Visible-Light-Induced Synthesis of Carbazoles by in situ Formation of Photosensitizing Intermediate

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General Considerations

General reagent information

All reagents and solvents including 'BuONO and 1,2-Dichloroethane (DCE) were purchased from Sigma-Aldrich, Alfa Aesar, TCI and Combi Blocks chemical companies. Flash column chromatography was performed using Merck silica gel 60 (70-230 mesh).

General analytical information

The 2,2'-diamino-1,1'-biaryls and carbazole products were characterized by ¹H, ¹³C NMR, and FT-IR spectroscopy. NMR spectra were recorded on a Bruker 600 MHz instrument (600 MHz for ¹H NMR and 151 MHz for ¹³C NMR). Copies of ¹H and ¹³C NMR spectra can be found at the end of the Supporting Information. ¹H NMR experiments are reported in units, parts per million (ppm), and were measured relative to residual chloroform (7.26 ppm) or residual DMSO (2.5 ppm) in the deuterated solvent. ¹³C NMR spectra are reported in ppm relative to deuterochloroform (77.23 ppm) or deuterated DMSO (39.5 ppm), and all were obtained with ¹H decoupling. Coupling constants were reported in Hz. FT-IR spectra were recorded on a Nicolet iS 10 ThermoFisher FT-IR spectrometer. Reactions were monitored by ¹H-NMR of the crude reaction mixture using bromoform as internal standard and products were detected by GC-MS using the Agilent GC 7890B/5977A inert MSD with Triple-Axis Detector. Mass spectral data of all unknown compounds were obtained from the Korea Basic Science Institute (Daegu) on a Jeol JMS 700 high resolution mass spectrometer. UV–Vis spectra were recorded on a Scinco S-3100 spectrophotometer equipped with a Peltier temperature controller. Melting points of unknown carbazole compounds were recorded on a Stuart SMP30 apparatus.

Optimization Table of Reaction Conditions

Table S1. Optimization of Reaction Conditions^a



entrv	nitrite source	solvent	variation	Yield (%) ^b	
	(equiv.)	(concentration)	Vanadon	2a	S1
1	^t BuONO (1.5)	MeCN (0.1 M)		47	14
2	^t BuONO (1.5)	DMF (0.1 M)		30	7
3	^t BuONO (1.5)	DMSO (0.1 M)			trace
4	^t BuONO (1.5)	MeOH (0.1 M)			trace
5	^t BuONO (1.5)	TFE (0.1 M)		48	20
6	^t BuONO (1.5)	DCM (0.1 M)		55	8
7	^t BuONO (1.5)	DCE (0.1 M)		68	11
8	^t BuONO (1.5)	DCE (0.1 M)	aerobic atmosphere	38	6
9	^t BuONO (1.5)	DCE (0.1 M)	no O2 ^c		12
10	^t BuONO (1.5)	DCE (0.1 M)	no visible light	trace	10
11	^t BuONO (1.5)	DCE (0.1 M)	using UV-cut @ 400 nm	52	9
12	^t BuONO (1.5)	DCE (0.1 M)	blue LEDs (21 W)	50	11
13	^t BuONO (1.5)	DCE (0.1 M)	white LEDs (21 W)	60	9
14	^t BuONO (1.5)	DCE (0.1 M)	CFL (15 W)	65	12
15	^t BuONO (1.5)	DCE (0.1 M)	CFL (100 W)	68	10
16	isoamyl nitrite (1.5)	DCE (0.1 M)		60	15
17	^t BuONO (2)	DCE (0.1 M)		66	12
18	^t BuONO (1.3)	DCE (0.1 M)		68	10
19	^t BuONO (1.1)	DCE (0.1 M)		56	8
20	^t BuONO (1.3)	DCE (0.2 M)		59	9
21	^t BuONO (1.3)	DCE (0.5 M)		43	8
22	^t BuONO (1.3)	DCE (0.05 M)		51	5
23	^t BuONO (1.3)	DCE (0.1 M)	20 h	66	10
24	^t BuONO (1.3)	DCE (0.1 M)	30 h	68	11
25	^t BuONO (1.3)	DCE (0.1 M)	TEA (1 equiv.)		
26	^t BuONO (1.3)	DCE (0.1 M)	lr(ppy) ₃ (1 mol%)	61	12
27	^t BuONO (1.3)	DCE (0.1 M)	lr(dFppy) ₃ (1 mol%)	65	9
28	^t BuONO (1.3)	DCE (0.1 M)	Ru(bpy) ₃ (1 mol%)	62	10
29	^t BuONO (1.3)	DCE (0.1 M)	Eosin Y (1 mol%)	60	15

^{*a*}Reaction scale: 0.1 mmol; ^{*b*}Yields were determined by ¹H NMR of crude reaction mixture using bromoform as the internal standard; ^{*c*}Molecular oxygen was effectively removed from the reaction mixture by repeated vacuum-freeze-thaw cycles.

Density Functional Theory (DFT) Calculation Details

All the molecules at ground state are optimized using hybrid B3LYP (Becke, three-parameter, Lee-Yang-Parr) exchange-correlation density functional theory and 6-311++G(d,p) basis. All calculations were performed using GAUSSIAN09 software, and the solvent effect was considered by the polarizable continuum model (IEF-PCM) implemented in GAUSSIAN09 package (with dichloromethane as a solvent). The absence of vibrational normal modes with imaginary frequencies was checked to verify the quality of minimization procedure. With this optimized

geometry at the ground state, a calculation using the time-dependent density functional theory (TD-DFT) with B3LYP functional was performed for the excitation energies (the number of calculated excited states set to 20) with the same level of basis set, 6-311++G(d,p), with IEF-PCM. To verify that our calculations are reasonable without any model dependence, we compare the TD-DFT results obtained from two different functionals, B3LYP/6-311++G(d,p) and PBE0/6-311+G(2d,p). These two functionals were known to provide reasonable spectroscopy data for organic dyes. In the figure below (Figure *S1*), we show the comparison for intermediates **I**, **II** and **III**. The agreement between them is fairly good and, therefore, we present only the results from the B3LYP functional in the main text and below in Figures *S2–S4*.



Comparison of TD-DFT results from different functionals, B3LYP and PBE0

Figures *S1*. Comparison of TD-DFT results from different functionals, B3LYP/6-311++G(d,p) and PBE0/6-311+G(2d,p) where (a), (b) and (c) represents Intermediate-II, Intermediate-III and Intermediate-III respectively. Results from the B3LYP and PBE0 functionals are shown in solid and dashed line respectively.

