

## **Polycyclic Ring Formation Using Bis-Diazolactams for Cascade Stitching**

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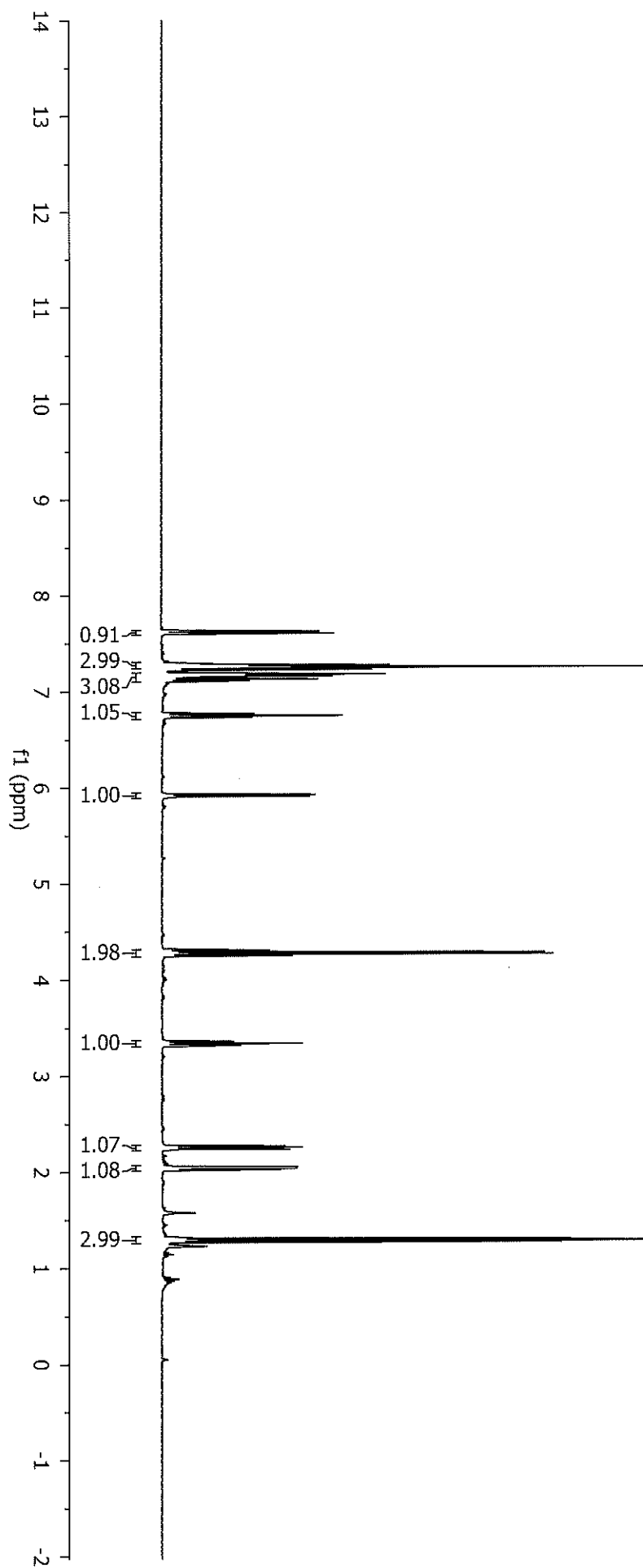
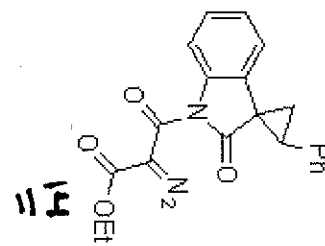
*Department of Chemistry, Emory University, Atlanta Georgia 30322 USA*

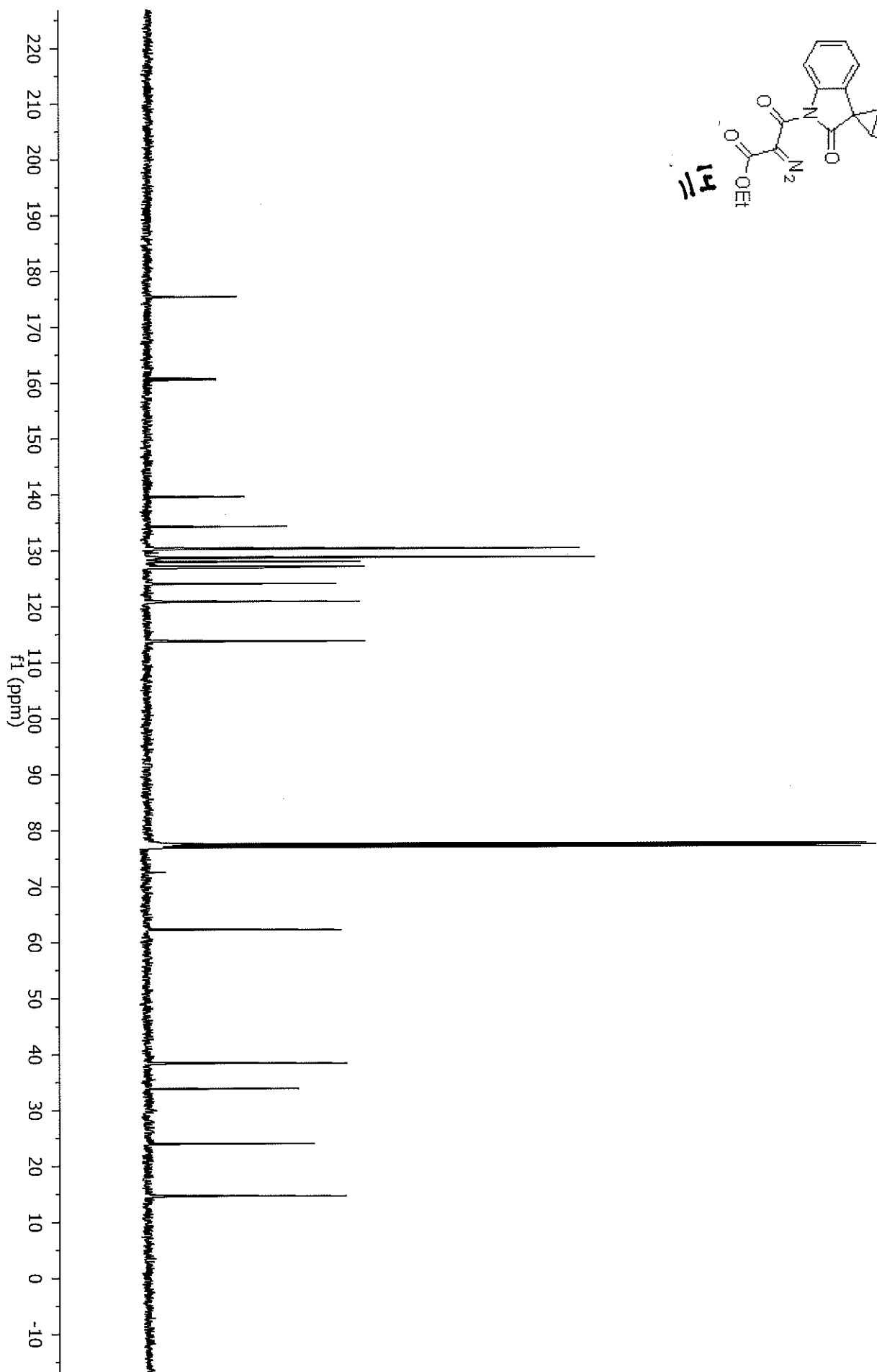
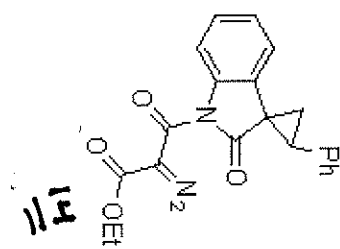
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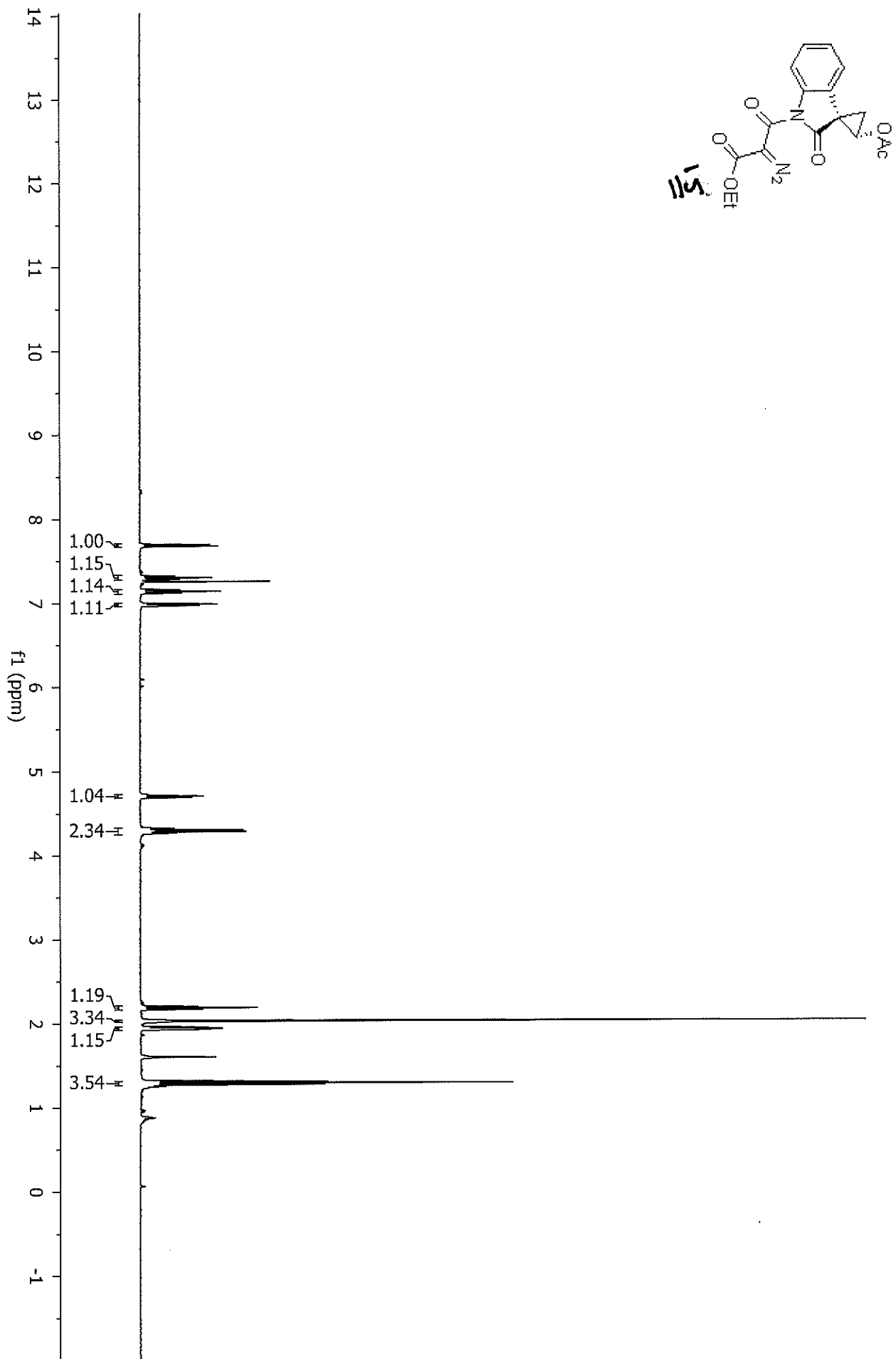
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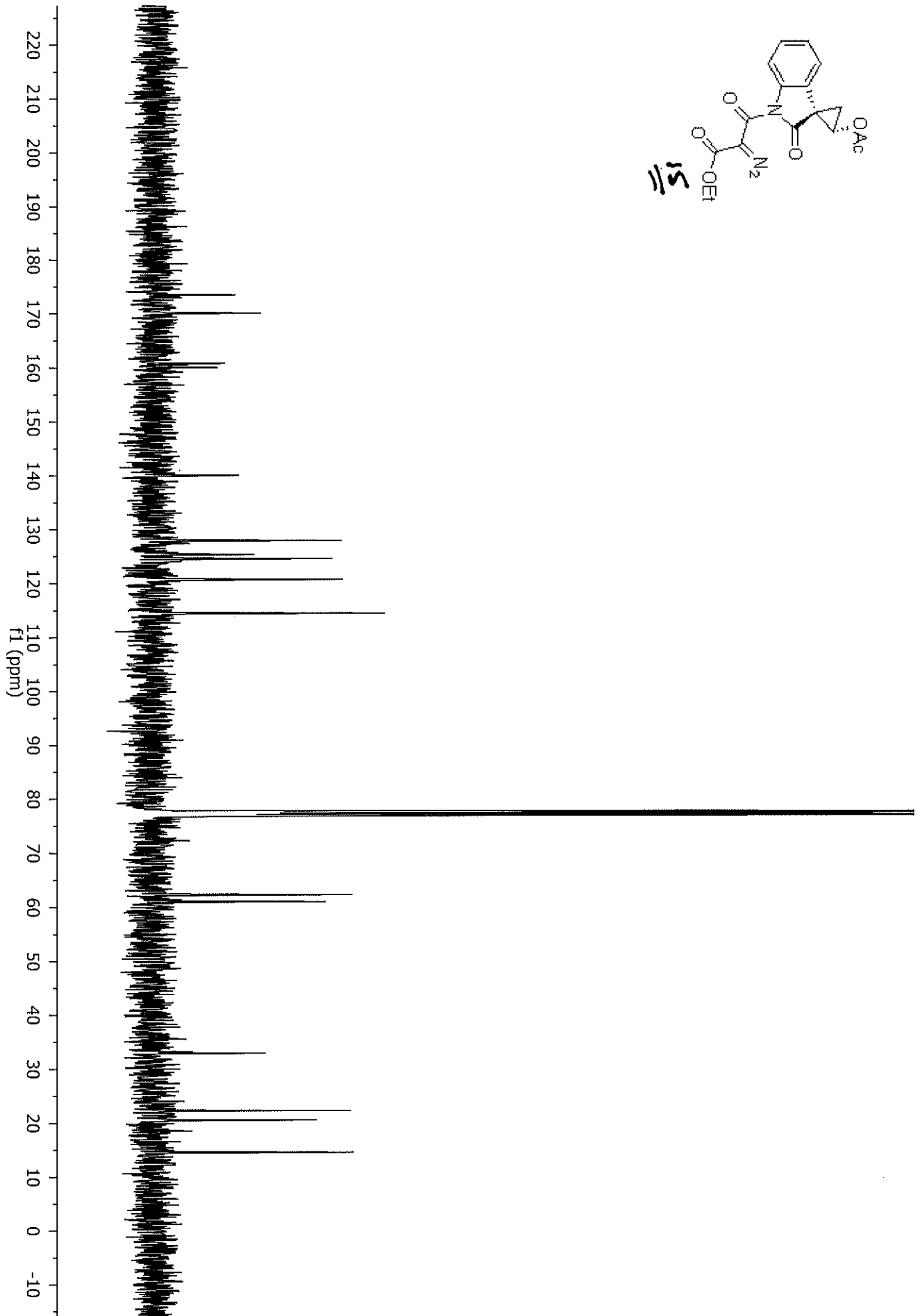
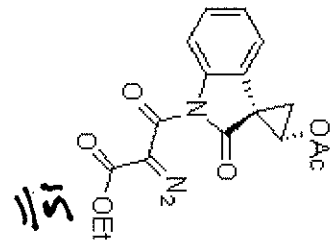
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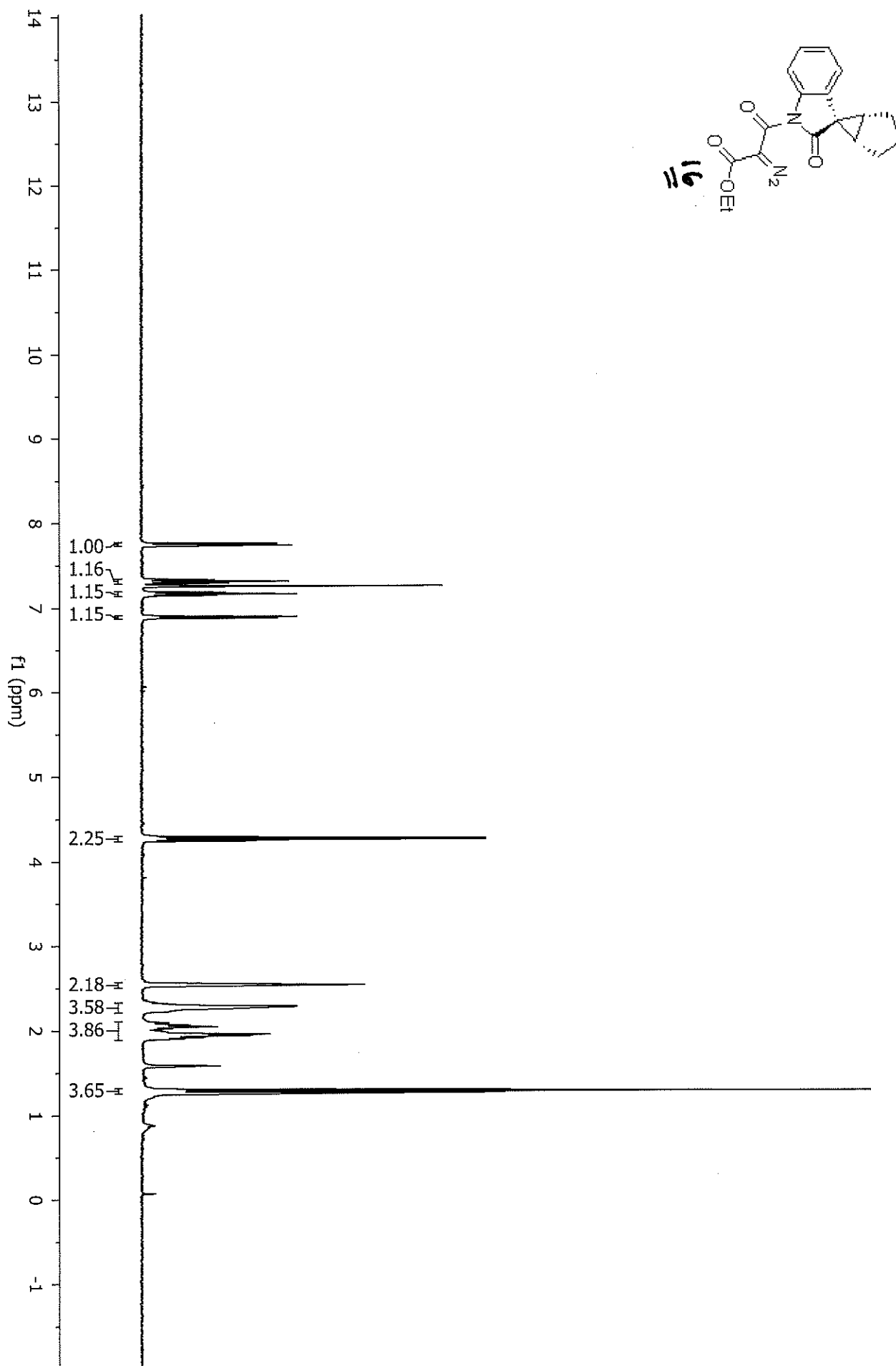
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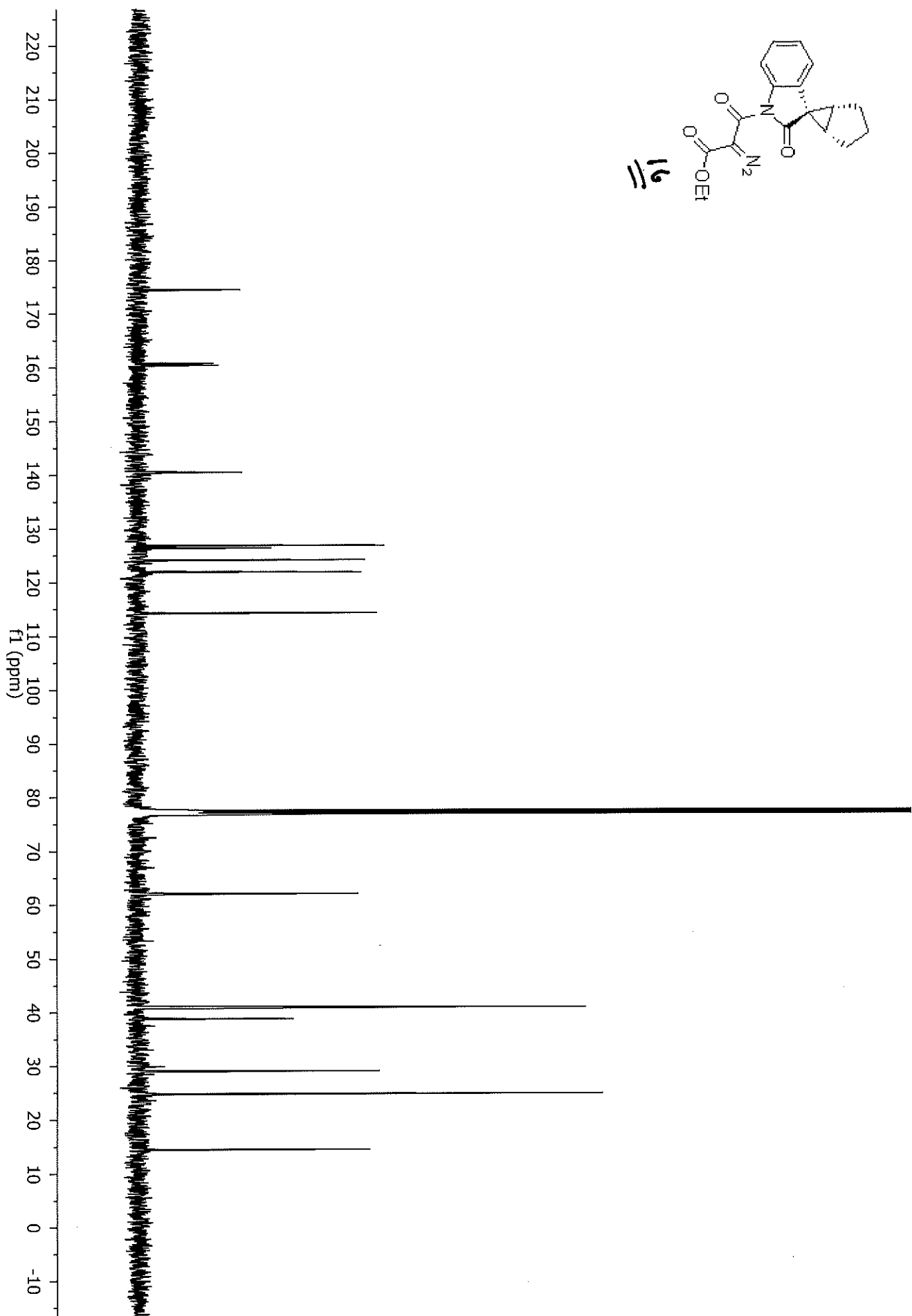
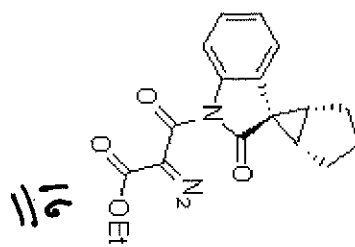






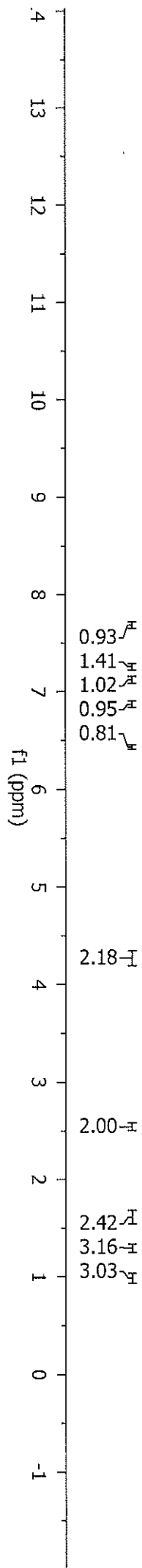
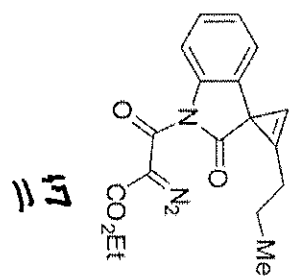




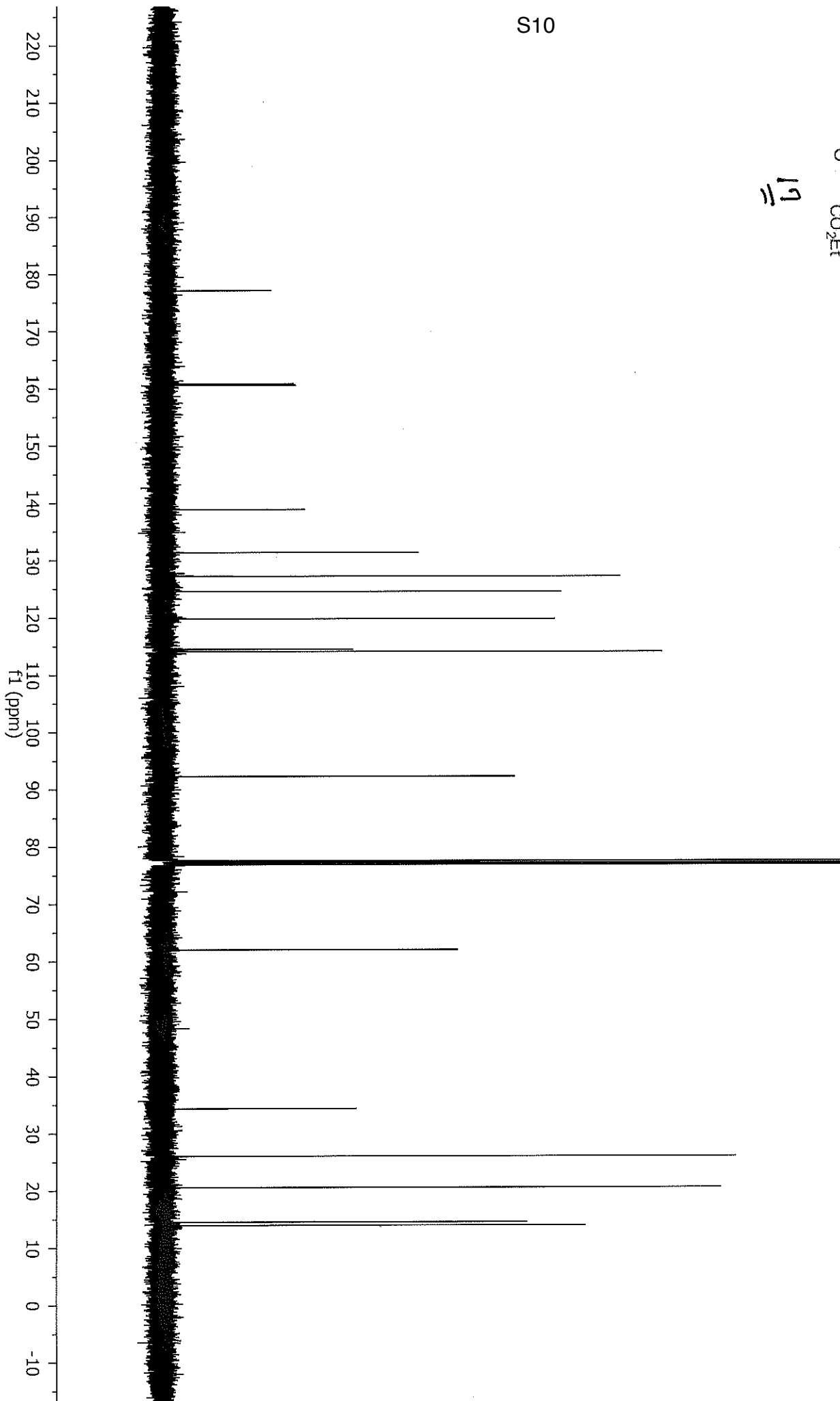
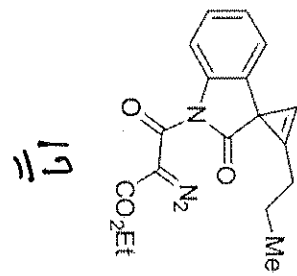




