Supporting Information

A Phenylselenium-Substituted BODIPY Fluorescent Turn-off Probe for Fluorescence Imaging of Hydrogen Sulfide in Living Cells

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Table S1. Spectroscopic/photophysical data of 1 (10 μ M) in addition of Na₂S.(Spectrum of 1+ Na₂S was recorded 20 min after Na₂S addition in TritonX-100/DMSO/HEPES buffer (0.01: 1: 9, v/v/v, 10 mM, pH 7.4) at 37 °C.)

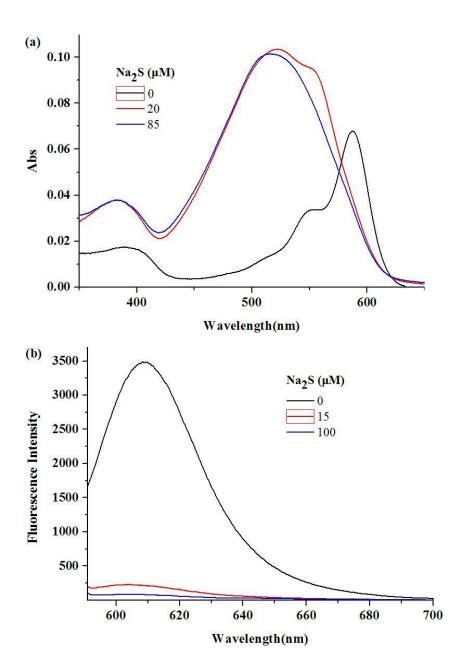
Compounds	Reation time/min	λ _{abs} (max)/ nm	$\lambda_{ems}(max)/nm$	ф
1	—	587	610	0.299
$1 + Na_2S$	20	516	602	0.008

Linear range	Detection	React	Reference
	Limit	time	
0-100 μM	2.46 µM	40min	Anal.Chem., 2016, 88,
			5476–5481
0-15 μM	0.05µM	10min	Chem. Commun.,2012,
			48,10529–10531.
0-8 µM	0.007µM	140 s	J. Am. Chem. Soc., 2015, 137,
			8490-8498
0-10 µM	0.01µM	15min	Chem. Commun., 2013, 49,
			403–405.
0-300 µM	0.0682µM	2min	Chem. Commun., 2015, 51,
			16225-16228
0-125 μM	0.38 µM	30min	Chem. Commun., 2015, 51,
			10463-10466.
0-16 µM	0.086 µM	90min	J. Am. Chem.
			Soc., 2015, 137,10216–10223
1.0-30 µM	0.052µM	60min	Anal. Chem., 2016, 88,
			1434–1439.
0-30 µM	0.85 µM	20min	Anal. Chem., 2015, 87,
			2678–2684
0-15 μM	0.0025 µM	20min	this work

Table S3. Photophysical Properties of **1** (10 μ M) in the absence and in addition of Na₂S in Triton X-100/DMSO/HEPES buffer (0.01: 1: 9, v/v/v, 10 mM, pH 7.4) at 37

Addition	Monitoe	τ_1/ns	τ_2/ns	τ_3/ns	α ₁ (%)	$\alpha_2(\%)$	$\alpha_{3}(\%)$
of Na ₂ S	d						
$/\mu M$	waveleng						
	th/ nm						
	590				1%	99%	-
0	610	2.73 ± 0.01	1.04 ± 0.01	-	4%	96%	-
	630				5%	95%	-
	650				5%	95%	-
	590				22%	45%	33%
10	610	3.26±0.01	1.31±0.01	0.49 ± 0.03	26%	54%	20%
	630				31%	51%	18%
	650				35%	44%	21%
	590				27%	40%	33%
25	610	3.34 ± 0.01	1.47 ± 0.01	0.48 ± 0.02	28%	47%	25%
	630				30%	47%	23%
	650				31%	45%	24%
	590				42%	42%	16%
100	610	2.99 ± 0.01	1.28 ± 0.01	0.44 ± 0.06	40%	43%	17%
	630				42%	43%	15%
	650				42%	47%	11%

^oC. Global analyses of decay times τ_1 , τ_2 and τ_3 , and the relative amplitude α_i (%), each spectrum was recorded 20 min after Na₂S addition at the same excitation wavelength (560 nm), but at different emission wavelength (590, 610, 630 and 650 nm).



