

Supporting Information

Enhanced energy storage density and high efficiency of lead-free

CaTiO₃-BiScO₃ dielectric ceramics

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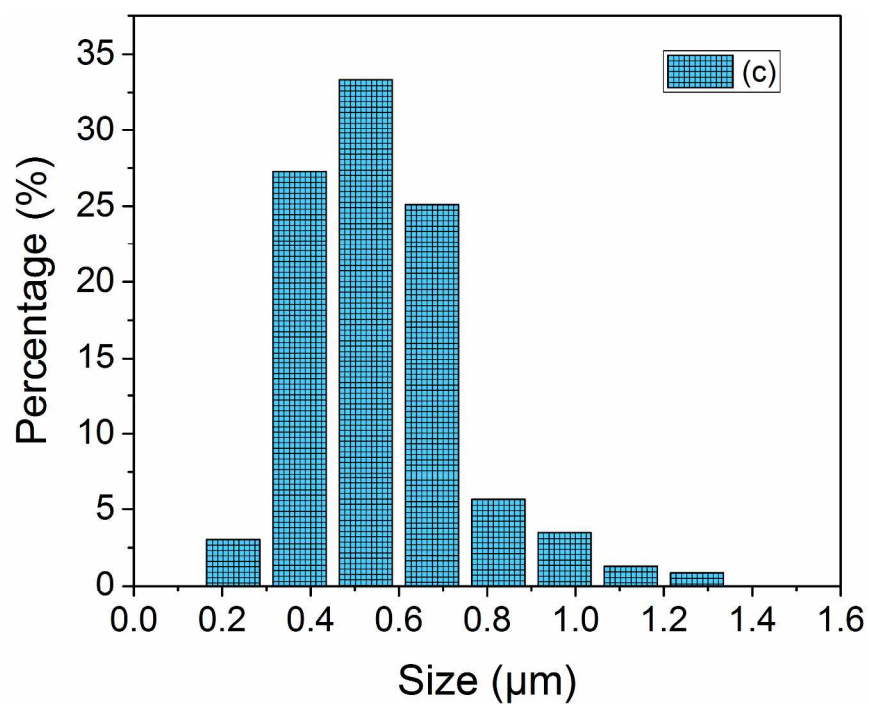


Figure S1 Grain size distribution counted from the SEM images of 0.9CaTiO₃-0.1BiScO₃ ceramics

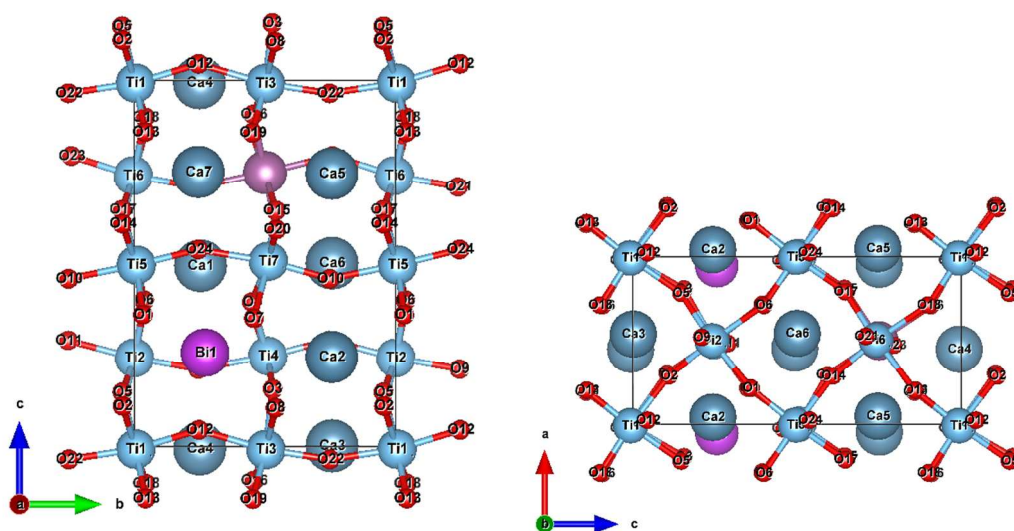


Figure S2 Relaxed structure of CaTiO₃-BiScO₃ lattice.

