

Supporting Information

A Bifunctional Iridium Catalyst Modified for Persistent Hydrogen Generation from Formic Acid: Understanding Deactivation via Cyclometalation of a 1,2-Diphenylethylenediamine Motif

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General information

All manipulations of oxygen and moisture-sensitive materials were performed under a purified argon atmosphere using standard Schlenk techniques. Solvents and formic acid were purchased from Kanto Chemical Co., Inc. and dried by refluxing over sodium benzophenone ketyl (1,2-dimethoxyethane) or B₂O₃ (formic acid), and distilled under argon before use. Deionized water was produced by a Millipore Elix 3 system. THF-*d*₈ was degassed by three freeze-pump-thaw cycles and purified by trap-to-trap distillation after being dried with CaH₂. The Ir (**1a**, **1b**) complexes derived from TfDPEN were prepared according to our previous paper¹. ¹H (399.8 MHz), ¹⁹F (376.2 MHz), and ¹³C{¹H} (100.5 MHz) NMR were recorded on a JEOL JNM-ECX400 spectrometer at 25 °C. The NMR chemical shifts were referenced to an external tetramethylsilane signal (0.0 ppm) by using the signals of residual proton impurities in the deuterated solvents for ¹H and ¹³C, and referenced to an external CF₃CO₂H signal (-76.5 ppm) for ¹⁹F NMR. Abbreviations for NMR are as follows: s = singlet, d = doublet, t = triplet, q = quartet, sept = septet, m = multiplet, or br = broad.

Representative procedures for the catalytic evolution of hydrogen from formic acid

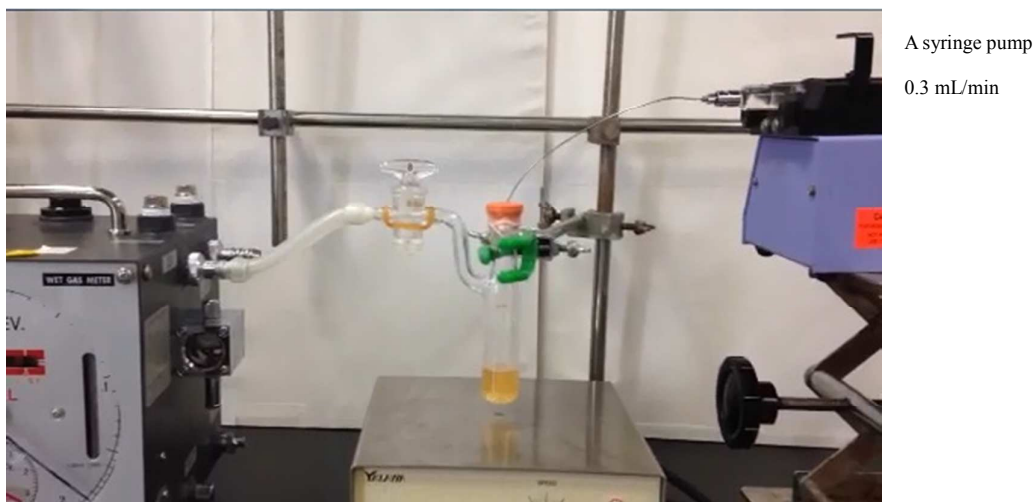
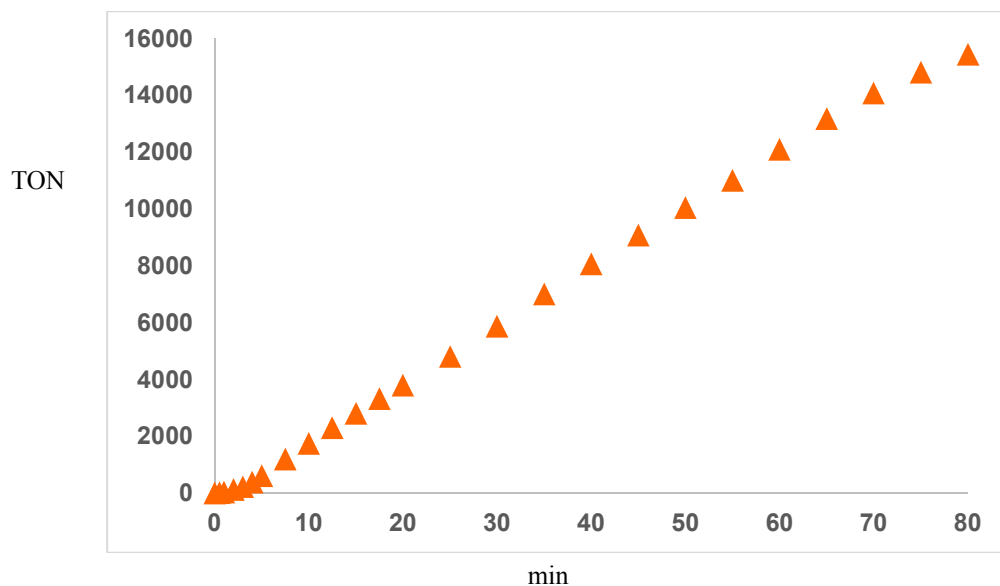


Figure S1. The experimental set up

To a solution of the iridium catalyst **1b** (20.9 mg, 0.030 mmol) in DME (3 mL) and degassed water (3 mL), formic acid was added for 10 min by using a syringe pump (0.3 mL/min). Using a mass-flowmeter, the rate of the gas evolution amount was measured.



S/C = 1.59×10^4 (**5a** 0.01 mmol, HCOOH 6 mL (0.075 mL/min), DME 6 mL, H₂O 6 mL)

Figure S2. Time versus TON plots for the catalytic acceptorless dehydrogenation from formic acid using complex **5b** at 60 °C.

Preparation and characterization data for Ir complexes

Thermal transformation of the hydrido-iridium complex **1b** to iridacycles **2** and **3**

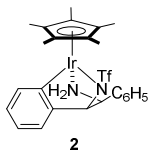
<in DME>

A solution of the hydrido-iridium complex **1b** (139.3 mg, 0.207 mmol) in DME (10 mL) was stirred under reflux conditions for 38 h. After the removal of the solvent, the crude product (**2:3** = 82:18, determined by ^1H NMR) was purified by silica gel flash column chromatography (eluted with hexane and diethyl ether (1:2); TLC information: R_f = 0.33 for **2**, and R_f = 0.22 for **3**) to give the pure complexes **2** (56.5 mg, 0.084 mmol, 41% isolated yield).

<in DME/H₂O mixed solvent>

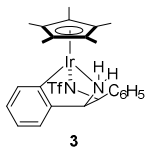
A solution of the hydrido-iridium complex **1b** (149.8 mg, 0.223 mmol) in DME and water mixed solvent (each 5 mL) was stirred at 80 °C for 50 min. After the removal of the DME, dichloromethane (15 mL) was introduced and water layer was separated. The crude product (129.3 mg, 0.193 mmol, 87% crude yield; **2:3** = 14:86, determined by ^1H NMR) was purified by silica gel flash column chromatography (eluted with hexane and diethyl ether (1:2)) to give the pure complexes **3** (98.4 mg, 0.147 mmol, 66% isolated yield).

*Cp*Ir[κ^3 (*N,N'*,*C*)-(S,S)-NH₂CHPhCHNTfC₆H₄)] (**2**)*



^1H NMR (THF-*d*₈, r.t.): δ 1.85 (s, 15H; C₅(CH₃)₅), 3.13 (d, $^3J_{\text{HH}} = 8.4$ Hz, 1H; NH₂CHPhCHNTfC₆H₄), 3.78 (d, $^3J_{\text{HH}} = 10.6$ Hz, 1H; NH₂CHPhCHNTfC₆H₄), 4.75 (s, 1H; NH₂CHPhCHNTfC₆H₄), 4.97 (br, 1H; NH₂CHPhCHNTfC₆H₄), 6.80-6.85 (m, 2H; aryl), 7.14-7.17 (m, 1H; aryl), 7.21-7.25 (m, 1H; aryl), 7.29-7.32 (m, 2H; aryl), 7.36-7.38 (m, 2H; aryl), 7.74-7.76 (m, 1H; aryl); ^{19}F NMR (THF-*d*₈, r.t.): δ -76.2 (s, CF₃); $^{13}\text{C}\{^1\text{H}\}$ NMR (CD₂Cl₂, r.t.): δ 9.02 (C₅(CH₃)₅), 54.4 (NH₂CHC₆H₅CHNTfC₆H₄), 78.0 (NH₂CHC₆H₅CHNTfC₆H₄), 86.9 (C₅(CH₃)₅), 120.7 (q, $^1J_{\text{CF}} = 327.8$ Hz; CF₃), 120.3, 123.2, 125.9, 126.9, 128.4, 128.7, 132.5, 139.5, 151.2, 152.0 (NH₂CHC₆H₅CHNTfC₆H₄). Elemental analysis calcd(%) for C₂₅H₂₈F₃IrN₂O₂S: C, 44.83; H, 4.21; N, 4.18; found: C, 44.45; H, 4.35; N, 4.20.

