# CO<sub>2</sub>-Mediated Formation of Chiral Carbamates from *meso*-Epoxides *via*

### **Polycarbonate Intermediates**

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### **Supporting Information**

## **Table of Content**

1. Determination of enantiomeric purity of carbamate derivatives	2
2. NMR spectrum of various carbamate derivatives	3
3. HPLC spectrum of various carbamate derivatives	17
4. Mechanistic understanding of ammonolysis reaction	35
5. NMR spectrum of ( <i>S</i> , <i>S</i> )-carbamate derivative for X-Ray analysis	42
6. Crystal structure determination	.43

#### 1. Determination of enantiomeric purity of carbamate derivatives

The enantiomeric excess of *trans*-carbamate derivatives **1a-1d**, **1g**, **1h**, **1i**, **2a-2c**, **3a**, **3c**, **4a**, **4b**, **5a**, **6a** and **7a** were determined by HPLC using a chiral column, the details were listed in Table S1.

Entry	substrate	Chiral column	Mobile phase ( <sup>n</sup> Hexane/ <sup>i</sup> PrOH)	λ (nm)	Retention time (min)
1	1a	Chiralpak AD	95/5	210	12.2/13.6
2	1b	Chiralpak AD	95/5	210	10.4/11.3
3	1c	Chiralpak AD	90/10	210	9.3/11.5
4	1d	Chiralpak AD	95/5	210	19.6/48.1
5	1g	Chiralpak AD	90/10	254	9.2/10.4
6	1h	Chiralpak AD	90/10	254	8.4/9.2
7	1i	Chiralpak AS-H	90/10	214	5.9/10.1
8	2a	Chiralpak AD	90/10	210	9.7/15.5
9	2b	Chiralcel AD	90/10	210	8.8/12.1
10	2c	Chiralpak AD	90/10	210	7.4/12.9
11	<b>3</b> a	Chiralpak AS-H	95/5	210	8.6/10.2
12	3c	Chiralpak AD	Chiralpak AD 90/10		
13	<b>4</b> a	Chiralpak AD	90/10	210	7.3/14.2
14	<b>4</b> b	Chiralpak AD	95/5	210	22.5/30.9
15	5a	Chiralpak AD	90/10	210	8.5/10.2
16	6a	Chiralpak AD	90/10	254	12.1/15.1
17	7a	Chiralpak AD 90/10 210			

 Table S1. HPLC resolution conditions and results of racemic compounds<sup>a</sup>

<sup>a</sup>flow rate is 1.0 mL/min and  $\lambda$  is maximum absorption wavelength.

The determination enantiomeric excess of **1e**, **1f** and **3b** was not determined because of the low ultraviolet response in HPLC analysis

## 2. NMR spectrum of various carbamate derivatives

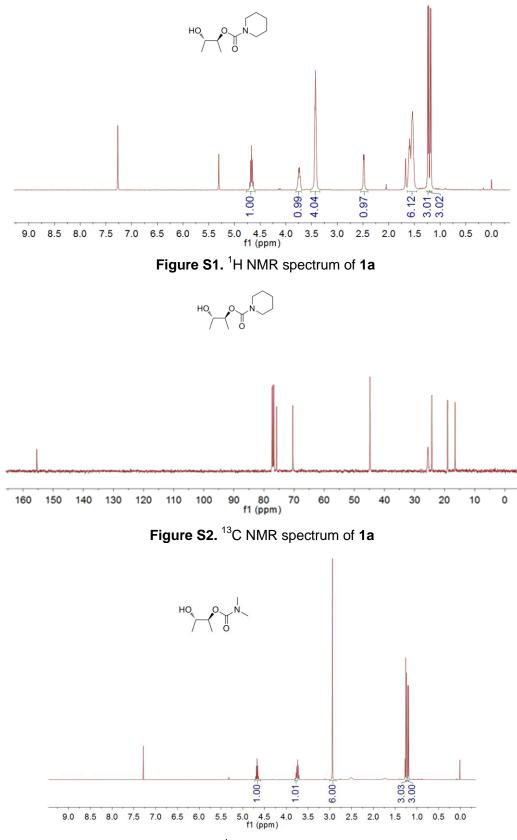


Figure S3. <sup>1</sup>H NMR spectrum of 1b

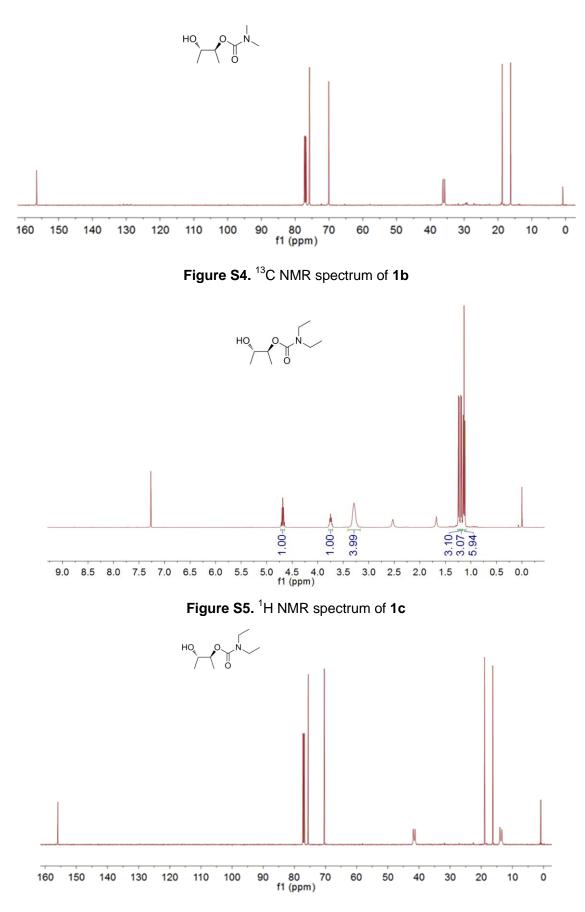
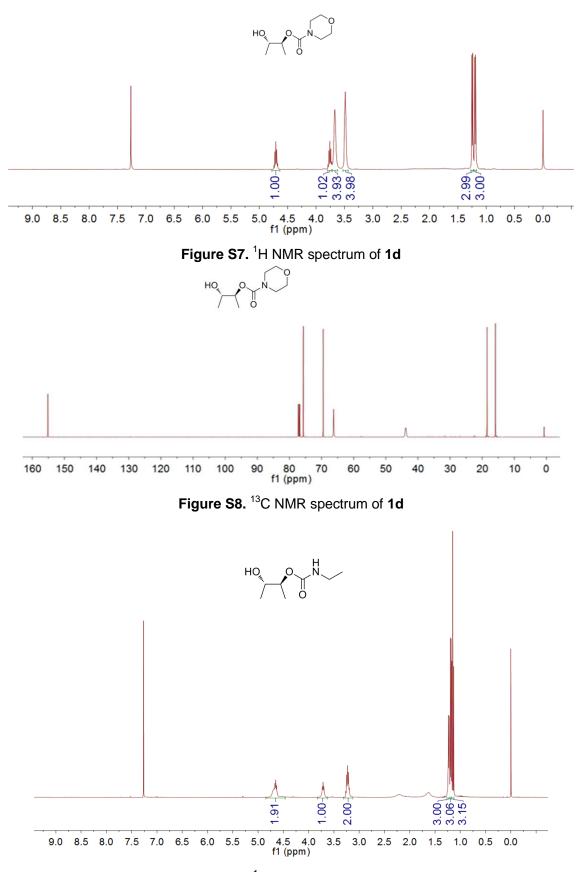


