## Supporting Information

# Unmasking Amides: Ruthenium-Catalyzed Protodecarbonylation of $N$ Substituted Phthalimide Derivatives 

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## 1. General Information.

All reagents were obtained from commercial sources and used as supplied. All reactions were carried out in flame-dried glassware under argon atmosphere unless otherwise noted. Catalytic experiments were performed in Schlenk-type flasks under argon atmosphere unless otherwise noted. Organic solutions were concentrated under reduced pressure using a rotary evaporator. Thin-layer chromatography (TLC) were carried out on 0.25 mm Merck silica gel (60-F254). Flash column chromatography was performed using silica gel Silica $60 \mathrm{M}, 0.04-0.063 \mathrm{~mm}$. $N$-methyl-2-pyrrolidone (NMP) was distilled under reduced pressure and stored under molecular sieves and argon atmosphere. Technical grade petroleum ether (40-60) and ethyl acetate were used for column chromatography. $\mathrm{CDCl}_{3}$ was stored under nitrogen over molecular sieves. NMR spectra were recorded on an AVANCE III 400 spectrometer. ${ }^{1} \mathrm{H}$ NMR spectra were referenced to residual protiated solvent ( $\delta=7.26 \mathrm{ppm}$ for $\mathrm{CDCl}_{3}, \delta=2.50 \mathrm{ppm}$ for DMSO- $d_{6}$ and $\delta$ $=2.05 \mathrm{ppm}$ for acetone- $d_{6}$ ) and ${ }^{13} \mathrm{C}$ chemical shifts are reported relative to deuterated solvents ( $\delta=77.0 \mathrm{ppm}$ for $\mathrm{CDCl}_{3}, \delta=39.5 \mathrm{ppm}$ for DMSO- $d_{6}$ and $\delta=29.8 \mathrm{ppm}$ for acetone- $d_{6}$. The peak patterns are indicated as follows: $s$, singlet; d , doublet; t , triplet; q , quartet; $m$, multiplet, and br. for broad. GC-MS analyses were performed with a GCMSQP2010S (Shimadzu) instrument with a GC-2010 equipped with a 30 m capillary column (Supelco, SLBTM-5ms, fused silica capillary column, $30 \mathrm{~m} \times 0.25 \mathrm{~mm} \times 0.25 \mathrm{~mm}$ film thickness), which was used with helium as the vector gas. The following GC conditions were used: initial temperature $80^{\circ} \mathrm{C}$ for 2 minutes, then rate $20^{\circ} \mathrm{C} / \mathrm{min}$ until $280^{\circ} \mathrm{C}$ and $280^{\circ} \mathrm{C}$ for 28 minutes. HRMS were recorded on a Waters Q-Tof 2 mass spectrometer at the corresponding facilities of the CRMPO, Centre Régional de Mesures Physiques de l'Ouest, Université de Rennes 1.

## 2. Preparation of substrates 1.

2.1. Method A: Phthalic anhydride ( $5 \mathrm{mmol}, 0.74 \mathrm{~g}, 1 \mathrm{eq}$. ) and the corresponding aniline ( $5 \mathrm{mmol}, 1 \mathrm{eq}$.) were refluxed in acetic acid ( 30 mL ) for 2-5 hours. Once at room temperature, water was added and the solid recovered by filtration. After drying under vacuum the desired phthalimide $\mathbf{1}$ was obtained.

2.2. Method B: Phthalimide ( $7 \mathrm{mmol}, 1.03 \mathrm{~g}, 1 \mathrm{eq}$.), potassium carbonate ( $14 \mathrm{mmol}, 2.59$ $\mathrm{g}, 2 \mathrm{eq}$ ) and the corresponding alkyl or benzyl halide ( $14 \mathrm{mmol}, 2 \mathrm{eq}$.) were heated at $40^{\circ} \mathrm{C}$ in $\mathrm{N}, \mathrm{N}$-dimethylformamide ( 6 mL ) for 18 hours. After solvents evaporation under vacuum, water was added to the reaction mixture followed by extraction with DCM. The combined organic phases were dried over $\mathrm{MgSO}_{4}$, filtered, and concentrated in vacuo. The desired phthalimide $\mathbf{1}$ was purified by silica gel column chromatography with a mixture of petroleum ether and ethyl acetate as eluent.

2.3. Method C: Phthalic anhydride ( $10 \mathrm{mmol}, 1.48 \mathrm{~g}, 1 \mathrm{eq}$.) and 1,3-diaminopropane ( 5 $\mathrm{mmol}, 0.5 \mathrm{eq}$.) were refluxed in acetic acid ( 15 mL ) for 8 hours. Once at room temperature, water was added and the solid recovered by filtration. After drying under vacuum the desired phthalimide $\mathbf{1}$ was obtained.

2.4. Method D: Hexahydrophthalic anhydride ( $10 \mathrm{mmol}, 1.54 \mathrm{~g}, 1 \mathrm{eq}$.) and aniline ( 10 mmol, 1 eq.) and THF ( 15 mL ) were added to a 100 mL round bottom flask. The solution was stirred for 30 min at $40^{\circ} \mathrm{C}$. Removal of the solvent using a rotary evaporator gave the corresponding carboxylic acid-amide as a white solid. The white solid was then heated
at $190{ }^{\circ} \mathrm{C}$ under Ar for 4 h . The desired phthalimide was purified by silica gel column chromatography with a mixture of petroleum ether and ethyl acetate as eluent.


## 3. Characterization data of substrates 1.


$N$-Methylphthalimide (1a): Prepared according to Method B starting from iodomethane in $88 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.78$ (dd, $J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.66(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.13(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{1}$

$N$-Butylphthalimide (1b): Prepared according to Method B starting from 1bromobutane in $98 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.82$ (dd, $J=5.6$ $\mathrm{Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.70-7.68(\mathrm{~m}, 2 \mathrm{H}), 3.67(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.68-1.61(\mathrm{~m}, 2 \mathrm{H}), 1.40-$ $1.31(\mathrm{~m}, 2 \mathrm{H}), 0.93(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{2}$


N -isopropylphthalimide (1c): Prepared according to Method B starting from 2bromopropane in $78 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.79$ (dd, $J=5.6$ $\mathrm{Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.68 (dd, $J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 4.57-4.47 (m, 1H), 1.48 (d, $J=7.2$ $\mathrm{Hz}, 6 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{3}$

$\boldsymbol{N}$-(1-Adamantyl)phthalimide (1d): Prepared according to Method A starting from amantadine in $35 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.75$ (dd, $J=5.2 \mathrm{~Hz}$, $3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.66(\mathrm{dd}, J=5.2 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.52(\mathrm{~d}, J=2.8 \mathrm{~Hz}, 6 \mathrm{H}), 2.17(\mathrm{~s}, 3 \mathrm{H})$, 1.81-1.70 (m, 6H) ppm. The spectral data match those previously reported. ${ }^{4}$


5-Phthalimidovaleronitrile (1e): Prepared according to Method B starting from 5bromovaleronitrile in $87 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.83$ (dd, $J=$ $5.6 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.72(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.72(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.42(\mathrm{t}, J$ $=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.88-1.81(\mathrm{~m}, 2 \mathrm{H}), 1.74-1.66(\mathrm{~m}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{3}$

$N$-[2-Methoxyethyl]phthalimide (1f): Prepared according to Method A starting from 2methoxyethanamine in $75 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.84$ (dd, $J$ $=5.2 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.70(\mathrm{dd}, J=5.2 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.89(\mathrm{t}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.63(\mathrm{t}$, $J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.34(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{5}$

$N$-Acetonylphthalimide (1g): Prepared according to Method B starting from chloroacetone in $50 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.86$ (dd, $J=5.2$
$\mathrm{Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.73$ (dd, $J=5.2 \mathrm{HZ}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.49(\mathrm{~s}, 2 \mathrm{H}), 2.26$ (s, 3H) ppm. The spectral data match those previously reported. ${ }^{6}$


Methyl phthalimidoacetate (1h): Prepared according to Method B starting from ethyl bromoacetate in $75 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.87$ (dd, $J=5.6$ $\mathrm{Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.73(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.42(\mathrm{~s}, 2 \mathrm{H}), 4.22(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H})$, $1.27(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{7}$

$N$-Phenylphthalimide (1i): Prepared according to Method A starting from aniline in $80 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.96(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.80 (dd, $J=5.2 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.52 (dd, $J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.34-7.27 (m, 3H) ppm . The spectral data match those previously reported. ${ }^{2}$

$\boldsymbol{N}$-( $\boldsymbol{p}$-Tolyl)phthalimide (1j): Prepared according to Method A starting from $p$-toluidine in $72 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.94(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.78(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.31(\mathrm{~s}, 4 \mathrm{H}), 2.41(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{2}$

$\boldsymbol{N}$-p-Anisylphthalimide (1k): Prepared according to Method A starting from $p$-anisidine in $80 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.94(\mathrm{dd}, J=5.2 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.78 (dd, $J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.34(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.02(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.85$ $(\mathrm{s}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{2}$


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$\boldsymbol{N}$-( $\boldsymbol{p}$-Fluorophenyl)phthalimide (11): Prepared according to Method A starting from $\boldsymbol{p}$ fluoroaniline in $82 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.86$ (dd, $J=5.2$ $\mathrm{Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.71 (dd, $J=5.2 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.34 (dd, $J=9.2 \mathrm{~Hz}, 4.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.11 (dd, $J=8.8 \mathrm{~Hz}, 8.8 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{19} \mathrm{~F}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=-113.8 \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{8}$

$\boldsymbol{p}$-Phthalimidoacetophenone (1m): Prepared according to Method A starting from $\boldsymbol{p}$ aminoacetophenone in $79 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=8.10(\mathrm{dd}, J$ $=6.8 \mathrm{~Hz}, 2.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.97(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.82(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}$, $2 \mathrm{H}), 7.63(\mathrm{dd}, J=6.8 \mathrm{~Hz}, 2.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.64(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{2}$

$\boldsymbol{N}$-(p-Ethoxycarbonylphenyl)phthalimide (1n): Prepared according to Method A starting from benzocaine in $50 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.19$
(dd, $J=6.8 \mathrm{~Hz}, 1.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.97$ (dd, $J=5.2 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.82(\mathrm{dd}, J=5.2 \mathrm{~Hz}, 2.8$ $\mathrm{Hz}, 2 \mathrm{H}), 7.59(\mathrm{dd}, J=6.8 \mathrm{~Hz}, 1.6 \mathrm{~Hz}, 2 \mathrm{H}), 4.41(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.41(\mathrm{t}, J=7.2 \mathrm{~Hz}$, $3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{9}$

$\boldsymbol{N}$-(p-Nitrophenyl)phthalimide (10): Prepared according to Method A starting from $p$ nitroaniline in $63 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\boldsymbol{\delta}=8.38$ (dd, $J=7.2 \mathrm{~Hz}$, $2.0 \mathrm{~Hz}, 2 \mathrm{H}), 8.00(\mathrm{dd}, J=5.2 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.85(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.78$ (dd, $J=7.2 \mathrm{~Hz}, 2.0 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{2}$

$\boldsymbol{N}$ - $\boldsymbol{p}$-Bromophenylphthalimide (1p): Prepared according to Method A starting from $\boldsymbol{p}$ bromoaniline in $88 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.96$ (dd, $J=5.2$ $\mathrm{Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.80$ (dd, $J=5.2 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.63$ (d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.36$ (d, $J=$ $8.8 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{10}$

$\boldsymbol{N}$ - $\boldsymbol{p}$-Iodophenylphthalimide (1q): Prepared according to Method A starting from $p$ iodoaniline in $80 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.96-7.94(\mathrm{~m}, 2 \mathrm{H})$, 7.84-7.79 (m, 4H), $7.23(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{11}$

$\boldsymbol{N}$-( $\boldsymbol{m}$-Methoxyphenyl)phthalimide (1r): Prepared according to Method A starting from $m$-anisidine in $76 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.95$ (dd, $J=5.2 \mathrm{~Hz}$, $3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.78$ (dd, $J=5.2 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.41$ (dd, $J=8.0 \mathrm{~Hz}, 8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.04-$ $7.02(\mathrm{~m}, 1 \mathrm{H}), 6.99-6.96(\mathrm{~m}, 1 \mathrm{H}), 6.94(\mathrm{dd}, J=2.4 \mathrm{~Hz}, 1.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.84(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{12}$