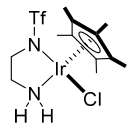
*Cp*Ir[κ^3 (*N,N'*,*C*)-(S,S)-NTfCHPhCHNH₂C₆H₄)] (**3**)*



^1H NMR (CD₂Cl₂, r.t.): δ 1.83 (s, 15H; C₅(CH₃)₅), 3.00 (br, 1H; NTfCHC₆H₅CHNH₂C₆H₄), 3.26 (br, 1H; NTfCHC₆H₅CHNH₂C₆H₄), 4.14 (s, 1H; NTfCHC₆H₅CHNH₂C₆H₄), 4.57 (s, 1H; NTfCHC₆H₅CHNH₂C₆H₄), 6.99-7.03 (m, 2H; aryl), 7.32-7.35 (m, 2H; aryl), 7.42-7.46 (m, 2H; aryl), 7.51-7.53 (m, 2H; aryl), 7.80-7.82 (m, 1H; aryl); ^{19}F NMR (CD₂Cl₂, r.t.): δ -76.0 (s, CF₃); $^{13}\text{C}\{^1\text{H}\}$ NMR (CD₂Cl₂, r.t.): δ 9.32 (C₅(CH₃)₅),

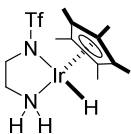
62.7 (NTfCHC₆H₅CHNH₂C₆H₄), 71.5 (NTfCHC₆H₅CHNH₂C₆H₄), 87.4 (C₅(CH₃)₅), 119.8 (q, ¹J_{CF} = 332.7 Hz; CF₃), 121.5, 123.0, 126.0, 126.8, 127.7, 129.1, 133.6, 142.0, 148.0, 150.2 (NTfCHC₆H₅CHNH₂C₆H₄). Elemental analysis calcd(%) for C₂₅H₂₈F₃IrN₂O₂S: C, 44.83; H, 4.21; N, 4.18; found: C, 45.01; H, 4.48; N, 4.29.

Cp*IrCl[κ²(N,N')-NTfCH₂CH₂NH₂]



¹H NMR (CD₂Cl₂, r.t.): δ 1.64 (s, 15H; C₅(CH₃)₅), 2.96 (br, 4H; NTfCH₂CH₂NH₂), 3.80 (br, 1H; NTfCH₂CH₂NH₂), 3.99 (br, 1H; NTfCH₂CH₂NH₂); ¹⁹F NMR (CD₂Cl₂, r.t.): δ -74.2 (s, CF₃); ¹³C{¹H} NMR (DMSO-*d*₆): δ 9.19 (C₅(CH₃)₅), 47.7 (NTfCH₂CH₂NH₂), 51.8 (NTfCH₂CH₂NH₂), 85.3 (C₅(CH₃)₅), 121.1 (q, ¹J_{CF} = 329.7 Hz; CF₃). Elemental analysis calcd(%) for C₁₄H₂₃Cl₃F₃IrN₂O₂S: C, 26.32; H, 3.63; N, 4.38; found: C, 26.66; H, 3.43; N, 4.41.

Cp*IrH[κ²(N,N')-NTfCH₂CH₂NH₂]



¹H NMR (THF-*d*₈, r.t.): δ -11.3 (s, 1H; IrH), 1.70 (s, 15H; C₅(CH₃)₅), 2.42 (m, 2H; NTfCH₂CH₂NH₂), 2.82 (br, 2H; NTfCH₂CH₂NH₂), 4.43 (br, 1H; NTfCH₂CH₂NH₂), 5.05 (br, 1H; NTfCH₂CH₂NH₂); ¹⁹F NMR (THF-*d*₈, r.t.): δ -72.6 (s, CF₃); ¹³C{¹H} NMR (THF-*d*₈, r.t.): δ = 9.28 (C₅(CH₃)₅), 49.1 (NTfCH₂CH₂NH₂), 54.0 (NTfCH₂CH₂NH₂), 85.7 (C₅(CH₃)₅), 122.6 (q, ¹J_{CF} = 331.6 Hz; CF₃). Elemental analysis calcd(%) for C₁₃H₂₂F₃IrN₂O₂S: C, 30.05; H, 4.27; N, 5.39; found: C, 30.28; H, 4.48; N, 5.46.

X-ray structure determination

All measurements were made on a Rigaku Saturn CCD area detector equipped with graphite-monochromated Mo- $K\alpha$ radiation ($\lambda = 0.71070 \text{ \AA}$) under nitrogen stream at 123 K. Indexing was performed from eighteen images. The crystal-to-detector distance was 45.05 mm. The data were collected to a maximum 2θ value of 55.0° . A total of 720 oscillation images were collected. A sweep of data was carried out using ω scans from -110.0 to 70.0° in 0.5° steps, at $\chi = 45.0^\circ$ and $\phi = 0.0^\circ$. A second sweep was performed using ω scans from -110.0 to 70.0° in 0.5° steps, at $\chi = 45.0^\circ$ and $\phi = 90.0^\circ$. Intensity data were collected for Lorentz-polarization effects as well as absorption. Structure solution and refinements were performed with the CrystalStructure program package. The heavy atom positions were determined by Direct methods (SIR2002), and the remaining non-hydrogen atoms were found by subsequent Fourier techniques. An empirical absorption correction based on equivalent reflections was applied to all data. All non-hydrogen atoms other than solvent molecules were refined anisotropically by full-matrix least-square techniques based on F^2 . The low-temperature data collection enabled hydrogens attached to the primary amine in **2**, **3**, **4**, and **5b** to be located from the Fourier difference map and refined isotropically. All other hydrogens either were not refined or were constrained to ride on their parent atom. Relevant crystallographic data are compiled in Tables S3 and S4.

Table S1. Comparison of selected bond lengths and angles of **2** and **3**.

Complex	Ir1-N1	Ir-N2	Ir-Cp*(av.)	∠N1-Ir1-N2
2^a	2.1595 Å	2.139 Å	2.1886 Å	75.95°
3	2.120(4) Å	2.166(4) Å	2.1926 Å	76.20(15) °

a) Mean value.

Table S2. Comparison of selected bond lengths and angles of **5b** and **1b¹**.

Complex	Ir1-N1	Ir-N2	Ir-Cp*(av.)	∠N1-Ir1-N2
5b	2.143(4) Å	2.136(5) Å	2.1908 Å	79.63(17)°
1b	2.126(2) Å	2.149(3) Å	2.195 Å	76.58(10) °

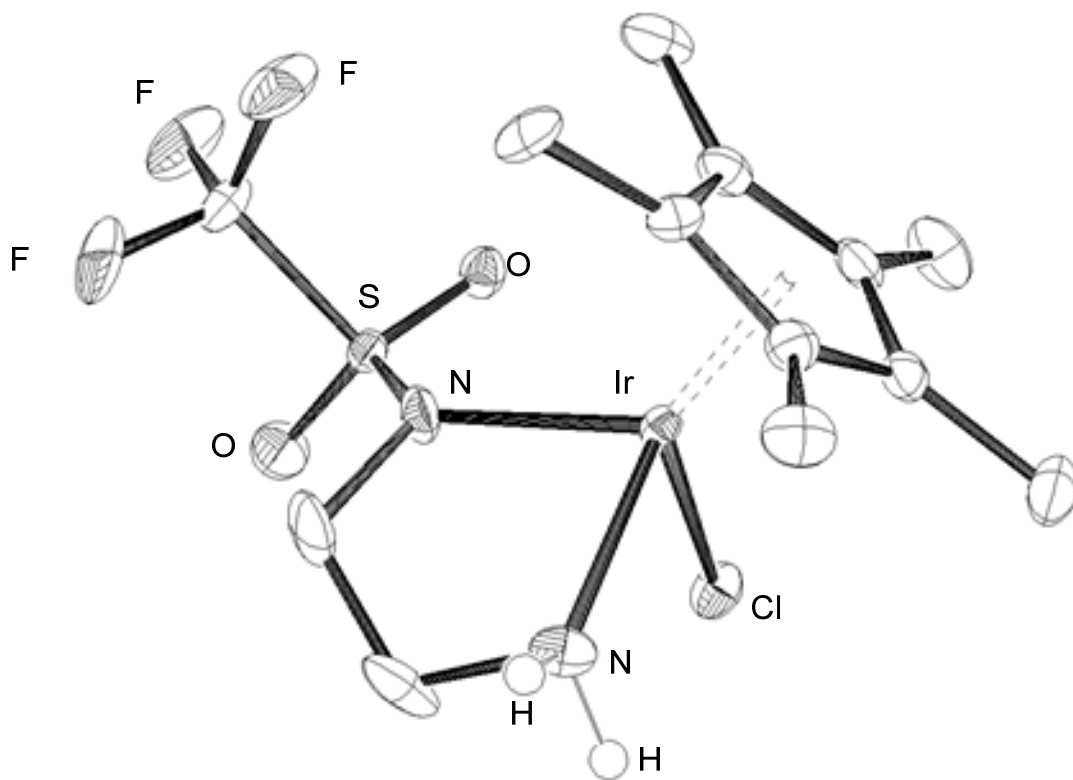


Figure S3. ORTEP drawing of a chloroiridium complex **4** derived from TfEN.