Wavelength	Osc. Strength (<i>f</i>)	Major Contributions	
437.82	0.0760	H-1 → L (99%)	
340.00	0.0430	$\textbf{H} \not \rightarrow \textbf{L+2} \ (98\%)$	
273.89	0.0469	H → L+3 (61%)	H-4 → L (23%)
272.13	0.0619	H-4 → L (58%)	H-3 → L (32%)
267.94	0.0985	H-3 → L (30%)	$H \not\rightarrow L+3 \ (28\%)$
255.14	0.0416	$\textbf{H} \not \rightarrow \textbf{L}{+}4~(77\%)$	H-3 → L (9%)
249.15	0.2385	H-1 → L+2 (74%)	
228.25	0.2609	$\textbf{H} \not \rightarrow \textbf{L+5} \ (49\%)$	$\textbf{H} \not \rightarrow \textbf{L+6} \ (27\%)$

Table S2. Selected TD-DFT [B3LYP/6-311++G(d,p)] calculated wavelength, oscillatorstrength and compositions of major electronic transitions of Intermediate-I.



Figure S2. Calculated [B3LYP/6-311++G(d,p)] absorption spectrum and structure of Intermediate-I

Wavelength	Osc. Strength (<i>f</i>)	Major Contributions	
328.81	0.0421	H-1 → L (73%)	H-2 → L (10%)
289.97	0.1929	$\textbf{H} \not \rightarrow \textbf{L+1} \ (50\%)$	H-2 → L (34%)
251.36	0.1747	$\textbf{H} \not \rightarrow \textbf{L+2} (30\%)$	H-3 → L (30%)
250.55	0.4347	H-1 → L+1 (39%)	$\textbf{H} \not \rightarrow \textbf{L+2} \ (26\%)$
233.92	0.1089	H-4 → L (33%)	H-2 → L+1 (32%)
228.54	0.0967	H-1 → L+2 (50%)	H-5 → L (13%)
211.33	0.1063	H-1 → L+3 (41%)	H-5 → L (14%)
207.95	0.1022	H-2 → L+2 (19%)	H-3 → L+1 (19%)
		H-1 → L+3 (17%)	

Table *S3*. Selected TD-DFT [B3LYP/6-311++G(d,p)] calculated wavelength, oscillator strength and compositions of major electronic transitions of Intermediate-**II**.



Figure *S***3.** Calculated [B3LYP/6-311++G(d,p)] absorption spectrum and structure of Intermediate-II.

Wavelength	Osc. Strength (f)	Major Contributions	
376.03	0.2562	$H \rightarrow L (93\%)$	
316.30	0.1183	H-1 \rightarrow L (83%)	$H \rightarrow L+1 \ (12\%)$
266.98	0.1488	H-2 → L (62%)	$H \rightarrow L+2 (31\%)$
258.45	0.7252	H-1 → L+1 (79%)	
238.47	0.1155	$\textbf{H} \not \rightarrow \textbf{L}{+}4~(70\%)$	H-2 → L+1 (20%)
230.69	0.2258	H-1 → L+2 (53%)	H-4 → L (35%)

Table S4. Selected TD-DFT [B3LYP/6-311++G(d,p)] calculated wavelength, oscillatorstrength and compositions of major electronic transitions of Intermediate-III.



Figure *S***4.** Calculated [B3LYP/6-311++G(d,p)] absorption spectrum and structure of Intermediate-III.

Experimental Details

1. Preparation of 2,2'-Diaminobiaryls, 1a-1n

For the synthesis of **1a-1n**, we followed a slightly modified synthetic procedure that was reported by the Ritter group for the synthesis of 2-(2-pyridinyl)aniline.¹

Representative procedure; Synthesis of 2,2'-Diaminobiphenyl 1a



An oven-dried 250 mL round bottom flask equipped with a magnetic stir bar was charged with 2-iodoaniline (5 mmol, 1.10 g) and 2-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)aniline (5 mmol, 1.09 g) in ethylene glycol dimethyl ether (25 mL) mixed with distilled water (25 mL). 5 mol% Pd(PPh₃)₄ (0.25 mmol, 0.29 g) and K₂CO₃ (20 mmol, 2.76 g) were added to it. The round bottom flask connected with a reflux column was placed in the oil bath and heated for 16 h at 120 °C. The progress of the reaction was monitored by TLC and gas chromatography. The flask was cooled to room temperature, the reaction mixture was diluted with dichloromethane, and the aqueous phase was extracted with dichloromethane. The combined organic layers were dried over anhydrous MgSO₄, and concentrated *in vacuo*. The residue was purified by silica gel flash column chromatography using dichloromethane as the eluent to give the desired product **1a** (0.69 g, 75%).

Other substrates **1b–1n** were synthesized similarly to the synthesis of **1a** by Suzuki-Miyaura coupling process using the corresponding substituted 2-haloanilines and 2-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)aniline.

2. Preparation of 2,2'-Diaminobiaryl² 5a–5d



Representative procedure; Synthesis of 1-(2-Aminophenyl)naphthalen-2-amine 5a

To an oven-dried two neck 100 mL round bottom flask equipped with a magnetic stir bar and a reflux condenser were charged tert-butyl 1-(naphthalen-2-yl)hydrazine-1carboxylate (3.87 mmol, 1.00 g), iodobenzene (3.2 mmol, 0.65 g), 5 mol% Pd(OAc)₂ (0.19 mmol, 0.04 g), 5 mol% P(^tBu)₃·HBF₄ (0.19 mmol, 0.06 g), Cs₂CO₃ (4.64 mmol, 1.51 g) and 30 mL of anhydrous toluene at rt. The resulting mixture was flushed with Ar several times and heated under reflux for 1.5 h. The progress of the reaction was monitored by TLC. The reaction mixture was cooled to room temperature and partitioned into dichloromethane and water. The combined organic layers were dried over anhydrous MgSO₄, and concentrated *in vacuo*. The residue was purified by silica gel flash column chromatography (hexane/EtOAc = 30/1) to give *tert*-butyl 1-(naphthalen-2-yl)-2phenylhydrazine-1-carboxylate (2.18 mmol, 0.73 g) in 68% yield. To a flask charged with the resulting hydrazide product were added 30 mL of ethanol and a few drops of conc. HCl (35%) at rt. The reaction mixture was then heated under reflux for 1 h. The progress of the reaction was monitored by TLC. The reaction mixture was cooled to rt and neutralized by adding $NaHCO_3(s)$. The mixture was then filtered, concentrated and purified by silica-gel chromatography (2/1 hexane/CH₂Cl₂ \rightarrow 9/1 EtOAc/MeOH) to afford 1-(2-aminophenyl)naphthalen-2-amine 5a (1.68 mmol, 0.39 g) in 77% yield.

3. Synthesis of Carbazole Compounds 2a-2n, 6a-6d, and 8



3.1. Representative procedure; Synthesis of N–H carbazole 2a

A resealable tube equipped with a magnetic stir bar was charged with [1,1'-biphenyl]-2,2'-diamine **1a** (0.3 mmol, 55 mg), in 1,2-dichloroethane (DCE) (3 mL, 0.1 M). Then 'BuONO [0.4 mmol, 52 μ L; considering the purity of 'BuONO (90% purity)] was added to the solution and stirred for 10 min. Oxygen gas was then bubbled through the reaction mixture for 10 minutes and the tube was sealed with a silicone septum screw cap. The tube was then placed under oxygen atmosphere (molecular oxygen balloon) and visiblelight irradiation of CFL (24 W) at room temperature. After 24 h, the solvent was evaporated and the reaction mixture was purified by flash column chromatography to furnish the pure *N*–H carbazole product **2a** (34 mg, 68%).

