

Synthesis of Monometallic (Au and Pd) and Bimetallic (AuPd) Nanoparticles Using Carbon Nitride (C₃N₄) Quantum Dots via the Photochemical Route for Nitrophenol Reduction

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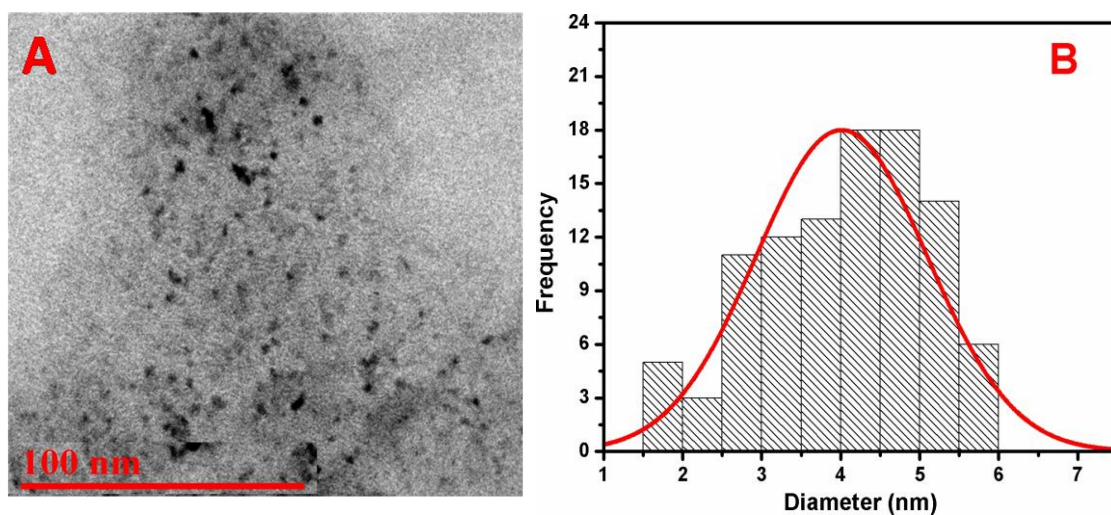


Figure S1: (a) TEM image of dialysed C_3N_4 QDs solution prepared by urea and trisodium citrate (b) size-distribution histogram (4.5 ± 1.1 nm)

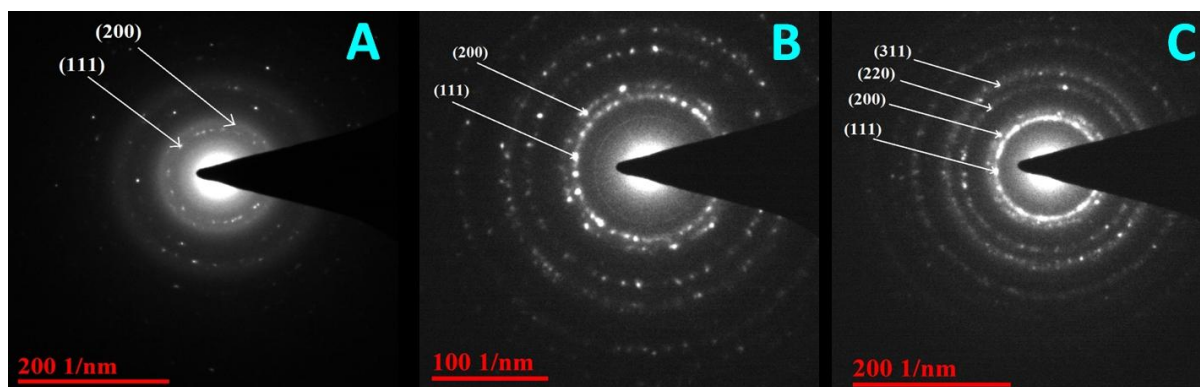


Figure S2: Selected area electron diffraction patterns in (a) Au, (b) Pd & (c) AuPd NPs.

