

**Uncatalyzed Carboboration of Seven-Membered-Ring *trans*-Alkenes: Formation of Air-Stable Trialkylboranes**

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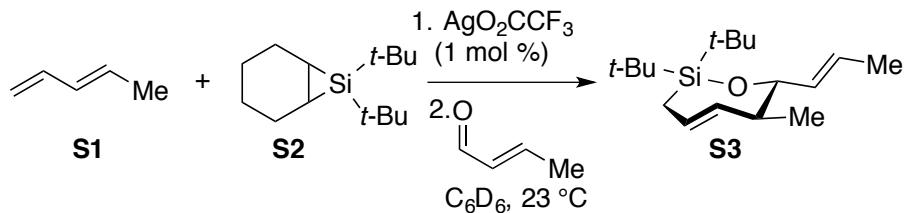
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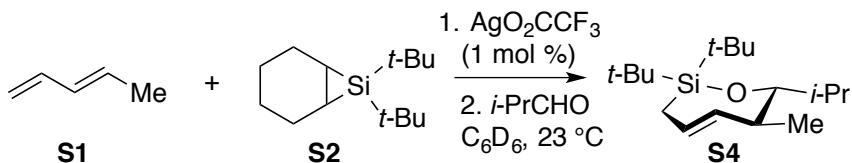
## I. General Experimental

All <sup>1</sup>H, <sup>13</sup>C, <sup>11</sup>B, <sup>19</sup>F, and <sup>29</sup>Si NMR spectra were recorded at ambient temperature using Bruker AVIII-400 (400, 100, 128 and 376 MHz, respectively), AV-500 (500 and 125 MHz, respectively) and AVIII-600 with TCI cryoprobe (600 and 150 MHz, respectively) spectrometers. The data were reported as follows: chemical shifts are reported in ppm from internal tetramethylsilane or from residual solvent peaks (<sup>1</sup>H NMR: CDCl<sub>3</sub> δ 7.26; C<sub>6</sub>D<sub>6</sub> δ 7.16; CD<sub>3</sub>OD δ 3.31 and <sup>13</sup>C NMR: CDCl<sub>3</sub> δ 77.2; C<sub>6</sub>D<sub>6</sub> δ 128.4) on the δ scale, multiplicity (s = singlet, b = broad, d = doublet, t = triplet, q = quartet, quint = quintet, m = multiplet), coupling constants (Hz), and integration. <sup>29</sup>Si NMR spectra (AV-500 99 MHz) were referenced against external tetramethylsilane (δ 0). <sup>19</sup>F NMR spectra were referenced against external hexafluorobenzene (δ -164.9) unless noted otherwise. <sup>11</sup>B NMR spectra were referenced against external boron trifluoride diethyl etherate (δ 0). Multiplicity of carbon peaks was determined using HSQC or DEPT experiments. Due to difficulties in isolation and purification of certain products, diagnostic <sup>1</sup>H and <sup>13</sup>C peaks were assigned using a combination of COSY, HSQC, HMBC, and nOe experiments. Infrared (IR) spectra were recorded using a Thermo Nicolet AVATAR Fourier Transform IR spectrometer using attenuated total reflectance (ATR). High-resolution mass spectra (HRMS) were recorded using an Agilent 6224 Accurate-Mass time-of-flight spectrometer with atmospheric pressure chemical ionization (APCI) or electrospray ionization (ESI) ionization sources. Melting points were reported uncorrected. Analytical thin layer chromatography was performed on Silicycle silica gel 60 Å F<sub>254</sub> plates. Liquid chromatography was performed using forced flow (flash chromatography) of the indicated solvent system on Silicycle silica gel (SiO<sub>2</sub>) 60 (230-400 mesh). THF, DMF, and CH<sub>2</sub>Cl<sub>2</sub> were dried by filtration through alumina according to the method of Grubbs.<sup>1</sup> All reactions were run under a nitrogen atmosphere in glassware that had been flame-dried under vacuum unless otherwise stated. Metal catalysts, silacyclop propane, and *trans*-oxasilacycloheptenes were stored and manipulated in a nitrogen-atmosphere dry box. All solvents and reagents were degassed before use. The synthesis and characterization of compounds **1**,<sup>2</sup> **S2**,<sup>3</sup> **S5**,<sup>4</sup> **S8**,<sup>5</sup> and **S27**<sup>6</sup> were reported previously. Unless otherwise noted, all reagents were commercially available. *trans*-Oxasilacycloheptenes were unable to be purified due to decomposition upon exposure to air and column chromatography. *trans*-Alkenes also undergo formal [1,3]-sigmatropic rearrangements at room temperature, preventing facile isolation.<sup>2</sup> All *trans*-oxasilacycloheptene concentrations were calculated by <sup>1</sup>H NMR spectroscopy relative to a known amount of internal standard (mesitylene) as described below. CCDC 1539169 (**2**) and CCDC 1539170 (*ent*-**10**) contain the supplementary crystallographic data for this paper. These data are provided free of charge by The Cambridge Crystallographic Data Centre.

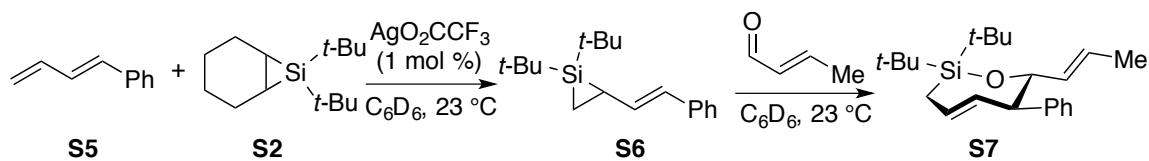
## II. Synthesis of *trans*-Oxasilacycloheptenes



**Representative Procedure for the One-Step Synthesis of *trans*-Alkenes (*trans*-Oxasilacycloheptene S3).** The synthesis of *trans*-oxasilacycloheptene S3 was adapted from a reported procedure.<sup>2</sup> To a solution of diene S1 (0.0075 mL, 0.075 mmol), cyclohexene silacyclopropane S2 (0.022 g, 0.098 mmol), and mesitylene (0.0020 mL, 0.014 mmol, internal standard) in C<sub>6</sub>D<sub>6</sub> (0.475 mL) in an NMR tube was added AgOCOCF<sub>3</sub> (0.025 mL, 0.0030 M in C<sub>6</sub>H<sub>6</sub> 1 mol %). After 10 min, crotonaldehyde (0.0063 mL, 0.076 mmol) was added. *trans*-Oxasilacycloheptene S3 was formed in 10 min in 56% yield over two steps based on <sup>1</sup>H NMR spectroscopic analysis of the area of a peak of the standard (δ 6.71) and the area of the allylic ether CH peak (~δ 3.60). *trans*-Oxasilacycloheptene S3 was unable to be purified and was used without further purification: <sup>1</sup>H NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>, diagnostic peaks) δ 5.58 (dqd, *J* = 15.2, 6.4, 0.9, 1H), 5.44 (ddq, *J* = 15.2, 6.7, 1.5, 1H), 4.98 (ddd, *J* = 16.8, 10.1, 0.9, 1H), 3.62–3.59 (m, 1H), 2.35–2.26 (m, 2H), 1.85 (dd, *J* = 12.0, 4.5, 1H), 1.57–1.55 (m, 3H), 1.08 (s, 9H), 1.00 (s, 9H); <sup>13</sup>C NMR (125 MHz, C<sub>6</sub>D<sub>6</sub>, diagnostic peaks) δ 133.4, (CH), 130.7 (CH), 126.2 (CH), 81.0 (CH), 49.1 (CH), 29.1 (CH<sub>3</sub>), 28.4 (CH<sub>3</sub>), 18.7 (CH<sub>2</sub>), 18.2 (CH<sub>3</sub>); HRMS (APCI) *m/z* calcd for C<sub>17</sub>H<sub>33</sub>OSi (M + H)<sup>+</sup> 281.2295, found 281.2297.

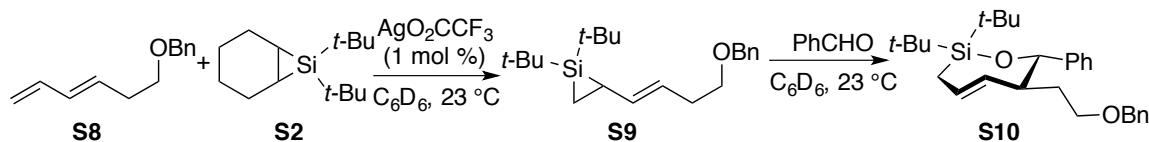


**trans**-Oxasilacycloheptene S4. *trans*-Oxasilacycloheptene S4 was prepared using the representative procedure for the synthesis of *trans*-alkenes using diene S1 (0.012 mL, 0.12 mmol), cyclohexene silacyclopropane S2 (0.034 g, 0.15 mmol), AgOCOCF<sub>3</sub> (0.036 mL, 0.010 M in C<sub>6</sub>D<sub>6</sub>, 0.3 mol %), and isobutyraldehyde (0.011 mL, 0.12 mmol). *trans*-Oxasilacycloheptene S4 was formed in 10 min in 63% yield over two steps based on <sup>1</sup>H NMR spectroscopic analysis of the area of a peak of the standard (δ 6.71) and the area of the ether CH peak (δ 3.15). *trans*-Oxasilacycloheptene S4 was unable to be purified and was used without further purification: <sup>1</sup>H NMR (400 MHz, C<sub>6</sub>D<sub>6</sub>, diagnostic peaks) δ 4.93 (dd, *J* = 16.9, 9.9, 1H), 3.15 (dd, *J* = 9.1, 1.6, 1H), 2.52–2.42 (m, 1H), 2.29–2.22 (m, 1H), 1.82 (dd, *J* = 12.0, 4.4, 1H), 1.07 (s, 9H), 0.99 (s, 9H), 0.93 (d, *J* = 6.6, 3H), 0.88 (d, *J* = 6.8, 3H), 0.72 (d, *J* = 7.0, 3H); <sup>13</sup>C NMR (100 MHz, C<sub>6</sub>D<sub>6</sub>, diagnostic peaks) δ 134.8 (CH), 131.6 (CH), 84.0 (CH), 45.4 (CH), 30.7 (CH), 29.1 (CH<sub>3</sub>), 28.5 (CH<sub>3</sub>), 18.9 (CH<sub>2</sub>), 16.9 (CH<sub>3</sub>), 15.6 (CH<sub>3</sub>), 15.1 (CH<sub>3</sub>); HRMS (APCI) *m/z* calcd for C<sub>17</sub>H<sub>35</sub>OSi (M + H)<sup>+</sup> 283.2452, found 283.2450.



**trans-Alkene (S7).** The synthesis of *trans*-oxasilacycloheptene **S7** was adapted from a reported procedure.<sup>2</sup> To a solution of diene **S5** (0.010 g, 0.077 mmol), cyclohexene silacyclopropane **S2** (0.029 g, 0.13 mmol), and mesitylene (0.0020 mL, 0.014 mmol, internal standard) in  $\text{C}_6\text{D}_6$  (0.50 mL) in an NMR tube was added  $\text{AgOCOCF}_3$  (0.025 mL, 0.030 M in  $\text{C}_6\text{D}_6$ , 1 mol %). Vinylsilacyclopropane **S6** was formed in 10 min in 72% yield based on  $^1\text{H}$  NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of the alkene CH peak ( $\delta$  6.43). Vinylsilacyclopropane **S6** was unable to be purified and was used without further purification:  $^1\text{H}$  NMR (500 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  7.35–7.33 (m, 2H), 7.19–7.16 (m, 2H), 7.05–7.01 (m, 1H), 6.55 (dd,  $J = 15.7, 7.7, 1\text{H}$ ), 6.43 (d,  $J = 15.7, 1\text{H}$ ), 2.00–1.94 (m, 1H), 1.07 (s, 9H), 1.00 (s, 9H), 0.90–0.84 (m, 2H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  135.0 (CH), 129.2 (CH), 127.7 (CH), 126.4 (CH), 125.9 (CH), 124.9 (CH), 30.9 ( $\text{CH}_3$ ), 29.9 ( $\text{CH}_3$ ), 19.6 (CH), 3.6 ( $\text{CH}_2$ );  $^{29}\text{Si}$  NMR (99 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  –47.1.

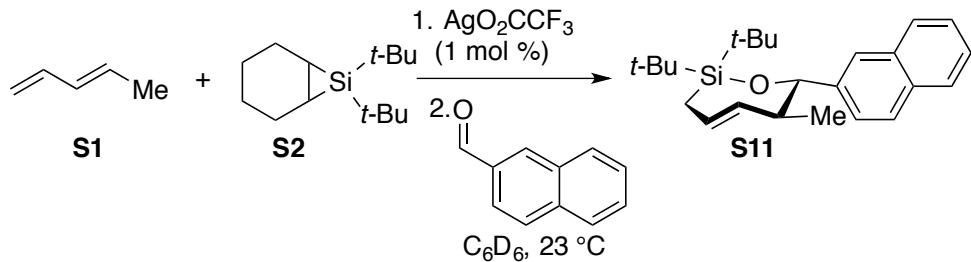
To a solution of vinylsilacyclopropane **S6** (0.015 g, 0.055 mmol) in  $\text{C}_6\text{D}_6$  (0.525 mL) was added crotonaldehyde (0.0065 mL, 0.079 mmol). *trans*-Oxasilacycloheptene **S7** was formed in 10 min in 86% yield from vinylsilacyclopropane **S6** based on  $^1\text{H}$  NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of the benzylic CH peak ( $\delta$  4.24). *trans*-Oxasilacycloheptene **S7** was unable to be purified and was used without further purification:  $^1\text{H}$  NMR (500 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  7.22–7.17 (m, 4H), 7.10–7.06 (m, 1H), 5.99–5.92 (m, 1H), 5.59–5.48 (m, 2H), 5.41–5.36 (m, 1H), 4.24 (ddt,  $J = 9.0, 5.2, 1.2, 1\text{H}$ ), 3.43 (t,  $J = 9.7, 1\text{H}$ ), 2.37–2.32 (m, 1H), 1.42 (dt,  $J = 6.5, 1.4, 3\text{H}$ ), 1.12 (s, 9H), 1.04 (s, 9H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  135.1 (CH), 132.1 (CH), 125.5 (CH), 79.9 (CH), 61.7 (CH), 29.1 ( $\text{CH}_3$ ), 28.4 ( $\text{CH}_3$ ), 19.1 ( $\text{CH}_2$ ), 18.1 ( $\text{CH}_3$ ); HRMS (APCI)  $m / z$  calcd for  $\text{C}_{22}\text{H}_{35}\text{OSi} (\text{M} + \text{H})^+$  343.2452, found 343.2447.



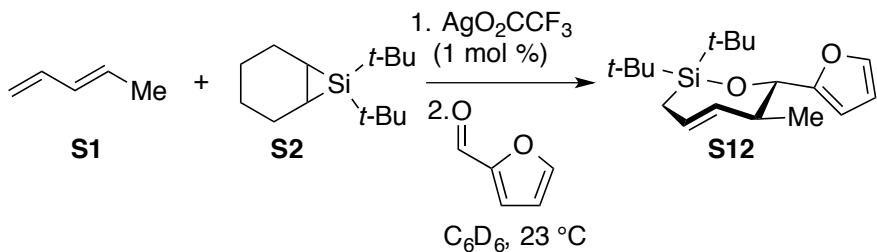
**trans-Alkene (S10).** The synthesis of *trans*-oxasilacycloheptene **S10** was adapted from a reported procedure.<sup>2</sup> To a solution of diene **S8** (0.014 g, 0.074 mmol), cyclohexene silacyclopropane **S2** (0.021 g, 0.094 mmol), and mesitylene (0.0020 mL, 0.014 mmol, internal standard) in  $\text{C}_6\text{D}_6$  (0.475 mL) in an NMR tube was added  $\text{AgOCOCF}_3$  (0.025 mL, 0.030 M in  $\text{C}_6\text{D}_6$ , 1 mol %). Vinylsilacyclopropane **S9** was formed in 10 min in 82% yield based on  $^1\text{H}$  NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of the alkene CH peak (~ $\delta$  5.51). Vinylsilacyclopropane **S9** was unable to be purified and was used without further purification:  $^1\text{H}$  NMR (500 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  7.32–7.30 (m, 2H), 7.11–7.08 (m, 1H), 5.85–5.80 (m, 1H),

5.54–5.48 (m, 1H), 4.35 (s, 2H), 4.43–3.40 (m, 2H), 2.46–2.42 (m, 2H), 1.83–1.77 (m, 1H), 1.09 (s, 9H), 0.90 (t,  $J = 11.2$ , 1H), 0.74 (dd,  $J = 11.2$ , 9.0);  $^{13}\text{C}$  NMR (125 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  135.3 (CH), 128.1 (CH), 127.8 (CH), 121.5 (CH), 73.3 ( $\text{CH}_2$ ), 71.5 ( $\text{CH}_2$ ), 34.3 ( $\text{CH}_2$ ), 31.0 ( $\text{CH}_3$ ), 30.0 ( $\text{CH}_3$ ), 17.9, (CH), 2.9 ( $\text{CH}_2$ );  $^{29}\text{Si}$  NMR (99 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  –48.5.

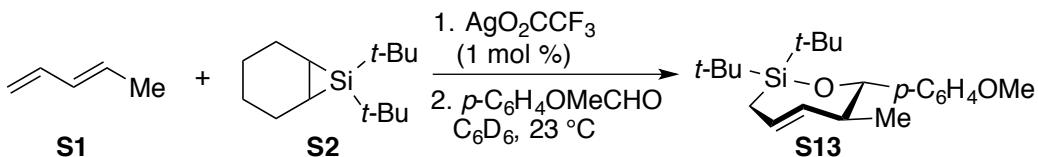
To a solution of vinylsilacyclopropane **S9** (0.021 g, 0.061 mmol) in  $\text{C}_6\text{D}_6$  (0.50 mL) was added benzaldehyde (0.0075 mL, 0.074 mmol). *trans*-Oxasilacycloheptene **S10** was formed in 10 min in 88% yield from vinylsilacyclopropane **S9** based on  $^1\text{H}$  NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of one of the ether  $\text{CH}_2$  peaks ( $\sim\delta$  3.35). *trans*-Oxasilacycloheptene **S10** was unable to be purified and was used without further purification:  $^1\text{H}$  NMR (500 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  6.05–5.97 (m, 1H), 5.00 (dd,  $J = 16.8$ , 10.1, 1H), 4.25 (s, 2H and m, 1H), 3.49–3.44 (m, 1H), 3.38–3.33 (m, 1H), 2.74–2.68 (m, 1H), 2.37–2.32 (m, 1H), 1.63–1.56 (m, 1H), 1.46–1.39 (m, 1H), 1.02 (s, 9H), 0.97 (s, 9H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  137.6 (CH), 129.2 (CH), 81.8 (CH), 73.4 ( $\text{CH}_2$ ), 70.0 ( $\text{CH}_2$ ), 54.5 (CH), 31.2 ( $\text{CH}_2$ ), 29.1 ( $\text{CH}_3$ ), 28.3 ( $\text{CH}_3$ ), 19.2 ( $\text{CH}_2$ );  $^{29}\text{Si}$  NMR (99 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  –1.7; HRMS (APCI)  $m/z$  calcd for  $\text{C}_{28}\text{H}_{41}\text{O}_2\text{Si} (\text{M} + \text{H})^+$  438.2899, found 438.2896.



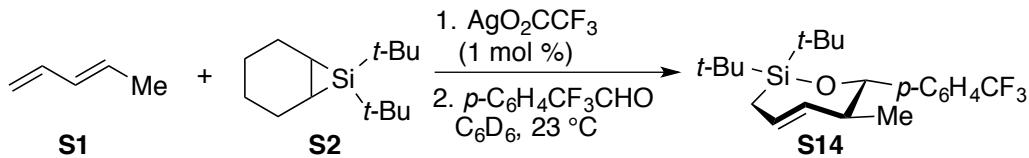
**trans**-Oxasilacycloheptene **S11**. *trans*-Oxasilacycloheptene **S11** was prepared using the representative procedure for the synthesis of *trans*-alkenes using diene **S1** (0.0080 mL, 0.080 mmol), cyclohexene silacyclopropane **S2** (0.024 g, 0.11 mmol),  $\text{AgOCOCF}_3$  (0.027 mL, 0.030 M in  $\text{C}_6\text{D}_6$ , 1 mol %), and 2-naphthaldehyde (0.012 g, 0.077 mmol). *trans*-Oxasilacycloheptene **S11** was formed in 10 min in 62% yield over two steps based on  $^1\text{H}$  NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of the benzylic CH peak ( $\delta$  4.30). *trans*-Oxasilacycloheptene **S11** was unable to be purified and was used without further purification:  $^1\text{H}$  NMR (400 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  7.73–7.61 (m, 4H), 7.29–7.24 (m, 2H), 7.23–7.17 (m, 1H), 6.10–6.01 (m, 1H), 5.12 (dd,  $J = 16.8$ , 9.8, 1H), 4.30 (d,  $J = 8.9$ , 1H), 2.70–2.61 (m, 1H), 2.41–2.35 (m, 1H), 1.07 (s, 9H), 1.01 (s, 9H), 0.91 (d,  $J = 6.6$ , 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  136.3 (CH), 130.7 (CH), 126.4 (CH), 83.2 (CH), 51.0 (CH), 29.1 ( $\text{CH}_3$ ), 28.3 ( $\text{CH}_3$ ), 19.0 ( $\text{CH}_2$ ), 16.5 ( $\text{CH}_3$ ); HRMS (APCI)  $m/z$  calcd for  $\text{C}_{24}\text{H}_{35}\text{OSi} (\text{M} + \text{H})^+$  367.2452, found 367.2448.



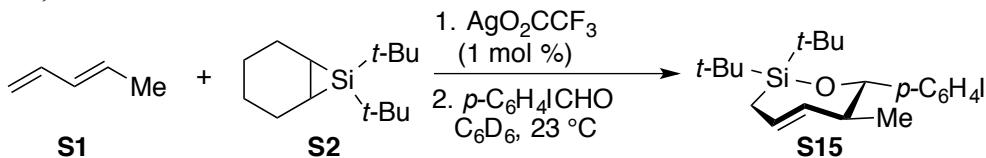
**trans-Oxasilacycloheptene S12.** *trans*-Oxasilacycloheptene **S12** was prepared using the representative procedure for the synthesis of *trans*-alkenes using diene **S1** (0.0075 mL, 0.075 mmol), cyclohexene silacyclopropane **S2** (0.022 g, 0.098 mmol),  $\text{AgOCOCF}_3$  (0.025 mL, 0.030 M in  $\text{C}_6\text{D}_6$ , 1 mol %), and furfural (0.0065 mL, 0.078 mmol). *trans*-Oxasilacycloheptene **S12** was formed in 10 min in 64% yield over two steps based on  $^1\text{H}$  NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of the benzylic CH peak ( $\delta$  4.27). *trans*-Oxasilacycloheptene **S12** was unable to be purified and was used without further purification:  $^1\text{H}$  NMR (400 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  7.08 (t,  $J$  = 1.3, 1H), 6.02, (d,  $J$  = 1.3, 2H), 6.01 (dd,  $J$  = 16.8, 13.2, 4.5, 0.8, 1H), 5.00 (ddd,  $J$  = 16.8, 10.0, 1.1, 1H) 4.27 (d,  $J$  = 9.3, 1H), 2.96–2.86 (m, 1H), 2.29 (ddd,  $J$  = 13.2, 12.0, 1.1, 1H), 1.86 (dd,  $J$  = 12.0, 4.5, 1H), 1.02 (s, 9H), 0.99 (s, 9H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  142.2 (CH), 136.7 (CH), 130.2 (CH), 110.6 (CH), 106.7 (CH), 75.7 (CH), 48.1 (CH), 29.0 ( $\text{CH}_3$ ), 28.2 ( $\text{CH}_3$ ), 19.0 ( $\text{CH}_2$ ); HRMS (APCI)  $m/z$  calcd for  $\text{C}_{18}\text{H}_{31}\text{OSi} (\text{M} + \text{H})^+$  307.2088, found 307.2078.



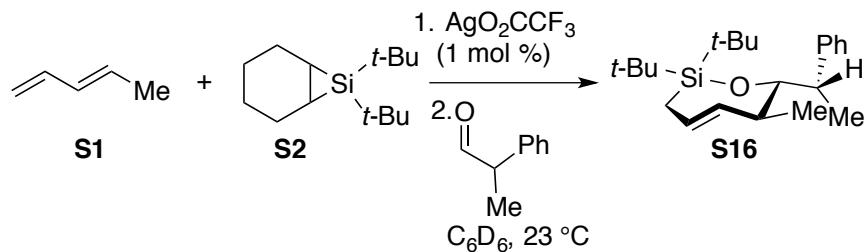
**trans-Oxasilacycloheptene S13.** *trans*-Oxasilacycloheptene **S13** was prepared using the representative procedure for the synthesis of *trans*-alkenes using diene **S1** (0.0070 mL, 0.070 mmol), cyclohexene silacyclopropane **S2** (0.018 g, 0.080 mmol),  $\text{AgOCOCF}_3$  (0.020 mL, 0.010 M in  $\text{C}_6\text{D}_6$ , 0.3 mol %), and *p*-methoxybenzaldehyde (0.0085 mL, 0.070 mmol). *trans*-Oxasilacycloheptene **S13** was formed in 10 min in 58% yield over two steps based on  $^1\text{H}$  NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of the methoxy  $\text{CH}_3$  peak ( $\delta$  3.32). *trans*-Oxasilacycloheptene **S13** was unable to be purified and was used without further purification:  $^1\text{H}$  NMR (600 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  7.17 (d,  $J$  = 8.7, 2H), 6.79 (d,  $J$  = 8.7, 2H), 6.04–5.98 (m, 1H), 5.10 (dd,  $J$  = 16.8, 10.0, 1H), 4.12 (d,  $J$  = 8.9, 1H), 3.32 (s, 3H), 2.60–2.54 (m, 1H), 2.37–2.33 (m, 1H), 1.06 (s, 9H), 1.00 (s, 9H), 0.92 (d,  $J$  = 6.6, 3H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  136.0 (CH), 131.1 (CH), 114.3 (CH), 82.7 (CH), 55.1 ( $\text{CH}_3$ ), 51.2 (CH), 29.1 ( $\text{CH}_3$ ), 28.3 ( $\text{CH}_3$ ), 19.0 ( $\text{CH}_2$ ), 16.6 ( $\text{CH}_3$ ); HRMS (APCI)  $m/z$  calcd for  $\text{C}_{21}\text{H}_{35}\text{O}_2\text{Si} (\text{M} + \text{H})^+$  347.2401, found 347.2398.



**trans-Oxasilacycloheptene S14.** *trans*-Oxasilacycloheptene **S14** was prepared using the representative procedure for the synthesis of *trans*-alkenes using diene **S1** (0.0080 mL, 0.080 mmol), cyclohexene silacyclopropane **S2** (0.024 g, 0.11 mmol),  $\text{AgOCOCF}_3$  (0.027 mL, 0.030 M in  $\text{C}_6\text{D}_6$ , 1 mol %), and *p*-(trifluoromethyl)benzaldehyde (0.011 mL, 0.081 mmol). *trans*-Oxasilacycloheptene **S14** was formed in 10 min in 70% yield over two steps based on  $^1\text{H}$  NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of the benzylic CH peak ( $\delta$  4.01). *trans*-Oxasilacycloheptene **S14** was unable to be purified and was used without further purification:  $^1\text{H}$  NMR (400 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  7.35 (d,  $J = 8.1$ , 2H), 7.08 (d,  $J = 8.1$ , 2H), 5.98–5.89 (m, 1H), 4.99 (dd,  $J = 16.7$ , 10.0, 1H), 4.01 (d,  $J = 8.9$ , 1H), 2.41–2.28 (m, 2H), 1.01 (s, 9H), 0.95 (s, 9H), 0.76 (d,  $J = 6.6$ , 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  136.5 (CH), 130.0 (CH), 127.7 (CH), 125.7 (q,  $J = 4.0$ , CH), 82.3 (CH), 51.2 (CH), 29.0 (CH<sub>3</sub>), 28.2 (CH<sub>3</sub>), 18.9 (CH<sub>2</sub>), 16.2 (CH<sub>3</sub>);  $^{19}\text{F}$  NMR (377 MHz,  $\text{C}_6\text{D}_6$ , internal reference *p*-CF<sub>3</sub>-C<sub>6</sub>H<sub>4</sub>CHO  $\delta$  -63.2)  $\delta$  -62.0; HRMS (APCI) *m/z* calcd for C<sub>21</sub>H<sub>32</sub>F<sub>3</sub>OSi (M + H)<sup>+</sup> 385.2169, found 385.2154.

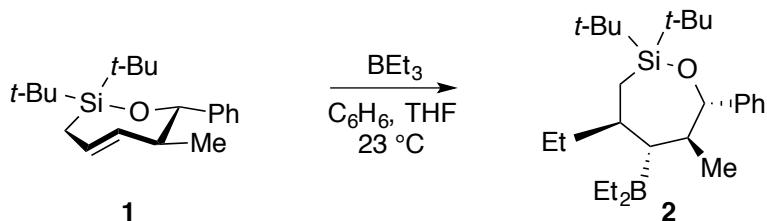


**trans-Oxasilacycloheptene S15.** *trans*-Oxasilacycloheptene **S15** was prepared using the representative procedure for the synthesis of *trans*-alkenes using diene **S1** (0.0080 mL, 0.080 mmol), cyclohexene silacyclopropane **S2** (0.023 g, 0.10 mmol),  $\text{AgOCOCF}_3$  (0.027 mL, 0.030 M in  $\text{C}_6\text{D}_6$ , 1 mol %), and *p*-iodobenzaldehyde (0.018 g, 0.078 mmol). *trans*-Oxasilacycloheptene **S15** was formed in 10 min in 57% yield over two steps based on  $^1\text{H}$  NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of the benzylic CH peak ( $\delta$  3.92). *trans*-Oxasilacycloheptene **S15** was unable to be purified and was used without further purification:  $^1\text{H}$  NMR (600 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  7.45 (d,  $J = 8.3$ , 2H), 6.76 (d,  $J = 8.3$ , 2H), 5.94–5.89 (m, 1H), 4.99 (dd,  $J = 16.8$ , 10.0, 1H), 3.92 (d,  $J = 8.9$ , 1H), 2.40–2.34 (m, 1H), 2.33–2.28 (m, 1H), 1.00 (s, 9H), 0.95 (s, 9H), 0.78 (d,  $J = 6.6$ , 3H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  137.8 (CH), 136.4 (CH), 131.0 (CH), 128.4 (CH), 82.3 (CH), 51.1 (CH), 29.0 (CH<sub>3</sub>), 28.2 (CH<sub>3</sub>), 18.9 (CH<sub>2</sub>), 16.3 (CH<sub>3</sub>); HRMS (APCI) *m/z* calcd for C<sub>20</sub>H<sub>30</sub>IOSi (M + H)<sup>+</sup> 443.1262, found 443.1253.



**trans-Oxasilacycloheptene S16.** *trans*-Oxasilacycloheptene **S16** was prepared using the representative procedure for the synthesis of *trans*-alkenes using diene **S1** (0.0080 mL, 0.080 mmol), cyclohexene silacyclopropane **S2** (0.030 g, 0.13 mmol),  $\text{AgOCOCF}_3$  (0.027 mL, 0.030 M in  $\text{C}_6\text{D}_6$  1 mol %), and 2-phenylpropionaldehyde (0.011 mL, 0.082 mmol). *trans*-Oxasilacycloheptene **S16** was formed in 10 min in 57% yield over two steps based on  $^1\text{H}$  NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of the benzylic CH peak ( $\delta$  3.39). *trans*-Oxasilacycloheptene **S16** was unable to be purified and was used without further purification:  $^1\text{H}$  NMR (600 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  7.21–7.18 (m, 2H), 7.08–7.06 (m, 2H), 6.85 (d,  $J$  = 7.7, 1H), 5.81–5.78 (m, 1H), 4.89 (dd,  $J$  = 16.8, 9.9, 1H), 3.39 (d,  $J$  = 9.2, 1H), 2.92 (q,  $J$  = 7.0, 1H), 2.57–2.51 (m, 1H), 2.24–2.19 (m, 1H), 1.81 (dd,  $J$  = 12.0, 4.2, 1H), 1.06 (s, 9H), 0.62 (s, 9H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  135.4 (CH), 130.8 (CH), 128.9 (CH), 128.7 (CH), 126.5 (CH), 85.0 (CH), 53.4 (CH), 45.9 (CH), 41.4 (CH), 28.6 ( $\text{CH}_3$ ), 28.5 ( $\text{CH}_3$ ), 19.3 ( $\text{CH}_2$ ); HRMS (APCI)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{37}\text{OSi} (\text{M} + \text{H})^+$  345.2608, found 345.2620.

### III. Synthesis of Trialkylboranes

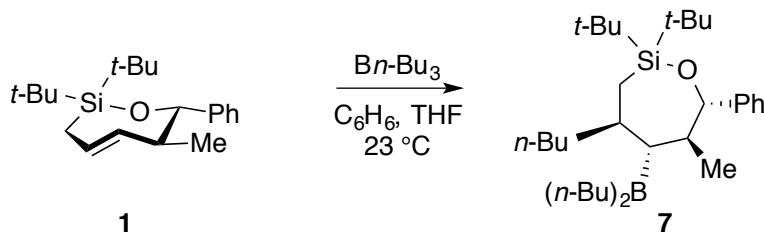


#### Representative Procedure for the Synthesis of Trialkylboranes (trialkylborane **2**).

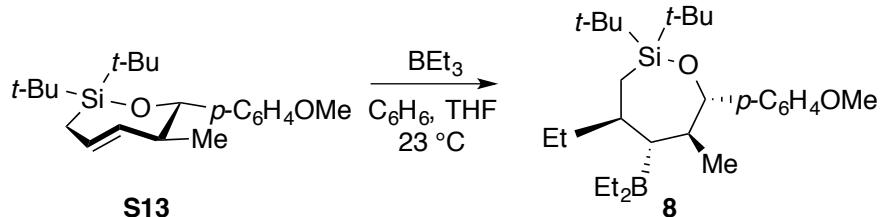
To a solution of *trans*-alkene **1** (0.132 g, 0.417 mmol) in  $\text{C}_6\text{H}_6$  (4.00 mL) was added triethylborane (3.60 mL, 1.0 M in THF, 3.6 mmol). After 2 h, 1 M NaOH (5.0 mL) was added, followed by 30%  $\text{H}_2\text{O}_2$  in  $\text{H}_2\text{O}$  (3.0 mL) (*caution the ensuing reaction was exothermic*). After 20 min,  $\text{Et}_2\text{O}$  (30 mL) and brine (15 mL) were added and the layers were separated. The aqueous layer was extracted with  $\text{Et}_2\text{O}$  (2  $\times$  30 mL). The combined organic layers were dried ( $\text{MgSO}_4$ ) and concentrated *in vacuo*. Purification by flash chromatography (hexanes) afforded trialkylborane **2** as a white solid (0.159 g, 92%): mp = 66–69 °C;  $^1\text{H}$  NMR (500 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  7.31 (d,  $J$  = 7.1, 2H), 7.21–7.17 (m, 2H), 7.10–7.07 (m, 1H), 4.64 (d,  $J$  = 9.5, 1H), 2.28–2.22 (m, 1H), 2.10–2.02 (m, 1H), 1.57 (t,  $J$  = 10.2, 1H), 1.23 (m, 5H and s, 9H), 1.08 (t,  $J$  = 7.6, 6H), 1.02 (s, 9H and m, 1H), 0.94–0.88 (m, 4H), 0.76 (dd,  $J$  = 15.5, 11.9, 1H), 0.17 (d,  $J$  = 7.0, 3H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{C}_6\text{D}_6$ , 25 °C)  $\delta$  145.8 (C), 128.7 (CH), 127.6 (CH), 127.5 (CH), 85.2 (CH), 52.7 (CH), 44.5 (CH), 39.5 (CH), 33.6 (CH<sub>2</sub>), 29.5 (CH<sub>3</sub>), 29.2 (CH<sub>3</sub>), 22.3 (C), 21.6 (C), 20.3 (CH<sub>3</sub>), 17.2 (CH<sub>2</sub>), 12.5 (CH<sub>3</sub>), 10.2 (CH<sub>3</sub>); (Some  $^{13}\text{C}$  shifts next to the boron atom are

too broad to resolve at 25 °C)  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ , –40 °C)  $\delta$  145.3 (C), 127.7 (CH), 126.8 (CH), 126.6 (CH), 83.7 (CH), 50.7 (CH), 42.6 (CH), 35.6 (CH), 36.2 (CH<sub>2</sub>), 28.7 (CH<sub>3</sub>), 28.5 (CH<sub>3</sub>), 22.0 (C), 21.9 (CH<sub>2</sub>), 21.0 (C), 19.6 (CH<sub>3</sub>), 16.0 (CH<sub>2</sub>), 15.7 (CH<sub>2</sub>), 12.4 (CH<sub>3</sub>), 9.9 (CH<sub>3</sub>), 9.8 (CH<sub>3</sub>);  $^{11}\text{B}$  NMR (128 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  32.1; IR (ATR) 1081, 1054, 823, 698  $\text{cm}^{-1}$ ; HRMS (APCI)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{37}\text{OSi} (\text{M} - \text{BEt}_2)^+$  345.2608, found 345.2610.

