

SUPPORTING INFORMATION

Exceptional Structural Compliance of the $\text{B}_{12}\text{F}_{12}^{2-}$ Superweak Anion

Dmitry V. Peryshkov and Steven H. Strauss

Abbreviations used in this document: $\text{Y}^{2-} = \text{B}_{12}\text{F}_{12}^{2-}$; $\odot = \text{B}_{12}$ centroid

Table of Contents

	page
Table S1. Bond valence parameters used in this work	S3
Table S2. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{CH}_3\text{CN})_8(\text{B}_{12}\text{F}_{12})$	S4
Table S3. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{CH}_3\text{CN})_5(\text{B}_{12}\text{F}_{12})$	S4
Table S4. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{CH}_3\text{CN})_4(\text{B}_{12}\text{F}_{12})$	S4
Table S5. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_3\text{CN})_4(\text{SO}_3\text{F})$	S5
Table S6. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{PhCH}_3)_6(\text{B}_{12}\text{F}_{12})$	S5
Table S7. Ag–X bond distances and bond valences for $\text{Ag}(\text{PhCH}_3)_3(\text{Li}(\text{Al}_2\text{F}_5(\text{C}(\text{SiMe}_3)_3)_2)_2)$	S5
Table S8. Ag–X bond distances and bond valences for $\text{Ag}(\text{PhCH}_3)_2(\text{Al}(\text{HFIP})_4)$	S6
Table S9. Ag–X bond distances and bond valences for $\text{Ag}(\text{H}_3\text{O})_4(\text{B}_{12}\text{F}_{12})$	S6
Table S10. Ag–X bond distances and bond valences for $\text{Ag}(\text{SO}_2)_6(\text{B}_{12}\text{F}_{12})$	S6
Table S11. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{CH}_2\text{Cl}_2)_4(\text{B}_{12}\text{F}_{12})$	S7
Table S12. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{CH}_2\text{Cl}_2)_4(\text{Pd}(\text{OTeF}_5)_4)$	S7
Table S13. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_2\text{Cl}_2)_2(\text{SbF}_6)$	S7
Table S14. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{CH}_2\text{Cl}_2)_6(\text{Ti}(\text{OTeF}_5)_6)$	S8
Table S15. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_2\text{Cl}_2)(\text{Al}(\text{HFTB})_4)$	S8
Table S16. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_3\text{CN})_2(\text{B}(\text{C}_6\text{H}_3(\text{CF}_3)_2)_4)$	S9
Table S17. Ag–X bond distances and bond valences for $\text{Ag}(\text{CHPh}_3)(1\text{-Me-CB}_{11}\text{F}_{12})$	S9
Table S18. Ag–X bond distances and bond valences for $\text{Ag}(\text{C}_6\text{H}_6)_3(\text{B}(\text{C}_6\text{F}_5)_4)$	S10
Table S19. Ag–X bond distances and bond valences for $\text{Ag}(\text{C}_6\text{D}_6)_3(\text{BF}_4)$	S10
Table S20. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_3\text{CN})_4(\text{B}(\text{C}_6\text{F}_5)_4)$	S11
Table S21. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_3\text{CN})_4(\text{ClO}_4)$	S11
Table S22. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_3\text{CN})_4(\text{Co}(\text{C}_2\text{B}_9\text{H}_{11})_2)$	S11

Figure S1. Thermogravimetric plots for Ag ₂ (CH ₂ Cl ₂) ₄ (Y) and Ag ₂ (CH ₃ CN) ₄ (Y)	S12
Figure S2. FTIR spectra of Ag ₂ (B ₁₂ Cl ₁₂) and Ag ₂ (Y) under high pressures of CO	S13
Figure S3. FTIR spectrum of Ag ₂ (CH ₃ CN) ₄ (Y)	S14
Table S23. Selected Gas-Phase Bond Dissociation Energies	S15
Figure S4. The [(CH ₃ CN) ₂ Ag(μ-CH ₃ CN-κ ¹ ,κ ² N)Ag(CH ₃ CN) ₂] ²⁺ complex in Ag ₂ (CH ₃ CN) ₄ (Y)	S16
Figure S5. Cation-anion packings in Ag ₂ (CH ₂ Cl ₂) ₄ (Y), Ag ₂ (PhCH ₃) ₆ (Y), and Ag ₂ (H ₂ O) ₄ (Y)	S17
Supporting Information References	S18

Table S1. Bond valence parameters used in this work for Ag–X distances (X = F, O, N, C, Cl)^a

cation	R_{AgF}	R_{AgO}	R_{AgN}	R_{AgC}	R_{AgCl}
Ag^+	1.80	1.805	1.85	1.85	2.09

^a The bond valence contribution for each Ag–X distance (i.e., $\text{bv}(\text{AgX})$; X = F, O, N, C, and Cl) is calculated using the formula $\text{bv}(\text{MX}) = \exp[(R_{\text{AgX}} - d_{\text{AgX}})/0.37]$, where d_{AgX} is the Ag–X distance. The R_{AgX} values were taken from Brese, N. E.; O'Keeffe, M. *Acta Cryst.* **1991**, *B47*, 192–197 except for R_{AgC} , which was derived for this work.

Table S2. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{CH}_3\text{CN})_8(\text{B}_{12}\text{F}_{12})$ (this work)

Ag ₂ (CH ₃ CN) ₈ (B ₁₂ F ₁₂) (this work)	bond	distance	bv param	bv	bv/ Σ (bv)
	Ag-N6A	2.263	1.850	0.328	0.290
	Ag-N2	2.225	1.850	0.363	0.321
	Ag-N1	2.295	1.850	0.300	0.266
	Ag-N3	2.419	1.850	0.215	0.190
	Ag-F	3.351	1.805	0.015	
	Ag-F	3.857	1.805	0.004	
Σ (bv) does not include bv values < 0.03			Σ (bv) =	1.206	
			ave. all 4 Ag-N	0.301	0.234
			ave. 3 strongest	0.330	0.253

Table S3. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{CH}_3\text{CN})_5(\text{B}_{12}\text{F}_{12})$ (this work)