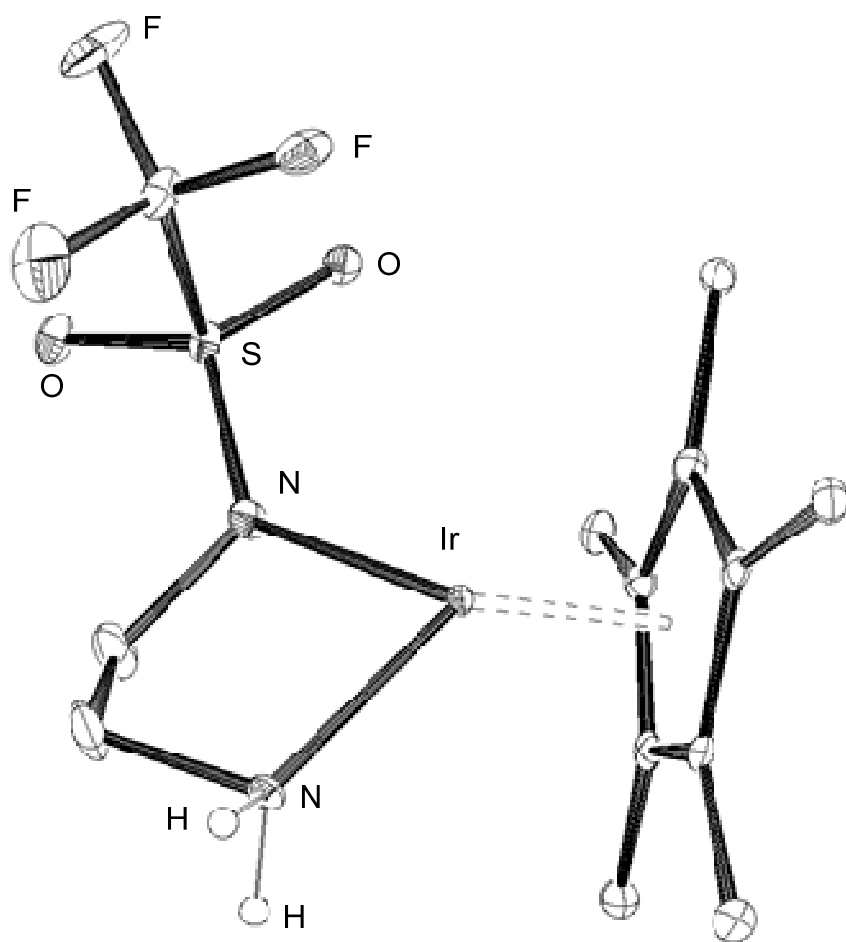


Figure S4. ORTEP drawing of a hydrido-iridium complex **5b** derived from TfEN.

Table S3. Crystallographic data for **2** and **3**

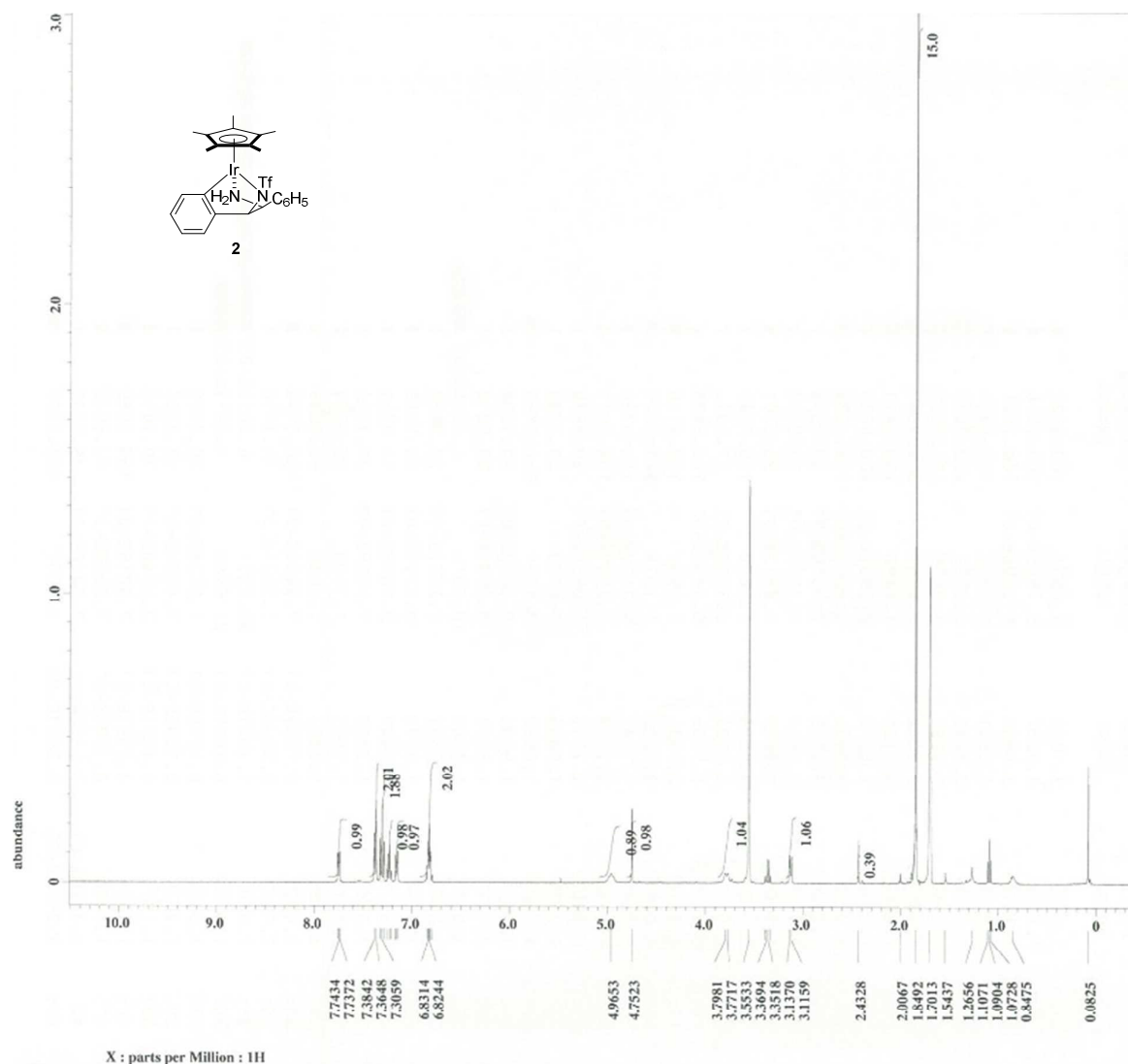
	2	3
Empirical Formula	C ₂₅ H ₂₈ F ₃ IrN ₂ O ₂ S	C ₂₅ H ₂₈ F ₃ IrN ₂ O ₂ S
Formula Weight	669.78	669.78
Crystal Color, Habit	yellow, prism	yellow, prism
Crystal System	monoclinic	orthohombic
Space Group	<i>C</i> 2 (#5)	<i>P</i> 2 ₁ 2 ₁ 2 ₁ (#19)
Lattice Parameters	<i>a</i> = 20.77760 Å <i>b</i> = 14.35170 Å <i>c</i> = 18.81300 Å	<i>a</i> = 11.1945(10) Å <i>b</i> = 12.8348(11) Å <i>c</i> = 17.138(2) Å <i>β</i> = 114.794(2) °
	<i>V</i> = 5234.74710 Å ³	<i>V</i> = 2462.4(4) Å ³
Z value	8	4
<i>D</i> _{calc}	1.700 g/cm ³	1.807 g/cm ³
<i>F</i> ₀₀₀	2624.00	1312.00
μ(MoKα)	52.401 cm ⁻¹	55.699 cm ⁻¹
Exposure Rate	10.0 sec./°	10.0 sec./°
No. of Reflections Measured	21609	20326
No. of unique reflections	9804	5646
No. Variables	670	336
<i>R</i> 1 (<i>I</i> > 2.00σ(<i>I</i>))	0.0359	0.0253
w <i>R</i> 2 (All reflections)	0.0935	0.0647
GOF on <i>F</i> ²	1.000	1.000

$$R1 = \Sigma ||F_o| - |F_c|| / \Sigma |F_o|, wR2 = [\Sigma (w (F_o^2 - F_c^2)^2) / \Sigma w(F_o^2)^2]^{1/2}$$

