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s1161a C18 H17 C1 N2 Pd

******** Complex 1a *********

SUPPLEMENTARY MATERIAL

BELONGING TO THE PAPER

Insertion of Allenes into Palladium-Carbon Bonds of Complexes

Containing Bidentate Nitrogen Ligands. Mechanism and Structural

Studies.

bу

J.G.P. Delis, J.H. Groen, K. Vrieze & P.W.N.M. van Leeuwen

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Table S1 - Crystal Data and Details of the Structure Determination for: s1161a C18 H17 C1 N2 Pd

Crystal Data

Formula Weight 403.22 Crystal System Triclinic Space group P-1 (No. 2) a, b, c [Angstrom] 9.2427(13) 11.211(2) 15.966(8) alpha, beta, gamma [deg] 88.19(3) 88.76(3) 89.728(12) V [Ang**3] 1653.2(9) Z 4 D(calc) [g/cm**3] 1.620 F(000) [Electrons] 808 Mu(MoKa) [/cm] 12.8 Crystal Size [mm] 0.08 x 0.20 x Data Collection 150 Radiation [Angstrom] MoKa (with monochromator) 0.71073 Theta Min-Max [Deg] 1.3, 27.5
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Scan type, Scan, [Deg] Omega/2Theta, 0.84 + 0.35 Tan(Theta)
Hor. and vert. aperture [mm] 3.03 4.00
Reference Reflection(s) -2, 2, 1 ; -2, -2, 5 ; -1, -3, 2
Dataset -12: 12 ; -14: 14 ; -13: 20
Tot., Uniq. Data, R(int) 9113, 7582, 0.0382
Observed data [I > 2.0 sigma(I)] 5495
DIFABS transmission range 0.676, 1.000
Refinement 7574 399
NIEL, Npar 7574, 555
R_{1} WR, S U.U.32, U.1002, 1.14
W = 1/[S 2 (FO 2)] + (0.0202F) 2 + 24.250F] WHELE F = (FO 2 + 2FC 2)/3 Move and Dv. Chift (Error
$Min and Max read done [o/hng^3] = 1.21 1.59$

Table S2 - Final Coordinates and Equivalent Isotropic Thermal Parameters of the non-Hydrogen atoms for: s1161a C18 H17 Cl N2 Pd

Atom	x	У	Z 	U(eq) [Ang^2]
pd(1)	0.55769(7)	0.86840(6)	0.18487(4)	0.0205(2)
Cl(1)	0.6792(2)	0.8072(2)	0.06133(13)	0.0255(6)
N(11)	0.3525(7)	0.8529(6)	0.1302(4)	0.019(2)
N(12)	0.1086(8)	0.7669(6)	0.3184(5)	0.024(2)
C(11)	0.3182(9)	0.9246(8)	0.0640(5)	0.021(2)
C(12)	0.1864(9)	0.9220(8)	0.0257(5)	0.023(3)
C(13)	0.0846(9)	0.8401(8)	0.0545(6)	0.026(3)
C(14)	0.1164(10)	0.7696(9)	0.1236(6)	0.029(3)
C(15)	0.2512(9)	0.7743(8)	0.1579(5)	0.020(2)
C(16)	0.2911(9)	0.6883(7)	0.2265(6)	0.022(3)
C(17)	0.3987(10)	0.6066(9)	0.2122(6)	0.033(3)
C(18)	0.4326(12)	0.5170(9)	0.2720(7)	0.043(4)
C(19)	0.3530(12)	0.5075(9)	0.3463(7)	0.040(3)
C(110)	0.2449(10)	0.5909(8)	0.3641(6)	0.027(3)
C (111)	0.1623(10)	0.5882(9)	0.4395(6)	0.030(3)
C(112)	0.0581(10)	0.6710(8)	0.4535(6)	0.030(3)
C(113)	0.0390(10)	0.7591(8)	0.3895(6)	0.028(3)
C(114)	0.2137(9)	0.6826(8)	0.3034(5)	0.022(2)
C (115)	0.6246(11)	0.8763(9)	0.3103(6)	0.030(3)
C(116)	0.7370(11)	0.9101(9)	0.2553(6)	0.033(3)
C (117)	0.4971(12)	0.9437(10)	0.2991(6)	0.040(4)
C(118)	0.6276(11)	0.7636(9)	0.3636(6)	0.036(3)
Pd(2)	1.08967(7)	0.18381(6)	0.29535(4)	0.0216(2)
Cl(2)	1.2252(2)	0.2157(2)	0.41718(14)	0.0279(7)
N(21)	0.8892(7)	0.1719(6)	0.3627(4)	0.0176(19)
N(22)	0.6499(8)	0.2365(7)	0.1833(5)	0.024(2)

Table S2 - Final Coordinates and Equivalent Isotropic Thermal Parameters of the non-Hydrogen atoms (continued) for: s1161a C18 H17 C1 N2 Fd

Atom	x	у	z	U(eq) [Ang^2]
C(21)	0.8743(10)	0.0880(8)	0.4246(5)	0.025(3)
C(22)	0.7475(11)	0.0655(8)	0.4675(6)	0.031(3)
C(23)	0.6270(11)	0.1327(9)	0.4480(6)	0.035(3)
C(24)	0.6404(10)	0.2198(8)	0.3858(6)	0.027(3)
C(25)	0.7728(9)	0.2384(7)	0.3450(5)	0.020(2)
C(26)	0.7936(8)	0.3391(7)	0.2812(5)	0.017(2)
C(27)	0.8717(10)	0.4369(8)	0.3018(5)	0.024(3)
C(28)	0.8805(10)	0.5377(8)	0.2469(6)	0.026(3)
C(29)	0.8132(9)	0.5376(7)	0.1722(6)	0.025(3)
C(210)	0.7358(9)	0.4373(8)	0.1479(5)	0.022(3)
C(211)	0.6669(10)	0.4326(9)	0.0700(6)	0.030(3)
C(212)	0.5937(10)	0.3328(9)	0.0507(6)	0.030(3)
C(213)	0.5876(10)	0.2365(8)	0.1104(6)	0.028(3)
C(214)	0.7251(9)	0.3358(7)	0.2031(5)	0.020(2)
C(215)	1.1446(10)	0.2080(9)	0.1657(6)	0.031(3)
C(216)	1.2652(10)	0.1715(11)	0.2127(6)	0.039(3)
C(217)	1.0243(11)	0.1315(9)	0.1751(5)	0.033(3)
C(218)	1.1328(13)	0.3321(10)	0.1261(7)	0.043(4)