Ag ₂ (CH ₃ CN) ₅ (B ₁₂ F ₁₂) (this work)	bond	distance, Å	bv param	bv	bv/ Σ (bv)
	Ag-N2	2.139	1.850	0.458	0.391
	Ag-N3	2.160	1.850	0.433	0.369
	Ag-N1	2.453	1.850	0.196	0.167
	Ag-F1	2.920	1.805	0.049	0.042
	Ag-F2	3.041	1.805	0.035	0.030
			Σ (bv) =	1.171	
			ave. all 3 Ag-N	0.362	0.309
			ave. 2 strongest	0.445	0.380

Table S4. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{CH}_3\text{CN})_4(\text{B}_{12}\text{F}_{12})$ (this work)

Ag ₂ (CH ₃ CN) ₄ (B ₁₂ F ₁₂) (this work)	bond	distance	bv param	bv	bv/ Σ (bv)
	Ag-N1	2.101	1.850	0.507	0.412
	Ag-N1A	2.101	1.850	0.507	0.412
	Ag-F1	2.797	1.805	0.068	0.056
	Ag-F1A	2.797	1.805	0.068	0.056
	Ag-F4	2.992	1.805	0.040	0.033
	Ag-F4A	2.992	1.805	0.040	0.033
			Σ (bv) =	1.233	

Table S5. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_3\text{CN})_4(\text{SO}_3\text{F})^{\text{a}}$

$\text{Ag}(\text{CH}_3\text{CN})_4(\text{SO}_3\text{F})$	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$
Dortbudak <i>ZNaturforsch B</i> 2014	Ag-N6A	2.248	1.850	0.341	0.270
	Ag-N2	2.270	1.850	0.321	0.255
	Ag-N1	2.281	1.850	0.312	0.247
	Ag-N3	2.313	1.850	0.286	0.227
			$\Sigma(\text{bv}) =$	1.261	1.000
			ave. all 4 Ag-N	0.315	0.250

^a Dortbudak et al., *Z. Naturforsch. B*. **2014**, *69*, 373–375 (ref 24 in the main text).

Table S6. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{PhCH}_3)_6(\text{B}_{12}\text{F}_{12})$ (this work)

$\text{Ag}_2(\text{PhCH}_3)_6(\text{B}_{12}\text{F}_{12})$ (this work)	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$
	Ag-C	2.408	1.85	0.221	0.234
	Ag-C	2.427	1.85	0.210	0.223
	Ag-C	2.512	1.85	0.167	0.177
	Ag-C	2.600	1.85	0.132	0.140
	Ag-C	2.652	1.85	0.114	0.121
	Ag-C	2.899	1.85	0.059	0.062
	Ag-F	2.991	1.80	0.040	0.042
			$\Sigma(\text{bv}) =$	0.944	
			ave. all six Ag-C	0.151	0.160
			ave. 4 strongest	0.183	0.194
			ave. 3 strongest	0.200	0.212

Table S7. Ag–X bond distances and bond valences for $\text{Ag}(\text{PhCH}_3)_3(\text{Li}(\text{Al}_2\text{F}_5(\text{C}(\text{SiMe}_3)_3)_2)_2)^{\text{a}}$

$\text{Ag}(\text{PhCH}_3)_3(\text{X})$	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$
Roesky et al. <i>OM</i> 1998	Ag-C	2.396	1.85	0.229	0.240
	Ag-C	2.408	1.85	0.221	0.232
$\text{X} = \text{Li}(\text{Al}_2\text{F}_5(\text{C}(\text{SiMe}_3)_3)_2)_2$	Ag-C	2.504	1.85	0.171	0.179
	Ag-C	2.544	1.85	0.153	0.161
	Ag-C	2.696	1.85	0.102	0.107
	Ag-C	2.790	1.85	0.079	0.083
			$\Sigma(\text{bv}) =$	0.954	
			ave. Ag-C bv	0.159	0.167
			ave. 4 strongest	0.193	0.202
			ave. 3 strongest	0.207	0.217

^a Roesky et al., *Organometallics* **1998**, *17*, 4326–4328 (ref 26 in the main text).

Table S8. Ag–X bond distances and bond valences for $\text{Ag}(\text{PhCH}_3)_2(\text{Al}(\text{HFIP})_4)^{\text{a}}$

$\text{Ag}(\text{PhCH}_3)_2(\text{Al}(\text{HFIP})_4)$	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$
Krossing <i>Chem. Eur. J.</i> 2001	Ag1-C	2.363	1.850	0.250	0.266	Ag2-C	2.416	1.850	0.217	0.221
	Ag1-C	2.435	1.850	0.206	0.219	Ag2-C	2.469	1.850	0.188	0.191
HFIP = OCH(CF ₃) ₂	Ag1-C	2.593	1.850	0.134	0.143	Ag2-C	2.484	1.850	0.180	0.184
	Ag1-C	2.647	1.850	0.116	0.123	Ag2-C	2.496	1.850	0.174	0.178
	Ag1-O	2.596	1.805	0.118	0.125	Ag2-O	2.541	1.805	0.137	0.139
	Ag1-O	2.599	1.805	0.117	0.124	Ag2-O	2.715	1.805	0.085	0.087
$\Sigma(\text{bv})$ does not include	Ag1-F	3.374	1.800	0.014		Ag2-F	3.397	1.800	0.013	
bv values < 0.03	Ag1-F	3.482	1.800	0.011		Ag2-F	3.505	1.800	0.010	
			$\Sigma(\text{bv}) =$	0.941				$\Sigma(\text{bv}) =$	0.981	
			ave. all 4 Ag-C	0.176	0.187			ave. all 4 Ag-C	0.190	0.194
			ave. 2 strongest	0.228	0.242			ave. 2 strongest	0.202	0.206

^a Krossing, *Chem. Eur. J.* **2001**, 7, 490–502 (ref 27 in the main text).

Table S9. Ag–X bond distances and bond valences for $\text{Ag}(\text{H}_3\text{O})_4(\text{B}_{12}\text{F}_{12})^{\text{a}}$

$\text{Ag}(\text{H}_2\text{O})_4(\text{B}_{12}\text{F}_{12})$	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$
Malishevski <i>Inorg. Chem.</i> 2016	Ag1-O1	2.371	1.805	0.217	0.250	Ag2-O2	2.376	1.805	0.214	0.240
	Ag1-O3	2.408	1.805	0.196	0.226	Ag2-O4	2.388	1.805	0.207	0.232
	Ag1-O2	2.411	1.805	0.194	0.224	Ag2-O3	2.467	1.805	0.167	0.188
	Ag1-O4	2.524	1.805	0.143	0.165	Ag2-O1	2.511	1.805	0.148	0.167
	Ag1-F8	2.734	1.800	0.080	0.092	Ag2-F11	2.742	1.800	0.078	0.088
	Ag1-F11	3.021	1.800	0.037	0.043	Ag2-F12	2.751	1.800	0.077	0.086
			$\Sigma(\text{bv}) =$	0.867				$\Sigma(\text{bv}) =$	0.891	
			ave. all 4 Ag-OH2	0.188				ave. all 4 Ag-OH2	0.184	

^a Malishevski et al., *Inorg. Chem.* **2016**, in press (doi 10.1021/om9803016; ref 1 in the main text).