Figure S6. <sup>13</sup>C NMR spectrum of 1c





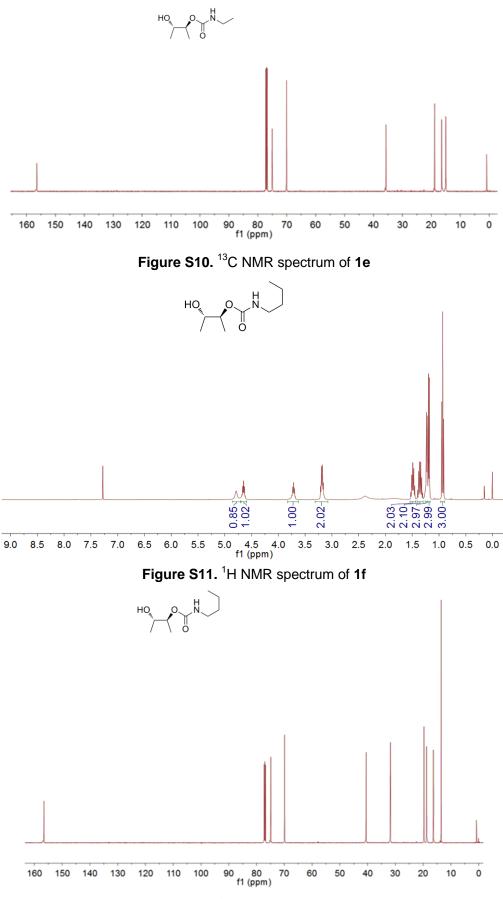
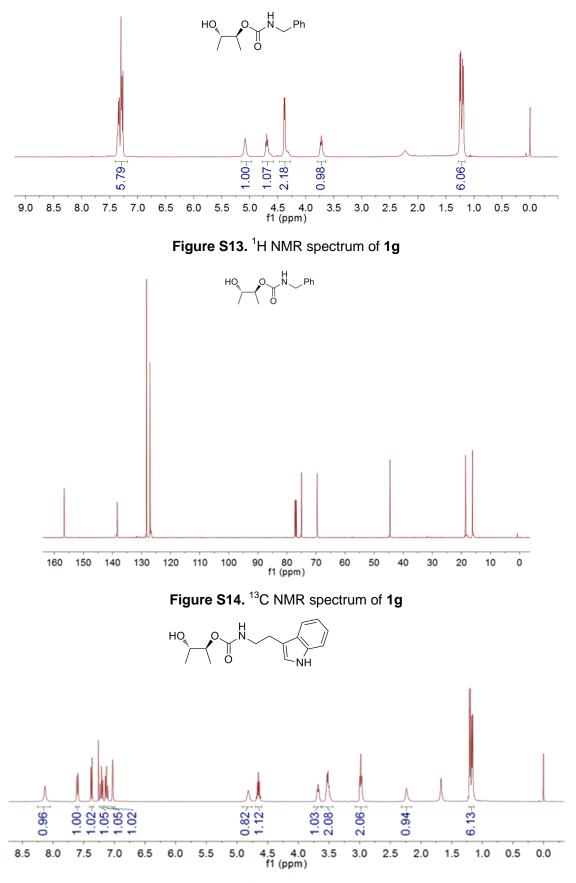
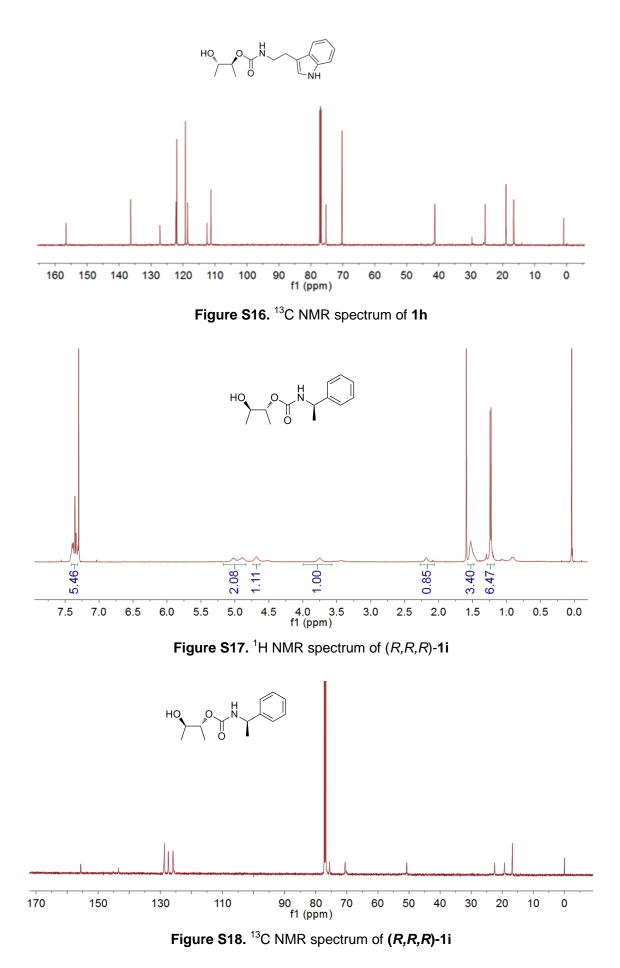


Figure S12. <sup>13</sup>C NMR spectrum of 1f







**S**8

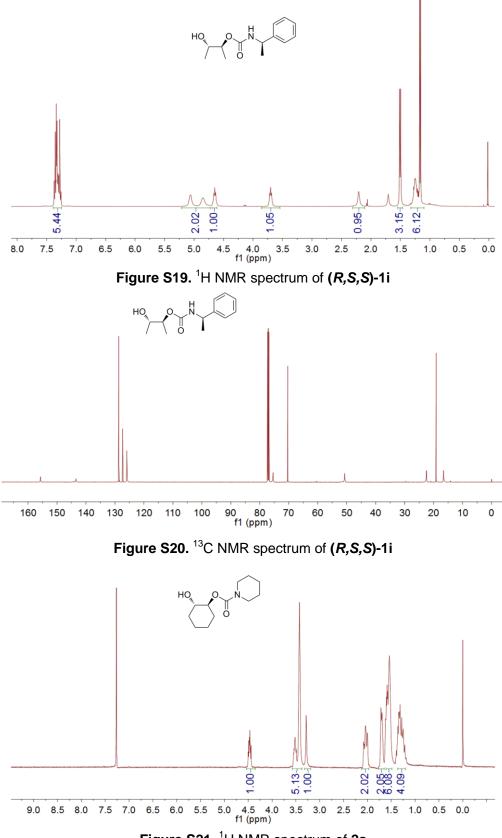


Figure S21. <sup>1</sup>H NMR spectrum of 2a

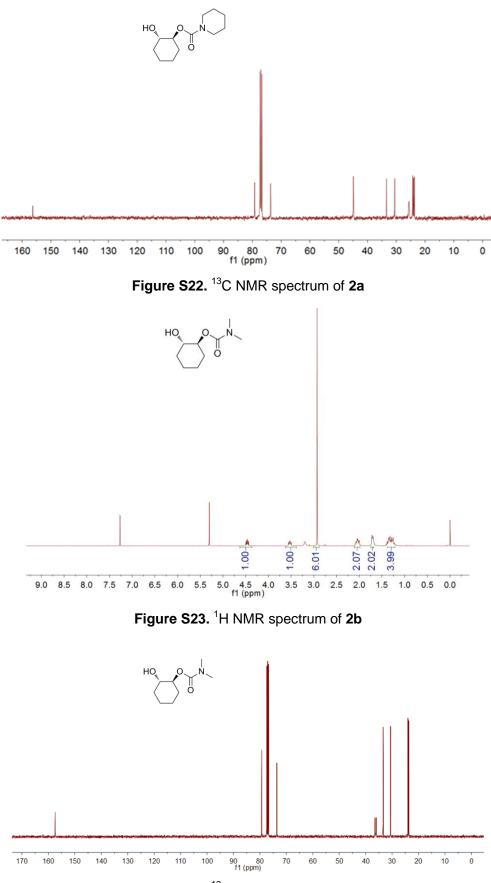


Figure S24. <sup>13</sup>C NMR spectrum of 2b

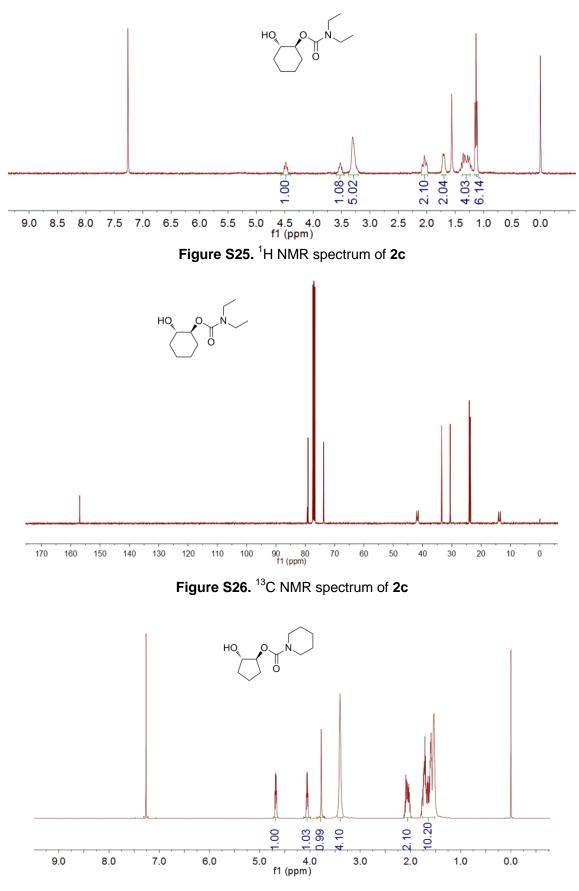
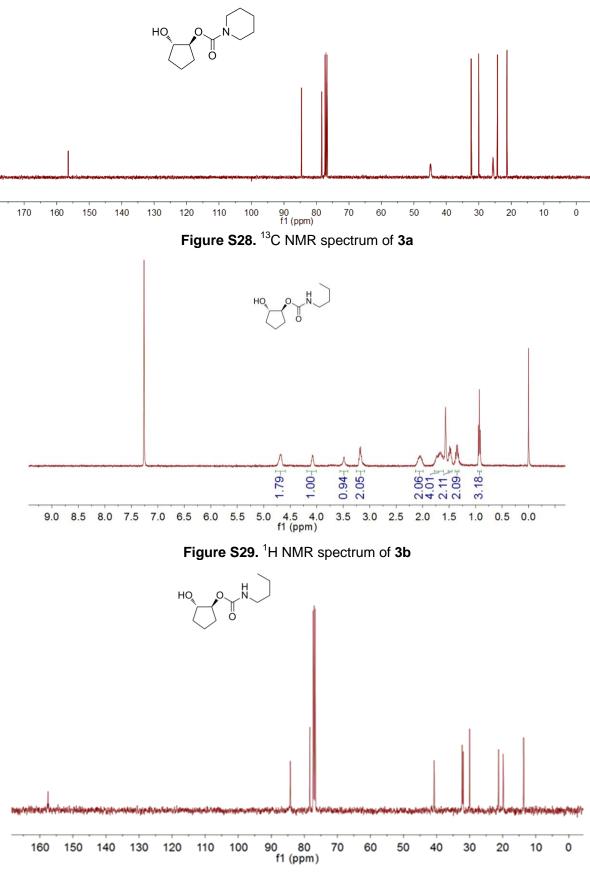