3.2. Representative procedure in 1 mmol scale; Synthesis of 3-methyl-9*H*-carbazole 2j

A resealable tube equipped with a magnetic stir bar was charged with 5-methyl-2,2'diaminebiphenyl **1j** (1 mmol, 198 mg), in 1,2-dichloroethane (DCE) (10 mL, 0.1 M). Then 'BuONO [1.4 mmol, 172 μ L; considering the purity of 'BuONO (90% purity)] was added to the solution and stirred for 15 min. Oxygen gas was then bubbled through the reaction mixture for 15 minutes and the tube was sealed with a silicone septum screw cap. The tube was then placed under oxygen atmosphere (molecular oxygen balloon) and visible-light irradiation of CFL (24 W) at room temperature. After 24 h, the solvent was evaporated and the reaction mixture was purified by flash column chromatography to furnish the pure 3-methyl-9*H*-carbazole **2j** (99 mg, 55%).

Analytic data for 2,2'-Diaminobiaryl Compounds



1a (2,2'-Diaminebiphenyl; CAS 1454-80-4)³: brown solid (0.69 g, 75%); ¹**H NMR (600 MHz, CDCl₃)** δ 7.20 (ddd, J = 8.4, 7.0, 1.8 Hz, 2H), 7.11 (dd, J = 7.8, 1.8 Hz, 2H), 6.82 (ddd, J = 7.8, 7.0, 1.2 Hz, 2H), 6.76 (dd, J = 8.4, 1.2 Hz, 2H), 3.69 (s, 4H); ¹³**C NMR (151 MHz, CDCl₃)** δ 144.1, 131.0, 128.8, 124.6,

118.8, 115.6; **IR** (**neat**): $v_{\text{max}} = 3436$, 3348, 3020, 1609, 1482, 1442, 1295, 745 cm⁻¹; $R_f 0.40$ (dichloromethane).

1b (3-Bromo-2,2'-diaminebiphenyl, CAS 306292-43-3)⁴: off-white solid (0.50 g, 38%); ¹H NMR (600 MHz, CDCl₃) δ 7.37 (dd, J = 8.4, 1.8 Hz, 1H), 7.10 (ddd, J = 8.4, 7.8, 1.8 Hz, 1H), 6.99 (dd, J = 7.8, 1.8 Hz, 1H), 6.96 (dd, J = 7.2, 1.2 Hz, 1H), 6.74 (ddd, J = 8.4, 7.2, 1.2 Hz, 1H), 6.63 (dd, J = 7.8, 1.8 Hz, 1H),

Br HZ, 1H), 6.74 (ddd, J = 8.4, 7.2, 1.2 HZ, 1H), 6.05 (dd, J = 7.8, 1.8 HZ, 1H), 6.59 (dd, J = 8.4, 7.8 HZ, 1H), 4.08 (bs, 2H), 3.54 (bs, 2H); ¹³C NMR (151 MHz, CDCl₃) δ 144.0, 142.0, 132.0, 130.8, 130.1, 129.2, 125.5, 123.8, 119.0, 118.7, 115.6, 109.7 ; **IR** (neat): v_{max} = 3459, 3368, 3023, 1738, 1610, 1443, 904, 725 cm⁻¹; **R**_f 0.65 (dichloromethane).



1c (5-Chloro-2,2'-diaminebiphenyl): pale red solid (0.60 g, 55%); ¹H NMR (600 MHz, CDCl₃) δ 7.20 (dd, *J* = 7.8, 6.8, 1.8 Hz, 1H), 7.14 (dd, *J* = 8.4, 2.4 Hz, 1H), 7.11 (d, *J* = 2.4 Hz, 1H), 7.09 (dd, *J* = 7.8, 1.8 Hz, 1H), 6.83 (ddd, *J* = 7.8, 6.8, 1.2 Hz, 1H), 6.78 (dd, *J* = 7.8, 1.2 Hz, 1H), 6.70 (d, *J* =

8.4 Hz, 1H), 3.68 (bs, 4H); ¹³C NMR (151 MHz, CDCl₃) δ 144.1, 143.0, 131.1, 130.8, 129.4, 128.8, 126.2, 123.4, 123.3, 119.1, 116.8, 115.9; **IR** (neat): $v_{max} = 3441$, 3359, 1613, 1485, 1295, 907, 730 cm⁻¹; **MS** m/z (EI) calc. for C₁₂H₁₁ClN₂ [M⁺] 218.0605, found 218.1; **R**_f 0.50 (dichloromethane).



1d (5-Fluoro-2,2'-diaminebiphenyl): brown solid (0.82 g, 80%); ¹**H** NMR (600 MHz, CDCl₃) δ 7.20 (dd, J = 8.4, 7.8 1.8 Hz, 1H), 7.10 (dd, J = 7.2, 1.2 Hz, 1H), 6.92 – 6.85 (m, 2H), 6.84 (ddd, J = 7.8, 7.2, 1.2 Hz, 1H), 6.79 (dd, J = 8.4, 0.6 Hz, 1H), 6.72 (dd, J = 9.0, 4.8 Hz, 1H), 3.73 (bs, 2H), 3.60

(bs, 2H); ¹³C NMR (151 MHz, CDCl₃) δ 156.4 (d, J = 237.1 Hz), 144.2, 140.4, 131.1, 129.4, 126.0 (d, J = 7.6 Hz), 123.8, 119.1, 117.5 (d, J = 22.7 Hz), 116.7 (d, J = 9.1 Hz), 116.0, 115.5 (d, J = 22.2 Hz); **IR (neat**): $v_{max} = 3353$, 1739, 1490, 1365, 751 cm⁻¹; **MS** m/z (EI) calc. for C₁₂H₁₁FN₂ [M⁺] 202.0906, found 202.1; **R**_f 0.33 (dichloromethane).

F₃CO

1e (5-Trifluoromethoxy-2,2'-diaminebiphenyl): brown solid (0.81 g, 60%); ¹**H** NMR (600 MHz, CDCl₃) δ 7.15 (dd, J = 7.8, 6.6 Hz, 1H), 7.05 (d, J = 7.8 Hz, 1H), 7.01 (d, J = 9.0 Hz, 1H), 6.99 (s, 1H), 6.79 (dd, J = 7.8, 6.6 Hz, 1H), 6.70 (d, J = 7.8 Hz, 1H), 6.65 (d, J = 9.0 Hz, 1H),

3.68 (bs, 4H); ¹³C NMR (151 MHz, CDCl₃) δ 144.1, 143.2, 141.2, 131.0, 129.4, 125.4, 124.0, 123.3, 121.9, 120.8 (q, J = 255.8 Hz), 119.0, 116.1, 115.9; **IR** (neat): $v_{max} = 1636$, 1457, 1279, 1151, 817, 750 cm⁻¹; **MS** m/z (EI) calc. for C₁₃H₁₁F₃N₂O [M⁺] 268.0818, found 268.1; R_f 0.82 (dichloromethane).



