

**CO<sub>2</sub> activation and synthesis of cyclic carbonates and quinazoline-2,4(1H,3H)-dione over amine and ionic liquids functionalized basic Nano-ZSM-5 catalysts**

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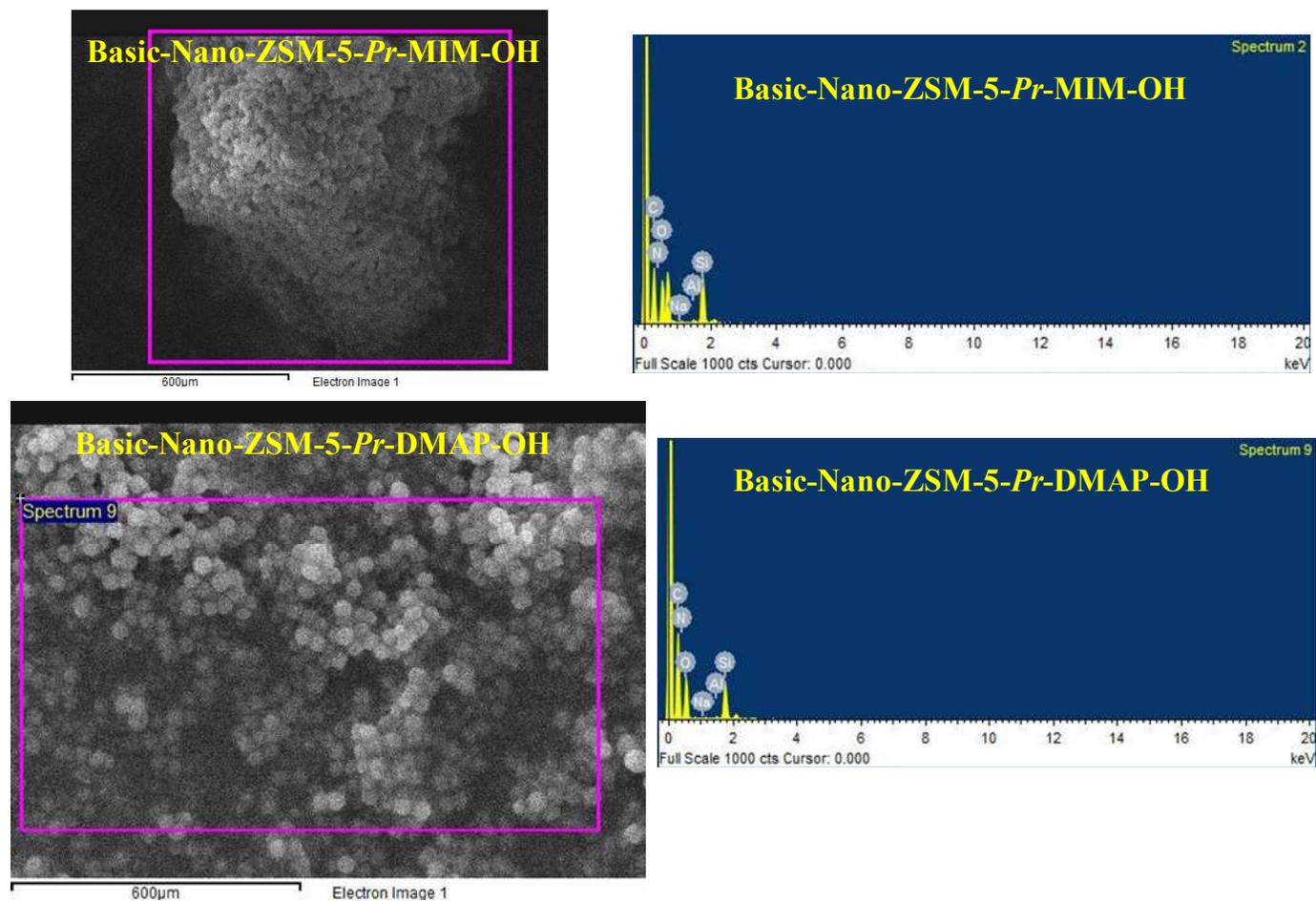
## Chemicals and materials