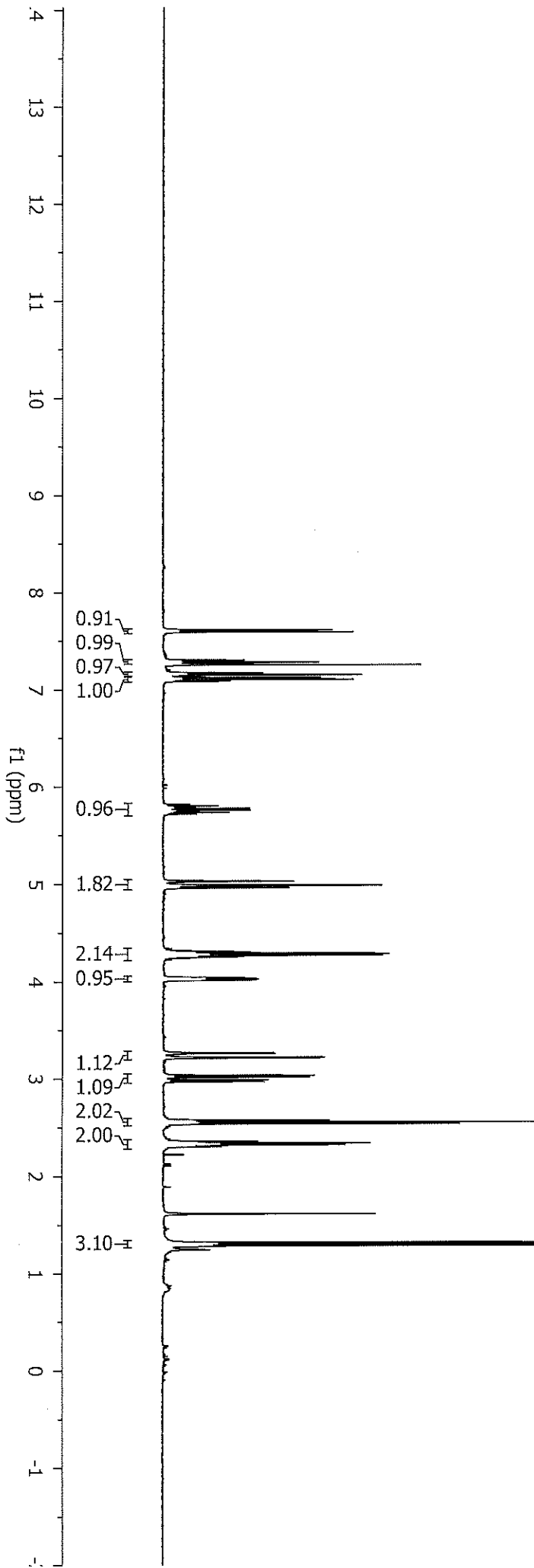
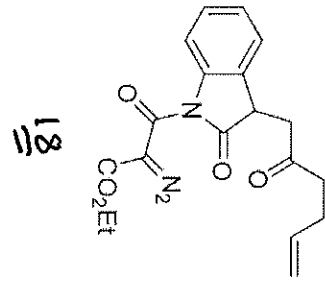
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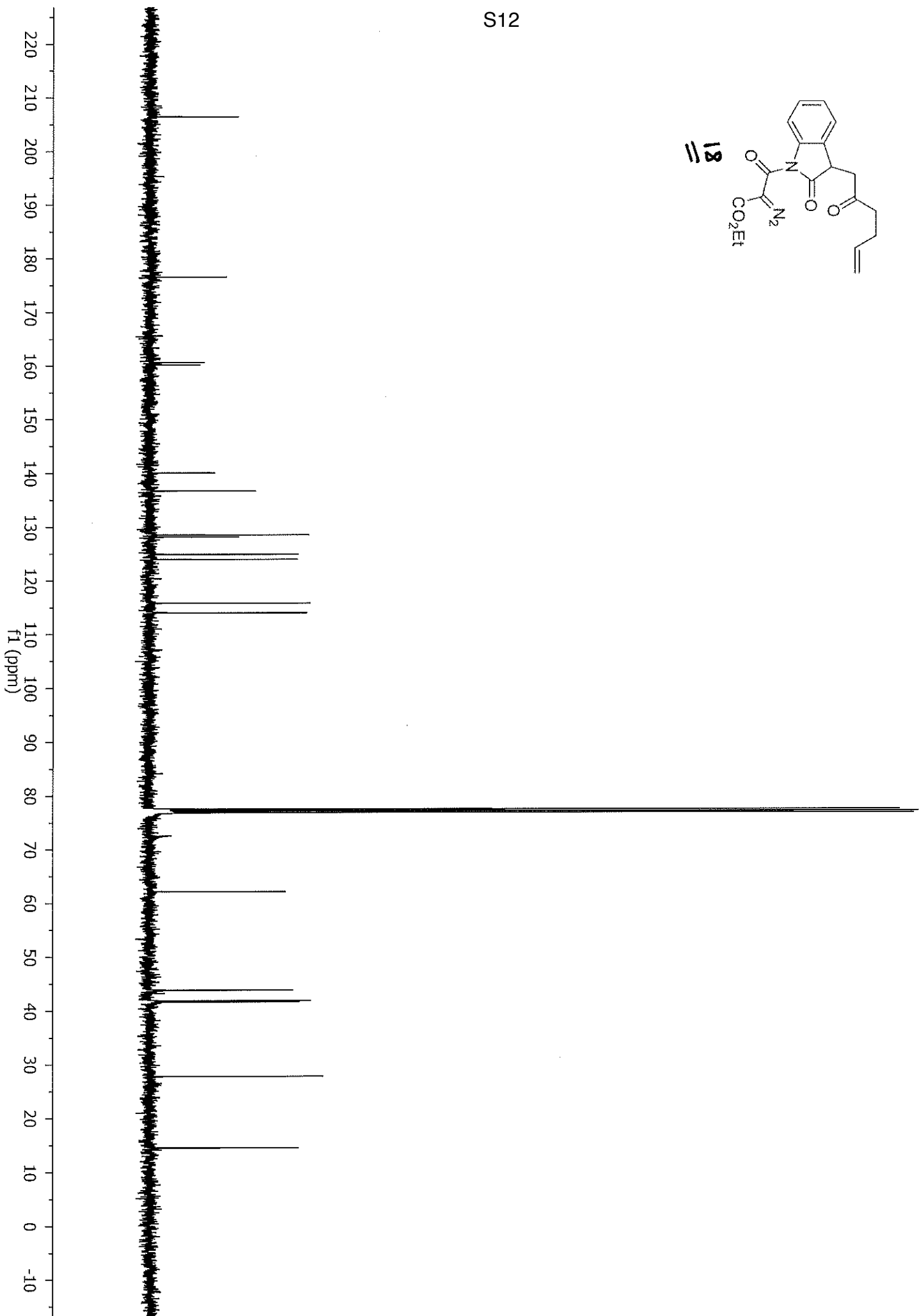
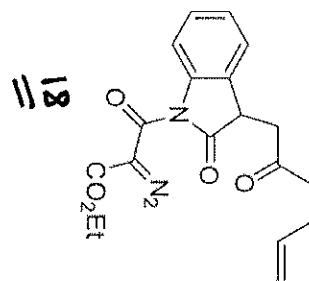
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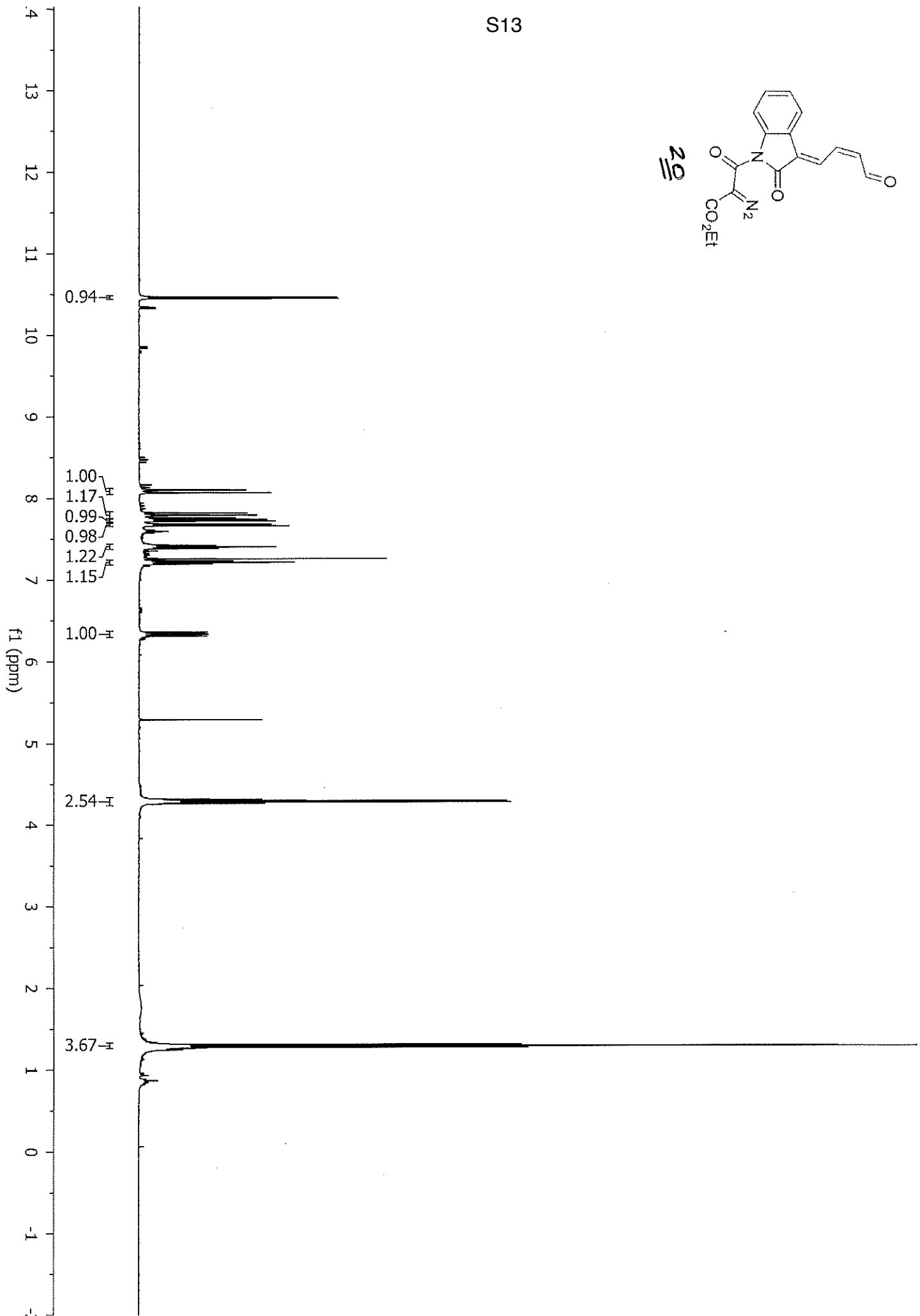
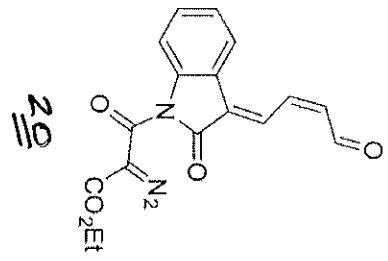
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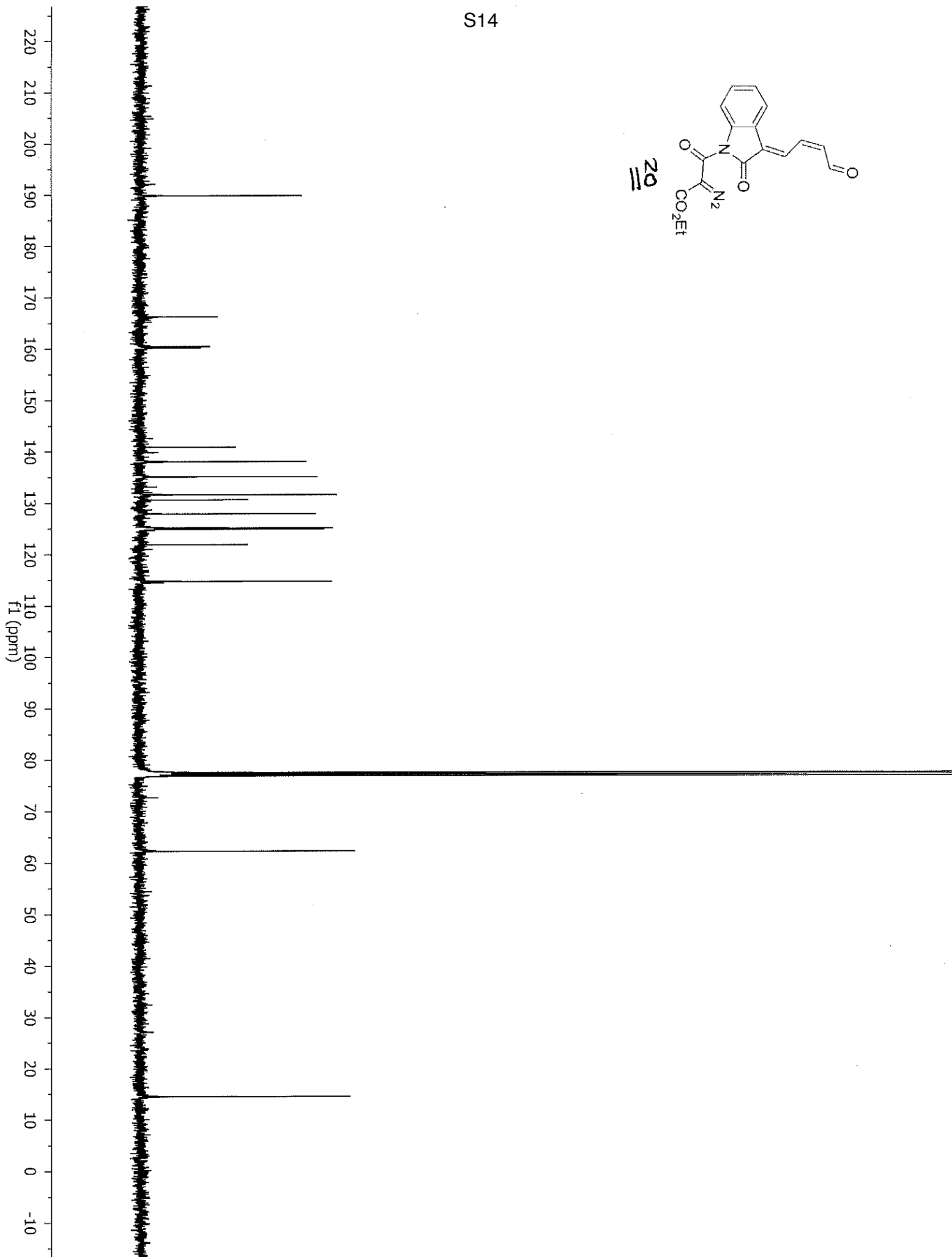
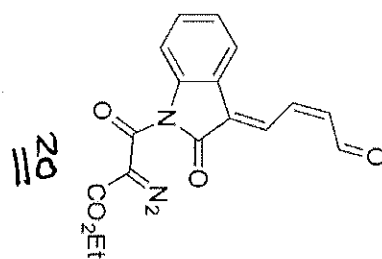
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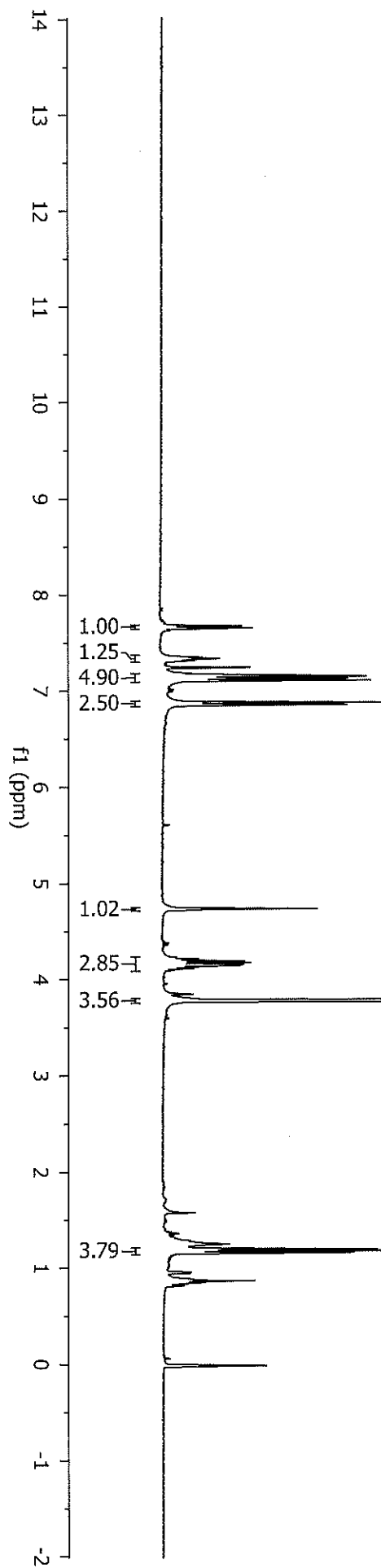
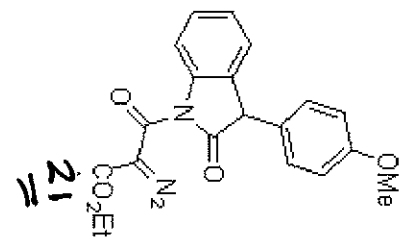


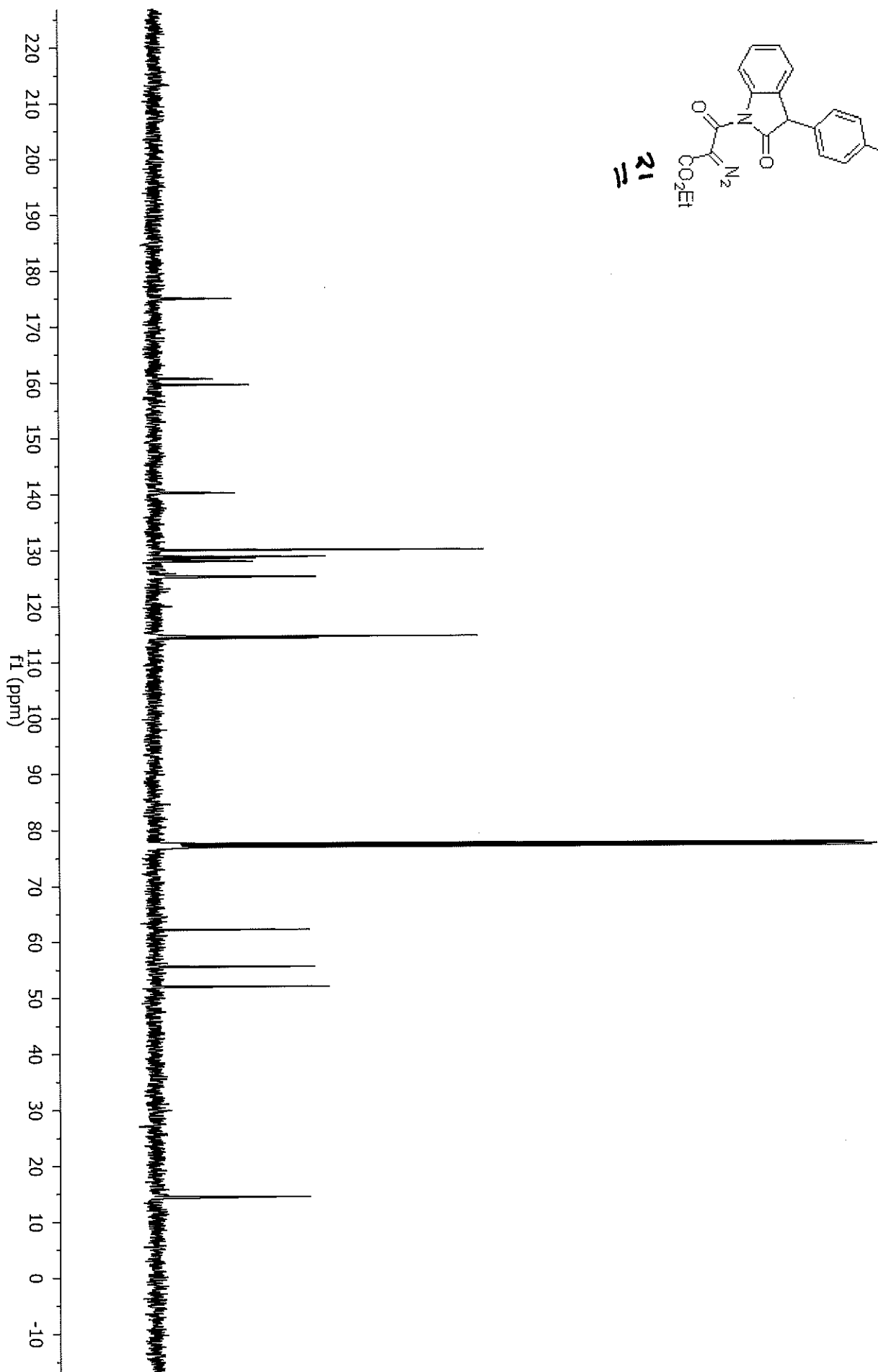
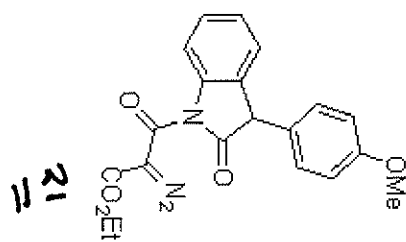
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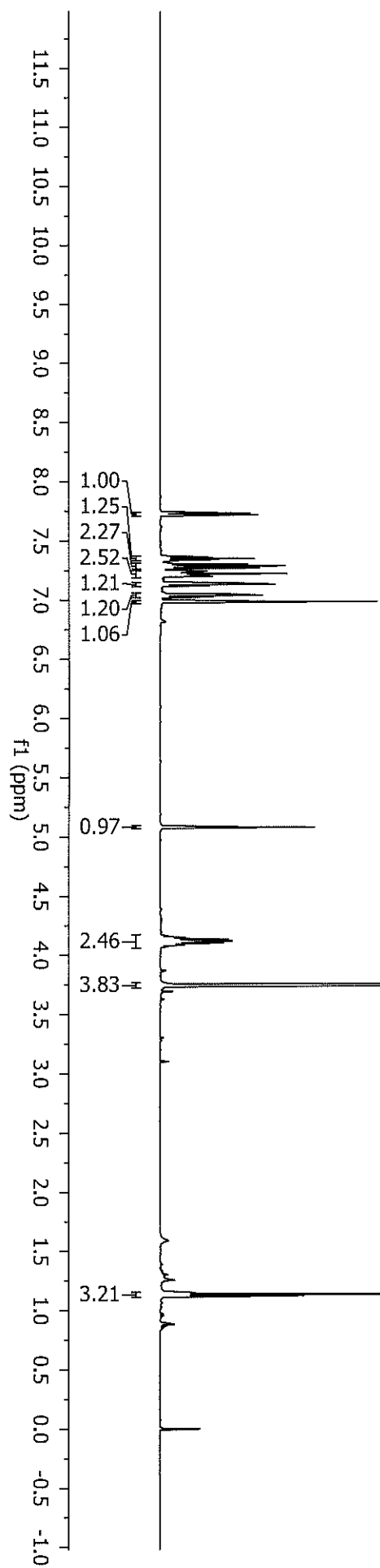
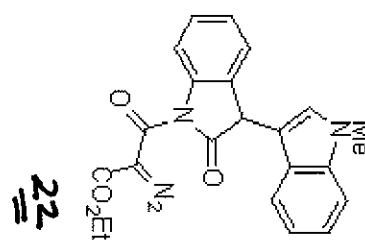
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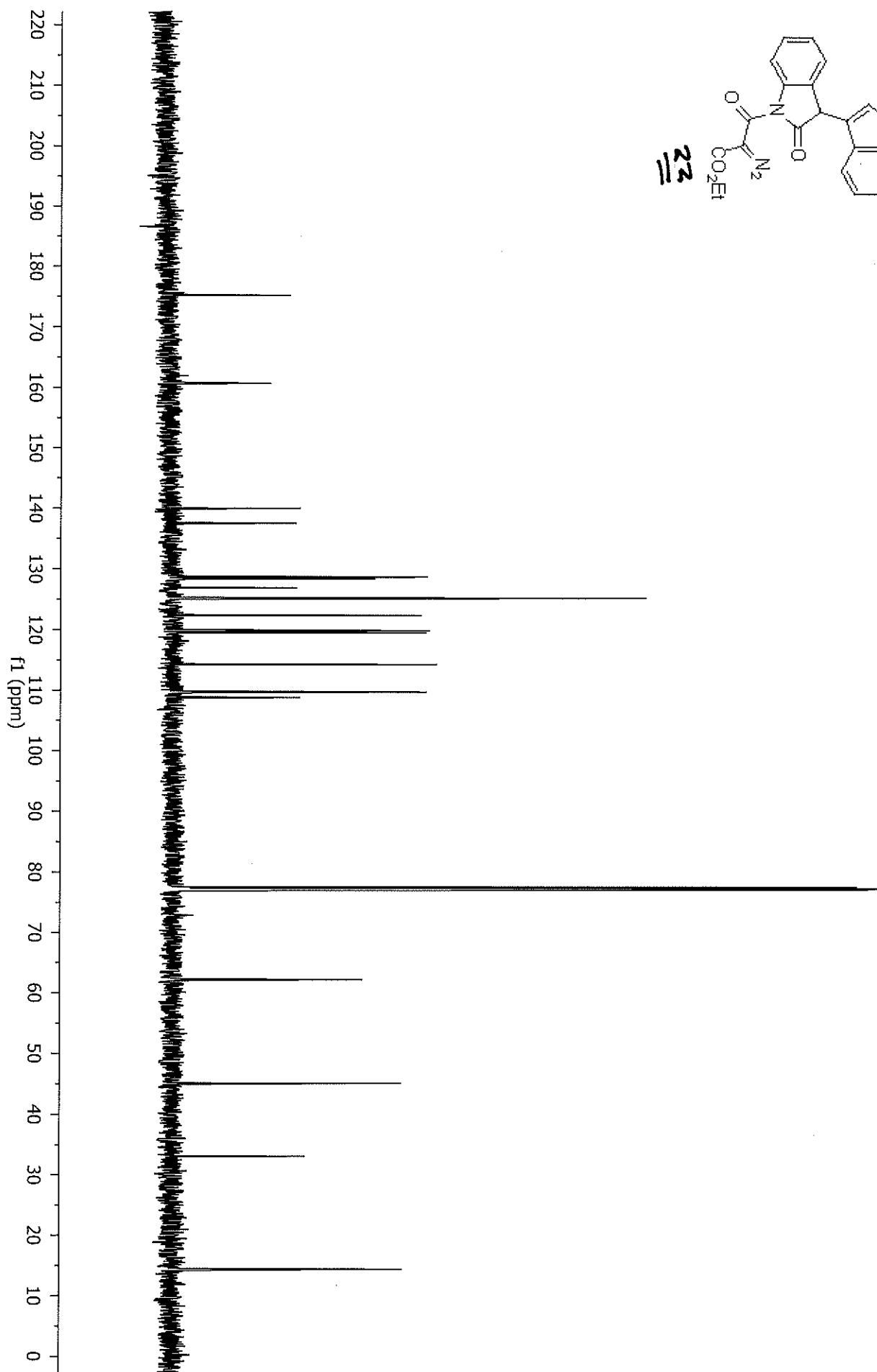
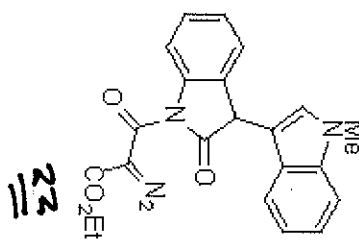