Table S10. Ag–X bond distances and bond valences for $\text{Ag}(\text{SO}_2)_6(\text{B}_{12}\text{F}_{12})^{\text{a}}$

$\text{Ag}(\text{SO}_2)_6(\text{B}_{12}\text{F}_{12})$	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$
Malishevski <i>Inorg. Chem.</i> 2016	Ag-O6	2.566	1.805	0.128	0.137
	Ag-O4	2.564	1.805	0.129	0.138
	Ag-O2	2.525	1.805	0.143	0.153
	Ag-O1A	2.661	1.805	0.099	0.106
	Ag-O3A	2.658	1.805	0.100	0.107
	Ag-O5A	2.790	1.805	0.070	0.075
	Ag-F1A	2.558	1.800	0.129	0.138
	Ag-F4	2.541	1.800	0.135	0.145
			$\Sigma(\text{bv}) =$	0.932	
			ave. all 6 Ag-OSO	0.111	0.119

^a Malishevski et al., *Inorg. Chem.* **2016**, in press (doi 10.1021/om9803016; ref 1 in the main text).

Table S11. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{CH}_2\text{Cl}_2)_4(\text{B}_{12}\text{F}_{12})$ (this work)

$\text{Ag}_2(\text{CH}_2\text{Cl}_2)_4(\text{B}_{12}\text{F}_{12})$ (this work)	bond	distance	bv param	bv	$\text{bv}/\sum(\text{bv})$
	Ag-Cl	2.659	2.090	0.215	0.224
	Ag-Cl	2.691	2.090	0.197	0.205
	Ag-Cl	2.834	2.090	0.134	0.139
	Ag-Cl	2.884	2.090	0.117	0.122
	Ag-F	2.634	1.800	0.105	0.109
	Ag-F	2.649	1.800	0.101	0.105
	Ag-F	2.679	1.800	0.093	0.097
			$\Sigma(\text{bv}) =$	0.962	
			ave. all 4 Ag-Cl	0.166	0.173
			ave. 2 strongest	0.206	0.214

Table S12. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{CH}_2\text{Cl}_2)_4(\text{Pd}(\text{OTeF}_5)_4)^{\text{a}}$

$\text{Ag}_2(\text{CH}_2\text{Cl}_2)_4(\text{Pd}(\text{OTeF}_5)_4)$ Colsman <i>J. Am. Chem. Soc.</i> 1990	bond	distance	bv param	bv	$\text{bv}/\sum(\text{bv})$
	Ag-Cl	2.775	2.090	0.157	0.171
	Ag-Cl	2.811	2.090	0.142	0.156
	Ag-Cl	2.859	2.090	0.125	0.137
	Ag-Cl	2.882	2.090	0.118	0.128
	Ag-O	2.532	1.805	0.140	0.153
	Ag-O	2.404	1.805	0.198	0.216
	Ag-F	3.030	1.800	0.036	0.039
			$\Sigma(\text{bv}) =$	0.916	
			ave. all 4 Ag-Cl	0.136	0.148
			ave. 2 strongest	0.150	0.164

^a Colsman et al., *J. Am. Chem. Soc.* **1990**, *112*, 2349–2362 (ref 12 in the main text).

Table S13. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_2\text{Cl}_2)_2(\text{SbF}_6)^{\text{a}}$

$\text{Ag}(\text{CH}_2\text{Cl}_2)_2(\text{SbF}_6)$ Decken <i>Chem. Eur. J.</i> 2009	bond	distance	bv param	bv	$\text{bv}/\sum(\text{bv})$
	Ag-Cl	2.903	2.090	0.111	0.114
	Ag-Cl	2.706	2.090	0.189	0.194
	Ag-Cl	2.657	2.090	0.216	0.222
	Ag-Cl	3.128	2.090	0.060	0.062
	Ag-F	2.601	1.800	0.115	0.118
	Ag-F	2.525	1.800	0.141	0.145
	Ag-F	2.524	1.800	0.141	0.145
			$\Sigma(\text{bv}) =$	0.974	
			ave. all 4 Ag-Cl	0.144	0.148
			ave. 2 strongest	0.203	0.208

^a Decken et al., *Chem. Eur. J.* **2009**, *15*, 6504–6517 (ref 31 in the main text).

Table S14. Ag–X bond distances and bond valences for $\text{Ag}_2(\text{CH}_2\text{Cl}_2)_6(\text{Ti}(\text{OTeF}_5)_6)$ ^a

$\text{Ag}_2(\text{CH}_2\text{Cl}_2)_6(\text{Ti}(\text{OTeF}_5)_6)$	bond	distance	R	bv	$\text{bv}/\Sigma(\text{bv})$
van Seggen <i>Inorg. Chem.</i> 1995	Ag-Cl	2.656	2.090	0.217	0.230
	Ag-Cl	2.702	2.090	0.191	0.203
	Ag-Cl	2.719	2.090	0.183	0.194
	Ag-Cl	2.856	2.090	0.126	0.134
	Ag-Cl	3.030	2.090	0.079	0.084
	Ag-Cl	3.049	2.090	0.075	0.079
	Ag-F	3.029	1.800	0.036	0.038
	Ag-F	3.033	1.800	0.036	0.038
			$\Sigma(\text{bv}) =$	0.942	
			ave. all 6 Ag-Cl	0.145	0.154
			ave. 3 strongest	0.197	0.209

^a van Seggen et al., *J. Am. Chem. Soc.* **1990**, *112*, 2349–2362 (ref 32 in the main text).

