# Enantioselective Synthesis of a Mitosane Core Assisted by Diversity-Based Catalyst Discovery 

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## Supporting Information

General Procedures. Proton NMR spectra were recorded on Varian 500, 400, or 300 spectrometers. Proton chemical shifts are reported in ppm ( $\delta$ ) relative to internal tetramethylsilane (TMS, $\delta 0.0$ ) or with the solvent reference relative to TMS employed as the internal standard $\left(\mathrm{CDCl}_{3}, \delta 7.26 \mathrm{ppm} ; \mathrm{d}_{6}\right.$-DMSO, $\left.\delta 2.50 \mathrm{ppm} ; \mathrm{C}_{6} \mathrm{D}_{6}, \delta 7.16 \mathrm{ppm}\right)$. Data are reported as follows: chemical shift (multiplicity [singlet (s), doublet (d), triplet (t), quartet (q), and multiplet (m)], coupling constants [Hz], integration). Carbon NMR spectra were recorded on Varian $500(125 \mathrm{MHz}), 400(100 \mathrm{MHz})$ or $300(75 \mathrm{MHz})$ spectrometers with complete proton decoupling. Carbon chemical shifts are reported in $\mathrm{ppm}(\delta)$ relative to TMS with the respective solvent resonance as the internal standard $\left(\mathrm{CDCl}_{3}, \delta 77.0\right)$. NMR data were collected at $25^{\circ} \mathrm{C}$, unless otherwise indicated. Infrared spectra were obtained on a Perkin-Elmer Spectrum 1000 spectrometer. Analytical thinlayer chromatography (TLC) was performed using Silica Gel 60 F254 precoated plates ( 0.25 mm thickness). TLC $\mathrm{R}_{f}$ values are reported. Visualization was accomplished by irradiation with a UV lamp and/or staining with ceric ammonium molybdenate (CAM) solution. Flash column chromatography was performed using Silica Gel 60 $(32-63 \mu \mathrm{~m})$ from Scientific Adsorbants Inc. Optical rotations were recorded on a Rudolf Research Analytical Autopol IV Automatic polarimeter at the sodium D line (path length 50 mm ). Elemental analyses were performed by Robertson Microlit (Madison, NJ). High resolution mass spectra were obtained at the Mass Spectrometry Facilities either of the University of Illinois (Urbana-Champaign, IL), Harvard University (Cambridge, MA) or Boston College (Chestnut Hill, MA). The method of ionization is given in parentheses.

Analytical normal phase HPLC was performed on a Hewlett-Packard 1100 Series chromatograph equipped with a diode array detector ( 214 nm and 254 nm ). All reactions were carried out under an argon or nitrogen atmosphere employing oven- and flame-dried glassware. All solvents were distilled from appropriate drying agents prior to use. Acetic anhydride was distilled prior to use and stored in a Schlenk tube for no more than 1 week.

(B) To a solution of isatin $\operatorname{ketal} \mathbf{A}(0.990 \mathrm{~g}, 5.18 \mathrm{mmol})$ in 10.0 mL of dry methylene chloride was added triethylamine ( $0.720 \mathrm{~mL}, 5.18 \mathrm{mmol}$ ) followed by di-tert-butyl dicarbonate ( $2.26 \mathrm{~g}, 10.4 \mathrm{mmol}$ ) and 4(dimethylamino)pyridine ( $0.630 \mathrm{~g}, 5.18 \mathrm{mmol}$ ). The solution was stirred for 6 h at ambient temperature under an argon atmosphere. The solution was diluted to 50 mL with methylene chloride, washed with citric acid (3 x 20 mL ) and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. After filtration, the solvent was removed under reduced pressure and the crude product was purified by flash chromatography ( $30 \%$ ethyl acetate:hexanes) to afford $1.30 \mathrm{~g}(86 \%)$ of the desired product. ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 400\right.$ $\mathrm{MHz}) \delta 7.90(\mathrm{~d}, \mathrm{~J}=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.43($ app. t, 2H), 7.21 (t, J = $7.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.57(\mathrm{~m}, 2 \mathrm{H})$, $4.35(\mathrm{~m}, 2 \mathrm{H}), 1.63(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{\mathbf{1 3}} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right) \delta 171.6,148.9,141.0,132.1$, 125.1, 124.9, 123.0, 115.7, 101.5, 84.8, 66.2, 28.4; TLC $\mathbf{R}_{\boldsymbol{f}} 0.30$ ( $15 \% \mathrm{EtOAc} / \mathrm{Hex}$ ); IR (film, $\mathrm{cm}^{-1}$ ) 2980, 2904, 1784, 1734, 1614, 1469; Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{NO}_{5}$ : C, 61.85; H, 5.88; N, 4.81. Found: C, 61.83; H, 5.81; N, 4.69; Exact mass calcd for [ $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{NO}_{5}$ +Na ] requires $m / z$ 314.1004. Found: 314.0992 (ESI+).