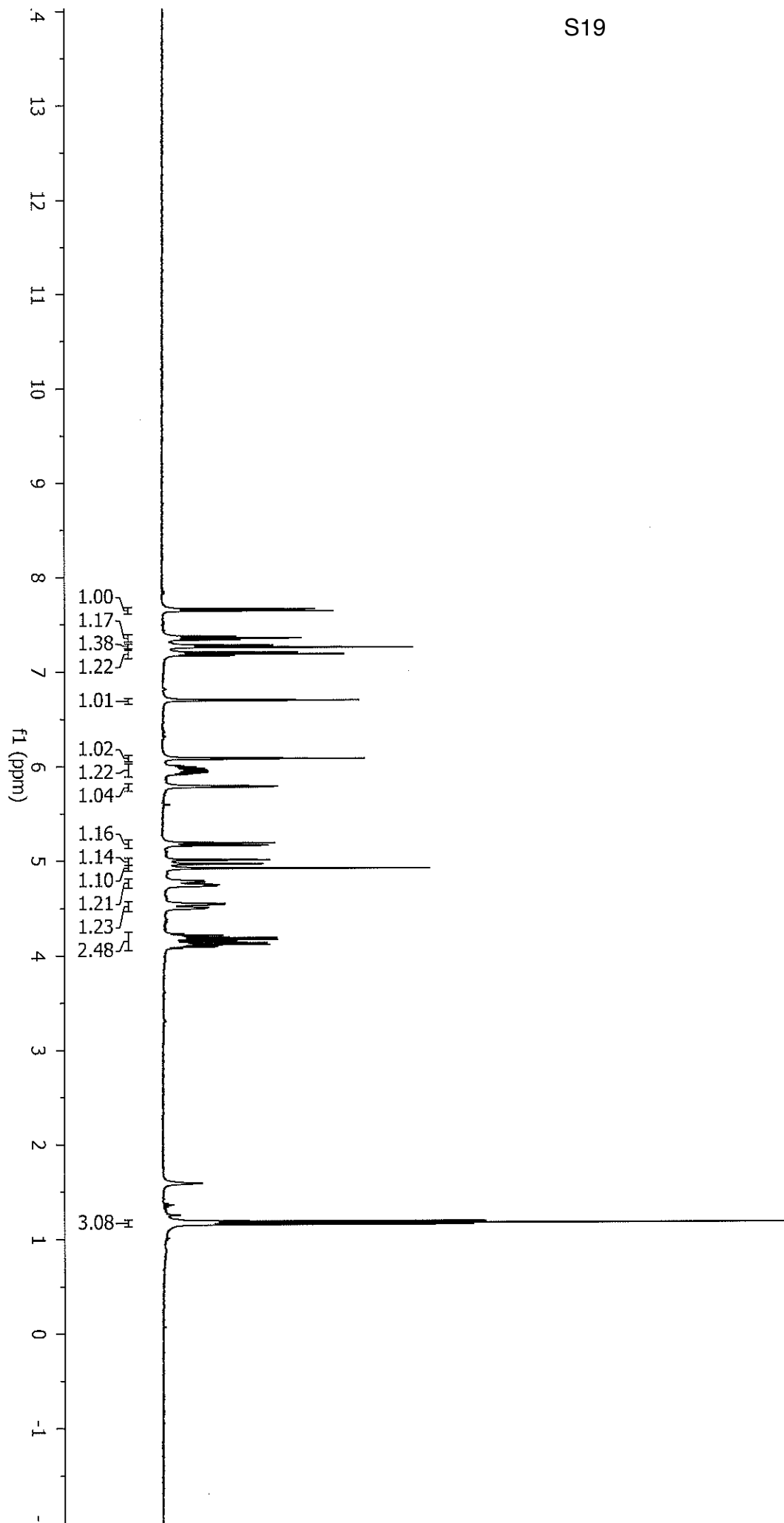




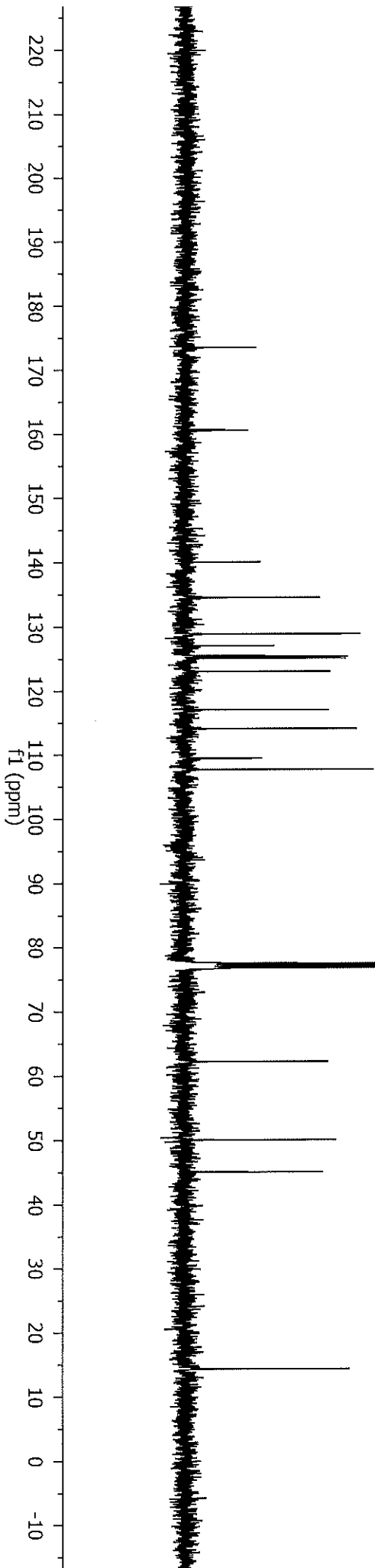
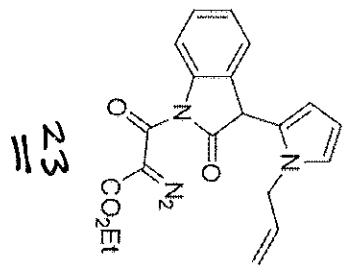




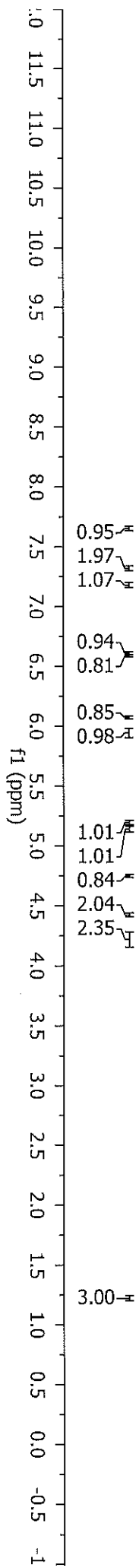
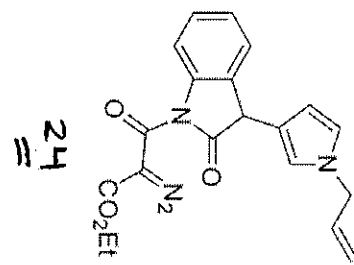




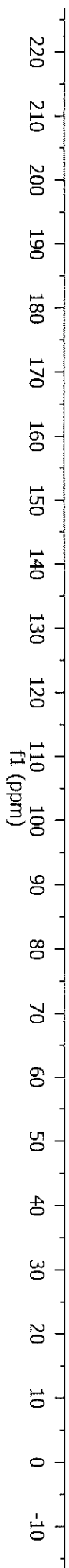
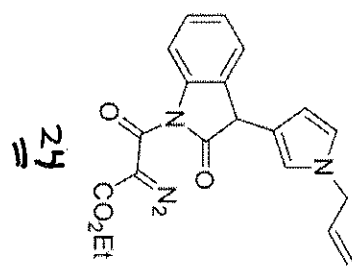
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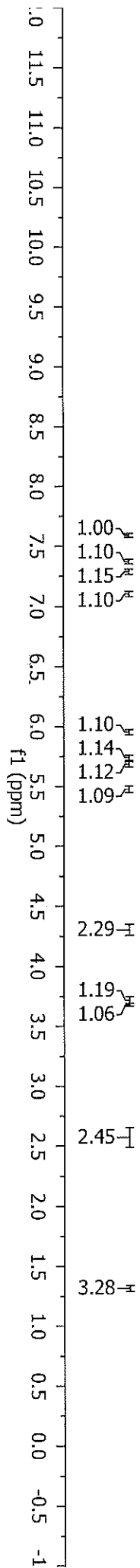
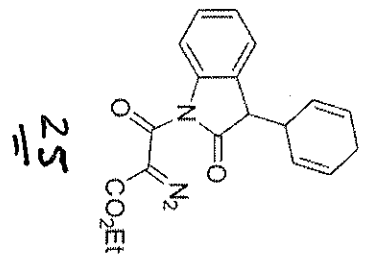
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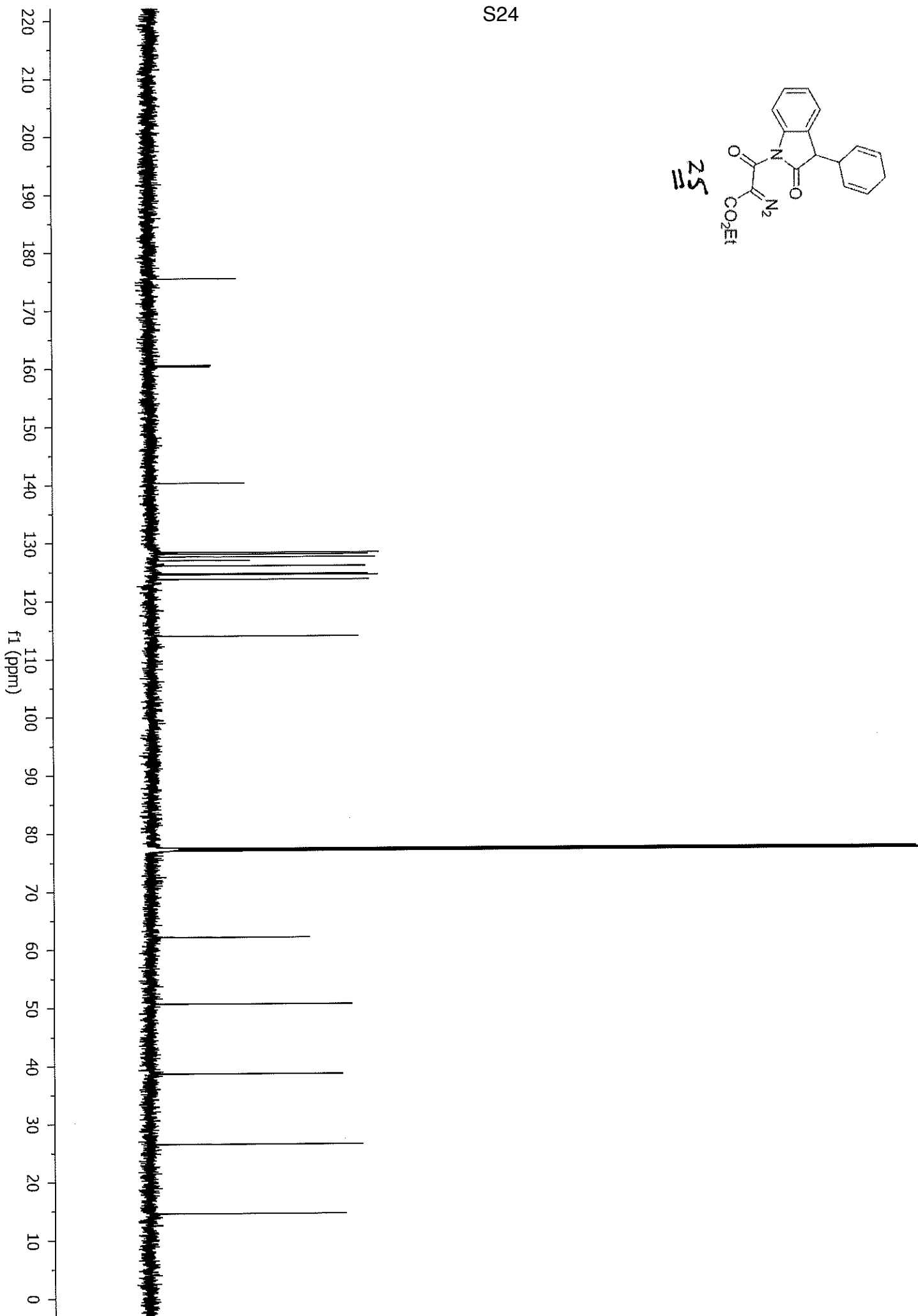
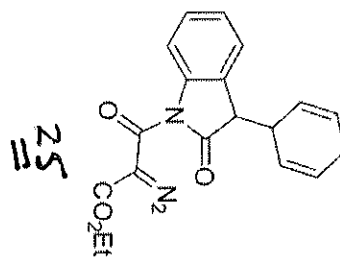
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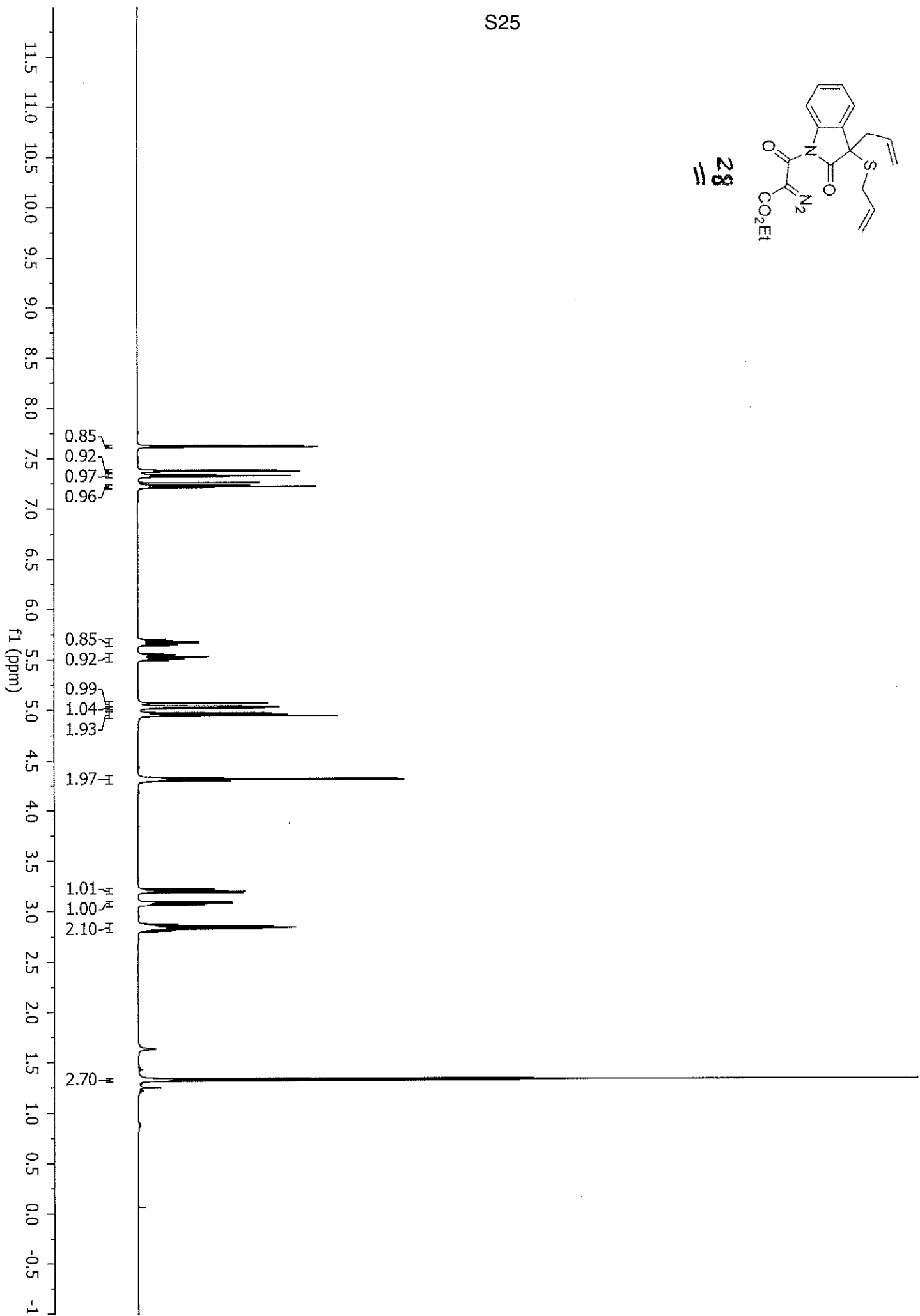
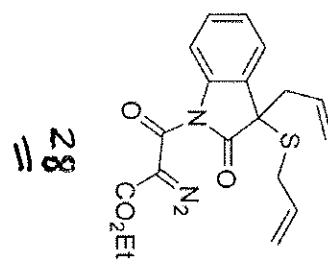


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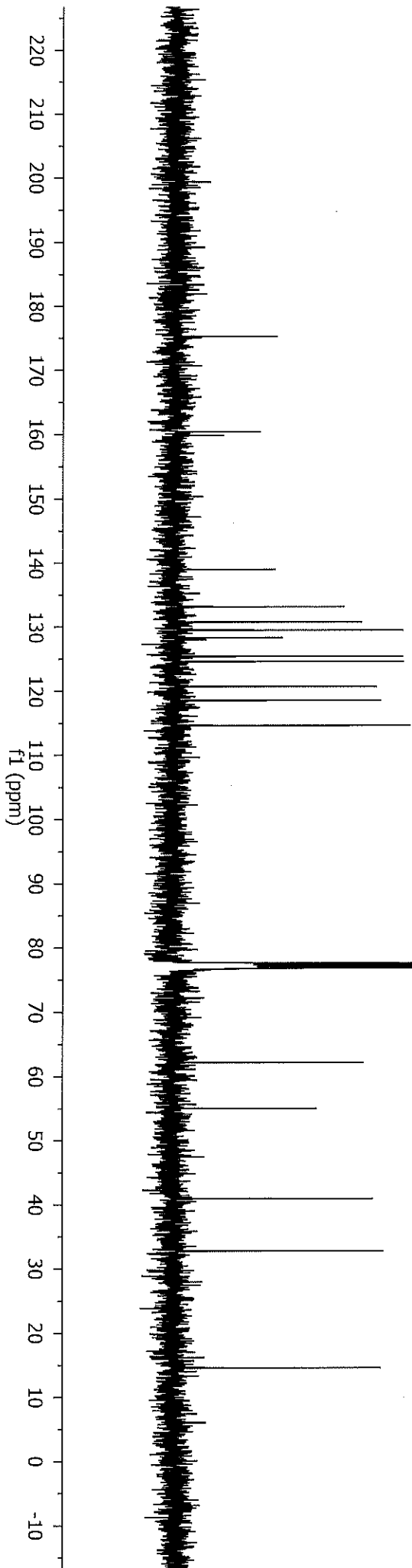
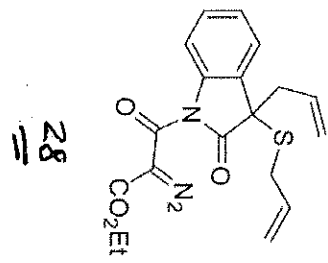


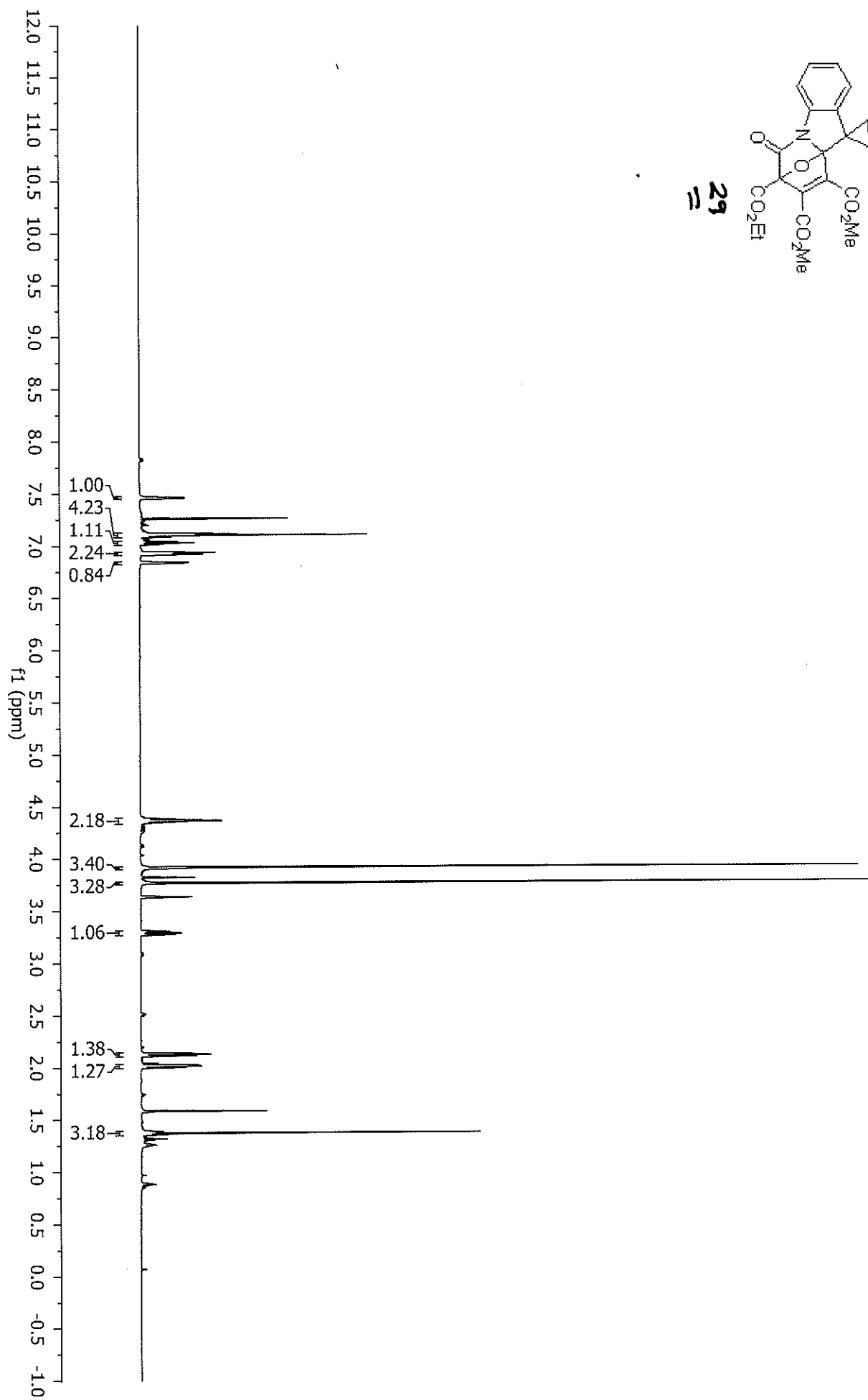
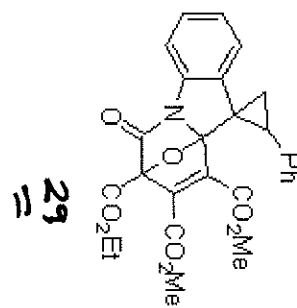