Figure S27. <sup>1</sup>H NMR spectrum of 3a





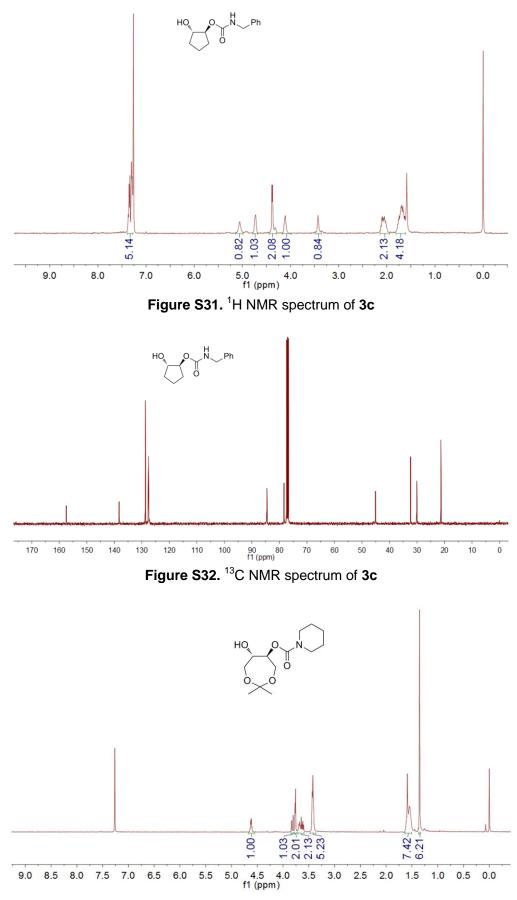


Figure S33. <sup>1</sup>H NMR spectrum of 4a

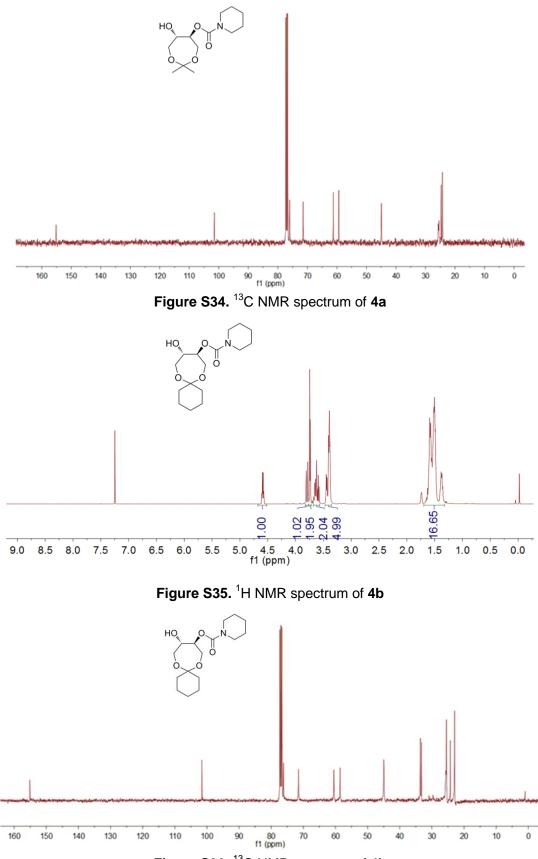


Figure S36. <sup>13</sup>C NMR spectrum of 4b

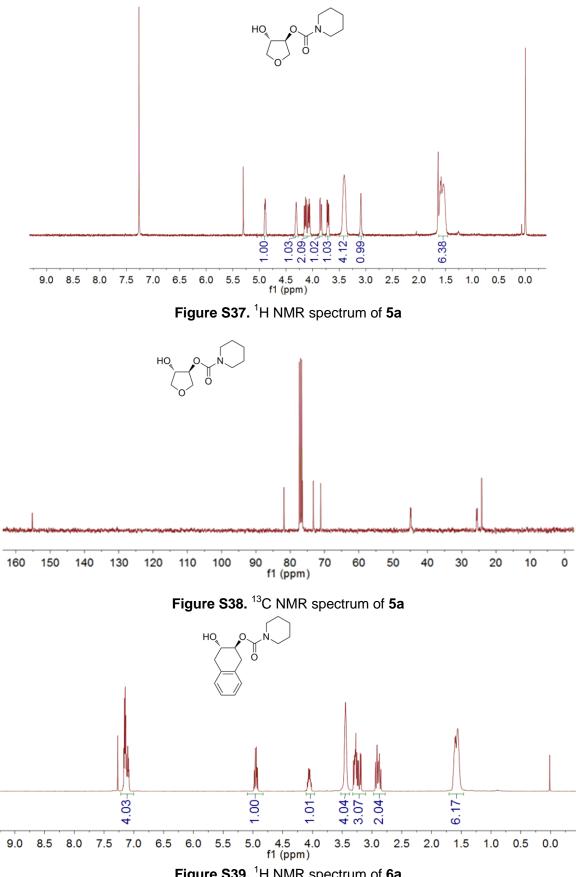


Figure S39. <sup>1</sup>H NMR spectrum of 6a

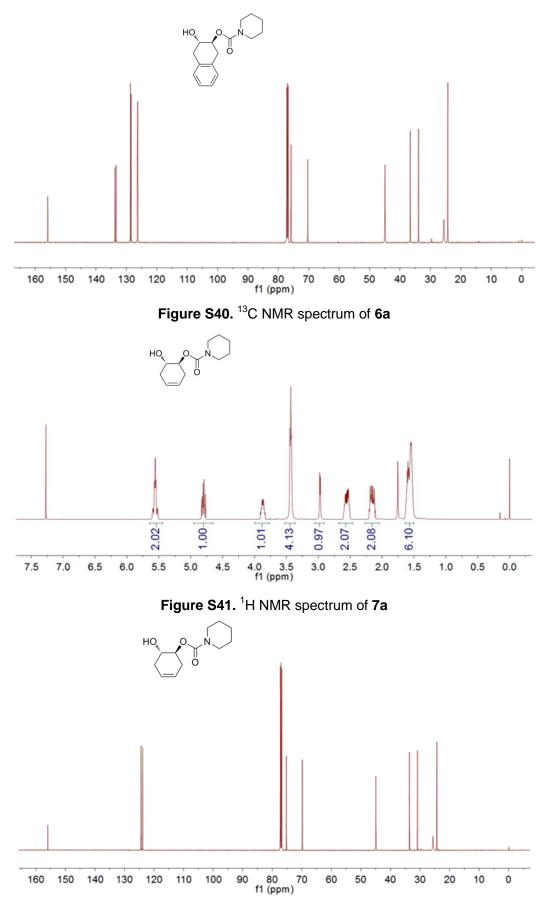


Figure S42. <sup>13</sup>C NMR spectrum of 7a

#### 3. HPLC spectrum of various carbamate derivatives

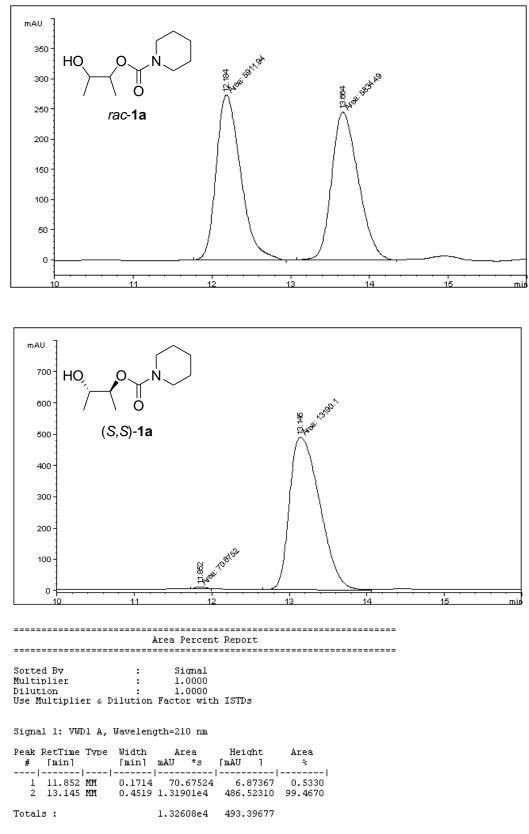


