

# Synthesis of Fused Imidazole-Containing Ring Systems *via* Dual Oxidative

## Amination of C(sp<sup>3</sup>)-H bonds

Georgette Castanedo,\* Yanzhou Liu, James J. Crawford and Marie-Gabrielle Braun

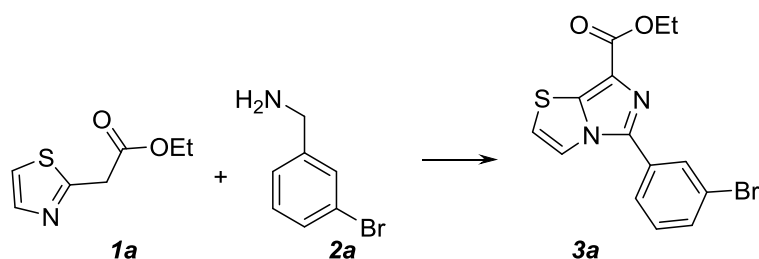
*Discovery Chemistry, Genentech, Inc., South San Francisco, CA 94080*

### SUPPORTING INFORMATION

#### Table of Contents

I. SI Schemes and Tables.....	pg. 1
II. NMRs.....	pg. 4

#### Reaction Optimization using ethyl 2-thiazol-2-ylacetate and 3-bromobenzylamine:



**SI Table 1: Halogen Source Optimization**

Entry	Iodine Source	Yield of compound <b>3a</b> (%) <sup>a,b</sup>	% conversion of <b>1a</b>
1	Iodine	64	100
2	NIS	84	100
3	Bu <sub>4</sub> NI	23	100
4	KI	25	100

<sup>a</sup> Yield of **3a**, determined by LCMS using naphthalene as a quantitative internal standard.

<sup>b</sup> Correction factor of 1.05 applied to LCMS yields of **3a** after determining difference in ratios of

product to starting material by LCMS/NMR

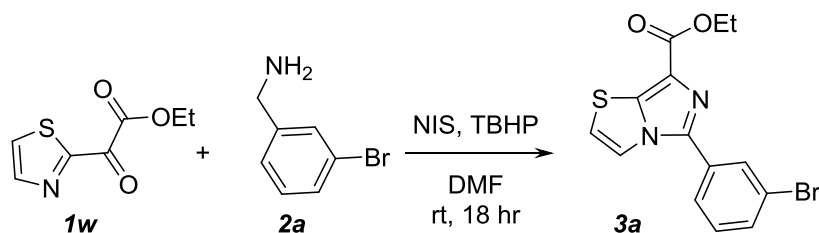
**SI Table 2: Solvent Optimization**

Entry	Solvent	Yield of compound <b>3a</b> (%) <sup>a,b</sup>	% conversion of <b>1a</b>
1	DMA	85	100
2	DMF	87	100
3	DMSO	86	100
4	ACN	92	100

<sup>a</sup> Yield of **3a**, determined by LCMS using naphthalene as a quantitative internal standard

<sup>b</sup> Correction factor of 1.05 applied to LCMS yields of **3a** after determining difference in ratios of product to starting material by LCMS/NMR

**SI Table 3:  $\beta$ -Keto ester as substrate**

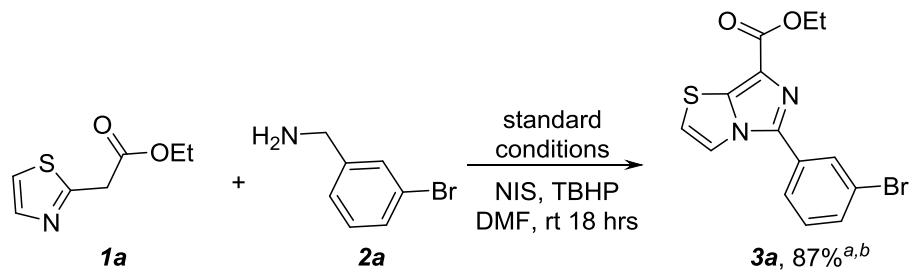


Entry	Substrate	Yield of compound <b>3a</b> (%) <sup>a,b</sup>	% conversion of <b>1w</b>
1	<b>1w</b>	89	100

<sup>a</sup> Yield of **3**, determined by LCMS using naphthalene as a quantitative internal standard

<sup>b</sup> Correction factor of 1.05 applied to LCMS yields of **3a** after determining difference in ratios of product to starting material by LCMS/NMR

**SI Table 4: Control Experiments**



<b>1a</b>	+	<b>no amine</b>	$\xrightarrow{\text{standard conditions}}$	<b>[O] of 1</b>
<b>1a</b>	+	<b>2a</b>	$\xrightarrow[\text{MINUS NIS}]{\text{standard conditions}}$	<b>no product</b>
<b>1a</b>	+	<b>2a</b>	$\xrightarrow[\text{MINUS TBHP}]{\text{standard conditions}}$	<b>3a, 59\%<sup>a,b</sup></b>

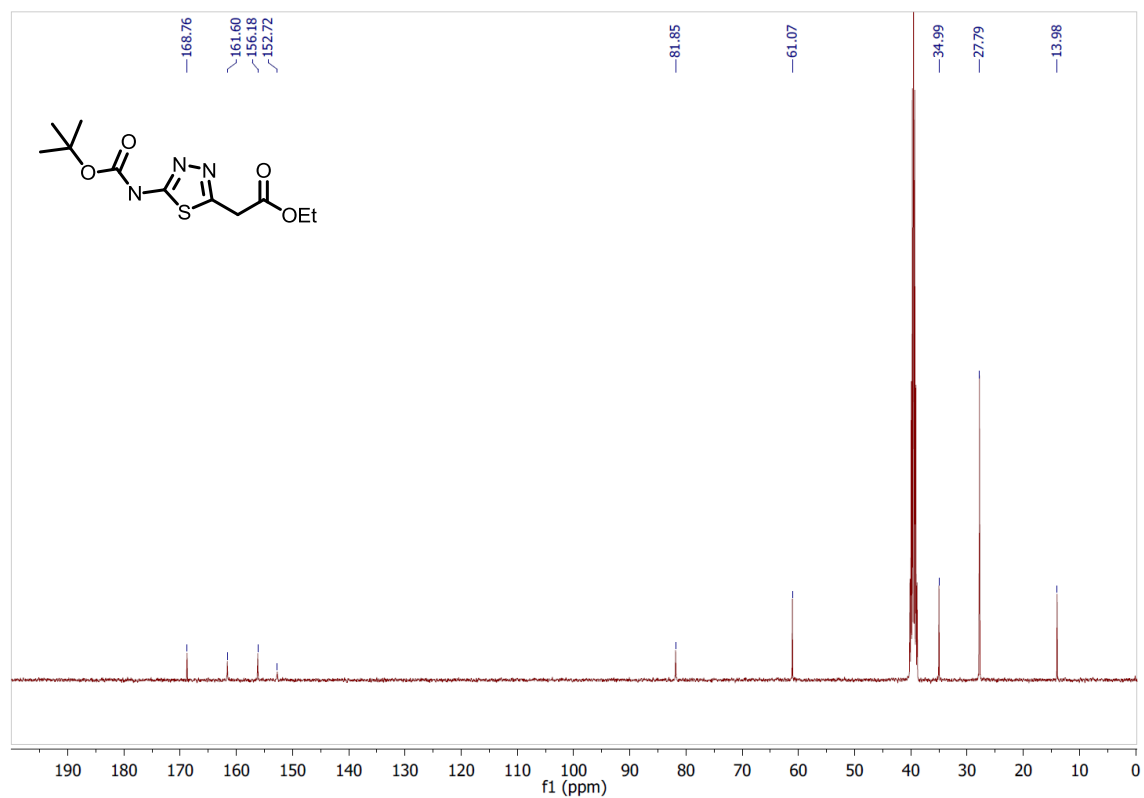
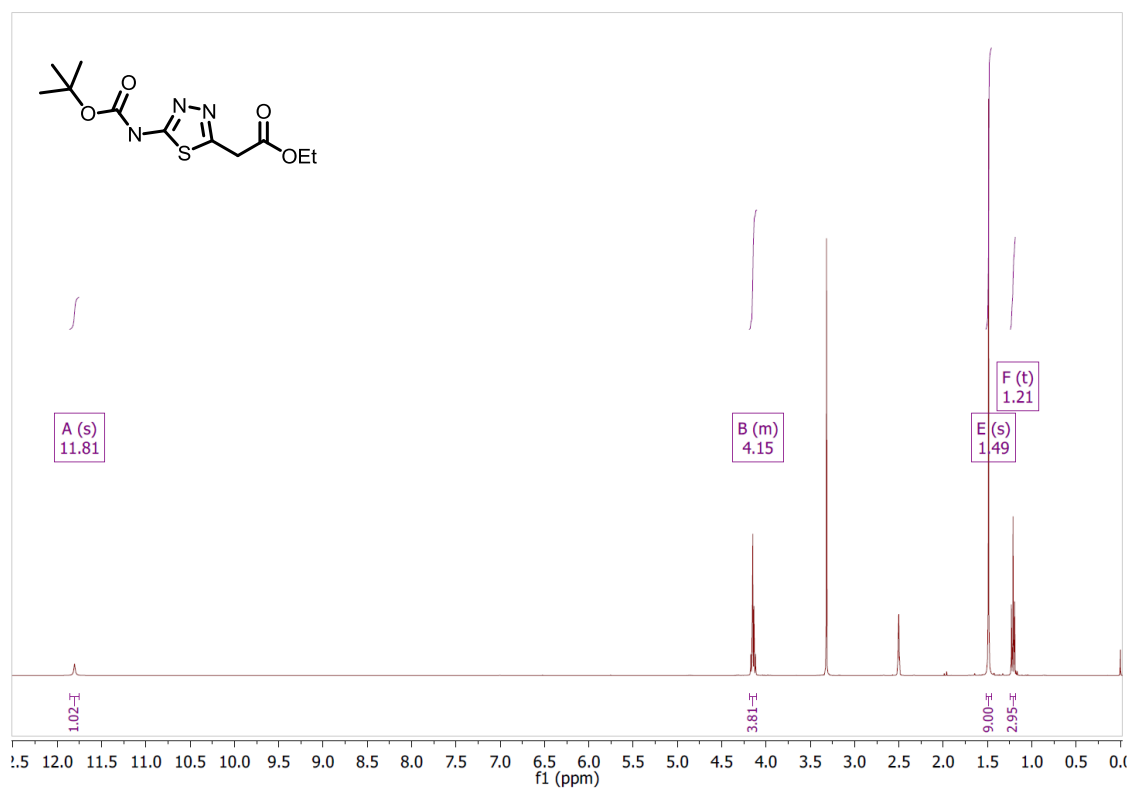
Entry	Reagent Missing	Yield of compound 3a (%) <sup>a,b</sup>	% conversion of 1
1	No Amine	0	95 <sup>c</sup>
2	No NIS	0	72
3	No TBHP	59	75

<sup>a</sup> Yield of **3a**, determined by LCMS using naphthalene as a quantitative internal standard

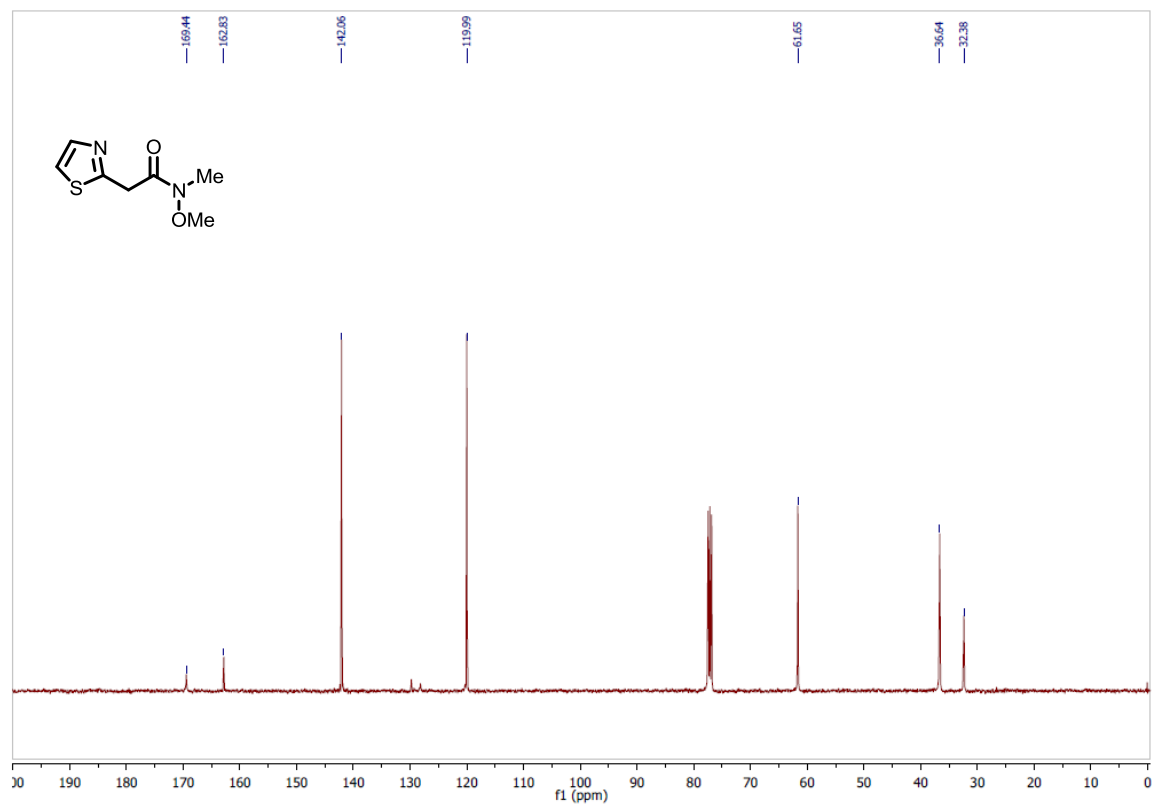
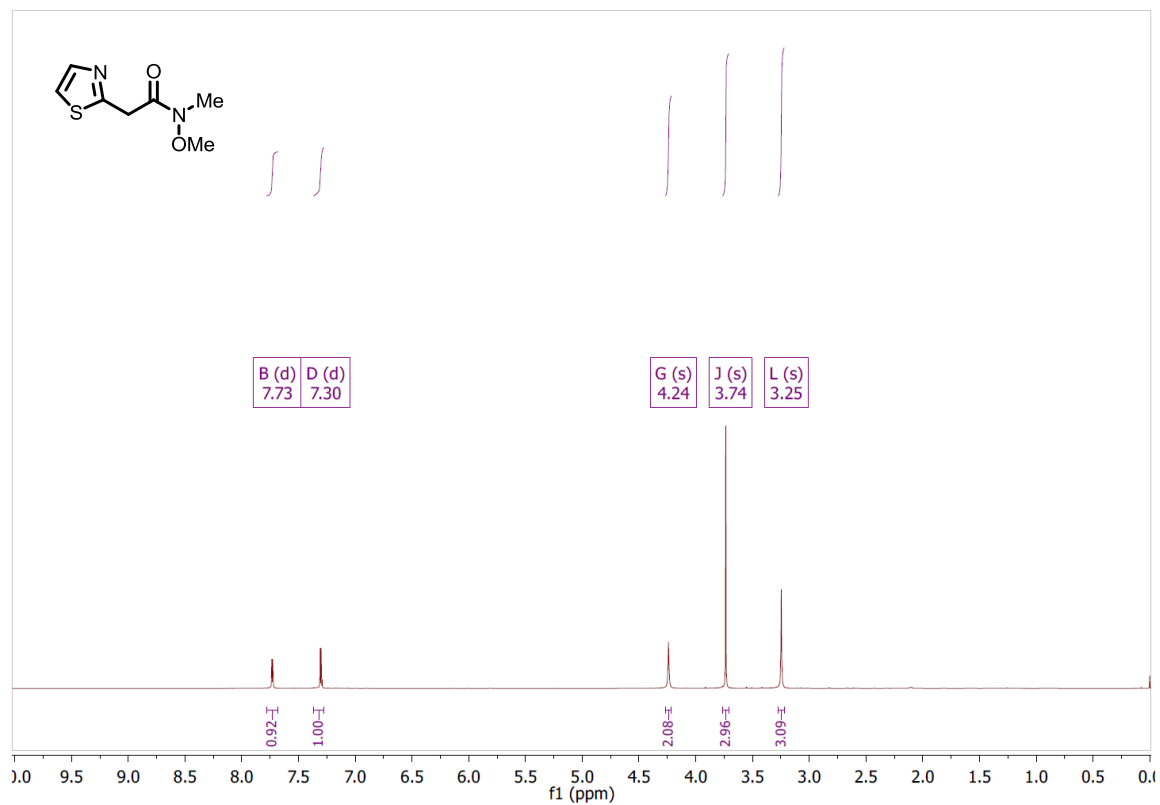
<sup>b</sup> Correction factor of 1.05 applied to LCMS yields of **3a** after determining difference in ratios of product to starting material by LCMS/NMR

<sup>c</sup> Mass for oxidation of **1a** to **1w** was seen by mass spec which is consistent with observations from Mohan, D.; Rao, S.; Ravi, C.; and Adimurthy, S. *Org. Biomol. Chem.* **2015**, *13*, 5602.

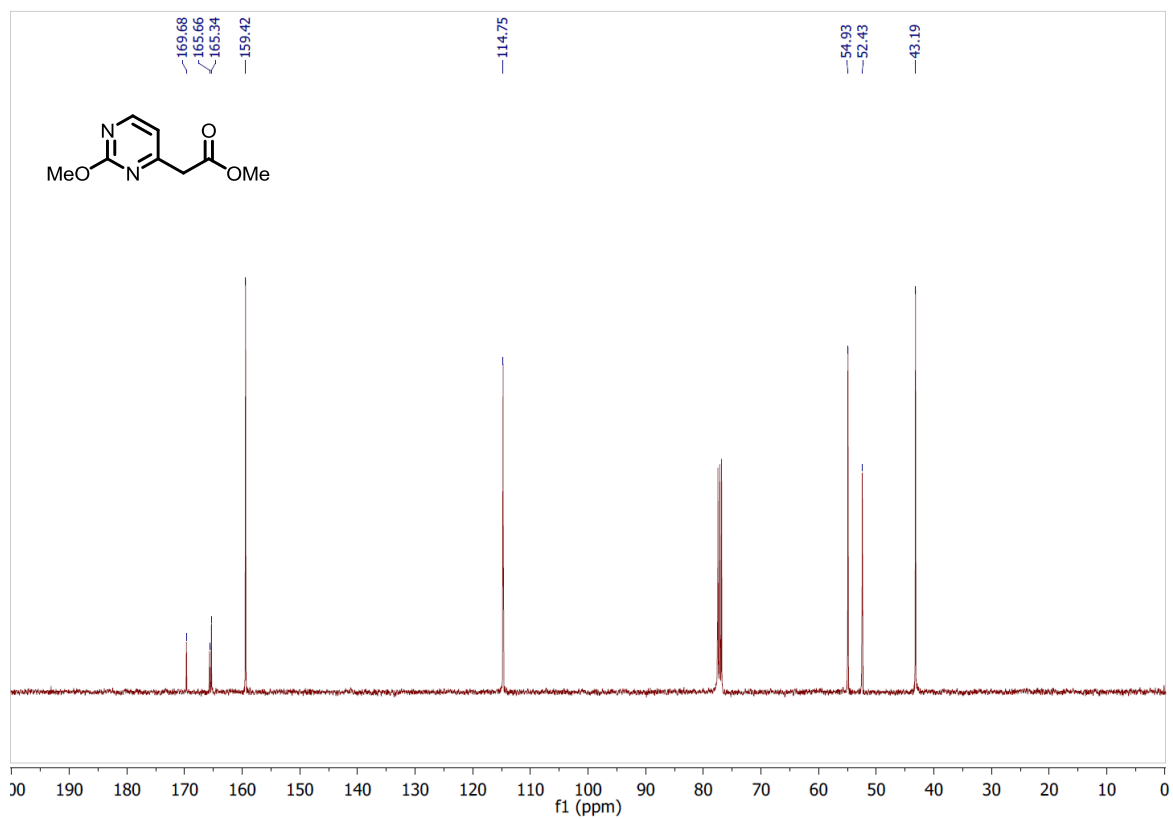
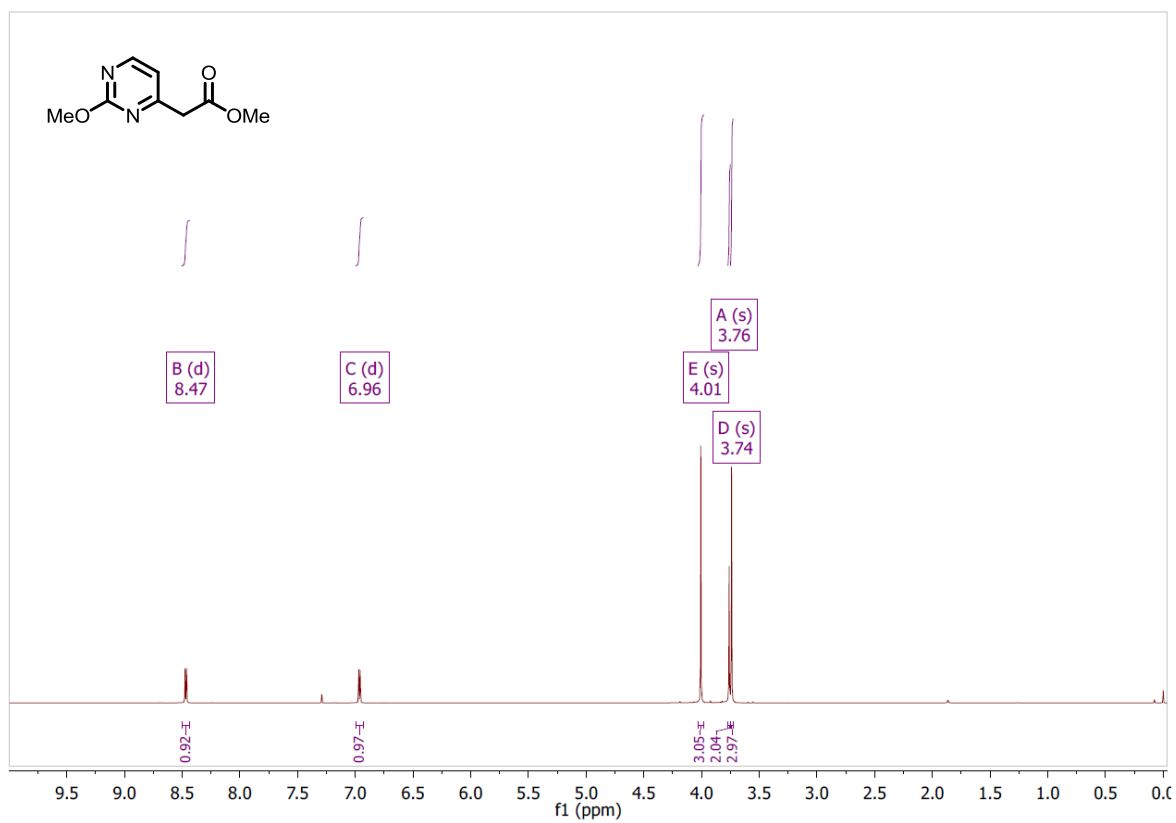
# **NMR Characterization Data** **Starting Material 1e**



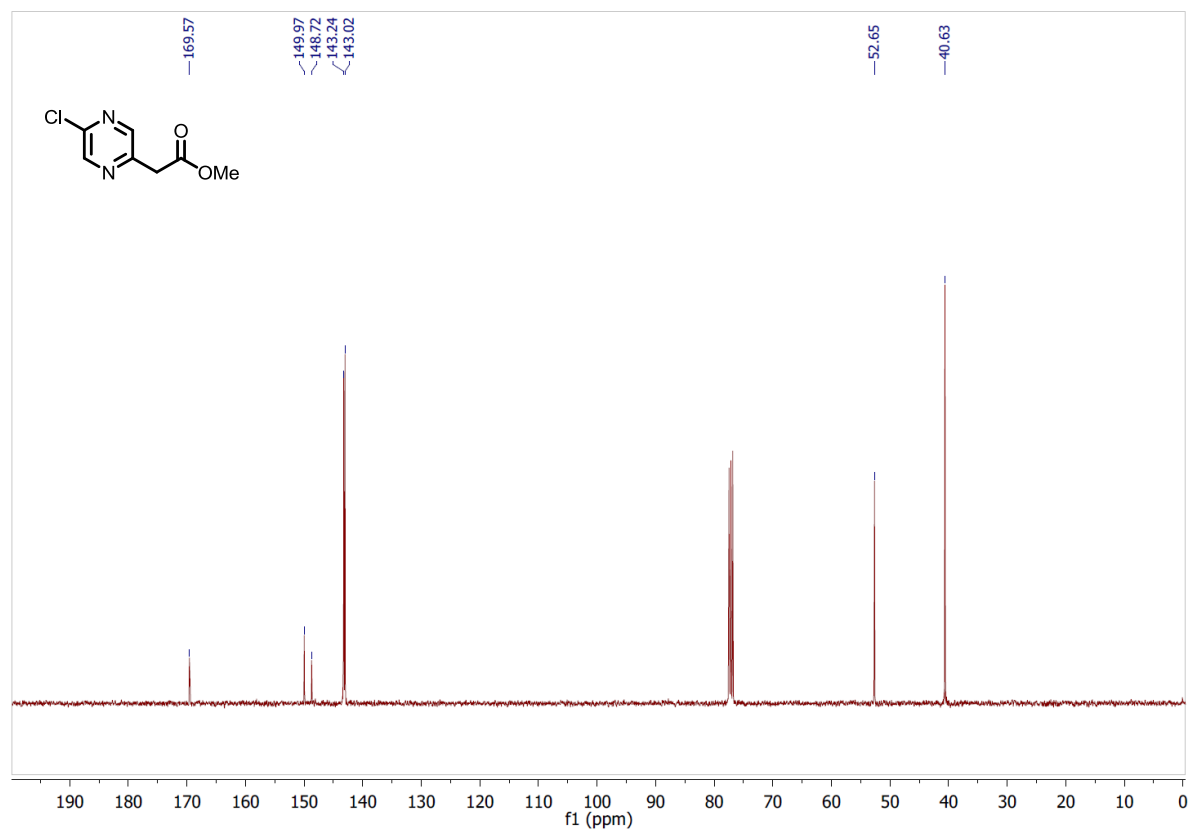
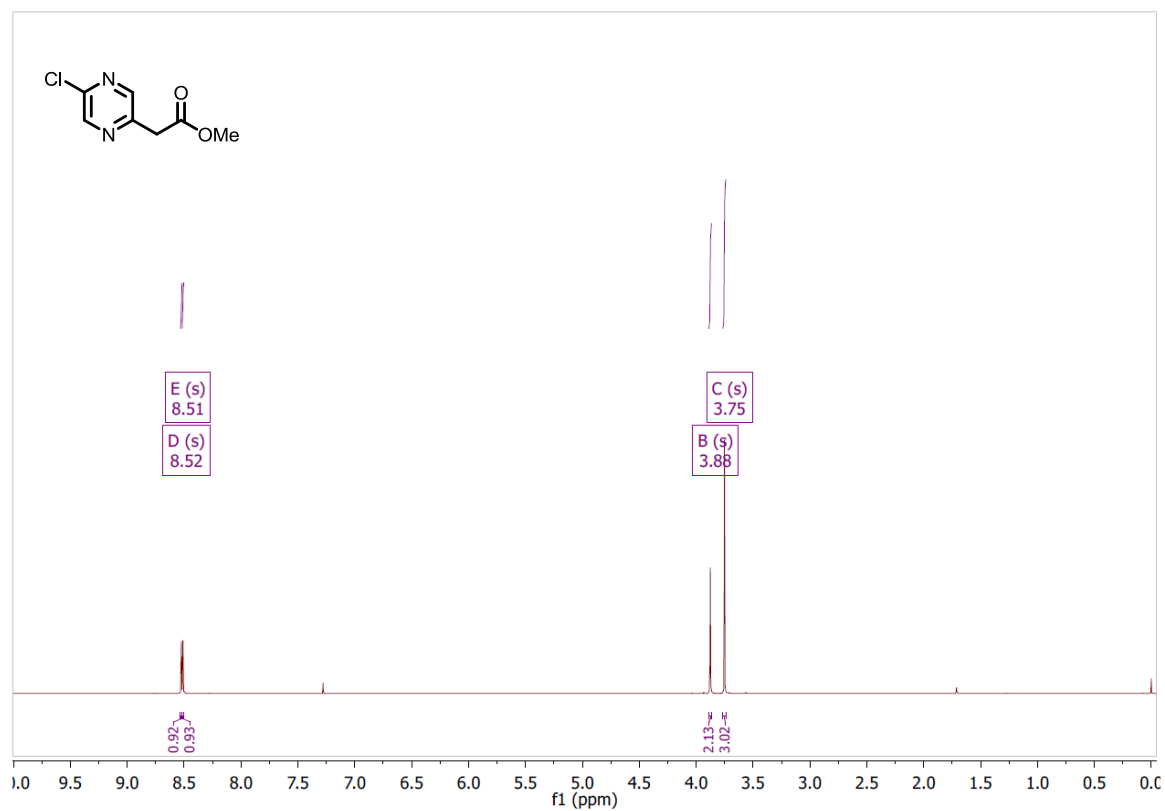
## Starting Material 1h