$\boldsymbol{N}$-(o-Chlorophenyl)phthalimide (1s): Prepared according to Method A starting from $o$-chloroaniline in $70 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.96(\mathrm{dd}, J=5.6$ $\mathrm{Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.80 (dd, $J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.59-7.55 (m, 1H), 7.46-7.38 (m, 2H), 7.37-7.35 (m, 1H) ppm. The spectral data match those previously reported. ${ }^{13}$

$1 t$
$\mathbf{N}$-o-Tolylphthalimide (1t): Prepared according to Method A starting from o-toluidine in $43 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.96(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.80 (dd, $J=5.6 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.39-7.31$ (m, 3H), 7.21 (d, $J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.21$ (s, $3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{4}$

$\mathbf{N}$-o-Methoxyphenylphthalimide (1u): Prepared according to Method A starting from $o$-anisidine in $85 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.94$ (dd, $J=5.6 \mathrm{~Hz}$,
$3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.77(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.46-7.42(\mathrm{~m}, 1 \mathrm{H}), 7.27-7.25(\mathrm{~m}, 1 \mathrm{H})$, 7.10-7.04 (m, 2H), $3.80(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{14}$

$\mathbf{N - 1 - N a p h t h y l p h t h a l i m i d e ~ ( 1 v ) : ~ P r e p a r e d ~ a c c o r d i n g ~ t o ~ M e t h o d ~ A ~ s t a r t i n g ~ f r o m ~ 1 - ~}$ aminonaphthalene in $79 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.02-7.99$ (m, $3 \mathrm{H}), 7.95$ (d, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.82 (dd, $J=5.6 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.66-7.59 (m, 2H), 7.56$7.48(\mathrm{~m}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{15}$


N-8-Quinolyl-phthalimide (1w): Prepared according to Method A starting from 8aminoquinoline in $76 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.86$ (dd, $J=4.4$ $\mathrm{Hz}, 1.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.23(\mathrm{dd}, J=8.4 \mathrm{~Hz}, 1.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.02-7.96(\mathrm{~m}, 3 \mathrm{H}), 7.81(\mathrm{dd}, J=5.2$ $\mathrm{Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.76(\mathrm{dd}, J=7.6 \mathrm{~Hz}, 1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.68(\mathrm{dd}, J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.44$ (dd, $J=8.0 \mathrm{~Hz}, 4.4 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{16}$

$N$-Benzylphthalimide (1x): Prepared according to Method B starting from benzyl bromide in $63 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.84$ (dd, $J=5.6 \mathrm{~Hz}, 3.2$ $\mathrm{Hz}, 2 \mathrm{H}$ ), 7.70 (dd, $J=5.2 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.43 (d, $J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.34-7.27(\mathrm{~m}, 3 \mathrm{H})$, $4.85(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{17}$

$\boldsymbol{N}$-[p-(methyl)benzyl]phthalimide (1y): Prepared according to Method B starting from p-methylbenzyl bromide in $86 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.83$ (dd, $J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.69(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.33(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H})$, $7.12(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 4.81(\mathrm{~s}, 2 \mathrm{H}), 2.30(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{3}$

$\boldsymbol{N}$-[p-(cyano)benzyl]phthalimide (1z): Prepared according to Method B starting from p-cyanobenzyl bromide in $86 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta=7.86$ (dd, $J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.73(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.61(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.52$ (d, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.88(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{14}$

$\boldsymbol{N}$-[p-(trifluoromethyl)benzyl]phthalimide (1aa): Prepared according to Method B starting from $p$-(triflouromethyl)benzyl bromide in $77 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.85(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.72(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H})$, 7.58-7.52 (m, 4H), $4.89(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{3}$


2-Phthalimidomethylpyridine (1ab): Prepared according to Method A starting from 2picolylamine in $73 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.53-8.51(\mathrm{~m}, 1 \mathrm{H})$, 7.88 (dd, $J=5.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.73 (dd, $J=5.6 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.65-7.61$ (m, 1H), 7.27 (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.15(\mathrm{dd}, J=7.6 \mathrm{~Hz}, 4.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.01(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{18}$


2-Phthalimidomethylthiophene (1ac): Prepared according to Method A starting from 2-thiophenemethylamine in $47 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.84$ (dd, $J=5.2 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.69 (dd, $J=5.2 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.20(\mathrm{dd}, J=5.2 \mathrm{~Hz}, 1.2$ $\mathrm{Hz}, 1 \mathrm{H}), 7.14(\mathrm{dd}, J=3.2 \mathrm{~Hz}, 0.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.92$ (dd, $J=5.2 \mathrm{~Hz}, 3.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.01$ (s, 2H) ppm . The spectral data match those previously reported. ${ }^{19}$


N,N-Diphthaloyl-1,3-propanediamine (1ad): Prepared according to Method C starting from 1,3-diaminopropane in $86 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.83$ $7.79(\mathrm{~m}, 4 \mathrm{H}), 7.71-7.67(\mathrm{~m}, 4 \mathrm{H}), 3.75(\mathrm{t}, J=7.2 \mathrm{~Hz}, 4 \mathrm{H}), 2.12-2.05(\mathrm{~m}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{20}$

$\boldsymbol{N}$-phenyl-m-fluorophthalimide (1ae): Prepared according to Method A starting from $m$-fluorophthalic anhydride in $91 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.82$ 7.77 (m, 2H), 7.53-7.49 (m, 2H), 7.47-7.40 (m, 4H) ppm. ${ }^{19} \mathrm{~F}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( 376 MHz , $\mathrm{CDCl}_{3}$ ): $\delta=-112.3 \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{21}$

$\boldsymbol{N}$-phenyl-m-chlorophthalimide (1af): Prepared according to Method A starting from $m$-chlorophthalic anhydride in $94 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.87$ (dd, $J=4.8 \mathrm{~Hz}, 3.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.71-7.70 (m, 2H), 7.53-7.48 (m, 2H), 7.45-7.41 (m, 3H) ppm . The spectral data match those previously reported. ${ }^{22}$

$\boldsymbol{N}$-phenyl- $\boldsymbol{m}$-nitrophthalimide (1ag): Prepared according to Method A starting from $m$ nitrophthalic anhydride in $78 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\boldsymbol{\delta}=8.21$ (dd, $J=7.6 \mathrm{~Hz}, 0.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.15(\mathrm{dd}, J=8.0 \mathrm{~Hz}, 0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.98(\mathrm{dd}, J=8.0 \mathrm{~Hz}, 8.0 \mathrm{~Hz}$, $1 \mathrm{H}), 7.54-7.50(\mathrm{~m}, 2 \mathrm{H}), 7.46-7.42(\mathrm{~m}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{23}$

$\boldsymbol{N}$-phenyl-m-methylphthalimide (1ah): Prepared according to Method A starting from $m$-methylphthalic anhydride in $95 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.78$
(d, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.63(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.54-7.49(\mathrm{~m}, 3 \mathrm{H}), 7.45-7.38(\mathrm{~m}, 3 \mathrm{H}), 2.75$ $(\mathrm{s}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{24}$

$\boldsymbol{N}$-phenyl-p-fluorophthalimide (1ai): Prepared according to Method A starting from $\boldsymbol{p}$ fluorophthalic anhydride in $89 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.96$ (dd, $J=8.4 \mathrm{~Hz}, 4.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.62 (dd, $J=7.2 \mathrm{~Hz}, 2.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.54-7.47 (m, 2H), 7.46$7.39(\mathrm{~m}, 4 \mathrm{H}) \mathrm{ppm} .{ }^{19} \mathrm{~F}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=-101.1 \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{21}$


1aj
$\boldsymbol{N}$-phenyl- $\boldsymbol{p}$-nitrophthalimide (1aj): Prepared according to Method A starting from $\boldsymbol{p}$ nitrophthalic anhydride in $98 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=8.77$ (dd, $J=6.0,0.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 8.67 (dd, $J=8.0 \mathrm{~Hz}, 2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.16$ (dd, $J=8.0 \mathrm{~Hz}, 0.8 \mathrm{~Hz}, 1 \mathrm{H})$, 7.56-7.52 (m, 2H), 7.48-7.43 (m, 3H) ppm. The spectral data match those previously reported. ${ }^{15}$

$\boldsymbol{N}$-phenyl- $\boldsymbol{p}$-methylphthalimide (1ak): Prepared according to Method A starting from p-methylphthalic anhydride in $97 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.83$ $(\mathrm{d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.75(\mathrm{dd}, J=0.8 \mathrm{~Hz}, 0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.59-7.56(\mathrm{~m}, 1 \mathrm{H}), 7.52-7.48(\mathrm{~m}$, $2 \mathrm{H}), 7.45-7.37(\mathrm{~m}, 3 \mathrm{H}), 2.55(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{19}$


1al
$N$-Phenylphthalamic acid (1al): Prepared according to a literature report starting from phthalic anhydride and aniline in quantitative yield. ${ }^{25}{ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ): $\delta$ $=9.15$ (br. s, 2H), 7.71-7.67 (m, 2H), 7.60-7.56 (m, 2H), 7.02 (dd, $J=7.3 \mathrm{~Hz}, 8.4 \mathrm{~Hz}$, $2 \mathrm{H}), 6.59(\mathrm{dd}, J=8.4 \mathrm{~Hz}, 1.0 \mathrm{~Hz}, 2 \mathrm{H}), 8.53(\mathrm{tt}, J=7.3 \mathrm{~Hz}, 1.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{26}$


1am
2-phenyl-1H-benzo[de]isoquinoline-1,3(2H)-dione (1am): Prepared according to Method A starting from 1,8-naphthalic anhydride in $78 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( 400 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.64$ (dd, $\left.J=7.2,1.2 \mathrm{~Hz}, 2 \mathrm{H}\right), 8.26$ (dd, $\left.J=8.4 \mathrm{~Hz}, 0.8 \mathrm{~Hz}, 2 \mathrm{H}\right), 7.78$ (dd, $J=8.0 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.59-7.54 (m, 2H), 7.51-7.47 (m, 1H), 7.34 (dd, $J=4.0 \mathrm{~Hz}$, $1.2 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{15}$

$\boldsymbol{N}$-( $\boldsymbol{p}$-Vinylbenzyl)phthalimide (1an): Prepared according to Method B starting from $p$-vinylbenzyl chloride in $77 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\boldsymbol{\delta}=7.83$ (dd, $J=5.6 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.68(\mathrm{dd}, J=5.6 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.40-7.34(\mathrm{~m}, 4 \mathrm{H}), 6.67$ (dd, $J=17.6 \mathrm{~Hz}, 11.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.70(\mathrm{dd}, J=17.6 \mathrm{~Hz}, 0.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.22(\mathrm{dd}, J=10.8 \mathrm{~Hz}, 0.8$ $\mathrm{Hz}, 1 \mathrm{H}), 4.82(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{27}$


I
3-Methyl-1-phenylmaleimide (I): Prepared according to Method A starting from methylmaleic anhydride in $77 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.48$ $7.43(\mathrm{~m}, 2 \mathrm{H}), 7.36-7.32(\mathrm{~m}, 3 \mathrm{H}), 6.46(\mathrm{q}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.15(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{28}$


II
$\boldsymbol{N}$-Phenylhomophthalimide (II): Prepared according to Method A starting from homophthalic anhydride in $95 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.25$ (dd, $J=8.0 \mathrm{~Hz}, 0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.67-7.63(\mathrm{~m}, 1 \mathrm{H}), 7.54-7.43(\mathrm{~m}, 4 \mathrm{H}), 7.35(\mathrm{~d}, J=7.2 \mathrm{~Hz}$, $1 \mathrm{H}), 7.23-7.20(\mathrm{~m}, 2 \mathrm{H}), 4.23(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{29}$


III
$\boldsymbol{N}$-Phenylphthalimidine (III): Prepared according to a literature report ${ }^{30}$ in $98 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.93(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.89-7.86(\mathrm{~m}, 2 \mathrm{H}), 7.62-$ 7.58 (m, 1H), 7.53-7.49 (m, 2H), 7.43 (dd, $J=8.4 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.18 (dd, $J=7.2 \mathrm{~Hz}$, $7.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.87(\mathrm{~s}, 2 \mathrm{H}) \mathrm{ppm}$. The spectral data match those previously reported. ${ }^{31}$


Hexahydro- $N$-phenylphthalimide (IV): Prepared according to Method D in $86 \%$ isolated yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.48-7.44$ (m, 2H), 7.39-7.35 (m, 1H), 7.30-7.27 (m, 2H), 3.06-3.00 (m, 2H), 1.95-1.85 (m, 4H), 1.53-1.50 (m, 4H) ppm. The spectral data match those previously reported. ${ }^{32}$

## 4. Reaction optimization.

4.1. General procedure: $\left[\mathrm{RuCl}_{2}(p \text {-cymene })\right]_{2}(0.004 \mathrm{mmol}, 2.5 \mathrm{mg}, 0.01 \mathrm{eq}$.$) , potassium$ carbonate ( $1.2 \mathrm{mmol}, 165.8 \mathrm{mg}, 3 \mathrm{eq}$.$) , distilled water ( 0.6 \mathrm{mmol}, 10.8 \mathrm{mg}, 10.8 \mu \mathrm{~L}, 1.5$ eq.), substrate 1 ( $0.4 \mathrm{mmol}, 1 \mathrm{eq}$. ) and $N$-methyl-2-pyrrolidone ( 2.0 mL ) were introduced in a flame-dried Schlenk tube under argon atmosphere. The reaction mixture was stirred at $150{ }^{\circ} \mathrm{C}$ during six hours. Then, the reaction mixture was cooled down to room temperature and diluted with water $(20 \mathrm{~mL})$. Then, $\mathrm{HCl}(1 \mathrm{M})$ was added until pH reached $c a .7$. The aqueous phase was extracted with ethyl acetate $(3 \times 20 \mathrm{~mL})$ and the combined organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$, filtered, and concentrated in vacuo. After solvents evaporation, the desired product 2 was purified by column chromatography with a mixture of petroleum ether and ethyl acetate as the eluent.

