

Supporting Information

Microporous Hexanuclear Ln(III) Cluster-based Metal-Organic Frameworks: Color Tunability for Barcode Application and Selective Removal of Methylene Blue

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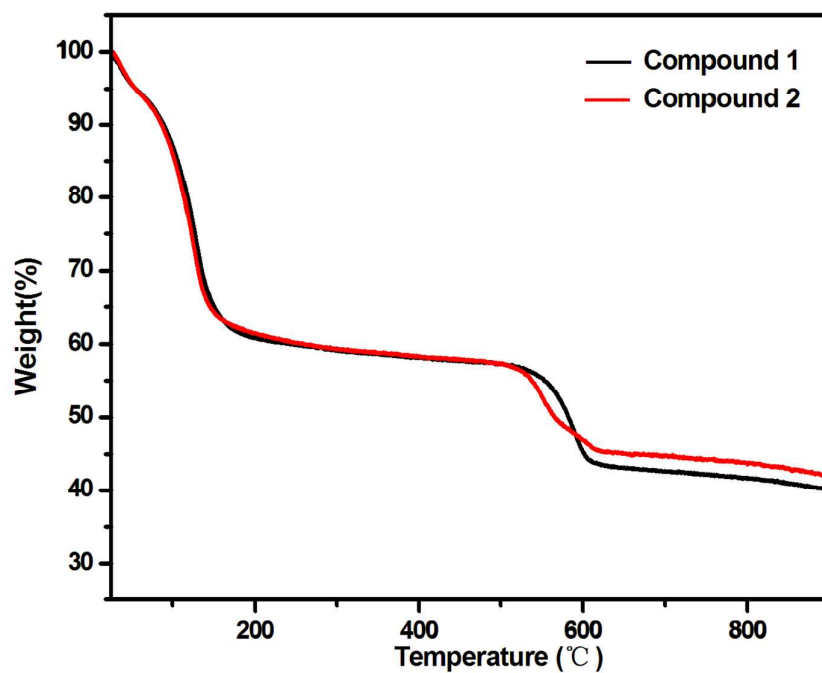


Figure S1 The thermal stability of **1** and **2** was examined by thermogravimetric analysis (TGA) in the temperature range of 25-900 °C with a heating rate of 5 °C min⁻¹ under a N₂ atmosphere.

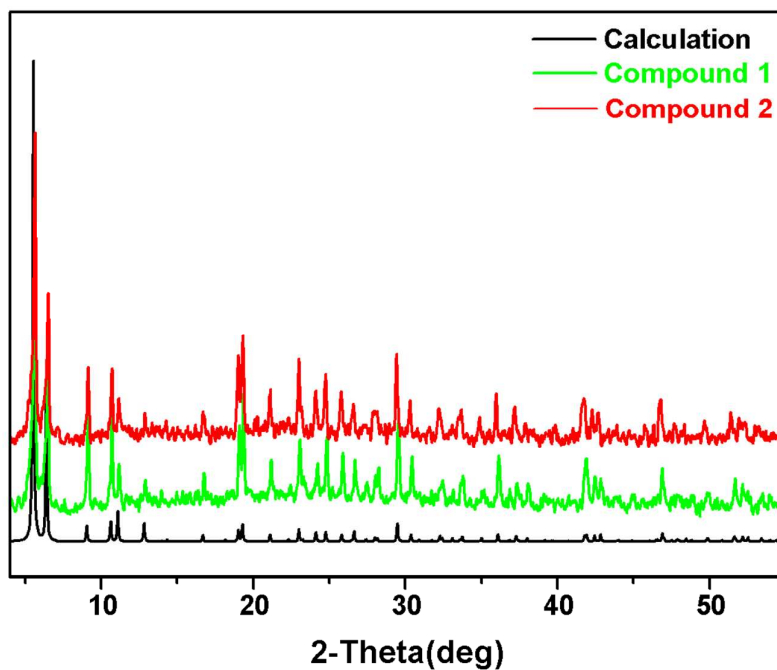


Figure S2 The measured PXRD patterns of complexes **1**, **2** in comparison to the simulated pattern of single-crystal data.

