

Supporting Information for

Copper(0)/Selectfluor System-promoted Oxidative Carbon-Carbon Bond Cleavage/Annulation of *o*-Aryl Chalcones: An Unexpected Synthesis of 9,10-Phenanthraquinone Derivatives

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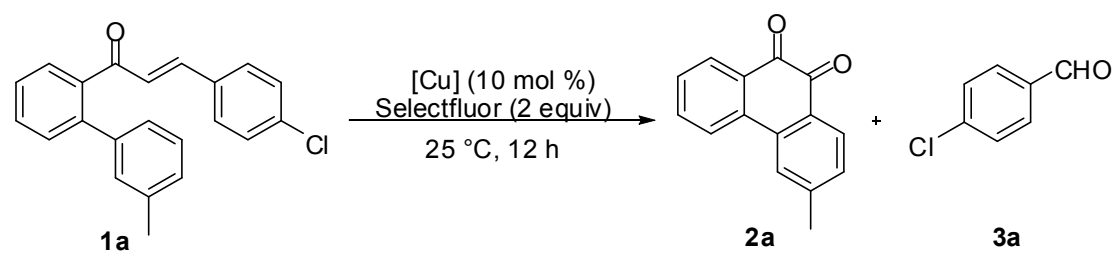
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Contents

1. Optimization of Reaction Conditions	S2–S5
2. Mechanistic Experiments	S6
2.1 Reaction of 1ze in CH ₃ CN-H ₂ O ¹⁸ (50:1, V/V)	S6–S7
2.2 Studies on the Intermolecular Kinetic Isotope Effects (KIE) Based on Substrate 1u and 1u-d₅	S7
3. ¹ H and ¹³ C NMR spectra of 2a-2zd	S8–S32
4. ¹ H spectrum of 1u-d₅	S32

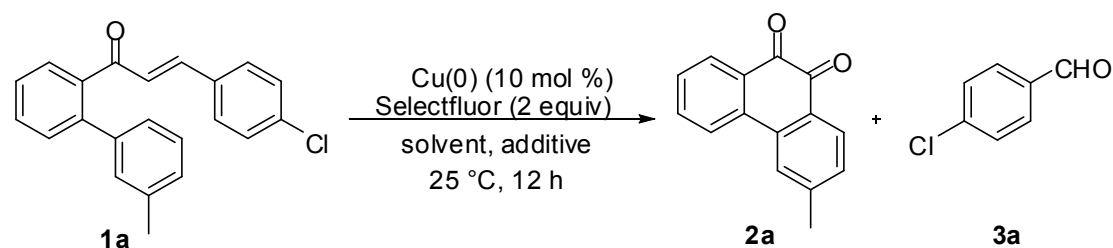
1. Optimization of Reaction Conditions

Table S1. Screening with different catalysts ^a



entry	[Cu]	solvent	yield of 2a (%) ^b
1	Cu(0) ^c	CH ₃ CN	70
2	Cu(0)	CH ₃ CN	75
3	Cu(0)	CH₃CN/H₂O = 50:1 (V/V)	80
4	Cu(0) ^d	CH ₃ CN/H ₂ O = 50:1 (V/V)	69
5	Cu(0) ^e	CH ₃ CN/H ₂ O = 50:1 (V/V)	55
6	--	CH ₃ CN/H ₂ O = 50:1 (V/V)	0
7	CuBr	CH ₃ CN/H ₂ O = 50:1 (V/V)	64
8	CuI	CH ₃ CN/H ₂ O = 50:1 (V/V)	57
9	CuCl	CH ₃ CN/H ₂ O = 50:1 (V/V)	55
10	CuCl ₂	CH ₃ CN/H ₂ O = 50:1 (V/V)	60
11	Cu(OAc) ₂	CH ₃ CN/H ₂ O = 50:1 (V/V)	50
12	Cu(NO ₃) ₂	CH ₃ CN/H ₂ O = 50:1 (V/V)	52

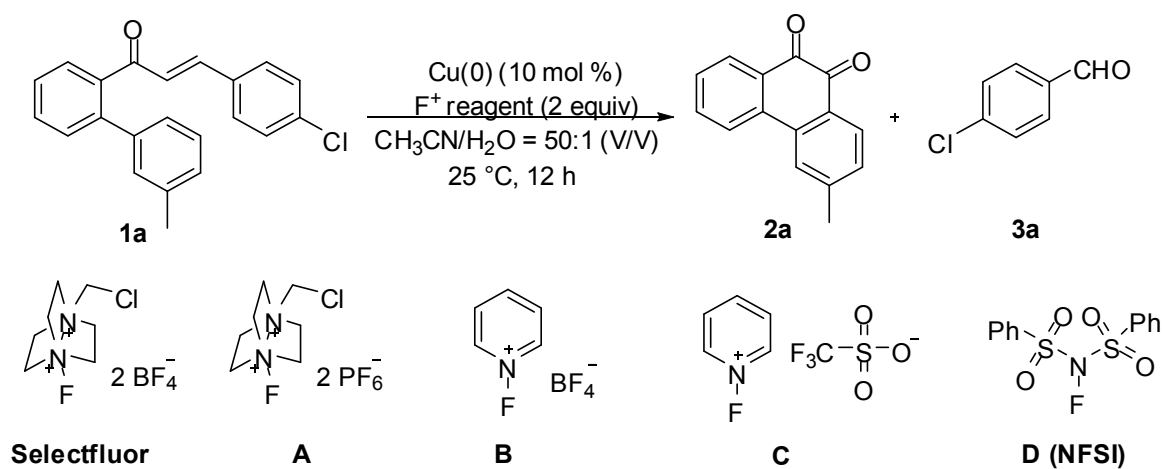
^a Reactions carried out with **1a** (0.3 mmol), copper catalyst (10 mol % based on **1a**), Selectfluor (2 equiv), solvent (3 mL), 25 °C for 12 h unless otherwise noted. ^b Isolated yields. ^c Using 5 mol % of Cu(0) powder. ^d Using 20 mol % of Cu(0) powder. ^e The reaction temperature is 50 °C.

Table S2. Screening with different solvents, additives and reaction times^a

entry	solvent	additive	yield of 2a (%), ^b
1	CH ₃ CN	--	75
2	CH ₃ CN	K ₂ CO ₃	68
3	CH ₃ CN	NaHCO ₃	64
4	CH ₃ CN	K ₂ CO ₃	67
5	CH ₃ CN	K ₃ PO ₄	57
6	CH₃CN/H₂O = 50:1 (V/V)	--	80
7	CH ₃ CN/H ₂ O = 25:1 (V/V)	--	65
8	CH ₃ CN/H ₂ O = 100:1 (V/V)	--	77
9	CH ₃ CN/H ₂ O = 200:1 (V/V)	--	75
10	CH ₃ CN/H ₂ O = 400:1 (V/V)	--	73
11	Acetone/H ₂ O = 50:1 (V/V)	--	0
12	EtOAc/H ₂ O = 50:1 (V/V)	--	0
13	MeOH/H ₂ O = 50:1 (V/V)	--	0
14	EtOH/H ₂ O = 50:1 (V/V)	--	0
15	DMF/H ₂ O = 50:1 (V/V)	--	0
16	toluene/H ₂ O = 50:1 (V/V)	--	0
17	CH ₂ Cl ₂ /H ₂ O = 50:1 (V/V)	--	0
18	1,4-dioxane/H ₂ O = 50:1 (V/V)	--	0
19	CH ₃ CN/H ₂ O = 50:1 (V/V)	--	70 ^c
20	CH ₃ CN/H ₂ O = 50:1 (V/V)	--	65 ^d
21	CH ₃ CN/H ₂ O = 50:1 (V/V)	--	30 ^e

^a Reactions carried out with **1a** (0.3 mmol), copper powder (10 mol % based on **1a**), Selectfluor (2 equiv), solvent (3 mL), 25 °C for 12 h unless otherwise noted. ^b Isolated yields. ^c The reaction time is 6 h. ^d The reaction time is 4 h. ^e The reaction time is 2 h.

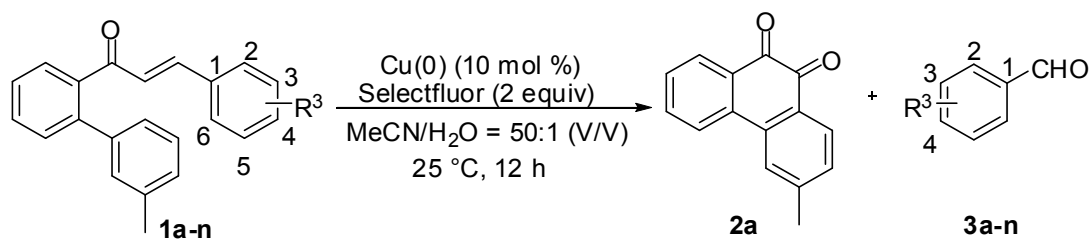
Table S3. Screening with different oxidants ^a



entry	F^+ reagent	yield of 2a (%) ^b
1	Selectfluor	80
2	Selectfluor	40 ^c
3	Selectfluor	62 ^d
4	A	40
5	B	trace
6	C	10
7	D	50

^a Reactions carried out with **1a** (0.3 mmol), copper powder (10 mol % based on **1a**), F^+ reagent (2 equiv), $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ mixture (50:1 (V/V), 3 mL), $25\text{ }^\circ\text{C}$ for 12 h unless otherwise noted. ^b Isolated yields. ^c The use of 1.0 equivalent of Selectfluor. ^d The use of 3.0 equivalents of Selectfluor.

Table S4. Investigation of the effect of different benzylidene moieties in **1a-n** on the formation of **2a**^a



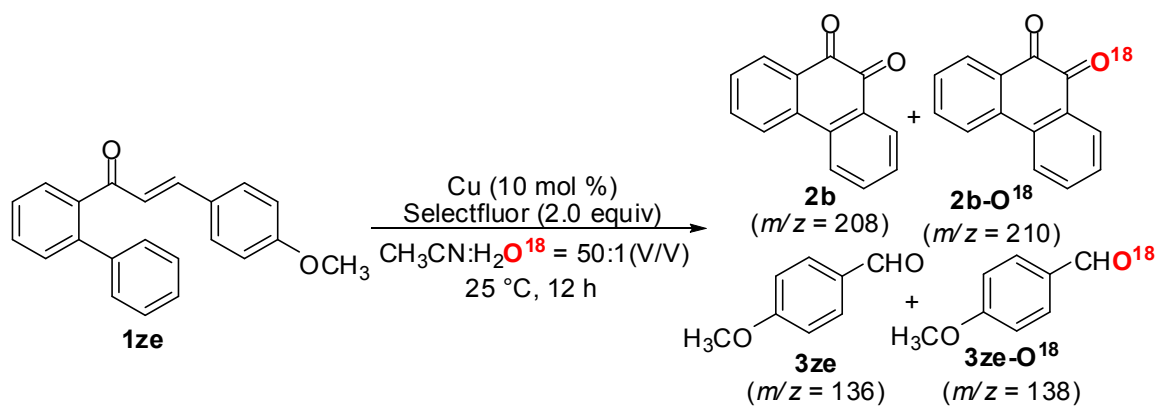
entry	R ³ (1a-n)	aldehyde (3a-n)	yield of 2a (%) ^b
1	4-Cl (1a)	3a	80
2	2-Cl (1a-1)	3a-1	66
3	3-Cl (1a-2)	3a-2	60
4	4-Br (1a-3)	3a-3	61
5	3,5-dichloro (1a-4)	3a-4	65
6	2,6-dichloro (1a-5)	3a-5	63
7	4-NO ₂ (1a-6)	3a-6	45
8	4-CH ₃ (1a-7)	3a-7	32
9	H (1a-8)	3a-8	60

^a Reactions carried out with **1a-n** (0.3 mmol), copper catalyst (10 mol % based on **1a-n**), Selectfluor (2 equiv), CH₃CN/H₂O = 50:1 (V/V) (3 mL), 25 °C for 12 h unless otherwise noted.

^b Isolated yields.

2. Mechanistic Experiments

2.1 Reaction of 1ze in CH₃CN-H₂O¹⁸ (50:1, V/V)



MS-analytical Results:

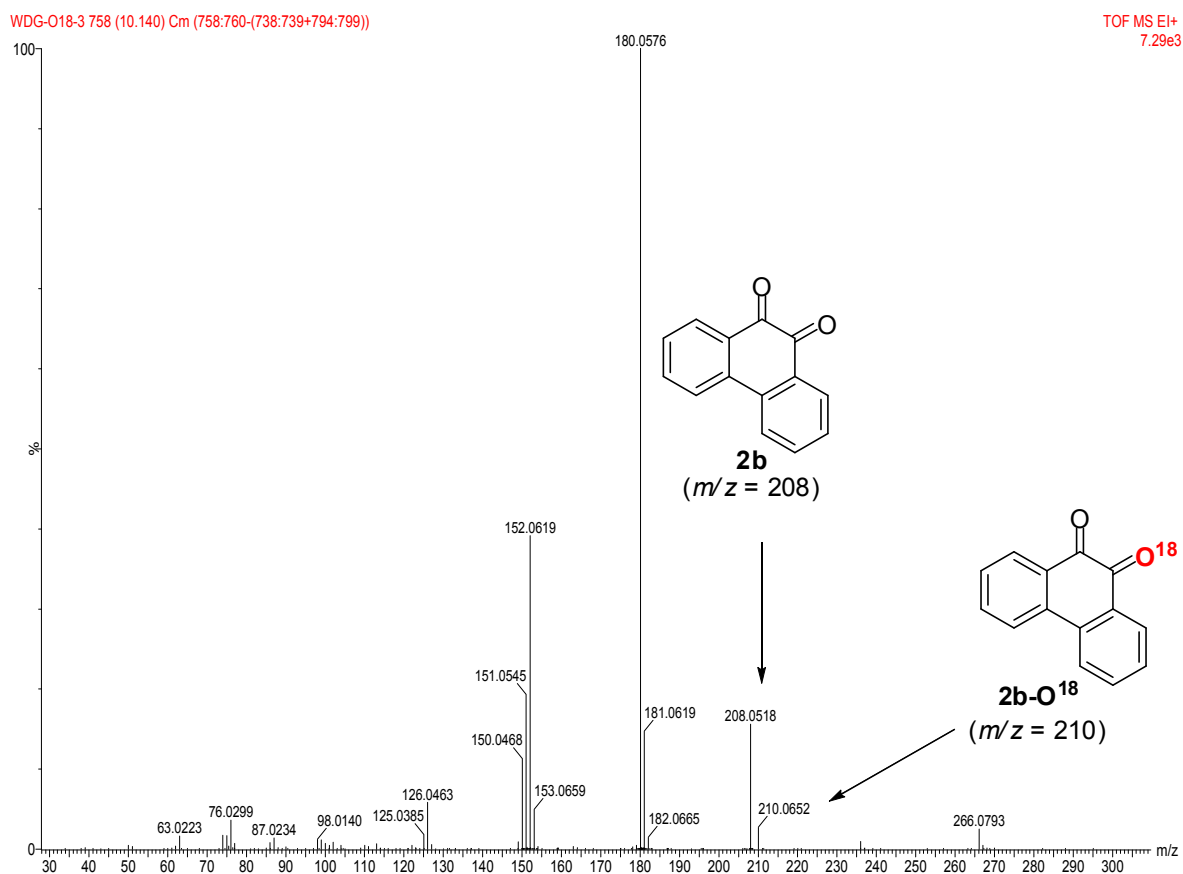


Figure S1. The MS-spectrum of 2b/2b-O¹⁸

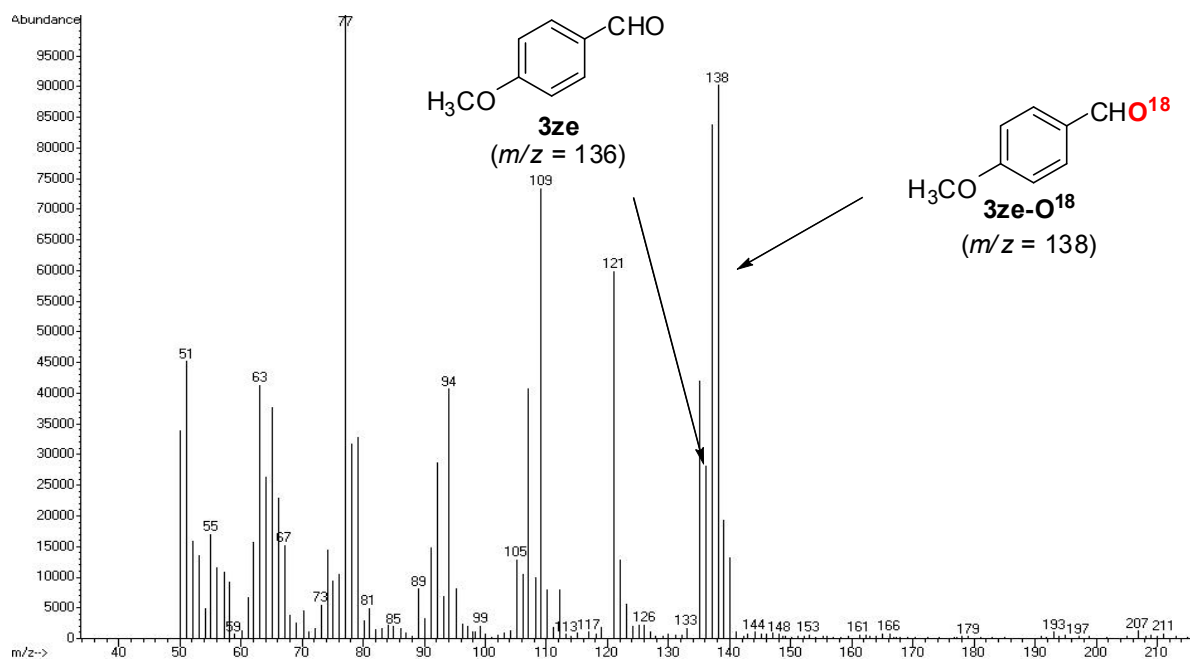


Figure S2. The MS-spectrum of **3ze/3ze-O¹⁸**

2.2 Studies on the Intermolecular Kinetic Isotope Effects (KIE) Based on Substrate **1u** and **1u-d₅**.

¹H NMR Analytical Results:

