

Supporting Information

Microporous Metal-Organic Framework Based on Ligand-Truncation Strategy with High Performance for Gas Adsorption and Separation

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Supplementary Figures

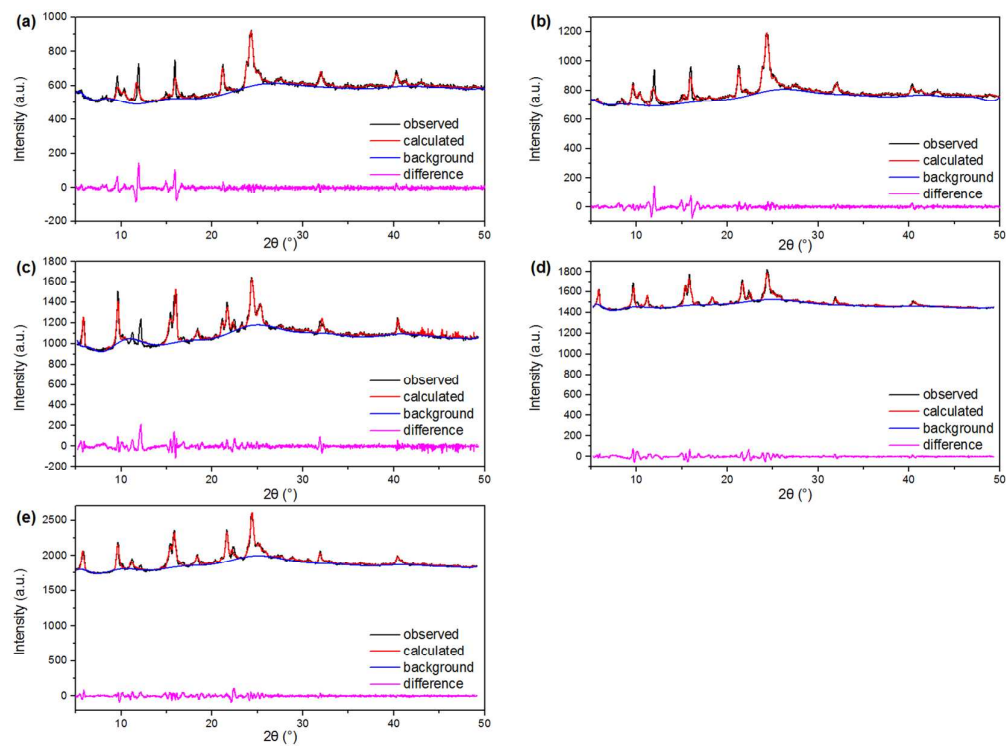


Figure S1. LeBail method fit to PXRD data for the samples of **1**. (a) as-synthesized, (b) activated, (c) after gas desorption, (d) immersed in CH_2Cl_2 (e) immersed in CH_3OH .

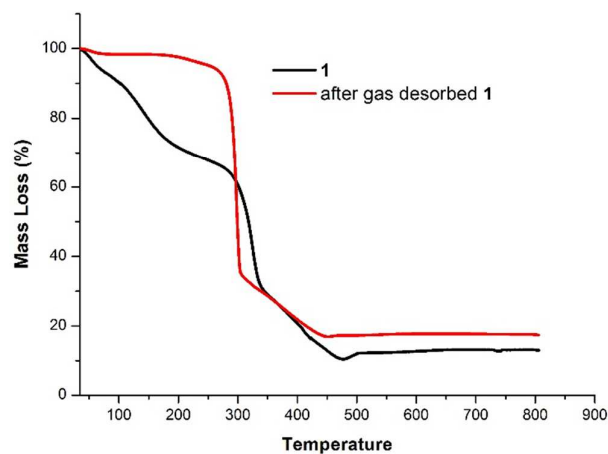


Figure S2. View of the TGA for as-synthesized and gas desorbed samples in **1**.

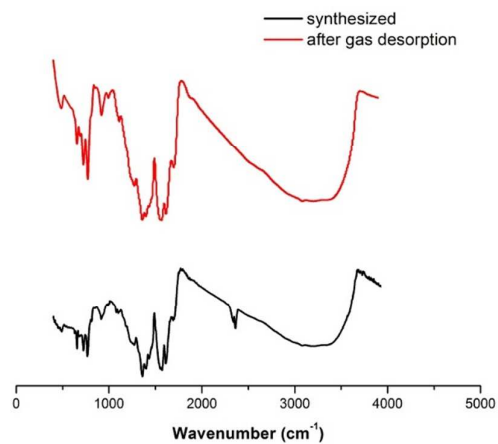


Figure S3. View of the IR for as-synthesized and gas desorbed samples in **1**.