AR grade chemicals were purchased and used without any purification. Tetraethyl orthosilicate (TEOS, 98%), tetrapropyl ammonium hydroxide (TPAOH, 40% aqueous solution), sodium aluminates (53 % Al<sub>2</sub>O<sub>3</sub>, 43% Na<sub>2</sub>O), *n*-propyltriethoxy silane, organic substrates used in the synthesis and catalytic reactions were procured from Sigma-Aldrich, India. Ammonia (28-30%) and solvents were obtained from Merck India Pvt. Ltd.

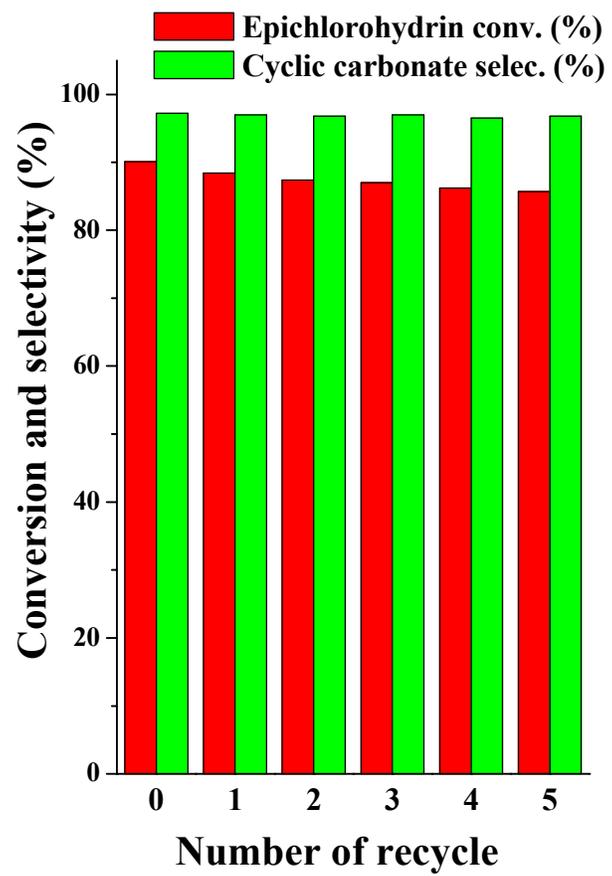
## Catalysts characterization

X-ray diffraction (XRD) patterns were recorded in the  $2\theta$  range of 5–80° with a scan speed of 2°/min on a PANalytical X'PERT PRO diffractometer using Cu K $\alpha$  radiation ( $\lambda=0.1542$  nm, 40 kV, 40 mA). Nitrogen adsorption measurements were performed at 77 K by Quantachrome Instruments, Autosorb-IQ volumetric adsorption analyzer. The material was degassed at 423 K in the degas port of the adsorption apparatus. The specific surface area of the material was calculated from the desorption data points obtained at  $P/P_0$  between 0.05–0.3 using the Brunauer-Emmett-Teller (BET) equation. The pore diameter was estimated using Barrett-Joyner-Halenda (BJH) and NLDFT methods. Scanning electron microscopy (SEM) measurements were carried out on a JEOL JSM-6610LV to investigate the morphology of the zeolites. The nanostructure was investigated using high-resolution transmission electron microscope (TECHNAI, FEI-G2) at an accelerating voltage of 300 kV at RSIC Centre, IIT Bombay. The sample was dispersed in ethanol using an ultrasonic bath, and dispersed sample was mounted on a carbon-coated Cu grid, dried, and used for TEM measurement. Thermogravimetric analysis (TGA) was performed on a TGA/DSC 1 STAR<sup>e</sup> SYSTEM from Mettler Toledo instrument with temperature increments of 10 K/min in air stream from ambient temperature (300 K) to 1273 K. Catalysts were also characterized by Fourier transform infrared (FT-IR) spectroscopy on Bruker FT-IR instrument. Solid-state NMR was carried out with <sup>29</sup>Si and <sup>13</sup>C frequencies of 79.4 and 100.5 MHz, respectively, on a 4 mm MAS probe using 400 MHz ECX JEOL NMR spectrometer. <sup>29</sup>Si cross-polarization magic-angle spinning (CPMAS)