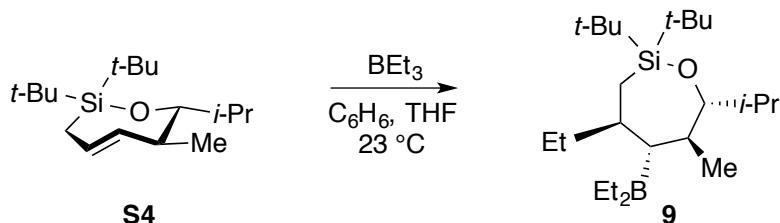
Note: The  $^1\text{H}$ ,  $^{13}\text{C}$ , and  $^{11}\text{B}$  NMR spectra are consistent with trialkylboranes having multiple conformational isomers, despite being a single stereoisomer. Subsequent oxidation of the boron–carbon bond results in a sinlge stereoisomer with one conformational isomer on the NMR time scale. Due to the slow rotation around the carbon–boron bonds, methylene carbons next to the boron atom are too broad to resolve at 25 °C in the  $^{13}\text{C}$  NMR spectra of most trialkylboranes. The methyl groups on the BEt<sub>2</sub> group are not diastereotopic at 25 °C in the  $^{13}\text{C}$  NMR spectrum (as evidenced by the  $^{13}\text{C}$  NMR spectrum at 25 °C), but are diastereotopic at –40 °C. Additionally, some trialkylboranes exhibit two  $^{11}\text{B}$  NMR resonances, likely due to the impeded rotation. Variable temperature NMR studies were unable to resolve these peaks, however.



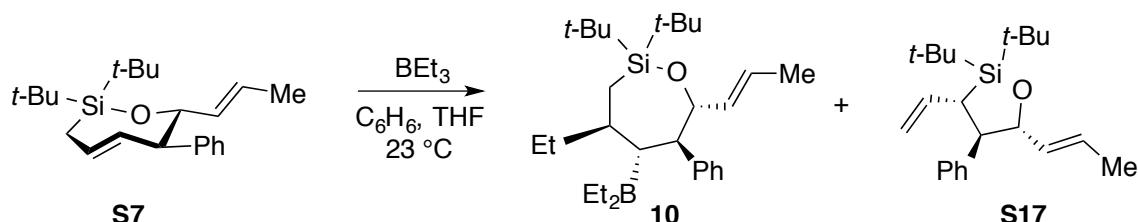
**Trialkylborane 7.** Trialkylborane **7** was prepared using the representative procedure for the synthesis of trialkylboranes using *trans*-alkene **1** (0.028 g, 0.088 mmol), tributylborane (0.150 mL, 0.615 mmol) (*caution: neat tributylborane reacts violently with oxygen*), and  $\text{C}_6\text{H}_6$  (1.00 mL). Purification by flash chromatography (hexanes) afforded trialkylborane **7** as a colorless oil (0.032 g, 73%):  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  7.29–7.25 (m, 4H), 7.22–7.19 (m, 1H), 4.52 (d,  $J = 7.9$ , 1H), 2.18–2.13 (m, 1H), 1.91–1.84 (m, 1H), 1.46–1.38 (m, 6H), 1.36–1.32 (m, 4H), 1.24–1.14 (m, 6H), 1.08 (m, 1H and s, 9H), 1.03–0.97 (m, 2H), 0.95 (s, 9H), 0.92–0.88 (m, 9H), 0.77–0.68 (m, 2H), 0.07 (d,  $J = 7.0$ , 3H);  $^{13}\text{C}$  NMR (125 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  145.8 (C), 128.7 (CH), 127.6 (CH), 127.5 (CH), 85.2 (CH), 53.6 (CH), 44.6 (CH), 41.3 (CH<sub>2</sub>), 38.1 (CH), 30.5 (CH<sub>2</sub>), 29.5 (CH<sub>3</sub>), 29.2 (CH<sub>3</sub>), 28.9 (CH<sub>2</sub>), 26.9 (CH<sub>2</sub>), 23.8 (CH<sub>2</sub>), 22.4 (C), 21.6 (C), 20.7 (CH<sub>3</sub>), 18.2 (CH<sub>2</sub>), 14.7 (CH<sub>3</sub>), 14.6 (CH<sub>3</sub>) (see note under trialkylborane **2**, p. S10);  $^{11}\text{B}$  NMR (128 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  54.0, 32.8 (see note under trialkylborane **2**, p. S10); IR (ATR) 1081, 1066, 823, 698  $\text{cm}^{-1}$ ; HRMS (APCI)  $m/z$  calcd for  $\text{C}_{24}\text{H}_{41}\text{OSi} (\text{M} - \text{B''Bu}_2)^+$  373.2921, found 373.2924.



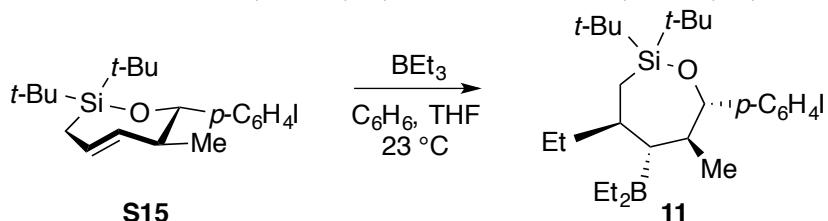
**Trialkylborane 8.** Trialkylborane **8** was prepared using the representative procedure for the synthesis of trialkylboranes using *trans*-alkene **S13** (0.012 g, 0.035 mmol), triethylborane (0.450 mL, 1.0 M in THF, 0.45 mmol), and C<sub>6</sub>H<sub>6</sub> (0.530 mL). Purification by flash chromatography (hexanes) afforded trialkylborane **8** as a colorless oil (0.013 g, 84%): <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.17 (d, *J* = 8.6, 2H), 6.82 (d, *J* = 8.6, 2H), 4.49 (d, *J* = 8.0, 1H), 3.80 (s, 3H), 2.11–2.06 (m, 1H), 1.89–1.83 (m, 1H), 1.45 (t, *J* = 10.2, 1H), 1.23–1.18 (m, 4H), 1.08 (m, 2H and s, 9H), 1.00 (t, *J* = 7.6, 6H), 0.95 (s, 9H), 0.91 (t, *J* = 8.0, 3H), 0.87–0.82 (m, 1H), 0.66 (dd, *J* = 15.5, 11.9, 1H), 0.07 (d, *J* = 7.0, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 144.51 (C), 137.9 (C), 127.8 (CH), 113.2 (CH), 83.8 (CH), 55.3 (CH<sub>3</sub>), 51.9 (CH), 43.9 (CH), 38.9 (CH), 32.9 (CH<sub>2</sub>), 29.0 (CH<sub>3</sub>), 28.8 (CH<sub>3</sub>), 22.0 (C), 21.1 (C), 19.8 (CH<sub>3</sub>), 16.6 (CH<sub>2</sub>), 12.2 (CH<sub>3</sub>), 9.7 (CH<sub>3</sub>) (see note under trialkylborane **2**, p. S10); <sup>11</sup>B NMR (128 MHz, CDCl<sub>3</sub>) δ 55.4; IR (ATR) 1244, 1074, 1041, 820 cm<sup>-1</sup>; HRMS (APCI) *m/z* calcd for C<sub>27</sub>H<sub>50</sub>BO<sub>2</sub>Si (M + H)<sup>+</sup> 445.3673, found 445.3664.



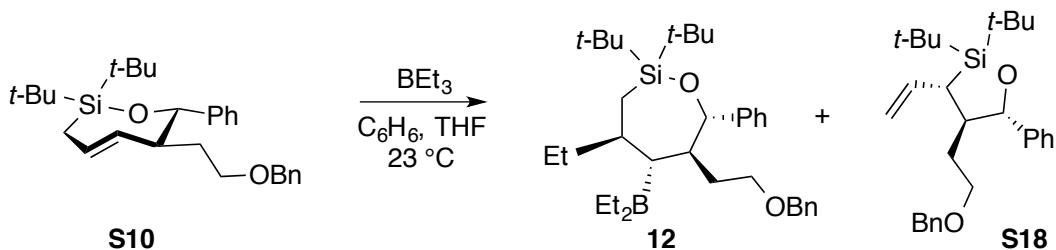
**Trialkylborane 9.** Trialkylborane **9** was prepared using the representative procedure for the synthesis of trialkylboranes using *trans*-alkene **S4** (0.021 g, 0.075 mmol), triethylborane (0.480 mL, 1.0 M in THF, 0.48 mmol), and C<sub>6</sub>H<sub>6</sub> (0.400 mL). Purification by flash chromatography (hexanes) afforded trialkylborane **9** as a colorless oil that could not be further purified (0.028 g, 97% unpurified yield): <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 3.49 (dd, *J* = 8.5, 1.8, 1H), 1.87–1.79 (m, 2H), 1.76–1.71 (m, 1H), 1.35 (t, *J* = 10.1, 1H), 1.16–1.13 (m, 4H), 1.03 (s, 9H and m, 2H), 1.01 (s, 9H and m, 8H), 0.90–0.85 (m, 4H), 0.81 (m, 1H and t, *J* = 6.7, 3H), 0.45 (m, 1H and d, *J* = 6.9, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 84.2 (CH), 50.7 (CH), 40.4 (CH), 39.1 (CH), 32.6 (CH<sub>2</sub>), 30.8 (CH), 29.0 (CH<sub>3</sub>), 28.6 (CH<sub>3</sub>), 22.4 (C), 21.7 (C), 21.5 (CH<sub>3</sub>), 18.5 (CH<sub>3</sub>), 17.5 (CH<sub>2</sub>), 13.9 (CH<sub>3</sub>), 12.4 (CH<sub>3</sub>), 9.7 (CH<sub>3</sub>) (see note under trialkylborane **2**, p. S10); <sup>11</sup>B NMR (128 MHz, C<sub>6</sub>D<sub>6</sub>) δ 54.2, 32.2 (see note under trialkylborane **2**, p. S10); IR (ATR) 1062, 820, 621 cm<sup>-1</sup>; HRMS (APCI) *m/z* calcd for C<sub>23</sub>H<sub>49</sub>BKOSi (M + K)<sup>+</sup> 419.3282, found 419.3277.



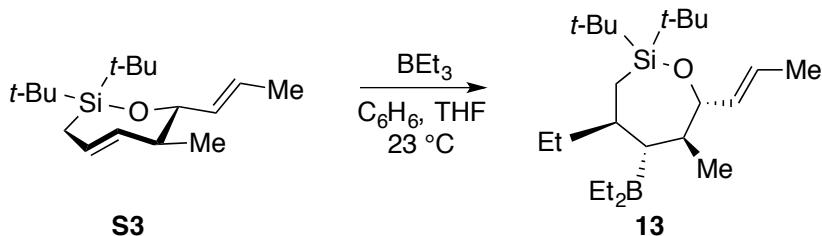
**Trialkylborane 10.** Trialkylborane **10** was prepared using the representative procedure for the synthesis of trialkylboranes using *trans*-alkene **S7** (0.015 g, 0.044 mmol), triethylborane (0.450 mL, 1.0 M in THF, 0.45 mmol), and C<sub>6</sub>H<sub>6</sub> (0.525 mL). <sup>1</sup>H NMR spectroscopic analysis of the unpurified reaction mixture after workup showed a mixture of trialkylborane **10** and alkene **S17** in a 77:23 (**10:S17**) ratio. (The spectral data are consistent with the data reported below for alkene **S17**, the result of rearrangement of the starting *trans*-alkene.) Purification by flash chromatography (hexanes-3:97 EtOAc:hexanes) afforded trialkylborane **10** as a white solid (0.010 g, 53%): mp = 75–77 °C; <sup>1</sup>H NMR (600 MHz, C<sub>6</sub>D<sub>6</sub>) δ 6.99–6.75 (m, 5H), 5.66–5.60 (m, 1H), 5.17 (ddd, *J* = 15.1, 5.6, 1.6, 1H), 4.70–4.68 (m, 1H), 2.91 (dd, *J* = 10.8, 8.4, 1H), 2.26–2.21 (m, 1H), 2.14 (t, *J* = 10.4, 1H), 1.35 (d, *J* = 6.6, 3H), 1.28 (s, 9H and m, 2H), 1.17 (s, 9H and m, 2H), 1.09–1.03 (m, 2H), 0.95–0.89 (m, 10H), 0.82 (dd, *J* = 15.5, 11.8, 1H); <sup>13</sup>C NMR (150 MHz, C<sub>6</sub>D<sub>6</sub>) δ 143.5 (C), 134.6 (CH), 131.3 (CH), 128.7 (CH), 126.8 (CH), 124.5 (CH), 81.3 (CH), 57.0 (CH), 52.5 (CH), 40.3 (CH), 33.6 (CH<sub>2</sub>), 29.5 (CH<sub>3</sub>), 29.3 (CH<sub>3</sub>), 22.6 (CH<sub>2</sub>), 22.5 (C), 21.7 (C), 18.1 (CH<sub>3</sub>), 18.0 (CH<sub>2</sub>), 17.8 (CH<sub>2</sub>), 12.9 (CH<sub>3</sub>), 10.1 (CH<sub>3</sub>), 9.7 (CH<sub>3</sub>); <sup>11</sup>B NMR (128 MHz, C<sub>6</sub>D<sub>6</sub>) δ 54.0; IR (ATR) 1600, 1474, 1070, 822 cm<sup>-1</sup>; HRMS (APCI) *m/z* calcd for C<sub>24</sub>H<sub>39</sub>OSi (M – BEt<sub>2</sub>)<sup>+</sup> 371.2765, found 371.2764. Anal. Calcd for C<sub>28</sub>H<sub>49</sub>BOSi: C, 76.33; H, 11.21. Found: C, 76.05; H, 11.27.



**Trialkylborane 11.** Trialkylborane **11** was prepared using the representative procedure for the synthesis of trialkylboranes using *trans*-alkene **S15** (0.019 g, 0.043 mmol), triethylborane (0.450 mL, 1.0 M in THF, 0.45 mmol), and C<sub>6</sub>H<sub>6</sub> (0.530 mL). Purification by flash chromatography (hexanes) afforded trialkylborane **11** as a white solid (0.018 g, 78%): mp = 91–93 °C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.60 (d, *J* = 8.4, 2H), 7.01 (d, *J* = 8.4, 2H), 4.47 (d, *J* = 8.0, 1H), 2.12–2.05 (m, 1H), 1.92–1.80 (m, 1H), 1.45 (t, *J* = 10.2, 1H), 1.24–1.16 (m, 4H), 1.11 (dd, *J* = 15.4, 1.1, 1H), 1.07 (s, 9H), 1.00 (m, 1H and t, *J* = 7.5, 6H), 0.92 (s, 9H and t, *J* = 6.3, 3H), 0.86–0.79 (m, 1H), 0.66 (dd, *J* = 15.4, 11.9, 1H), 0.07 (d, *J* = 7.0, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 145.2 (C), 137.1 (CH), 129.0 (CH), 92.1 (C), 83.8 (CH), 51.9 (CH), 43.7 (CH), 38.8 (CH), 32.9 (CH<sub>2</sub>), 28.9 (CH<sub>3</sub>), 28.7 (CH<sub>3</sub>), 22.0 (C), 21.1 (C), 19.7 (CH<sub>3</sub>), 16.6 (CH<sub>2</sub>), 12.1 (CH<sub>3</sub>), 9.6 (CH<sub>3</sub>) (see note under trialkylborane **2**, p. S10); <sup>11</sup>B NMR (128 MHz, CDCl<sub>3</sub>) δ 33.3; IR (ATR) 1078, 1006, 848, 823 cm<sup>-1</sup>; HRMS (APCI) *m/z* calcd for C<sub>22</sub>H<sub>36</sub>IOSi (M – BEt<sub>2</sub>)<sup>+</sup> 471.1575, found 471.1577.

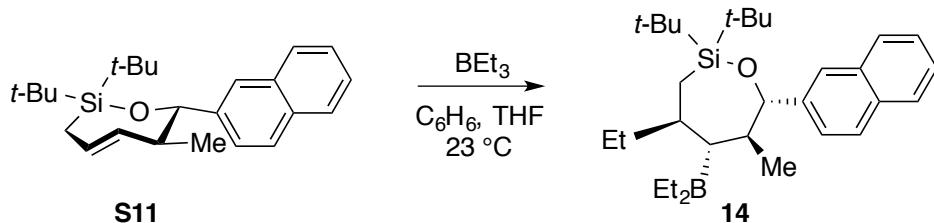


**Trialkylborane 12.** Trialkylborane **12** was prepared using the representative procedure for the synthesis of trialkylboranes using *trans*-alkene **S10** (0.048 g, 0.11 mmol), triethylborane (0.900 mL, 1.0 M in THF, 0.90 mmol), and C<sub>6</sub>H<sub>6</sub> (1.00 mL). <sup>1</sup>H NMR spectroscopic analysis of the unpurified reaction mixture after workup showed a mixture of trialkylborane **12** and alkene **S18** in a 41:59 (**12:S18**) ratio. (The spectral data are consistent with the data reported below for alkene **S18**, the result of rearrangement of the starting *trans*-alkene.) Purification by flash chromatography (1:99 EtOAc:hexanes) afforded trialkylborane **12** as a colorless oil (0.015 g, 26%): <sup>1</sup>H NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>) δ 7.32 (d, *J* = 7.2, 2H), 7.15–7.11 (m, 6H), 7.09–7.03 (m, 2H), 4.74 (d, *J* = 8.2, 1H), 4.07 (d, *J* = 12.2, 1H), 4.02 (d, *J* = 12.2, 1H), 2.76 (td, *J* = 9.2, 6.6, 1H), 2.53 (td, *J* = 9.2, 4.7, 1H), 2.27–2.21 (m, 1H), 2.16–2.11 (m, 1H), 1.79 (t, *J* = 10.4, 1H), 1.43–1.28 (m, 6H), 1.21 (s, 9H), 1.15 (m, 2H and t, *J* = 7.5, 6H), 0.99 (s, 9H), 0.94–0.92 (m, 4H), 0.74 (dd, *J* = 15.5, 11.8, 1H); <sup>13</sup>C NMR (150 MHz, C<sub>6</sub>D<sub>6</sub>) δ 145.4 (C), 139.7 (C), 128.8 (CH), 128.7 (CH), 128.1 (CH), 127.9 (CH), 127.85 (CH), 127.78 (CH), 83.6 (CH), 73.1 (CH<sub>2</sub>), 68.5 (CH<sub>2</sub>), 51.3 (CH), 47.1 (CH), 39.7 (CH), 34.5 (CH<sub>2</sub>), 33.1 (CH<sub>2</sub>), 29.5 (CH<sub>3</sub>), 29.3 (CH<sub>3</sub>), 22.2 (C), 21.5 (C), 17.2 (CH<sub>2</sub>), 12.7 (CH<sub>3</sub>), 10.2 (CH<sub>3</sub>) (see note under trialkylborane **2**, p. S10); <sup>11</sup>B NMR (128 MHz, C<sub>6</sub>D<sub>6</sub>) δ 54.3, 32.9 (see note under trialkylborane **2**, p. S10); IR (ATR) 1306, 1082, 823, 699 cm<sup>-1</sup>; HRMS (APCI) *m/z* calcd for C<sub>30</sub>H<sub>45</sub>O<sub>2</sub>Si (M – BEt<sub>2</sub>)<sup>+</sup> 465.3183, found 465.3181.

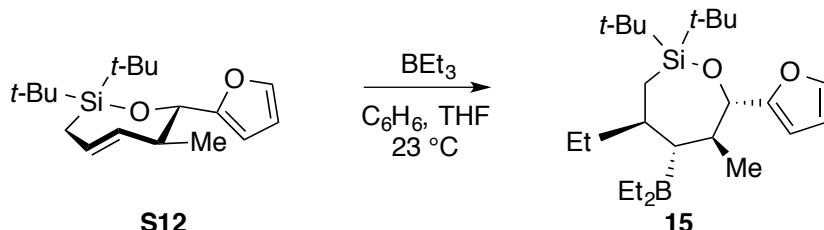


**Trialkylborane 13.** Trialkylborane **13** was prepared using the representative procedure for the synthesis of trialkylboranes using *trans*-alkene **S3** (0.022 g, 0.079 mmol), triethylborane (0.975 mL, 1.0 M in THF, 0.98 mmol), and C<sub>6</sub>H<sub>6</sub> (1.00 mL). Purification by flash chromatography (hexanes) afforded trialkylborane **13** as a colorless oil (0.020 g, 67%): <sup>1</sup>H NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>) δ 5.63–5.51 (m, 2H), 4.12–4.09 (m, 1H), 2.18–2.10 (m, 1H), 1.87–1.79 (m, 1H), 1.59 (d, *J* = 6.1, 3H), 1.45 (t, *J* = 10.3, 1H), 1.22 (m, 3H and s, 9H), 1.12 (s, 9H and m, 10H), 0.92–0.89 (m, 3H), 0.68 (dd, *J* = 15.4, 11.9, 1H), 0.46 (d, *J* = 7.0, 3H) (Some of trialkylborane **13** decomposed to alkene **S19** (vide infra) and is present in the <sup>1</sup>H NMR spectrum); <sup>13</sup>C NMR (125 MHz, C<sub>6</sub>D<sub>6</sub>) δ 135.1 (CH), 124.8 (CH), 83.2 (CH), 52.7 (CH), 43.2 (CH), 39.4 (CH), 33.6 (CH<sub>2</sub>), 29.6 (CH<sub>3</sub>), 29.3 (CH<sub>3</sub>), 22.5 (C), 21.6 (C), 19.8 (CH<sub>3</sub>), 18.2 (CH<sub>3</sub>), 17.2 (CH<sub>2</sub>), 12.5 (CH<sub>3</sub>), 10.2 (CH<sub>3</sub>) (see note under

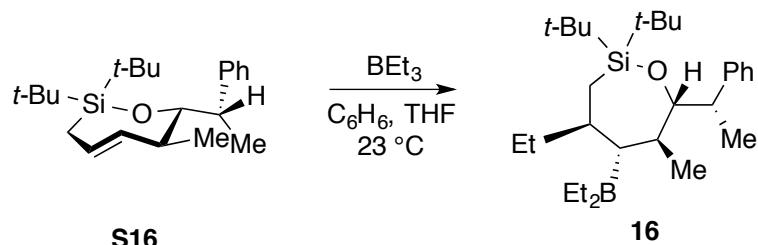
trialkylborane **2**, p. S10);  $^{11}\text{B}$  NMR (128 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  54.0, 32.0 (see note under trialkylborane **2**, p. S10); IR (ATR) 1079, 1048, 821, 643  $\text{cm}^{-1}$ ; HRMS (APCI)  $m/z$  calcd for  $\text{C}_{19}\text{H}_{37}\text{OSi} (\text{M} - \text{BEt}_2)^+$  309.2608, found 309.2616.



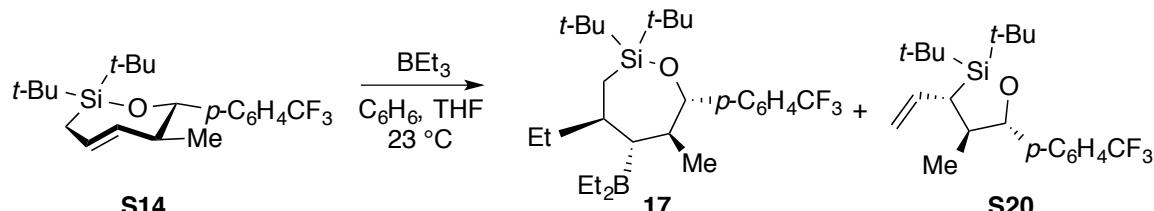
**Trialkylborane 14.** Trialkylborane **14** was prepared using the representative procedure for the synthesis of trialkylboranes using *trans*-alkene **S11** (0.018 g, 0.050 mmol) and triethylborane (0.300 mL, 1.0 M in THF, 0.30 mmol). Purification by flash chromatography (hexanes) afforded trialkylborane **14** as a colorless oil (0.020 g, 87%):  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.83–7.79 (m, 3H), 7.60 (s, 1H), 7.52 (dd,  $J = 8.4, 1.4$ , 1H), 7.48–7.41 (m, 2H), 4.71 (d,  $J = 8.0$ , 1H), 2.19–2.11 (m, 1H), 2.07–1.97 (m, 1H), 1.54 (t,  $J = 10.2$ , 1H), 1.26–1.17 (m, 5H), 1.11 (s, 9H), 1.02 (t,  $J = 7.6$ , 6H and m, 2H), 0.96 (s, 9H and m, 3H), 0.73 (dd,  $J = 15.5, 11.9$ , 1H), 0.09 (d,  $J = 7.0$ , 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  142.9 (C), 133.2 (C), 132.9 (C), 128.0 (CH), 127.83 (CH), 127.78 (CH), 125.8 (CH), 125.5 (CH), 125.40 (CH), 125.38 (CH), 84.5 (CH), 52.0 (CH), 43.5 (CH), 38.9 (CH), 33.0 ( $\text{CH}_2$ ), 29.0 ( $\text{CH}_3$ ), 28.7 ( $\text{CH}_3$ ), 22.0 (C), 21.2 (C), 19.8 ( $\text{CH}_3$ ), 16.7 ( $\text{CH}_2$ ), 12.2 ( $\text{CH}_3$ ), 9.7 ( $\text{CH}_3$ ) (see note under trialkylborane **2**, p. S10);  $^{11}\text{B}$  NMR (128 MHz,  $\text{CDCl}_3$ )  $\delta$  33.5; IR (ATR) 1076, 823, 668  $\text{cm}^{-1}$ ; HRMS (APCI)  $m/z$  calcd for  $\text{C}_{26}\text{H}_{39}\text{OSi} (\text{M} - \text{BEt}_2)^+$  395.2765, found 395.2765.



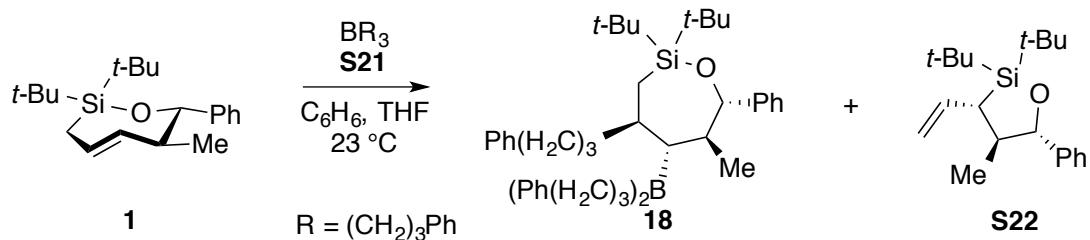
**Trialkylborane 15.** Trialkylborane **15** was prepared using the representative procedure for the synthesis of trialkylboranes using *trans*-alkene **S12** (0.011 g, 0.035 mmol), triethylborane (0.45 mL, 1.0 M in THF, 0.45 mmol), and  $\text{C}_6\text{H}_6$  (0.500 mL). Purification by flash chromatography (hexanes) afforded trialkylborane **15** as a colorless oil (0.009 g, 64%):  $^1\text{H}$  NMR (400 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  7.09 (dd,  $J = 1.8, 0.7$ , 1H), 6.14 (dd,  $J = 3.1, 0.7$ , 1H), 6.09 (dd,  $J = 3.1, 1.8$ , 1H), 4.79 (d,  $J = 8.2$ , 1H), 2.37–2.27 (m, 1H), 2.25–2.17 (m, 1H), 1.52 (t,  $J = 10.4$ , 1H), 1.20 (s, 9H and m, 3H), 1.16–1.12 (m, 2H), 1.06 (m, 6H and s, 9H), 1.00–0.94 (m, 1H), 0.90–0.83 (m, 4H), 0.70 (dd,  $J = 15.5, 11.9$ , 1H), 0.32 (d,  $J = 7.0$ , 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  158.4 (C), 141.7 (CH), 110.5 (CH), 106.3 (CH), 78.6 (CH), 52.4 (CH), 43.5 (CH), 39.3 (CH), 33.5 ( $\text{CH}_2$ ), 29.4 ( $\text{CH}_3$ ), 29.1 ( $\text{CH}_3$ ), 22.4 (C), 21.5 (C), 19.0 ( $\text{CH}_3$ ), 17.1 ( $\text{CH}_2$ ), 12.4 ( $\text{CH}_3$ ), 10.1 ( $\text{CH}_3$ ) (see note under trialkylborane **2**, p. S10);  $^{11}\text{B}$  NMR (128 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  54.0; IR (ATR) 1504, 1009, 823, 729  $\text{cm}^{-1}$ ; HRMS (APCI)  $m/z$  calcd for  $\text{C}_{20}\text{H}_{35}\text{O}_2\text{Si} (\text{M} - \text{BEt}_2)^+$  335.2401, found 335.2397.



**Trialkylborane 16.** Trialkylborane **16** was prepared using the representative procedure for the synthesis of trialkylboranes using *trans*-alkene **S16** (0.013 g, 0.038 mmol), triethylborane (0.400 mL, 1.0 M in THF, 0.40 mmol), and C<sub>6</sub>H<sub>6</sub> (0.530 mL). Purification by flash chromatography (hexanes) afforded trialkylborane **16** as a white solid that could not be further purified (0.011 g, 65% unpurified yield): mp = 89–92 °C; <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.27–7.26 (m, 4H), 7.17–7.14 (m, 1H), 3.69 (dd, *J* = 8.5, 1.8, 1H), 3.02 (qd, *J* = 7.0, 1.8, 1H), 1.91–1.83 (m, 2H), 1.39 (t, *J* = 10.1, 1H), 1.26 (d, *J* = 7.0, 3H and m, 4H), 1.04 (m, 6H and s, 9H), 1.01–0.98 (m, 1H), 0.87–0.82 (m, 5H), 0.63 (d, *J* = 6.7, 3H), 0.46 (s, 9H), 0.35 (dd, *J* = 15.3, 12.1, 1H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 146.4 (C), 129.1 (CH), 127.9 (CH), 125.8 (CH), 84.7 (CH), 50.6 (CH), 42.0 (CH), 40.5 (CH), 39.1 (CH), 32.5 (CH<sub>2</sub>), 28.9 (CH<sub>3</sub>), 27.9 (CH<sub>3</sub>), 21.8 (C), 21.6 (C), 19.2 (CH<sub>3</sub>), 17.7 (CH<sub>2</sub>), 12.4 (CH<sub>3</sub>), 10.6 (CH<sub>3</sub>), 9.8 (CH<sub>3</sub>) (see note under trialkylborane **2**, p. S10); <sup>11</sup>B NMR (128 MHz, C<sub>6</sub>D<sub>6</sub>) δ 53.8; IR (ATR) 1105, 1091, 822, 698 cm<sup>-1</sup>; HRMS (APCI) *m/z* calcd for C<sub>24</sub>H<sub>41</sub>OSi (M – BEt<sub>2</sub>)<sup>+</sup> 373.2921, found 373.2926.



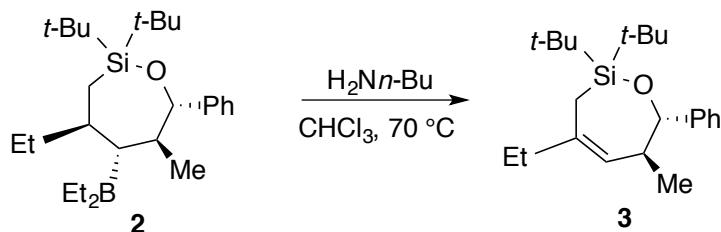
**Trialkylborane 17.** Trialkylborane **17** was prepared using the representative procedure for the synthesis of trialkylboranes using *trans*-alkene **S14** (0.022 g, 0.056 mmol), triethylborane (0.450 mL, 1.0 M in THF, 0.45 mmol), and C<sub>6</sub>H<sub>6</sub> (0.530 mL). <sup>1</sup>H NMR spectroscopic analysis of the unpurified reaction mixture after workup showed a mixture of trialkylborane **17** and alkene **S20** in a 70:30 (**17:S20**) ratio. (The spectral data are consistent with the data reported below for alkene **S20**, the result of rearrangement of the starting *trans*-alkene.) Purification by flash chromatography (hexanes) afforded trialkylborane **17** as a colorless oil (0.015 g, 56%): <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.54 (d, *J* = 8.0, 2H), 7.37 (d, *J* = 8.0, 2H), 4.59 (d, *J* = 8.0, 1H), 2.13–2.08 (m, 1H), 1.93–1.87 (m, 1H), 1.48 (t, *J* = 10.2, 1H), 1.26–1.16 (m, 4H), 1.14–1.11 (m, 1H), 1.08 (s, 9H), 1.01 (t, *J* = 7.6, 6H and m, 1H), 0.94 (s, 9H), 0.91 (t, *J* = 6.8, 3H), 0.87–0.81 (m, 1H), 0.68 (dd, *J* = 15.5, 11.9, 1H), 0.07 (d, *J* = 7.0, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 149.3 (C), 129.0 (q, *J* = 32.0, C), 127.2 (CH), 125.0 (q, *J* = 3.8, CH), 124.4 (q, *J* = 270.1, C), 83.8 (CH), 51.9 (CH), 43.7 (CH), 38.8 (CH), 32.9 (CH<sub>2</sub>), 28.9 (CH<sub>3</sub>), 28.7 (CH<sub>3</sub>), 22.0 (C), 21.1 (C), 19.7 (CH<sub>3</sub>), 16.6 (CH<sub>2</sub>), 12.1 (CH<sub>3</sub>), 9.6 (CH<sub>3</sub>) (see note under trialkylborane **2**, p. S10); <sup>11</sup>B NMR (128 MHz, CDCl<sub>3</sub>) δ 33.4; <sup>19</sup>F NMR (376 MHz, CDCl<sub>3</sub>) δ -65.4; IR (ATR) 1324, 1126, 849, 822 cm<sup>-1</sup>; HRMS (APCI) *m/z* calcd for C<sub>23</sub>H<sub>36</sub>F<sub>3</sub>OSi (M – BEt<sub>2</sub>)<sup>+</sup> 413.2482, found 413.2466.



**Trialkylborane 18.** Trialkylborane **18** was prepared following the representative procedure for the synthesis of trialkylboranes using *trans*-alkene **1** (0.030 g, 0.094 mmol), trialkylborane **S21**<sup>7</sup> (0.150 g, 0.407 mmol); (*caution: neat trialkylboranes reacts violently with oxygen*), and C<sub>6</sub>H<sub>6</sub> (0.97 mL). <sup>1</sup>H NMR spectroscopic analysis of the unpurified reaction mixture after workup showed a mixture of trialkylborane **18** and alkene **S22** in a 73:37 (**18:S22**) ratio. The spectral data are consistent with the data reported previously for alkene **S22**,<sup>2</sup> the result of rearrangement of the starting *trans*-alkene. Purification by flash chromatography (1:99 EtOAc:hexanes) afforded trialkylborane **18** as a colorless oil (0.019 g, 30%): <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.31–7.24 (m, 8H), 7.22–7.16 (m, 10H), 7.10–7.08 (m, 2H), 4.47 (d, *J* = 7.9, 1H), 2.59–2.42 (m, 6H), 2.15–2.07 (m, 1H), 1.85–1.77 (m, 2H), 1.71–1.57 (m, 3H), 1.44–1.39 (m, 2H), 1.14 (t, *J* = 8.0, 4H), 1.07 (s, 9H and m, 2H), 0.92 (m, 1H and s, 9H), 0.72–0.65 (m, 2H), 0.01 (d, *J* = 7.0, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 145.2 (C), 143.0 (C), 142.4 (C), 128.62 (CH), 128.58 (CH), 128.5 (CH), 128.4 (CH), 127.9 (CH), 126.9 (CH), 126.8 (CH), 125.9 (CH), 125.8 (CH), 84.1 (CH), 52.1 (CH), 43.8 (CH), 40.4 (CH<sub>2</sub>), 39.5 (CH<sub>2</sub>), 37.3 (CH), 36.4 (CH<sub>2</sub>), 29.4 (CH<sub>2</sub>), 29.0 (CH<sub>3</sub>), 28.7 (CH<sub>3</sub>), 27.9 (CH<sub>2</sub>), 21.9 (C), 21.1 (C), 19.9 (CH<sub>3</sub>), 17.5 (CH<sub>2</sub>) (see note under trialkylborane **2**, p. S10); <sup>11</sup>B NMR (128 MHz, C<sub>6</sub>D<sub>6</sub>) δ 32.9 (see note under trialkylborane **2**, p. S10); IR (ATR) 1080, 1029, 823, 697 cm<sup>-1</sup>; HRMS (APCI) *m/z* calcd for C<sub>29</sub>H<sub>43</sub>OSi (M – C<sub>16</sub>H<sub>22</sub>B)<sup>+</sup> 435.3078, found 435.3071.

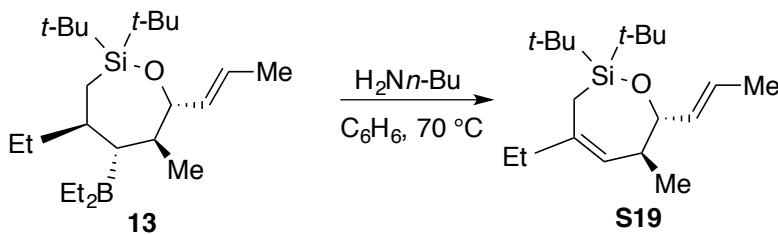
A compound similar to alkene **3**, but with a 1-phenylpropyl group, resulting from the decomposition of **18**, was also isolated (0.007 g, 17%, clear oil): <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, diagnostic peaks) δ 5.09–5.07 (m, 1H), 4.73 (d, *J* = 9.3, 1H), 2.69–2.60 (m, 8H), 1.07 (s, 9H), 1.00 (s, 9H), 0.56 (d, *J* = 7.2, 3H).