Table S15. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_2\text{Cl}_2)(\text{Al}(\text{HFTB})_4)$ ^a

$\text{Ag}(\text{CH}_2\text{Cl}_2)(\text{Al}(\text{HFTB})_4)$	bond	distance	R	bv	$\text{bv}/\Sigma(\text{bv})$
Krossing <i>Chem. Eur. J.</i> 2001	Ag-Cl	2.613	2.090	0.243	0.244
	Ag-Cl	2.874	2.090	0.120	0.121
	Ag-F	2.857	1.800	0.057	0.058
	Ag-F	2.931	1.800	0.047	0.047
	Ag-F	2.924	1.800	0.048	0.048
	Ag-F	2.849	1.800	0.059	0.059
	Ag-O	2.377	1.805	0.213	0.214
	Ag-O	2.386	1.805	0.208	0.209
			$\Sigma(\text{bv}) =$	0.996	
			ave. both Ag-Cl	0.182	0.183

^a Krossing et al., *Chem. Eur. J.* **2001**, *7*, 490–502 (ref 27 in the main text).

Table S16. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_3\text{CN})_2(\text{B}(\text{C}_6\text{H}_3(\text{CF}_3)_2)_4)^{\text{a}}$

$\text{Ag}(\text{CH}_3\text{CN})_2(\text{B}(\text{C}_6\text{H}_3(\text{CF}_3)_2)_4)$	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$
Zhang <i>New J. Chem. 2005</i>	Ag-N	2.097	1.850	0.513	0.416
	Ag-N	2.097	1.850	0.513	0.416
	Ag-F	3.132	1.805	0.028	0.022
	Ag-F	3.132	1.805	0.028	0.022
				$\Sigma(\text{bv}) =$	1.081

^a Zhang et al., *New J. Chem. Eur. J. 2005*, 29, 366–370 (ref 25 in the main text). There are two unique $\text{Ag}(\text{CH}_3\text{CN})_2^+$ cations in the structure. The second cation has two symmetry related Ag–N bonds of 2.066(4) Å and two symmetry related Ag–F bonds of 2.131 Å, giving a $\Sigma(\text{bv})$ value of 1.171.

Table S17. Ag–X bond distances and bond valences for $\text{Ag}(\text{CHPh}_3)(1\text{-Me-CB}_{11}\text{F}_{12})^{\text{a}}$

$\text{Ag}(\text{CHPh}_3)(1\text{-Me-CB}_{11}\text{F}_{12})$	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$
Ivanov <i>Cryst. Growth Design 2004</i>	Ag-C	2.550	1.85	0.151	0.149
	Ag-C	2.551	1.85	0.150	0.149
	Ag-C	2.407	1.85	0.222	0.220
	Ag-C	2.468	1.85	0.188	0.186
	Ag-C	2.451	1.85	0.197	0.195
	Ag-C	2.692	1.85	0.103	0.102
			$\Sigma(\text{bv}) =$	1.011	
			ave. all six Ag-C	0.169	
			ave. 3 strongest	0.202	

^a Ivanov et al., *Crystal Growth Des. 2004*, 4, 249–254 (ref 28 in the main text).

Table S18. Ag–X bond distances and bond valences for $\text{Ag}(\text{C}_6\text{H}_6)_3(\text{B}(\text{C}_6\text{F}_5)_4)^{\text{a}}$

Ag(C ₆ H ₆) ₃ (B(C ₆ F ₅) ₄)	bond	distance	bv param	bv	bv/ Σ (bv)
Ogawa <i>Organometallics</i> 2005	Ag-C	2.391	1.85	0.232	0.231
	Ag-C	2.398	1.85	0.227	0.226
	Ag-C	2.400	1.85	0.226	0.225
	Ag-C	2.646	1.85	0.116	0.116
	Ag-C	2.660	1.85	0.112	0.111
	Ag-C	2.737	1.85	0.091	0.091
				$\Sigma(\text{bv}) =$	1.005
			ave. all six Ag-C	0.167	0.160
			ave. 3 strongest	0.228	0.194

^a Ogawa et al., *Organometallics* **2005**, 24, 4842–4844 (ref 29 in the main text).

Table S19. Ag–X bond distances and bond valences for $\text{Ag}(\text{C}_6\text{D}_6)_3(\text{BF}_4)^{\text{a}}$

Ag(C ₆ D ₆) ₃ (BF ₄)	bond	distance	bv param	bv	bv/ Σ (bv)
Batsanov <i>JOMC</i> 1998	Ag-C	2.480	1.85	0.182	0.190
	Ag-C	2.487	1.85	0.179	0.187
	Ag-C	2.525	1.85	0.161	0.169
	Ag-C	2.536	1.85	0.157	0.164
	Ag-C	2.880	1.85	0.062	0.065
	Ag-C	2.889	1.85	0.060	0.063
	Ag-F	2.488	1.80	0.156	0.163
			$\Sigma(\text{bv}) =$	0.957	
			ave. all six Ag-C	0.134	0.160
			ave. 4 strongest	0.170	0.194

^a Batsanov et al., *J. Organometal. Chem.* **1998**, 550, 59–61 (ref 30 in the main text).

Table S20. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_3\text{CN})_4(\text{B}(\text{C}_6\text{F}_5)_4)^{\text{a}}$

$\text{Ag}(\text{CH}_3\text{CN})_4(\text{B}(\text{C}_6\text{F}_5)_4)$	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$
Zhang <i>New J. Chem.</i> 2005	Ag-N	2.265	1.850	0.326	0.264
	Ag-N	2.279	1.850	0.314	0.254
	Ag-N	2.281	1.850	0.312	0.253
	Ag-N	2.298	1.850	0.298	0.242
			$\Sigma(\text{bv}) =$	1.249	

^a Zhang et al., *New J. Chem.* **2005**, 29, 366–370 (ref 25 in the main text).

Table S21. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_3\text{CN})_4(\text{ClO}_4)^{\text{a}}$

$\text{Ag}(\text{CH}_3\text{CN})_4(\text{ClO}_4)$	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$
Jones <i>Z. Krist.</i> 1993	Ag-N	2.240	1.850	0.349	0.283
	Ag-N	2.251	1.850	0.338	0.274
	Ag-N	2.247	1.850	0.342	0.277
	Ag-N	2.273	1.850	0.319	0.259
			$\Sigma(\text{bv}) =$	1.348	

^a Jones et al., *Z. Krist.* **1993**, 208, 213–218 (ref 68 in the main text).