U(eq) = 1/3 of the trace of the orthogonalized U

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Jable S3 - Hydrogen Atom Positions and Isotropic ThermalParametersfor: sll61aC18 H17 C1 N2 Pd

Atom	x	У	z	U(iso) [Ang^2]
H(11)	0.3873(9)	0.9785(8)	0.0432(5)	0.0250
H(12)	0.1664(9)	0.9745(8)	-0.0188(5)	0.0280
H(13)	-0.0033(9)	0.8329(8)	0.0278(6)	0.0300
H(14)	0.0466(10)	0.7185(9)	0.1472(6)	0.0340
H(17)	0.4506(10)	0.6106(9)	0.1616(6)	0.0390
H(18)	0.5083(12)	0.4641(9)	0.2617(7)	0.0510
H(19)	0.3717(12)	0.4453(9)	0.3846(7)	0.0480
H(111)	0.1796(10)	0.5288(9)	0.4801(6)	0.0360
H(112)	0.0022(10)	0.6699(8)	0.5026(6)	0.0360
H(113)	-0.0306(10)	0.8173(8)	0.3995(6)	0.0340
H(161)	0.8188(11)	0.8558(9)	0.2520(6)	0.0400
H(162)	0.7647(11)	0.9934(9)	0.2549(6)	0.0400
H(171)	0.5062(12)	1.0296(10)	0.3020(6)	0.0470
H(172)	0.4100(12)	0.9125(10)	0.3266(6)	0.0470
H(181)	0.688(6)	0.706(2)	0.337(2)	0.0540
H(182)	0.666(7)	0.7803(16)	0.4174(17)	0.0540
H(183)	0.5312(15)	0.733(4)	0.371(4)	0.0540
H(21)	0.9552(10)	0.0428(8)	0.4389(5)	0.0310
H(22)	0.7425(11)	0.0057(8)	0.5091(6)	0.0380
H(23)	0.5391(11)	0.1196(9)	0.4761(6)	0.0420
H(24)	0.5608(10)	0.2663(8)	0.3711(6)	0.0330
H(27)	0.9195(10)	0.4368(8)	0.3525(5)	0.0290
H(28)	0.9324(10)	0.6043(8)	0.2619(6)	0.0310
H(29)	0.8184(9)	0.6048(7)	0.1366(6)	0.0300
H(211)	0.6717(10)	0.4974(9)	0.0322(6)	0.0370
H(212)	0.5486(10)	0.3277(9)	-0.0006(6)	0.0360

Table S3 - Hydrogen Atom Positions and Isotropic Thermal Parameters (continued) for: slibla Cl8 H1/ Cl N2 Pd

Atom	x	У	z	U(iso) [Ang^2]
H(213)	0.5360(10)	0.1690(8)	0.0969(6)	0.0330
H(261)	1.3410(10)	0.2301(11)	0.2180(6)	0.0470
H(262)	1.3018(10)	0.0917(11)	0.2030(6)	0.0470
H(271)	1.0403(11)	0.0486(9)	0.1619(5)	0.0390
H(272)	0.9325(11)	0.1630(9)	0.1559(5)	0.0390
H(281)	1.167(8)	0.3309(18)	0.0689(16)	0.0650
H(282)	1.191(7)	0.3863(17)	0.157(3)	0.0650
H(283)	1.0336(17)	0.358(3)	0.128(4)	0.0650

The Temperature Factor has the Form of Exp(-T) Where T = 8*(Pi**2)*U*(Sin(Theta)/Lambda)**2 for Isotropic Atoms

			- 8 -			
Table S	4 - (An)isotr for: s11	opic Ther 61a C18	mal Paramet H17 Cl N2 P	ers (contin d	ued)	
Atom	U(1,1) or U	U(2,2)	U(3,3)	U(2,3)	U(1,3)	U(1,2)
C(21)	0.029(5)	0.023(4)	0.024(5)	0.003(4)	-0.004(4)	0.002(
C(22)	0.044(6)	0.022(5)	0.028(5)	0.010(4)	-0.004(4)	-0.008(
C(23)	0.033(5)	0.040(6)	0.032(5)	-0.005(4)	0.011(4)	-0.011(
C(24)	0.023(5)	0.017(4)	0.041(5)	0.001(4)	0.006(4)	-0.002 (
C(25)	0.023(4)	0.022(4)	0.016(4)	-0.004(3)	0.001(3)	-0.004
C(26)	0.016(4)	0.020(4)	0.015(4)	-0.002(3)	0.003(3)	0.004
C(27)	0.029(5)	0.021(4)	0.023(4)	0.002(3)	-0.004(4)	-0.007
C(28)	0.028(5)	0.017(4)	0.032(5)	-0.002(4)	-0.003(4)	-0.013
C(29)	0.027(5)	0.015(4)	0.034(5)	0.006(4)	0.000(4)	-0.004
C(210)	0.017(4)	0.032(5)	0.018(4)	0.004(3)	0.003(3)	0.002
C(211)	0.023(5)	0.039(6)	0.029(5)	0.003(4)	0.002(4)	0.002
C(212)	0.023(5)	0.048(6)	0.020(4)	0.001(4)	-0.002(4)	-0.002
C(213)	0.025(5)	0.028(5)	0.031(5)	-0.012(4)	0.003(4)	-0.008
C(214)	0.014(4)	0.020(4)	0.027(4)	-0.005(3)	0.002(3)	0.002
C(215)	0.029(5)	0.045(6)	0.019(4)	-0.001(4)	0.011(4)	0.010
C(216)	0.022(5)	0.059(7)	0.036(6)	-0.016(5)	0.011(4)	0.012
C(217)	0.037(5)	0.043(6)	0.019(4)	-0.019(4)	-0.002(4)	0.013
C(218)	0.048(7)	0.051(7)	0.028(5)	0.009(5)	0.012(5)	0.005

The Temperature Factor has the Form of Exp(-T) Where T = 8*(Pi**2)*U*(Sin(Theta)/Lambda)**2 for Isotropic Atoms T = 2*(Pi**2)*Sumij(h(i)*h(j)*U(i,j)*Astar(i)*Astar(j)), for Anisotropic Atoms. Astar(i) are Reciprocal Axial Lengths and h(i) are the Reflection Indices.