### 4.2. Screening of reaction conditions (Table S1).


[a] Reaction conditions: 1a ( 0.2 mmol ), catalysts ( $1 \mathrm{~mol} \%$ ), Base ( $300 \mathrm{~mol} \%$ ) and Additive ( $150 \mathrm{~mol} \%$ ) are stirred in 1 mL of solvents for 6 h under Ar. [b] Determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy against an internal standard ( $1,3,5$-trimethoxybenzene). The isolated yield is shown in parentheses. [c] $\mathrm{RuCl}_{3} \bullet \mathrm{nH}_{2} \mathrm{O}$ as the catalyst. [d] $\mathrm{Ru}_{3}(\mathrm{CO})_{12}$ as the catalyst. [e] 2 equivalent $\mathrm{K}_{2} \mathrm{CO}_{3}$. [f] Reaction performed at $130^{\circ} \mathrm{C}$. [g] Undisitillted solvent. [h] Under air.

## 5. Scale-up experiment.



Fig. S1. Reaction studied for scale-up experiments.
$\left[\mathrm{RuCl}_{2} \text { ( } p \text {-cymene) }\right]_{2}(0.055 \mathrm{mmol}, 33.6 \mathrm{mg}, 0.01 \mathrm{eq}$.), potassium carbonate $(16.5 \mathrm{mmol}$, $2.28 \mathrm{~g}, 3 \mathrm{eq}$. ), distilled water ( $8.3 \mathrm{mmol}, 0.15 \mathrm{~g}, 0.15 \mathrm{~mL}, 1.5 \mathrm{eq}$.), substrate $\mathbf{1 i}(5.5 \mathrm{mmol}$, $1.23 \mathrm{~g}, 1 \mathrm{eq}$.$) and \mathrm{N}$-methyl-2-pyrrolidone ( 20 mL ) were introduced in a flame-dried Schlenk tube under argon atmosphere. The reaction mixture was stirred at $150^{\circ} \mathrm{C}$ during six hours. Then, the reaction mixture was cooled down to room temperature and water ( 200 mL ) was added. Then, $\mathrm{HCl}(1 \mathrm{M})$ was added until pH reached ca. 7. The aqueous layer was extracted with ethyl acetate and the combined organic phases were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered, and concentrated in vacuo. After solvents evaporation, the desired product $\mathbf{2 i}$ was obtained in $97 \%$ yield ( 1.05 g ) by column chromatography with a mixture of petroleum ether and ethyl acetate as the eluent.

## 6. Mechanistic investigations.

6.1. GC-Gas analysis of the reaction mixture. $\left[\mathrm{RuCl}_{2}(p \text {-cymene })\right]_{2}(0.01 \mathrm{mmol}, 6.1 \mathrm{mg}$, 0.01 eq.), potassium carbonate ( $3 \mathrm{mmol}, 414.6 \mathrm{mg}, 3$ eq.), distilled water ( $1.5 \mathrm{mmol}, 27$ $\mathrm{mg}, 27 \mu \mathrm{~L}, 1.5 \mathrm{eq}$.), substrate $\mathbf{1 a}$ ( $1 \mathrm{mmol}, 161 \mathrm{mg}, 1 \mathrm{eq}$. ) and $N$-methyl-2-pyrrolidone ( 5 mL ) were introduced in a flame-dried Schlenk tube under argon atmosphere. The reaction mixture was stirred at $150{ }^{\circ} \mathrm{C}$ during six hours. Then, the reaction mixture was cooled down to room temperature and analysed by GC-gas analysis (see details below) indicating the major presence of $\mathrm{H}_{2}$ and $\mathrm{CO}_{2}$ besides traces of $\mathrm{CH}_{4}$ and CO . Air $\left(\mathrm{O}_{2}\right.$ and $\left.\mathrm{N}_{2}\right)$ was observed because the analysis could not be done under completely argon atmosphere. TLC and ${ }^{1} \mathrm{H}$ NMR spectroscopy analysis of the reaction mixture indicated the full conversion of 1a into 2a.

The same study was performed in a reaction lacking the substrate 1a.

Analysis method: Gas phase chromatography apparatus $\mu \mathrm{GC} 3000$ SRA.
Column chromatography: Molecular sieves 5A-30m.
Oven temperature: $100^{\circ} \mathrm{C}$.
Vector gas: Helium.
Detector: Cathetometer.


Sample S - PoraplotQ-H2:2830; CH4 :192; CO2 : 4201


Sample S - Molecularsieve5A - H2: 2939; O2: 23321; N2:160013; CO: 227
Fig. S2. GC-Gas spectrum (using two columns) of the reaction mixture.

Sample NOS-PoraplotQ-H2 : 3864; CH4: 98; CO2: 73


Fig. S3. GC-Gas spectrum (using two columns) of a reaction performed without substrate $1 a$.

Atmospheric air - PoraplotQ column - CO2 400ppm - CO2 : 39,42


Atmospheric air - MolecularSieve 5A column -


HE : 5,61; Ne : 9,82; O2:55638; N2: 174998
Fig. S4. GC-Gas spectrum (using two columns) of a blank analysis.
6.2. Deuteration experiments. $\left[\mathrm{RuCl}_{2}(p \text {-cymene })\right]_{2}(0.001 \mathrm{mmol}, 0.6 \mathrm{mg}, 0.01 \mathrm{eq}$.$) ,$ potassium carbonate ( $0.3 \mathrm{mmol}, 41.5 \mathrm{mg}, 3 \mathrm{eq}$.$) , substrate 1 \mathrm{a}$ ( $0.1 \mathrm{mmol}, 16 \mathrm{mg}, 1 \mathrm{eq}$.), $N$-methyl-2-pyrrolidone $(0.9 \mathrm{~mL})$ and $\mathrm{D}_{2} \mathrm{O}(0.1 \mathrm{~mL})$ were introduced in a flame-dried Schlenk tube under argon atmosphere. The reaction mixture was stirred at $150^{\circ} \mathrm{C}$ during six hours. Then, the reaction mixture was cooled down to room temperature and water $(10 \mathrm{~mL})$ was added. Then, $\mathrm{HCl}(1 \mathrm{M})$ was added until pH reached $c a$. 7. The aqueous phase was extracted with ethyl acetate ( $3 \times 10 \mathrm{~mL}$ ) and the combined organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$, filtered, and concentrated in vacuo. After solvents evaporation, the reaction mixture was analysed without any further purification by ${ }^{1} \mathrm{H}$ NMR spectroscopy (using 1,3,5-trimethoxybenzene as internal standard) indicating the exclusive presence of product 2a- $d$ in $63 \%$ yield (see spectra below).




Fig. S5. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ spectrum of $\mathbf{2 a}$ in a reaction without using $\mathrm{D}_{2} \mathrm{O}$.


Fig. S6. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ spectrum of $\mathbf{2 a}-d$ showing the disappearance of the peak at 7.77-7.74 ppm.
6.3. Hydrogenation experiments. $\left[\mathrm{RuCl}_{2} \text { ( } p \text {-cymene) }\right]_{2}(0.002 \mathrm{mmol}, 1.2 \mathrm{mg}, 0.01 \mathrm{eq}$.$) ,$ potassium carbonate ( $0.6 \mathrm{mmol}, 83 \mathrm{mg}, 3 \mathrm{eq}$. ), distilled water ( $0.3 \mathrm{mmol}, 5.4 \mathrm{mg}, 5.4 \mu \mathrm{~L}$, 1.5 eq.), substrate $\mathbf{1 a}(0.2 \mathrm{mmol}, 32 \mathrm{mg}, 1 \mathrm{eq}$.) and $N$-methyl-2-pyrrolidone ( 1 mL ) were introduced in a flame-dried Schlenk tube under argon atmosphere. Then, the reaction mixture was flushed with vaccum $/ \mathrm{H}_{2}$ over 3 cycles. The Schlenk tube was connected to a balloon filled with $\mathrm{H}_{2}$ and the reaction mixture was stirred at $150{ }^{\circ} \mathrm{C}$ during six hours. Then, the reaction mixture was cooled down to room temperature and water ( 10 mL ) was added. Then $\mathrm{HCl}(1 \mathrm{M})$ was added until pH reached $c a$. 7. The aqueous phase was extracted with ethyl acetate ( $3 \times 10 \mathrm{~mL}$ ) and the combined organic phases were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered, and concentrated in vacuo. After solvents evaporation, ${ }^{1} \mathrm{H}$ NMR spectroscopy analysis (using 1,3,5-trimethoxybenzene as internal standard) indicated $<10 \%$ formation of 2a and $>90 \%$ presence of $\mathbf{1 a}$.
6.4. TEMPO experiments. $\left[\operatorname{RuCl}_{2}(p-c y m e n e)\right]_{2}(0.002 \mathrm{mmol}, 1.2 \mathrm{mg}, 0.01 \mathrm{eq}$.$) ,$ potassium carbonate ( $0.6 \mathrm{mmol}, 83 \mathrm{mg}, 3 \mathrm{eq}$. ), distilled water ( $0.3 \mathrm{mmol}, 5.4 \mathrm{mg}, 5.4 \mu \mathrm{~L}$, 1.5 eq.), substrate $\mathbf{1 a}(0.2 \mathrm{mmol}, 32 \mathrm{mg}, 1 \mathrm{eq}$.$) , TEMPO ( 0.2 \mathrm{mmol}, 31 \mathrm{mg}, 1 \mathrm{eq}$.$) and N$ -methyl-2-pyrrolidone ( 1 mL ) were introduced in a flame-dried Schlenk tube under argon atmosphere and the reaction mixture was stirred at $150{ }^{\circ} \mathrm{C}$ during six hours. Then, the reaction mixture was cooled down to room temperature and analysed by GC-MS indicating the presence of $[\mathbf{1 a}+(2 \times$ TEMPO $)]$ at $m / z=474$ besides the main presence of 1a (see spectra below). The reaction mixture was further diluted with water $(10 \mathrm{~mL})$ and $\mathrm{HCl}(1 \mathrm{M})$ was added until pH reached $c a .7$. Then, the aqueous phase was extracted with ethyl acetate ( $3 \times 10 \mathrm{~mL}$ ) and the combined organic phases were dried over anhydrous $\mathrm{MgSO}_{4}$, filtered, and concentrated in vacuo. After solvents evaporation, the desired product $\mathbf{2 a}$ ( $24 \%$ isolated yield) was purified by column chromatography with a mixture of petroleum ether and ethyl acetate as the eluent.

