Supporting Information

# Process Safety Considerations for the Use of 1 M Borane Tetrahydrofuran Complex Under General Purpose Plant Conditions

Alexandre M. Monteiro<sup>a</sup> and Roy C. Flanagan<sup>b</sup>

<sup>a</sup>Process Safety, GlaxoSmithKline R&D, Collegeville, PA 19426

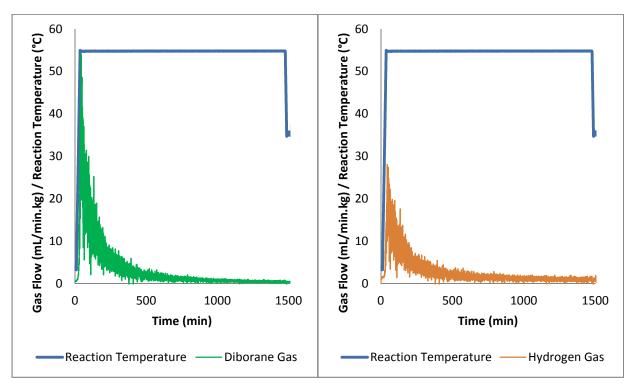
<sup>b</sup>Process Safety, GlaxoSmithKline R&D, Zebulon, NC 27597

#### Table of Contents

1.	Headspace mass spectroscopy traces for BTHF aging studies	S-2
2.	Heat flow trace for the addition of acetone to fresh BTHF	S-4
3.	ARC traces for fresh BTHF	S-5

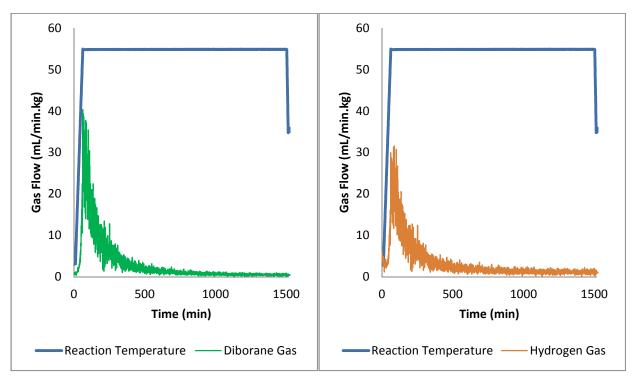
## Headspace mass spectroscopy traces for BTHF aging studies

Note: Gas flow data normalized to the amount of fresh BTHF charged.

