

Hexagonal $\text{Ce}_{1-x}\text{Ln}_x(\text{OH})\text{CO}_3$ as highly efficient precursors of nanocrystalline $\text{Ln}(\text{III-IV})$ -substituted ceria.

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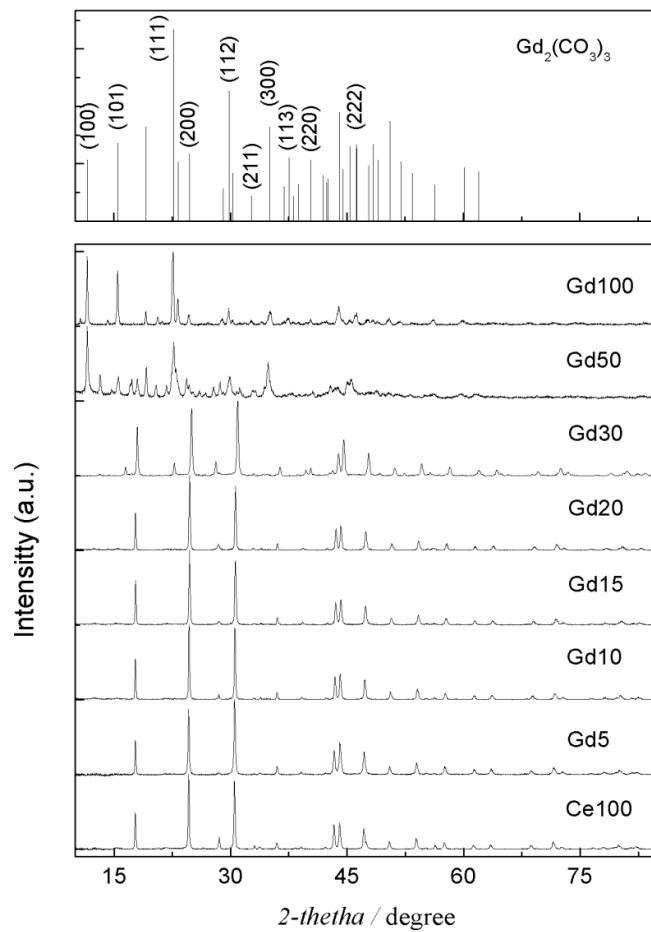


Figure S1. Lower panel: PXRD patterns of binary precursors $\text{Ce}(\text{III})\text{-Gd}(\text{III})$ with increasing $\text{Gd}(\text{III})$ content. Upper panel depicts the main reflections of $\text{Gd}_2(\text{CO}_3)_3$ (PDF 37-0559).

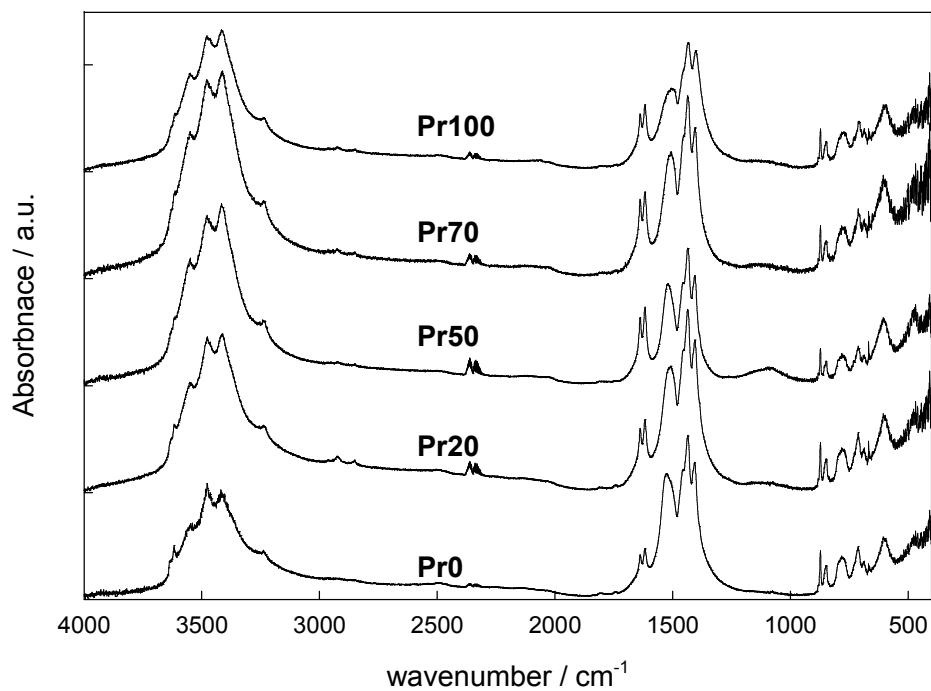


Figure S2. FTIR spectra recorded fro representative $\text{Ce}_{1-x}\text{Pr}_x(\text{OH})\text{CO}_3$.