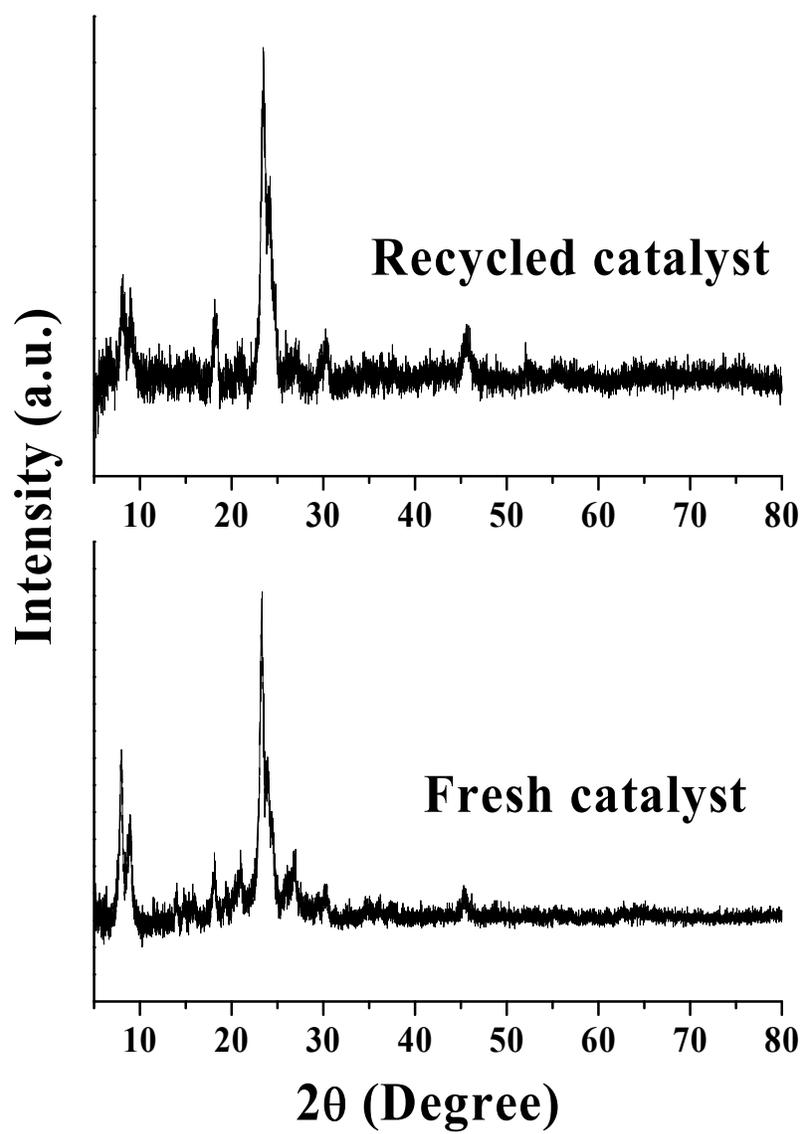
spectrum was measured at room temperature with the following condition: magic-angle spinning at 8 kHz; contact time of 5 ms and a repetition delay of 4.05s; 16454 scans (18 h) which was referenced to tetramethylsilane.  $^{13}\text{C}$  CPMAS spectrum was measured with a contact time of 1ms with recycle delay of 4s, 10000 scans (11 h). Finally, temperature-programmed desorption (TPD) experiments were conducted on a Quantachrome ChemBET<sup>TM</sup> TPR/TPD instrument. For  $\text{CO}_2$ -TPD, initially, the sample was pre-treated in He (50 mL/min) at 873 K for 1 h. After cooling to 323 K, carbon dioxide (partial pressure 100 Torr) was passed through the samples for 1 h. Then, the sample was subsequently flushed by He stream (50 mL/min) at 323 K for 1 h to remove physically adsorbed  $\text{CO}_2$ . TPD experiments were carried out in the range of 323-573 K at a heating rate of 10 K/min.