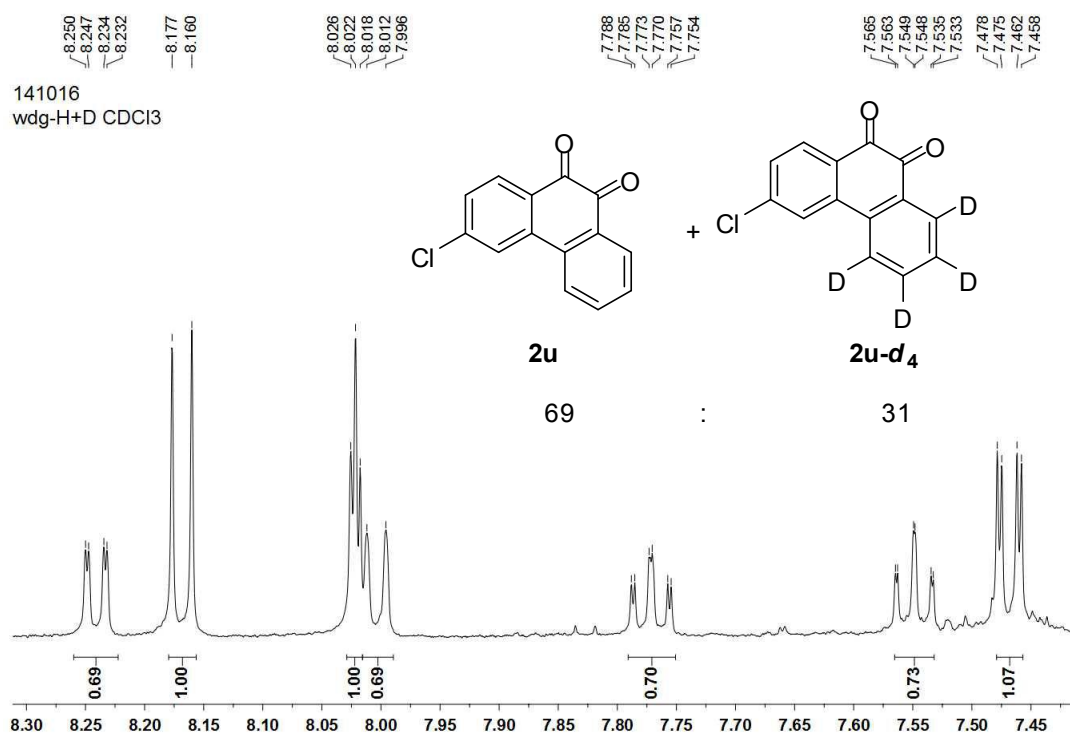
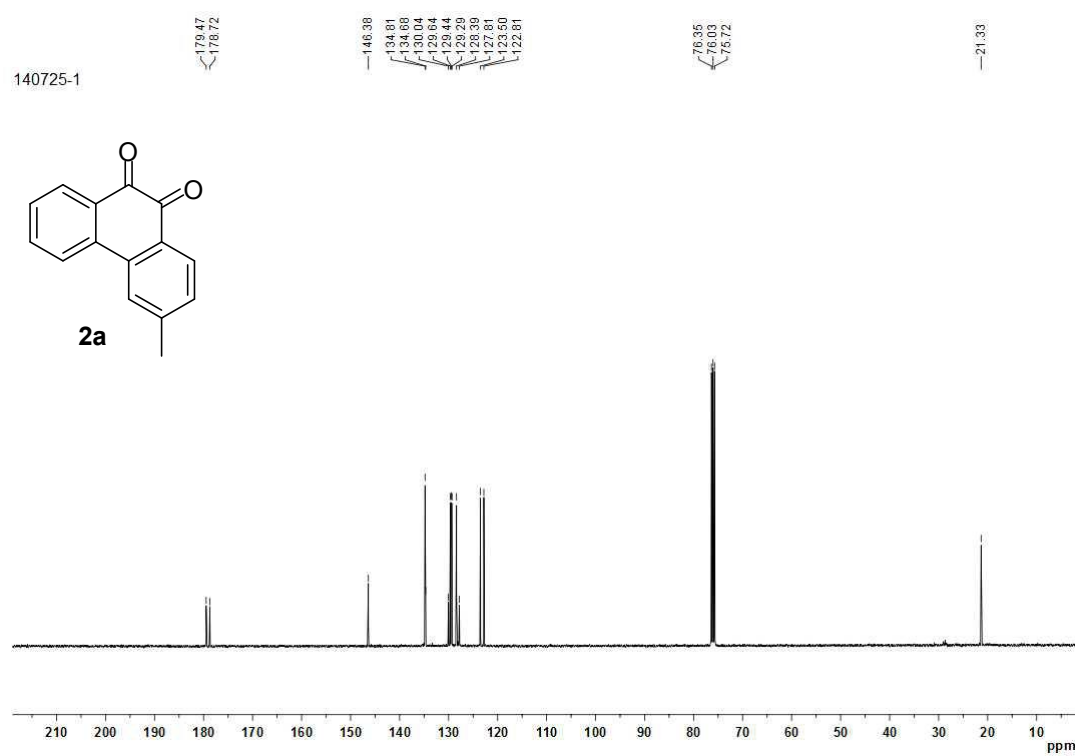
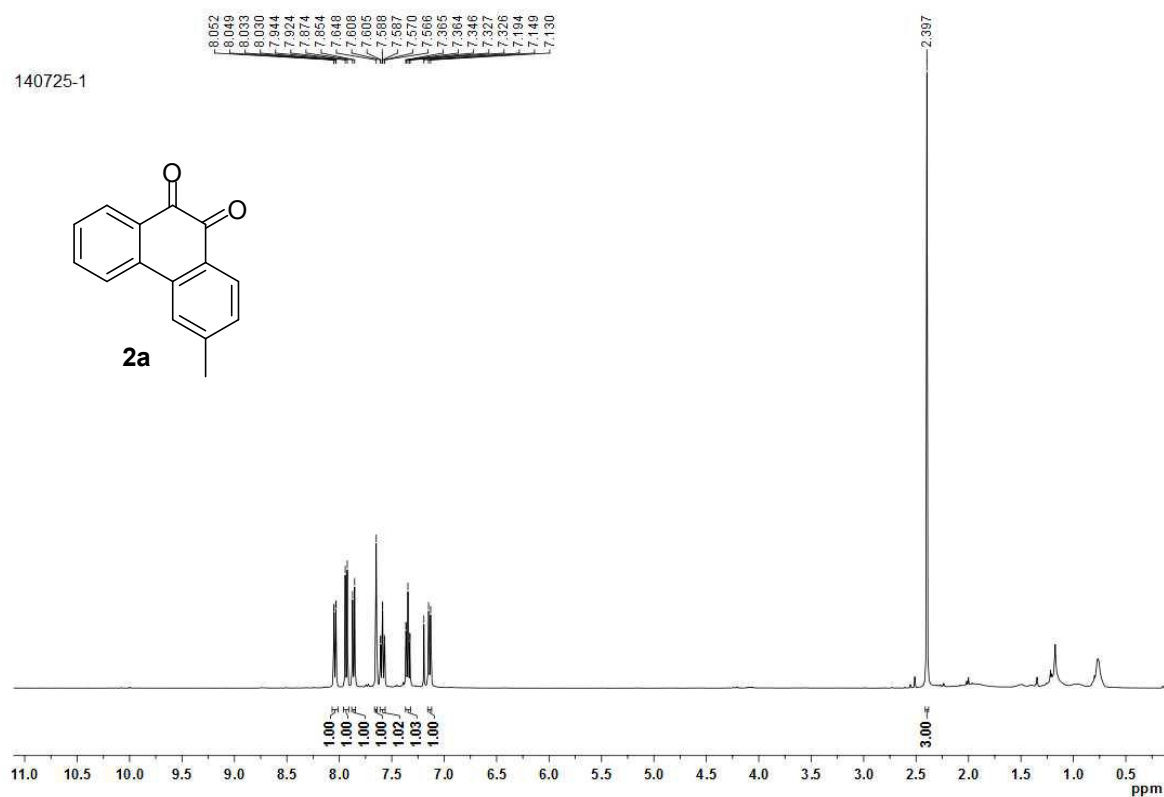
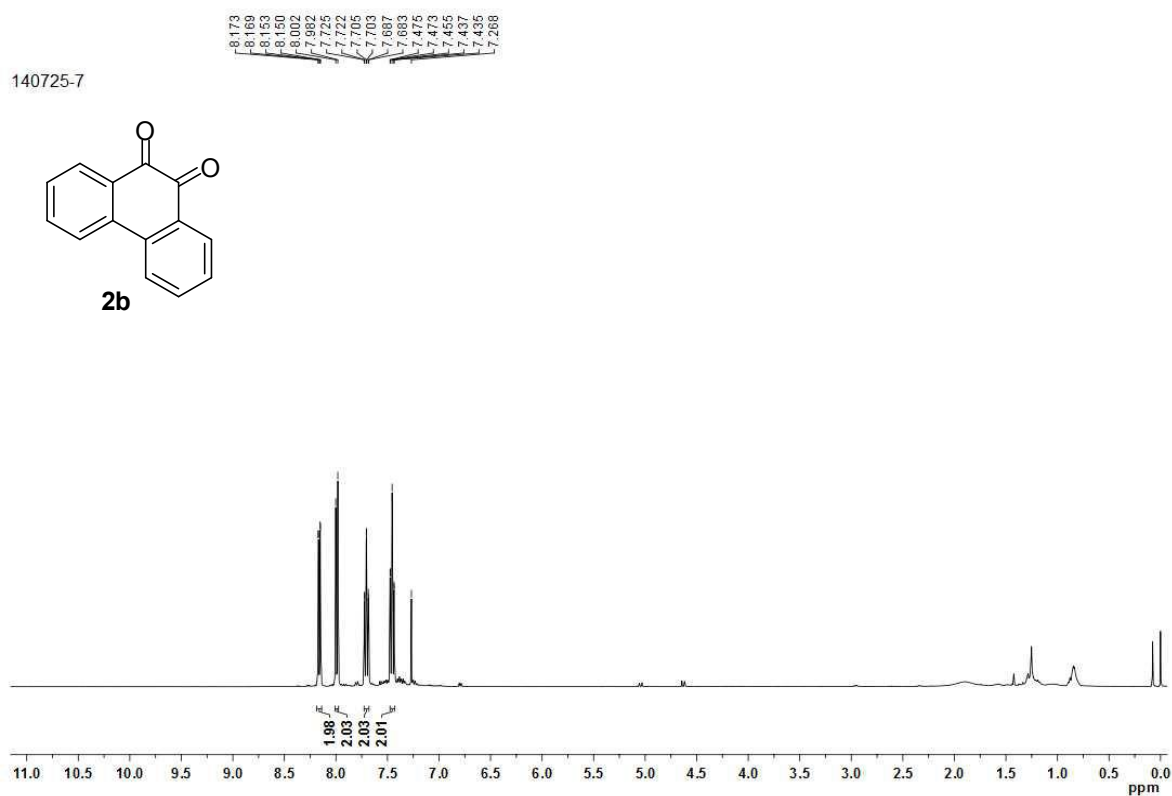
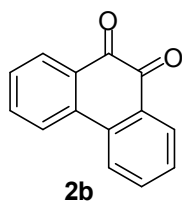


Figure S3. ¹H NMR spectrum of the mixture of **2u** and **2u-d₄**

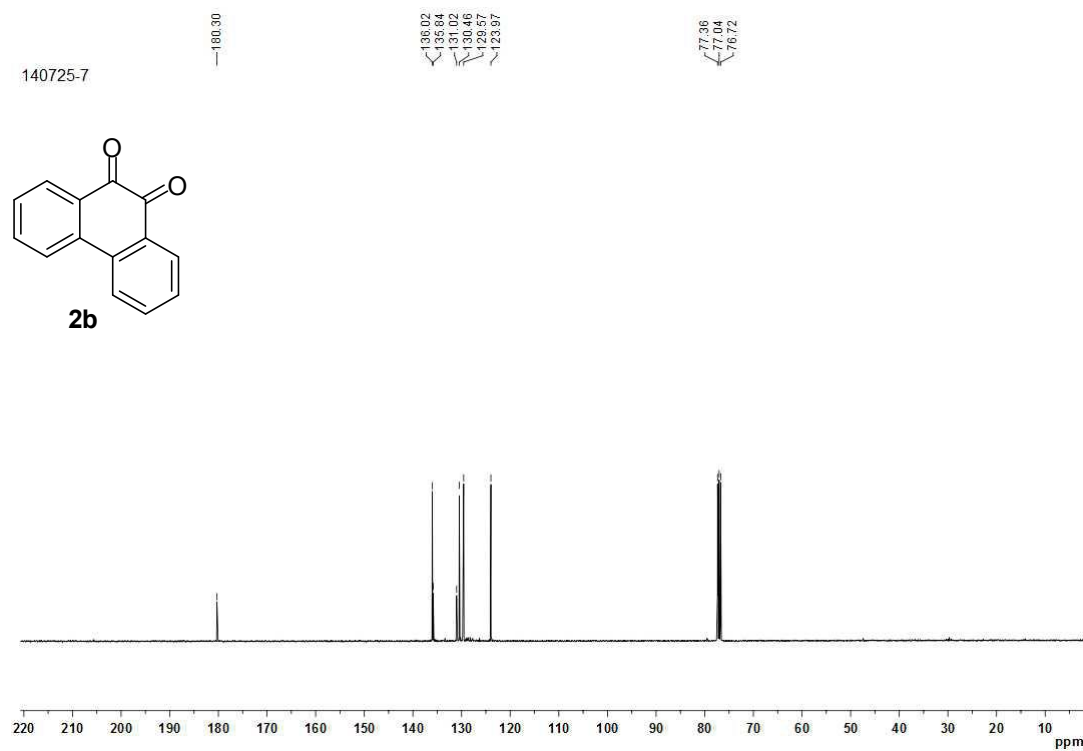
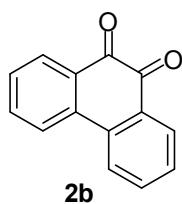
3. ^1H and ^{13}C NMR spectra of 2a-2zd



