

Table S-1. Kinetic data for exchange of $\text{PPh}_3\text{-d}_{15}$ into $\text{CpRe(PPh}_3)_2\text{H}_2$.

Exchange of $\text{PPh}_3\text{-d}_{15}$ into $\text{CpRe(PPh}_3)_2\text{H}_2$ by ^{31}P NMR by photolysis

$[\text{CpReP}_2\text{H}_2] = 0.00514$
$[\text{PPh}_3\text{-d}_{15}] = 0.244$

time(sec)	time(min)	$[\text{PPh}_3\text{-d}_{15}]/[\text{PPh}_3]$	$[\text{PPh}_3]$	$\ln(1 - [\text{PPh}_3]/2C_0)$
2700	45	603	0.000405	-0.040157844
4500	75	348	0.000701	-0.070642657
7200	120	216	0.00113	-0.116405904
10800	180	152	0.001605	-0.16978527
14400	240	119	0.00205	-0.2224653
19800	330	87	0.002805	-0.318582323

statistics:	
-0.0009615	0.0023085
1.643E-05	0.0031525
0.9988337	0.0039427
3425.5339	4
0.0532508	6.218E-05

$k_{\text{obs}} = 1.603\text{E-}05 \text{ s}^{-1}$
95% limit = $3.103\text{E-}07$
$k_2 = 6.568\text{E-}05 \text{ M}^{-1}\text{s}^{-1}$
$k(\text{HD})/k(\text{PPh}_3) = 36.752959$

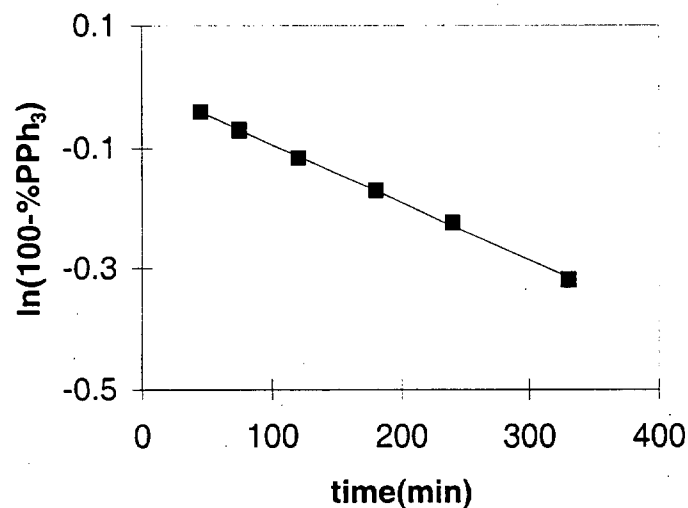


Table S-2. Kinetic data for H/D exchange between pentane and C₆D₆ catalyzed by CpRe(PPh₃)₂H₂.

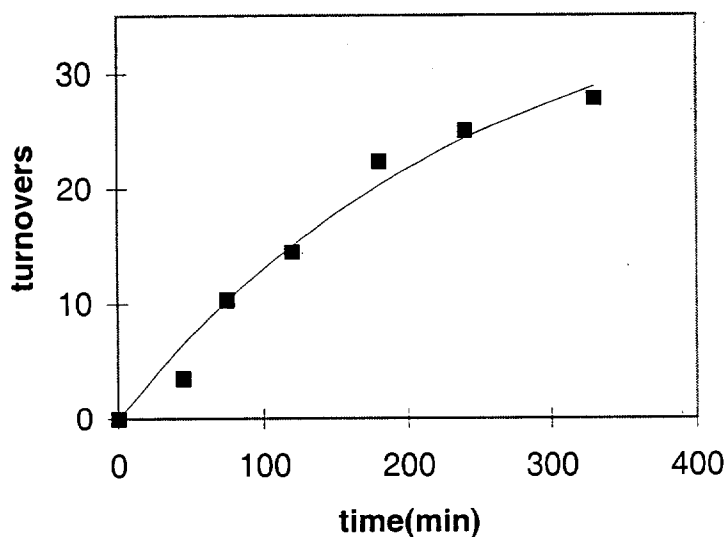
Photochemical H/D exchange between pentane and C₆D₆ by CpRe(PPh₃)₂H₂.

[pentane]=0.88M

[CpReP₂H₂]=0.00514

time(sec)	time(min)	turnovers	calc	error^2
0	0	0	0	0
2700	45	3.5	6.500766	9.004595
4500	75	10.3	10.228	0.005184
7200	120	14.5	15.04353	0.295424
10800	180	22.4	20.25324	4.608574
14400	240	25	24.3511	0.42107
19800	330	27.8	28.91808	1.250102
				15.58495

k=	0.004001
C _{inf} =	39.45423
k _{obs} =	6.67E-05 s ⁻¹
(above fit using solver)	
linest statistics (5 points):	
0.002124	-0.56573
0.000153	0.960128
0.984606	1.273874
191.8859	3
311.3837	4.868263



k _{obs} =	0.002124 s ⁻¹
95%error=	0.000184
k ₂ =	0.002414 M ⁻¹ s ⁻¹

Table S-3. Kinetic data for H/D exchange between propane and C₆D₆ catalyzed by CpRe(PPh₃)₂H₂.

