

# **Supporting Information for Kinetics and Mechanism of the Oxidation of Thiourea Dioxide by Iodine in a Slightly Acidic Medium**

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Table S1: Initial Concentrations of the Reactants in Different Kinetic Runs

No.	[TDO] <sub>0</sub> /mM	[I <sub>2</sub> ] <sub>0</sub> /mM	[I <sup>-</sup> ] <sub>0</sub> /mM	pH	ages of TDO/s
1–3	0.3	1.23	0.1	4.0	362, 78289, 170892
4–6	0.3	1.23	0.1	4.0	255062, 349479, 431574
7–11	0.6	0.61	0.1	4.0	397, 11905, 34974, 80732, 117522
12–15	0.6	0.61	0.1	4.0	203178, 260952, 346865, 434120
16–18	0.8	1.20	0.01, 0.02, 0.04	3.8	362
19–21	0.8	1.20	0.06, 0.08, 0.1	3.8	362
22–24	0.3	1.22	0.01, 0.02, 0.04	3.6	362
25–27	0.3	1.22	0.06, 0.08, 0.1	3.6	362
28–30	0.3, 0.6, 0.8	1.19	0.1	3.6	362
31–34	1.2, 1.6, 2.1, 3.0	1.19	0.1	3.6	362
35–37	3.0, 2.1, 1.6	1.19	0.1	3.8	362
38–41	1.2, 0.8, 0.6, 0.3	1.19	0.1	3.8	362
42–44	3.0, 2.1, 1.6	1.21	0.1	4.0	362
45–48	1.2, 0.8, 0.6, 0.3	1.21	0.1	4.0	362
49–51	3.0, 2.1, 1.2	1.19	0.1	4.2	362
52–54	0.8, 0.6, 0.3	1.19	0.1	4.2	362
55–57	3.0, 2.1, 1.2	1.19	0.1	4.4	362
58–61	1.0, 0.8, 0.6, 0.3	1.19	0.1	4.4	362
62–64	0.6	2.895, 2.172, 1.448	0.1	3.6	362
65–68	0.6	0.965, 0.734, 0.302, 0.151	0.1	3.6	362
69–71	0.6	2.413, 1.800, 1.206	0.1	3.8	362
72–75	0.6	0.814, 0.573, 0.302, 0.161	0.1	3.8	362
76–78	0.6	2.413, 1.800, 1.216	0.1	4.0	362
79–82	0.6	0.814, 0.613, 0.312, 0.151	0.1	4.0	362
83–85	0.6	2.350, 1.810, 1.186	0.1	4.4	362
86–88	0.6	0.804, 0.603, 0.292	0.1	4.4	362

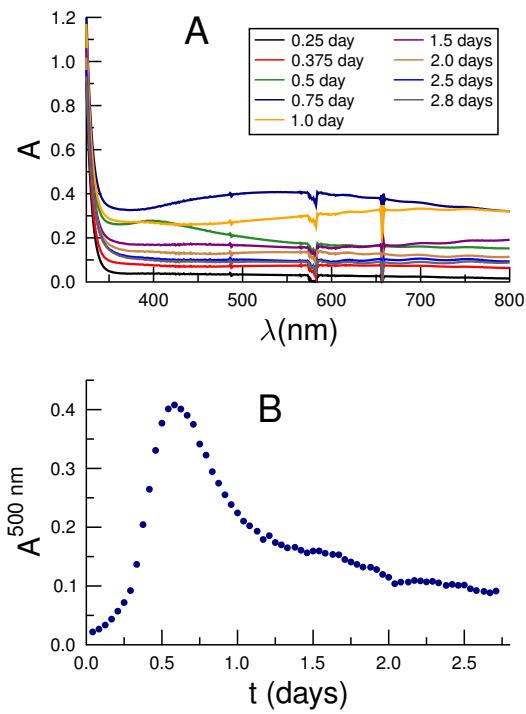


Figure S1: A: Spectral changes of TDO solution at  $\text{pH} = 2$  during the course of ageing.  $[\text{TDO}]_0 = 0.22 \text{ M}$ . B: Absorbance–time plot at  $\lambda = 500 \text{ nm}$ .