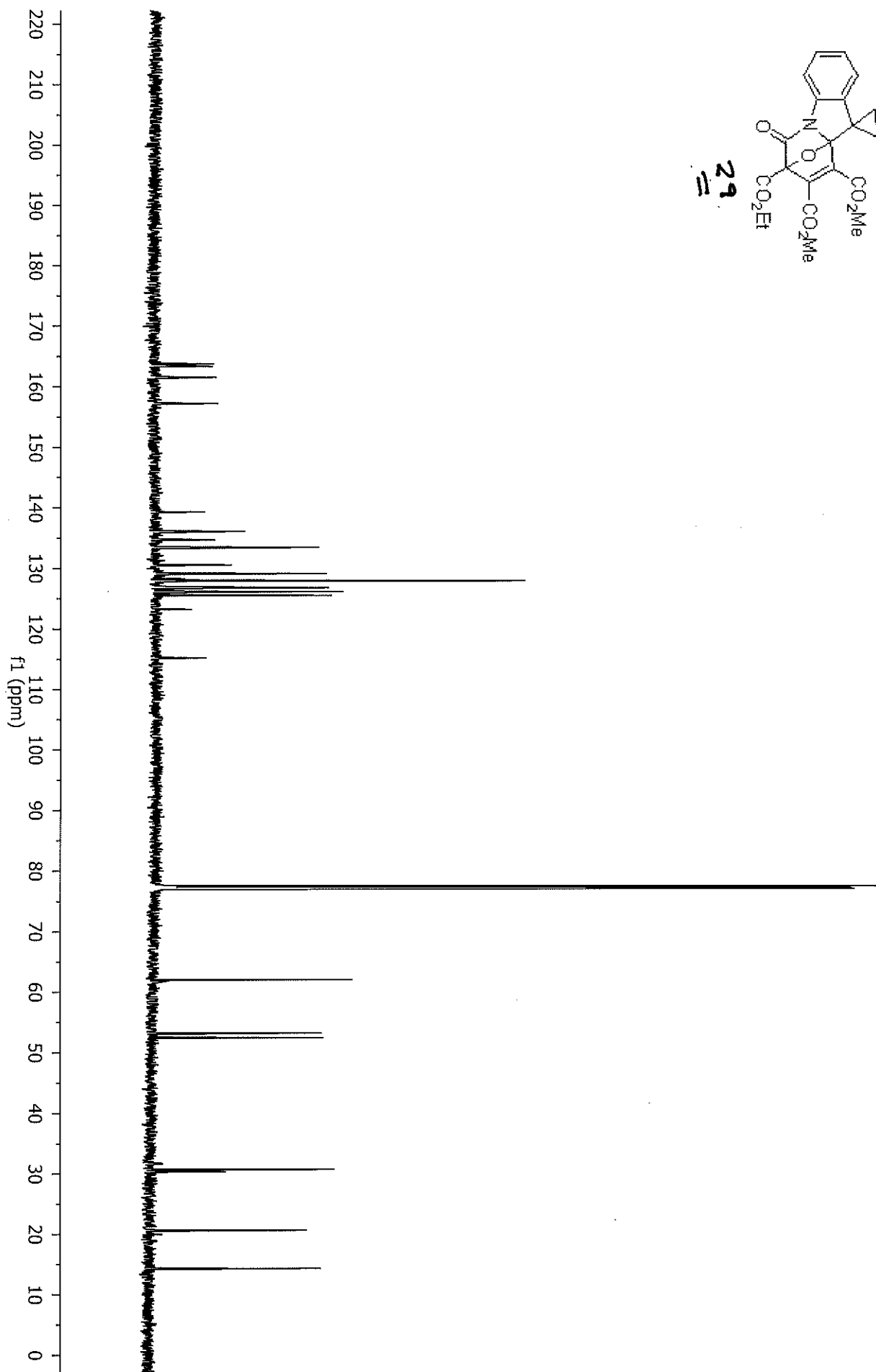
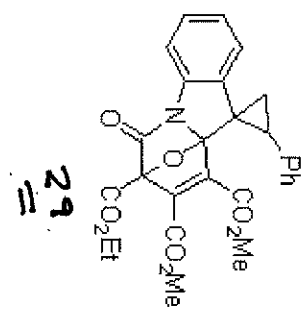
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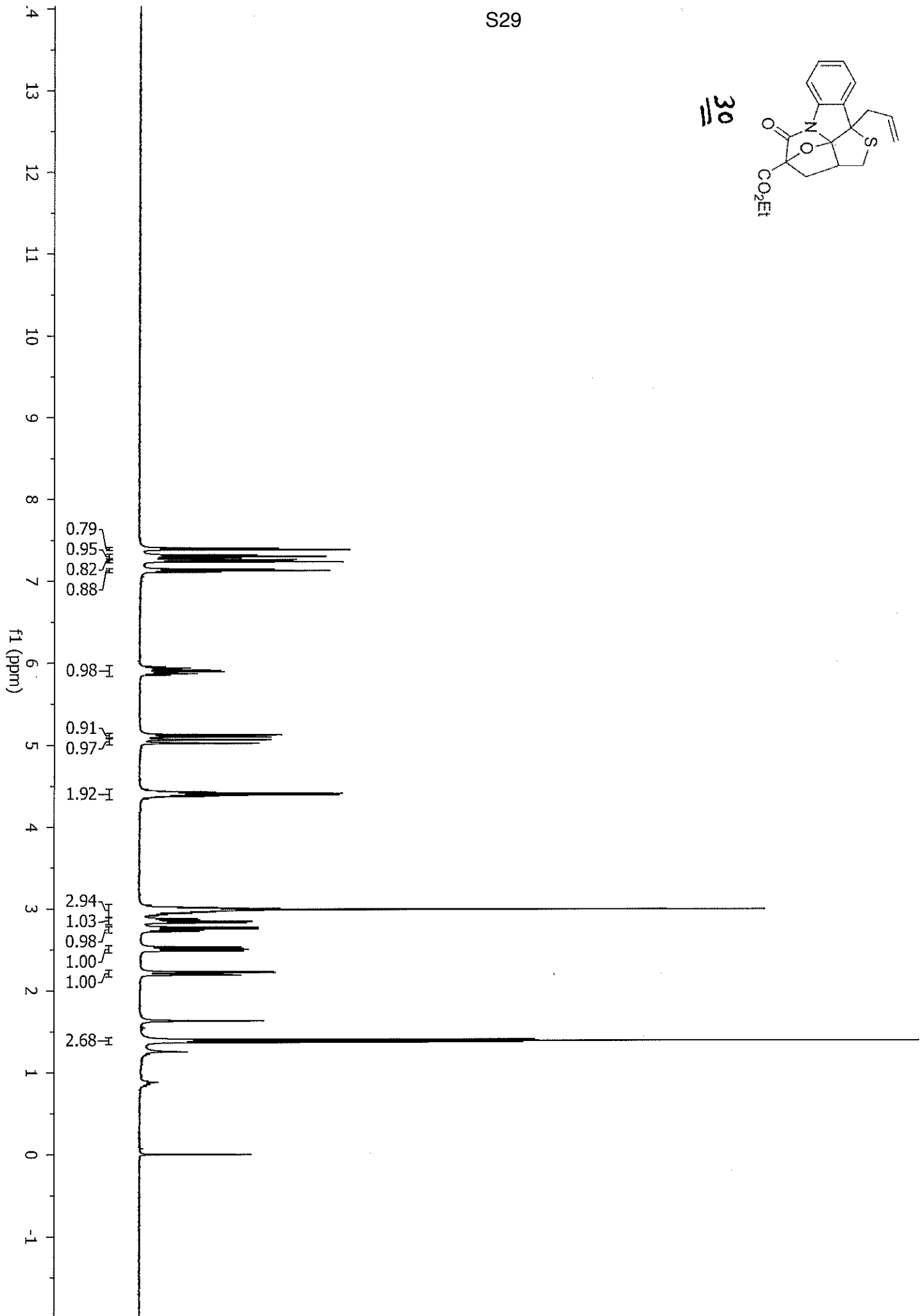
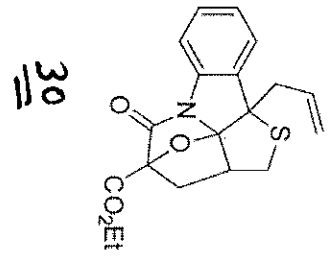
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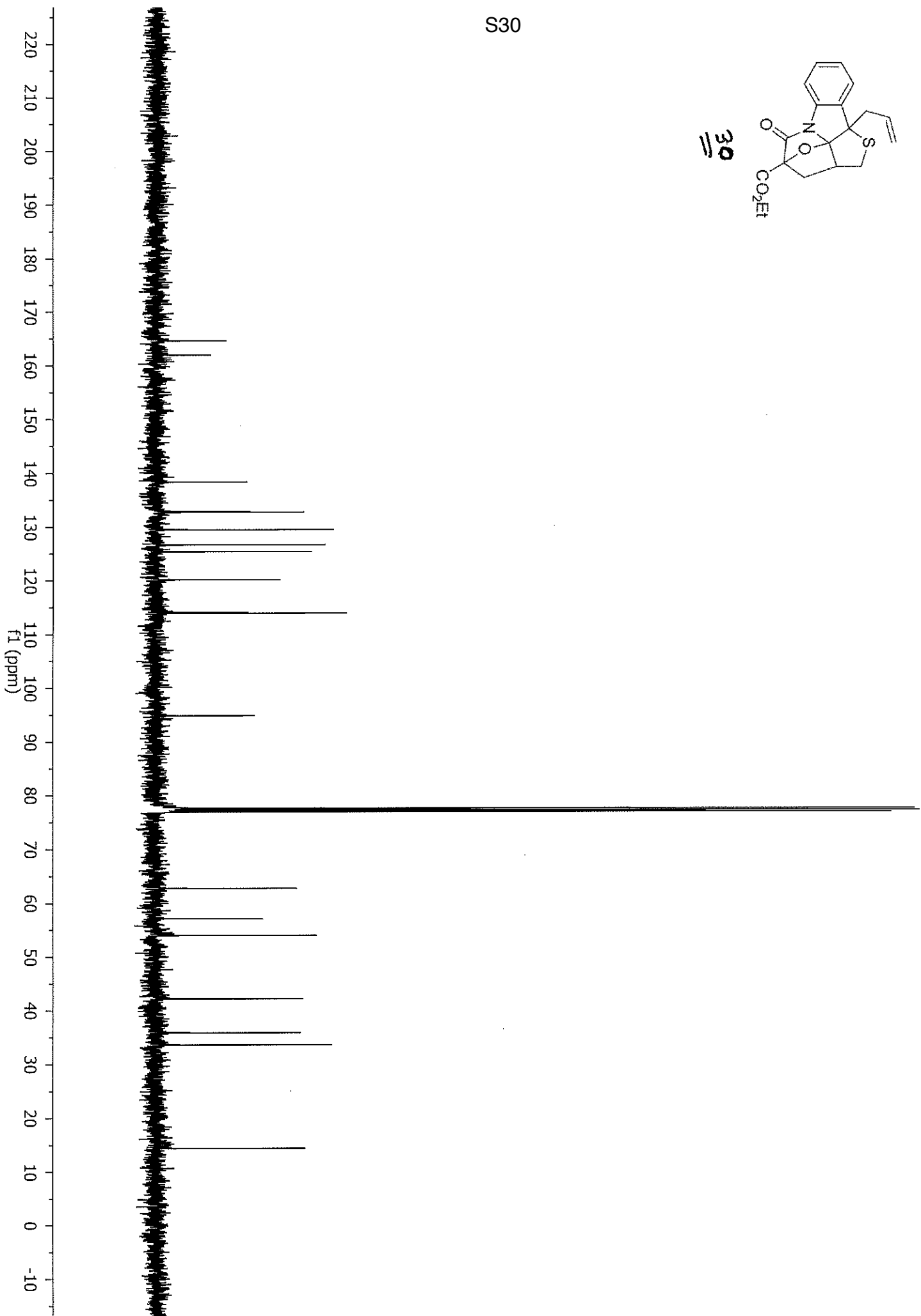
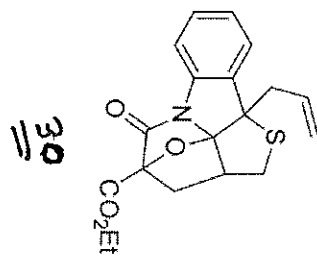




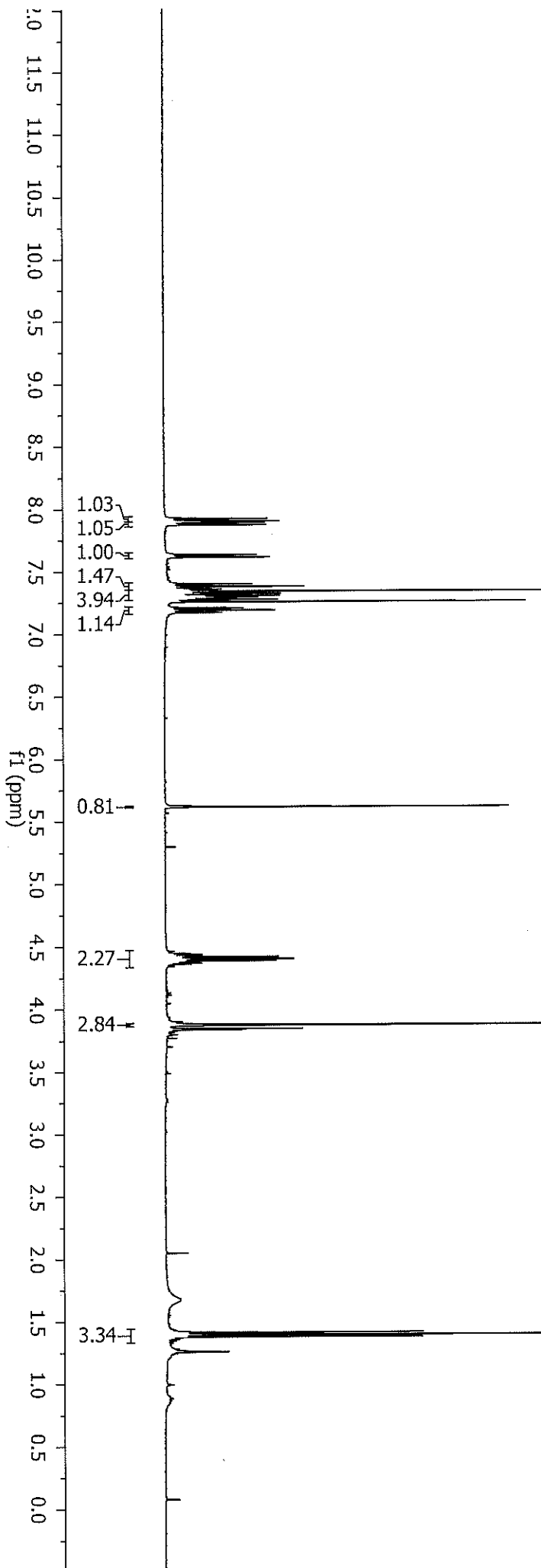
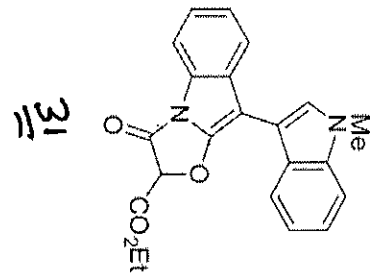
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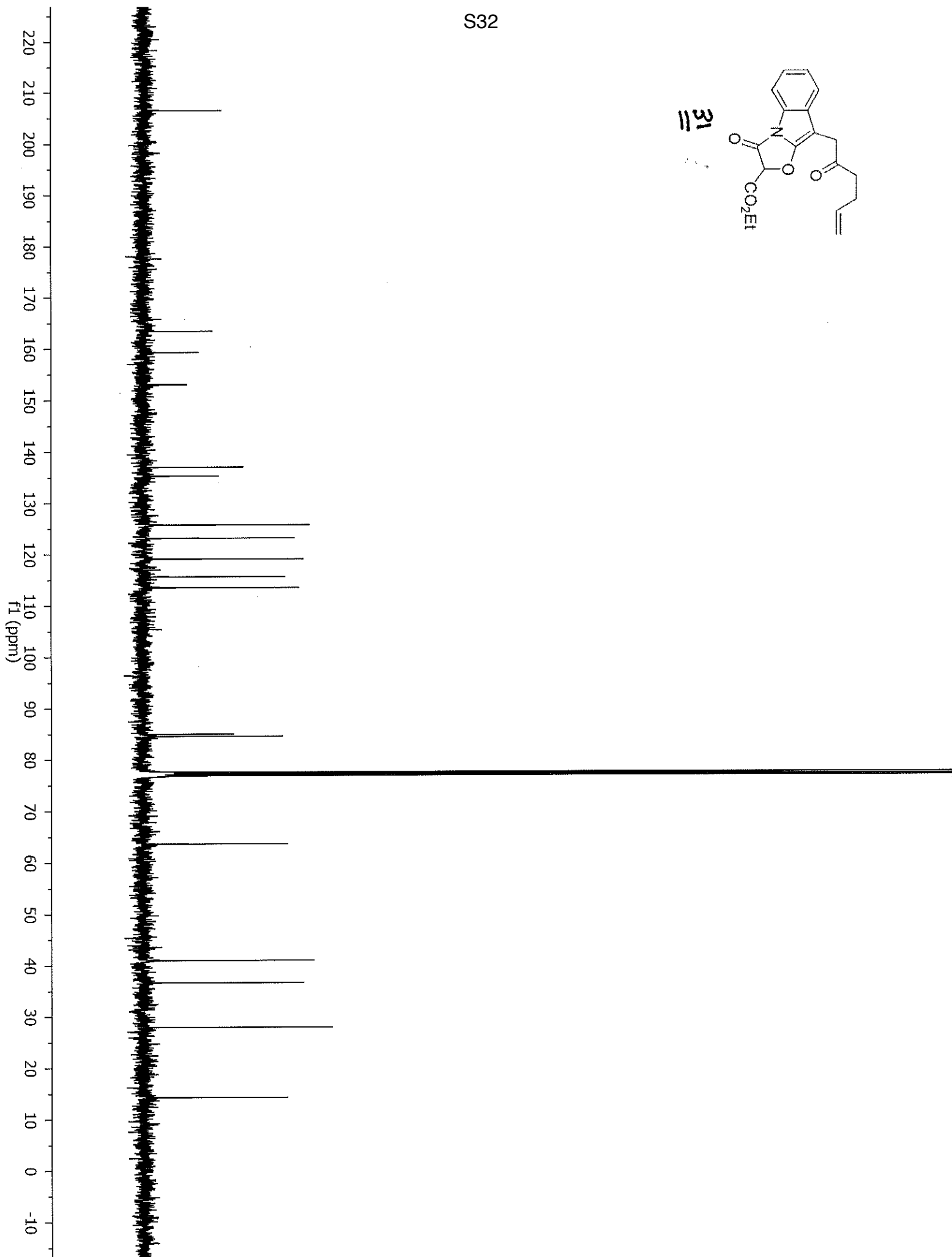
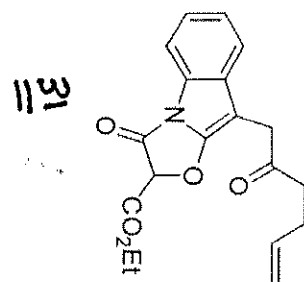


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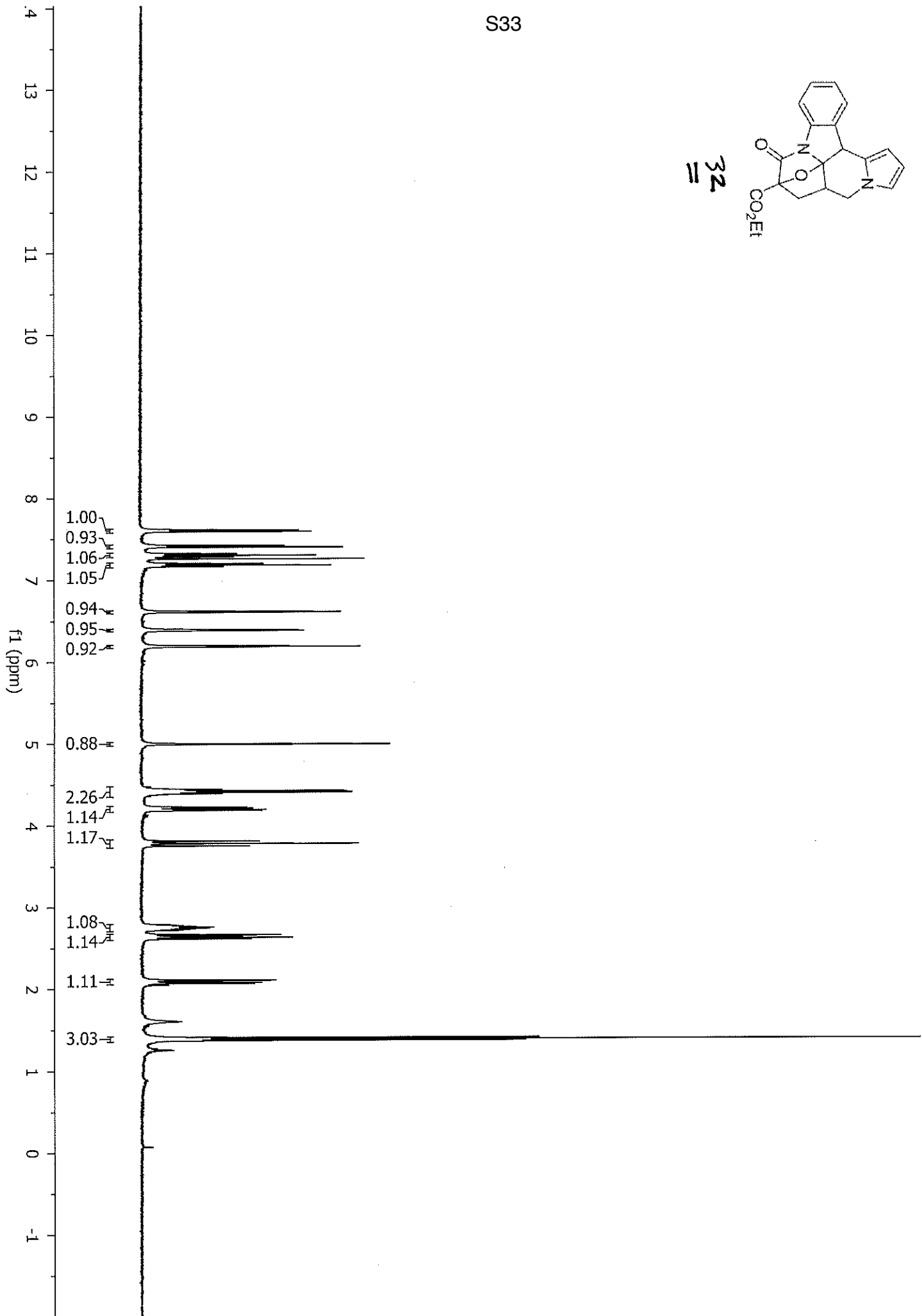
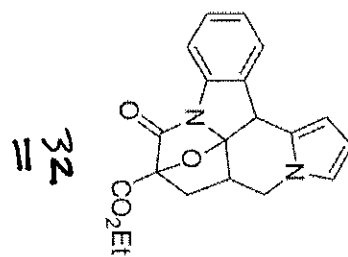
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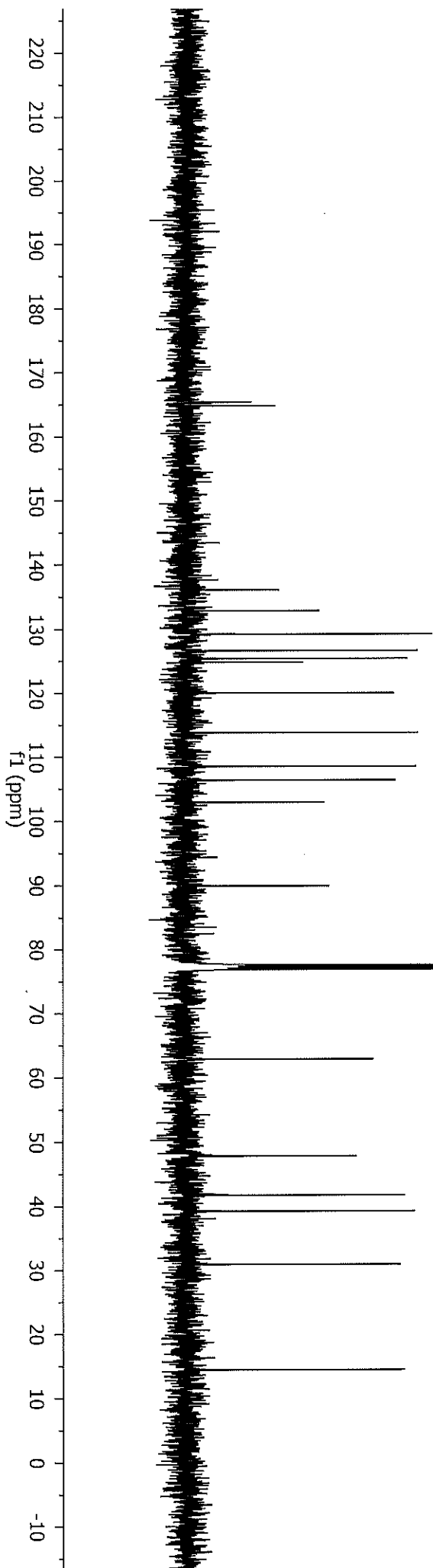
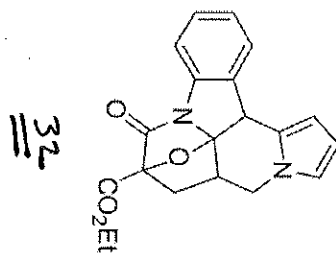