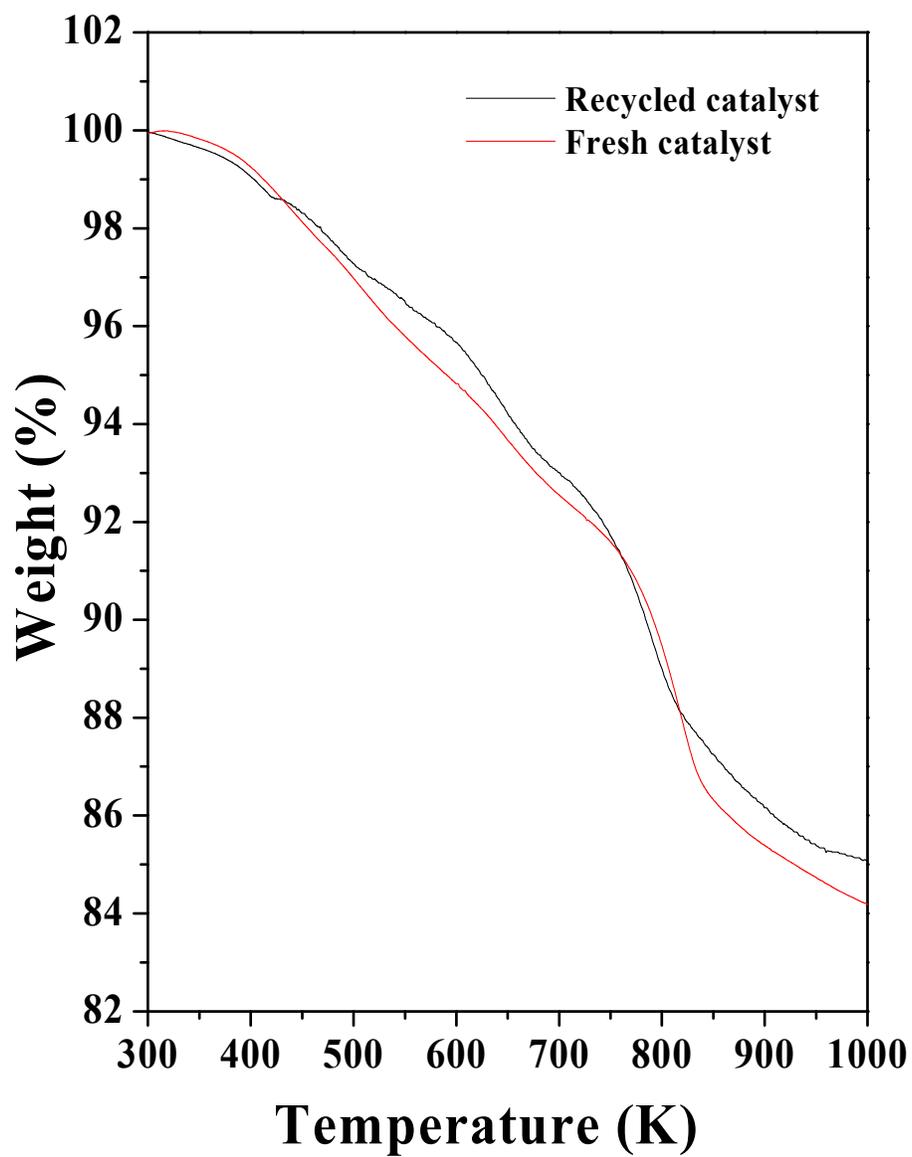
**Figure S1.** EDAX spectra of Basic-Nano-ZSM-5-Pr-DMAP-OH and Basic-Nano-ZSM-5-Pr-MIM-OH.