GC-MS results:
Sample Infommation
Analyzed
Sample Type
Level \#
Sample Name
Sample ID
$:$

Data File
Org Data File
Method File
Org Method File
Report File
Tuning File
Modified by
Modified

17/08/2017 12:00:52
: Unlnown
$\therefore 1$
YYC-2-10-WUP
YYC-2-10-WUP

| Data File | : D: laarafalRGyyelYYC-2-10-WUP.ogd |
| :---: | :---: |
| Org Data File | : D: laarafaRGyyel YYC-2-10-WUP.ogd |
| Method File | : D:Methode-STd-40min qgm |
| Org Method File | : D:Methode-STd-40minggm |
| Report File |  |
| Tuning File | : C:IGCMSsolution'Systeml Tunel 12016-09-22.qgt |
| Modified by | : Admin |
| Modified | : 17/08/2017 12-40:52 |



Line\#:1 R Time:9.1(Scan\#:617)
MassPeaks:816
RawMode:Single 9.1(617) BasePeak:161(157774)
BGMode:None Group 1 - Event 1


Fig. S7. GC-MS peak belonging to starting material 1a.

Line\#: 2 R.Time:18.4(Scan\#\#2004)
MassPeaks: 830
RawMode:Single 18.4(2004) BasePeak:474(28842)
BGMode:None Group 1 - Event 1


Fig. S8. GC-MS peak belonging to [1a $+(2 \times$ TEMPO $)]$.
Line\#3 R Time:23.1(Scan\#:2718)
MassPeaks:795
RawMode:Single 23.1(2718) BacePe
k:474(25700)


Fig. S9. GC-MS peak belonging to [1a $+(2 \times$ TEMPO $)]$.

### 6.5. Mercury tests.

Test A: $\left[\operatorname{RuCl}_{2}(p \text {-cymene })\right]_{2}(0.002 \mathrm{mmol}, 1.2 \mathrm{mg}, 0.01 \mathrm{eq}$.), potassium carbonate ( 0.6 mmol, 83 mg , 3 eq.), distilled water ( $0.3 \mathrm{mmol}, 5.4 \mathrm{mg}, 5.4 \mu \mathrm{~L}, 1.5 \mathrm{eq}$. ), substrate $\mathbf{1 a}(0.2$ mmol, $32 \mathrm{mg}, 1 \mathrm{eq}$. ), 1 drop of mercury and $N$-methyl-2-pyrrolidone ( 1 mL ) were introduced in a flame-dried Schlenk tube under argon atmosphere. The reaction mixture was stirred at $150^{\circ} \mathrm{C}$ during six hours. Then, the reaction mixture was cooled down to room temperature and water ( 10 mL ) was added. Then $\mathrm{HCl}(1 \mathrm{M})$ was added to the mixture until pH reached $c a$. 7. The aqueous phase was extracted with ethyl acetate ( 3 x 10 mL ) and the combined organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$, filtered, and concentrated in vacuo. After solvents evaporation, the desired product 2a was obtained in $92 \%$ isolated yield by column chromatography with a mixture of petroleum ether and ethyl acetate as the eluent.

Test B: $\left[\mathrm{RuCl}_{2}(p \text {-cymene })\right]_{2}(0.002 \mathrm{mmol}, 1.2 \mathrm{mg}, 0.01 \mathrm{eq}$.), potassium carbonate ( 0.6 $\mathrm{mmol}, 83 \mathrm{mg}, 3 \mathrm{eq}$. ), distilled water ( $0.3 \mathrm{mmol}, 5.4 \mathrm{mg}, 5.4 \mu \mathrm{~L}, 1.5 \mathrm{eq}$.$) , substrate \mathbf{1 a}(0.2$ mmol, $32 \mathrm{mg}, 1 \mathrm{eq}$. ), and $N$-methyl-2-pyrrolidone ( 1 mL ) were introduced in a flamedried Schlenk tube under argon atmosphere. The reaction mixture was stirred at $150^{\circ} \mathrm{C}$ during during one hour. Then, at $150^{\circ} \mathrm{C}$ and under argon atmosphere, a drop of mercury was added to the reaction mixture, which was further stirred at $150^{\circ} \mathrm{C}$ during five hours (total reaction time was six hours). Then, the reaction mixture was cooled down to room temperature and water ( 10 mL ) was added. Then $\mathrm{HCl}(1 \mathrm{M})$ was added to the mixture until pH reached $c a$. 7 . The aqueous phase was extracted with ethyl acetate ( 3 x 10 mL ) and the combined organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$, filtered, and concentrated in vacuo. After solvents evaporation, the desired product 2a was obtained in $88 \%$ isolated yield by column chromatography with a mixture of petroleum ether and ethyl acetate as the eluent.

### 6.6. Study on a plausible intermediate (1al).


$\left[\operatorname{RuCl}_{2} \text { ( } p \text {-cymene) }\right]_{2}(0.002 \mathrm{mmol}, 1.2 \mathrm{mg}, 0.01 \mathrm{eq}$.), potassium carbonate $(0.6 \mathrm{mmol}, 83$ $\mathrm{mg}, 3 \mathrm{eq}$.$) , distilled water ( 0.3 \mathrm{mmol}, 5.4 \mathrm{mg}, 5.4 \mu \mathrm{~L}, 1.5 \mathrm{eq}$.$) , substrate \mathbf{1 a l}(0.2 \mathrm{mmol}$, $48.2 \mathrm{mg}, 1 \mathrm{eq}$.$) and N$-methyl-2-pyrrolidone ( 1.0 mL ) were introduced in a flame-dried Schlenk tube under argon atmosphere. The reaction mixture was stirred at $150^{\circ} \mathrm{C}$ during six hours. Then, the reaction mixture was cooled down to room temperature and diluted with water $(10 \mathrm{~mL})$. Then, $\mathrm{HCl}(1 \mathrm{M})$ was added until pH reached $c a$. 7. The aqueous phase was extracted with ethyl acetate ( $3 \times 10 \mathrm{~mL}$ ) and the combined organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$, filtered, and concentrated in vacuo. After solvents evaporation, the crude mixture was analysed by ${ }^{1} \mathrm{H}$ NMR spectroscopy indicating the exclusive formation of benzoic acid together with aniline (see spectrum below).


Fig. S10. 1H NMR spectrum of the reaction mixture after catalysis using substrate 1al.

### 6.7. Study of the size of the phthalimide ring (1am).


$\left[\operatorname{RuCl}_{2} \text { ( } p \text {-cymene) }\right]_{2}(0.004 \mathrm{mmol}, 2.5 \mathrm{mg}, 0.01 \mathrm{eq}$.$) , potassium carbonate (1.2 \mathrm{mmol}$, $165.8 \mathrm{mg}, 3$ eq.), distilled water ( $0.6 \mathrm{mmol}, 10.8 \mathrm{mg}, 10.8 \mu \mathrm{~L}, 1.5 \mathrm{eq}$.), substrate 1 am ( $0.4 \mathrm{mmol}, 109.3 \mathrm{mg}, 1 \mathrm{eq}$.) and N -methyl-2-pyrrolidone ( 2.0 mL ) were introduced in a flame-dried Schlenk tube under argon atmosphere. The reaction mixture was stirred at $150^{\circ} \mathrm{C}$ during six hours. Then, the reaction mixture was cooled down to room temperature and diluted with water $(20 \mathrm{~mL})$. Then, $\mathrm{HCl}(1 \mathrm{M})$ was added until pH reached $c a .7$. The aqueous phase was extracted with ethyl acetate ( $3 \times 20 \mathrm{~mL}$ ) and the combined organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$, filtered, and concentrated in vacuo. After solvents evaporation, the crude mixture was analysed by ${ }^{1} \mathrm{H}$ NMR spectroscopy indicating the exclusive presence of starting material 1am.

### 6.8. Indirect evidence of hydrogen formation in the catalysis.


$\left[\operatorname{RuCl}_{2}(p \text {-cymene })\right]_{2}(0.004 \mathrm{mmol}, 2.5 \mathrm{mg}, 0.01 \mathrm{eq}$.$) , potassium carbonate (1.2 \mathrm{mmol}$, $165.8 \mathrm{mg}, 3$ eq.), distilled water ( $0.6 \mathrm{mmol}, 10.8 \mathrm{mg}, 10.8 \mu \mathrm{~L}, 1.5 \mathrm{eq}$. ), substrate $1 \mathrm{an}(0.4$ mmol, $105.3 \mathrm{mg}, 1 \mathrm{eq}$.) and $N$-methyl-2-pyrrolidone ( 2.0 mL ) were introduced in a flamedried Schlenk tube under argon atmosphere. The reaction mixture was stirred at $150^{\circ} \mathrm{C}$ during six hours. Then, the reaction mixture was cooled down to room temperature and diluted with water ( 20 mL ). Then, $\mathrm{HCl}(1 \mathrm{M})$ was added until pH reached $c a .7$. The aqueous phase was extracted with ethyl acetate ( $3 \times 20 \mathrm{~mL}$ ) and the combined organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$, filtered, and concentrated in vacuo. After solvents evaporation, the crude mixture was analysed by GC-MS indicating the presence of the starting material 1am and its hydrogenated version at $m / z=265$ (see GC-MS spectrum below).

| Data File | :D:laarafalRGyyclYYC-2-28-RE-2.qgd |
| :--- | :--- |
| Org Data File | :D:larafaRGyyclYYC-2-28-RE-2.qgd |
| Method File | :D:Methode-STd-40minggm |
| Org Method File | :D:Methode-STd-40min.qgm |
| Report File | :C:GCMSsolutionSystemTune12016-09-22.qgt |
| Tuning File | :Admin |
| Modified by | :14/09/2017 19:14:42 |
| Modified |  |




Fig. S11. GC-MS peak belonging to the hydrogenated starting material.

## 7. Characterization data of products 2-3.



2a
$N$-Methylbenzamide (2a): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $5: 1$ to $2: 1, v / v$ ) in $93 \%$ yield $(50.3 \mathrm{mg})$ as a colourless solid using petroleum ether. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.77-7.74(\mathrm{~m}, 2 \mathrm{H}), 7.49-7.45(\mathrm{~m}, 1 \mathrm{H})$, 7.42-7.38 (m, 2H), 6.42 (br. s, 1H), 2.98 (d, $J=4.8 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( 100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=168.3,134.5,131.2,128.3,126.8,26.7 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=8.7 \mathrm{~min} ; \mathrm{MS}$ $(E I): m / z=134\left(\mathrm{M}^{+}, 48\right), 105(100), 77(91), 51(34)$. The spectral data match those previously reported. ${ }^{33}$

$N$-Butylbenzamide (2b): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $88 \%$ yield ( 62.4 mg ) as a yellow oil. ${ }^{1} \mathrm{H}$ NMR ( 400 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.76$ (d, $\left.J=7.6 \mathrm{~Hz}, 2 \mathrm{H}\right), 7.46(\mathrm{dd}, J=7.2 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.39$ (dd, $J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.36$ (br. s, 1H), 3.43 (td, $J=6.8 \mathrm{~Hz}, 6.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 1.62-1.54 (m, 2 H ), 1.43-1.34 (m, 2H), $0.93(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta=167.5,134.7,131.0,128.2,126.8,39.7,31.6,20.0,13.6 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=9.9 \mathrm{~min} ; \mathrm{MS}$ $(E I): m / z=177\left(\mathrm{M}^{+}, 8\right), 105(100), 77(41)$. The spectral data match those previously reported. ${ }^{34}$

$N$-Isopropylbenzamide (2c): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to 2:1, $v / v$ ) in $87 \%$ yield ( 56.8 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=7.76-7.73$ (m, 2H), 7.50-7.46 (m, 1H), 7.44-7.39 (m, 2H), 5.94
(br. s, 1H), 4.33-4.25(m, 1H), $1.26(\mathrm{~d}, J=6.4 \mathrm{~Hz}, 6 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ): $\delta=166.7,135.0,131.2,128.5,126.8,41.9,22.8 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=8.7 \mathrm{~min} ; \mathrm{MS}$ $(E I): m / z=163\left(\mathrm{M}^{+}, 25\right), 105(100), 77(39)$. The spectral data match those previously reported. ${ }^{35}$

$N$-(p-Cyanobutyl)benzamide (2e): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to $2: 1, v / v$ ) in $91 \%$ yield ( 73.6 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\boldsymbol{\delta}=7.75$ (dd, $J=8.0 \mathrm{~Hz}, 1.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.48-7.43 (m, 1H), 7.39-7.35 (m, 2H), 6.80 (br. s, 1H), 3.45-3.40 (m, 2H), 2.35 (t, $J=6.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 1.76$1.63(\mathrm{~m}, 4 \mathrm{H})$ ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=167.7,134.2,131.4,128.4$, 126.8, 119.5, 38.6, 28.6, 22.6, 16.6 ppm. HRMS (ESI) calcd. for [M+Na] ${ }^{+} \mathrm{C}_{12} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{ONa}$ 225.09983 , found $225.0996(1 \mathrm{ppm})$. The spectral data match those previously reported. ${ }^{36}$