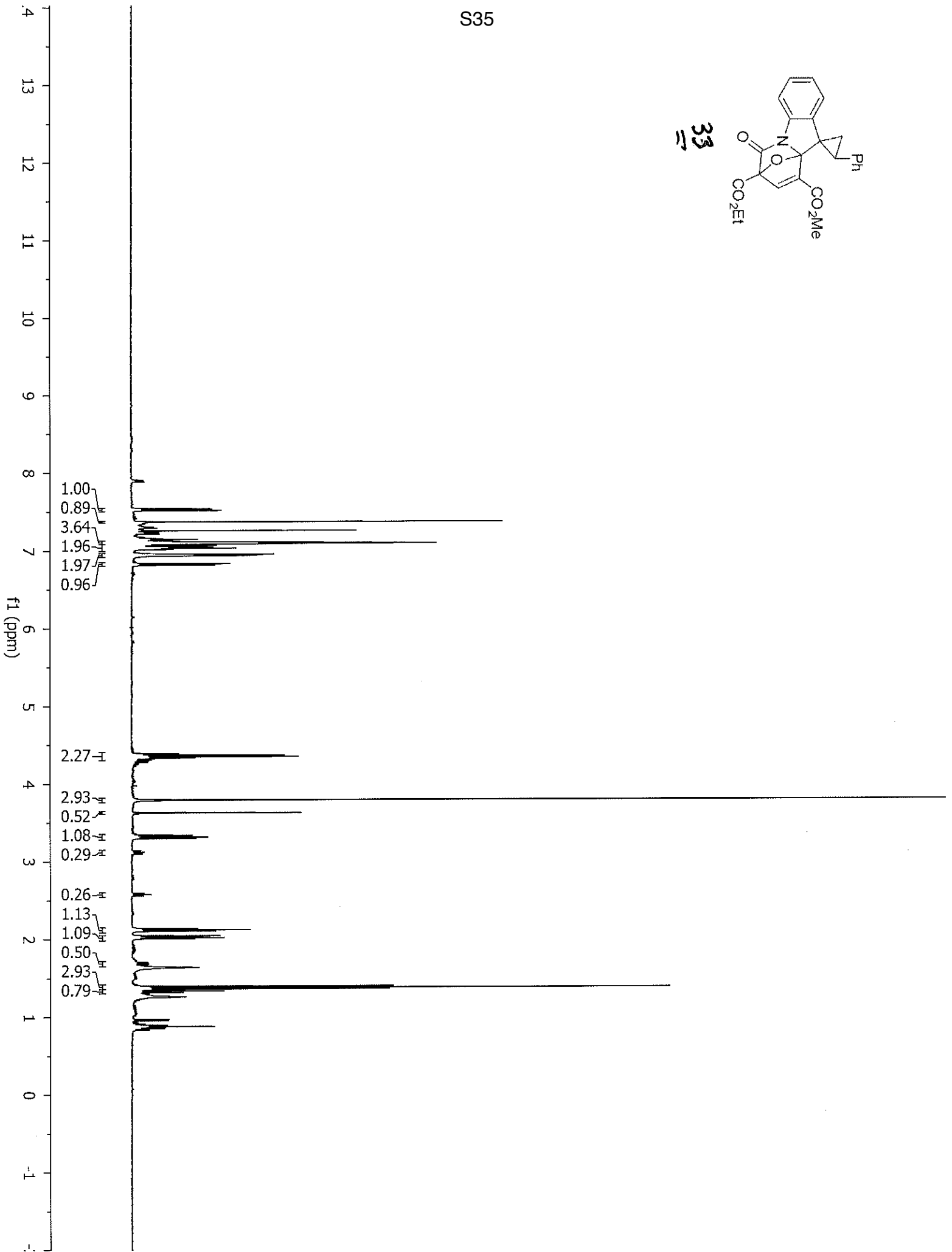
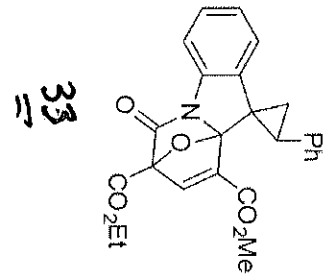
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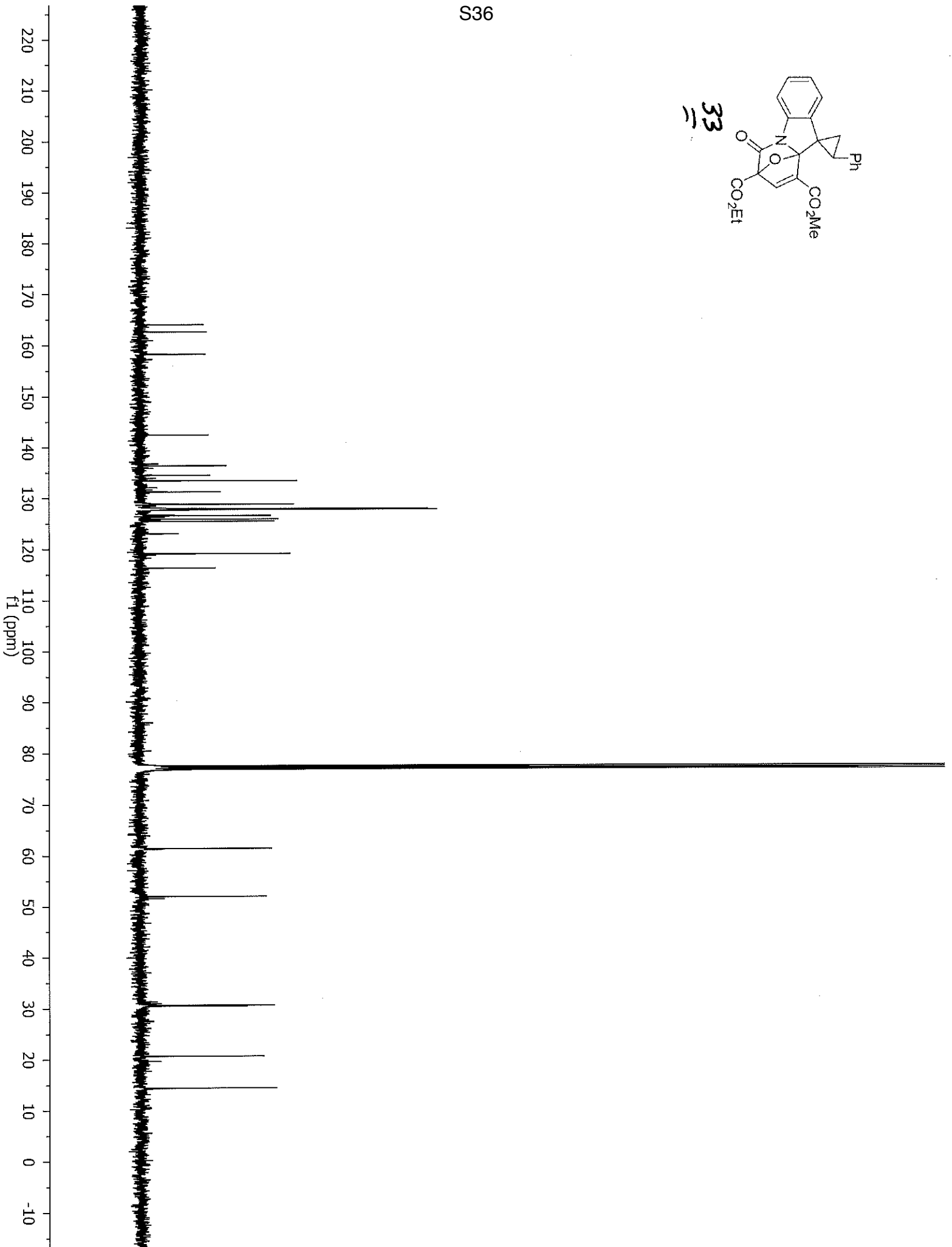
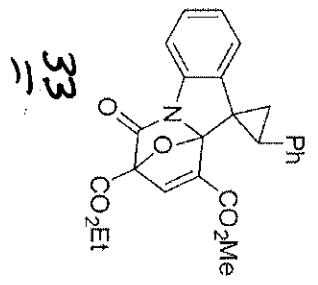
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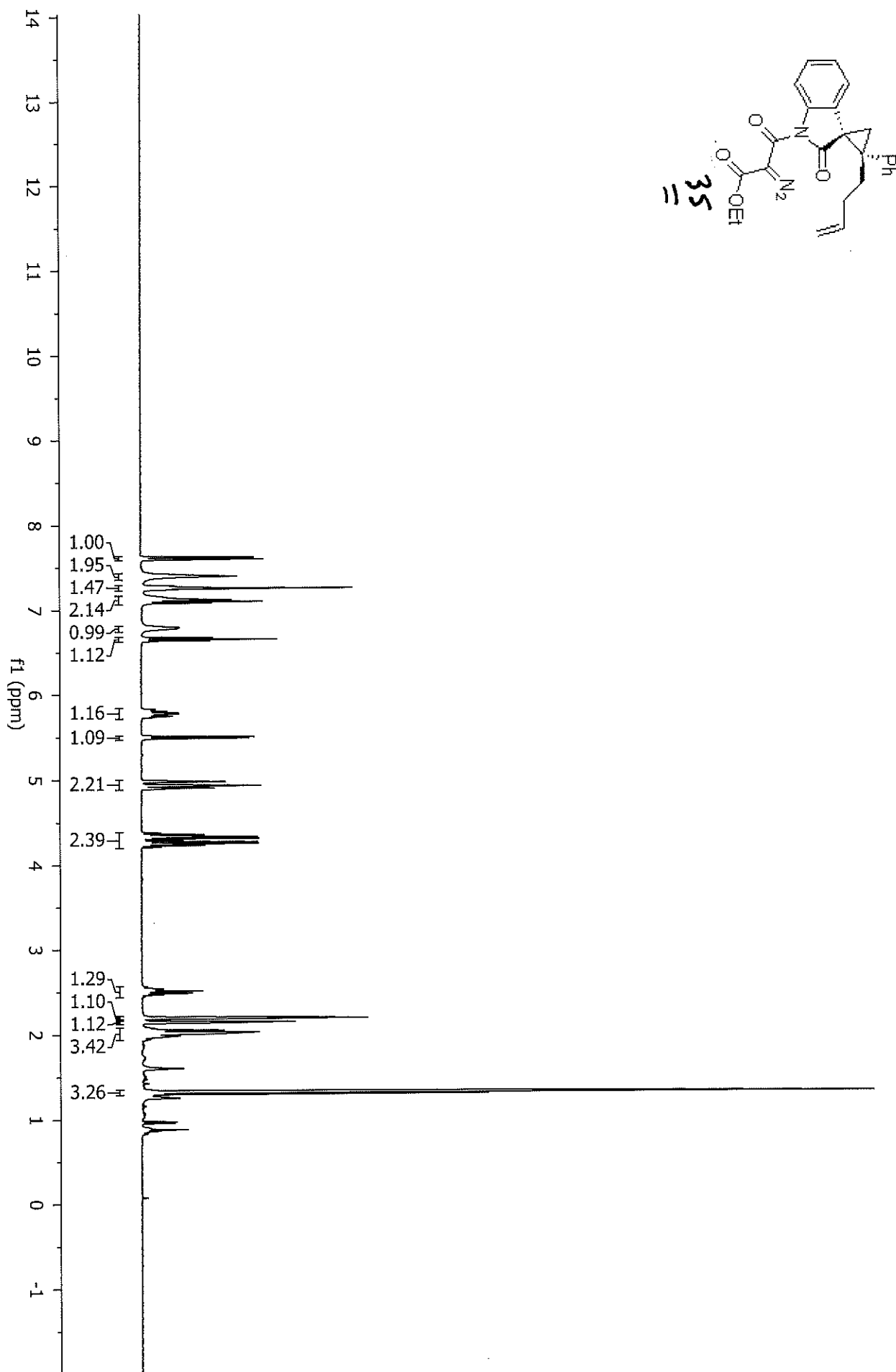


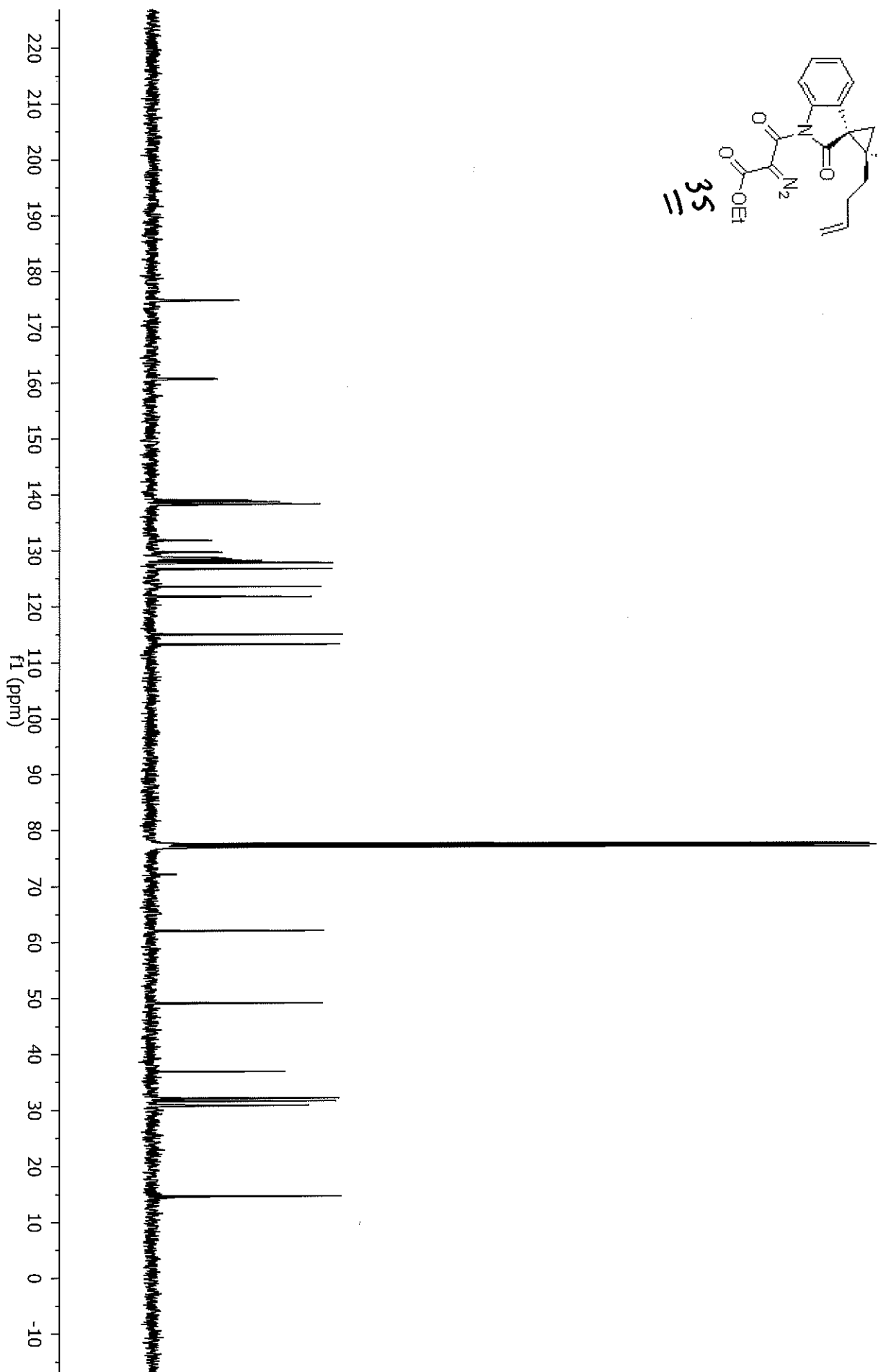
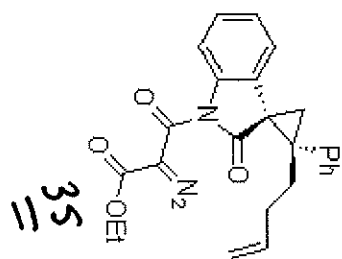
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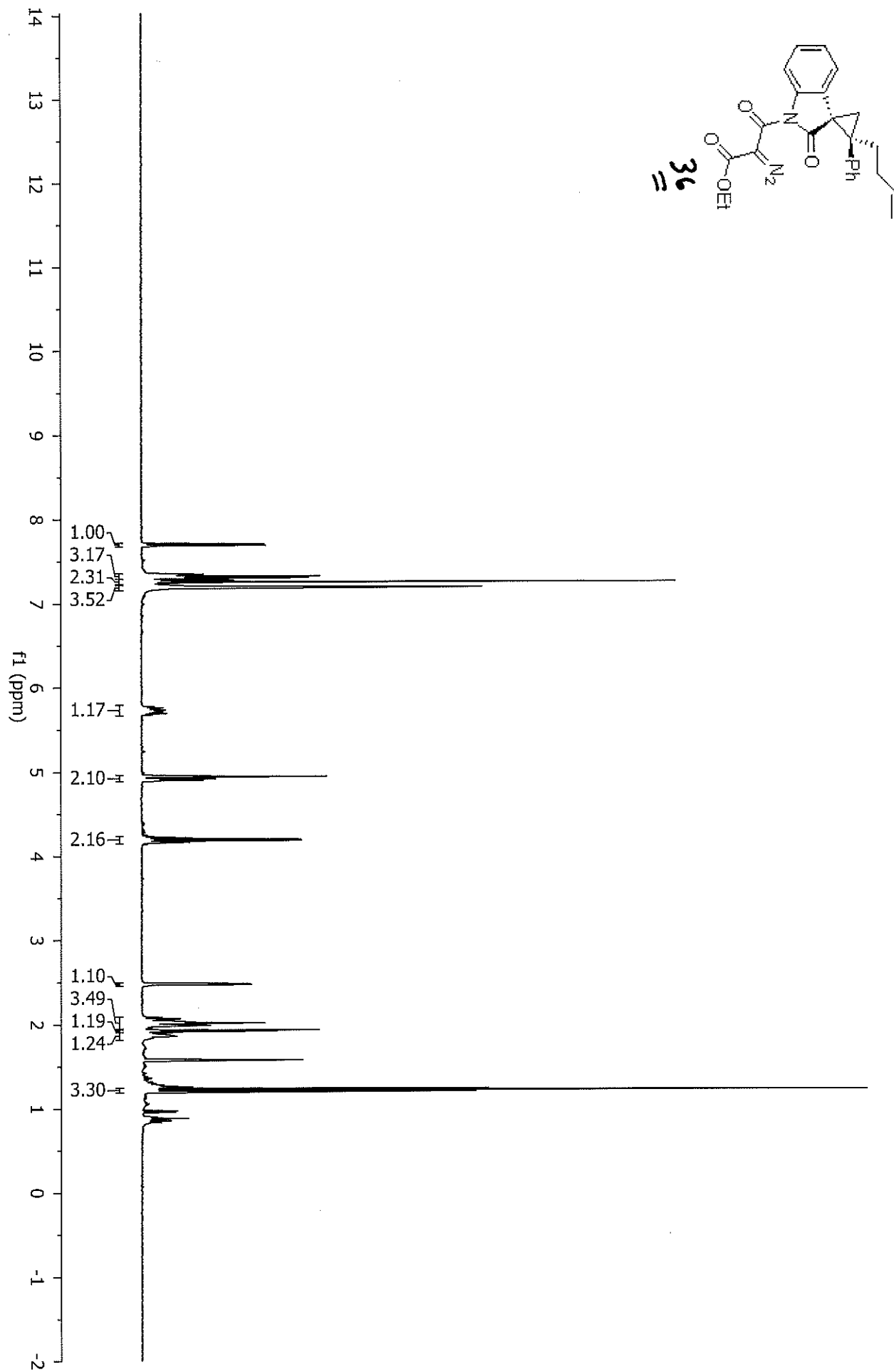


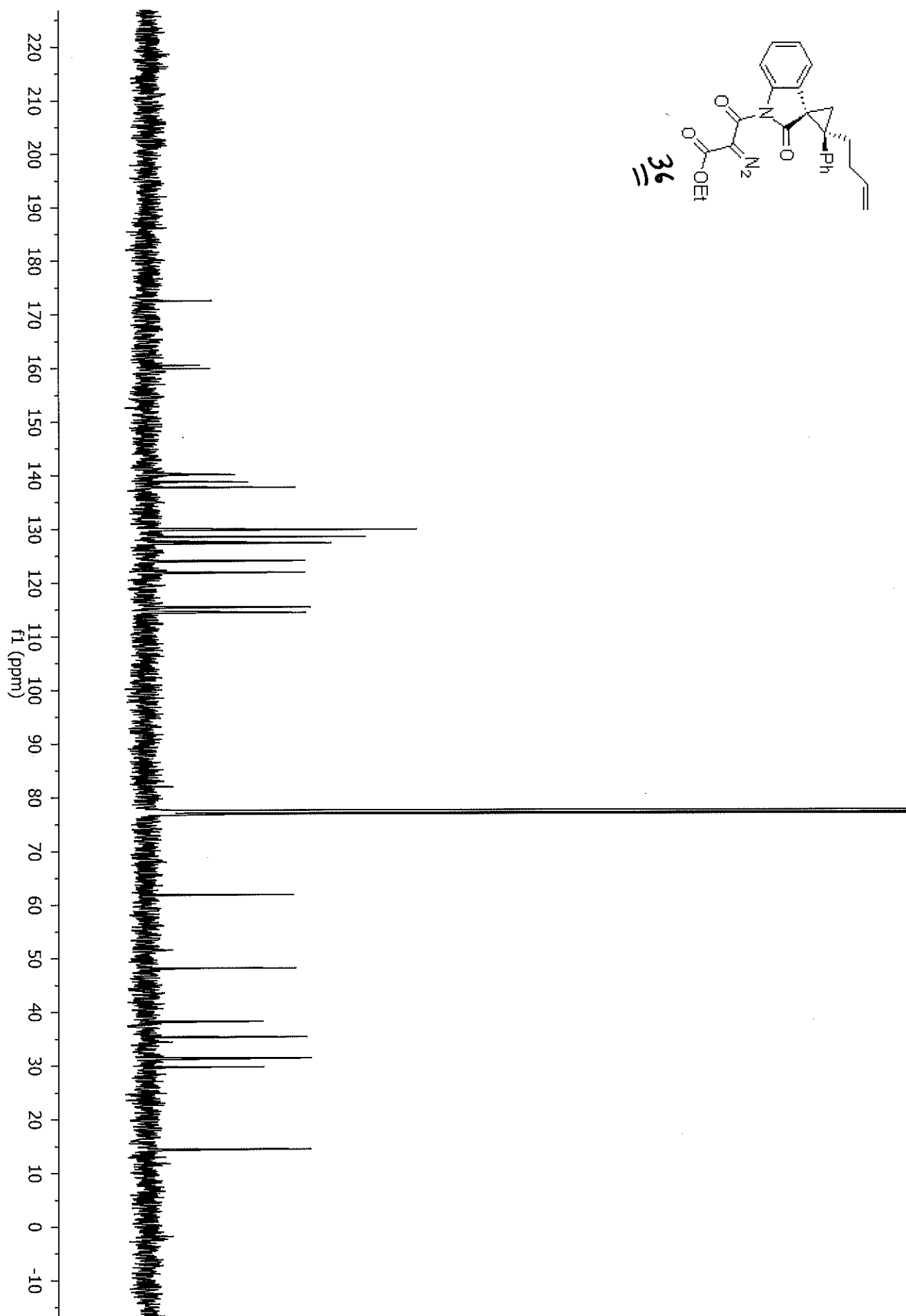
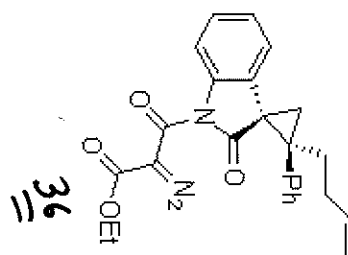
S36



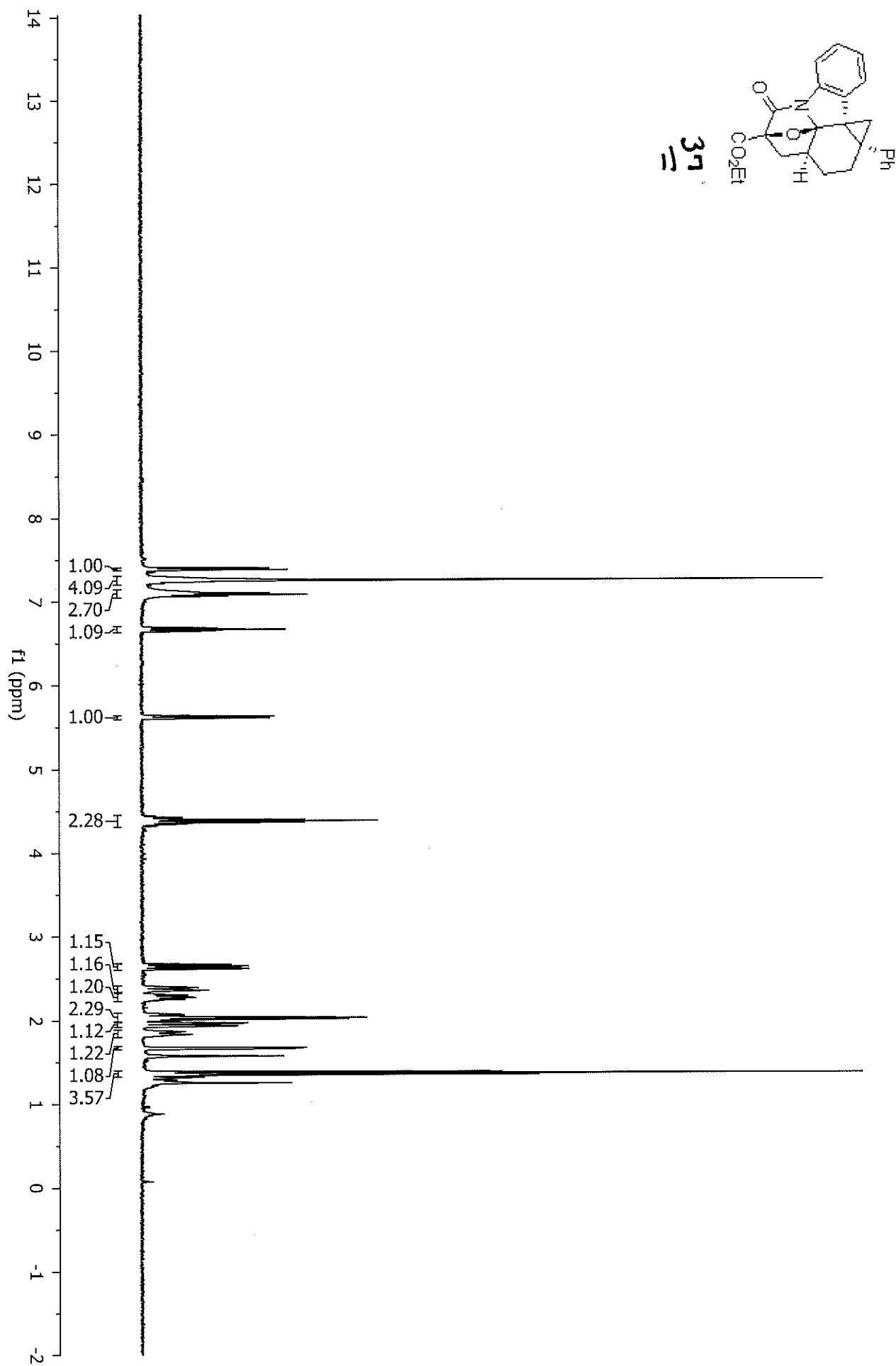


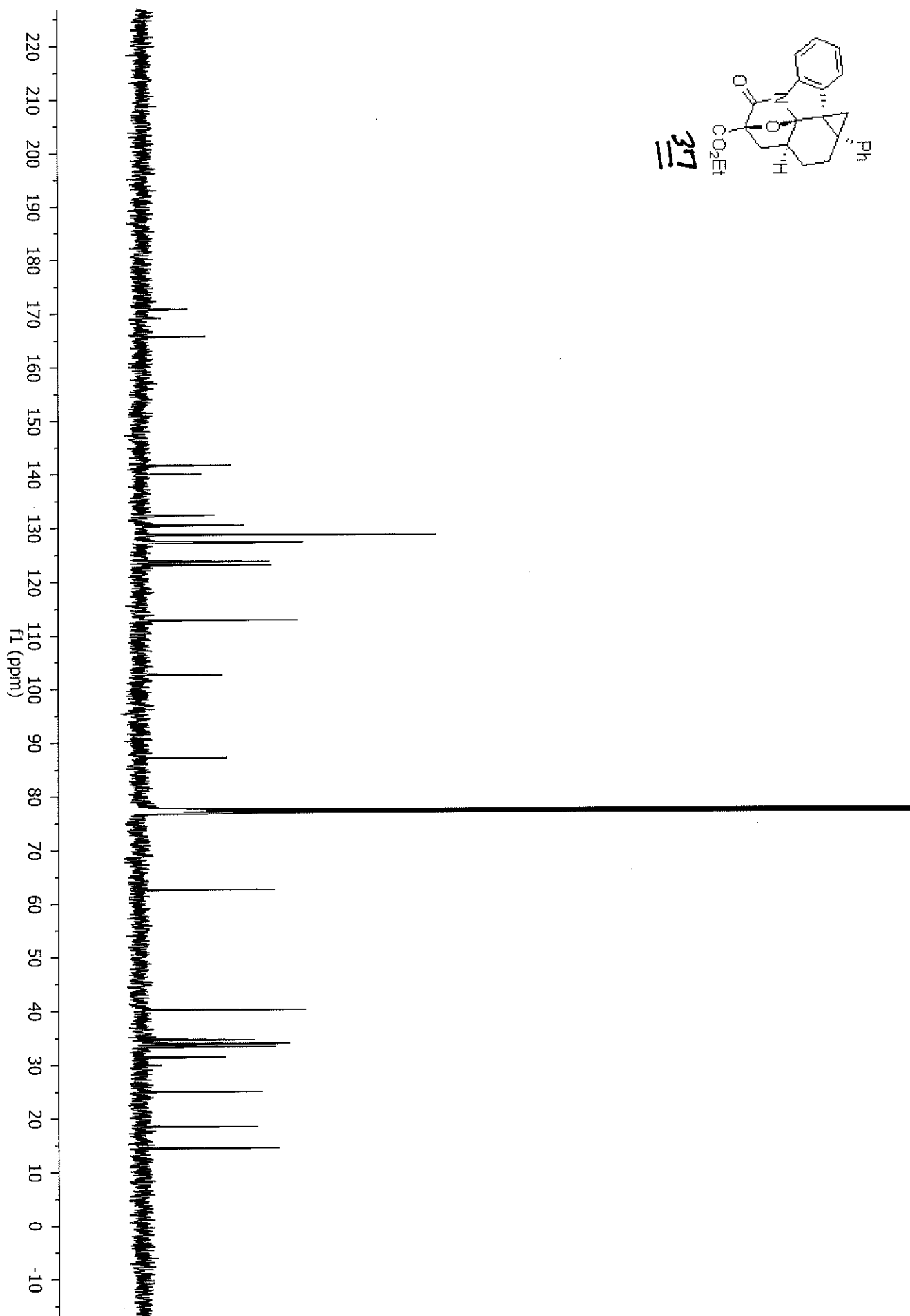
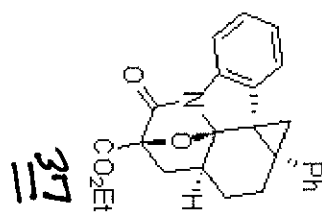