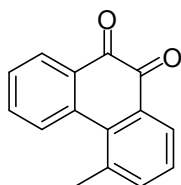
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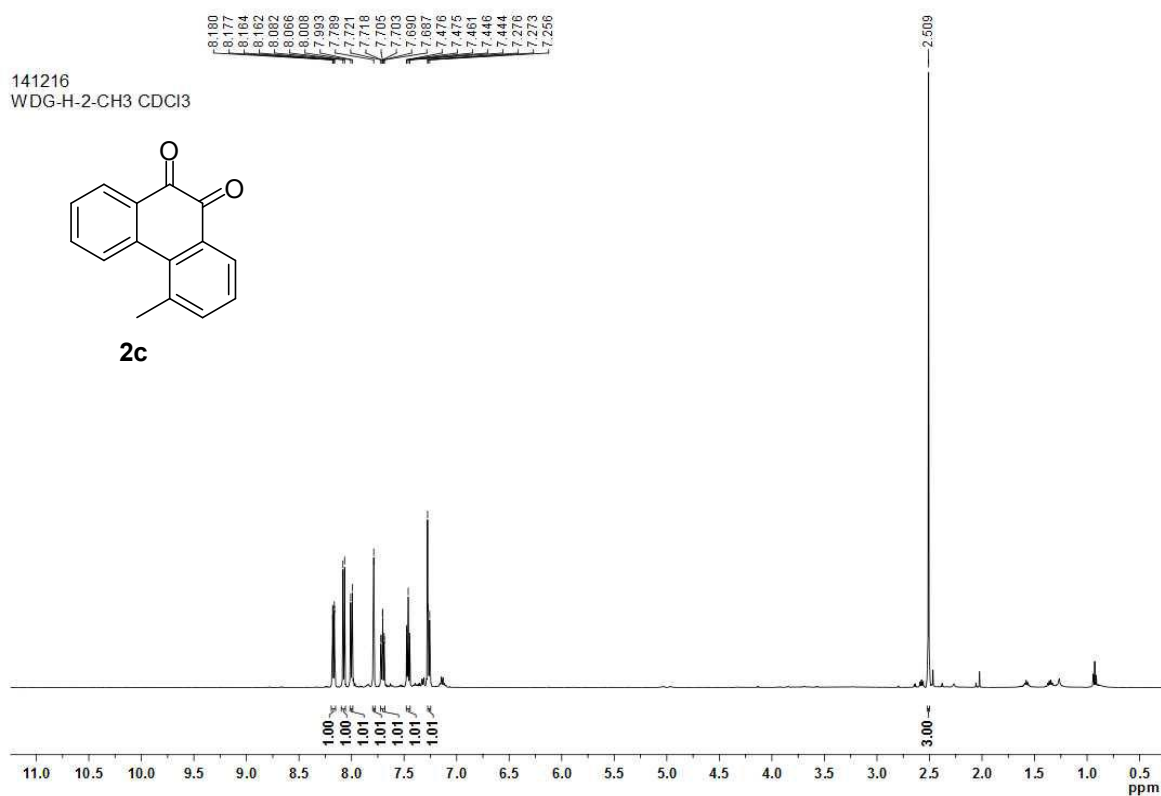
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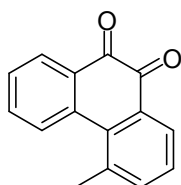
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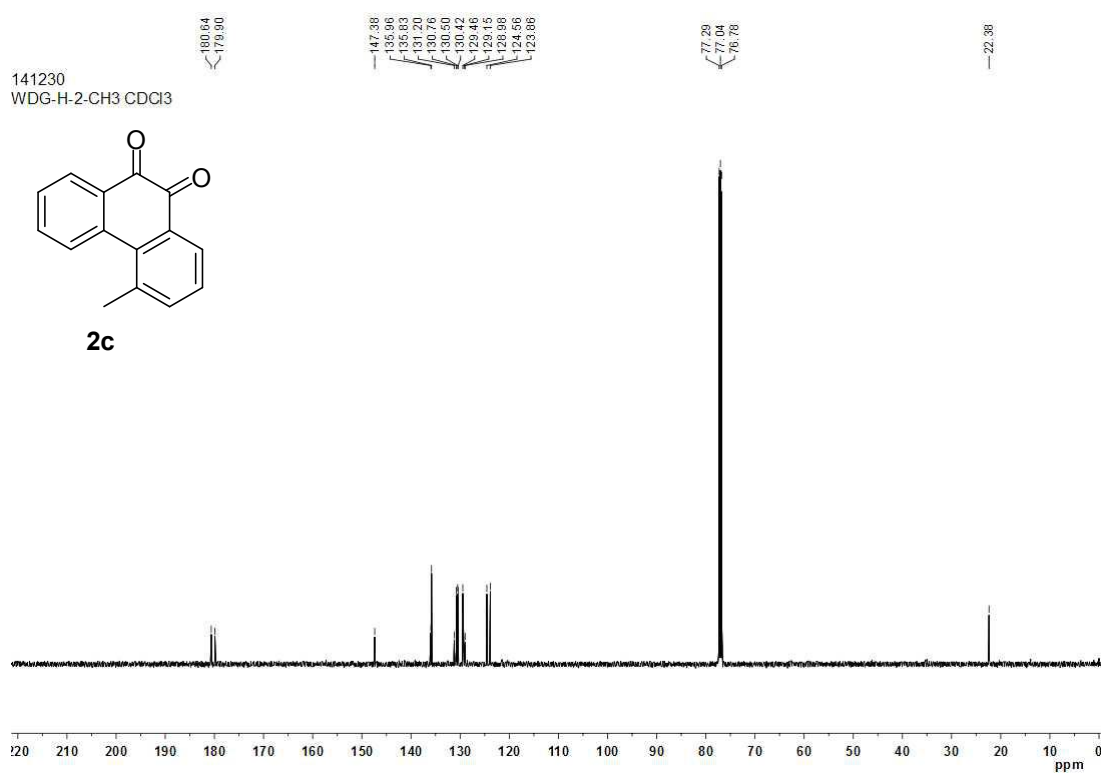
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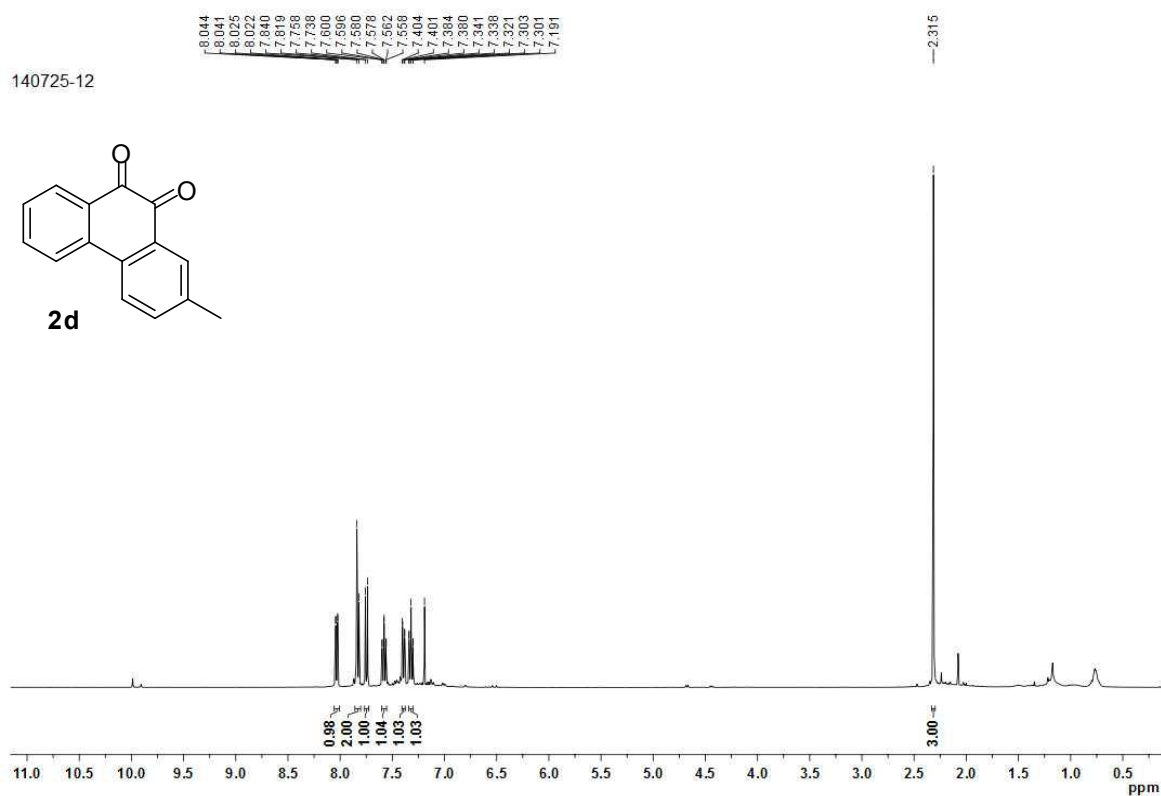
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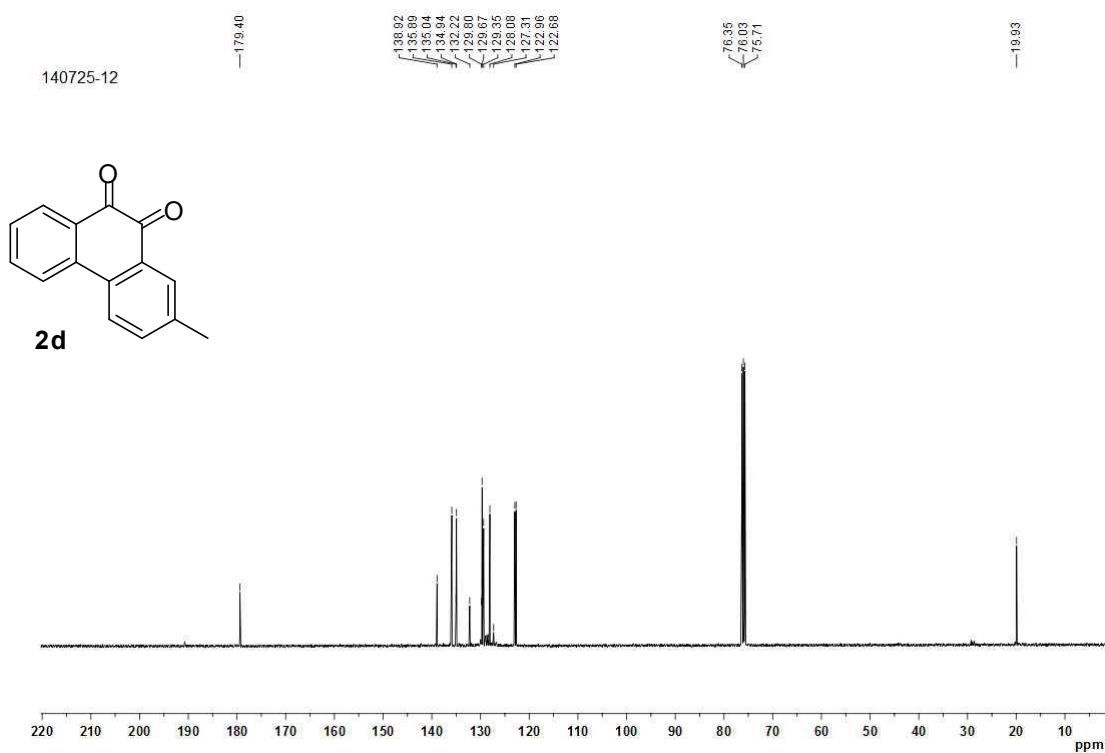
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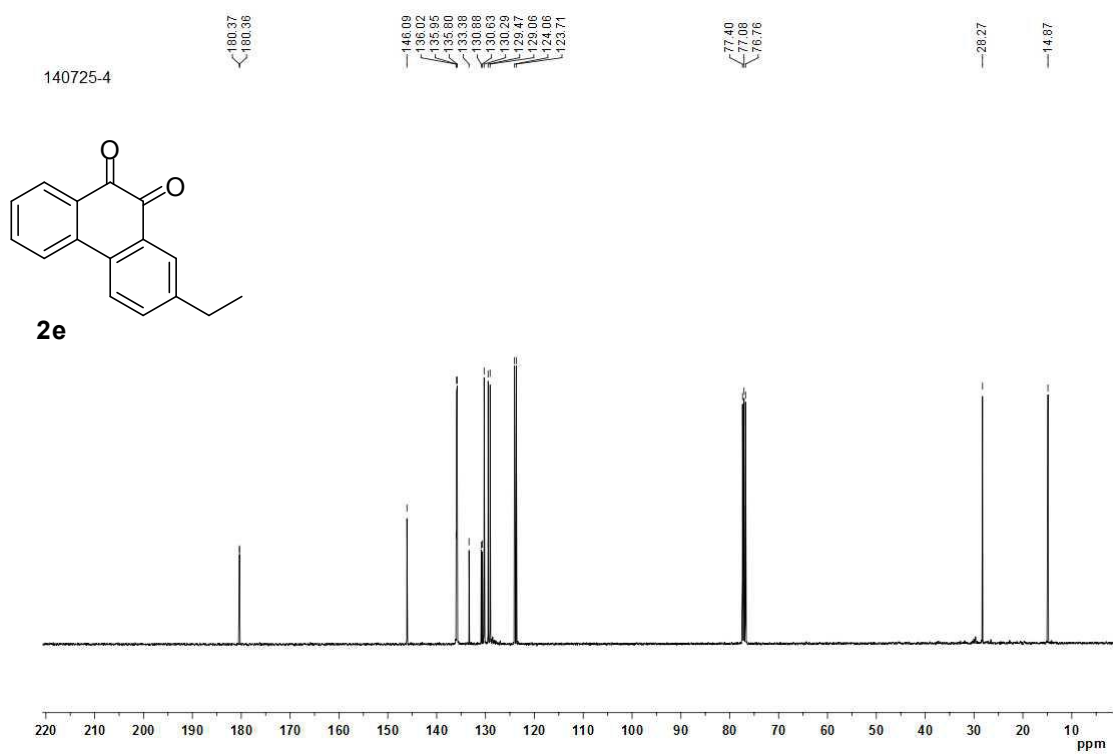
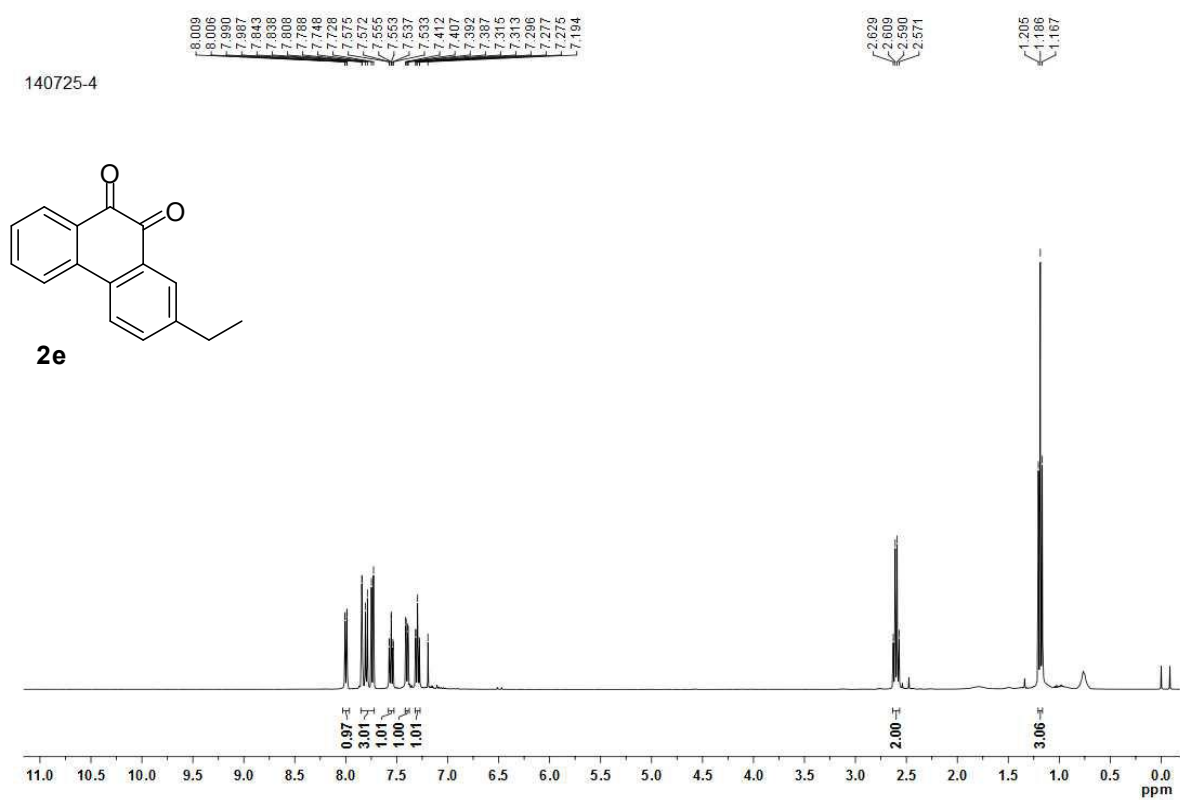


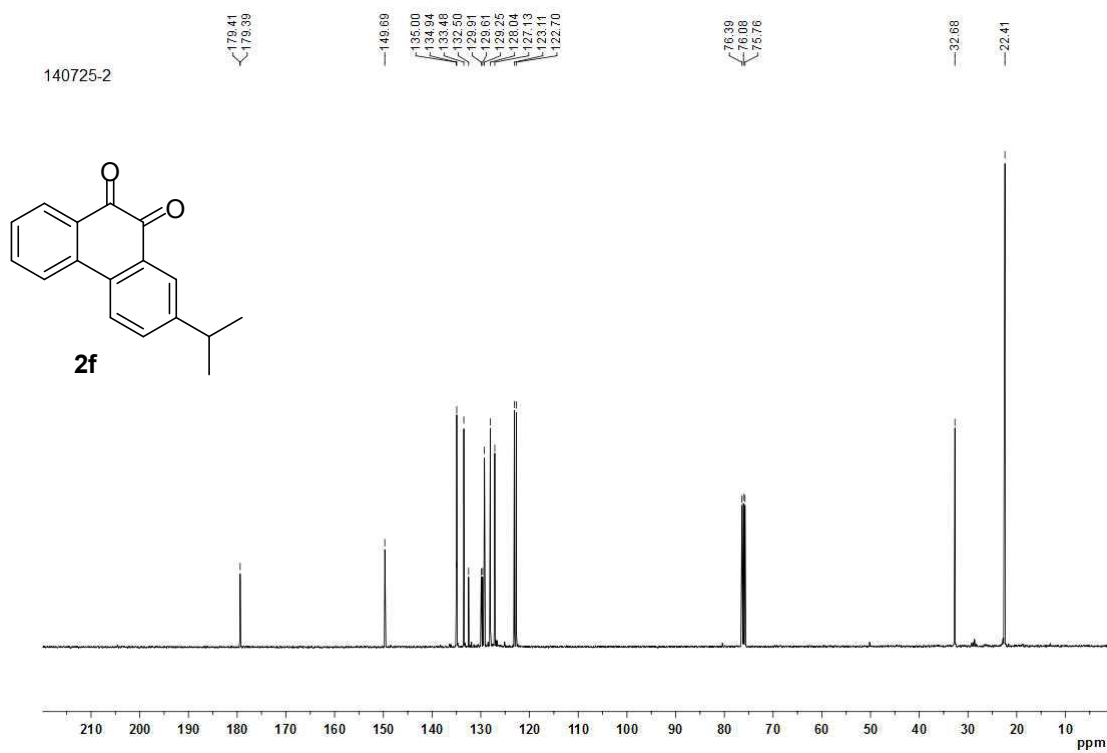
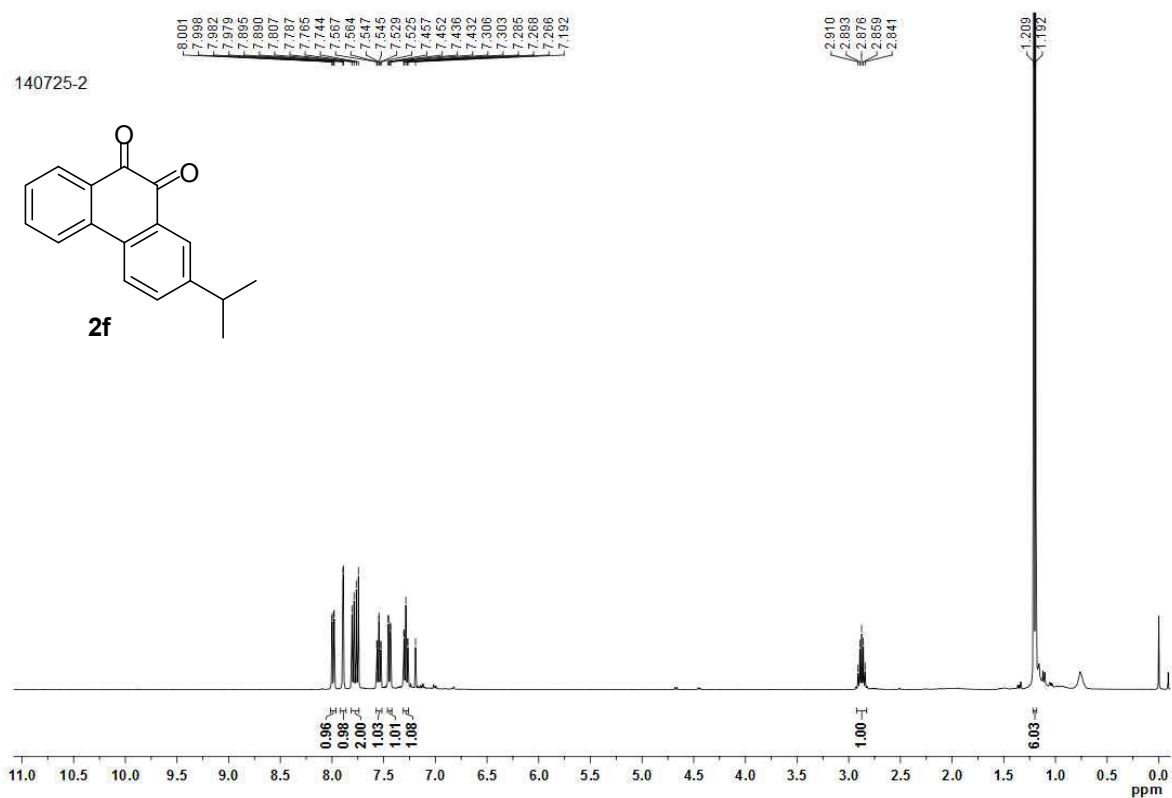
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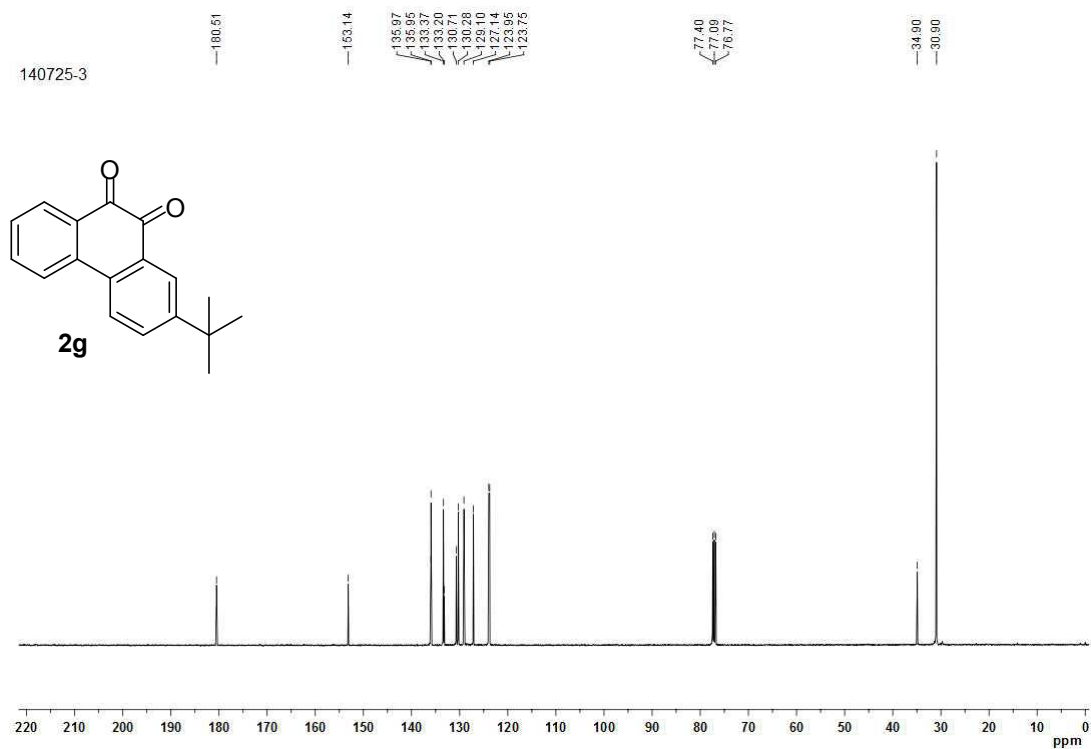
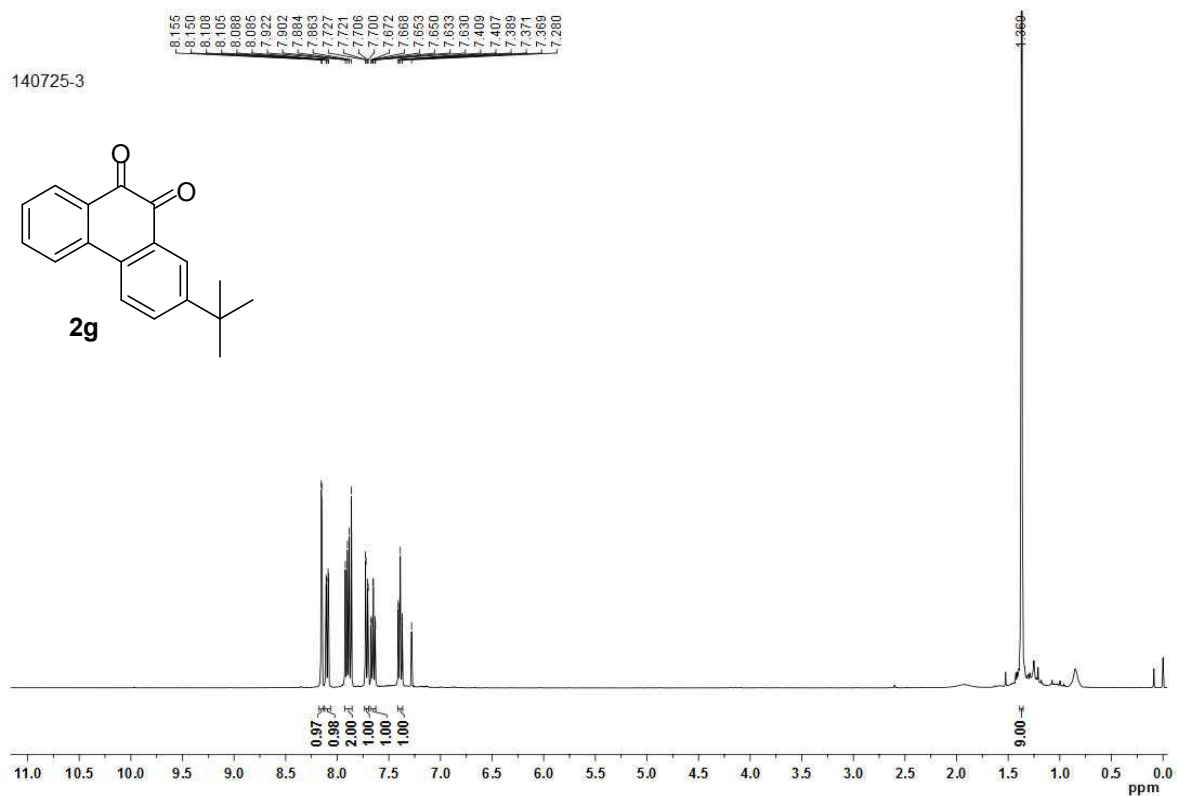