1f (1-(2',6-Diamino-[1,1'-biphenyl]-3-yl)ethan-1-one): brown solid (0.91 g, 80%); ¹**H** NMR (600 MHz, CDCl₃) δ 7.77 (dd, J = 8.4, 2.0 Hz, 1H), 7.74 (d, J = 2.0 Hz, 1H), 7.18 (ddd, J = 7.8, 7.8, 1.2 Hz, 1H), 7.08 (dd, J = 7.2, 1.2 Hz, 1H), 6.82 (ddd, J = 7.8, 7.2, 1.8 Hz, 1H), 6.77

(d, J = 7.8 Hz, 1H), 6.73 (d, J = 8.4 Hz, 1H), 4.29 (bs, 2H), 3.68 (bs, 2H), 2.48 (s, 3H); ¹³C NMR (151 MHz, CDCl₃) δ 196.6, 149.2, 144.2, 132.6, 131.1, 130.0, 129.4, 127.9, 123.2, 119.0, 115.8, 114.5 (overlapped peaks are present), 26.2; **IR** (neat): $v_{max} = 3353$, 1659, 1612, 1293, 1231, 904, 723 cm⁻¹; MS m/z (EI) calc. for C₁₄H₁₄N₂O₂ [M⁺] 226.1101, found 226.1; R_f 0.23 (dichloromethane).



1g (Methyl 2',6-diamino-[1,1'-biphenyl]-3-carboxylate): brown solid (0.95 g, 78%); ¹H NMR (600 MHz, CDCl₃) δ 7.85 (dd, J = 8.4, 1.8 Hz, 1H), 7.82 (d, J = 1.8 Hz, 1H), 7.18 (dd, J = 7.8, 7.8 Hz, 1H), 7.08 (d, J = 7.8 Hz, 1H), 6.82 (dd, J = 7.8, 7.8 Hz, 1H), 6.76 (d, J = 7.8 Hz, 1H)

1H), 6.71 (d, J = 8.4 Hz, 1H), 4.19 (bs, 2H), 3.83 (s, 3H), 3.66 (bs, 2H); ¹³C NMR (151 MHz, CDCl₃) δ 167.3, 148.9, 144.2, 133.2, 131.1, 130.9, 129.3, 123.4, 123.3, 119.8, 118.9, 115.8, 114.5, 51.7; **IR** (neat): $v_{max} = 3366$, 1698, 1615, 1307, 1235, 904, 723 cm⁻¹; **MS** m/z (EI) calc. for C₁₄H₁₄N₂O₂ [M⁺] 242.1050, found 242.1; **R**_f 0.33 (dichloromethane).



1h (2',6-Diamino-[1,1'-biphenyl]-3-carbonitrile): gray solid (0.73 g, 69%);
¹H NMR (600 MHz, CDCl₃) δ 7.44 (dd, J = 7.6, 1.8 Hz, 1H), 7.41 (d, J = 1.8 Hz, 1H), 7.22 (ddd, J = 8.4, 7.6, 1.2 Hz, 1H), 7.06 (dd, J = 7.8, 1.8 Hz, 1H), 6.85 (ddd, J = 7.8, 7.6, 1.2 Hz, 1H), 6.80 (dd, J = 8.4, 1.2 Hz, 1H),

6.76 (d, J = 7.6 Hz, 1H), 4.22 (bs, 2H), 3.64 (bs, 2H); ¹³C NMR (151 MHz, CDCl₃) δ 148.5,

144.1, 135.5, 133.2, 131.1, 130.0, 124.6, 122.0, 120.1, 119.4, 116.1, 115.2, 100.9; **IR** (neat): v_{max} = 3362, 2215, 1616, 1490, 669 cm⁻¹; **MS** m/z (EI) calc. for C₁₃H₁₁N₃ [M⁺] 209.0953, found 209.1; **R**_f 0.30 (dichloromethane).



1j (5-Methyl-2,2'-diaminebiphenyl): brown solid (0.40 g, 40%) ¹**H NMR** (**600 MHz, CDCl**₃) δ 7.18 (dd, *J* = 7.8, 7.8 Hz, 1H), 7.13 (d, *J* = 7.8 Hz, 1H), 7.00 (d, *J* = 7.8 Hz, 1H), 6.96 (s, 1H), 6.84 (dd, *J* = 7.8, 7.8 Hz, 1H), 6.78 (d, *J* = 7.8 Hz, 1H), 6.71 (d, *J* = 7.8 Hz, 1H), 3.66 (bs, 4H), 2.28 (s,

3H); ¹³C NMR (151 MHz, CDCl₃) δ 144.3, 141.7, 131.7, 131.2, 129.5, 128.9, 128.2, 125.0, 124.9, 118.9, 115.9, 115.7, 20.6; **IR** (neat): $v_{max} = 3434$, 3349, 1613, 1491, 1297, 751 cm⁻¹; **MS** m/z (EI) calc. for C₁₃H₁₄N₂ [M⁺] 198.1157, found 198.1; R_f 0.40 (dichloromethane).



1k (4-Methoxy-2,2'-diaminebiphenyl, CAS 54147-86-3)⁵: brown solid (0.27 g, 25%); ¹H NMR (600 MHz, CDCl₃) δ 7.17 (ddd, J = 8.2, 7.8, 1.8 Hz, 1H), 7.11 (dd, J = 7.8, 1.8 Hz, 1H), 7.04 (d, J = 8.0 Hz, 1H), 6.82 (ddd, J = 7.8, 7.8, 1.2 Hz, 1H), 6.78 (dd, J = 8.2, 1.2 Hz, 1H), 6.42 (dd, J

= 8.0, 2.4 Hz, 1H), 6.35 (d, J = 2.4 Hz, 1H), 3.81 (s, 3H), 3.72 (bs, 4H); ¹³C NMR (151 MHz, CDCl₃) δ 160.5, 145.5, 144.6, 132.0, 131.6, 128.8, 124.5, 118.9, 117.6, 115.6, 104.6, 101.2, 55.4; IR (neat): v_{max} = 3445, 3358, 1610, 1487, 1296, 1205, 752 cm⁻¹; R_f 0.35 (dichloromethane).



11 (5-Chloro-5'-methoxy-2,2'-diaminebiphenyl): gray solid (0.43 g, 37%); **¹H NMR (600 MHz, CDCl₃)** δ 7.13 (dd, J = 8.4, 2.4 Hz, 1H), 7.11 (d, J = 2.4 Hz, 1H), 7.01 (dd, J = 7.8, 1.8 Hz, 1H), 6.91 (d, J = 1.8 Hz, 1H), 6.70 (d, J = 7.8 Hz, 1H), 6.69 (d, J = 8.4 Hz, 1H), 3.66 (bs, 4H), 2.28 (s, 3H); **¹³C NMR (151 MHz, CDCl₃)** δ 143.0, 141.5, 131.4, 130.7, 130.0, 128.6,

128.4, 126.5, 123.6, 123.2, 116.8, 116.1, 20.6; **IR** (neat): $v_{max} = 3435$, 3352, 1616, 1487, 1285, 813 cm⁻¹; **MS** m/z (EI) calc. for C₁₃H1₃ClN₂ [M⁺] 232.0762, found 232.1; **R**_f 0.55 (dichloromethane).