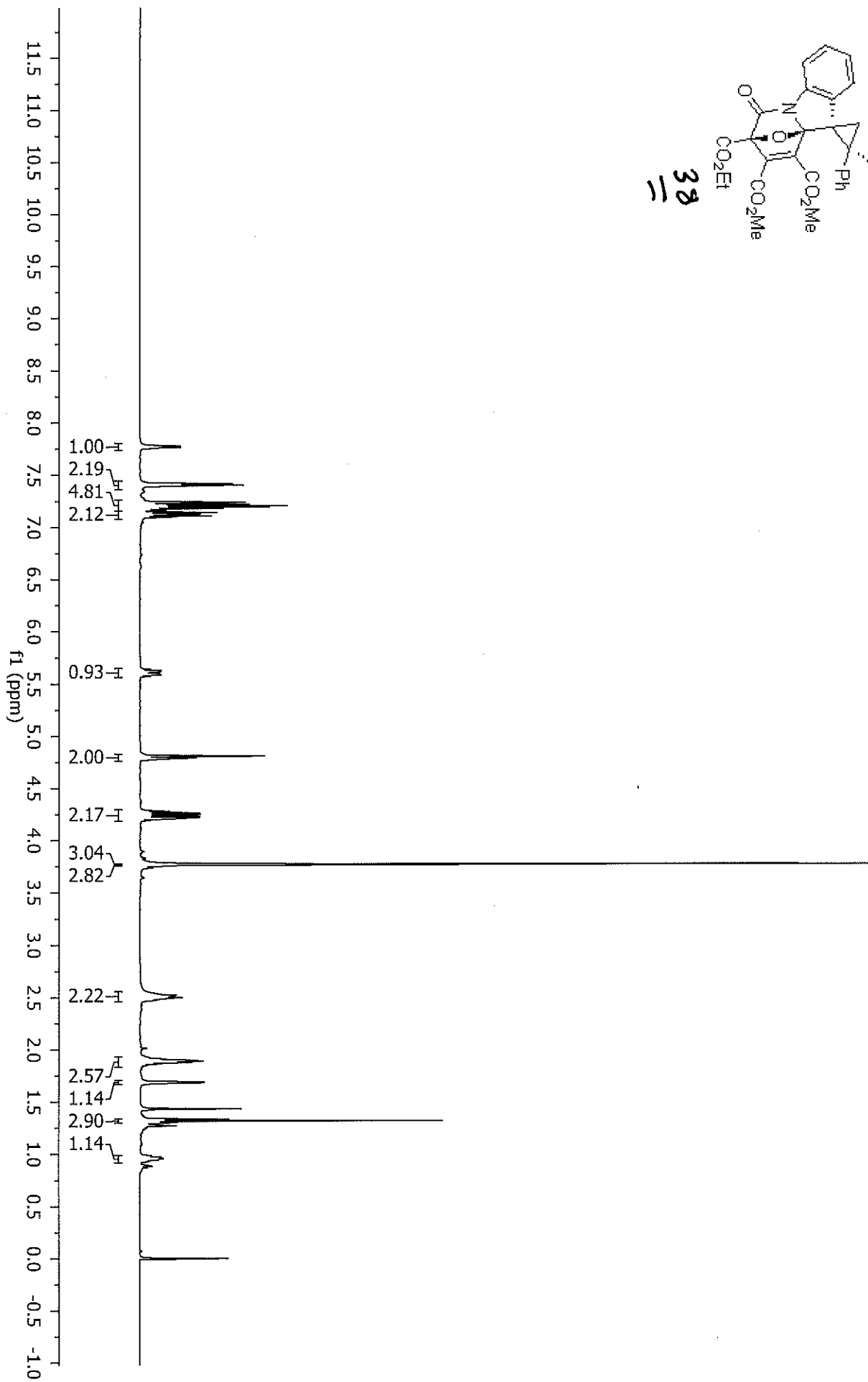
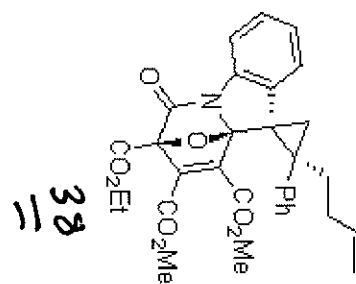


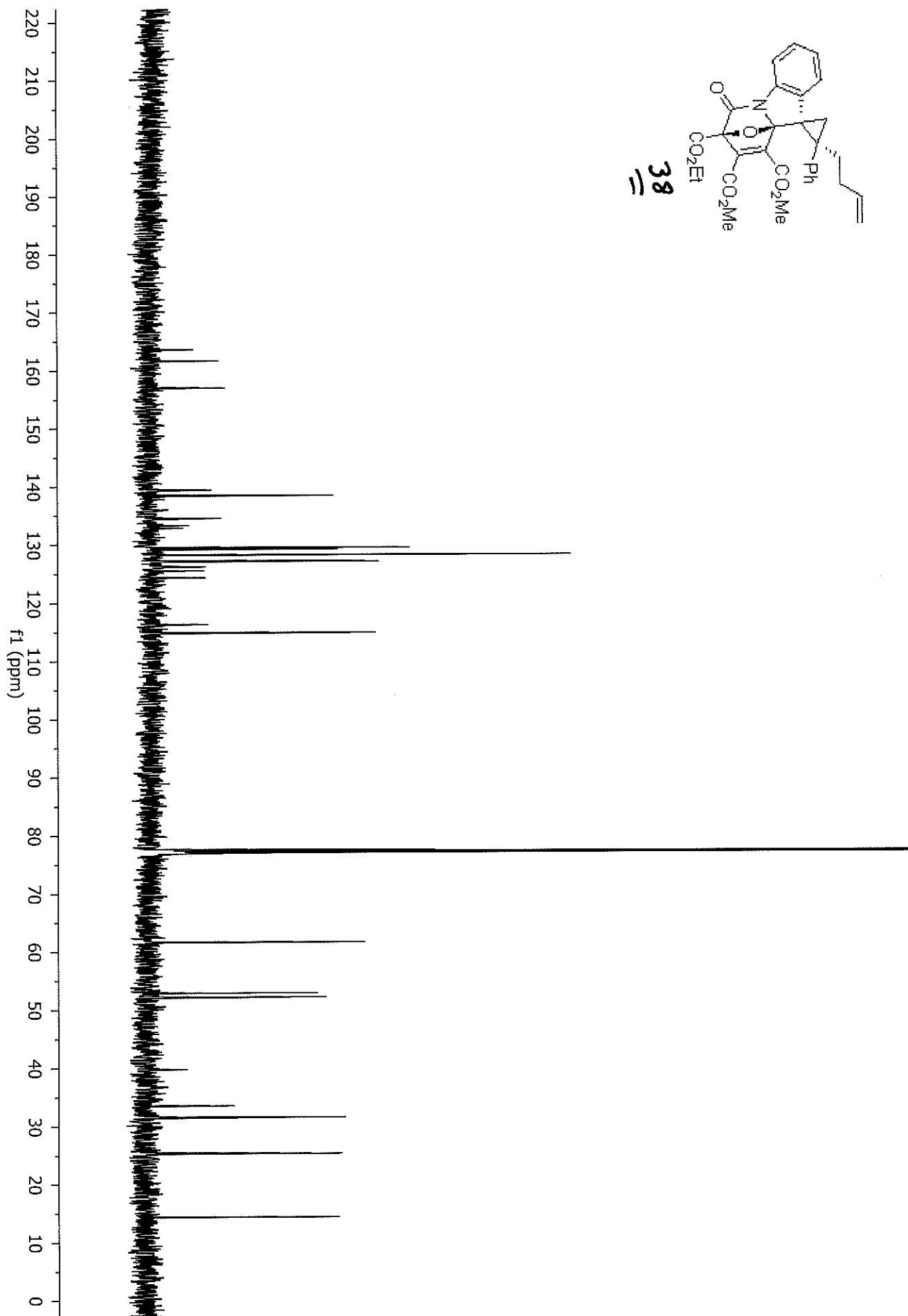
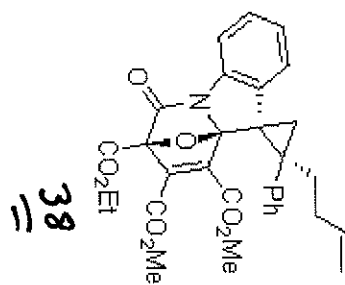


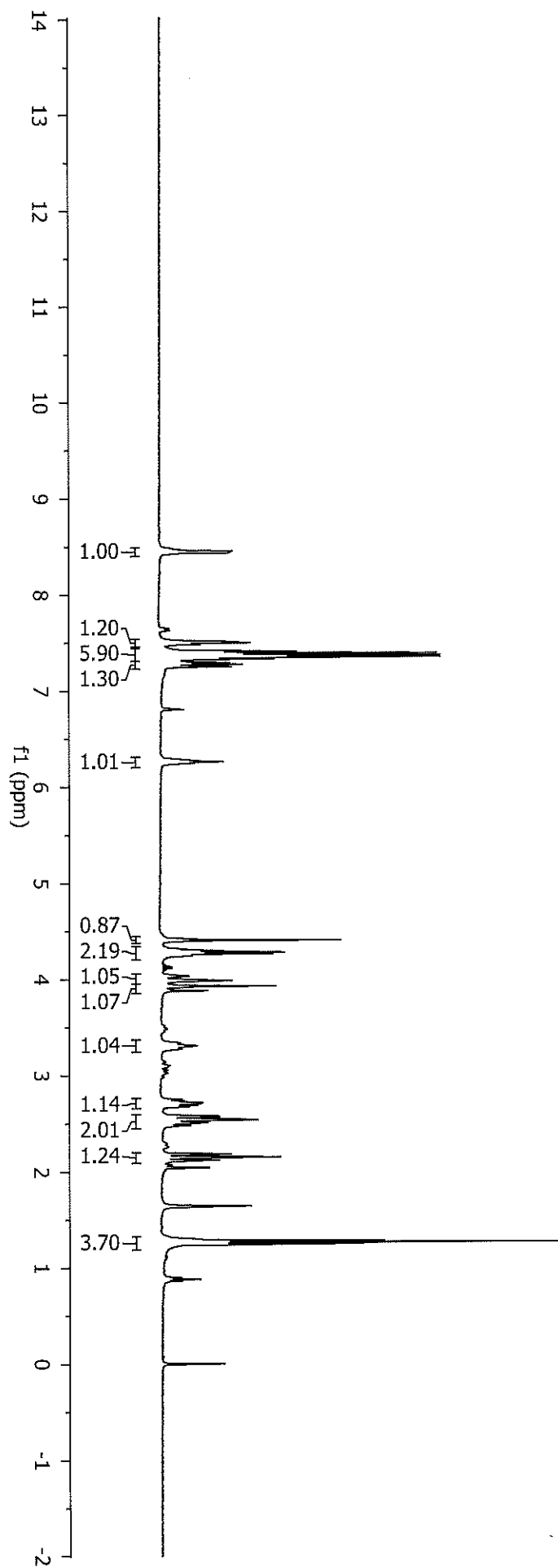
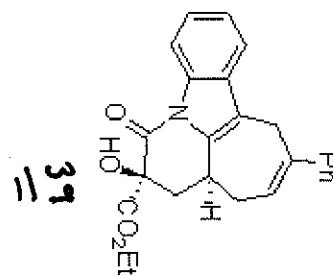


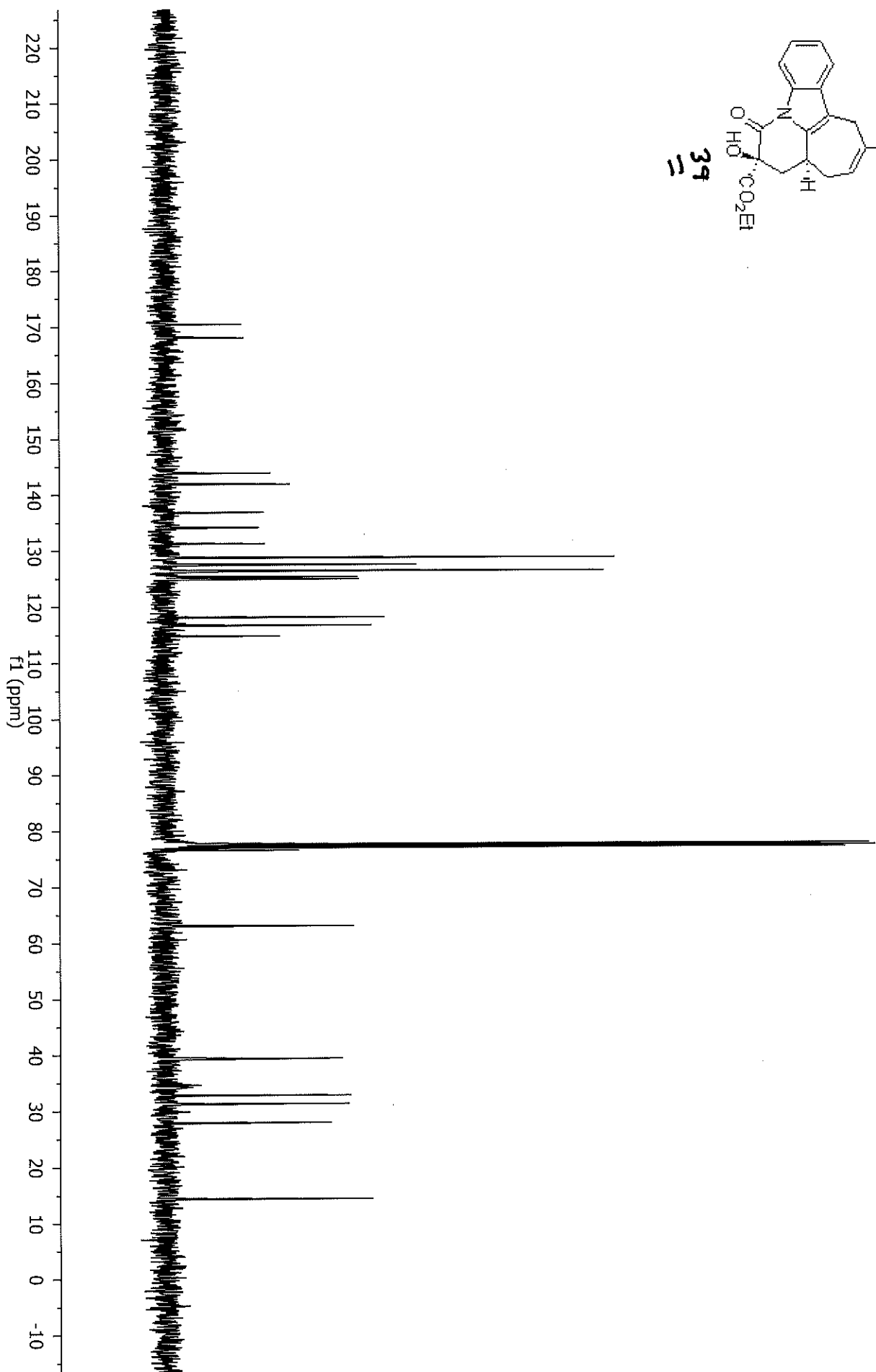
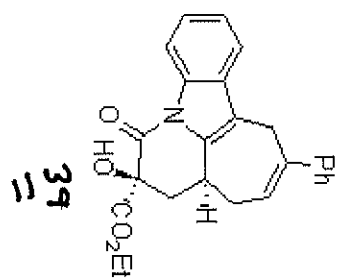


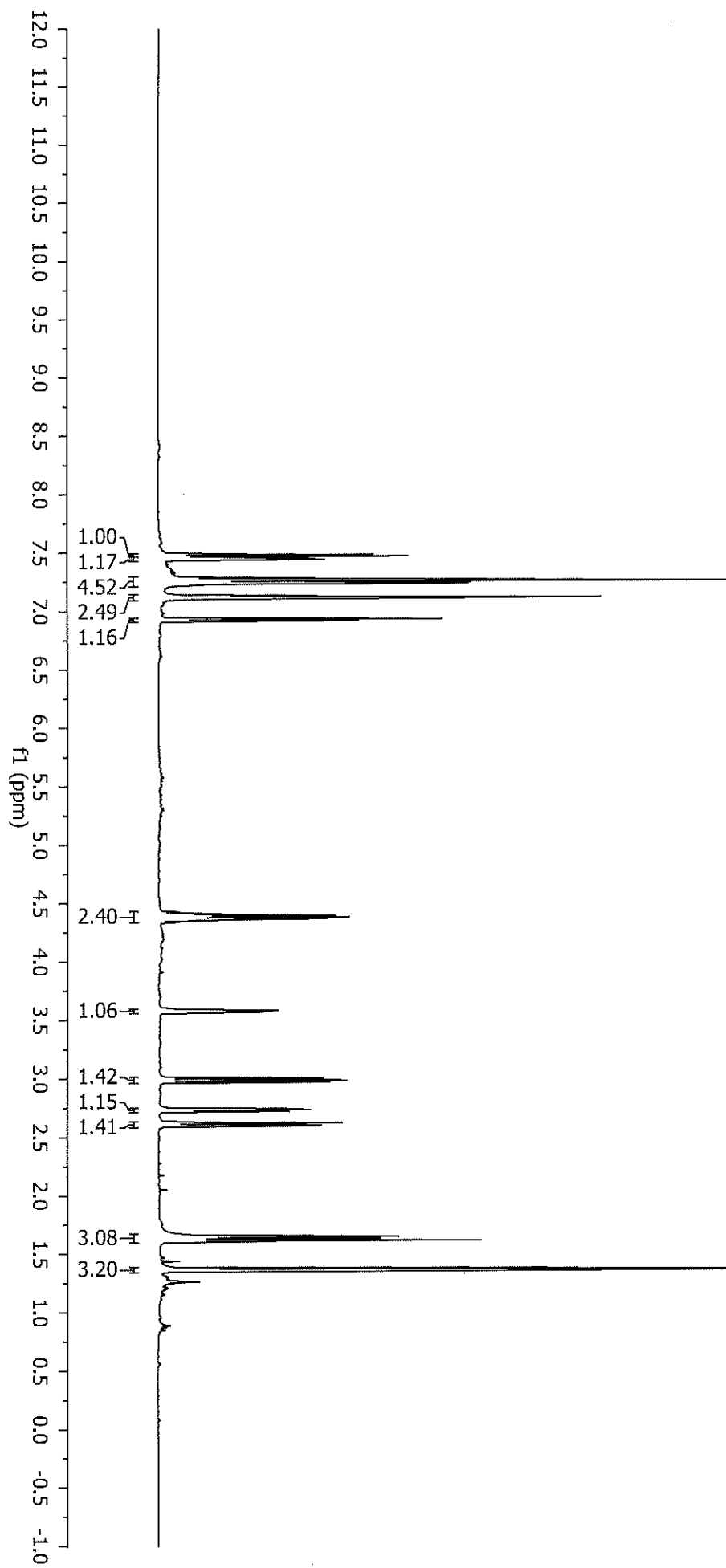


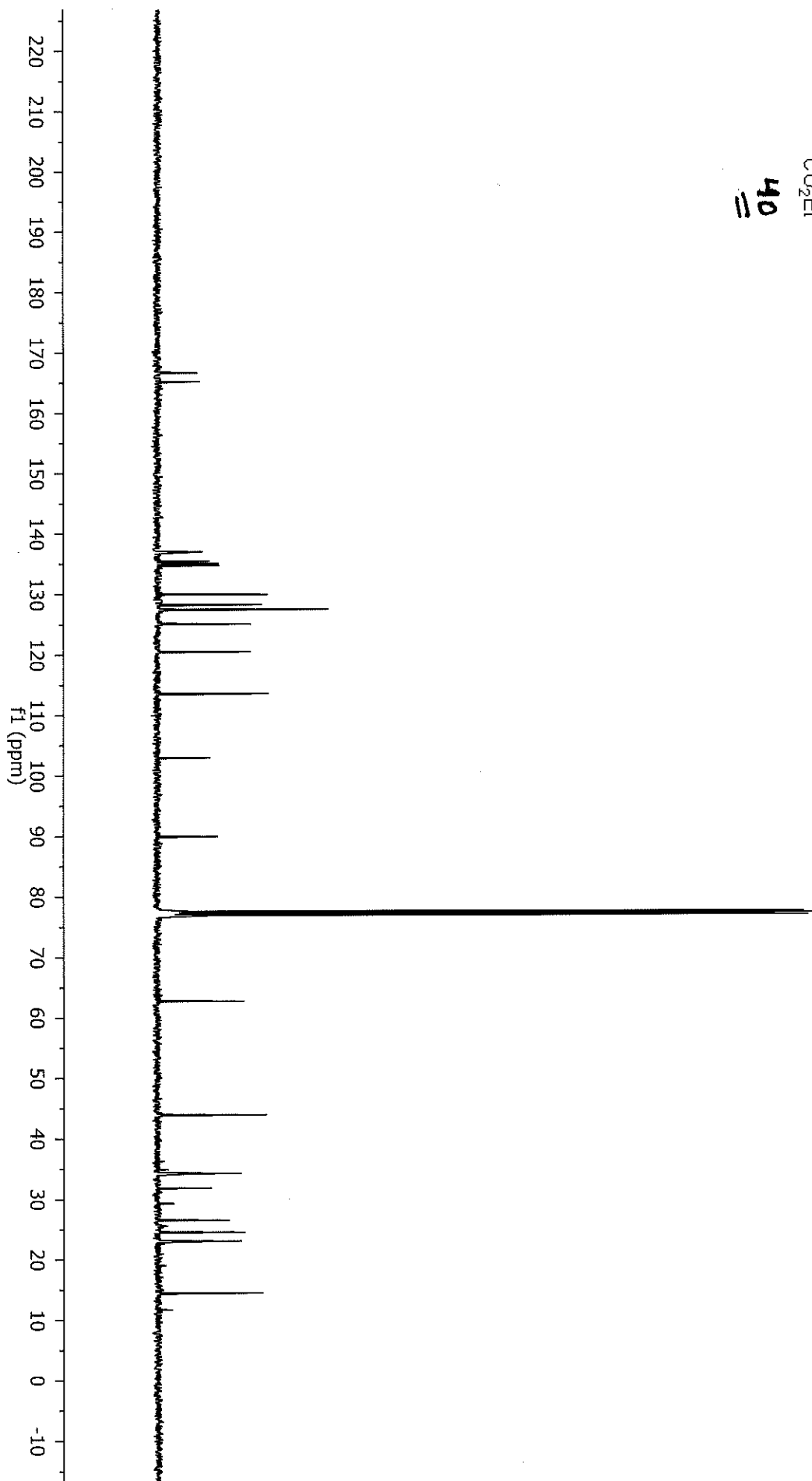
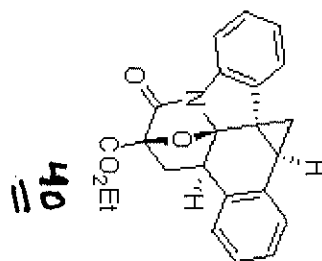




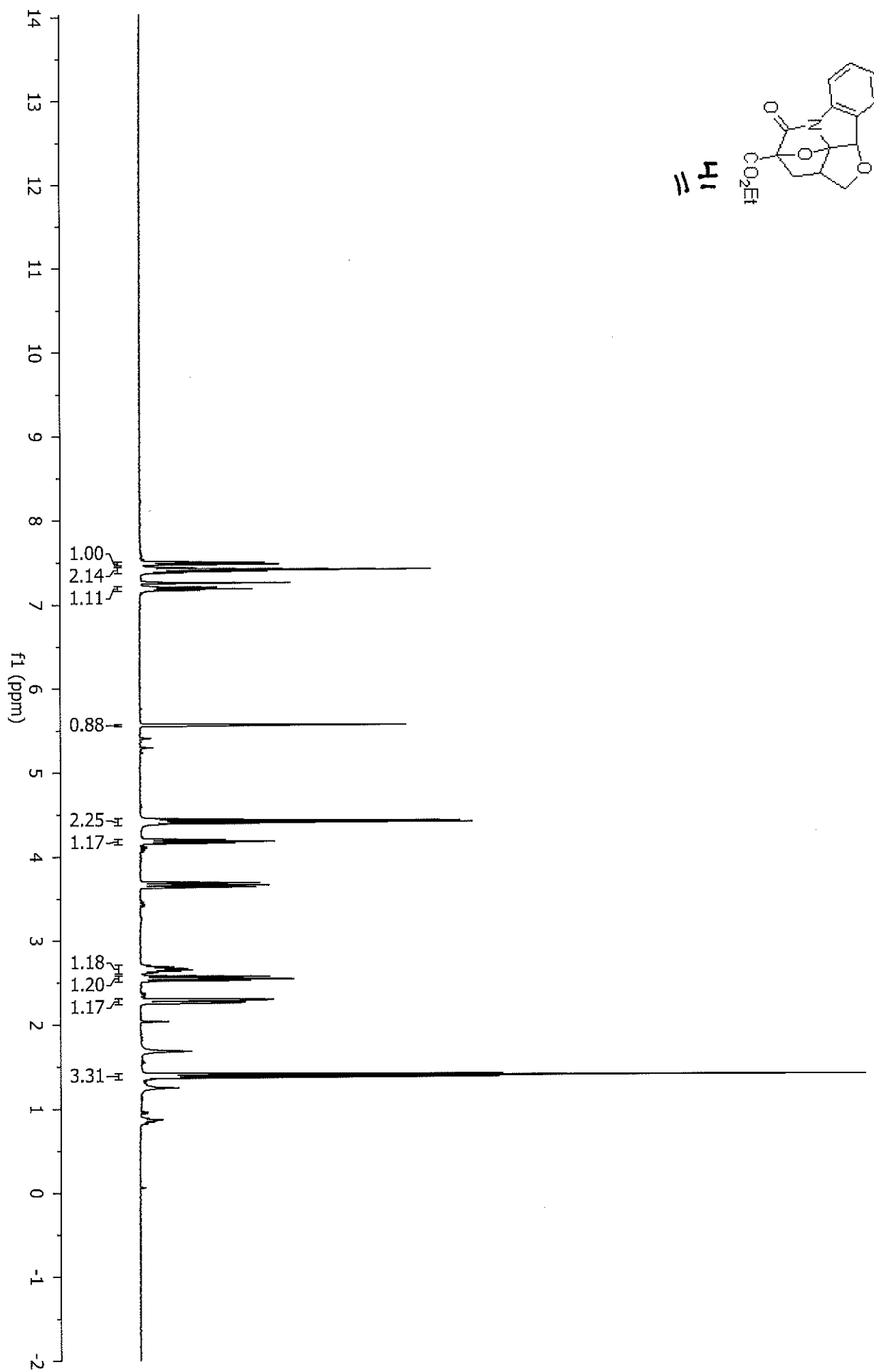


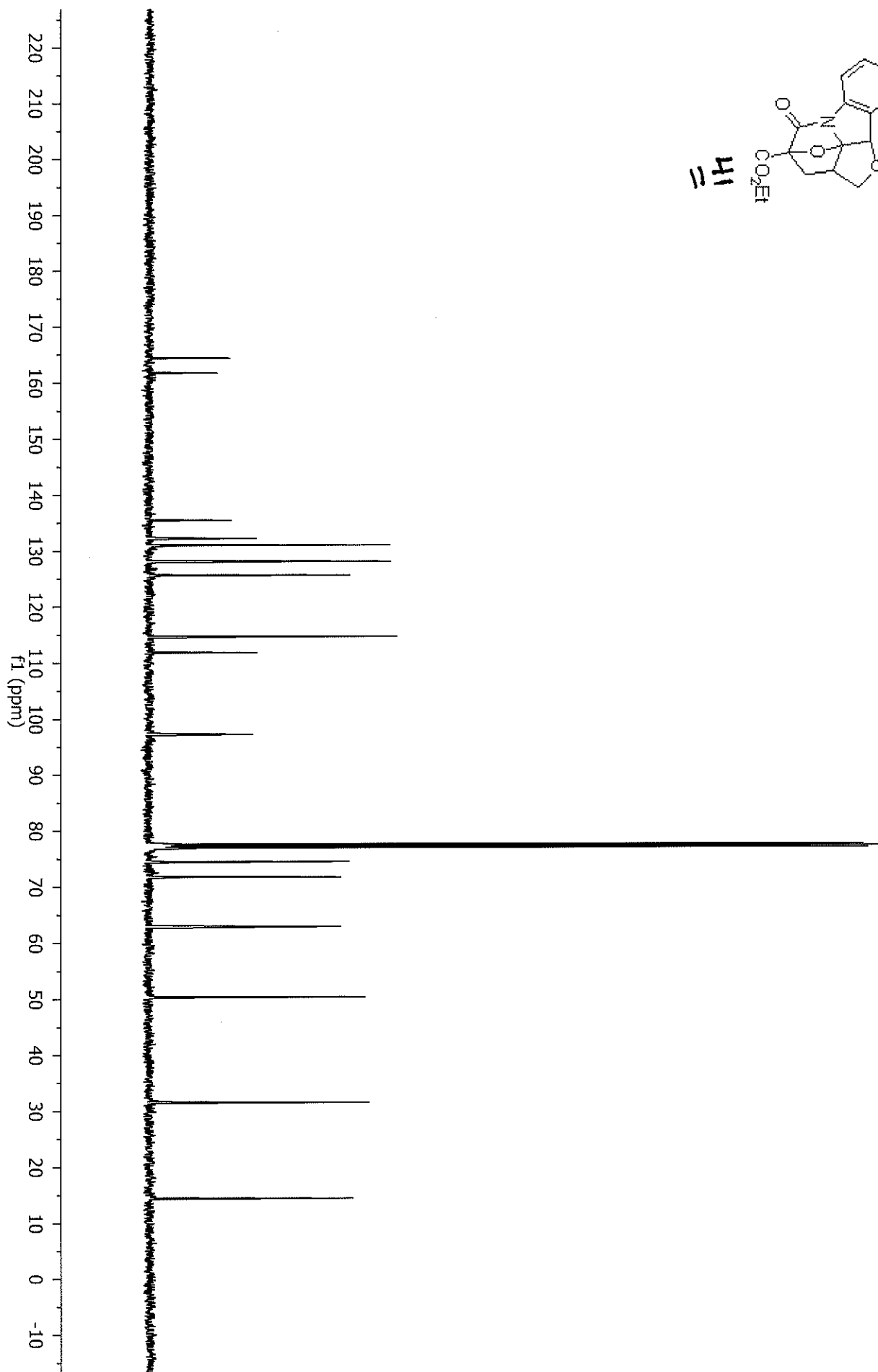
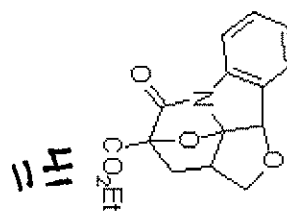


