Hydrochlorination of Ruthenaphosphaalkenyls: Unexpectedly Facile Access to Alkylchlorohydrophosphane Complexes.

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Computational Details

Calculations were performed using Gaussian 09W, Revision C.01,³⁹ running on an Intel Core i5-2500 (quad, 3.3 GHz), equipped with 4 GB RAM; results were visualized using GaussView 5.0. Geometries were optimized with the hybrid density functional B3LYP, using the RECP basis set Lanl2dz for Ru and 6-31G** for all other atoms. Minima were characterized by frequency calculations at the same level of theory. Single point energy calculations were subsequently performed with the B3LYP functional, using Lanl2dz for Ru and the 6-311+G** basis set for all other atoms; NBO calculations were performed at the same level of theory.

The supplemental file RuPHCl.xyz contains the computed Cartesian coordinates of compounds 1, 2a, 3, 4a, 6 and proposed intermediary A. The file may be opened as a text file to read the coordinates, or opened directly by a molecular modelling program such as Mercury (version 3.3 or later, http://www.ccdc.cam.ac.uk/pages/Home.aspx) for visualization and analysis.

Table S1: Electronic energies of compounds 1, 2a, 3, 4a, 6 and intermediary A (defined as 'X'), and summated energies with one and two equivalents of HCl.

	E(X)		E(X + HCI)		E(X + 2HCl)	
X	/a.u.	/kcal mol ⁻¹	/a.u.	/kcal mol ⁻¹	/a.u.	/kcal mol ⁻¹
HCl	-460.8338442	-289177.846				
1	-3278.596984	-2057352.394	-3739.430829	-2346530.239	-4200.264673	-2635708.085
2 a	-3530.044816	-2215138.422	-3990.878660	-2504316.268	-4451.712504	-2793494.113
Α	-3990.904856	-2504332.706	-4451.738700	-2793510.552		
3	-3739.457921	-2346547.240	-4200.291765	-2635725.085		
4a	-4451.791213	-2793543.504				
6	-4200.343814	-2635757.747				

Table S2: Comparative electronic energies (kcal mol⁻¹) and ΔE_{react} for additions of HCl to 'X'.

	E(X + nHCl)	E(Y)	ΔE_react
$X + nHCI \rightarrow Y$	/kcal mol ⁻¹	/kcal mol ⁻¹	/kcal mol ⁻¹
$1 + HCI \rightarrow 3$	-2346530.239	-2346547.240	-17.001
$2a + HCI \rightarrow A$	-2504316.268	-2504332.706	-16.438
$A + HCI \rightarrow 4a$	-2793510.552	-2793543.504	-32.952
$3 + HCI \rightarrow 6$	-2635725.085	-2635757.747	-32.661
$2a + 2HCl \rightarrow 4a$	-2793494.113	-2793543.504	-49.390
1 + 2HCl → 6	-2635708.085	-2635757.747	-49.662

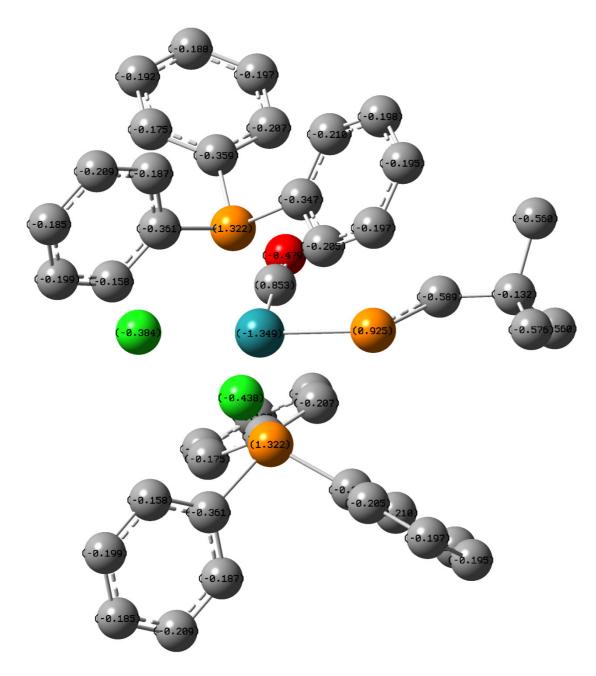


Figure S1: Distribution of NBO charge density for compound 3.