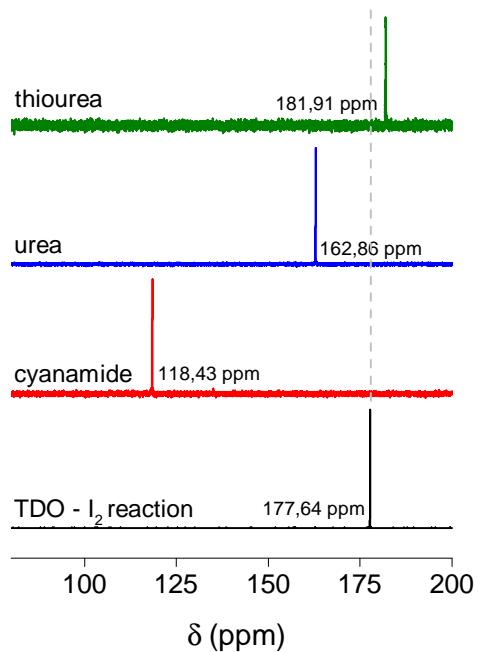


Figure S2:  $^{13}\text{C}$ -NMR spectrum of the end product of the TDO–iodine reaction. Conditions are as follows:  $[\text{TDO}]_0 = 0.1385 \text{ M}$ ,  $\text{T}_{\text{I}_2}^0 = 0.032 \text{ M}$ ,  $\text{T}_{\text{I}^-}^0 = 0.1 \text{ M}$ . Reference spectrum of urea, thiourea, cyanamide are also indicated.