Table S4 - (An)isotropic Thermal Parameters for: s1161a C18 H17 C1 N2 Pd

Atom	U(1,1) or U	U U(2,2)	U(3,3)	U(2,3)	U(1,3)	U(1,2)
Pd(1)	0.0196(3)	0.0195(3)	0.0224(4)	0.0009(3)	-0.0041(3)	-0.0030(3)
Cl(1)	0.0214(10)	0.0305(11)	0.0242(11)	0.0059(9)	0.0021(8)	-0.0028(8)
N(11)	0.015(3)	0.021(4)	0.023(4)	-0.009(3)	-0.001(3)	-0.002(3)
N(12)	0.022(4)	0.019(4)	0.031(4)	-0.005(3)	0.000(3)	0.001(3)
C(11)	0.021(4)	0.025(4)	0.016(4)	0.001(3)	0.001(3)	-0.004(3)
C(12)	0.026(5)	0.023(4)	0.021(4)	0.003(3)	-0.003(3)	0.002(3)
C(13)	0.015(4)	0.035(5)	0.027(5)	-0.006(4)	-0.002(3)	-0.005(4)
C(14)	0.025(5)	0.033(5)	0.028(5)	-0.001(4)	0.000(4)	-0.003(4)
C(15)	0.018(4)	0.024(4)	0.017(4)	-0.005(3)	0.007(3)	-0.003(3)
C(16)	0.019(4)	0.013(4)	0.034(5)	-0.001(3)	0.003(3)	-0.005(3)
C(17)	0.033(5)	0.031(5)	0.033(5)	0.008(4)	0.014(4)	0.006(4)
C(18)	0.044(6)	0.033(6)	0.049(7)	0.015(5)	0.025(5)	0.017(5)
C(19)	0.043(6)	0.033(5)	0.042(6)	0.020(5)	0.014(5)	0.013(5)
C(110)	0.024(5)	0.022(4)	0.036(5)	0.005(4)	0.003(4)	-0.006(4)
C(111)	0.034(5)	0.030(5)	0.026(5)	0.004(4)	0.000(4)	0.002(4)
C(112)	0.030(5)	0.031(5)	0.029(5)	-0.011(4)	0.007(4)	-0.007(4)
C(113)	0.024(5)	0.027(5)	0.034(5)	-0.015(4)	0.001(4)	-0.002(4)
C(11 4)	0.020(4)	0.020(4)	0.027(4)	0.000(3)	-0.001(3)	-0.003(3)
C(115)	0.033(5)	0.032(5)	0.027(5)	-0.011(4)	-0.008(4)	0.003(4)
C(116)	0.036(6)	0.030(5)	0.035(5)	-0.004(4)	-0.009(4)	-0.008(4)
C(117)	0.038(6)	0.048(7)	0.034(6)	-0.011(5)	-0.012(5)	0.008(5)
C(118)	0.035(6)	0.046(6)	0.027(5)	0.005(4)	-0.006(4)	0.009(5)

 Pd(2)
 0.0203(3)
 0.0239(4)
 0.0205(3)
 -0.0016(3)
 0.0008(3)
 0.0041(3)

 C1(2)
 0.0261(11)
 0.0328(12)
 0.0246(11)
 0.0042(9)
 -0.0035(9)
 -0.0013(9)

 N(21)
 0.025(4)
 0.016(3)
 0.012(3)
 -0.002(3)
 -0.001(3)
 0.000(3)

 N(22)
 0.022(4)
 0.026(4)
 0.024(4)
 -0.003(3)
 0.003(3)
 -0.006(3)

Table S5 - Bond Distances (Angstrom) (continued) for: s1161a C18 H17 Cl N2 Pd

C(26)	-C(214)	1.412(11)	C(24)	-Н(24)	0.930(13)
C(27)	-C(28)	1.410(13)	C(27)	-н(27)	0.931(12)
C(28)	-C(29)	1.357(13)	C(28)	-H(28)	0.930(13)
C(29)	-C(210)	1.405(12)	C(29)	-H(29)	0.930(12)
C(210)	-C(211)	1.412(12)	C(211)	-H(211)	0.930(14)
C(210)	-C(214)	1.420(12)	C(212)	-Н(212)	0.931(13)
C(211)	-C(212)	1.357(14)	C(213)	-Н(213)	0.930(13)
C(212)	-C(213)	1.418(14)	C(216)	-H(261)	0.970(15)
C(215)	-C(216)	1.409(13)	C(216)	-Н(262)	0.971(17)
C(215)	-C(217)	1.410(14)	C(217)	-н(271)	0.969(14)
C(215)	-C(218)	1.514(15)	C(217)	-н(272)	0.970(14)
C(21)	-H(21)	0.930(13)	C(218)	-н(281)	0.96(4)
C(22)	-Н(22)	0.929(13)	C(218)	-н(282)	0.97(5)
C(23)	-н(23)	0.930(14)	C(218)	-H(283)	0.96(2)

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Table S5 - Bond Distances (Angstrom)for:\$1161aC18H17C1N2Pd

Pd(1)	-Cl(1)	2.367(2)	C(14)	-H(14)	0.930(14)
Pd(1)	-N(11)	2.115(7)	C(17)	-H(17)	0.931(13)
Pd(1)	-C(115)	2.113(10)	C(18)	-H(18)	0.930(15)
Pd(1)	-C(116)	2.086(10)	C(19)	-н(19)	0.931(15)
Pd(1)	-C(117)	2.098(10)	C(111)	-H(111)	0.930(14)
N(11)	-C(11)	1.350(11)	C(112)	-H(112)	0.929(13)
N(11)	-C(15)	1.347(11)	C(113)	-H(113)	0.929(13)
N(12)	-C(113)	1.293(12)	C(116)	-H(161)	0.970(14)
N(12)	-C(114)	1.375(11)	C(116)	-H(162)	0.969(14)
C(11)	-C(12)	1.376(12)	C(117)	-H(171)	0.970(16)
C(12)	-C(13)	1.379(12)	C(117)	-H(172)	0.970(15)
C(13)	-C(14)	1.373(13)	C(118)	-H(181)	0.96(4)
C(14)	-C(15)	1.374(12)	C(118)	-H(182)	0.96(4)
C(15)	-C(16)	1.487(12)	C(118)	-H(183)	0.96(2)
C(16)	-C(17)	1.369(13)	Pd(2)	-C1(2)	2.374(3)
C(16)	-C(114)	1.407(12)	Pd(2)	-N(21)	2.125(7)
C(17)	-C(18)	1.403(15)	Pd(2)	-C(215)	2.130(10)
C(18)	-C(19)	1.384(16)	Pd(2)	-C(216)	2.076(9)
C(19)	-C(110)	1.396(14)	Pd(2)	-C(217)	2.126(8)
C(110)	-C(111)	1.411(13)	N(21)	-C(21)	1.348(11)
C(110)	-C(114)	1.423(13)	N(21)	-C(25)	1.336(10)
C(111)	-C(112)	1.354(13)	N(22)	-C(213)	1.310(12)
C(112)	-C(113)	1.412(13)	N(22)	-C(214)	1.364(11)
C(115)	-C(116)	1.392(14)	C(21)	-C(22)	1.365(13)
C(115)	-C(117)	1.408(15)	C(22)	~C(23)	1.377(14)
C(115)	-C(118)	1.501(14)	C(23)	-C(24)	1.375(14)
C(11)	-H(11)	0.930(12)	C(24)	-C(25)	1.387(12)
C(12)	-H(12)	0.929(12)	C(25)	-C(26)	1.507(11)
C (13)	-H(13)	0.931(12)	C(26)	-C(27)	1.368(12)

