

Supporting Information

Identifying Coordination Geometries of Metal Aquaiions in Water: Application to the case of Lanthanoid and Actinoid Hydrates

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Intermolecular ion-water potential

The monoatomic cation, M(III), (M being La, Lu, Ac or Cf) is described by a positive charge, $Z_M = 4$, and a mobile negative charge density, ρ_M with a total charge, $q_M = -1$, joined to the nucleus by a spring of force constant, k_M (see Figure S1). The intra-atomic energy is defined by:

$$U_{intra} = \frac{1}{2}k_M \cdot r^2 \quad (1)$$

where r is the distance between the nucleus and its associated mobile charge density. In the absence of an external field, the equilibrium position of the oscillator is located on the nucleus and $U_{intra} = 0$.

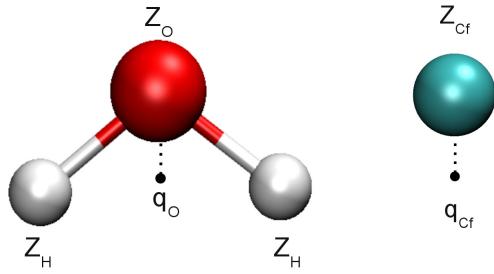


Figure S1: Schematic representation of MCDHO model for the water molecule and the metal cation.

The incorporation of the MCDHO model to the new potential needs the consideration of the following intermolecular terms for the M(III)-H₂O interaction:

- Classical interaction between the mobile charge densities, q_O and q_M , given by a two-exponential function:

$$U_{inter}(q_O, q_M) = A_{MO} \cdot e^{-\alpha_{MO} \cdot r_{MO}} + B_{MO} \cdot e^{-\beta_{MO} \cdot r_{MO}} \quad (2)$$

where r_{MO} is the distance between the mobile charge densities. A_{MO} , α_{MO} , B_{MO} and β_{MO} are fitting parameters.

- Classical interaction between the M nucleus, Z_M , and those of the water molecule, Z_i ($i \equiv O, H$), is given by a two-exponential function as well:

$$U_{inter}(Z_i, Z_M) = C_{Mi} \cdot e^{-\gamma_{Mi} \cdot R_i} + D_{Mi} \cdot e^{-\delta_{Mi} \cdot R_i} \quad (3)$$

where R_i is the distance between the M nucleus and each i -th nucleus of the water molecule, and C_{Mi} , γ_{Mi} , D_{Mi} and δ_{Mi} are fitting parameters.

- Electrostatic interaction between the water mobile charge density, q_O and the M nucleus, Z_M :

$$U_{inter}(q_O, Z_M) = \frac{q_O Z_M}{r'} \left[1 - \left(\frac{r'}{\lambda'} + 1 \right) e^{-2r'/\lambda'} \right] \quad (4)$$

where r' is the distance between the center of ρ_O and the M nucleus and λ' is the intermolecular screening described in the original MCDHO model[?].

- Electrostatic interaction between the M mobile charge density, q_M , and each of the charges on the water molecule nuclei, $Z_i(i \equiv O, H)$:

$$U_{inter}(Z_i, q_M) = \frac{Z_i q_M}{r_i} \left[1 - \left(\frac{r_i}{\lambda'_M} + 1 \right) e^{-2r_i/\lambda'_M} \right] \quad (5)$$

where r_i is the distance from the ρ_M center to Z_i and λ'_M is the corresponding intermolecular screening.

Thus, the interaction energy for a cluster with N water molecules is computed by the expression:

$$U = \sum_{S=1}^N \left(\sum_{i \in S} \sum_{j \in T} [U_{inter}(Z_i, Z_j) + U_{inter}(q_i, q_j) + U_{inter}(q_i, Z_j) + U_{inter}(q_j, Z_i)] + \sum_{i \in S} \frac{1}{2} k_i \cdot r_{ii}^2 + \frac{1}{2} k_M \cdot r^2 \right) \quad (6)$$

where S runs over the water molecules and T over the M.

Table S1: Fitted parameters of the potentials (in a.u.).

