Supporting Information

of

A green and sustainable technology for high-efficiency and low-damage manipulation of densely crosslinked proteins

Helan Xu,[†] Kaili Song,^{†,‡} Bingnan Mu,[†] Yiqi Yang*,^{†,l,§}

[†]Department of Textiles, Merchandising and Fashion Design, 234, HECO Building, University of Nebraska-Lincoln, Lincoln, NE 68583-0802, United States

[‡]Key Laboratory of Science and Technology of Eco-Textiles, Ministry of Education, Donghua University, Shanghai 201620, China

Department of Biological Systems Engineering, 234, HECO Building, University of Nebraska-Lincoln, Lincoln, NE 68583-0802,

United States

§Nebraska Center for Materials and Nanoscience, 234, HECO Building, University of Nebraska-Lincoln, Lincoln, NE 68583-0802,

United States

* Corresponding Author. Tel: +001 402 472 5197; Fax: +001 402 472 0640; E-mail: yyang2@unl.edu

Table S1 demonstrates the difference in potential hazards of active ingredients of the commercial product used in this research and those of our product. It could be seen that the chemicals used in the developed formula showed less potential harm comparing to those used in the commercial product.

Table S1. Comparison of active ingredients in commercial product and our formula

Chemical		Potential hazard	Source
Commercial product	Hydrogen	Eye: Produces irritation, characterized by a burning sensation, redness, tearing,	https://fscimage.fishersci
	peroxide	inflammation, and possible corneal injury. Vapors may cause eye irritation.	.com/msds/40069.htm
		Skin: May cause skin irritation. May cause skin discoloration. May cause	
		reddening of the skin.	
		Ingestion: May cause irritation of the digestive tract. May lead to distention of	
		the esophagus and stomach.	
		Inhalation: May cause respiratory tract irritation. Irritation may lead to chemical	
		pneumonitis and pulmonary edema.	1
	Ammonium	The material can produce chemical burns to the eye following direct contact.	http://datasheets.scbt.co
	tnioglycolate	Vapors or mists may be extremely irritating. If applied to the eyes, this material	m/sc-251234.pdf
		causes severe eye damage.	
		Inhalation of thioglycolic acid mists may cause olfactory paralysis, weakness, shortness of breath and liver damage.	
	Ammonium	The substance is corrosive to the eyes, the skin and the respiratory tract.	https://www.cdc.gov/nio
			sh/ipcsneng/neng0215.ht
	llydioxide		ml
		The effects may be delayed.	1111
		Lungs may be affected by repeated or prolonged exposure to the vapor or	
		aerosol.	
	Cysteine	Cause skin irritation	https://www.fishersci.co
Our product		Cause serious eye irritation	m/shop/msdsproxy?prod
		May cause respiratory irritation	uctName=C56225∏
			uctDescription=L+CYS
			TEINE+HCL+REAGE
			NT+25G&catNo=C562-
			25&vendorId=VN00033

		897&storeId=10652
Urea	The substance irritates the eyes, the skin and the respiratory tract.	https://www.cdc.gov/nio
	Repeated or prolonged contact with skin may cause dermatitis.	sh/ipcsneng/neng0595.ht
		ml
Citric acid	The substance is irritating to the eyes, the skin and the respiratory tract.	https://www.cdc.gov/nio
		sh/ipcsneng/neng0855.ht
		ml
Sodium	Decomposes at high temperatures (>248 °C) generating flammable and toxic	https://fscimage.fishersci
hypophosphite	phosphine gas. May cause eye, skin, and respiratory tract irritation.	.com/msds/21325.htm
		http://www.dcfinechemi
		cals.com/files2/Hojas%2
		0de%20seguridad%20(E
		N)/111780-SDS-EN.pdf