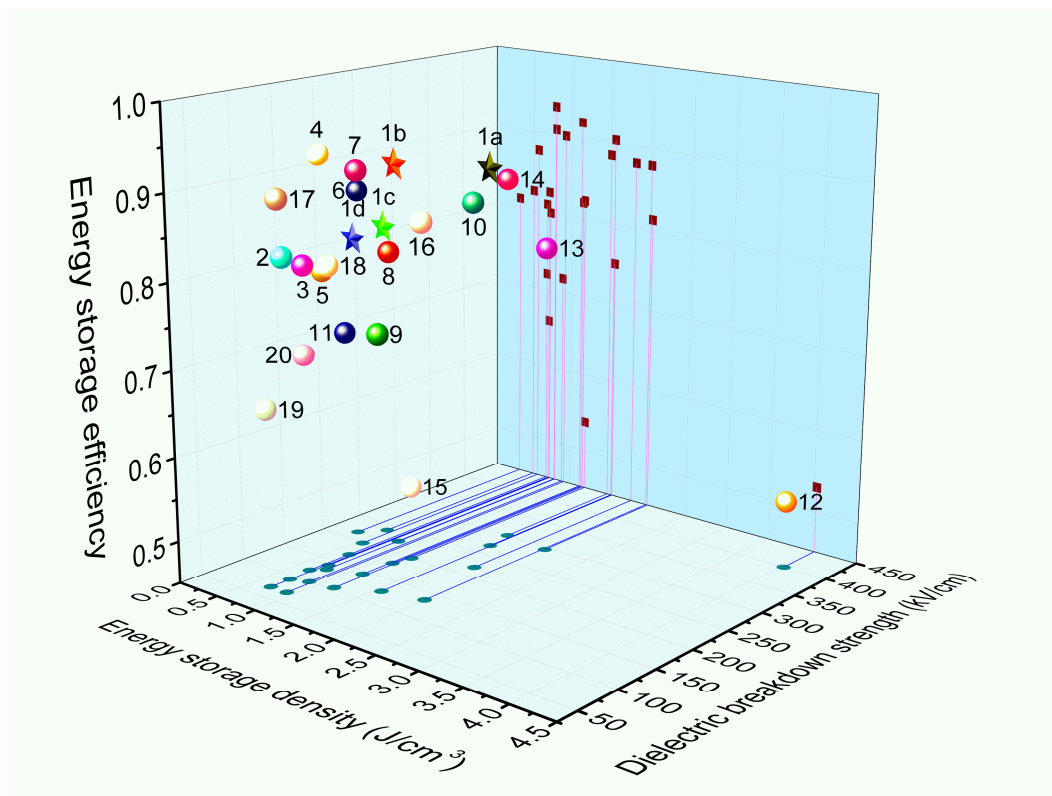


Figure S3 Comparison dielectric breakdown strength, energy storage density and energy storage efficiency of different lead-free dielectric ceramics.

Table S1 Specific results of dielectric breakdown strength, energy storage density and energy storage efficiency of different lead-free dielectric ceramics.