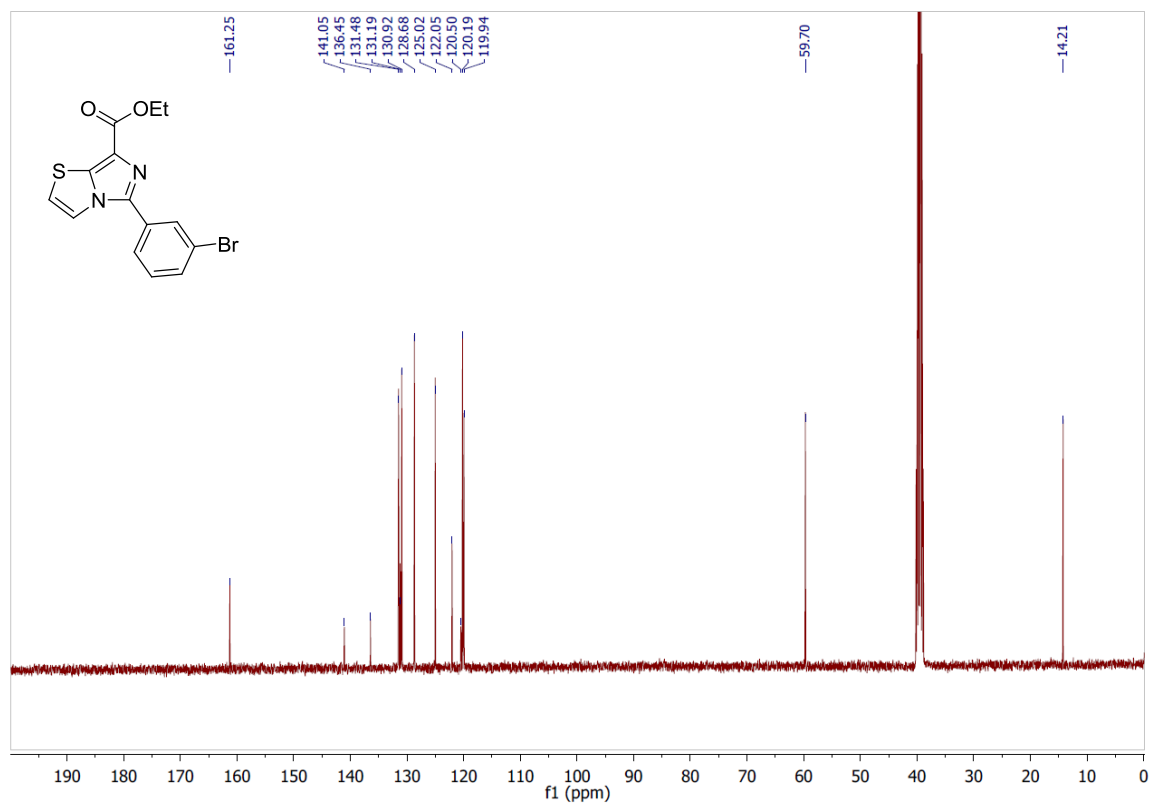
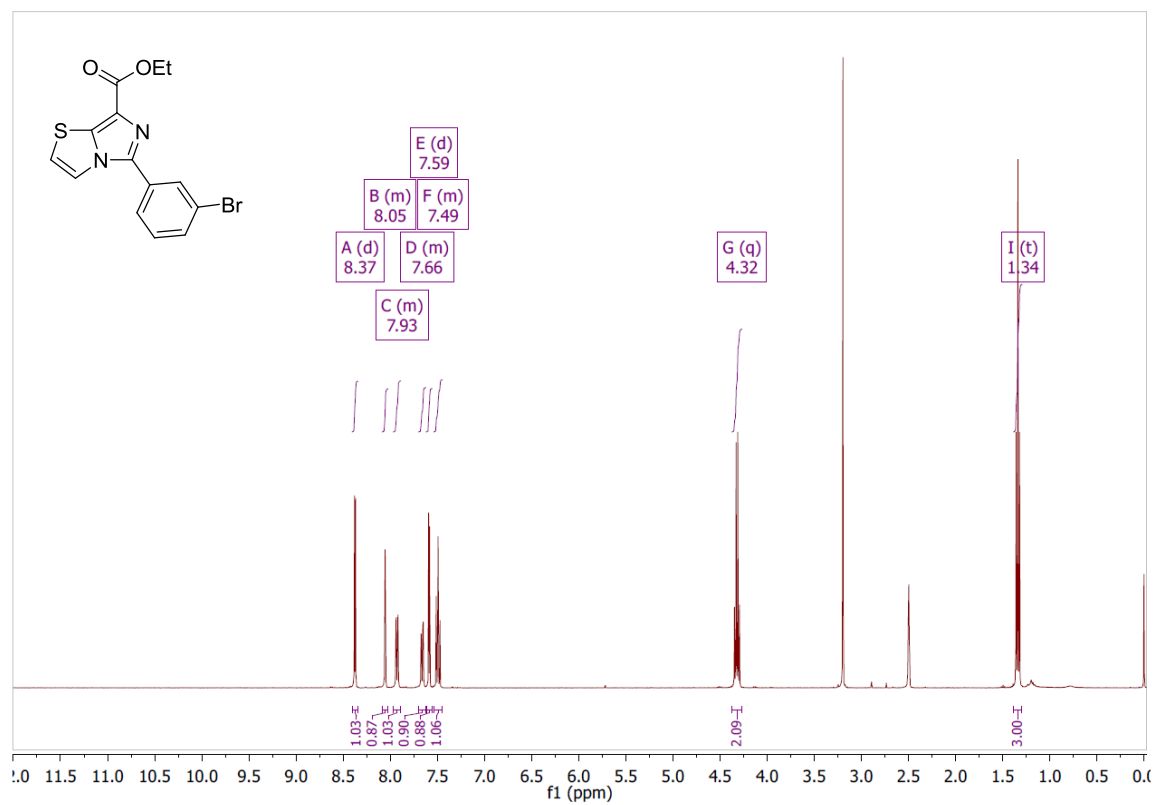
## Starting Material *1p*



## Starting Material 1q

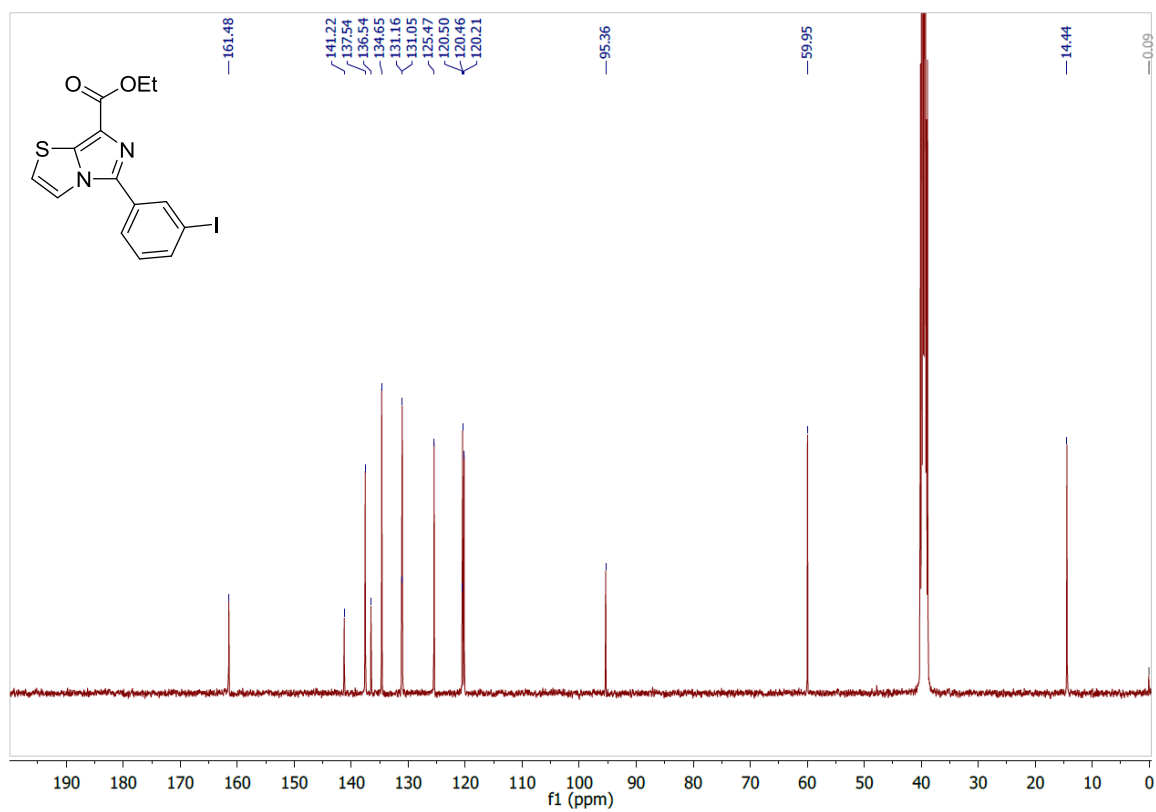
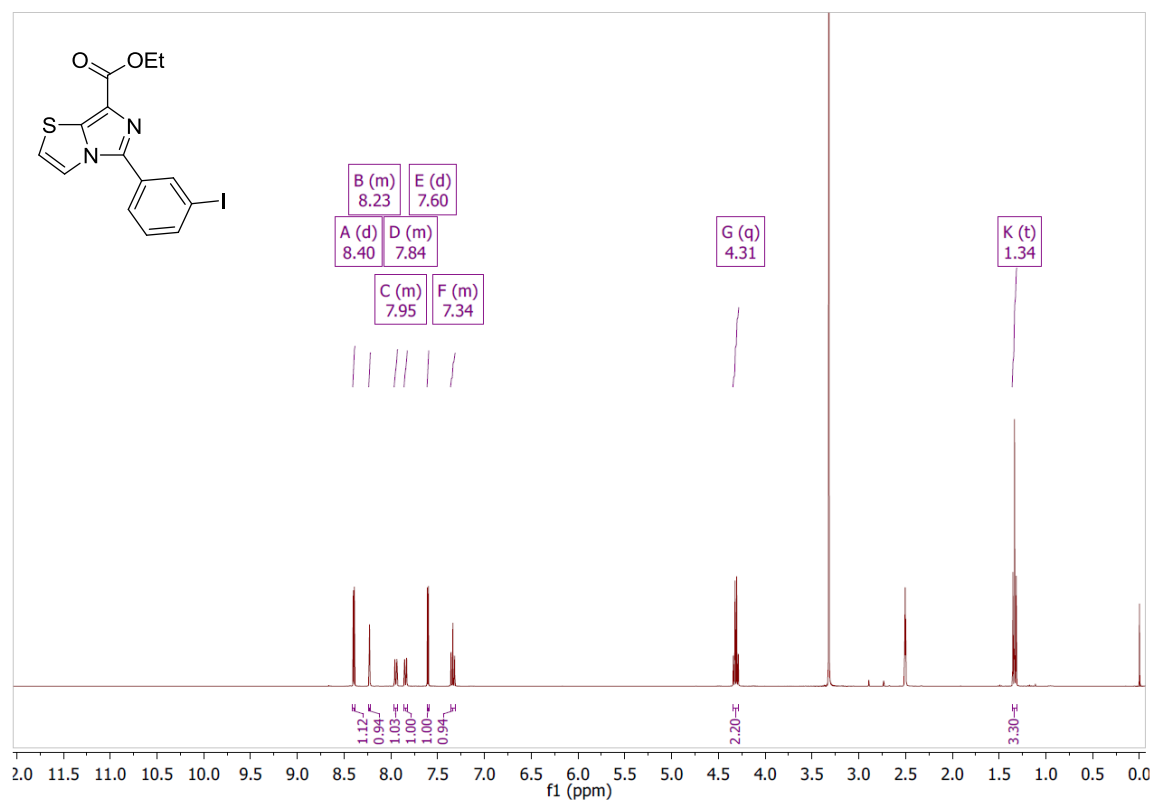


# Compound 3a

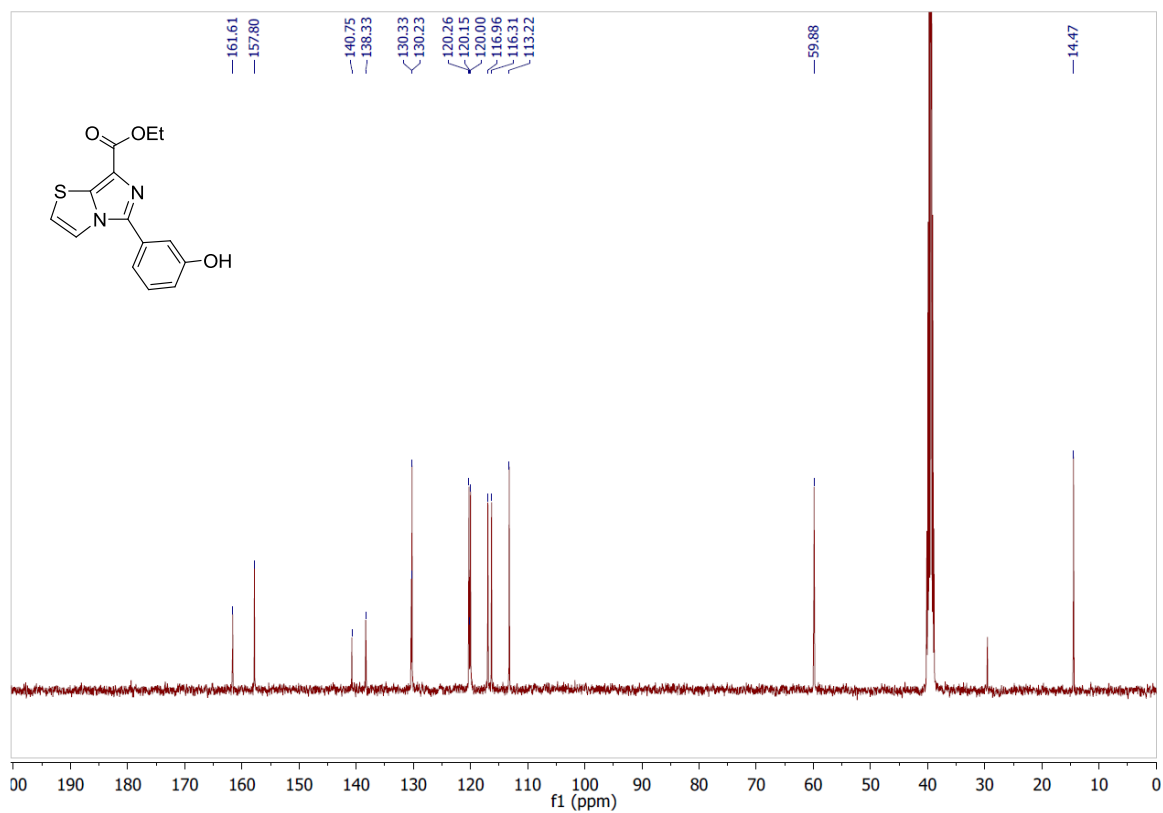
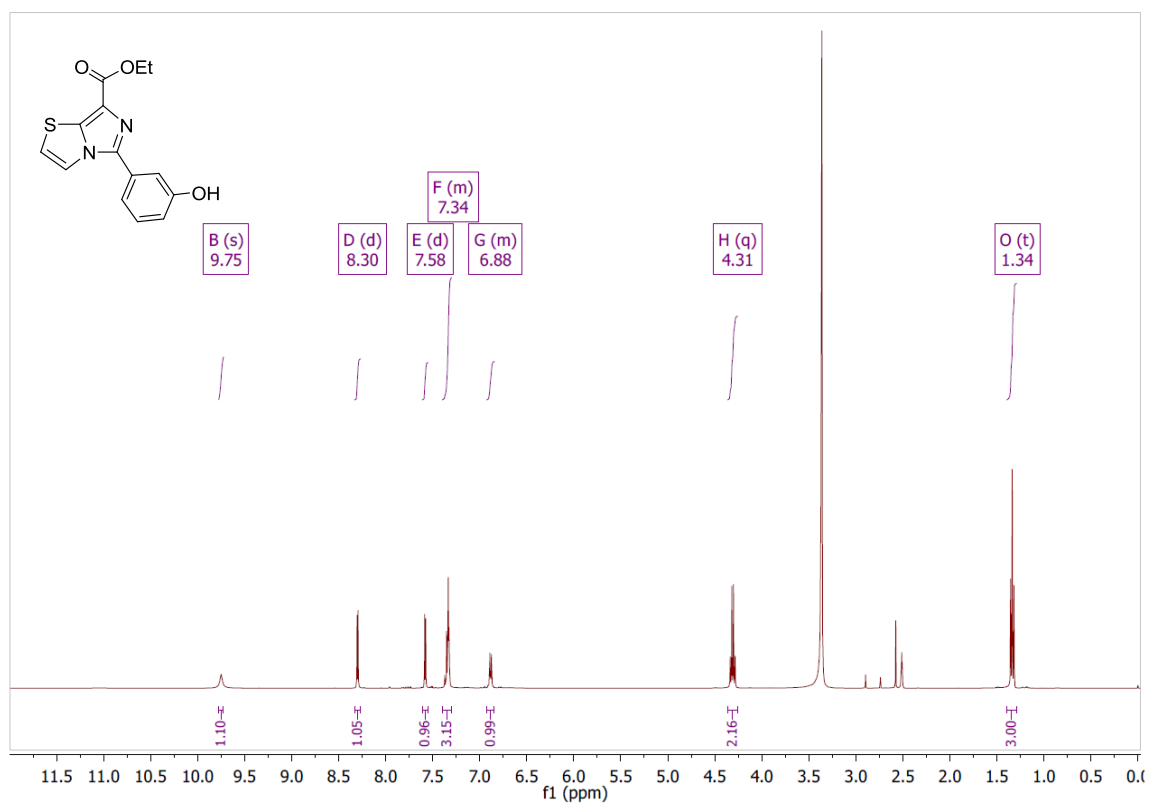




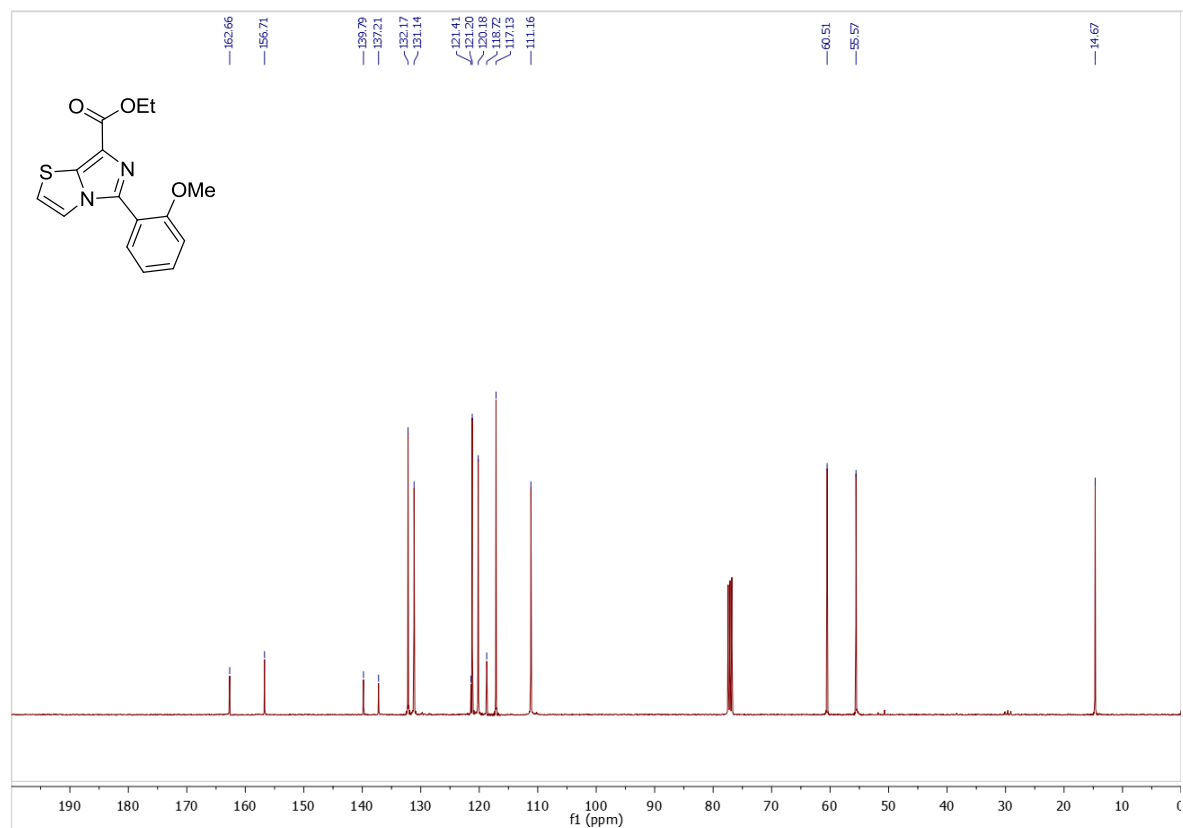
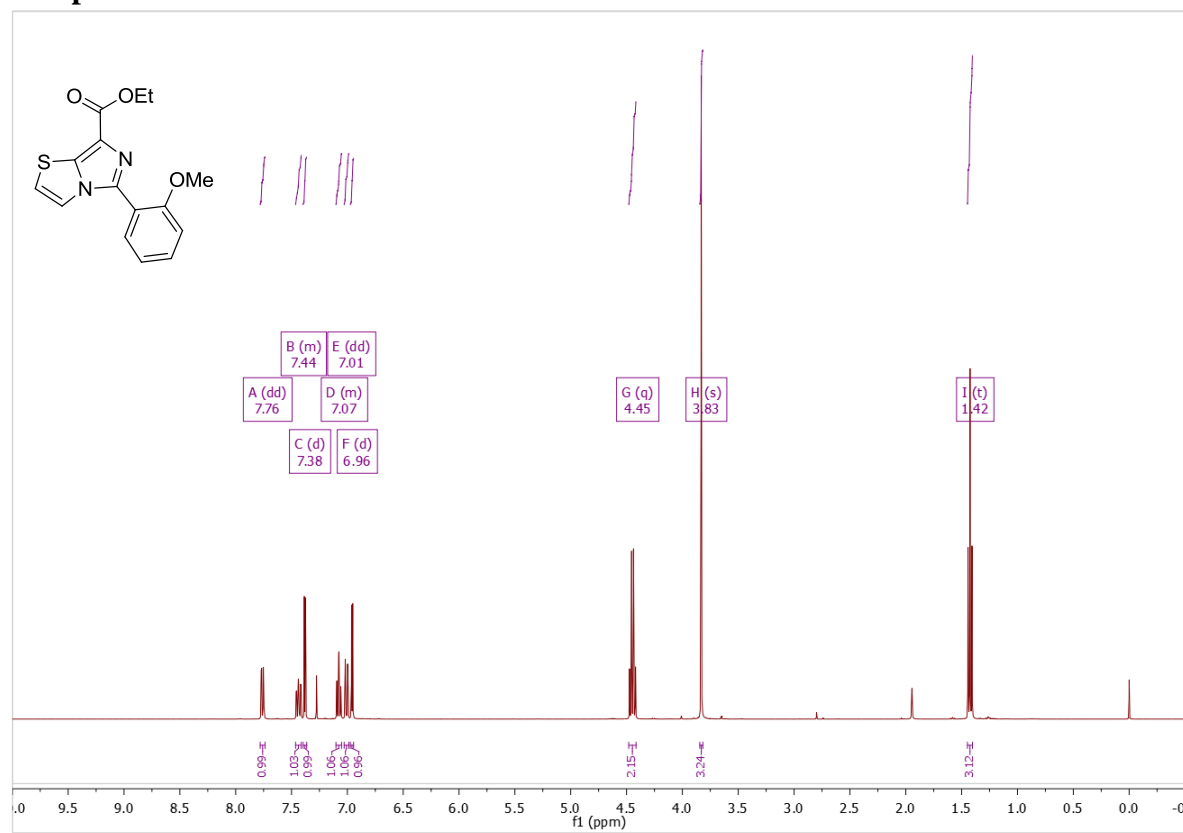
# Compound 3b



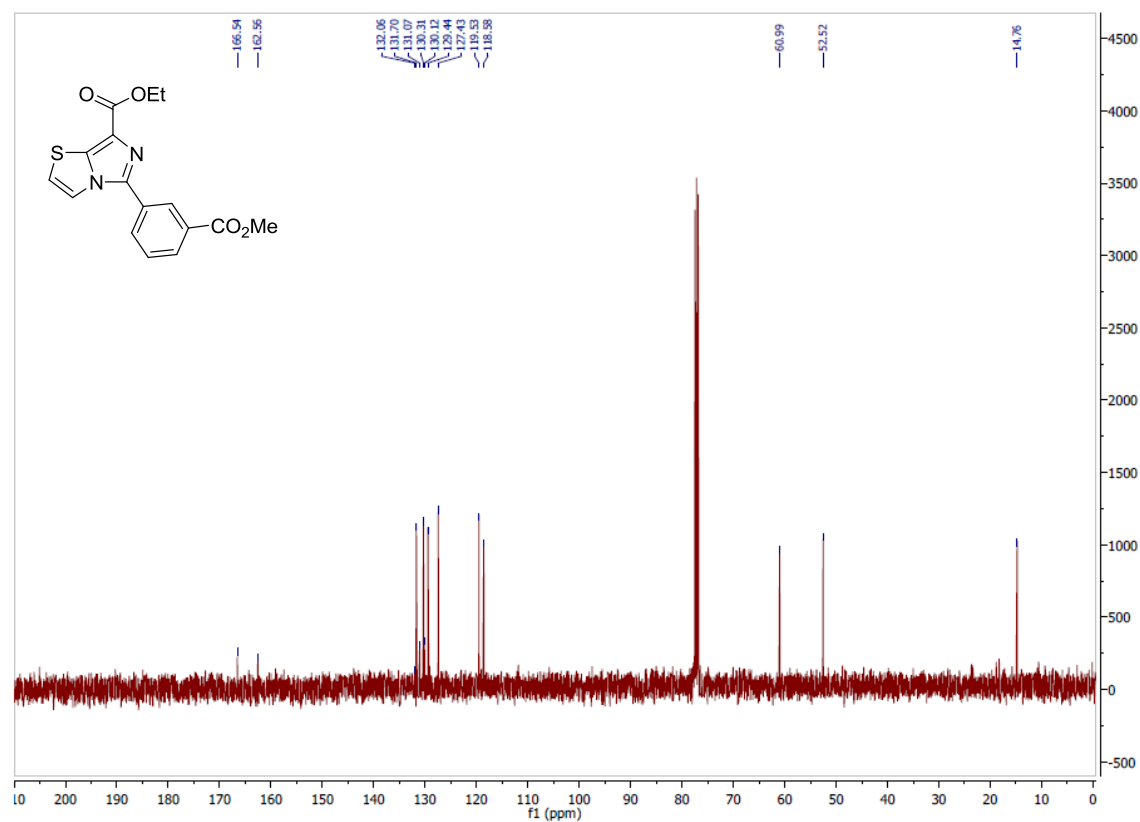
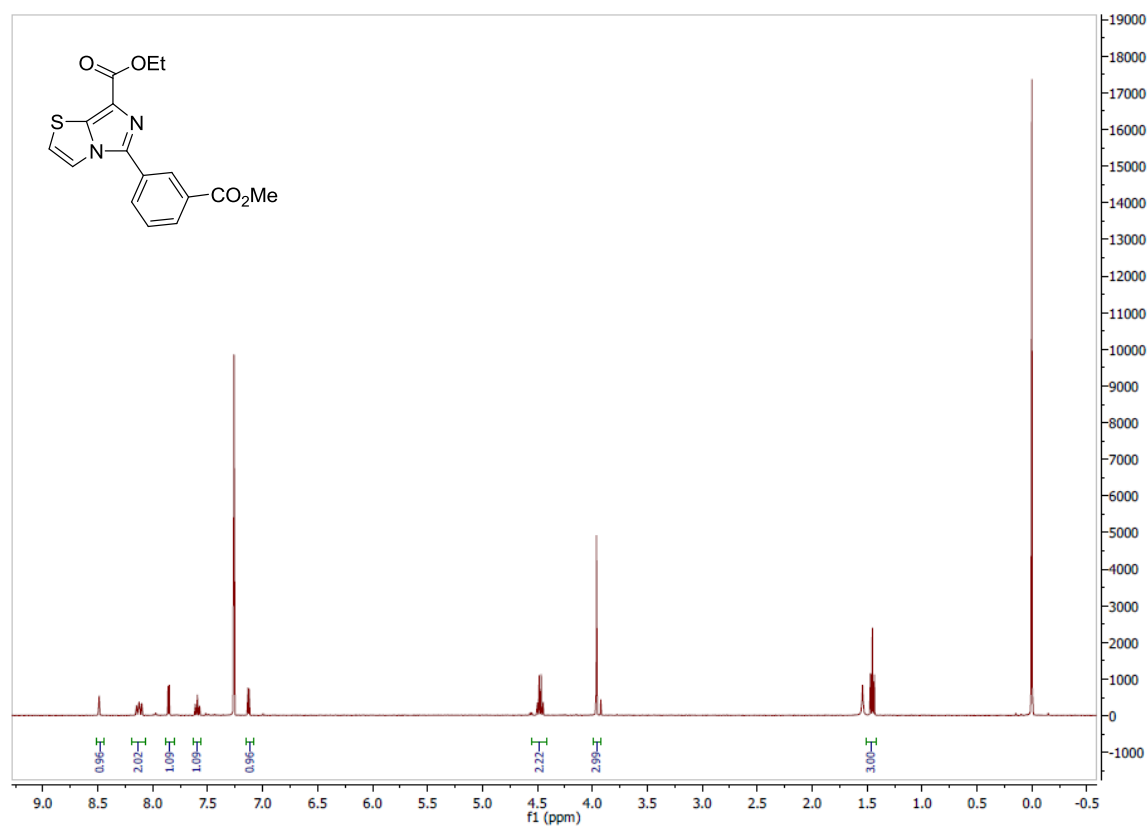
# Compound 3c



