

## **SUPPORTING INFORMATION**

# **Absorption of H<sub>2</sub>S and CO<sub>2</sub> in Aqueous Solutions of Tertiary-Amine Functionalized Protic Ionic Liquids**

*Kuan Huang<sup>a,b</sup>\*, Jia-Yin Zhang<sup>a</sup>, Xing-Bang Hu<sup>b</sup>, You-Ting Wu<sup>b</sup>\**

<sup>a</sup>Poyang Lake Key Laboratory of Environment and Resource Utilization (Nanchang University), Ministry of Education; School of Resources Environmental and Chemical Engineering, Nanchang University, Nanchang, Jiangxi 330031, China.

<sup>b</sup>School of Chemistry and Chemical Engineering, Nanjing University, Nanjing, Jiangsu 210023, China.

## Characterization results

**[TMEDA][AcO]**  $^1\text{H}$  NMR (300 MHz, D<sub>2</sub>O, 298.2 K, TMS),  $\delta$  (ppm): 1.72 (3H, s), 2.40 (12H, s), 2.81 (4H, s);  $^{13}\text{C}$  NMR (300 MHz, D<sub>2</sub>O, 298.2 K, TMS),  $\delta$  (ppm): 23.30, 43.53, 53.41, 181.34; Elemental Analysis, calculated for C<sub>8</sub>H<sub>20</sub>N<sub>2</sub>O<sub>2</sub>: C 54.51 %, H 11.44 %, N 15.89 %, found: C 54.71 %, H 11.22 %, N 16.13 %; MS: 117.19 for C<sub>6</sub>H<sub>17</sub>N<sub>2</sub><sup>+</sup> (calculated: 117.14), 59.08 for C<sub>2</sub>H<sub>3</sub>O<sub>2</sub><sup>-</sup> (calculated: 59.01); TGA, decomposition temperature: 438 K.

**[TMPDA][AcO]**  $^1\text{H}$  NMR (300 MHz, D<sub>2</sub>O, 298.2 K, TMS),  $\delta$  (ppm): 1.90 (5H, m), 2.56 (12H, s), 2.79 (4H, t);  $^{13}\text{C}$  NMR (300 MHz, D<sub>2</sub>O, 298.2 K, TMS),  $\delta$  (ppm): 21.57, 23.40, 43.22, 55.57, 181.22; Elemental Analysis, calculated for C<sub>9</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>: C 56.81 %, H 11.65 %, N 14.72 %, found: C 57.02 %, H 11.66 %, N 14.58 %; MS: 132.10 for C<sub>7</sub>H<sub>19</sub>N<sub>2</sub><sup>+</sup> (calculated: 132.16), 59.08 for C<sub>2</sub>H<sub>3</sub>O<sub>2</sub><sup>-</sup> (calculated: 59.01); TGA, decomposition temperature: 441 K.

**[BDMAEE][AcO]**  $^1\text{H}$  NMR (300 MHz, D<sub>2</sub>O, 298.2 K, TMS),  $\delta$  (ppm): 1.92 (3H, s), 2.60 (12H, s), 3.00 (4H, t), 3.76 (4H, t);  $^{13}\text{C}$  NMR (300 MHz, D<sub>2</sub>O, 298.2 K, TMS),  $\delta$  (ppm): 23.39, 43.30, 57.07, 65.68, 181.28; Elemental Analysis, calculated for C<sub>10</sub>H<sub>24</sub>N<sub>2</sub>O<sub>3</sub>: C 54.52 %, H 10.98 %, N 12.72 %, found: C 54.63 %, H 11.17 %, N 12.55 %; MS: 161.12 for C<sub>8</sub>H<sub>21</sub>N<sub>2</sub>O<sup>+</sup> (calculated: 161.16), 59.08 for C<sub>2</sub>H<sub>3</sub>O<sub>2</sub><sup>-</sup> (calculated: 59.01); TGA, decomposition temperature: 449 K.

**[BMEE][AcO]**  $^1\text{H}$  NMR (300 MHz, D<sub>2</sub>O, 298.2 K, TMS),  $\delta$  (ppm): 1.92 (3H, s), 3.04 (12H, m), 3.79 (4H, t), 3.89 (8H, t);  $^{13}\text{C}$  NMR (300 MHz, D<sub>2</sub>O, 298.2 K, TMS),  $\delta$  (ppm): 21.57, 23.40, 43.22, 55.57, 181.22; Elemental Analysis, calculated for C<sub>14</sub>H<sub>28</sub>N<sub>2</sub>O<sub>5</sub>: C 55.24 %, H 9.27 %, N 9.20 %, found: C 54.98 %, H 9.36 %, N 9.43 %; MS: 245.22 for C<sub>12</sub>H<sub>25</sub>N<sub>2</sub>O<sub>3</sub><sup>+</sup> (calculated: 245.19), 59.08 for C<sub>2</sub>H<sub>3</sub>O<sub>2</sub><sup>-</sup> (calculated: 59.01); TGA, decomposition temperature: 453 K.