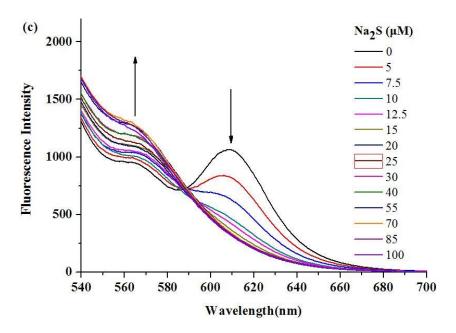


Figure S1. (a) Absorption and (b) fluorescent spectra of **1** (10 μ M) in the absence and presence of different concentration of Na₂S. Each data point was acquired 20 min after addition in Triton X-100/DMSO/HEPES buffer (0.01: 1: 9, v/v/v, 10 mM, pH 7.4) at 37 °C. $\lambda_{ex} = 582$ nm. (c) fluorescence spectral changes of **1** (10 μ M) upon addition of Na₂S. Each spectrum was recorded 20 min after Na₂S addition. $\lambda_{ex} = 500$ nm.

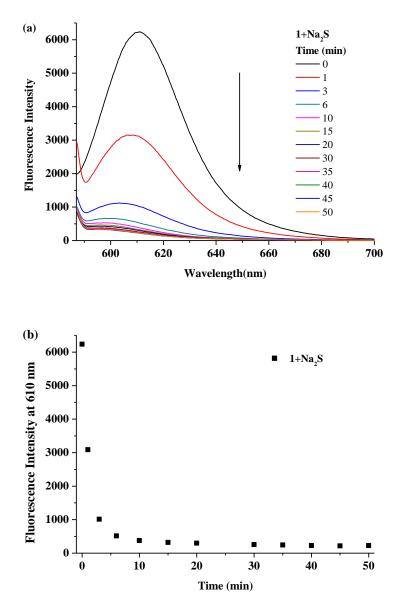
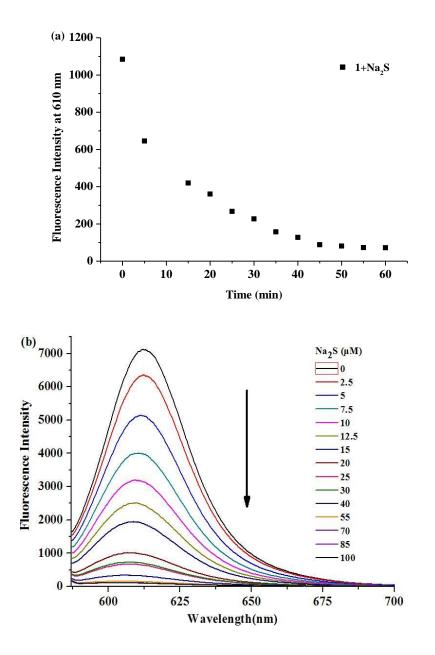


Figure S2. (a) Time-dependent fluorescence spectra of **1** (10 μ M) with 10 equiv. of Na₂S in Triton X-100/DMSO/HEPES buffer (0.01: 1: 9, v/v/v, 10 mM, pH 7.4) at 37 °C. (b) is time course of the response at 610 nm of (a), $\lambda_{ex} = 582$ nm.