#### IV. Transformations of Trialkylboranes

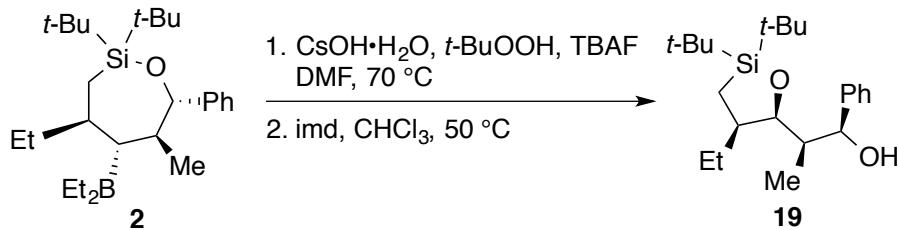


**Alkene 3.** A solution of trialkylborane **2** (0.034 g, 0.082 mmol) and H<sub>2</sub>N<sup>n</sup>Bu (0.130 mL, 1.19 mmol) in CHCl<sub>3</sub> (2.0 mL) was heated to 70 °C. After 16 h, the reaction mixture was cooled and CH<sub>2</sub>Cl<sub>2</sub> (5 mL) and H<sub>2</sub>O (5 mL) were added. The layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (5 mL). The combined organic layers were

washed with brine (10 mL), dried with  $\text{MgSO}_4$ , and concentrated *in vacuo*. Purification by flash chromatography (hexanes) afforded alkene **3** as a colorless oil (0.019 g, 68%):  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.32–7.31 (m, 4H), 7.25–7.21 (m, 1H), 5.02–5.01 (m, 1H), 4.70 (d,  $J = 9.2$ , 1H), 2.72–2.62 (m, 1H), 2.21–2.14 (m, 2H), 1.99 (d,  $J = 15.2$ , 1H), 1.53 (d,  $J = 15.2$ , 1H), 1.06 (m, 3H and s, 9H), 0.99 (s, 9H), 0.53 (d,  $J = 7.1$ , 3H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  145.2 (C), 139.3 (C), 128.7 (CH), 128.3 (CH), 127.8 (CH), 127.6 (CH), 82.3 (CH), 43.0 (CH), 35.6 ( $\text{CH}_2$ ), 29.2 ( $\text{CH}_3$ ), 28.7 ( $\text{CH}_3$ ), 23.1 (C), 21.5 (C), 20.3 ( $\text{CH}_3$ ), 15.5 ( $\text{CH}_2$ ), 13.3 ( $\text{CH}_3$ ); IR (ATR) 1603, 1096, 823  $\text{cm}^{-1}$ ; HRMS (APCI)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{37}\text{OSi} (\text{M} + \text{H})^+$  345.2608, found 345.2604. Anal. Calcd for  $\text{C}_{22}\text{H}_{36}\text{OSi}$ : C, 76.68; H, 10.53. Found: C, 76.53; H, 10.76.

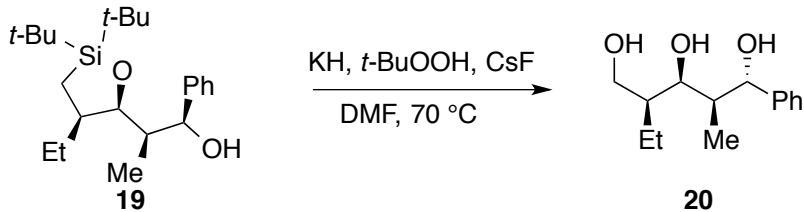


**Alkene S19.** A solution of trialkylborane **13** (0.014 g, 0.037 mmol) and  $\text{H}_2\text{N}''\text{Bu}$  (0.060 mL, 0.61 mmol) in  $\text{C}_6\text{H}_6$  (1.5 mL) was heated to 70 °C. After 18 h, the reaction mixture was cooled and  $\text{Et}_2\text{O}$  (15 mL) and  $\text{H}_2\text{O}$  (15 mL) were added. The layers were separated and the aqueous layer was extracted with  $\text{Et}_2\text{O}$  (15 mL). The combined organic layers were washed with brine (2 x 15 mL), dried with  $\text{MgSO}_4$ , and concentrated *in vacuo*. Purification by flash chromatography (hexanes) afforded alkene **S19** as a colorless oil (0.012 g, 100%):  $^1\text{H}$  NMR (600 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  5.55–5.54 (m, 2H), 5.00 (d,  $J = 5.7$ , 1H), 4.14–4.12 (m, 1H), 2.52–2.49 (m, 1H), 2.17–2.09 (m, 2H), 1.94 (d,  $J = 15.2$ , 1H), 1.56 (s, 3H), 1.35 (d,  $J = 15.2$ , 1H), 1.12 (s, 18H), 1.07–1.05 (m, 3H), 0.87 (d,  $J = 7.0$ , 3H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  139.9 (C), 134.7 (CH), 127.5 (CH), 126.5 (CH), 80.7 (CH), 40.8 (CH), 34.9 ( $\text{CH}_2$ ), 29.2 ( $\text{CH}_3$ ), 28.8 ( $\text{CH}_3$ ), 22.8 (C), 21.6 (C), 19.7 ( $\text{CH}_3$ ), 18.0 ( $\text{CH}_3$ ), 15.6 ( $\text{CH}_2$ ), 13.2 ( $\text{CH}_3$ ); IR (ATR) 1673, 1363, 1096, 822  $\text{cm}^{-1}$ ; HRMS (APCI)  $m/z$  calcd for  $\text{C}_{19}\text{H}_{37}\text{OSi} (\text{M} + \text{H})^+$  309.2608, found 309.2610. Anal. Calcd for  $\text{C}_{19}\text{H}_{36}\text{OSi}$ : C, 73.95; H, 11.76. Found: C, 74.25; H, 11.77.

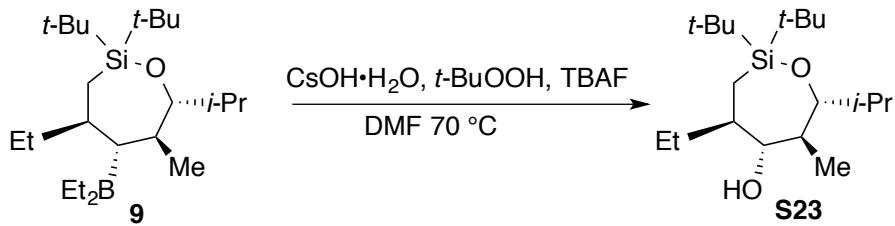


**Alcohol 19.** A solution of trialkylborane **2** (0.091 g, 0.22 mmol),  $t\text{-BuOOH}$  (0.620 mL, 5–6 M in decane, 3 mmol),  $\text{CsOH}\cdot\text{H}_2\text{O}$  (0.517 g, 3.08 mmol), and TBAF (1.1 mL, 1.0 M in THF, 1.1 mmol) in DMF (3.4 mL) was heated to 70 °C. After 19 h, the reaction mixture was cooled and  $\text{Et}_2\text{O}$  (5 mL) and saturated aqueous  $\text{Na}_2\text{S}_2\text{O}_3$  (5 mL) were added. The layers were separated and the aqueous layer was extracted with  $\text{Et}_2\text{O}$  (2 x 5 mL). The combined organic layers were washed with  $\text{H}_2\text{O}$  (10 mL) and brine (10 mL). These

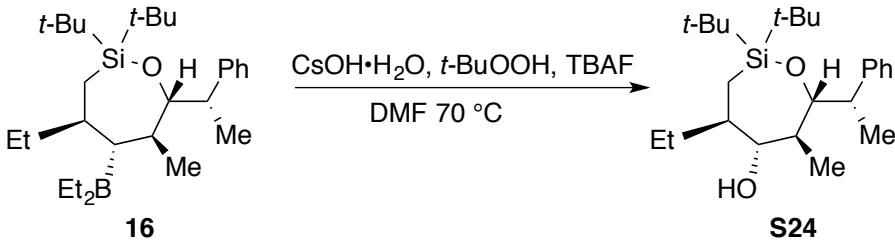
layers were dried with MgSO<sub>4</sub>, and concentrated *in vacuo*. To a solution of the resulting oil in CHCl<sub>3</sub> (6 mL) was added imidazole (0.114 g, 1.67 mmol) and the reaction mixture was heated to 50 °C. After 3 d, the reaction mixture was cooled and CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and H<sub>2</sub>O (10 mL) were added. The layers were separated and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 x 10 mL). The combined organic layers were washed with H<sub>2</sub>O (20 mL) and brine (20 mL), dried with MgSO<sub>4</sub> and concentrated *in vacuo*. Purification by flash chromatography (3:97 EtOAc:hexanes) afforded alcohol **19** as a yellow oil (0.053 g, 66% over two steps): <sup>1</sup>H NMR (400 MHz, C<sub>6</sub>D<sub>6</sub>) δ 7.44–7.42 (m, 2H), 7.25–7.22 (m, 2H), 7.12–7.08 (m, 1H), 4.82 (dd, *J* = 6.5, 4.2, 1H), 4.34 (dd, *J* = 7.9, 1.9, 1H), 3.88 (d, *J* = 6.5, 1H), 2.02–1.95 (m, 1H), 1.87–1.77 (m, 1H), 1.37–1.27 (m, 1H), 1.17 (d, *J* = 6.9, 3H and m, 1H), 1.06 (s, 9H), 0.97 (s, 9H), 0.75 (t, *J* = 7.3, 3H and m, 1H), 0.36 (dd, *J* = 15.1, 7.4, 1H); <sup>13</sup>C NMR (100 MHz, C<sub>6</sub>D<sub>6</sub>) δ 146.3 (C), 128.7 (CH), 127.4 (CH), 126.8 (CH), 80.5 (CH), 79.2 (CH), 44.0 (CH), 42.6 (CH), 29.1 (CH<sub>3</sub>), 28.6 (CH<sub>3</sub>), 25.5 (CH<sub>2</sub>), 21.7 (C), 20.3 (C), 14.6 (CH<sub>3</sub>), 13.8 (CH<sub>3</sub>), 11.0 (CH<sub>2</sub>); IR (ATR) 3458, 1062, 970, 823 cm<sup>-1</sup>; HRMS (APCI) *m/z* calcd for C<sub>22</sub>H<sub>38</sub>NaO<sub>2</sub>Si (M + Na)<sup>+</sup> 385.2533, found 385.2525. Anal. Calcd for C<sub>22</sub>H<sub>38</sub>O<sub>2</sub>Si: C, 72.87; H, 10.55. Found: C, 72.72; H, 10.72.



**Triol 20.** To a solution of KH (0.012 g, 0.30 mmol) in DMF (0.40 mL) was added 'BuOOH (0.065 mL, 5–6 M in decane, 0.4 mmol), followed by alcohol **19** (0.008 g, 0.022 mmol) in DMF (0.20 mL). After 10 min, CsF (0.019 g, 0.13 mmol) was added and the reaction mixture was heated to 70 °C. After 17 h, the reaction mixture was cooled to 23 °C and Et<sub>2</sub>O (5 mL) and saturated aqueous Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (5 mL) were added. The layers were separated and the aqueous layer was extracted with Et<sub>2</sub>O (2 × 5 mL). The combined organic layers were washed with H<sub>2</sub>O (10 mL) and brine (10 mL). These layers were dried with MgSO<sub>4</sub> and concentrated *in vacuo*. Purification by flash chromatography (1:1 EtOAc:hexanes) afforded alcohol **20** as a colorless oil (0.003 g, 57%): <sup>1</sup>H NMR (600 MHz, CD<sub>3</sub>OD) δ 7.33–7.31 (m, 4H), 7.25–7.23 (m, 1H), 4.58 (d, *J* = 7.9, 1H), 3.95 (d, *J* = 8.7, 1H), 3.56 (dd, *J* = 11.3, 4.5, 1H), 3.51 (dd, *J* = 11.3, 4.6, 1H), 2.04–1.99 (m, 1H), 1.76–1.69 (m, 1H), 1.56–1.51 (m, 1H), 1.47–1.40 (m, 1H), 0.94 (t, *J* = 7.5, 3H), 0.70 (d, *J* = 7.0, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 143.9 (C), 128.5 (CH), 127.6 (CH), 126.1 (CH), 78.7 (CH), 72.6 (CH), 63.2 (CH<sub>2</sub>), 45.3 (CH), 41.5 (CH), 19.6 (CH<sub>2</sub>), 11.9 (CH<sub>3</sub>), 11.7 (CH<sub>3</sub>); IR (ATR) 3338, 1017 cm<sup>-1</sup>; HRMS (ESI) *m/z* calcd for C<sub>14</sub>H<sub>22</sub>NaO<sub>3</sub> (M + Na)<sup>+</sup> 261.1461, found 261.1460. Anal. Calcd for C<sub>14</sub>H<sub>22</sub>O<sub>3</sub>: C, 70.56; H, 9.30. Found: C, 70.30; H, 9.20.

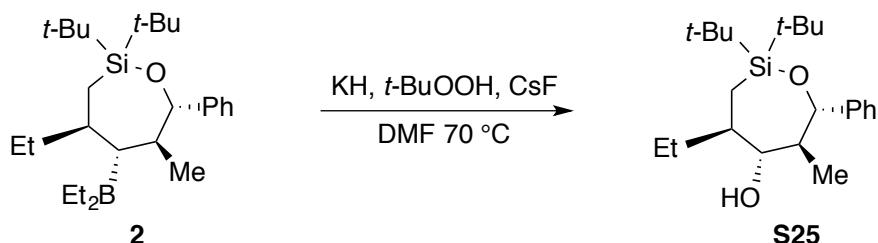


**Alcohol S23.** To a solution of trialkylborane **9** (0.028 g, 0.074 mmol),  $^t\text{BuOOH}$  (0.210 mL, 5–6 M in decane, 1 mmol), and  $\text{CsOH}\cdot\text{H}_2\text{O}$  (0.174 g, 1.04 mmol) in DMF (1.2 mL) was added TBAF (0.370 mL, 1.0 M in THF, 0.37 mmol). After 30 min,  $\text{Et}_2\text{O}$  (5 mL) and saturated aqueous  $\text{Na}_2\text{S}_2\text{O}_3$  (2.5 mL) were added. The layers were separated and the aqueous layer was extracted with  $\text{Et}_2\text{O}$  ( $2 \times 5$  mL). The combined organic layers were washed with  $\text{H}_2\text{O}$  (10 mL) and brine (10 mL). These layers were dried with  $\text{MgSO}_4$  and concentrated *in vacuo*. Purification by flash chromatography (5:95 EtOAc:hexanes) afforded alcohol **S23** as a colorless oil (0.019 g, 76% over two steps):  $^1\text{H}$  NMR (600 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  3.43 (dd,  $J = 9.0, 1.8$ , 1H), 2.78–2.75 (m, 1H), 1.88–1.83 (m, 1H), 1.80–1.74 (m, 1H), 1.73–1.68 (m, 1H), 1.54–1.49 (m, 1H), 1.48–1.41 (m, 1H), 1.10 (s, 9H), 1.08 (d,  $J = 6.7$ , 3H), 1.03 (s, 9H), 0.97 (t,  $J = 7.4$ , 3H), 0.92–0.91 (m, 6H), 0.78 (br s, 1H and dd,  $J = 15.4, 1.5$ , 1H), 0.56 (dd,  $J = 15.4, 12.5$ , 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  81.3 (CH), 80.2 (CH), 47.8 (CH), 43.5 (CH), 32.1 (CH), 29.3 (CH<sub>3</sub>), 29.2 (CH<sub>2</sub>), 29.0 (CH<sub>3</sub>), 22.7 (C), 22.2 (CH<sub>3</sub>), 21.9 (C), 15.4 (CH<sub>3</sub>), 14.5 (CH<sub>3</sub>), 13.2 (CH<sub>2</sub>), 10.7 (CH<sub>3</sub>); IR (ATR) 3351, 1139, 1062, 823  $\text{cm}^{-1}$ ; HRMS (APCI) *m/z* calcd for  $\text{C}_{19}\text{H}_{41}\text{O}_2\text{Si}$  ( $M + \text{H}$ )<sup>+</sup> 329.2870, found 329.2862. Anal. Calcd for  $\text{C}_{19}\text{H}_{40}\text{O}_2\text{Si}$ : C, 69.45; H, 12.27. Found: C, 69.64; H, 12.34.



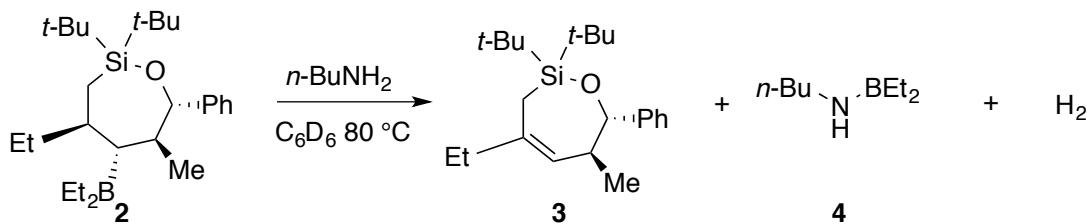
**Alcohol S24.** To a solution of trialkylborane **16** (0.011 g, 0.025 mmol),  $^t\text{BuOOH}$  (0.070 mL, 5–6 M in decane, 0.4 mmol), and  $\text{CsOH}\cdot\text{H}_2\text{O}$  (0.058 g, 0.35 mmol) in DMF (0.400 mL) was added TBAF (0.125 mL, 1.0 M in THF, 0.13 mmol). After 100 min,  $\text{Et}_2\text{O}$  (5 mL) and saturated aqueous  $\text{Na}_2\text{S}_2\text{O}_3$  (2.0 mL) were added. The layers were separated and the aqueous layer was extracted with  $\text{Et}_2\text{O}$  ( $2 \times 5$  mL). The combined organic layers were washed with  $\text{H}_2\text{O}$  (10 mL) and brine (10 mL). These layers were dried with  $\text{MgSO}_4$  and concentrated *in vacuo*. Purification by flash chromatography (5:95 EtOAc:hexanes) afforded alcohol **S24** as a white solid (0.008 g, 53% over two steps): mp = 126–128 °C;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.27–7.26 (m, 4H), 7.17–7.13 (m, 1H), 3.75 (dd,  $J = 9.0, 1.9$ , 1H), 3.16 (qd,  $J = 7.0, 1.9$ , 1H), 2.99 (t,  $J = 8.9$ , 1H), 1.91–1.85 (m, 1H), 1.78–1.65 (m, 2H), 1.54 (br s, 1H), 1.50–1.40 (m, 1H), 1.29 (d,  $J = 7.0$ , 3H), 1.18 (d,  $J = 6.7$ , 3H), 1.01 (s, 9H), 0.94 (t,  $J = 7.4$ , 3H), 0.73 (dd,  $J = 15.6, 1.5$ , 1H), 0.51 (dd,  $J = 15.6, 12.5$ , 1H), 0.45 (s, 9H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  145.9 (C), 129.1 (CH), 128.0 (CH), 126.0 (CH), 81.2 (CH), 79.9 (CH), 47.2 (CH), 42.6 (CH), 42.3 (CH), 28.7 (CH<sub>3</sub>), 28.6

(CH<sub>2</sub>), 27.8 (CH<sub>3</sub>), 21.8 (C), 21.5 (C), 15.9 (CH<sub>3</sub>), 12.6 (CH<sub>2</sub>), 10.7 (CH<sub>3</sub>), 9.9 (CH<sub>3</sub>); IR (ATR) 3348, 1041, 1029, 1009, 824 cm<sup>-1</sup>; HRMS (APCI) *m/z* calcd for C<sub>24</sub>H<sub>43</sub>O<sub>2</sub>Si (M + H)<sup>+</sup> 391.3027, found 391.3038.



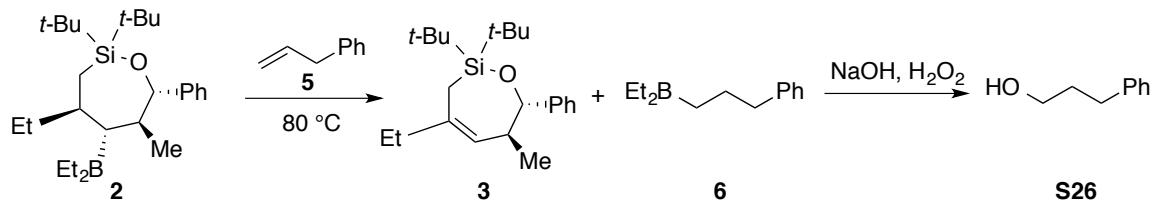
**Alcohol S25.** To a solution of KH (0.025 g, 0.55 mmol) in DMF (0.70 mL) was added 'BuOOH (0.110 mL, 5–6 M in decane, 0.6 mmol), followed by trialkylborane **2** (0.016 g, 0.039 mmol) in DMF (0.40 mL) and THF (0.15 mL). After 10 min, CsF (0.037 g, 0.24 mmol) was added and the reaction mixture was heated to 70 °C. After 14 h, the reaction mixture was cooled to 23 °C, Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (0.037 g) was added, and the reaction mixture was concentrated *in vacuo*. The reaction mixture was partitioned in Et<sub>2</sub>O (10 mL) and brine (10 mL). The layers were separated and the aqueous layer was extracted with Et<sub>2</sub>O (2 x 5 mL). The combined organic layers were dried with MgSO<sub>4</sub> and concentrated *in vacuo*. Purification by flash chromatography (5:95 EtOAc:hexanes) afforded alcohol **S25** as a white solid (0.010 g, 71%): mp = 99–100 °C; <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.31–7.29 (m, 2H), 7.26–7.22 (m, 3H), 4.51 (d, *J* = 8.5, 1H), 3.05 (t, *J* = 9.2, 1H), 2.11–2.04 (m, 1H), 1.85–1.74 (m, 2H), 1.59–1.53 (m, 2H), 1.06 (s, 9H), 0.99 (t, *J* = 7.4, 3H), 0.93 (s, 9H), 0.85–0.84 (m, 2H), 0.60 (d, *J* = 6.8, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>) δ 145.0 (C), 128.1 (CH), 127.0 (2 CH), 81.8 (CH), 78.7 (CH), 50.3 (CH), 42.0 (CH), 28.9 (CH<sub>3</sub> and CH<sub>2</sub>), 28.6 (CH<sub>3</sub>), 21.8 (C), 21.0 (C), 17.4 (CH<sub>3</sub>), 11.2 (CH<sub>2</sub>), 9.6 (CH<sub>3</sub>); IR (ATR) 3346, 1078, 1057, 909, 823 cm<sup>-1</sup>; HRMS (ESI) *m/z* calcd for C<sub>22</sub>H<sub>37</sub>O (M – OH)<sup>+</sup> 345.2608, found 345.2606.

## V. Control Experiments



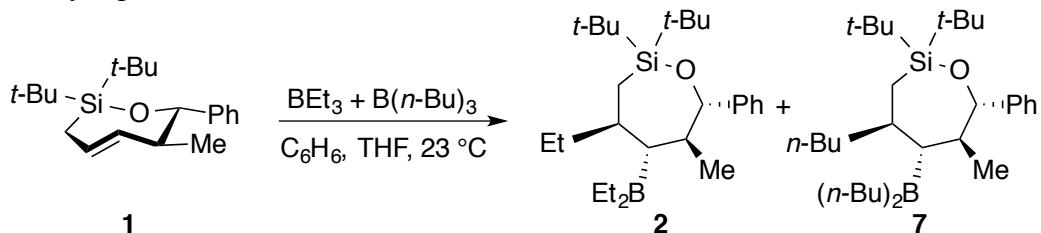
**Amine 4 and Alkene 3.** To a solution of trialkylborane **2** (0.018 g, 0.043 mmol) and mesitylene (0.0020 mL, 0.014 mmol, internal standard) in C<sub>6</sub>D<sub>6</sub> (0.600 mL) in a NMR tube was added <sup>7</sup>BuNH<sub>2</sub> (0.050 mL, 0.50 mmol). The NMR tube was flame-sealed under vacuum. After 24 h, the reaction mixture was heated to 80 °C. After 24 h, alkene **3** was formed in 95% yield based on <sup>1</sup>H NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of the benzylic CH peak ( $\delta$  4.70). The spectral data are consistent with the data reported above for alkene **3**. Amine **4** was formed in 85% yield based on <sup>1</sup>H NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71).

and the area of the NH peak ( $\delta$  3.77):  $^1\text{H}$  NMR (600 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  3.77 (br s, 1H), 2.84–2.80 (m, 2H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  42.8 ( $\text{CH}_2$ ), 36.2 ( $\text{CH}_2$ ), 9.7 ( $\text{CH}_3$ );  $^{11}\text{B}$  NMR (128 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  45.9.

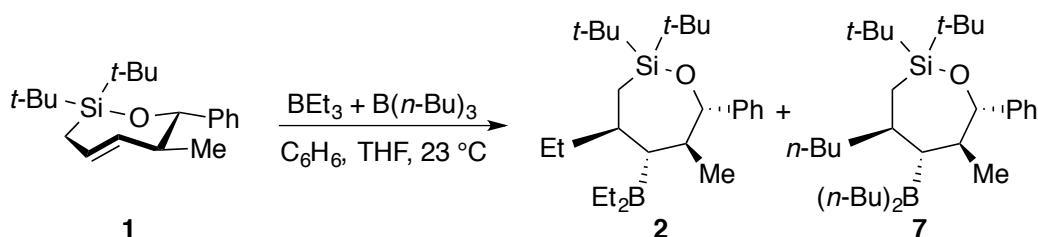


**Alkene 3 and Trialkylborane 6.** To a solution of trialkylborane **2** (0.013 g, 0.031 mmol) and mesitylene (0.0020 mL, 0.014 mmol, internal standard) in  $\text{C}_6\text{D}_6$  (0.550 mL) in a NMR tube was added allylbenzene **5** (0.006 mL, 0.045 mmol). The NMR tube was flame-sealed under vacuum. After 24 h, the reaction mixture was heated to 80 °C. After 24 h, alkene **3** was formed in 96% yield based on  $^1\text{H}$  NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of the benzylic CH peak ( $\delta$  4.70). The spectral data are consistent with the data reported above for alkene **3**. Trialkylborane **6** was formed in 74% yield based on  $^1\text{H}$  NMR spectroscopic analysis of the area of a peak of the standard ( $\delta$  6.71) and the area of the benzylic  $\text{CH}_2$  peak ( $\delta$  2.51):  $^1\text{H}$  NMR (600 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  2.51 (t,  $J$  = 7.6, 2H), 1.69–1.63 (m, 2H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{C}_6\text{D}_6$ , diagnostic peaks)  $\delta$  39.8 ( $\text{CH}_2$ ), 27.9 ( $\text{CH}_2$ ), 27.3 ( $\text{CH}_2$ ), 20.0 ( $\text{CH}_2$ ), 8.7 ( $\text{CH}_3$ );  $^{11}\text{B}$  NMR (128 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  85.9.

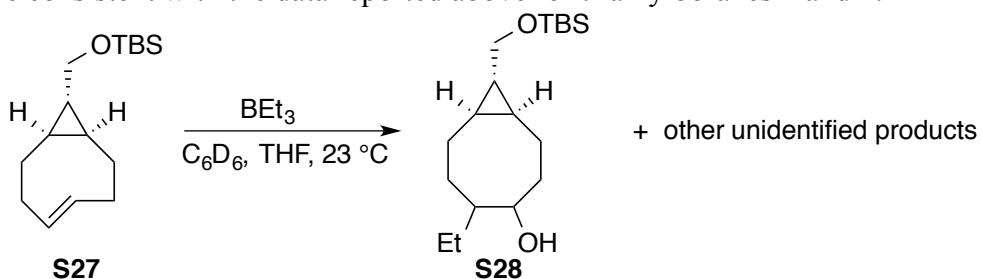
**Alcohol S26.** Oxidation of trialkylborane **6** was performed to confirm the identity of **6**. The above reaction mixture was transferred to a flask containing 1 M NaOH (1.0 mL) and 30%  $\text{H}_2\text{O}_2$  (0.5 mL) using  $\text{Et}_2\text{O}$  (2 mL). The layers were separated and the aqueous layer was extracted with  $\text{Et}_2\text{O}$  (2  $\times$  3 mL). The combined organic layers were dried ( $\text{MgSO}_4$ ) and concentrated *in vacuo*. The spectral data for alcohol **S26** matched the data previously reported.<sup>8</sup>



**Crossover experiment with  $\text{BEt}_3$  (5 equiv) and  $\text{B}(n\text{-Bu})_3$  (5 equiv).** To a solution of *trans*-alkene **1** (0.007 g, 0.021 mmol) in  $\text{C}_6\text{H}_6$  (0.200 mL) was added tri-*n*-butylborane (0.024 mL, 0.098 mmol), immediately followed by triethylborane (0.100 mL, 1.0 M in THF, 0.10 mmol). After 2 h, 1 M NaOH (0.80 mL) was added, followed by 30%  $\text{H}_2\text{O}_2$  in  $\text{H}_2\text{O}$  (0.40 mL) (*caution the ensuing reaction was exothermic*). After 20 min,  $\text{Et}_2\text{O}$  (5 mL) and brine (5 mL) were added and the layers were separated. The aqueous layer was extracted with  $\text{Et}_2\text{O}$  (2  $\times$  5 mL). The combined organic layers were dried ( $\text{MgSO}_4$ ) and concentrated *in vacuo*.  $^1\text{H}$  NMR spectroscopic analysis of the unpurified reaction mixture showed a mixture of trialkylboranes **2** and **7** in a 83:17 (**2:7**) ratio. The spectral data are consistent with the data reported above for trialkylboranes **2** and **7**.

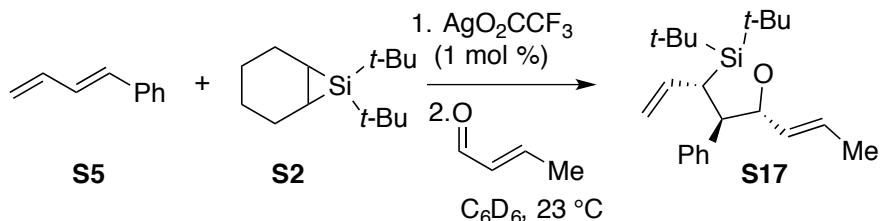


**Crossover experiment with  $\text{BEt}_3$  (5 equiv) and  $\text{B}(n\text{-Bu})_3$  (20 equiv).** To a solution of *trans*-alkene **1** (0.007 g, 0.021 mmol) in  $\text{C}_6\text{H}_6$  (0.200 mL) was added tri-*n*-butylborane (0.100 mL, 0.410 mmol), immediately followed by triethylborane (0.100 mL, 1.0 M in THF, 0.10 mmol). After 2 h, 1 M NaOH (0.80 mL) was added, followed by 30%  $\text{H}_2\text{O}_2$  in  $\text{H}_2\text{O}$  (0.40 mL) (*caution the ensuing reaction was exothermic*). After 20 min,  $\text{Et}_2\text{O}$  (5 mL) and brine (5 mL) were added and the layers were separated. The aqueous layer was extracted with  $\text{Et}_2\text{O}$  (2 x 5 mL). The combined organic layers were dried ( $\text{MgSO}_4$ ) and concentrated *in vacuo*.  $^1\text{H}$  NMR spectroscopic analysis of the unpurified reaction mixture showed a mixture of trialkylboranes **2** and **7** in a 59:41 (**2:7**) ratio. The spectral data are consistent with the data reported above for trialkylboranes **2** and **7**.

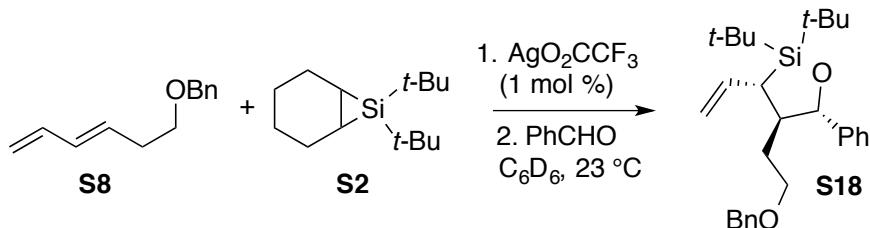


**Alcohol S28.** To a solution of *trans*-alkene **S27** (0.017 g, 0.062 mmol) in C<sub>6</sub>D<sub>6</sub> (0.610 mL) in a J. Young NMR tube was added triethylborane (0.370 mL, 1.0 M in THF, 0.37 mmol). The reaction mixture was monitored by <sup>1</sup>H NMR spectroscopy until *trans*-alkene **S28** was consumed. After 13 d, 1 M NaOH (1.0 mL) was added, followed by 30% H<sub>2</sub>O<sub>2</sub> in H<sub>2</sub>O (0.50 mL) (*caution the ensuing reaction was exothermic*). After 20 min, Et<sub>2</sub>O (5 mL) and brine (5 mL) were added and the layers were separated. The aqueous layer was extracted with Et<sub>2</sub>O (2 x 10 mL). The combined organic layers were dried (MgSO<sub>4</sub>) and concentrated *in vacuo*. Purification by flash chromatography (hexanes) afforded alcohol **S28** as a clear oil that could not be purified further (0.004 g, 21% unpurified yield): <sup>1</sup>H NMR (600 MHz, C<sub>6</sub>D<sub>6</sub>) δ 3.46–3.44 (m, 2H), 3.24–3.21 (m, 1H), 2.01–1.96 (m, 1H), 1.85–1.80 (m, 2H), 1.75–1.68 (m, 1H), 1.62–1.56 (m, 2H), 1.52–1.42 (m, 1H), 1.32–1.19 (m, 3H), 1.01 (s, 9H), 0.88–0.79 (m, 4H), 0.57–0.52 (m, 1H), 0.48–0.43 (m, 2H), 0.39–0.34 (m, 1H), 0.08 (s, 6H); <sup>13</sup>C NMR (150 MHz, C<sub>6</sub>D<sub>6</sub>) δ 75.1 (CH), 67.1 (CH<sub>2</sub>), 44.2 (CH), 40.7 (CH<sub>2</sub>), 33.5 (CH<sub>2</sub>), 27.4 (CH), 26.5 (CH<sub>3</sub>), 26.3 (CH<sub>2</sub>), 24.9 (CH<sub>2</sub>), 24.0 (CH<sub>2</sub>), 21.51 (CH), 21.50 (CH), 18.9 (C), 13.7 (CH<sub>3</sub>), -4.6 (CH<sub>3</sub>); IR (ATR) 3406, 1074, 836 cm<sup>-1</sup>; HRMS (APCI) *m/z* calcd for C<sub>18</sub>H<sub>36</sub>NaO<sub>2</sub>Si (M + Na)<sup>+</sup> 335.2377, found 335.2374.

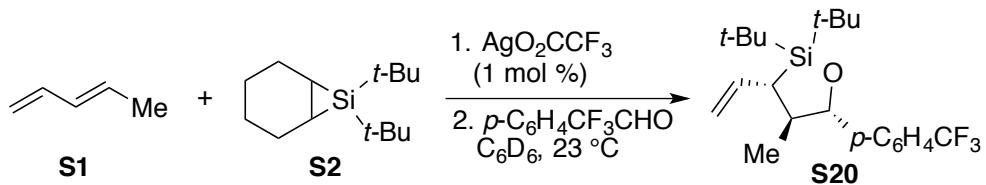
## VI. Rearrangement of Seven-Membered-Ring *trans*-Alkenes



**Alkene S17.** The synthesis of alkene **S17** was adapted from a reported procedure.<sup>2</sup> To a solution of diene **S5** (0.011 g, 0.084 mmol) and cyclohexene silacyclop propane **S2** (0.021 g, 0.094 mmol) in  $\text{C}_6\text{H}_6$  (0.50 mL) was added  $\text{AgOCOCF}_3$  (0.024 mL, 0.010 M in  $\text{C}_6\text{H}_6$ , 0.3 mol %). After 10 min, crotonaldehyde (0.007 mL, 0.08 mmol) was added. After 1 d, the reaction mixture was concentrated *in vacuo*. Purification by flash chromatography (3:97 EtOAc:hexanes) afforded alkene **S17** as a colorless oil (0.015 g, 52% over three steps):  $^1\text{H}$  NMR (400 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  7.18–7.11 (m, 4H), 7.07–7.02 (m, 1H), 5.89 (ddd,  $J$  = 17.0, 10.3, 8.3, 1H), 5.71–5.62 (m, 1H), 5.59–5.53 (m, 1H), 4.83 (dt,  $J$  = 17.0, 1.5, 1H), 4.78–4.74 (m, 1H), 4.47 (ddt,  $J$  = 10.3, 5.2, 1.0, 1H), 3.04 (dd,  $J$  = 12.6, 10.3, 1H), 2.73 (ddt,  $J$  = 12.6, 8.3, 1.0, 1H), 1.47 (dt,  $J$  = 6.2, 1.3, 3H), 1.20 (s, 9H), 1.17 (s, 9H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  142.1 (C), 138.1 (CH), 132.6 (CH), 129.0 (CH), 128.9 (CH), 127.2 (CH), 126.0 (CH), 113.4 (CH<sub>2</sub>), 84.4 (CH), 58.4 (CH), 40.4 (CH), 28.6 (CH<sub>3</sub>), 28.4 (CH<sub>3</sub>), 22.4 (C), 22.2 (C), 18.0 (CH<sub>3</sub>); IR (ATR) 1628, 1472, 837  $\text{cm}^{-1}$ ; HRMS (ESI)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{33}\text{Si}$  ( $\text{M} - \text{OH}$ )<sup>+</sup> 325.2346, found 325.2347.



