

Supporting Information for

Energy-Renormalization for Achieving Temperature Transferable Coarse-Graining of Polymer Dynamics

Wenjie Xia, Jake Song, Cheol Jeong, David D. Hsu, Frederick R. Phelan Jr.,

Jack F. Douglas*, Sinan Keten*

*To whom correspondence should be addressed. E-mails: jack.douglas@nist.gov (J.F.D),
s-keten@northwestern.edu (S.K.)

Supporting Figures

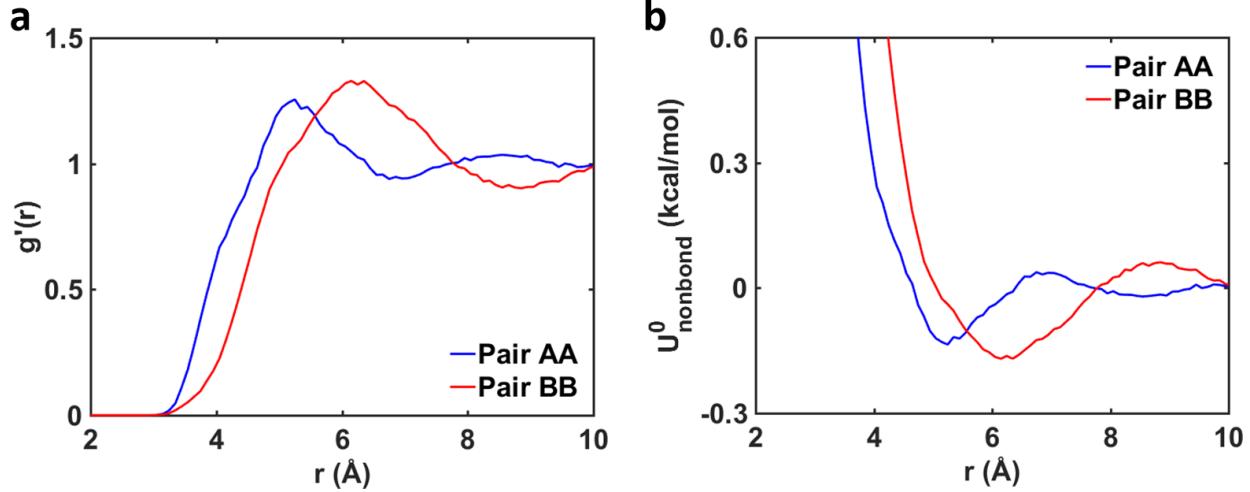


Figure S1. (a) All-atomistic radial distribution functions (RDF) $g'(r)$ between effective bead centers *AA* and *BB* used in the CG mapping scheme, and (b) their respective potentials U^0_{nonbond} obtained via the Inverse Boltzmann method. The initial nonbonded LJ parameters are determined: $\sigma_{AA}^0 = 4.60 \text{ \AA}$, $\sigma_{BB}^0 = 5.06 \text{ \AA}$, $\varepsilon_{AA}^0 = 0.136 \text{ kcal/mol}$, and $\varepsilon_{BB}^0 = 0.174 \text{ kcal/mol}$.

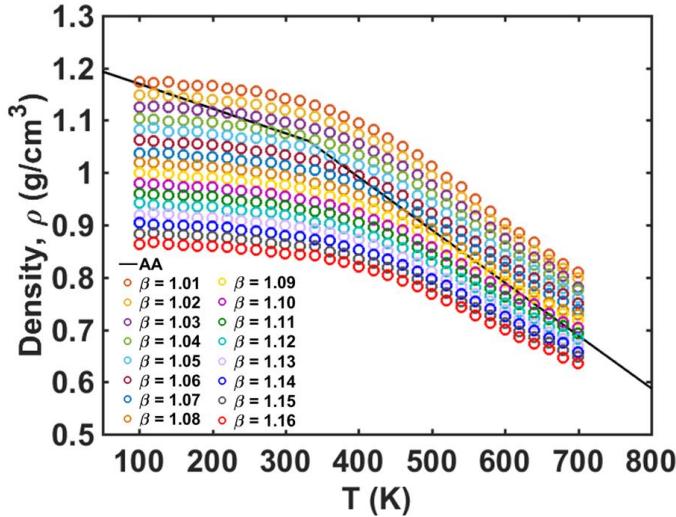


Figure S2. Density sweeps as function of temperature for AA (solid line) and CG (open symbols) models with varying β parameters from 1.01 to 1.16. Temperature-dependent β can be determined by matching the density results of AA model, leading to linear scaling relationship.