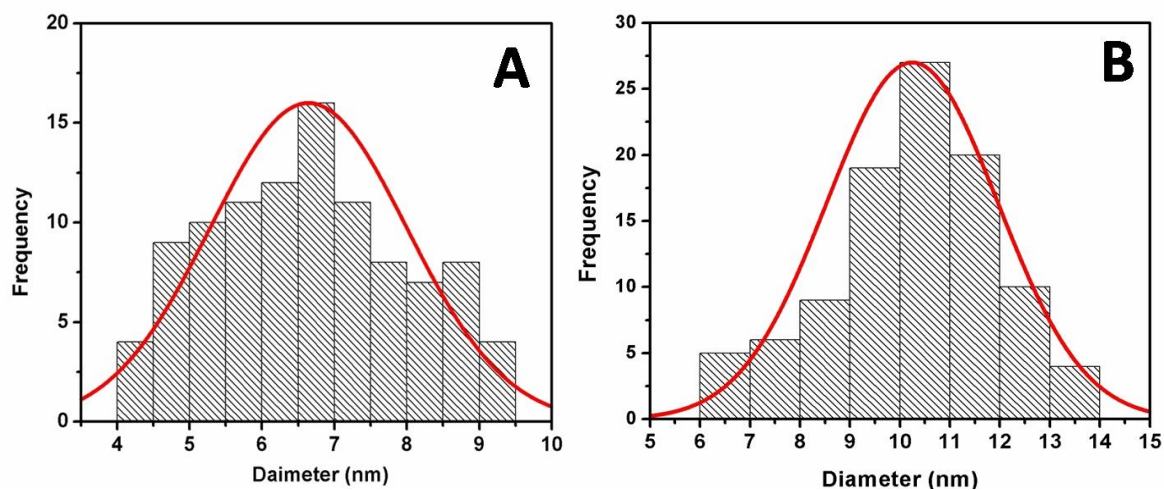


Figure S3: (a) Size-distribution histogram of Au NPs (6.8 ± 1.3 nm) and (b) size-distribution histogram of Pd NPs (10.1 ± 1.2 nm)