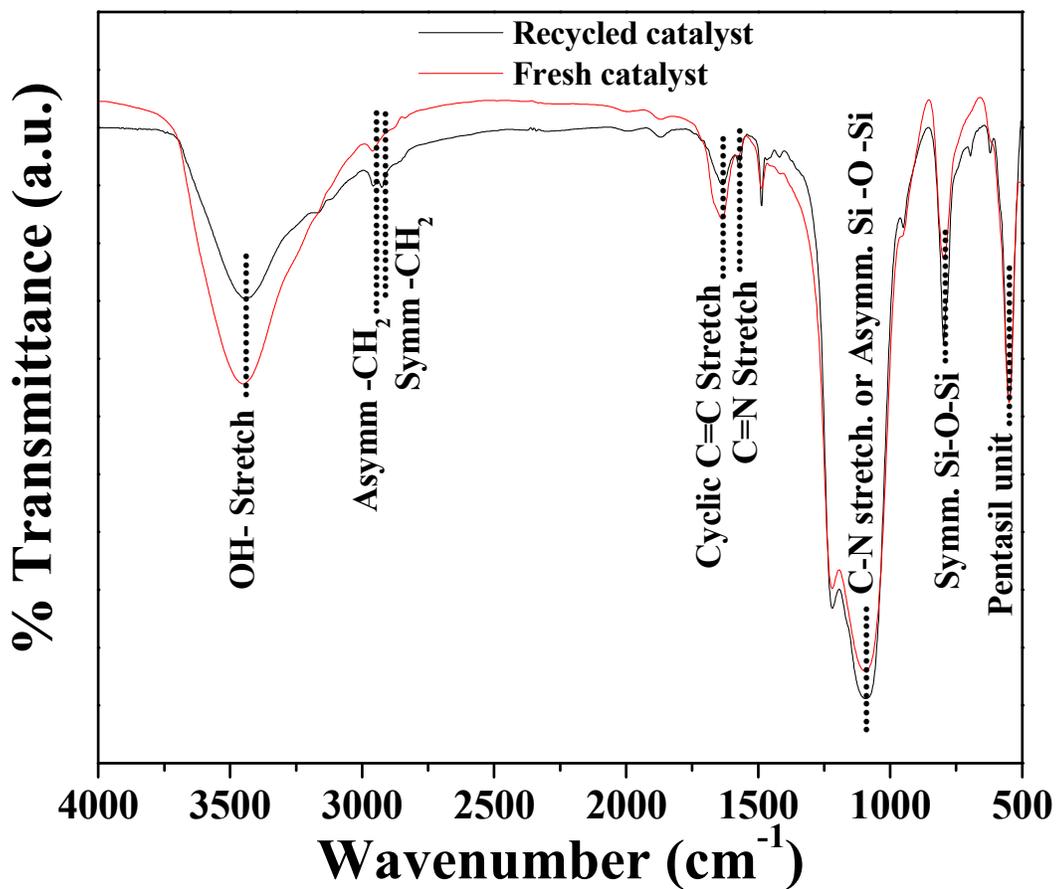
**Figure S2.** Recycling data obtained by the reaction between CO<sub>2</sub> and epichlorohydrin using Basic-Nano-ZSM-5-*Pr*-MIM-OH.



