
Supporting Information for Concise Synthesis of Key Intermediates of Pyriftalid and Paquinimod via Hydrogenation Method

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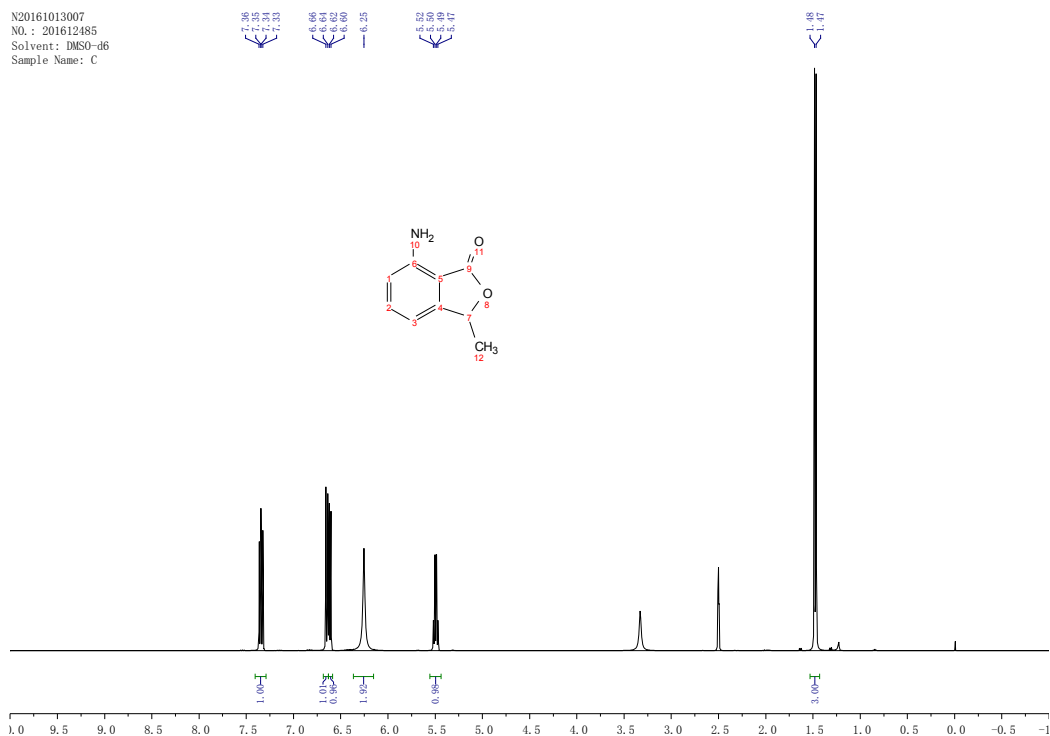
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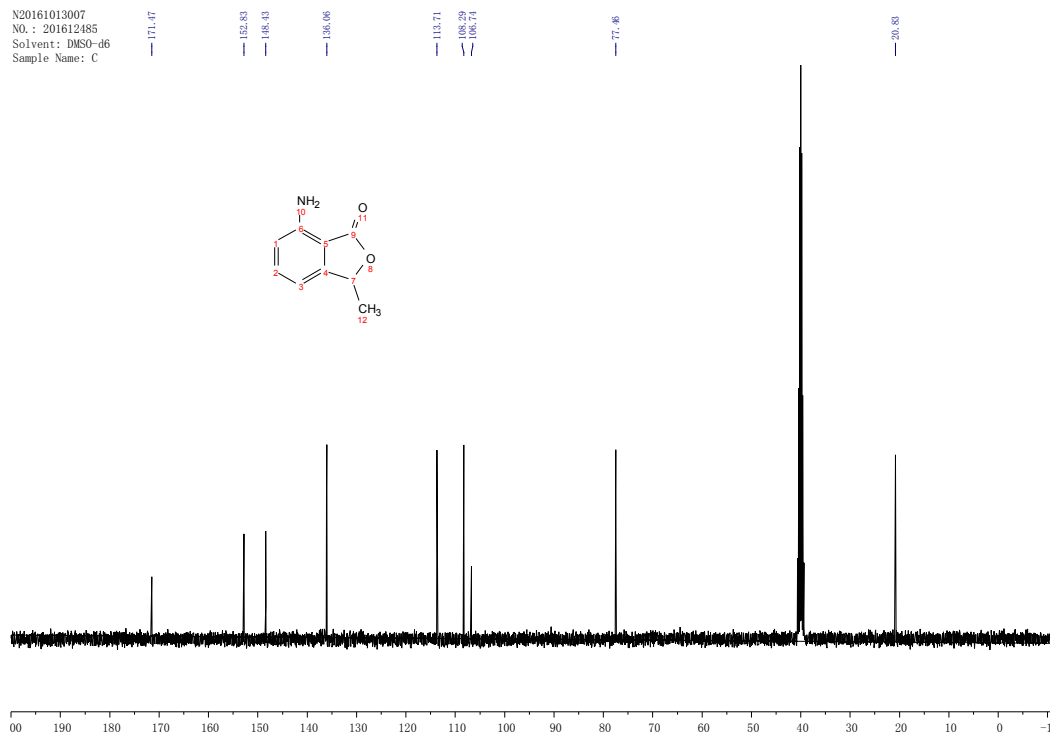