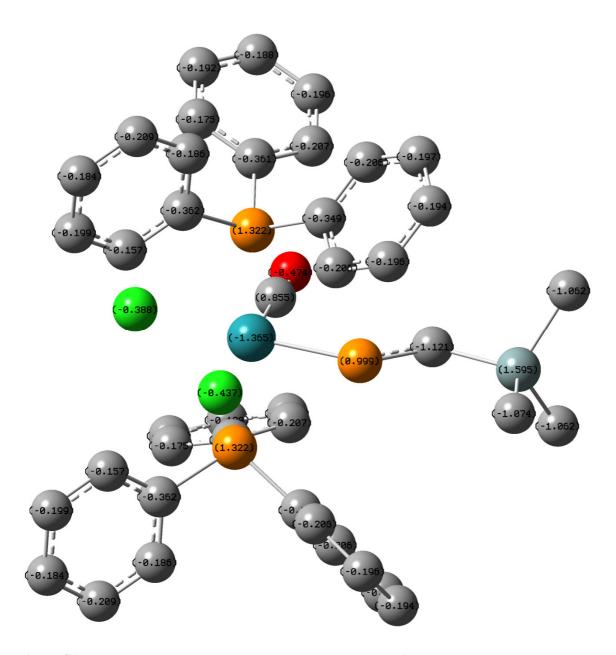


Figure S2: Distribution of NBO charge density for intermediary A.

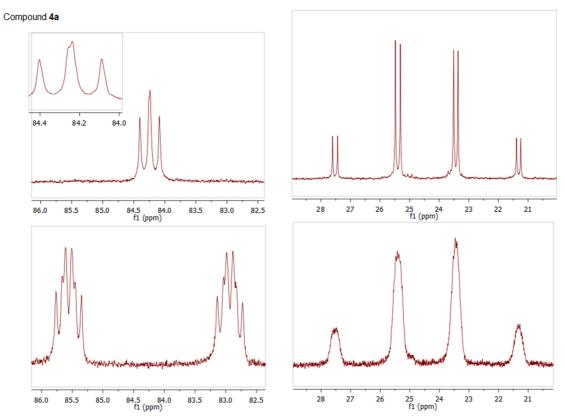


Figure S3. ³¹P{¹H} (top) and ³¹P (bottom) NMR signatures for [Ru(P(H)ClCH₂SiMe₃)Cl₂(CO)(PPh₃)₂] (**4a**) plot on equivalent scales. The multiplicity of the chlorophosphane resonances is illustrated as inset for the proton decoupled spectrum.

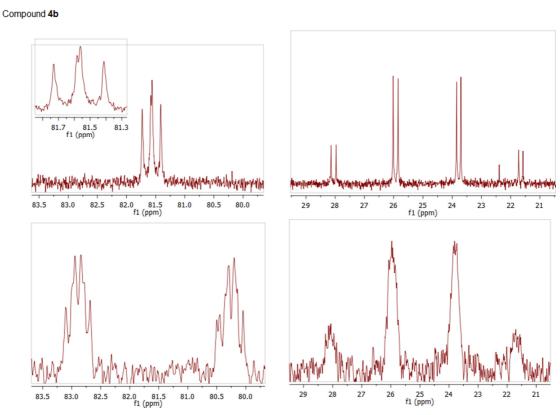


Figure S4. ³¹P{ ¹H} (top) and ³¹P (bottom) NMR signatures for [Ru(P(H)ClCH₂SiMe₂Ph)Cl₂(CO)(PPh₃)₂] (**4b**) plot on equivalent scales. The multiplicity of the chlorophosphane resonances is illustrated as inset for the proton decoupled spectrum.