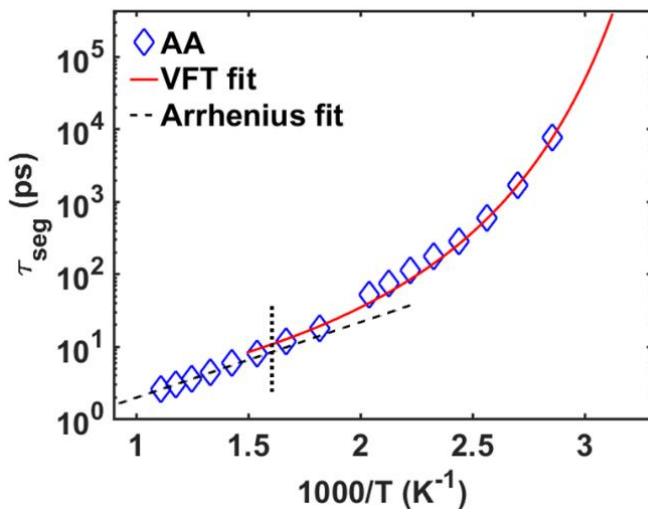


Figure S3. τ_{seg} as a function of temperature for the AA model. The solid curve shows the VFT fit of the data for AA model. The dashed slope shows the Arrhenius fit of τ_{seg} at high temperatures. The vertical dotted line marks T_A .

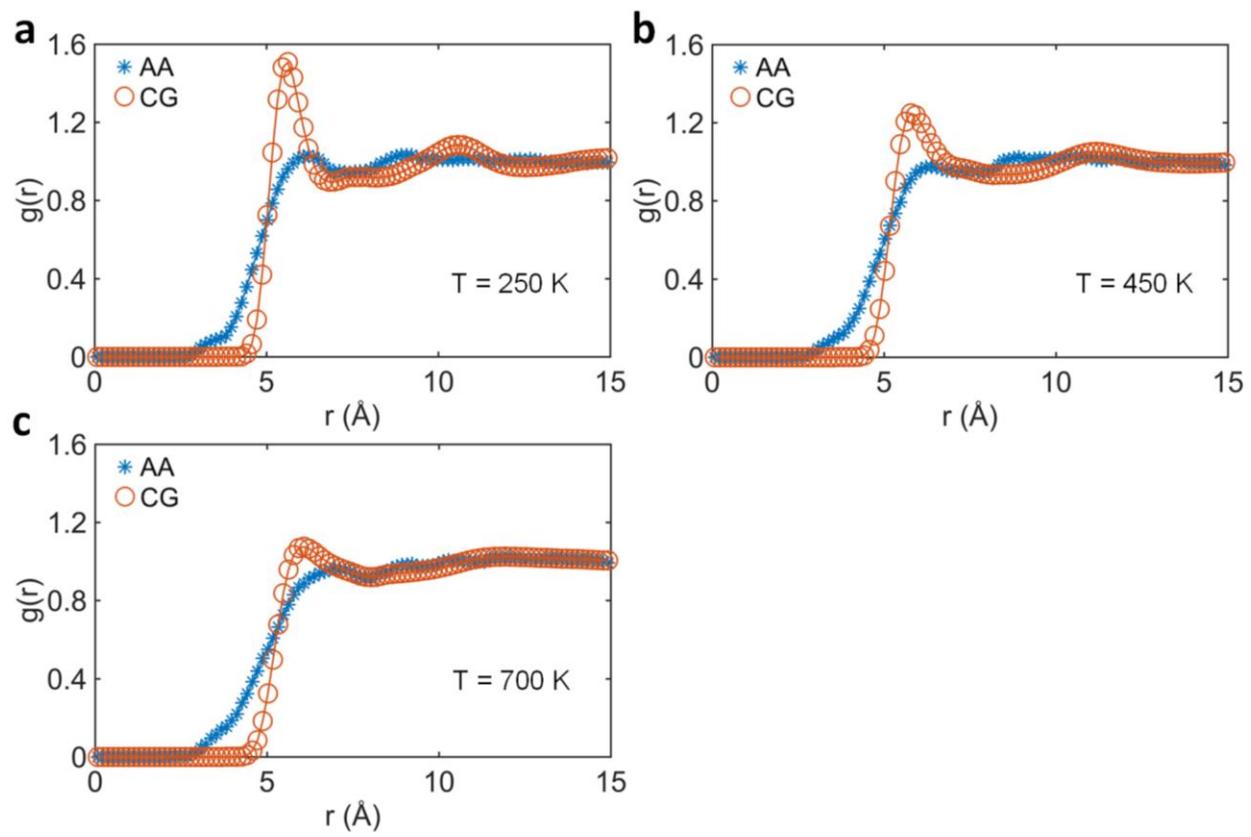


Figure S4. Comparison of the RDF $g(r)$ of all the CG bead sites within the AA and CG systems at different temperatures: (a) $T = 250$ K, (b) $T = 450$ K, and (c) $T = 700$ K.