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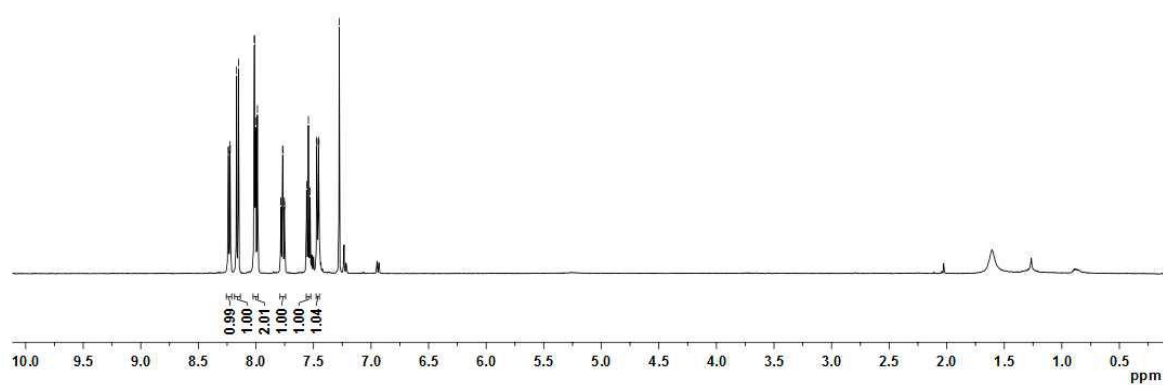
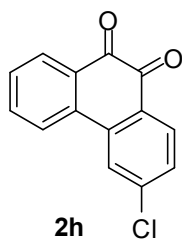




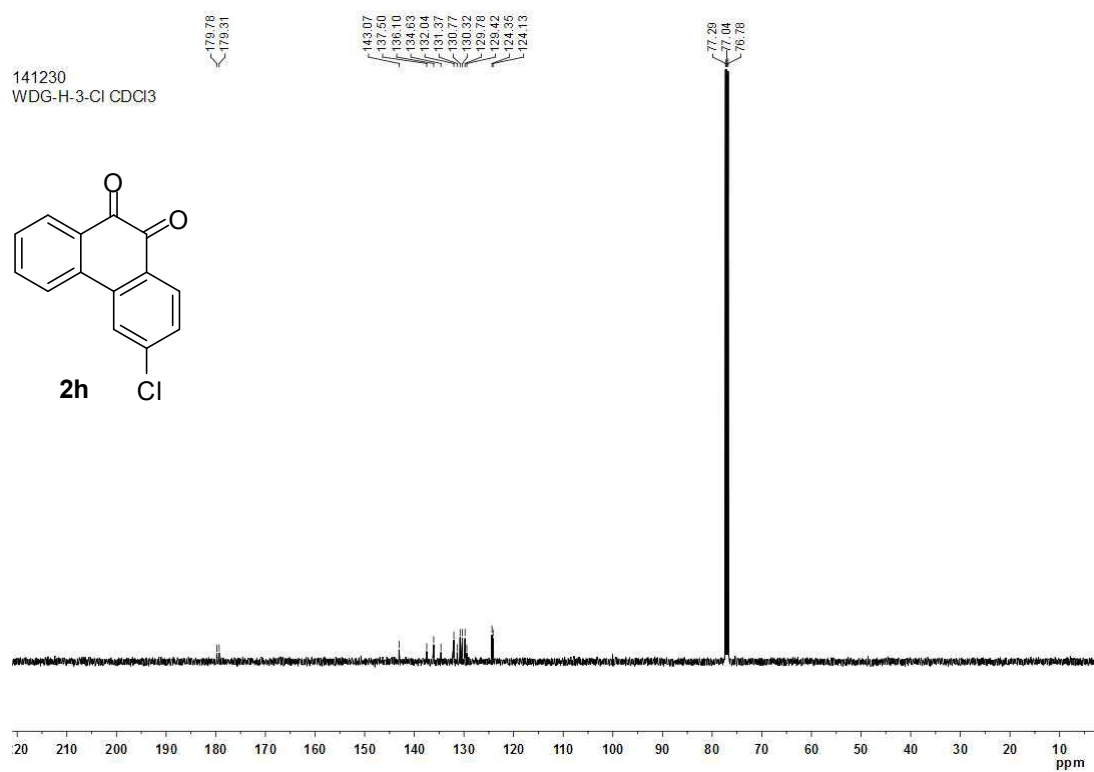
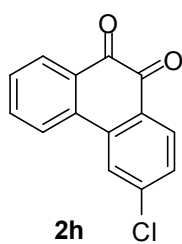




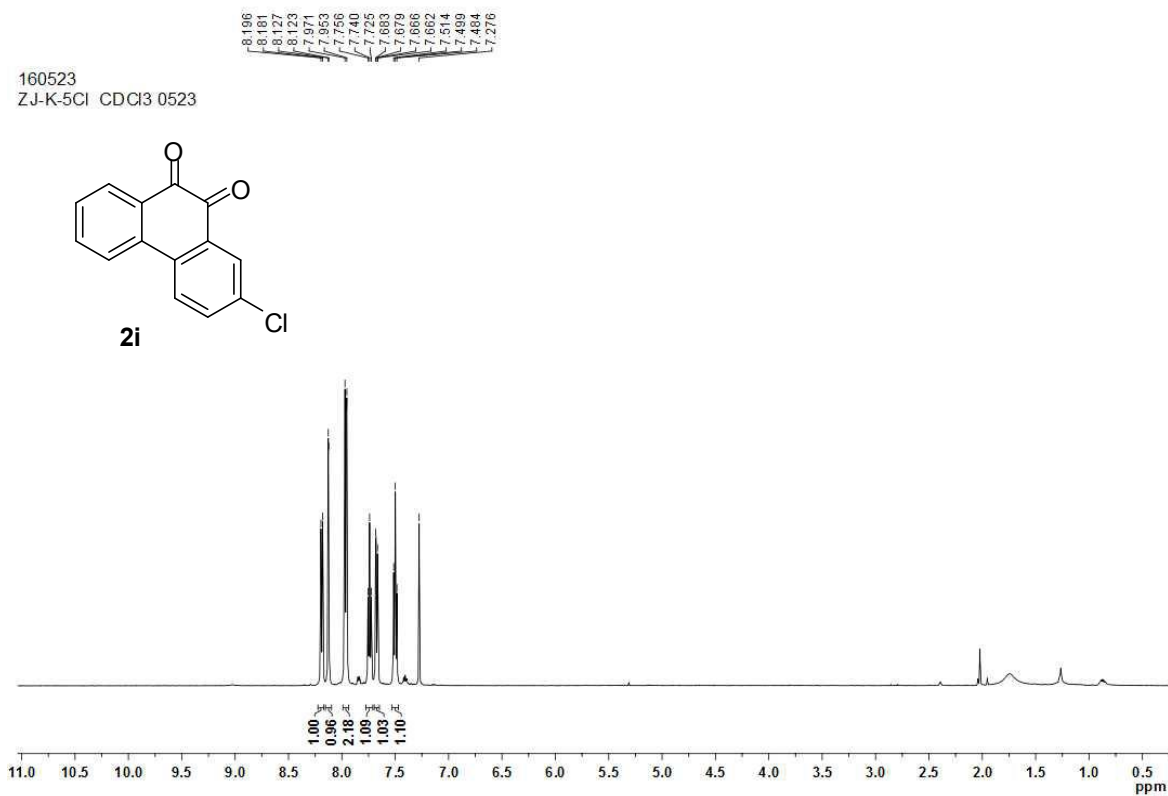
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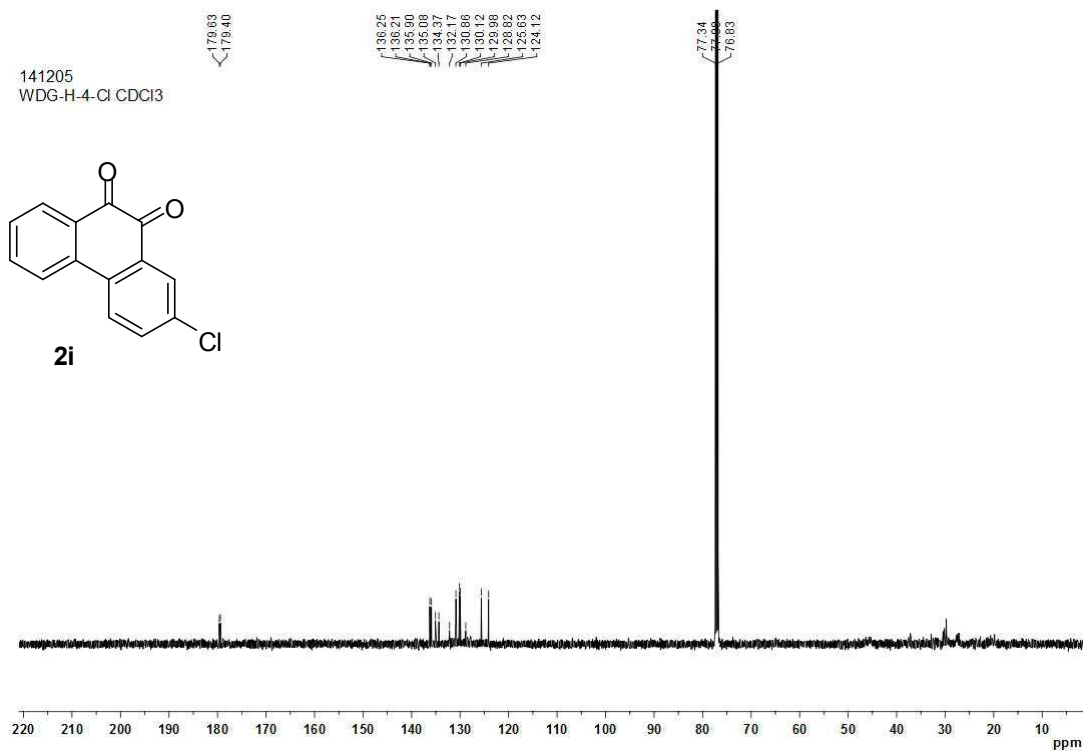
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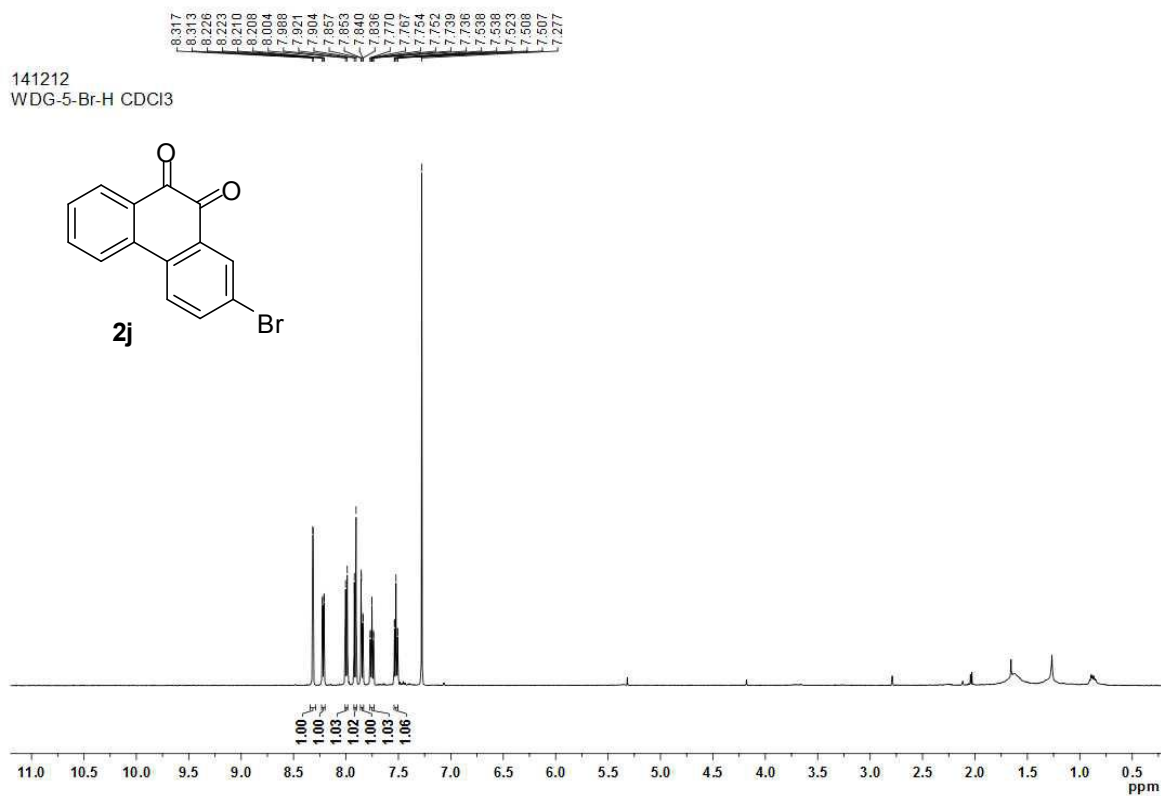
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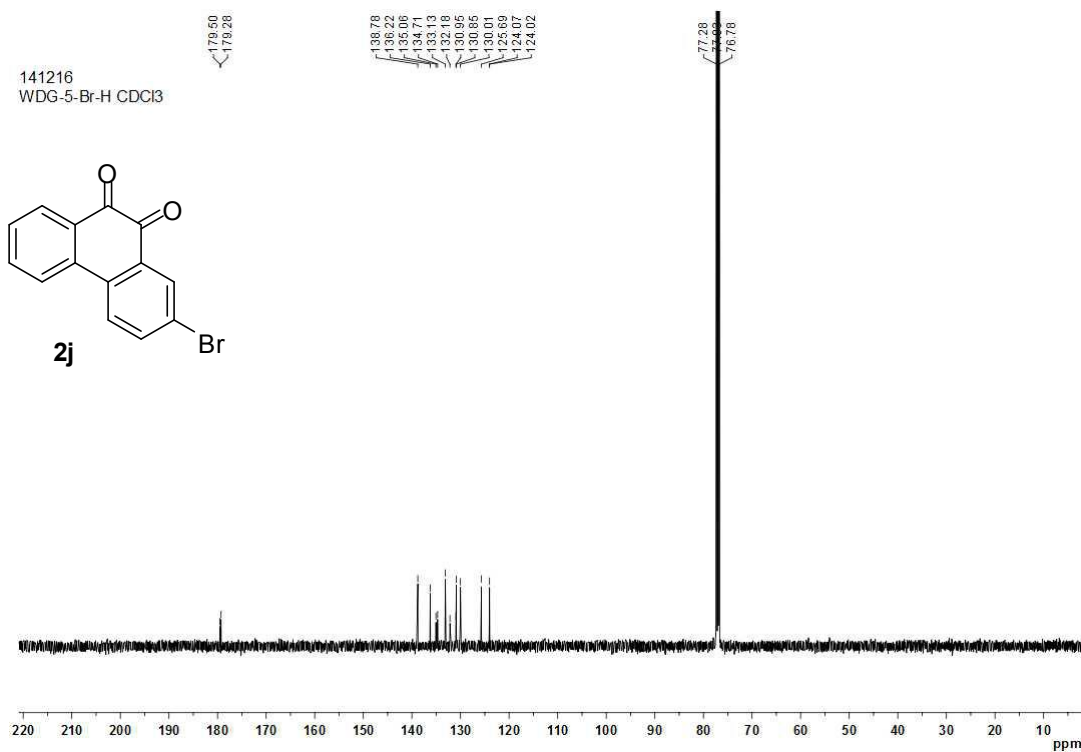
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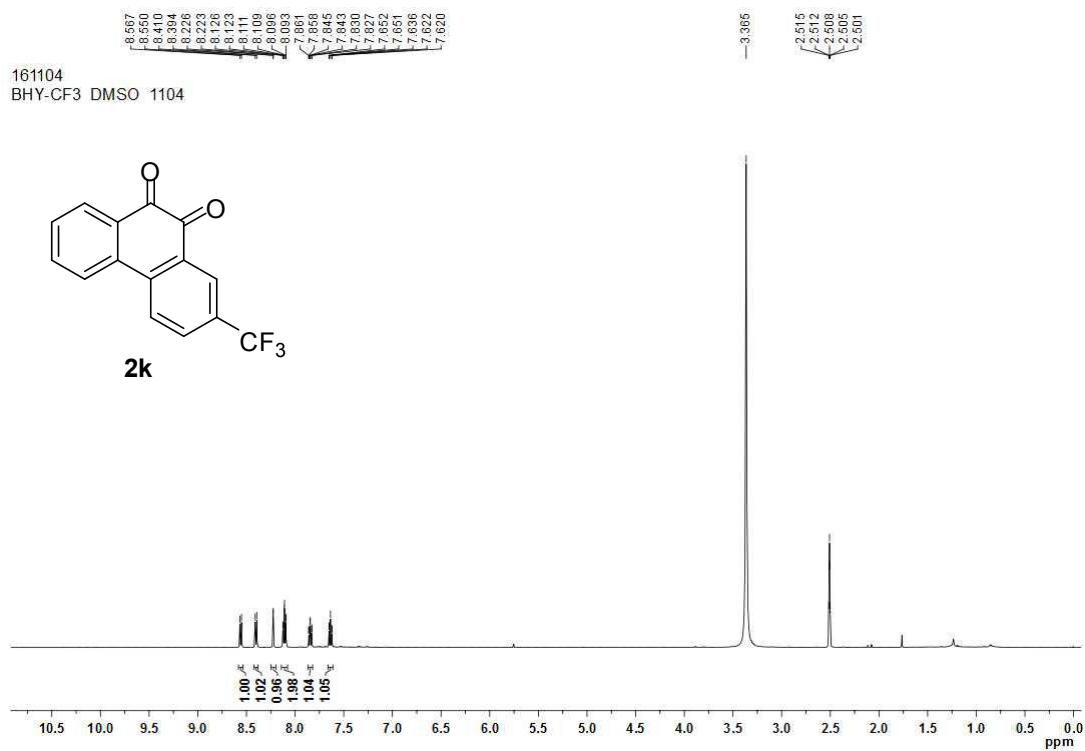
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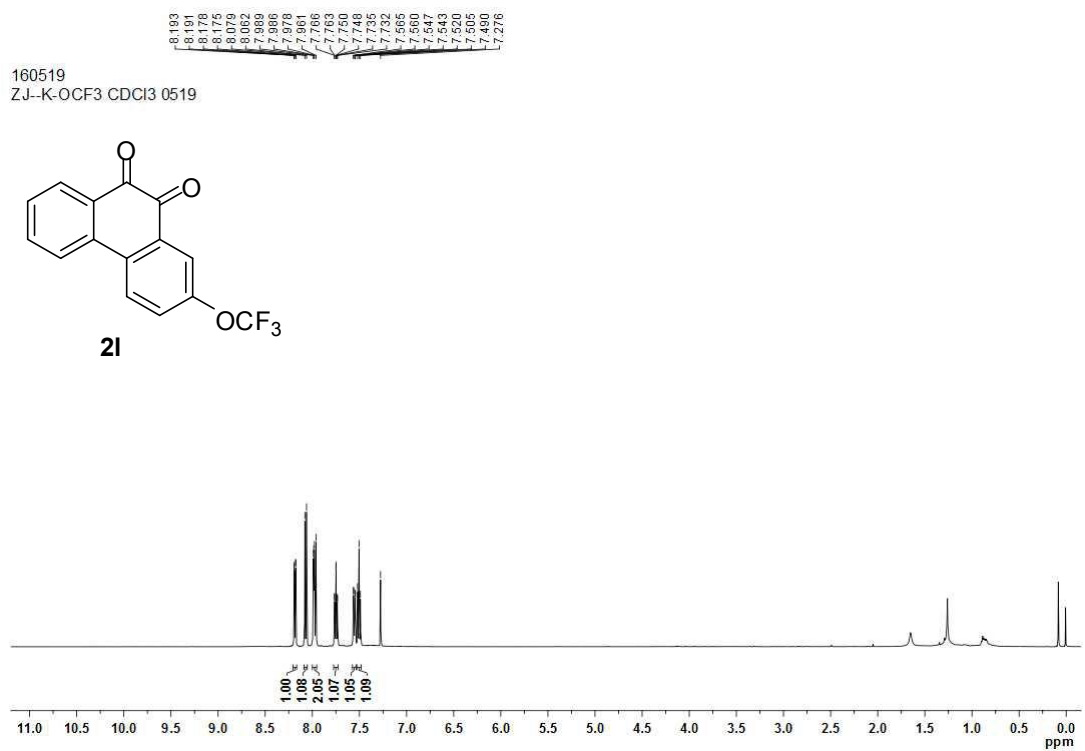
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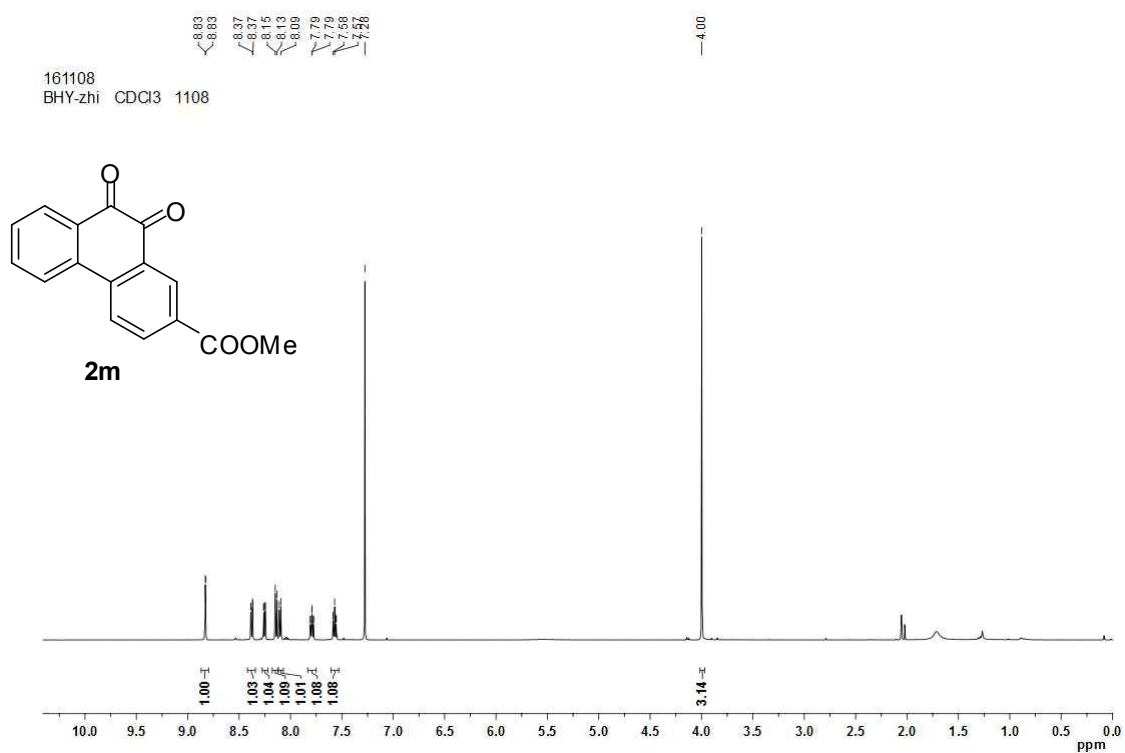
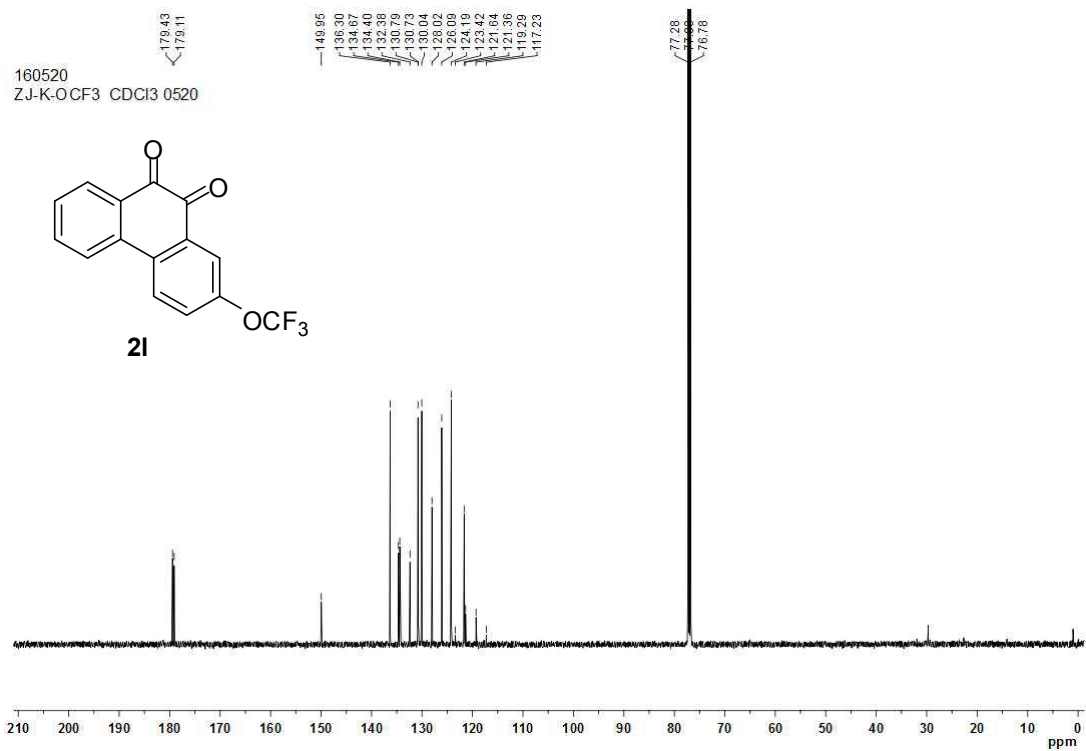