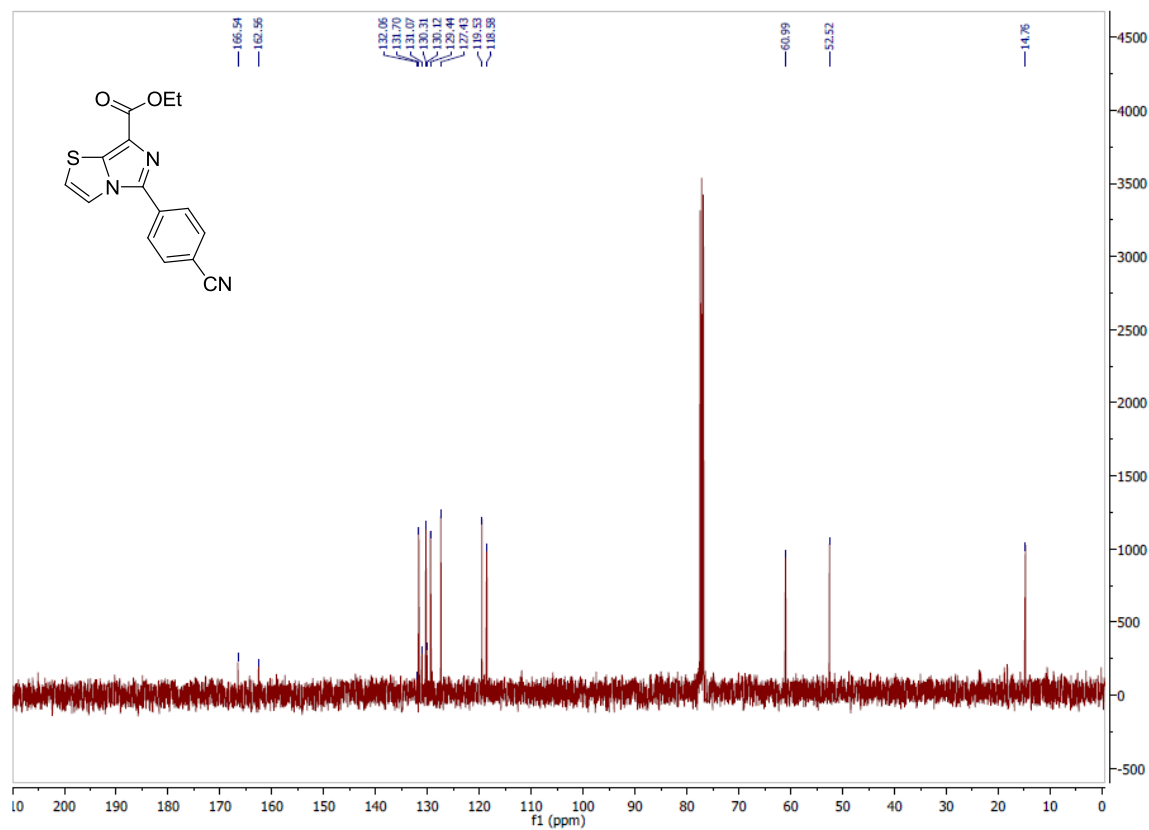
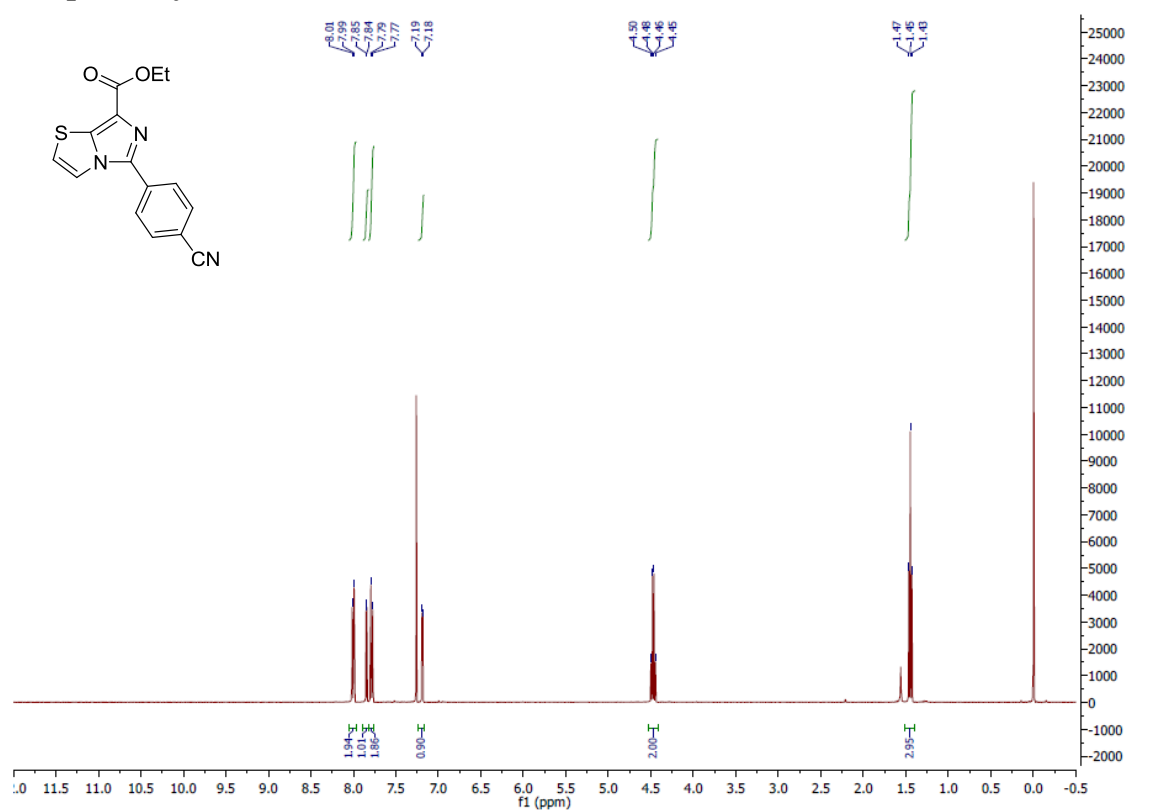
# Compound 3d



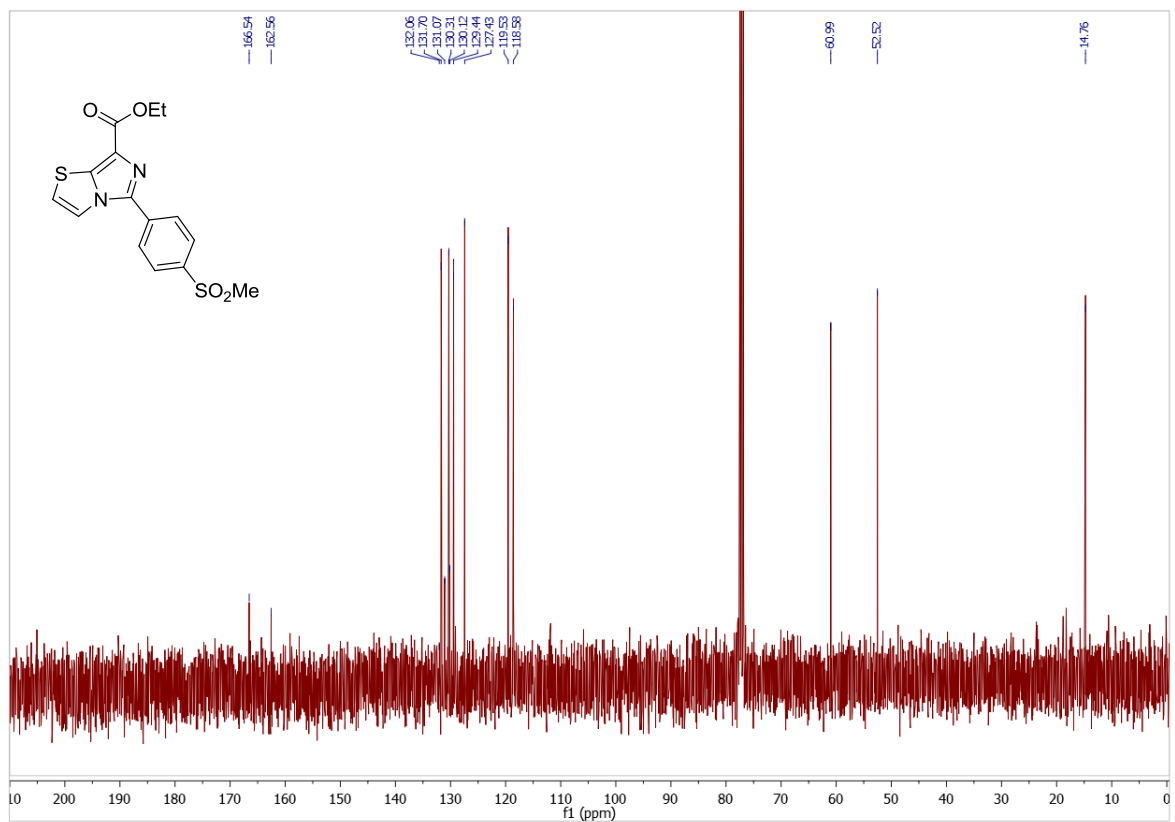
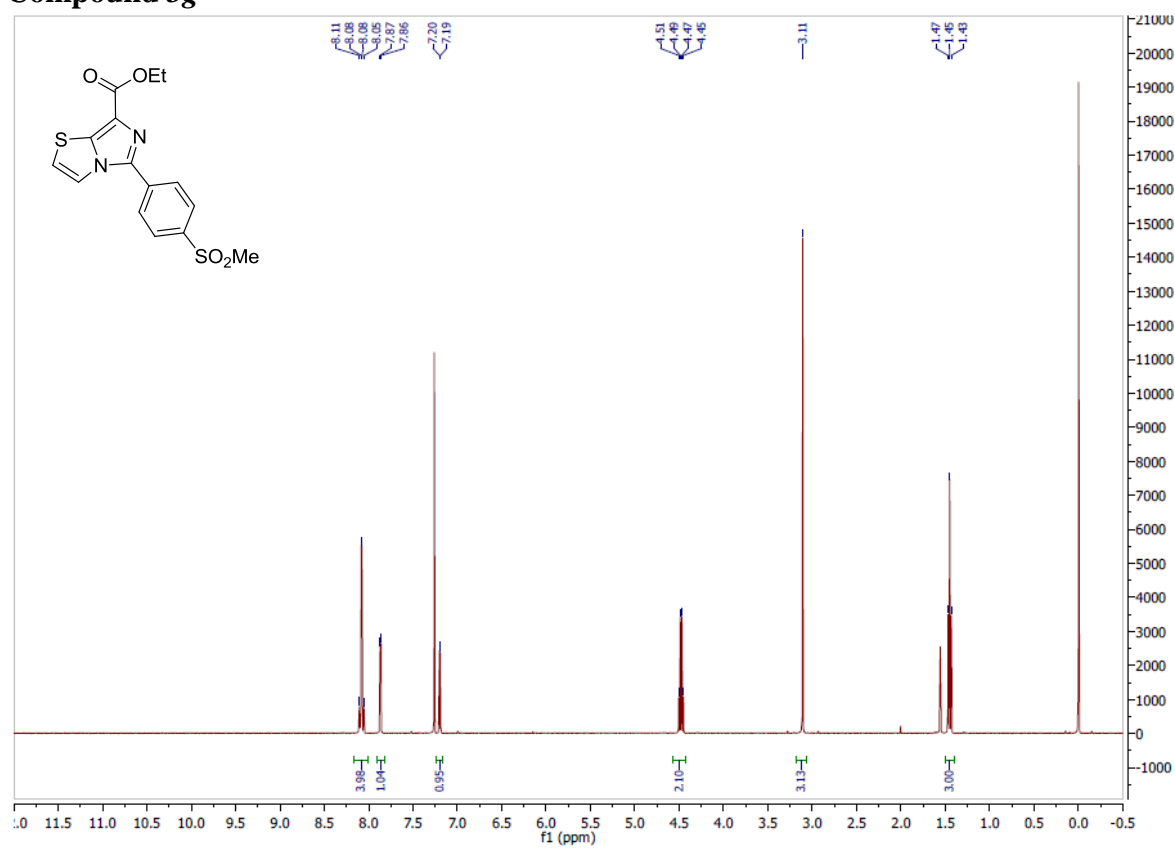
# Compound 3e



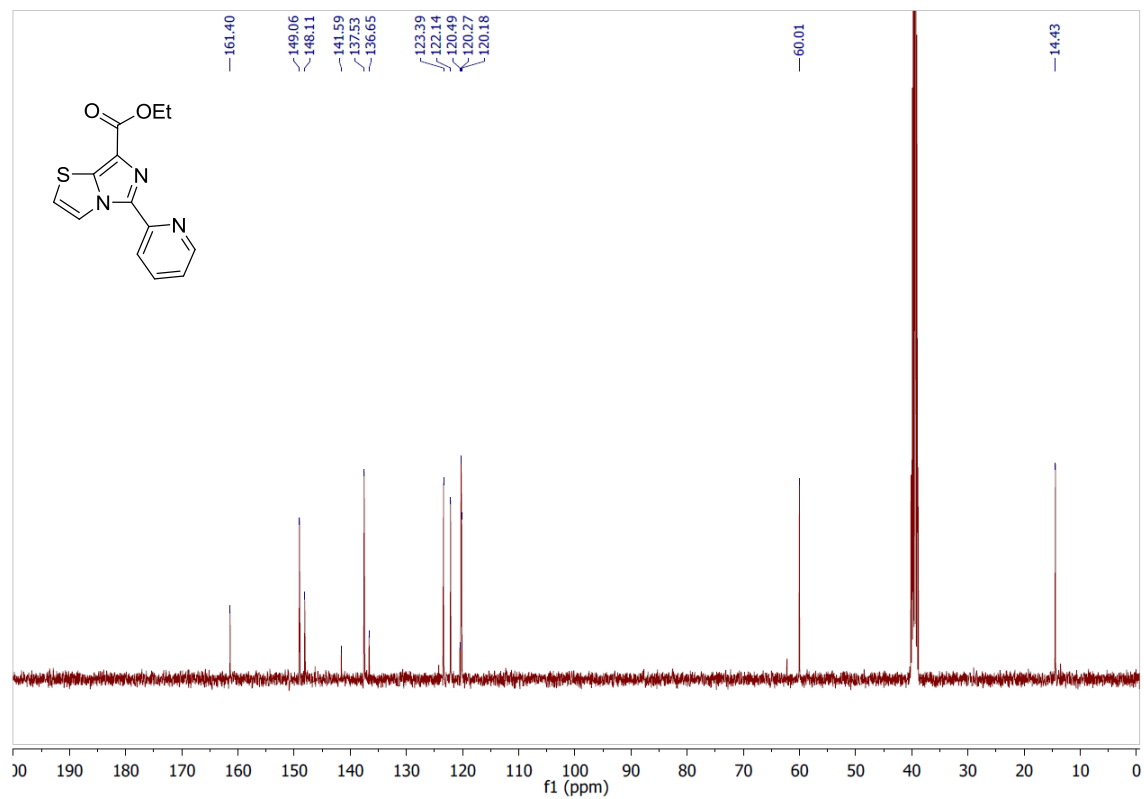
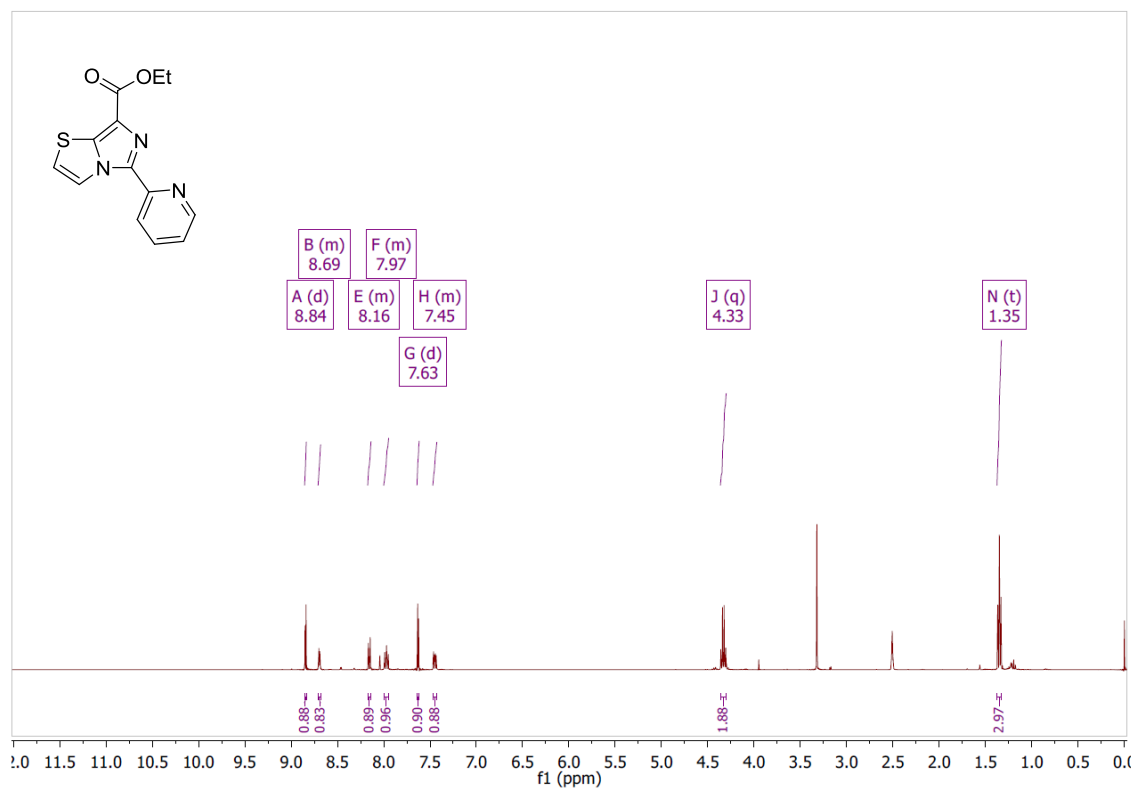
# Compound 3f



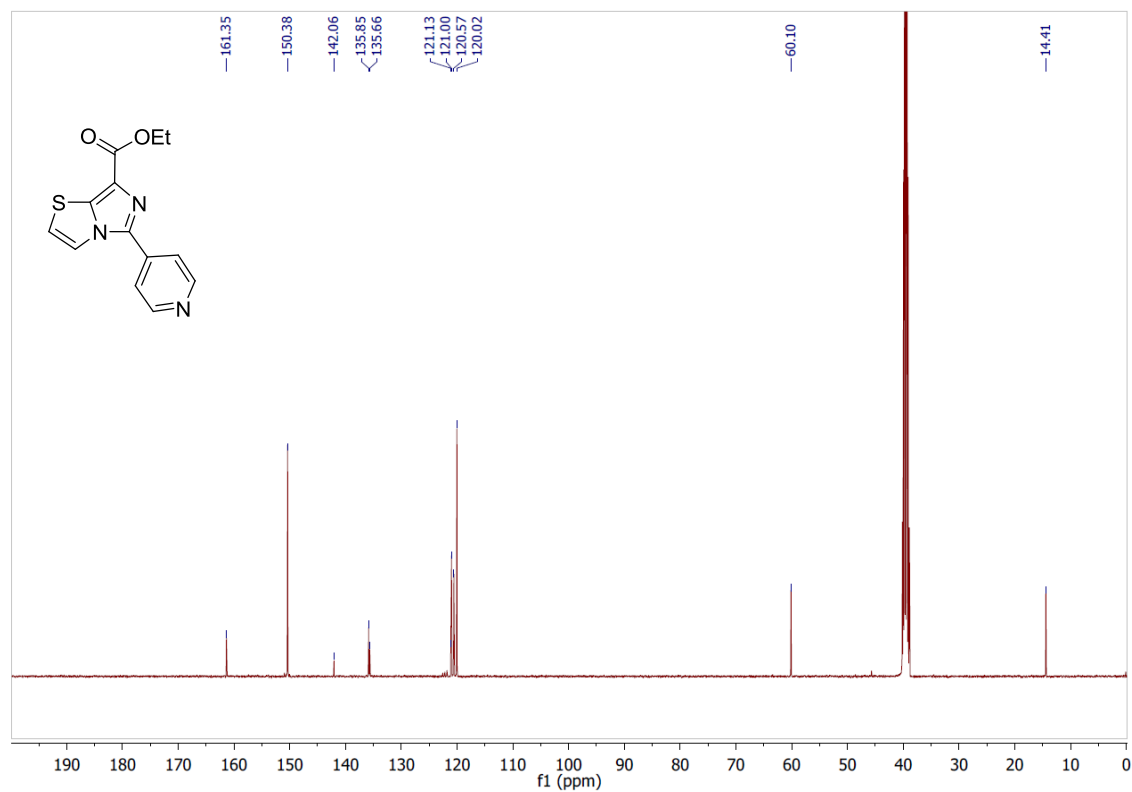
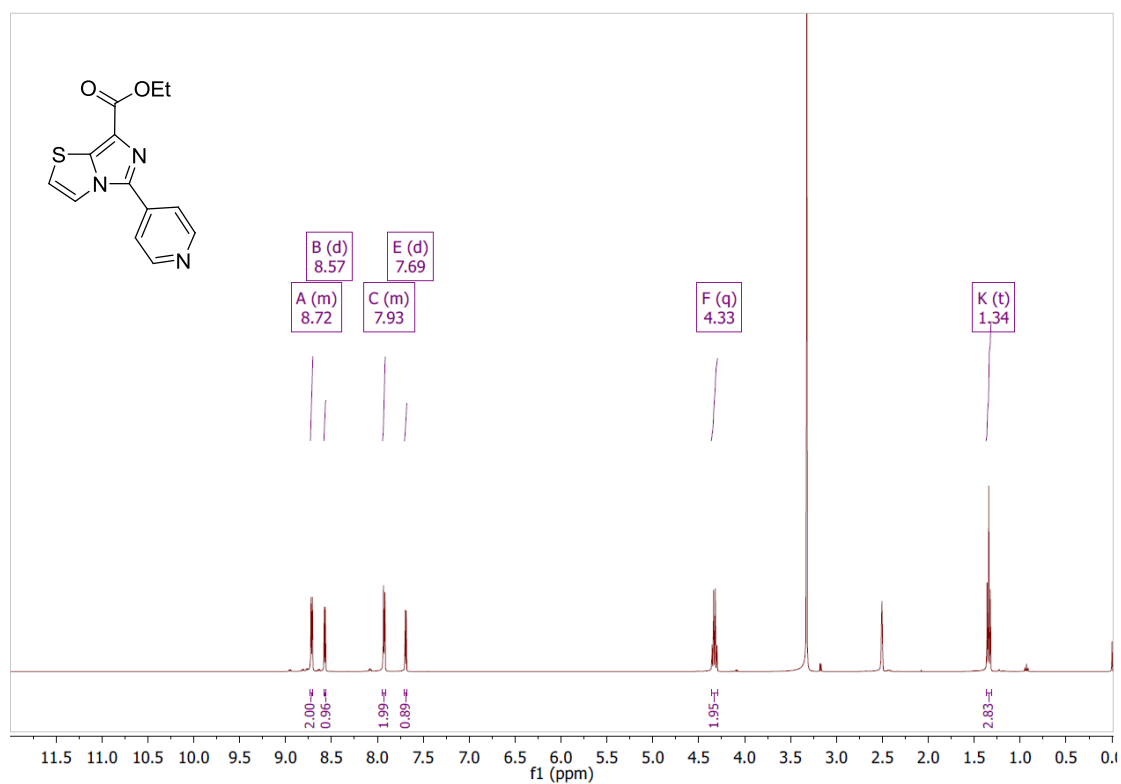
# Compound 3g



# Compound 3h

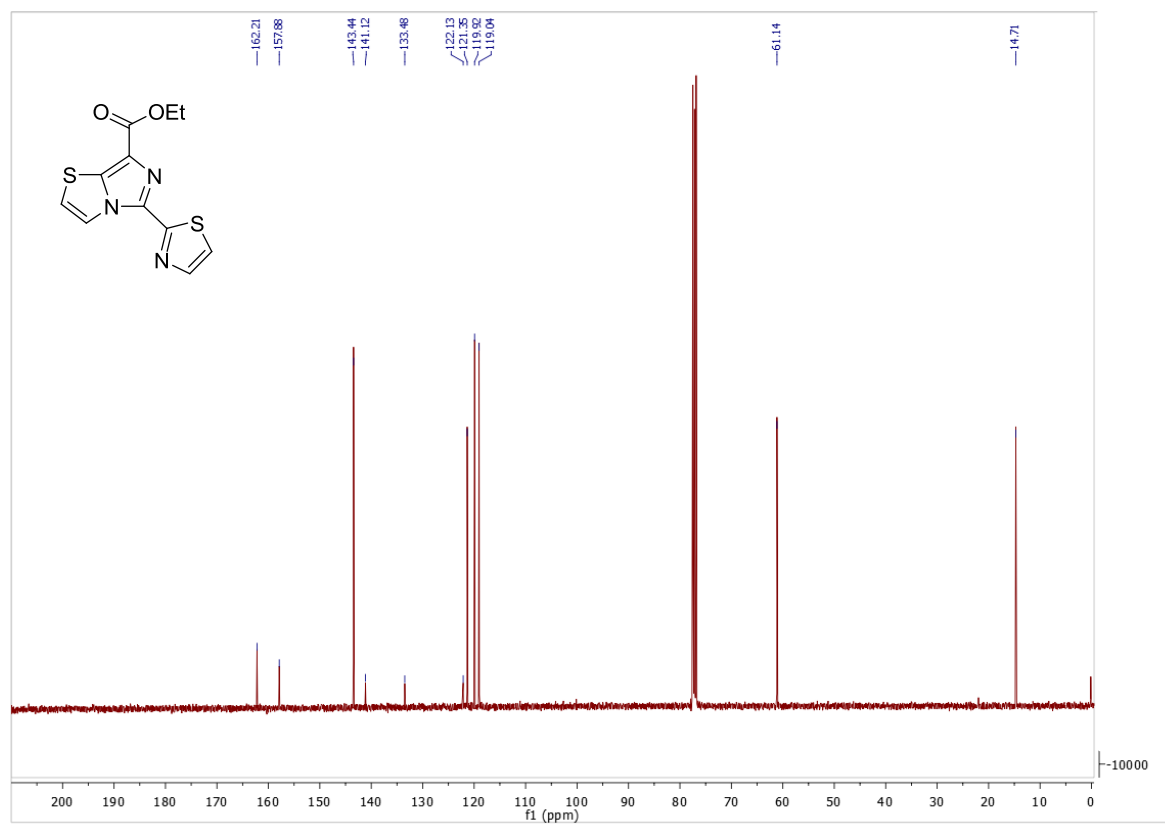
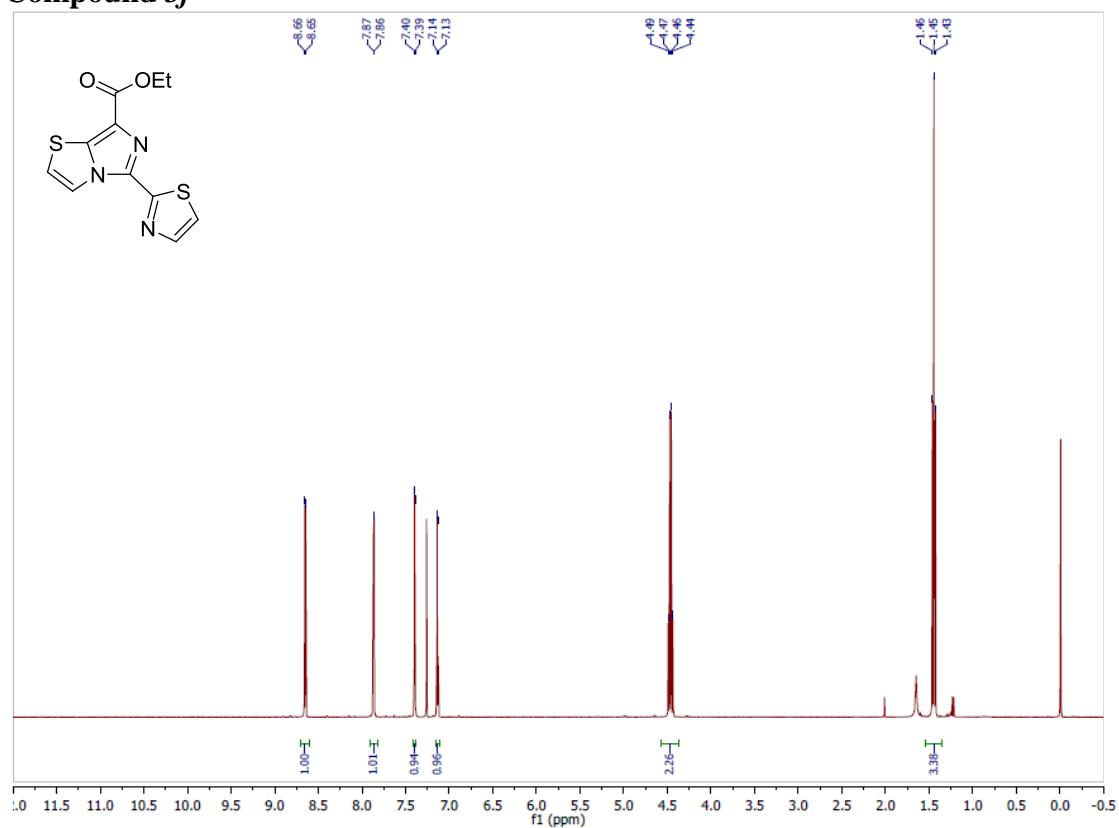