Table S4. Crystallographic data for **4·CH₂Cl₂** and **5b**

	4·CH₂Cl₂	5b
Empirical Formula	C ₁₄ H ₂₃ Cl ₃ F ₃ IrN ₂ O ₂ S	C ₁₃ H ₂₂ F ₃ IrN ₂ O ₂ S
Formula Weight	638.98	519.60
Crystal Color, Habit	yellow, prism	orange, prism
Crystal System	monoclinic	tetragonal
Space Group	<i>P</i> 2 ₁ / <i>c</i> (#14)	<i>I</i> 4 ₁ <i>cd</i> (#110)
Lattice Parameters	<i>a</i> = 11.168(3) Å <i>b</i> = 11.884(3) Å <i>c</i> = 16.488(4) Å <i>β</i> = 104.162(3) ° <i>V</i> = 2121.7(9) Å ³	<i>a</i> = 16.724(2) Å <i>c</i> = 23.136(3) Å <i>V</i> = 6470.7(12) Å ³
Z value	4	16
<i>D</i> _{calc}	2.000 g/cm ³	2.133 g/cm ³
<i>F</i> ₀₀₀	1232.00	4000.00
μ(MoKα)	68.226 cm ⁻¹	84.433 cm ⁻¹
Exposure Rate	6.0 sec./°	2.0 sec./°
No. of Reflections Measured	17038	25200
No. of unique reflections	4866	3673
No. Variables	258	221
<i>R</i> 1 (<i>I</i> > 2.00σ(<i>I</i>))	0.0381	0.0234
w <i>R</i> 2 (All reflections)	0.1014	0.0574
GOF on <i>F</i> ²	1.000	1.000

$$R1 = \Sigma ||F_o| - |F_c|| / \Sigma |F_o|, wR2 = [\Sigma (w (F_o^2 - F_c^2)^2) / \Sigma w(F_o^2)^2]^{1/2}$$