Figure S43. HPLC spectrum of 1a

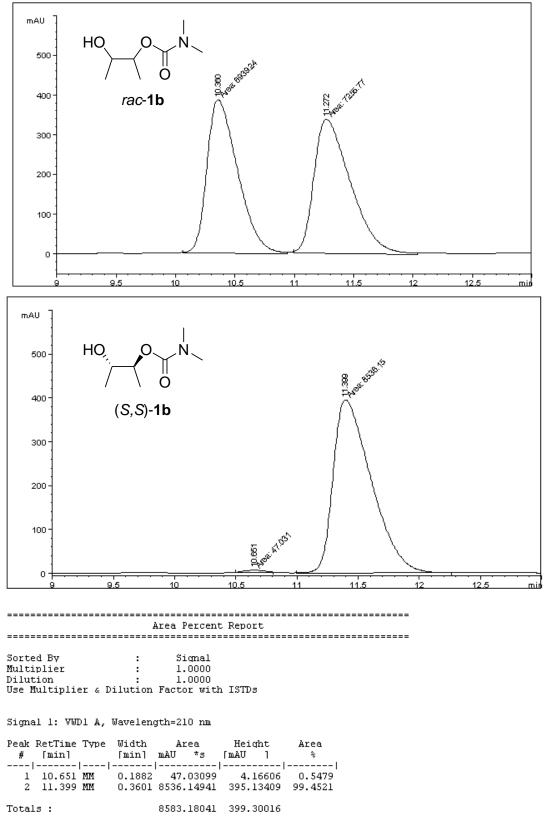


Figure S44. HPLC spectrum of 1b

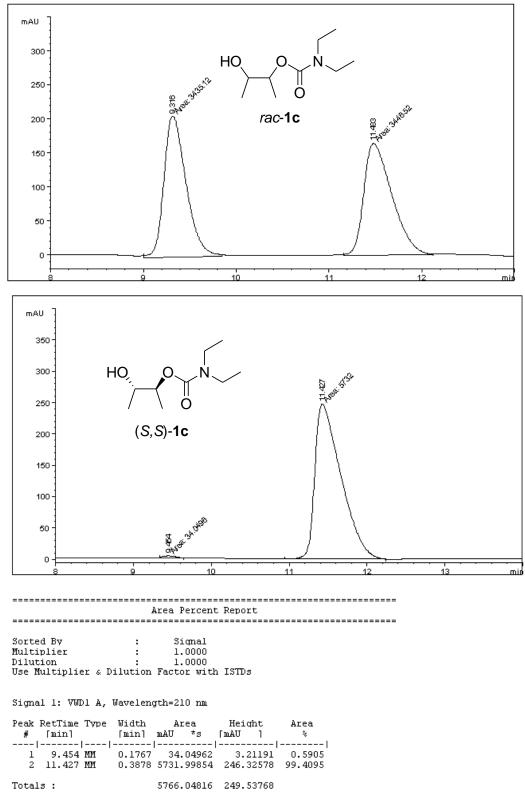


Figure S45. HPLC spectrum of 1c

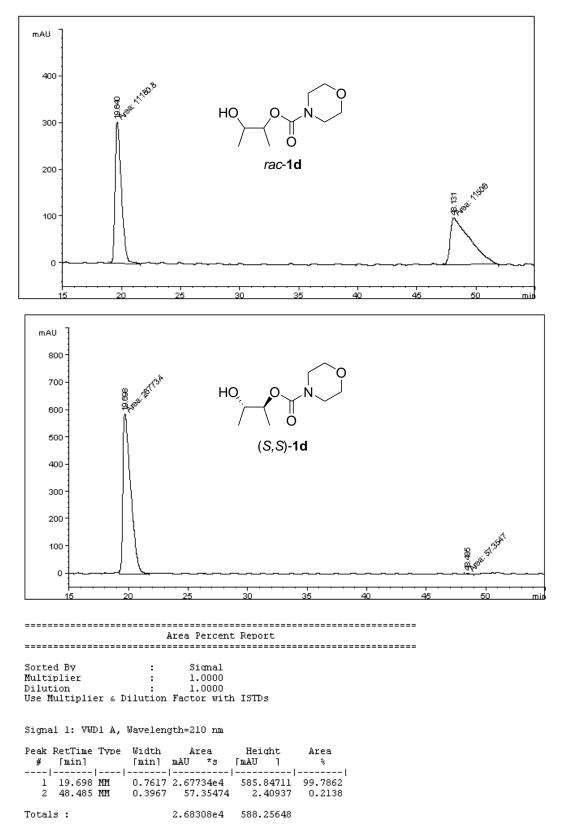


Figure S46. HPLC spectrum of 1d

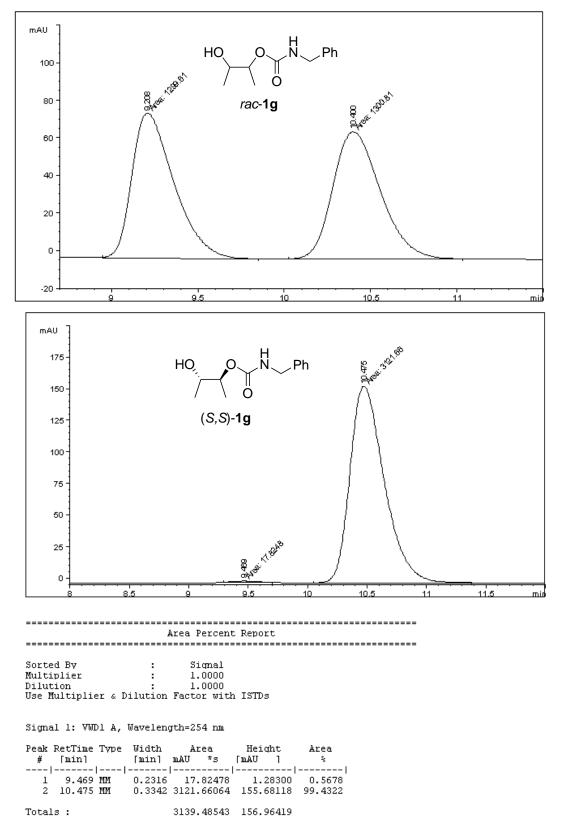


Figure S47. HPLC spectrum of 1g

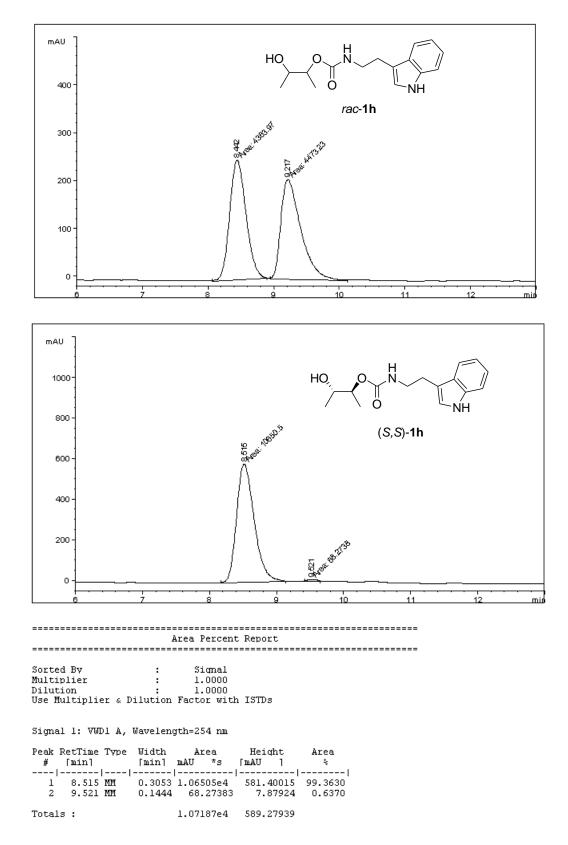


Figure S48. HPLC spectrum of 1h

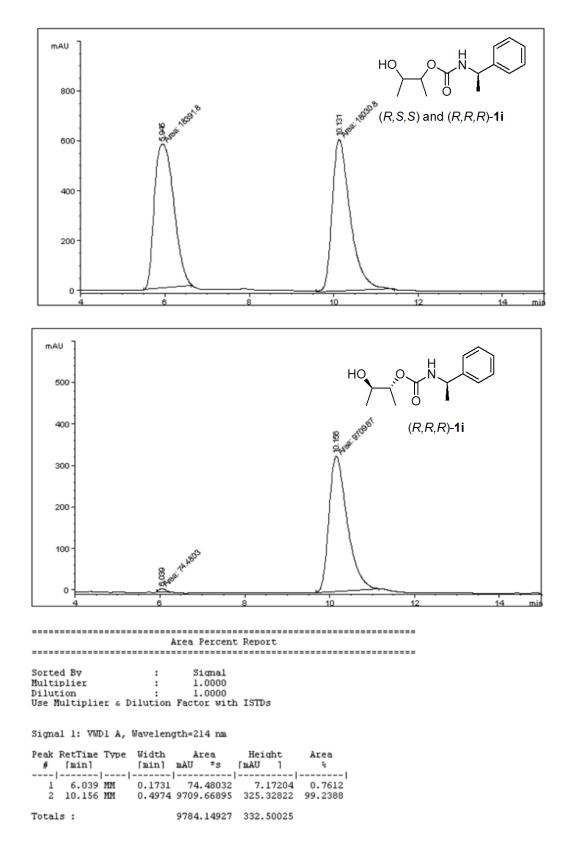


Figure S49. HPLC spectrum of (R,R,R)-1i

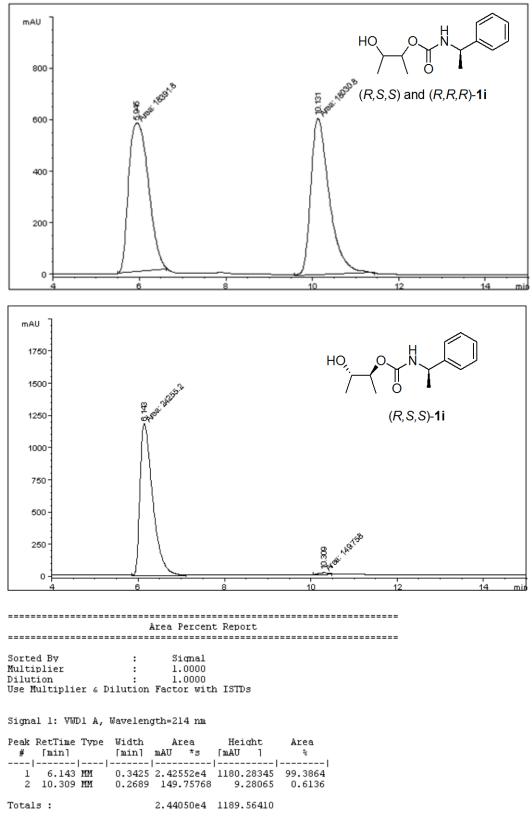