# Compound 3i

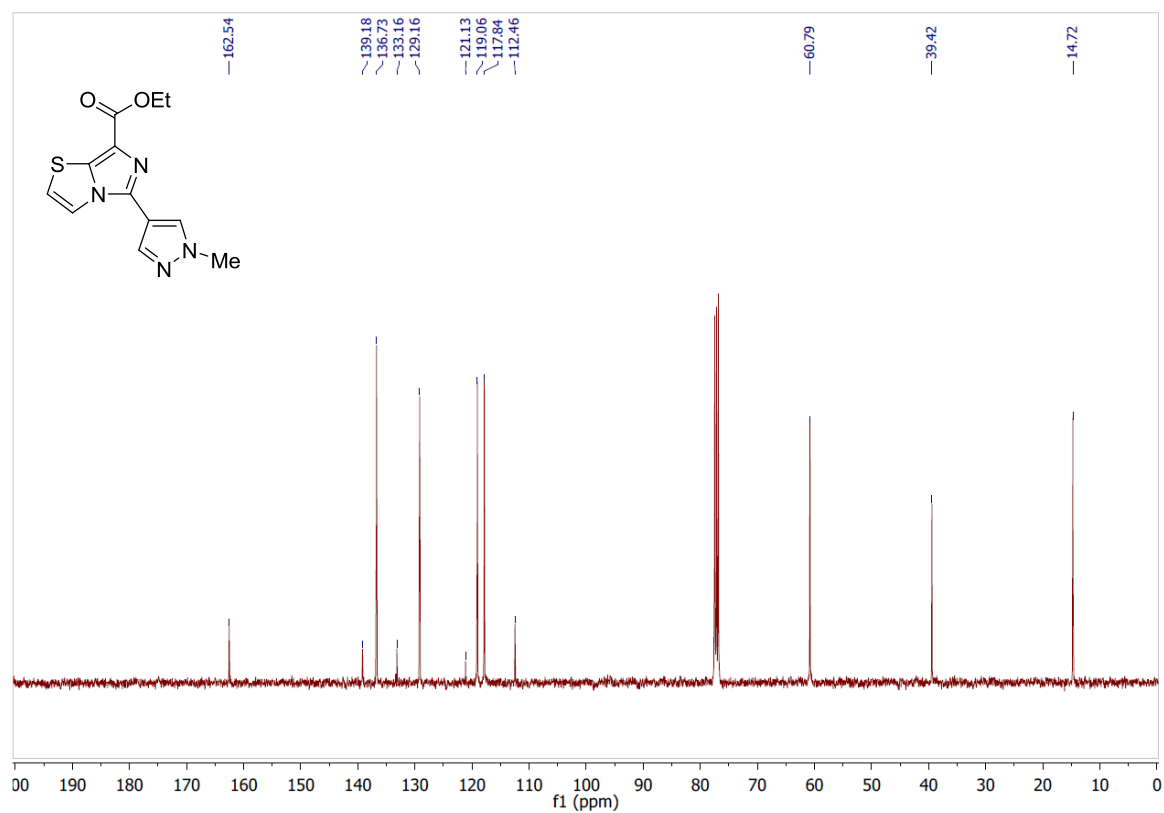
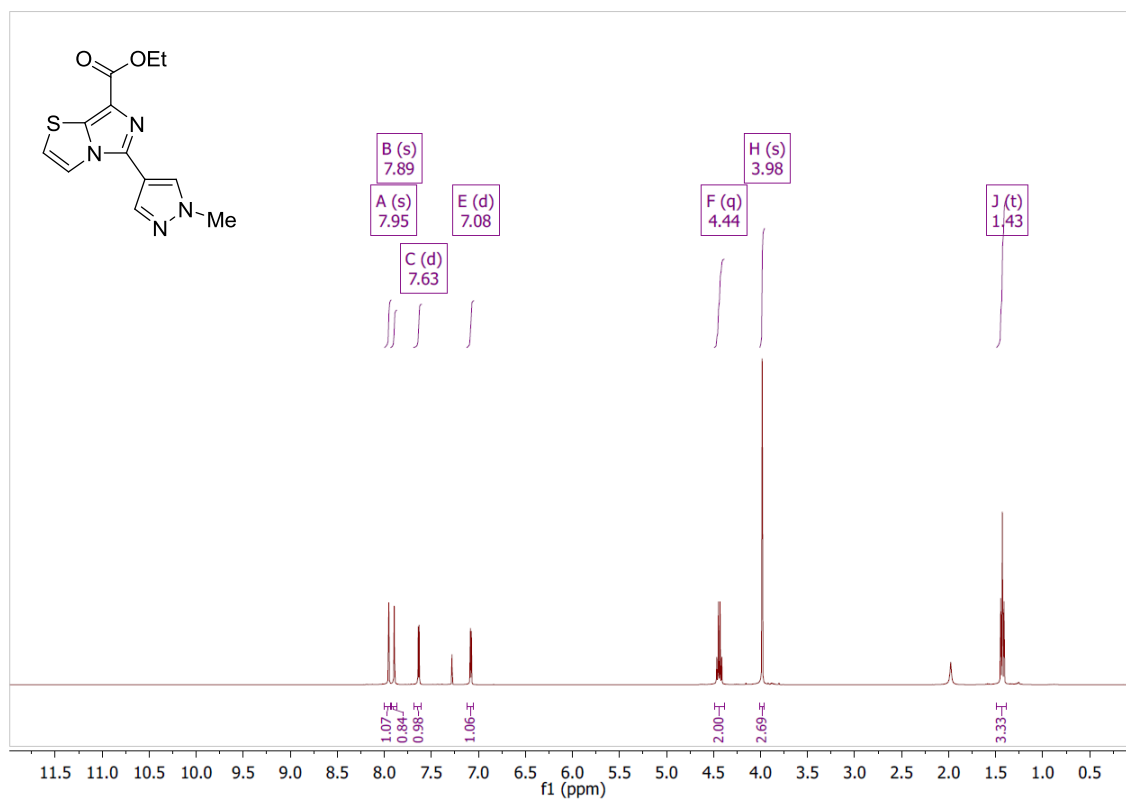




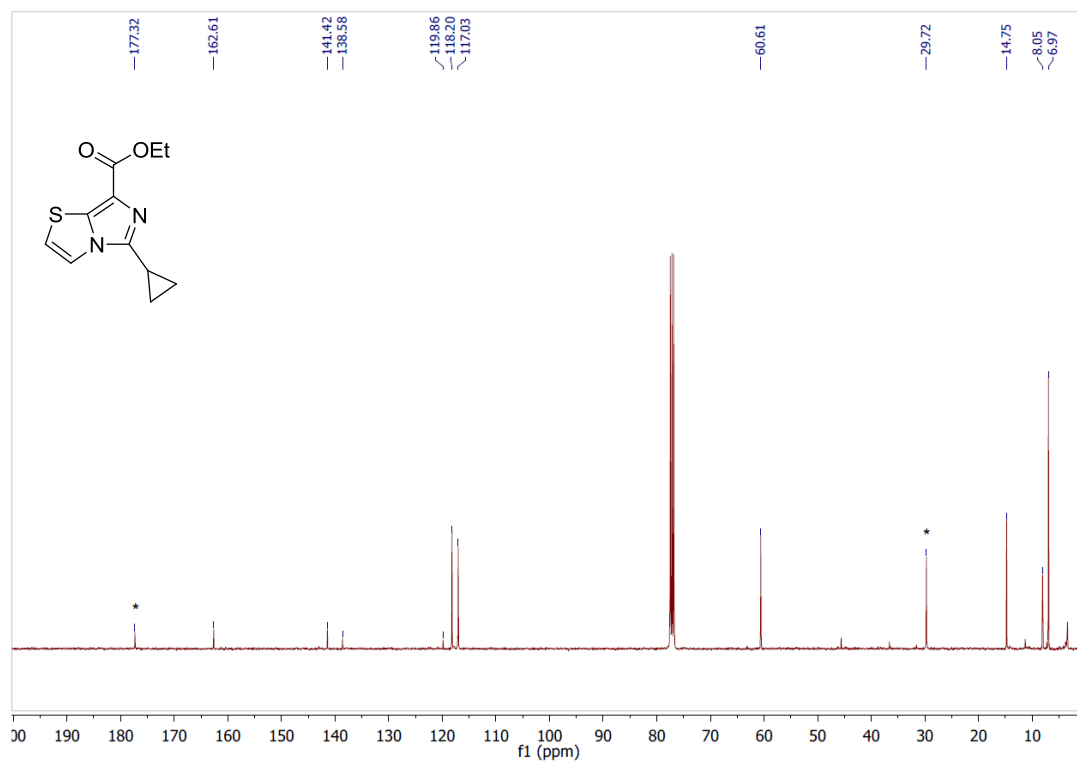
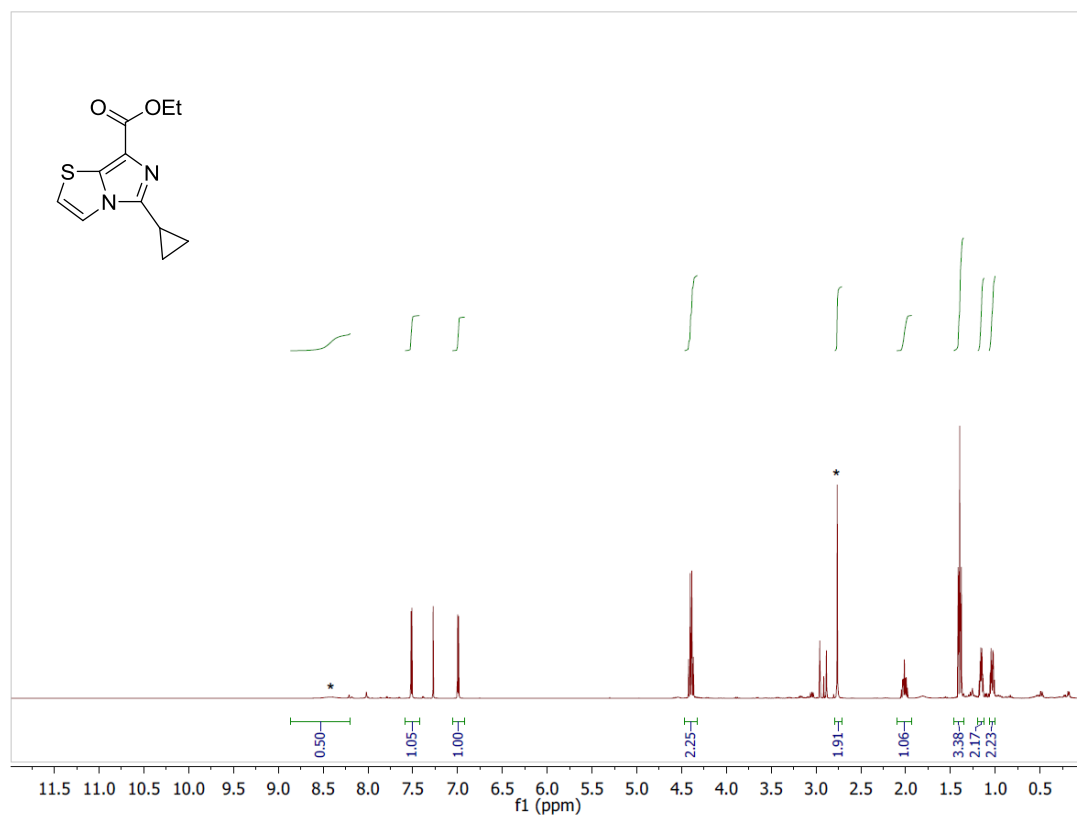
# Compound 3j



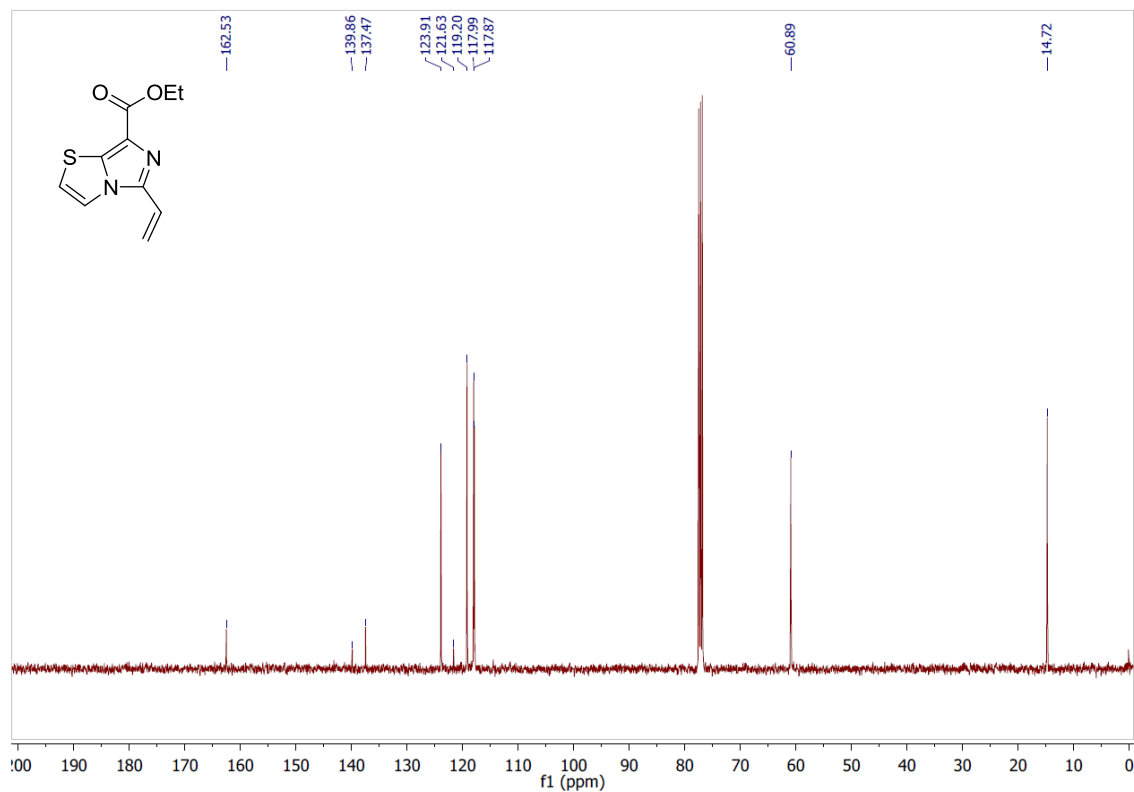
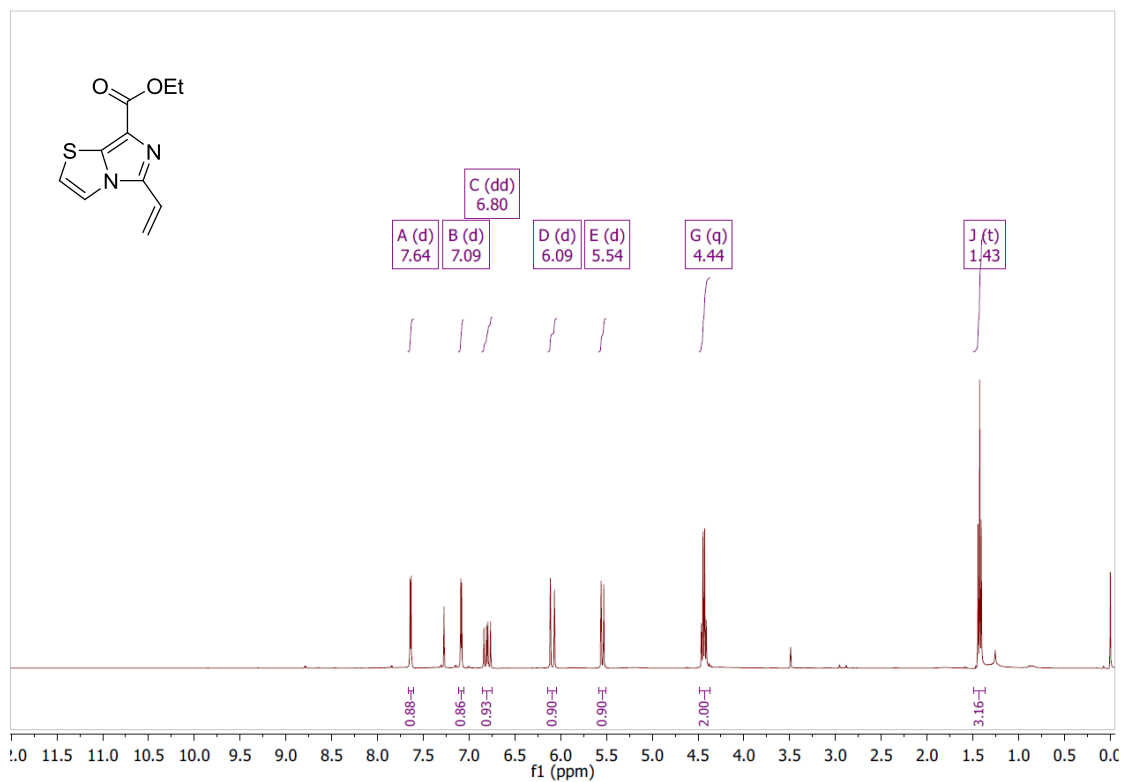
# Compound 3k



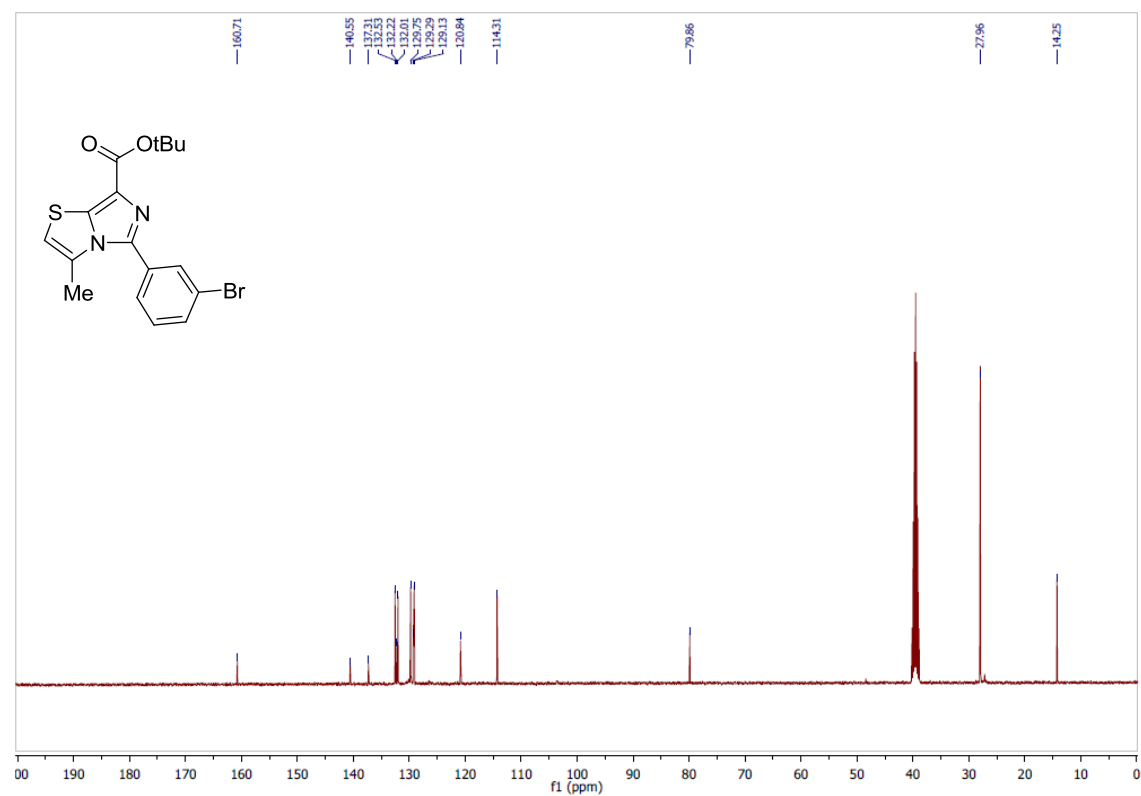
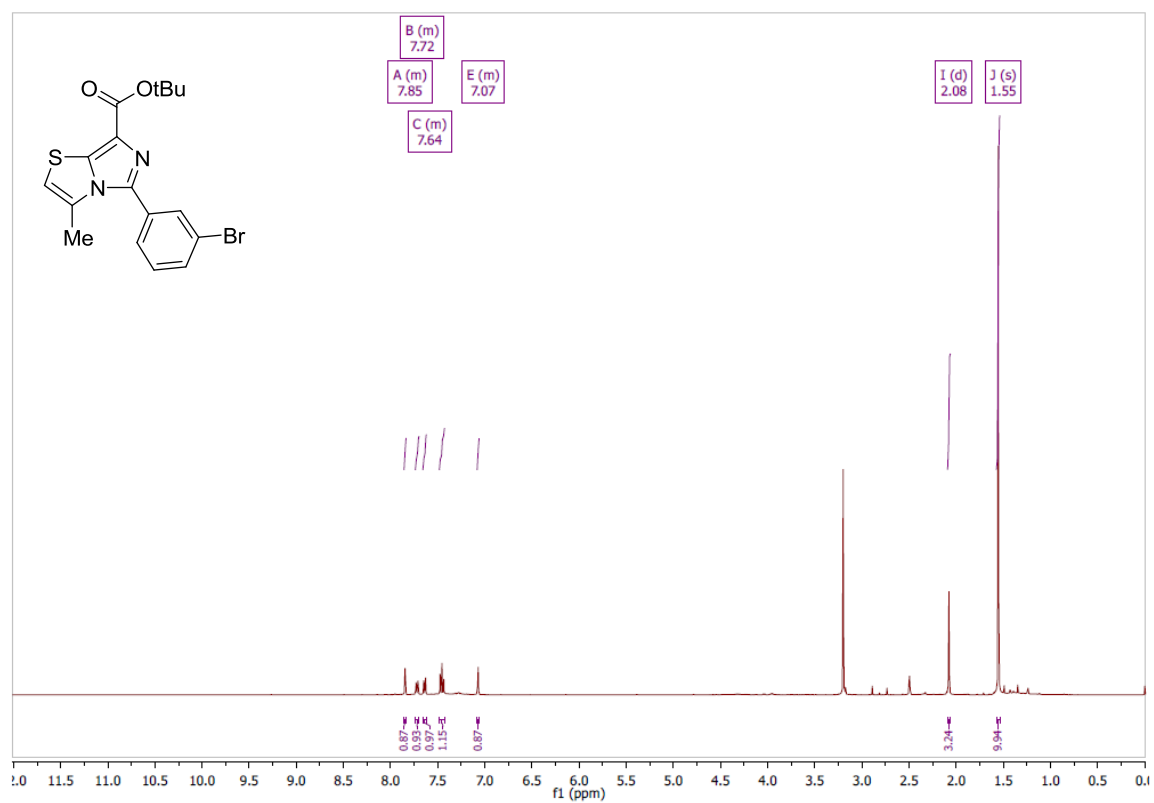
# Compound 3l



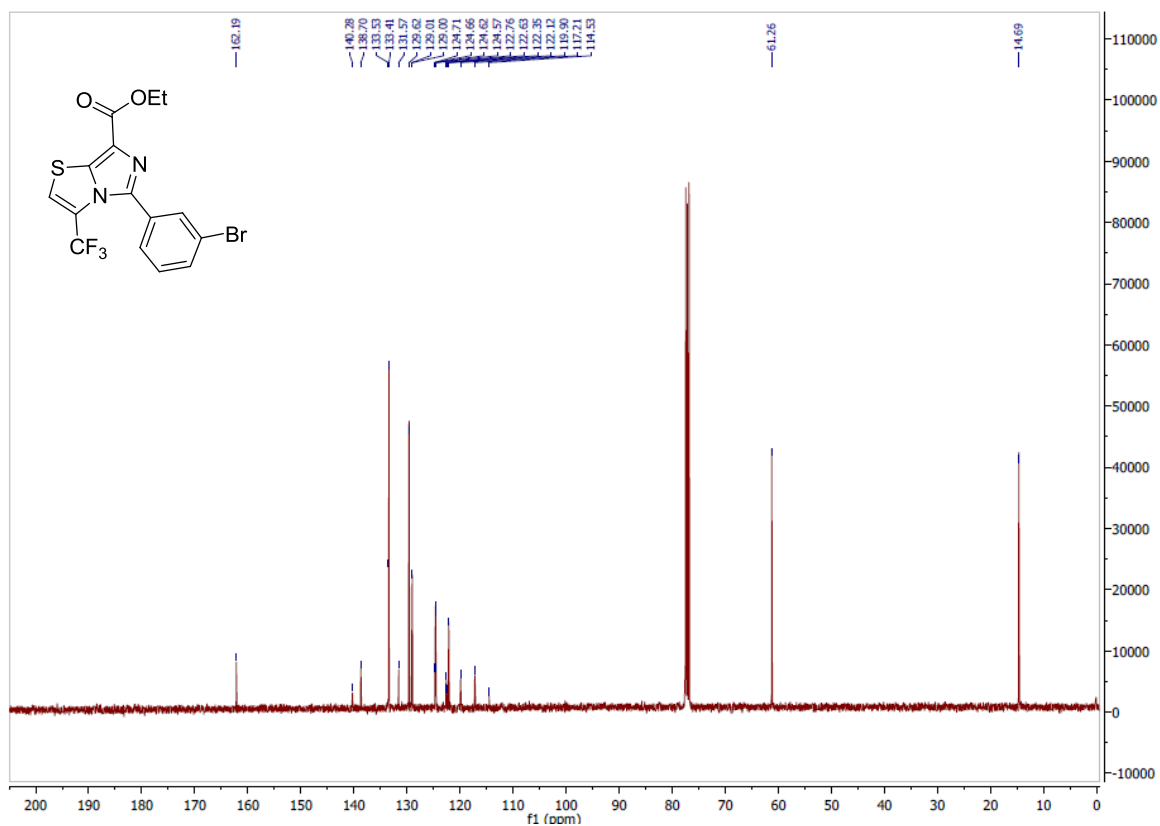
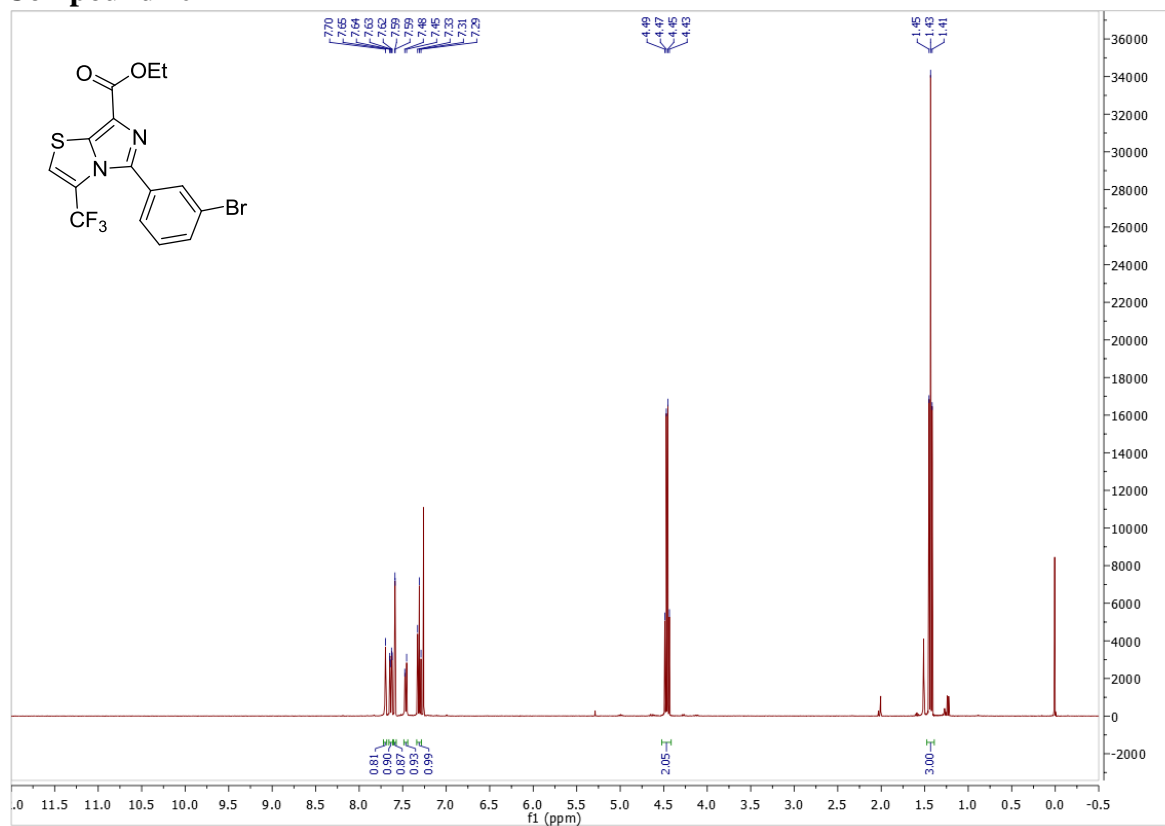
# Compound 3m