Table S22. Ag–X bond distances and bond valences for $\text{Ag}(\text{CH}_3\text{CN})_4(\text{Co}(\text{C}_2\text{B}_9\text{H}_{11})_2)^{\text{a}}$

$\text{Ag}(\text{CH}_3\text{CN})_4(\text{Co}(\text{C}_2\text{B}_9\text{H}_{11})_2)$	bond	distance	bv param	bv	$\text{bv}/\Sigma(\text{bv})$
Xie <i>JOMC</i> 2000	Ag-N1	2.31	1.850	0.288	0.229
	Ag-N1'	2.31	1.850	0.288	0.229
	Ag-N2	2.19	1.850	0.399	0.316
	Ag-N2'	2.19	1.850	0.399	0.316
			$\Sigma(\text{bv}) =$	1.375	1.090
			ave. all 4 Ag-N	0.344	0.250

^a Xie et al., *J. Organometal. Chem.* **2000**, 613, 99–104 (ref 69 in the main text).

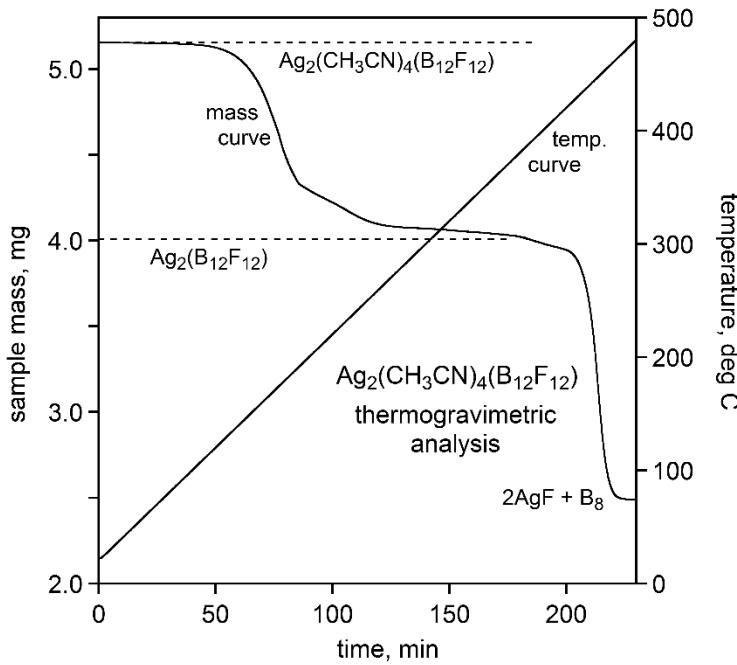
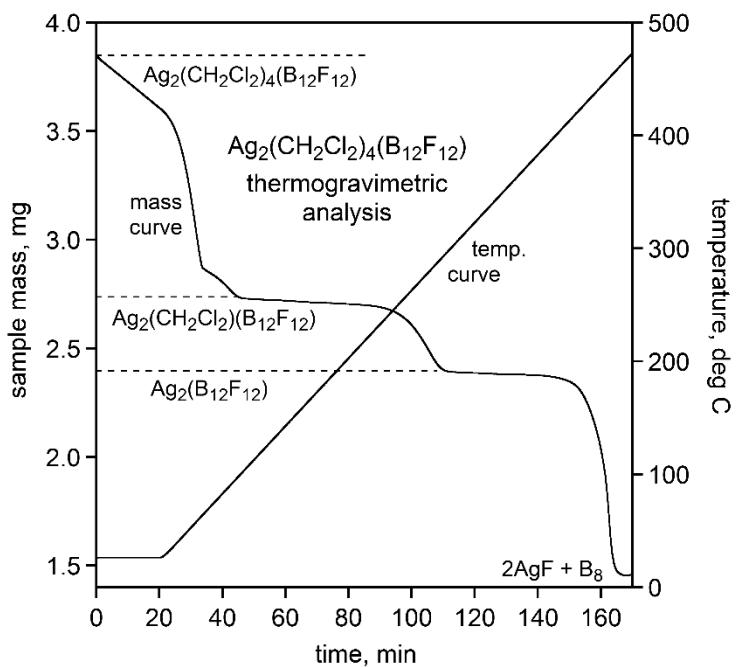


Figure S1. Thermogravimetric plots for $\text{Ag}_2(\text{CH}_2\text{Cl}_2)_4(\text{B}_{12}\text{F}_{12})$ (top) and $\text{Ag}_2(\text{CH}_3\text{CN})_4(\text{B}_{12}\text{F}_{12})$ (bottom). The final mass designated $2\text{AgF} + \text{B}_8$ is consistent with this composition. However, the chemical nature of the final decomposed product was not investigated experimentally.

