

## Supporting Information

### New 4V-class and zero-strain cathode material for Na ion batteries

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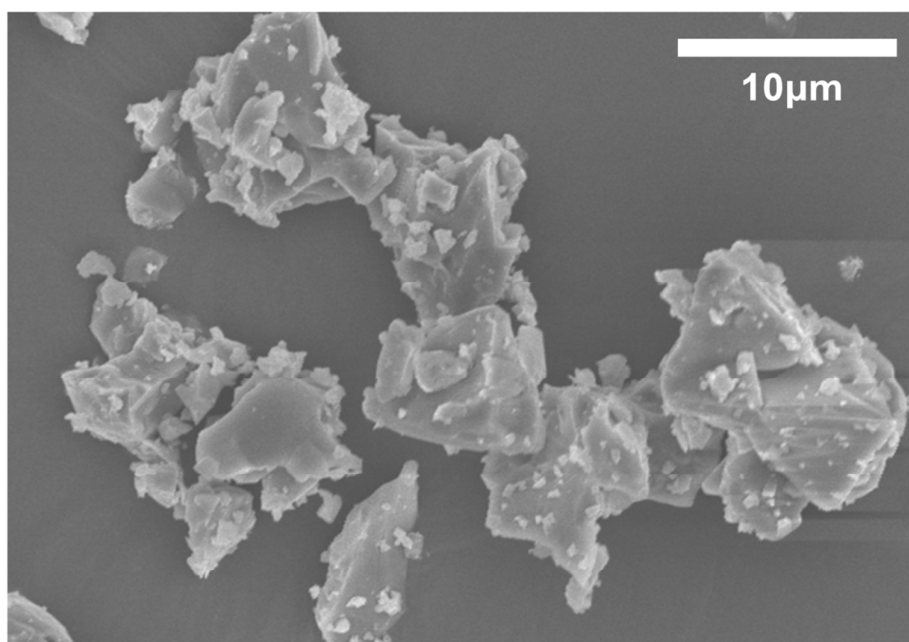
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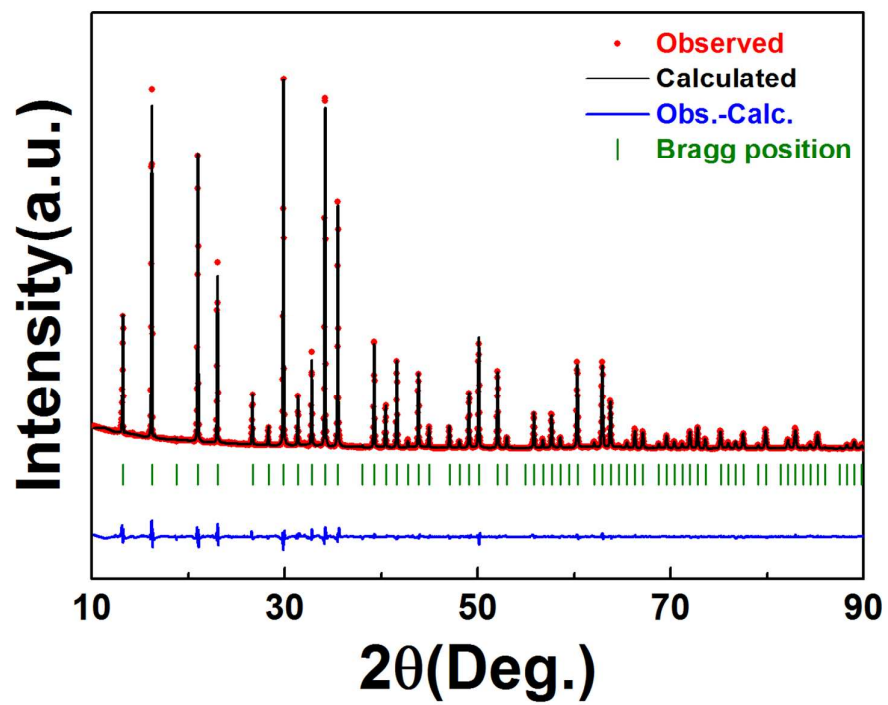
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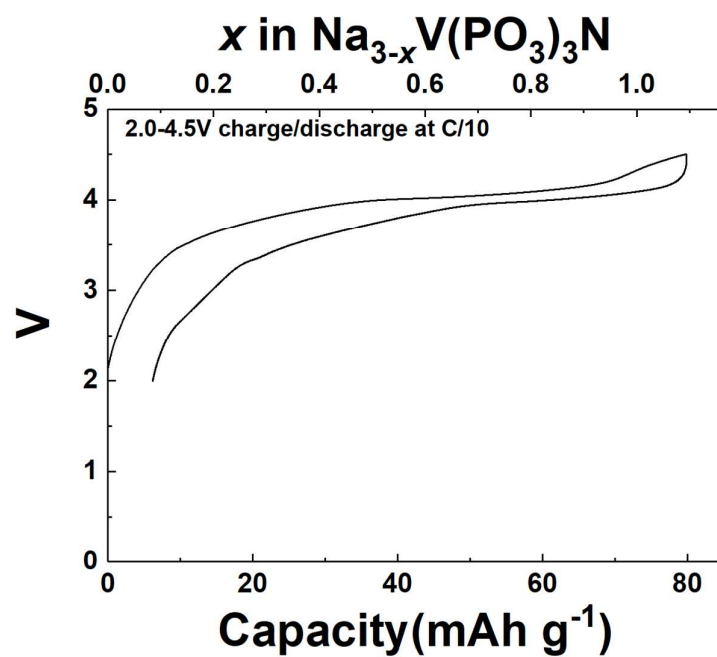
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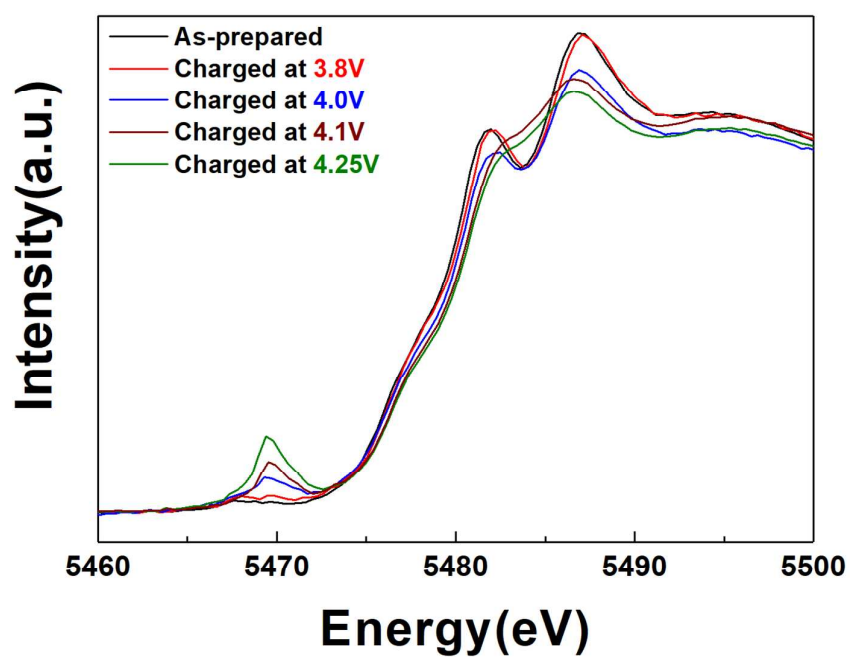
**Supporting Figure S1** SEM image of Na<sub>3</sub>V(PO<sub>3</sub>)<sub>3</sub>N.