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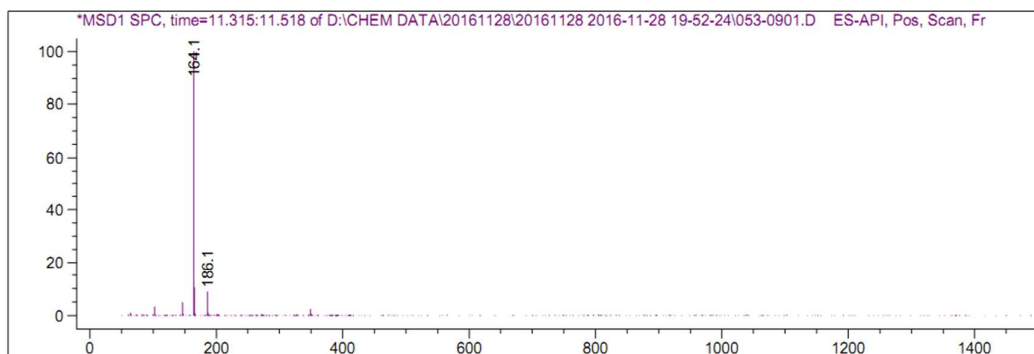
1. ¹ H NMR, ¹³ C NMR, MS data of 1	S2-S3
2. ¹ H NMR, ¹³ C NMR, MS data of 2	S3-S4
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N20161013007
 NO.: 201612485
 Solvent: DMSO-d6
 Sample Name: C