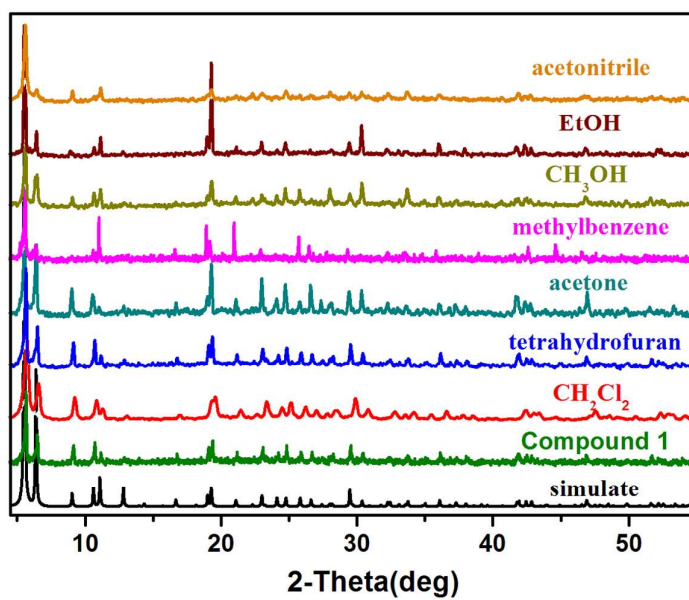


Figure S3 The measured PXRD patterns of solvent-stable to the compound **1** in comparison to the simulated pattern of single-crystal data.

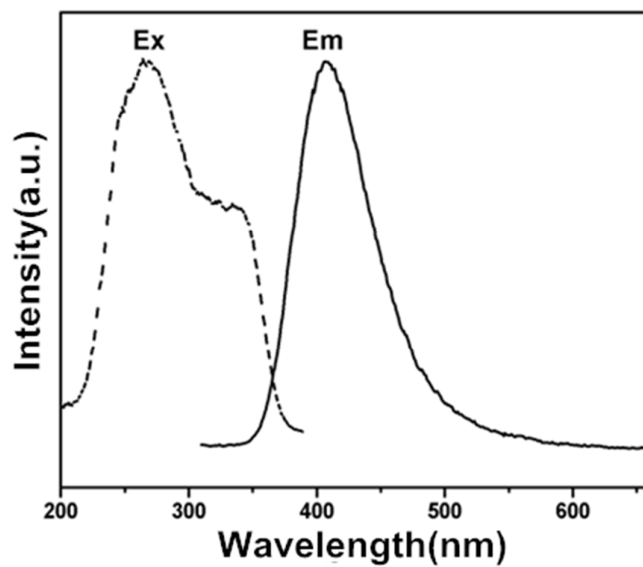


Figure S4 Excitation and emission spectra of the H₂BPDC.

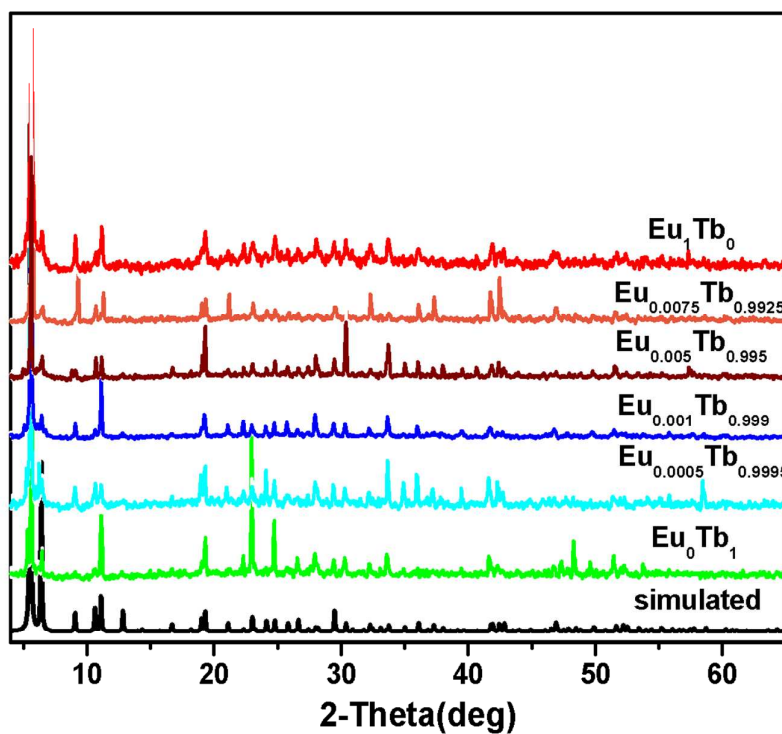


Figure S5 The measured PXRD patterns of bimetallic Eu_xTb_{1-x} series of complexes **1**, **2**, **3-6** in comparison to the simulated pattern of single-crystal **1**.