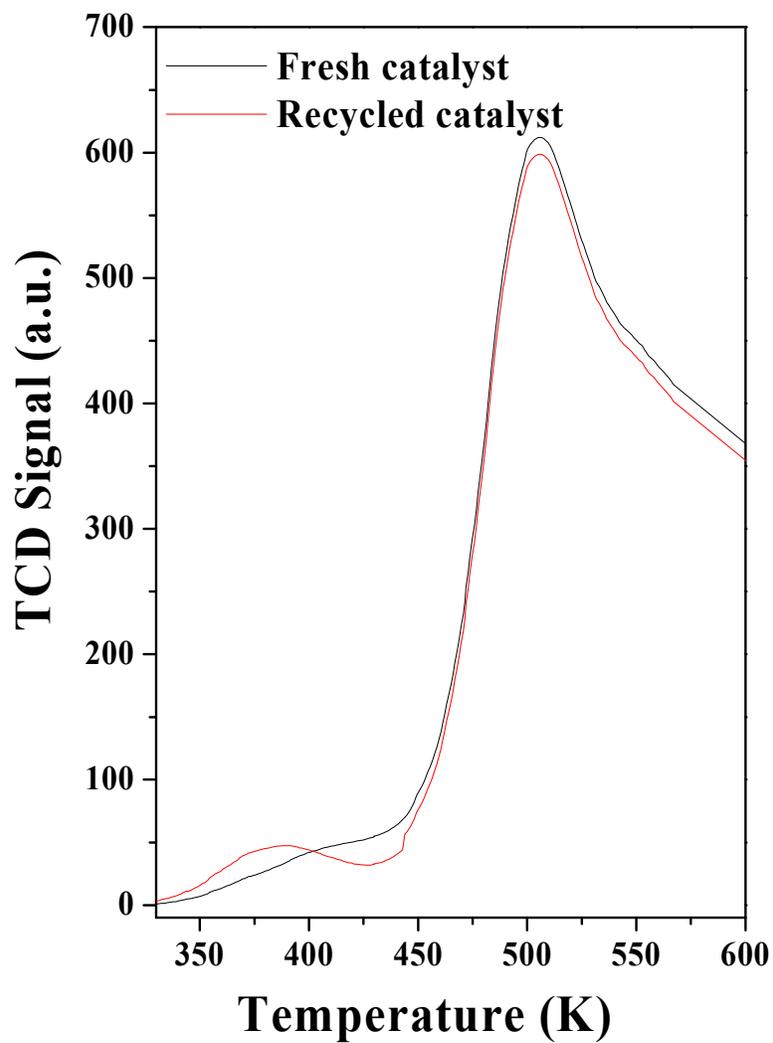
**Figure S3.** XRD pattern of fresh and recycled Basic-Nano-ZSM-5-*Pr*-MIM-OH obtained after 5<sup>th</sup> recycle.



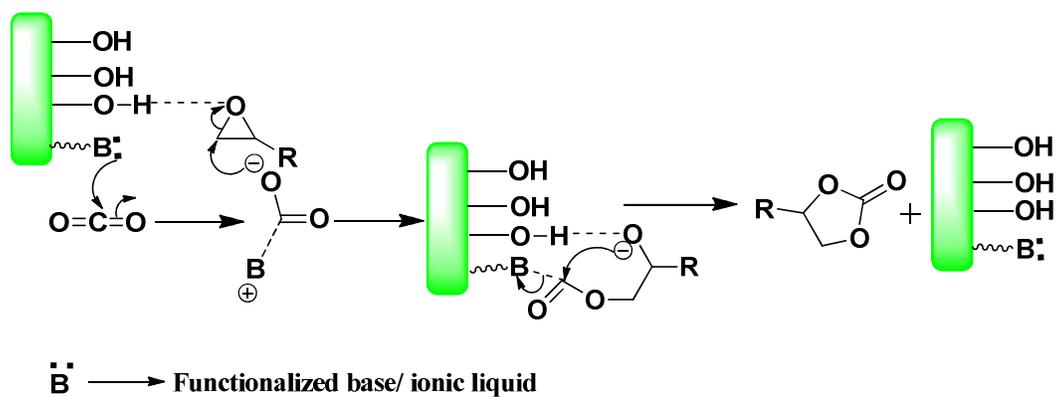
**Figure S4.** Thermograms of fresh and recycled Basic-Nano-ZSM-5-Pr-MIM-OH obtained after 5<sup>th</sup> recycle.