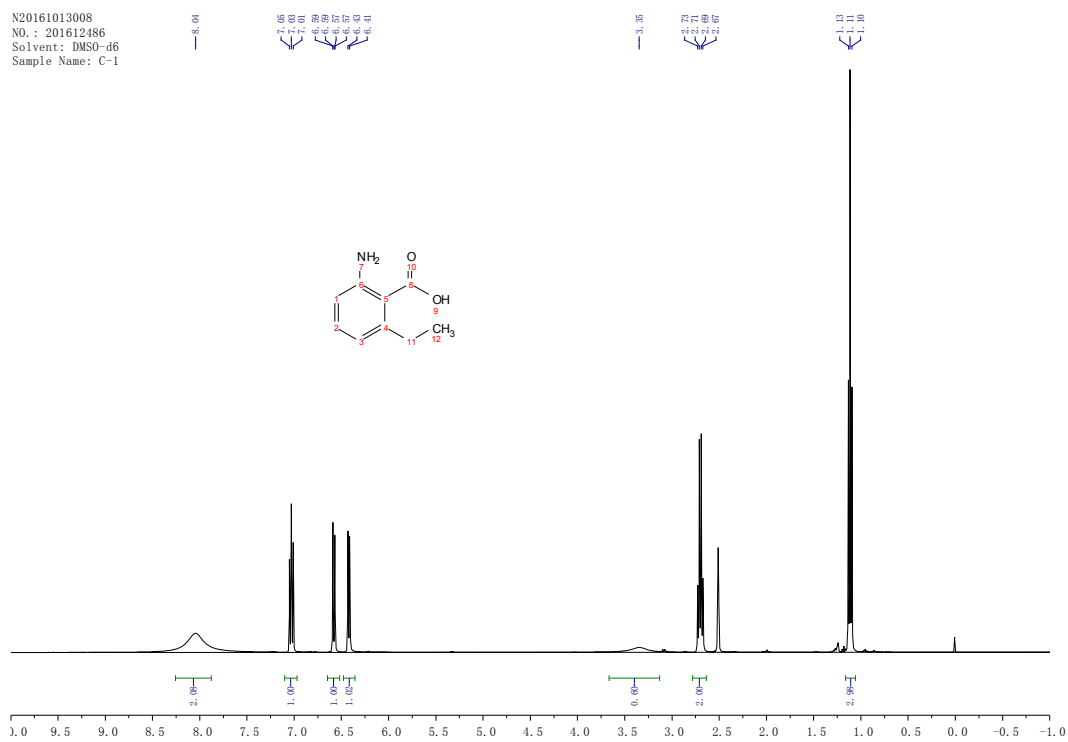
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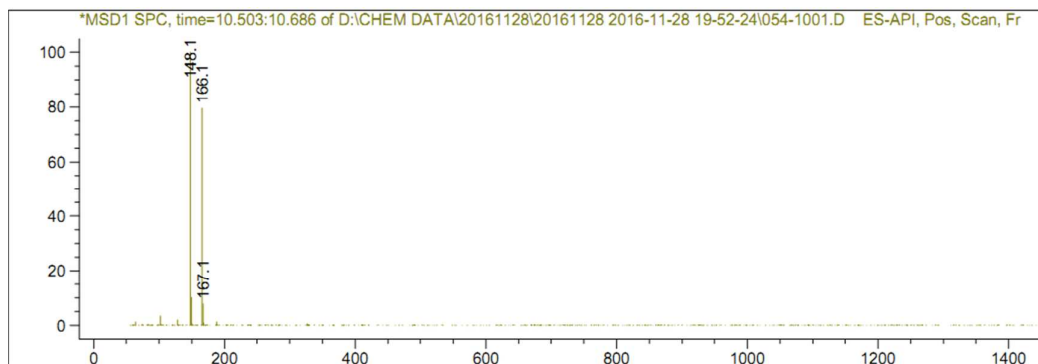
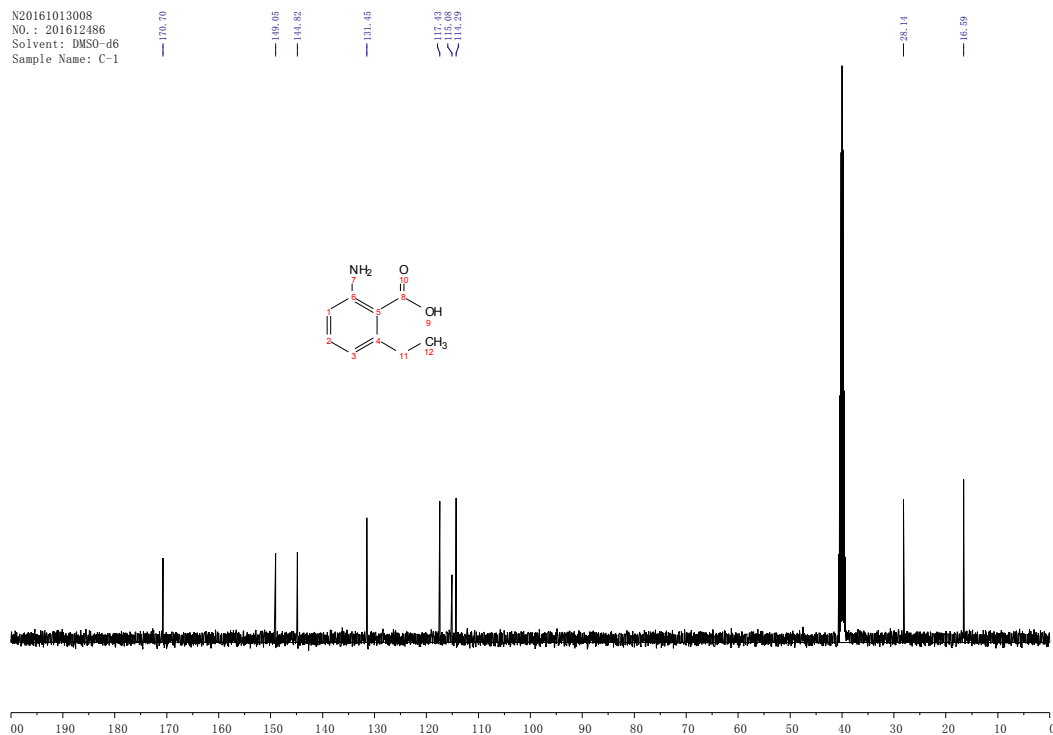
MS spectrum of compound 1