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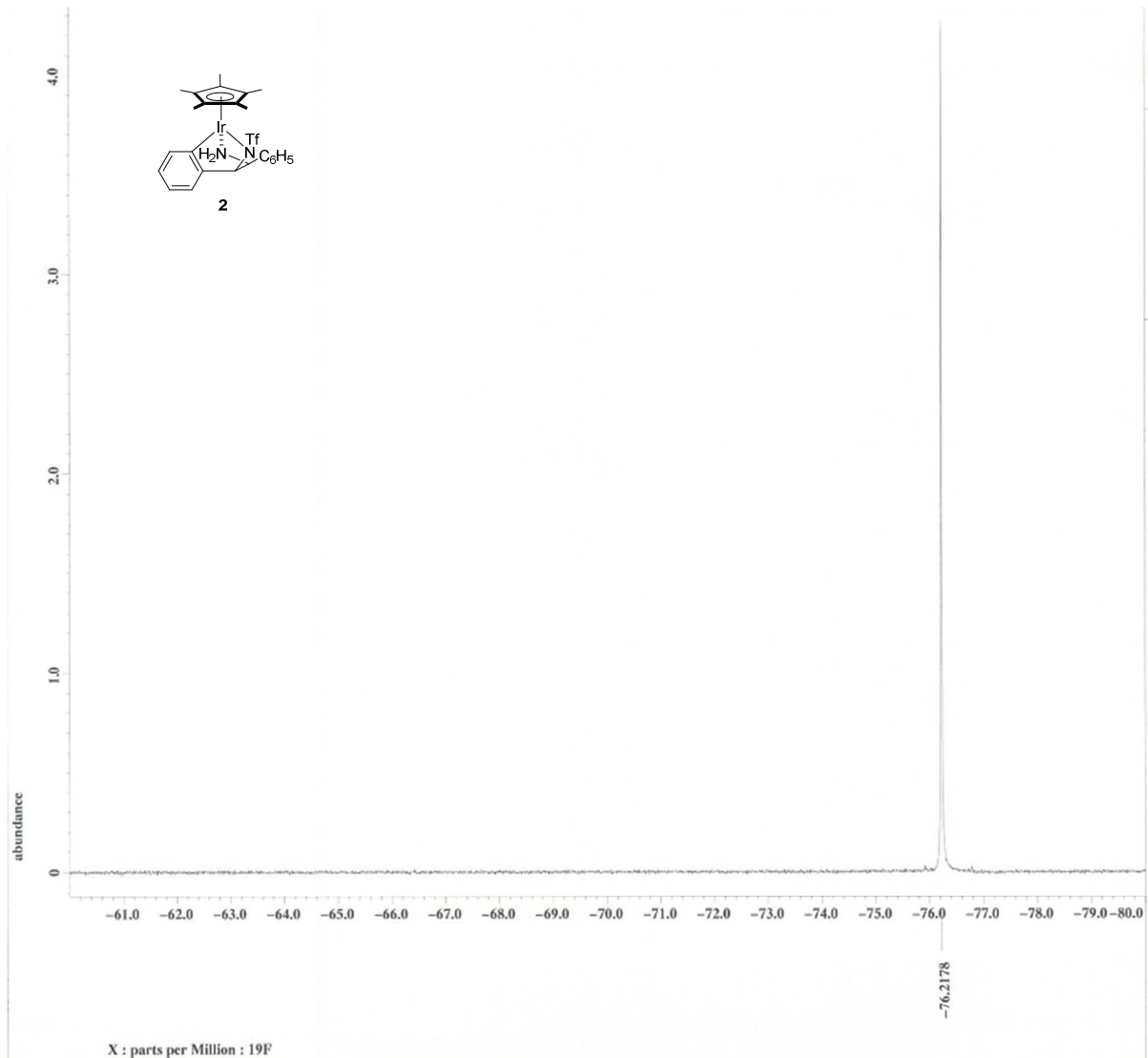
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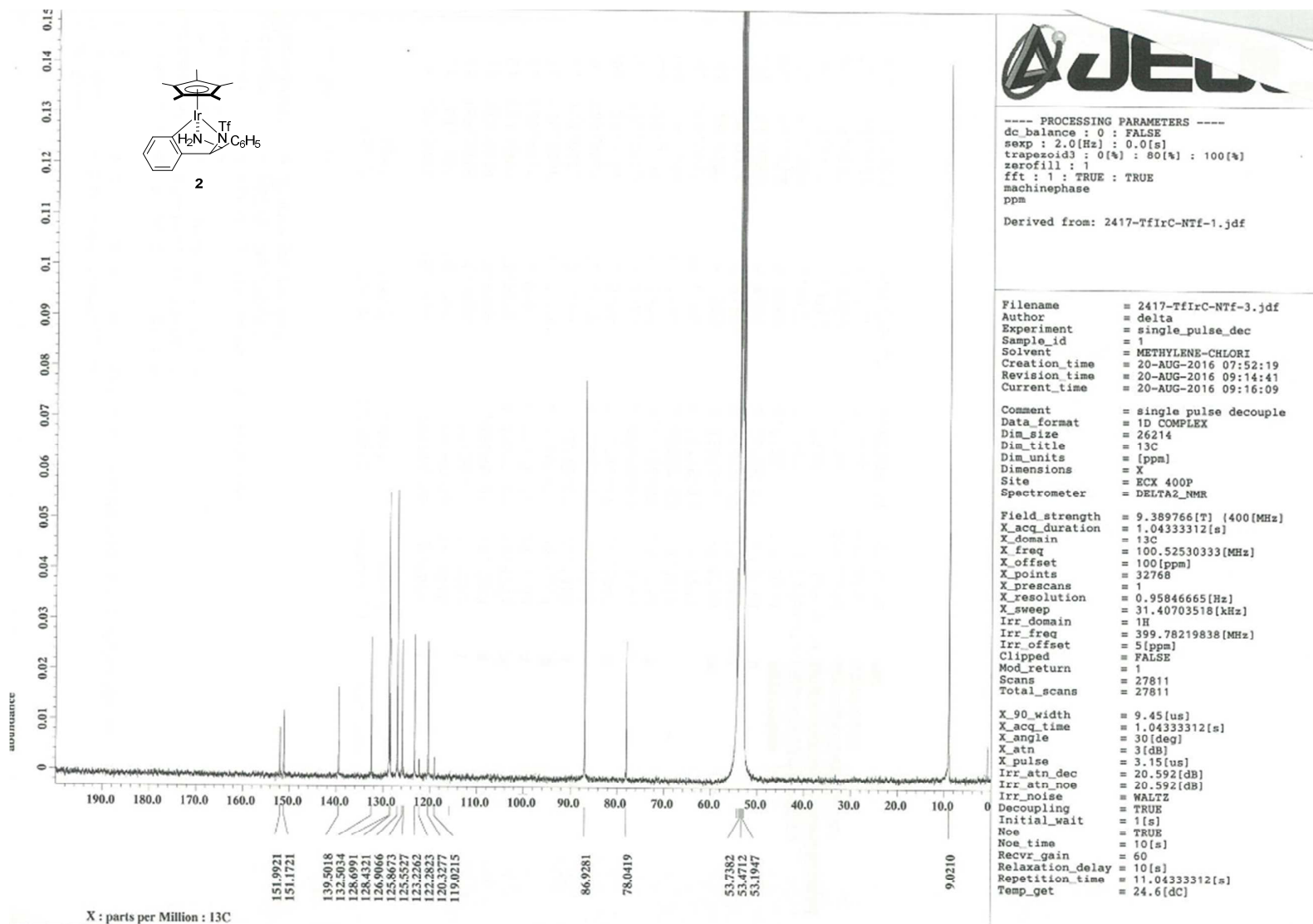
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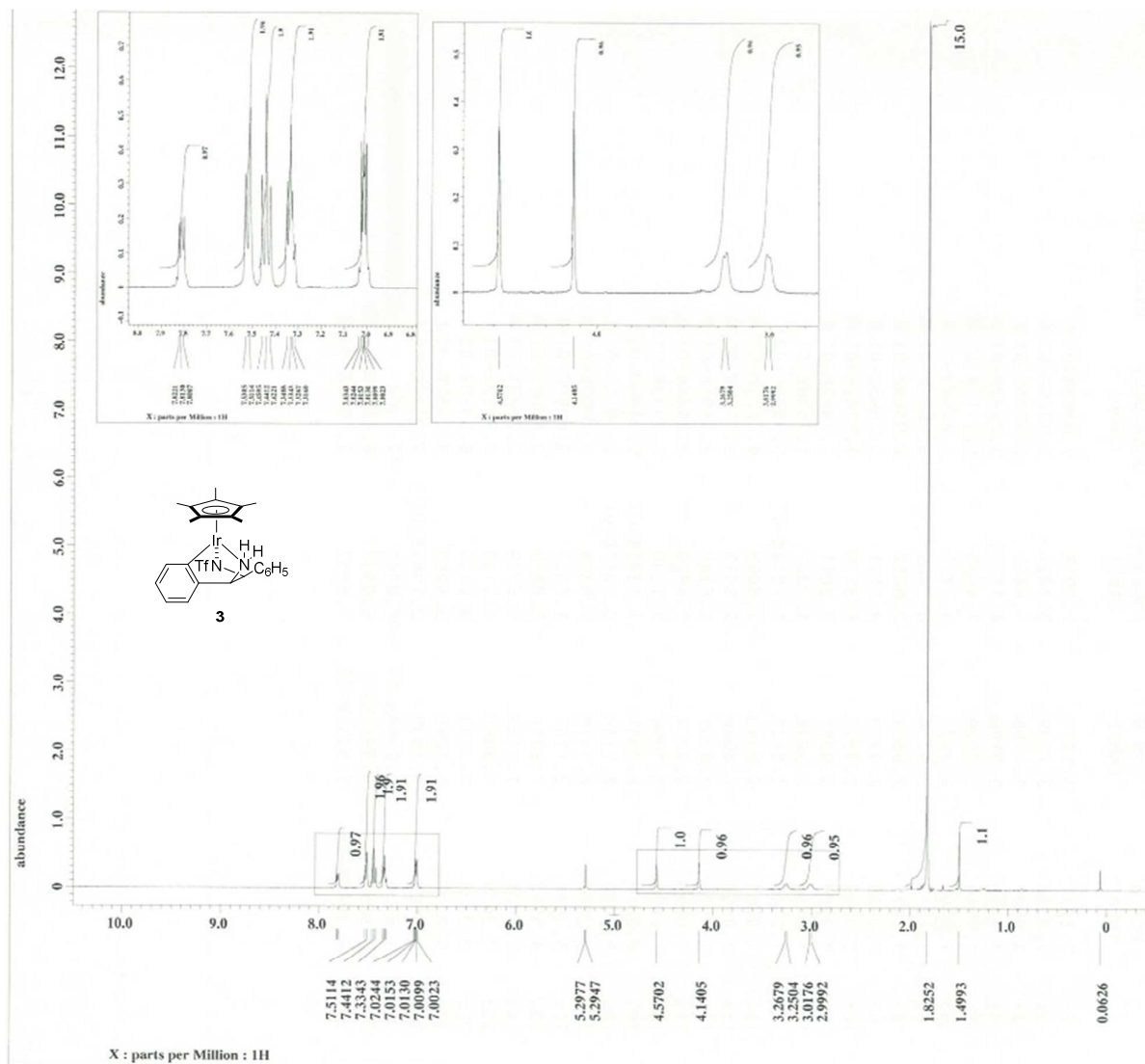
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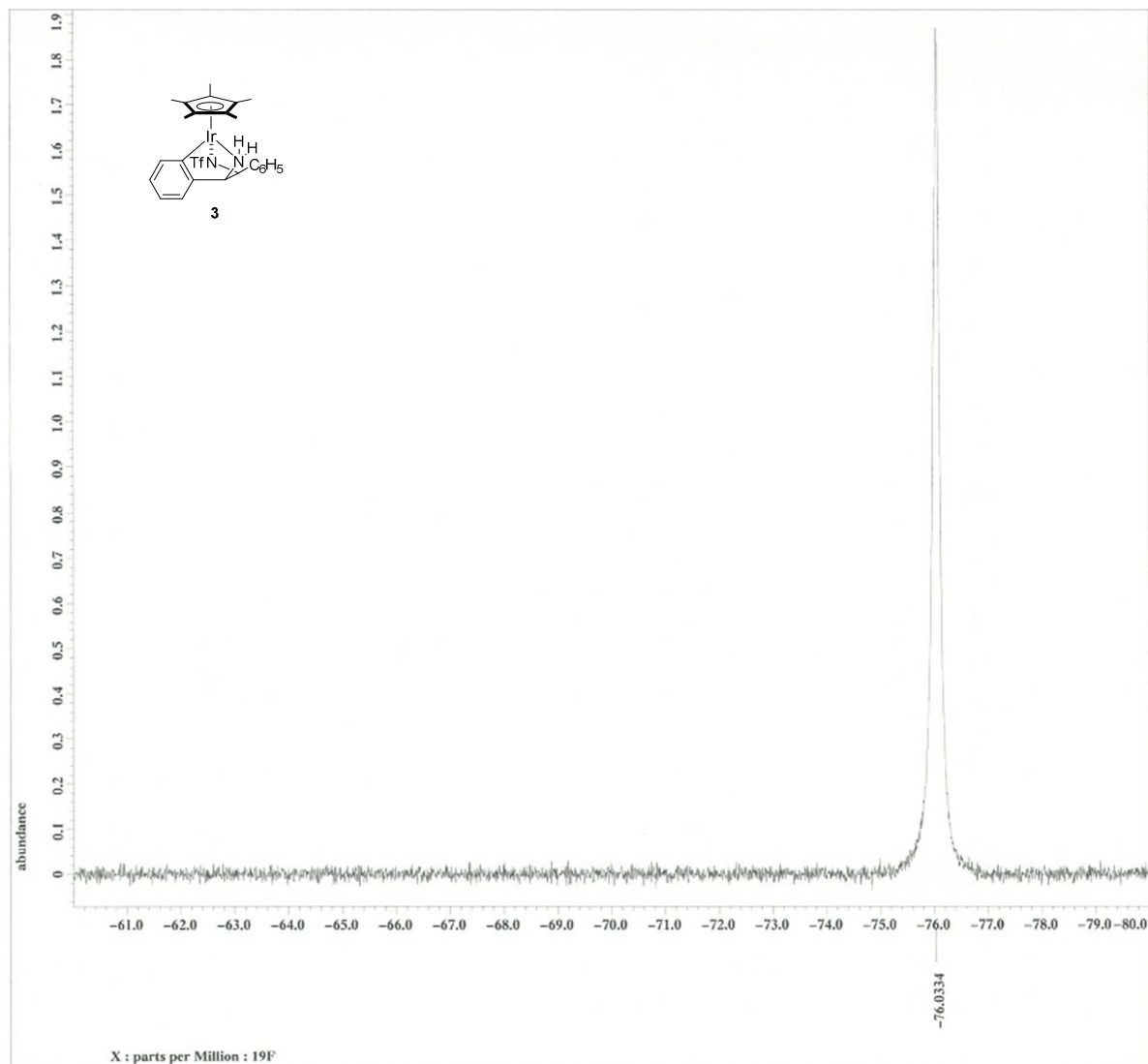
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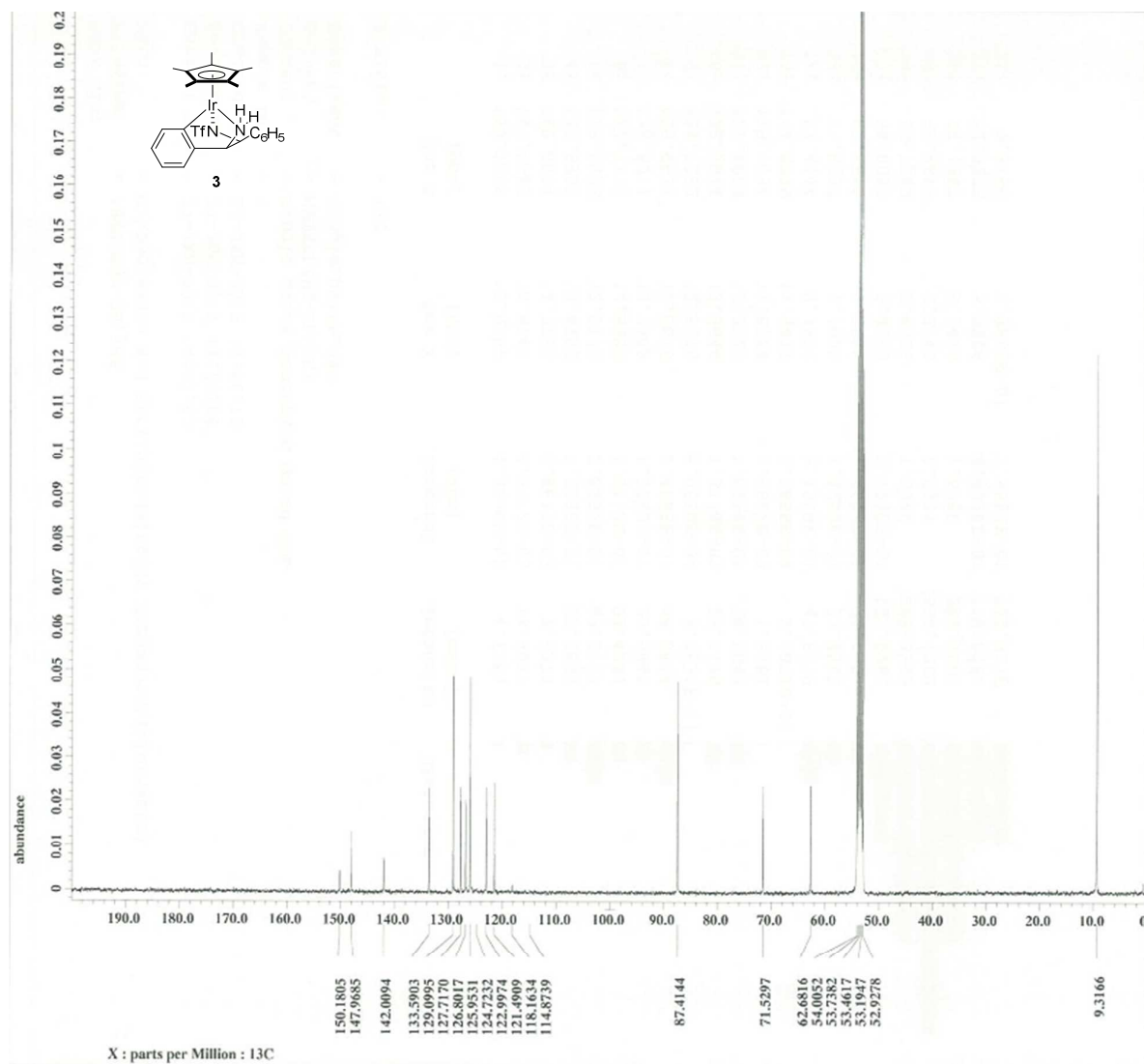
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 phase : 80 : 0 : 50[%]
 phase : 0 : 0 : 50[%]
 phase : -80 : 0 : 50[%]
 base_correct : None : 0 : Smooth
 Derived from: 1847-19F-1.jdf