Effect of [propane] on H/D exchange with C₆D₆ by 0.0128 M CpRe(PPh₃)₂H₂ (365 nm).

time(min)	1.28 M	2.56 M	3.84 M	simulated data			errors		
0	0	0	0	0	0	0	0	0	0
10				1.163336	2.312984	3.151113			
20				2.27538	4.515533	6.157827			
30				3.338392	6.612919	9.026758			
40	0	6	11	4.354535	8.610163	11.76422	18.96198	6.812952	0.584032
60				6.254388	12.32312	16.86856			
80				7.990402	15.68999	21.51581			
100				9.576705	18.74302	25.7469			
120	11	20	28	11.02621	21.51148	29.59909	0.000687	2.28456	2.557089
150				12.96949	25.18791	34.74034			
180				14.66689	28.36248	39.20669			
210				16.14953	31.1037	43.08673			
240	19	36	48	17.44457	33.47073	46.45744	2.419347	6.397227	2.379483
300				19.56382	37.27953	51.92951			
360				21.18071	40.11944	56.05923			
420				22.41433	42.23693	59.17588			
480	24	44	62	23.35552	43.81578	61.52798	0.415351	0.033938	0.222799
580				24.45506	45.60628	64.23738			
680				25.15551	46.70403	65.93231			
780				25.60172	47.37705	66.99262			
880	25	47	67	25.88597	47.78968	67.65593	0.784942	0.623591	0.43024
1000				26.09446	48.08003	68.13296			
							22.5823	16.15227	6.173644

C _{inf} =	26.38486	48.44354	68.76416
k=	0.004509	0.004892	0.004691 min ⁻¹

(fit using solver)

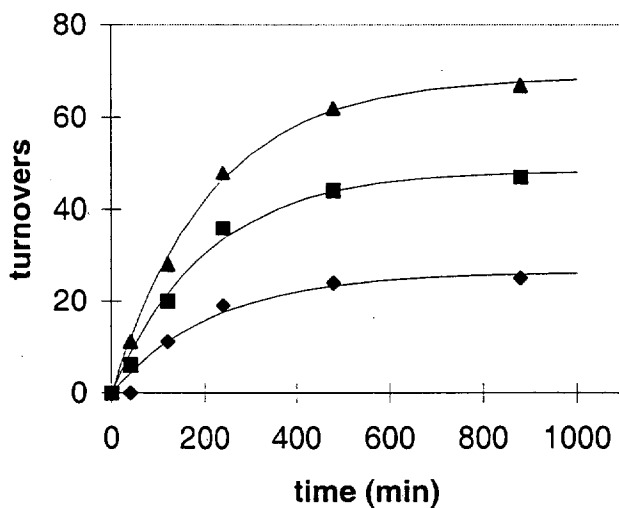
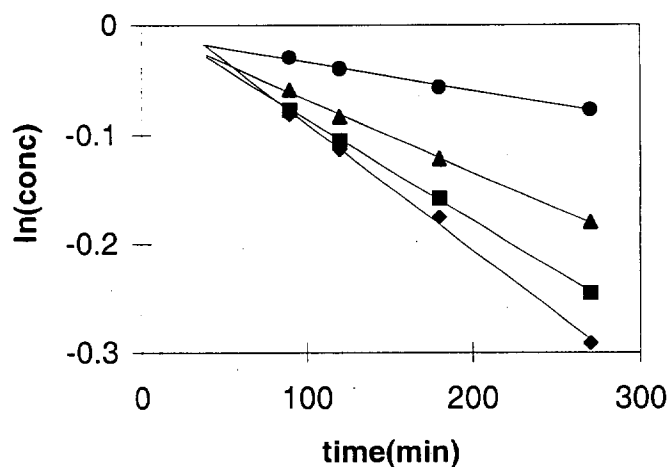


Table S-4. Kinetic data for exchange of PMe_3 into $\text{CpRe}(\text{PPh}_3)_2\text{H}_2$.