Supporting Tables

Table S1. Functional Form of Force Field and Potential Parameters for CG Model of PS.

Interaction	Potential form	Parameters		
A-A Bond Length	$U_{bondAA}(l) = k(l - l_0)^2$	$k = 80.69 \text{ kcal/mol}\cdot\text{\AA}^2$	$l_0 = 2.568 \text{ \AA}$	
A-B Bond Length	$U_{bondAB}(l) = k(l - l_0)^2$	$k = 224.6 \text{ kcal/mol}\cdot\text{\AA}^2$	$l_0 = 2.871 \text{ \AA}$	
A-A-A Angle	$U_{angleAAA}(\theta) = -k_b T \ln \left[a_1 \exp \left(\frac{\theta - \theta_1}{b_1} \right)^2 + a_2 \exp \left(\frac{\theta - \theta_2}{b_2} \right)^2 + a_3 \exp \left(\frac{\theta - \theta_3}{b_3} \right)^2 \right]$	$a_1 = 2.277e-1$ $b_1 = 10.76^\circ$ $\theta_1 = 177.1^\circ$	$a_2 = 5.588e-1$ $b_2 = 19.27^\circ$ $\theta_2 = 148.8^\circ$	$a_3 = -5.497e-1$ $b_3 = 18.78^\circ$ $\theta_3 = 148.5^\circ$
A-A-B Angle	$U_{angleAAB}(\theta) = -k_b T \ln \left[a_1 \exp \left(\frac{\theta - \theta_1}{b_1} \right)^2 + a_2 \exp \left(\frac{\theta - \theta_2}{b_2} \right)^2 + a_3 \exp \left(\frac{\theta - \theta_3}{b_3} \right)^2 \right]$	$a_1 = 1.451e-1$ $b_1 = 6.762^\circ$ $\theta_1 = 141.5^\circ$	$a_2 = 1.767e-2$ $b_2 = 9.745^\circ$ $\theta_2 = 93.22^\circ$	$a_3 = 3.199e-2$ $b_3 = 11.01^\circ$ $\theta_3 = 134.7^\circ$
A-A-A-A Dihedral Angle	$U_{dihedralAAA}(\phi) = A \cos(\phi)$	A (atactic) = 0.5 (kcal/mol)		
B-A-A-B Dihedral Angle	$U_{dihedralBAAB}(\phi) = \sum_{k=1}^5 A_k \cos^{k-1}(\phi)$	$meso$ $A_1 = 4.236 \text{ (kcal/mol)}$ $A_2 = -1.274 \text{ (kcal/mol)}$ $A_3 = 0.5337 \text{ (kcal/mol)}$ $A_4 = -0.8179 \text{ (kcal/mol)}$ $A_5 = -0.4622 \text{ (kcal/mol)}$ $\varepsilon_{AA} = \alpha(T) \times 0.136 \text{ (kcal/mol)}$ $\sigma_{AA} = \beta(T) \times 4.60 \text{ (\AA)}$		
Non-bonded	$U_{nonbond} = 4\varepsilon(T) \left[\left(\frac{\sigma(T)}{r} \right)^{12} - \left(\frac{\sigma(T)}{r} \right)^6 \right] + S_{LJ}(r)$	$racemo$ $A_1 = 3.762 \text{ (kcal/mol)}$ $A_2 = 0.3631 \text{ (kcal/mol)}$ $A_3 = 2.128 \text{ (kcal/mol)}$ $A_4 = -0.9691 \text{ (kcal/mol)}$ $A_5 = -1.880 \text{ (kcal/mol)}$ $\varepsilon_{BB} = \alpha(T) \times 0.174 \text{ (kcal/mol)}$ $\sigma_{BB} = \beta(T) \times 5.06 \text{ (\AA)}$		

Table S2. Functional Form of Temperature Dependent Energy-Renormalization Factors for the CG Nonbonded Potential.

Scaling factor	Functional form	Parameters
$\alpha(T)$	$\alpha(T) = \frac{\alpha_A - \alpha_g}{1 + \exp[-k(T - T_T)]} + \alpha_g$	$\alpha_A = 2.33, \alpha_g = 3.60$ $T_T = 475.2 \text{ K}$ $k = 0.0185 \text{ K}^{-1}$
$\beta(T)$	$aT + b$	$a = 1.718 \times 10^{-4} \text{ K}^{-1}, b = 0.9905$

Table S3. Comparison of the ratios of characteristic temperatures of simulation results described in the main text and prediction of the generalized entropy theory (GET) for flexible-stiff class of polymer.

Predictions	T_A/T_g	T_g/T_0	T_A/T_0	T_l/T_0
AA	2.14	1.11	2.38	2.22
CG	2.15	1.11	2.39	2.23
GET	1.70	1.09	1.86	-