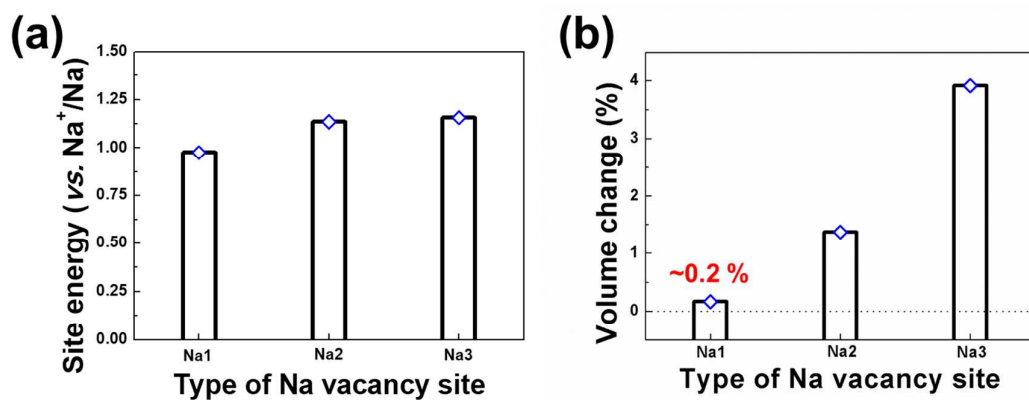
**Supporting Figure S2** Refined XRD pattern of  $\text{Na}_3\text{V}(\text{PO}_3)_3\text{N}$  ( $R_p = 7.52\%$ ,  $R_I = 8.53\%$ ,  $R_F = 7.39\%$ ,  $\chi^2 = 8.78\%$ ).



