

Supporting Information

Ab Initio Modeling of Structure and Properties of Single and Mixed Alkali Silicate Glasses

Khagendra Baral¹, Aize Li² and Wai-Yim Ching^{1*}

1. University of Missouri-Kansas City, Department of Physics and Astronomy, Kansas City, Missouri, 64110, USA
2. Corning Incorporated, Corning, New York, 14870, USA

Corresponding author: Chingw@umkc.edu

This supporting information present the Figures for the 30% doping similar to the 20% doping in the main text.

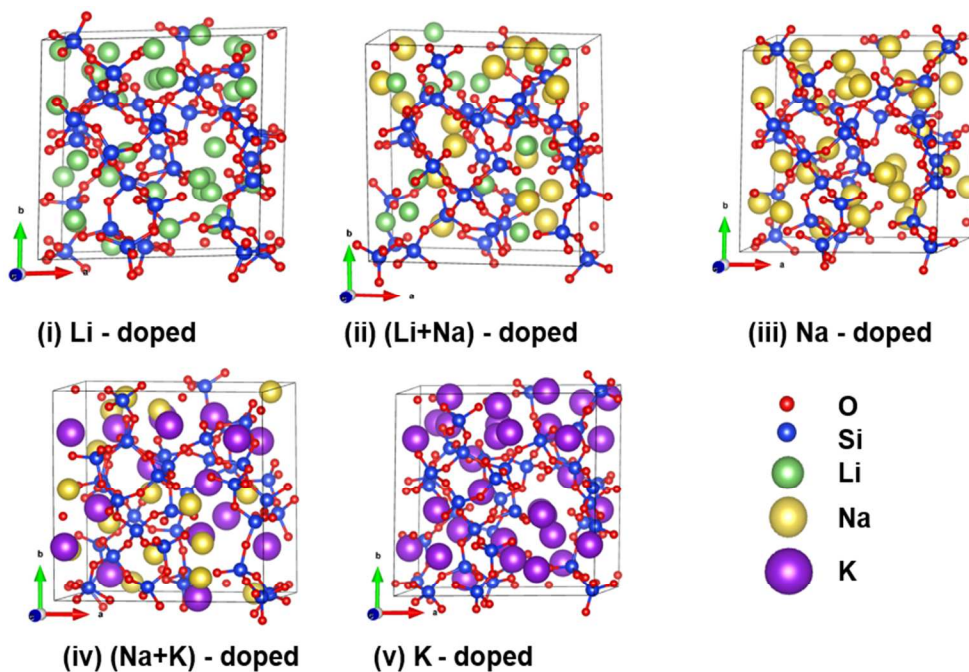


Figure S1: Ball and stick figures of 30 mol. % alkali oxide doped silicate glasses.

