membranes in the present of different  $\text{Cu}^{2+}$  (the amplify inset).



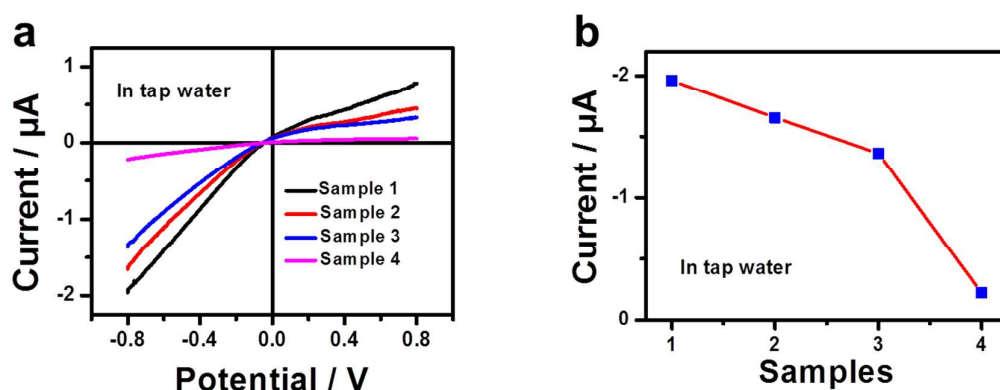
**Figure S5.** The *I-V* profiles of the PGA modified nanochannel-ionchannel hybrid for detection of  $\text{Cu}^{2+}$  using different methods. (a)  $\text{Cu}^{2+}$  ions were added in the nanochannel side of PAA membrane (b)  $\text{Cu}^{2+}$  ions were added in the ionchannel side of PAA membrane.



**Figure S6.**  $\text{Cu}^{2+}$ -PGA recognition kinetics for different systems. (a)  $\text{Cu}^{2+}$  ions were added in the nanochannel side of PAA membrane (b)  $\text{Cu}^{2+}$  ions were added in the ionchannel side of PAA membrane. The dashed line means the time needed for PGA- $\text{Cu}^{2+}$  recognition equilibrium. The  $\text{Cu}^{2+}$  concentration is  $3.37 \times 10^{-10} \mu\text{M}$ .



**Figure S7.** The current values versus of  $\text{Cu}^{2+}$  concentrations.



**Figure S8.** (a) The  $I$ - $V$  profiles of different concentrations of  $\text{Cu}^{2+}$  in tap water samples. (Sample 1: no additional  $\text{Cu}^{2+}$  in tap water; Sample 2: added  $5 \times 10^{-7} \mu\text{M}$  in tap water; Sample 3: added  $5 \times 10^{-3} \mu\text{M}$  in tap water; Sample 4: added  $5 \times 10^5 \mu\text{M}$  in tap water). (b) The ionic current at -0.8 V versus different tap water samples.

**Table S1.** Results of the detection of  $\text{Cu}^{2+}$  in real tap water samples.

Sample	Added $\text{Cu}^{2+}$ ( $\mu\text{M}$ )	Measured $\text{Cu}^{2+}$ ( $\mu\text{M}$ )	Recovery (%)
Sample 1	0	$2.27 \times 10^{-8}$	/
Sample 2	$5 \times 10^{-7}$	$4.99 \times 10^{-7}$	99.9
Sample 3	$5 \times 10^{-3}$	$4.67 \times 10^{-3}$	93.4
Sample 4	$5 \times 10^5$	$5.63 \times 10^5$	112.6