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¹H NMR Spectral Data for the (CO)₅MnC(O)CH₃ (2a)-Catalyzed PhSiH₃ Hydrosilation of FpC(O)CH₃ (1a).

The chemical shift regions that were assigned for each of 3b-5b (Table 1) agree with the results of COSY NMR experiments (500 MHz spectrometer). These results helped establish a down-field progression of the methine quartets, the Cp singlets, and the methyl doublets in their ¹H NMR spectra in going from **5b** to **4b** to **3b**. This spectral dispersion is illustrated in Figure 1 (Scan B) for the **2a**-catalyzed PhSiH₃ hydrosilation of **1a**.

The ¹H NMR spectrum of isolated [FpCH(CH₃)O]₃SiPh (**5b**) is reproduced as Scan C in Figure 1. It resolves four methine quartets between δ 6.20-6.40 (although the overlapping downfield pair are separated by only 2.2 Hz), four Cp singlets, δ 4.41-4.17, and four methyl doublets, δ 2.07-1.97. The results of COSY experiments further correlated these methine and methyl absorptions and were used to assign the four diastereomers labeled A-D in Table 1.

This NMR spectral data are consistent with a constitutionally symmetrical molecule that has three stereogenic FpCHMeO groups branching off of an achiral silicon center. The resulting four stereoisomers constitute a diastereomeric pair of enantiomers [RRR(SSS) and RRS(SSR)].³² Differences in the relative populations of these diastereomers as well as the magnetic (non)equivalency of their FpCHMeO groups accounts for both the presence of four ¹H and ¹³C NMR spectral signals and their 1:1:1:1 relative intensities for the FpCHMeO groups. For example, the ¹³C{¹H} NMR spectrum of **5b** reveals four resonances for the Cp, methine, and methyl absorptions, in addition to eight carbonyl resonances for the diastereotopic Fe(CO)₂ centers.

The presence of a C₃ rotational axis on the RRR(SSS) enantiomers of **5b** ensures symmetrically and magnetically equivalent FpCHMeO groups and e.g. a single Cp resonance. Since the RRS(SSR) diastereomers retain three chiral centers that are not related by any symmetry operation, three magnetically nonequivalent FpCHMeO groups, e.g. three Cp resonances, are expected. These diastereomers by virtue of the constitutional symmetry also represent a relative population of six equivalent or enantiometric stereoisomers vs. the two RRS(SSR) enantiomers, hence the net 1:1:1:1 relative intensities for the Cp (and other FpCHMeO) ¹H and ¹³C NMR spectral resonances of **5b**. © 1998 American Chemical Society Organometallics V17 Page 1993 Mao Supplemental Page 2

Figure 1. ¹H NMR Spectra of (CO)₅MnC(O)CH₃ (2a)-Catalyzed PhSiH₃ Hydrosilation of FpC(O)CH₃ (1a).

<u>Scan A</u> (200 MHz) for the reaction mixture of **1a** (0.91 mmol), **2a** (4.6 mol%), and PhSiH₃ (1.09 mmol) in 600 mg of C₆D₆ after 8 h. Intense singlets at δ 4.22 and 4.17 represent unreacted **1a** (also, δ 2.40) and PhSiH₃, respectively, and the singlet at δ 2.26 pinpoints remaining **2a**. Methine quartets for [FpCH(CH₃)O]₃SiPh (**5b**) (6.36-6.18) and for [FpCH(CH₃)O]₂SiHPh (**4b**) (6.07, 6.06) partially overlap, and that of FpCH(CH₃)OSiH₂Ph (**3b**) is at δ 5.73. The higher field Cp resonances for **4b** (δ 4.31, 4.29) appear as a shoulder on the downfield Cp resonances for **5b** (δ 4.44-4.33); the Cp resonance for **3b** (δ 4.24) overlaps that of **2a**. The upfield methyl doublet of **5b** (δ 2.06-1.98) overlaps the downfield doublet of **4b** (δ 1.93, 1.85); and the methyl doublet of **3b** appears at δ 1.78. Assignments for the methine and methyl ligands were confirmed by COSY NMR experiments.