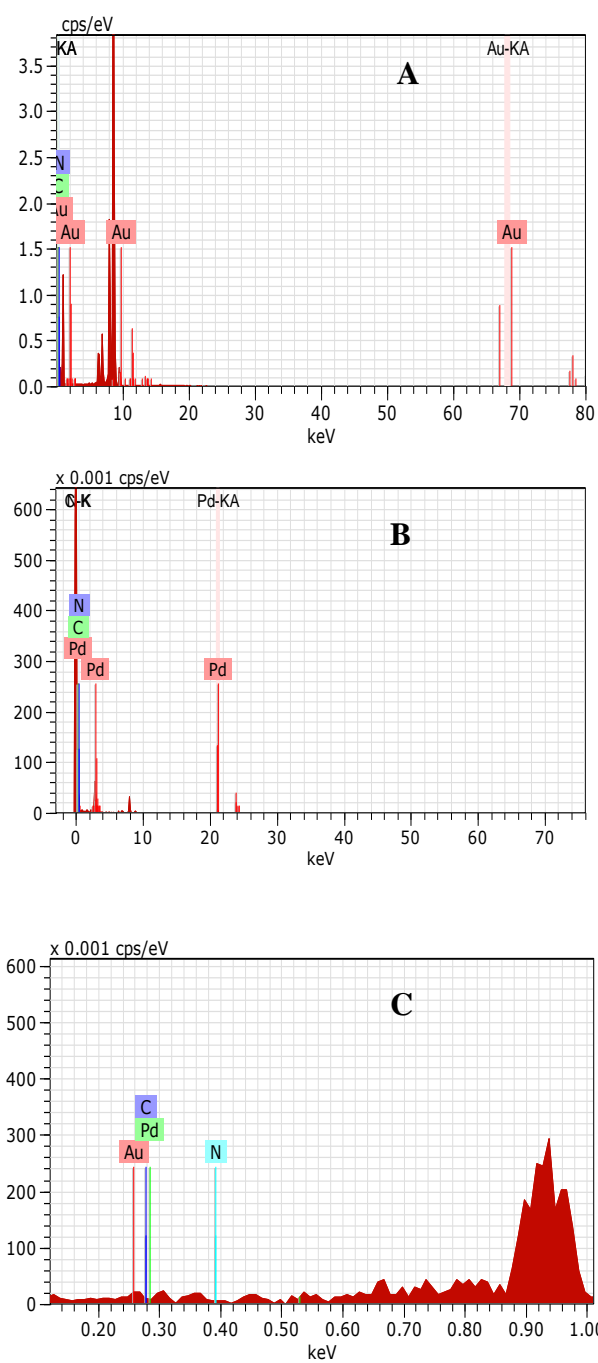


Figure S4: EDS spectra of (a) Au, (b) Pd & (c) AuPd NPs

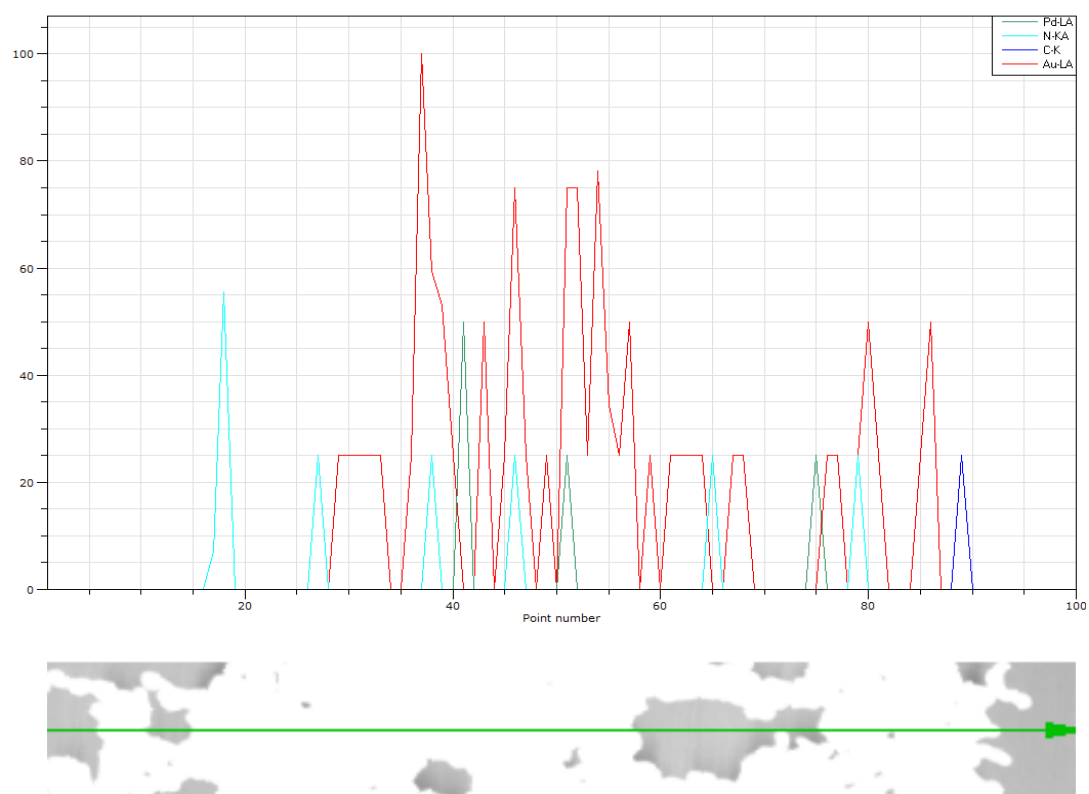


Figure S5: Line spectrum of AuPd NPs solution shows the presence of both Au and Pd nanoparticles with C and N from g-C₃N₄.