$N$-(2-Methoxyethyl)benzamide (2f): Isolated by column chromatography $\left(\mathrm{SiO}_{2}\right.$, petroleum ether/ethyl acetate, 10:1 to $2: 1, v / v$ ) in $98 \%$ yield ( 70.3 mg ) as a colourless oil. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.76-7.74(\mathrm{~m}, 2 \mathrm{H}), 7.46-7.42(\mathrm{~m}, 1 \mathrm{H}), 7.39-7.34(\mathrm{~m}$, 2 H ), 6.76 (br. s, 1H), 3.62-3.58 (m, 2H), 3.52-3.50 (m, 2H), 3.33 (s, 3H) ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=167.4,134.4,131.2,128.3,126.8,71.0,58.6,39.5 \mathrm{ppm}$. HRMS (ESI) calcd. for [M+Na] ${ }^{+} \mathrm{C}_{10} \mathrm{H}_{13} \mathrm{NO}_{2} \mathrm{Na} 202.08385$, found 202.0837 ( 1 ppm ). The spectral data match those previously reported. ${ }^{37}$


2g
$\boldsymbol{N}$-(2-Oxopropyl)benzamide (2g): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $73 \%$ yield ( 51.2 mg ) as a yellow solid. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.80(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.50-7.47(\mathrm{~m}, 1 \mathrm{H}), 7.41(\mathrm{dd}, J=7.6 \mathrm{~Hz}$, $7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.04 (br. s, 1H), 4.31 (d, $J=4.4 \mathrm{~Hz}, 2 \mathrm{H}$ ), 2.23 (s, 3H) ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=203.0,167.2,133.6,131.7,128.5,127.0,50.2,27.3 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}$ $=11.2 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=177\left(\mathrm{M}^{+}, 10\right), 135(45), 105(100), 77(55)$. The spectral data match those previously reported. ${ }^{38}$


Ethyl benzamidoacetate (2h): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $50 \%$ yield $(41.4 \mathrm{mg})$ as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.82-7.80(\mathrm{~m}, 2 \mathrm{H}), 7.54-7.49(\mathrm{~m}, 1 \mathrm{H}), 7.46-7.42(\mathrm{~m}, 2 \mathrm{H}), 6.69$ (br. s, 1H), 4.29-4.23 (m, 4H), $1.31(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta=170.1,167.4,133.7,131.8,128.6,127.0,61.6,41.9,14.1 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=10.9$ $\mathrm{min} ; \mathrm{MS}(\mathrm{EI}): m / z=207\left(\mathrm{M}^{+}, 7\right), 105(100), 77(35)$. The spectral data match those previously reported. ${ }^{39}$


2i
$N$-Phenylbenzamide (2i): Isolated by column chromatography $\left(\mathrm{SiO}_{2}\right.$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $98 \%$ yield ( 77.1 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.89-7.85(\mathrm{~m}, 3 \mathrm{H}), 7.65(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.57-7.52(\mathrm{~m}, 1 \mathrm{H})$, 7.49-7.45 (m, 2H), 7.37 (dd, $J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.18-7.13 (m, 1H) ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=165.8,137.9,135.0,131.8,129.1,128.8,127.0,124.6$, $120.2 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=12.0 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=197\left(\mathrm{M}^{+}, 28\right), 105(100), 77$ (55). The spectral data match those previously reported. ${ }^{40}$


2j
$\boldsymbol{N}$-(p-Tolyl)benzamide (2j): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $96 \%$ yield ( 81.4 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.86$ (d, $J=7.2 \mathrm{~Hz}, 3 \mathrm{H}$ ), 7.52 (d, $J=8.4 \mathrm{~Hz}, 3 \mathrm{H}$ ), 7.48-7.44 (m, 2 H ), 7.16 (d, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), $2.34(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=$ $165.6,135.3,135.1,134.2,131.7,129.5,128.7,127.0,120.3,20.9 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=12.7$ $\min ; \mathrm{MS}(\mathrm{EI}): m / z=211\left(\mathrm{M}^{+}, 20\right), 105(100), 77(79)$. The spectral data match those previously reported. ${ }^{41}$

$\boldsymbol{N}$-( $\boldsymbol{p}$-Methoxyphenyl)benzamide (2k): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $96 \%$ yield ( 87.1 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.86$ (d, $J=7.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.80 (br. s, 1H), 7.557.52 (m, 3H), 7.47 (dd, $J=7.2 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 2 \mathrm{H}), 6.90(\mathrm{dd}, J=6.8 \mathrm{~Hz}, 2.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.81$ (s, 3H) ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=165.6,156.6,135.0,131.7,131.0$, $128.7,127.0,122.1,114.2,55.5 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=13.6 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=227\left(\mathrm{M}^{+}, 21\right)$, 105 (100), 77 (59). The spectral data match those previously reported. ${ }^{41}$

$\boldsymbol{N}$-( $\boldsymbol{p}$-Fluorophenyl)benzamide (21): Isolated by column chromatography $\left(\mathrm{SiO}_{2}\right.$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $96 \%$ yield ( 82.3 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.79$ (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.72 (br. s, 1H), 7.557.51 (m, 2H), 7.48 (d, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.44-7.40(\mathrm{~m}, 2 \mathrm{H}), 7.00(\mathrm{dd}, J=8.4 \mathrm{~Hz}, 8.4 \mathrm{~Hz}$, $2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( 100 MHz, DMSO- $d_{6}$ ): $\delta=165.5,158.3$ (d, $J_{C-F}=239.0 \mathrm{~Hz}$ ), $135.5\left(\mathrm{~d}, J_{C-F}=2.3 \mathrm{~Hz}\right), 134.8,131.6,128.4,127.6,122.2\left(\mathrm{~d}, J_{C-F}=7.7 \mathrm{~Hz}\right), 115.2\left(\mathrm{~d}, J_{C-}\right.$
$\left.{ }_{F}=22.2 \mathrm{~Hz}\right) \mathrm{ppm} .{ }^{19} \mathrm{~F}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=-117.6 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=12.0 \mathrm{~min}$; MS (EI): $m / z=215\left(\mathrm{M}^{+}, 12\right), 105(100), 77(80), 51(28)$. The spectral data match those previously reported. ${ }^{42}$


4-Benzamidoacetophenone ( $\mathbf{2 m}$ ): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/dichloromethane, $4: 1$ to $1: 2, v / v$ ) in $98 \%$ yield $(93.4 \mathrm{mg})$ as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\boldsymbol{\delta}=8.00(\mathrm{dd}, J=6.8 \mathrm{~Hz}, 2.0 \mathrm{~Hz}, 3 \mathrm{H}), 7.90-7.88(\mathrm{~m}, 2 \mathrm{H}), 7.77$ (dd, $J=6.8 \mathrm{~Hz}, 2.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.59(\mathrm{dd}, J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.53-7.49(\mathrm{~m}, 2 \mathrm{H}), 2.60$ (s, 3 H ) ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=196.9,165.7,142.2,134.5,133.2$, $132.3,129.8,128.9,127.1,119.2,26.4 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=15.6 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=239\left(\mathrm{M}^{+}\right.$, 8), 105 (100), 77 (69). The spectral data match those previously reported. ${ }^{43}$


2n
Ethyl $\boldsymbol{p}$-[(phenylcarbonyl)amino]benzoate (2n): Isolated by column chromatography $\left(\mathrm{SiO}_{2}\right.$, petroleum ether/ethyl acetate, $10: 1$ to $\left.2: 1, v / v\right)$ in $94 \%$ yield $(100.9 \mathrm{mg})$ as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.37$ (br. s, 1 H ), 8.01 (dd, $J=6.8 \mathrm{~Hz}$, $6.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.85(\mathrm{dd}, J=7.2 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.75$ (dd, $J=7.2 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.52$ (dd, $J=7.2 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.43$ (dd, $J=7.2 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 2 \mathrm{H}), 4.34(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H})$, $1.38(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=166.2,166.0,142.2$, $134.5,132.1,130.7,128.7,127.1,126.0,119.2,60.9,14.3 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=15.8 \mathrm{~min} ; \mathrm{MS}$ $(E I): m / z=269\left(\mathrm{M}^{+}, 19\right), 105(100), 77(38)$. The spectral data match those previously reported. ${ }^{44}$

$\boldsymbol{N}$-(p-Nitrophenyl)benzamide (20): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $80 \%$ yield ( 77.5 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.27$ (d, $J=9.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 8.05 (br. s, 1H), 7.87 $(\mathrm{dd}, J=8.4 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 4 \mathrm{H}), 7.62(\mathrm{dd}, J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.54(\mathrm{dd}, J=7.6 \mathrm{~Hz}, 7.6$ $\mathrm{Hz}, 2 \mathrm{H}$ ) ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( 100 MHz , acetone- $d_{6}$ ): $\delta=167.0,146.4,144.1,135.5$, $133.0,129.4,128.6,125.5,120.6 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=16.3 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=242\left(\mathrm{M}^{+}, 10\right)$, 105 (100), 77 (53). The spectral data match those previously reported. ${ }^{45}$


2p
$\boldsymbol{N}$-( $\boldsymbol{p}$-Bromophenyl)benzamide (2p): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) as a mixture of $\mathbf{2 p}: 2 \mathbf{2 i}$ in a ratio $74: 26$ according to ${ }^{1} \mathrm{H}$ NMR spectroscopy analysis. GC: $\mathrm{t}_{\mathrm{R}}=14.6 \mathrm{~min}$; MS (EI): $m / z=275\left(\mathrm{M}^{+}\right.$, 10), 105 (100), 77 (45), 51 (10). The spectral data of $\mathbf{2 p}$ match those previously reported. ${ }^{46}$


2q
$N$-(p-Iodophenyl)benzamide (2q): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) as a mixture of $\mathbf{2 q}: 2 \mathbf{i}$ in a ratio $24: 76$ according to ${ }^{1} \mathrm{H}$ NMR spectroscopy analysis. GC: $\mathrm{t}_{\mathrm{R}}=14.6 \mathrm{~min}$; MS (EI): $m / z=323\left(\mathrm{M}^{+}\right.$, 30), 105 (100), 77 (50), 51 (10). The spectral data of $\mathbf{2 q}$ match those previously reported. ${ }^{47}$

$\boldsymbol{N}$-( $\boldsymbol{m}$-Methoxyphenyl)benzamide (2r): Isolated by column chromatography $\left(\mathrm{SiO}_{2}\right.$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $97 \%$ yield ( 88.5 mg ) as a brown solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\boldsymbol{\delta}=8.22$ (br. s, 1H), 7.83 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.49 (dd, $J=$ $7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.43-7.38(\mathrm{~m}, 3 \mathrm{H}), 7.21(\mathrm{dd}, J=8.0 \mathrm{~Hz}, 8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.14(\mathrm{~d}, J=$ $7.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.69(\mathrm{dd}, J=8.0 \mathrm{~Hz}, 2.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.77(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( 100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=166.0,160.1,139.2,134.8,131.7,129.6,128.6,127.0,112.5,110.4$, $105.9,55.2 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=13.4 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): \mathrm{m} / \mathrm{z}=227\left(\mathrm{M}^{+}, 19\right), 105(100), 77(66)$. The spectral data match those previously reported. ${ }^{48}$


2 s
$N$-(o-Chlorophenyl)benzamide (2s): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $63 \%$ yield ( 58.3 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\boldsymbol{\delta}=8.48$ (dd, $J=8.4 \mathrm{~Hz}, 0.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 8.37 (br. s, 1H), 7.84 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.50$ (dd, $J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.45-7.41$ (m, 2H), 7.33 (dd, $J=8.0 \mathrm{~Hz}, 0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.25(\mathrm{dd}, J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.02-6.97(\mathrm{~m}, 1 \mathrm{H}) \mathrm{ppm}$. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=165.2,134.6,134.5,132.1,128.9,128.8,127.8$, $127.0,124.7,123.0,121.5 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=12.3 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=231\left(\mathrm{M}^{+}, 5\right), 196(27)$, 105 (100), 77 (77), 51 (27). The spectral data match those previously reported. ${ }^{49}$