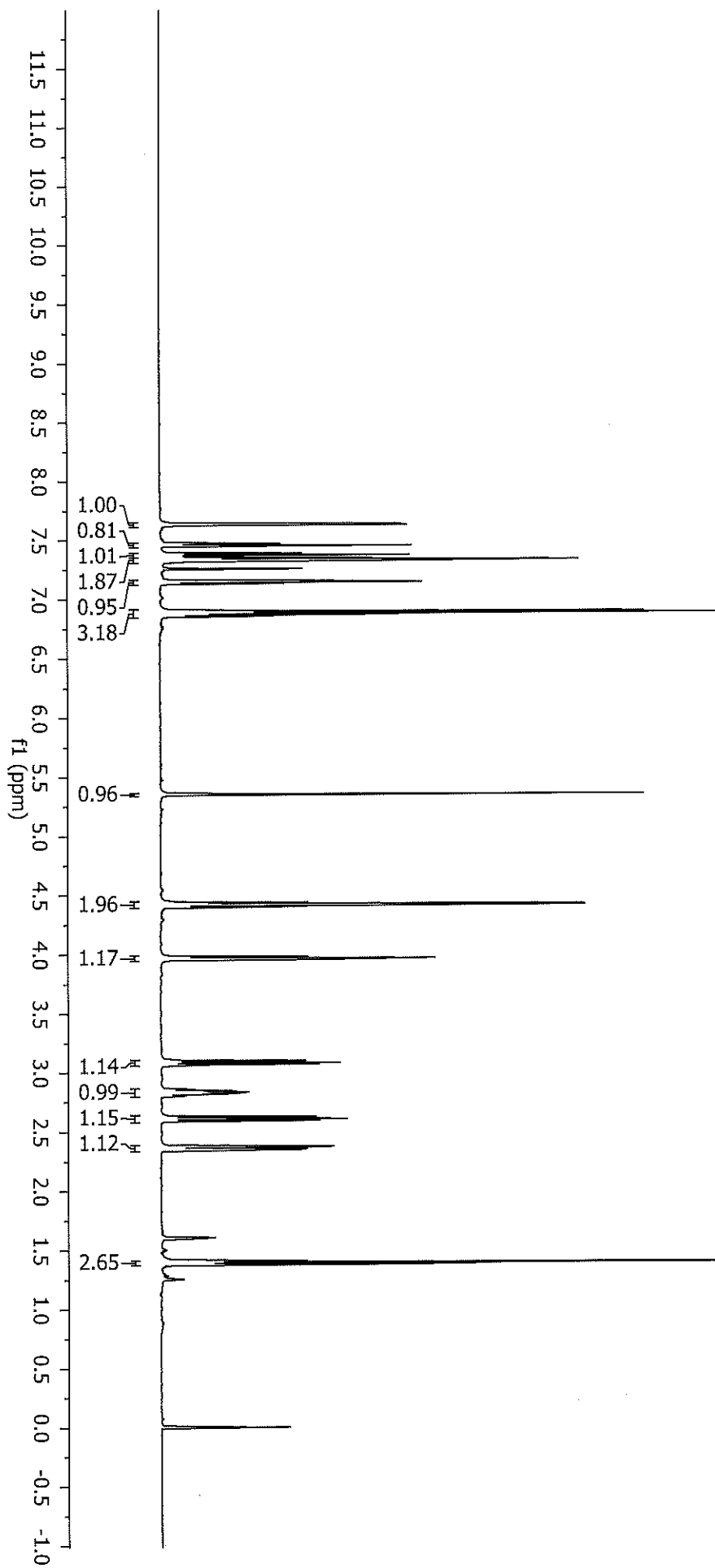
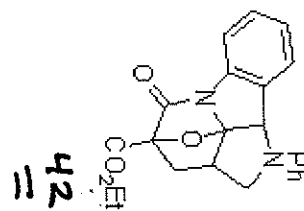


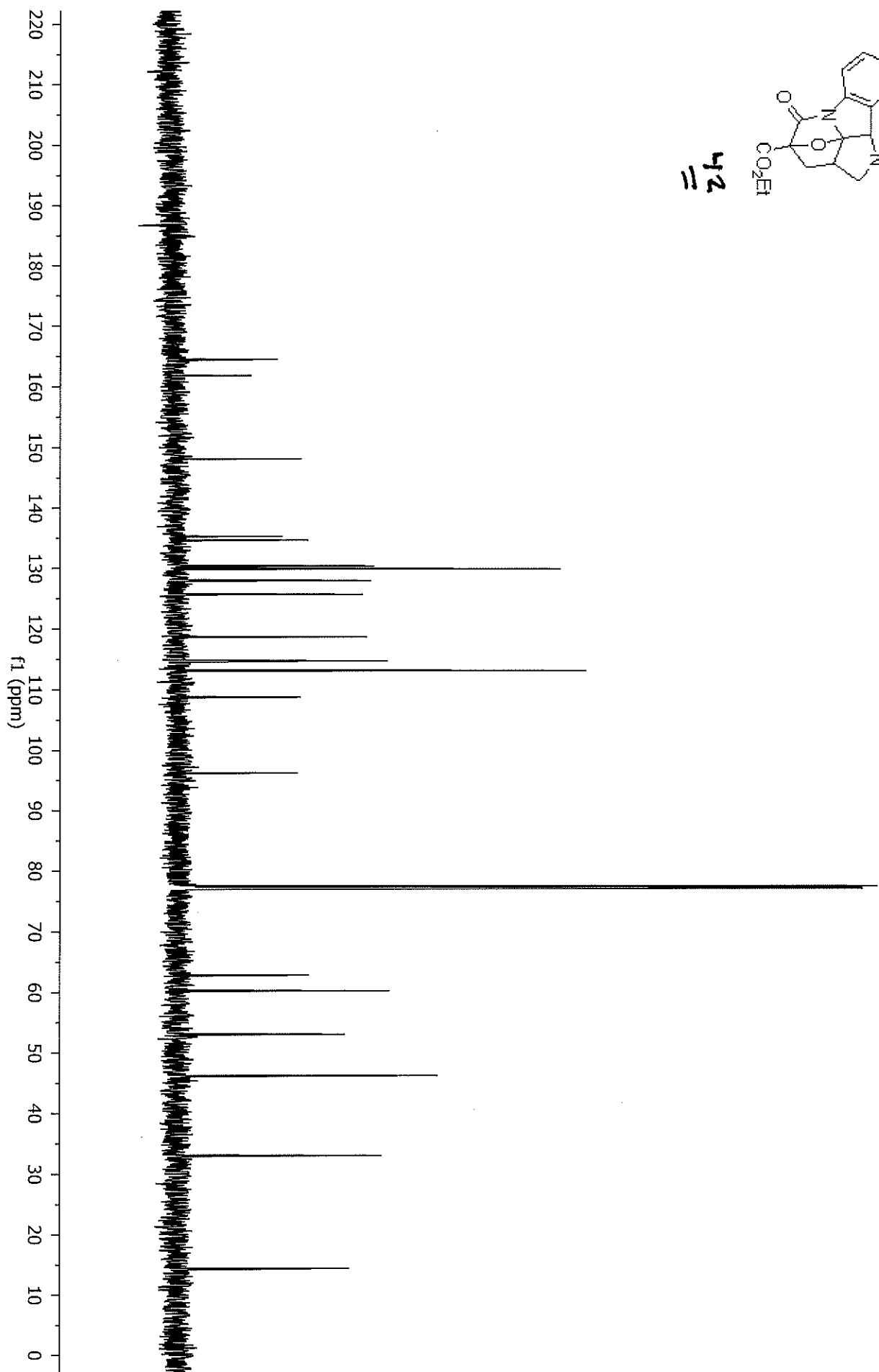
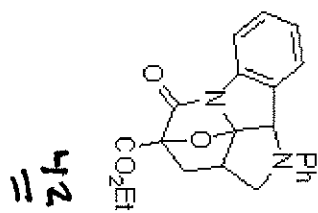


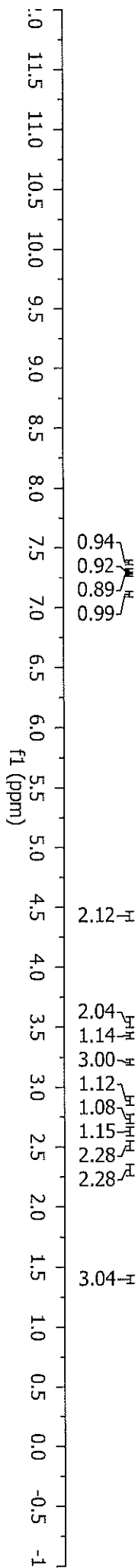
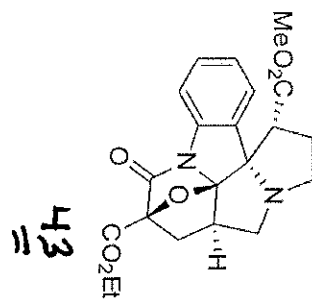


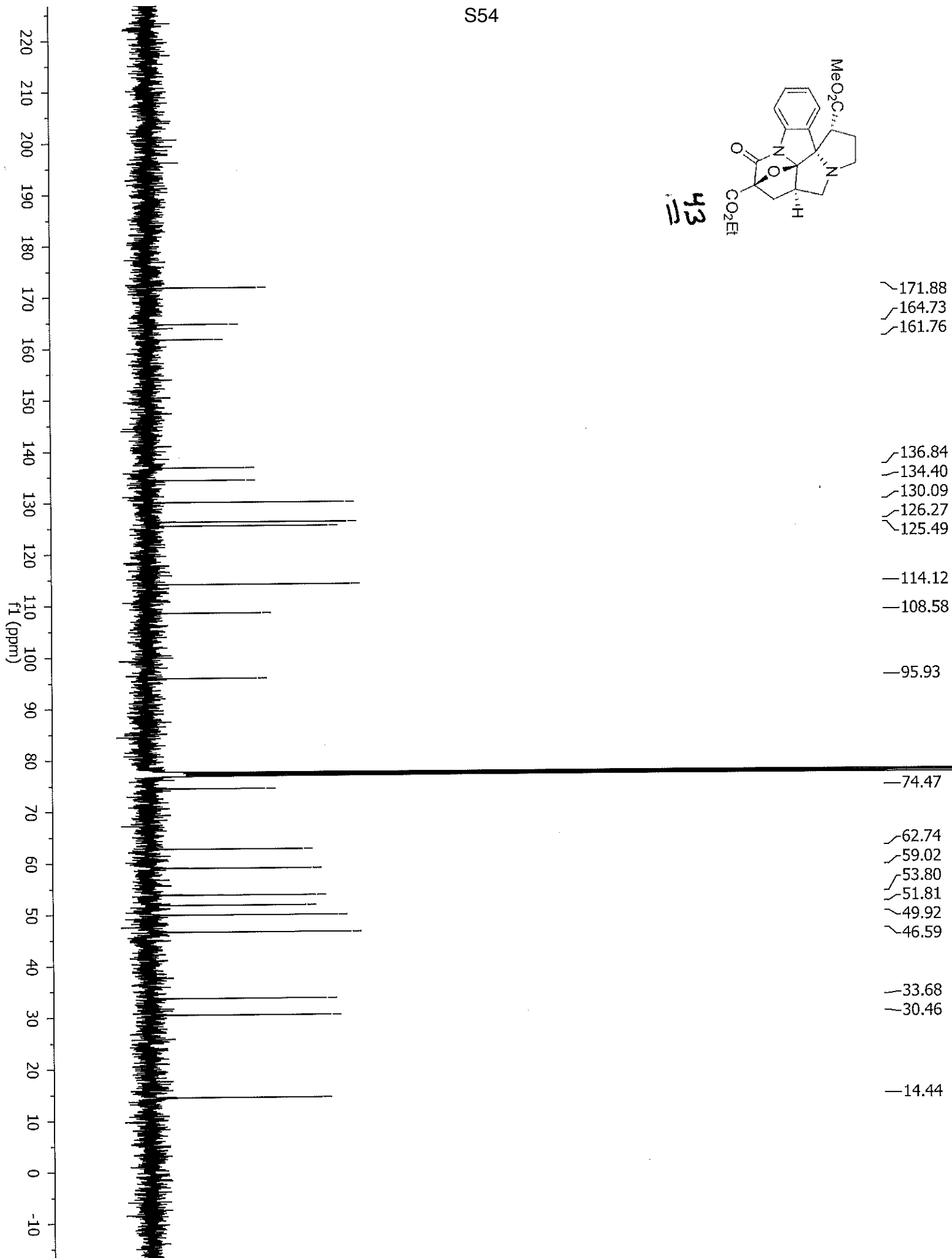
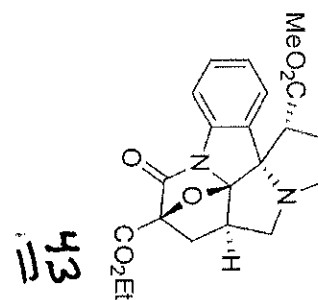




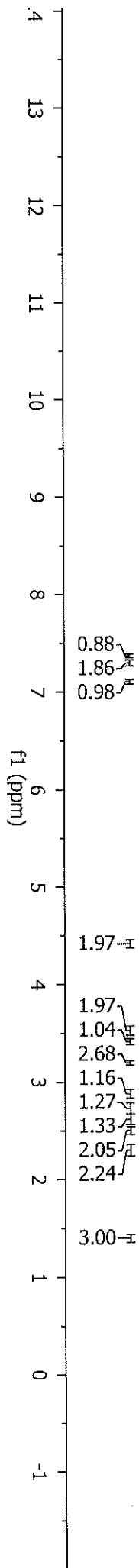
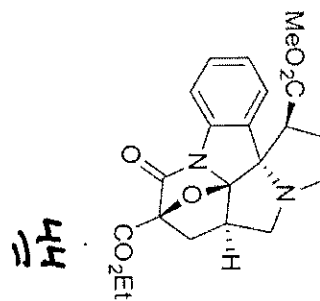


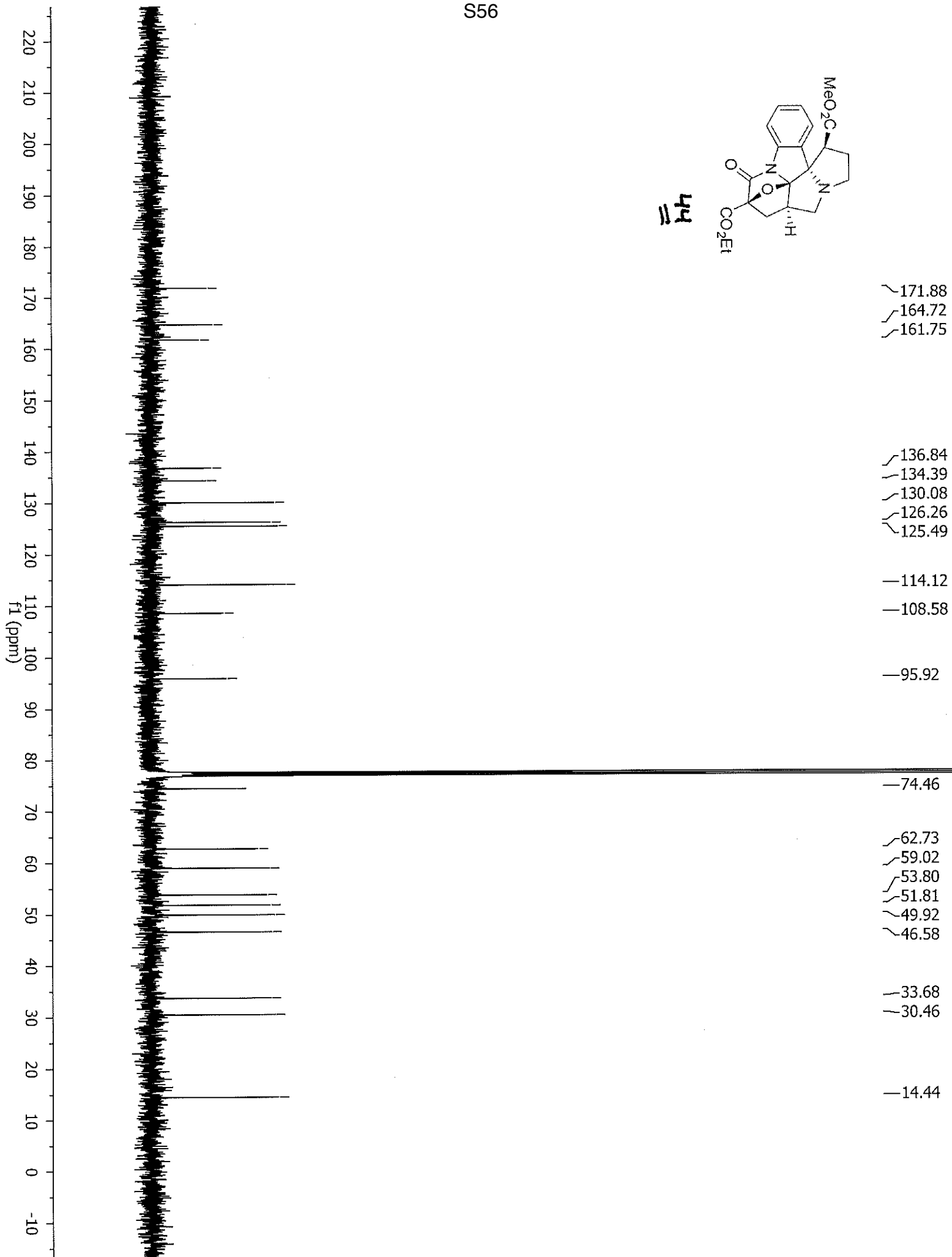
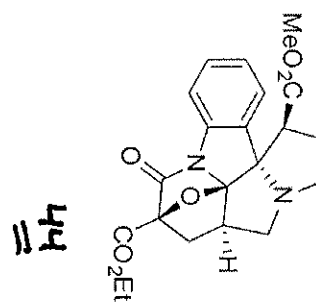






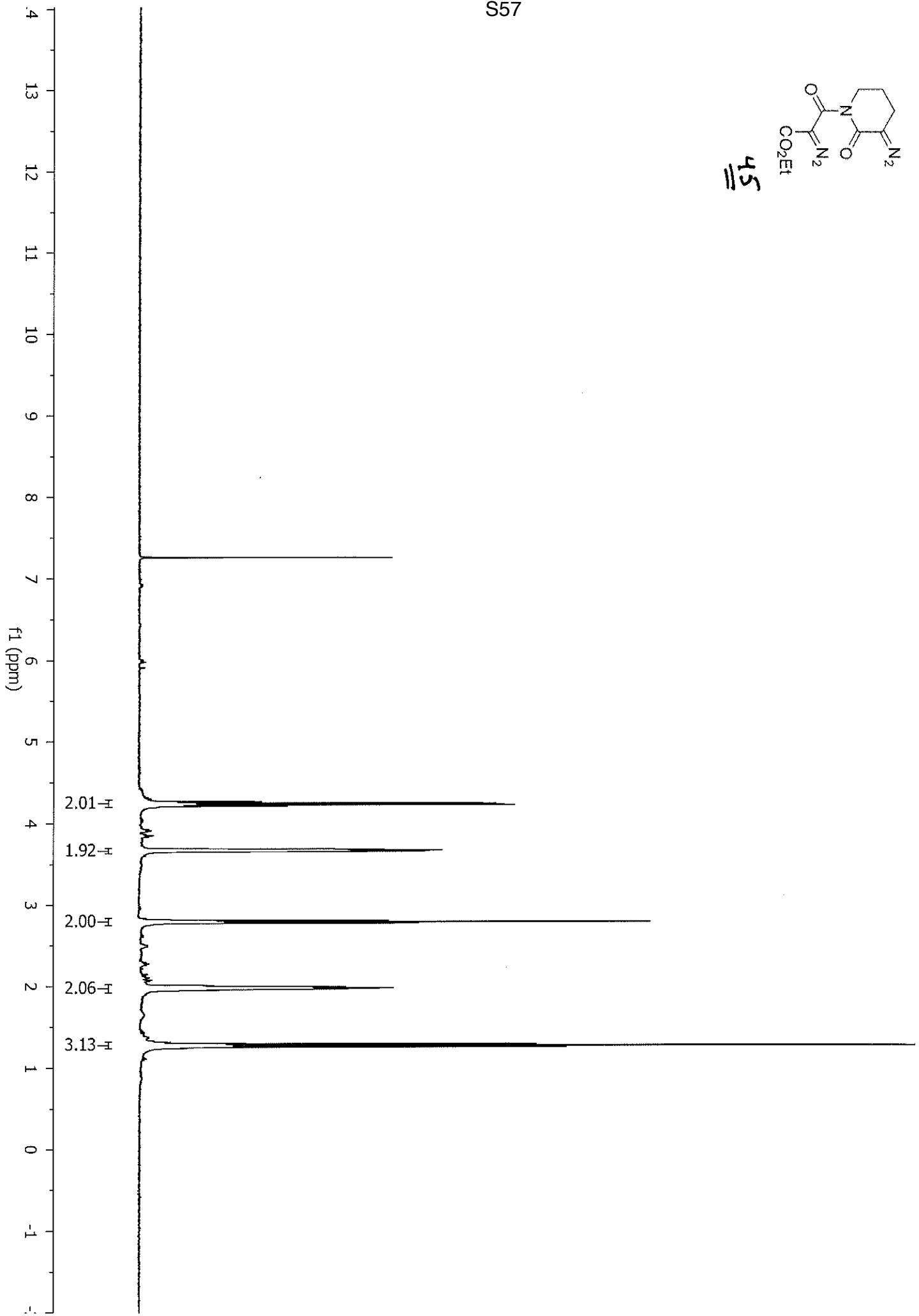
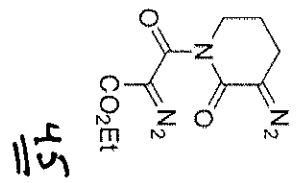
S55



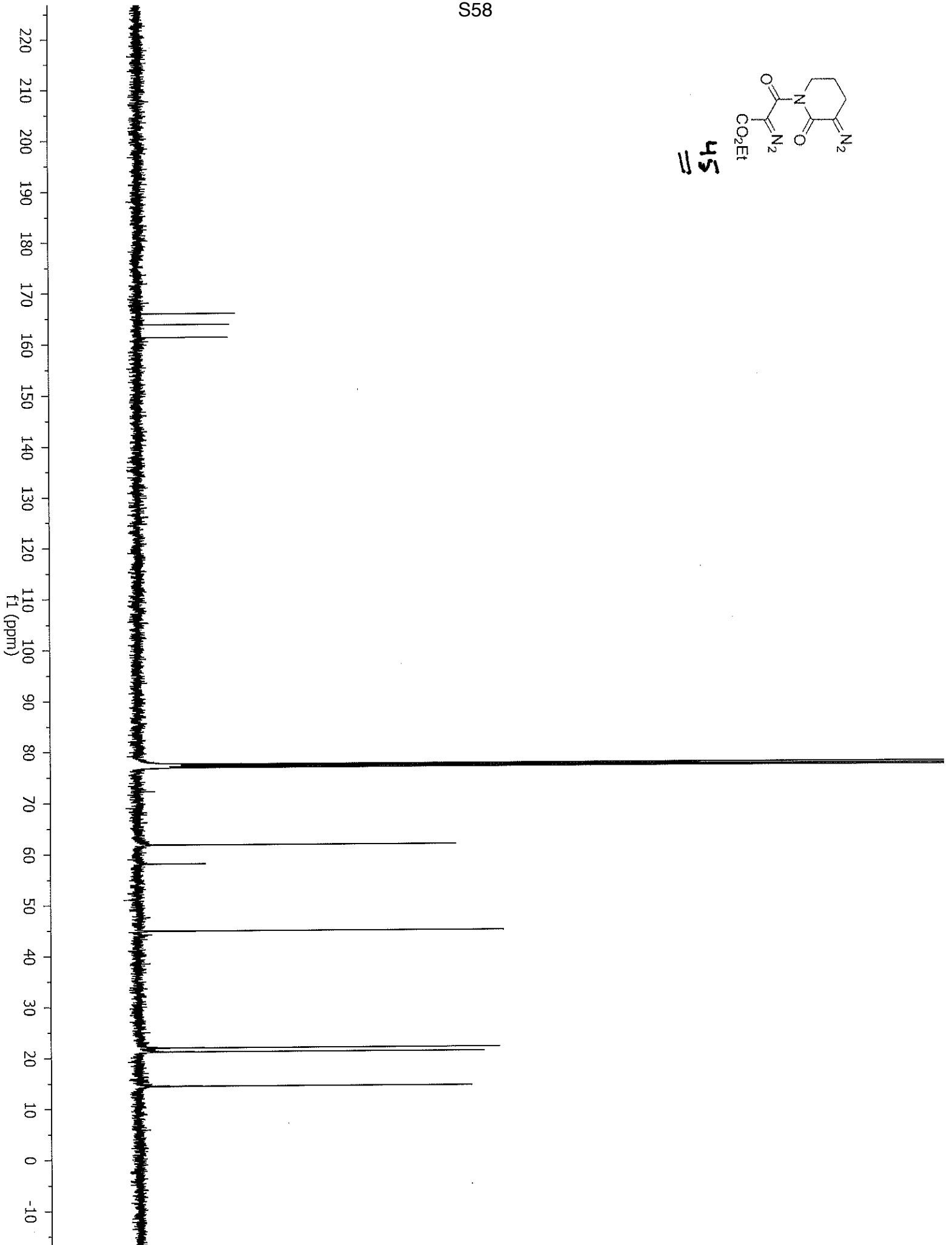
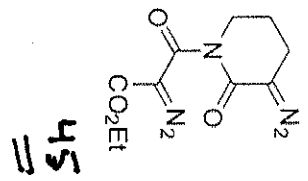


