

## *Supporting Information*

### **Effects of High Pressure on Phospholipid Bilayers**

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#### **Electron density profiles**

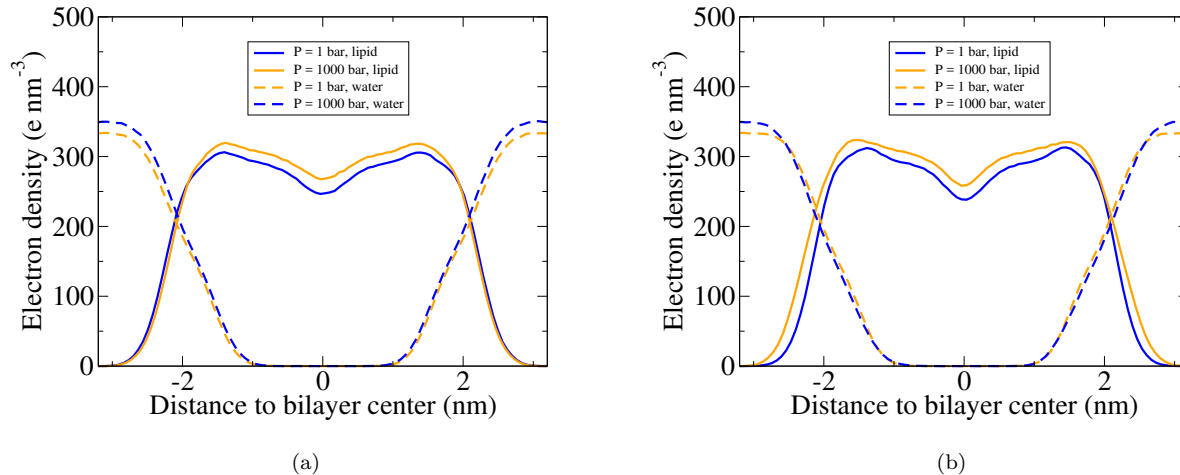


Figure S1: Electron density of the individual water and lipid components for (a) DOPC and (b) POPC bilayers at 1 bar (blue) and 1000 bar (orange).

### Headgroup orientation

We calculated the angle between the lipid head P-N vector and the z-axis (i.e. the bilayer normal). For DOPC, the angle is  $69.53 \pm 0.20$  degrees at 1 bar and  $69.24 \pm 0.23$  degrees at 1000 bar, while for POPC it is  $69.38 \pm 0.22$  degrees at 1 bar and  $69.54 \pm 0.22$  degrees at 1000 bar. Statistically, the differences for both pairs are not significant ( $p > 0.05$ ). Incidentally, the absence of a significant effect on the headgroup orientation is consistent with the results obtained for the dipole potential, when one considers that the P-N vector is the main contributor (in terms of magnitude) to the overall dipole potential.

### Hydrogen bonds

The table below reports the average number of hydrogen bonds per lipid formed between lipid head groups and water molecules (“Head-Water”), and between lipid head groups themselves (“Head-Head”):

Number of H bonds	DOPC		POPC	
	1 bar	1000 bar	1 bar	1000 bar
Head-Water	$6.87 \pm 0.04$	$6.91 \pm 0.03$	$6.60 \pm 0.03$	$6.66 \pm 0.03$
Head-Head	$0.34 \pm 0.02$	$0.34 \pm 0.01$	$0.38 \pm 0.01$	$0.39 \pm 0.01$