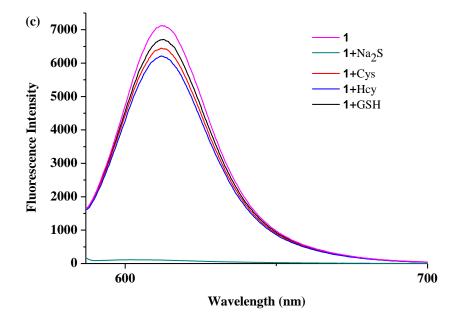


Figure S3. (a) Time course of the response at 610 nm of time-dependent fluorescence spectra of **1** (10 μ M) with 10 equiv. of Na₂S. (b) Fluorescence spectral changes of **1** (10 μ M) upon addition of Na₂S. (c) Emission response of **1** (10 μ M) upon addition of 10 equiv. of biothiols and Na₂S. Each spectrum was recorded 50 min after Na₂S addition in DMSO/HEPES buffer (1: 1, v/v, 10 mM, pH 7.4) at 37 °C, $\lambda_{ex} = 582$ nm.)

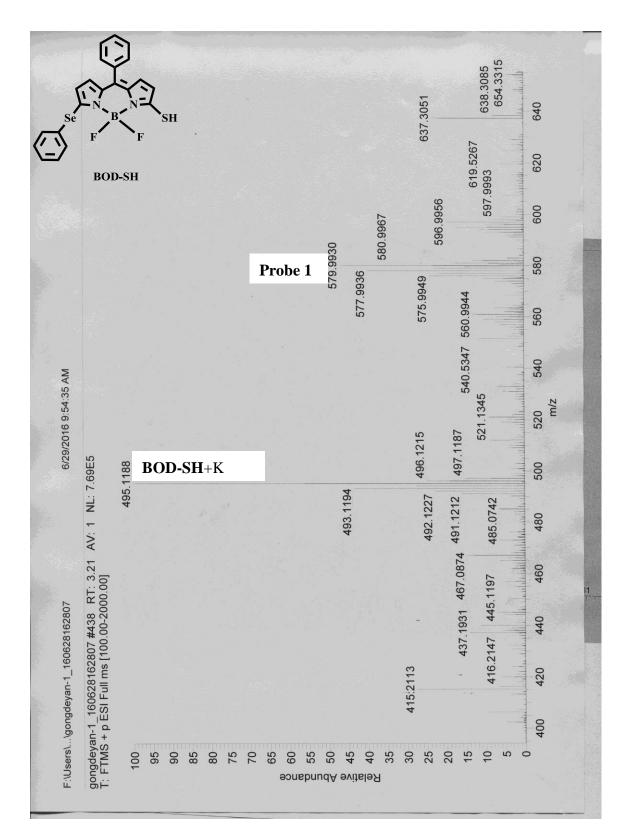


Figure S4. MS Spectrum of **1** in EtOH with Na₂S (1 equiv.) at 2 h reaction time at 25 °C. The predicted product is **BOD-SH** (the molecular weight of [**BOD-SH**] ($C_{21}H_{15}BF_2N_2SSe$) is 456.0). The peaks at m/z 495.1 (calcd = 495.0) corresponding to [**BOD-SH**+K].

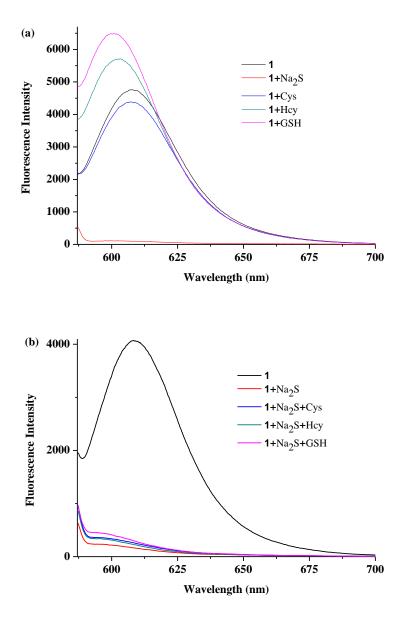


Figure S5. (a) Emission response of 1 (10 μ M) upon addition of 10 equiv. of biothiols and Na₂S. (b) the competition graph of 1 (10 μ M) in the presence of 10 equiv.of biothiols with Na₂S. Each data point was acquired 50 min after addition at 37 °C. λ_{ex} = 582 nm.