	La	Lu	Ac	Cf
k_M	1.01107	1.12522	1.712844	0.960727
λ'_M	0.75010	0.54790	0.292704	0.635312
A_{MO}	60.04400	253.49280	328.983209	568.073911
α_{MO}	1.32059	1.78930	1.951373	2.214706
B_{MO}	-10.36074	-56.92027	0.182997	-0.138243
β_{MO}	0.99410	1.20578	0.402498	0.475451
C_{MH}	41.50598	103.56302	12.181625	11.141950
γ_{MH}	2.65323	3.79588	0.565244	0.933736
D_{MH}	0.00214	0.00132	-11.812755	-13.336449
δ_{MH}	1.83328	2.29914	0.555489	0.999286

Details of MD simulations

- 1 cation + 1000 water molecules.
- Cubic box with length chosen to get experimental water density = 0.997 g cm⁻³ at simulation conditions.
- NVT ensemble with T=300K
- PBC and Ewald sum.
- MD simulations have been carried out using the Dynamical shell model (P.J.Mitchell and D. Fincham, *J. Phys. Condens. Matter* **1993**, 5, 1031-1038) to account for the polarizable MCDHO model.
- A modified version of the DL-POLY Classic code (W. Smith, T.R. Forester and I.T. Todorov, DL-POLY v. 2.19; STFC Daresbury Laboratory, Daresbury, Warrington WA4 4AD, Cheshire, UK 2012).
- 1ns production time.

Potential Energy Surface Building

The interaction energies to be fitted are obtained from ab initio calculations obtained at the level indicated in the following table.

Table S2: Quantum mechanical methods^a

M(III)	DFT	Pseudopotential ^b
La	M06	ECP46MBW
Lu	M06	ECP60MBW
Ac	BP86	ECP78MBW
Cf	BP86	ECP87MBW

(a) Basis sets from:

M. Dolg et al. Theor.Chem.Acc. **1989**, 75, 173-194; **1993**, 85, 441-450

J. Yang et al. Theor.Chem.Acc. **2005**, 113, 212-224.

(b) Pseudopotentials taken from <http://www.theochem.uni-stuttgart.de>

It is worth pointing out that the quasi-relativistic effective core potential (RECP) were specifically developed by Dolg and col.[6] for the lanthanides and actinides trivalent cations. Their recommended basis sets for the valence orbitals were used and aug-cc-pvdz basis sets for the O and H atoms.

Table S3: M–O Distances (Å) and hydration numbers corresponding to the first and second hydration shells, and hydration energies.

Cation	$R_{\text{M-O}_\text{I}}$				N_I			$R_{\text{M-O}_\text{II}}$			N_II			ΔH_{hydr}		
	This work	Theor.	Exptal.	This work	Theor.	Exptal.	This work	Theor.	Exptal.	This work	Theor.	Exptal.	This work	Theor.	Exptal.	
La	2.62	2.65 ³	2.58 ¹	9.1	9.5 ³	9.1 ^{1,8}	4.72	5.0 ³	4.63 ⁶	20	23 ³	18 ⁶	-745 ±60	-792 ⁹	-792 ¹¹	
		2.56 ⁴	2.55-2.64 ²		8.9 ⁴	9 ²		4.68 ⁴			15 ⁴					
		2.52 ⁷	2.54 ⁵		9.0 ⁷	9.2 ⁵		4.65 ⁷			19 ⁷					
		2.52-2.65 ⁶				9 ⁶										
		2.60 ⁸														
Lu	2.37	2.32 ⁷	2.35 ⁸	8	8 ⁷	8 ⁸	4.59	4.50 ⁷		18	18 ⁷		-838 ±60	-896 ⁹	-883 ¹¹	
Ac	2.65			8.7			4.69			22						
Cf	2.47	2.37-2.49 ¹³	2.42-2.55 ¹²	7.8		9 ^{12,16}	4.69			19						
			2.42 ¹⁴			8.5 ¹⁴										