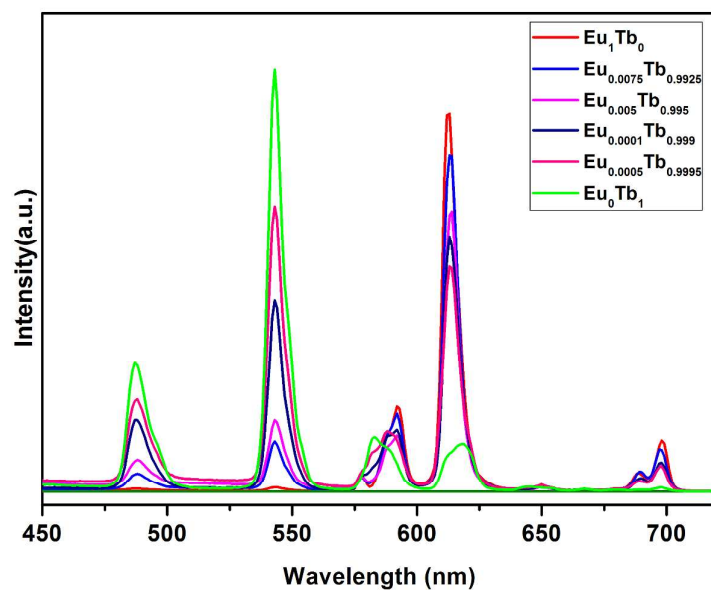


Figure S6 Solid-state emission spectra of **1** and $\text{Eu}_x\text{Tb}_{1-x}$ -MOF excited with 300 nm UV light.

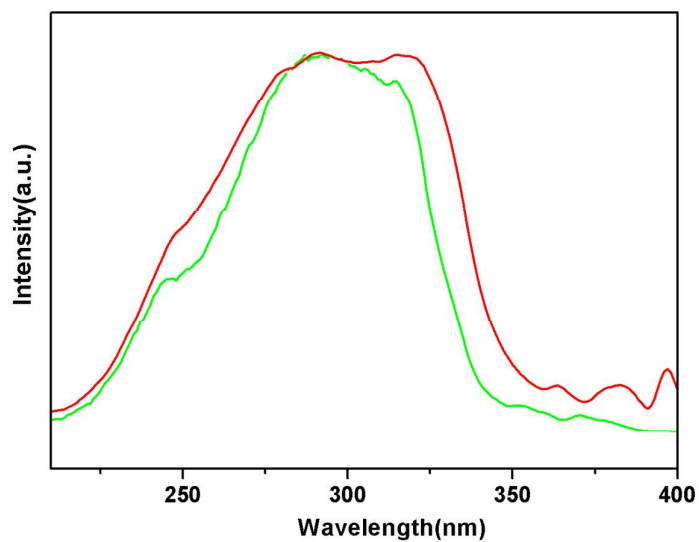


Figure S7 Solid-state excitation spectra at room temperature. Represented by $\text{Eu}_{0.0075}\text{Tb}_{0.9925}$ -**6** when monitored at respective Eu^{3+} and Tb^{3+} characteristic emission peaks.