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<u>Scan B</u> (200 MHz) is for a similar reaction mixture after 12 h; anisole (δ 3.35) is present as an internal integration standard. In addition to **5b** and remaining PhSiH₃, **6a** (δ 4.12, 1.58, 1.35) is clearly evident.

<u>Scan C</u> (200 MHz with 500 MHz insets) of $[FpCH(CH_3)O]_3$ SiPh (5b) after isolation by size exclusion chromatography.

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¹H NMR Spectral Data for the (PPh₃)(CO)₄MnC(O)CH₃ (2c)-Catalyzed PhSiH₃ Hydrosilation of FpC(O)CH₃ (1a).

The precatalyst (PPh₃)(CO)₄MnC(O)CH₃ (2c) differed in activity from 2a and 2b in that PhSiH₃ hydrosilation of 1a yielded [FpCH(CH₃)O]₂SiHPh (4b) as the major product. The use of 2c (3.0 mol%) as the precatalyst consumed 1a within 0.5 hour and selectively yielded 4b and 6a as the only NMR detectable products. With 0.50 equivalents of PhSiH₃ under otherwise identical conditions, 1a transformed to mixtures containing 73% 4b plus 6% 6a. Figure 2 illustrates the ¹H NMR spectrum for this reaction; the δ 4.3-4.0 region shows the Cp resonances of 4b (δ 4.25-4.18), the absence of PhSiH₃ (δ 4.22), and the Cp resonance for 6a (δ 4.02).

An eight-line multiplet appears in the ¹H NMR spectrum of **4b** (500 MHz) at δ 6.10-6.00 that is a composite of four partially overlapping methine quartets. These methine quartets are coupled to the two multiplets that appear for the methyl groups, an apparent triplet at δ 1.96 (which contains two methyl doublets, δ 1.96 and 1.95) and two partially overlapping doublets centered at δ 1.87. COSY NMR experiments further revealed spin correlations between the δ 6.04 and 6.01 methine quartets with the δ 1.96 and 1.95 methyl doublets, and between the 6.07 and 6.05 quartets with the δ 1.88 and 1.87 doublets. HMQC experiments connected these four methyl doublets and four methine quartets with sets of four ¹³C NMR absorptions between δ 70.20-69.40 and between δ 35.73-35.43, respectively. Although four Cp resonances also were detected in the ¹H NMR spectrum, only three were observed in the ¹³C{¹H} NMR spectra (the δ 86.24 absorption, however, is broadened). The SiH group absorbs as two singlets at δ 5.45 and 5.40 with 1:3 relative intensities. ©1998 American Chemical Society Organometallics V17 Page 1993 Mao Supplemental Page 5

Figure 2. ¹H NMR Spectra of $(PPh_3)(CO)_4MnC(O)CH_3(2c)$ -Catalyzed PhSiH₃ Hydrosilation of FpC(O)CH₃ (1a).

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<u>Scan A</u> (500 MHz) for the reaction mixture of **1a** (0.20 mmol), **2c** (3.0 mol%), PhSiH₃ (0.10 mmol), and toluene (0.09 mmol) in 500 mg of C₆D₆ after 0.5 h. The weak singlet at δ 4.22 indicates residual PhSiH₃, and the absence of singlets at δ 4.17, 2.40 demonstrates that all **1a** has been consumed. Insets show expanded absorptions for [FpCH(CH₃)O]₂SiHPh (**4b**): Cp region (δ 4.25-4.18); four overlapping methine quartets (δ 6.10-6.00) converted to four singlets after double irradiation at δ 2.1; two partially overlapping methyl doublets (δ 1.96, 1.95) and two methyl doublets (δ 1.88, 1.87) converted to four singlets after double

<u>Scan B</u> (500 MHz with expanded insets) of 4b after isolation by size exclusion chromatography.



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Hydrosilation of $(CO)_5 MnC(O)CH_3$ (2a) with PhSiH₃.