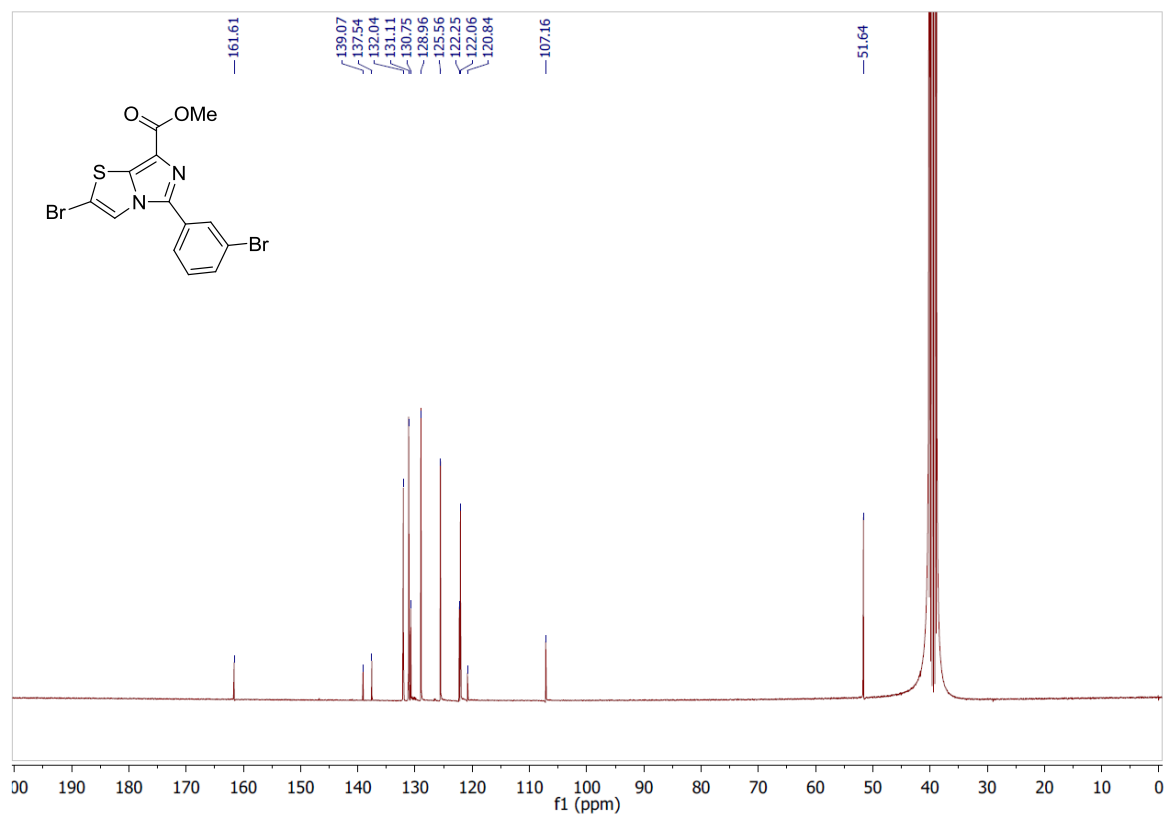
# Compound 4b



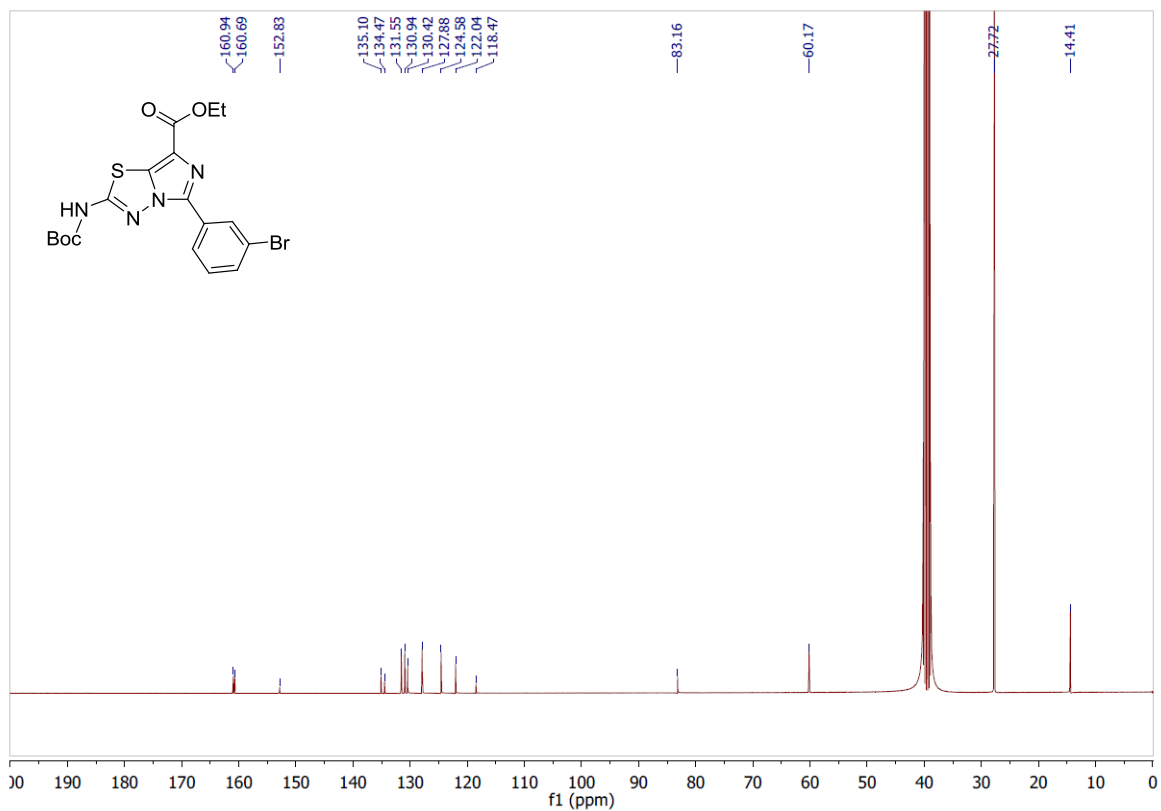
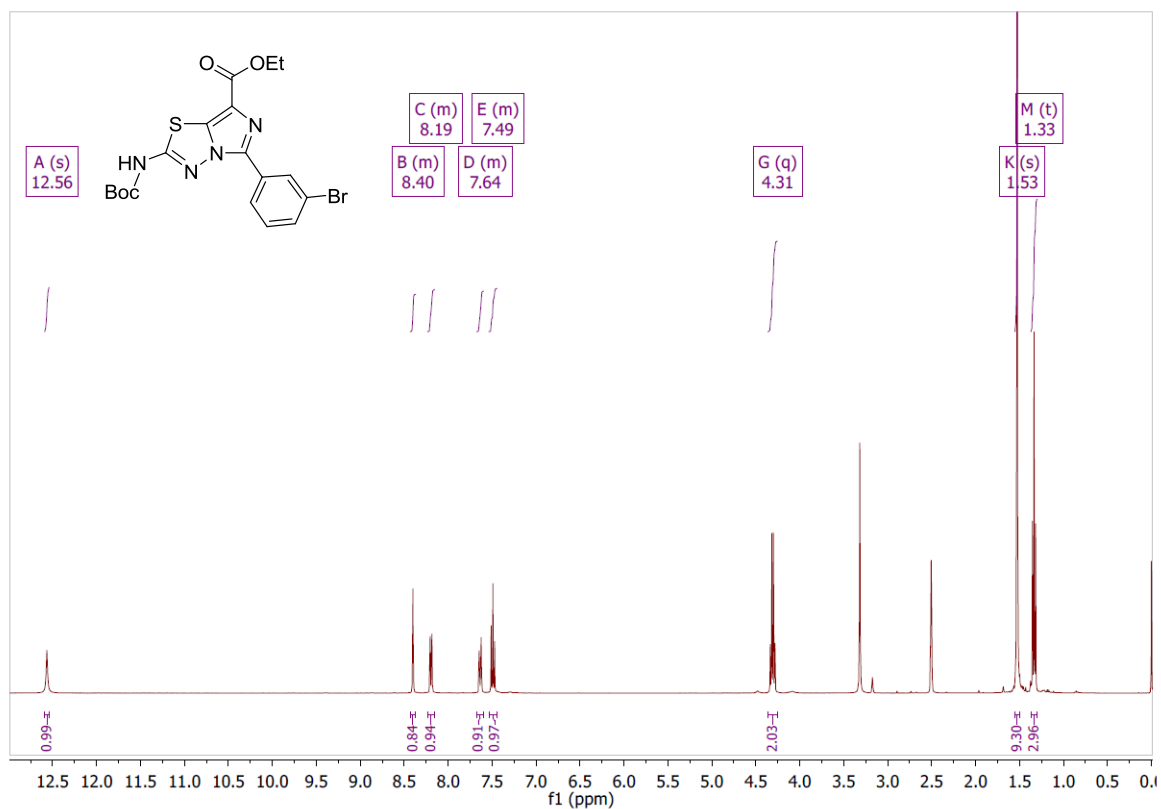
# Compound 4c



# Compound 4d

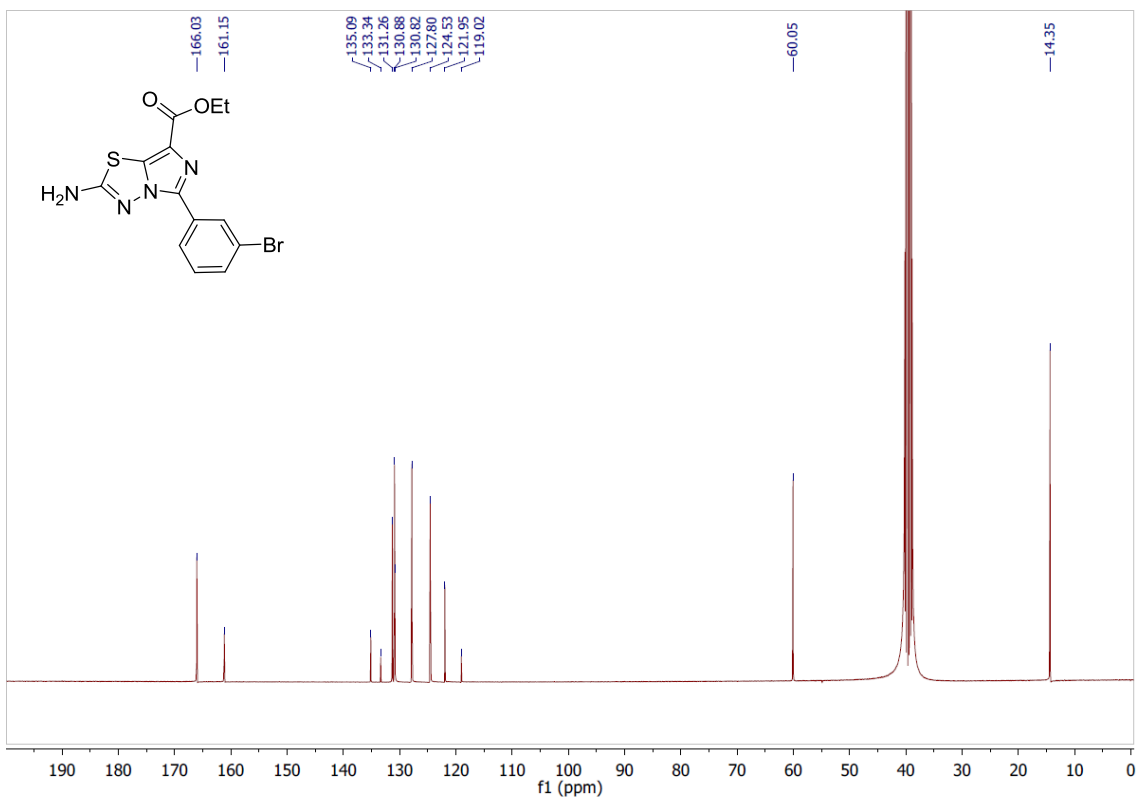
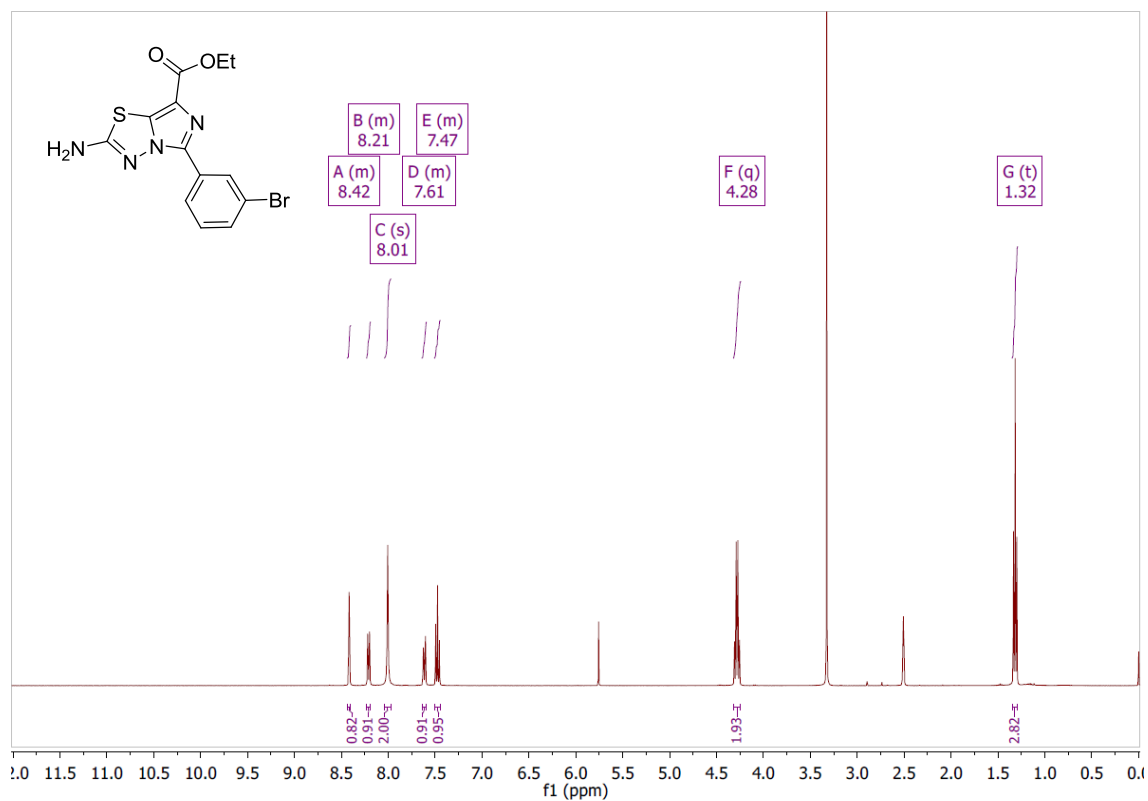


# Compound 4e

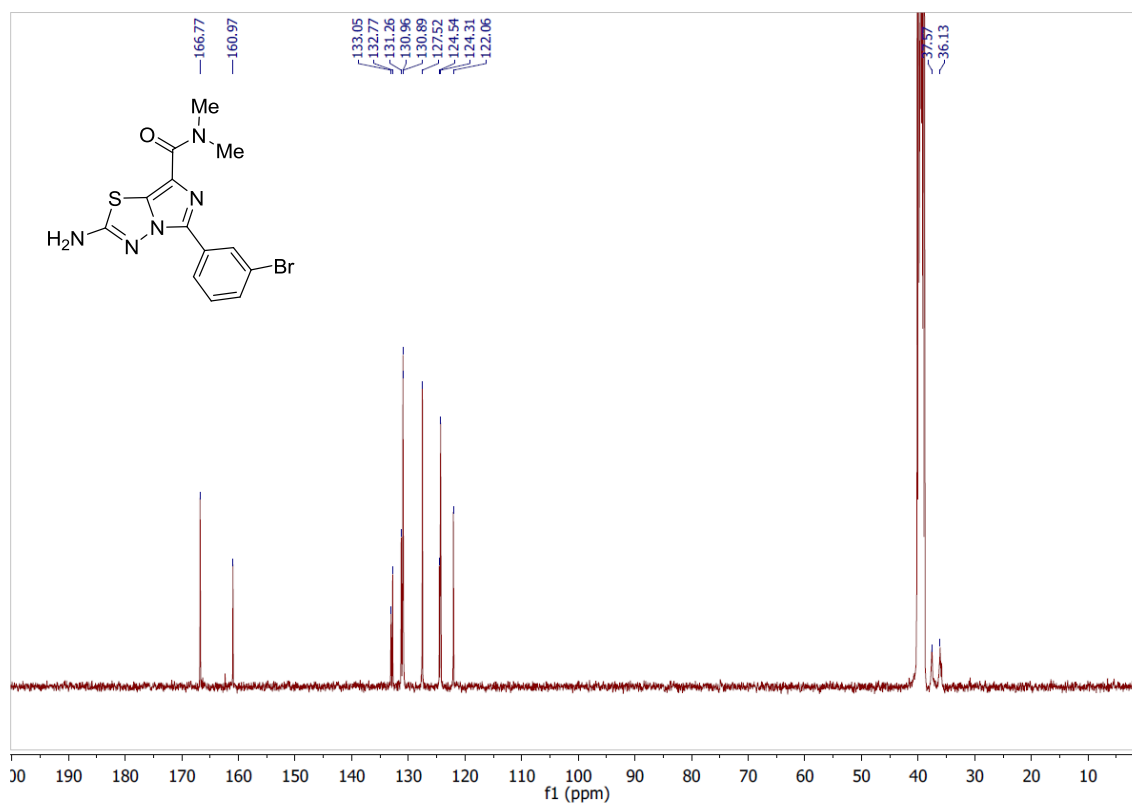
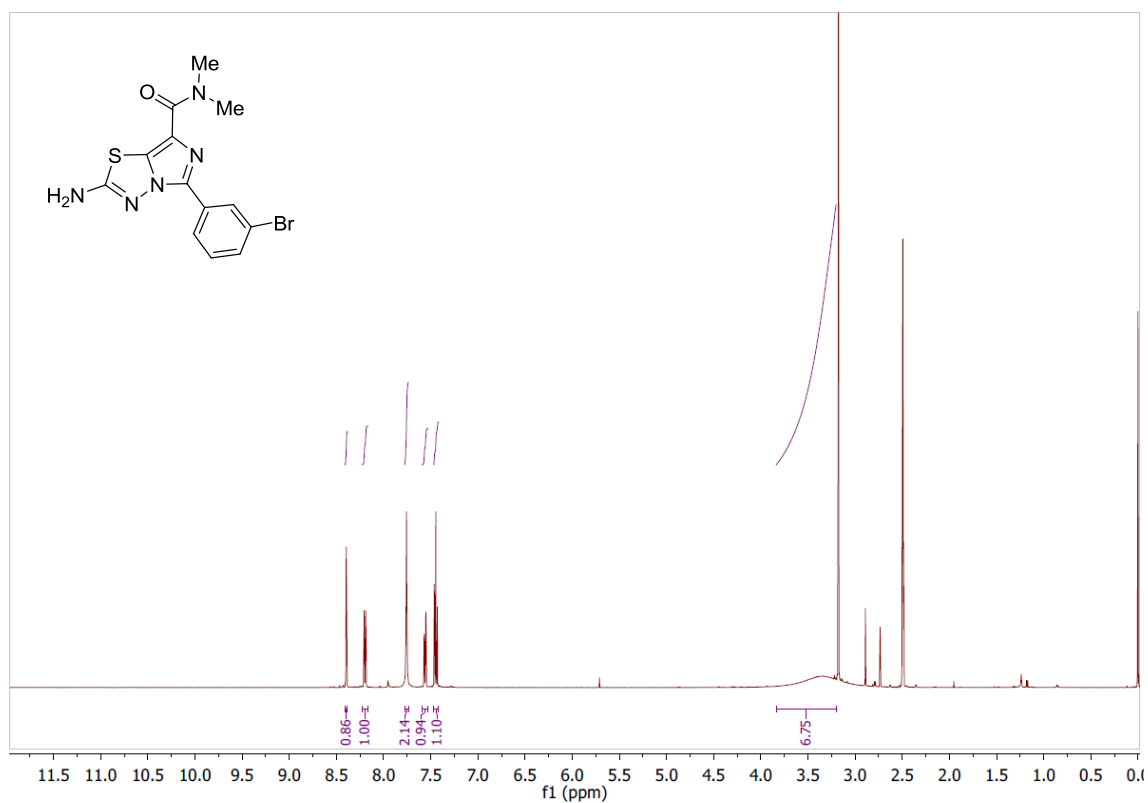




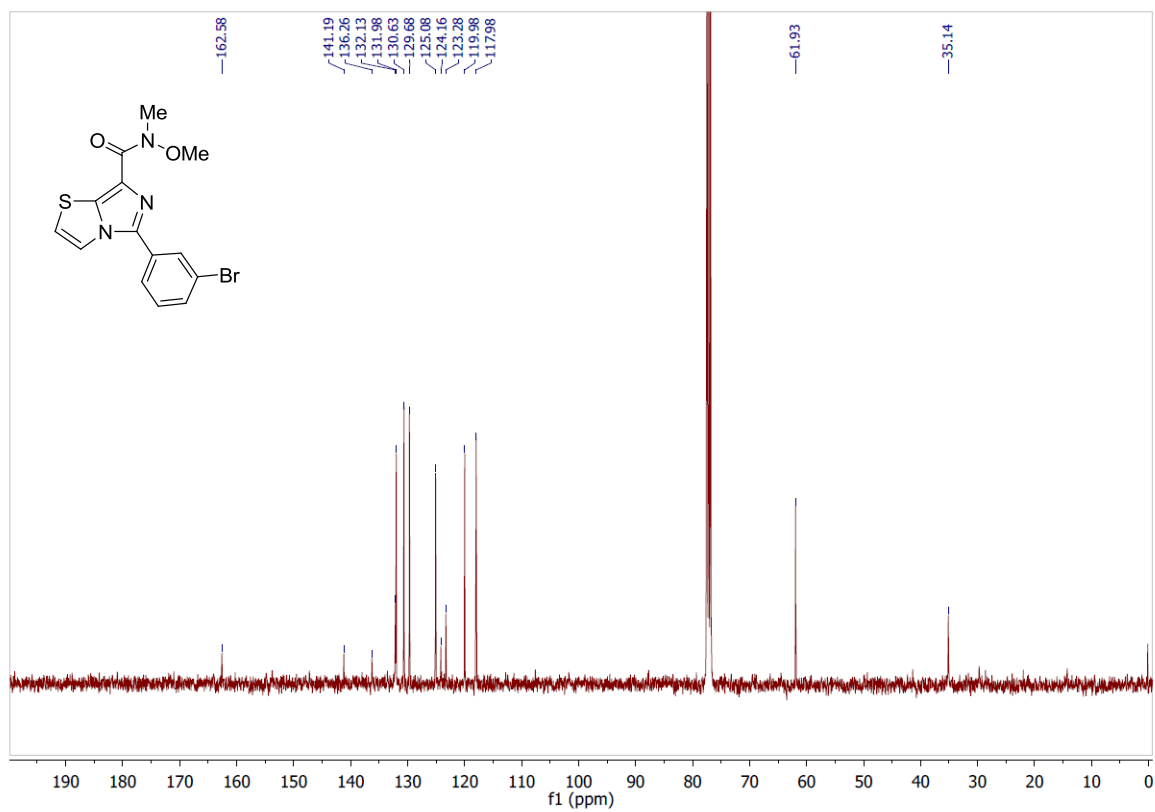
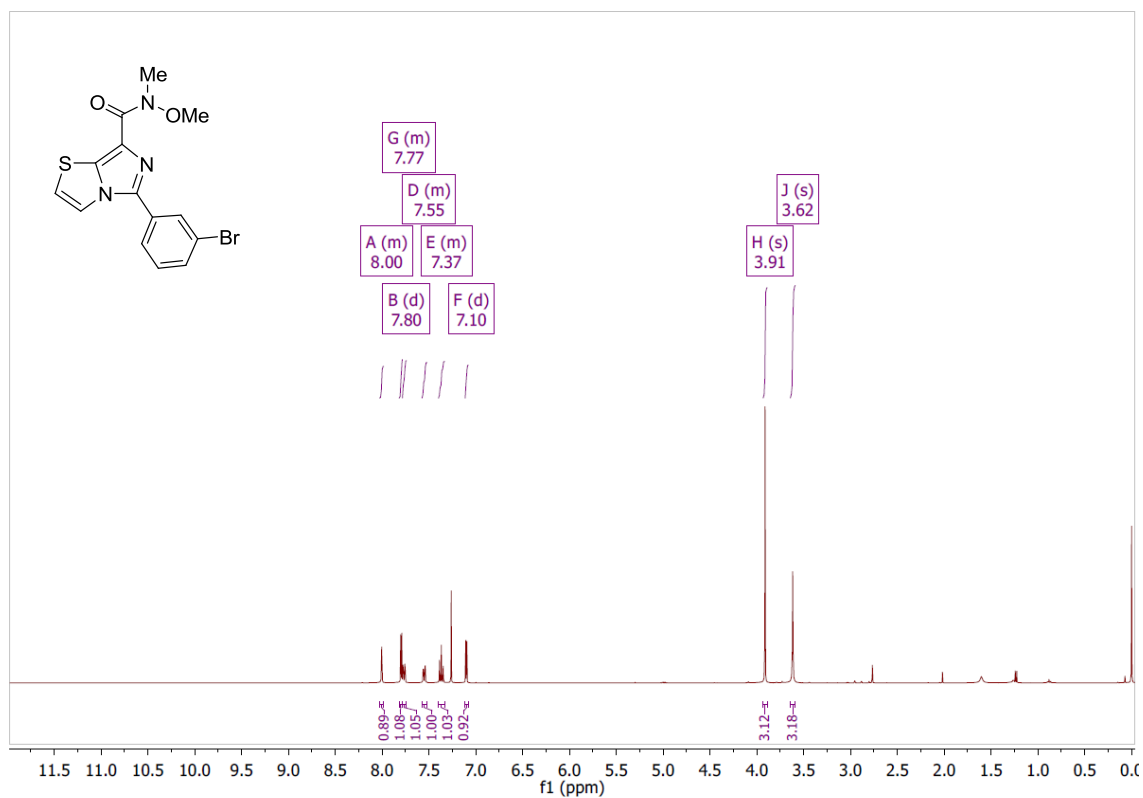
# Compound 4f