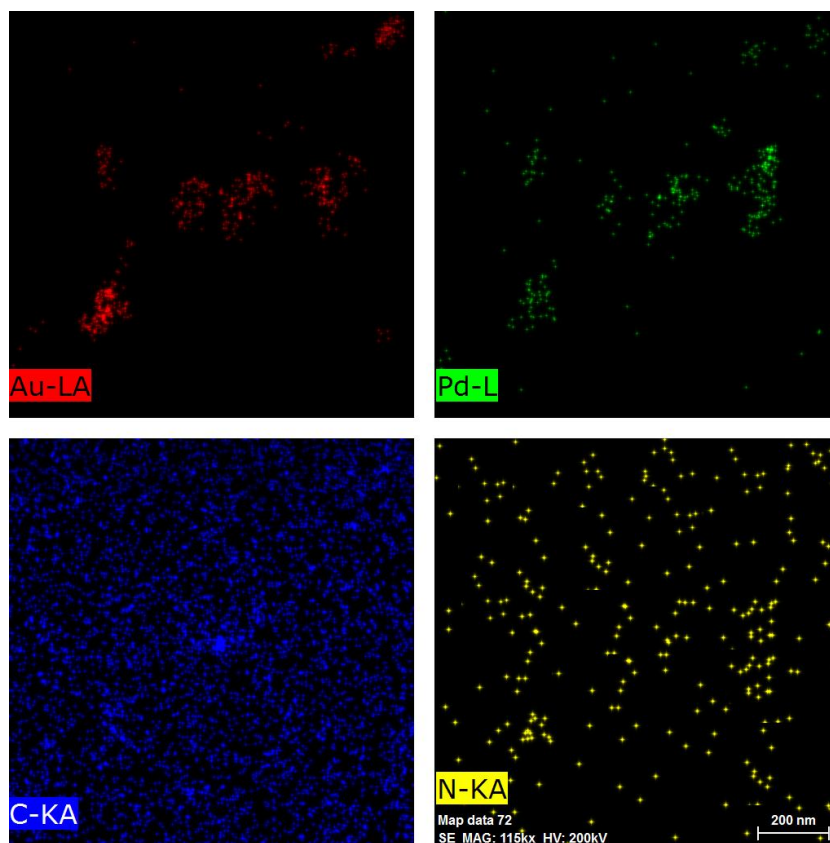


Figure S6: EDS mapping of AuPd alloy NPs shows the presence of both Au and Pd nanoparticles with C and N from C_3N_4 .

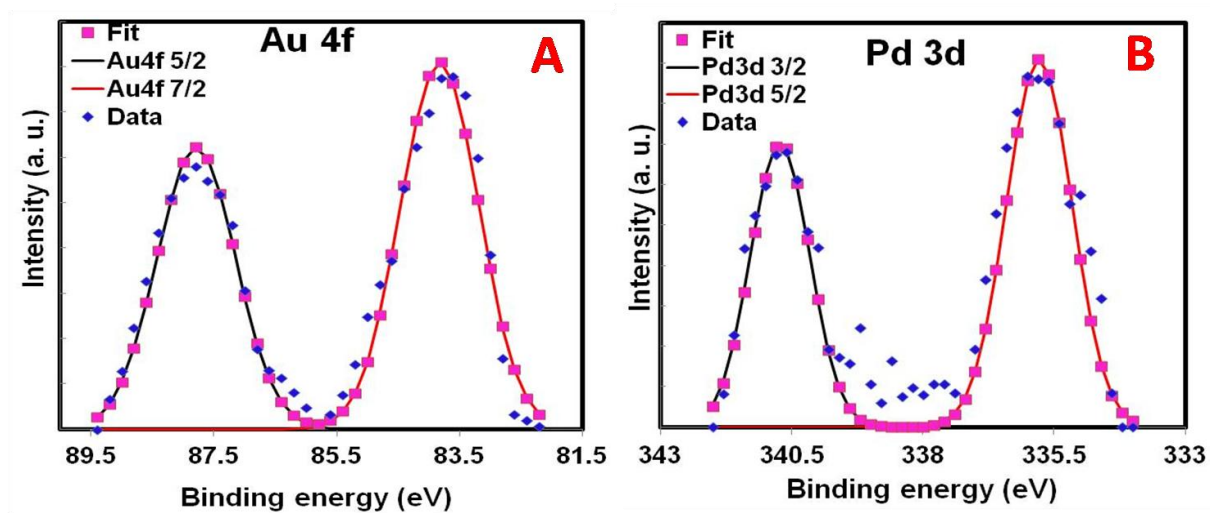


Figure S7: High resolution XPS spectra (a) Au 4f and (b) Pd 3d for Au and Pd NPs on g-C₃N₄ structure, respectively.