Filename = 1847-19F-6.jdf
 Author = delta
 Experiment = single_pulse.ex2
 Sample_id = 1
 Solvent = METHYLENE-CHLORI
 Creation_time = 22-JAN-2017 14:49:24
 Revision_time = 22-JAN-2017 16:11:26
 Current_time = 22-JAN-2017 16:11:41
 Comment = single_pulse
 Data_format = 1D REAL
 Dim_size = 52428
 Dim_title = 19F
 Dim_units = [ppm]
 Dimensions = X
 Site = ECK 400P
 Spectrometer = DELTA2_NMR
 Field_strength = 9.389766[T] (400[MHz])
 X_acq_duration = 0.34603008[s]
 X_domain = 19F
 X_freq = 376.17105393[MHz]
 X_offset = 0[ppm]
 X_points = 65536
 X_prescans = 1
 X_resolution = 2.88992217[Hz]
 X_sweep = 189.3939393[kHz]
 Irr_domain = 19F
 Irr_freq = 376.17105393[MHz]
 Irr_offset = 5[ppm]
 Tri_domain = 19F
 Tri_freq = 376.17105393[MHz]
 Tri_offset = 5[ppm]
 Clipped = FALSE
 Mod_return = 1
 Scans = 7
 Total_scans = 7
 X_90_width = 16.1[us]
 X_acq_time = 0.34603008[s]
 X_angle = 45[deg]
 X_atn = 3[dB]
 X_pulse = 8.05[us]
 Irr_mode = Off
 Tri_mode = Off
 Dante_presat = FALSE
 Initial_wait = 1[s]
 Recvr_gain = 52
 Relaxation_delay = 5[s]
 Repetition_time = 5.34603008[s]
 Temp_get = 36.5[degC]



----- PROCESSING PARAMETERS -----
 dc_balance : 0 : FALSE
 secp : 2.0[Hz] : 0.0[s]
 trapezoid3 : 0[%] : 80[%] : 100[%]
 fft : 1 : TRUE : TRUE
 machinephase
 ppm