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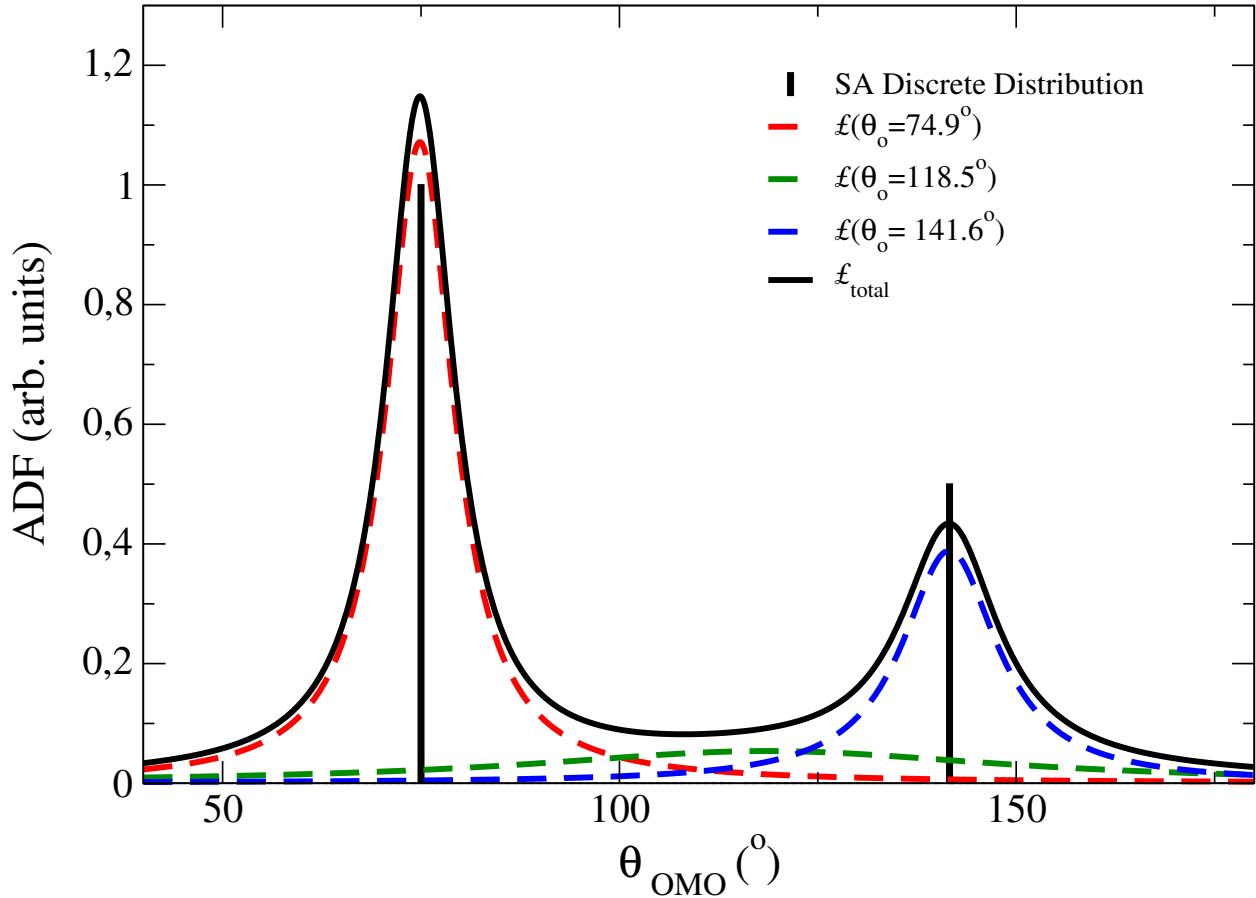


Figure S2: Linear combination of Lorentzian functions centered on the OMO angles of a SA polyhedron.