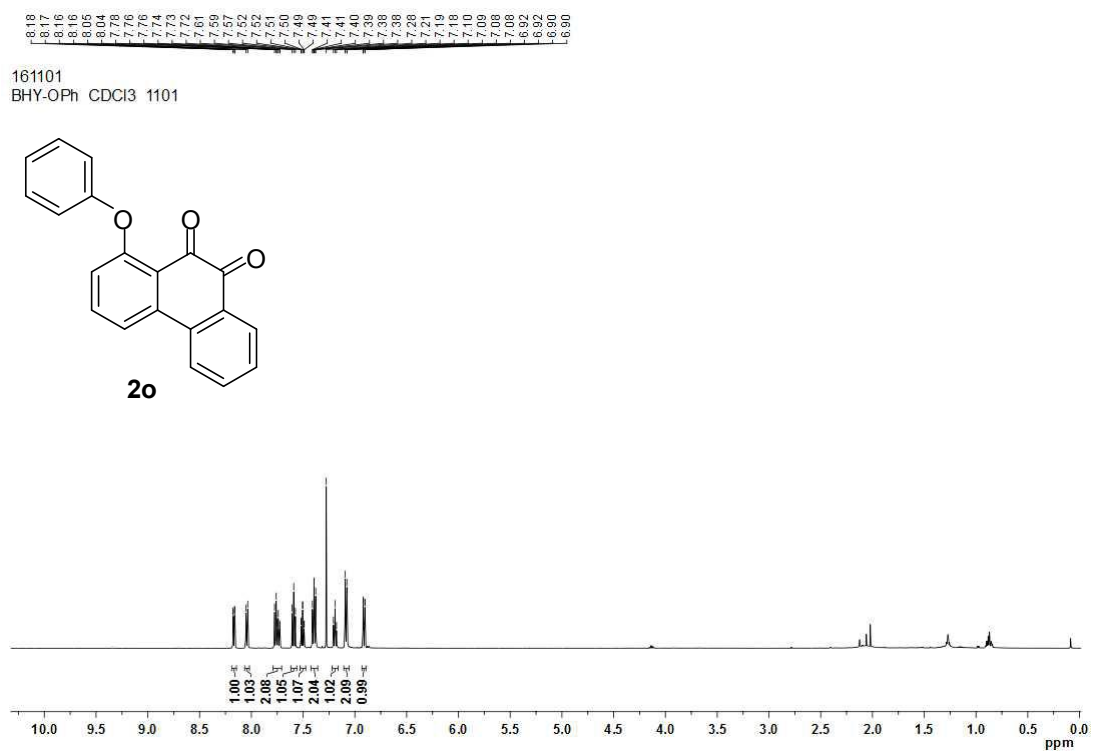
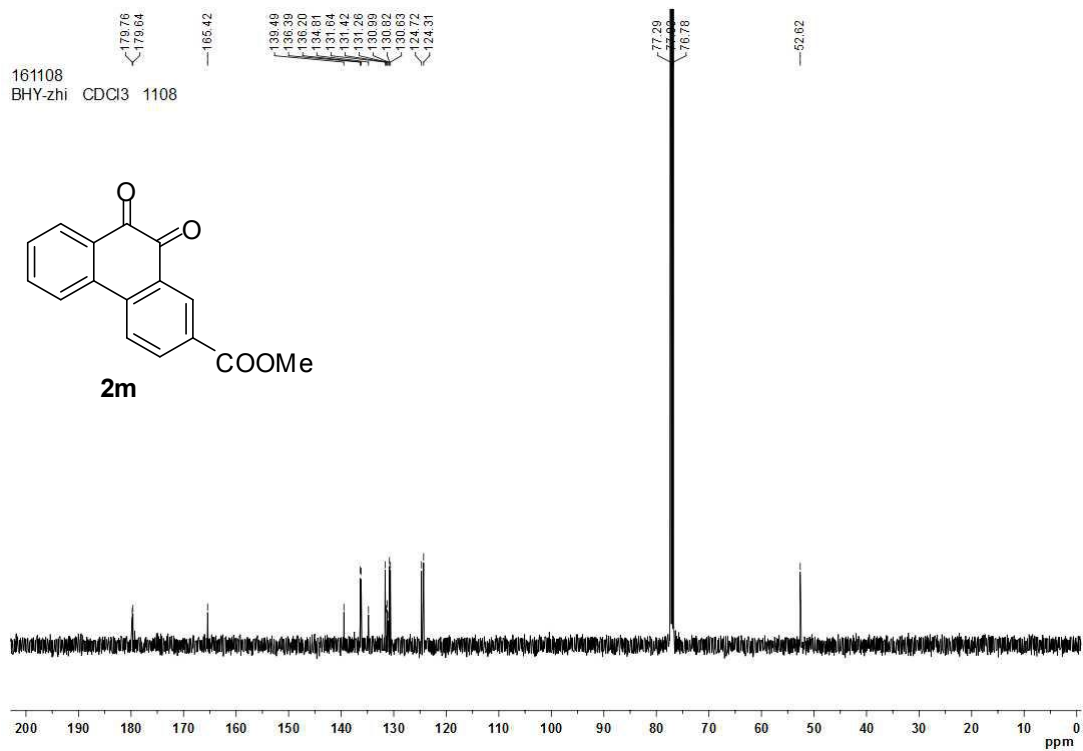
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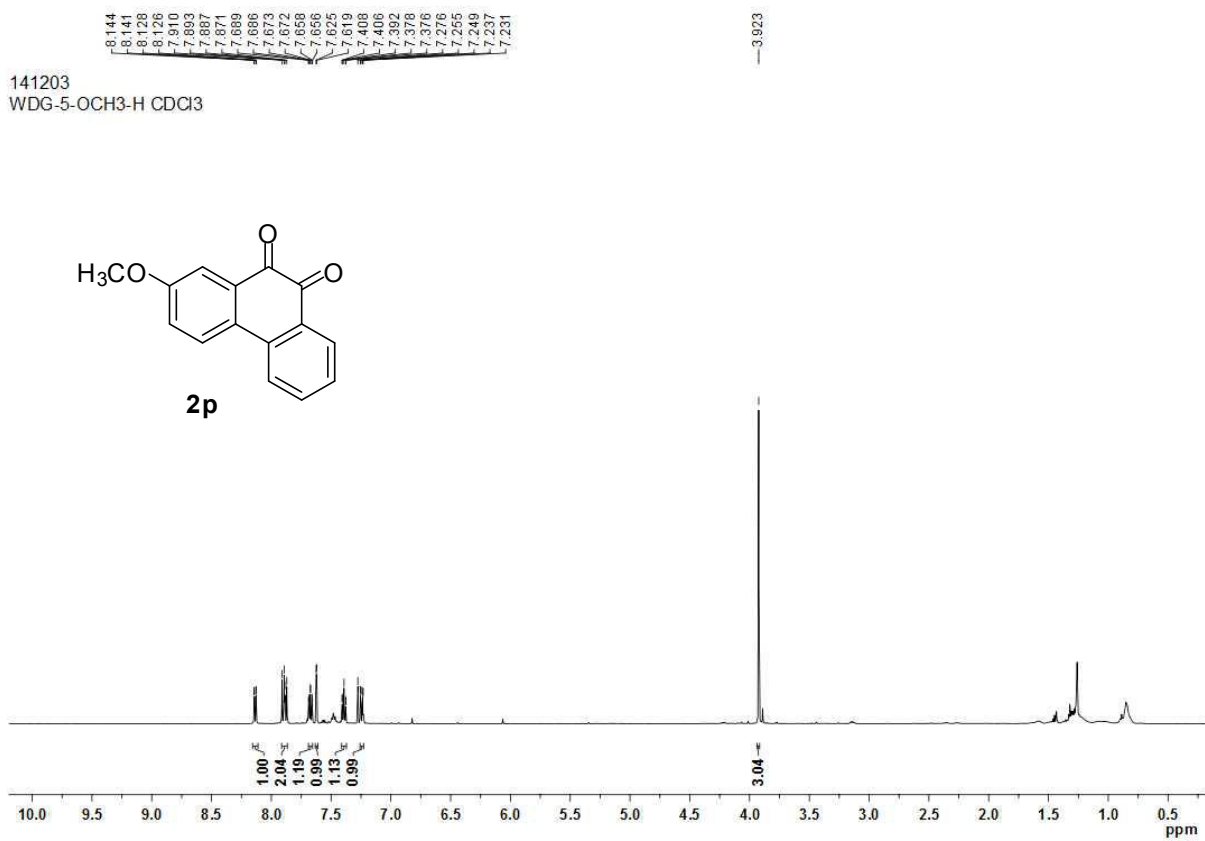
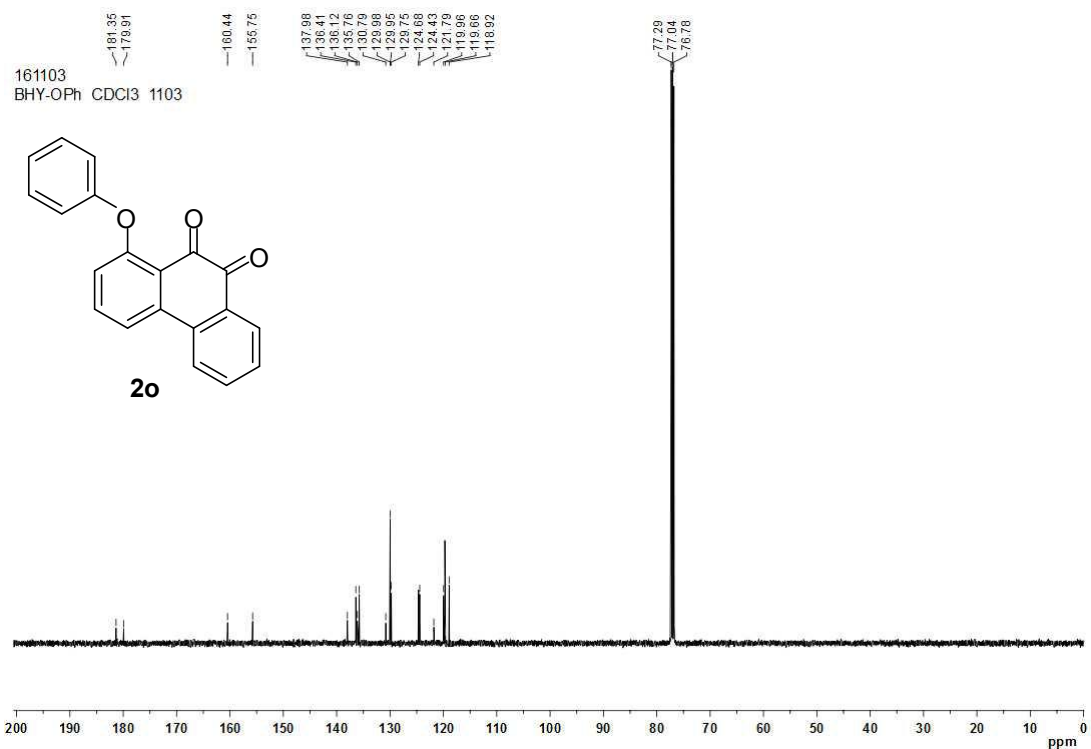