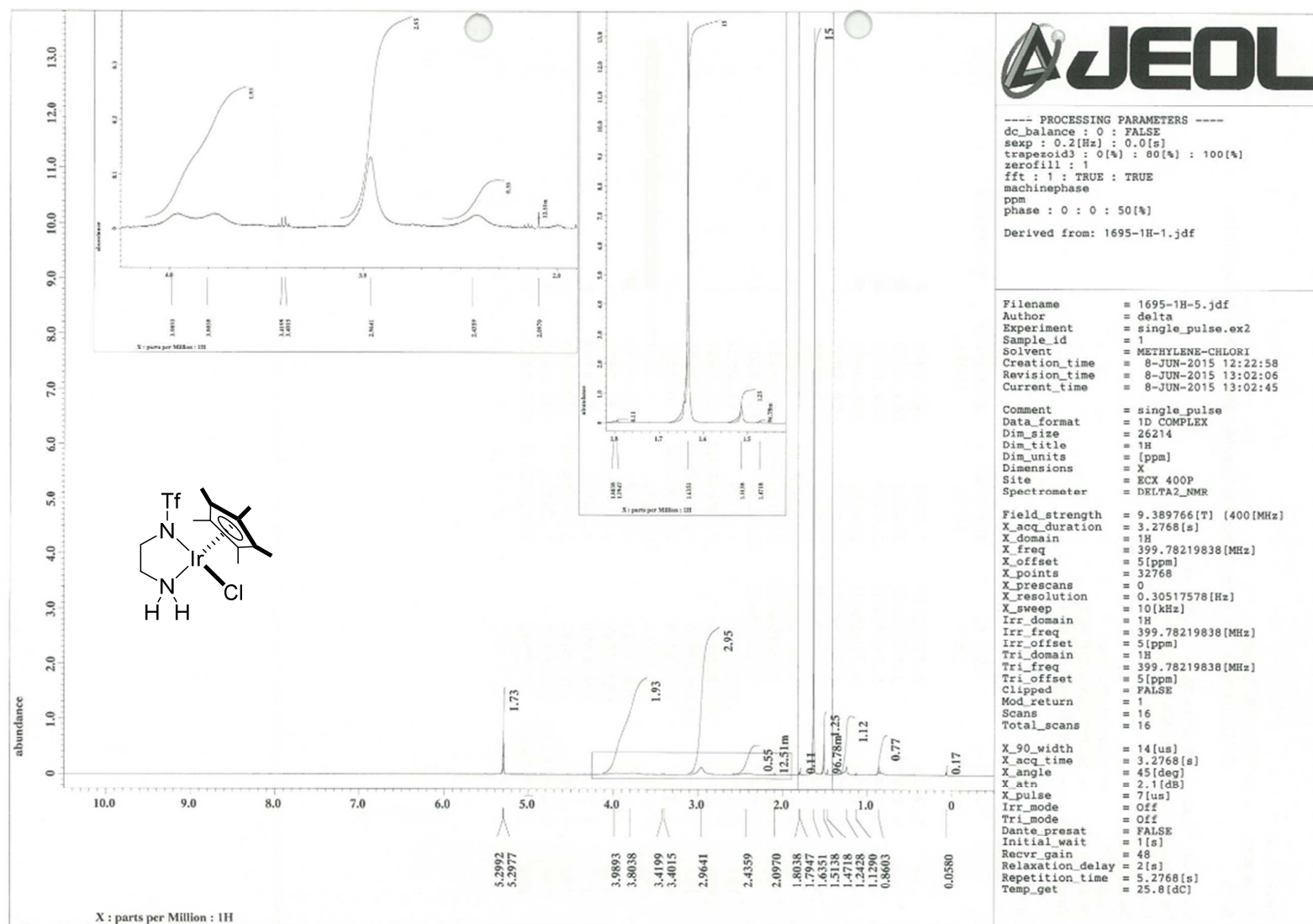
Derived from: 1847-13C-1.jdf

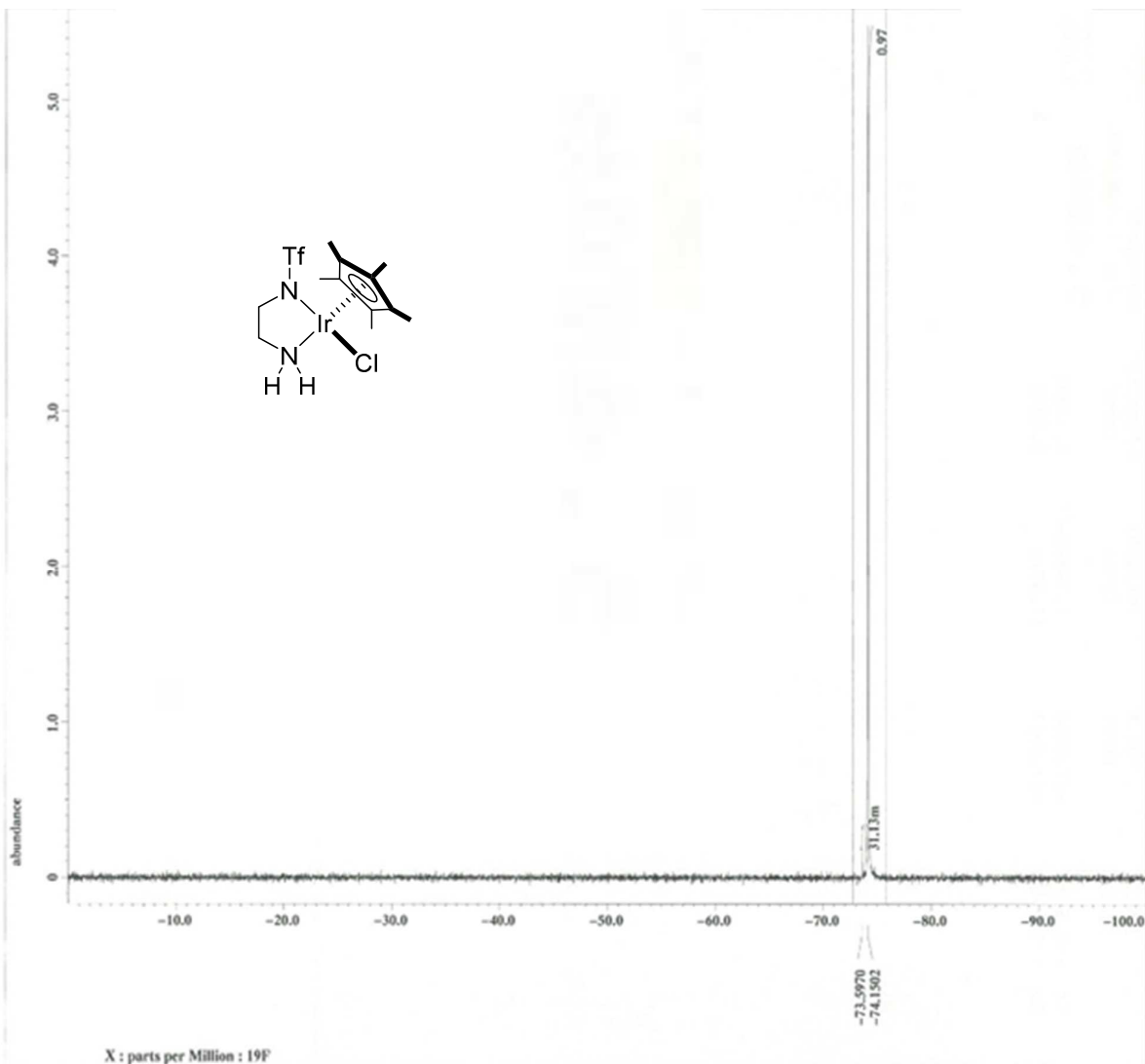
Filename = 1847-13C-25.jdf
 Author = delta
 Experiment = single_pulse_dec
 Sample_id = 1
 Solvent = METHYLENE-CHLORI
 Creation_time = 22-JAN-2017 14:44:19
 Revision_time = 27-JAN-2017 14:22:04
 Current_time = 27-JAN-2017 14:24:12

Comment = single pulse decouple
 Data_format = 1D COMPLEX
 Dim_size = 26214
 Dim_title = 13C
 Dim_units = [ppm]
 Dimensions = X
 Site = RCX 400P
 Spectrometer = DELTA2_NMR

Field_strength = 9.389766[T] {400[MHz]}
 X_acq_duration = 1.04333312[s]
 X_domain = 13C
 X_freq = 100.52530333[MHz]
 X_offset = 100[ppm]
 X_points = 32768
 X_prescans = 2
 X_resolution = 0.95846665[Hz]
 X_sweep = 31.40703518[kHz]
 Irr_domain = 1H
 Irr_freq = 399.78219838[MHz]
 Irr_offset = 5[ppm]
 Clipped = FALSE
 Mod_return = 1
 Scans = 16131
 Total_scans = 16131

X_90_width = 9.45[us]
 X_acq_time = 1.04333312[s]
 X_angle = 30[deg]
 X_atn = 3[db]
 X_pulse = 3.15[us]
 Irr_atn_dec = 20.592[db]
 Irr_atn_noe = 20.592[db]
 Irr_noise = WALTZ
 Decoupling = TRUE
 Initial_wait = 1[s]
 Noe = TRUE
 Noe_time = 3[s]
 Recvr_gain = 54
 Relaxation_delay = 3[s]
 Repetition_time = 4.04333312[s]
 Temp_get = 42[dc]





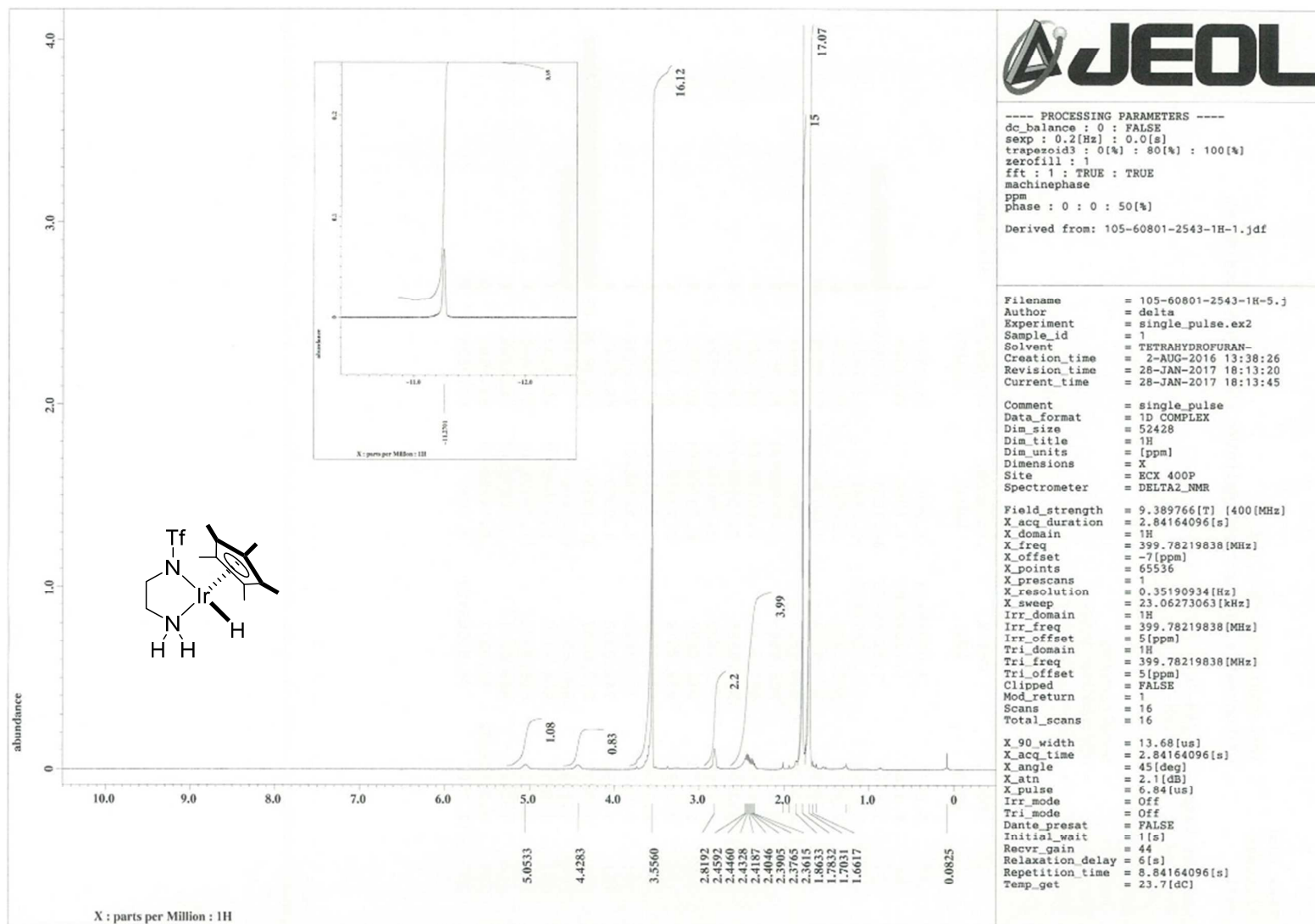
----- PROCESSING PARAMETERS -----
 dc_balance : 0 : FALSE
 secp : 0.2[Hz] : 0.0[s]
 trapezoid3 : 0[%] : 80[%] : 100[%]
 zerofill : 1
 fft : 1 : TRUE : TRUE
 machinephase
 ppm
 reference : -77.638[ppm] : -76.5[ppm]
 phase : 0 : 0 : 50[%]
 phase : 0 : 0 : 50[%]
 phase : -4 : 0 : 50[%]
 base_correct : None : 0 : Smooth