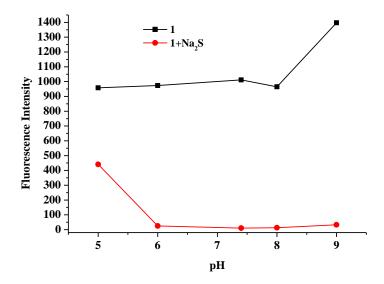


Figure S6. Fluorescence intensity of **1** at 610 nm in the absence and presence of 10 equiv. of Na₂S as a function of pH. **1** = 10 μ M, λ_{ex} = 582 nm. Each data point was acquired 50 min after addition of Na₂S in Triton X-100/DMSO/HEPES buffer (0.01: 1: 9, v/v/v, 10 mM, pH 7.4) at 37 °C.

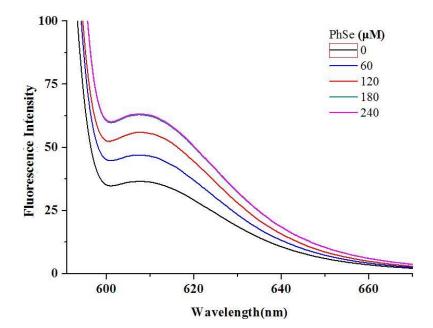
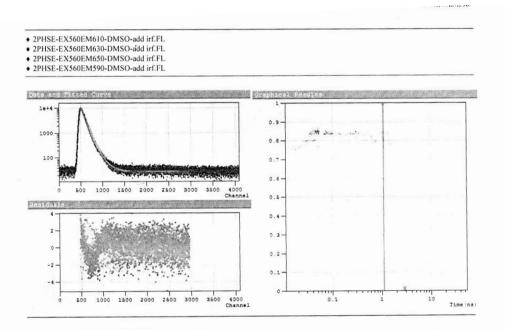
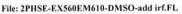


Figure S7. The fluorescence spectral changes of the **BOD-SH** mixture solution upon addition of different concentration of benzeneselenol. Each data point was acquired 20 min after addition of benzeneselenol in Triton X-100/DMSO/HEPES buffer (0.01: 1: 9, v/v/v, 10 mM, pH 7.4) at 37 °C, $\lambda_{ex} = 582$ nm.



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Slobal Analysis (Reconvolution)

Fitting rar Global X X ⁴	: 1	475; 2950] .314 .100	channels			
	Bi	ΔB _i	f _i (%)	Δf _i (%)	τ _i (ns)	$\Delta \tau_i (ns)$
1	0.0282	4.2e-5	96.28	0.2803	1.037 linked	0.0013
2	0.0004	2.7e-5	3.7242	0.2645	2.733 linked	0.0072
Shift Decay Backgrou IRF Back	ind :	0.0026 ns (= 27.27 (= 12.40	±0ns) ±0)			

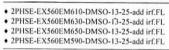
File: 2PHSE-EX560EM630-DMSO-add irf.FL

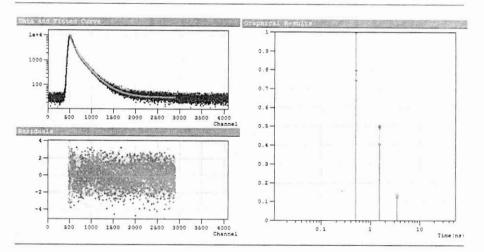
Global Analysis (Reconvolution)