Although the PhSiH₃ hydrosilation of **1a** required the presence of a precatalyst such as $(CO)_5MnC(O)CH_3$ (**2a**), this precatalyst largely remained intact until the substrate was consumed. In order to probe this reactivity of **2a**, we briefly studied its hydrosilation by PhSiH₃. Treatment of **2a** with just one equivalent of PhSiH₃ - no other precatalyst was added - initiated an immediate and exothermic reaction. This reaction consumed all of the manganese acetyl within 15 minutes, as ascertained by NMR spectroscopy. The vigor of this reaction, however, was matched by its product decomposition in these dark brown solutions, and the resulting paramagnetic broadening precluded further analysis of the NMR spectra.

In order to circumvent this decomposition, we studied the PhSiH₃ hydrosilation of 2 a at 5 °C. The substrate was consumed within an hour, and continued NMR spectral monitoring of the orange solutions was possible for several hours before the onset of paramagnetic broadening. These ¹H and ¹³C {¹H} NMR spectra are consistent with a reaction that converted 2a to a mixture of $[(CO)_5MnCH(CH_3)O]_xSiH_{3-x}Ph$ (3c, x = 1; 4c, x = 2; 5c, x = 3) and then only slowly to $(CO)_5MnCH_2CH_3$ (6c) (12%, 2.5 h). The ethoxyphenylsilanes H_xSi(EtO)_{3-x}Ph (x = 1-3) were not detected, although over time the reaction mixtures degraded and left moderate yields of Mn₂(CO)₁₀ as the only identifiable organomanganese species.¹d



¹H NMR spectra of the **3c-4c-5c** mixtures exhibited complex multiplets at δ 5.49 and 5.27 for methine hydrogens that are coupled to a series of six methyl doublets, δ 1.99-1.82. Five of these methyl doublets are immediately apparent at 500 MHz (confirmed by results of double irradiation of the methine multiplets), and we inferred the sixth from relative intensities as two overlapping doublets at δ 1.99. Other major absorptions were accounted for by a SiH singlet (δ 5.21), unreacted PhSiH₃ (δ 4.22), and the ethyl multiplets for **6c** (δ 1.38, q; 0.93, t, J = 7.7 Hz). The ¹³C {¹H} NMR spectra likewise are dominated by the presence of closely spaced methine and methyl resonances, δ 69-70 and δ 34-35, respectively, phenyl absorptions, and the ethyl carbons for **6c** (δ 21.06 and 0.05).

Assignment of the mono-, bis-, and tris-manganese α -siloxyethyl complexes 3c-5cwas complicated by the absence of adequate NMR spectral dispersion as was observed for their Fp analogs 3b-5b. Thus absorptions of 3c-5c within the methine and methyl chemical shift regions overlap extensively in their ¹H and ¹³C NMR spectra. Individual methine quartets in the ¹H NMR spectrum of 4c and 5c superimpose (even at 500 MHz), and ¹³C NMR absorptions within the methine and methyl chemical shift regions overlapped to the extent that we only detected four of the methine and five of the methyl resonances.

The results of COSY NMR experiments, however, correlated the methyl doublets with the methine complex multiplets at δ 5.49 and 5.27. Three of the doublets at δ 1.99 and 1.92 couple with the downfield methine multiplet, and the remaining three doublets at δ 1.95, 1.89, and 1.82 correlate with the upfield methine multiplet. We *tentatively* assigned the latter two doublets to [(CO)₅MnCH(CH₃)O]₂SiHPh (4c) on the basis of (a) their relative intensities with respect to the SiH singlet and (b) the upfield chemical shifts of both its methine and methyl absorptions. This NMR spectral data thus represents a 1:1.7 mixture of 4c and 5c after 1-2 h.