N20161013008
NO.: 201612486
Solvent: DMSO-d6
Sample Name: C-1

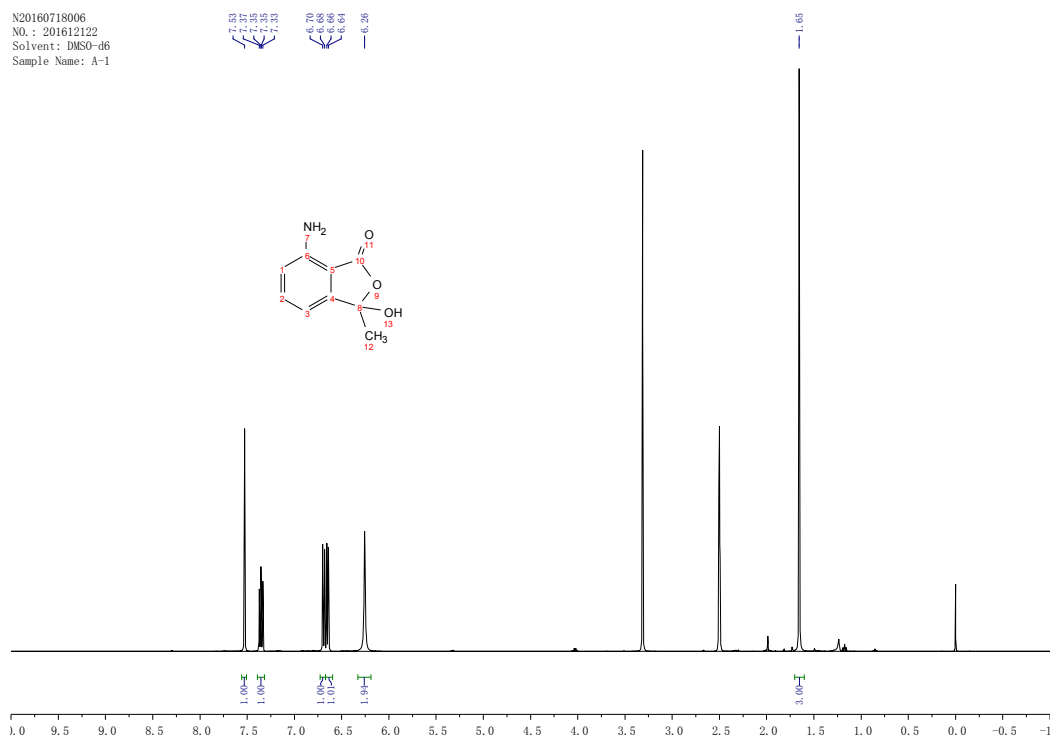


¹H NMR spectrum of compound 2

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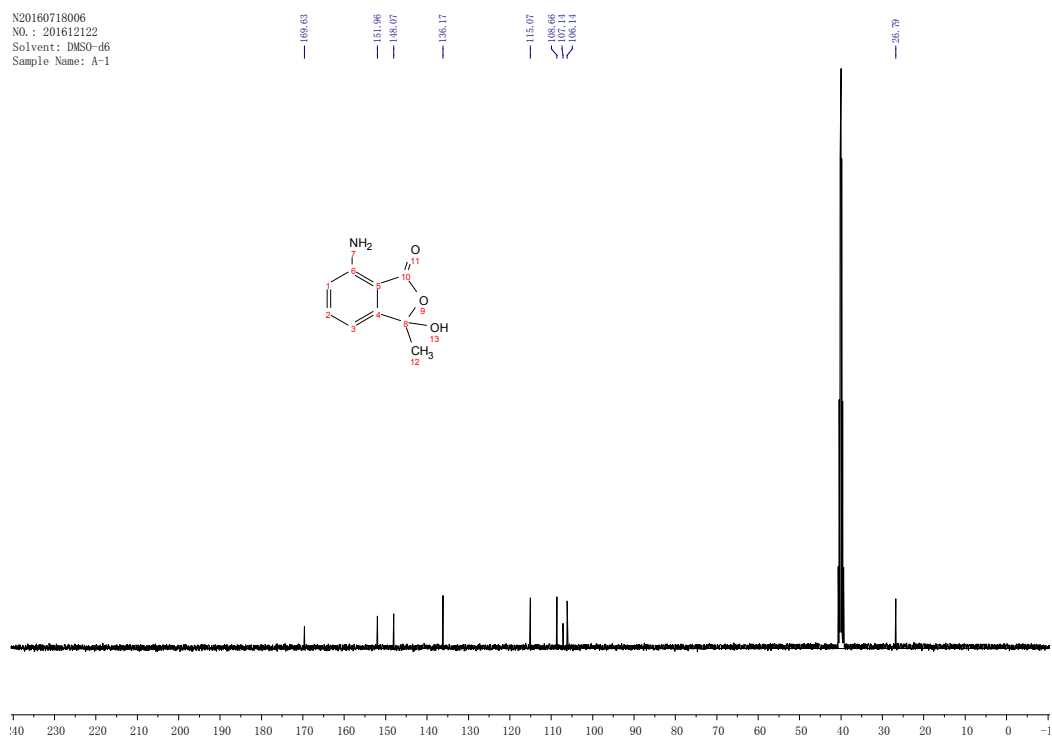


