

Supporting Information

A Fluorene-Based Two-Dimensional Covalent Organic Framework with Thermoelectric Properties through Doping

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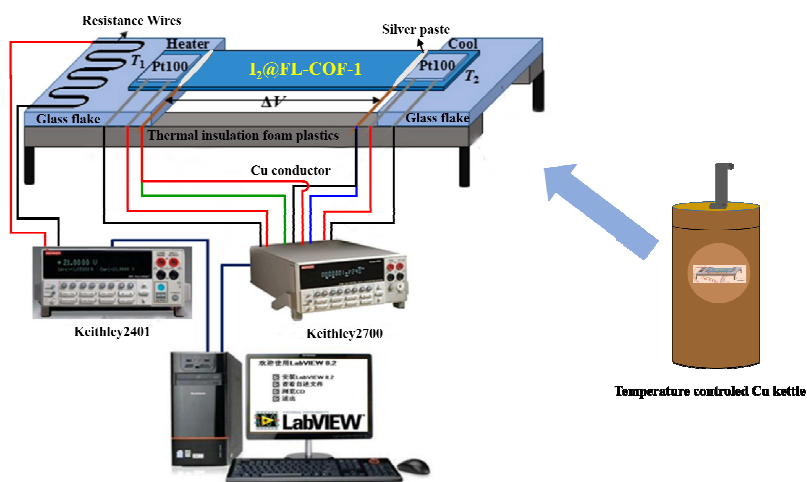
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| Scheme S1 Schematic illustration of the thermopower measurement. | S-3 |
| Figure S1 FTIR spectra of FL-COF-1. | S-4 |
| Figure S2 TEM images of FL-COF-1. | S-5 |
| Figure S3 TGA data of FL-COF-1. | S-6 |
| Figure S4 BET surface area plot of FL-COF-1. | S-7 |
| Figure S5 PXRD pattern of I ₂ @FL-COF-1. | S-8 |
| Figure S6 N ₂ sorptions and BET surface area plot of I ₂ @FL-COF-1. | S-9 |
| Figure S7 FTIR and XRD data of the dedoped I ₂ @FL-COF-1. | S-10 |
| Figure S8 Thermoelectric voltage as a function of ΔT for I ₂ @FL-COF-1. | S-11 |
| Figure S9 Raman spectrum of I ₂ @FL-COF-1. | S-12 |



Scheme S1. Schematic illustration of the thermopower measurement.

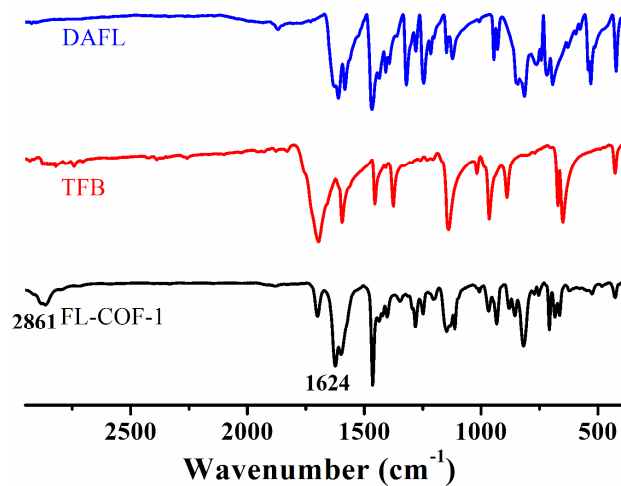


Figure S1. FTIR spectra of FL-COF-1 (black), TFB (red), and DAFL (blue). The appearance of an intensive band at 1624 cm⁻¹ in FL-COF-1 confirmed the formation of new imine bonds (C=N).

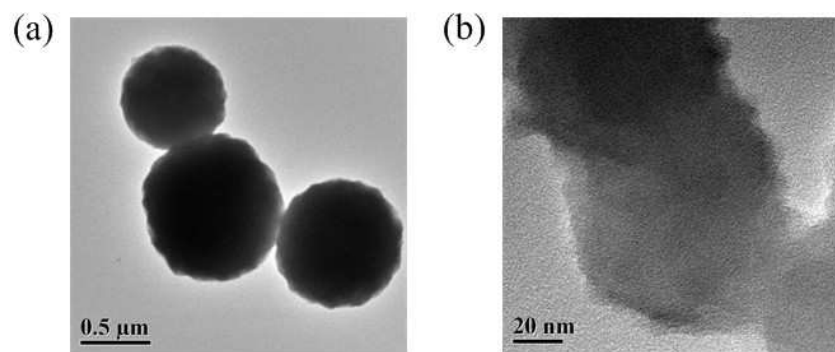


Figure S2. TEM images of FL-COF-1 at different magnifications.