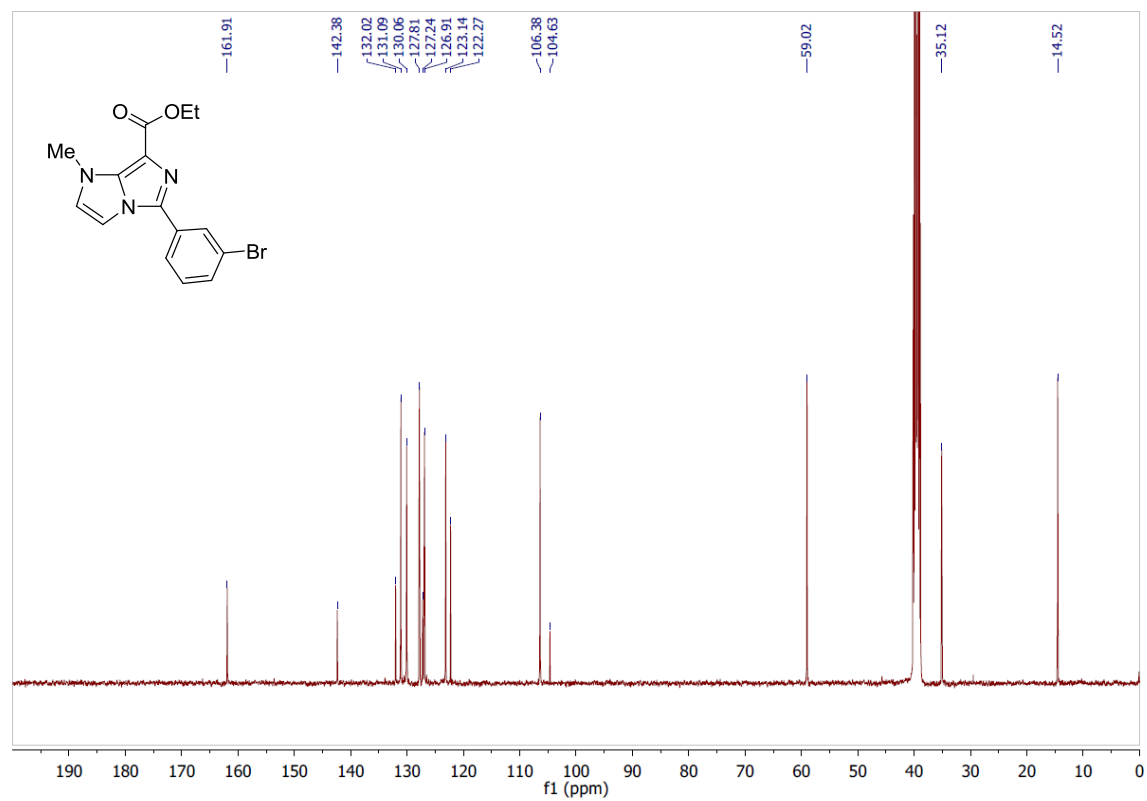
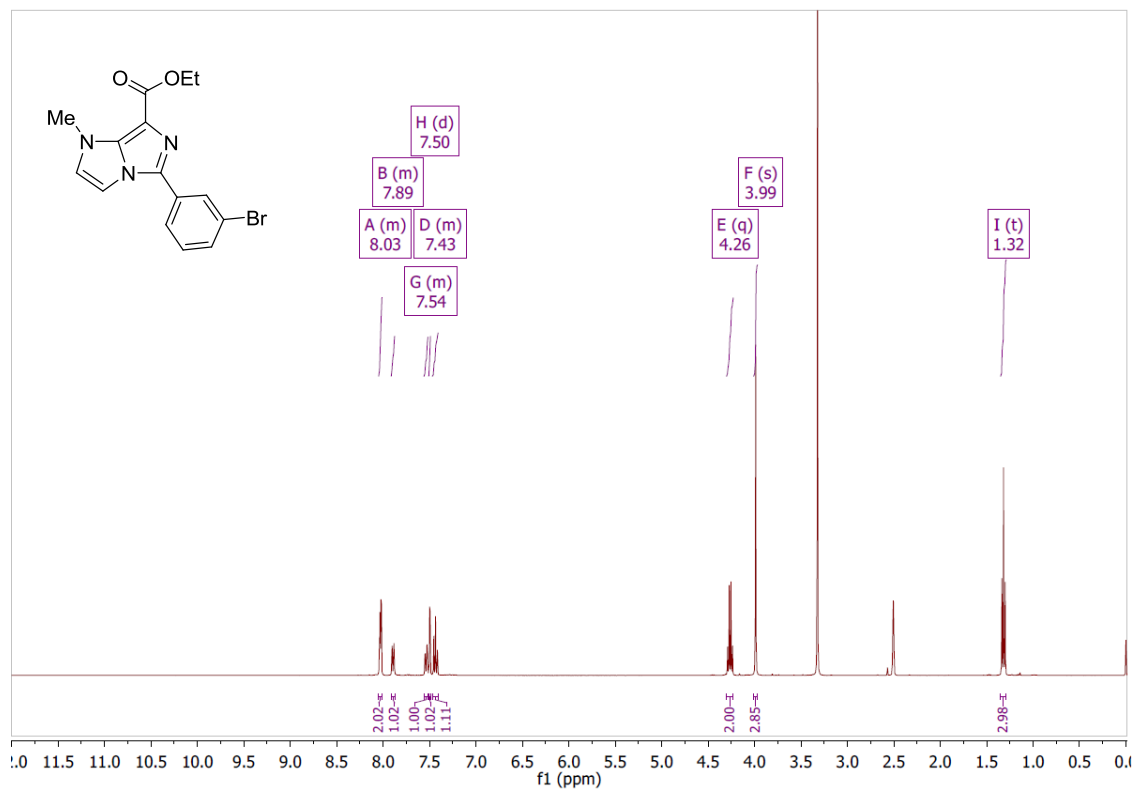
# Compound 4g



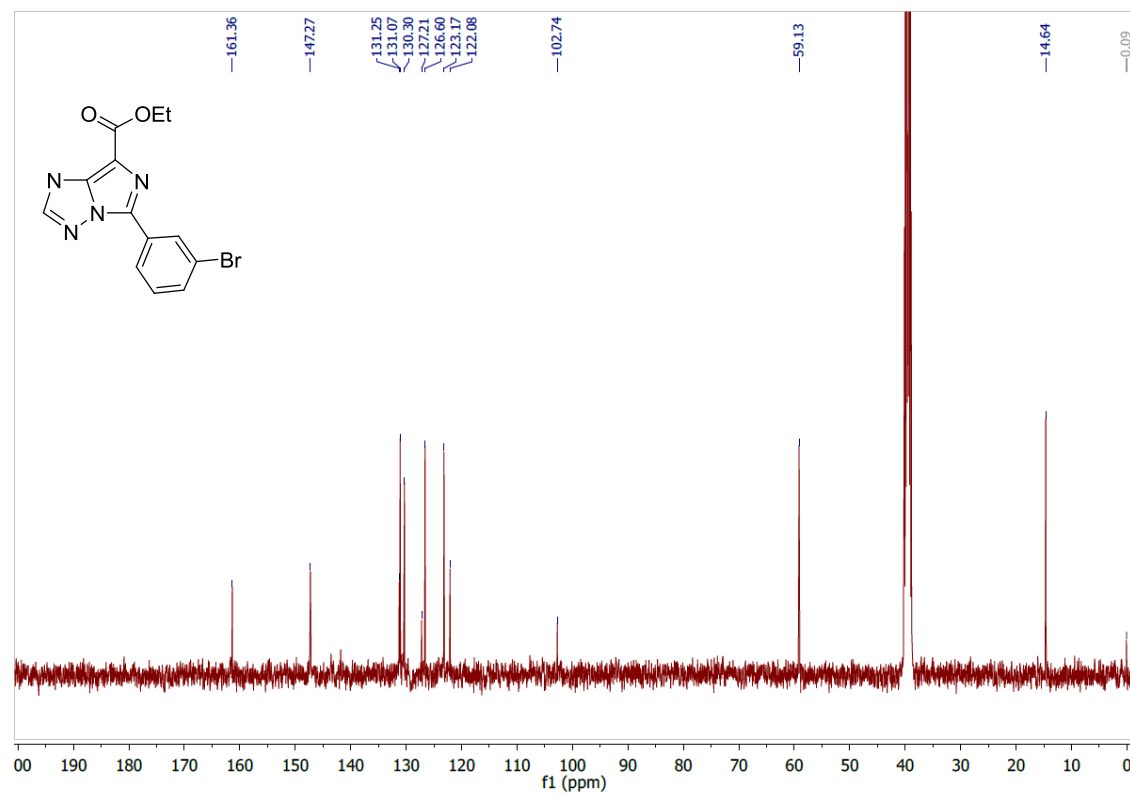
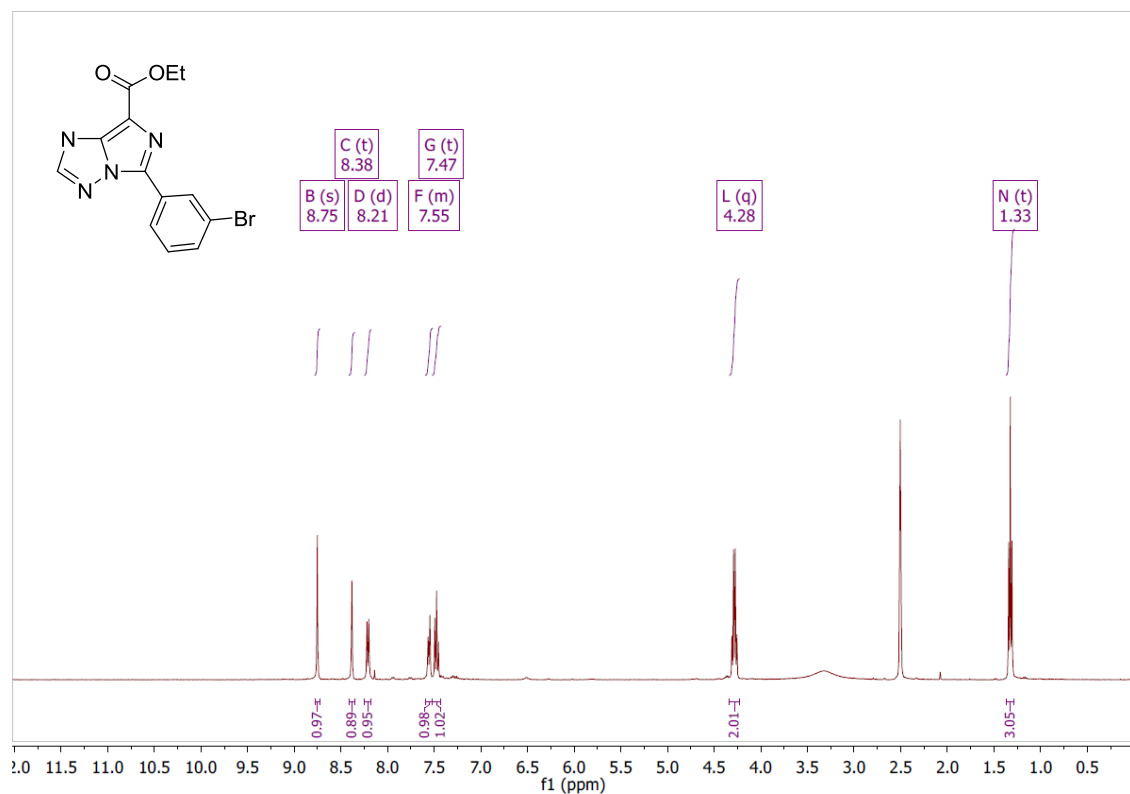
# Compound 4h



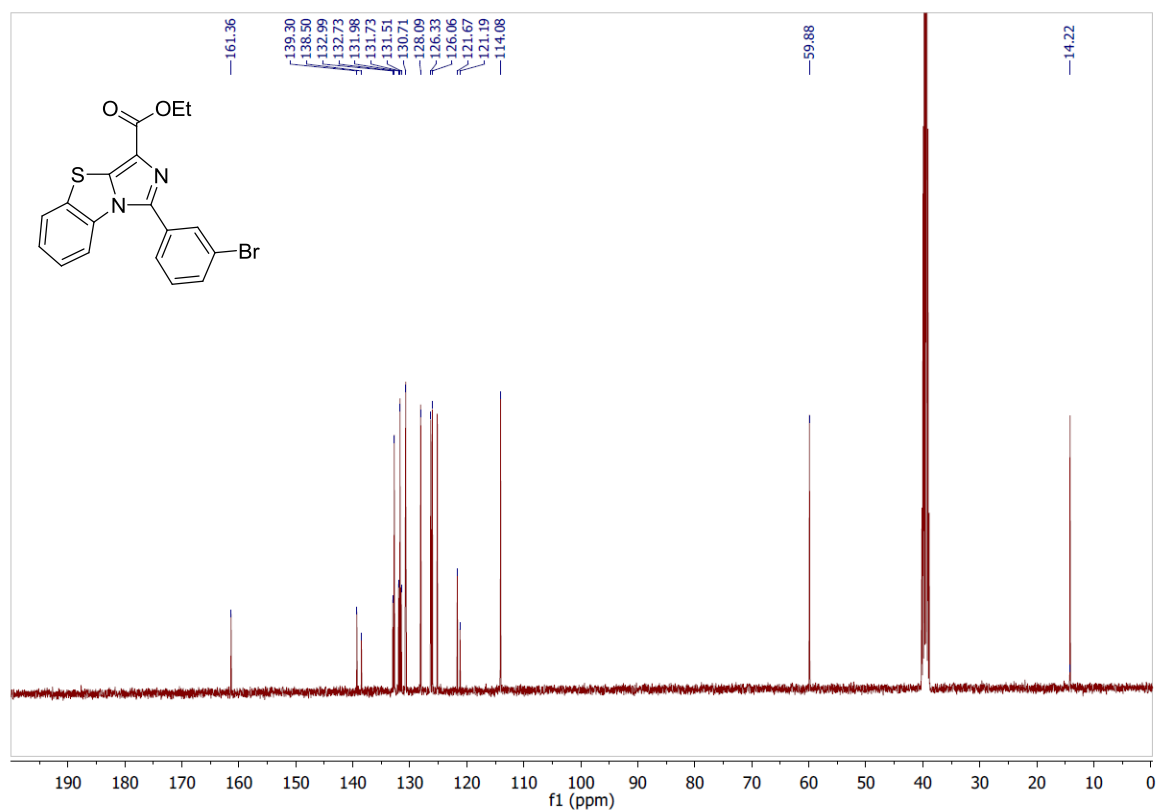
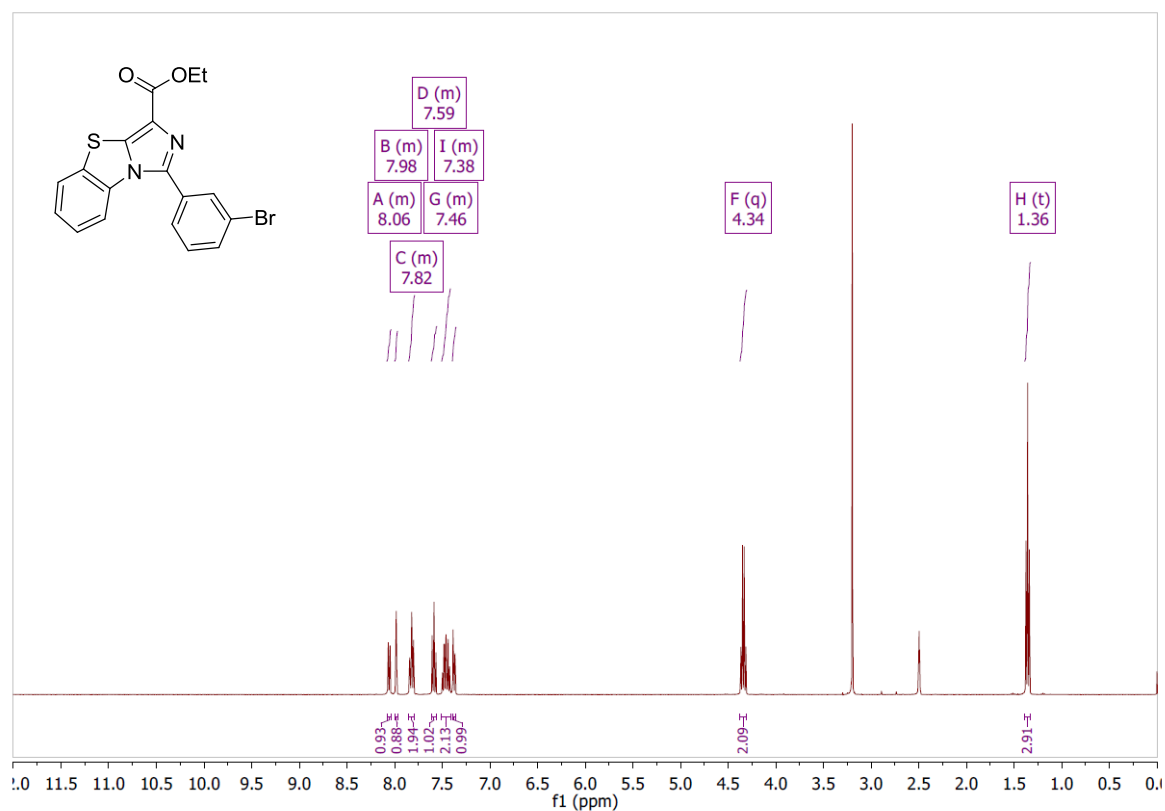
# Compound 4i



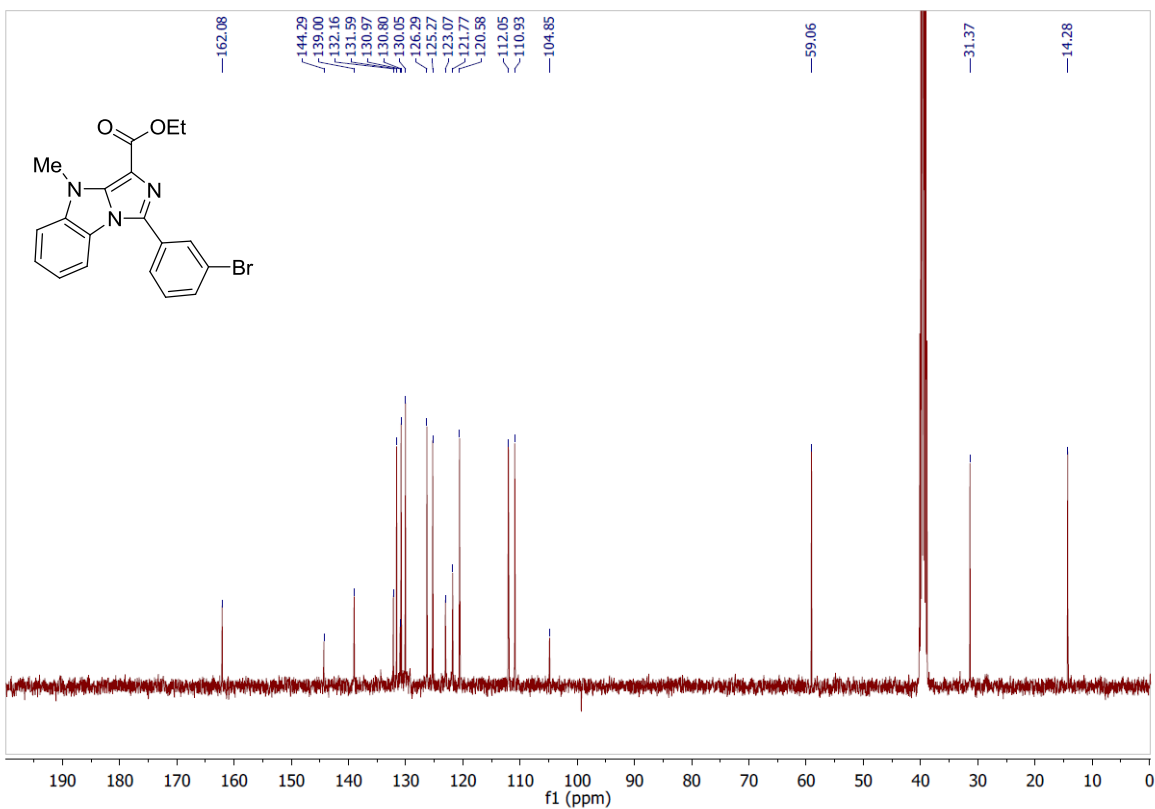
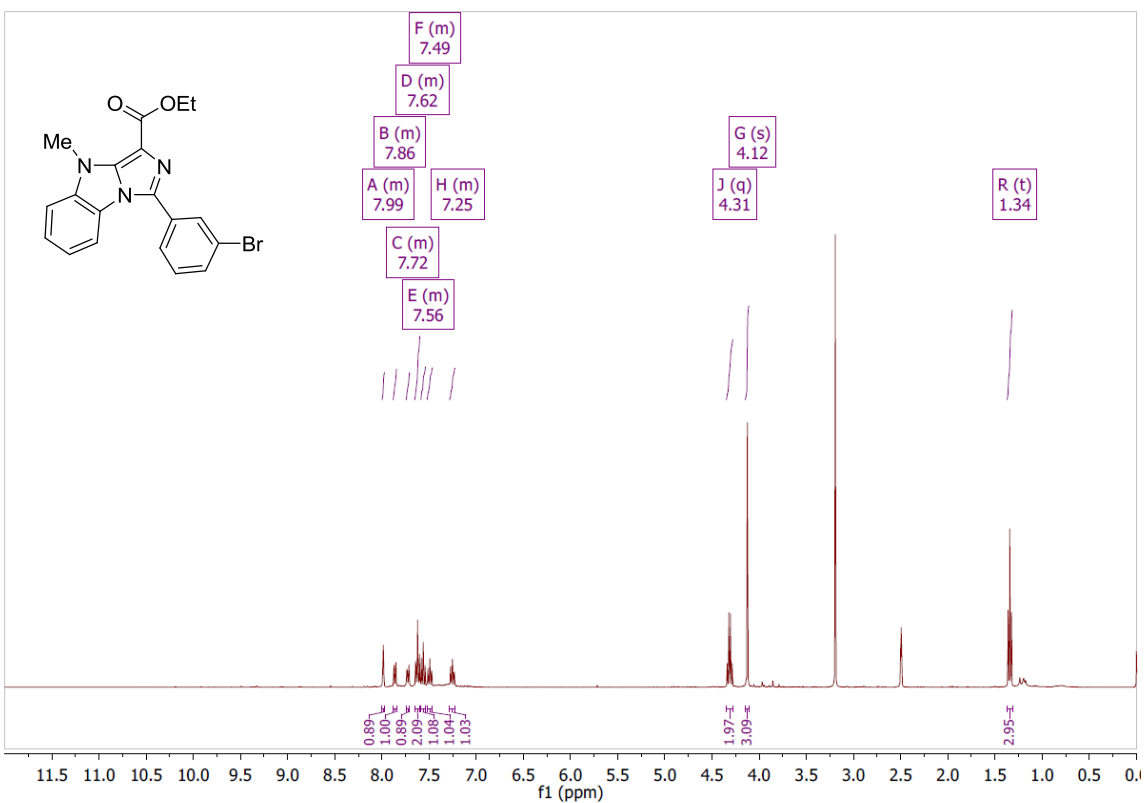
# Compound 4j



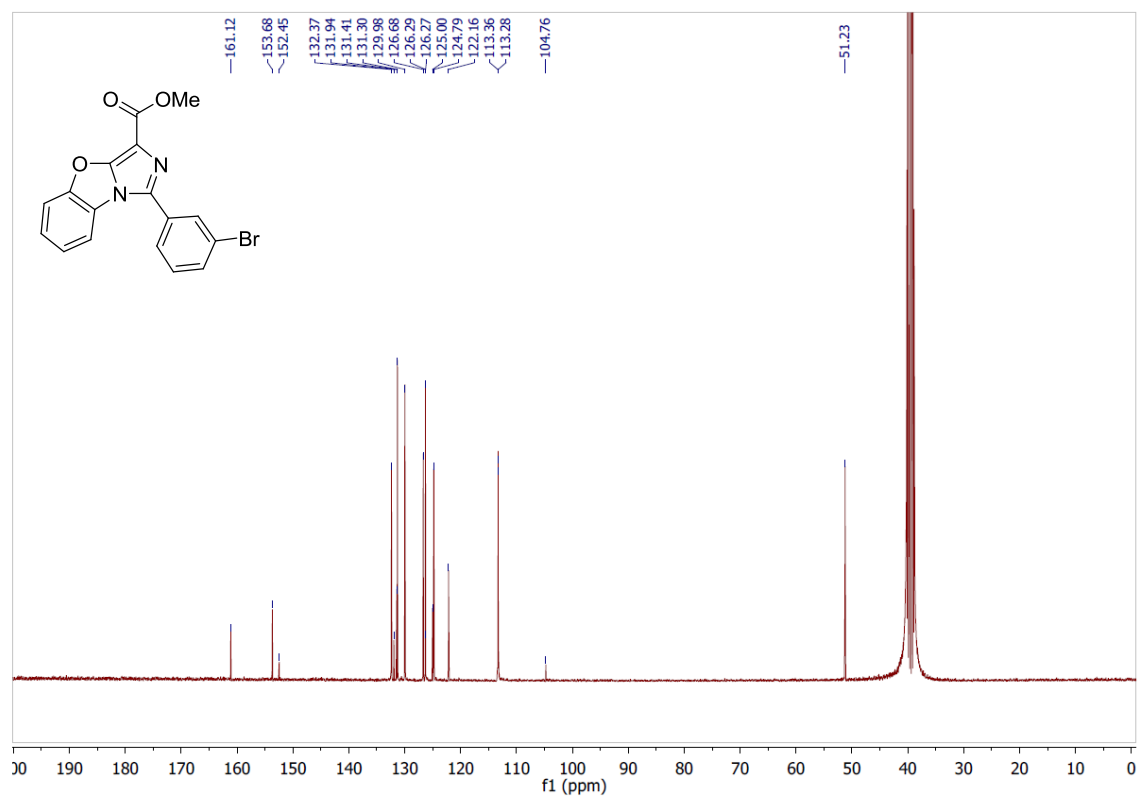
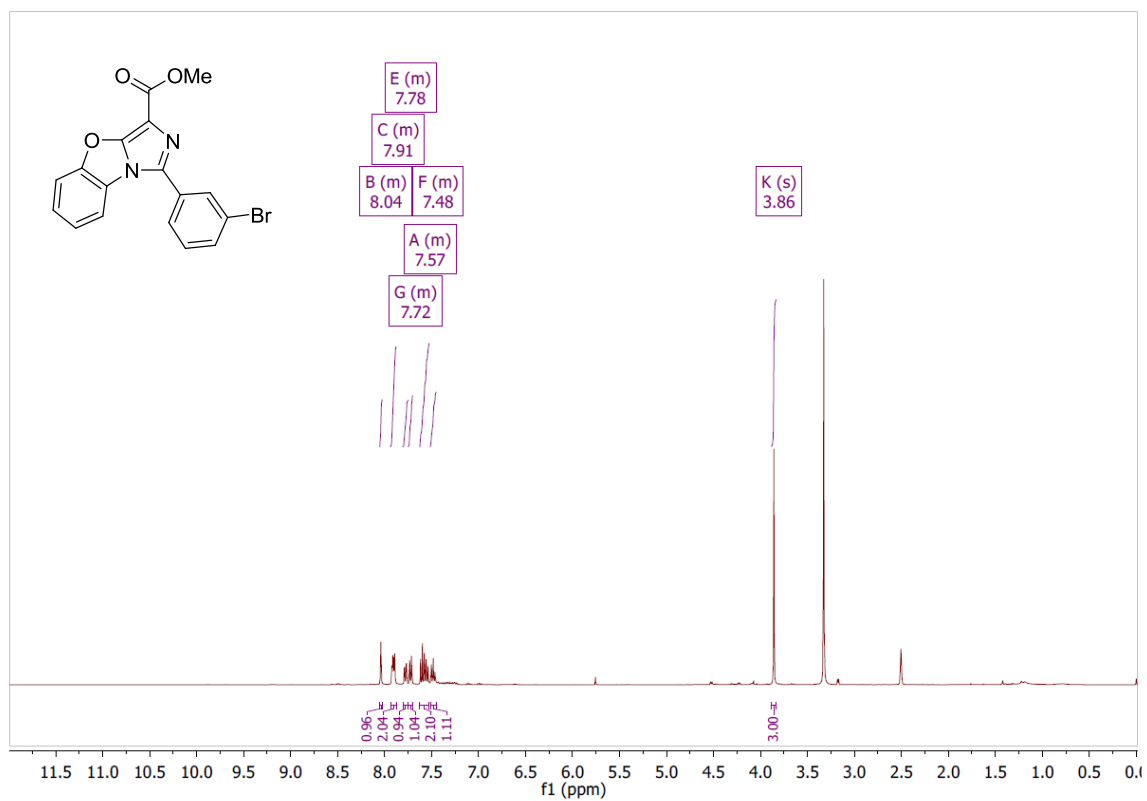
# Compound 4k