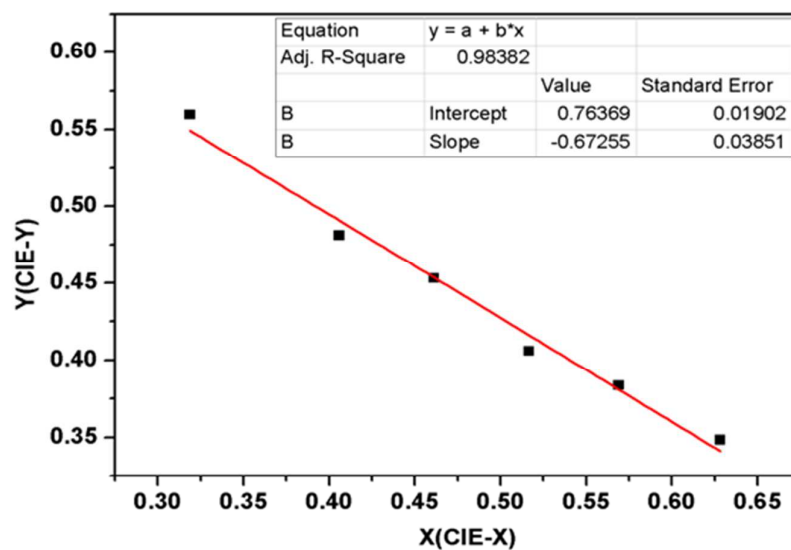


Figure S8 Coordinates of mixed Ln(III)-based MOFs, **3**, **4**, **5** and **6** in comparison to compound **1** and **2** when excited at 300 nm, showing variation depending on $\text{Tb}^{3+}/\text{Eu}^{3+}$ atomic ratio. By changing the content of Tb^{3+} and Eu^{3+} , the emission color of these materials can be fine-tuned in the RG spectra region.

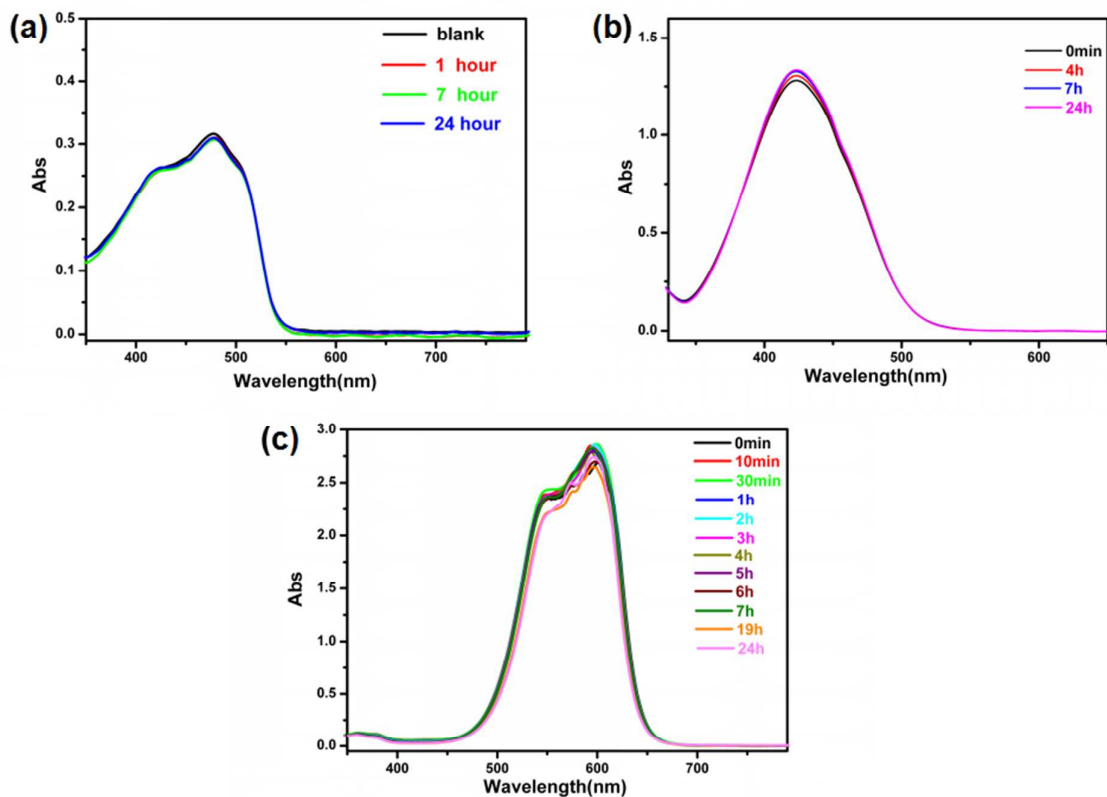


Fig. S9 UV-vis spectra of DMF solution of MO^- , SD^0 and CV^+ in the presence of Ln(III)-based MOF **1**.

