1	*** Supporting Information ***									
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3	Title: Extracting metallic nanoparticles from soils for quantitative analysis:									
4	Method development using engineered silver nanoparticles and SP-ICP-MS									
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12 13 14 15 16	This document contains:									
17	1. Characterization of the Nanomaterials									
18	 SP-ICP-MS Analytical Details Environmental data in a finite provide the finite provided and the fini									
19	3. Experimental details of the Spike-Recovery Analysis									
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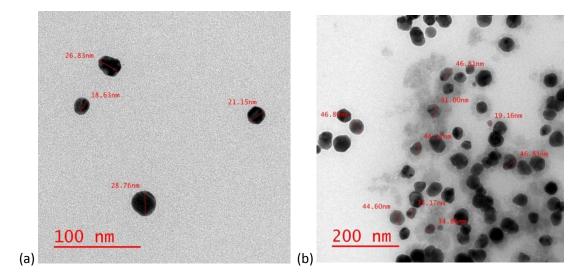
21 Characterization of the Nanomaterials

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AgNP (25 nm), nanoComposix. TEM analysis of the 20 mg L^{-1} stock suspension showed that 23 most particles fell within the supplier's size specifications of 23.1 ± 6.9 nm (Fig. S1a). Dynamic 24 light scattering (DLS) analysis of the same 25 nm stock suspension of the particles agreed with 25 the TEM results and showed a mono-modal size distribution with a z-average hydrated diameter 26 of 26.3 nm (0.3 nm standard deviation) with a polydispersivity index of 0.11. SP-ICP-MS 27 analysis performed on a diluted suspension (10 to 100 ng L⁻¹) revealed that 89% of the total Ag 28 mass consisted of detectable particles (i.e. ≥ 19 nm) showing a mono-modal size distribution 29 30 with mode and mean diameters of 24.8 and 30.4 nm.

AgNP (40 nm), nanoComposix. TEM analysis of the 15 g L^{-1} stock suspension showed that most 31 particles fell within the supplier's size specifications of 38.6 ± 9.8 nm with a minor presence of 32 particles approximately \leq 19 nm (Fig. S1b). DLS analysis was conducted on a diluted suspension 33 (15 mg L^{-1}) and showed a bimodal particle size distribution with 9% of the total light intensity 34 arising from the smaller ($\sim < 15$ nm) particle size fraction (Fig. S2). Note: z-average particle 35 sizes derived from poly-modal distributions are not accurate. SP-ICP-MS analysis performed on 36 an initial diluted suspension (~ 10 to 100 ng L^{-1}) revealed that 93% of the total Ag mass 37 consisted of detectable particles showing a mono-modal size distribution with mode and mean 38 particle diameters of 35.1 and 43.4 nm (Fig. S3). However, analysis of the suspension 48 h later 39 showed that only 35% of the total Ag mass consisted of detectable particles and mode and mean 40 particle sizes were slightly decreased to 31.2 and 39.8 nm. Presumably, the 65% of the Ag mass 41 comprising the "dissolved" fraction of the SP-ICP-MS analysis consisted not only of dissolved 42 Ag, but of particles below the particle size detection limit of 19 nm. This shift in the particle size 43 distribution and increase of Ag in the dissolved fraction after 48 h was likely because of particle 44

dissolution. It should be noted that the spiked soil samples used in this study were amended with the original 15 g L^{-1} 40 nm stock suspension.

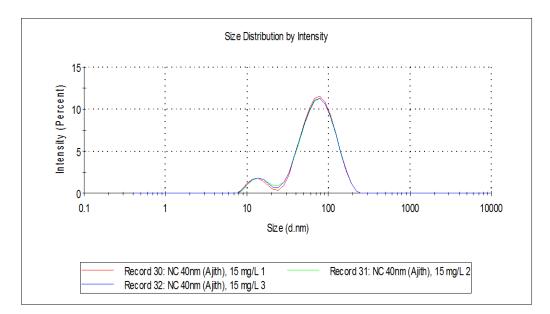


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Figure S1. TEM images of the 25 nm AgNP prepared from a 20 mg L^{-1} aqueous suspension (a), and the

- 51 40 nm AgNP prepared from a 15 g L^{-1} aqueous suspension (b). Imaging was conducted using an FEITM
- 52 Tecnai G2 F20 TEM (FEI, Hillsboro OR) at the Carleton University Nano Imaging Facility.
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Figure S2. Particle size distribution by DLS (based on intensity) of the 40 nm AgNP determined on three replicate samples of a 15 mg L^{-1} suspension.

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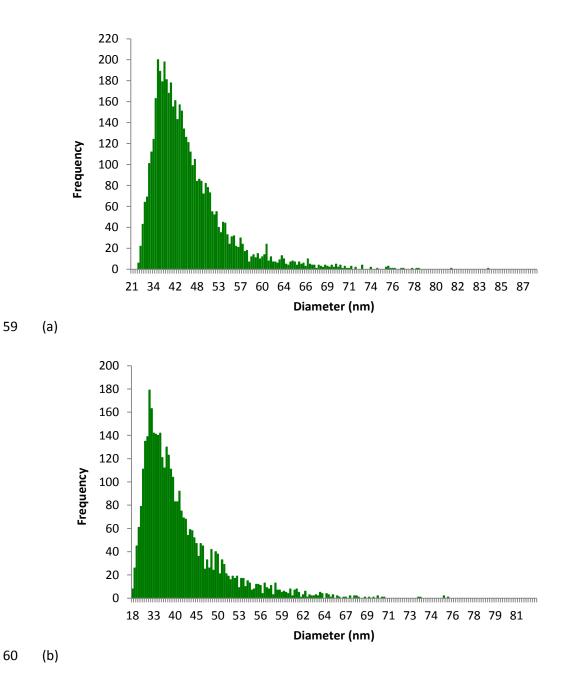


Figure S3. SP-ICP-MS particle size histograms of the 40 nm AgNP material performed on suspensions in the 10 to 100 ng L^{-1} range analyzed fresh (a) and analysed after 48 h (b).