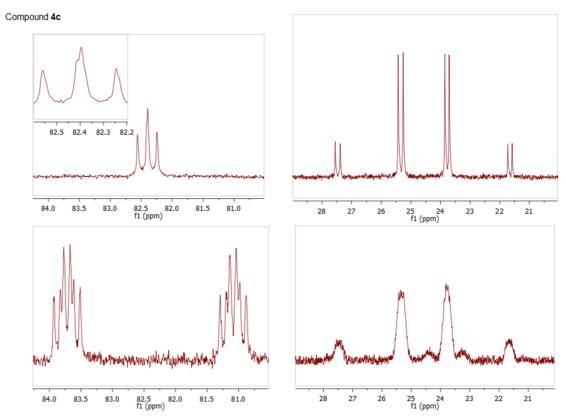


Figure S5. ³¹P{¹H} (top) and ³¹P (bottom) NMR signatures for [Ru(P(H)ClCH₂SiMe₂Tol)Cl₂(CO)(PPh₃)₂] (**4c**) plot on equivalent scales. The multiplicity of the chlorophosphane resonances is illustrated as inset for the proton decoupled spectrum.

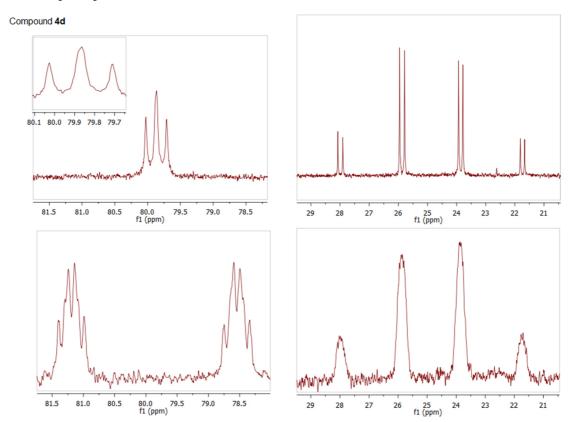


Figure S6. ³¹P{¹H} (top) and ³¹P (bottom) NMR signatures for [Ru(P(H)ClCH₂SiMe₂C₆H₄CF₃)Cl₂(CO)-(PPh₃)₂] **(4d)** plot on equivalent scales. The multiplicity of the chlorophosphane resonances is illustrated as inset for the proton decoupled spectrum.

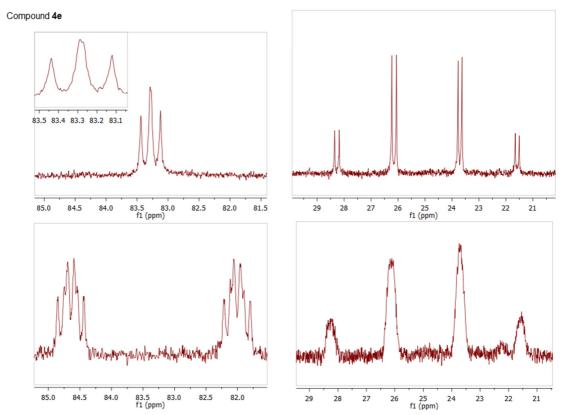


Figure S7. ³¹P{¹H} (top) and ³¹P (bottom) NMR signatures for [Ru(P(H)ClCH₂SiMe₂n-Bu)Cl₂(CO)(PPh₃)₂] (**4e**) plot on equivalent scales. The multiplicity of the chlorophosphane resonances is illustrated as inset for the proton decoupled spectrum.