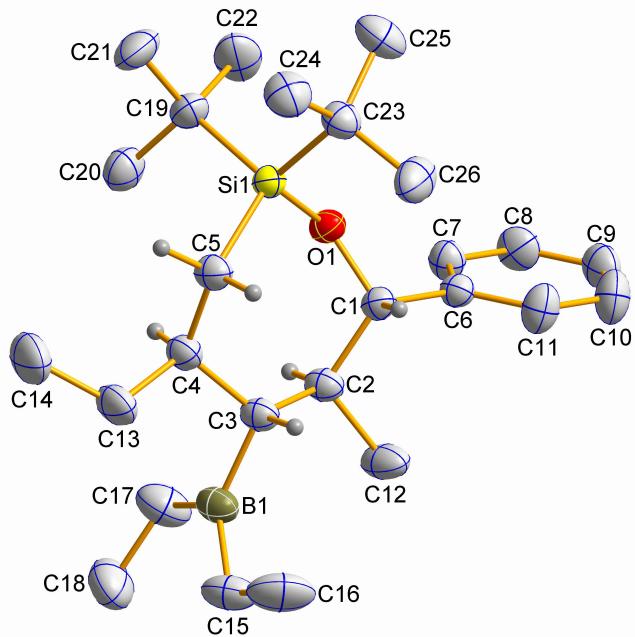
**Alkene S18.** The synthesis of alkene **S18** was adapted from a reported procedure.<sup>2</sup> To a solution of diene **S8** (0.014 g, 0.074 mmol) and cyclohexene silacyclop propane **S2** (0.021 g, 0.094 mmol) in  $\text{C}_6\text{D}_6$  (0.50 mL) was added  $\text{AgOCOCF}_3$  (0.025 mL, 0.030 M in  $\text{C}_6\text{H}_6$ , 1 mol %). After 10 min, benzaldehyde (0.0075 mL, 0.074 mmol) was added. After 1 d, the reaction mixture was concentrated *in vacuo*. Purification by flash chromatography (3:97 EtOAc:hexanes) afforded alkene **S18** as a colorless oil (0.020 g, 63% over three steps):  $^1\text{H}$  NMR (600 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  7.52 (d,  $J$  = 7.7, 2H), 7.24 (d,  $J$  = 7.6, 2H), 7.21–7.18 (m, 2H), 7.14–7.07 (m, 4H), 6.00–5.94 (m, 1H), 5.00 (d,  $J$  = 17.1, 1H), 4.89 (d,  $J$  = 10.3, 1H), 4.64 (d,  $J$  = 9.5, 1H), 4.18 (s, 2H), 3.29–3.23 (m, 2H), 2.36–2.27 (m, 2H), 1.75–1.71 (m, 2H), 1.21 (s, 9H), 1.13 (s, 9H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{C}_6\text{D}_6$ )  $\delta$  144.2 (C), 140.1 (CH), 139.6 (C), 128.84 (CH), 128.82 (CH), 128.7 (CH), 128.1 (CH), 127.91 (CH), 127.87 (CH), 112.6 (CH<sub>2</sub>), 85.4 (CH), 73.2 (CH<sub>2</sub>), 68.1 (CH<sub>2</sub>), 49.6 (CH), 38.5 (CH), 31.6 (CH<sub>2</sub>), 28.59 (CH<sub>3</sub>), 28.58 (CH<sub>3</sub>), 22.5 (C), 22.2 (C); IR (ATR) 1625, 1472, 1002, 822  $\text{cm}^{-1}$ ; HRMS (APCI)  $m/z$  calcd for  $\text{C}_{28}\text{H}_{41}\text{O}_2\text{Si}$  ( $\text{M} + \text{H}$ )<sup>+</sup> 437.2870, found 437.2873.



**Alkene S20.** The synthesis of alkene **S20** was adapted from a reported procedure.<sup>2</sup> To a solution of diene **S1** (0.0080 mL, 0.080 mmol) and cyclohexene silacyclopropane **S2** (0.022 g, 0.098 mmol) in  $\text{C}_6\text{H}_6$  (0.50 mL) was added  $\text{AgOCOCF}_3$  (0.022 mL, 0.010 M in  $\text{C}_6\text{H}_6$ , 0.3 mol %). After 10 min, *p*-(trifluoromethyl)benzaldehyde (0.010 mL, 0.073 mmol) was added. After 4 d, the reaction mixture was concentrated *in vacuo*. Purification by flash chromatography (3:97 EtOAc:hexanes) afforded alkene **S20** as a colorless oil (0.009 g, 29% over three steps):  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.61 (d,  $J = 8.2$ , 2H), 7.52 (d,  $J = 8.2$ , 2H), 5.93 (ddd,  $J = 17.0$ , 10.3, 8.8, 1H), 5.02–4.95 (m, 2H), 4.37 (d,  $J = 10.0$ , 1H), 2.09–2.04 (m, 1H), 1.98–1.87 (m, 1H), 1.17 (s, 9H), 1.12 (s, 9H), 0.90 (d,  $J = 6.3$ , 3H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  147.3 (C), 137.9 (CH), 129.9 (q,  $J = 32.1$ , C), 126.9 (CH), 125.4 (q,  $J = 3.6$ , CH), 124.4 (q,  $J = 270.7$ , C), 113.3 (CH<sub>2</sub>), 85.8 (CH), 47.4 (CH), 40.2 (CH), 28.1 (CH<sub>3</sub>), 28.0 (CH<sub>3</sub>), 22.0 (C), 21.8 (C), 15.8 (CH<sub>3</sub>);  $^{19}\text{F}$  NMR (376 MHz,  $\text{CDCl}_3$ )  $\delta$  -65.6; IR (ATR) 1622, 1473, 1323, 1067, 823  $\text{cm}^{-1}$ ; HRMS (APCI) *m/z* calcd for  $\text{C}_{21}\text{H}_{31}\text{F}_3\text{NaOSi}$  ( $\text{M} + \text{Na}$ )<sup>+</sup> 407.1988, found 407.1982. Anal. Calcd for  $\text{C}_{21}\text{H}_{31}\text{F}_3\text{OSi}$ : C, 65.59; H, 8.13. Found: C, 65.84; H, 8.27.

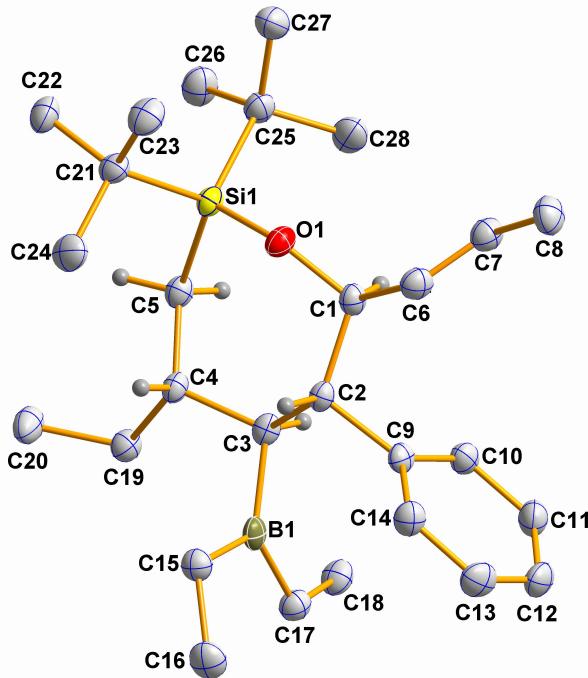
## VII. X-ray Crystallographic Data

### A. X-ray Data Collection, Structure Solution, and Refinement for Trialkylborane **2** (CCDC 1539169)



**Single-crystal structure determination.** The X-ray intensity data of the crystal were recorded on a Bruker D8 APEX-II CCD system using graphite-monochromated and 0.5 mm-MonoCap-collimated Mo-K $\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ) with the  $\omega$  scan method at 100 K. The temperature was controlled by an Oxford Cryosystems 700+ Cooler. The dataset was processed with the INTEGRATE program of the APEX2 software for reduction and cell refinement.<sup>9</sup> Multi-scan absorption corrections were applied by using the SCALE program for the area detector. The structure was solved by intrinsic phasing methods (SHELXT)<sup>10</sup> and the structure model was completed and refined using the full-matrix least-square methods on  $F^2$  (SHELXL).<sup>11</sup> Non-hydrogen atoms were refined with anisotropic displacement parameters, and hydrogen atoms on carbons were placed in idealized positions (C-H = 0.95-1.00  $\text{\AA}$ ) and included as riding with  $U_{\text{iso}}(\text{H}) = 1.2$  or 1.5  $U_{\text{eq}}(\text{non-H})$ .

**B. X-ray Data Collection, Structure Solution, and Refinement for Trialkylborane *ent*-**10** (CCDC 1539170)**



**Single-crystal structure determination.** The X-ray intensity data of the crystal were recorded on a Bruker D8 APEX-II CCD system using graphite-monochromated and 0.5 mm-MonoCap-collimated Mo-K $\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ) with the  $\omega$  scan method at 100 K. The temperature was controlled by an Oxford Cryosystems 700+ Cooler. The dataset was processed with the INTEGRATE program of the APEX2 software for reduction and cell refinement.<sup>9</sup> Multi-scan absorption corrections were applied by using the SCALE program for the area detector. The structure was solved by intrinsic phasing methods (SHELXT)<sup>10</sup> and the structure model was completed and refined using the full-matrix least-square methods on  $F^2$  (SHELXL).<sup>11</sup> Non-hydrogen atoms were refined with anisotropic displacement parameters, and hydrogen atoms on carbons were placed in idealized positions (C-H = 0.95-1.00  $\text{\AA}$ ) and included as riding with  $U_{\text{iso}}(\text{H}) = 1.2$  or 1.5  $U_{\text{eq}}(\text{non-H})$ .

## VIII. Stereochemical Proofs

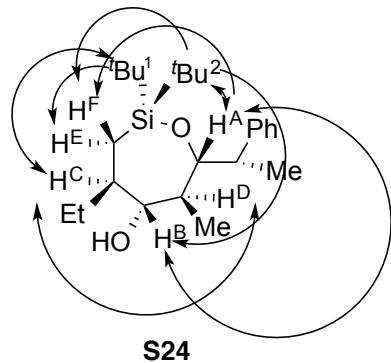
### A. Assignment by analogy

The relative stereochemistry of *trans*-alkenes **S3**, **S4**, **S7**, **S10–S16** and alkenes **S17**, **S18**, and **S20** were assigned by analogy to previously reported structures.<sup>2</sup> The relative stereochemistry of trialkylboranes **7–18** were assigned by analogy to X-ray crystallographic structures of compounds **2** and *ent*-**10**. The relative stereochemistry of alcohol **20** was assigned by analogy to alcohols **S23** and **S24** (nOe data reported below).

### B. General Procedure for DPFGSE-nOe Experiments

All DPFGSE-nOe data were collected on degassed C<sub>6</sub>D<sub>6</sub> samples with a mixing time of 0.50 seconds. All peaks in the <sup>1</sup>H NMR spectra were assigned using <sup>1</sup>H/<sup>1</sup>H COSY, <sup>1</sup>H/<sup>13</sup>C HSQC, and <sup>1</sup>H NMR chemical shifts.

### C. DPFGSE-nOe Data



#### Relevant nOe Data for Alcohol S24 (C<sub>6</sub>D<sub>6</sub>)

H<sup>A</sup> irradiated: H<sup>B</sup> (2.1%), H<sup>F</sup> (0.7%)

H<sup>B</sup> irradiated: H<sup>A</sup> (2.2%)

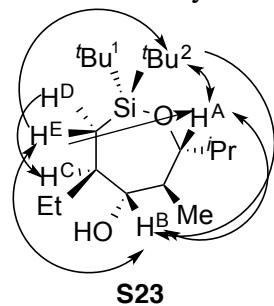
H<sup>C</sup> irradiated: H<sup>D</sup> (2.1%), 'Bu<sup>1</sup> (3.3%)

H<sup>D</sup> irradiated: H<sup>C</sup> (2.4%)

'Bu<sup>1</sup> irradiated: H<sup>C</sup> (0.8%), H<sup>D</sup> (0.4%), H<sup>E</sup> (0.5%)

'Bu<sup>2</sup> irradiated: H<sup>A</sup> (0.9%), H<sup>B</sup> (0.6%), H<sup>F</sup> (0.8%)

Note: There was no nOe observed between H<sup>A</sup> and H<sup>C</sup>. These data indicate the relative stereochemistry shown above.



#### Relevant nOe Data for Alcohol S23 (C<sub>6</sub>D<sub>6</sub>)

H<sup>A</sup> irradiated: H<sup>B</sup> (1.6%)

H<sup>B</sup> irradiated: H<sup>A</sup> (1.6%)

H<sup>D</sup> irradiated: H<sup>C</sup> (1.2%)

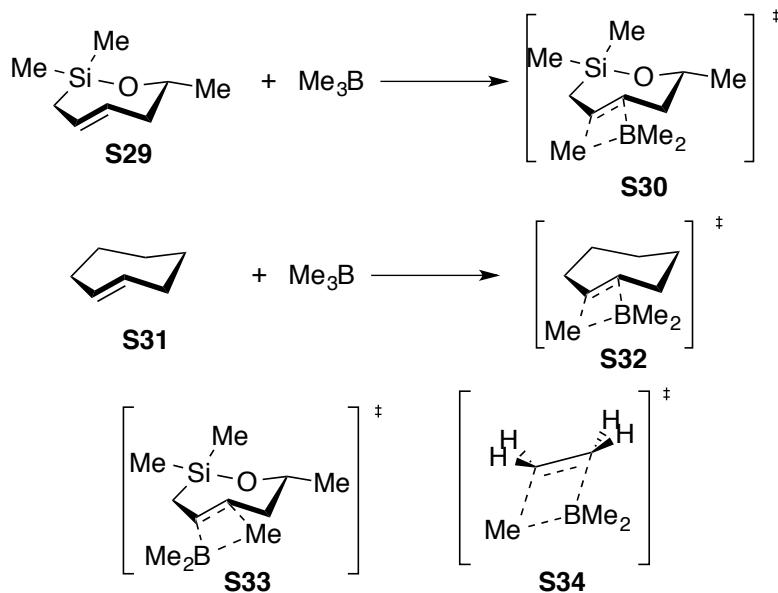
H<sup>E</sup> irradiated: H<sup>A</sup> (0.7%), H<sup>B</sup> (2.0%), 'Bu<sup>2</sup> (2.7%)

'Bu<sup>2</sup> irradiated: H<sup>A</sup> (0.7%), H<sup>B</sup> (0.4%)

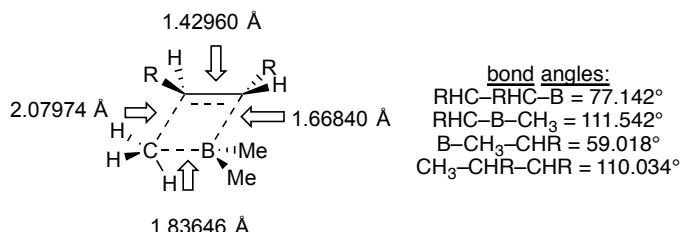
Note: There was no nOe observed between H<sup>A</sup> and H<sup>D</sup>. These data indicate the relative stereochemistry shown above.

## IX. Computational Investigations of the Carboboration Reaction

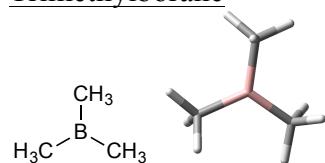
Computational studies of the carboboration reaction of *trans*-oxasilacycloheptenes were performed on model compounds. A conformational search of a model compound using the Merck Molecular Force Field (MMFF) as implemented in Spartan16<sup>12</sup> produced structure **S29** that resides in a conformation that resembles the X-ray crystal structure of a *trans*-oxasilacycloheptene.<sup>13</sup> This structure was further minimized using semi-empirical methods (PM3), then Hartree–Fock methods (HF/6-31G(d)) and then density functional methods (B3LYP/6-31G(d)). Additional optimization of **S29** was performed with Gaussian09<sup>14</sup> (M06-2X/6-311+G(2d,p)) using the integrated equation formalism of the polarizable continuum model (IEFPCM) with THF as the solvent; the structures from the different methods were not significantly different. The minimization protocol was repeated for trimethylborane, ethylene, and *trans*-cycloheptene **S31**. All structures were determined to be energy minima by vibrational calculations, which showed no imaginary frequencies.



Transition structures for carboboration reactions were optimized. The transition structures were first identified at the B3LYP/6-31G(d) level without solvent (as implemented in Spartan16) by reducing the distance between the boron atom and the alkene. At approximately 2.0 Å, the energy of the system, which had slowly risen, dropped significantly. The highest energy structures were minimized at the B3LYP/6-31G(d) level to obtain transition structures. All transition structures structurally resembled those identified for hydroboration.<sup>15</sup> These structures were optimized further using Gaussian09 (M06-2X/6-311+G(2d,p)) with the polarizable continuum model (IEFPCM) with THF as the solvent. The transition state for the formation of **S33**, shown below, is representative. That these structures were transition states was confirmed by vibration calculations, which showed only one imaginary frequency (421 cm<sup>-1</sup> in the case of **S33**). Transition states were also located for the other carboboration reactions.

**Summary of Computational Data**

Transition State Description and Number	Transition State Structure	$\Delta\Delta E^\ddagger$ (kcal/mol)
ethylene and trimethylborane (S34)		21.4
<i>trans</i> -cycloheptene and trimethylborane (S32)		1.13
Major regioisomer and trimethylborane (S30)		3.42
Minor regioisomer and trimethylborane (S33)		8.60

Trimethylborane

Gaussian09: MO6-2X/6-311+G(2d,p), THF (IEPCM) (note: convergence required the “calcfc” keyword)

Energy: -144.56553199

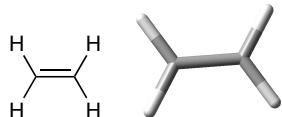
Number of imaginary vibrations: 0

Table S1: Cartesian Coordinates for Trimethylborane

Center	Atomic	X (Å)	Y (Å)	Z (Å)

Number	Number			
1	5	-0.002519	-0.002270	-0.002306
2	6	-1.547399	-0.286728	-0.004649
3	1	-1.822489	-1.338963	-0.090029
4	1	-1.964146	0.095807	0.936065
5	1	-2.051388	0.280371	-0.793330
6	6	1.021381	-1.192839	0.008940
7	1	1.027780	-1.613025	-1.006428
8	1	2.048073	-0.913572	0.251342
9	1	0.702467	-2.010617	0.660518
10	6	0.524681	1.477920	-0.005029
11	1	1.001986	1.668609	0.964613
12	1	1.318374	1.612244	-0.746694
13	1	-0.240035	2.240378	-0.160098

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ethylene

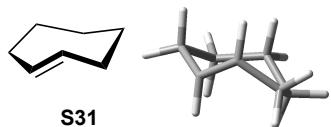
Gaussian09: MO6-2X/6-311+G(2d,p), THF (IEPCM)

Energy: -78.56741412

Number of imaginary vibrations: 0

Table S2: Cartesian Coordinates for Ethylene

Center Number	Atomic Number	X (Å)	Y (Å)	Z (Å)
1	6	0.661928	0.000000	0.000004
2	6	-0.661928	0.000000	-0.000004
3	1	1.229122	0.923425	-0.000010
4	1	1.229122	-0.923425	0.000023
5	1	-1.229122	-0.923425	0.000010
6	1	-1.229122	0.923425	-0.000023

*trans*-cycloheptene

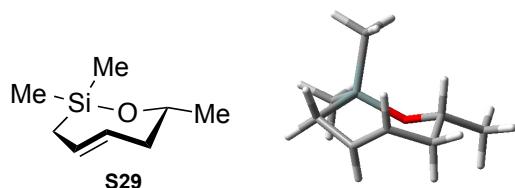
Gaussian09: MO6-2X/6-311+G(2d,p), THF (IEPCM)

Energy: -273.85136156 au

Number of imaginary vibrations: 0

Table S3: Cartesian Coordinates for *trans*-Cycloheptene

Center Number	Atomic Number	X (Å)	Y (Å)	Z (Å)
1	1	0.366651	-1.192109	1.516442
2	6	0.569537	-1.241617	0.448757
3	6	-0.471866	-1.298059	-0.377507
4	1	-0.277247	-1.259531	-1.450226
5	6	1.802945	-0.520468	-0.018030
6	1	2.580732	-0.505908	0.745942
7	1	2.230129	-0.952674	-0.924917
8	6	-1.735290	-0.631832	0.068887
9	1	-1.912206	-0.801981	1.133790
10	1	-2.633225	-0.907269	-0.486698
11	6	1.280657	0.926316	-0.321130
12	1	1.106295	1.003942	-1.398230
13	1	2.060024	1.653479	-0.079358
14	6	-1.393533	0.870076	-0.171050
15	1	-2.189564	1.486616	0.256728
16	1	-1.402325	1.055856	-1.250149
17	6	-0.033860	1.367129	0.393683
18	1	0.023297	1.132081	1.461294
19	1	-0.064107	2.458236	0.333714

trans-oxasilacycloheptene model compound

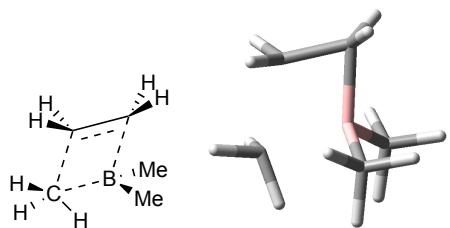
Gaussian09: MO6-2X/6-311+G(2d,p), THF (IEPCM)

Energy: -679.16022095 au

Number of imaginary vibrations: 0

Table S4: Cartesian Coordinates for **S29**

Center Number	Atomic Number	X (Å)	Y (Å)	Z (Å)
1	14	1.178548	-0.253742	0.014051
2	8	-0.346764	-0.773024	-0.452129
3	6	1.399488	1.638059	-0.297801
4	6	-2.149763	0.867118	-0.007475
5	6	-1.583284	-0.563681	0.233788
6	6	2.346153	-1.276848	-1.011076
7	6	1.386466	-0.543256	1.846331
8	1	1.712053	2.100181	0.641148
9	1	2.089605	1.903652	-1.096891
10	1	-3.031642	1.002285	0.622138
11	1	-2.450694	0.947324	-1.056037
12	1	2.209942	-2.341069	-0.810416
13	1	2.171437	-1.103567	-2.074820
14	1	3.381839	-1.011299	-0.788802
15	1	2.425878	-0.361005	2.129021
16	1	1.134861	-1.573608	2.107406
17	1	0.756704	0.123624	2.438573
18	6	-2.557574	-1.630169	-0.227600
19	1	-3.521174	-1.504520	0.269125
20	1	-2.173933	-2.625584	-0.002719
21	1	-2.708284	-1.550080	-1.305990
22	1	-1.408644	-0.672836	1.311668
23	6	-1.008217	1.783474	0.300783
24	1	-0.757378	1.923315	1.351813
25	6	-0.039435	1.864371	-0.615310
26	1	-0.319138	1.645348	-1.644733

transition state for addition of trimethylborane to ethylene

Gaussian09: MO6-2X/6-311+G(2d,p), THF (IEPCM)

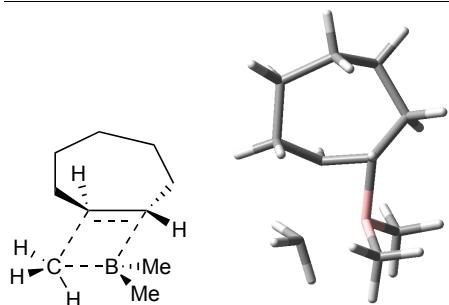
Energy: -223.09893940

Number of imaginary vibrations: 1 (-511.65 cm<sup>-1</sup>)

Table S5: Cartesian Coordinates for Carboboration of Ethylene

Center Number	Atomic Number	X (Å)	Y (Å)	Z (Å)
1	5	0.336308	-0.143723	0.076425
2	6	1.025291	-1.023351	-1.072564
3	1	1.535102	-1.887425	-0.635602
4	1	1.772920	-0.452910	-1.634703
5	1	0.302322	-1.412776	-1.797702
6	6	1.246332	0.642781	1.136328
7	1	1.829046	-0.071271	1.726092
8	1	0.656148	1.237203	1.840695
9	1	1.952101	1.321469	0.644508
10	6	-0.363418	1.329846	-0.853206
11	1	-0.551585	2.177672	-0.200917
12	1	-1.062051	1.334177	-1.685150
13	1	0.631288	1.442634	-1.284339
14	1	-0.905080	-2.028230	0.529373
15	6	-0.862006	-0.996624	0.852514
16	6	-1.533228	-0.034966	0.033769
17	1	-0.881952	-0.837591	1.922449
18	1	-1.967036	-0.387830	-0.890681
19	1	-2.070581	0.777375	0.502805

transition state for addition of trimethylborane to *trans*-cycloheptene



Gaussian09: MO6-2X/6-311+G(2d,p), THF (IEPCM)

Energy: -418.41508651

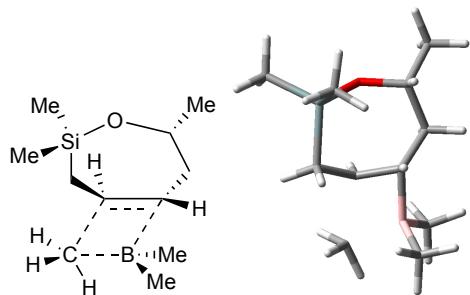
Number of imaginary vibrations: 1 (-383.28 cm<sup>-1</sup>)

Table S6: Cartesian Coordinates for Carboboration of *trans*-Cycloheptene

Center Number	Atomic Number	X (Å)	Y (Å)	Z (Å)
1	5	-1.694092	-0.168282	0.087693
2	6	-1.887529	1.436830	-0.698545
3	1	-2.966531	1.296187	-0.798786
4	1	-1.510974	1.678247	-1.691247
5	1	-1.717846	2.276579	-0.027253
6	6	-2.490839	0.075169	1.467451
7	1	-2.028277	0.849975	2.087580
8	1	-2.525653	-0.840943	2.064706
9	1	-3.523604	0.387195	1.273414
10	6	-2.288272	-1.230184	-0.963528
11	1	-2.288934	-2.243217	-0.548942
12	1	-1.729343	-1.270789	-1.904901
13	1	-3.324570	-0.982918	-1.219773
14	1	0.092038	-0.261576	1.533706
15	6	-0.086025	-0.545437	0.499878
16	6	0.010571	0.537940	-0.393863
17	1	0.110724	0.263840	-1.441158
18	6	0.691667	-1.779702	0.086169
19	1	0.701847	-2.494890	0.911066
20	1	0.226634	-2.281567	-0.764983
21	6	0.809326	1.729195	0.014574
22	1	0.606940	1.971688	1.060585
23	1	0.622622	2.614448	-0.596106
24	6	2.139446	-1.357985	-0.278073
25	1	2.196052	-1.172542	-1.355736
26	1	2.819545	-2.189861	-0.080758
27	6	2.279242	1.269039	-0.160179
28	1	2.919364	2.040835	0.273116

29	1	2.505415	1.245363	-1.231135
30	6	2.674202	-0.097690	0.443817
31	1	2.408822	-0.126989	1.505075
32	1	3.765462	-0.140712	0.406857

transition state for addition of trimethylborane to *trans*-oxasilacycloheptene, observed regiosomer



Gaussian09: MO6-2X/6-311+G(2d,p), THF (IEPCM)

Energy: -823.72030262

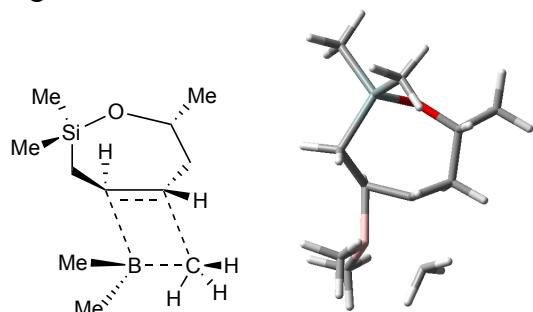
Number of imaginary vibrations: 1 (-421.12 cm<sup>-1</sup>)

Table S7: Cartesian Coordinates for Carboboration of **S30**, Observed Regiosomer

Center Number	Atomic Number	X (Å)	Y (Å)	Z (Å)
1	14	-1.825701	-0.849551	0.010507
2	8	-1.648464	0.722038	-0.503243
3	6	1.011774	0.454313	0.484493
4	6	0.777330	-0.603563	-0.448081
5	6	-0.122195	-1.732570	-0.096967
6	6	0.452472	1.803799	0.092533
7	6	-1.072971	1.798773	0.245582
8	6	-3.030120	-1.632011	-1.167270
9	6	-2.420460	-0.904335	1.776236
10	1	0.768558	0.182299	1.513002
11	1	0.708848	-0.268075	-1.477606
12	1	0.116366	-2.133513	0.891272
13	1	-0.113948	-2.542139	-0.828555
14	1	0.864284	2.604995	0.713018
15	1	0.699662	2.034884	-0.948272
16	1	-4.005817	-1.147374	-1.099463
17	1	-2.672366	-1.544091	-2.194972
18	1	-3.155436	-2.692466	-0.939418
19	1	-2.622794	-1.939957	2.059832
20	1	-3.343553	-0.332443	1.893265
21	1	-1.677252	-0.507269	2.470349
22	6	-1.702745	3.090232	-0.237934

23	1	-1.288750	3.941817	0.304608
24	1	-2.782586	3.071487	-0.088729
25	1	-1.500445	3.223952	-1.302561
26	1	-1.312690	1.656190	1.307356
27	5	2.555017	-0.045013	0.093791
28	6	3.276168	0.965689	-0.927403
29	1	2.751889	1.063588	-1.884363
30	1	4.294541	0.632104	-1.155132
31	1	3.350186	1.971226	-0.500402
32	6	2.539271	-1.653564	-0.792149
33	1	2.119043	-1.854314	-1.776168
34	1	2.341802	-2.492581	-0.129358
35	1	3.619109	-1.569027	-0.932157
36	6	3.331044	-0.471426	1.436572
37	1	4.311206	-0.909519	1.216636
38	1	2.772219	-1.202967	2.029775
39	1	3.492950	0.403661	2.074253

transition state for addition of trimethylborane to *trans*-oxasilacycloheptene, other regioisomer



Gaussian09: MO6-2X/6-311+G(2d,p), THF (IEPCM)

Energy: -823.71204962

Number of imaginary vibrations: 1 (-453.43 cm<sup>-1</sup>)

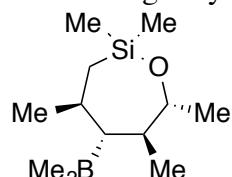
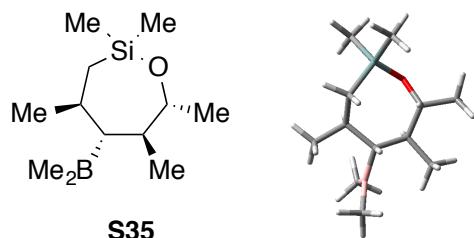
Table S8: Cartesian Coordinates for Carboboration of **S33**, Other Regioisomer

Center Number	Atomic Number	X (Å)	Y (Å)	Z (Å)
1	14	-1.782264	-0.930967	-0.018151
2	8	-1.678364	0.659479	-0.530780
3	6	0.966489	0.541095	0.398935
4	6	0.868172	-0.548122	-0.509861
5	6	-0.046558	-1.695069	-0.102322
6	6	0.389921	1.870791	0.021975
7	6	-1.138999	1.750002	0.205138
8	6	-2.977292	-1.733014	-1.197448
9	6	-2.414659	-0.973412	1.741206

10	1	0.807404	0.270630	1.440852
11	1	0.704131	-0.224701	-1.534876
12	1	0.235666	-2.091377	0.876944
13	1	0.013280	-2.520924	-0.814872
14	1	0.747330	2.687340	0.653185
15	1	0.615113	2.102879	-1.022484
16	1	-3.963948	-1.271072	-1.130395
17	1	-2.621005	-1.637415	-2.225176
18	1	-3.079236	-2.796759	-0.972109
19	1	-2.594550	-2.010320	2.035894
20	1	-3.353400	-0.423940	1.839240
21	1	-1.693195	-0.553015	2.445616
22	6	-1.838416	3.017958	-0.247538
23	1	-1.462405	3.883849	0.299368
24	1	-2.911851	2.933785	-0.077553
25	1	-1.665460	3.173607	-1.314030
26	1	-1.335958	1.595870	1.274306
27	5	2.499864	-0.447763	-0.125301
28	6	2.945243	-1.588156	0.913335
29	1	2.749123	-2.586465	0.509428
30	1	2.427788	-1.525744	1.876949
31	1	4.018571	-1.525119	1.124167
32	6	3.335923	-0.263534	-1.486508
33	1	3.262612	-1.166518	-2.100467
34	1	4.397895	-0.084893	-1.282226
35	1	2.976019	0.571509	-2.095862
36	6	2.929789	1.124481	0.718885
37	1	3.954286	0.766168	0.834399
38	1	2.594307	1.433333	1.706733
39	1	2.939099	1.977688	0.045039

### Computational Investigations of NBO Calculations

Additional calculations were performed to evaluate the bonding of the borane to explain the unusual chemical shift of the trialkylborane products. These studies were performed on a model compound, borane **S35**. A conformational search using the Merck Molecular Force Field (MMFF, with keywords “montecarlo” and “findboats” resulting in more conformers) as implemented in Spartan16<sup>12</sup> produced 13 low-energy conformers. One structure, the lowest energy one, was found to occupy a conformation that resembles the X-ray crystal structure of the product **2**. These structures were refined using semi-empirical methods (PM3), then Hartree–Fock methods (HF/3-31G(d)), which produced five unique conformers. These structures were optimized with Gaussian09<sup>14</sup> (M06-2X/6-311+G(2d,p)). The structure resembling the X-ray crystal structure was >0.9 kcal/mol lower in energy than all others. This structure was determined to be an energy minima by vibrational calculations, which showed no imaginary frequencies.

**S35**borane model compound (**S35**)**S35**

Gaussian09: M06-2X/6-311+G(2d,p)

Energy: -863.087061105 au

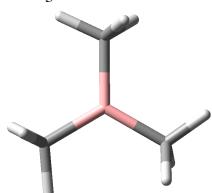
Number of imaginary vibrations: 0

Table S9: Cartesian Coordinates for **S35**

Standard orientation:

		Coordinates (Angstroms)		
Center Number	Atomic Number	X	Y	Z
1	8	1.539515	0.835524	0.663553
2	6	-1.277836	-0.101018	-0.389199
3	6	0.874306	-1.479241	-0.814397
4	6	-0.466738	-1.382640	-0.055304
5	14	2.246062	-0.511594	0.013509

6	6	0.654069	1.622090	-0.126475
7	6	-0.796916	1.246252	0.214125
8	6	2.966234	-1.453261	1.449541
9	6	3.579040	-0.043309	-1.210886
10	1	0.731703	-1.160866	-1.852942
11	1	1.192659	-2.527218	-0.866774
12	1	3.646735	-0.818071	2.019686
13	1	3.524170	-2.327078	1.106464
14	1	2.179876	-1.793960	2.125878
15	1	4.395358	0.484381	-0.713494
16	1	3.994688	-0.932623	-1.690645
17	1	3.185545	0.604837	-1.997247
18	6	1.011201	3.076226	0.140379
19	1	2.083374	3.206108	-0.009388
20	1	0.778015	3.336127	1.175832
21	1	0.482717	3.758196	-0.525336
22	6	-1.766543	2.351269	-0.228444
23	1	-1.560861	3.298787	0.267768
24	1	-2.804595	2.093830	-0.002978
25	1	-1.702510	2.510471	-1.309533
26	6	-1.311042	-2.623932	-0.375852
27	1	-2.238922	-2.660713	0.203399
28	1	-0.757416	-3.538688	-0.156820
29	1	-1.572940	-2.636918	-1.438434
30	5	-2.780356	-0.307581	0.090179
31	6	-3.959412	-0.358538	-0.947484
32	1	-4.946641	-0.175033	-0.518210
33	1	-3.977325	-1.377267	-1.359138
34	1	-3.804234	0.304873	-1.801667
35	6	-3.095694	-0.443324	1.626854
36	1	-2.243420	-0.712890	2.253230
37	1	-3.918200	-1.136941	1.820155
38	1	-3.450752	0.536480	1.973917
39	1	-1.257290	0.010096	-1.484256
40	1	-0.270307	-1.383626	1.024279
41	1	-0.830230	1.171131	1.308203
42	1	0.817576	1.419162	-1.197541

**BMe<sub>3</sub>:**

Gaussian09: MO6-2X/6-311+G(2d,p)

Energy: -144.564980142 au

Number of imaginary vibrations: 0

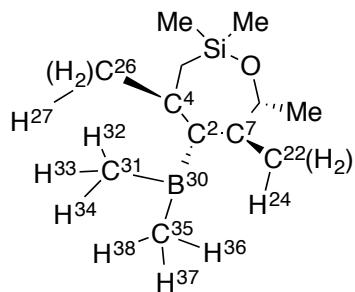
Table S10: Cartesian Coordinates for BMe<sub>3</sub>:

		Coordinates (Angstroms)		
Center Number	Atomic Number	X	Y	Z
1	6	0.637229	1.434376	-0.004682
2	1	-0.067021	2.254342	-0.150473
3	1	1.141985	1.588737	0.957363
4	1	1.430835	1.510990	-0.754608
5	6	-1.563668	-0.166486	-0.004731
6	1	-1.950506	0.228651	0.943357
7	1	-2.027530	0.451606	-0.779655
8	1	-1.919981	-1.192097	-0.108419
9	5	-0.001405	-0.001087	-0.003723
10	6	0.925029	-1.268994	0.007897
11	1	0.556601	-2.040065	0.689418
12	1	0.874554	-1.719525	-0.992609
13	1	1.976555	-1.070583	0.223339

Computational studies of <sup>11</sup>B NMR chemical shifts were performed using the method of Wrackmeyer. Employing the same computational models used in that work (GIAO method, B3LYP/6-311+(d,p)) reproduced the chemical shift for BMe<sub>3</sub> from that paper,<sup>16</sup> but that method suggested that the product should have a chemical shift of δ 88 ppm, which is not consistent with the experimental results.