Number	Lead-free dielectric ceramics for energy storage	Energy storage efficiency	Dielectric breakdown strength (kV/cm)	Energy storage density (J/cm ³)
1a	0.9CaTiO ₃ -0.1BiScO ₃ , this work	0.904	270.0	1.55
1b	0.95CaTiO ₃ -0.05BiScO ₃ , this work	0.906	223.4	0.80
1c	0.8CaTiO ₃ -0.2BiScO ₃ , this work	0.825	237.7	0.500
1d	0.7CaTiO ₃ -0.3BiScO ₃ , this work	0.812	216.6	0.32
2	0.91BaTiO ₃ -0.09BiYbO ₃ ¹	0.826	93	0.71
3	Core-shell BaTiO ₃ @BiScO ₃ ²	0.81	120	0.68
4	0.85BaTiO ₃ -0.15Bi(Zn _{2/3} Nb _{1/3})O ₃ ³	0.935	131	0.79
5	BaTiO ₃ coated with Al ₂ O ₃ and SiO ₂ ⁴	0.8	139.2	0.725
6	Ba _{0.4} Sr _{0.6} TiO ₃ -based ceramics ⁵	0.877	198.8	0.564
7	0.85BaTiO ₃ -0.15Bi(Mg _{2/3} Nb _{1/3})O ₃ ⁶	0.92	143.5	1.13
8	(Ba _{0.4} Sr _{0.6})TiO ₃ ⁷	0.82	180	1.15
9	BiFeO ₃ -BaTiO ₃ -Ba(Mg _{1/3} Nb _{2/3})O ₃ ⁸	0.75	125	1.56
10	0.88BaTiO ₃ -0.12Bi(Mg _{2/3} Nb _{1/3})O ₃ ⁹	0.88	220	1.81
11	Ba(Zr _{0.2} Ti _{0.8})O ₃ -(Ba _{0.85} Ca _{0.15})TiO ₃ ¹⁰	0.72	170	0.68
12	0.8(K _{0.5} Na _{0.5})NbO ₃ -0.2Sr(Sc _{0.5} Nb _{0.5})O ₃ ¹¹	0.814	295	2.02
13	95 wt%Ba _{0.4} Sr _{0.6} TiO ₃ -5 wt% MgO composite ¹²	0.885	300	1.5
14	SiO ₂ -coated BaTiO ₃ ¹³	0.538	200	1.2
15	0.10Bi(Mg _{1/2} Ti _{1/2})O ₃ -0.90(Na _{1/2} Bi _{1/2})TiO ₃ ¹⁴	0.88	135	2
16	0.10Bi(Mg _{1/2} Ti _{1/2})O ₃ -0.90(Na _{1/2} Bi _{1/2})TiO ₃ ¹⁴	0.9	70	0.92
17	0.90(Na _{0.5} Bi _{0.5})TiO ₃ -0.10KNbO ₃ ¹⁵	0.823	104	1.17
18	0.9[(0.92Na _{0.5} Bi _{0.5})TiO ₃ -0.08BaTiO ₃] -0.1NaNbO ₃ ¹⁶	0.66	70	0.71
19	0.94(0.94(Na _{1/2} Bi _{1/2})TiO ₃ -0.06BaTiO ₃) -0.06KNbO ₃ ¹⁷	0.718	100	0.89

Table S2 Static dielectric tensor of CaTiO_3 and $\text{CaTiO}_3\text{-BiScO}_3$ from density functional perturbation theory.

Static dielectric tensor	CaTiO_3	$\text{CaTiO}_3\text{-BiScO}_3$
κ_{11}	6.92	142.7239
κ_{22}	37.78	95.47216
κ_{33}	6.85	71.08469
κ_{12}	0	-3.15454
κ_{13}	0	-17.5119
κ_{23}	0	12.48837

Table S3 Calculated Born effective charge tensors of CaTiO₃-BiScO₃ lattice in unit e. The different ions with the number 1~40 were corresponding to the different position shown in Figure S2

<i>Ions</i>		<i>X</i>	<i>Y</i>	<i>Z</i>	<i>Ions</i>		<i>X</i>	<i>Y</i>	<i>Z</i>
<i>O1</i>					<i>O11</i>				
	1	-3.870	0.334	1.636		1	-1.784	-0.155	-0.046
	2	0.306	-2.096	0.195		2	0.059	-5.539	-0.224
	3	1.466	0.258	-3.424		3	0.009	-0.222	-1.705
<i>O2</i>					<i>O12</i>				
	1	-3.570	-0.128	-1.688		1	-1.625	0.027	0.386
	2	-0.102	-1.902	-0.135		2	-0.071	-4.422	-0.427
	3	-1.340	-0.173	-3.098		3	0.081	-0.502	-3.090
<i>O3</i>					<i>O13</i>				
	1	-3.480	0.187	1.192		1	-3.401	-0.056	1.614
	2	0.265	-2.409	0.363		2	0.012	-1.716	-0.083
	3	0.918	0.448	-3.001		3	1.558	0.002	-3.595
<i>O4</i>					<i>O14</i>				
	1	-3.453	0.181	-1.089		1	-3.653	-0.042	-1.725
	2	0.021	-2.247	-0.366		2	0.070	-1.710	0.041
	3	-0.906	-0.533	-3.244		3	-1.634	-0.076	-3.589
<i>O5</i>					<i>O15</i>				
	1	-3.760	0.157	1.541		1	-3.021	-0.032	1.439
	2	0.084	-2.136	-0.216		2	-0.073	-1.884	-0.108
	3	1.345	-0.185	-3.043		3	1.391	-0.088	-3.567
<i>O6</i>					<i>O16</i>				
	1	-3.753	-0.063	-1.412		1	-3.520	-0.356	-1.313
	2	-0.052	-2.023	0.222		2	-0.370	-1.915	-0.101
	3	-1.343	0.376	-3.240		3	-1.332	-0.106	-2.930
<i>O7</i>					<i>O17</i>				
	1	-3.942	-0.838	1.075		1	-3.771	0.142	1.611
	2	-0.665	-2.229	-0.176		2	0.097	-1.723	0.032
	3	0.900	-0.223	-2.797		3	1.577	-0.025	-3.438
<i>O8</i>					<i>O18</i>				
	1	-3.264	-0.031	-1.566		1	-3.656	0.168	-1.535
	2	-0.016	-1.852	0.144		2	0.112	-1.684	-0.034
	3	-1.312	0.110	-3.151		3	-1.580	0.049	-3.321