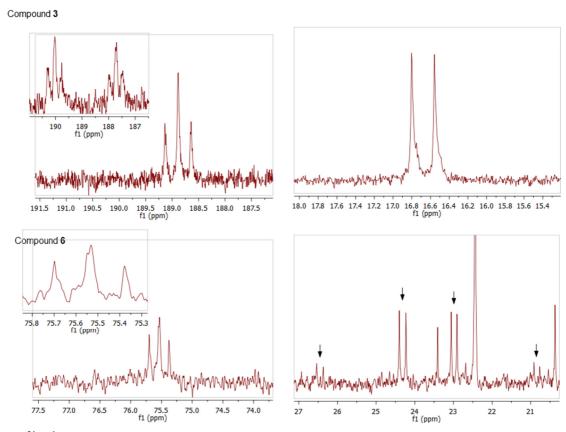


Figure S8. $^{31}P\{^{1}H\}$ NMR signatures for $[Ru(P(H)=CHtBu)Cl_{2}(CO)(PPh_{3})_{2}]$ (3) (top) with the proton coupled phosphacarbon resonance as inset, and $Ru(P(H)ClCH_{2}tBu)Cl_{2}(CO)(PPh_{3})_{2}]$ (6) (bottom). The multiplicity (dd) of the chlorophosphane resonances is illustrated as inset for compound **6**.

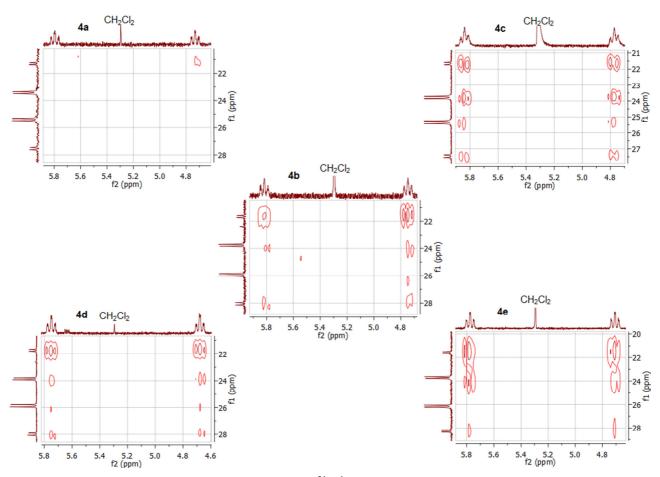


Figure S9: Expansions of the PPh₃ / PH region of ³¹P-¹H HMBC plots for compounds **4a-e**. The relative cross-peak integrals of correlations for the two separate PPh₃ units is in each case at least 5:1, indicative of a statistically significant difference in the magnitude of the H-P-Ru-P (³J) scalar interactions.

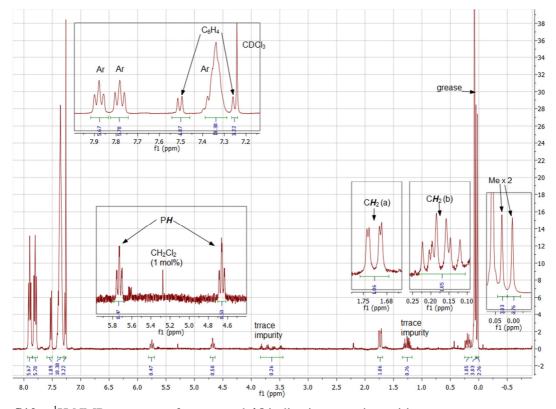


Figure S10a: ¹H-NMR spectrum of compound 4d indicating trace impurities.

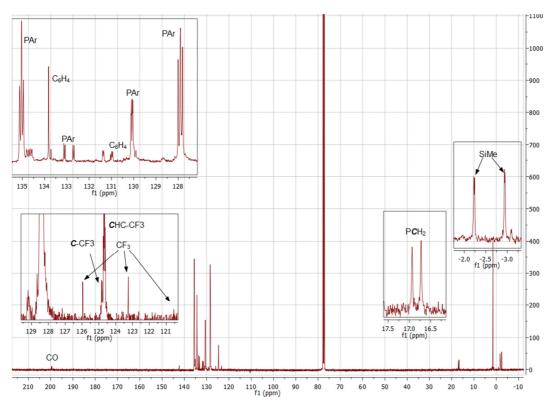


Figure S10b: ¹³C{ ¹H}-NMR spectrum of compound **4d**.