# Compound 4l

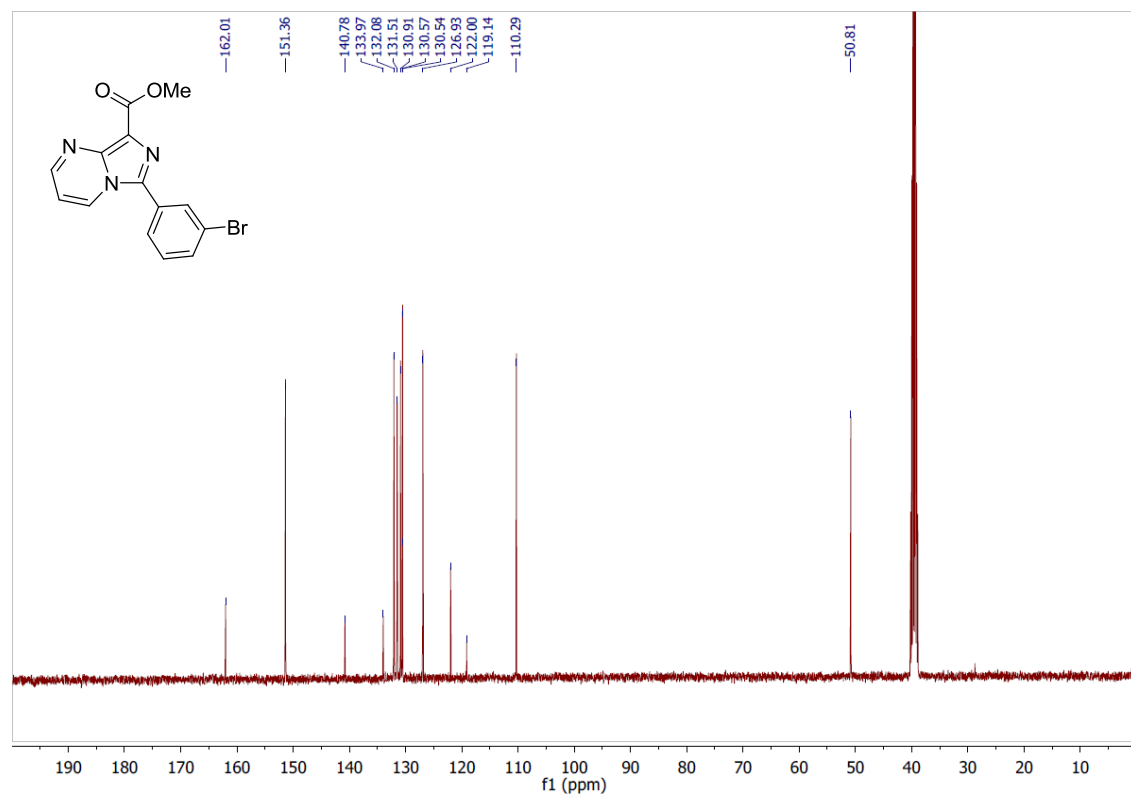
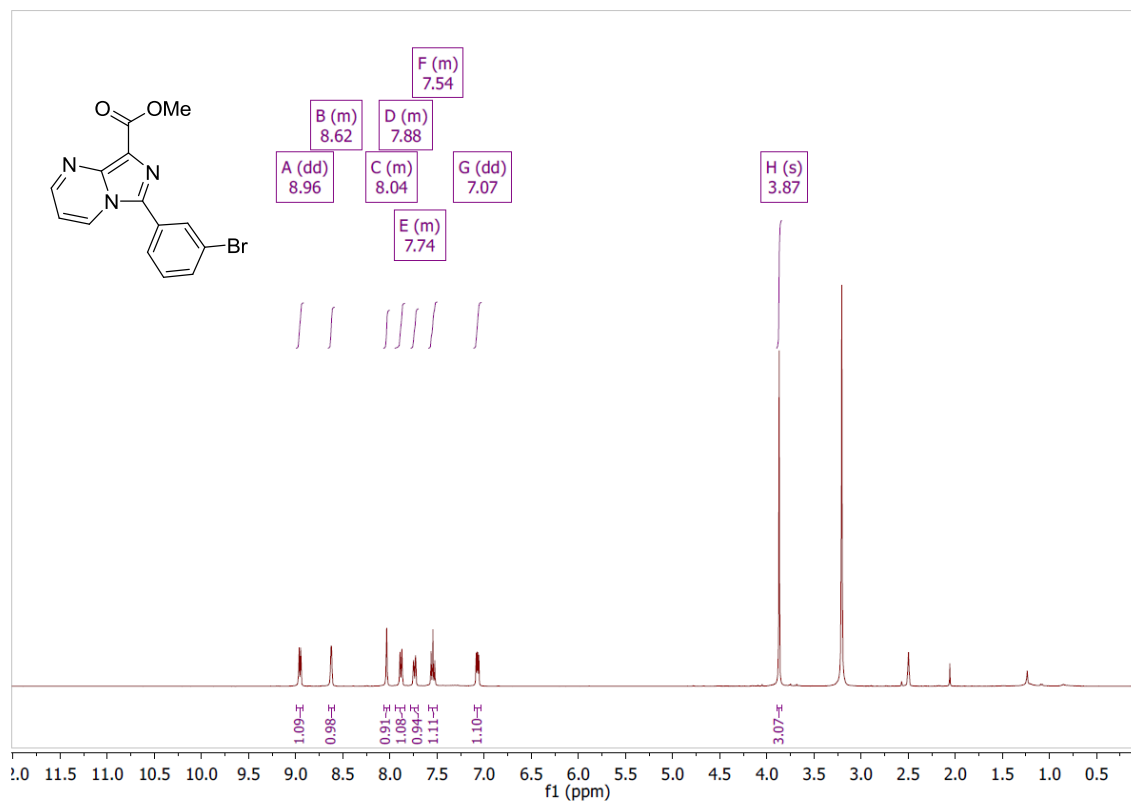


# Compound 4m

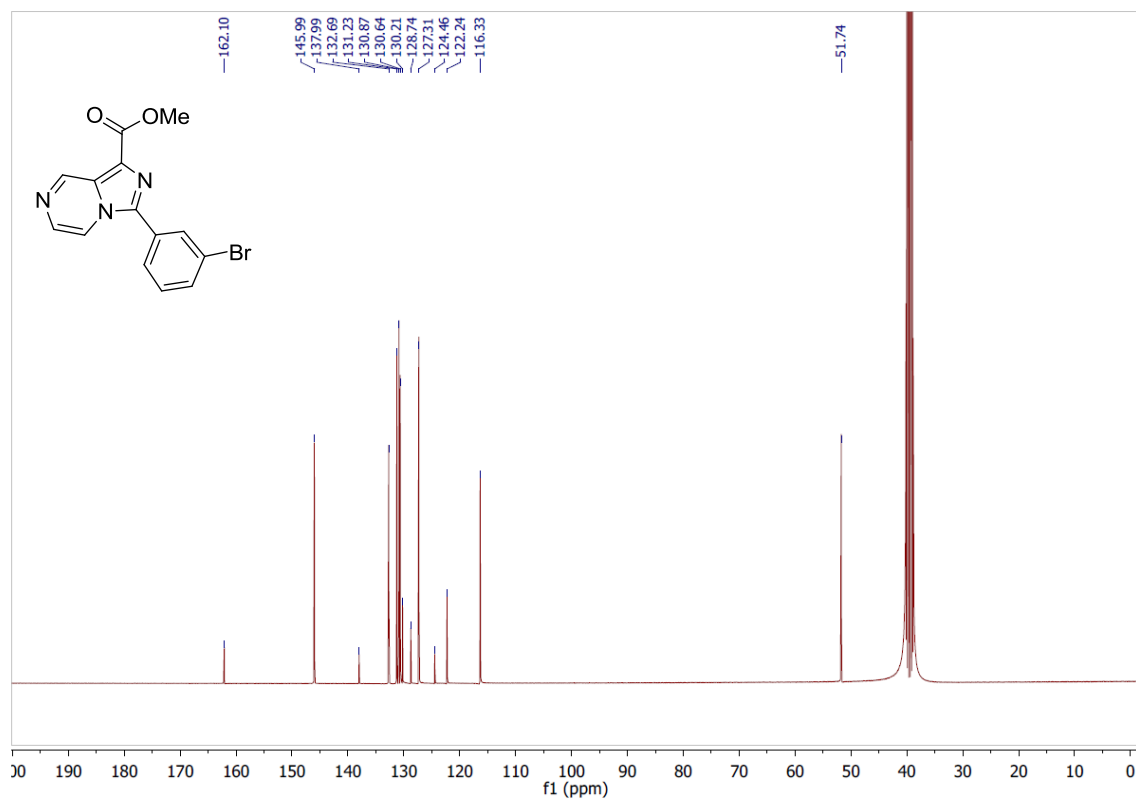
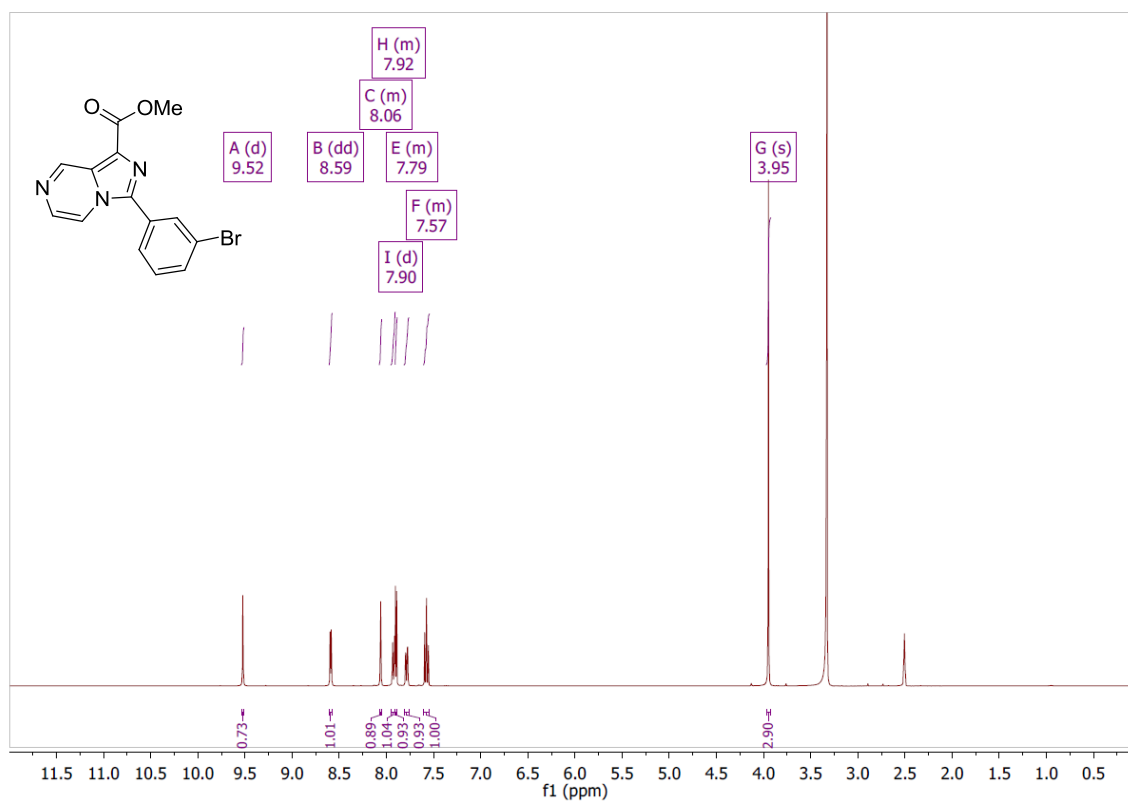




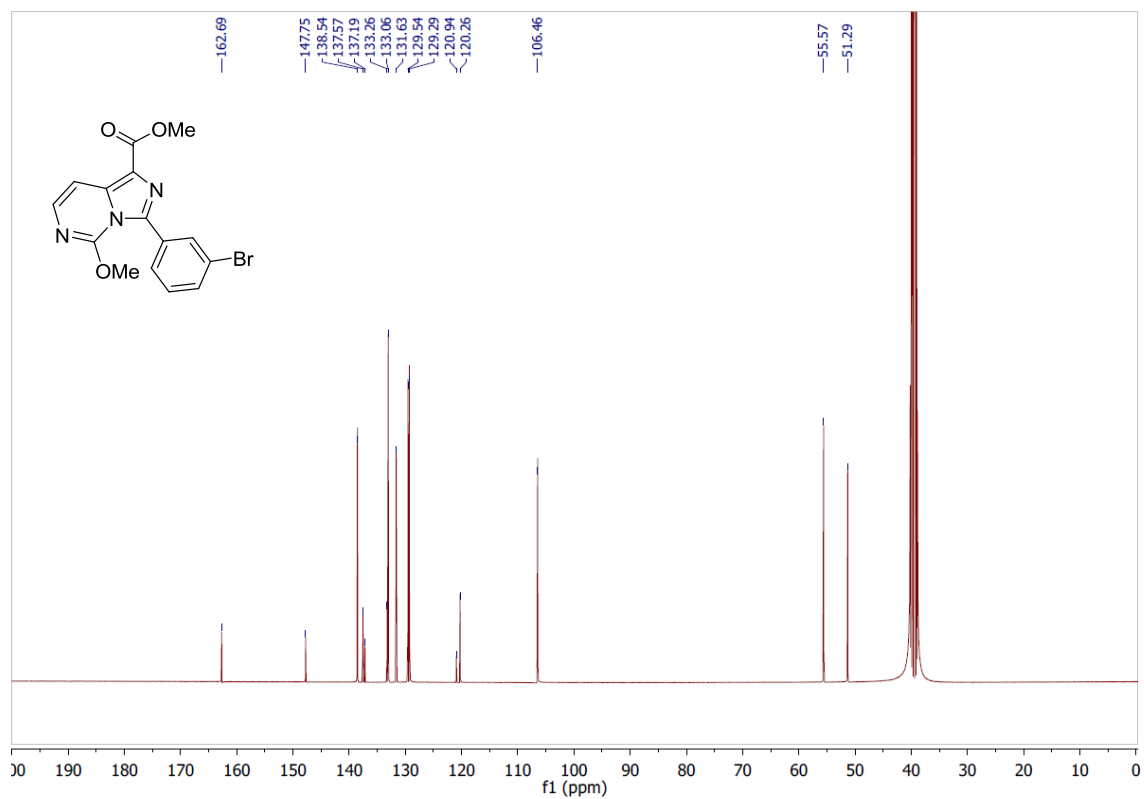
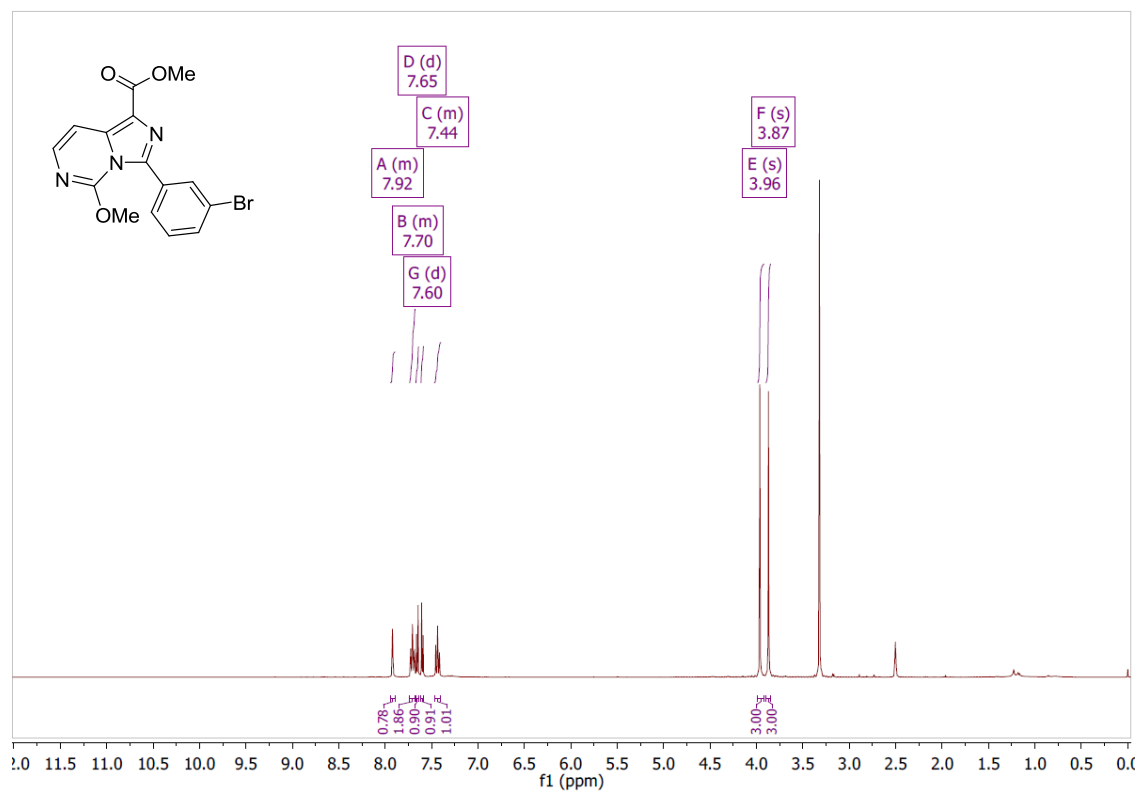
# Compound 4n



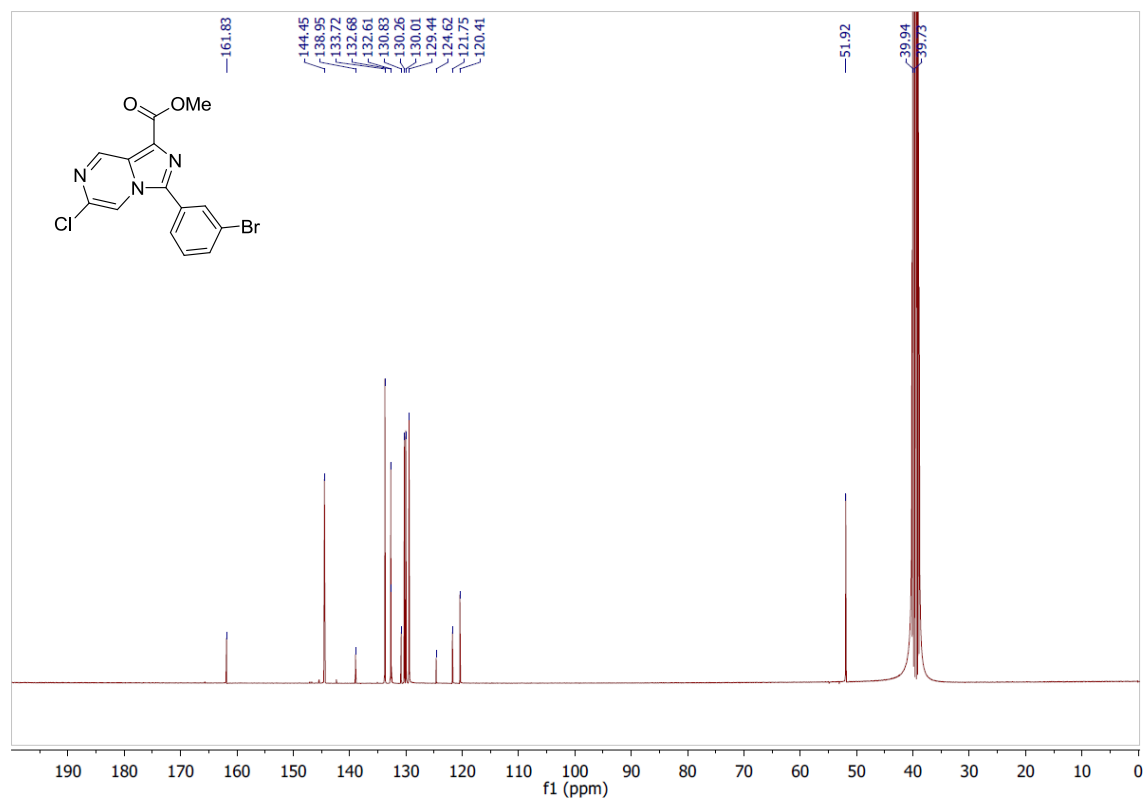
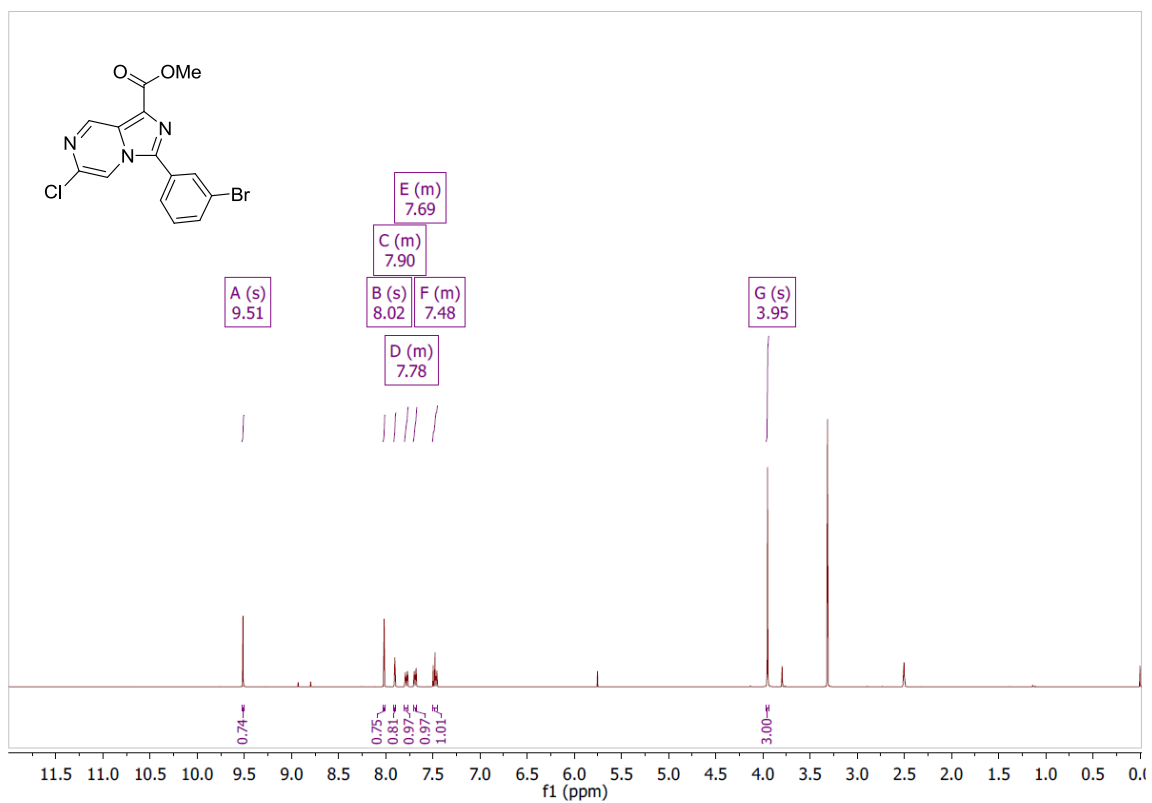
# Compound 4o



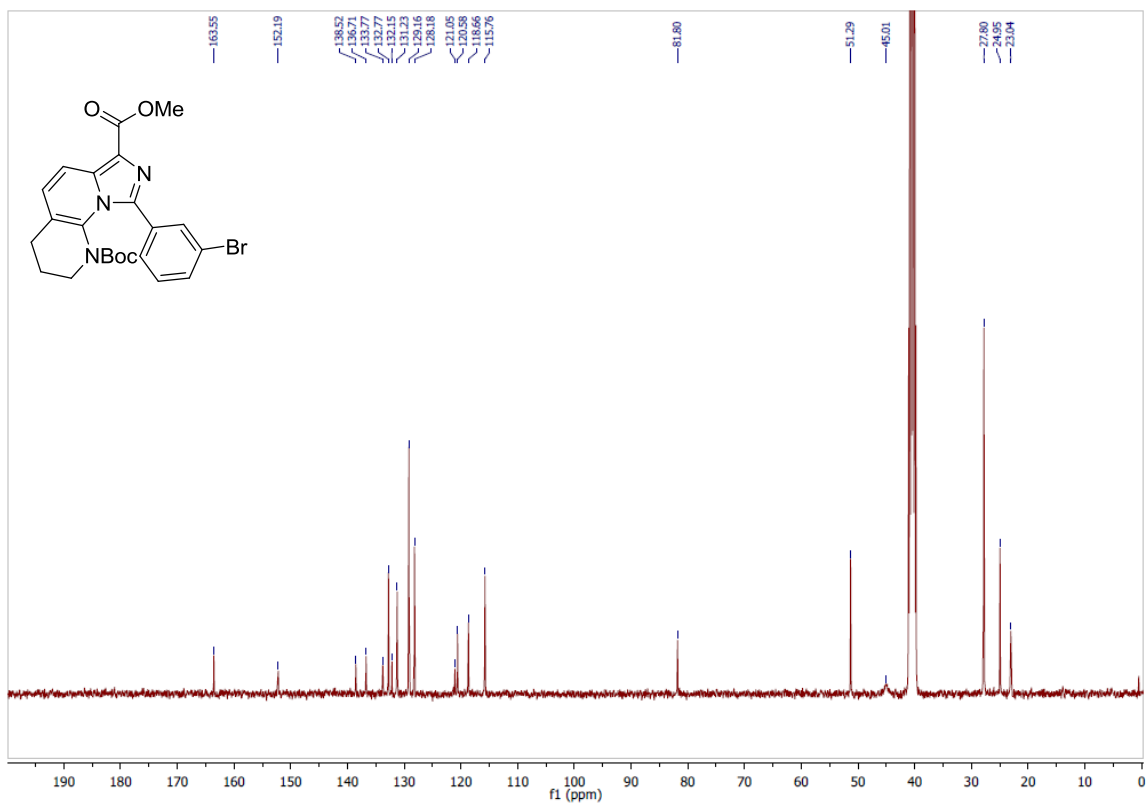
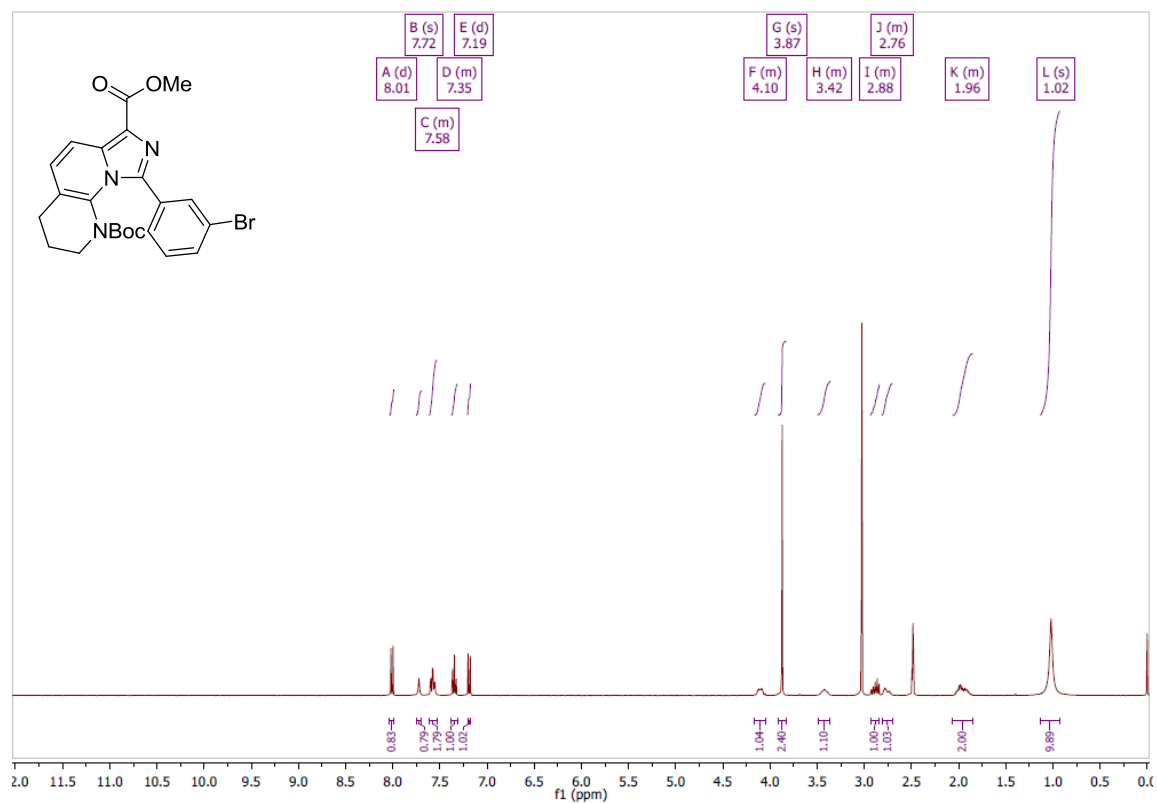
# Compound 4p