Figure S50. HPLC spectrum of (R,S,S)-1i

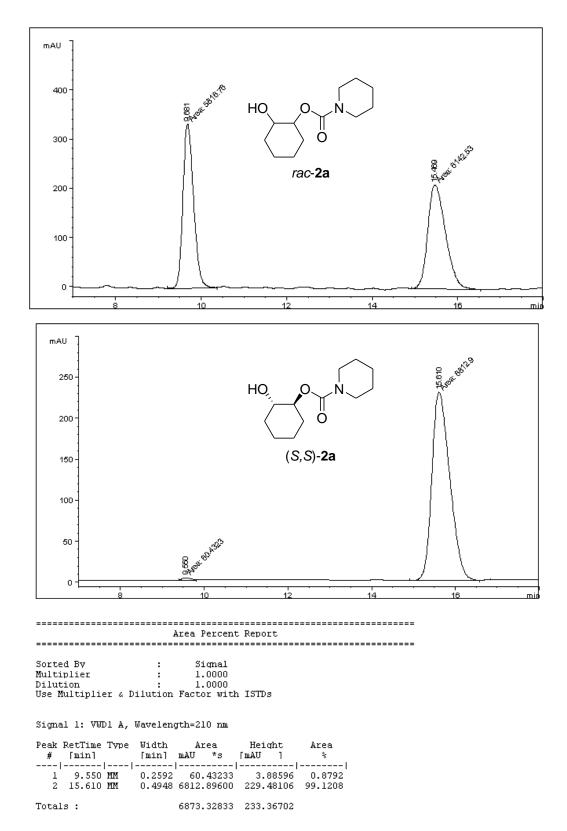
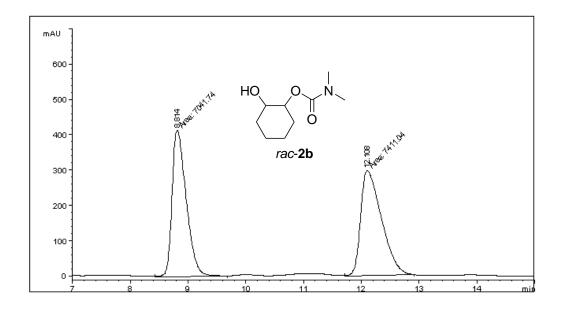
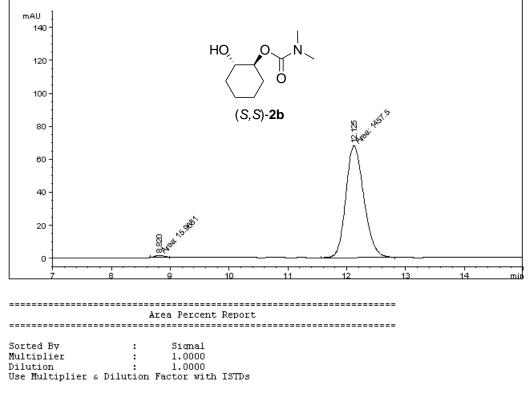


Figure S51. HPLC spectrum of 2a





Signal 1: VWD1 A, Wavelength=210 nm

			Width [min]					
1	8.820	MM	0.1878	15.	96810	1.4	1725	1.0837 98.9163
4	14.145	1111	0.3370	1457.	JU2J2	00.0	4010	30.9103
Total	з:			1473.	47042	69.4	5741	

Figure S52. HPLC spectrum of 2b

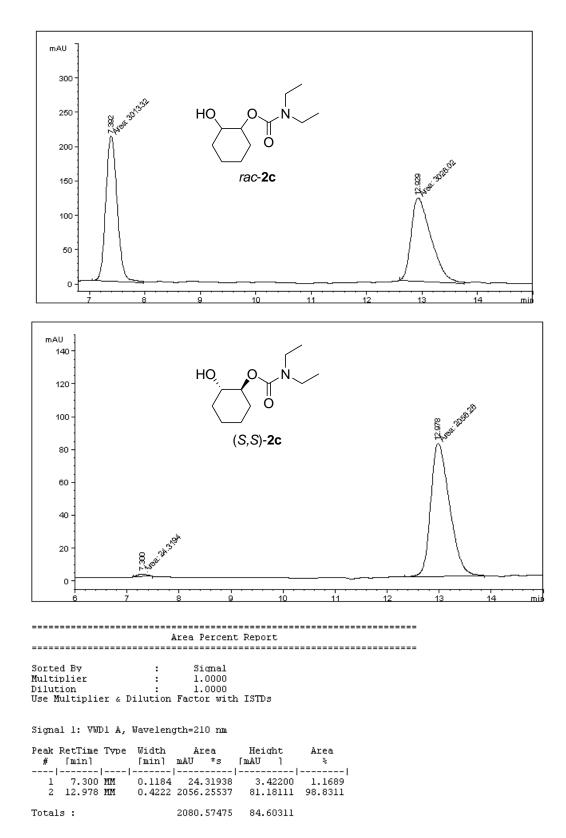
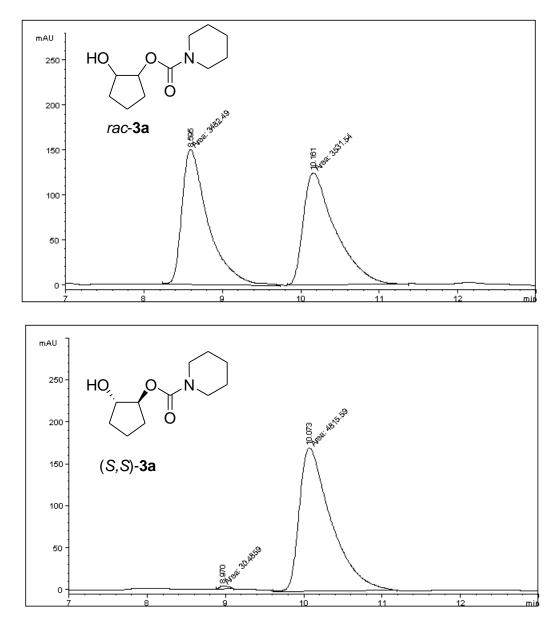


