

**Modeling the Concentration of Volatile and Semi-Volatile Contaminants in Direct
Contact Membrane Distillation (DCMD) Product Water:**

Supplementary Information

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Supplementary information consisting of one table describing a list of compounds used to model overall mass transfer coefficients and equilibrium ratio values for treatment by the large-scale DCMD system. This table extends across two pages.

SI Table 1. List of compounds used to model overall mass transfer coefficients and equilibrium ratio values for treatment by the large-scale DCMD system. Henry's constants calculated using the method described in the manuscript.

Compound	Chemical Formula	Molecular Weight (g/mol)	Activity at Infinite Dilution¹	Henry's constant at 25°C [PaL/mol] Log₁₀(H)
Ammonia	NH ₃	17.03	1.00E+00	4.26
Ethanol	C ₂ H ₆ O	46.07	4.09E+00	2.81
Acetone	C ₃ H ₆ O	58.08	1.01E+01	3.74
Butane	C ₄ H ₁₀	58.12	4.60E+04	8.31
Propanol	C ₃ H ₈ O	60.10	1.73E+01	3.01
Methyl ethyl ketone	C ₄ H ₈ O	72.11	2.93E+01	3.79
Butanol	C ₄ H ₁₀ O	74.12	5.42E+01	2.93
Benzene	C ₆ H ₆	78.11	2.47E+03	5.68
Pyridine	C ₅ H ₅ N	79.10	2.10E+01	3.02
Dichloromethane	CH ₂ Cl ₂	84.93	3.57E+02	5.57
Diethyl ketone	C ₅ H ₁₀ O	86.13	9.36E+01	3.89
Methyl propyl ketone	C ₅ H ₁₀ O	86.13	8.30E+01	3.92
Methyl tert-butyl ether	C ₅ H ₁₂ O	88.15	1.47E+02	4.95
Pentanol	C ₅ H ₁₂ O	88.15	2.26E+02	3.10
Toluene	C ₇ H ₈	92.14	8.92E+03	5.79
Epichlorohydrin	C ₃ H ₅ ClO	92.53	8.64E+01	3.58
Aniline	C ₆ H ₇ N	93.13	1.21E+02	2.13
Phenol	C ₆ H ₆ O	94.11	5.69E+01	1.61
1,1-Dichloroethylene	C ₂ H ₂ Cl ₂	96.94	2.30E+03	6.51
cis-1,2-Dichloroethylene	C ₂ H ₂ Cl ₂	96.94	8.40E+02	5.60
trans-1,2-Dichloroethylene	C ₂ H ₂ Cl ₂	96.94	1.19E+03	5.96
Cyclohexanone	C ₆ H ₁₀ O	98.15	7.99E+01	2.86
1,2-Dichloroethane	C ₂ H ₄ Cl ₂	98.96	6.36E+02	5.07
Methyl butyl ketone	C ₆ H ₁₂ O	100.16	3.50E+02	3.80
Hexanol	C ₆ H ₁₄ O	102.18	9.52E+02	3.51
Styrene	C ₈ H ₈	104.15	1.85E+04	5.45
1,4 Xylene	C ₈ H ₁₀	106.17	3.18E+04	5.83
Ethylbenzene	C ₈ H ₁₀	106.17	3.33E+04	5.91
Xylenes (total)	C ₈ H ₁₀	106.17	5.55E+04	5.95
2-Methylphenol	C ₇ H ₈ O	108.14	9.49E+01	0.49
3-Methylphenol	C ₇ H ₈ O	108.14	9.49E+01	1.68
4-Methylphenol	C ₇ H ₈ O	108.14	9.49E+01	1.38
Phenoxymethanol	C ₇ H ₈ O	108.14	3.10E+03	1.03
Phenylmethanol	C ₇ H ₈ O	108.14	5.48E+01	4.45
Chlorobenzene	C ₆ H ₅ Cl	112.56	1.15E+04	5.55

1,2-Dichloropropane	C3H6Cl2	112.99	2.20E+03	5.44
Octane	C8H18	114.23	1.47E+07	8.72
Heptanol	C7H16O	116.20	3.94E+03	3.47
Acetophenone	C6H8O	120.15	1.28E+03	3.08
1,2,4-Trimethylbenzene	C9H12	120.19	1.17E+05	6.06
Propylbenzene	C9H12	120.19	1.28E+05	5.58
2-Phenylethanol	C8H10O	122.17	2.77E+02	1.76
2,4-Dimethylphenol	C8H10O	122.17	8.56E+02	2.29
Octanol	C8H18O	130.23	1.57E+04	3.60
Trichloroethylene	C2HCl3	131.39	6.62E+03	5.99
1,1,1-Trichloroethane	C2H3Cl3	133.40	5.72E+03	6.22
1,1,2-Trichloroethane	C2H3Cl3	133.40	1.61E+03	4.97
1,3-Dimethoxybenzene	C8H10O2	138.17	6.31E+03	0.85
2-Phenoxyethanol	C8H10O2	138.17	2.59E+02	2.46
Di-tertbutyl ketone	C9H18O	142.24	2.39E+03	4.24
Decane	C10H22	142.29	3.99E+08	9.18
o-Dichlorobenzene	C6H4Cl2	147.00	5.88E+04	5.32
p-Dichlorobenzene	C6H4Cl2	147.00	1.06E+05	5.80
Carbon tetrachloride	CCl4	153.82	1.05E+04	6.45
Tetrachloroethylene	C2Cl4	165.83	4.35E+04	6.30
Dodecane	C12H26	170.34	2.50E+09	8.95
1,2,4-Trichlorobenzene	C6H3Cl3	181.45	7.42E+05	5.89
Ethylene dibromide	C2H4Br2	187.86	2.65E+03	4.93
Nonylphenol	C15H24O	220.36	1.75E+06	2.59
Hexachlorocyclopentadiene	C5Cl6	272.77	1.88E+07	6.45

(1) Yaws, C. L. *Yaws' Handbook of Properties for Aqueous Systems*; Knovel, 2012.