A natural bond order calculation was performed on both **S35** and BMe<sub>3</sub> using the computational methods employed to find minima. The results of those calculations are summarized in Tables S11 and S12.

Table 11. Second Order Perturbation Theory Analysis of Fock Matrix in NBO Basis for S35



					E(2)	E(j)-E(i)	F(I,j)
Donor NBO (i)			Acceptor NBO (j)			a.u.	a.u.
3. BD	C2	C4	64. LP*	B30	2.61	0.71	0.040
4. BD	C2	C7	64. LP*	B30	3.45	0.71	0.046
11. BD	C4	C26	64. LP*	B30	1.48	0.74	0.031
18. BD	C7	C22	64. LP*	B30	1.24	0.74	0.028
19. BD	C7	H41	64. LP*	B30	0.62	0.61	0.018
30. BD	C22	H24	64. LP*	B30	8.01	0.64	0.066
32. BD	C26	H27	64. LP*	B30	8.52	0.64	0.068
34. BD	C26	H29	64. LP*	B30	1.08	0.64	0.024
37. BD	C31	H32	64. LP*	B30	0.86	0.62	0.021
38. BD	C31	H33	64. LP*	B30	13.61	0.61	0.084
39. BD	C31	H34	64. LP*	B30	2.68	0.62	0.038
40. BD	C35	H36	64. LP*	B30	0.56	0.62	0.017
41. BD	C35	H37	64. LP*	B30	3.31	0.62	0.042
42. BD	C35	H38	64. LP*	B30	12.92	0.61	0.082

Table 12. Second Order Perturbation Theory Analysis of Fock Matrix in NBO Basis for BMe<sub>3</sub>

					E(2)	E(j)- E(i)	F(I,j)
Donor NBO (i)			Acceptor NBO (j)		kcal/mol	a.u	a.u.
2. BD	C1	H3	17. LP*	B9	8.99	0.61	0.068
3. BD	C1	H4	17. LP*	B9	5.15	0.62	0.052
5. BD	C5	H6	17. LP*	B9	9.27	0.61	0.069
6. BD	C5	H7	17. LP*	B9	4.87	0.62	0.050
10. BD	C10	H11	17. LP*	B9	3.74	0.62	0.044
11. BD	C10	H12	17. LP*	B9	10.54	0.61	0.074

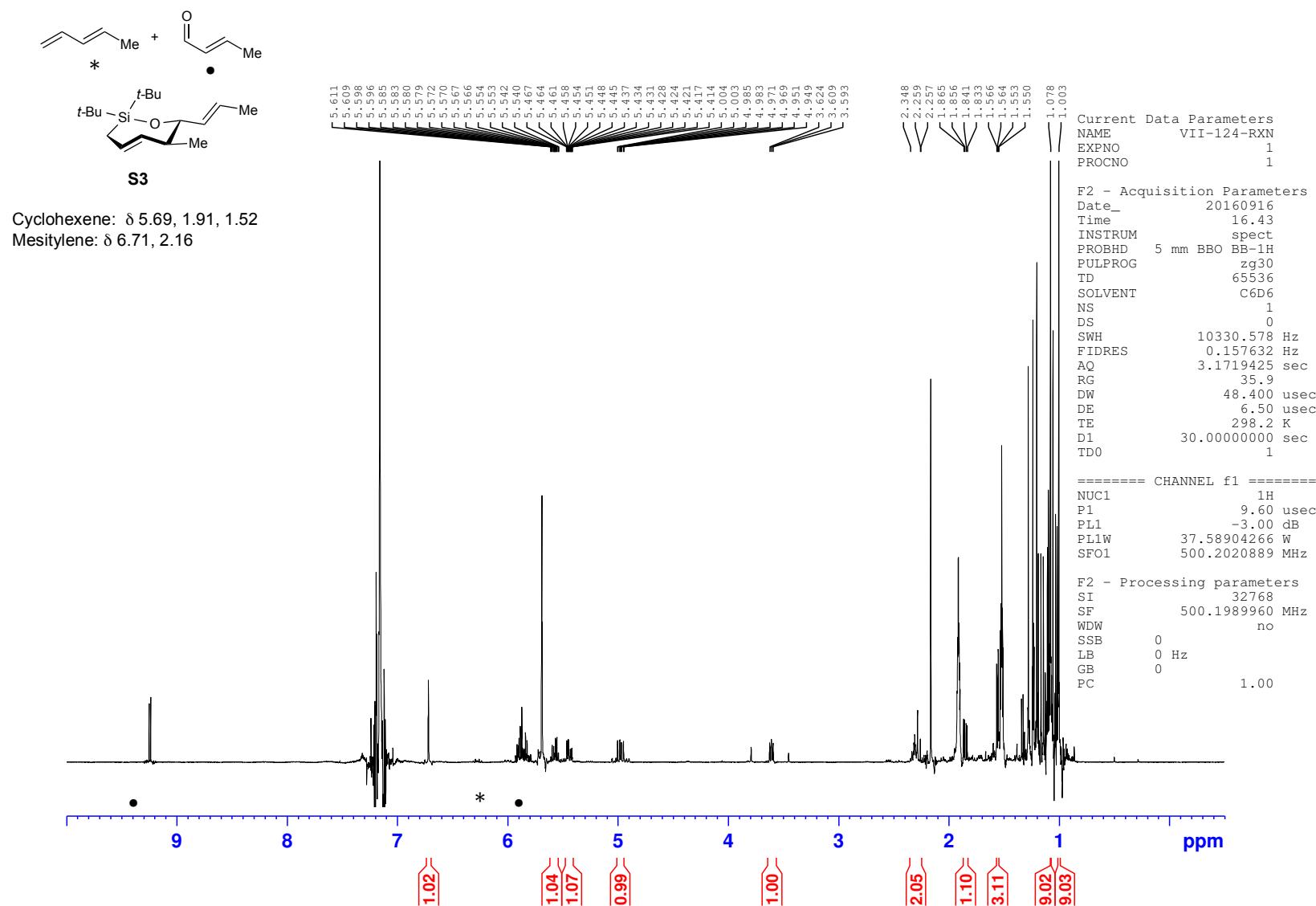
The NBO analysis suggests an explanation for the unusually high upfield chemical shift for the  $^{11}\text{B}$  atom of borane **S35**. As the electron density of a boron atom increases, the chemical shift should move upfield, so contributions to electron density of the boron atom was evaluated by NBO analysis, and the chemical shifts were compared to  $\text{BMe}_3$ . The hyperconjugative interactions of  $\sigma_{\text{CH}} \rightarrow n_{\text{B}}$  were of a similar magnitude between these compounds. Where  $\text{BMe}_3$  had a third set of those interactions, borane **S35** had  $\sigma_{\text{CC}} \rightarrow n_{\text{B}}$  interactions. Borane **S35**, however, also had large  $\sigma_{\text{CH}} \rightarrow n_{\text{B}}$  interactions from remote C–H bonds, which are held rigidly over the boron atom. These interactions (underlined in Table 11) donate electron density at boron and thus could be responsible for the higher chemical shifts. No unusual chemical shifts were present for those remote methyl groups, but that interaction could be averaged by bond rotation, making it difficult to detect. These interactions of remote carbon–hydrogen bonds to electron-deficient atoms resemble interactions observed in alkyl-substituted carbocations.<sup>17</sup>

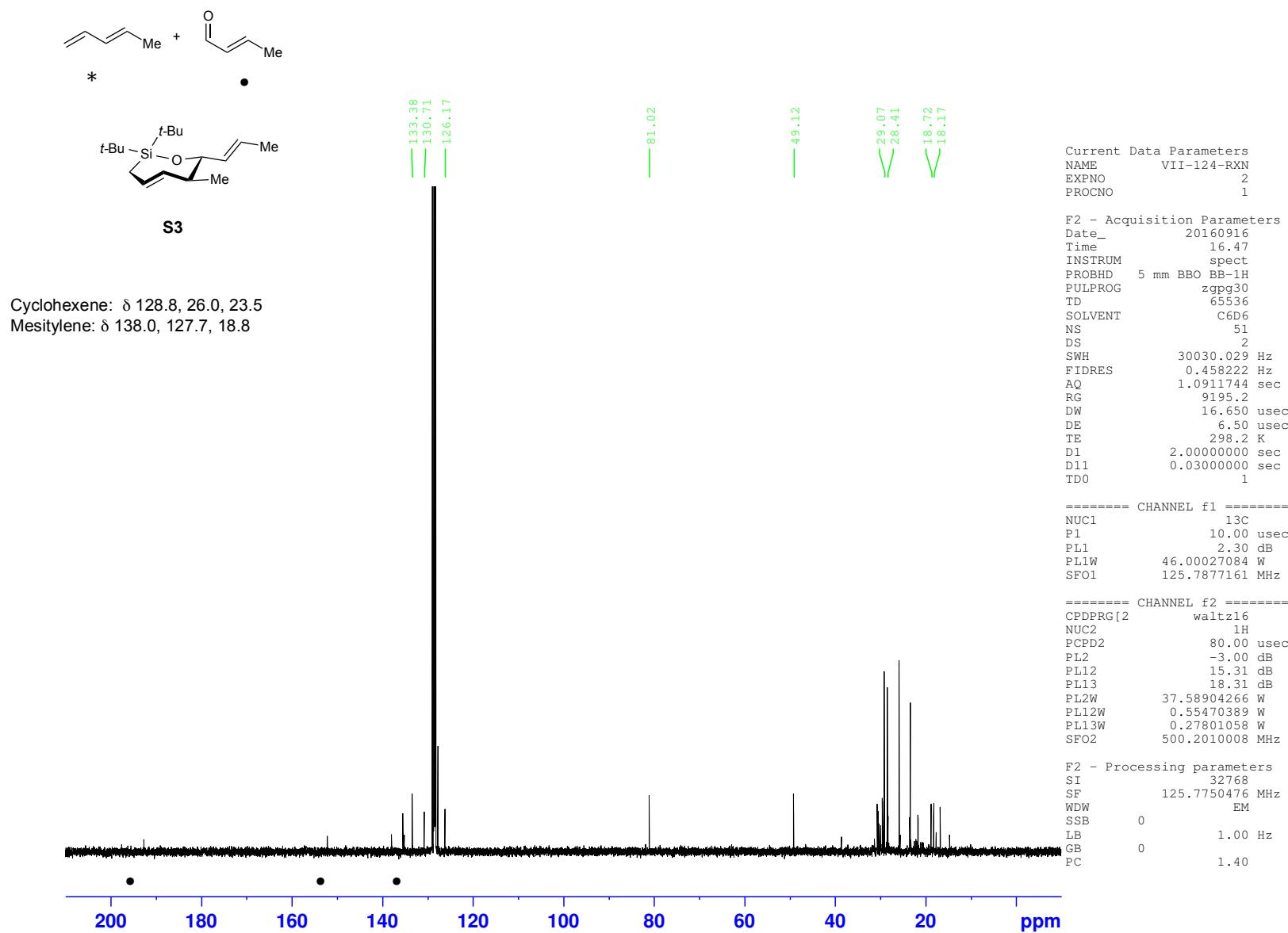
## X. References

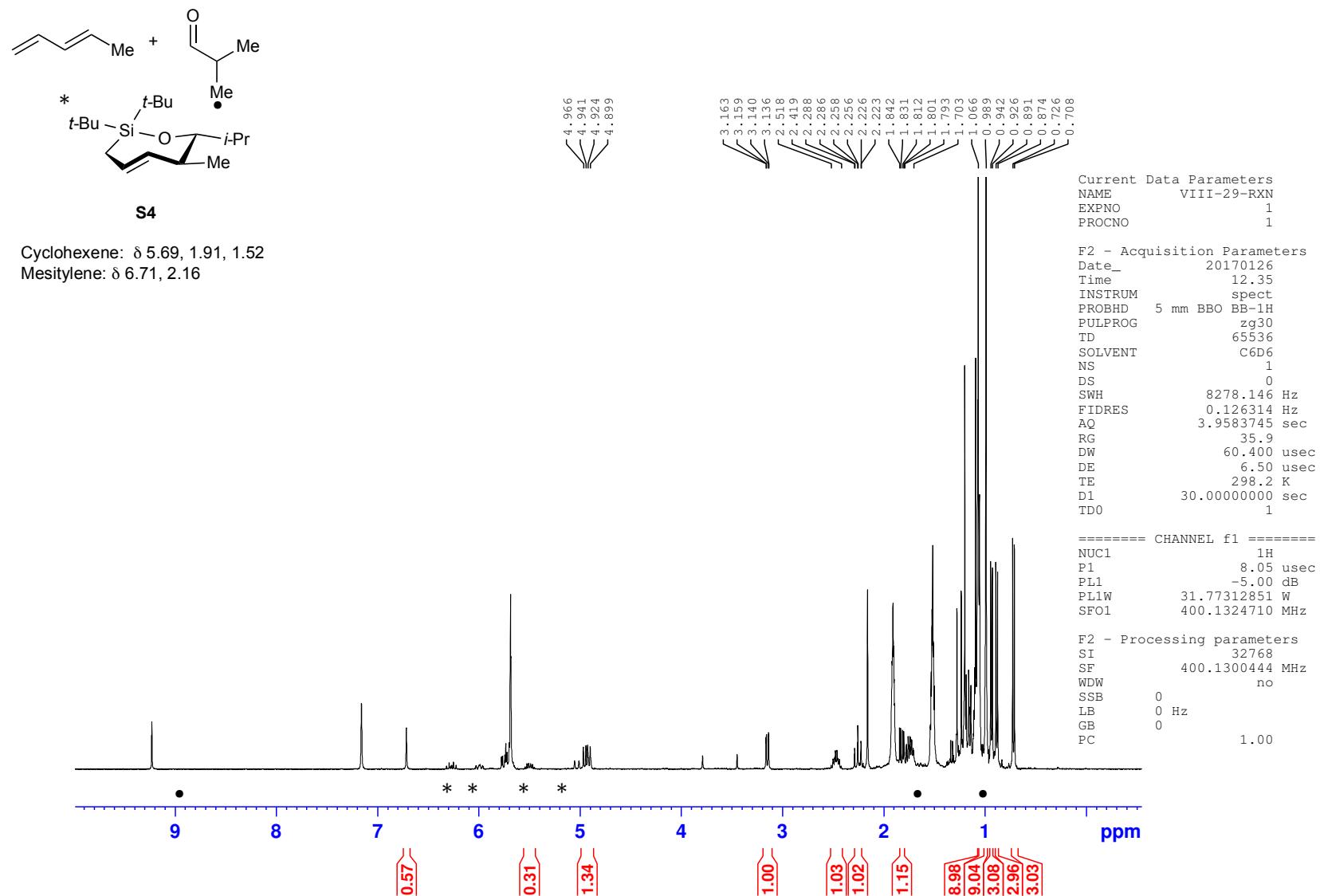
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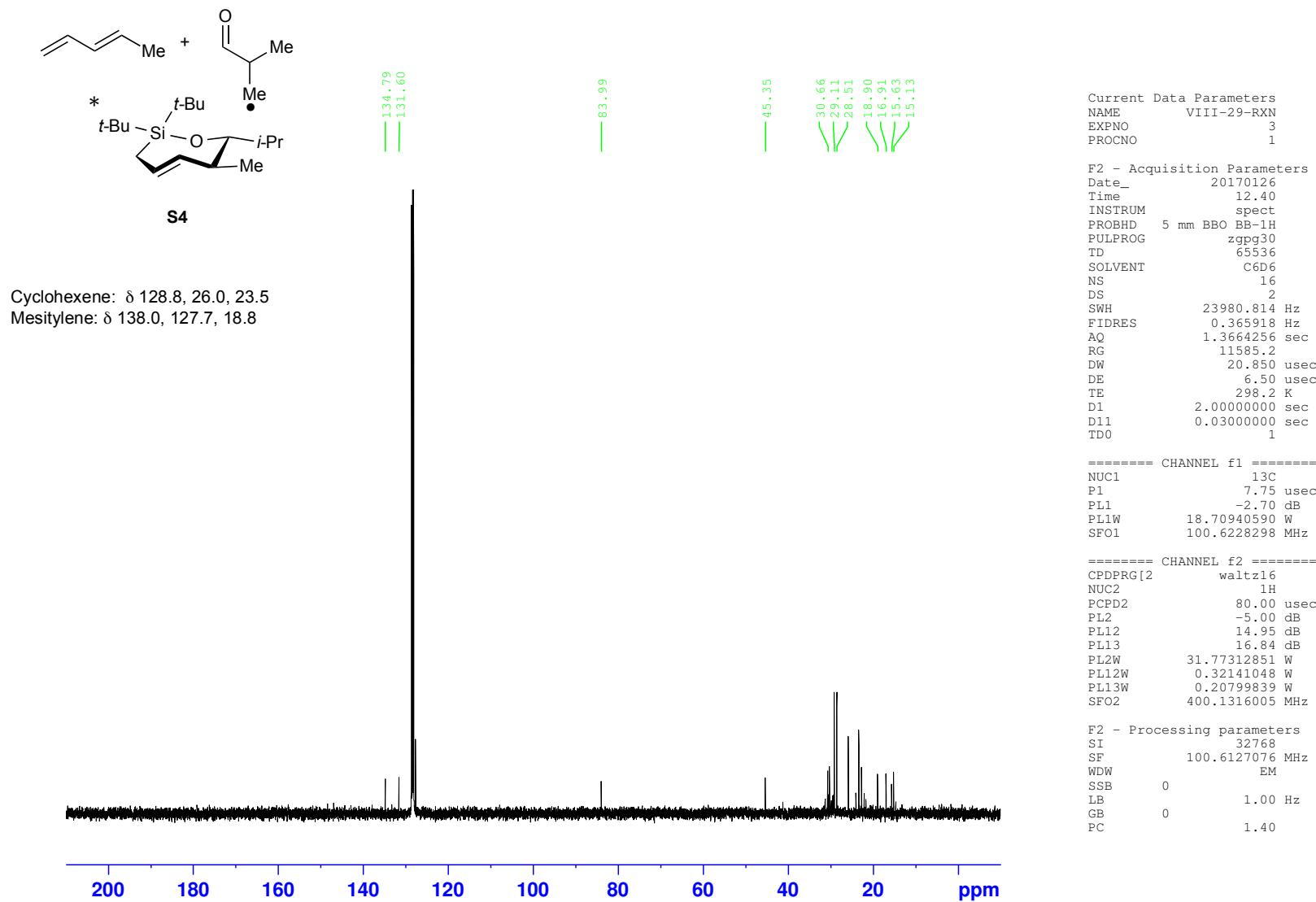
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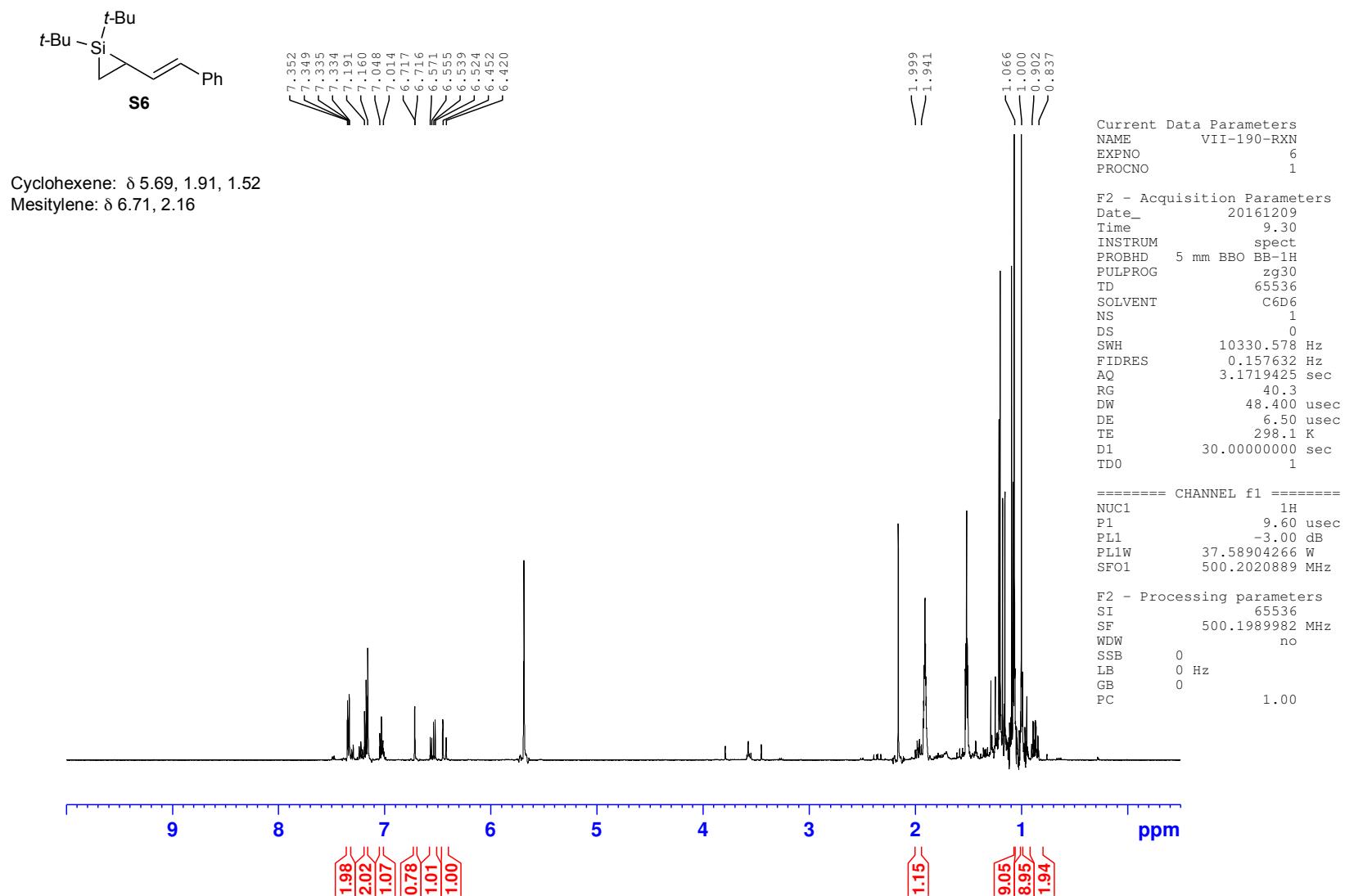
**XII.  $^1\text{H}$  and  $^{13}\text{C}$  NMR Spectra** (Side products, starting materials, and standards from *in situ* reactions listed above relevant spectra.)

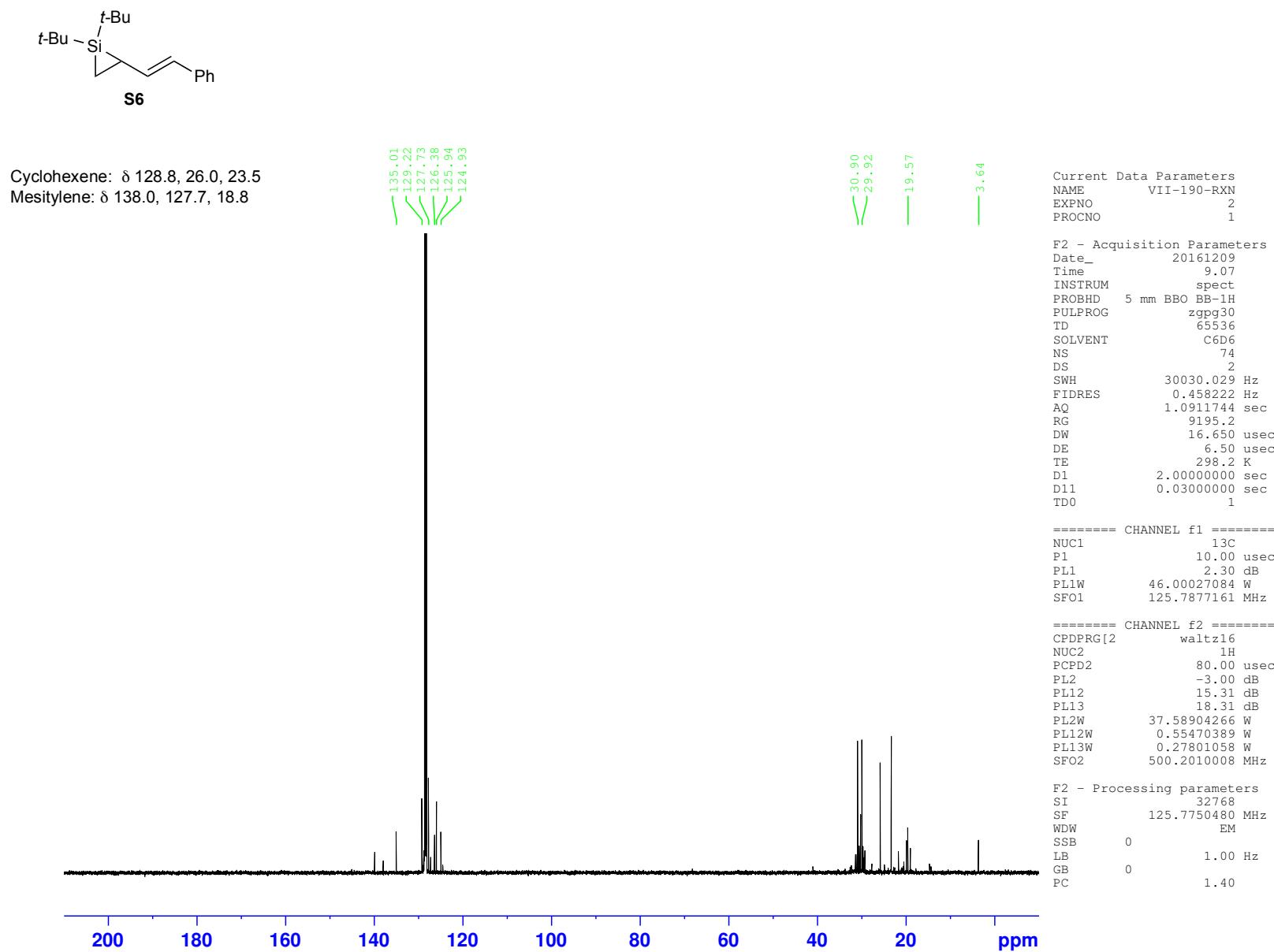


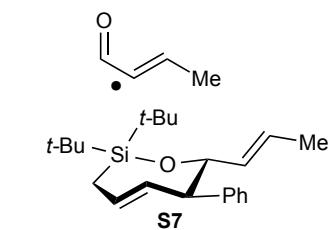




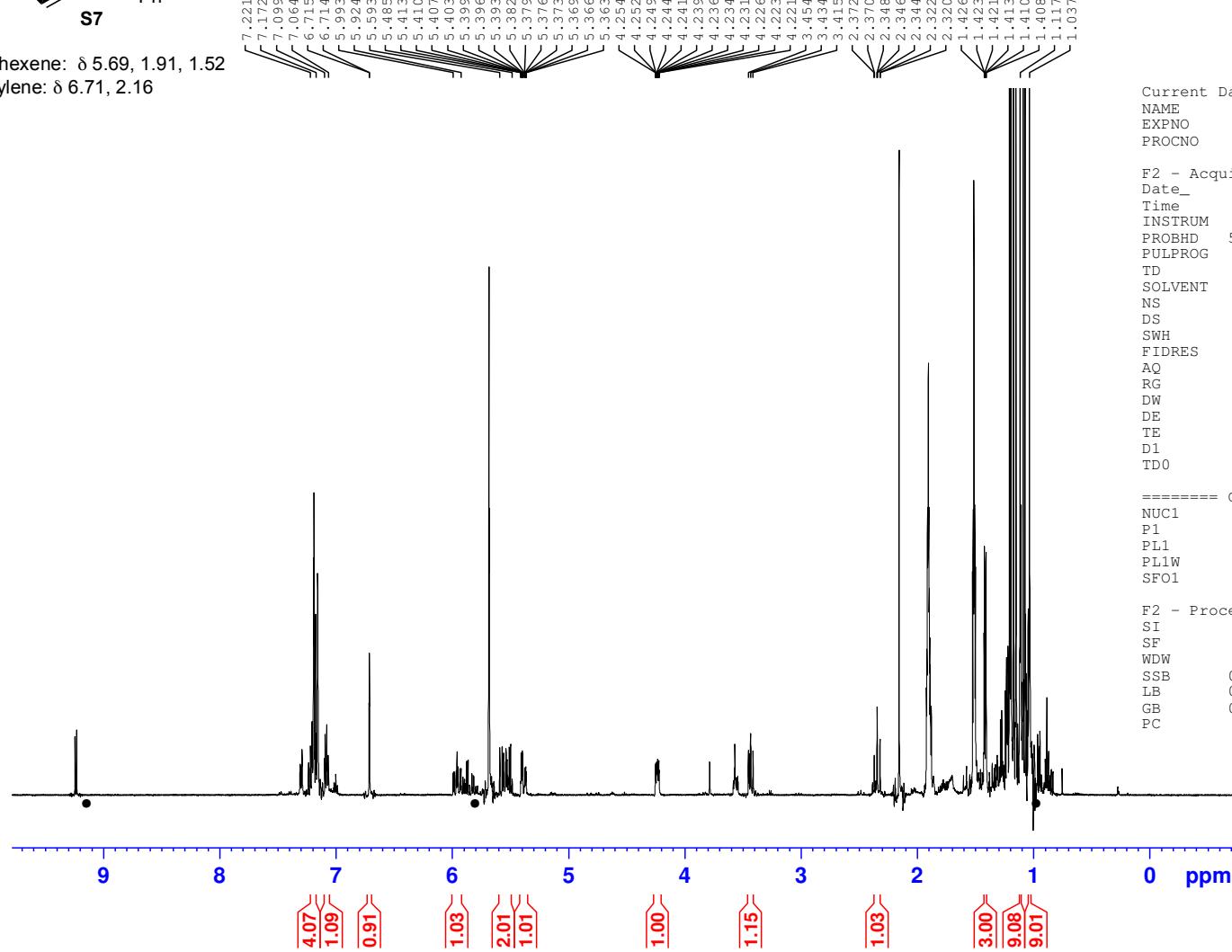








Cyclohexene:  $\delta$  5.69, 1.91, 1.52  
Mesitylene:  $\delta$  6.71, 2.16



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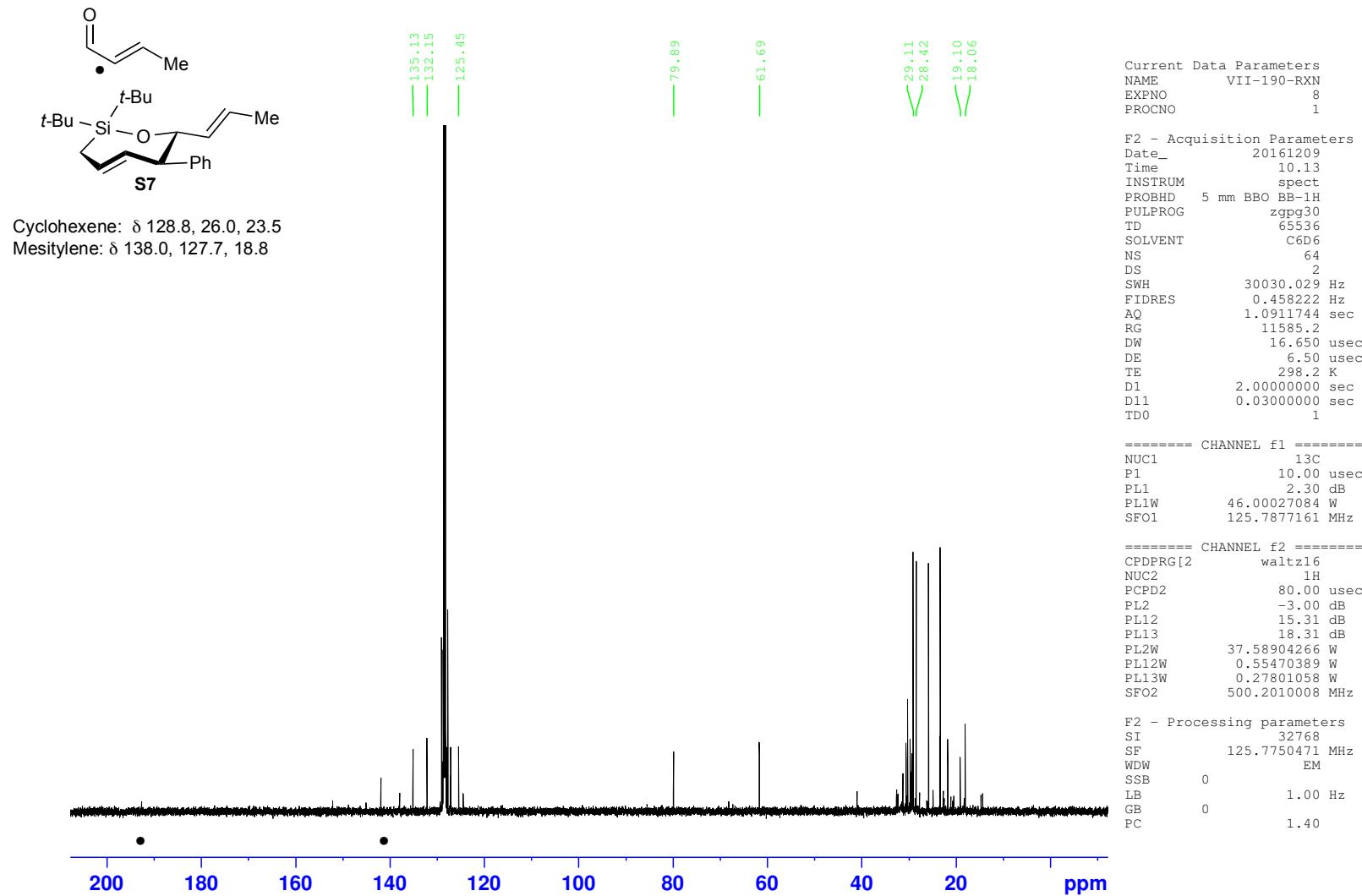
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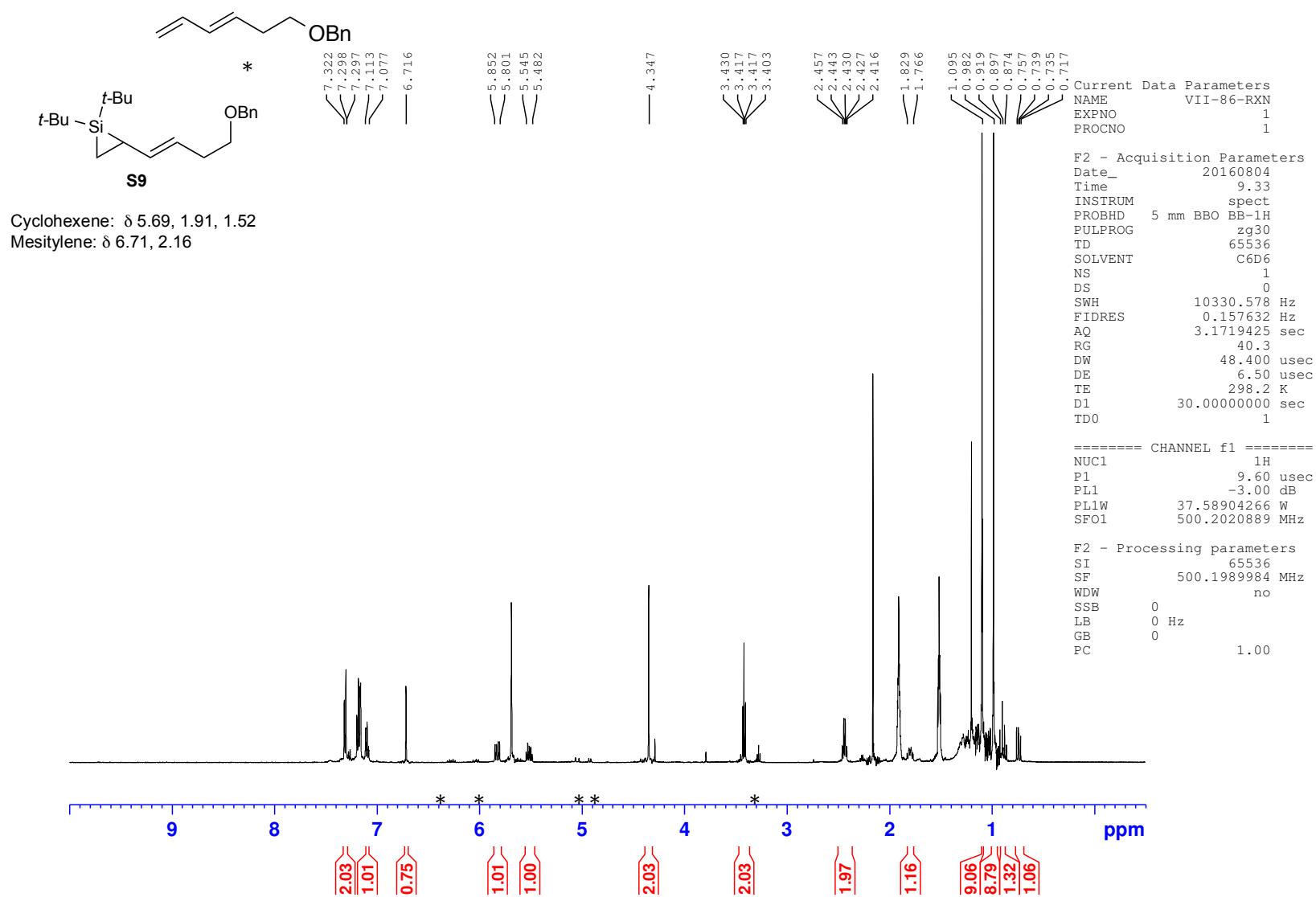
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FIDRES        0.157632 Hz
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DE             6.50 usec
TE             298.2 K
D1             30.0000000 sec
TD0              1