1m (4,4'-Dimethoxy-2,2'-diaminebiphenyl): brown solid (0.43 g, 35%); ¹H NMR (600 MHz, CDCl₃) δ 7.01 (d, J = 8.0 Hz, 2H), 6.41 (dd, J = 8.0, 2.6 Hz, 2H), 6.33 (d, J = 2.6 Hz, 2H), 3.80 (s, 6H), 3.75 (bs, 4H); ¹³C NMR (151 MHz, CDCl₃) δ 160.2, 145.7,

132.2, 117.2, 104.3, 101.0, 55.2; **IR (neat)**: $v_{\text{max}} = 3367$, 1614, 1497, 1205, 904, 723 cm⁻¹; **MS** m/z (EI) calc. for C₁₄H₁₆N₂O₂ [M⁺] 244.1212, found 244.1; *R*_f 0.30 (dichloromethane).



1n (2-(2-Aminophenyl)pyridin-3-amine, CAS 144190-32-9)⁶: brown oil (0.50 g, 54%); ¹**H NMR (600 MHz, CDCl**₃) δ 8.07 (dd, J = 4.2, 1.8 Hz, 1H), 7.29 (dd, J = 7.8, 1.8 Hz, 1H), 7.15 (ddd, J = 7.8, 7.2, 1.2 Hz, 1H), 7.04 – 7.00 (m, 2H), 6.80 (ddd, J = 7.8, 7.2, 1.2 Hz, 1H), 6.76 (dd, J = 8.4, 1.2 Hz, 1H), 3.89 (bs,

4H); ¹³C NMR (151 MHz, CDCl₃) δ 144.9, 143.7, 140.8, 139.6, 129.9, 129.6, 123.2, 123.15, 123.12, 118.6, 116.9; **IR** (neat): $v_{max} = 3324$, 3191, 1611, 1441, 1302, 905, 724 cm⁻¹; R_f 0.13 (dichloromethane).



5a (1-(2-Aminophenyl)naphthalen-2-amine, CAS 93013-27-5)⁷: gray solid (0.39 g, 77%); ¹H NMR (400 MHz, CDCl₃) δ 7.72 – 7.67 (m, 2H), 7.27 – 7.19 (m, 4H), 7.10 (dd, J = 1.6, 1.2 Hz, 1H), 7.03 (d, J = 8.8 Hz, 1H), 6.90 - 6.84 (m, 2H), 3.64 (bs, 4H); ¹³C NMR (100 MHz, CDCl₃) δ 144.8, 141.7,

133.3, 131.7, 130.7, 129.0, 128.9, 128.0, 127.9, 126.5, 123.8, 122.2, 121.2, 118.9, 118.1, 115.6, 115.3; **IR** (**CHCl**₃): $v_{max} = 3381$, 3052, 1620, 1497, 1392, 1291, 1150 cm⁻¹; R_f 0.31 (dichloromethane).



5b (1-(2-Amino-5-(tert-butyl)phenyl)naphthalen-2-amine, CAS 756822-75-0)⁷: gray solid (0.51 g, 80%); ¹H NMR (**400 MHz, CDCl**₃) δ 7.71 (d, *J* = 7.6 Hz, 1H), 7.69 (d, *J* = 8.8 Hz, 1H), 7.30 – 7.19 (m, 4H), 6.81 (d, *J* = 8.4 Hz, 1H), 3.62 (bs, 4H); ¹³C NMR (**100 MHz, CDCl**₃) δ 142.1, 141.8, 141.7, 133.4, 128.9, 128.5, 128.0, 127.9, 126.4, 125.7, 124.0, 122.1, 120.9, 118.1,

116.0, 115.4, 34.2, 31.7; **IR** (**CHCl**₃): $v_{max} = 3489$, 3369, 3203, 3052, 2961, 2868, 1620, 1501, 1285, 1258, 1210 cm⁻¹.



5c (Methyl 4-amino-3-(2-aminonaphthalen-1-yl)benzoate, CAS 756822-77-2)⁷: off-white solid (0.50 g, 79%); ¹H NMR (400 MHz, CDCl₃) δ 7.94 (dd, J = 2.0, 2.0 Hz, 1H), 7.84 (d, J = 2.0 Hz, 1H), 7.73 – 7.70 (m, 2H), 7.30 – 7.20 (m, 3H), 7.03 (d, J = 8.8 Hz, 1H), 6.83 (d, J = 8.4 Hz, 1H), 3.96 (bs, 2H), 3.82 (bs, 2H); ¹³C NMR (100 MHz, CDCl₃) 166.9, 149.3, 141.9, 134.0, 133.1, 131.2, 129.5, 128.00, 127.97, 126.7, 123.5, 122.3, 120.14, 120.12, 118.1, 114.4, 113.8, 51.7; **IR** (**CHCl**₃): $v_{max} = 3472$, 3368, 3213, 3054, 1703, 1617, 1304, 1254, 1147 cm⁻¹; \mathbf{R}_f 0.21 (dichloromethane).



5d (1-(4-Amino-[1,1'-biphenyl]-3-yl)naphthalen-2-amine, CAS 756822-76-1)⁷: gray solid (0.47 g, 69%); ¹**H NMR (400 MHz, CDCl**₃) δ 7.74 – 7.70 (m, 2H), 7.56 – 7.52 (m, 3H), 7.41 (d, *J* = 2.4 Hz, 1H), 7.37 – 7.33 (m, 3H), 7.31 – 7.26 (m, 1H), 7.25 – 7.20 (m, 2H), 7.05 (d, *J* = 8.8 Hz, 1H), 6.93 (d, *J* = 8.0 Hz, 1H), 3.83 (bs, 2H), 3.60 (bs, 2H); ¹³C NMR (100 MHz, CDCl₃) 114.3,

141.8, 140.6, 133.3, 131.6, 130.2, 129.1, 128.5, 128.0, 127.9, 127.5, 126.6, 126.14, 126.10, 123.8, 122.2, 121.6, 118.1, 116.0, 115.1; **IR** (**CHCl**₃) $v_{max} = 3465$, 3374, 3201, 3053, 1615, 1512, 1484, 1303, 1263, 1147 cm⁻¹.

Analytic data for Carbazole Compounds



2a (*N*-H Carbazole, CAS 86-74-8)⁸: white solid (34 mg, 68%); ¹H NMR (600 MHz, CDCl₃) δ 8.09 (d, *J* = 7.8 Hz, 2H), 8.04 (bs, 1H), 7.44 - 7.41 (m, 4H), 7.25 (ddd, *J* = 8.4, 6.0, 2.4 Hz, 2H); ¹³C NMR (151 MHz, CDCl₃) δ 139.7,

126.0, 123.6, 120.5, 119.7, 110.8; **IR** (**neat**): $v_{max} = 3418$, 1450, 1335, 748, 724 cm⁻¹; *R*_f 0.50 (hexane/EtOAc, 4/1).