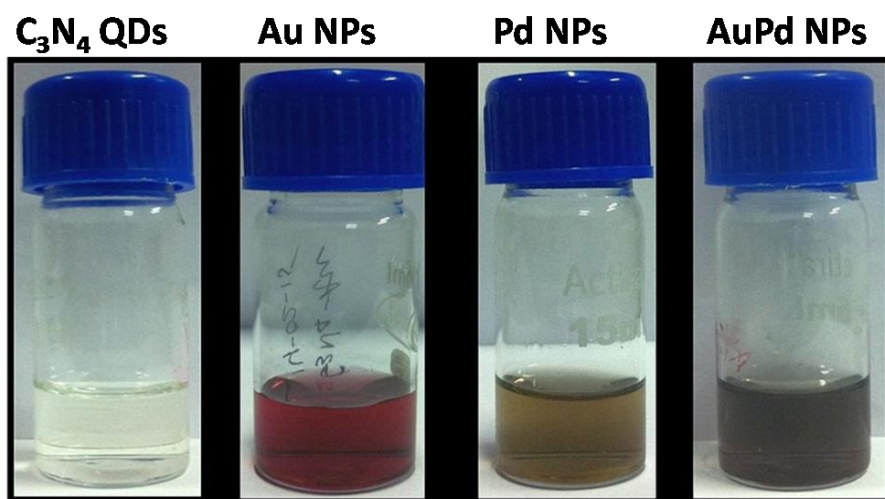


Figure S8: Digital photograph of C_3N_4 QDs, Au, Pd and AuPd NPs

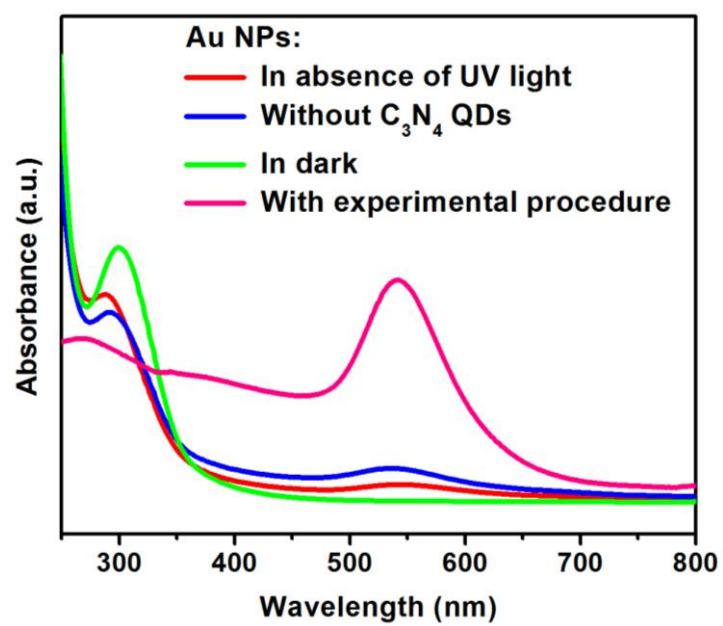


Figure S9: UV-vis of Au solution synthesized in different conditions (in absence of UV light, without C₃N₄ QDs, in dark, and with experimental procedure)

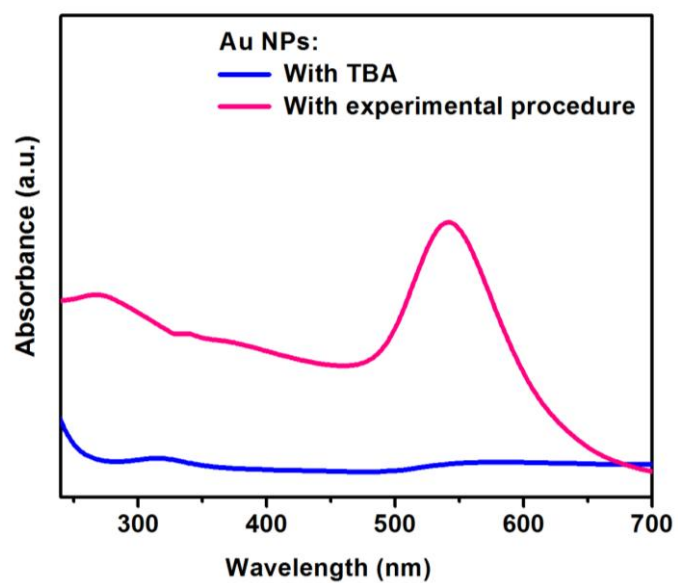


Figure S10: UV-vis of Au NPs solution synthesized in presence of an electron scavenger, Tert-butyl alcohol (TBA) and with experimental procedure (without electron scavenger).

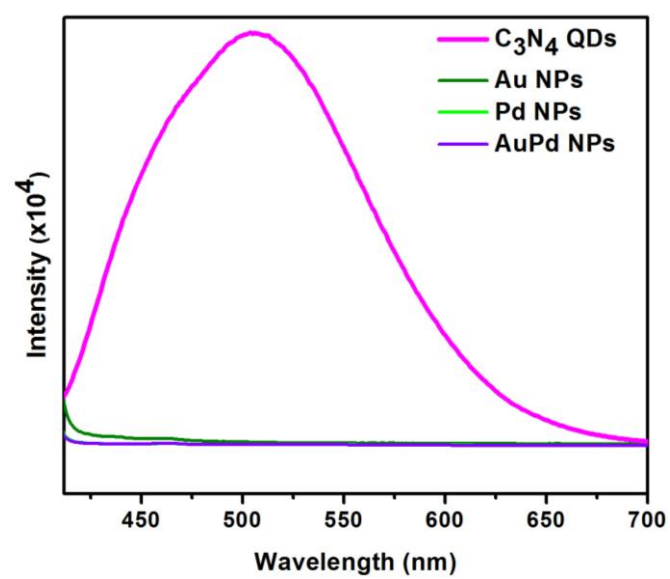


Figure S11: Comparative photoluminescence spectra of C₃N₄ QD and Au, Pd, and AuPd NPs.