Fitting rar Global X X ²	· :	[475; 2950] 1.314 1.454	channels			
	B	ΔB _i	f _i (%)	Δf _i (%)	τ _i (ns)	$\Delta \tau_i (ns)$
1	0.0281	5.5e-5	94.85	0.3920	1.037 linked	0.0015
2	0.0006	3.9e-5	5.1468	0.4331	2.733 linked	0.0082
Shift Decay Backgrou IRF Back	ind :	0.0074 ns (= 23.17 (= 12.40	±0ns) ±0)			

		onvolution				
Fitting ran Global X		475; 2950] .314	channels			
χ^2		.243				
~	Bi	ΔΒ	f ; (%)	Δf_i (%)	τ _i (ns)	$\Delta \tau_{i}$ (ns)
1	0.0282	5.3e-5	94.56	0.3410	1.037 linked	0.0014
2	0.0282	3.8e-5	5.4400	0.3890	2.733 linked	0.0076
Shift		0.0082 ns (=				
Decay Backgrou IRF Back	na	26.88 (= 12.40	±0)			
Backgrou IRF Back	na ground : -EX560EM nalysis (Rec nge : [12.40 1590-DM	SO-add i	rf.FL		
Backgrou IRF Back ile: 2PHSE Global A Fitting ra	na ground : -EX560EM nalysis (Rec nge : [t' :]	12.40 1590-DM convolution (475; 2950]	SO-add i	rf.FL		
Backgrou IRF Back ile: 2PHSE Global A Fitting ra Global	na ground : -EX560EM nalysis (Rec nge : [t' :]	12.40 M590-DM convolution (475; 2950] 1.314	SO-add i	rf.FL Δf _i (%)	τ _i (ns)	Δτ _i (ns)
Backgrou IRF Back ile: 2PHSE Global A Fitting ra Global	-EX560EN nalysis (Rec nge : [c ⁴ : 1 : 1	12.40 M590-DM convolution (475; 2950) 1.314 1.458	SO-add i		τ ₁ (ns) 1.037 linked	

Figure S8. Global analysis of decay times result of **1** (10 μ M) in Triton X-100/DMSO/HEPES buffer (0.01: 1: 9, v/v/v, 10 mM, pH 7.4) at 37 °C, each spectrum was recorded at the same excitation wavelength (560 nm), but at different emission wavelength (590, 610, 630 and 650 nm).





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* Global Analysis (Reconvolution)

Fitting ra Global) χ ⁴	¢ : I	475; 2890] .109 .057	channels			
	Bi	ΔB _i	f _i (%)	Δf _i (%)	τ _i (ns)	$\Delta \tau_i$ (ns)
1	0.0177	0.0002	24.90	1.6047	0.483 linked	0.0246
2	0.0109	0.0001	47.00	0.9672	1.473 linked	0.0098
3	0.0029	10.0e-5	28.10	1.0211	3.339 linked	0.0025
Shift Decay Backgrou IRF Back	ind : 2	-0.1332 ns (29.74 (12.40	± 0 ns) ± 0)			

File: 2PHSE-EX560EM630-DMSO-13-25-add irf.FL

Global Analysis (Reconvolution)

			in the second seco	475; 2890] (.109 .087	· : I	Fitting rar Global χ χ ²
$\Delta \tau_i (ns)$	τ _i (ns)	Δf _i (%)	f _i (%)	ΔBi	Bi	
nked 0.0250	0.483 linked	1.5532	23.07	0.0002	0.0165	1
nked 0.0099	1.473 linked	0.9732	47.45	0.0001	0.0111	2
nked 0.0026	3.339 linked	1.0904	29.48	0.0001	0.0031	3
lir	3.339	1.0904		0.0001 0.1182 ns (:	and a second second second	3 Shift Decay