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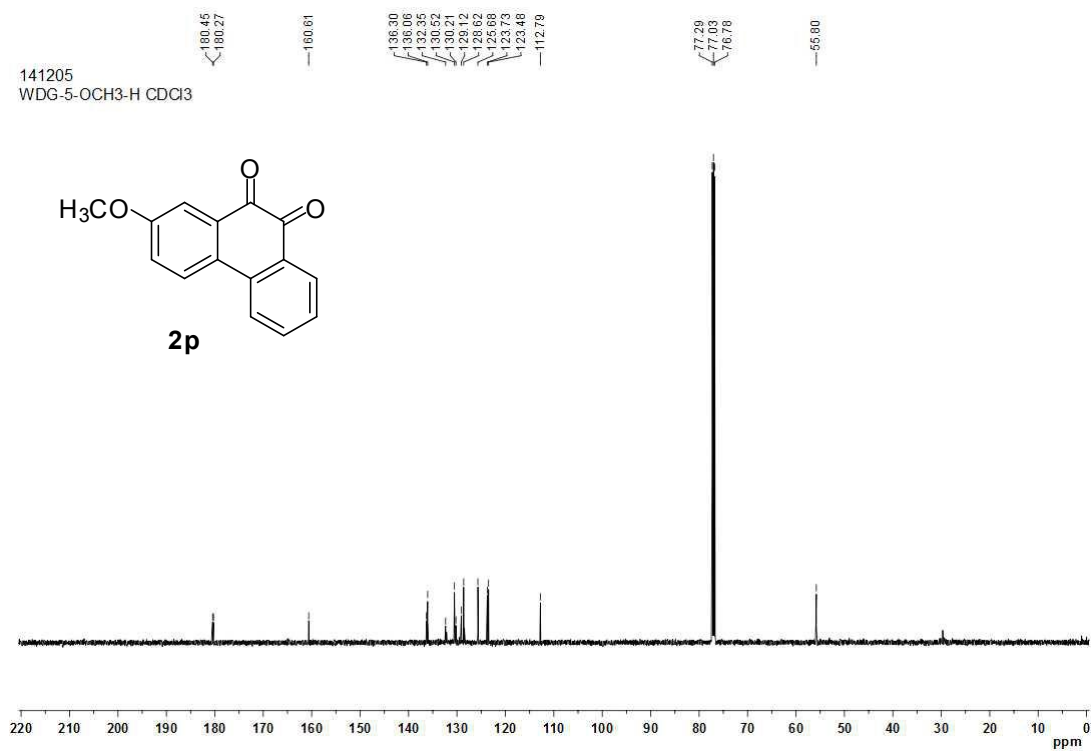




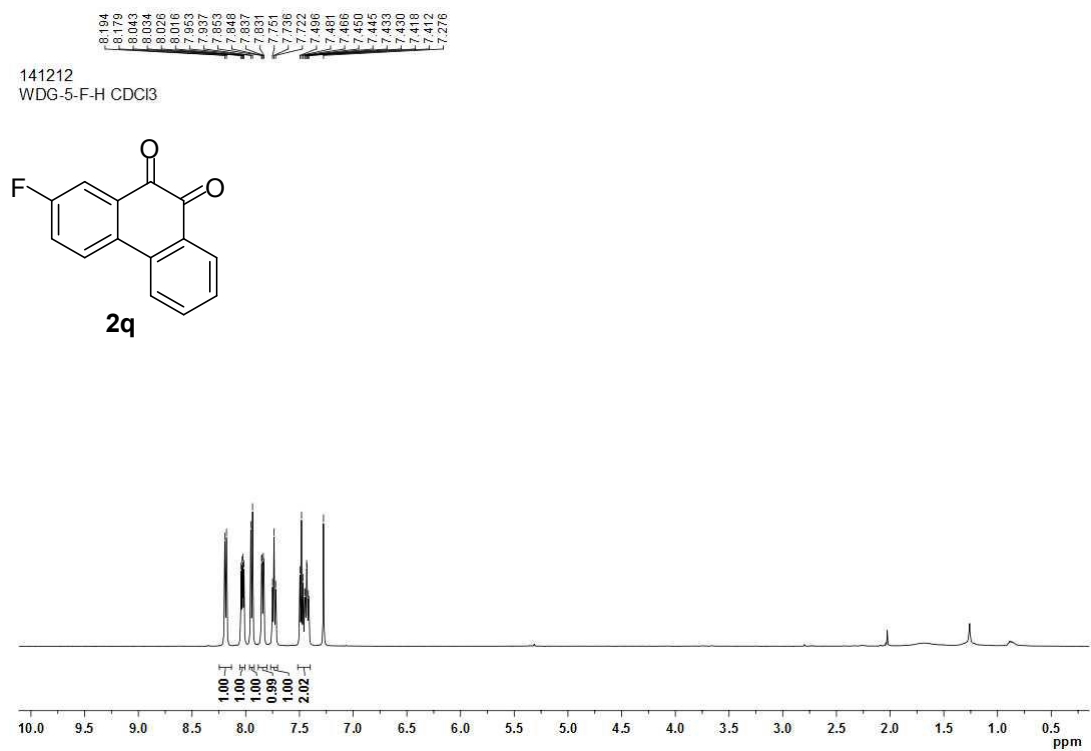




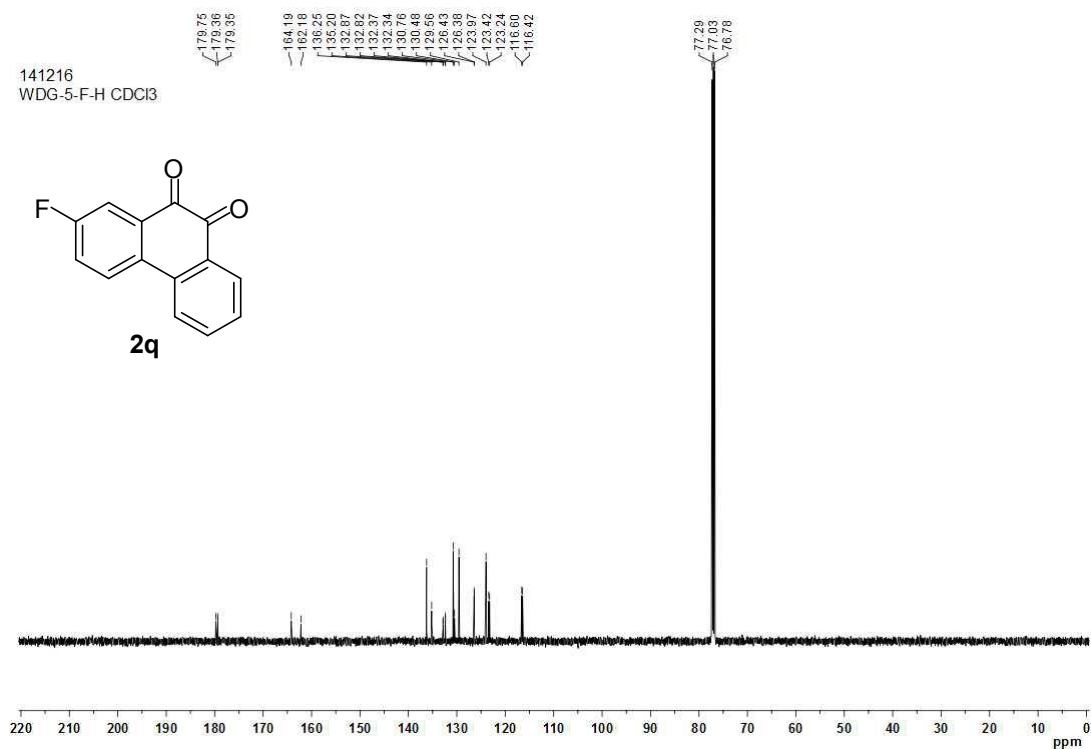
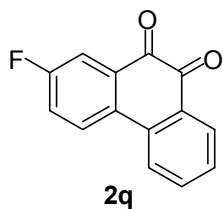
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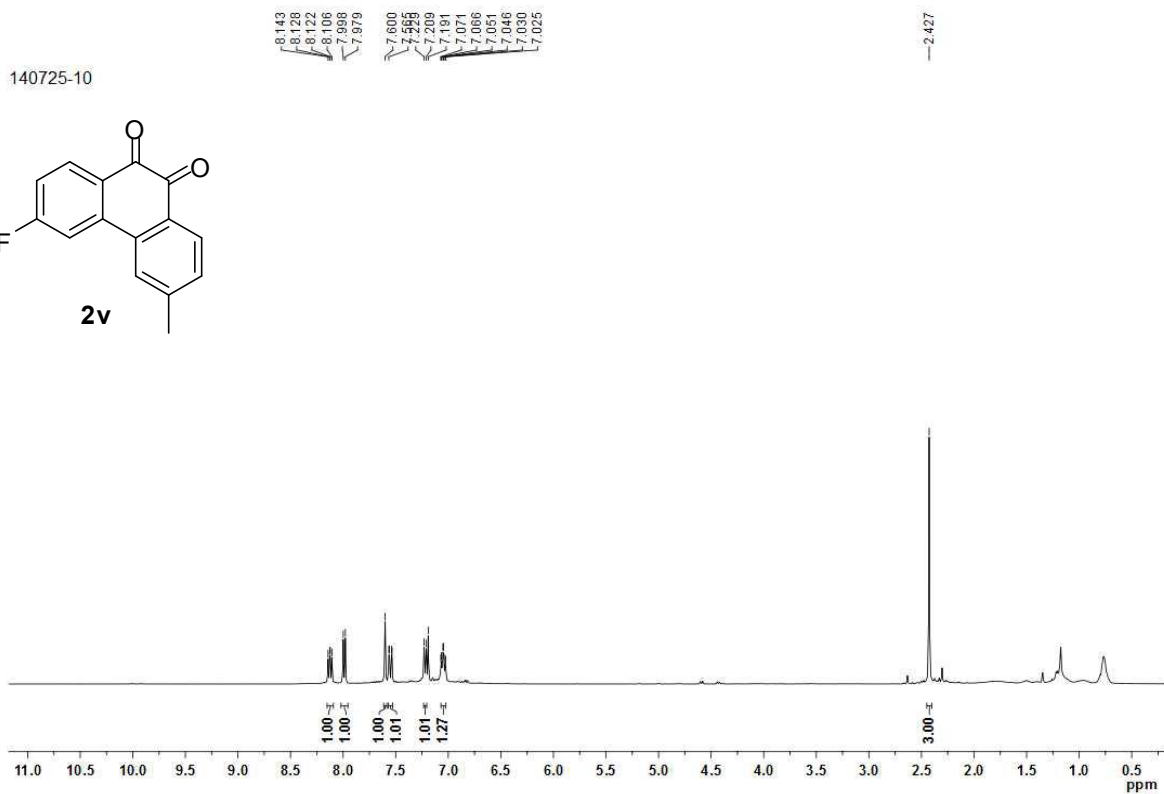
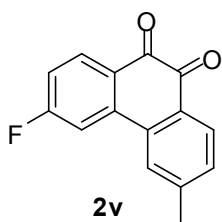
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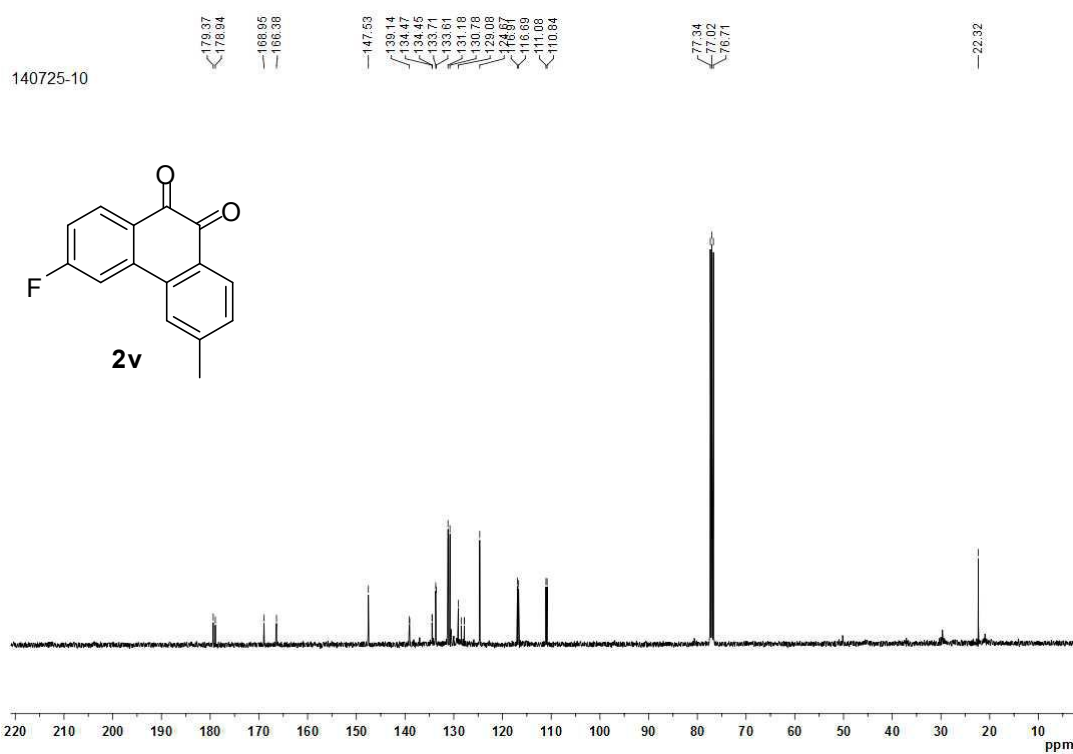
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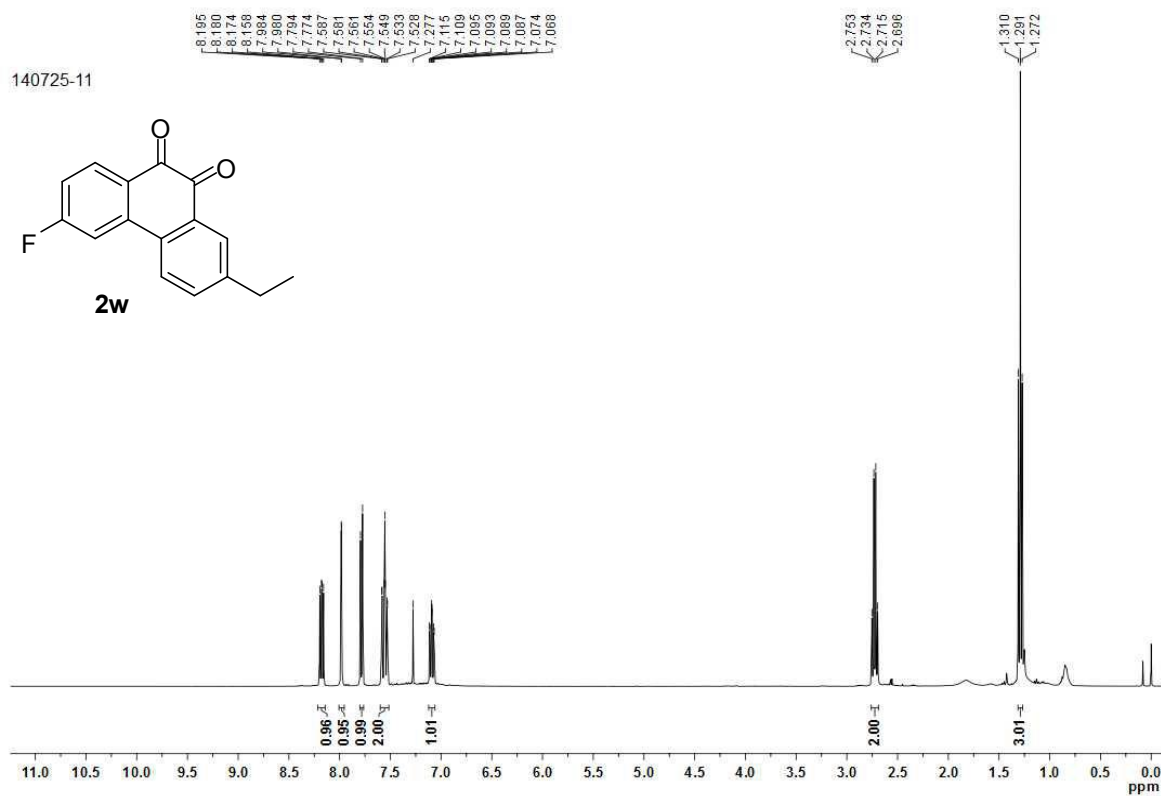
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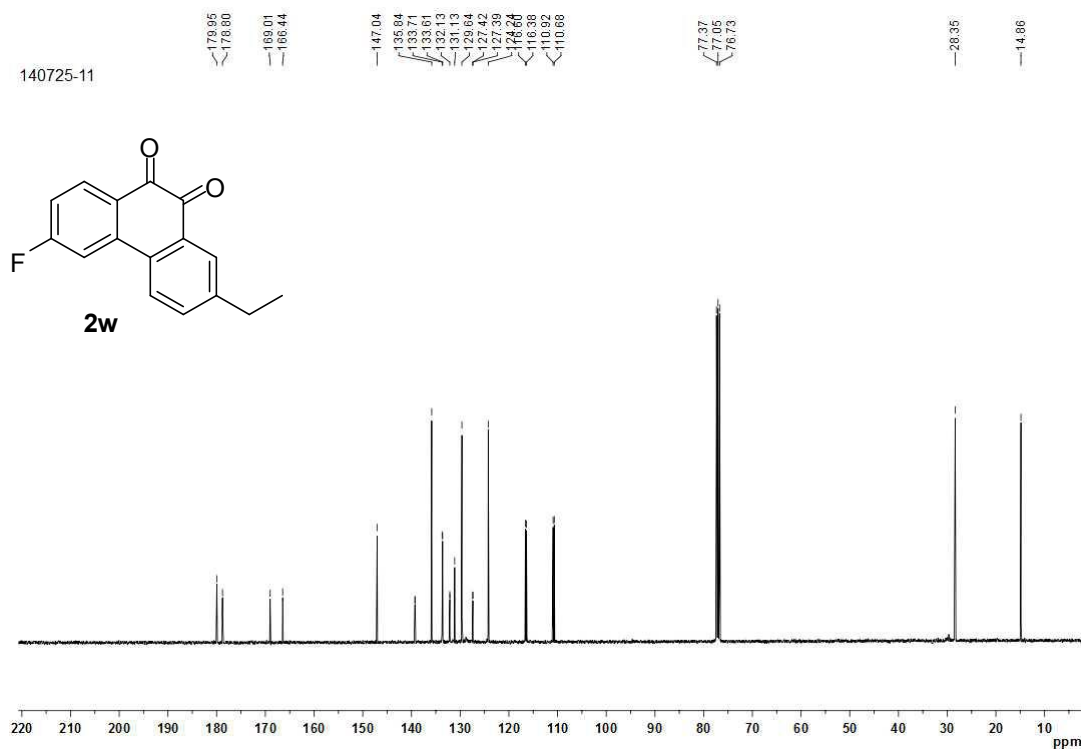
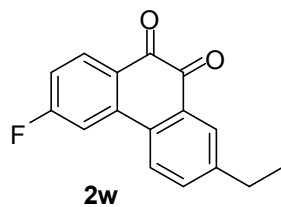
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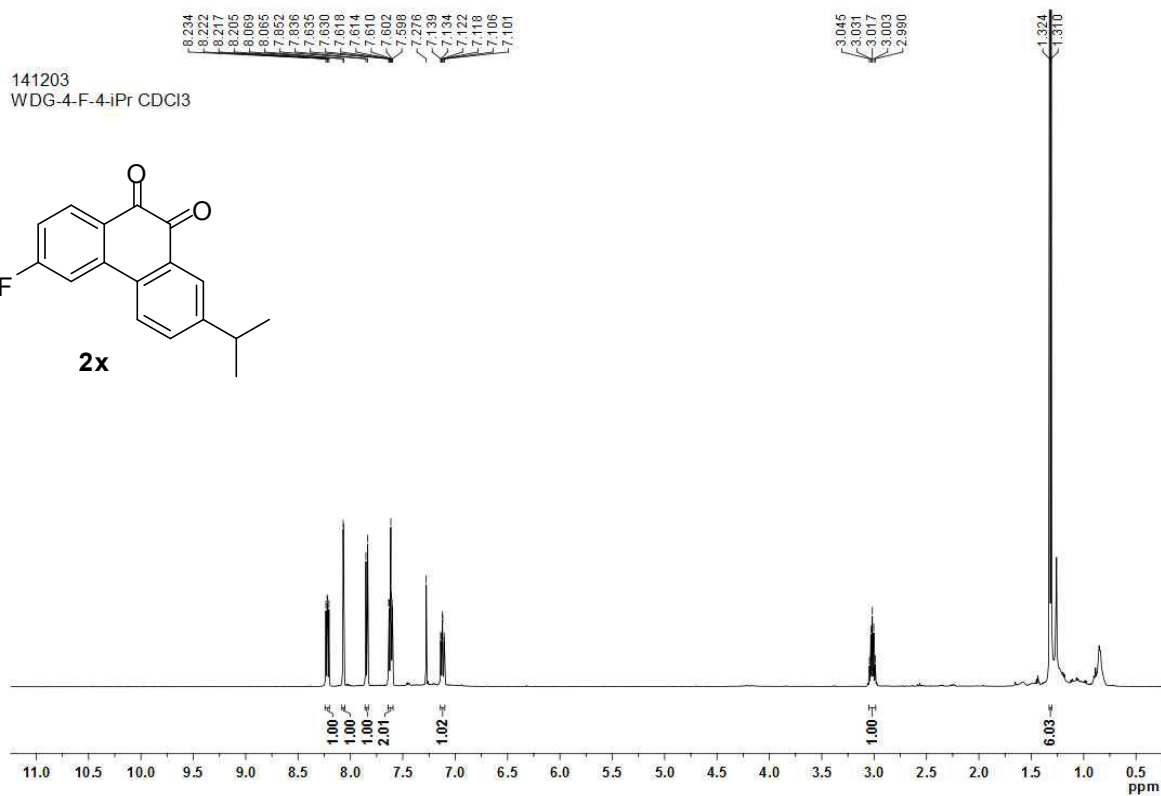
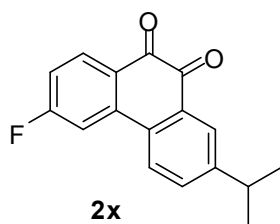
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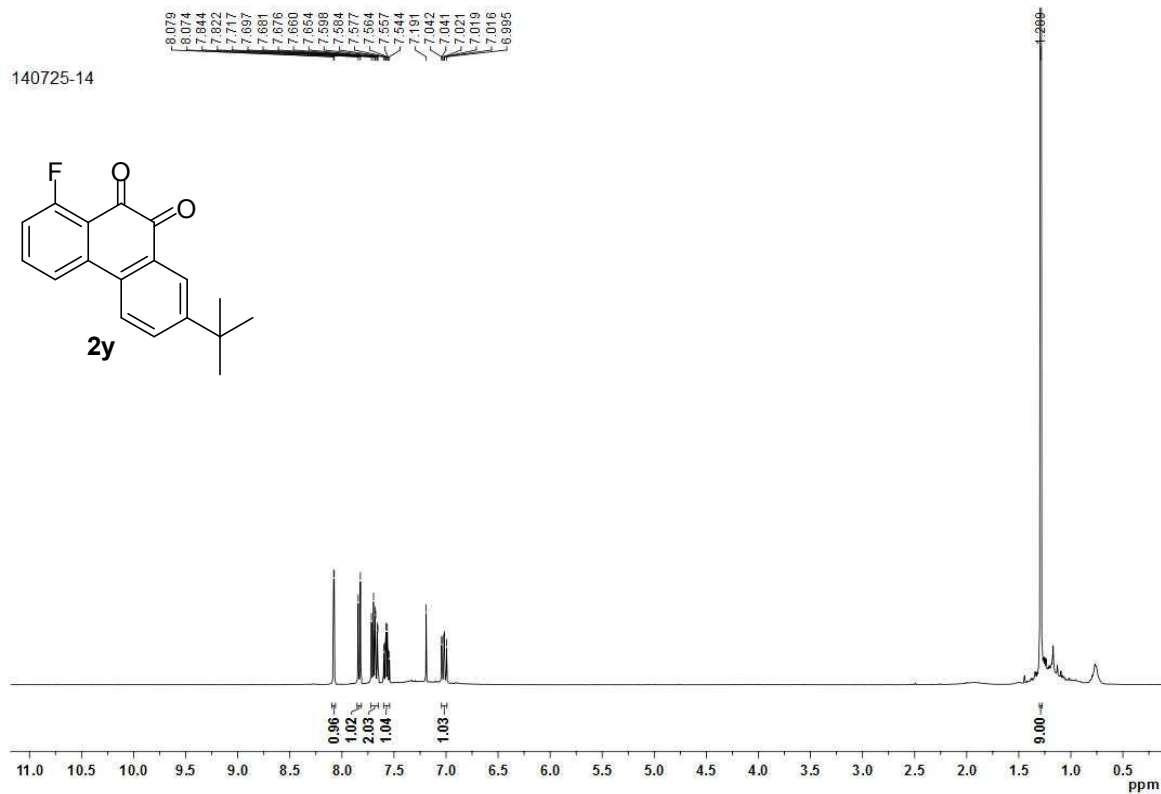
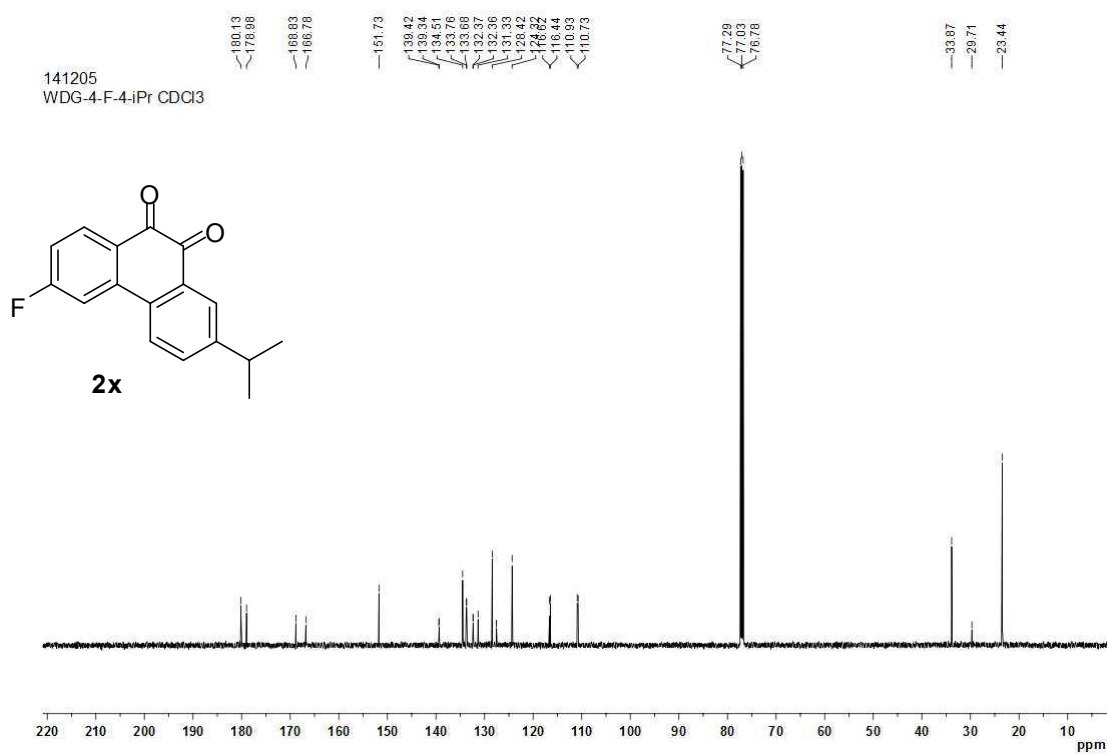