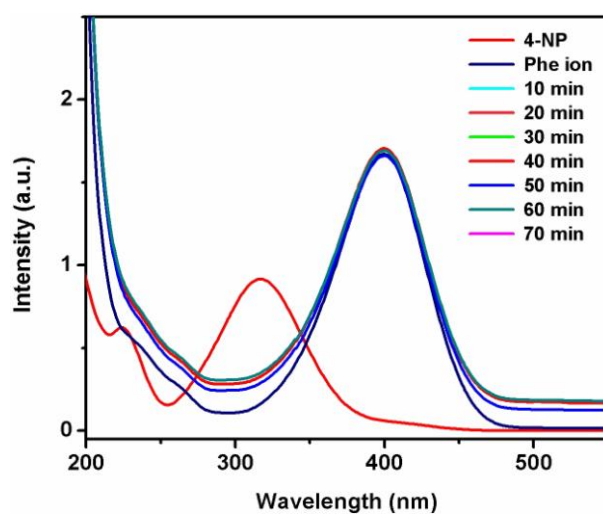


Figure S12. UV-vis spectra of 4-nitrophenol and 4-nitrophenolate ion after addition of NaBH_4 in the absence of any catalyst

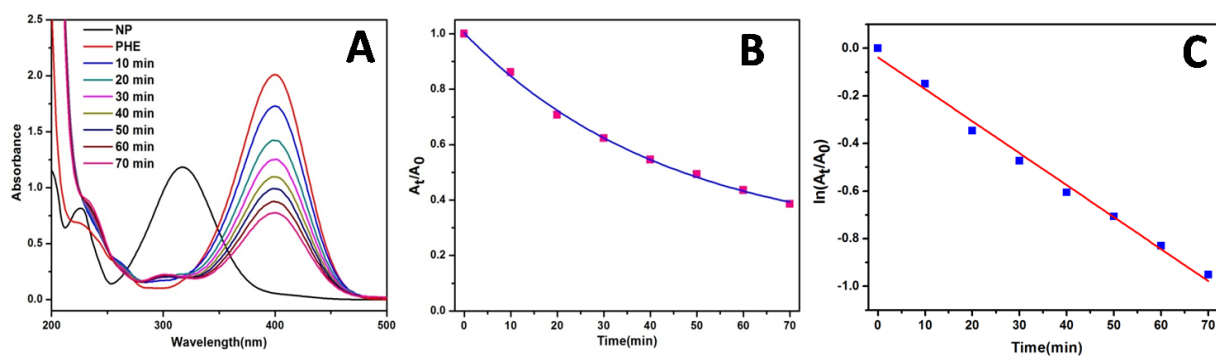


Figure S13. UV-vis spectra of (a) 4-nitrophenol reduction in presence of C_3N_4 (b) A_t/A_0 vs. time (min) plot (c) $\ln(A_t/A_0)$ vs. time (min) plot. Conditions: $[4\text{-NP}] = 10^{-4}$ M and amount of catalyst = 30 μL .

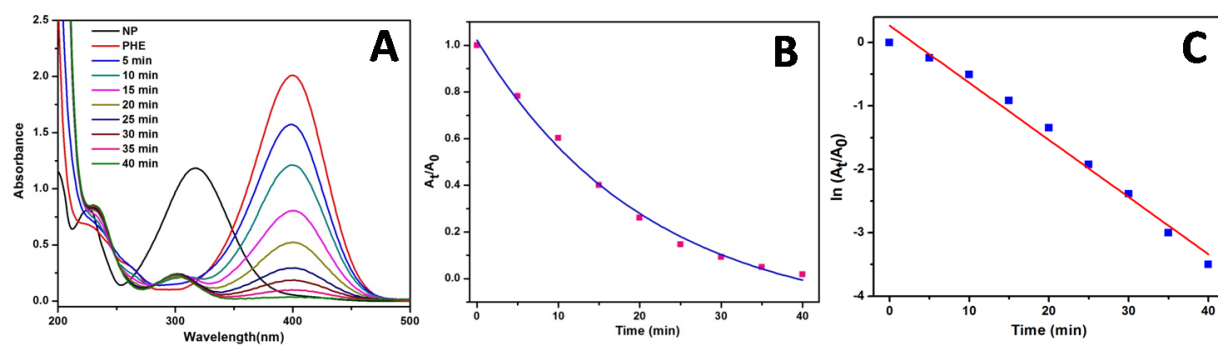


Figure S14. UV-vis spectra of (a) 4-nitrophenol reduction in presence of Au NPs (b) A_t/A_0 vs. time (min) plot (c) $\ln(A_t/A_0)$ vs. time (min) plot. Conditions: $[4\text{-NP}] = 10^{-4}$ M and amount of catalyst = 30 μL .