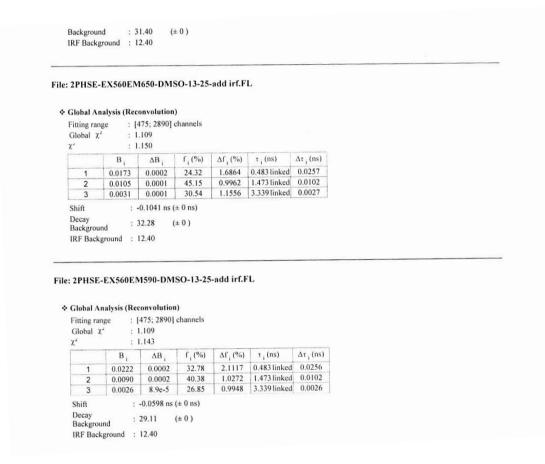


Figure S9. Global analysis of decay times result of **1** (10 μ M) in addition of 25 μ M Na₂S in Triton X-100/DMSO/HEPES buffer (0.01: 1: 9, v/v/v, 10 mM, pH 7.4) at 37 °C, each spectrum was recorded 20 min after Na₂S addition at the same excitation wavelength (560 nm), but at different emission wavelength (590, 610, 630 and 650 nm).

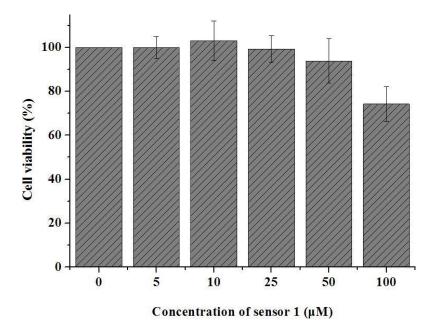
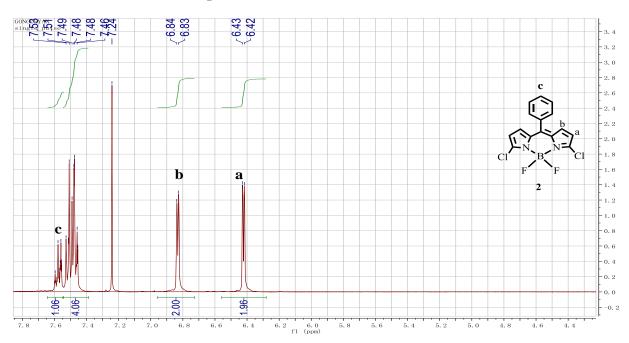


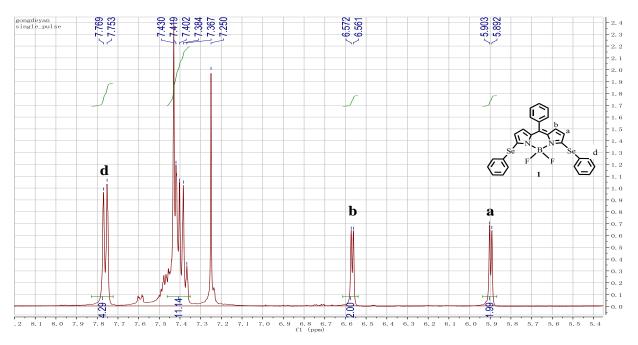
Figure S10. Effects of 1 at varied concentrations on the viability of BHK cells. The

cell viability data were checked three times.