Figure S53. HPLC spectrum of 2c



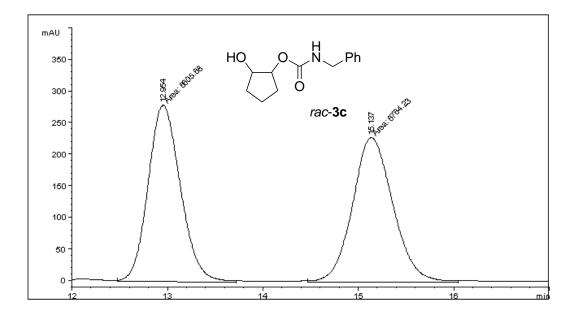
# Area Percent Report

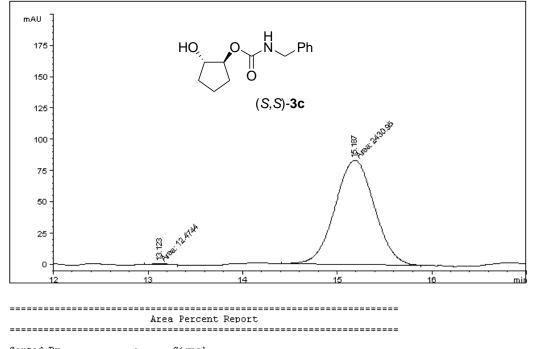
Sorted By : Signal Multiplier : 1.0000 Dilution : 1.0000 Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=210 nm

	RetTime [min]							Area %
								0.6291
2	10.073	MM	0.4702	4815.	59180	170.6	59839	99.3709
Total	s :			4846.	07771	174.1	4121	

Figure S54. HPLC spectrum of 3a





Sorted By : Signal Multiplier : 1.0000 Dilution : 1.0000 Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=210 nm

	RetTime [min]		Width [min]			Heid [mAU		Area %
1	13.123	MM	0.1860	12.	47439	1.1	1766	0.5105
2	15.187	MM	0.4855	2430.	95068	83.4	14405	99.4895
Total	з:			2443.	42507	84.5	56171	