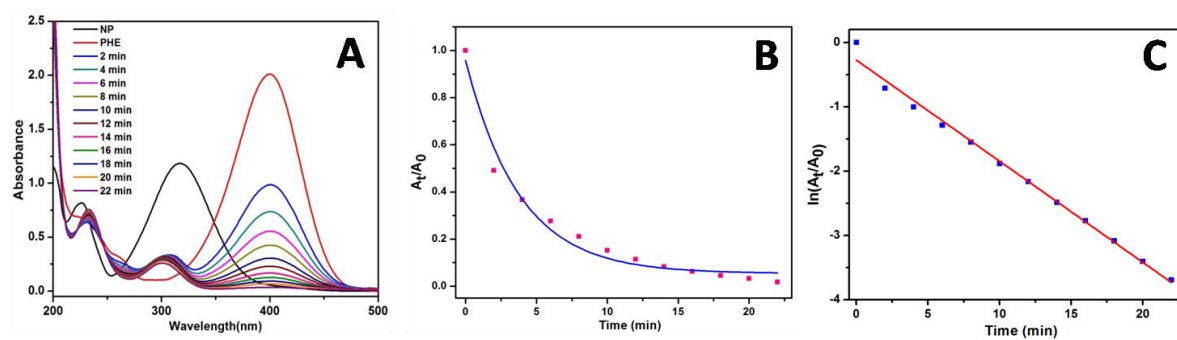


Figure S15. UV-vis spectra of (a) 4-nitrophenol reduction in presence of Pd NPs (b) A_t/A_0 vs. time (min) plot (c) $\ln(A_t/A_0)$ vs. time (min) plot. Conditions: $[4\text{-NP}] = 10^{-4}$ M and amount of catalyst = 30 μL .

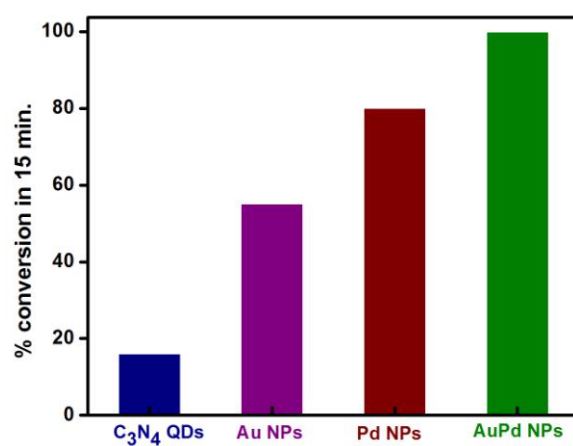


Figure S16. Comparative study of percent conversion from 4-NP to 4-AP in 15 min C₃N₄ QDs, Au, Pd & AuPd NPs