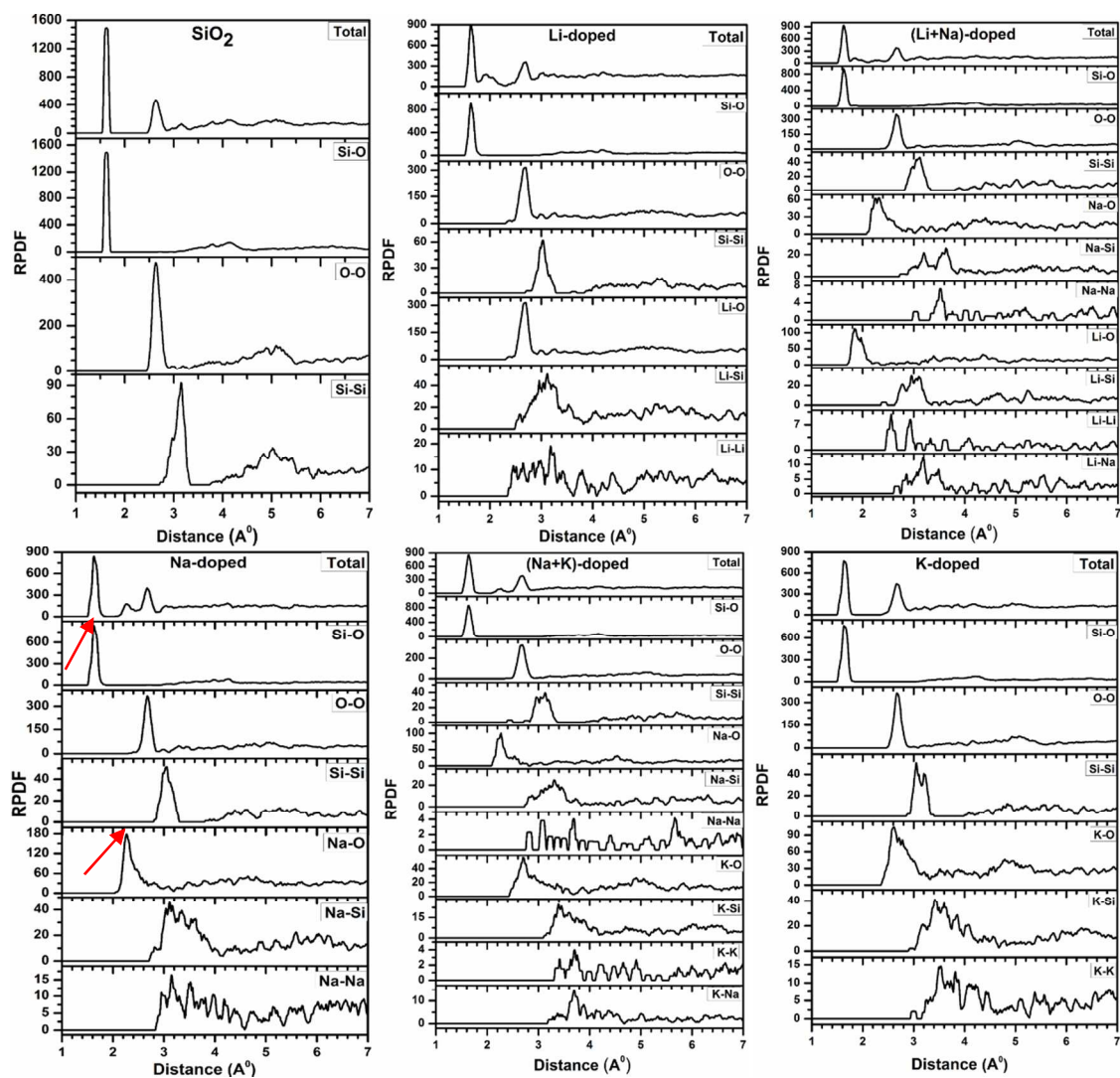


Figure S2: Radial pair distribution functions for 30 mol. % alkali oxide doped silicate glasses. The arrow in the Na-doped panel indicates the position of the experimental (33.33 mol. %) peak from ref. [28].