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SP-ICP-MS analysis requires that samples are sufficiently dilute such that analytical coincidence is avoided (i.e., two or more particles making up a plasma event). Based on our operating parameters, we estimated that coincidence could be avoided if sample particle concentrations were $\leq 1.2 \times 10^5 \text{ mL}^{-1}$; therefore target particle concentrations for calibration suspensions and samples were approximately 10⁵ particles mL⁻¹. Data processing was performed within the SyngistixTM Nano-Application module. The software derives the intensity threshold used to discriminate a particle signal from a dissolved signal using an iterative "mean + 3σ " computation, similar to that described by Hadioui et al.¹ As quantification of both the dissolved and nano-particulate forms of Ag are of interest, capturing both dissolved and particle signals is necessary; however, it is doubtful that one dilution/analysis can optimally capture both signals. Therefore samples were analyzed using a dilution series previously described in Schwertfeger et al.² Briefly, soil extracts were first diluted to capture the dissolved analyte reading within the dissolved calibration range and above the limit of quantitation (LoO). A second or "optimal" dilution was used to optimize the particle signal by (i) reducing the dissolved signal to near background levels. (ii) reducing particle counts (i.e., $< 1.2 \times 10^5 \text{ mL}^{-1}$) to minimize coincidence. and (iii) ensuring that the number of particles detected was sufficient for statistical power for describing the particle distributions (generally \geq 1000 particles), which sometimes resulted in our

extending the sampling time. Finally, a third dilution was analyzed, (usually 2x the optimal dilution), which was used as a quality control. For example, a 2x dilution should not affect particle intensity, but particle counts should be approximately half, otherwise samples were sub-

sampled again, checked for matrix interferences, and re-analyzed. Each dilution was prepared 85

such that the ionic strength of the sample matrix was maintained. This was done by using an

- 87 appropriate dilution of the extracting reagent. Dilutions were analyzed immediately.
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89 **References**

90 (1) Hadioui, M.; Peyrot, C.;Wilkinson, K. J., Improvements to Single Particle ICPMS by the Online
91 Coupling of Ion Exchange Resins. *Anal. Chem.* 2014, *86* (10), 4668-4674.
92 (2) Schwertfeger, D. M.; Velicogna, J.; Jesmer, A.; Scroggins, R. P.;Princz, J. I., SP-ICP-MS Analysis
93 of Metallic Nanoparticles in Environmental Samples with Large Dissolved Analyte Fractions. *Anal.*94 *Chem.* 2016. DOI: 10.1021/acs.analchem.6b02716

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96 Spike-Recovery Analysis

97 Experimental Set-up. Ag recovery for the optimal extraction procedure was investigated using a control soil sample to which AgNP was added. Triplicate samples were prepared by adding an 98 aliquot of a 150 mg L⁻¹ AgNP (40 nm) suspension to 50-mL test tubes containing 0.5 g soil and 99 50 mL 2.5 mM TSPP to achieve a nominal total Ag concentrations of 1.5 mg L^{-1} which 100 101 corresponded to 150 µg Ag per g soil. The spiked samples were treated as regular samples and followed the finalized method protocol: 30 min mixing, 1.5 min ultrasonication with the wand, 102 18 h settling, followed by sampling extracts between a depth of 0.5 to 1.0 cm from sample 103 surface and diluting 100x in NanopureTM water within minutes prior to initial SP-ICP-MS 104 analysis. Triplicate AgNP-spiked method blanks were also prepared in the same manner as 105 above, except they did not contain any soil (i.e., TSPP reagent only). The reference sample to 106 which Ag recovery in the spiked samples/blanks was assessed was a 1.5 mg L⁻¹ AgNP 107 suspension prepared fresh from the 15 g L^{-1} stock suspension and analyzed immediately. 108 Recovery was assessed based on three parameters: particle number concentration (particles mL⁻ 109 ¹), particulate Ag mass ($\mu g L^{-1}$) and total Ag mass ($\mu g L^{-1}$) (i.e., particulate Ag mass + dissolved 110 Ag mass) obtained by SP-ICP-MS. 111



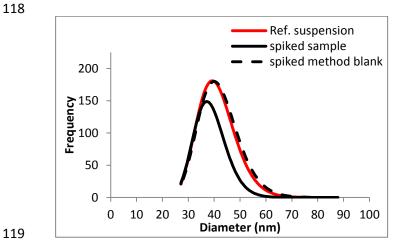
Table S1. SP-ICP-MS results for the spike-recovery samples. Bracketed values show the standard

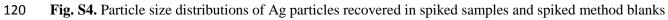
115 deviation derived from three replicate samples.

Sample	nominal total Ag conc.	measured total Ag conc.		particulate Ag mass conc.		U U		Particle # conc.		mean part. size
	$\mu g L^{-1}$	μg L ⁻¹		μg L ⁻¹		μg L ⁻¹		particles mL ⁻¹		nm
AgNP Ref. suspension	1500	1443	(86)	1201	(32)	242	(19)	2.4E+09	(5E+07)	43.4
spiked soil sample	1500	1397	(165)	1018	(48)	379	(118)	2.0E+09	(7E+07)	42.7
spiked method blank	1500	1523	(127)	1218	(38)	345	(165)	2.5E+09	(6E+07)	42.8

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- 121 compared to distribution of the reference sample. Note: these distributions represent one of the three
- 122 replicates samples.