N -o-Tolylbenzamide (2t): Isolated by column chromatography $\left(\mathrm{SiO}_{2}\right.$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $89 \%$ yield ( 75.1 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.93-7.87$ (m, 3H), 7.74 (br. s, 1 H ), 7.58-7.54 (m, 1H), 7.51-7.47 (m, 2H), 7.27-7.22 (m, 2H), 7.14-7.10(m, 1H), $2.33(\mathrm{~s}, 3 \mathrm{H})$ ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR (100
$\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=165.7,135.7,134.8,131.7,130.5,129.7,128.7,127.0,126.7,125.4$, $123.4,17.7 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=12.3 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=211\left(\mathrm{M}^{+}, 28\right), 105(100), 77(50)$. The spectral data match those previously reported. ${ }^{50}$


2u
$\boldsymbol{N}$-(o-Methoxyphenyl)benzamide (2u): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $78 \%$ yield ( 70.9 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.55$ (dd, $\left.J=7.6 \mathrm{~Hz}, 1.6 \mathrm{~Hz}, 2 \mathrm{H}\right), 7.90(\mathrm{dd}, J=$ $6.8 \mathrm{~Hz}, 1.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.57-7.48(\mathrm{~m}, 3 \mathrm{H}), 7.11-7.01(\mathrm{~m}, 2 \mathrm{H}), 6.93(\mathrm{dd}, J=8.0 \mathrm{~Hz}, 1.2 \mathrm{~Hz}$, $1 \mathrm{H}), 3.93$ (s, 3 H ) ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=165.2,148.1,135.3,131.6$, 128.7, 127.8, 127.0, 123.8, 121.2, 119.8, 109.9, $55.8 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=12.8 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}):$ $m / z=227\left(\mathrm{M}^{+}, 29\right), 105(100), 77(46)$. The spectral data match those previously reported. ${ }^{48}$

$\boldsymbol{N}$-(1-Naphthalenyl)benzamide (2v): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to 2:1, $v / v)$ in $92 \%$ yield ( 90.7 mg ) as a pink solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.33$ (br. s, 1H), 7.96-7.93 (m, 3 H ), 7.89-7.87 (m, 2 H ), 7.73 (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.57(\mathrm{dd}, J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.51-7.45(\mathrm{~m}, 5 \mathrm{H}) \mathrm{ppm}$. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=166.3,134.7,134.1,132.4,131.8,128.73,128.70$, $127.6,127.2,126.3,126.1,126.0,125.6,121.4,120.8 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=16.0 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}):$ $m / z=247\left(\mathrm{M}^{+}, 8\right), 105(100), 77(62)$. The spectral data match those previously reported. ${ }^{35}$


2w
$N-8$-Quinolinylbenzamide (2w): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $33 \%$ yield ( 32.7 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=10.75$ (br. s, 1H), 8.95 (dd, $J=7.6 \mathrm{~Hz}, 1.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), 8.85 (dd, $J=$ $4.0 \mathrm{~Hz}, 1.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.18(\mathrm{dd}, J=8.0 \mathrm{~Hz}, 1.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.09(\mathrm{dd}, J=8.0 \mathrm{~Hz}, 1.6 \mathrm{~Hz}, 2 \mathrm{H})$, 7.62-7.53 (m, 5H), 7.47 (dd, $J=8.4 \mathrm{~Hz}, 4.0 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ): $\delta=165.4,148.2,138.8,136.4,135.1,134.6,131.8,128.8$ (x 2), 128.0, 127.4, $127.3,121.6,116.5 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=15.6 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=248\left(\mathrm{M}^{+}, 15\right), 105(100), 77$ (72). The spectral data match those previously reported. ${ }^{18}$


2x
$N$-Benzylbenzamide (2x): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to 2:1, v/v) in 90\% yield ( 76.2 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.81-7.78$ (m, 2H), 7.52-7.48 (m, 1H), 7.44-7.40 (m, 2H), 7.57$7.46(\mathrm{~m}, 4 \mathrm{H}), 7.32-7.27(\mathrm{~m}, 1 \mathrm{H}), 6.52$ (br. s, 1H), $4.64(\mathrm{~d}, J=5.6 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=167.4,138.3,134.2,131.2,128.4,128.2,127.5,127.1$, 126.9, $43.7 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=12.5 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=211\left(\mathrm{M}^{+}, 47\right), 105(100), 77(57)$. The spectral data match those previously reported. ${ }^{51}$ Crystals suitable for X-ray diffraction studies were grown by slow diffusion of $n$-heptane into a concentrated solution of $\mathbf{2 x}$ in dichloromethane at room temperature.

$\boldsymbol{N}$ - $\boldsymbol{p}$-Methylbenzylbenzamide (2y): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $98 \%$ yield ( 88.4 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.68(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.36 (dd, $J=7.2 \mathrm{~Hz}, 7.2$ $\mathrm{Hz}, 1 \mathrm{H}), 7.28$ (dd, $J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.11(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.03(\mathrm{~d}, J=8.0 \mathrm{~Hz}$, 2 H ), 6.65 (br. s, 1H), 4.45 (d, $J=5.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.23(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR (100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=167.3,137.1,135.2,134.4,131.3,129.3,128.4,127.8,126.9,43.7$, $21.0 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=12.9 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=225\left(\mathrm{M}^{+}, 24\right)$, 105 (100), 77 (60). The spectral data match those previously reported. ${ }^{52}$

$2 z$
$N$-[(p-Cyanophenyl)methyl]benzamide (2z): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $90 \%$ yield $(58.2 \mathrm{mg}$ ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.79$ (d, $J=7.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.58 (dd, $J=$ $6.8 \mathrm{~Hz}, 1.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.54-7.49 (m, 1H), 7.44-7.40 (m, 4H), 6.89 (br. s, 1H), 4.66 (d, $J=$ $6.0 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=167.6,143.9$, 133.7, 132.4, $131.9,128.6,128.1,127.0,118.7,111.2,43.4 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=15.2 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): \mathrm{m} / \mathrm{z}=$ $236\left(\mathrm{M}^{+}, 18\right), 105$ (100), 77 (72), 51 (30). The spectral data match those previously reported. ${ }^{53}$

$N$-[[p-(Trifluoromethyl)phenyl]methyl]benzamide (3a): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to 2:1, $v / v$ ) in $99 \%$ yield (110.9 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.80-7.78(\mathrm{~m}, 2 \mathrm{H}), 7.56(\mathrm{~d}, J$ $=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.53-7.48(\mathrm{~m}, 1 \mathrm{H}), 7.43-7.39(\mathrm{~m}, 4 \mathrm{H}), 6.84$ (br. s, 1H), $4.65(\mathrm{~d}, J=6.0 \mathrm{~Hz}$, $2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=167.6,142.4,134.0,131.7,129.7\left(\mathrm{q}, J_{C}\right.$. $\left.{ }_{F}=32.9 \mathrm{~Hz}\right), 128.6,127.8,127.0,125.6\left(\mathrm{q}, J_{C-F}=3.8 \mathrm{~Hz}\right), 124.0\left(\mathrm{q}, J_{C-F}=270.4 \mathrm{~Hz}\right)$, $43.4 \mathrm{ppm} .{ }^{19} \mathrm{~F}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=-62.5 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=12.3 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}):$ $m / z=279\left(\mathrm{M}^{+}, 10\right), 105(100), 77(64)$. The spectral data match those previously reported. ${ }^{54}$

$\boldsymbol{N}$-o-Picolylbenzamide (3b): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $88 \%$ yield ( 74.7 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR
( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.54(\mathrm{~d}, J=4.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.88-7.85(\mathrm{~m}, 2 \mathrm{H}), 7.68-7.64(\mathrm{~m}, 2 \mathrm{H})$, $7.51-7.46(\mathrm{~m}, 1 \mathrm{H}), 7.44-7.40(\mathrm{~m}, 2 \mathrm{H}), 7.30(\mathrm{dd}, J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.19(\mathrm{dd}, J=7.2$ $\mathrm{Hz}, 1.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.74(\mathrm{~d}, J=4.8 \mathrm{~Hz}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=$ $167.3,156.2,148.9,136.8,134.3,131.4,128.5,127.0,122.4,122.1,44.7 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=$ $12.7 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=212\left(\mathrm{M}^{+}, 4\right), 107(100), 77(76), 51(44)$. The spectral data match those previously reported. ${ }^{55}$

$\boldsymbol{N}$-(2-Thienylmethyl)benzamide (3c): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $10: 1$ to $2: 1, v / v$ ) in $60 \%$ yield ( 52.3 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.79-7.77$ (m, 2H), 7.51-7.47 (m, 1H), 7.41 (dd, $J=7.2 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.23(\mathrm{dd}, J=5.2 \mathrm{~Hz}, 0.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.03(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.96$ (dd, $J=5.2 \mathrm{~Hz}, 3.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.62 (br. s, 1H), 4.80 (d, $J=5.6 \mathrm{~Hz}, 2 \mathrm{H}$ ) ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=167.1,140.8,134.1,131.6,128.5,127.0,126.9,126.2$, $125.3,38.8 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=12.6 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=217\left(\mathrm{M}^{+}, 42\right), 105(100), 77(44)$. The spectral data match those previously reported. ${ }^{56}$


3d
$N, N^{\prime}$-Trimethylenebis(benzamide) (3d): Isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to $2: 1, v / v$ ) in $93 \%$ yield ( 106.2 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.88-7.86$ (m, 4H), 7.52-7.48 (m, 2H), 7.45-7.41 (m, 4H), 7.29 (br. s, 2H), 3.58-3.53 (m, 4H), 1.84-1.78 (m, 2H) ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR (100 MHz, DMSO- $d_{6}$ ): $\delta=166.2,134.6,131.0,128.2,127.1,37.0,29.3 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=19.3$ $\min ; \mathrm{MS}(\mathrm{EI}): m / z=282\left(\mathrm{M}^{+}, 8\right), 134(27), 105(100), 77(46)$. The spectral data match those previously reported. ${ }^{57}$

$\boldsymbol{m}$-Fluoro- $\boldsymbol{N}$-phenylbenzamide (3e): Starting from 1ae and isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to 2:1, $v / v$ ) in 93\% yield (80.1 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.88$ (br. s, 1 H ), 7.64-7.56 $(\mathrm{m}, 4 \mathrm{H}), 7.48-7.43(\mathrm{~m}, 1 \mathrm{H}), 7.41-7.35(\mathrm{~m}, 2 \mathrm{H}), 7.27-7.24(\mathrm{~m}, 1 \mathrm{H}), 7.22-7.14(\mathrm{~m}, 1 \mathrm{H})$ ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=164.2\left(\mathrm{~d}, J_{C-F}=36.8 \mathrm{~Hz}\right), 161.6,137.6$, $137.2\left(\mathrm{~d}, J_{C-F}=6.9 \mathrm{~Hz}\right), 130.4\left(\mathrm{~d}, J_{C-F}=7.7 \mathrm{~Hz}\right), 129.1,124.8,122.4\left(\mathrm{~d}, J_{C-F}=3.0 \mathrm{~Hz}\right)$, $120.3,118.8\left(\mathrm{~d}, J_{C-F}=21.4 \mathrm{~Hz}\right), 114.5\left(\mathrm{~d}, J_{C-F}=23.0 \mathrm{~Hz}\right) \mathrm{ppm} .{ }^{19} \mathrm{~F}\left\{{ }^{1} \mathrm{H}\right\}$ NMR $(376 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta=-111.3 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=12.6 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=215\left(\mathrm{M}^{+}, 30\right), 123(100), 95$ (52), 75 (15). The spectral data match those previously reported. ${ }^{58}$

$\boldsymbol{m}$-Chloro- $\boldsymbol{N}$-phenylbenzamide (3f): Starting from 1af and isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to 2:1, $v / v$ ) in 65\% yield (59.9 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.93$ (br. s, 1 H ), $7.83(\mathrm{~s}, 1 \mathrm{H})$, 7.72 (d, $J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.62(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.50(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.41-7.34$ (m, 3 H ), 7.16 (dd, $J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=164.5$, 137.6, 136.7, 134.9, 131.8, 130.0, 129.1, 127.4, 125.1, 124.9, $120.4 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=13.6$ $\min ; \mathrm{MS}(\mathrm{EI}): m / z=231\left(\mathrm{M}^{+}, 15\right), 139(100), 111(48), 75(27)$. The spectral data match those previously reported. ${ }^{59}$