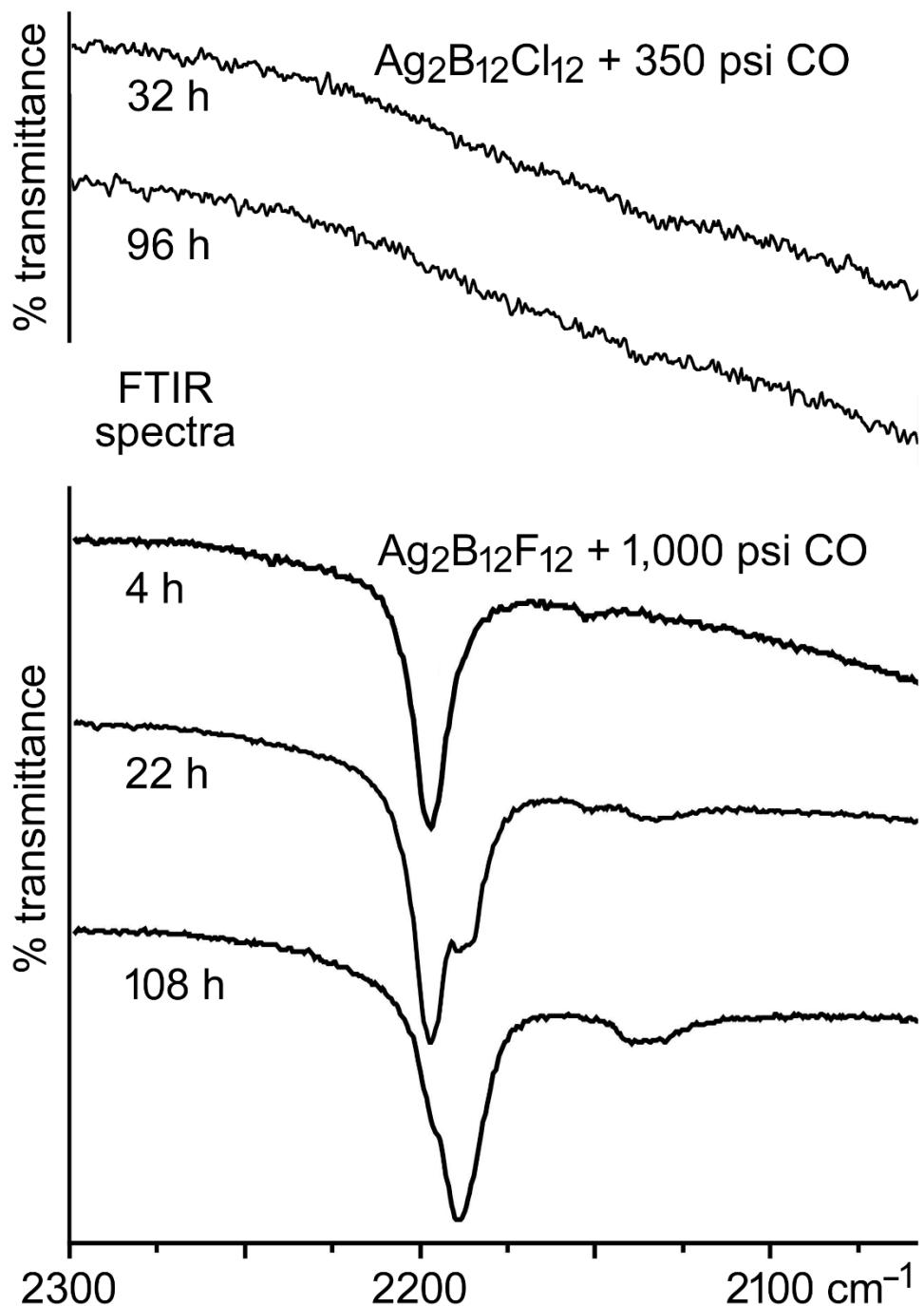


Figure S2. FTIR spectra of Nujol mulls of $\text{Ag}_2(\text{B}_{12}\text{Cl}_{12})$ (top) and $\text{Ag}_2(\text{B}_{12}\text{F}_{12})$ (bottom) under high pressures of CO in a high-pressure transmission IR cell of local design (see text for details). No $\text{Ag}(\text{CO})_n^+$ species were observed when the sample was $\text{Ag}_2(\text{B}_{12}\text{Cl}_{12})$.

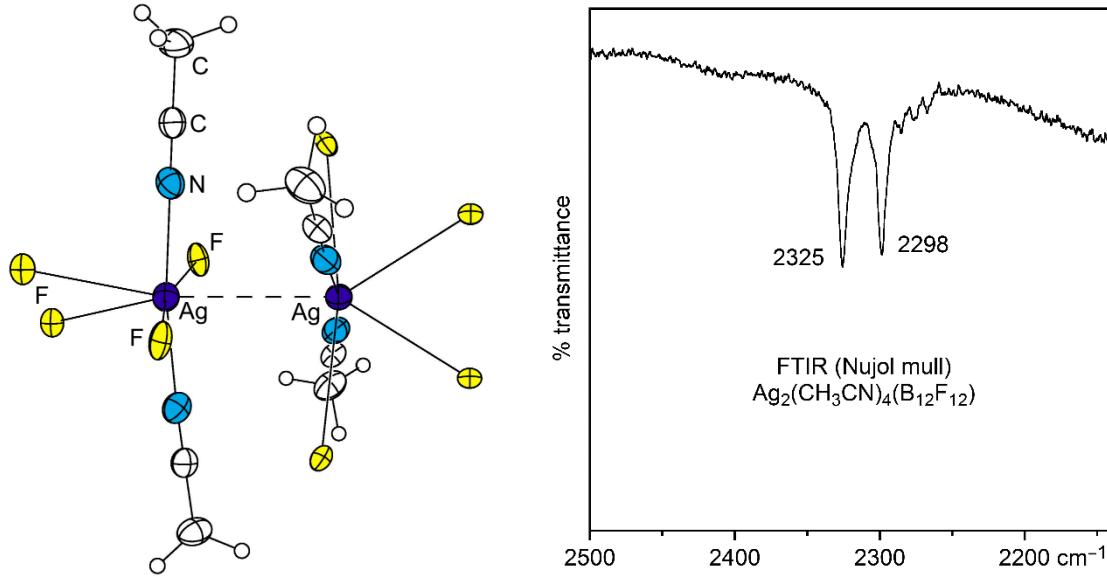


Figure S3. FTIR spectrum of a Nujol mull of $\text{Ag}(\text{CH}_3\text{CN})_4(\text{B}(\text{C}_6\text{F}_5)_4)$. Isolated linear $\text{Ag}(\text{CH}_3\text{CN})_2^+$ complexes should give rise to only one IR $\nu(\text{CN})$ band. The presence of two bands indicates that the pair of $\text{Ag}(\text{CH}_3\text{CN})_2^+$ complexes with a $\text{Ag}\cdots\text{Ag}$ separation of 3.131 Å are acting as a vibrationally-coupled unit, giving rise to two IR $\nu(\text{CN})$ bands and presumably two Raman $\nu(\text{CN})$ bands. The dihedral angle formed by the two AgN_2 planes is 58.7°.

Table S23. Selected Gas-Phase Bond Dissociation Energies^a

species	BDE, kJ mol ⁻¹	reference (in this PDF)
H ₂ O–Na ⁺	100(5)	1
H ₂ O–K ⁺	75(4)	1
H ₂ O–Cu ⁺	157(8)	2
H ₂ O–Ag ⁺	131(8)	3
(C ₆ H ₆)–Na ⁺	93(6)	4
(C ₆ H ₆)–K ⁺	73(4)	4
(C ₆ H ₆)–Cu ⁺	217(6)	5
(C ₆ H ₆)–Ag ⁺	156(7)	6
(C ₆ H ₅ CH ₃)–K ⁺	80(5)	7
(C ₆ H ₅ CH ₃)–Cu ⁺	219(10)	5
(CH ₃ CN)–Na ⁺	102(4)	8
(CH ₃ CN)–K ⁺	102(4)	8
(CH ₃ CN)–Cu ⁺	238(3)	9
(CH ₃ CN)–Ag ⁺	162(6)	10
OC–Na ⁺	31(8)	11
OC–K ⁺	19(5)	11
OC–Cu ⁺	149(7)	12
OC–Ag ⁺	89(5)	12

^a Most of these values are from threshold collision-induced dissociation experiments and are adjusted for 0 K. See individual references for details.

