Supporting information

Development of Ga salt of molybdophosphoric acid for biomass conversion to levulinic acid

Vijay Bhooshan Kumar, Indra Neel Pulidindi, Rahul Kumar Mishra, and Aharon

Gedanken*

Department of Chemistry and Bar-Ilan Institute for Nanotechnology & Advanced Materials,

Bar Ilan University, Ramat-Gan 52900, Israel

*Corresponding author email: gedanken@mail.biu.ac.il

Fax: 972-3-7384053; Tel: 972-3-5318315

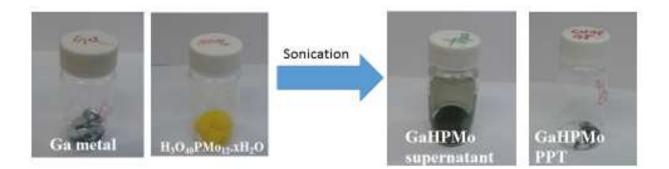


Figure S1. Pictorial representation of Ga metal, HPMo, GaHPMo (supernatant) and Ga@HPMo (precipitate)



Figure S2. Various components of the autoclave (the metal rod on the right is used for opening

and closing the reactor).

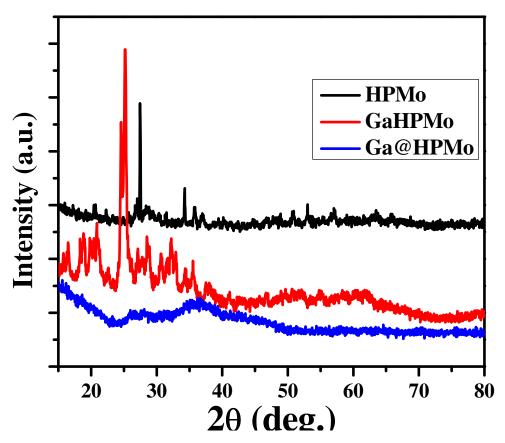


Figure S3. XRD patterns of HPMo, GaHPMo and Ga@HPMo

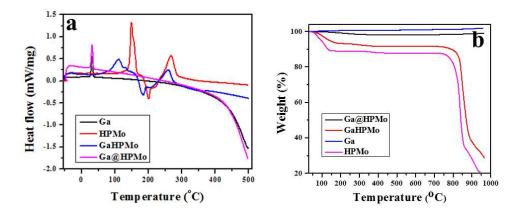


Figure S4. (a) DSC (b) TGA traces of Ga metal, HPMo, GaHPMo, and Ga@HPMo.

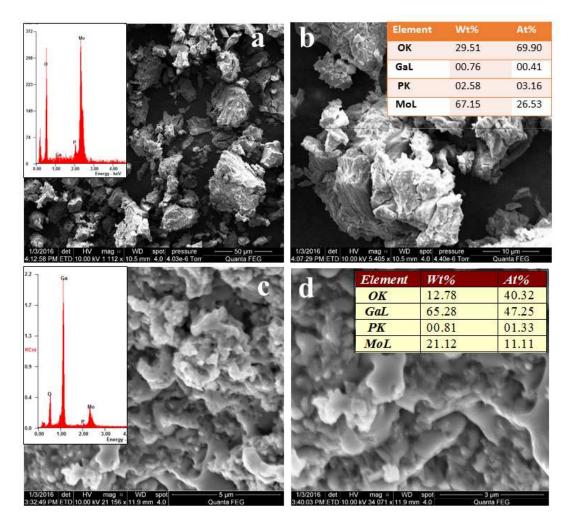


Figure S5. SEM images of (a, b) GaHPMo (inset: EDS, (c, d) Ga@HPMo (inset: EDS).

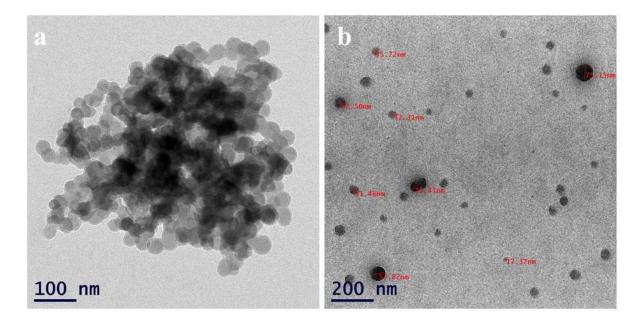


Figure. S6. TEM images of (a) Ga@HPMo (b) GaHPMo.

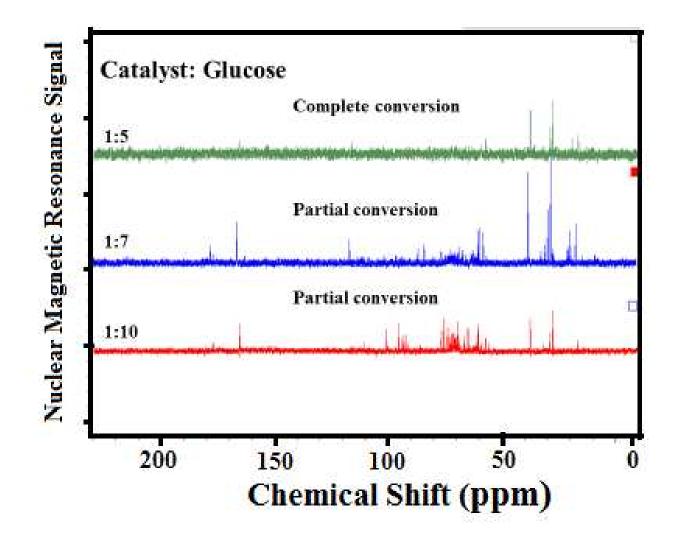


Fig. S7 ¹³C NMR spectra of the reaction product obtained from the hydrothermal reaction (6 h at 150 $^{\circ}$ C) with different ratios of the catalyst (GaHPMo) and glucose.

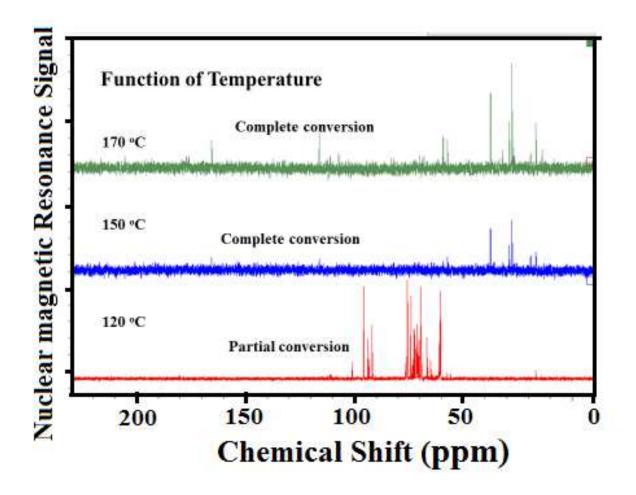


Fig. S8.¹³C NMR spectrum of the product obtained at deferent reaction temperatures.