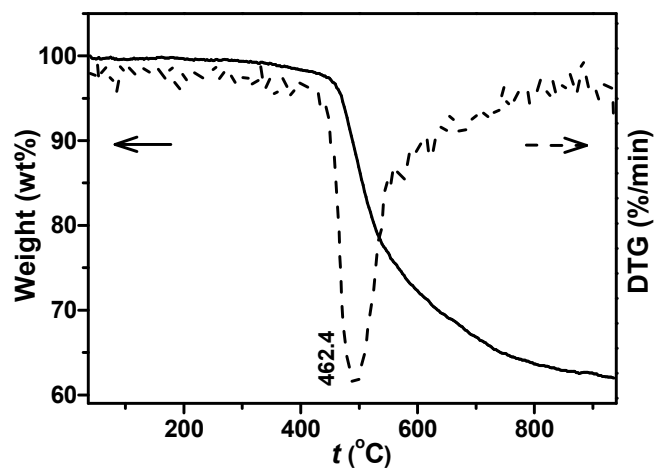


Figure S3. TGA and DTG plots of FL-COF-1 under a nitrogen atmosphere.

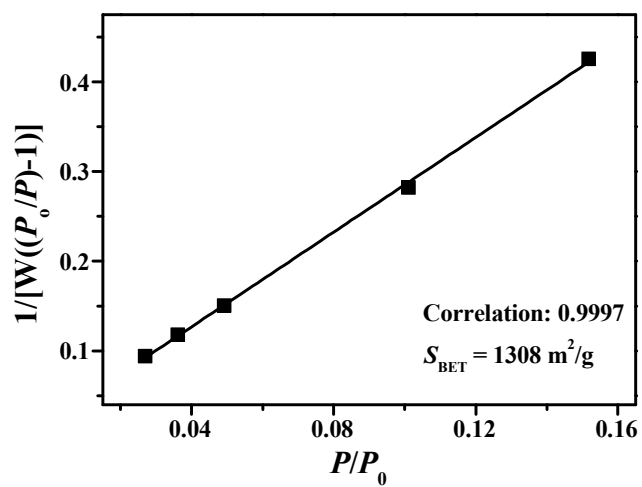


Figure S4. BET linear plot and surface area of FL-COF-1 calculated from the N₂ adsorption isotherm at 77 K.

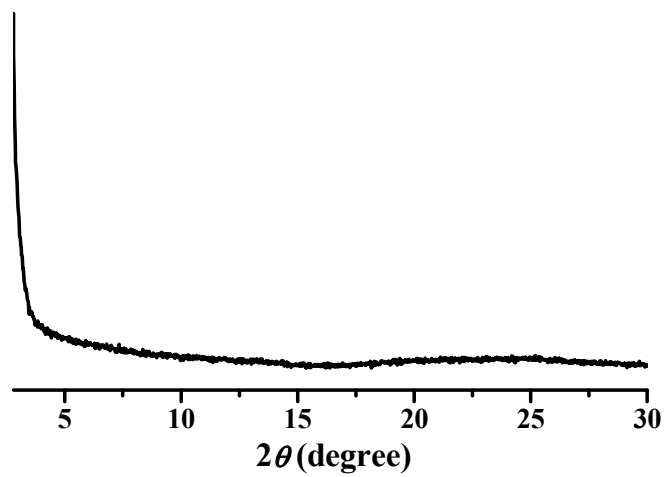


Figure S5. The PXRD pattern of I₂@FL-COF-1.

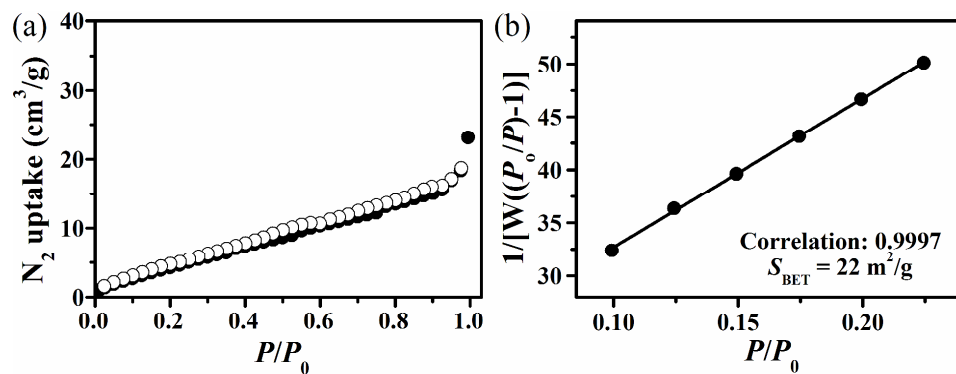


Figure S6. (a) N_2 sorption isotherms measured at 77 K and (b) BET surface area plot of $I_2@FL-COF-1$.