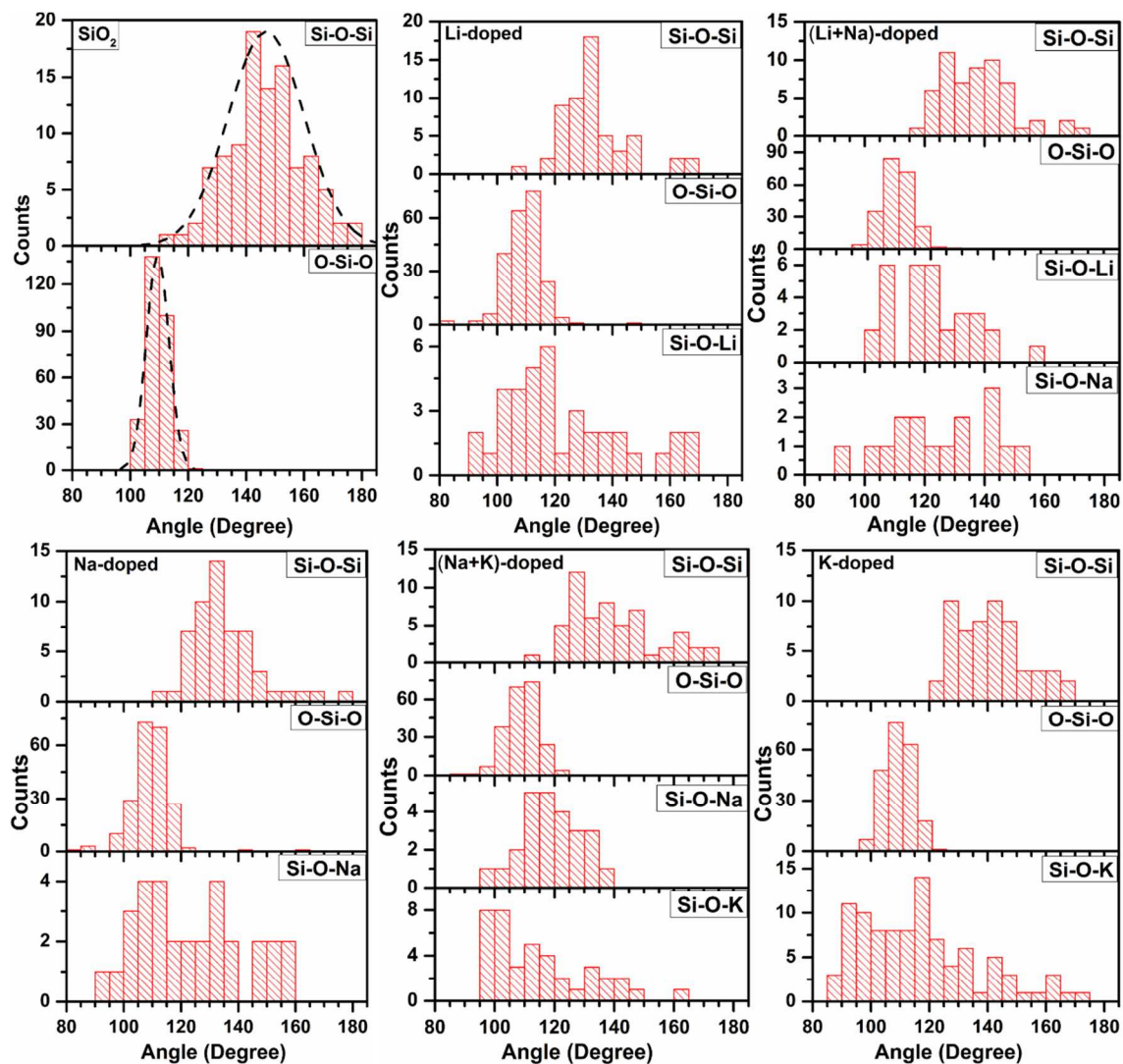


Figure S3: Bond angle distribution for 30 mol. % alkali oxide mixed silicate glasses.