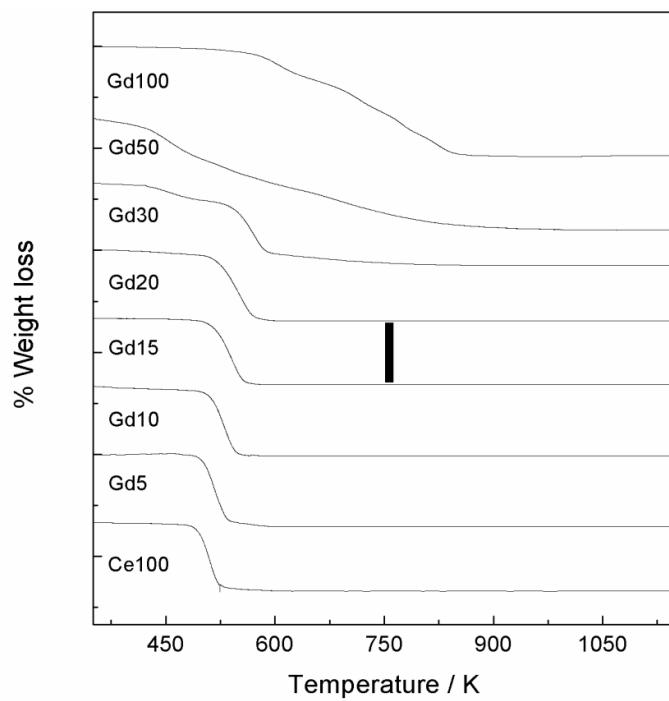


Figure S3. Thermogravimetric trace of $\text{Ce}_{1-x}\text{Gd}_x(\text{OH})\text{CO}_3$ samples recorded at 1 K min^{-1} under air atmosphere.

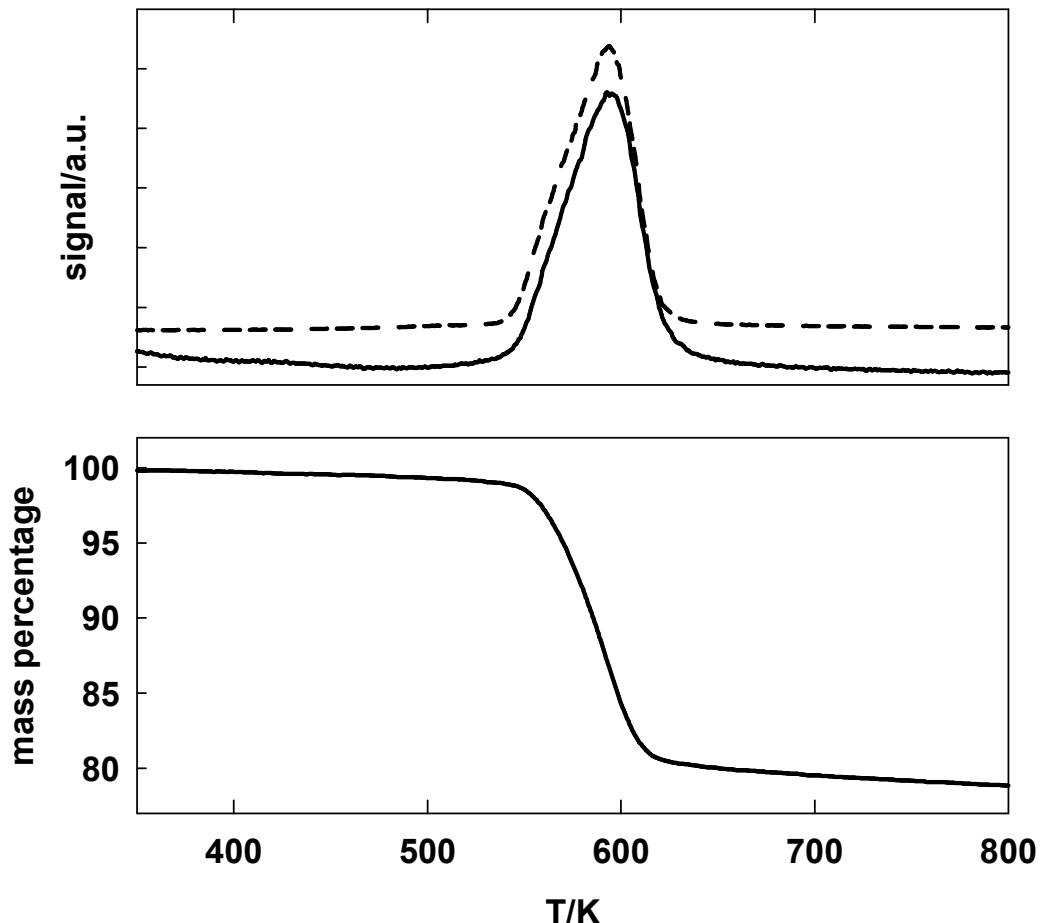


Figure S4. Upper panel: evolution of m/z 44 (dashed line) and 18 (full line) and mass (lower panel) as a function of temperature, during decomposition of sample Gd20 recorded at 5 K min^{-1} under air atmosphere.