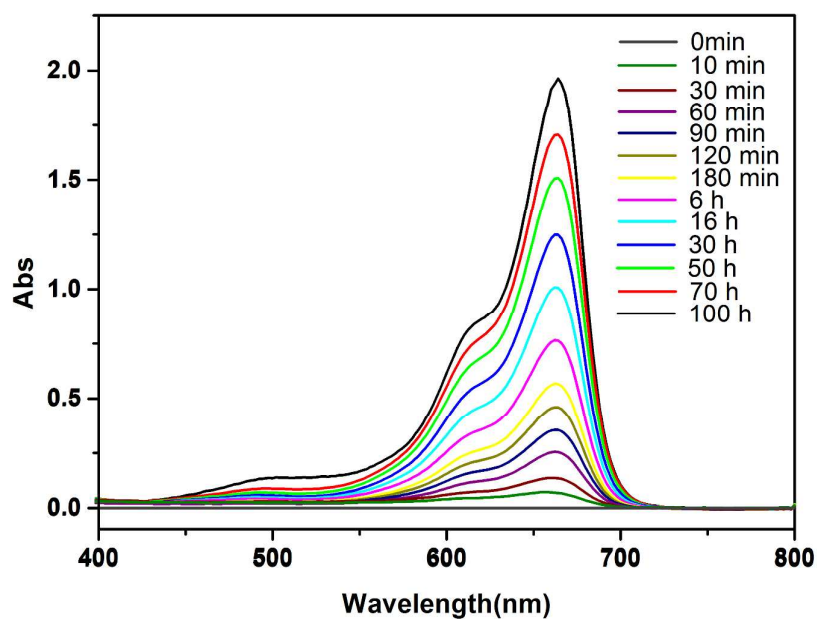


Fig. S10 UV-vis spectra of MB^+ release from $\text{MB}^+@1$ in the DMF solution.

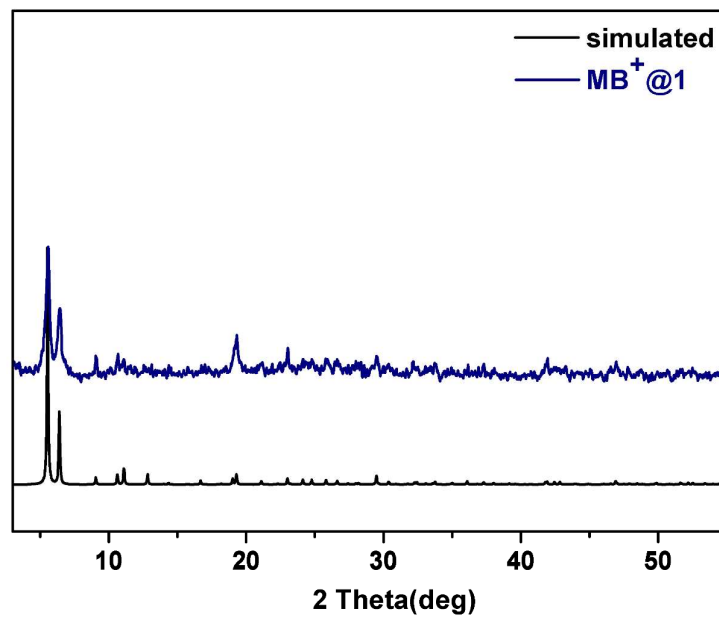


Fig. S11 Powder XRD patterns for simulated **1** and $\text{MB}^+@1$ after MB^+ release.