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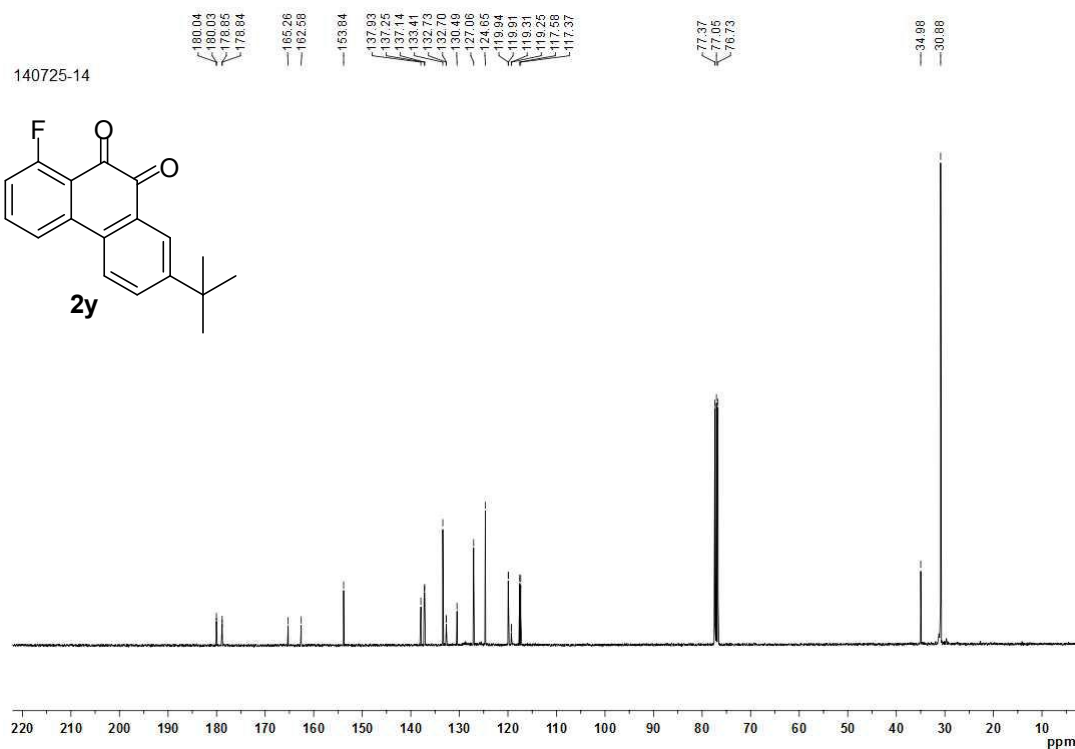
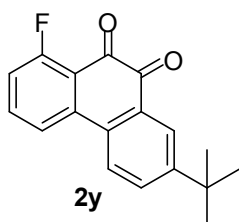


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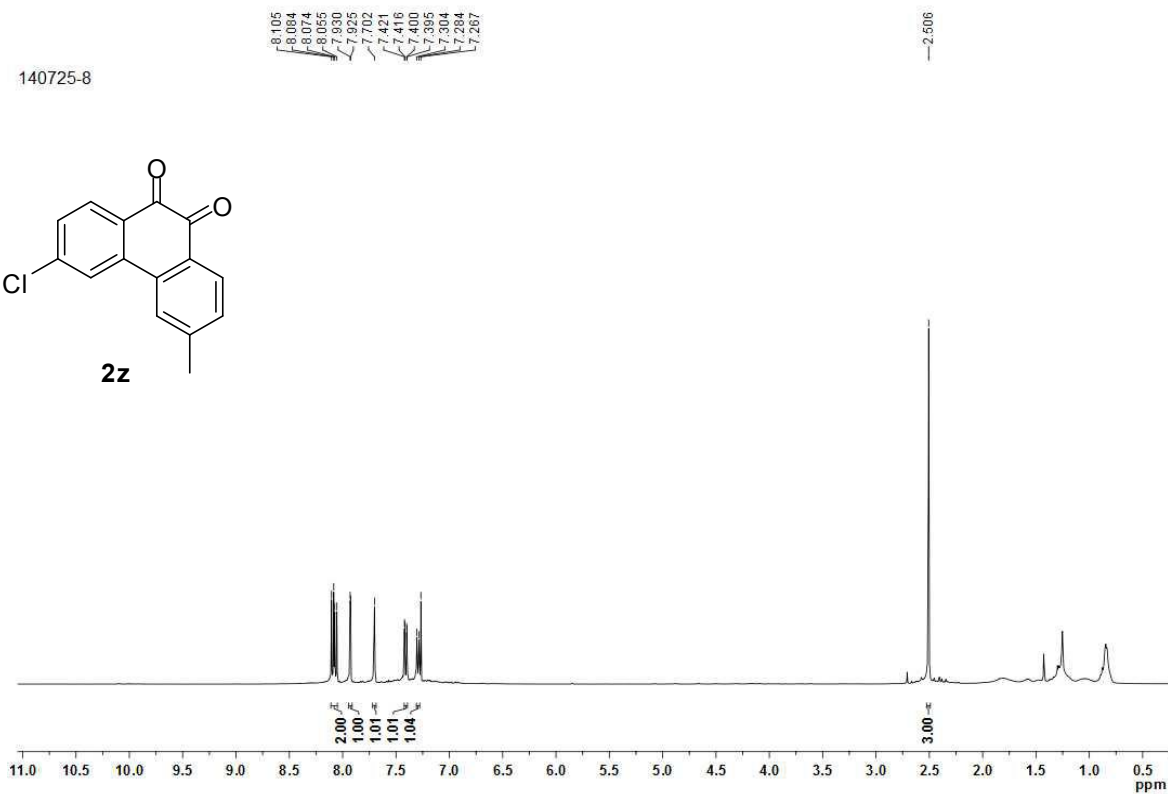
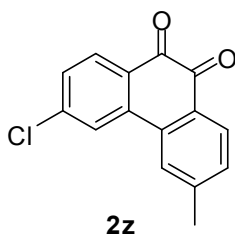




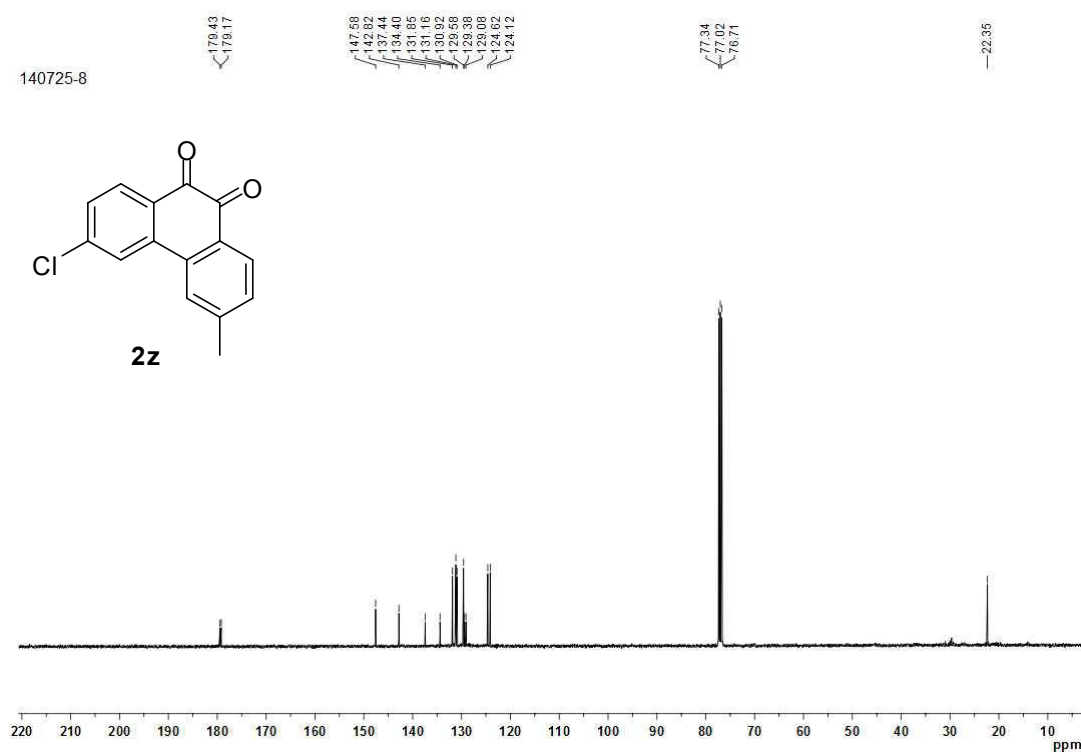
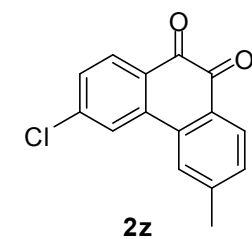
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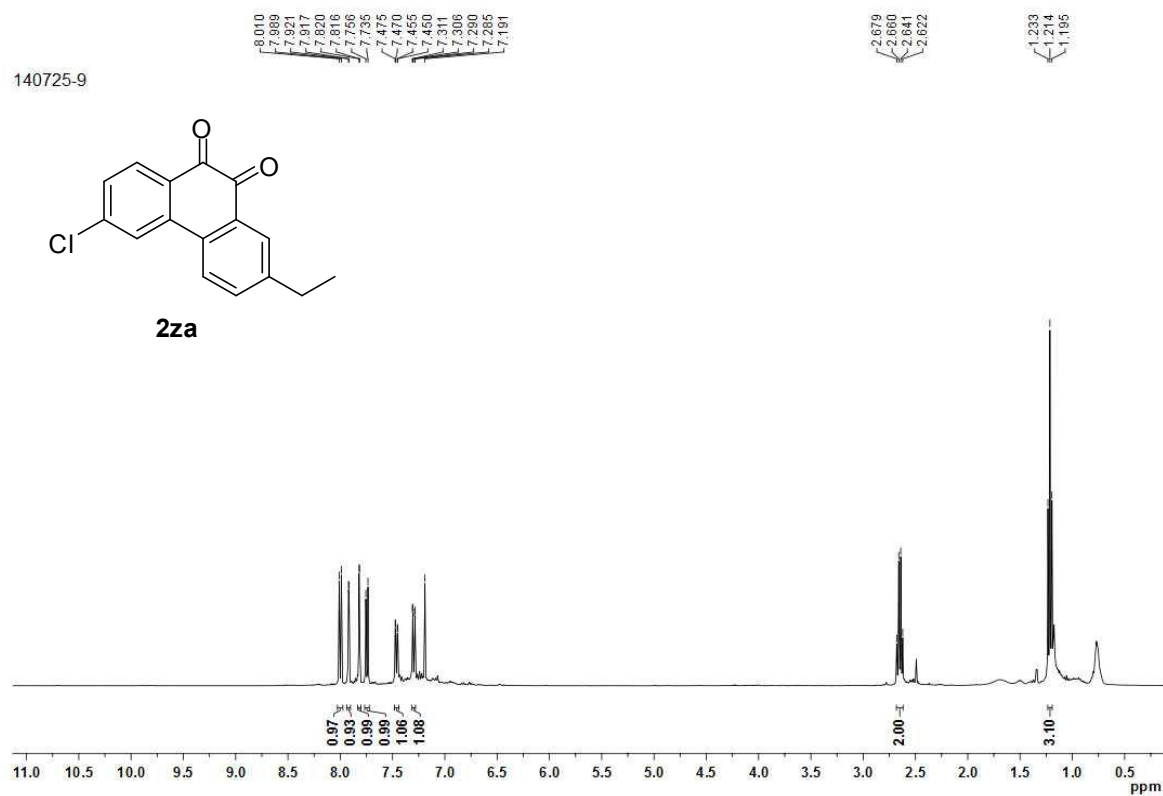
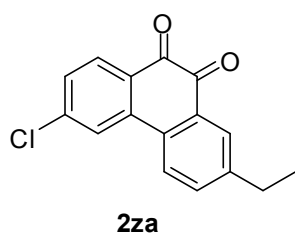
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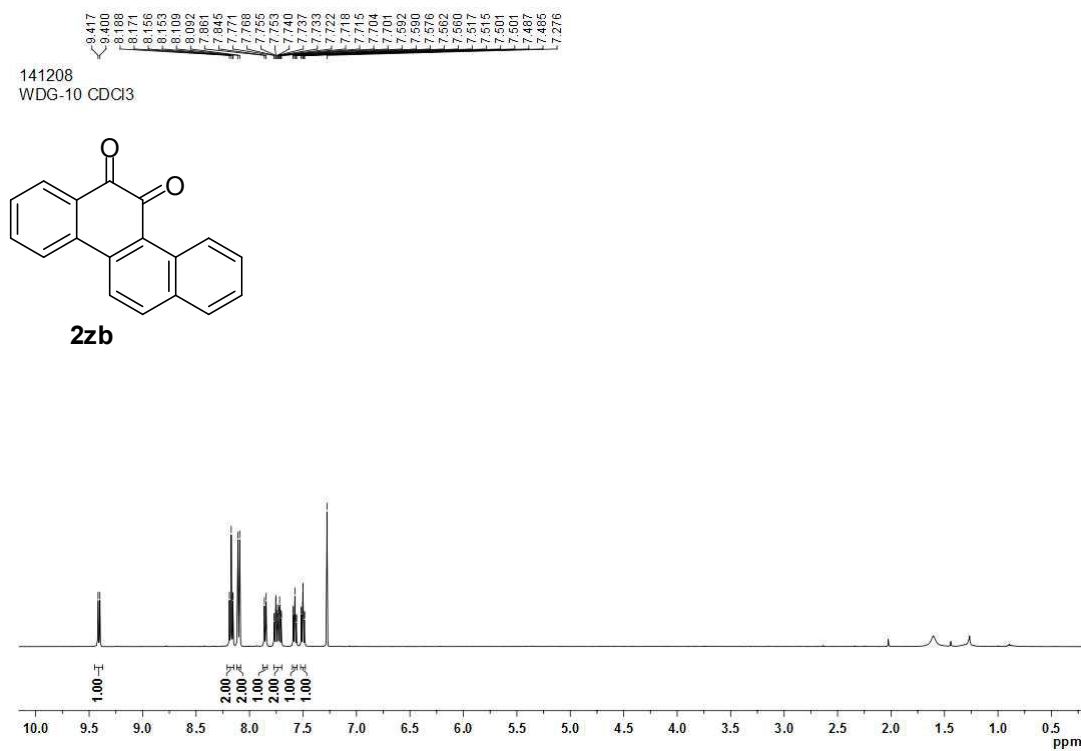
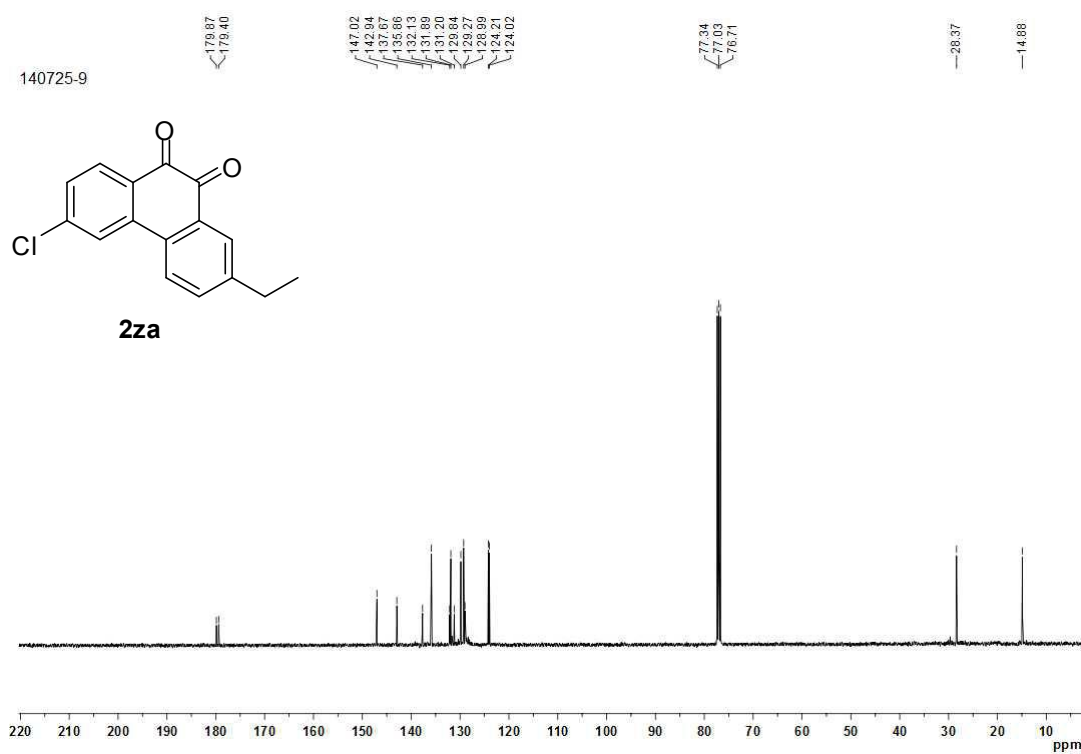