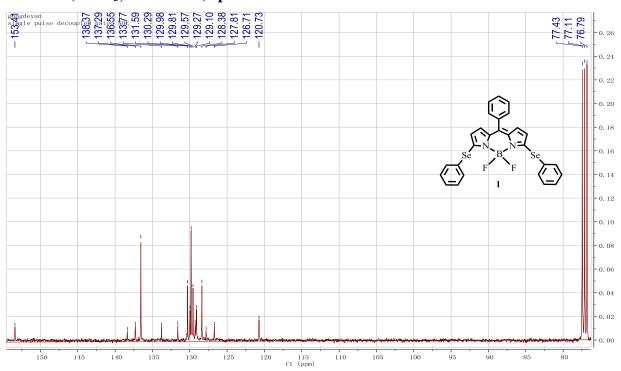
¹H NMR (CDCl₃, 400 MHz) spectrum of 2

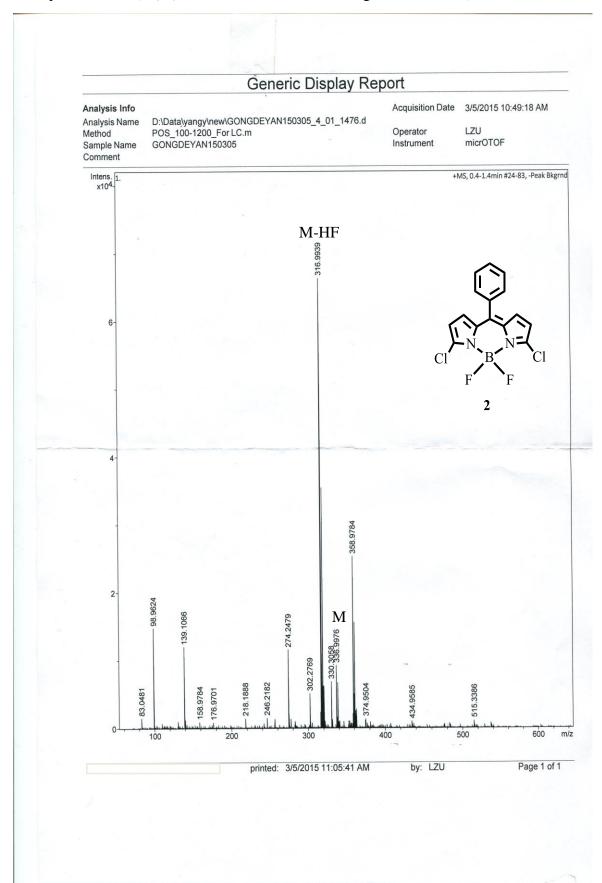


¹H NMR (CDCl₃, 400 MHz) spectrum of 1

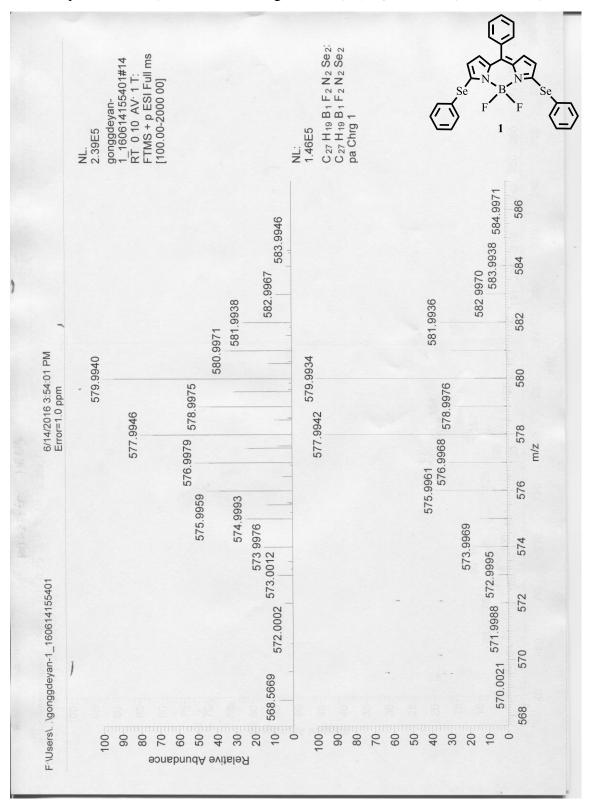


¹³C NMR (CDCl₃, 400/4 MHz) spectrum of 1





MS Spectrum of 2 (C₁₅H₉BCl₂F₂N₂, the molecular weight of 2 is 336.9).



HRMS Spectrum of 1 (the molecular weight of [1] $(C_{17}H_{13}BClF_2N_4O)$ is 579.9940).