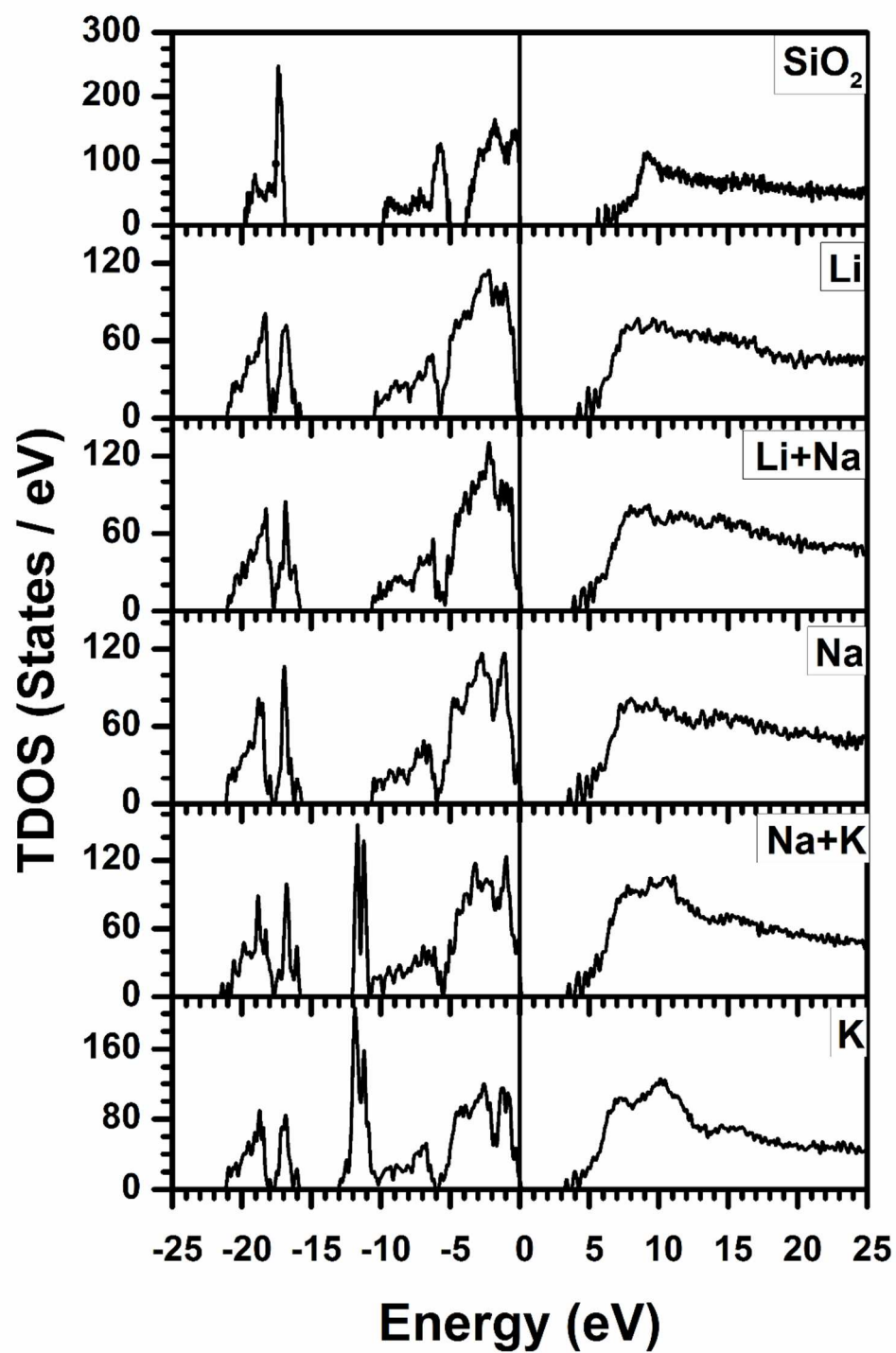


Figure S4: Electronic total density of states for 30 mol. % alkali oxide added silicate glasses.