N20160718006
 NO.: 201612122
 Solvent: DMSO-d6
 Sample Name: A-1

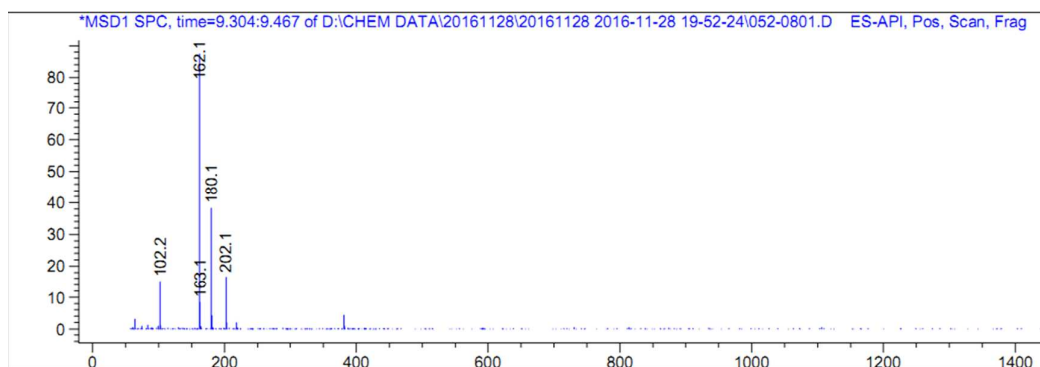


¹H NMR spectrum of compound 4

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 Sample Name: A-1

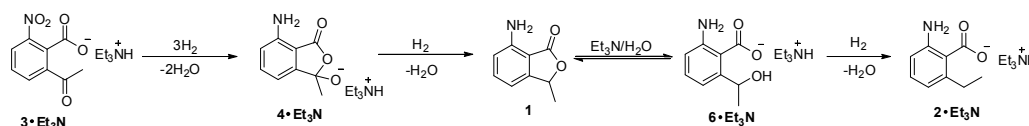


¹³C NMR spectrum of compound 4



MS spectrum of compound **4**

Reaction process data about Raney-Ni for Reduction of Triethylammonium salt of **3** to prepare **2**.



^aReaction Process detected by HPLC.

^b Time (h)	HPLC (%) ^c				
	3	4	1	6	2
1	26.17	54.90	10.98	1.70	/
2	ND	0.43	71.52	12.69	14.64
4	ND	ND	34.59	1.03	63.85
6	ND	ND	10.84	0.54	87.67
7	ND	ND	6.90	1.19	91.69

^a Reaction conditions: triethylammonium salt of **3** (6.20 g), Et₃N (4.04 g), Raney-Ni (1.00 g), H₂O (45 mL), 120°C (Bath temperature), 15-20atm;

^b The reaction was collected data by hour;

^c Compositions of reaction mixture were calculated from HPLC area.

As showed in the table above, the nitro reduction intermediates **4** was observed as the main product in 1h; compound **1** was the main product in 2h; as time prolonged, **2** became the main product (4h, 6h, 7h). The nitro reduction is faster compared to the following steps, and the reaction steps proceed mostly by sequence.