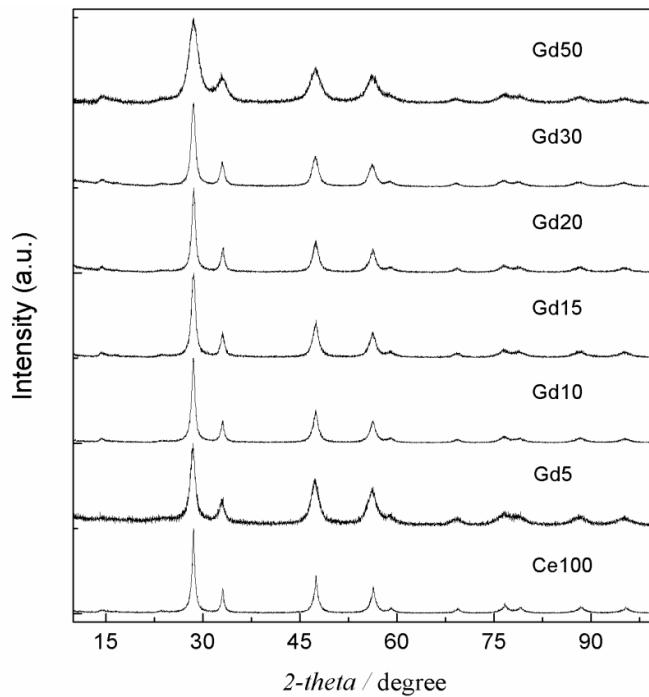


Figure S5. PXRD patterns of the oxides obtained after decomposition of $\text{Ce}_{1-x}\text{Gd}_x(\text{OH})\text{CO}_3$ precursors in air at 723 K.

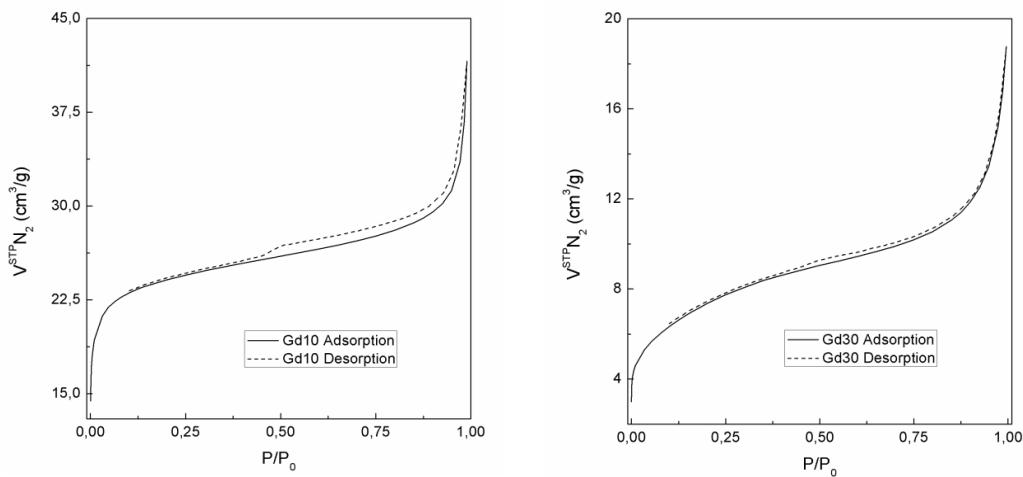


Figure S6. Typical N_2 sorption isotherms of the oxides obtained after decomposition of samples Gd10 (left) and Gd30 (right) at 723 K.

Table S1. Specific surface area obtained from N₂ isotherms recorded at 77 K.

Sample	Langmuir Surface Area (m ² /g)	Sample	Langmuir Surface Area (m ² /g)
Ce100	90	Sm15	91
Pr5	111	Sm20	80
Pr10	116	Sm30	58
Pr15	118	Sm50	15
Pr20	73	Gd10	111
Pr30	100	Gd15	116
Pr50	15	Gd20	110
Sm5	129	Gd30	39
Sm10	105		