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Table	S6 - Bond for	d Angles : s1161a	(Degrees) C18 H17 Cl	N2 Pd				Table S	6 - Bon for	d Angles : s1161a	(Degrees) C18 H17 Cl	(continu N2 Pd	ued)		
Cl(1)	-Pd(1)	-N(11)	92.22(18)	C(111)	-C(110)	-C(114)	117.7(8)	C(110)	-C(111)	-H(111)	119.6(11)	C(215)	-Pd(2)	-C(216)	39.1(
Cl(1)	-Pđ(1)	-C(115)	131.9(3)	C(110)	-C(111)	-C(112)	120.8(9)	C(112)	-C(111)	-H(111)	119.6(11)	C(215)	-Pd(2)	-C(217)	38.7(
Cl(1)	-Pd(1)	-C(116)	99.1(3)	C(111)	-C(112)	-C(113)	116.4(9)	C(111)	-C(112)	-H(112)	121.8(11)	C(216)	-Pd(2)	-C(217)	68.6(
Cl(1)	-Pd(1)	-C(117)	166.2(3)	N(12)	-C(113)	-C(112)	126.6(8)	C(113)	-C(112)	-H(112)	121.8(11)	Pd(2)	-N(21)	-C(21)	118.6(
N(11)	-Pd(1)	-C(115)	132.9(3)	N(12)	-C(114)	-C(16)	118.8(8)	N(12)	-C(113)	-H(113)	116.8(11)	Pd(2)	-N(21)	-C(25)	124.2(
N(11)	-Pd(1)	-C(116)	167.6(3)	N(12)	-C(114)	-C(110)	121.1(8)	C(112)	-C(113)	-H(113)	116.7(11)	C(21)	-N(21)	-C(25)	117.0(
N(11)	-Pd(1)	-C(117)	100.1(3)	C(16)	-C(114)	-C(110)	120.1(8)	Pd(1)	-C(116)	-H(161)	116.4(10)	C(213)	-N(22)	-C(214)	117.8(
C(115)	-Pd(1)	-C(116)	38.7(4)	Pd(1)	-C(115)	-C(116)	69.6(6)	Pd(1)	-C(116)	-H(162)	116.5(10)	N(21)	-C(21)	-C(22)	123.9(
C(115)	-Pd(1)	-C(117)	39.1(4)	Pd(1)	-C(115)	-C(117)	69.9(6)	C(115)	-C(116)	-H(161)	116.4(11)	C(21)	-C(22)	-C(23)	118.9(
C(116)	-Pd(1)	-C(117)	68.1(4)	Pd(1)	-C(115)	-C(118)	119.1(7)	C(115)	-C(116)	-н(162)	116.4(11)	C(22)	-C(23)	-C(24)	118.1(
Pd(1)	-N(11)	-C(11)	119.6(5)	C(116)	-C(115)	-C(117)	113.6(9)	H(161)	-C(116)	-H(162)	113.4(13)	C(23)	-C(24)	-C(25)	120.0(
Pd(1)	-N(11)	-C(15)	123.5(5)	C(116)	-C(115)	-C(118)	123.0(9)	Pd(1)	-C(117)	-H(171)	116.5(9)	N(21)	-C(25)	-C(24)	122.0(
C(11)	-N(11)	-C(15)	116.9(7)	C(117)	-C(115)	-C(118)	122.2(9)	Pd(1)	-C(117)	-H(172)	116.5(10)	N(21)	-C(25)	-C(26)	116.7(
C(113)	-N(12)	-C(114)	117.4(8)	Pd(1)	-C(116)	-C(115)	71.7(6)	C(115)	-C(117)	-H(171)	116.5(12)	C(24)	-C(25)	-C(26)	121.2(
N(11)	-C(11)	-C(12)	123.4(8)	Pd(1)	-C(117)	-C(115)	71.1(6)	C(115)	-C(117)	-H(172)	116.5(12)	C(25)	-C(26)	-C(27)	119.3(
C(11)	-C(12)	-C(13)	118.8(8)	N(11)	-C(11)	-H(11)	118.3(9)	H(171)	-C(117)	-H(172)	113.5(14)	C(25)	-C(26)	-C(214)	120.2(
C(12)	-C(13)	-C(14)	118.2(8)	C(12)	-C(11)	-H(11)	118.3(10)	C(115)	-C(118)	-H(181)	109(2)	C(27)	-C(26)	-C(214)	120.4(
C(13)	-C(14)	-C(15)	120.1(9)	C(11)	-C(12)	-H(12)	120.5(10)	C(115)	-C(118)	-H(182)	109.4(15)	C(26)	-C(27)	-C(28)	120.6(
N(11)	-C(15)	-C(14)	122.3(8)	C(13)	-C(12)	-H(12)	120.6(10)	C(115)	-C(118)	-H(183)	110(3)	C(27)	-C(28)	-C(29)	120.0(
N(11)	-C(15)	-C(16)	117.6(7)	C(12)	-C(13)	-H(13)	120.8(11)	H(181)	-C(118)	-H(182)	109(4)	C(28)	-C(29)	-C(210)	120.9(
C(14)	-C(15)	-C(16)	120.0(8)	C(14)	-C(13)	-H(13)	120.9(11)	H(181)	-C(118)	-H(183)	110(4)	C(29)	-C(210)	-C(211)	122.7(
C(15)	-C(16)	-C(17)	119.1(8)	C(13)	-C(14)	-H(14)	120.0(11)	H(182)	-C(118)	-Н(183)	109(5)	C(29)	-C(210)	-C(214)	119.5(
C(15)	-C(16)	-C(114)	121.7(7)	C(15)	-C(14)	-H(14)	119.9(11)	C1(2)	-Pđ(2)	-N(21)	93.71(18)	C(211)	-C(210)	-C(214)	117.8(
C(17)	-C(16)	-C(114)	119.0(8)	C(16)	-C(17)	-H(17)	119.2(11)	C1(2)	-Pd(2)	-C(215)	131.2(3)	C(210)	-C(211)	-C(212)	119.5(
C(16)	-C(17)	-C(18)	121.6(9)	C(18)	-C(17)	-H(17)	119.2(11)	C1(2)	-Pd(2)	-C(216)	96.7(3)	C(211)	-C(212)	-C(213)	118.5(
C(17)	-C(18)	-C(19)	119.8(10)	C(17)	-C(18)	-H(18)	120.1(13)	C1(2)	-Pd(2)	-C(217)	163.6(3)	N(22)	-C(213)	-C(212)	124.2(
C(18)	-C(19)	-C(110)	120.3(10)	C(19)	-C(18)	-H(18)	120.1(13)	N(21)	-Pd(2)	-C(215)	133.1(3)	N(22)	-C(214)	-C(26)	119.4(
C(19)	-C(110)	-C(111)	123.2(9)	C(18)	-C(19)	-H(19)	119.9(13)	N(21)	-Pd(2)	-C(216)	168.0(3)	N(22)	-C(214)	-C(210)	122.1(
C(19)	-C(110)	-C(114)	119.1(9)	C(110)	-C(19)	-H(19)	119.9(13)	N(21)	-Pd(2)	-C(217)	100.3(3)	C(26)	-C(214)	-C(210)	118.5(