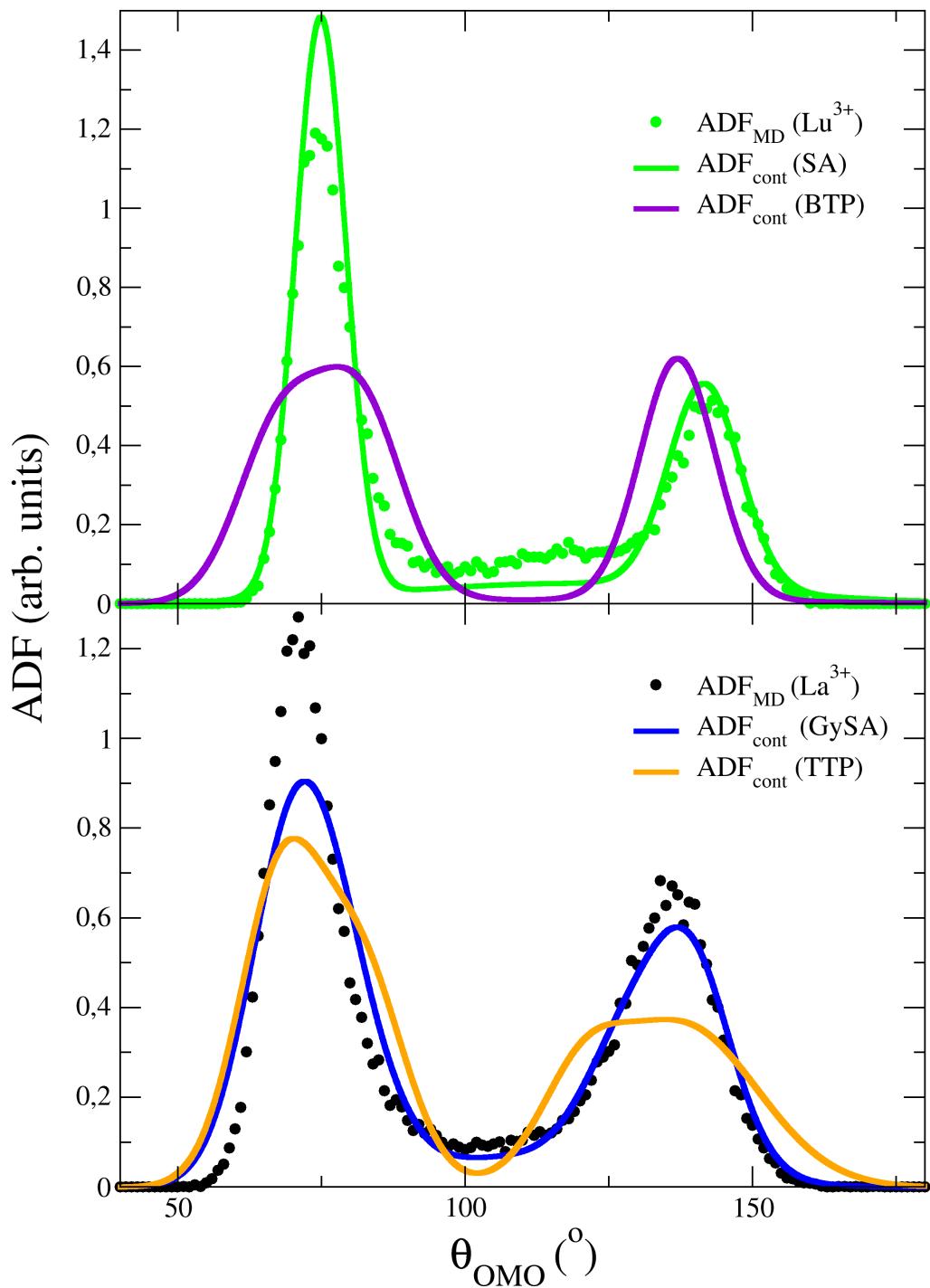


Figure S3: Top: Comparison of the simulation ADFs for Lu^{3+} (green dots) and the ADF_{cont} built from Gaussians for the SA (green line) and BTP (magenta line) 8-coordination polyhedra Bottom: Comparison of the simulation ADFs for La^{3+} (black dots) and the ADF_{cont} s built from Gaussians for the GySA (blue line) and TTP (orange line) 9-coordination polyhedra

Table S4: Parameters to build the Lorentzian functions associated to the different angles of the polyhedron representative of the octa-fold and ennea-fold coordination.

Data from reference polyhedra							
SA		Cube		TPP		GySA	
θ_o^{ref}	Degeneration	θ_o^{ref}	Degeneration	θ_o^{ref}	Degeneration	θ_o^{ref}	Degeneration
74.86	16	70.53	12	67.79	12	66.8	4
118.53	4	109.47	12	81.79	9	68.5	4
141.59	8	180.0	8	120.00	3	73.8	8
				135.59	6	81.1	4
				139.11	6	105.4	2
						127.3	4
						133.7	2
						139.1	8

Data taken from QM calculations and MD simulations used to derive Eq. (3)							
Cf(SA)				Lu(SA)			
θ_o^{QM}	\bar{k}	θ_o^{ADF}	$\Delta\theta_{FWHH}$	θ_o^{QM}	\bar{k}	θ_o^{ADF}	$\Delta\theta_{FWHH}$
77.3	0.0356	74.0	12.5	77.6	0.0592	74.0	11.7
115.4	0.0070	—	—	115.1	0.0084	—	—
142.3	0.0414	143.0	16.2	142.3	0.0419	143.0	15.4

La/Ac(TPP)			
θ_o^{QM}	\bar{k}	θ_o^{ADF}	$\Delta\theta_{FWHH}$
69.7	0.0363	{70.6	15.3
80.0	0.0344	—	16.1
120.0	0.0367	—	15.1
134.1	0.0191	{136.2	29.0
139.3	0.0207	—	26.8

Lorentzian parameters employed to construct Figs. 3 and 4 using Eqs. 1 and 3.						
SA						
Cf				Lu		
θ_0	74.86	118.53	141.59	74.86	118.53	141.59
A	16	4	8	16	4	8
Γ	17.2	87.6	14.8	10.4	73.0	14.6