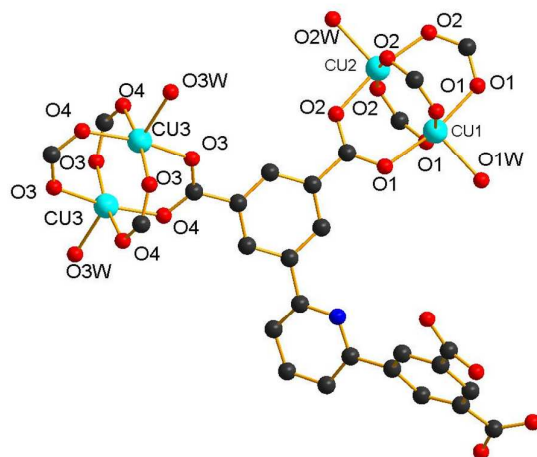


Figure S4. View of coordination environment of metal ions and ligands in **1**.

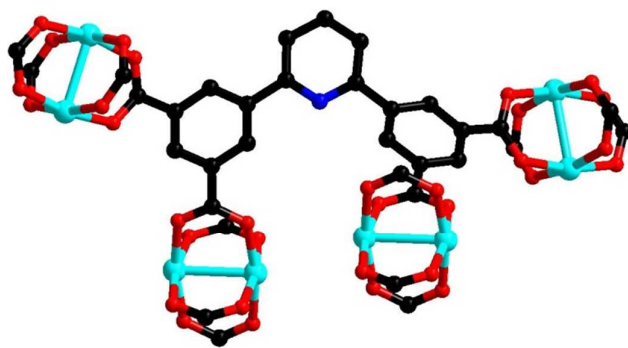


Figure S5. The L^4 ligands can be regarded as 4-c node linked by Cu_2 paddlewheel units

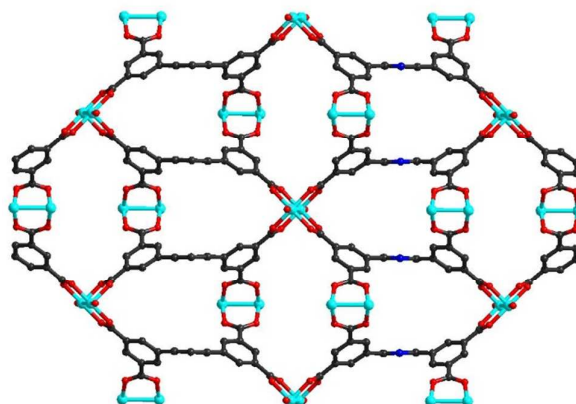


Figure S6. Two linked cages in the 3D network of **1**.

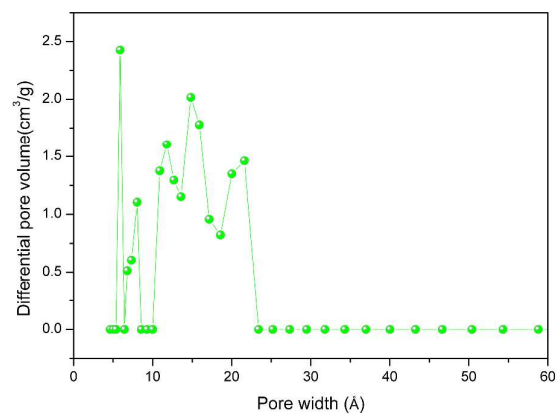


Figure S7. Pore size distributions analyzed by NLDFT methods.