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Table S7 - Torsion Angles (Degrees) for: s1161a C18 H17 Cl N2 Pd

Cl(1)	-Pd(1)	-N(11)	-C(11)	-64.3(6)
Cl(1)	-Pd(1)	-N(11)	-C(15)	115.8(6)
C(115)	-Pd(1)	-N(11)	-C(11)	133.9(6)
C(115)	-Pd(1)	-N(11)	-C(15)	-46.0(8)
C(117)	-Pd(1)	-N(11)	-C(11)	109.5(7)
C(117)	-Pd(1)	-N(11)	-C(15)	-70.4(7)
Cl(1)	-Pd(1)	-C(115)	-C(116)	38.4(7)
Cl(1)	-Pd(1)	-C(115)	-C(117)	164.7(5)
Cl(1)	-Pd(1)	-C(115)	-C(118)	-78.9(8)
N(11)	-Pd(1)	-C(115)	-C(116)	-166.4(5)
N(11)	-Pd(1)	-C(115)	-C(117)	-40.1(8)
N(11)	-Pd(1)	-C(115)	-C(118)	76.3(9)
C(116)	-Pd(1)	-C(115)	-C(117)	126.2(9)
C(116)	-Pd(1)	~C(115)	-C(118)	-117.3(10)
C(117)	-Pd(1)	-C(115)	-C(116)	-126.2(9)
C(117)	-Pd(1)	-C(115)	-C(118)	116.5(10)
Cl(1)	~Pd(1)	-C(116)	-C(115)	-152.1(5)
C(117)	-Pd(1)	-C(116)	-C(115)	33.2(6)
N(11)	-Pd(1)	-C(117)	-C(115)	151.4(6)
C(116)	-Pd(1)	-C(117)	-C(115)	-32.9(6)
Pd(1)	-N(11)	-C(11)	-C(12)	-178.8(7)
C(15)	-N(11)	-C(11)	-C(12)	1.2(12)
Pd(1)	-N(11)	-C(15)	-C(14)	177.1(7)
Pd(1)	-N(11)	-C(15)	-C(16)	-4.7(10)
C(11)	-N(11)	-C(15)	-C(14)	-2.8(12)
C(11)	-N(11)	-C(15)	-C(16)	175.4(7)
C(114)	-N(12)	-C(113)	-C(112)	-1.1(14)
C(113)	-N(12)	-C(114)	-C(16)	179.8(8)
C(113)	-N(12)	-C(114)	-C(110)	-0.3(14)

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Table	S6 - Bond for	d Angles : s1161a	(Degrees) C18 H17 Cl I	(continu N2 Pd	ued)		
Pd(2)	-C(215)	-C(216)	68.4(5)	C(210)	-C(211)	-H(211)	120.2(11)
Pd(2)	-C(215)	-C(217)	70.5(5)	C(212)	-C(211)	-H(211)	120.3(11)
Pd(2)	-C(215)	-C(218)	118.4(7)	C(211)	-C(212)	-H(212)	120.7(12)
C(216)	-C(215)	-C(217)	114.2(9)	C(213)	-C(212)	-H(212)	120.8(11)
C(216)	-C(215)	-C(218)	122.1(9)	N(22)	-C(213)	-Н(213)	117.9(11)
C(217)	-C(215)	-C(218)	122.1(9)	C(212)	-C(213)	-Н(213)	117.9(11)
Pd(2)	-C(216)	-C(215)	72.5(6)	Pd(2)	-C(216)	-H(261)	116.4(10)
Pd(2)	-C(217)	-C(215)	70.8(5)	Pd(2)	-C(216)	-H(262)	116.3(10)
N(21)	-C(21)	-H(21)	118.0(10)	C(215)	-C(216)	-H(261)	116.3(12)
C(22)	-C(21)	-Н(21)	118.1(10)	C(215)	-C(216)	-H(262)	116.2(11)
C(21)	-C(22)	-H(22)	120.5(12)	H(261)	-C(216)	-H(262)	113.4(13)
C(23)	-C(22)	-H(22)	120.5(12)	Pd(2)	-C(217)	-H(271)	116.5(8)
C(22)	-C(23)	-н(23)	121.0(12)	Pd(2)	-C(217)	-Н(272)	116.5(9)
C(24)	-C(23)	-Н(23)	120.9(12)	C(215)	-C(217)	-H(271)	116.6(11)
C(23)	-C(24)	-H(24)	120.0(11)	C(215)	-C(217)	-Н(272)	116.5(11)
C(25)	-C(24)	-H(24)	120.0(11)	H(271)	-C(217)	-н(272)	113.5(13)
C(26)	-C(27)	-Н(27)	119.7(10)	C(215)	-C(218)	-H(281)	109.4(18)
C(28)	-C(27)	-н(27)	119.7(10)	C(215)	-C(218)	-H(282)	109(2)
C(27)	-C(28)	-н(28)	120.0(11)	C(215)	-C(218)	-н(283)	110(3)
C(29)	-C(28)	-H(28)	120.0(11)	H(281)	-C(218)	-H(282)	110(5)
C(28)	-C(29)	-Н(29)	119.6(10)	H(281)	-C(218)	-H(283)	110(6)
C(210)	-C(29)	-Н(29)	119.6(10)	H(282)	-C(218)	-H(283)	109(4)