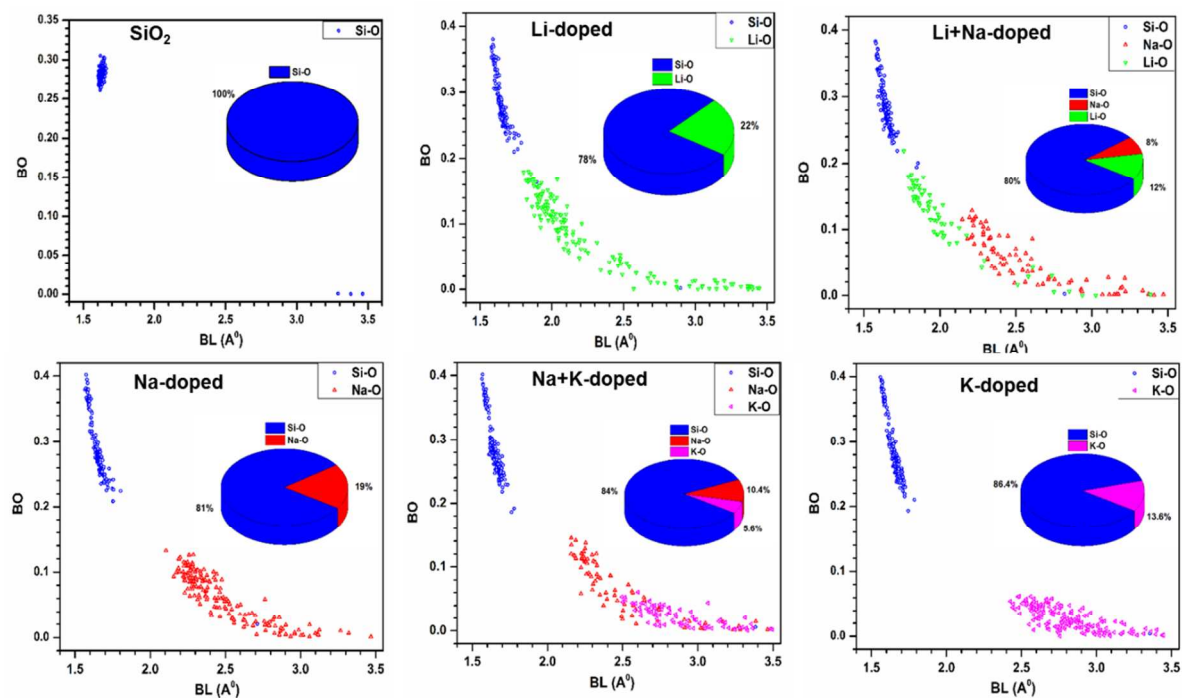


Figure S5: Distribution of bond order versus bond length for 30 mol. % alkali oxide doped glasses.

Inset shows the contribution form different types of bonds.

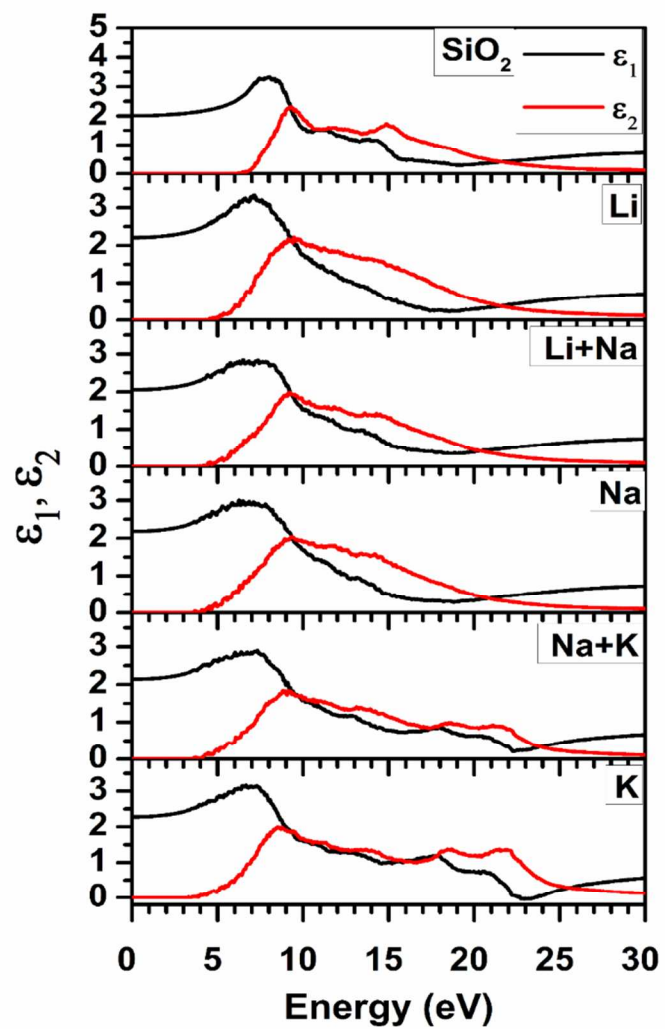


Figure S6: Calculated dielectric constants for the 30 mol. % alkali oxide doped glasses.