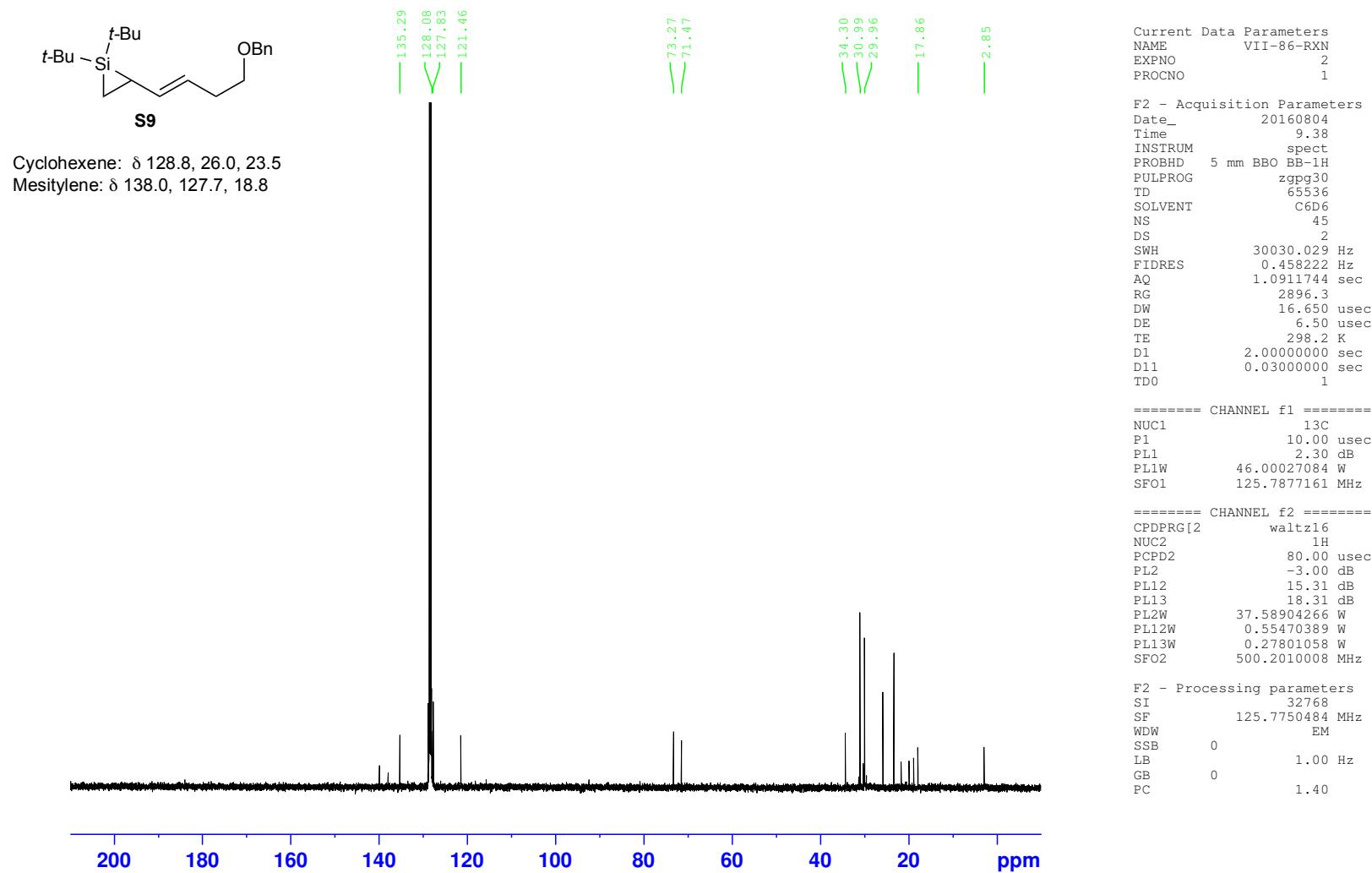
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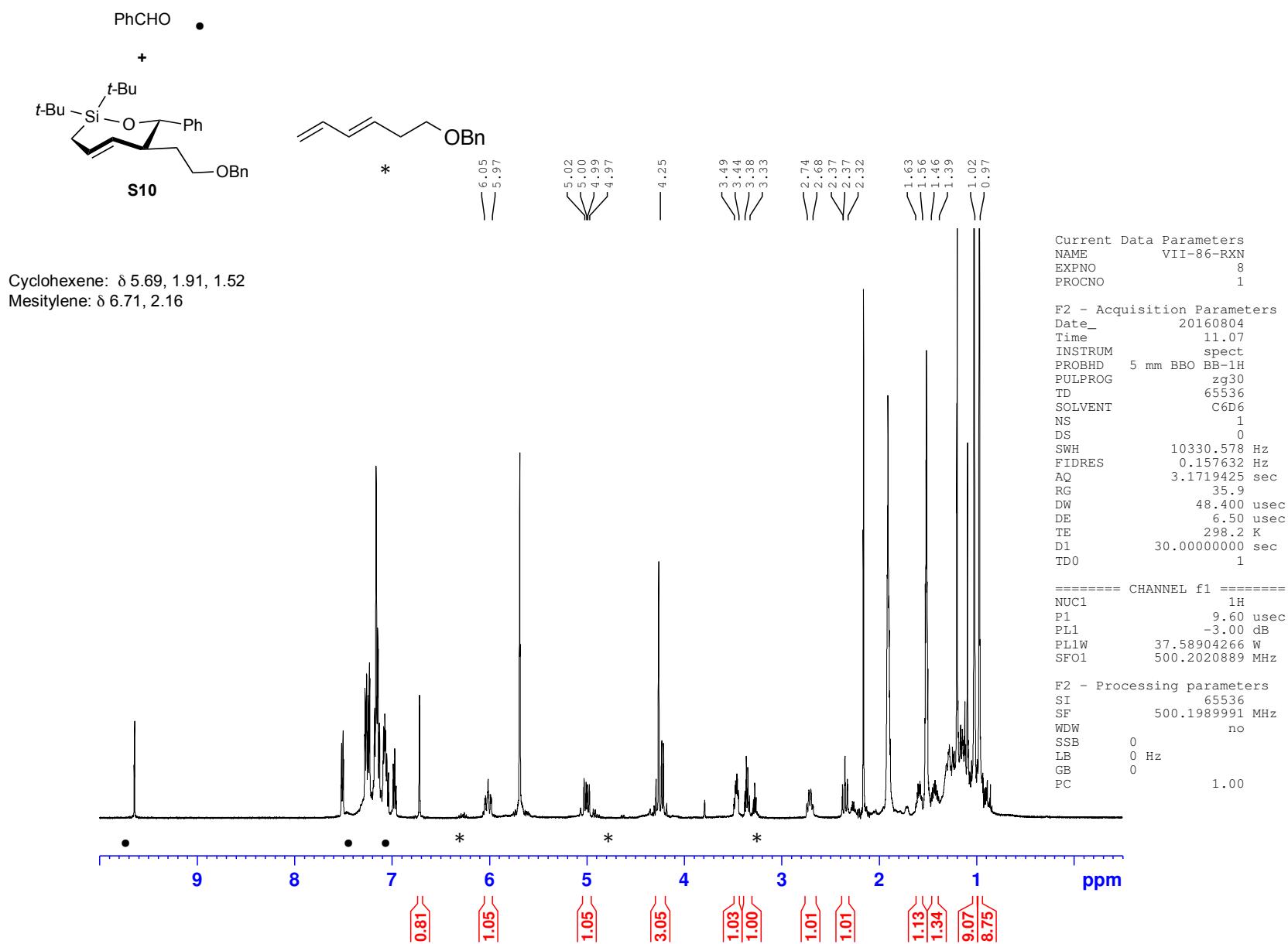
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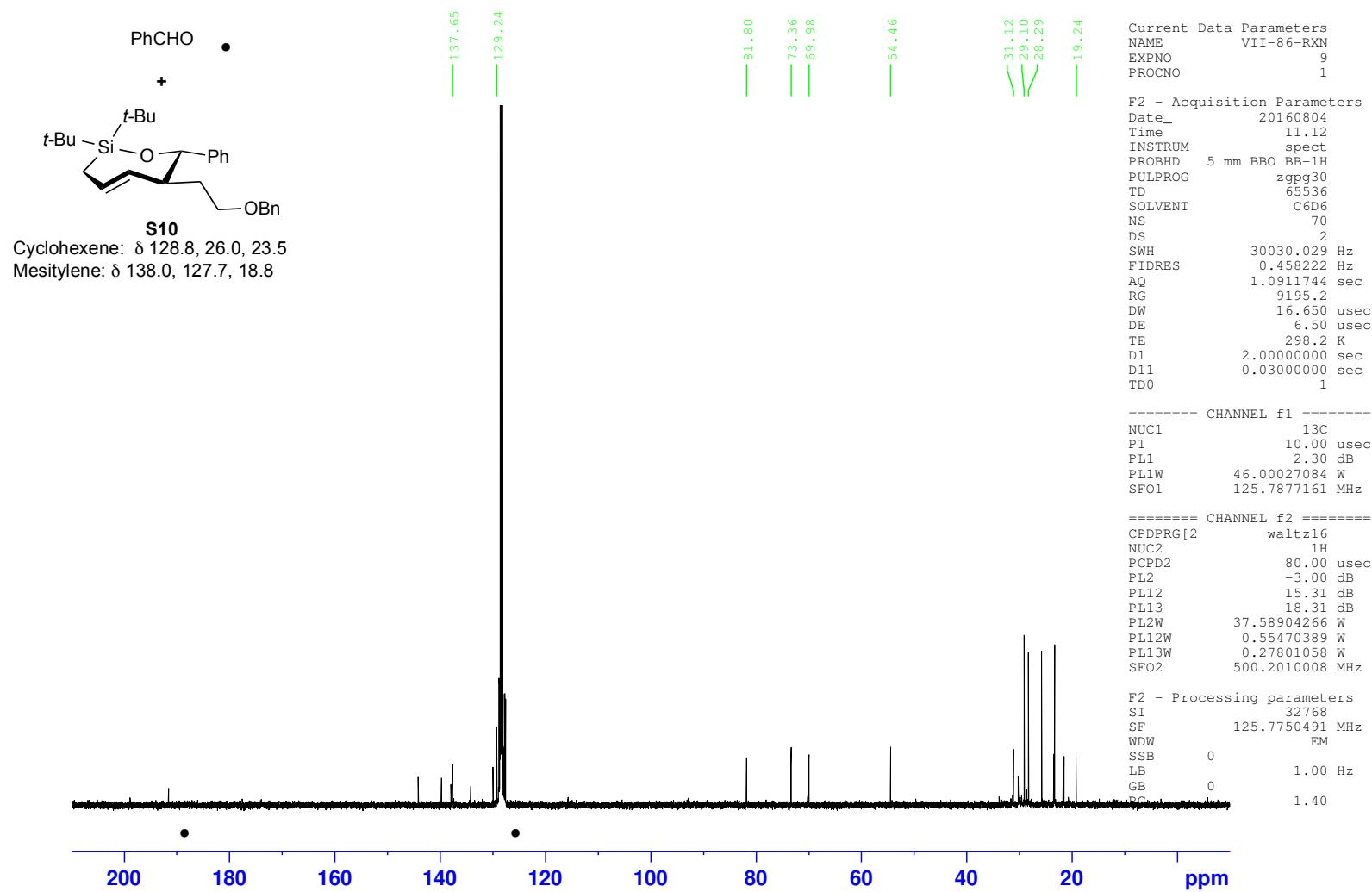
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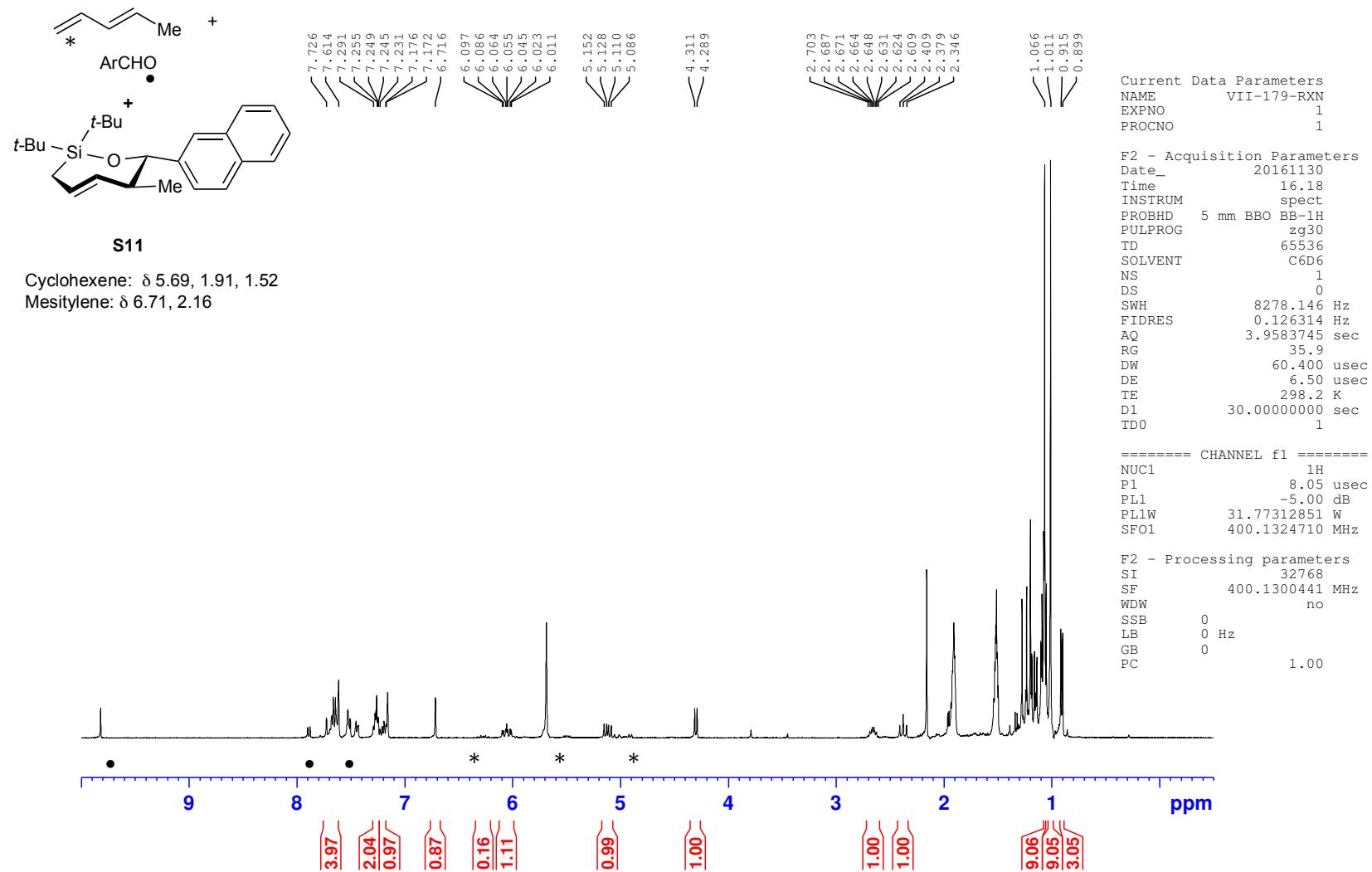