Table S1 Crystal data and structure refinement for **1** and **2**

Compounds	1	2
Empirical formula	C42 H28 O16 Tb3	C42 H28 O16 Eu3
Formula weight	1265.40	1244.52
Temperature	100(2) K	100(2) K
Wavelength	0.71073 Å	1.54184 Å
Crystal system	Cubic	Cubic
Space group	<i>Fm</i> -3m	<i>Fm</i> -3m
Unit cell dimensions	$a = 27.5460(11) \text{ Å}$ $\alpha = 90^\circ$.	$a = 27.5460(11) \text{ Å}$ $\alpha = 90^\circ$.
	$b = 27.5460(11) \text{ Å}$ $\beta = 90^\circ$.	$b = 27.5460(11) \text{ Å}$ $\beta = 90^\circ$.
	$c = 27.5460(11) \text{ Å}$ $\gamma = 90^\circ$.	$c = 27.5460(11) \text{ Å}$ $\gamma = 90^\circ$.
Volume	20901(3) Å ³	20901(3) Å ³
Z	8	8
Density (calculated)	0.804 Mg/m ³	0.791 Mg/m ³
Absorption coefficient	2.037 mm ⁻¹	12.962 mm ⁻¹

F(000)	4824	4776
Crystal size	0.150 x 0.140 x 0.130 mm ³	0.150 x 0.130 x 0.100 mm ³
Theta range for data collection	2.958 to 28.272°.	4.540 to 73.938°.
Index ranges	-36<= <i>h</i> <=36, -36<= <i>k</i> <=36, -36<= <i>l</i> <=36	-33<= <i>h</i> <=32, -14<= <i>k</i> <=23, -30<= <i>l</i> <=22
Reflections collected	87687	5684
Independent reflections	1343 [R(int) = 0.0669]	1087 [R(int) = 0.0298]
Completeness to theta = 25.03°	99.5 %	99.1 %
Absorption correction	Semi-empirical from equivalents	Semi-empirical from equivalents
Max. and min. transmission	0.7457 and 0.6544	1.00000 and 0.72208
Refinement method	Full-matrix least-squares on F ²	Full-matrix least-squares on F ²
Data / restraints / parameters	1343 / 27 / 37	1087 / 45 / 43
Goodness-of-fit on F ²	1.132	1.088
Final R indices [I>2sigma(I)] ^a	R1 = 0.0655, wR2 = 0.1652	R1 = 0.0432, wR2 = 0.1180
R indices (all data)	R1 = 0.0881, wR2 = 0.1970	R1 = 0.0471, wR2 = 0.1218
Largest diff. peak and hole	3.348 and -2.414 e.Å ⁻³	1.951 and -0.950 e.Å ⁻³
$^a R_1 = \sum F_o - F_c / \sum F_o ; wR_2 = \sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]^{1/2}$		

Table S2 The ⁵D₄ of Tb³⁺ and ⁵D₀ of Eu³⁺ lifetimes for Ln(III)-based MOFs **1**, **2**, **3-6**. The decay curves are monitored at 547 nm and 617 nm and excited at 300 nm.

Mixed-Tb _x Eu _{1-x} -MOFs	The mole ratio of Eu (based on RE)	
	⁵ D ₄ of Tb ³⁺ /μs	⁵ D ₀ of Eu ³⁺ /μs
1	144.179	—
2	—	781.5773
3	38.4892	917.135
4	31.0129	945.127
5	25.5415	972.682
6	19.9444	975.511

Table S3 The ICP-OES analysis results of mixed Eu_xTb_{1-x}-MOFs **3-6**.

Mixed-Tb _x Eu _{1-x} -MOFs	The mole ratio of Eu		Molecular formula
	(based on Ln(III))		
	Theoretical	Measured	
3	0.0005	0.0005	Eu _{0.0005} Tb _{0.9995} C ₈₈ H ₈₀ N ₂ O ₁₈
4	0.001	0.0009	Eu _{0.0009} Tb _{0.9991} C ₈₈ H ₈₀ N ₂ O ₁₈
5	0.005	0.0053	Eu _{0.0053} Tb _{0.9967} C ₈₈ H ₈₀ N ₂ O ₁₈
6	0.0075	0.0076	Eu _{0.0076} Tb _{1.9924} C ₈₈ H ₈₀ N ₂ O ₁₈

Table S4 The chromaticity coordinates of the six binary Eu_xTb_{1-x} complexes when excited with 300 nm.

Mixed-Tb _x Eu _{1-x} -MOFs(Samples)	CIE(X, Y)
1	0.3187, 0.5598
3	0.4058, 0.481
4	0.4612, 0.4531
5	0.5164,0.406
6	0.569, 0.3838
2	0.6282, 0.3485