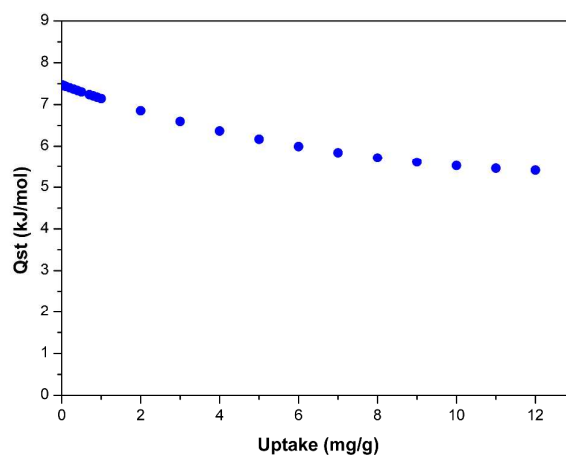


Figure S8. The Q_{st} of hydrogen for **1**

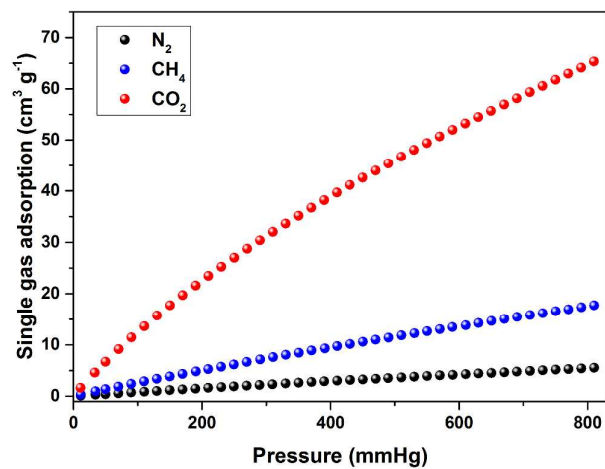


Figure S9. The single gas adsorption at 293 K

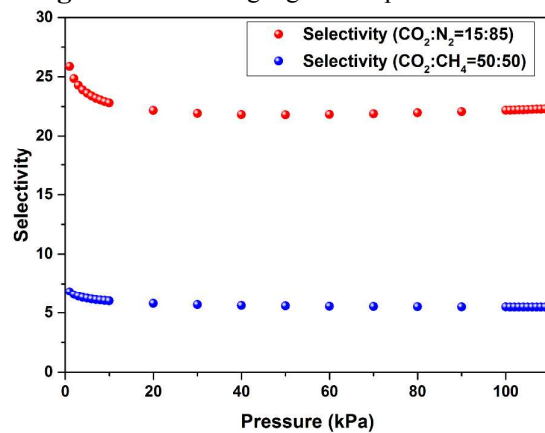


Figure S10. Adsorption selectivity of CO₂ over CH₄ or N₂ at 293 K

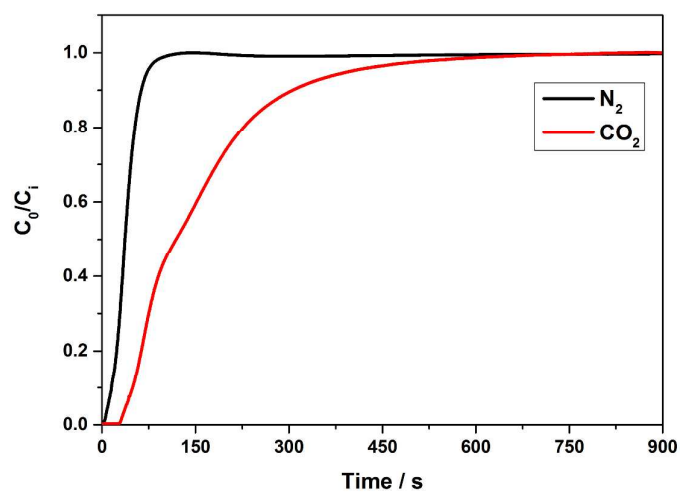


Figure S11. Breakthrough curves of **1** at 293 K and 1 bar

Table S1. Crystal data and structure refinement information for compound **1**

Crystal system	Cubic
Space group	orthorhombic
<i>a</i> , Å	13.8346(8)
<i>b</i> , Å	20.5675(11)
<i>c</i> , Å	30.1981(16)
<i>V</i> , Å ³	8592.7(8)
<i>Z</i>	8
<i>D</i> _{calcd} , g/cm ³	1.100
<i>μ</i> , mm ⁻¹	1.340
<i>F</i> (000)	3616
<i>θ</i> Range, deg	2.2287- 25.5585
Reflection collected	14531
Independent reflections (<i>R</i> _{int})	0.0918
Reflections with <i>I</i> > 2σ(<i>I</i>)	4072
<i>R</i> ₁ , <i>wR</i> ₂ (<i>I</i> > 2σ(<i>I</i>))*	0.0528, 0.1474
<i>R</i> ₁ , <i>wR</i> ₂ (all data)**	0.0854, 0.1615

$$* R = \sum(F_o - F_c)/\sum(F_o), ** wR_2 = \{\sum[w(F_o^2 - F_c^2)]/\sum(F_o^2)\}^{1/2}.$$

Table S2. Selected bond distances (Å) and angles (°) of structure **1**

Cu1-O1 ^{#1}	1.968(3)	Cu1-O1w ^{#2}	2.139(6)
Cu1-Cu2	2.6672(14)	Cu2-O2 ^{#3}	1.973(3)
Cu2-O2w	2.096(6)	Cu3-O4	1.956(3)
Cu3-O3 ^{#4}	1.976(3)	Cu3-Cu3	2.6559(16)
O1-Cu1-O1	91.0(2)	O1-Cu1-O1w	95.53(9)
O2-Cu2-O2	90.3(2)	O2-Cu2-O2w	96.13(9)
O2w-Cu2-Cu1	180	O4-Cu3-O4	87.57(19)
O4-Cu3-O3	168.26(15)	O3-Cu3-O3	90.21(19)
O4-Cu3-O3w	97.13(16)	O3-Cu3-O3w	94.56(15)

#1: -x, -y+1/2, z; #2: -x, y+1/2, -z; #3: x, -y, -z; #4: x+1/2, y+1/2, z+1/2

Table S3. The pattern matching analysis data for **1** by the LeBail route

	As-synthesized	Dehydrated	After gas adsorption	Immersed in CH ₂ Cl ₂	Immersed in CH ₃ OH
a (Å)	13.9158	13.8691	13.8376	13.7962	13.8631
b (Å)	20.7448	20.7278	20.6047	20.2590	20.6908
c (Å)	30.2311	30.2448	30.1585	30.0488	30.4623
α (°)	90	90	90	90	90
β (°)	90	90	90	90	90
γ (°)	90	90	90	90	90
V (Å ³)	8727.10	8694.68	8598.78	8398.59	8737.78
R _p	1.338	0.999	1.349	0.510	0.621
R _{wp}	2.389	1.655	2.133	0.816	0.935
R _{exp}	4.098	3.556	2.996	2.576	2.274
χ^2	0.340	0.217	0.507	0.100	0.169