Table S7 - Torsion Angles (Degrees) (continued) for: s1161a C18 H17 Cl N2 Pd

C(116)	-C(115)	-C(117)	-Pd(1)	55.6(8)
C(118)	-C(115)	-C(117)	-Pd(1)	-112.4(9)
Cl(2)	-Pd(2)	-N(21)	-C(21)	-58.1(6)
Cl(2)	-Pd(2)	-N(21)	-C(25)	125.6(6)
C(215)	-Pd(2)	-N(21)	-C(21)	137.0(6)
C(215)	-Pd(2)	-N(21)	-C(25)	-39.3(8)
C(217)	-Pd(2)	-N(21)	-C(21)	113.4(6)
C(217)	-Pd(2)	-N(21)	-C(25)	-62.9(7)
C1(2)	-Pd(2)	-C(215)	-C(216)	33.6(8)
Cl(2)	-Pd(2)	-C(215)	-C(217)	161.1(5)
Cl(2)	-Pd(2)	-C(215)	-C(218)	-82.2(8)
N(21)	-Pd(2)	-C(215)	-C(216)	-166.6(6)
N(21)	-Pd(2)	-C(215)	-C(217)	-39.1(7)
N(21)	-Pd(2)	-C(215)	-C(218)	77.6(9)
C(216)	-Pd(2)	-C(215)	-C(217)	127.5(9)
C(216)	-Pd(2)	-C(215)	-C(218)	-115.8(10)
C(217)	-Pd(2)	-C(215)	-C(216)	-127.5(9)
C(217)	-Pd(2)	-C(215)	-C(218)	116.7(10)
Cl(2)	-Pd(2)	-C(216)	-C(215)	-155.2(6)
C(217)	-Pd(2)	-C(216)	-C(215)	32.2(6)
N(21)	-Pd(2)	-C(217)	-C(215)	152.1(5)
C(216)	-Pd(2)	-C(217)	-C(215)	-32.5(6)
Pd(2)	-N(21)	-C(21)	-C(22)	-174.5(7)
C(25)	-N(21)	-C(21)	-C(22)	2.1(12)
Pd(2)	-N(21)	-C(25)	-C(24)	173.9(6)
Pd(2)	-N(21)	-C(25)	-C(26)	-8.9(10)
C(21)	-N(21)	-C(25)	-C(24)	-2.4(12)
C(21)	-N(21)	-C(25)	-C(26)	174.7(7)
C(214)	-N(22)	-C(213)	-C(212)	0.2(14)

Table S7 - Torsion Angles (Degrees) (continued) for: s1161a C18 H17 C1 N2 Pd

N(11)	-C(11)	-C(12)	-C(13)	-1.8(13)
C(11)	-C(12)	-C(13)	-C(14)	3.9(13)
C(12)	-C(13)	-C(14)	-C(15)	-5.6(14)
C(13)	-C(14)	-C(15)	-N(11)	5.2(14)
C(13)	-C(14)	-C(15)	-C(16)	-173.0(8)
N(11)	-C(15)	-C(16)	-C(17)	-63.7(11)
N(11)	-C(15)	-C(16)	-C(114)	121.4(9)
C(14)	-C(15)	-C(16)	-C(17)	114.6(10)
C(14)	-C(15)	-C(16)	-C(114)	-60.4(12)
C(15)	-C(16)	-C(17)	-C(18)	-174.5(9)
C(114)	-C(16)	-C(17)	-C(18)	0.5(14)
C(15)	-C(16)	-C(114)	-N(12)	-6.9(12)
C(15)	-C(16)	-C(114)	-C(110)	173.2(8)
C(17)	-C(16)	-C(114)	-N(12)	178.2(8)
C(17)	-C(16)	-C(114)	-C(110)	-1.7(13)
C(16)	-C(17)	-C(18)	-C(19)	2.2(15)
C(17)	-C(18)	-C(19)	-C(110)	-3.8(16)
C(18)	-C(19)	-C(110)	-C(111)	-178.4(10)
C(18)	-C(19)	-C(110)	-C(114)	2.6(15)
C(19)	-C(110)	-C(111)	-C(112)	-179.7(10)
C(114)	-C(110)	-C(111)	-C(112)	-0.6(14)
C(19)	-C(110)	-C(114)	-N(12)	-179.7(10)
C(19)	-C(110)	-C(114)	-C(16)	0.2(16)
C(111)	-C(110)	-C(114)	-N(12)	1.2(13)
C(111)	-C(110)	-C(114)	-C(16)	-178.9(9)
C(110)	-C(111)	-C(112)	-C(113)	-0.7(14)
C(111)	-C(112)	-C(113)	-N(12)	1.7(14)
C(117)	-C(115)	-C(116)	-Pd(1)	-55.7(8)
C(118)	-C(115)	-C(116)	-Pd(1)	112.1(9)