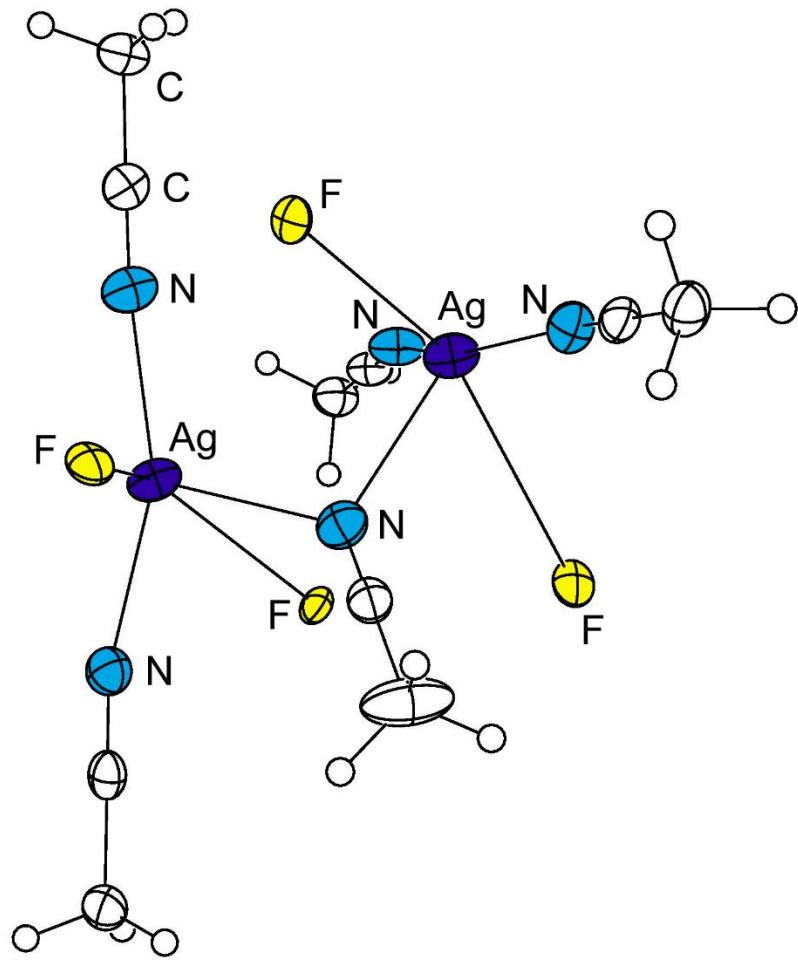


Figure S4. Drawing of the $[(\text{CH}_3\text{CN})_2\text{Ag}(\mu\text{-CH}_3\text{CN}-\kappa^1,\kappa^2\text{N})\text{Ag}(\text{CH}_3\text{CN})_2]^{2+}$ dinuclear complex and the four F atoms bonded to it in the structure of $\text{Ag}_2(\text{CH}_3\text{CN})_5(\text{B}_{12}\text{F}_{12})$.

