

Supporting information for

“A solvent-based atomistic theory for doping
colloidal–synthesized quantum dots via cation
exchange”

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Table S1 lists the crystal structure and formation energies of 3d transition metal selenide and CdSe. We take the total energy of bulk metals (E_M^{bulk}) and selenium ($E_{\text{Se}}^{\text{bulk}}$) as the reference for the chemical potential. So, the chemical potential (μ) can be written as $\mu \equiv E^{\text{bulk}} + \mu^*$.

Table S1. The space group, lattice parameters and formation energies of bulk metal selenides. The experimental lattice parameters are listed for comparison.

	Space Group	a (Å)		c (Å)		ΔH_f^{bulk} (eV)
		Expt.	Calc.	Expt.	Calc.	
CdSe ¹	$F\bar{4}3m$	6.08	6.21	--	--	-1.264
VSe ²	$P6_3mc$	3.58	3.75	5.98	5.83	-0.809
CrSe ²	$P6_3mc$	3.68	3.78	6.02	6.13	-0.623
MnSe ²	$Fm\bar{3}m$	5.82	5.40	--	--	-0.664
FeSe ²	$P6_3mc$	3.63	3.84	5.95	5.03	-0.203
CoSe ²	$P6_3mc$	3.63	3.62	5.30	5.21	-0.282
NiSe ²	$P6_3mc$	3.66	3.71	5.35	5.23	-0.498
CuSe ³	$P6_3/mmc$	3.94	3.95	17.25	17.42	-0.168

The formation energies of bulk metal-selenide were calculated as:

$$\Delta H_f^{\text{bulk}}(\text{MSe}) = E^{\text{bulk}}(\text{MSe}) - E_M^{\text{bulk}} - E_{\text{Se}}^{\text{bulk}}.$$

Considering the equilibrium condition $\mu_M + \mu_{\text{Se}} = \mu_{\text{MSe}}$, we can write the formation energy of bulk metal-selenide as:

$$\mu_M^* + \mu_{\text{Se}}^* = \Delta H_f^{\text{bulk}}(\text{MSe}).$$

As discussed in the text, we here consider the Se-rich condition, where μ_{Se}^* reaches its upper limit ($\mu_{\text{Se}}^* = 0$) and $\mu_{\text{Cd}}^* = -1.264$ eV. Similarly, under Se-rich condition, $\mu_{\text{TM}}^* = \Delta H_f^{\text{bulk}}(\text{TMSe})$.

Reference

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