Table S7 - Torsion Angles (Degrees) (continued) for: s1161a C18 H17 Cl N2 Pd

C(213)	-N(22)	-C(214)	-C(26)	179.9(10)
C(213)	-N(22)	-C(214)	-C(210)	0.6(12)
N(21)	-C(21)	-C(22)	-C(23)	-0.9(14)
C(21)	-C(22)	-C(23)	-C(24)	0(5)
C(22)	-C(23)	-C(24)	-C(25)	-0.4(15)
C(23)	-C(24)	-C(25)	-N(21)	1.7(13)
C(23)	-C(24)	-C(25)	-C(26)	-175.4(8)
N(21)	-C(25)	-C(26)	-C(27)	-70.7(10)
N(21)	-C(25)	-C(26)	-C(214)	113.5(8)
C(24)	-C(25)	-C(26)	-C(27)	106.5(10)
C(24)	-C(25)	-C(26)	-C(214)	-69.3(11)
C(25)	-C(26)	-C(27)	-C(28)	-173.4(8)
C(214)	-C(26)	-C(27)	-C(28)	2.5(13)
C(25)	-C(26)	-C(214)	-N(22)	-5.3(12)
C(25)	~C(26)	-C(214)	-C(210)	174.0(7)
C(27)	-C(26)	-C(214)	-N(22)	178.9(8)
C(27)	-C(26)	-C(214)	-C(210)	-1.8(12)
C(26)	-C(27)	-C(28)	-C(29)	-1.1(14)
C(27)	-C(28)	-C(29)	-C(210)	-0.9(14)
C(28)	-C(29)	-C(210)	-C(211)	-178.6(9)
C(28)	-C(29)	-C(210)	-C(214)	1.5(13)
C(29)	-C(210)	-C(211)	-C(212)	-179.9(10)
C(214)	-C(210)	-C(211)	-C(212)	0.0(16)
C(29)	-C(210)	-C(214)	-N(22)	179.2(8)
C(29)	-C(210)	-C(214)	-C(26)	-0.2(12)
C(211)	-C(210)	-C(214)	-N(22)	-0.7(12)
C(211)	-C(210)	-C(214)	-C(26)	179.9(12)
C(210)	-C(211)	-C(212)	-C(213)	0.7(14)
C(211)	-C(212)	-C(213)	-N(22)	-0.9(15)

.

Table S7 - Torsion Angles (Degrees) (continued) for: s1161a C18 H17 Cl N2 Pd

-55.1(8)	-Pd(2)	-C(216)	-C(215)	C(217)	
110.8(9)	-Pd(2)	-C(216)	-C(215)	C(218)	
54.0(8)	-Pd(2)	-C(217)	-C(215)	C(216)	
-112.0(9)	-Pd(2)	-C(217)	-C(215)	C(218)	

Table S8: ¹³C NMR data for the allyl complexes (8-PQ)Pd(η^3 -allyl)X.^{*a*}

	C2	C3	C4	C5	C6	C7	C8	С9	C10	C11	C12	C13	C14	C15	C16	C17	C18	R
1a	152.5	122.1	137.1	130.0	127.0	132.9	128.5	136.9	124.0	151.5	159.2	139.9	145.7	128.8	59.6	59.6	128.2	Me: 22.6
1 b	151.1	121.8	136.9	b	126.9	132.9	129.9	136.5	123.6	151.2	158.5	139.5	145.7	128.6	b	b	124.5	Me: 24.0 ^c
2a	b	121.7	136.8	129.7	126.8	132.6	128.2	136.5	123.6	151.2	b	139.6	145.6	128.6	b	b	127.5	Me: 22.4
2 b	149.5	b	135.3	b	125.3	131.2	128.3	134.8	b	149.0	b	137.8	144.2	127.1	b	b	122.7	Me: ^{<i>b</i>,<i>d</i>}
3a	151.8	122.3	137.4	130.6	126.7	132.3	128.6	137.1	124.2	152.3	159.1	139.7	145.7	128.9	58.6	58.6	118.3	Me: 26.2 ^e
3 b	151.5	122.2	137.0	130.3	126.8	132.3	b	137.2	124.1	b	b	139.9	145.8	128.9	b	b	114.5	Me: 26.0 ^f
4 a	152.0	122.3	137.5	132.4	127.0	133.9	128.8	137.3	124.4	152.5	159.4	139.7	146.0	129.1	60.3	60.3	121.3	Ph g
5a	151.9	122.5	138.0	130.9	127.1	132.4	128.2	137.7	124.4	152.0	Ь	135.5	145.5	129.0	58.9	58.9	117.5	<i>i</i> -Pr ^h
12	158.2	122.0	140.9	132.0	127.8	134.5	127.9	140.4	124.9	154.3	153.1	132.9	142.2	129.8	63.6	63.6	120.7	-
13	158.2	122.0	140.8	132.0	127.8	134.5	127.8	140.3	124.9	154.2	153.3	137.5	142.4	129.7	62.6	62.6	133.0	Me: 23.2
13	154.0	122.1	141.1	132.0	127.0	135.0	128.5	140.6	125.4	150.8	154.4	134.2	143.2	130.2	62.8	62.8	130.2	Me: 23.7
	<u>134.7</u> 4 /=	$\frac{122.7}{3}$		<u>152.5</u>	Record	led at 7	5.48 M	Hz in (at 21 °C	C: <i>b</i> : no	t obser	ved; c: 1	<i>n</i> -Buty	: 31.1,	29.3, 2	.2.7 (CH	H ₂), 14.3
5 6	<u>15</u>) 2 −N		u.	THale d	n_Rut	vl· 31 f	5 29 8	212 (CH ₂).	12.8 (C	(Ha): e:	C(0)N	1e: 195	.3: f: n	-Butyl:	32.6, 3	0.2, 22.5
۲ ⁰	//14 7 13/ ₁₂	Pd-			-113), <i>u</i>		J.) 164	5, 27.0, 5 A (C(I	ΔM_{e}	a. Ph.	130.5 ((Cantha)	129.0	(Cmoto) 133	9 (Cpar	a). Ca:	136.0: 191.9
	8	=N /11	17	(C		+.∠ (CΓ	13), 10. D 26 ().+ (<u>C</u> ()	20 1/C	אין איז. יע.) ר	$(\mathbf{O})\mathbf{M}_{\mathbf{A}}$	$\sim 0r(n0)$, 127.0	Cineta	, 1000		a/, ~y.	
	9			(<u>C</u>	$2(\mathbf{O})$ Me	(), n: l	rr: 30.0	л (СП),	20.1(C	.113), <u>C</u>		. 202.5						

Insertions of Allenes into Palladium-Carbon Bonds of Complexes Containing Bidentate Nitrogen Ligands. Structural and Mechanistic Studies Johannes G. P. Delis, Johannes H. Groen, Kees Vrieze, Piet W. N. M. van Leeuwen Nora Veldman, Anthony L. Spek