Effect of $[\text{PMe}_3]$ on photosubstitution of $\text{CpRe}(\text{PPh}_3)_2\text{H}_2$

		[Re] ₀ = 0.00514								
		[PMe ₃]=	.574 M	.109 M	.027 M	.0053 M	.574 M	.109 M	.027 M	.0053 M
time(sec)	time(min)	conc	conc	conc	conc	ln(conc)	ln(conc)	ln(conc)	ln(conc)	ln(conc)
0	0	0.00514	0.00514	0.00514	0.00514	0	0	0	0	0
5400	90	0.00474	0.00476	0.00485	0.00499	-0.08102	-0.07681	-0.05807	-0.02962	
7200	120	0.00459	0.00463	0.00473	0.00494	-0.11317	-0.1045	-0.08313	-0.03969	
10800	180	0.00431	0.00439	0.00455	0.00486	-0.17612	-0.15772	-0.12193	-0.05601	
16200	270	0.00384	0.00402	0.00429	0.00476	-0.29158	-0.24577	-0.18077	-0.07681	
with zeroes						k=	1.80E-05	1.52E-05	1.12E-05	4.74E-06
						95% error=	1.63E-06	5.19E-07	3.20E-07	4.70E-07
without zeroes						k=	1.95E-05	1.56E-05	1.12E-05	4.32E-06
						95% error=	1.53E-06	5.47E-07	6.73E-07	4.62E-07

without zeroes:



with zeroes:

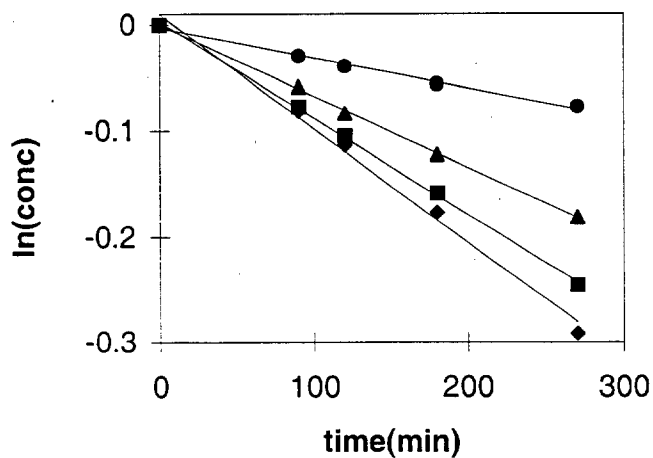
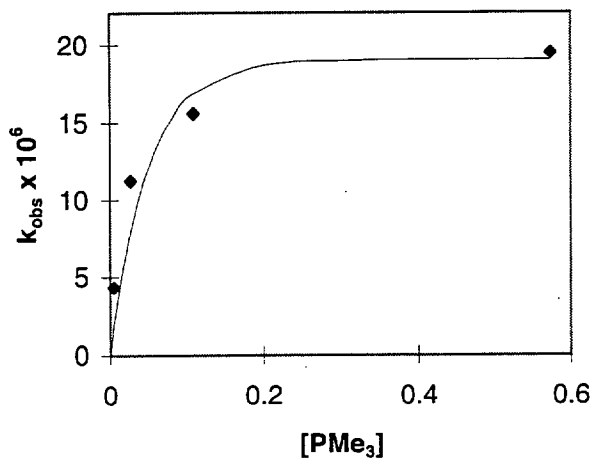


Table S-5. Plots of saturation kinetics for exchange of PMe_3 into $\text{CpRe}(\text{PPh}_3)_2\text{H}_2$. k_{obs} vs $[\text{PMe}_3]$ for substitution in $\text{CpRe}(\text{PPh}_3)_2\text{H}_2$.

$[\text{PMe}_3]$	$k_{\text{obs}} \times 10^6$	k_{calc}	$1/[\text{PMe}_3]$	$1/k_{\text{obs}}$	error
1E-08	1E-08	3.8E-06	1E+08	1E+08	1E-11
0.002		0.745001			
0.0045		1.635307			
0.0053	4.318828	1.910932	188.68	0.232	5.798
0.01		3.444116			
0.015		4.924454			
0.021		6.51611			
0.027	11.21078	7.927783	37.037	0.089	10.778
0.04		10.46275			
0.06		13.27731			
0.08		15.16397			
0.109	15.61649	16.85221	9.1743	0.064	1.527
0.2		18.652			
0.32		18.96843			
0.45		18.99766			
0.574	19.47859	18.9998	1.7422	0.051	0.2292
$C_{\text{inf}} =$		19			
$k =$		20	Solver values		
					18.332



Double Reciprocal plot:

$[\text{PMe}_3]$	k_{obs}	$1/[\text{PMe}_3]$	$1/k$
0.0053	4.32E-06	188.68	2E+05
0.027	1.12E-05	37.037	89200
0.109	1.56E-05	9.1743	64035
0.574	1.95E-05	1.7422	51338

			95% error
$k_1/k_2 =$	slope =	948.32	42.17
$1/k_1 =$	intercept =	52928.47	4059

$k_1 =$	1.89E-05	1.4E-06
$k_1/k_2 =$	0.017917	0.0008