Supporting Figure S3 Charge/discharge curve of  $\text{Na}_3\text{V}(\text{PO}_3)_3\text{N}$  from 2.0-4.5V at C/10



**Supporting Figure S4** *Ex-situ* XANES spectra of  $\text{Na}_{3-x}\text{V}(\text{PO}_3)_3\text{N}$  samples with various Na amounts in the structure



**Supporting Figure S5** (a) Site energy of a single Na vacancy from  $\text{Na}_3\text{V}(\text{PO}_3)_3\text{N}$  plotted as a type of Na site. (b) Volume change of the  $\text{Na}_3\text{V}(\text{PO}_3)_3\text{N}$  structure upon the removal of all Na ions in each Na site.

| Atom | Multiplicity | x         | y          | z         | B <sub>iso</sub> | Occupancy |
|------|--------------|-----------|------------|-----------|------------------|-----------|
| P1   | 12           | 0.3327(5) | 0.0826(4)  | 0.2444(5) | 0.95(6)          | 1         |
| V1   | 4            | 0.0798(3) | -0.0798(3) | 0.4202(3) | 0.2(13)          | 1         |
| Na1  | 4            | 0.0094(6) | 0.0094(6)  | 0.0094(6) | 1.56(10)         | 0.989(7)  |
| Na2  | 4            | 0.3901(8) | 0.3901(8)  | 0.3901(8) | 1.56(10)         | 0.997(3)  |
| Na3  | 4            | 0.7022(7) | 0.2022(7)  | 0.2978(7) | 1.56(10)         | 0.998(2)  |
| O1   | 12           | 0.2693(4) | -0.0261(4) | 0.3488(4) | 1.03(5)          | 1         |
| O2   | 12           | 0.3711(4) | 0.0009(4)  | 0.1112(4) | 0.92(4)          | 1         |
| O3   | 12           | 0.4506(3) | 0.1671(4)  | 0.3065(4) | 1.19(5)          | 1         |
| N1   | 4            | 0.1956(3) | 0.1956(3)  | 0.1956(3) | 0.88(5)          | 1         |

***a* = 9.44905(7) Å**

**Supporting Table T1** Atomic information of Na<sub>3</sub>V(PO<sub>3</sub>)<sub>3</sub>N.

|                              | <b>Na<sub>3</sub>V(PO<sub>3</sub>)<sub>3</sub>N</b> | <b>Na<sub>2</sub>V(PO<sub>3</sub>)<sub>3</sub>N</b> |
|------------------------------|-----------------------------------------------------|-----------------------------------------------------|
| <b>Na1-O<sub>6</sub> (Å)</b> | 2.60 (3), 2.52 (3)                                  | 2.80 (3), 2.58 (3)                                  |
| <b>V-O<sub>6</sub> (Å)</b>   | 2.04 (6)                                            | 1.96 (6)                                            |

**Supporting Table T2** Na1-O and V-O Bond lengths of Na<sub>3</sub>V(PO<sub>3</sub>)<sub>3</sub>N and Na<sub>2</sub>V(PO<sub>3</sub>)<sub>3</sub>N.



|           | <b>Na<sub>3</sub>V(PO<sub>3</sub>)<sub>3</sub>N</b><br>(x, y, z) (Å) | <b>Na<sub>2</sub>V(PO<sub>3</sub>)<sub>3</sub>N</b><br>(x, y, z) (Å) | <b>Displacement</b><br>(x, y, z) (Å) |
|-----------|----------------------------------------------------------------------|----------------------------------------------------------------------|--------------------------------------|
| <b>V1</b> | (0.761, 8.832, 4.035)                                                | (0.796, 8.802, 4.003)                                                | (0.035, -0.030, -0.032)              |
| <b>V2</b> | (4.035, 0.761, 8.832)                                                | (4.003, 0.796, 8.802)                                                | (-0.032, 0.035, -0.030)              |
| <b>V3</b> | (8.832, 4.035, 0.761)                                                | (8.802, 4.003, 0.796)                                                | (-0.030, -0.032, 0.035)              |
| <b>V4</b> | (5.557, 5.557, 5.557)                                                | (5.594, 5.594, 5.594)                                                | (0.038, 0.038, 0.038)                |

**Supporting Table T3** Cartesian coordinates of four V ions in the unit cell of Na<sub>3</sub>V(PO<sub>3</sub>)<sub>3</sub>N and Na<sub>2</sub>V(PO<sub>3</sub>)<sub>3</sub>N. The displacements vectors of V ions upon desodiation is also tabulated. Due to the symmetry of Na<sub>3</sub>V(PO<sub>3</sub>)<sub>3</sub>N and desodiated Na<sub>2</sub>V(PO<sub>3</sub>)<sub>3</sub>N, the direction of V ion movement cancels out each other, resulting in the negligible vector sum of net V ion displacements.