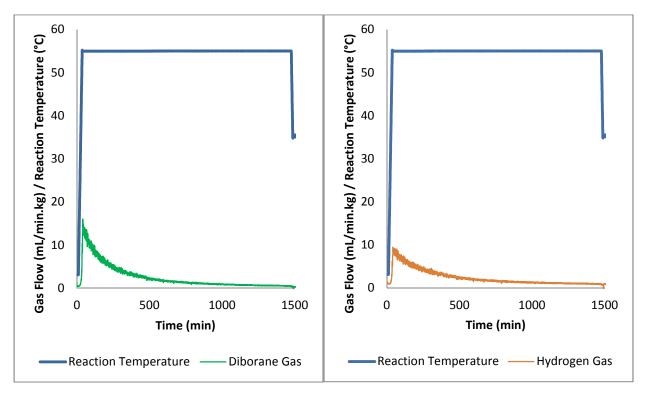




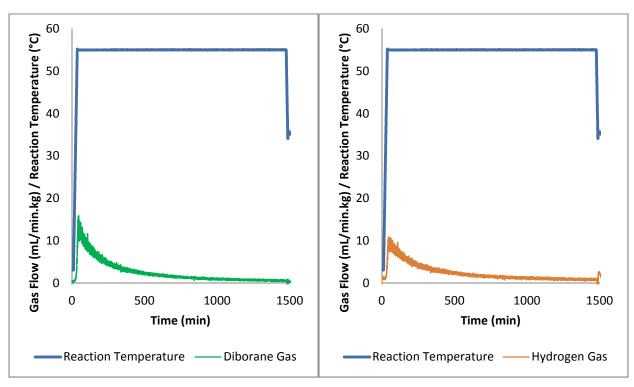






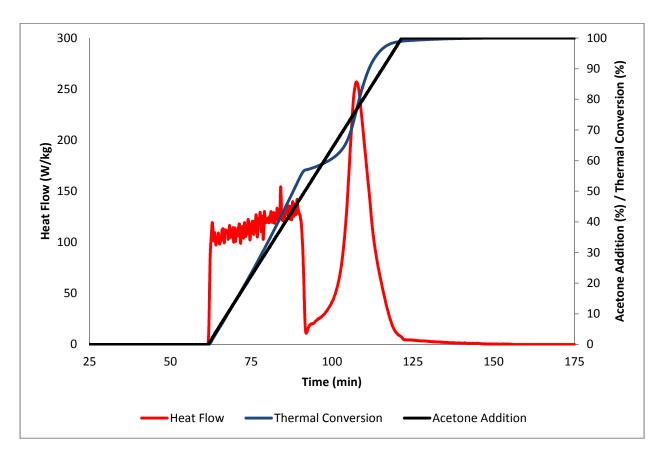






#### Heat flow trace for the addition of acetone to fresh BTHF

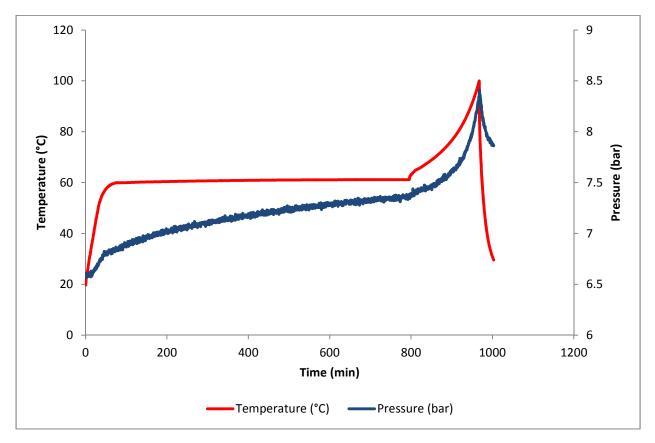
Note: Heat flow data is normalized to the amount of BTHF charged



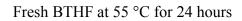
The destruction of fresh BTHF is performed at 35 °C and uses an excess of acetone. Two distinct exothermic events are observed in the overall heat flow profile during the addition of acetone, with heat flow quickly returning to baseline after the addition is complete. The reaction mixture is then aged, followed by the addition of isopropyl alcohol. No additional heat output or significant gas evolution is observed during the age period or IPA addition, confirming the complete destruction of BTHF. Water is added, driving the reaction to the final boric acid species.

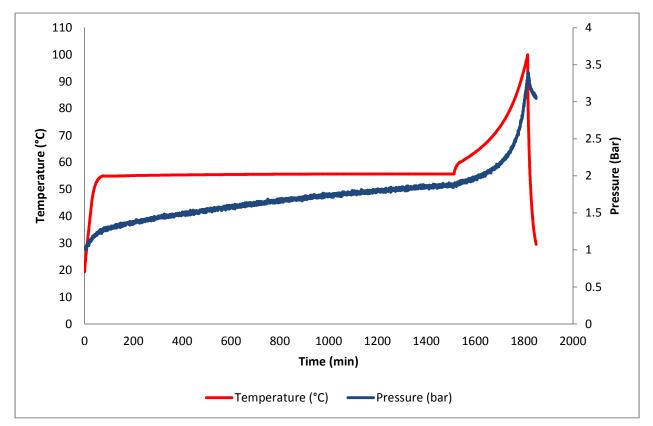
### **ARC traces for fresh BTHF**

Fresh BTHF at 60 °C for 12 hours



On completing the isothermal aging period of 12 hours at 60 °C, the sample went into exothermic tracking immediately upon achieving the next heating step.





On completing the isothermal aging period of 24 hours at 55 °C, the sample went into exothermic tracking immediately upon achieving the next heating step.