<i>O9</i>	1	-2.637	-0.194	-0.189	<i>O19</i>	1	-3.112	0.114	1.360
	2	-0.285	-4.677	0.102		2	0.112	-1.954	-0.182
	3	-0.214	-0.105	-1.549		3	1.416	-0.128	-3.486
<i>O10</i>	1	-1.656	0.052	0.126	<i>O20</i>	1	-3.641	-0.154	-1.399
	2	0.092	-5.296	0.280		2	-0.112	-1.939	-0.131
	3	-0.025	0.249	-1.906		3	-1.392	-0.077	-3.066
<i>O21</i>	1	-1.948	-0.029	-0.179	<i>Ti7</i>	1	5.697	0.070	-0.123
	2	-0.090	-4.625	0.106		2	-0.063	6.337	-0.311
	3	0.006	0.059	-2.181		3	0.631	0.636	5.977
<i>O22</i>	1	-1.607	0.039	0.118	<i>Ca1</i>	1	2.647	-0.109	-0.238
	2	0.046	-5.529	0.163		2	0.044	2.233	-0.191
	3	0.039	0.003	-1.864		3	-0.101	-0.263	2.467
<i>O23</i>	1	-2.004	-0.099	-0.201	<i>Ca2</i>	1	2.503	-0.026	0.059
	2	-0.101	-4.436	0.302		2	-0.016	2.707	-0.121
	3	0.049	0.357	-2.469		3	0.069	0.055	2.108
<i>O24</i>	1	-1.721	-0.048	0.116	<i>Ca3</i>	1	2.352	-0.023	-0.147
	2	-0.157	-4.748	-0.292		2	-0.061	2.487	-0.180
	3	-0.036	-0.161	-2.182		3	-0.160	-0.054	2.673
<i>Ti1</i>	1	6.461	-0.025	-0.403	<i>Ca4</i>	1	2.744	0.042	-0.015
	2	0.238	6.295	0.858		2	0.016	2.383	-0.046
	3	0.396	-0.016	6.140		3	-0.166	0.062	2.349
<i>Ti2</i>	1	6.902	0.491	0.055	<i>Ca5</i>	1	2.312	-0.006	0.129
	2	0.192	6.572	-0.308		2	0.084	2.415	0.004
	3	-0.542	0.910	5.382		3	0.225	0.181	2.789
<i>Ti3</i>	1	5.871	0.113	-0.353	<i>Ca6</i>	1	2.618	0.137	-0.203
	2	0.018	6.261	-0.668		2	-0.016	2.599	0.253
	3	0.316	0.403	5.780		3	-0.141	0.253	2.463
<i>Ti4</i>	1	6.784	-0.179	0.128	<i>Ca7</i>	1	2.134	0.087	0.166
	2	-0.236	6.214	0.358		2	0.092	2.385	0.182
	3	-0.142	-0.422	4.801		3	0.262	-0.083	3.065
<i>Ti5</i>	1	6.524	0.081	-0.451	<i>Bi</i>	1	4.742	0.029	0.537

Ti6	2	0.293	6.294	0.470	Sc	2	0.138	4.572	0.268
	3	0.446	-0.704	6.431		3	0.322	-0.127	3.899
	1	6.611	0.203	0.487		1	4.934	-0.05384	0.50282
	2	0.281	5.854	-0.648		2	-0.186	5.05437	0.60782
	3	-0.494	0.468	6.451		3	-0.553	-0.6098	5.24396

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