2b (1-Bromo-9*H*-carbazole, CAS 16807-11-7)⁹: off-white solid (45 mg, 61%); ¹H NMR (600 MHz, CDCl₃) δ 8.25 (bs, 1H), 8.06 (d, *J* = 7.8 Hz, 1H), 8.02 (d, *J* = 7.8 Hz, 1H), 7.57 (d, *J* = 7.8 Hz, 1H), 7.51 (d, *J* = 7.8 Hz, 1H), 7.47 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.28 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.13 (dd, *J* = 7.8, 8.6 Hz, 1H), 7.28 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.13 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.28 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.13 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.28 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.13 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.28 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.13 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.28 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.13 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.28 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.13 (dd, J = 7.8, 6.6 Hz, 1H), 7.8 (dd, J = 7.8, 6.6 Hz, 1H), 7.8 (dd, J = 7.8, 6.6 Hz, 1H), 7.13 (dd, J = 7.8, 6.6 Hz, 1H), 7.8 (dd, J = 7.8, 6.6 Hz, 1H), 7.8 (dd, J = 7.8, 6.6 Hz, 1H), 7.8 (dd, J = 7.8, 7.8 Hz), 7.8 (dd, J = 7.8, 7.8 Hz), 7.8 (dd, J = 7.8 Hz), 7.8 (dd, J =

7.8 Hz, 1H); ¹³C NMR (151 MHz, CDCl₃) δ 139.3, 138.3, 128.2, 126.8, 124.9, 123.9, 121.1, 120.8, 120.3, 119.6, 111.3, 104.3; **IR** (neat): $v_{max} = 3409$, 1496, 1454, 1320, 905, 749 cm⁻¹; R_f 0.60 (hexane/EtOAc, 4/1).



2c (3-Chloro-9*H*-carbazole, CAS 2732-25-4)¹⁰: off-white solid (36 mg, 60%); ¹H NMR (600 MHz, CDCl₃) 8.04 (bs, 1H), 8.03 (s, 1H), 8.02 (d, *J* = 6.6 Hz, 1H), 7.46-7.41 (m, 2H), 7.37 (dd, *J* = 8.4, 1.8 Hz, 1H), 7.34 (d, *J*

= 8.4 Hz, 1H), 7.25 (ddd, J = 8.4, 6.6, 1.8 Hz, 1H); ¹³C NMR (151 MHz, CDCl₃) δ 140.2, 137.9, 126.8, 126.1, 125.2, 124.8, 122.7, 120.7, 120.3, 120.0, 111.74, 111.0; **IR** (neat): $v_{max} = 3405$, 1471, 1270, 904, 726 cm⁻¹; R_f 0.45 (hexane/EtOAc, 4/1).

2d (3-Floro-9*H*-carbazole, CAS 391-45-7)¹⁰: off-white solid (31 mg, 56%); ¹H NMR (600 MHz, CDCl₃) δ 8.03 (d, J = 7.8 Hz, 1H), 8.00 (bs, 1H), 7.73 (dd, J = 8.4, 2.4 Hz, 1H), 7.46 - 7.42 (m, 2H), 7.35 (dd, J = 9.0, 4.2 Hz, 1H),7.24 (ddd, J = 8.4, 6.0, 2.4 Hz, 1H), 7.16 (ddd, J = 9.0, 8.4, 2.4 Hz, 1H); ¹³C **NMR** (151 MHz, CDCl₃) δ 158.5, 156.9, 140.7, 136.0, 126.6, 123.7 (d, J = 127.6 Hz), 120.8,

119.7, 113.8 (d, J = 25.5 Hz), 111.3 (d, J = 9.06 Hz), 111.1, 106.2 (d, J = 23.7 Hz); **IR (neat)**: $v_{\text{max}} = 3421, 1496, 1456, 1168, 806, 746, 724 \text{ cm}^{-1}; \mathbf{R}_f 0.42 \text{ (hexane/EtOAc, 4/1)}.$



2e (3-Trifluoromethyoxy-9*H*-carbazole): brown solid (52 mg, 70%); mp 158 – 159 °C ¹H NMR (600 MHz, CDCl₃) δ 8.05 (bs, 1H), 8.04 (d, J = 7.8 Hz, 1H), 7.90 (s, 1H), 7.45 (dd, J = 7.8, 6.6 Hz, 1H), 7.41 (d, *J* = 7.8 Hz, 1H), 7.36 (d, *J* = 9.0 Hz, 1H), 7.27 (dd, *J* = 9.0, 7.8 Hz, 1H),

7.25 (d, J = 6.6 Hz, 1H); ¹³C NMR (151 MHz, CDCl₃) δ 142.8, 140.5, 137.9, 126.9, 124.0, 123.1, 120.8, 120.6 (q, J = 255.8 Hz), 120.1, 119.7, 113.4, 111.2, 111.1; **IR** (neat): $v_{max} = 3417$, 1457, 1258, 1163, 903, 723 cm⁻¹; **HRMS** m/z (EI) calc. for C₁₃H₈F₃NO [M⁺] 251.0558, found 251.05568; R_f 0.45 (hexane/EtOAc, 4/1).



2f (1-(9H-Carbazol-3-yl)ethan-1-one, CAS 3215-37-0)¹¹: white solid (38 mg, 60%); ¹H NMR (600 MHz, CDCl₃) δ 8.74 (s, 1H), 8.36 (bs, 1H), 8.14 (d, J = 7.8 Hz, 1H), 8.10 (d, J = 9.0 Hz, 1H), 7.48–7.47 (m, 2H), 7.45 (d, J = 8.4 Hz, 1H), 7.32-7.29 (m, 1H), 2.73 (s, 3H); ¹³C

NMR (151 MHz, CDCl₃) δ 198.0, 142.6, 140.2, 129.7, 126.9, 126.8, 123.7, 123.4, 122.1, 120.8, 120.7, 111.2, 110.5, 26.9; **IR** (neat): $v_{max} = 3286$, 1660, 1601, 1361, 1247, 735 cm⁻¹; R_f 0.42 (hexane/EtOAc, 2/1).



2g (Methyl 9H-carbazole-3-carboxylate, CAS 97931-41-4)¹²: offwhite solid (42 mg, 62%); ¹H NMR (600 MHz, CDCl₃) δ 8.82 (s, 1H), 8.47 (bs, 1H), 8.14 (d, J = 8.4 Hz, 1H), 8.12 (dd, J = 7.8 Hz, 1H), 7.48–7.43 (m, 2H), 7.41 (d, J = 8.4 Hz, 1H), 7.29 (dd, J = 8.4, 6.6 Hz,

1H), 3.99 (s, 3H); ¹³C NMR (151 MHz, CDCl₃) δ 168.2, 142.5, 140.2, 127.6, 126.8, 123.5,

123.3, 123.1, 121.5, 120.8, 120.5, 111.1, 110.4, 52.2; **IR** (neat): $v_{max} = 3296$, 1687, 1604, 1332, 1246, 728 cm⁻¹; R_f 0.28 (hexane/EtOAc, 4/1).