BTP						
Cf					Cube	
θ_0	67.79	81.79	120.	135.59	139.11	70.53
A	8	9	1	6	4	12
Γ	17.2	17.2	87.6	14.8	14.8	17.2

TPP				
La/Ac				
θ_0	67.79	81.79	120.00	135.59
A	12	9	3	6
Γ	16.7	16.9	17.8	32.1

GySA								
La/Ac					GySA			
θ_0	66.8	68.5	73.8	81.1	105.4	127.3	133.7	139.1
A	4	4	8	4	2	4	2	8
Γ	16.9	16.9	16.9	17.9	32.1	16.7	32.1	16.7

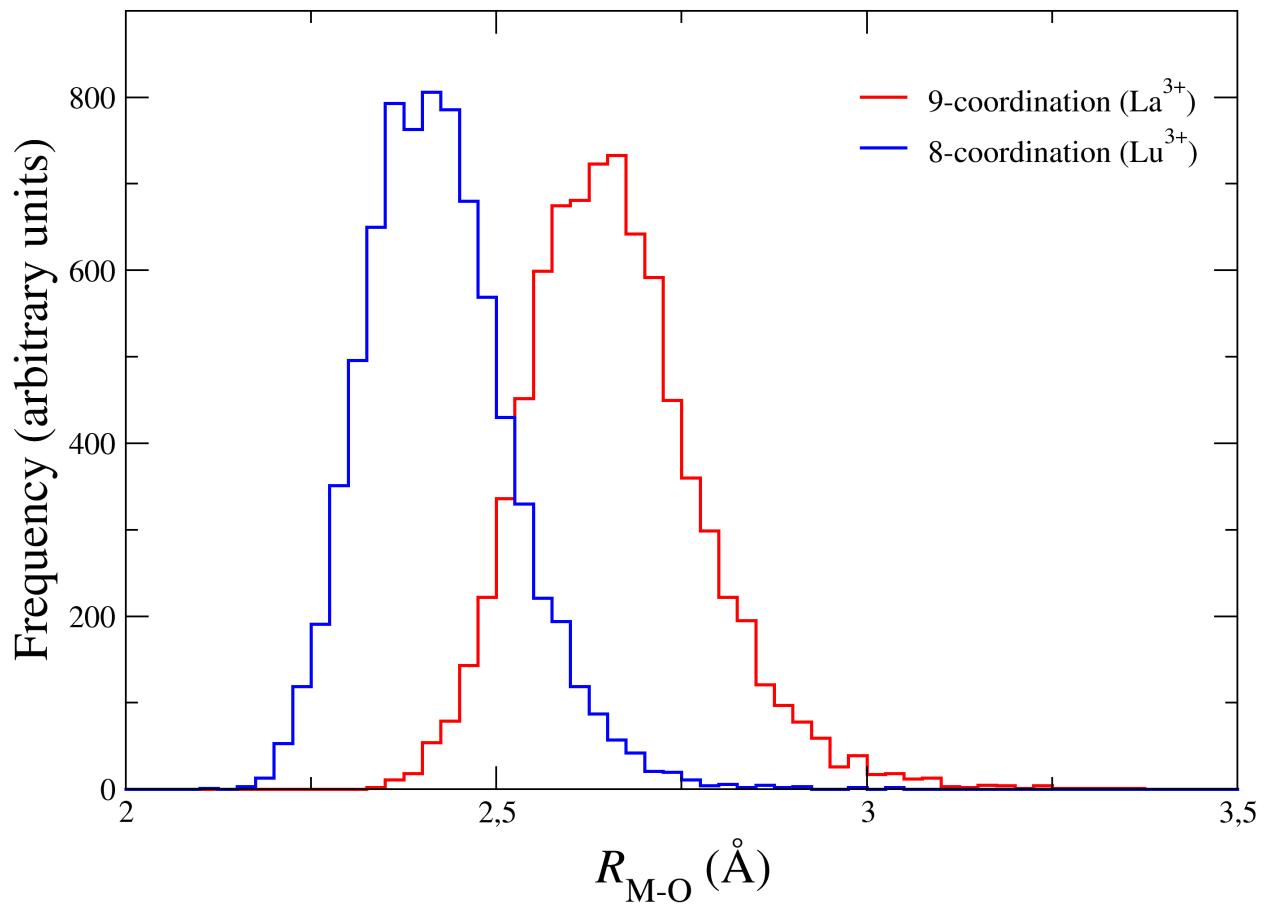


Figure S4: Histogram of M-O distances for the 8-coordinated (Lu^{3+}) and 9-coordinated (La^{3+}).