$\boldsymbol{m}$-Nitro- $N$-phenylbenzamide (3g): Starting from $\mathbf{1 a g}$ and isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to 2:1, $v / v$ ) in $87 \%$ yield ( 87.1 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.69(\mathrm{~s}, 1 \mathrm{H}), 8.39(\mathrm{~d}, J=8.0$
$\mathrm{Hz}, 1 \mathrm{H}), 8.25(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.01$ (br. s, 1H), 7.71-7.64 (m, 3H), 7.39 (dd, $J=7.6$ $\mathrm{Hz}, 7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.20(\mathrm{dd}, J=7.2 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ): $\delta=163.3,148.2,137.2,136.6,133.4,130.1,129.2,126.4,125.3,121.8,120.5$ ppm. GC: $\mathrm{t}_{\mathrm{R}}=15.2 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=242\left(\mathrm{M}^{+}, 43\right), 212(26), 150(100), 120(79), 104$ (45), 92 (51), 65 (30). The spectral data match those previously reported. ${ }^{60}$

$\boldsymbol{o}$ - and $\boldsymbol{m}$-methyl- $\boldsymbol{N}$-phenylbenzamide (3h): Starting from 1ah and isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to 2:1, $v / v$ ) in 92\% yield (77.8 mg ) as a mixture of isomers in a $38: 62$ ratio of $\boldsymbol{o} \mathbf{- 3 h}: \boldsymbol{m} \mathbf{- 3 h}$ according to ${ }^{1} \mathrm{H}$ NMR spectroscopy analysis. GC: $\mathrm{t}_{\mathrm{R}}=12.9 \mathrm{~min}$ (both isomers appear together); $\mathrm{MS}(\mathrm{EI}): \mathrm{m} / \mathrm{z}=$ $211\left(\mathrm{M}^{+}, 10\right), 119(100), 91(50), 65(30)$. The spectral data match those previously reported. ${ }^{61}$

$\boldsymbol{m}$ - and $\boldsymbol{p}$-Fluoro- $\boldsymbol{N}$-phenylbenzamide (3i): Starting from 1ai and isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to 2:1, $v / v$ ) in $82 \%$ yield (70.9 mg ) as a mixture of isomers in a $59: 41$ ratio of $\boldsymbol{m}-\mathbf{3 i} \mathbf{i} \boldsymbol{p}$-3i according to ${ }^{19} \mathrm{~F}$ NMR spectroscopy analysis. GC: $\mathrm{t}_{\mathrm{R}}=12.1 \mathrm{~min}$ (both isomers appear together); $\mathrm{MS}(\mathrm{EI}): \mathrm{m} / \mathrm{z}=$ $215\left(\mathrm{M}^{+}, 25\right), 123$ (100), 95 (40), 75 (10). The spectral data match those previously reported. ${ }^{47,62}$

m-3j
$\boldsymbol{m}$-Nitro- $\boldsymbol{N}$-phenylbenzamide ( $\boldsymbol{m}-\mathbf{3 j}$ ): Starting from 1aj and isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to 2:1, $v / v$ ) in 22\% yield (21.3 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.69(\mathrm{~s}, 1 \mathrm{H}), 8.39(\mathrm{~d}, J=8.0$ $\mathrm{Hz}, 1 \mathrm{H}), 8.25(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.01$ (br. s, 1H), 7.71-7.64 (m, 3H), $7.39(\mathrm{dd}, J=7.6$
$\mathrm{Hz}, 7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.20 (dd, $J=7.2 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ): $\delta=163.3,148.2,137.2,136.6,133.4,130.1,129.2,126.4,125.3,121.8,120.5$ ppm. GC: $\mathrm{t}_{\mathrm{R}}=15.2 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=242\left(\mathrm{M}^{+}, 43\right), 212(26), 150(100), 120(79), 104$ (45), 92 (51), 65 (30). The spectral data match those previously reported. ${ }^{60}$

$p-3 \mathrm{j}$
$\boldsymbol{p}$-Nitro- N -phenylbenzamide ( $\boldsymbol{p}-\mathbf{3 j}$ ): Starting from 1aj and isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to 2:1, $v / v$ ) in 27\% yield (25.7 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , acetone- $d_{6}$ ): $\delta=9.82$ (br. s, 1H), 8.37 (d, $J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 8.24(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.84(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.38(\mathrm{dd}, J=8.0 \mathrm{~Hz}$, $8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.16(\mathrm{dd}, J=7.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\} \mathrm{NMR}(100 \mathrm{MHz}$, acetone$\left.d_{6}\right): \delta=164.7,150.6,142.0,139.9,129.8,129.6,125.1,124.4,121.2 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=15.2$ min; MS (EI): $m / z=242\left(\mathrm{M}^{+}, 42\right), 212(27), 150(100), 120(78), 104$ (45), $92(51), 65$ (30). The spectral data match those previously reported. ${ }^{61}$

$\boldsymbol{m}$ - and $\boldsymbol{p}$-methyl- $\boldsymbol{N}$-phenylbenzamide (3k): Starting from 1ak and isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, 10:1 to 2:1, $v / v$ ) in 98\% yield (89.0 mg ) as a mixture of isomers in a 57:43 ratio of $\boldsymbol{m} \mathbf{- 3 k}: \boldsymbol{p}-\mathbf{3 k}$ according to ${ }^{1} \mathrm{H}$ NMR spectroscopy analysis. GC: $\mathrm{t}_{\mathrm{R}}=12.9 \mathrm{~min}$ (both isomers appear together); $\mathrm{MS}(\mathrm{EI}): \mathrm{m} / \mathrm{z}=$ $211\left(\mathrm{M}^{+}, 10\right), 119(100), 91(50), 65(30)$. The spectral data match those previously reported. ${ }^{61,63}$


Benzamide: Starting from phthalimide and isolated by column chromatography $\left(\mathrm{SiO}_{2}\right.$, petroleum ether/ethyl acetate, 5:1 to $1: 1, v / v$ ) in $17 \%$ yield ( 8.3 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR (400 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=7.83-7.80(\mathrm{~m}, 2 \mathrm{H}), 7.54-7.50(\mathrm{~m}, 1 \mathrm{H}), 7.46-7.42(\mathrm{~m}$,

2H), 6.26 (br. s, 2H) ppm. ${ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=169.7$, 133.4, 131.9, $128.6,127.3 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=8.8 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=121\left(\mathrm{M}^{+}, 46\right), 105(62), 77(100), 51$ (51). The spectral data match those previously reported. ${ }^{64}$


Benzoic acid: Starting from phthalic anhydride and isolated by column chromatography ( $\mathrm{SiO}_{2}$, petroleum ether/ethyl acetate, $4: 1$ to $1: 2, v / v$ ) in $51 \%$ yield ( 25.1 mg ) as a colourless solid. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=12.08$ (br. s, 1 H ), $8.16-8.13$ (m, 2 H ), 7.65-7.61 $(\mathrm{m}, 1 \mathrm{H}), 7.52-7.47(\mathrm{~m}, 2 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}\left\{{ }^{1} \mathrm{H}\right\}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=172.5,133.8$, $130.2,129.3,128.5 \mathrm{ppm} . \mathrm{GC}: \mathrm{t}_{\mathrm{R}}=7.0 \mathrm{~min} ; \mathrm{MS}(\mathrm{EI}): m / z=122\left(\mathrm{M}^{+}, 71\right), 105(94), 77$ (100), 51 (76). The spectral data match those previously reported. ${ }^{65}$

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## 9. X-ray crystallographic data for $2 x$ (CCDC-1577926).

$\left(\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{NO}\right) ; M=211.25$. APEXII, Bruker-AXS diffractometer, Mo-K $\alpha$ radiation $(\lambda=$ $0.71073 \AA$ ) , $T=150 \mathrm{~K}$; monoclinic $P 2{ }_{1} / n$ (I.T.\#14), $\mathrm{a}=7.9741(8), \mathrm{b}=24.748(3), \mathrm{c}=$

${ }^{1}$. The structure was solved by dual-space algorithm using the SHELXT program [1], and then refined with full-matrix least-square methods based on $F^{2}$ (SHELXL) [2]. All nonhydrogen atoms were refined with anisotropic atomic displacement parameters. Except nitrogen linked hydrogen atoms that were introduced in the structural model through Fourier difference maps analysis, H atoms were finally included in their calculated positions. A final refinement on $F^{2}$ with 2544 unique intensities and 149 parameters converged at $\omega R\left(F^{2}\right)=0.1120(R(F)=0.0439)$ for 2221 observed reflections with $I>$ $2 \sigma(I)$.
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### 9.1. Structural data for 2 x .

Empirical formula
Formula weight
Temperature
Wavelength
Crystal system, space group
Unit cell dimensions

Volume
$Z$, Calculated density
Absorption coefficient
F(000)
Crystal size
Crystal color
Theta range for data collection
h_min, h_max
k_min, k_max
1_min, 1_max
Reflections collected / unique
Reflections [ $\mathrm{I}>2 \sigma$ ]
Completeness to theta_max
Absorption correction type
Max. and min. transmission
Refinement method
Data / restraints / parameters
${ }^{\mathrm{b}} \mathrm{S}$ (Goodness-of-fit)
Final $R$ indices [ $[>2 \sigma$ ]
$R$ indices (all data)
Largest diff. peak and hole
${ }^{b} S=\left\{\sum\left[w\left(\mathrm{~F}_{\mathrm{o}}{ }^{2}-\mathrm{F}_{\mathrm{c}}{ }^{2}\right)^{2}\right] /(n-p)\right\}^{1 / 2}$
${ }^{c} R 1=\sum| | F_{o}\left|-\left|F_{c}\right|\right| / \sum\left|F_{o}\right|$
${ }^{d} w R 2=\left\{\sum\left[w\left(F_{o}{ }^{2}-F_{c}{ }^{2}\right)^{2}\right] / \sum\left[w\left(F_{o}{ }^{2}\right)^{2}\right]\right\}^{1 / 2}$
$w=1 /\left[\sigma\left(F_{0}{ }^{2}\right)+\mathrm{a} P^{2}+\mathrm{b} P\right]$ where $P=\left[2 F_{\mathrm{c}}{ }^{2}+\operatorname{MAX}\left(F_{0}^{2}, 0\right)\right] / 3$
9.2. Atomic coordinates, site occupancy (\%) and equivalent isotropic displacement parameters $(\AA 2) . U(e q)$ is defined as one third of the trace of the orthogonalized Uij tensor.

| Atom | x | y | z | occ. | $\mathrm{U}(\mathrm{eq})$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| C1 | $0.3255(2)$ | $0.34549(6)$ | $0.3349(3)$ | 1 | $0.0285(4)$ |
| H1 | 0.391412 | 0.327271 | 0.222448 | 1 | 0.034 |
| C2 | $0.2761(2)$ | $0.39830(7)$ | $0.2968(3)$ | 1 | $0.0336(4)$ |
| H2 | 0.308436 | 0.416411 | 0.158844 | 1 | 0.040 |
| C3 | $0.1793(2)$ | $0.42470(7)$ | $0.4605(3)$ | 1 | $0.0341(4)$ |
| H3 | 0.146982 | 0.461188 | 0.435725 | 1 | 0.041 |
| C4 | $0.1295(2)$ | $0.39827(7)$ | $0.6599(3)$ | 1 | $0.0311(4)$ |
| H4 | 0.061139 | 0.416355 | 0.769875 | 1 | 0.037 |
| C5 | $0.1794(2)$ | $0.34538(6)$ | $0.6992(3)$ | 1 | $0.0255(4)$ |
| H5 | 0.145689 | 0.327260 | 0.836479 | 1 | 0.031 |
| C6 | $0.27898(19)$ | $0.31889(6)$ | $0.5372(3)$ | 1 | $0.0226(4)$ |
| C7 | $0.3465(2)$ | $0.26299(6)$ | $0.5745(3)$ | 1 | $0.0259(4)$ |
| O8 | $0.47488(17)$ | $0.24814(5)$ | $0.4742(3)$ | 1 | $0.0487(4)$ |
| N9 | $0.26593(18)$ | $0.23075(5)$ | $0.7208(2)$ | 1 | $0.0242(3)$ |
| H9 | $0.169(3)$ | $0.2412(9)$ | $0.790(4)$ | 1 | 0.050 |
| C10 | $0.3270(2)$ | $0.17662(6)$ | $0.7724(3)$ | 1 | $0.0266(4)$ |
| H10A | 0.298256 | 0.167825 | 0.934790 | 1 | 0.032 |
| H10B | 0.450836 | 0.176616 | 0.759846 | 1 | 0.032 |
| C11 | $0.25799(19)$ | $0.13267(6)$ | $0.6173(3)$ | 1 | $0.0236(4)$ |
| C12 | $0.2780(2)$ | $0.07900(7)$ | $0.6840(3)$ | 1 | $0.0308(4)$ |
| H12 | 0.332086 | 0.070949 | 0.826492 | 1 | 0.037 |
| C13 | $0.2205(2)$ | $0.03725(7)$ | $0.5463(3)$ | 1 | $0.0371(5)$ |
| H13 | 0.235951 | 0.000845 | 0.593934 | 1 | 0.044 |
| C14 | $0.1402(2)$ | $0.04842(7)$ | $0.3390(3)$ | 1 | $0.0352(4)$ |
| H14 | 0.101047 | 0.019839 | 0.243381 | 1 | 0.042 |
| C15 | $0.1178(2)$ | $0.10152(7)$ | $0.2726(3)$ | 1 | $0.0309(4)$ |
| H15 | 0.061415 | 0.109434 | 0.131743 | 1 | 0.037 |
| C16 | $0.1768(2)$ | $0.14343(6)$ | $0.4097(3)$ | 1 | $0.0273(4)$ |
| H16 | 0.161586 | 0.179772 | 0.361142 | 1 | 0.033 |