Figure S55. HPLC spectrum of 3c

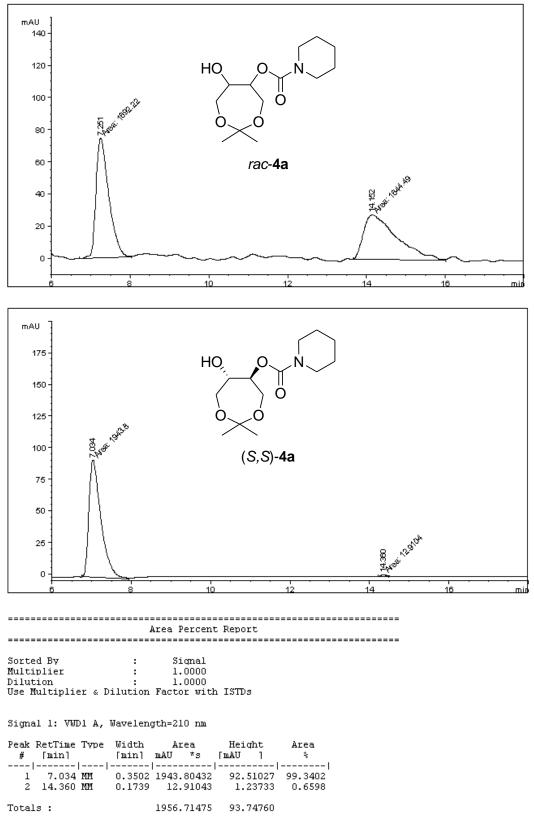


Figure S56. HPLC spectrum of 4a

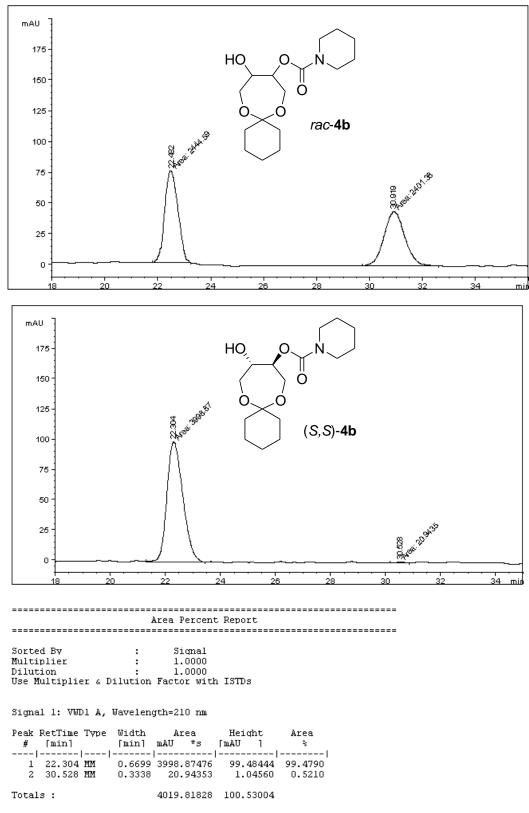


Figure S57. HPLC spectrum of 4b

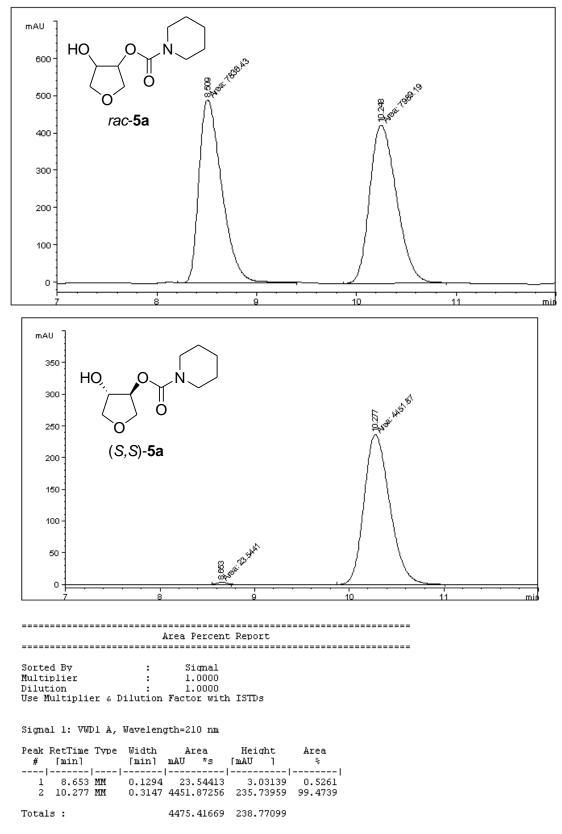


Figure S58. HPLC spectrum of 5a

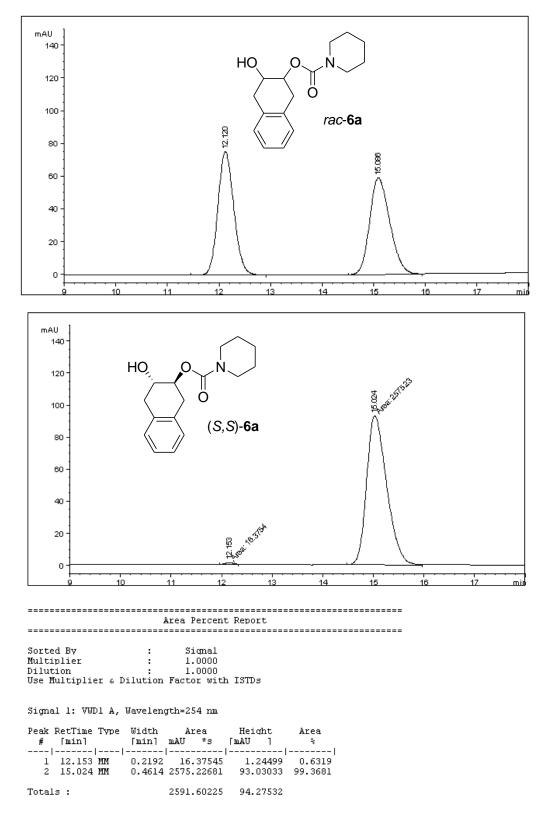


Figure S59. HPLC spectrum of 6a

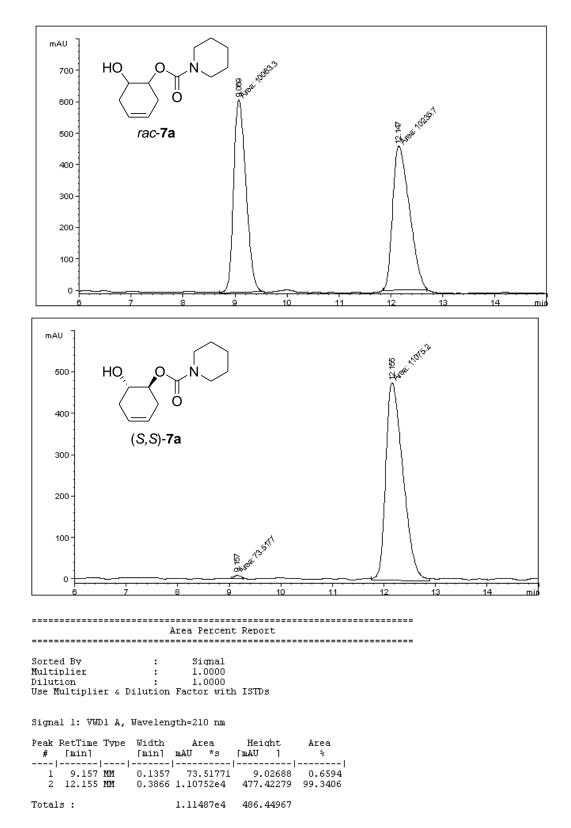
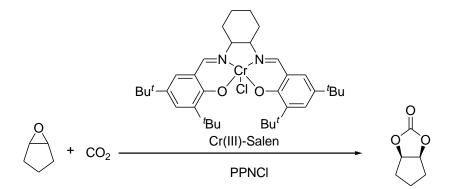


Figure S60. HPLC spectrum of 7a

## 4. Mechanistic understanding of ammonolysis reaction



Scheme S1. Synthesis of cis-carbamates from CO<sub>2</sub>/cyclopentene oxide

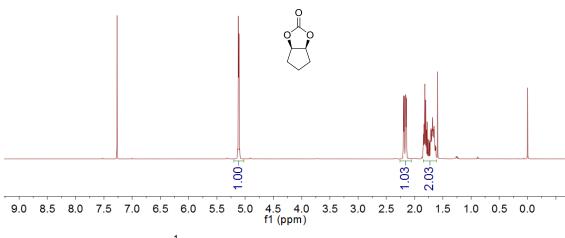


Figure S61. <sup>1</sup>H NMR spectrum of *cis*-cyclopentene carbonate

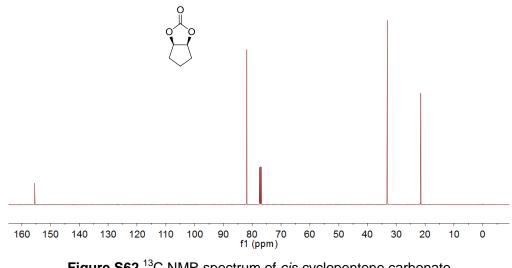
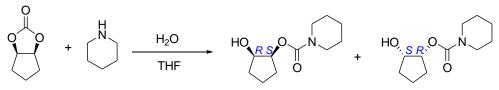


Figure S62.<sup>13</sup>C NMR spectrum of *cis*-cyclopentene carbonate



Scheme S2. Synthesis of cis-carbamates from aminolysis of carbonates

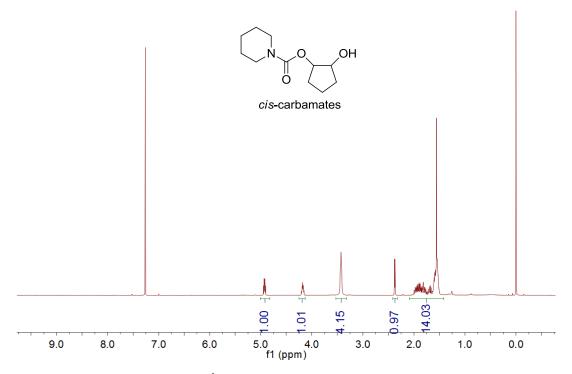


Figure S63. <sup>1</sup>H NMR spectrum of *cis*-carbamates.

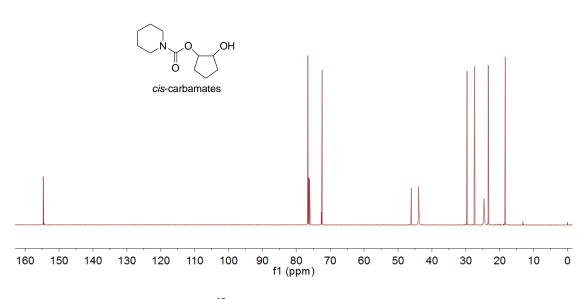
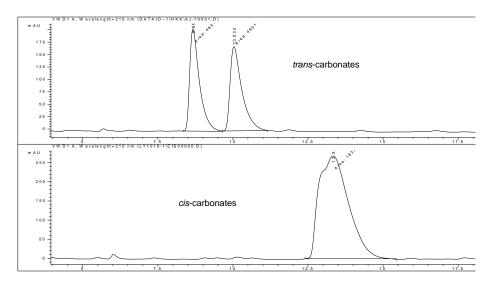
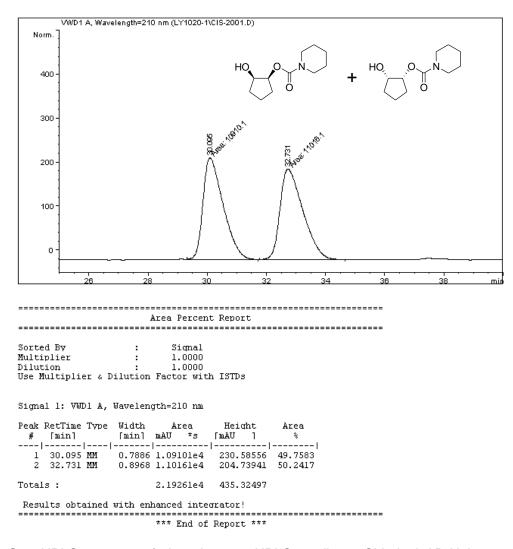


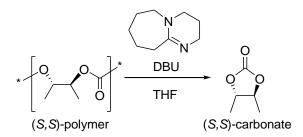
Figure S64. <sup>13</sup>C NMR spectrum of *cis*-carbamates.



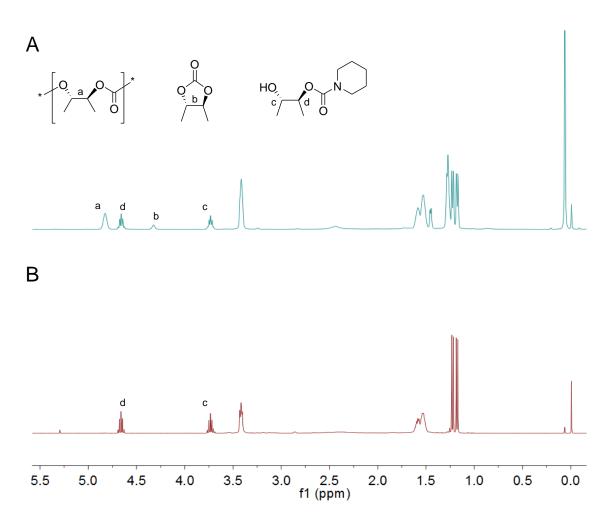
**Figure S65.** HPLC spectrum of *cis*-carbamate and *trans*-carbamate (HPLC conditons: Chiralpak AS-H, hexane/<sup>i</sup>PrOH = 95/5, flow rate = 1.0 mL/min, maximum absorption wavelength  $\lambda$  = 210 nm).



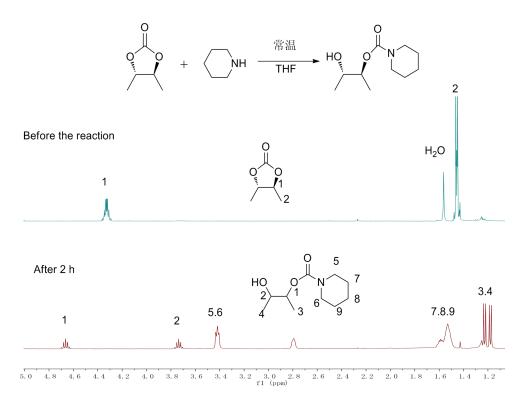
**Figure S66.** HPLC spectrum of *cis*-carbamate. HPLC conditons: Chiralpak AD-H, hexane/<sup>*i*</sup>PrOH = 98/2, flow rate = 0.8 mL/min, maximum absorption wavelength  $\lambda$  = 210 nm.