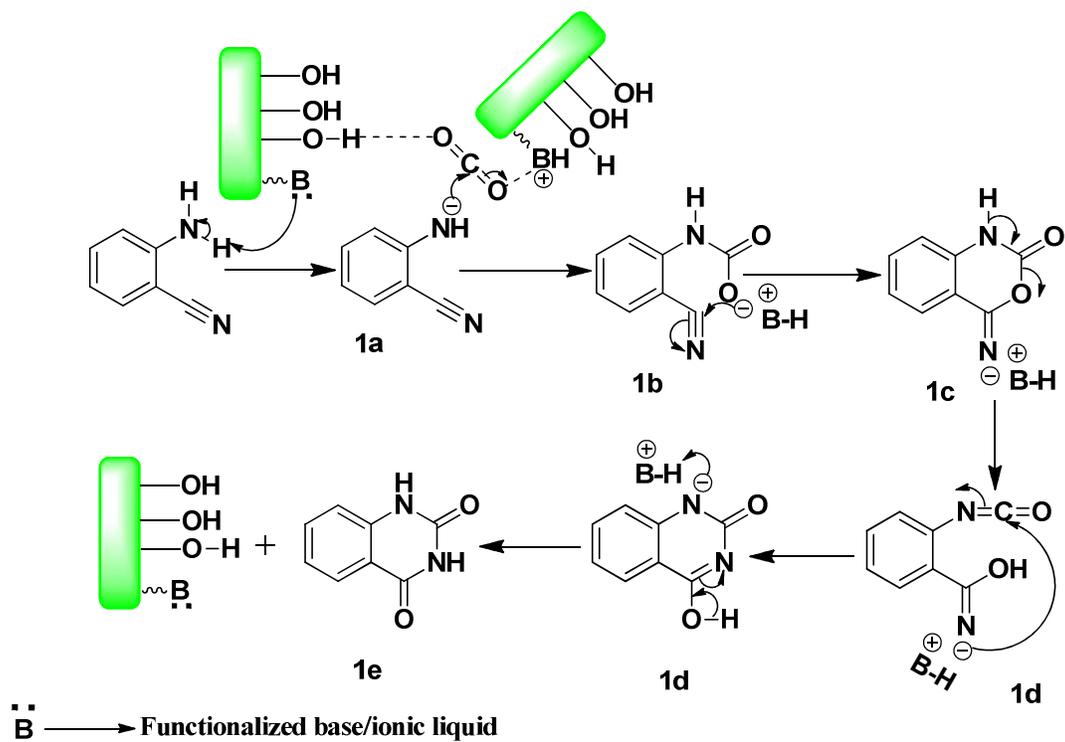
**Figure S5.** FT-IR spectra of fresh and recycled Basic-Nano-ZSM-5-Pr-MIM-OH obtained after 5<sup>th</sup> recycle.



**Figure S6.** TPD profiles of fresh and recycled Basic-Nano-ZSM-5-*Pr*-MIM-OH obtained after 5<sup>th</sup> recycle.



**Scheme S1.** Proposed mechanism for the synthesis of cyclic carbonate by the cycloaddition reaction of CO<sub>2</sub> and epoxide over functionalized Basic-Nano-ZSM-5.



**Scheme S2.** Proposed mechanism for the synthesis of quinazoline-2,4(1H,3H)-dione by the cycloaddition reaction of  $\text{CO}_2$  and 2-aminobenzonitrile over functionalized Basic-Nano-ZSM-5.