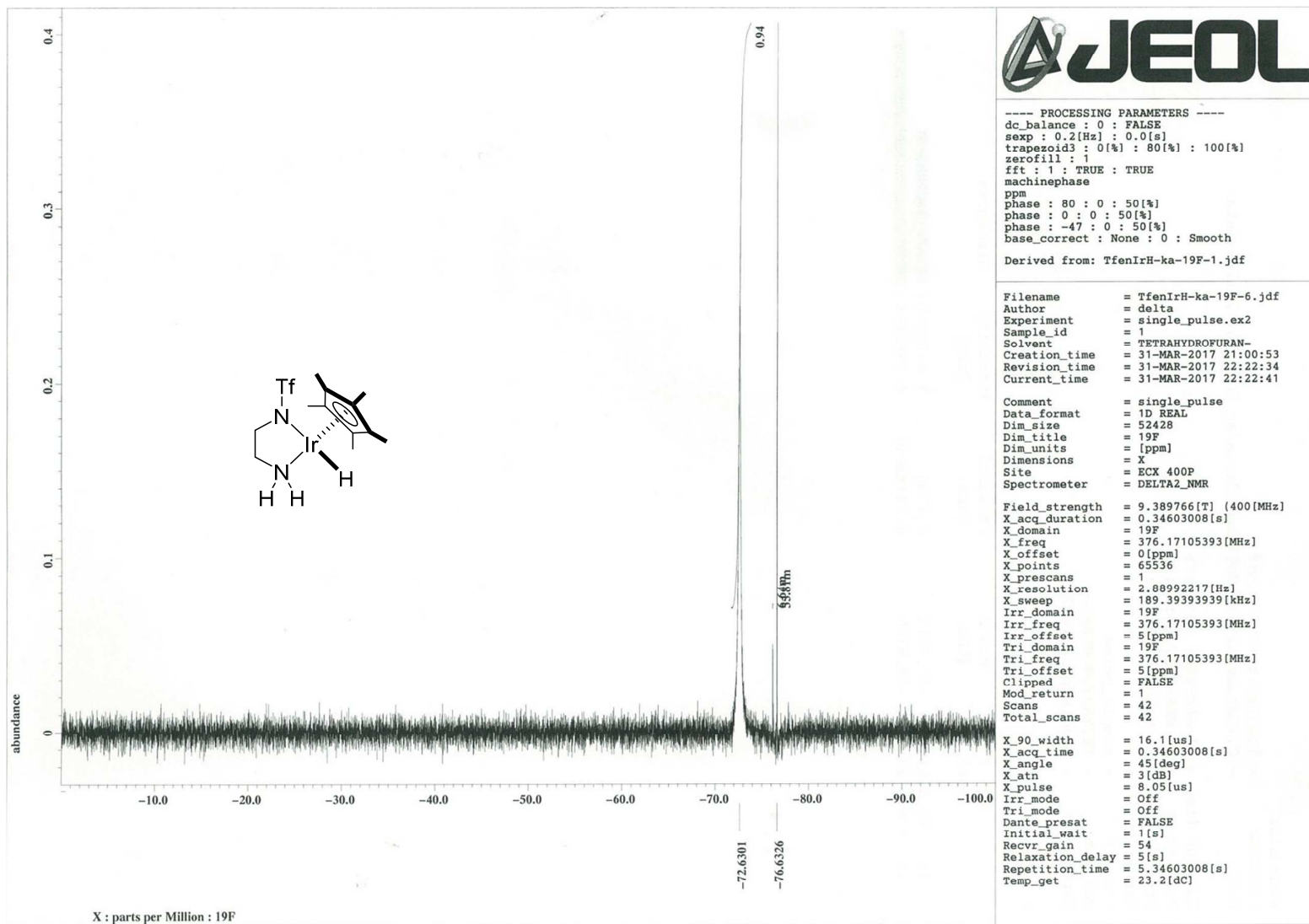
Filename = 1695-19F-4.jdf
 Author = delta
 Experiment = single_pulse.ex2
 Sample_id = 1
 Solvent = METHYLENE-CHLOR1
 Creation_time = 8-JUN-2015 12:20:10
 Revision_time = 8-JUN-2015 12:57:28
 Current_time = 8-JUN-2015 12:57:52

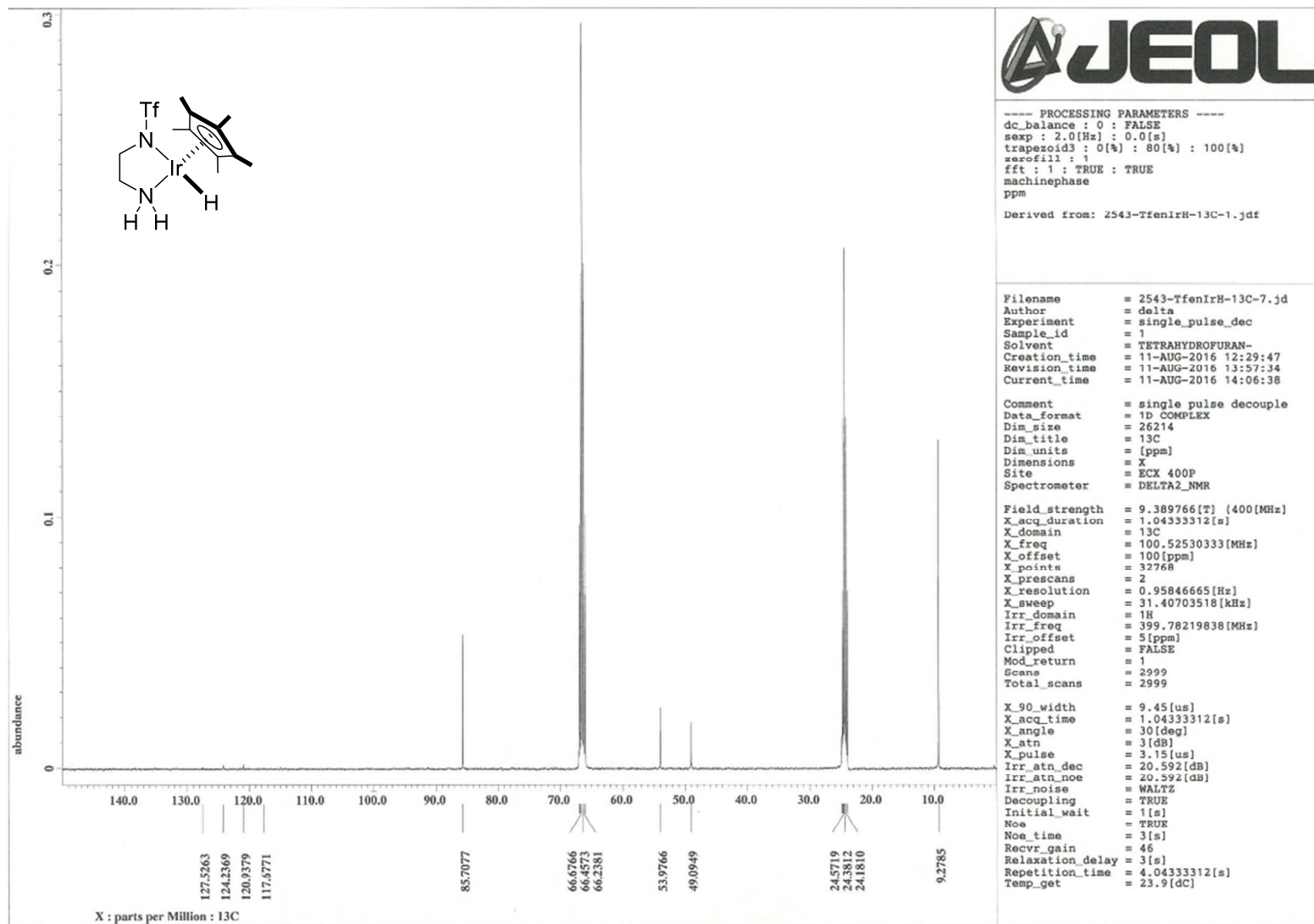
Comment = single_pulse
 Data_format = 1D_REAL
 Dim_size = 26214
 Dim_title = 19F
 Dim_units = [ppm]
 Dimensions = X
 Site = ECX 400P
 Spectrometer = DELTA2_NMR

Field_strength = 9.389766[T] {400[MHz]
 X_acq_duration = 0.17301504[s]
 X_domain = 19F
 X_freq = 376.17105393[MHz]
 X_offset = 0[ppm]
 X_points = 32768
 X_prescans = 1
 X_resolution = 5.77984434[Hz]
 X_sweep = 189.3939393[kHz]
 Irr_domain = 19F
 Irr_freq = 376.17105393[MHz]
 Irr_offset = 5[ppm]
 Tri_domain = 19F
 Tri_freq = 376.17105393[MHz]
 Tri_offset = 5[ppm]
 Clipped = FALSE
 Mod_return = 1
 Scans = 12
 Total_scans = 12

X_90_width = 16.1[us]
 X_acq_time = 0.17301504[s]
 X_angle = 45[deg]
 X_atn = 3[db]
 X_pulse = 8.05[us]
 Irr_mode = Off
 Tri_mode = Off
 Dante_presat = FALSE
 Initial_wait = 1[s]
 Recvr_gain = 54
 Relaxation_delay = 5[s]
 Repetition_time = 5.17301504[s]
 Temp_get = 25.9[degC]







Reference

- 1 A. Matsunami, Y. Kayaki, T. Ikariya, *Chem. Eur. J.* **2015**, *21*, 13513.