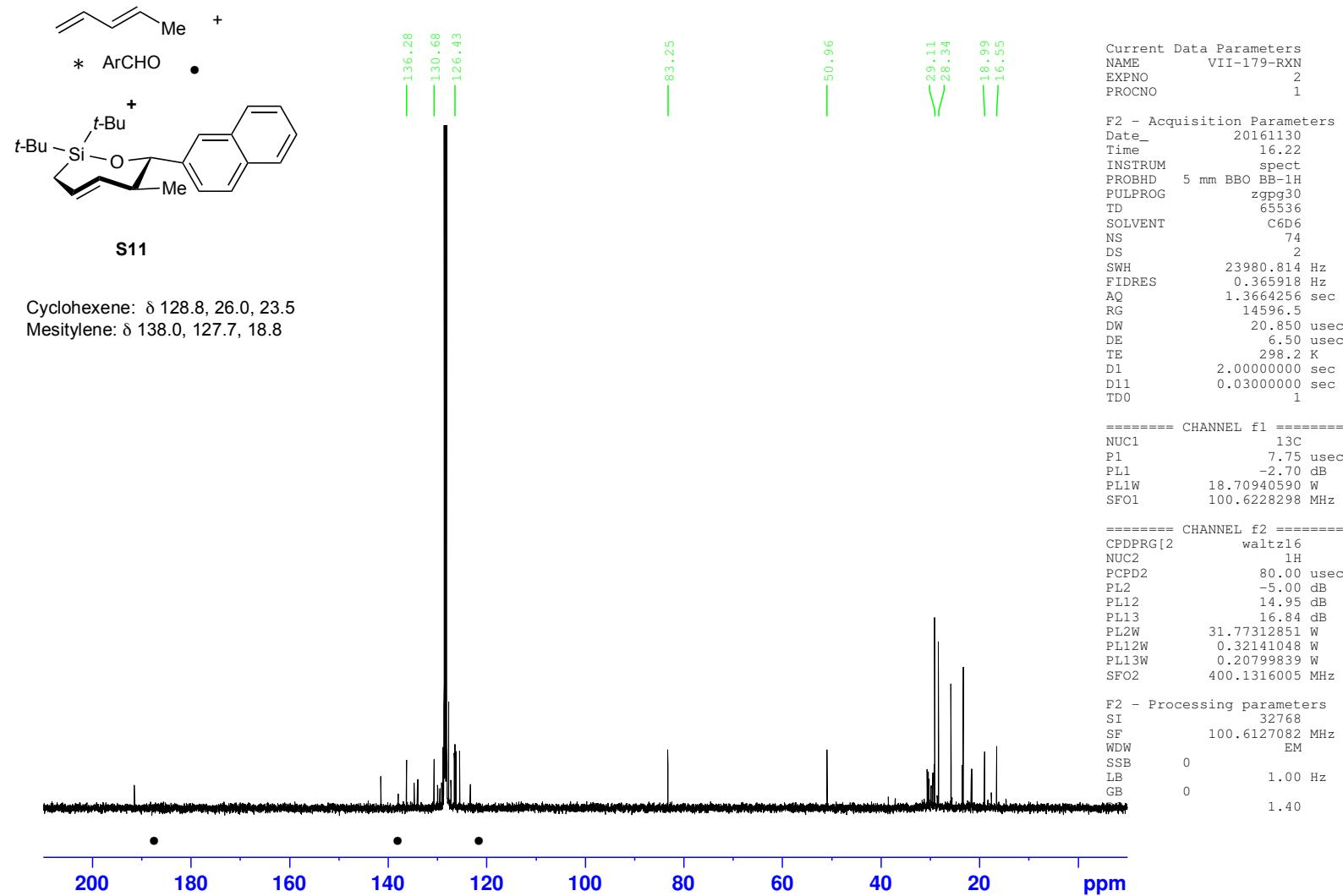


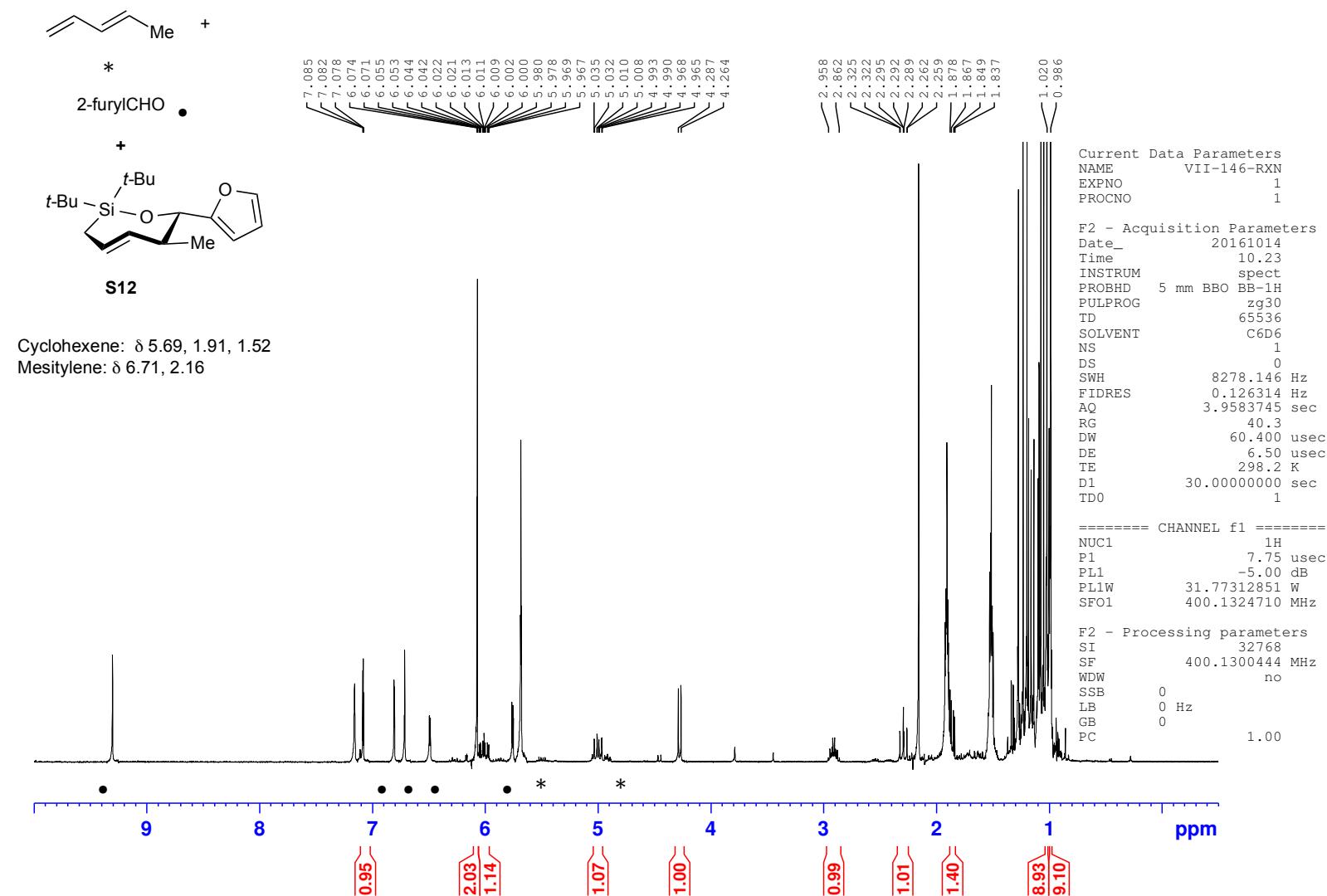


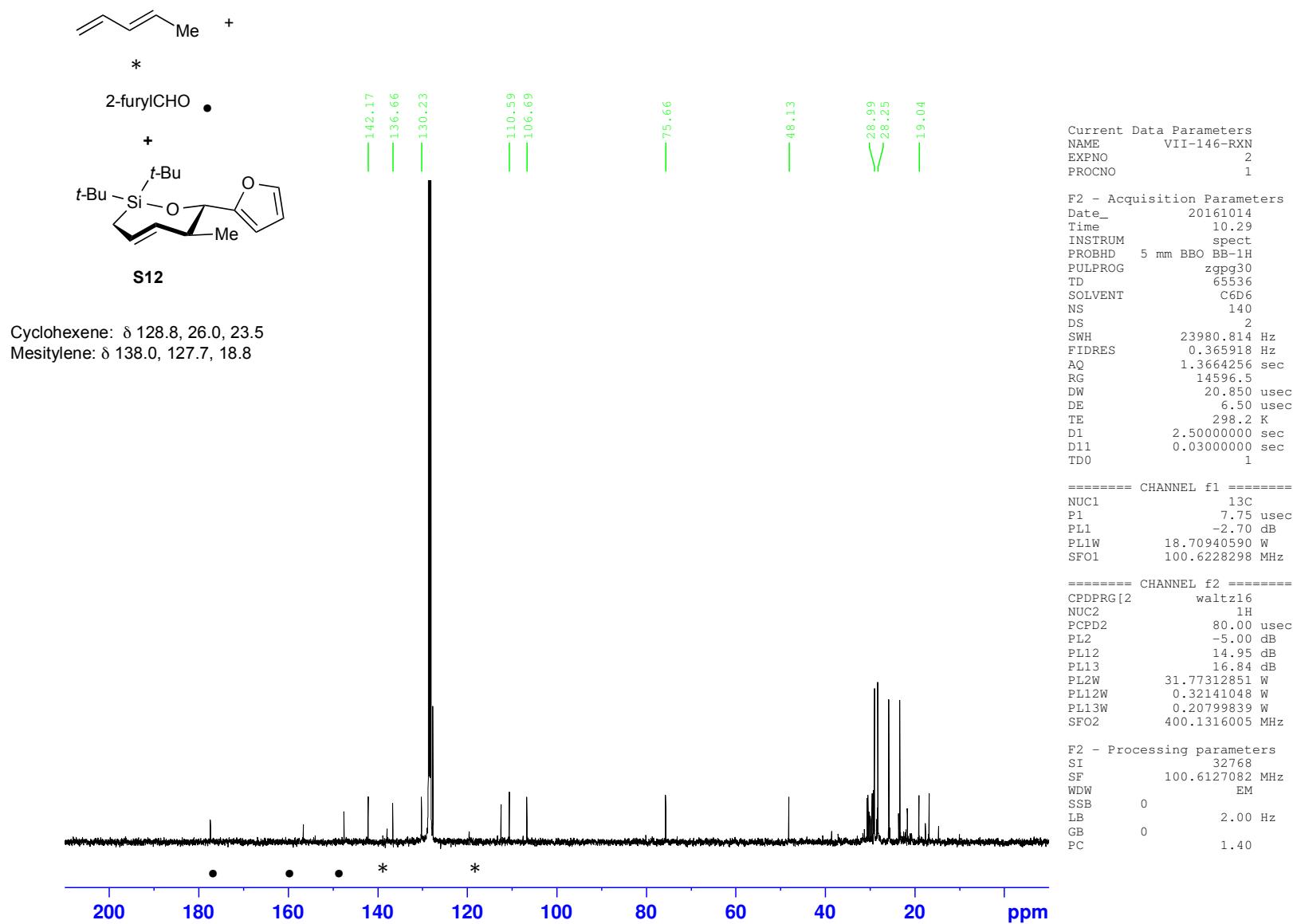


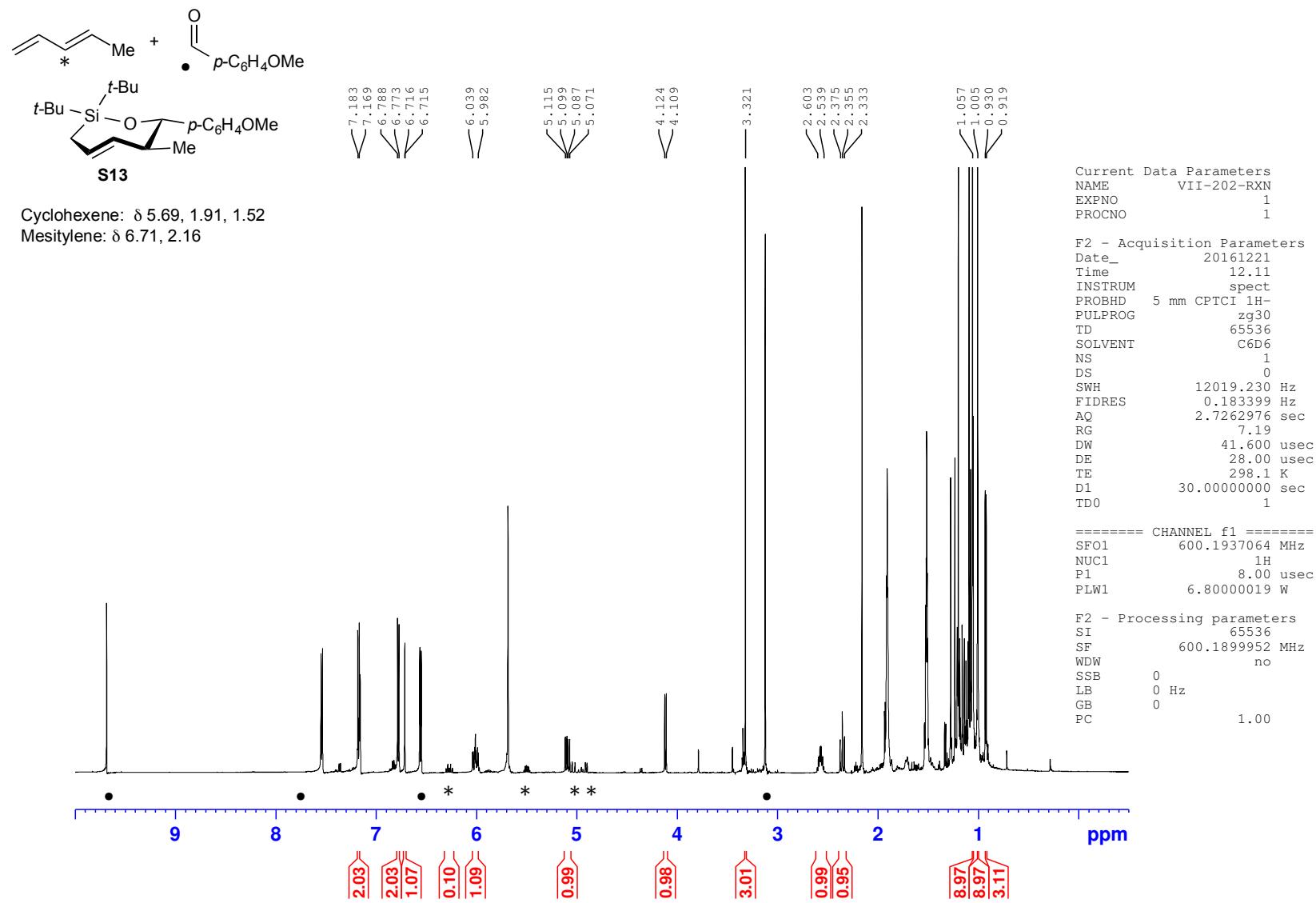


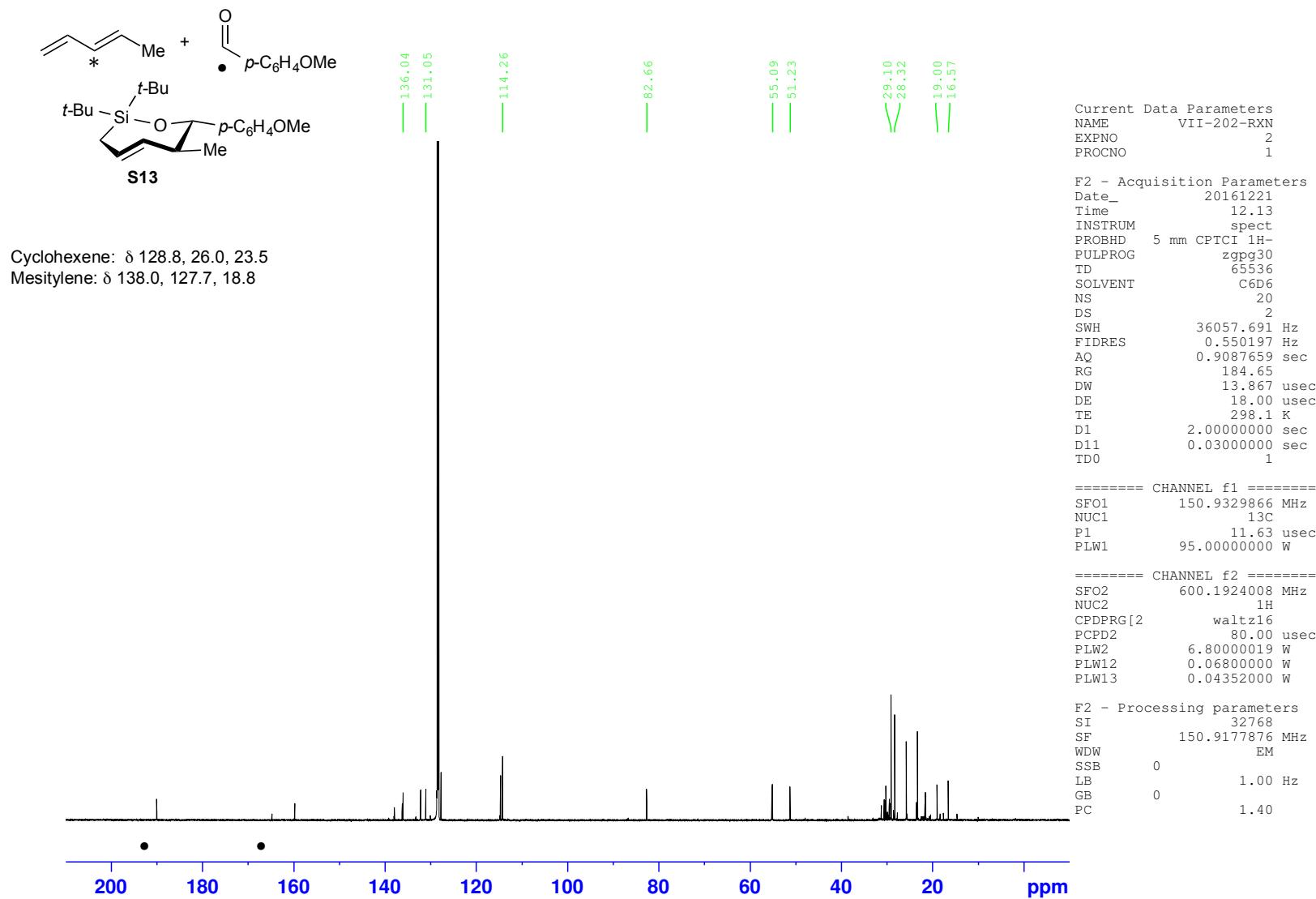


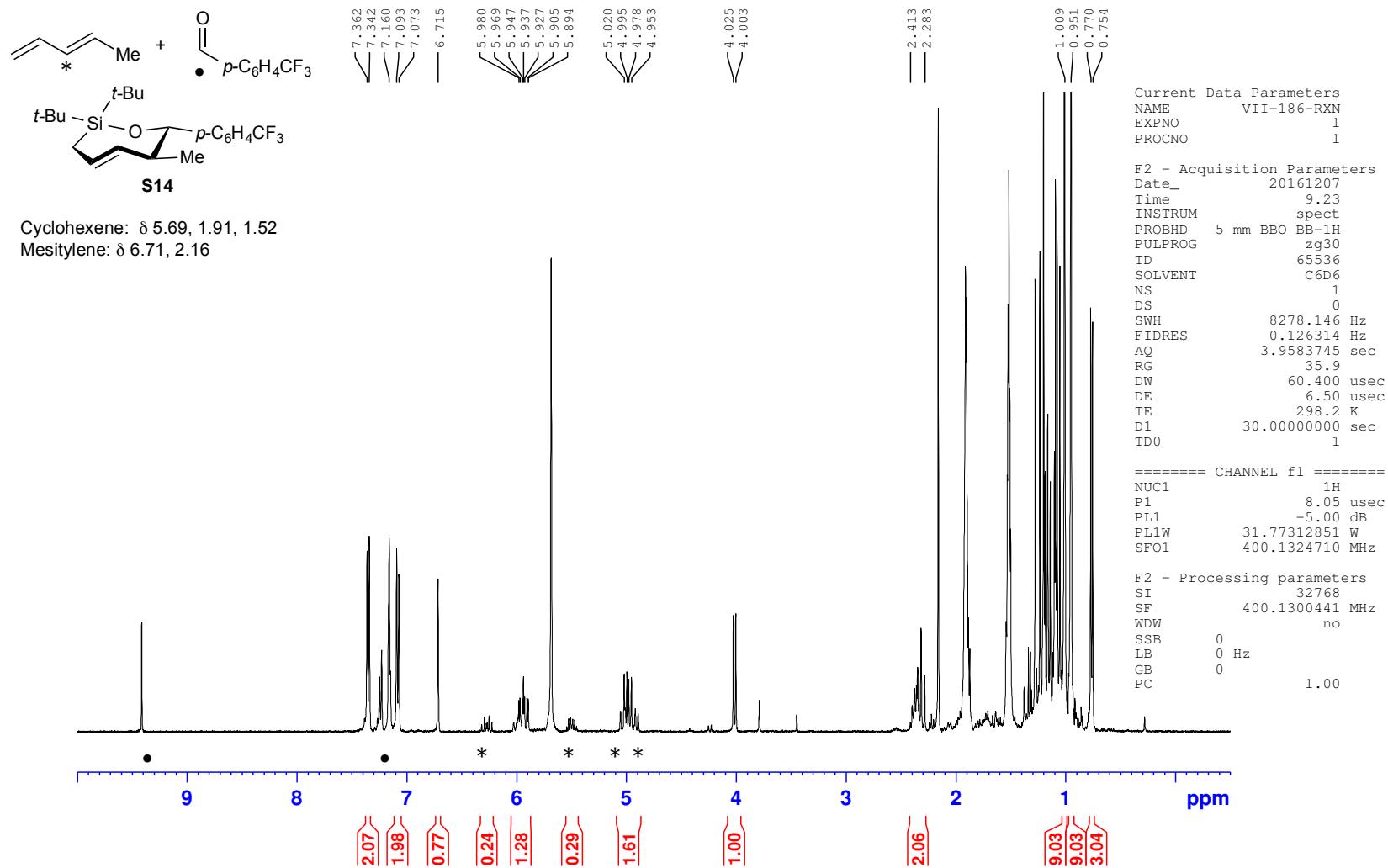


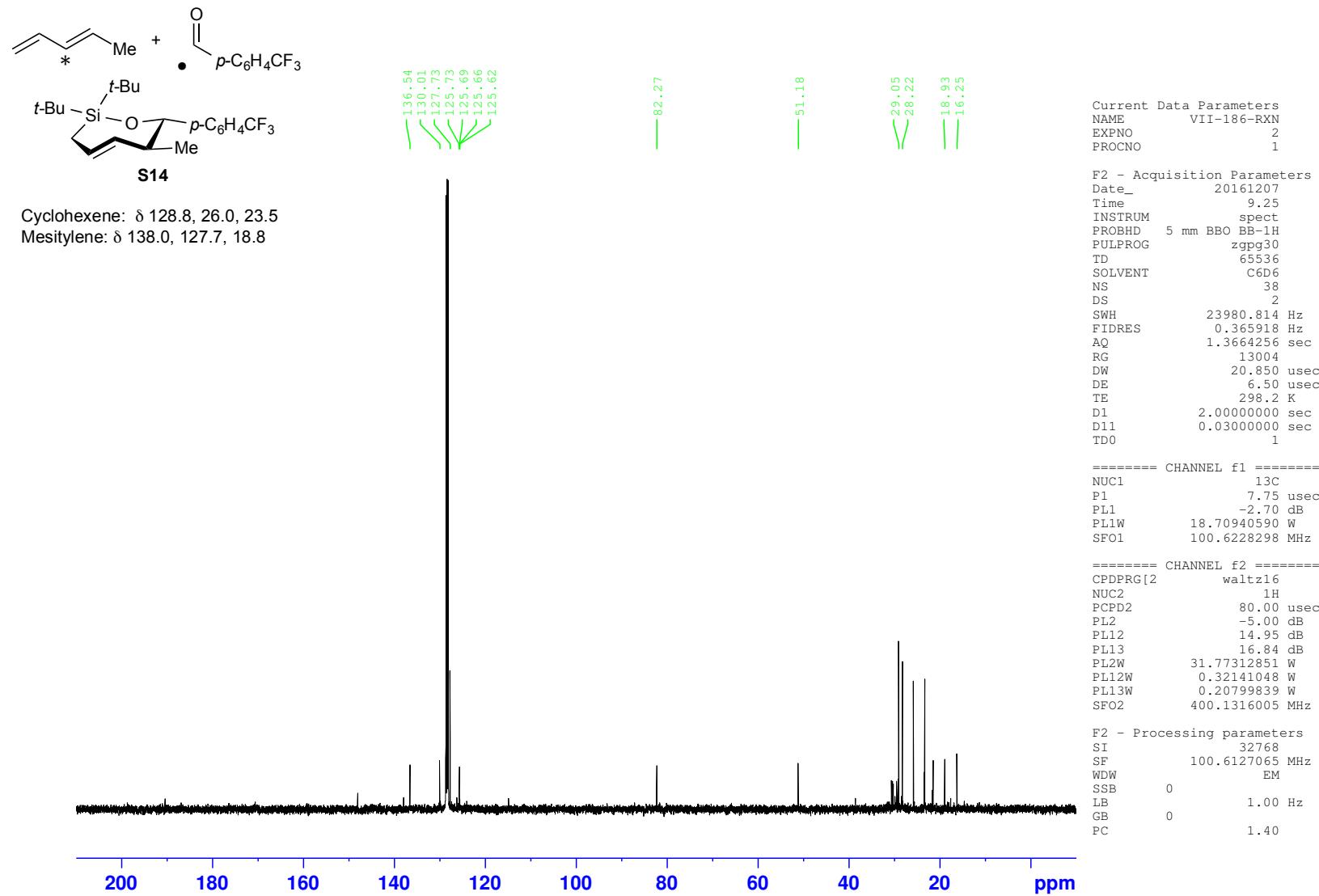


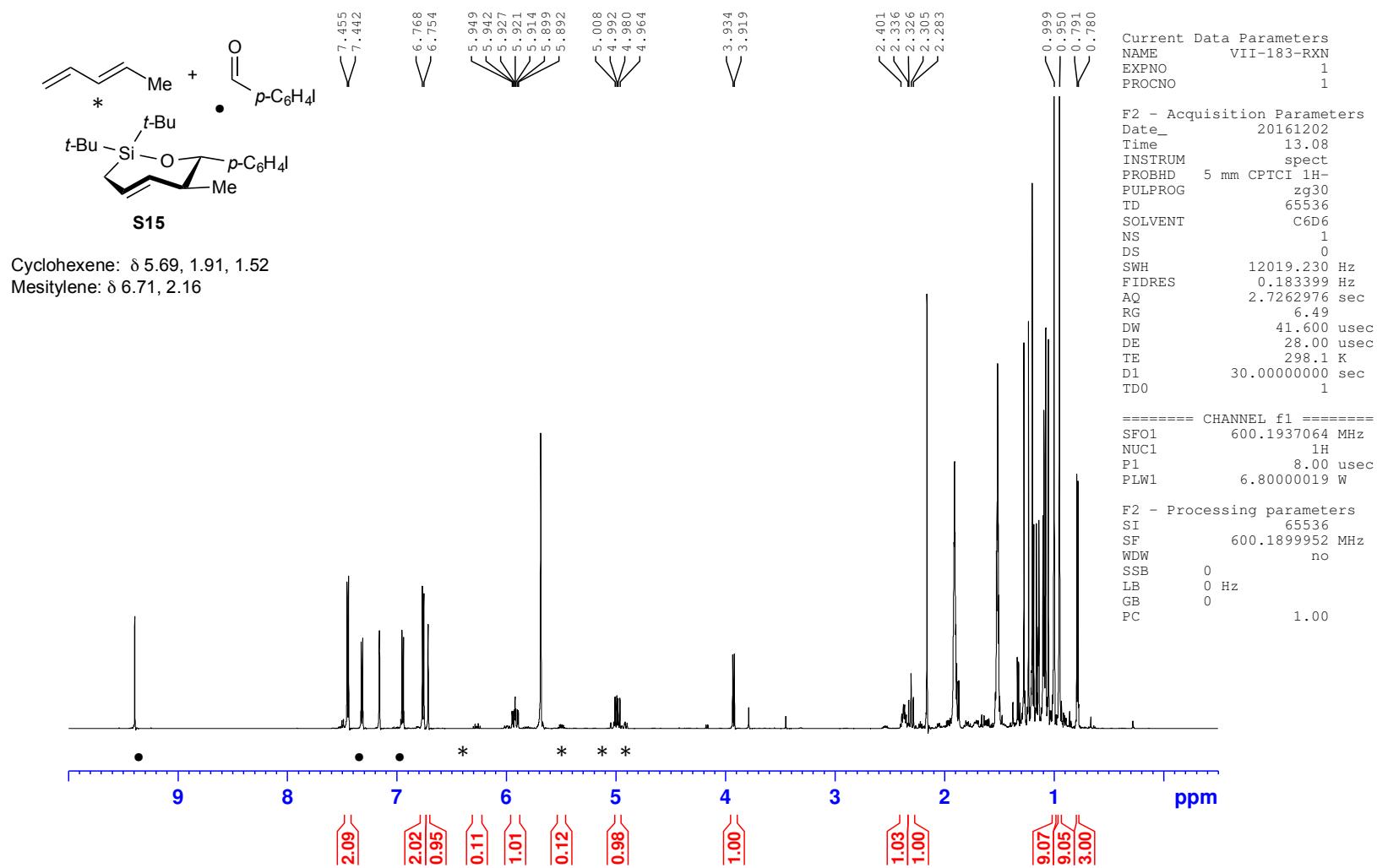


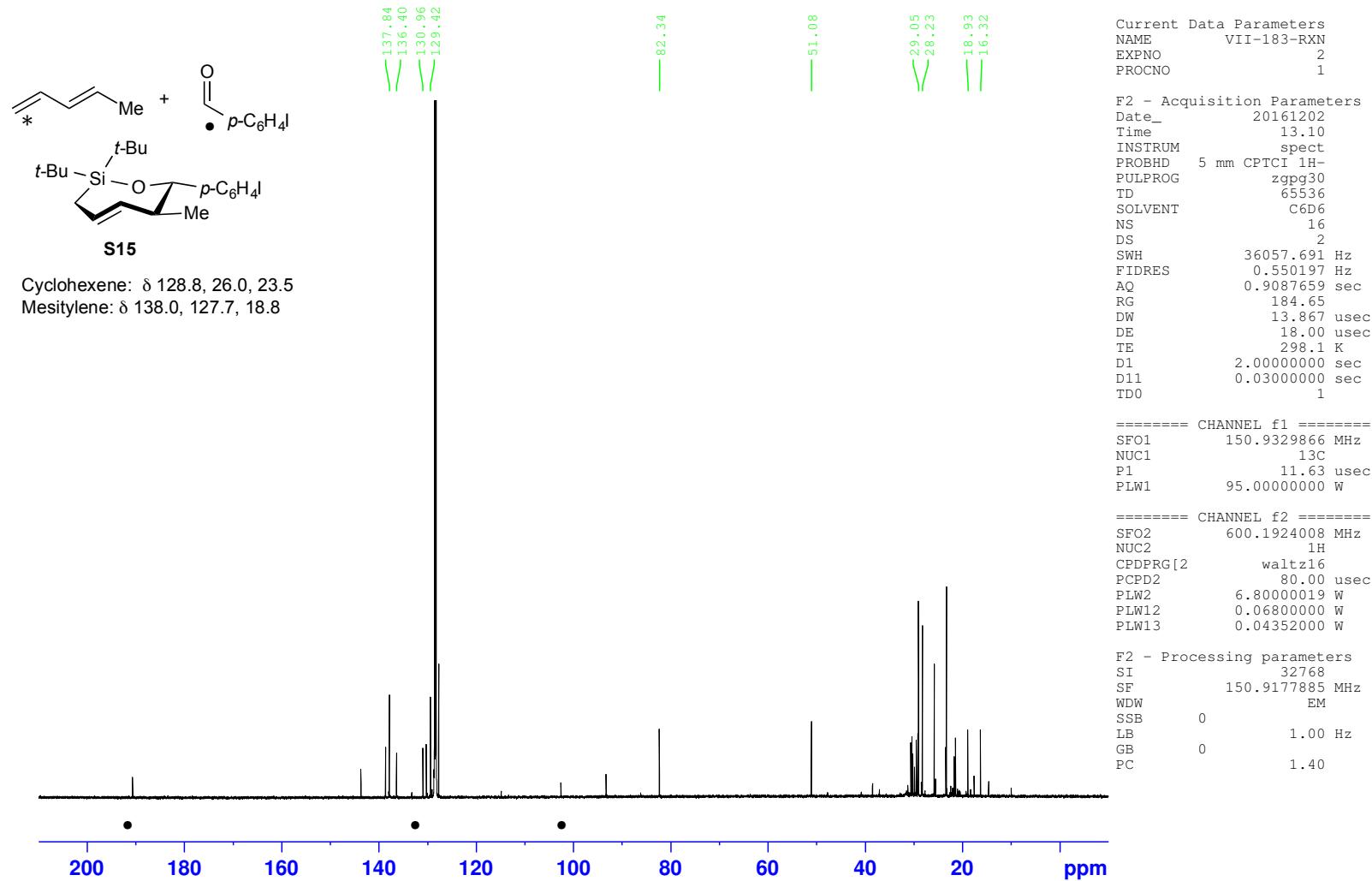


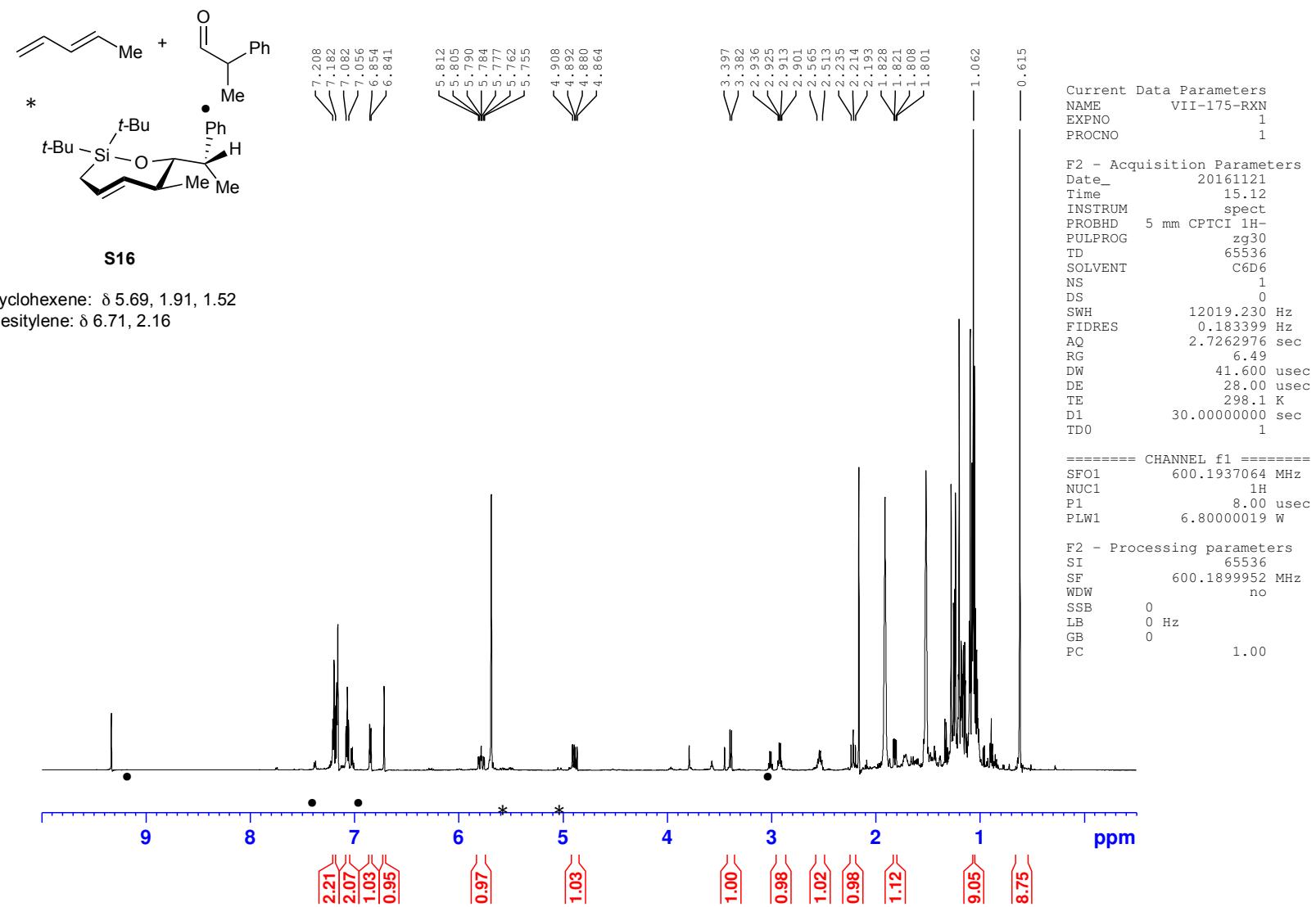


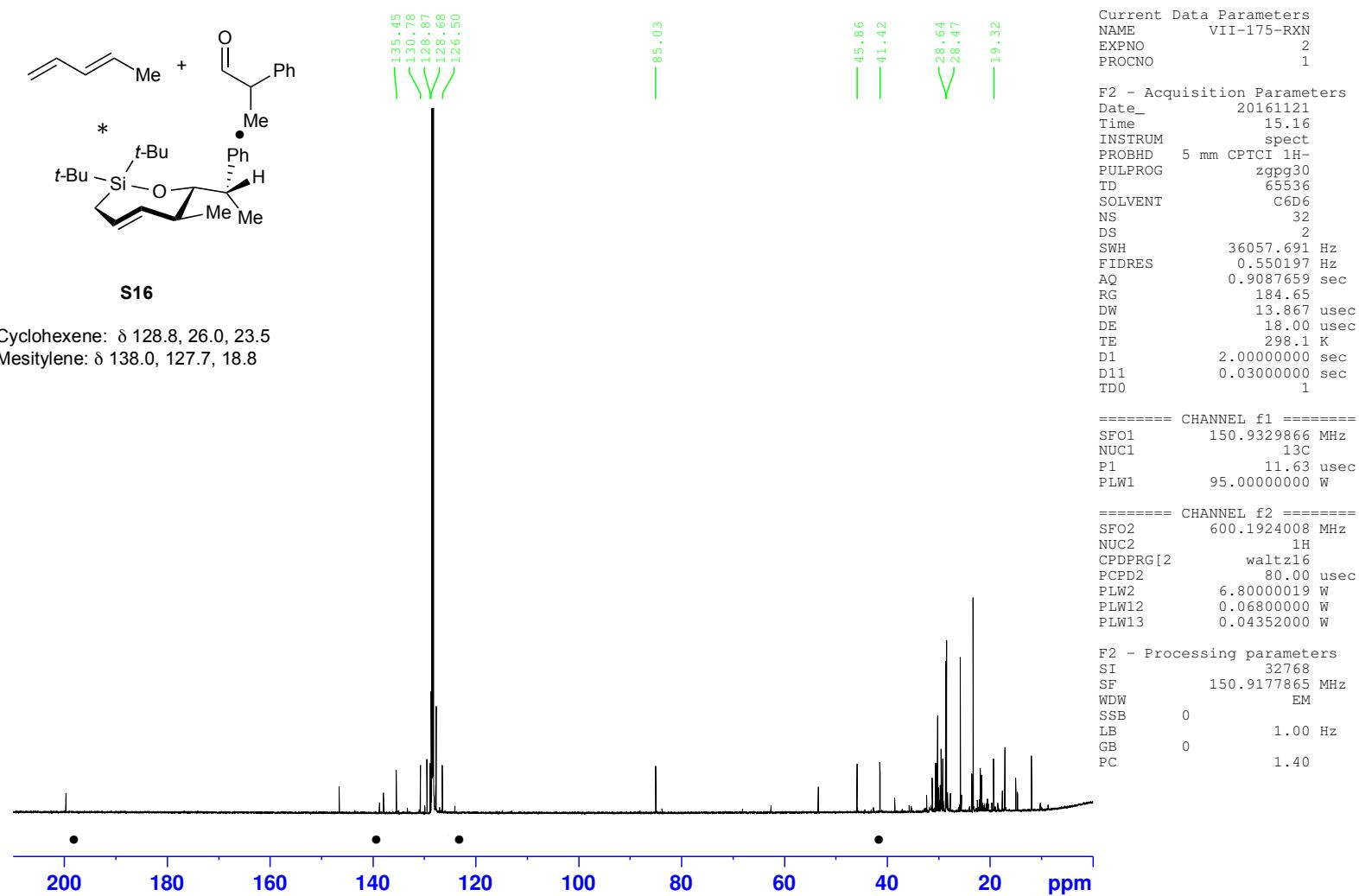


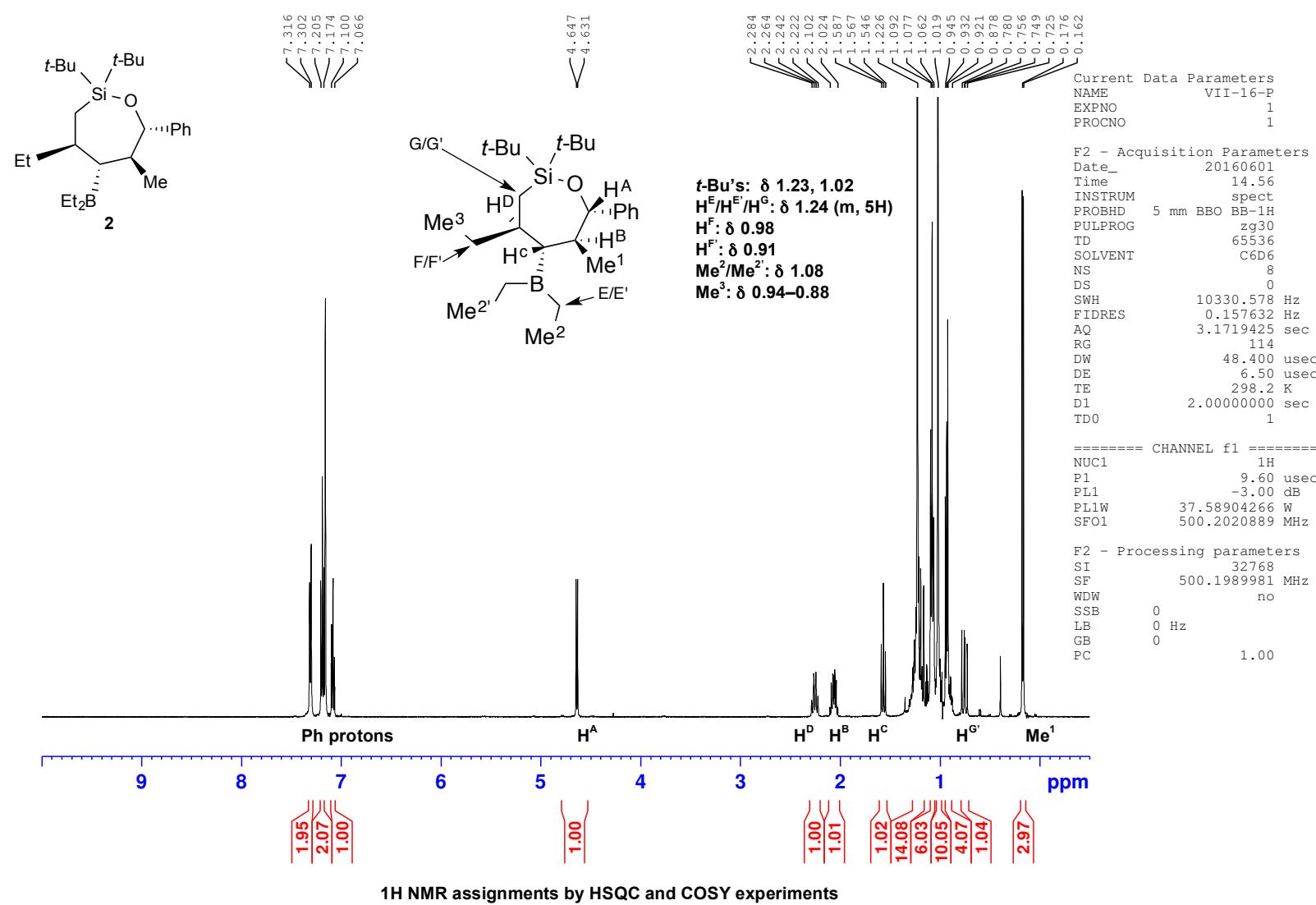


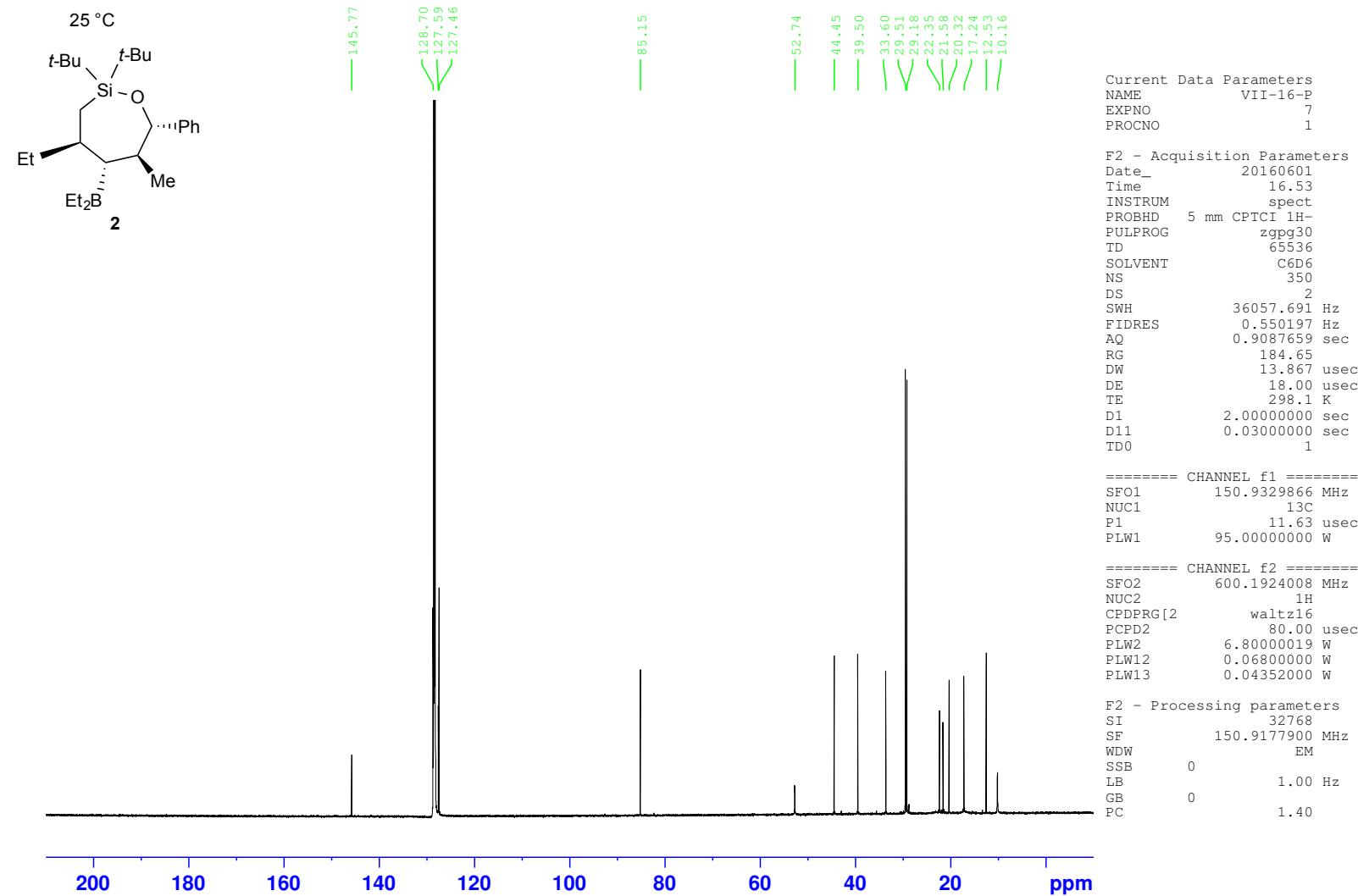


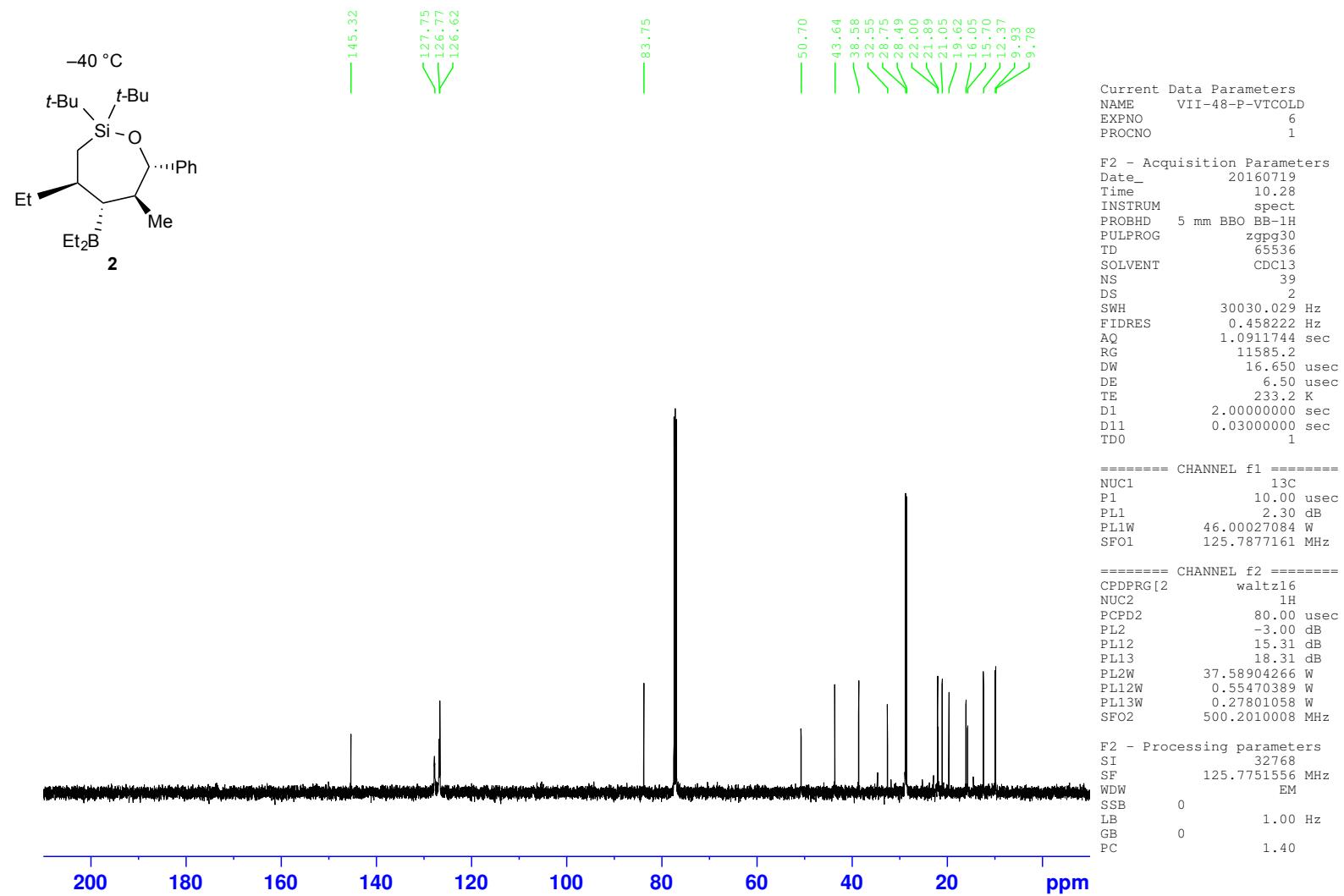


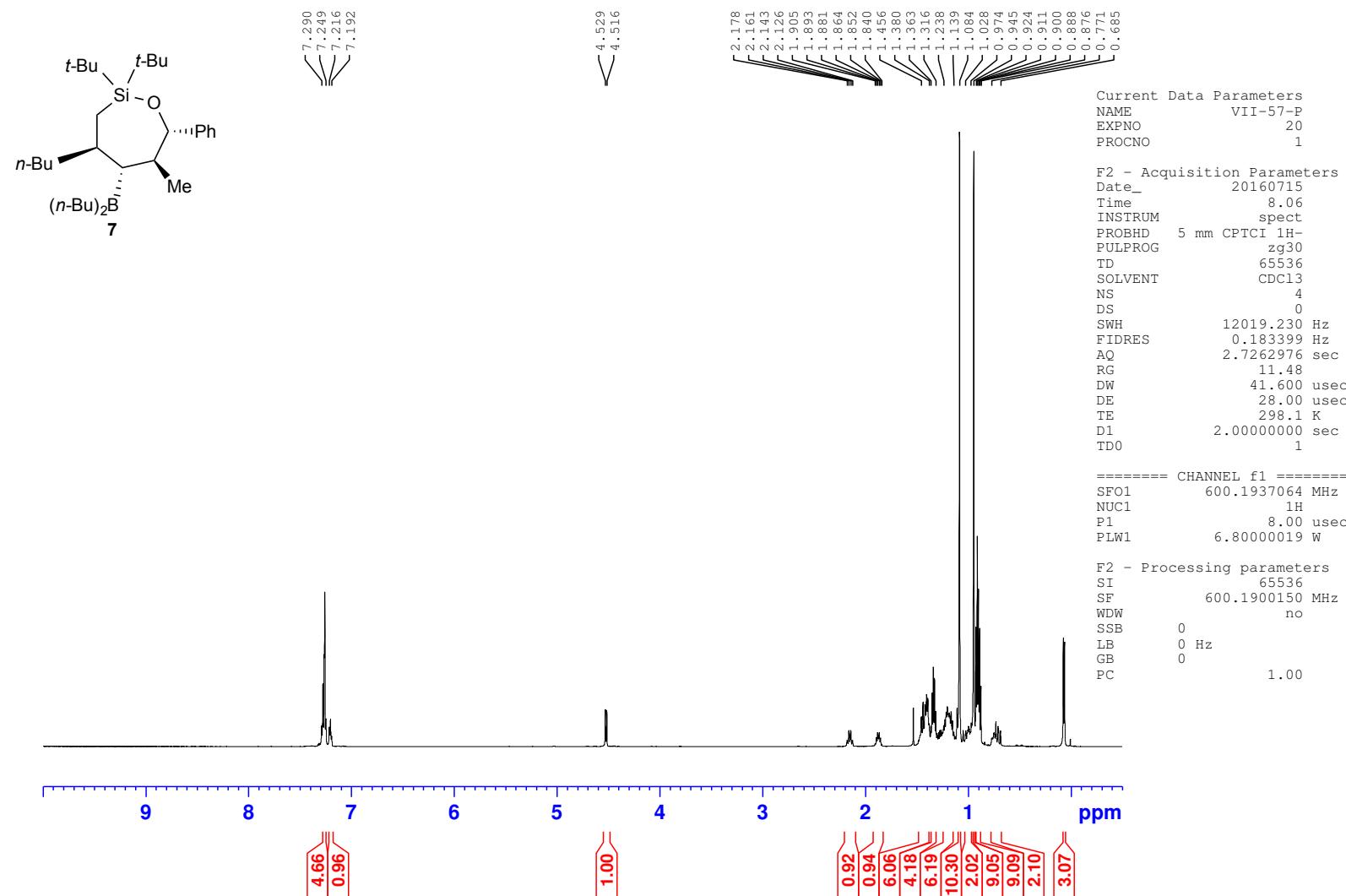


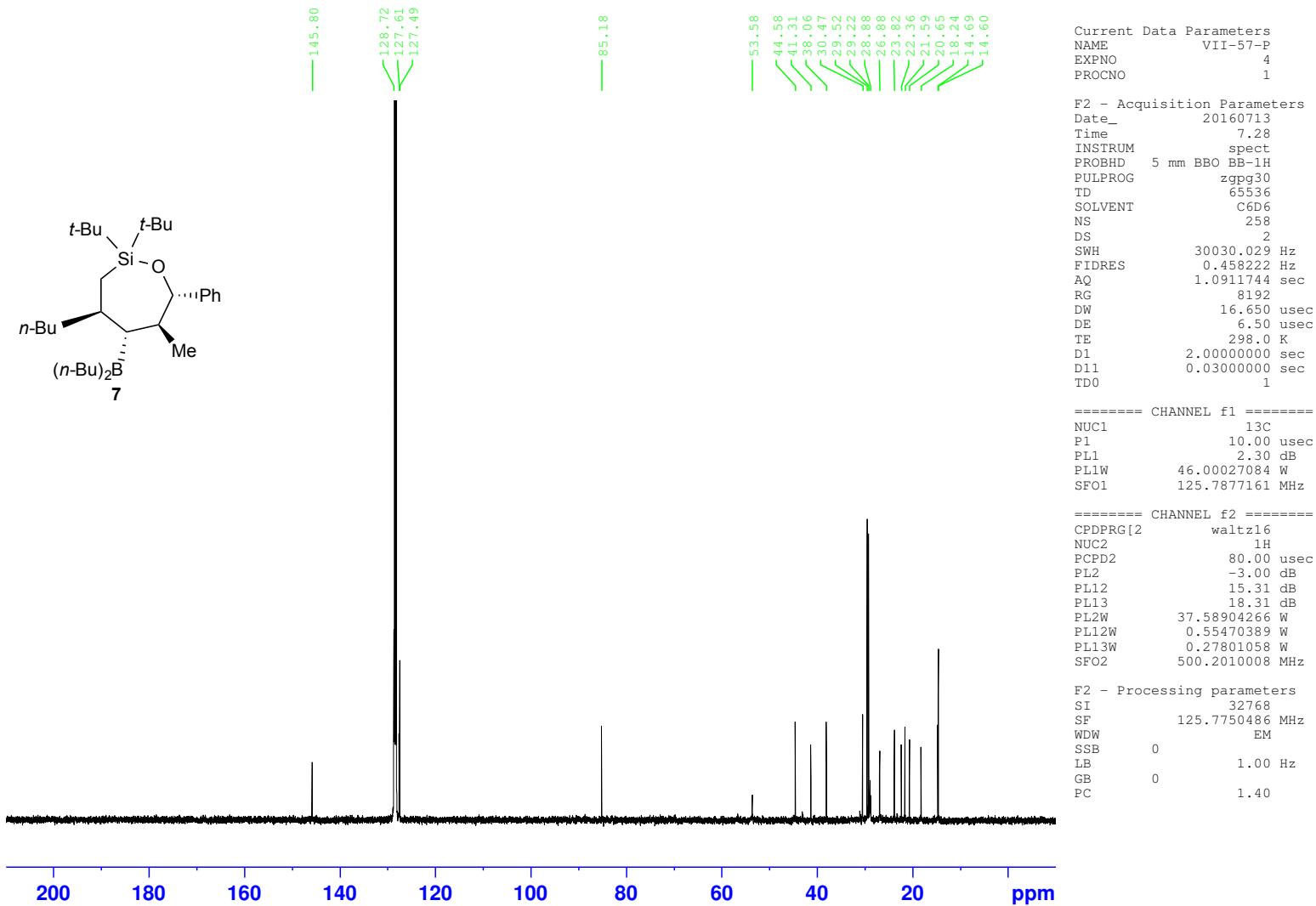


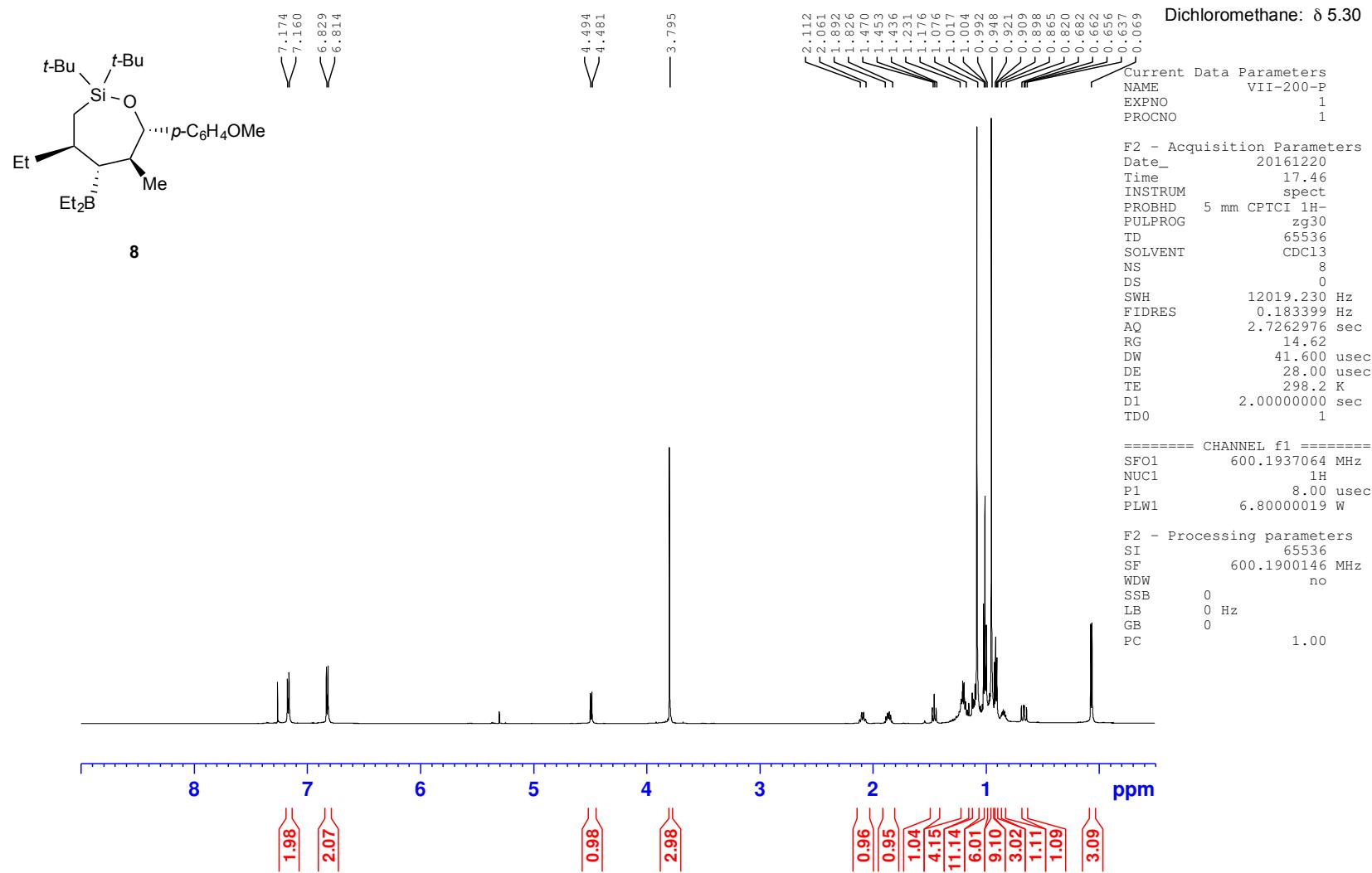


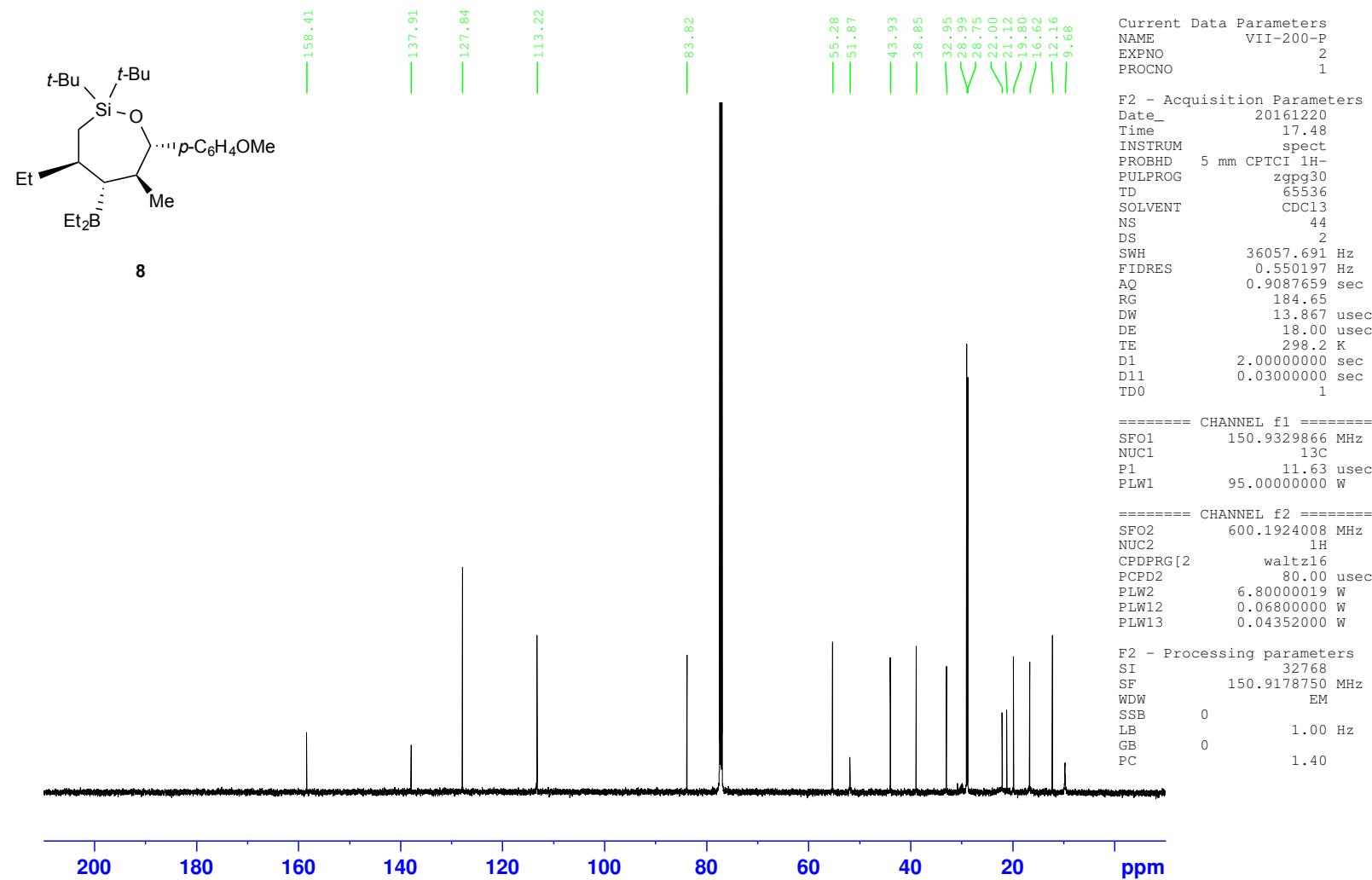


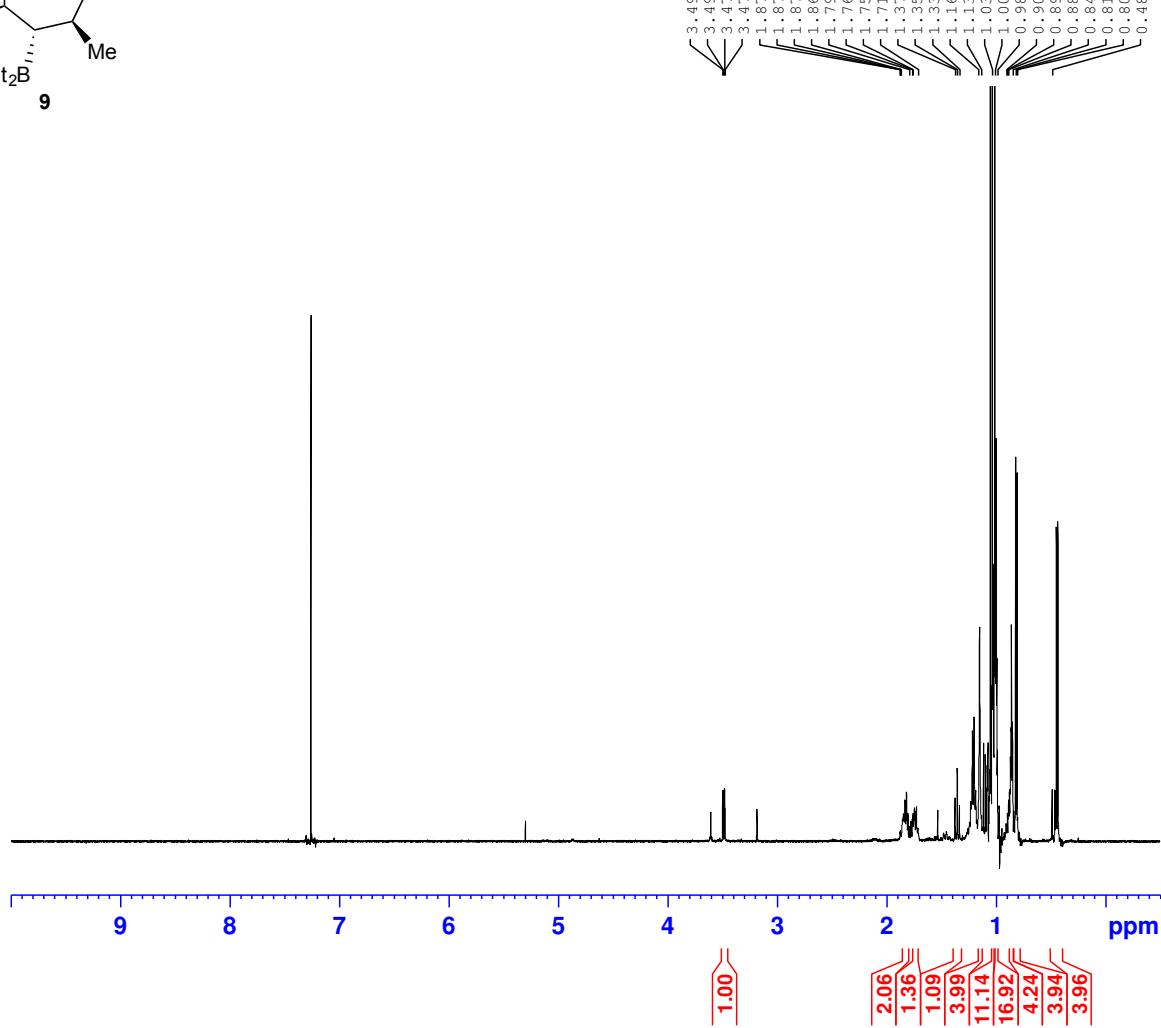