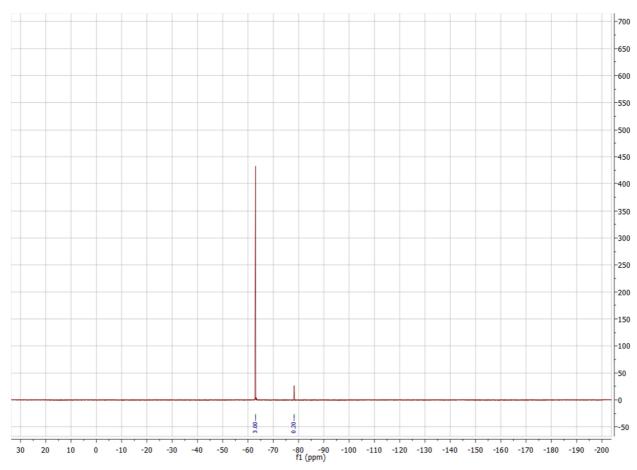


Figure S10c: ¹⁹F-NMR spectrum of compound **4d**, indicating trace impurity.

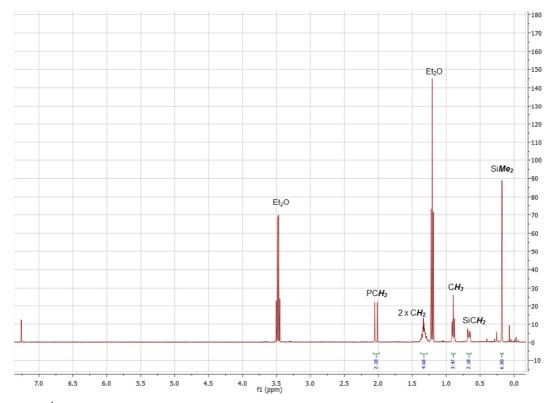


Figure S11a: ¹H-NMR spectrum of nBuMe₂SiCH₂PCl₂ indicating residual solvent.

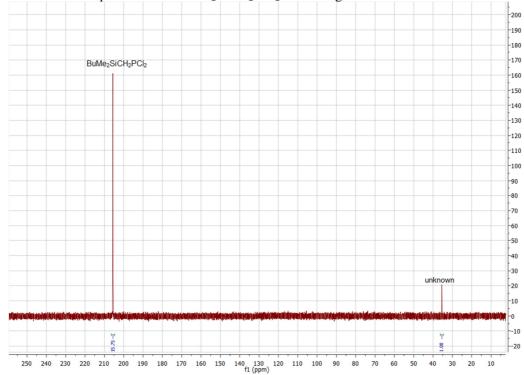


Figure S11b: ³¹P-NMR spectrum of nBuMe₂SiCH₂PCl₂ indicating trace impurity.

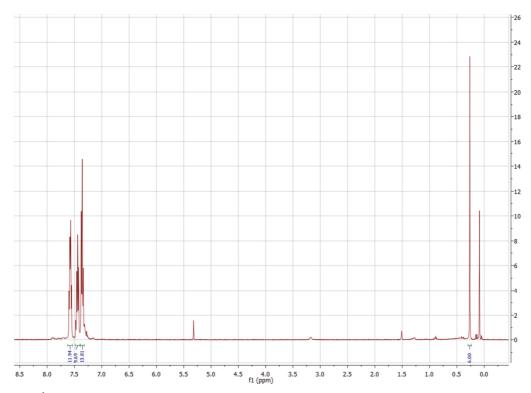


Figure S12a: ¹H-NMR spectrum of compound 2d

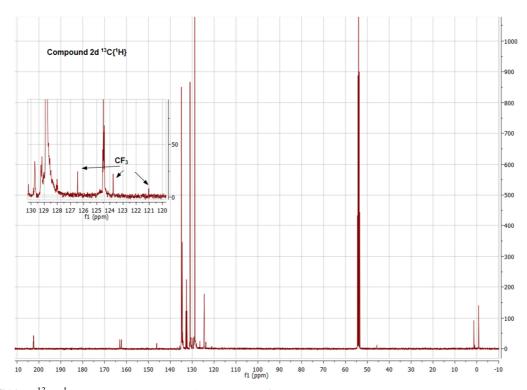


Figure S12b: $^{13}C\{^{1}H\}$ -NMR spectrum of compound **2d**.