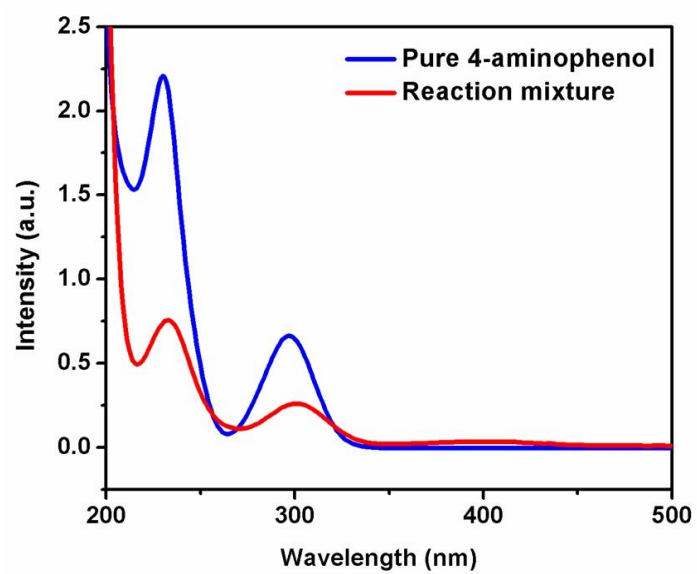


Figure S17. UV-vis spectra of pure 4-aminophenol and reaction mixture

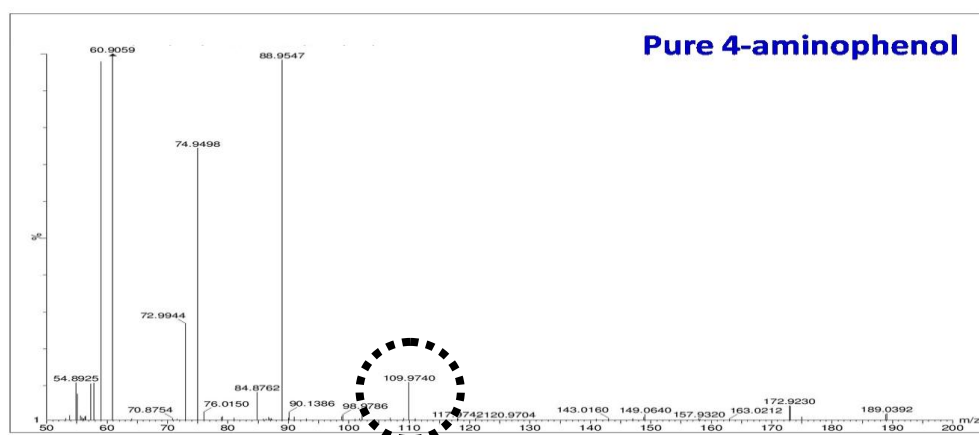
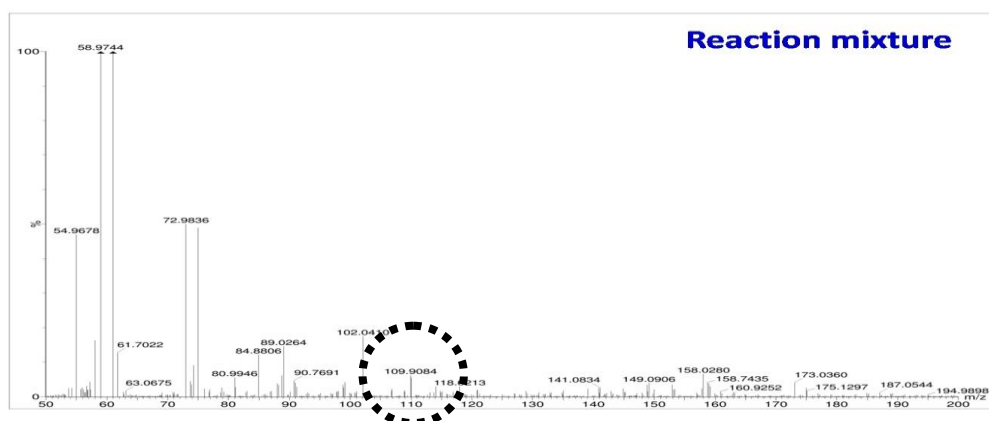
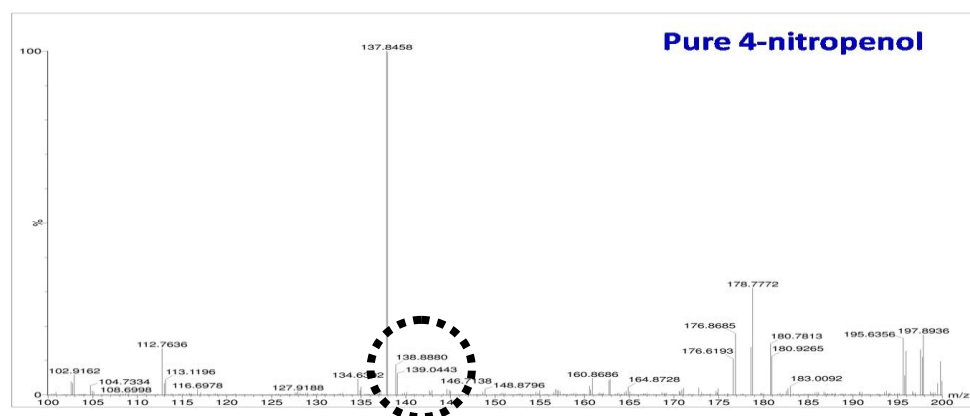


Figure S18. Mass spectra of pure 4-nitrophenol, reaction mixture and pure 4-aminophenol.

The molecular ion peak for all is encircled.