9.3. Anisotropic displacement parameters $\left(\AA^{2}\right)$. The anisotropic displacement factor exponent takes the form: $-2 \pi^{2}\left[h^{2} a^{* 2} U_{11}+\ldots+2 h k a^{*} b^{*} U_{12}\right]$.

| Atom | U11 | U22 | U33 | U 23 | U 13 | U 12 |
| :--- | :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| C1 | $0.0312(9)$ | $0.0295(9)$ | $0.0248(9)$ | $0.0005(7)$ | $0.0045(8)$ | $-0.0074(7)$ |
| C2 | $0.0381(10)$ | $0.0357(9)$ | $0.0271(9)$ | $0.0081(8)$ | $-0.0022(8)$ | $-0.0105(8)$ |
| C3 | $0.0338(9)$ | $0.0232(8)$ | $0.0453(11)$ | $0.0027(8)$ | $-0.0055(9)$ | $-0.0024(7)$ |
| C4 | $0.0298(9)$ | $0.0266(8)$ | $0.0369(10)$ | $-0.0065(8)$ | $0.0029(8)$ | $-0.0001(7)$ |
| C5 | $0.0249(8)$ | $0.0243(8)$ | $0.0274(8)$ | $-0.0017(7)$ | $0.0046(8)$ | $-0.0046(7)$ |
| C6 | $0.0205(8)$ | $0.0238(7)$ | $0.0236(8)$ | $-0.0016(6)$ | $0.0040(7)$ | $-0.0055(6)$ |
| C7 | $0.0224(8)$ | $0.0277(8)$ | $0.0276(9)$ | $-0.0013(7)$ | $0.0113(7)$ | $-0.0022(6)$ |
| O8 | $0.0417(8)$ | $0.0381(7)$ | $0.0664(10)$ | $0.0105(7)$ | $0.0407(8)$ | $0.0091(6)$ |
| N9 | $0.0231(7)$ | $0.0216(6)$ | $0.0280(8)$ | $0.0008(5)$ | $0.0119(6)$ | $0.0003(5)$ |
| C10 | $0.0280(8)$ | $0.0245(8)$ | $0.0273(9)$ | $0.0031(7)$ | $0.0038(7)$ | $0.0011(7)$ |
| C11 | $0.0199(7)$ | $0.0248(8)$ | $0.0261(8)$ | $0.0028(6)$ | $0.0077(7)$ | $0.0013(6)$ |
| C12 | $0.0338(9)$ | $0.0259(8)$ | $0.0327(9)$ | $0.0049(7)$ | $-0.0035(8)$ | $0.0044(7)$ |
| C13 | $0.0436(11)$ | $0.0227(8)$ | $0.0449(11)$ | $0.0010(8)$ | $-0.0025(10)$ | $0.0039(8)$ |
| C14 | $0.0370(10)$ | $0.0296(9)$ | $0.0389(11)$ | $-0.0051(8)$ | $0.0000(9)$ | $0.0003(7)$ |
| C15 | $0.0306(9)$ | $0.0341(9)$ | $0.0280(9)$ | $-0.0014(8)$ | $0.0009(8)$ | $0.0039(7)$ |
| C16 | $0.0292(8)$ | $0.0252(8)$ | $0.0275(9)$ | $0.0035(7)$ | $0.0071(8)$ | $0.0045(7)$ |

### 9.4. Bond lengths $[\AA]$.

| C1 | - C2 | $=1.383(2)$ | C1 | - C6 | = $1.392(2)$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| C1 | - H1 | $=0.9500$ | C2 | - C3 | $=1.385(3)$ |
| C2 | - H2 | $=0.9500$ | C3 | - C4 | $=1.383(3)$ |
| C3 | - H3 | $=0.9500$ | C4 | - C5 | $=1.387(2)$ |
| C4 | - H4 | $=0.9500$ | C5 | - C6 | $=1.392(2)$ |
| C5 | - H5 | $=0.9500$ | C6 | - C7 | $=1.500(2)$ |
| C7 | - O8 | $=1.233(2)$ | C7 | - N9 | $=1.328(2)$ |
| N9 | - C10 | $=1.456(2)$ | N9 | - H9 | $=0.91(2)$ |
| C10 | - C11 | $=1.513(2)$ | C10 | - H10A | $=0.9900$ |
| C10 | - H10B | $=0.9900$ | C11 | - C16 | $=1.388(3)$ |
| C11 | - C12 | $=1.392(2)$ | C12 | - C13 | = 1.382(3) |
| C12 | - H12 | $=0.9500$ | C13 | - C14 | $=1.385(3)$ |
| C13 | - H13 | $=0.9500$ | C14 | - C15 | $=1.380(2)$ |
| C14 | - H14 | $=0.9500$ | C15 | - C16 | $=1.387(2)$ |
| C15 | - H15 | $=0.9500$ | C16 | - H16 | $=0.9500$ |

### 9.5. Angles [ ${ }^{\circ}$ ].

| C2 | - C1 | - $\mathrm{C} 6=120.33(16)$ |
| :---: | :---: | :---: |
| C2 | - C1 | - H1 $=119.80$ |
| C6 | - C1 | - H1 $=119.80$ |
| C1 | - C2 | $-\mathrm{C} 3=119.72(17)$ |
| C1 | - C2 | $-\mathrm{H} 2=120.10$ |
| C3 | - C2 | - $\mathrm{H} 2=120.10$ |
| C4 | - C3 | $-\mathrm{C} 2=120.41$ (16) |
| C4 | - C3 | - H3 = 119.80 |
| C2 | - C3 | - H3 $=119.80$ |
| C3 | - C4 | - $\mathrm{C} 5=120.03(17)$ |
| C3 | - C4 | $-\mathrm{H} 4=120.00$ |
| C5 | - C4 | $-\mathrm{H} 4=120.00$ |
| C4 | - C5 | - C6 = 119.86(16) |
| C4 | - C5 | - H5 $=120.10$ |
| C6 | - C5 | - H5 = 120.10 |
| C1 | - C6 | - $\mathrm{C} 5=119.62(15)$ |
| C1 | - C6 | $-\mathrm{C} 7=117.47$ (14) |
| C5 | - C6 | $-\mathrm{C} 7=122.85(15)$ |
| O8 | - C7 | - $\mathrm{N} 9=121.47$ (16) |
| O8 | - C7 | - $\mathrm{C} 6=120.37(15)$ |
| N9 | - C7 | - $\mathrm{C} 6=118.15$ (14) |
| C7 | - N9 | $-\mathrm{C} 10=121.40(14)$ |
| C7 | - N9 | - H9 = 121.60(14) |
| C10 | - N9 | - $\mathrm{H} 9=117.00(14)$ |
| N9 | - C10 | - $\mathrm{C} 11=114.75$ (14) |
| N9 | - C10 | - $\mathrm{H} 10 \mathrm{~A}=108.60$ |
| C11 | - C10 | - $\mathrm{H} 10 \mathrm{~A}=108.60$ |
| N9 | - C10 | - $\mathrm{H} 10 \mathrm{~B}=108.60$ |
| C11 | - C10 | - $\mathrm{H} 10 \mathrm{~B}=108.60$ |
| H10A | - C10 | - $\mathrm{H} 10 \mathrm{~B}=107.60$ |
| C16 | - C11 | $-\mathrm{C} 12=118.40$ (16) |
| C16 | - C11 | $-\mathrm{C} 10=122.89(14)$ |
| C12 | - C11 | $-\mathrm{C} 10=118.71(15)$ |
| C13 | - C12 | $-\mathrm{C} 11=121.06(17)$ |
| C13 | - C12 | - $\mathrm{H} 12=119.50$ |
| C11 | - C12 | $-\mathrm{H} 12=119.50$ |
| C12 | - C13 | $-\mathrm{C} 14=120.07(16)$ |
| C12 | - C13 | - $\mathrm{H} 13=120.00$ |
| C14 | - C13 | - $\mathrm{H} 13=120.00$ |
| C15 | - C14 | $-\mathrm{C} 13=119.31(17)$ |
| C15 | - C14 | - $\mathrm{H} 14=120.30$ |
| C13 | - C14 | - $\mathrm{H} 14=120.30$ |
| C14 | - C15 | $-\mathrm{C} 16=120.65(17)$ |
| C14 | - C15 | - $\mathrm{H} 15=119.70$ |
| C16 | - C15 | - $\mathrm{H} 15=119.70$ |
| C15 | - C16 | $-\mathrm{C} 11=120.49(15)$ |
| C15 | - C16 | - H16 $=119.80$ |
| C11 | - C16 | - $\mathrm{H} 16=119.80$ |

### 9.6. Torsion angles $\left[{ }^{\circ}\right]$.

| C6 | - C1 | - C2 | - C3 | $=-0.30(3)$ |
| :---: | :---: | :---: | :---: | :---: |
| C1 | - C2 | - C3 | - C4 | $=-1.10$ (3) |
| C2 | - C3 | - C4 | - C5 | $=1.40$ (3) |
| C3 | - C4 | - C5 | - C6 | $=-0.30(3)$ |
| C2 | - C1 | - C6 | - C5 | $=1.30$ (3) |
| C2 | - C1 | - C6 | - C7 | $=-175.97$ (16) |
| C4 | - C5 | - C6 | - C1 | $=-1.00$ (3) |
| C4 | - C5 | - C6 | - C7 | $=176.11(16)$ |
| C1 | - C6 | - C7 | - O 8 | $=22.30(2)$ |
| C5 | - C6 | - C7 | - O 8 | $=-154.94(18)$ |
| C1 | - C6 | - C7 | - N9 | $=-158.93$ (15) |
| C5 | - C6 | - C7 | - N9 | $=23.90$ (2) |
| O8 | - C7 | - N9 | - C10 | $=1.00(3)$ |
| C6 | - C7 | - N9 | - C10 | $=-177.80(14)$ |
| C7 | - N9 | - C10 | - C11 | $=-90.94(19)$ |
| N9 | - C10 | - C11 | - C16 | $=15.00$ (2) |
| N9 | - C10 | - C11 | - C12 | $=-165.61(15)$ |
| C16 | - C11 | - C12 | - C13 | $=0.80$ (3) |
| C10 | - C11 | - C12 | - C13 | $=-178.63$ (16) |
| C11 | - C12 | - C13 | - C14 | $=-0.50(3)$ |
| C12 | - C13 | - C14 | - C15 | $=-0.40(3)$ |
| C13 | - C14 | - C15 | - C16 | $=0.90$ (3) |
| C14 | - C15 | - C16 | - C11 | $=-0.70(3)$ |
| C12 | - C11 | - C16 | - C15 | $=-0.20(2)$ |
| C10 | - C11 | - C16 | - C15 | $=179.18(15)$ |

10. NMR spectra of products 2-3.




























##  促 Thnifl




























${ }^{1} \mathrm{H}$ NMR（ $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ）spectrum of $\mathbf{2 s}$ ．





















##    1717














성N M




A. 28






(1)


