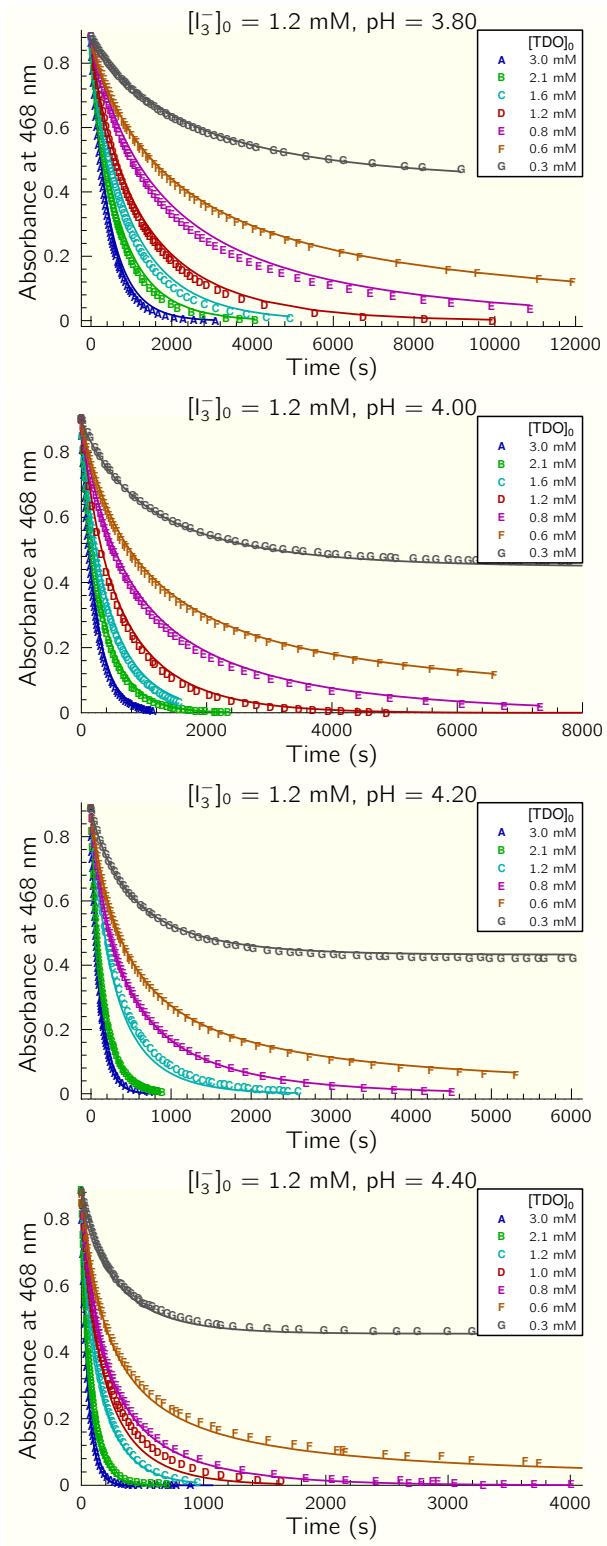


Figure S3: Measured and calculated kinetic curves. TDO dependence.

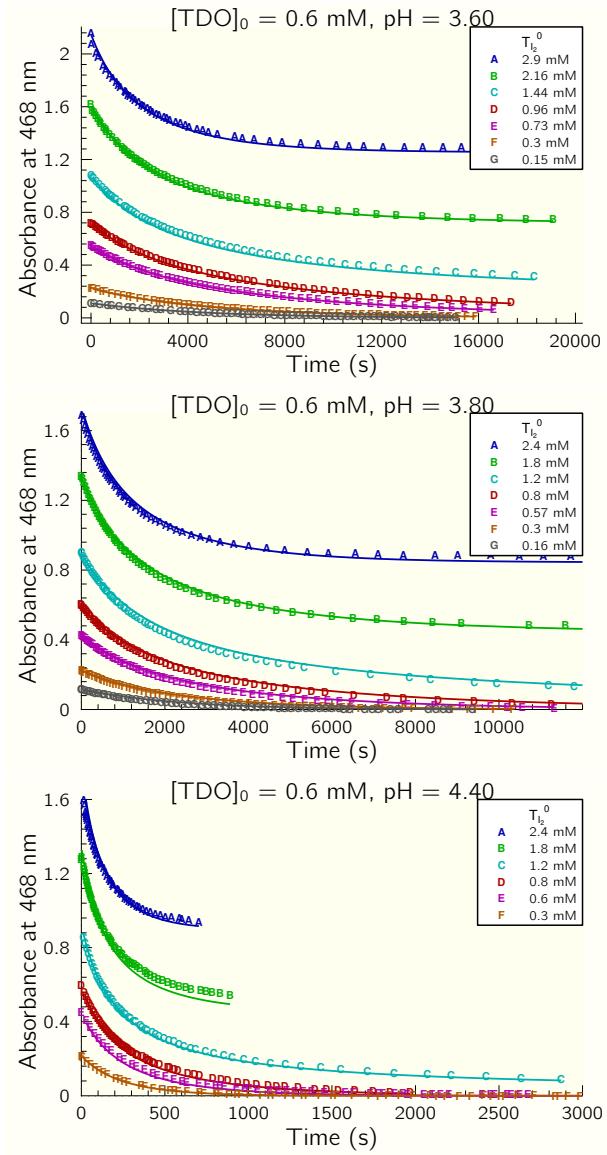


Figure S4: Measured and calculated kinetic curves. Iodine dependence.

