Supporting information

Atomic and local electronic structures of $Ca_2AlMnO_{5+\delta}$ as an oxygen storage material

Genki Saito^{a*}, Yuji Kunisada^a, Kazuki Hayami^a, Takahiro Nomura^a, and Norihito Sakaguchi^a

^aCenter for Advanced Research of Energy and Materials, Hokkaido University, Kita 13 Nishi 8, Kitaku, Sapporo 060-8628, Japan



Fig. S1. XRD patterns of the synthesized powders.

	Ca_2AIMnO_5	$Ca_2AIMnO_{5.5}$	
	<i>I 2 m</i> b	I m m a	
<i>a,</i> Å	5.24256(11)	5.25478(15)	
<i>b,</i> Å	15.0051(3)	29.4266(9)	
<i>c,</i> Å	5.47213(11)	5.37314(15)	
V, Å ³	430.46(58)	830.85(05)	
R wp, %	4.94	5.45	
R p, %	3.73	4.07	
Re,%	4.70	4.46	
S	1.0508	1.2229	
X ²	1.1043	1.4954	

Table S1. Structural parameters from Rietveld refinement.

Table S2. Atomic parameters for Ca₂AlMnO₅ and CaAlMnO_{5.5}.

Ca ₂ AlMnO ₅ (<i>I m a</i> 2)						
Atom	Occupancy	х	У	Z	Uiso, Å ²	
Са	1.0	0.1121	0.5266	0.0072	0.00200	
Mn	1.0	0	0	0	0.00550	
Al	1.0	0.25	0.0726	0.0396	0.00520	
01	1.0	0.0115	0.2450	0.2440	0.00740	
02	1.0	0.1485	0.5660	0.4812	0.00400	
03	1.0	0.25	0.1427	0.3801	0.00500	
Ca ₂ AlMnO _{5.5} (<i>I m m a</i>)						
Atom	Occupancy	х	у	Z		
Ca1	1.0	0	0.0604	0.5106	_	
Ca2	1.0	0	0.1826	0.5259		
Mn	1.0	0	0.1254	0.0075		
Al1	0.5	0.0100	0.25	0.0440		
Al2	1.0	0	0	0		
01	0.5	0.1220	0.25	0.3680		
02	1.0	0	0.0695	0.0611		
03	1.0	0	0.6993	0.0270		
04	1.0	0.25	0.1392	0.25		
05	1.0	0.25	0.5031	0.25		
06	1.0	0.25	0.6259	0.25		



Fig. S2. Scheme of electron beam channeling at Al_t (tetrahedral) and Al_o (octahedral)

sites.



Fig. S3. Observed planar fault in $Ca_2AlMnO_{5.5}$, wherein two Al_t planes continuously arranged as a stacking error.



Fig. S4. Calculated O K-edge ELNES at each oxygen site in (a) Ca₂AlMnO₅ and (b)

Ca₂AlMnO_{5.5}.