2h (9*H*-Carbazole-3-carbonitrile, CAS 57102-93-9)⁸: off-white solid (37 mg, 64%); ¹H NMR (600 MHz, DMSO-d₆) δ 11.85 (s, 1H), 8.70 (s, 1H), 8.24 (d, *J* = 7.8 Hz, 1H), 7.75 (d, *J* = 8.4 Hz, 1H), 7.63 (d, *J* = 8.4 Hz, 1H), 7.57 (d, *J* = 7.8 Hz, 1H), 7.49 (dd, *J* = 7.8, 7.2 Hz, 1H), 7.24 (dd, *J* = 7.8,

7.2 Hz, 1H); ¹³C NMR (151 MHz, DMSO-d₆) δ 142.1, 140.7, 129.0, 127.4, 126.0, 123.1, 122.0, 121.4, 121.0, 120.3, 112.5, 112.0, 100.6; **IR** (neat): $v_{max} = 3305$, 2224, 1734, 1206, 669 cm⁻¹; R_f 0.47 (hexane/EtOAc, 2/1).

O₂N

Me

2i (3-Nitro-9*H*-carbazole, CAS 3077-85-8)¹³: pale yellow solid (32 mg, 50%); ¹H NMR (600 MHz, CDCl₃) δ 8.44 (bs, 1H), 8.37 (s, 1H), 8.17 – 8.13 (m, 3H), 7.55 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.52 (d, *J* = 7.8 Hz, 1H), 7.33 (dd, *J* = 7.8, 6.6 Hz, 1H) ; ¹³C NMR (151 MHz, CDCl₃) δ 146.1, 141.8,

138.4, 128.7, 128.5, 122.2, 121.7, 121.0, 120.5, 115.2, 111.5, 107.0; **IR** (neat): $v_{max} = 3377$, 2923, 1512, 1341, 1323, 903, 733 cm⁻¹; **R**_f 0.39 (hexane/EtOAc, 4/1).

2j (3-Methyl-9*H*-carbazole, CAS 4630-20-0)⁸: off-white solid (35 mg, 65%); ¹H NMR (600 MHz, CDCl₃) δ 8.04 (d, *J* = 7.8 Hz, 1H), 7.90 (bs, 1H), 7.86 (s, 1H), 7.42 – 7.36 (m, 2H), 7.29 (d, *J* = 7.8, 1H), 7.22 (d, *J* = 7.8 Hz, 1H), 7.21 (d, *J* = 7.2 Hz, 1H), 2.53 (s, 3H), ¹³C NMR (151 MHz, 1H), 7.21 (d, *J* = 7.2 Hz, 1H), 2.53 (s, 2H), 2.53 (s, 2H),

CDCl₃) δ 140.0, 137.9, 129.0, 127.4, 125.8, 123.7, 123.4, 120.45, 120.43, 119.4, 110.8, 110.4, 21.6; **IR (neat)**: $v_{max} = 3405$, 2918, 1458, 904, 727 cm⁻¹; R_f 0.54 (hexane/EtOAc, 4/1).



2k (2-Methoxy-9*H*-carbazole, CAS 6933-49-9)¹⁰: gray solid (15 mg, 26%); ¹**H** NMR (600 MHz, CDCl₃) δ 7.97 (d, *J* = 8.4 Hz, 1H), 7.96 (bs, 1H), 7.93 (d, *J* = 8.4 Hz, 1H), 7.38 (d, *J* = 7.8 Hz, 1H), 7.34 (dd, *J* = 7.8, 7.2 Hz, 1H), 7.21 (dd, *J* = 7.8, 7.2 Hz, 1H), 6.91 (s, 1H), 6.86 (d,

J = 7.8 Hz, 1H); 3.90 (s, 3H); ¹³C NMR (151 MHz, CDCl₃) δ 125.3, 141.0, 139.7, 124.8, 121.3, 119.8, 119.7, 117.5, 110.5, 108.4, 94.9, 55.9, 29.3; IR (neat): $v_{max} = 3387$, 2917, 1737, 1462, 903, 727 cm⁻¹; R_f 0.36 (hexane/EtOAc, 4/1).



2l (3-Chloro-6-methyl-9*H*-carbazole): off-white solid (42 mg, 66%); mp 163 – 164 °C; ¹H NMR (600 MHz, CDCl₃) δ 7.99 (s, 1H), 7.94 (bs, 1H), 7.82 (s, 1H), 7.34(d, *J* = 8.4 Hz, 1H), 7.31 (d, *J* = 8.4 Hz, 1H), 7.30 (d, *J* = 8.4 Hz, 1H), 7.27 (d, *J* = 8.4 Hz, 1H), 2.53 (s, 3H); ¹³C NMR

(**151 MHz, CDCl**₃) δ 138.5, 138.2, 129.4, 128.1, 125.9, 124.9, 124.6, 122.9, 120.6, 120.2, 111.7, 110.7, 21.6; **IR (neat)**: $v_{max} = 3392$, 2920, 2852, 1446, 1274, 811, 731 cm⁻¹; **HRMS** *m/z* (EI) calc. for C₁₃H₁₀ClN [M⁺] 215.0502, found 215.0500; **R**_f 0.46 (hexane/EtOAc, 4/1).



2m (2,7-Dimethoxy-9*H*-carbazole, CAS 61822-18-2)¹⁴: palebrown solid (20 mg, 30%); ¹H NMR (600 MHz, DMSO-d₆) δ 10.95 (bs, 1H), 7.83 (d, *J* = 8.4 Hz, 2H), 6.92 (s, 2H), 6.72 (dd, *J* = 8.4 Hz, 2H), 3.80 (s, 6H); ¹³C NMR (151 MHz, DMSO-d₆) δ

158.0, 141.4, 120.3, 116.9, 107.7, 95.1, 55.7; **IR** (neat): $v_{max} = 3379$, 1610, 1456, 1023, 1000, 823, 761 cm⁻¹; R_f 0.23 (hexane/EtOAc, 4/1).



2n (5*H*-Pyrido[3,2-*b*]indole, CAS 245-08-9)⁸: brown solid (22 mg, 45%); ¹**H NMR (600 MHz, DMSO-d₆)** δ 11.42 (bs, 1H), 8.45 (d, *J* = 4.2 Hz, 1H), 8.18 (d, *J* = 7.8 Hz, 1H), 7.88 (d, *J* = 8.4 Hz, 1H), 7.54 (d, *J* = 8.4 Hz, 1H), 7.50 (dd, *J* = 8.4, 7.2 Hz, 1H), 7.38 (dd, *J* = 8.4, 4.2 Hz, 1H), 7.24 (dd, *J* = 7.8, 7.2

Hz, 1H); ¹³C NMR (151 MHz, DMSO-d₆) δ 141.6, 140.9, 133.3, 127.8, 122.0, 120.6, 120.5, 119.7, 118.4, 112.1; **IR** (neat): $v_{max} = 2921$, 1627, 1458, 1394, 889 cm⁻¹; R_f 0.26 (hexane/EtOAc, 2/1).