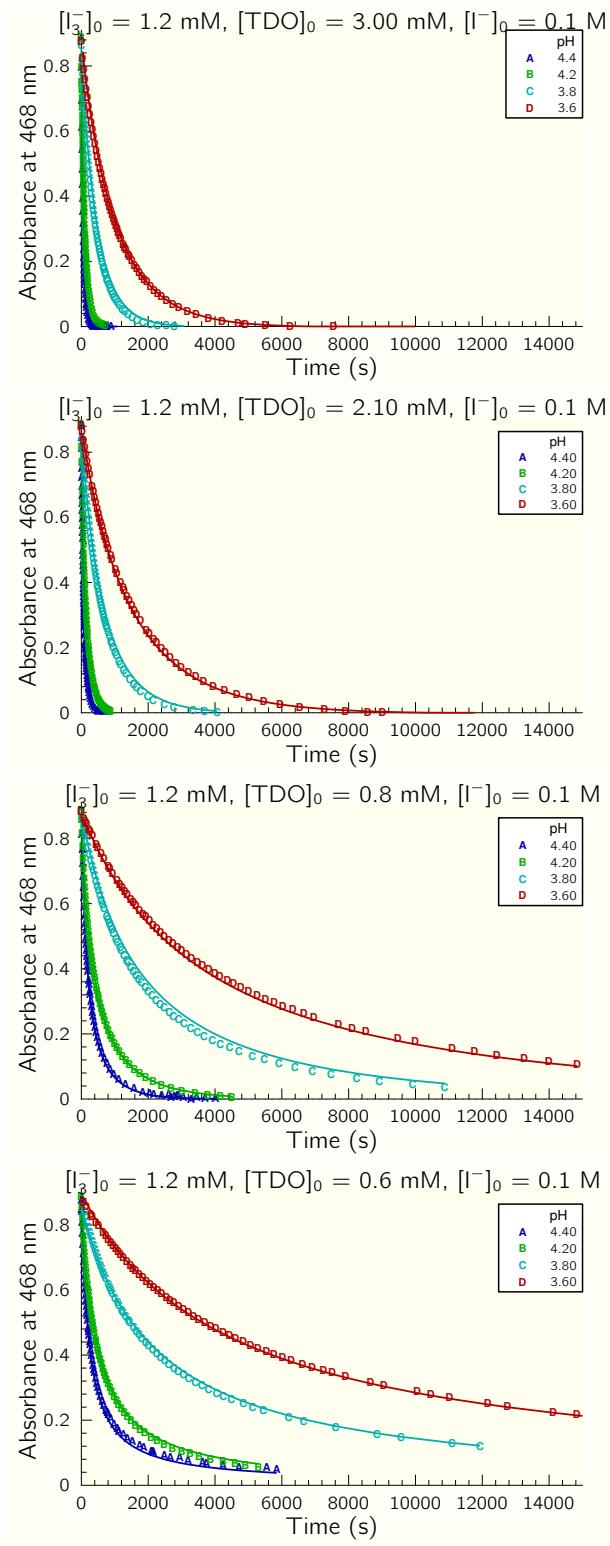


Figure S5: Measured and calculated kinetic curves. pH dependence.

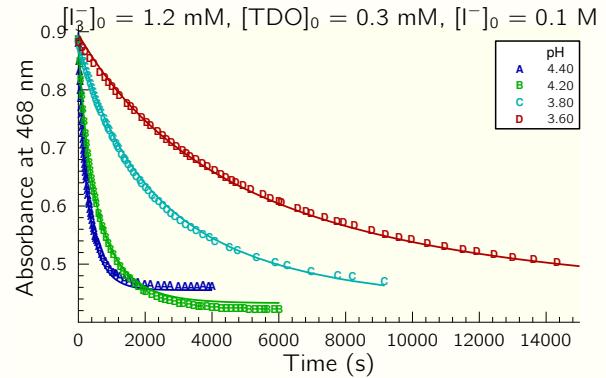


Figure S6: Measured and calculated kinetic curves. pH dependence in iodine excess.

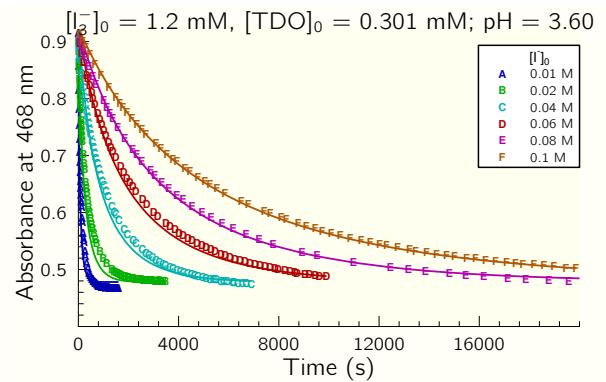


Figure S7: Measured and calculated kinetic curves. Iodide ion dependence.