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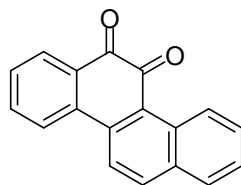


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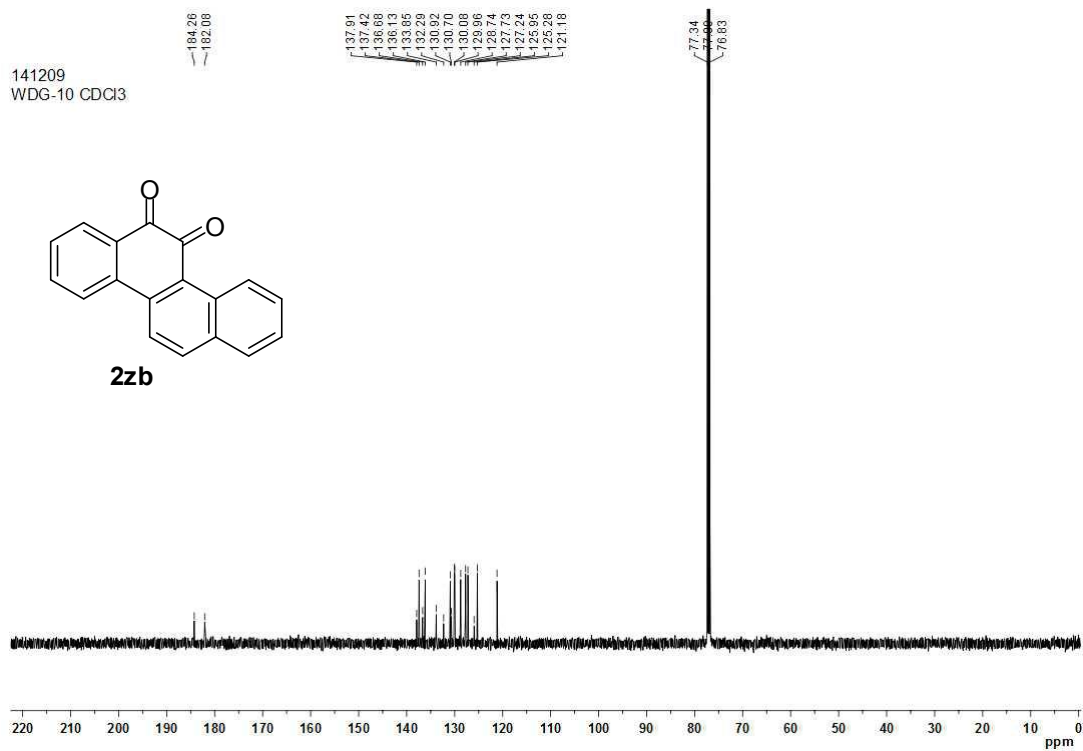




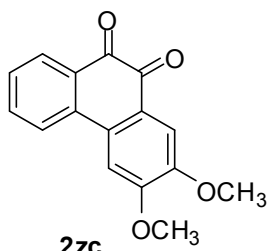
141209
WDG-10 CDCl₃



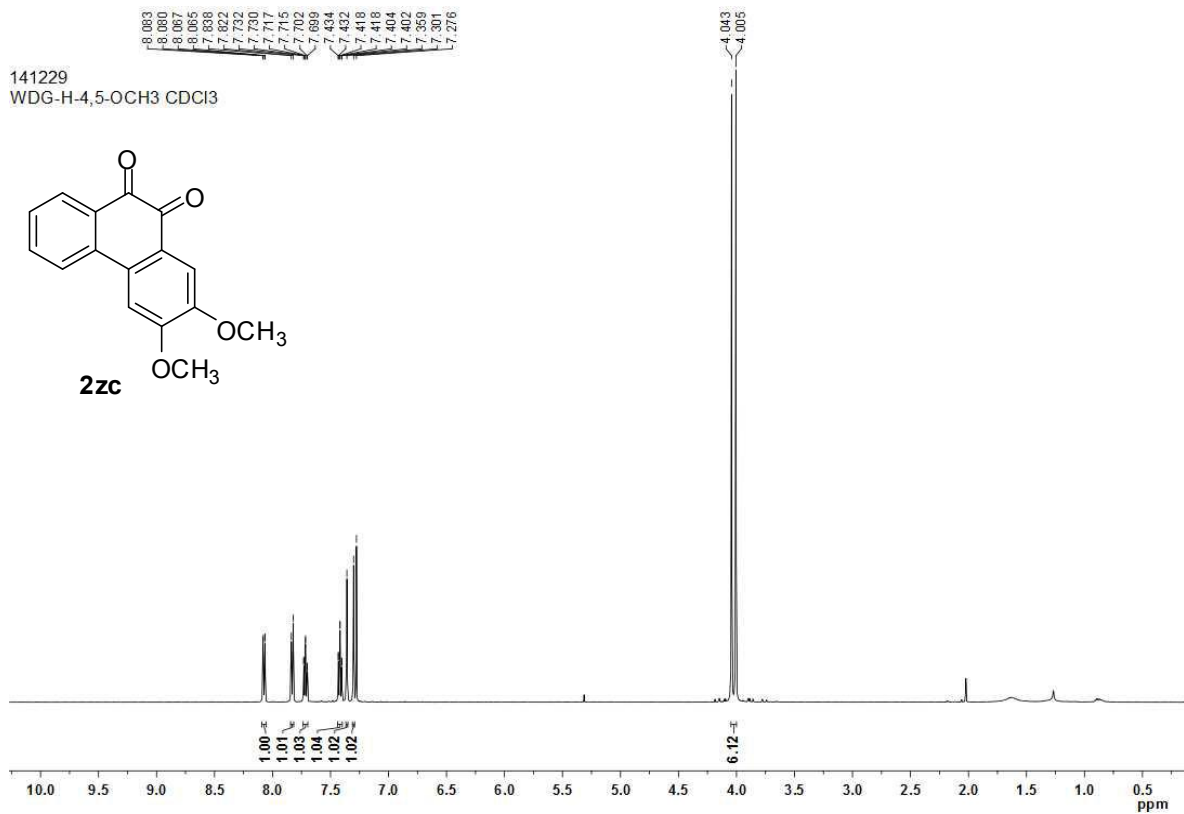
2zb

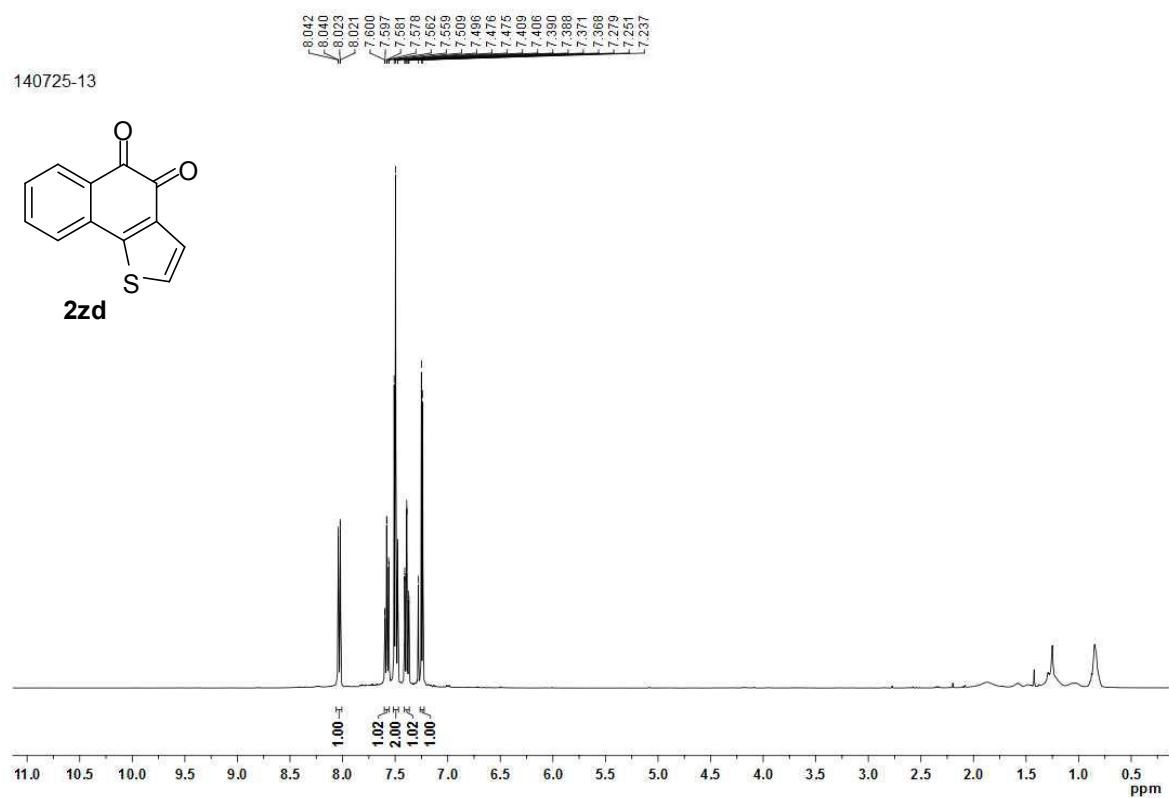
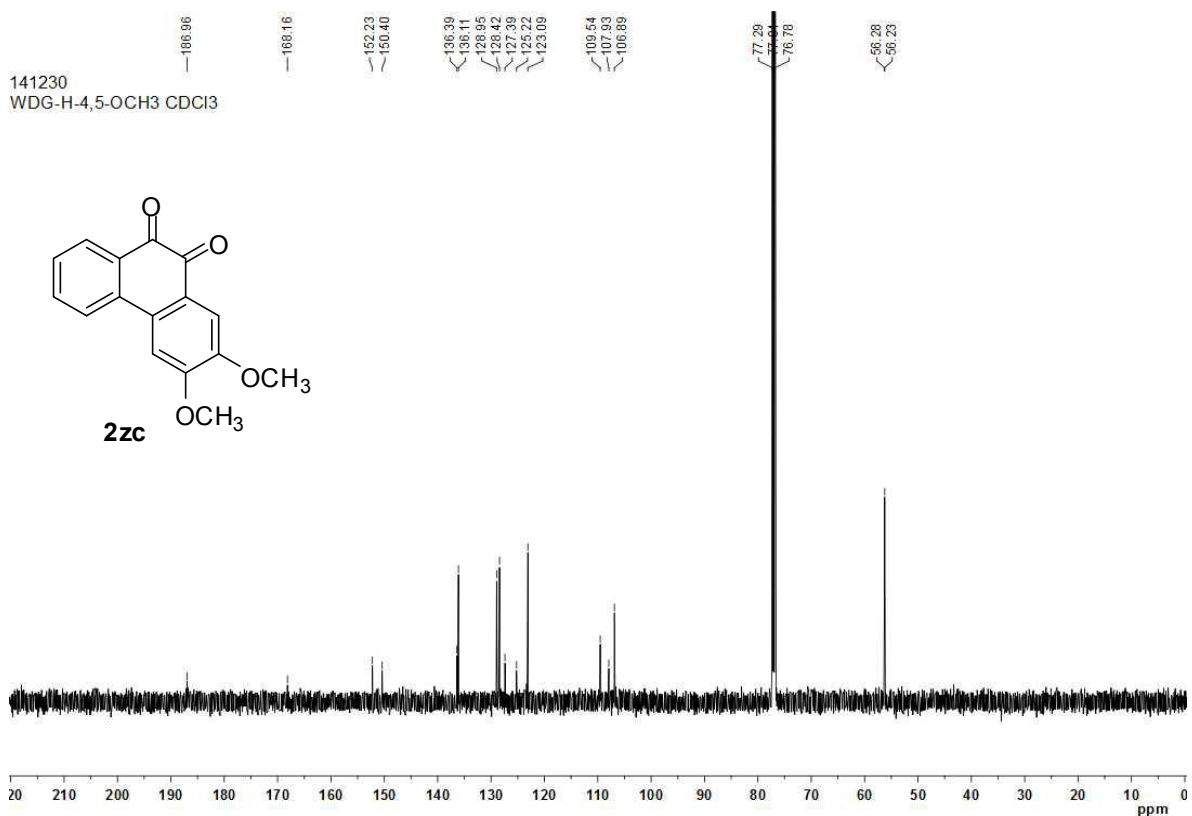


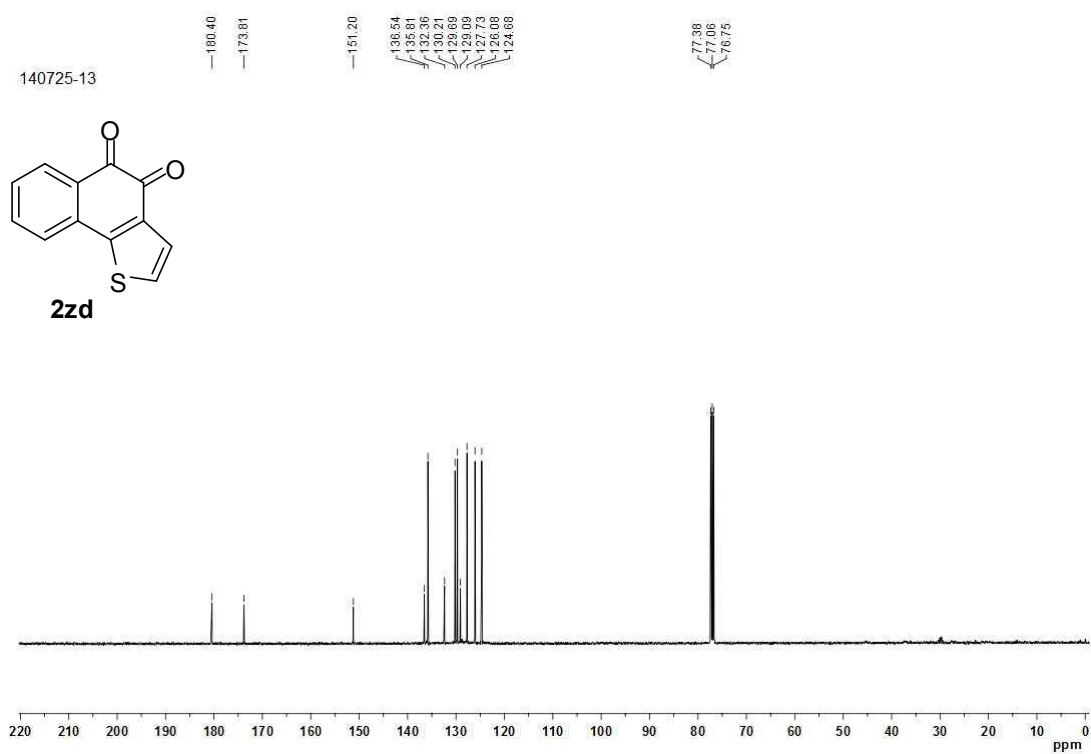
141229
WDG-H-4,5-OCH₃ CDCl₃



2zc







4. ¹H spectrum of 1u-d₅