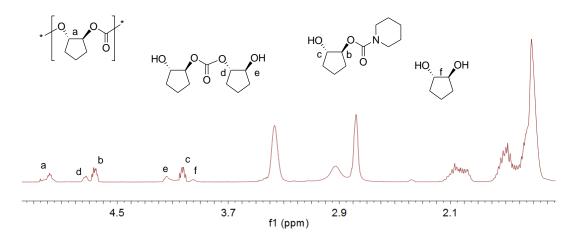
Scheme S3. Synthesis of trans-2-butene carbonate



**Figure S67.** <sup>1</sup>H NMR spectrum of the reaction mixture of poly(*trans*-2-butene carbonate) with piperidine in CDCl<sub>3</sub>. A) the reaction was carried out in absence of water; B) the reaction was carried out in presence of water. (Reaction condition: CO<sub>2</sub> polymers (464 mg, 4 mmol), amine (4 mmol, 1 equiv to per carbonate unit), water (4 mmol, 1 equiv to per carbonate unit) were dissolved in 10 mL THF at room temperature.The reaction mixture was stirred at 60 °C for 1 h. A small amount of mixture was removed for <sup>1</sup>H NMR analysis after the removal of the solvent under reduced pressure



**Figure S68.** <sup>1</sup>H NMR spectrum of the reaction of *trans*-2-butene carbonate with piperidine in CDCl<sub>3</sub>. Reaction condition: *trans*-2-butene carbonate (0.2 mmol), piperidine (0.2 mmol, 1 equiv to per carbonate unit) and water (0.2 mmol, 1 equiv to per carbonate unit) were dissolved in 5 mL THF. The reaction mixture was stirred for 2 h at room temperature. A small amount of mixture was removed for <sup>1</sup>H NMR analysis after the removal of the solvent under reduced pressure



**Figure S69.** <sup>1</sup>H NMR spectrum of the reaction mixture of poly(cyclopentene carbonate) with piperidine in CDCl<sub>3</sub>. Reaction condition: CO<sub>2</sub> polymers (512 mg, 4.0 mmol), amine (20 mmol, 5 equiv to per carbonate unit), water (20 mmol, 5 equiv to per carbonate unit) were dissolved in 10 mL THF at room temperature The reaction mixture was stirred at 60 °C for 24 h. A small amount of mixture was removed for <sup>1</sup>H NMR analysis after the removal of the solvent under reduced pressure

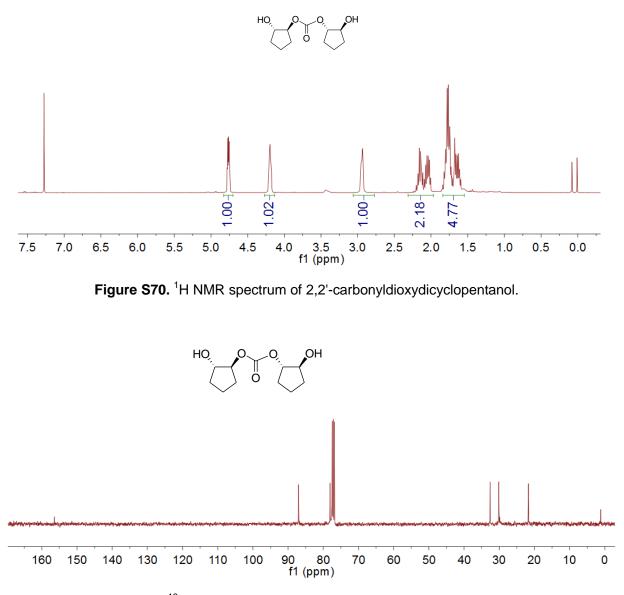
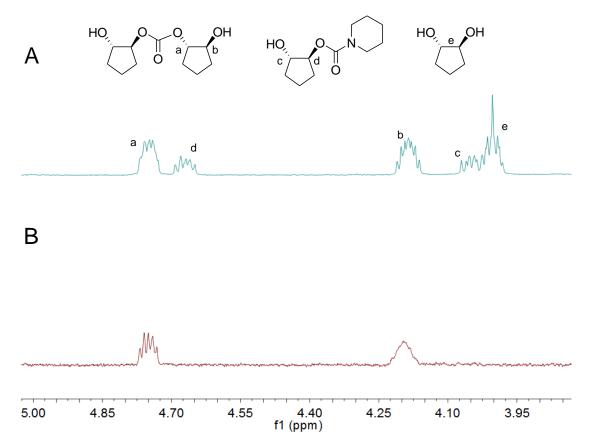
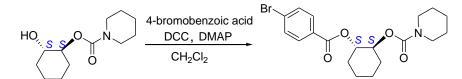


Figure S71. <sup>13</sup>C NMR spectrum of 2,2'-carbonyldioxydicyclopentanol.



**Figure S72.** The aminolysis reaction of 2,2'-carbonyldioxydicyclopentanol with piperidine in presence (top, plot-A) and absence (bottom, plot-B) of water. Reaction condition: 2,2'-carbonyldioxydicyclohexanol (0.2 mmol), piperidine (0.2 mmol) and water (0.2 mmol) were dissolved in 5 mL THF at room temperature. The reaction mixture was stirred at room temperature for 24 h. A small amount of mixture was removed for <sup>1</sup>H NMR analysis after the removal of the solvent under reduced pressure

### 5. NMR spectrum of (S,S)-carbamate derivative for X-Ray analysis



Scheme S4. Synthesis of (S,S)-carbamate derivative for X-Ray analysis

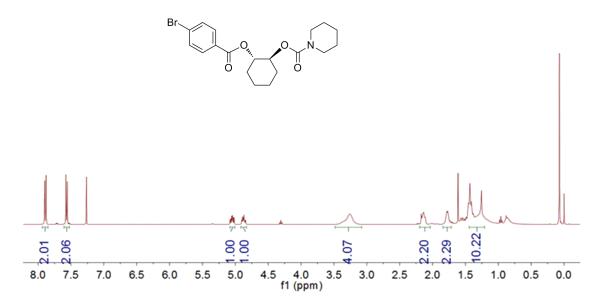


Figure S73. <sup>1</sup>H NMR spectrum of (*S*,*S*)-carbamate derivative for X-Ray analysis

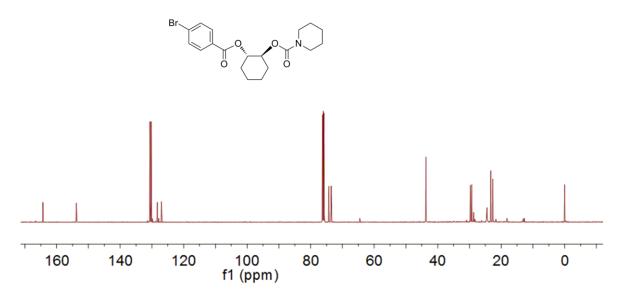


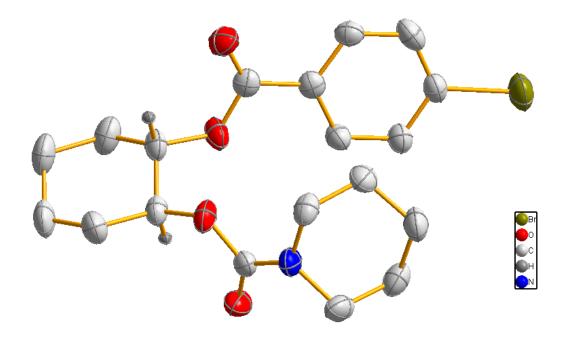
Figure S74. <sup>13</sup>C NMR spectrum of (*S*,*S*)-carbamate derivative for X-Ray analysis

#### 6. Crystal structure determination

**Crystal structure determination.** The single crystal of (*S*,*S*)-carbamate derivative (CCDC: 1491321) suitable for X-ray structural analysis was obtained from a hexane at -7 °C. Diffraction data were collected at 200 K on a CCD diffractometer using graphite-monochromated Mo K $\alpha$  radiation ( $\lambda = 0.71073$  Å). The structures were solved by direct methods and refined by full-matrix least squares on  $F^2$ . All nonhydrogen atoms were refined anisotropically, and the hydrogen atoms were included in idealized positions. All calculations were performed using the SHELXTL crystallographic software packages. The data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data\_request/cif.

	( <i>S</i> , <i>S</i> )-carbamate derivative
structure	
mol formula	C19 H24 Br N O4
mol wt	410.30
cryst syst	Orthorhombic
space group	P2(1)2(1)2(1)
$a/ m \AA$	9.586(4)
b/Å	10.266(4)
$c/{ m \AA}$	19.842(8)
a/deg	90.00
$\beta$ /deg	90.00
γ/deg	90.00
V/Å <sup>3</sup>	1952.8(14)
Z	4
abs coeff/mm <sup>-1</sup>	2.127
R <sub>int</sub>	0.0496
R1 (I $\geq 2\sigma$ )	0.0438 (2616)
wR2 $(I > 2\sigma)$	0.982
GOF	1.007

Table S2. Summary of crystal data and structural refinement details of carbamate derivative



**Figure S75.** X-Ray molecular structures of (S,S)-carbamate derivative. Hydrogen atoms and uncoordinated solvent omitted for clarity, and thermal ellipsoids are at the 30% probability level.