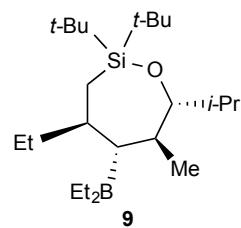












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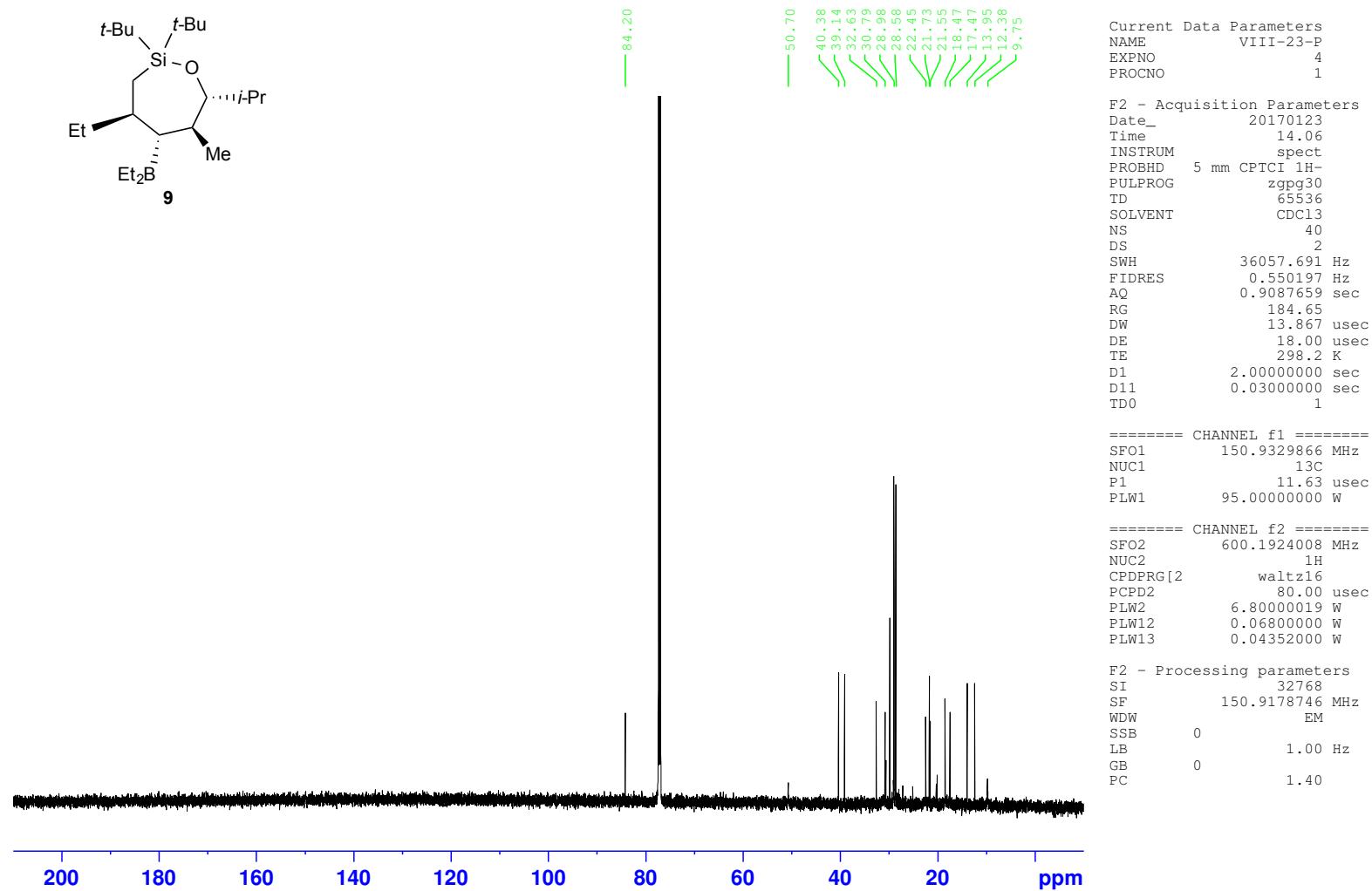
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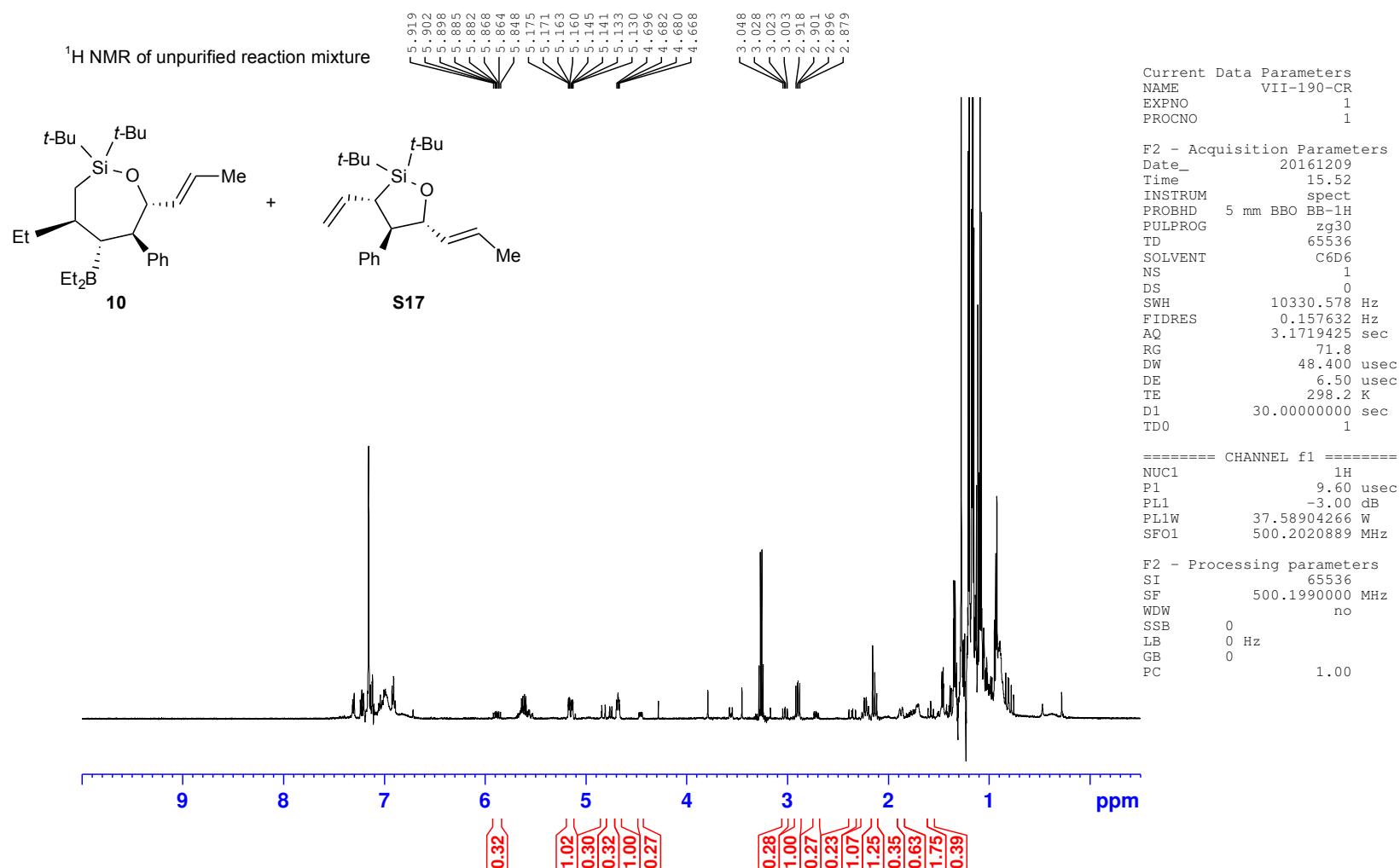
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FIDRES        0.157632 Hz
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DE             6.50 usec
TE             298.2 K
D1             2.0000000 sec
TD0              1

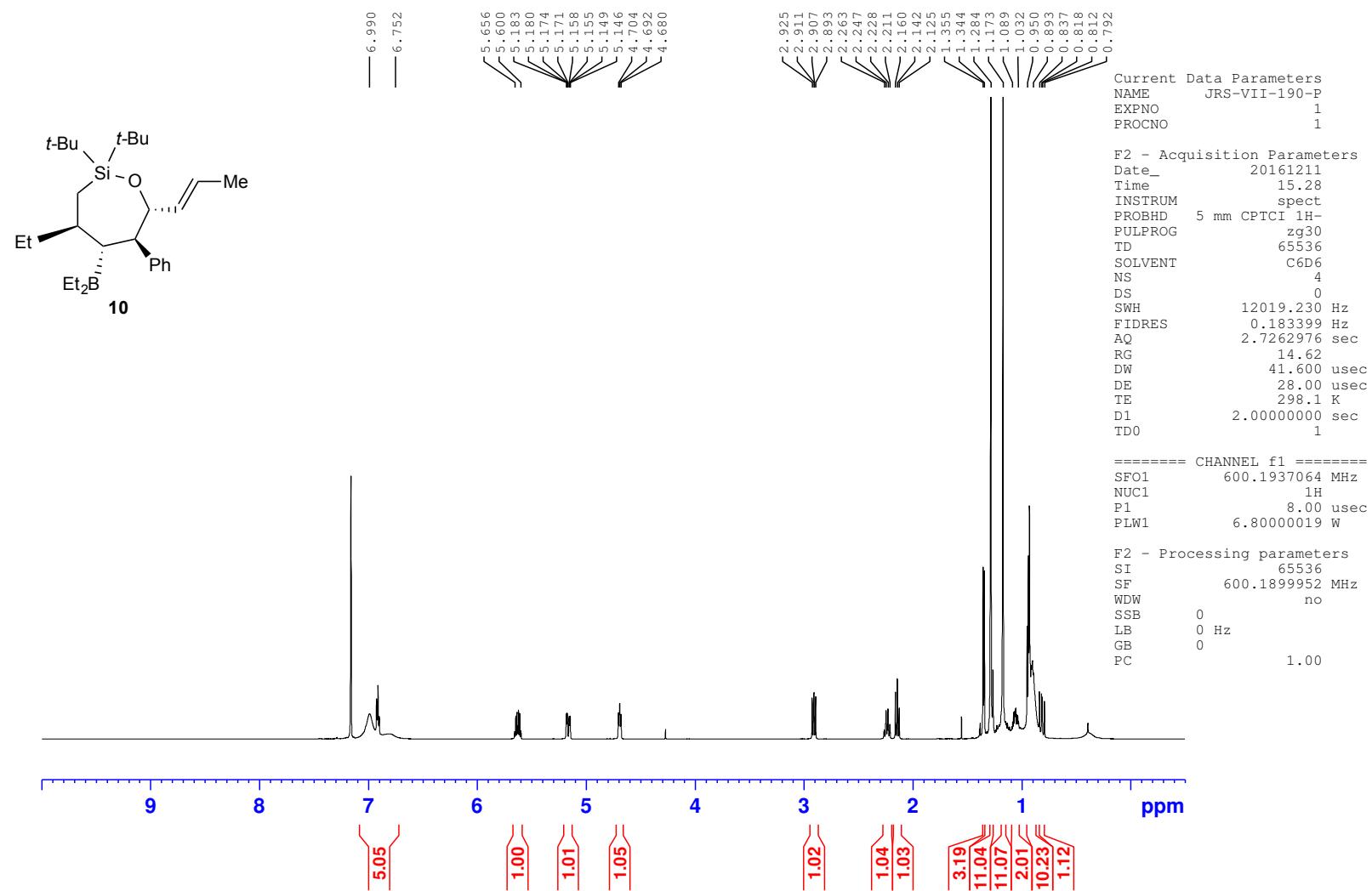
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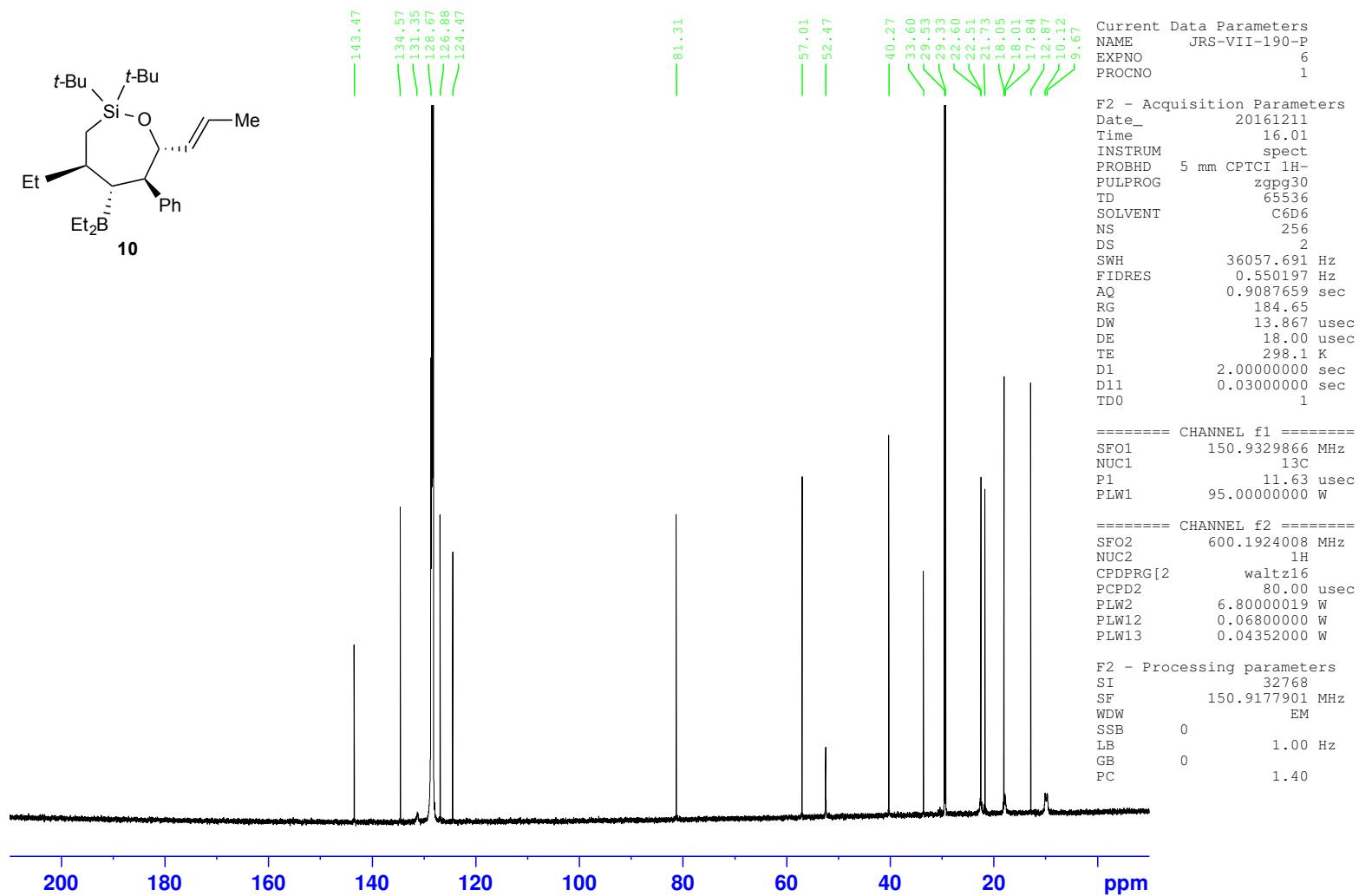
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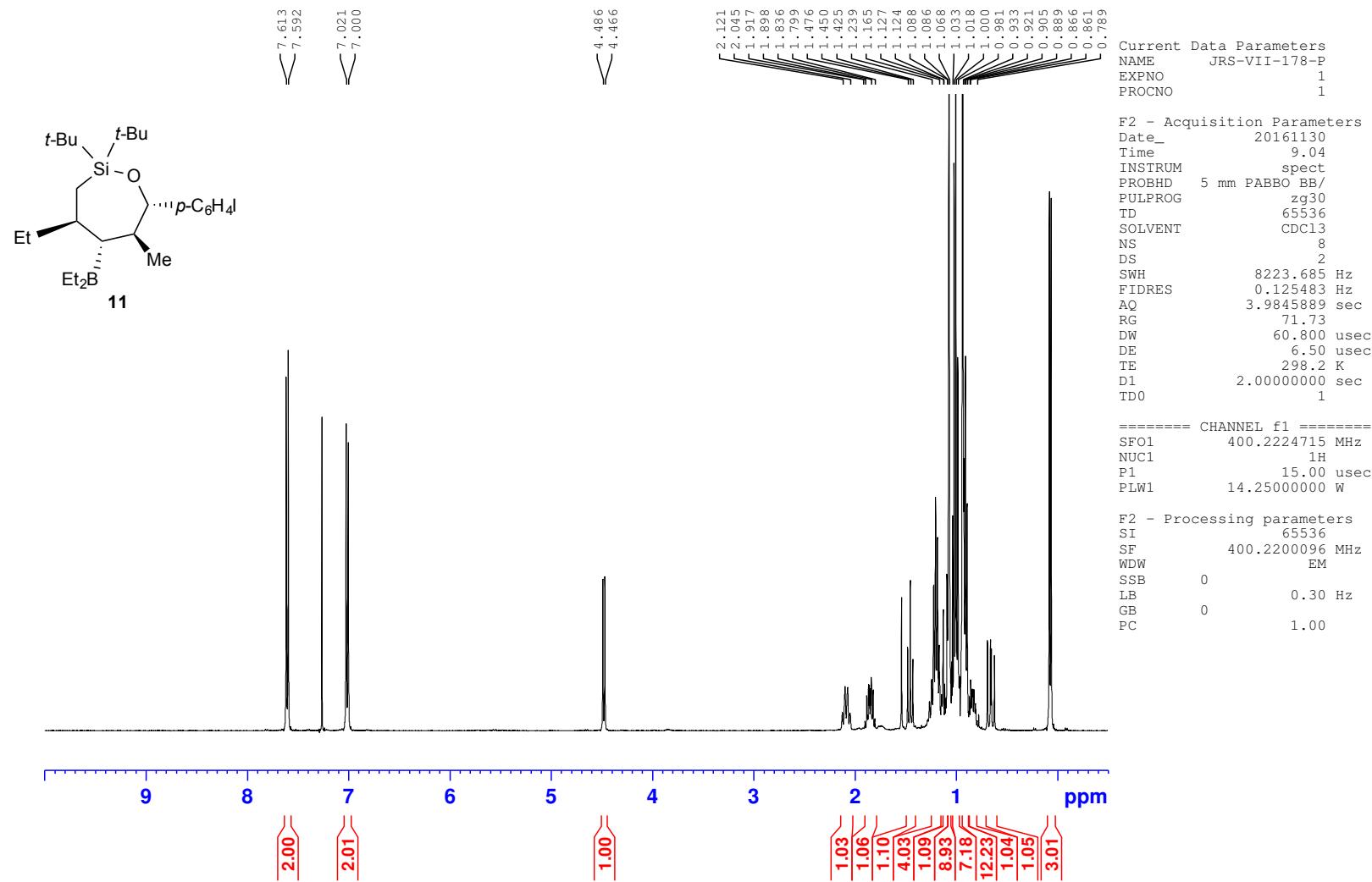
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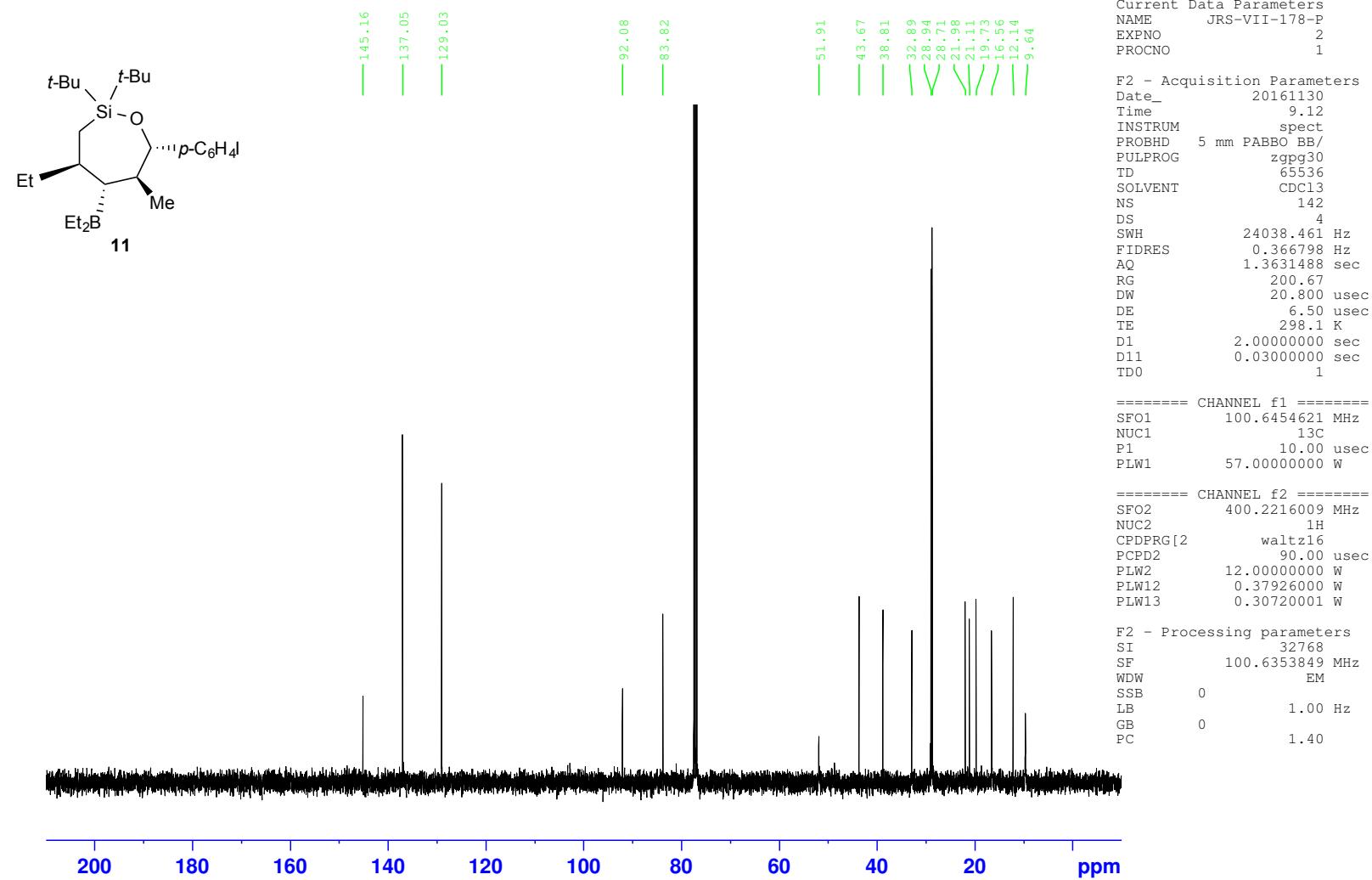












<sup>1</sup>H NMR of unpurified reaction mixture