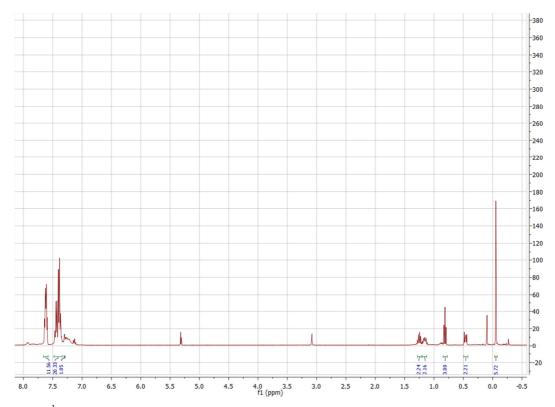


Figure S13a: ¹H-NMR spectrum of compound 2e

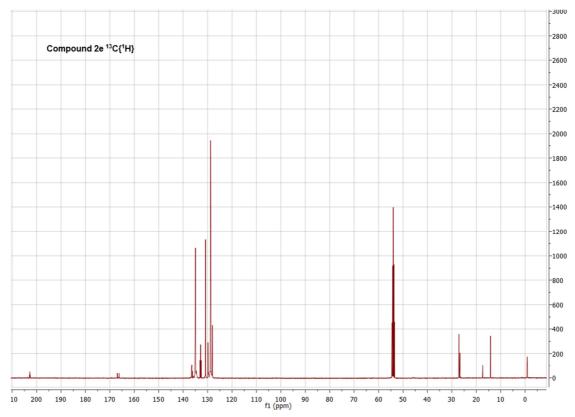


Figure S13b: ¹³C{¹H}-NMR spectrum of compound **2e**.

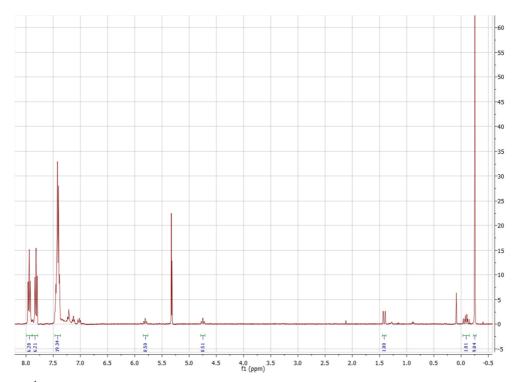


Figure S14a: ¹H-NMR (CD₂Cl₂) spectrum of compound 4a

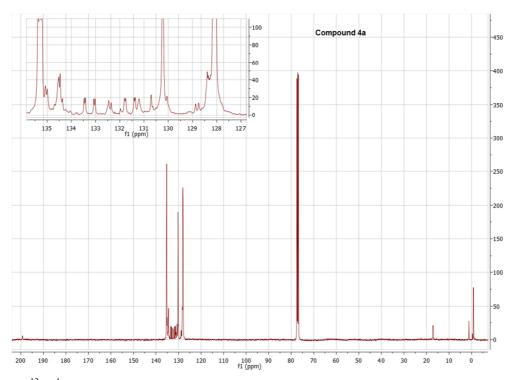


Figure S14b: $^{13}C\{^{1}H\}$ -NMR (CDCl₃) spectrum of compound 4a.