Compound	solvent	Т	[1,2-heptadiene]	k _{obs} *10 ²
		(K)	*10 ² (M) ^b	(s ⁻¹) ^c
(8-PQ)Pd(Me)Cl	CH_2Cl_2	303.0	2.638	0.53(1)
(1)			5.270	0.98(2)
			7.916	1.59(3)
			10.55	2.22(4)
		298.0	2.638	0.376(7)
			5.270	0.68(1)
			7.916	1.00(2)
			10.55	1.52(3)
			13.19	1.83(4)
			15.83	2.29(4)
		293.0	2.638	0.262(5)
			5.270	0.394(8)
			7.916	0.61(1)
			10.55	0.84(2)
			13.19	1.10(2)
			15.83	1.32(3)
			18.47	1.51(3)
		288.0	5.270	0.231(5)
			7.916	0.340(7)
			10.55	0.452(9)
			13.19	0.56(1)
			15.83	0.72(1)
		285.0	5.270	0.167(3)
			7.916	0.237(5)
			10.55	0.318(6)
			13.19	0.405(8)
			15.83	0.52(1)

Table S9: Observed reaction rates for the reaction of $(N \cap N)Pd(R)X$ (1-3, 6-8,10) with 1,2-heptadiene (with Esd's in Parentheses)^{*a*}.

	THF	293.0	2.638	0.225(4)
			5.270	0.373(7)
			7.916	0.54(1)
			10.55	0.69(1)
			13.19	0.86(2)
			15.83	1.04(2)
			18.47	1.27(3)
	CH ₃ CN	293.0	2.638	0.320(6)
			5.270	0.61(1)
			7.916	0.92(2)
			10.55	1.28(3)
			13.19	1.67(3)
			15.83	2.02(4)
			18.47	2.41(5)
(8-PQ)Pd(Me)Br	CH_2Cl_2	303.0	2.638	1.20(2)
(2)			5.270	2.20(4)
			7.916	3.22(6)
			10.55	4.26(8)
		298.0	2.638	0.61(1)
			5.270	1.29(2)
			7.916	1.97(4)
			10.55	2.58(5)
			13.19	3.13(6)
		293.0	2.638	0.49)1)
			5.270	0.99(2)
			7.916	1.49(3)
			10.55	1.97(4)
			13.19	2.49(5)
			15.83	2.82(6)
			18.47	3.50(7)
		288.0	2.638	0.334(6)
			5.270	0.63(1)
			7.916	0.95(2)
			10.55	1.22(2)
			13.19	1.53(3)

		283.0	2.638	0.175(3)
			5.270	0.368(7)
			7.916	0.54(1)
			10.55	0.70(1)
			13.19	0.90(2)
(8-PQ)Pd(C(O)Me)Cl	CH_2Cl_2	294.0	1.319	5.1(1)
(3)			2.638	9.4(2)
			3.957	15.4(3)
			5.270	19.1(4)
			6.589	24.1(5)
		288.0	1.319	3.81(8)
			2.638	7.2(1)
			3.957	10.6(2)
			5.270	14.5(3)
			6.589	18.9(4)
		283.0	1.319	2.71(6)
			2.638	5.0(1)
			3.957	7.5(2)
			5.270	9.9(2)
			6.589	12.5(2)
			7.916	15.0(3)
		278.0	1.319	1.92(4)
			2.638	3.61(7)
			3.957	5.2(1)
			5.270	7.3(1)
			6.589	8.5(2)
			7.916	10.3(2)
		273.0	1.319	1.34(3)
			2.638	2.71(5)
			3.957	4.31(9)
			5.270	5.5(1)
			6.589	6.8(1)
			7.916	7.8(2)

(p-An-BIAN)Pd(C(O)Me)Cl C	CH_2Cl_2	295.0	2.638	1.28(3)
(6)			5.270	2.36(5)
			7.916	3.62(7)
			10.55	4.44(9)
			13.19	6.5(1)
		288.0	2.638	0.82(2)
			5.270	1.59(3)
			7.916	2.31(5)
· · · · · · · · · · · · · · · · · · ·			10.55	3.14(6)
			13.19	4.10(8)
			15.83	4.8(1)
		283.0	2.638	0.71(1)
			5.270	1.23(2)
			7.916	1.90(4)
			10.55	2.36(5)
			13.19	3.19(6)
		278.0	2.638	0.46(1)
			5.270	0.89(2)
			7.916	1.35(3)
			10.55	1.76(4)
			13.19	2.18(4)
(p-An-BIAN)Pd(C(O)Ph)Cl C	CH_2Cl_2	305.0	5.270	0.241(5)
(7)			7.916	0.334(7)
			10.55	0.48(1)
			13.19	0.51(1)
			15.83	0.57(1)
			18.47	0.69(1)
		300.0	5.270	0.216(4)
			7.916	0.276(5)
			10.55	0.369(7)
			13.19	0.433(9)
			15.83	0.47(1)
			18.47	0.52(1)

		295.0	5.270	0.141(3)
			7.916	0.191(4)
			10.55	0.237(5)
			13.19	0.281(6)
			15.83	0.339(7)
			18.47	0.364(7)
		290.0	5.270	0.086(2)
			7.916	0.137(3)
			10.55	0.177(4)
			13.19	0.222(4)
			15.83	0.253(5)
			18.47	0.281(6)
(i-Pr-DAB)Pd(C(O)Me)Cl	CH_2Cl_2	288.0	1.319	6.7(1)
			2.638	12.0(2)
			3.958	15.8(3)
			5.270	20.7(4)
		283.0	1.319	4.50(9)
			2.638	9.3(2)
			3.957	12.3(2)
			5.270	16.0(3)
			6.589	20.6(4)
		278.0	1.319	3.25(7)
			2.638	5.9(1)
			3.957	9.4(2)
			5.270	11.6(2)
			6.589	13.9(3)
		274.0	1.319	2.46(5)
			2.638	4.8(1)
			3.957	7.4(1)
			5.270	9.1(2)
			6.589	11.5(2)
(i-Pr-PyCa)Pd(C(O)Me)Cl	CH_2Cl_2	288.0	7.916	0.266(5)
(10)			10.55	0.318(6)
			13.19	0.422(8)
			15.83	0.52(1)
			18.47	0.60(1)

a: [Pd] = 1.38 mM, b: accuracy in temperature is $0.5 \degree C$, c: the standard deviation in the k_{obs} is 2%.