6a (7*H*-Benzo[*c*]carbazole, CAS 205-25-4)¹⁵: off-white solid (37 mg, 58%); ¹**H** NMR (600 MHz, CDCl₃) δ 8.79 (d, *J* = 8.4 Hz, 1H), 8.57 (d, *J* = 7.8 Hz, 1H), 8.42 (bs, 1H), 8.01 (d, *J* = 8.4 Hz, 1H), 7.87 (d, *J* = 8.4 Hz, 1H), 7.72 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.63 (d, *J* = 8.4 Hz, 1H), 7.58 (d, *J* = 8.4 Hz, 1H), 7.48 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.46 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.39 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.46 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.39 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.46 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.39 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.46 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.39 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.46 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.49 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.46 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.49 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.46 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.49 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.40 (dd, J = 8.4, 7.8

8.4, 7.8 Hz, 1H); ¹³C NMR (151 MHz, CDCl₃) δ 138.7, 137.3, 130.2, 129.43, 129.40, 127.6, 127.1, 124.6, 124.2, 123.5, 123.2, 122.3, 120.5, 115.7, 112.8, 111.3; **IR** (neat): $v_{max} = 3411$, 3050, 2924, 1467, 1359, 805, 741 cm⁻¹; R_f 0.38 (hexane/EtOAc, 4/1).



6b (10-(*tert*-Butyl)-7*H*-benzo[*c*]carbazole, CAS 756822-83-0)¹⁵: brown solid (45 mg, 55%); ¹**H NMR (600 MHz, CDCl**₃) δ 8.79 (d, *J* = 8.4 Hz,

1H), 8.57 (s, 1H), 8.36 (s, 1H), 8.00 (d, J = 9.0 Hz, 1H), 7.84 (d, J = 9.0 Hz, 1H), 7.73 (dd, J = 8.4, 7.2 Hz, 1H), 7.61 (d, J = 9.0 Hz, 1H), 7.55 (d, J = 9.0 Hz, 1H), 7.51 (d, J = 8.4 Hz, 1H), 7.48 (dd, J = 8.4, 7.2 Hz, 1H), 1.54 (s, 9H); ¹³C NMR (151 MHz, CDCl₃) δ 143.4, 137.6, 136.8, 130.2, 129.4, 129.4, 127.3, 127.0, 124.1, 123.4, 123.0, 122.7, 118.2, 115.8, 112.9, 110.8, 35.1, 32.3; **IR (neat)**: v_{max} = 3413, 2959, 1480, 1365, 1286, 908, 805, 735 cm⁻¹; R_f 0.43 (hexane/EtOAc, 4/1).



6c (Methyl 7*H*-benzo[*c*]carbazole-10-carboxylate)¹⁵: brown solid (50 mg, 61%); mp 243-244 °C; ¹H NMR (**600** MHz, DMSO-d₆) δ 12.23 (s, 1H), 9.15 (s, 1H), 8.71 (d, *J* = 8.4 Hz, 1H), 8.07 (d, *J* = 7.8 Hz, 1H), 8.06 (dd, *J* = 8.4, 7.8 Hz, 1H), 7.98 (d, *J* = 9.0 Hz, 1H), 7.80 (d, *J* = 9.0 Hz, 1H), 7.77 (d, *J* = 7.2 Hz, 1H), 7.73 (d, *J* = 8.4 Hz, 1H),

7.51 (dd, J = 8.4, 7.2 Hz, 1H), 3.94 (s, 3H).¹³C NMR (151 MHz, DMSO-d₆) δ 167.5, 141.9, 138.9, 129.8, 129.5, 129.3, 128.4, 127.8, 125.5, 123.7, 123.6, 123.0, 122.9, 121.2, 114.6, 113.9, 112.0, 52.4; **IR** (neat): $v_{max} = 3321$, 2921, 1697, 1616, 796 cm⁻¹; **HRMS** m/z (EI) calc. for C₁₈H₁₃NO₂ [M⁺] 275.0946, found 275.0948; *R*_f 0.22 (hexane/EtOAc, 4/1).



6d (10-Phenyl-7*H*-benzo[*c*]carbazole, CAS 756822-84-1)¹⁵: off-white solid (61 mg, 70%); ¹**H NMR** (**600 MHz**, **CDCl**₃) δ 8.84 (d, *J* = 8.4 Hz, 1H), 8.76 (s, 1H), 8.42 (bs, 1H), 8.02 (d, *J* = 8.4 Hz, 1H), 7.87 (d, *J* = 9.0 Hz, 1H), 7.80 (d, *J* = 7.8 Hz, 2H), 7.74 (dd, *J* = 7.8, 6.6 Hz, 1H), 7.70 (d, *J* = 9.0 Hz, 1H), 7.60 (dd, *J* = 8.4, 7.8 Hz, 2H), 7.55–7.49 (m, 3H), 7.40 (dd,

J = 8.4, 7.2 Hz, 1H); ¹³C NMR (151 MHz, CDCl₃) δ 142.7, 138.1, 137.8, 134.1, 130.2, 129.5, 129.4, 129.0, 127.9, 127.8, 127.1, 126.8, 124.8, 124.3, 123.5, 123.3, 120.9, 115.8, 112.8, 111.5; IR (neat): $v_{max} = 3411, 3049, 2924, 1473, 1288, 906, 804, 734$ cm⁻¹; R_f 0.37 (hexane/EtOAc, 4/1).



8 (7*H*-Dibenzo[*c*,*g*]carbazole, CAS 194-59-2)¹⁵: brown solid (20 mg, 25%); ¹**H** NMR (600 MHz, CDCl₃) δ 9.22 (d, *J* = 8.4 Hz, 2H), 8.78 (bs, 1H), 8.04 (d, *J* = 7.8 Hz, 2H), 7.87 (d, *J* = 8.4 Hz, 2H), 7.70 (dd, *J* = 8.4, 6.6 Hz, 2H), 7.67 (d, *J* = 8.4 Hz, 2H), 7.53 (dd, *J* = 7.8, 6.6 Hz, 2H); ¹³C NMR (151 MHz, CDCl₃) δ 136.4, 130.2, 129.44, 129.39, 127.1, 125.7, 125.4, 123.5,

118.1, 112.8; **IR** (**neat**): $v_{max} = 3408$, 2924, 1529, 1381, 1322, 801, 740 cm⁻¹; R_f 0.34 (hexane/EtOAc, 4/1).

Analytic data for Benzocinnoline N-imide (III)



131.7, 130.5, 129.1, 128.1, 124.4, 122.5, 122.4, 121.7, 121.4, 116.2 (overlapped peaks present); **IR** (**neat**): $v_{max} = 3189$, 2923, 1467, 1384, 1326, 763, 715 cm⁻¹; **HRMS** *m*/*z* (EI) calc. for C₁₂H₉N₃ [M⁺] 195.0796, found 195.0796; *R*_f 0.45 (hexane/EtOAc, 4/1).

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