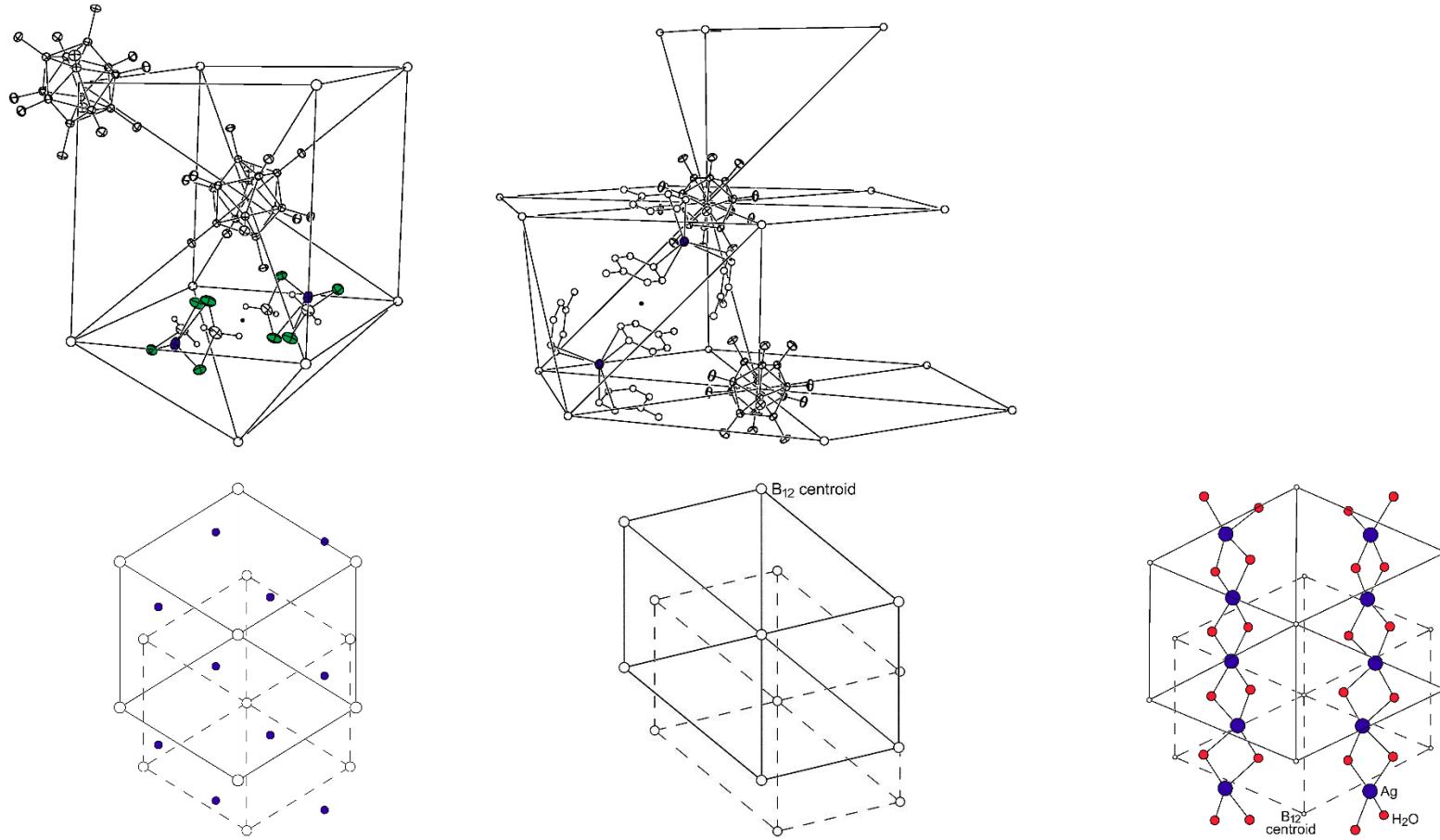


Figure S5. Packing of $\text{Ag}(\text{L})_n^+$ cations and $\text{B}_{12}\text{F}_{12}^{2-}$ anions (Y^{2-}) in the structures of $\text{Ag}_2(\text{CH}_2\text{Cl}_2)_4(\text{Y})$ (left; this work; 50% thermal ellipsoids except for H atoms), $\text{Ag}_2(\text{PhCH}_3)_6(\text{Y})$ (middle; this work; 50% thermal ellipsoids except for C atoms; H atoms omitted for clarity), and $\text{Ag}_2(\text{H}_2\text{O})_4(\text{Y})$ (right; ref 13; H atoms omitted for clarity). In all three structures the B_{12} centroids (\odot s) form rigorously-planar, idealized close-packed arrays that are not stacked in a close-packed manner in the third dimension. Instead, they are offset (i) by ca. one-half of the $\odot \cdots \odot$ distance along one of their respective $\odot \cdots \odot \cdots \odot$ vectors and (ii) by a much smaller displacement perpendicular to the aforementioned $\odot \cdots \odot \cdots \odot$ vector.

References

- (1) Džidić, I.; Kebarle, P. Hydration of the Alkali Ions in the Gas Phase. Enthalpies and Entropies of Reactions $M^+(H_2O)n-1 + H_2O = M^+(H_2O)_n$. *J. Phys. Chem.* **1970**, *74*, 1466-1474.
- (2) Dalleska, N. F.; Honma, K.; Sunderlin, L. S.; Armentrout, P. B. Solvation of Transition Metal Ions by Water. Sequential Binding Energies of $M^+(H_2O)_x$ ($x = 1-4$) for $M = Ti$ to Cu Determined by Collision-Induced Dissociation. *J. Am. Chem. Soc.* **1994**, *116*, 3519-3528.
- (3) Koizumi, H.; Larson, M.; Muntean, F.; Armentrout, P. B. Sequential bond energies of $Ag^+(H_2O)_n$ and $Ag^+(dimethyl\ ether)_n$, $n = 1-4$, determined by threshold collision-induced dissociation. *Int. J. Mass Spectrom.* **2003**, *228*, 221-235.
- (4) Amicangelo, J. C.; Armentrout, P. B. Absolute Binding Energies of Alkali-Metal Cation Complexes with Benzene Determined by Threshold Collision-Induced Dissociation Experiments and ab Initio Theory. *J. Phys. Chem. A* **2000**, *104*, 11420-11432.
- (5) Ruan, C.; Yang, Z.; Rodgers, M. T. Influence of the d orbital occupation on the nature and strength of copper cation- π interactions: threshold collision-induced dissociation and theoretical studies. *Phys. Chem. Chem. Phys.* **2007**, *9*, 5902-5918.
- (6) Chen, Y.-M.; Armentrout, P. B. Collision-induced dissociation of $Ag(C_6H_6)^+$. *Chem. Phys. Lett.* **1993**, *210*, 123-128.
- (7) Amunugama, R.; Rodgers, M. T. Influence of Substituents on Cation- π Interactions. 1. Absolute Binding Energies of Alkali Metal Cation-Toluene Complexes Determined by Threshold Collision-Induced Dissociation and Theoretical Studies. *J. Phys. Chem. A* **2002**, *106*, 5529-5539.
- (8) Davidson, W. R.; Kebarle, P. Ionic Solvation by Aprotic Solvents. Gas Phase Solvation of the Alkali Ions by Acetonitrile. *J. Am. Chem. Soc.* **1976**, *98*, 6125-6133.
- (9) Vitale, G.; Valina, A. B.; Huang, H.; Amunugama, R.; Rodgers, M. T. Solvation of Copper Ions by Acetonitrile. Structures and Sequential Binding Energies of $Cu^+(CH_3CN)_x$, $x = 1-5$, from Collision-Induced Dissociation and Theoretical Studies. *J. Phys. Chem. A* **2001**, *105*, 11351-11364.
- (10) Shoeib, T.; El Aribi, H.; Siu, K. W. M.; Hopkinson, A. C. A Study of Silver (I) Ion-Organonitrile Complexes: Ion Structures, Binding Energies, and Substituent Effects. *J. Phys. Chem. A* **2001**, *105*, 710-719.
- (11) Walter, D.; Sievers, M. R.; Armentrout, P. B. Alkali ion carbonyls: sequential bond energies of $Li^+(CO)_x$ ($x = 1-3$), $Na^+(CO)_x$ ($x = 1, 2$) and $K^+(CO)$. *Int. J. Mass Spectrom. Ion Proc.* **1998**, *175*, 93-106.

- (12) Meyer, F.; Y.-M., C.; Armentrout, P. B. Sequential Bond Energies of Cu(CO)_x⁺ and Ag(CO)_x⁺ ($x = 1-4$). *J. Am. Chem. Soc.* **1995**, *117*, 4071-4081.
- (13) Malischewski, M.; Peryshkov, D. V.; Bukovsky, E. V.; Seppelt, K.; Strauss, S. H. Structures of M₂(SO₂)₆B₁₂F₁₂ (M = Ag, K) and Ag₂(H₂O)₂B₁₂F₁₂: Comparison of the Coordination of SO₂ vs. H₂O and of B₁₂F₁₂²⁻ vs. Other Weakly Coordinating Anions to Metal Ions in the Solid State. *Inorg. Chem.* **2016**, 0000 (in press; doi: 0010.1021/acs.inorgchem.0006b01980).