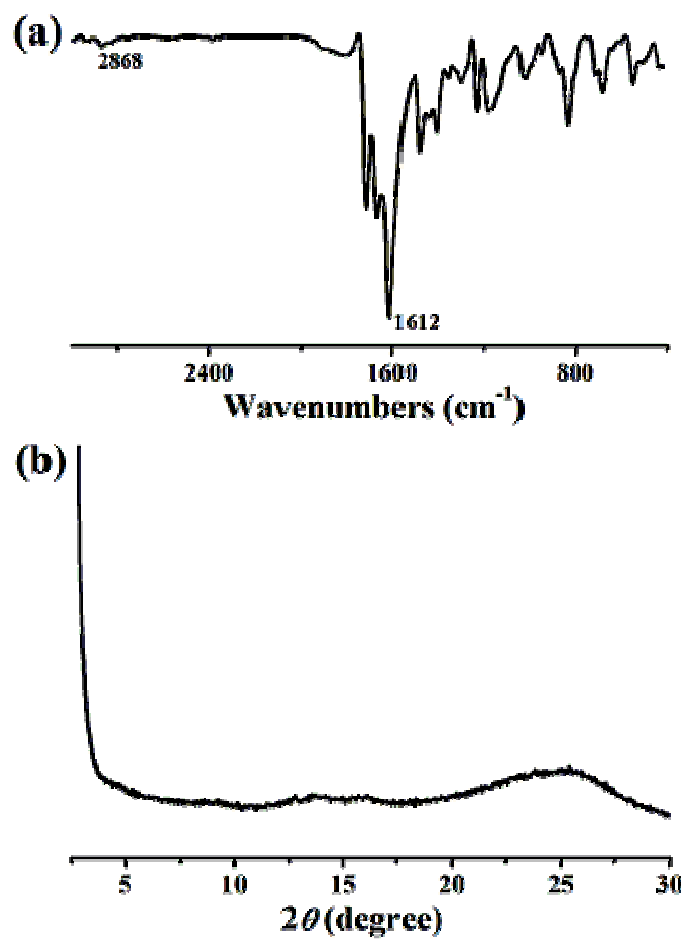


Figure S7. (a) FTIR and (b) XRD data of the dedoped I₂@FL-COF-1.

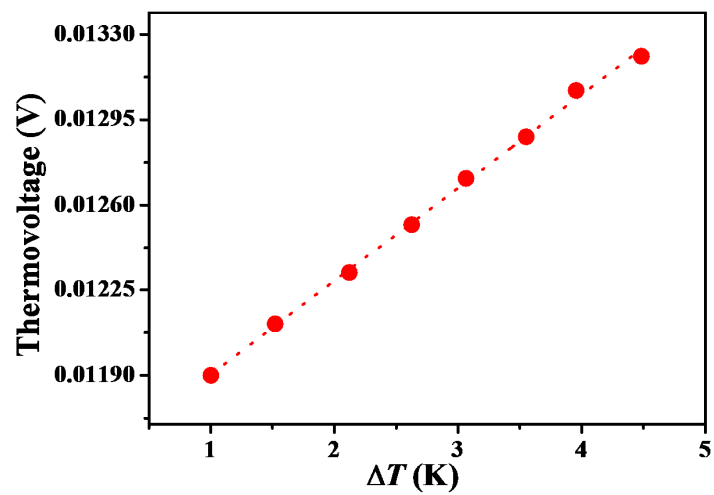


Figure S8. Thermoelectric voltage as a function of ΔT for $\text{I}_2@\text{FL-COF-1}$.

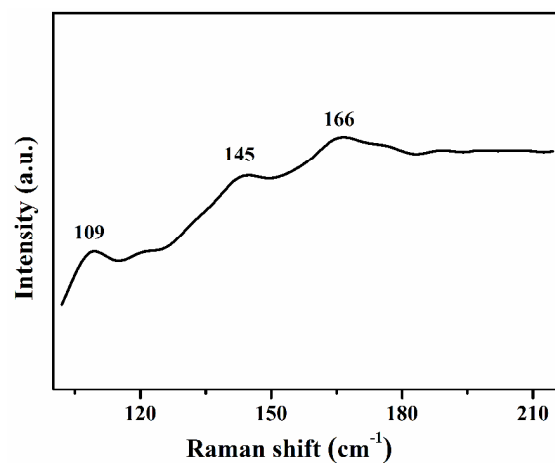


Figure S9. Raman spectrum of $\text{I}_2@\text{FL-COF-1}$. The characteristic peak around 109 cm^{-1} corresponded to the perturbed iodine molecules, whereas the peaks at 145 and 164 cm^{-1} reflected the existence of I^{5-} from I^- ions and di-iodine molecules $[\text{I}^-(\text{I}_2)_2]$.