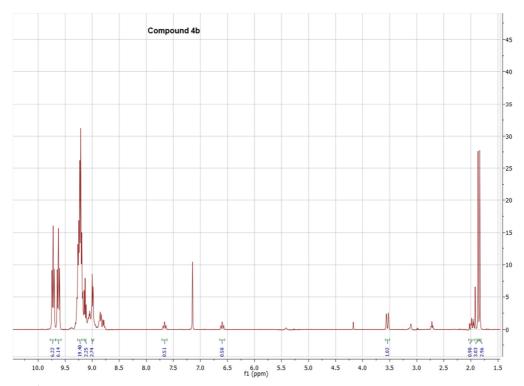


Figure S15a: ¹H-NMR (CD₂Cl₂) spectrum of compound 4b

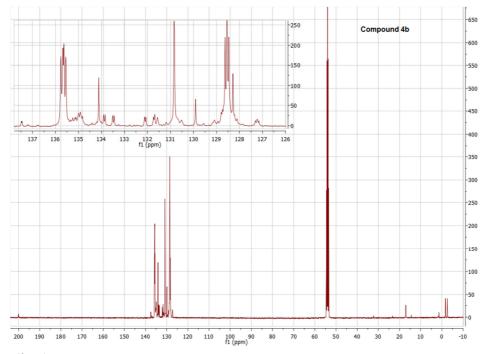


Figure S15b: ¹³C{¹H}-NMR (CD₂Cl₂) spectrum of compound **4b**.

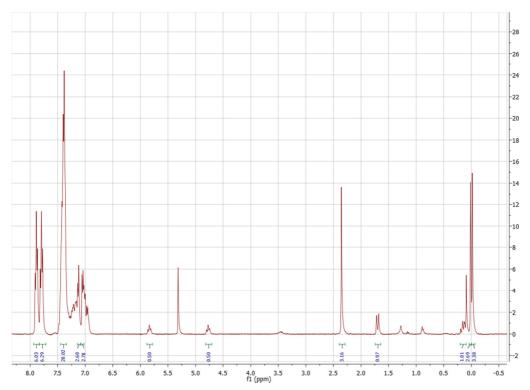


Figure S16a: $^{1}\text{H-NMR}$ (CD $_{2}$ Cl $_{2}$) spectrum of compound 4c

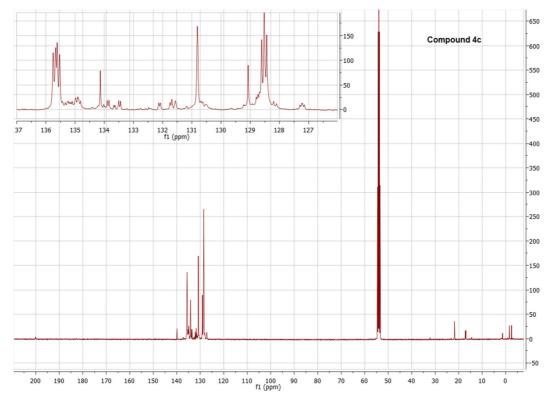


Figure S16b: ¹³C{¹H}-NMR (CD₂Cl₂) spectrum of compound **4c**.

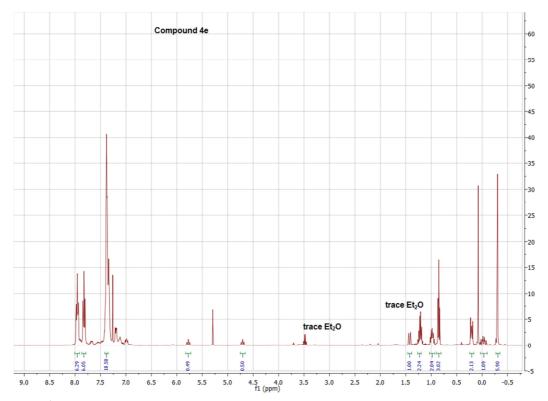


Figure S17a: ¹H-NMR (CD₂Cl₂) spectrum of compound 4e

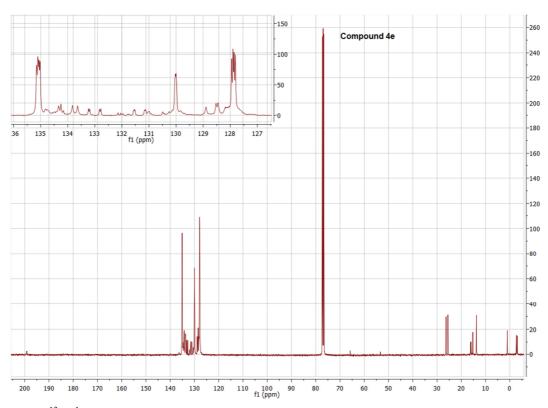


Figure S17b: $^{13}C\{^{1}H\}$ -NMR ($CD_{2}Cl_{2}$) spectrum of compound 4e.