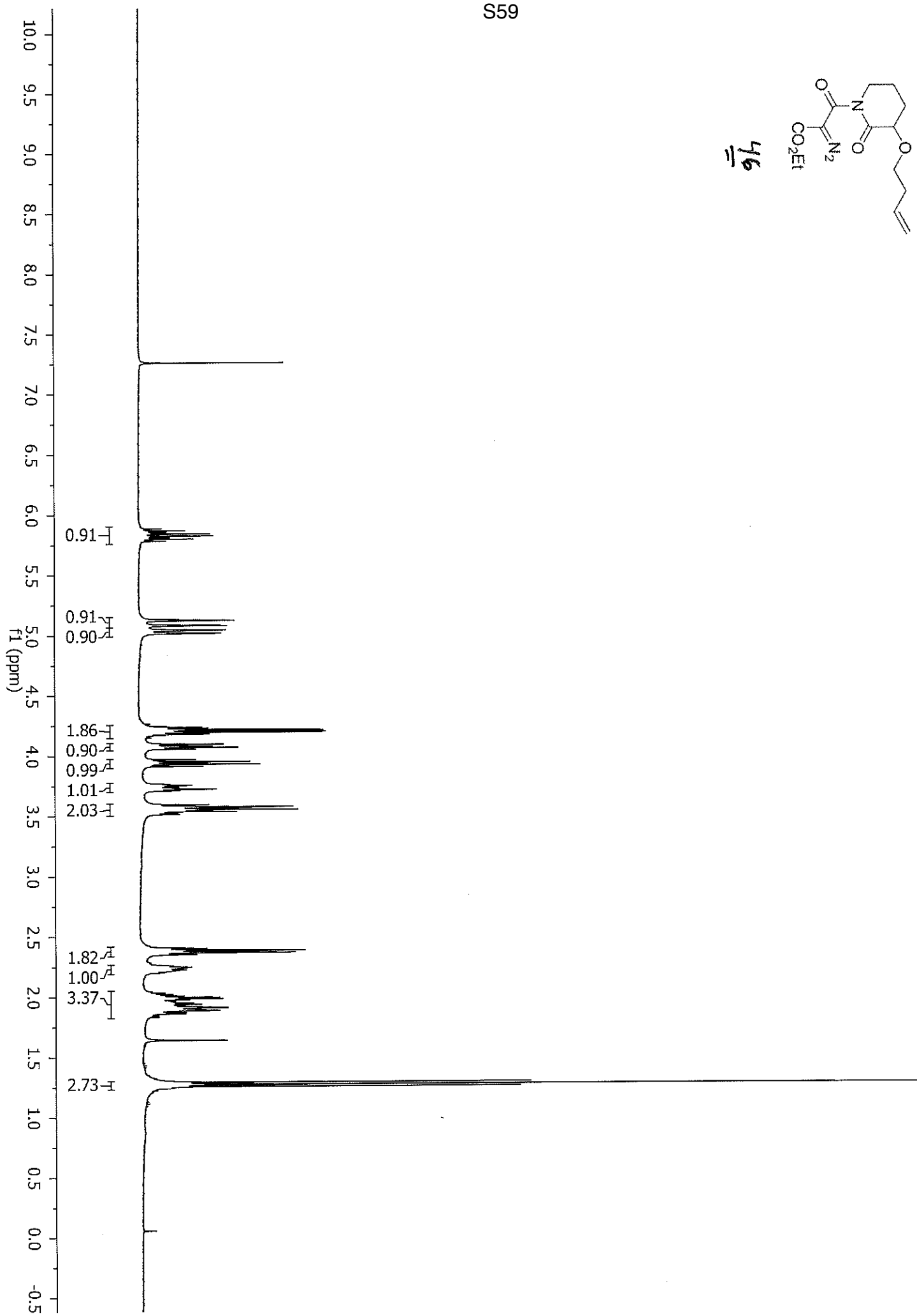
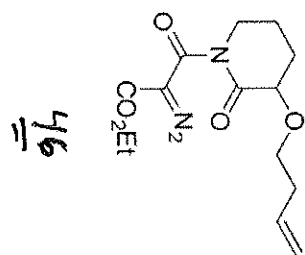


S57

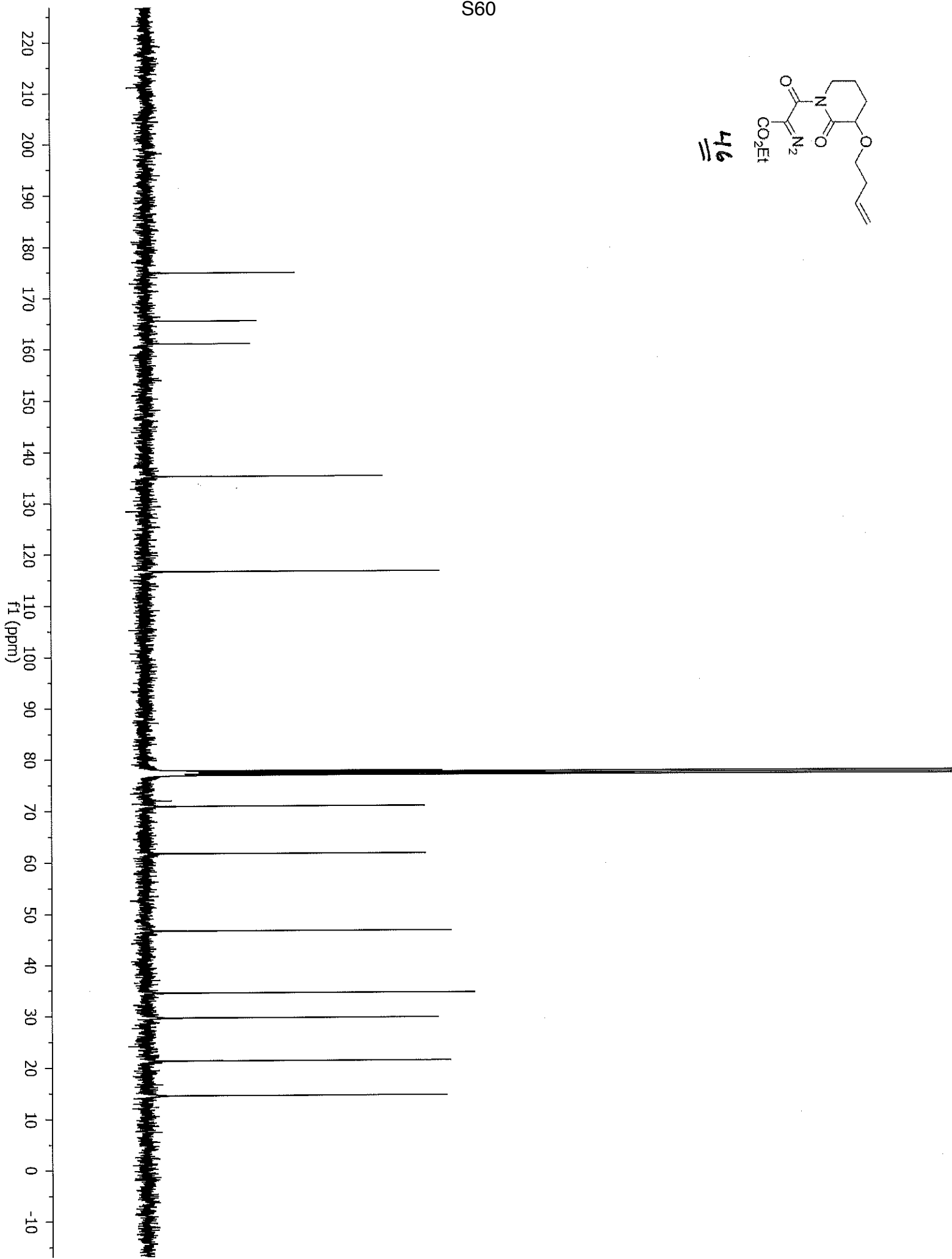
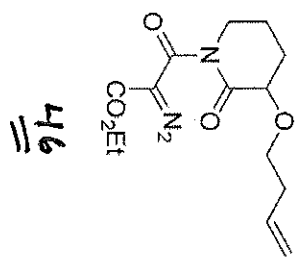


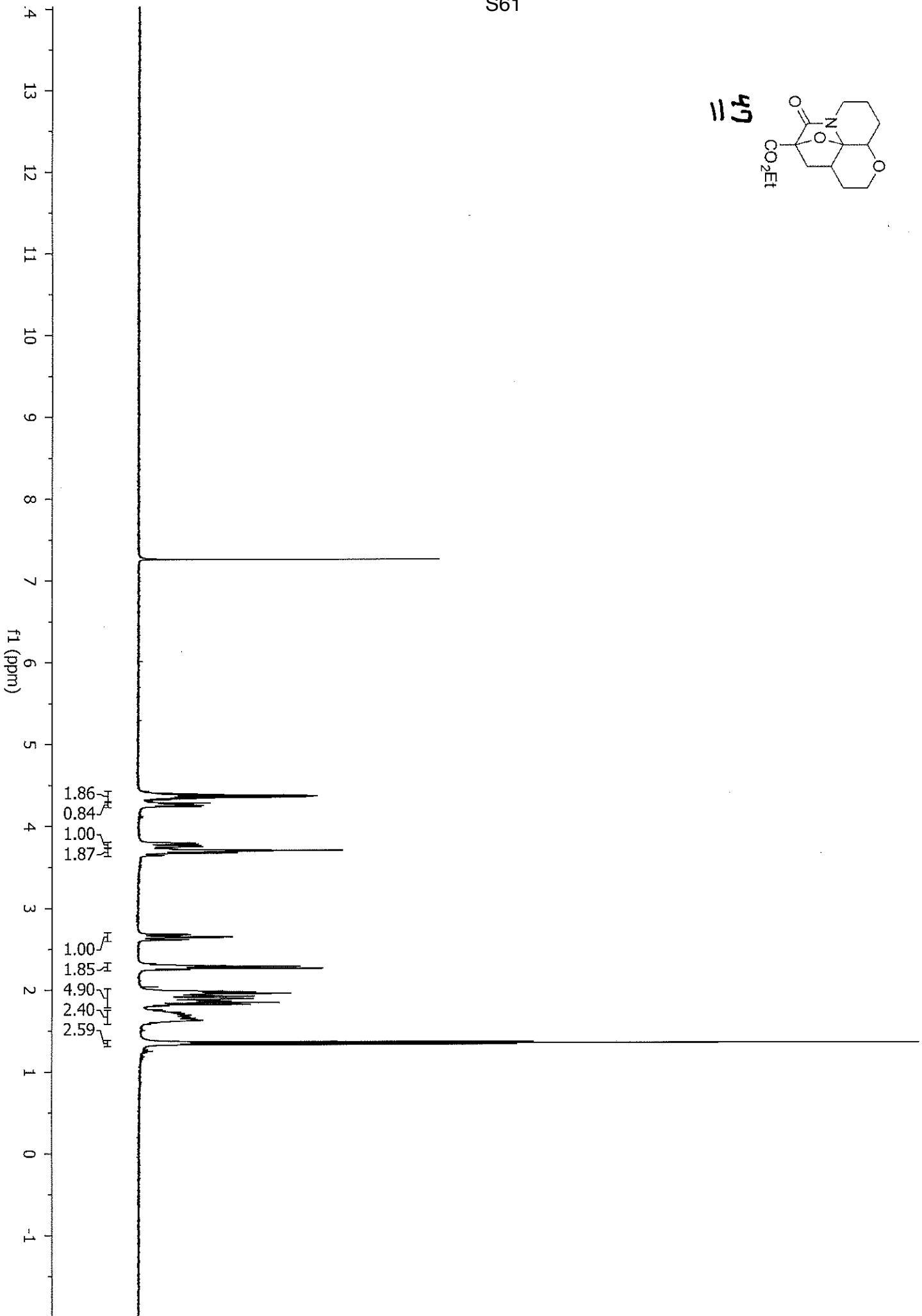
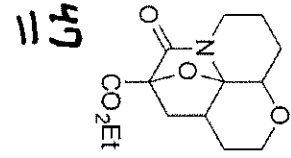
S58

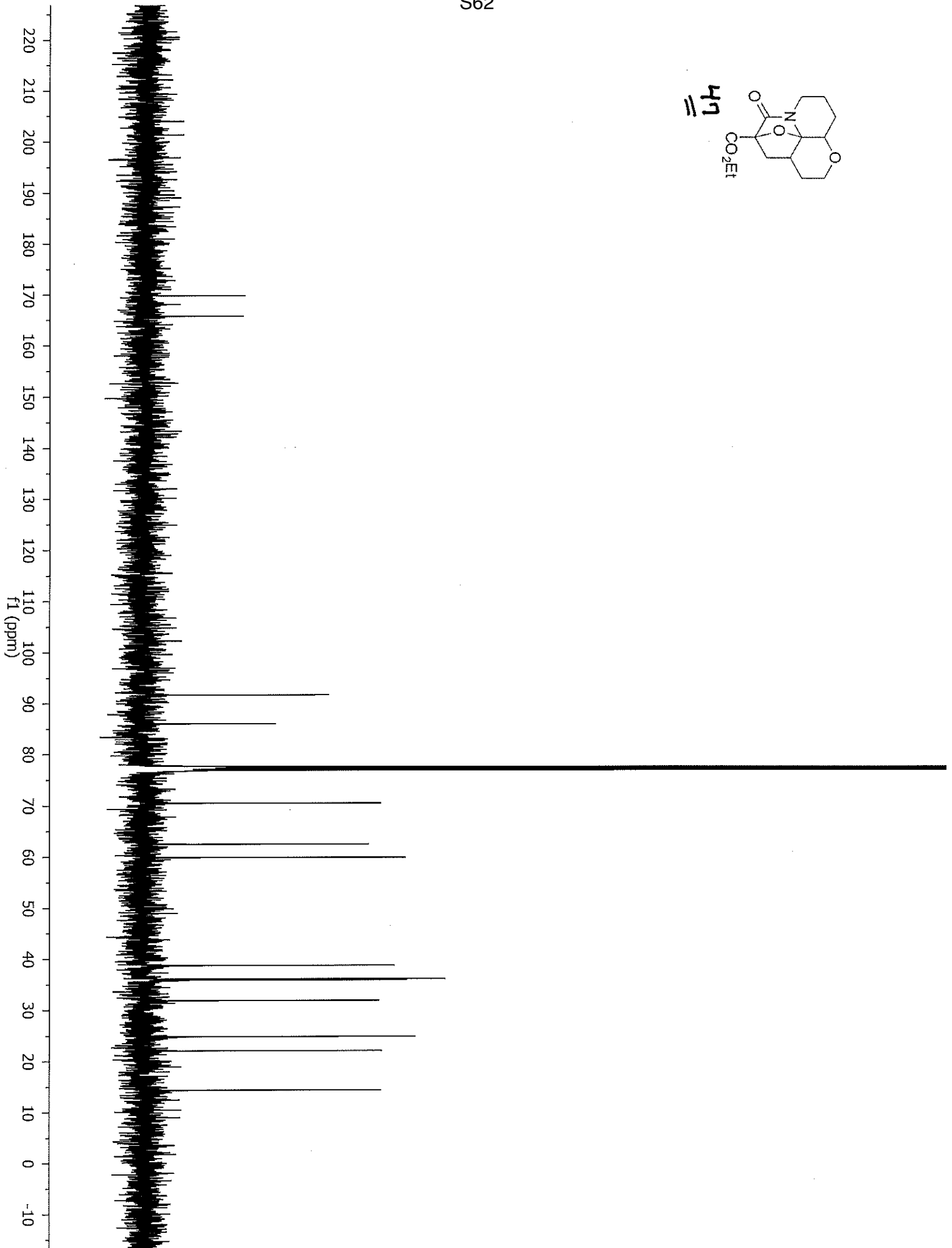
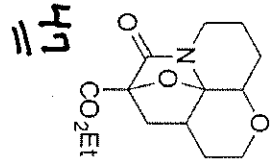




S60

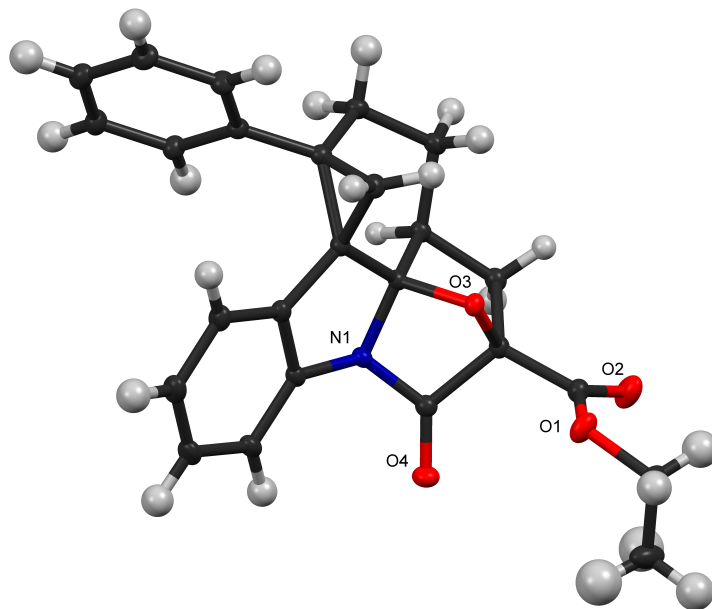






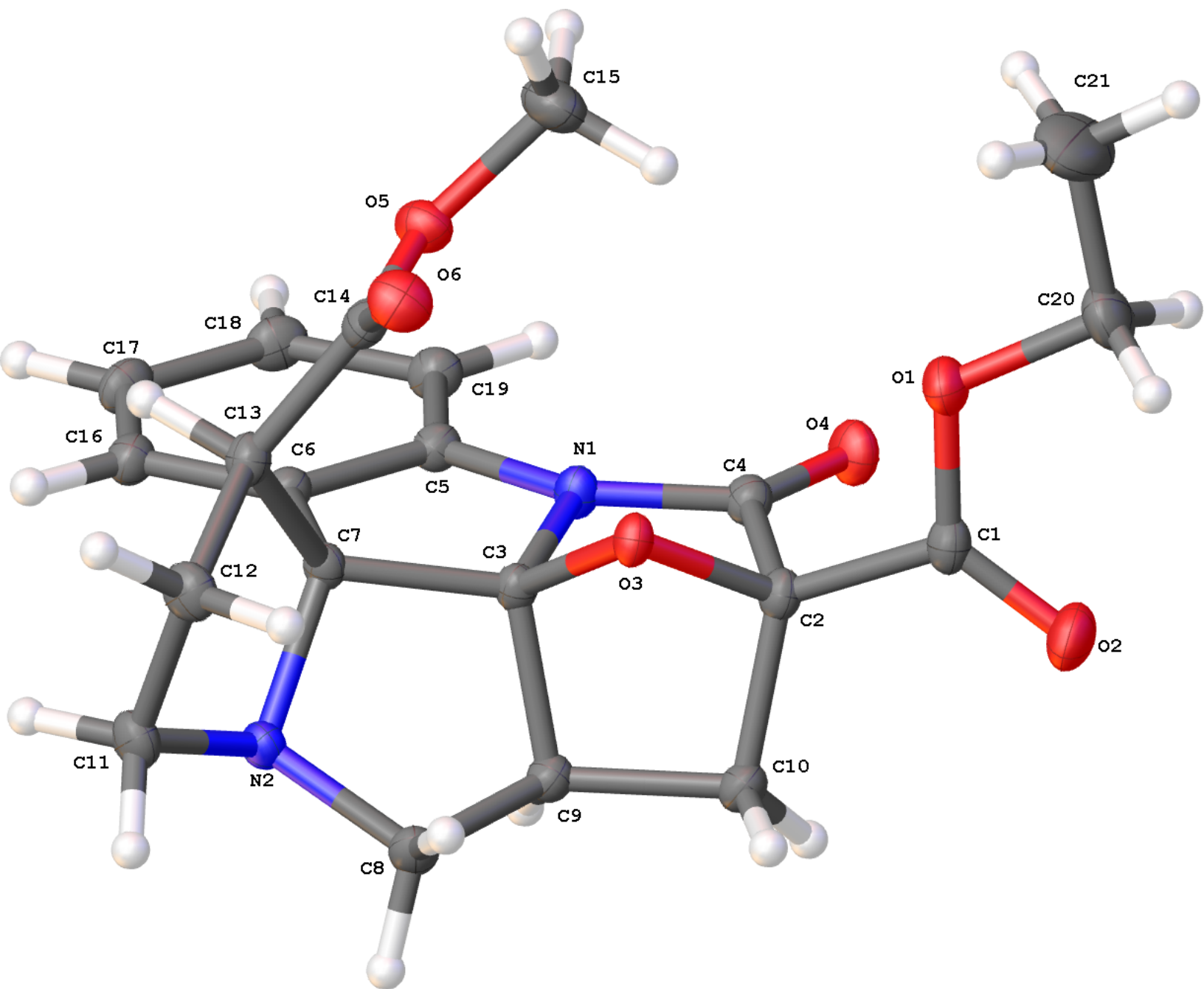
Thermal ellipsoid plot for compound **37**

The ellipsoid contour percent probability level = 50%



Thermal ellipsoid plot for compound **44**

The ellipsoid contour percent probability level = 50%





Crystal Data and Experimental for Compounds **37** and **44**

Single clear colorless prism-shaped crystals of both compounds **37** and **44** were crystallised from ether/hexane. A suitable crystal ( $0.73 \times 0.68 \times 0.42 \text{ mm}^3$ ) was selected and mounted on a loop with paratone oil on a Bruker APEX-II CCD diffractometer. The crystal was cooled to  $T = 100(2) \text{ K}$  during data collection. The structures were solved with the **ShelXT-2014/4** (Sheldrick, 2015) structure solution program using the Charge Flipping solution method and by using **Olex2** (Dolomanov et al., 2009) as the graphical interface. The model was refined with version 2014/7 of **ShelXL-2014/7** (Sheldrick, 2014) using Least Squares minimisation.

A clear colourless prism-shaped crystal with dimensions  $0.73 \times 0.68 \times 0.42 \text{ mm}^3$  was mounted on a loop with paratone oil. Data were collected using a Bruker APEX-II CCD diffractometer equipped with a Oxford Cryosystems low-temperature device, operating at  $T = 100(2) \text{ K}$ .

Data were measured using  $f$  and  $w$  scans using  $\text{MoK}_\alpha$  radiation (fine-focus sealed tube, 45 kV, 35 mA). The total number of runs and images was based on the strategy calculation from the program **APEX2** (Bruker). The maximum resolution that was achieved was  $Q = 37.858^\circ$ .

The diffraction patterns were indexed using **SAINT** (Bruker, V8.27A, 2012) and the unit cells were refined using **SAINT** (Bruker, V8.27A, 2012) on 9990 reflections, 24 % of the observed reflections. Data reduction, scaling and absorption corrections were performed using **SAINT** (Bruker, V8.27A, 2012) and **SADABS-2012/1** (Bruker, 2012). The values of  $wR_2(\text{int})$  were 0.0824 before and 0.0598 after correction. The ratio of minimum to maximum transmission is 0.8393. The  $//2$  correction factor is 0.0015. The final completeness is 99.9% out to  $37.858^\circ$  in  $Q$ . The absorption coefficient  $\mu$  of this material is  $0.091 \text{ mm}^{-1}$  at this wavelength ( $\lambda = 0.71073 \text{ \AA}$ ) and the minimum and maximum transmissions are 0.8393 and 1.0000.

The structures were solved and the space group  $P-1$  (# 2) determined by the **ShelXT-2014/4** (Sheldrick, 2015) structure solution program using Charge Flipping and refined by Least Squares using version 2014/7 of **ShelXL-2014/7** (Sheldrick, 2014). All non-hydrogen atoms were refined anisotropically. Hydrogen atom positions were calculated geometrically and refined using the riding model.

Table 1. Crystal data and structure refinement for compound **37**

Empirical formula	C <sub>25</sub> H <sub>23</sub> NO <sub>4</sub>
Formula weight	401.44
Temperature/K	172.3
Crystal system	triclinic
Space group	P-1
a/Å	9.0316(2)
b/Å	10.4052(2)
c/Å	10.9083(3)
$\alpha$ /°	99.9550(10)
$\beta$ /°	97.1890(10)
$\gamma$ /°	97.4110(10)
Volume/Å <sup>3</sup>	989.77(4)
Z	2
$\rho_{\text{calc}}$ /mg/mm <sup>3</sup>	1.347
m/mm <sup>-1</sup>	0.091
F(000)	424.0
Crystal size/mm <sup>3</sup>	0.735 × 0.675 × 0.418
2 $\Theta$ range for data collection	3.834 to 75.716°
Index ranges	-15 ≤ h ≤ 15, -17 ≤ k ≤ 17, -18 ≤ l ≤ 18
Reflections collected	40859
Independent reflections	10498[R(int) = 0.0476]
Data/restraints/parameters	10498/0/363
Goodness-of-fit on F <sup>2</sup>	1.028
Final R indexes [I ≥ 2 $\sigma$ (I)]	R <sub>1</sub> = 0.0466, wR <sub>2</sub> = 0.1265
Final R indexes [all data]	R <sub>1</sub> = 0.0535, wR <sub>2</sub> = 0.1355
Largest diff. peak/hole / e Å <sup>-3</sup>	0.74/-0.37

Table 2. Crystal data and structure refinement for compound **44**

Empirical formula	C <sub>21</sub> H <sub>22</sub> N <sub>2</sub> O <sub>6</sub>	
Formula weight	398.40	
Temperature	110(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P 1 2 <sub>1</sub> /c 1	
Unit cell dimensions	a = 12.7648(13) Å	$\alpha = 90^\circ$ .
	b = 9.0440(9) Å	$\beta = 92.9810(10)^\circ$ .
	c = 16.8162(18) Å	$\gamma = 90^\circ$ .
Volume	1938.7(3) Å <sup>3</sup>	
Z	4	
Density (calculated)	1.365 Mg/m <sup>3</sup>	
Absorption coefficient	0.101 mm <sup>-1</sup>	
F(000)	840	
Crystal size	0.672 x 0.458 x 0.158 mm <sup>3</sup>	
Theta range for data collection	2.426 to 30.995°.	
Index ranges	-18 ≤ h ≤ 18, -12 ≤ k ≤ 13, -24 ≤ l ≤ 24	
Reflections collected	22767	
Independent reflections	6142 [R(int) = 0.0318]	
Completeness to theta = 25.242°	100.0 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.7463 and 0.5813	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	6142 / 27 / 337	
Goodness-of-fit on F <sup>2</sup>	1.126	
Final R indices [I > 2σ(I)]	R1 = 0.0503, wR2 = 0.1192	
R indices (all data)	R1 = 0.0683, wR2 = 0.1451	
Extinction coefficient	n/a	
Largest diff. peak and hole	0.607 and -0.309 e.Å <sup>-3</sup>	