The hydrosilation of 2a with PhSiH₃ differed appreciably from those reactions using R₂SiH₂ or R₃SiH, which furnished α -siloxyethyl complexes as the final organomanganese products.¹ With Et₂SiH₂, for example, mixtures of (CO)₅MnCH(OSiHEt₂)CH₃ (3d) and [(CO)₅MnCH(CH₃)O]₂SiEt₂ (4d) resulted, whereas reactions using Ph₂SiH₂ provided only (CO)₅MnCH(OSiHPh₂)CH₃ (4e).^{1b} Although unstable, these dihydrosilane adducts of 2a were unambiguously identified by NMR spectroscopy. [Although the RhCl(PPh₃)₃-catalyzed Ph₂SiH₂ hydrosilation of 2a was reported³ to give 4e, the presence of RhCl(PPh₃)₃ is superfluous.] Treatment of 2a with Me₂PhSiH, on the other hand, afforded the fully characterized (CO)₅MnCH(OSiMe₂Ph)CH₃ (eq 1), which is stable in the absence of excess HSiMe₂Ph.^{1d} Its reaction with additional Me₂PhSiH gave degradation product(s) that serve as active catalysts for the hydrosilation of ketones,^{5c} esters,^{5b} 1a,^{1b,c,e} and even 2a.^{1 d}

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Unlike the reaction between 2a and PhSiH₃, those using (excess) R_2SiH_2 or R_3SiH do not provide the manganese ethyl complex 6c.

The hydrosilation of 2a and its use as a precatalyst for the hydrosilation of 1a are related.^{1d,e,4a} The observation that most of the precatalyst 2a remained intact as it promoted the PhSiH₃ hydrosilation of 1a is consistent with our previously observed hydrosilation chemistry. In kinetics studies to be published, we will document that both 1a and especially FpC(O)Ph (which is a poor hydrosilation substrate) inhibit the Me₂PhSiH hydrosilation of 2a as well as the 2a-catalyzed hydrosilation of 1a.^{1e}

Hydrosilation of (CO)5MnC(O)CH₃ (2a) with PhSiH₃. An NMR tube containing 100 mg of 2a (0.42 mmol) and 45 mg of C₆H₅OCH₃ (0.42 mmol) in 600 mg of C₆D₆ was cooled to 5 °C before PhSiH₃ (118 mg, 1.09 mmol) was added by syringe. The resulting light yellow solution turned orange over 1 hour. Within this hour, the starting 2a was consumed, as determined by ¹H and ¹³C {¹H} NMR spectroscopy. ¹H NMR double irradiation and COSY experiments of this solution were carried out in order to facilitate the tentative assignments, including that of (CO)₅MnCH₂CH₃:^{40 1}H NMR (C₆D₆) δ 1.38 (q, J = 7.6 Hz, MnCH₂), 0.93 (t, CH₃); ¹³C NMR δ 21.06 (CH₃), 0.05 (MnCH₂). (CO)₅MnCH(CH₃)O]₂SiPhH (4c) ¹H NMR $(C_6D_6) \delta$ 7.91 (m, 2H, Ph-4c), 7.30 (m, 2H, Ph-4c), 7.22 (m, 4H, Ph-4c+5c), 5.49 (m, width=90 Hz, MnCH-4c+5c), 5.27 (m, width=80 Hz, MnCH-4c+5c), 5.21 (s, SiH, 4c), 1.89 (d, J=5.9 Hz, CH₃), 1.82 (d, J=5.9 Hz, CH₃); ¹³C NMR δ 212.3 (br s, CO, 4c+5c), 70.05, 69.88, 69.70, 69.34 (MnC, 4c+5c), 35.64, 35.35, 35.06, (CH₃, 4c+5c), 34.90, 34.81 (CH₃, 4c+5c). (CO)₅MnCH(CH₃)O]₃SiPh (5c) ¹H NMR (C₆D₆) δ 7.76 (m, 2H, Ph-5c), 7.22 (m, 4H, Ph-4c+5c), 5.49 (m, width=90 Hz, MnCH-4c+5c), 5.27 (m, width=80 Hz, MnCH-4c+5c), 1.99 (d, J=6.3 Hz, CH₃), 1.99 (d, J=6.3 Hz, CH₃), 1.95 (d, J=6.3 Hz, CH₃), 1.92 (d, J=6.3 Hz, CH₃). Results of COSY experiments: methine multiplet at δ 5.49 correlates with methyl doublets δ 1.99, 1.92 (5c), and methine multiplet at δ 5.27 correlates with methyl doublets δ 1.95, 1.99, 1.92 (5c).

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