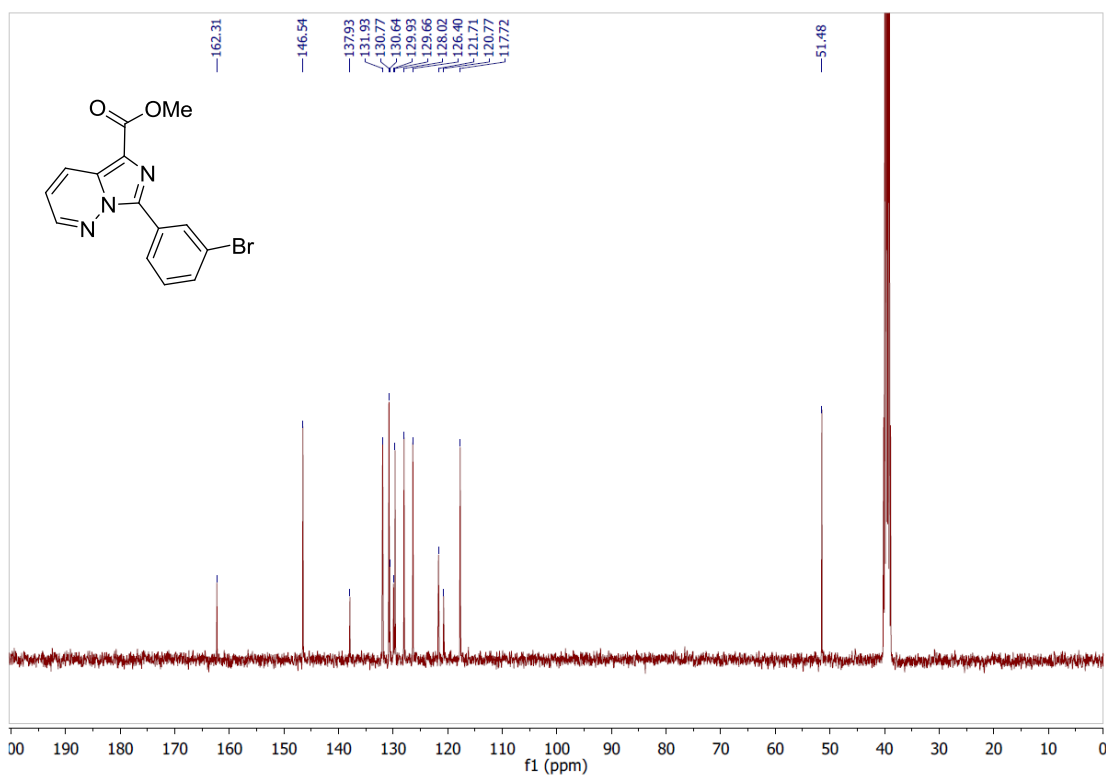
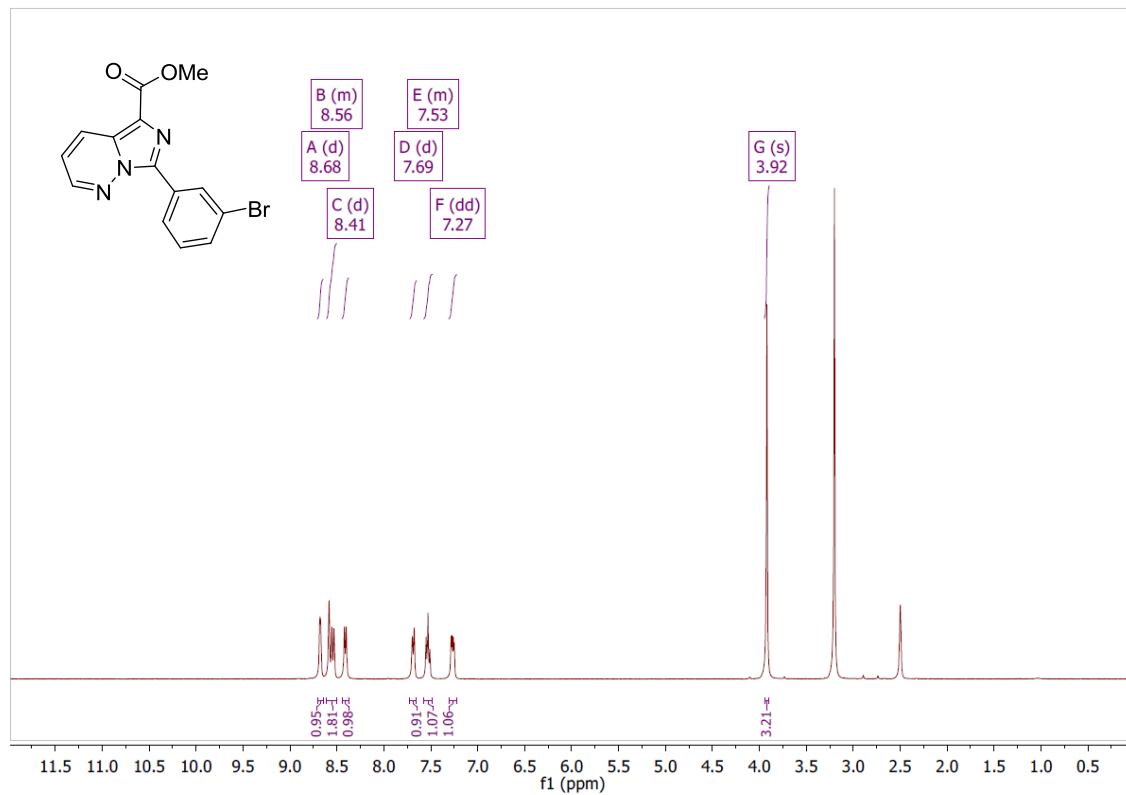
# Compound 4q



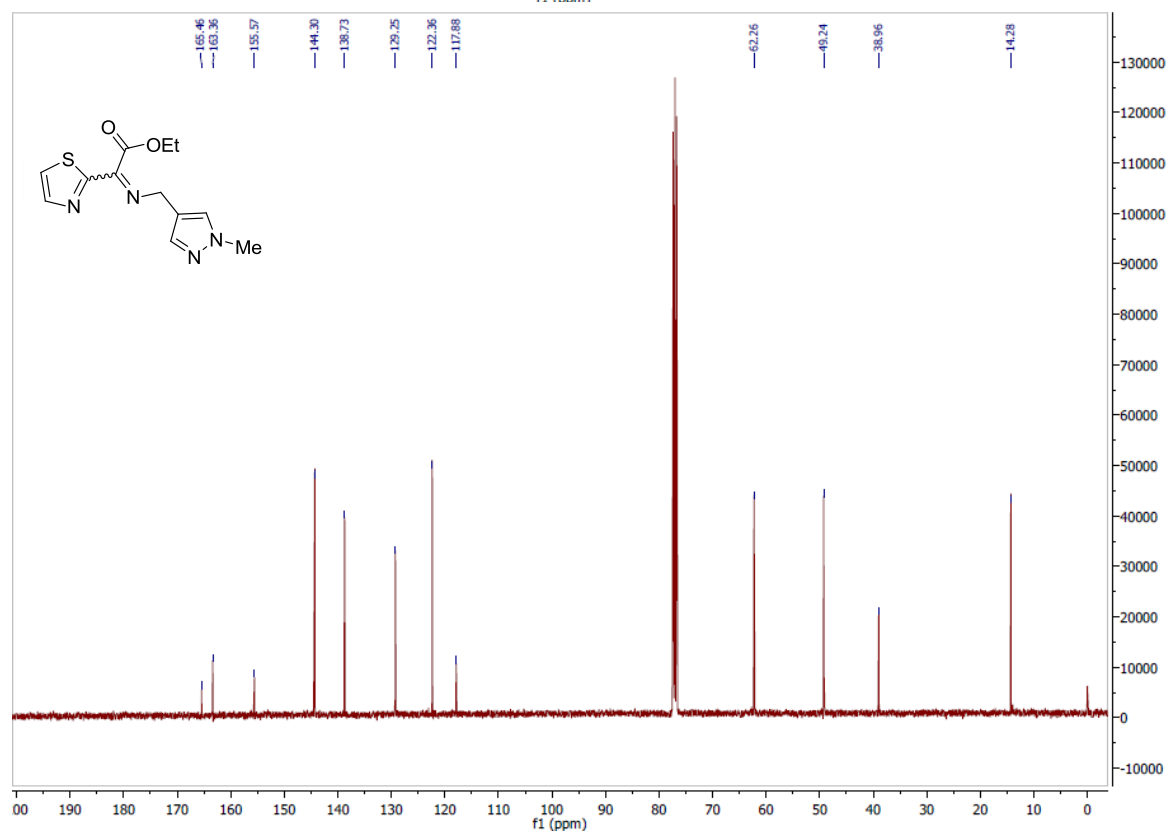
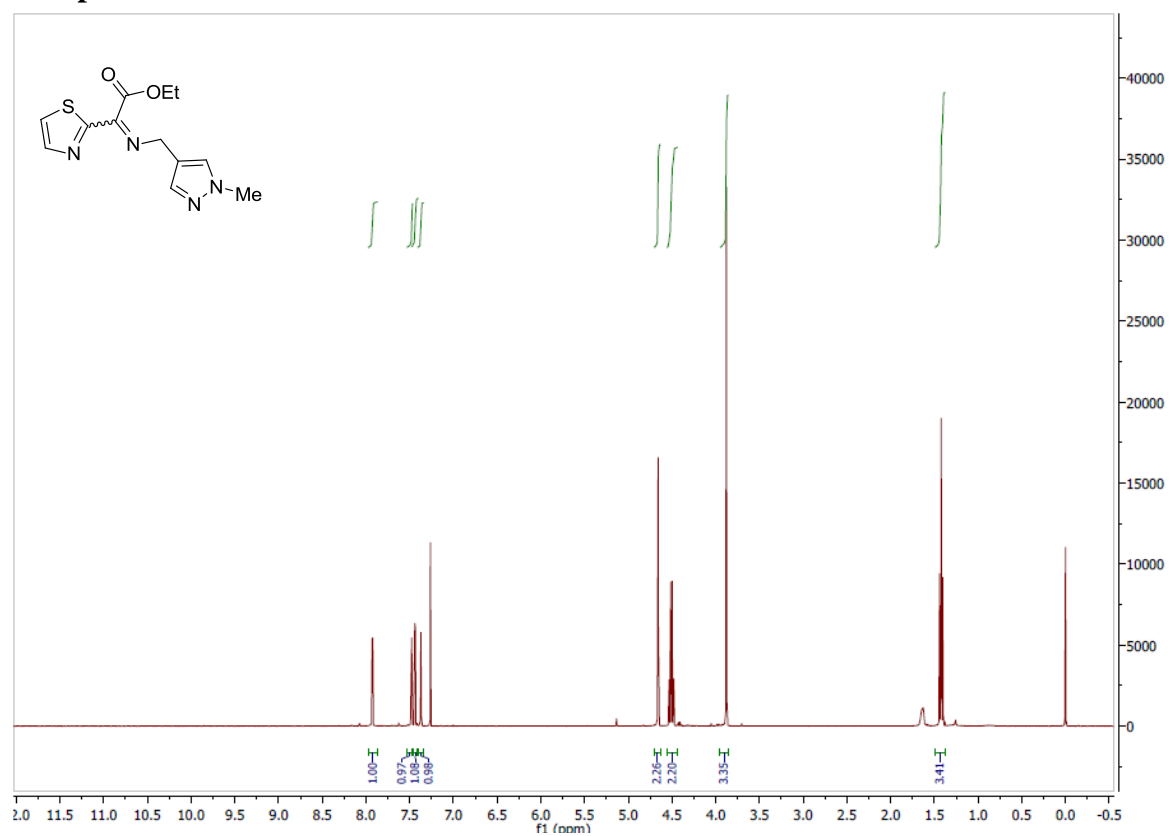
# Compound 4r



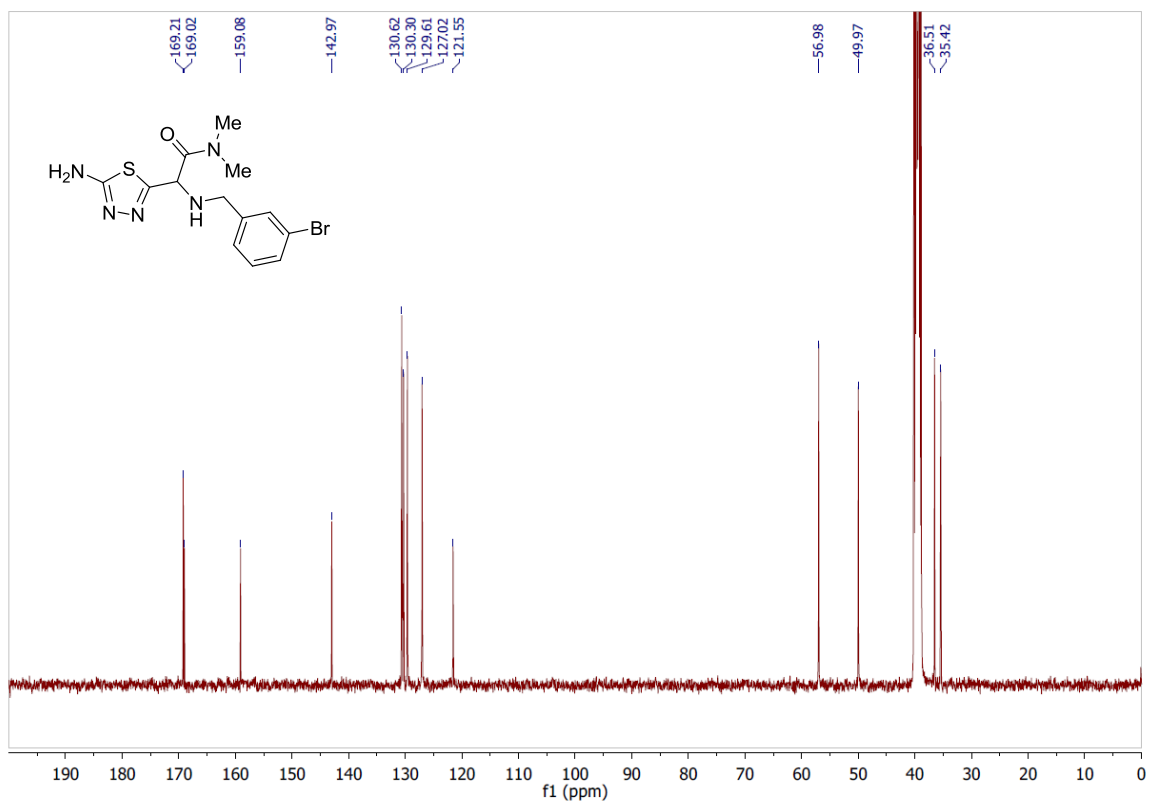
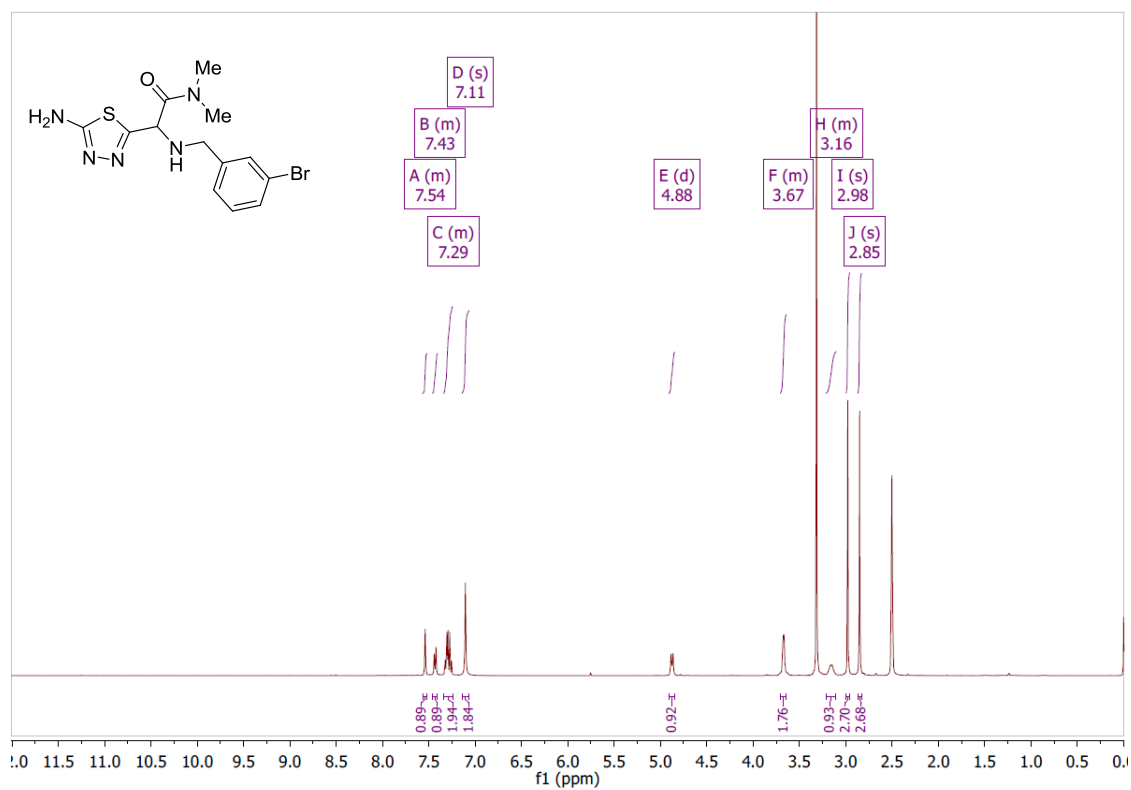
# Compound 4t



## Compound 5

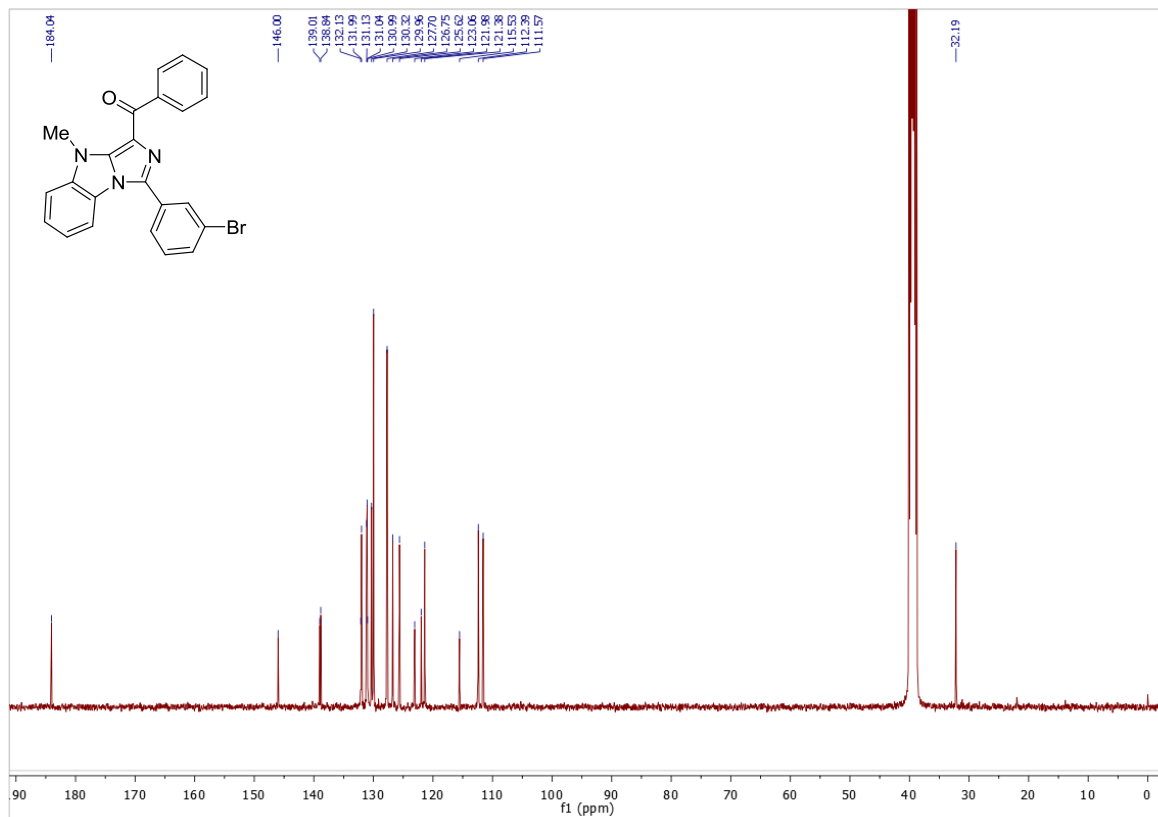
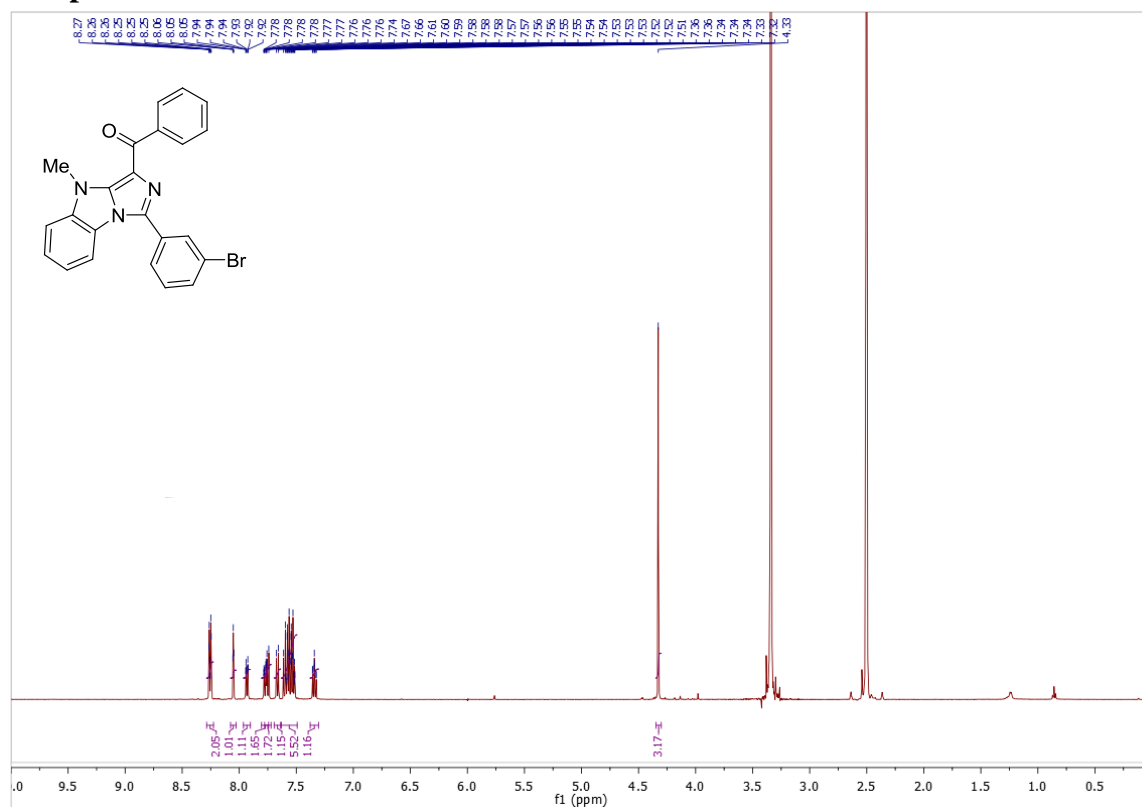


# Compound 6

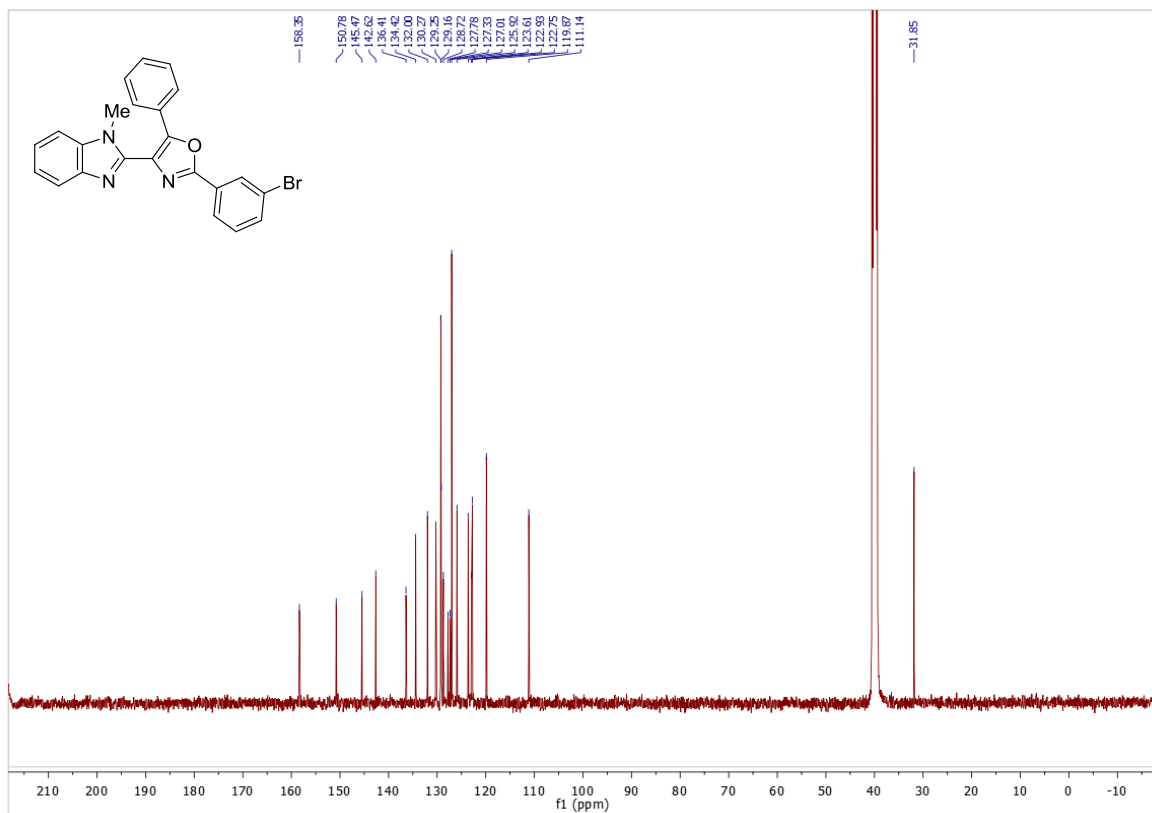
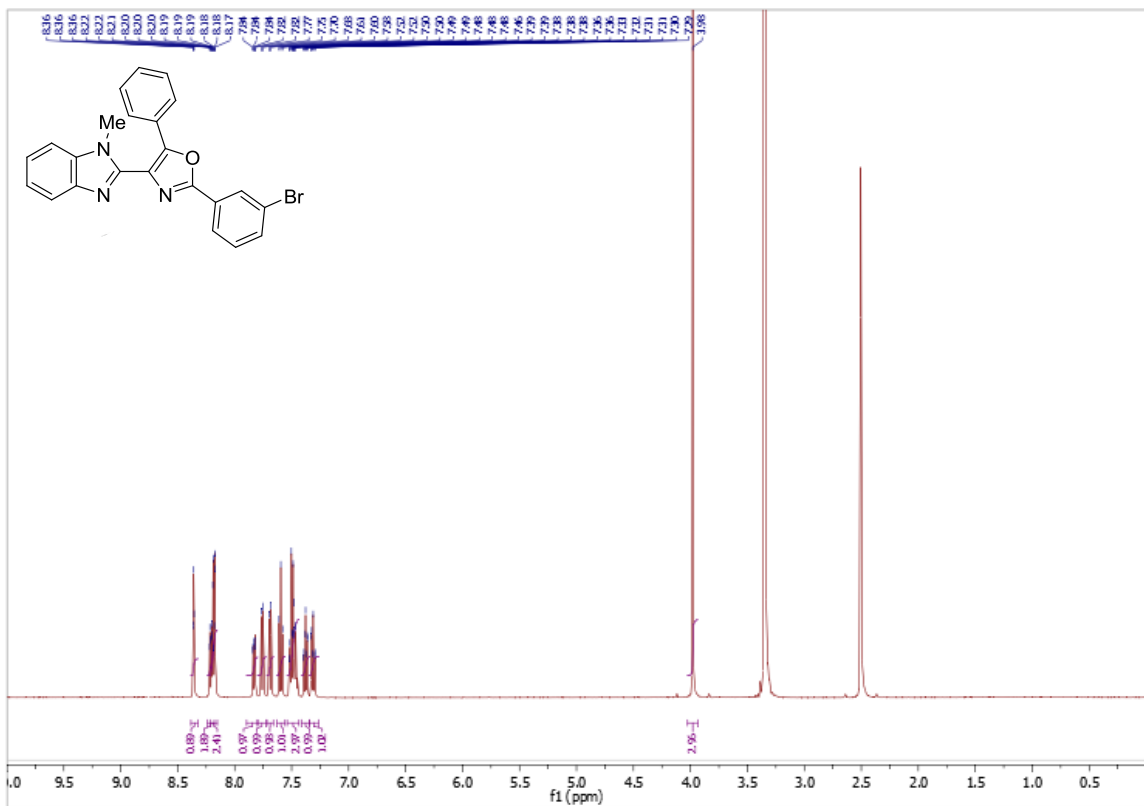




# Compound 4u



### Compound 7



## Compound 8