(C): To a solution of BOC isatin B ( $25.4 \mathrm{~g}, 87.3 \mathrm{mmol}$ ) in $300 \mathrm{~mL}(5: 2)$ THF/ $\mathrm{H}_{2} \mathrm{O}$ was added $\mathrm{NaOH}(6.98 \mathrm{~g}, 175 \mathrm{mmol})$. The solution was heated to reflux for 1 h then cooled to ambient temp. The volitiles were removed under reduced pressure and the resulting slurry was partitioned between $10 \%$ citric acid $(700 \mathrm{~mL})$ and methylene chloride ( 1 L ). The acidic aqueous layer was extracted with methylene chloride ( $2 \times 1 \mathrm{~L}$ ) and the organic layers were combined and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. After filtration, the solvent was removed under reduced pressure to afford 27.0 g ( $99 \%$ ) of pure product. ${ }^{\mathbf{1}} \mathbf{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 10.40(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.90(\mathrm{br} \mathrm{d}, \mathrm{J}=7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.84(\mathrm{~s}, 1 \mathrm{H}), 7.60(\mathrm{~d}$, $\mathrm{J}=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.33(\mathrm{t}, \mathrm{J}=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.08(\mathrm{t}, \mathrm{J}=7.7,1 \mathrm{H}), 4.21-4.16(\mathrm{~m}, 2 \mathrm{H}), 4.13-4.07$ $(\mathrm{m}, 2 \mathrm{H}), 1.48(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right) \delta 172.4,153.7,136.2,130.1,126.4$, 123.3, 122.4, 105.1, 80.8, 65.9, 28.5; TLC $\mathbf{R}_{\boldsymbol{f}} 0.34$ ( $30 \% \mathrm{EtOAc} / \mathrm{Hex}$ ); IR (film, $\mathrm{cm}^{-1}$ ) 3402, 2980, 2936, 2904, 2615, 1734, 1595, 1513, 1444 ; Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{19} \mathrm{NO}_{6}$ : C, 58.25; H, 6.19; N, 4.53. Found: C, 58.14; H, 6.21; N, 4.35.


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(4): To a solution of carboxylic acid $\mathbf{C}(23.6, \mathrm{~g} 76.3 \mathrm{mmol})$ in 400 mL dry DMF was added allyl bromide ( $65.9 \mathrm{~mL}, 763 \mathrm{mmol}$ ). The resulting solution was cooled to $0^{\circ} \mathrm{C}$ and sodium hydride ( $5.49 \mathrm{~g}, 229$ $\mathrm{mmol})$ was added in small portions. After 10 min at $0^{\circ} \mathrm{C}$ the reaction was allowed to reach ambient temp. The reaction was stirred for 3 h at ambient temp under a nitrogen atmosphere. The reaction was cooled to $0{ }^{\circ} \mathrm{C}$ and a dilute aqueous solution of ammonium chloride was added dropwise to the reaction. When gas evolution ceased, dilute ammonium chloride (1 L) was added to the mixture. The solution was separated into two portions and the following work-up was repeated twice. The mixture was poured into a separatory funnel
and the aqueous solution was washed with diethyl ether ( $5 \times 500 \mathrm{~mL}$ ). The combined ether layer was then washed with 500 mL of brine. The organics were combined and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. After filtration, the solvent was removed under reduced pressure and the crude product was purified by flash chromatography ( $20 \%$ ethyl acetate:hexanes) to afford $25.9 \mathrm{~g}(87 \%)$ of the desired product. ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.77$ (br s, 1 H ), 7.33 (br m, 2H), 7.10 (br s, 1H), 6.00-5.80 (m, 2H), 5.30-5.00 (m, 4H), 4.73-4.68 (br dd, J=13.4, $5.7 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.50 (br s, 2H), 4.26 (br s, 1H), 4.17-4.08 (m, 3H), 3.67-3.58 (dd, J=15.6, 7.1 Hz, 1H), $1.46(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathbf{C} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right) \delta 167.8,154.8$, $140.2,134.2,131.7,131.0,129.7,127.3,127.2,118.3,117.2,105.0,80.2,66.5,66.1$, 65.7, 54.3, 28.6; TLC $\mathbf{R}_{\boldsymbol{f}} 0.27$ ( $20 \% \mathrm{EtOAc} / \mathrm{Hex}$ ); IR (film, $\mathrm{cm}^{-1}$ ) 3081, 2974, 2898, 1765, 1696, 1646; Anal. Calcd. for $\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{NO}_{6}$ : C, 64.77; H, 6.99; N, 3.60. Found: C, 64.92; H, 6.99; N, 3.42; Exact mass calcd for $\left[\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{NO}_{6}\right]+\mathrm{H}$ requires $\mathrm{m} / \mathrm{z} 390.1917$. Found 390.1899 (ESI+).
(D): To a solution of allyl ester $4(24.9 \mathrm{~g}, 63.7 \mathrm{mmol})$ in 400 mL of $1: 1$
 THF/ $\mathrm{H}_{2} \mathrm{O}$ was added $\mathrm{NaOH}(5.12 \mathrm{~g}, 128 \mathrm{mmol})$. The solution was heated to reflux for 5 h then cooled to ambient temp. The solvent was partially removed under reduced pressure and the resulting slurry was partitioned between $10 \%$ citric acid $(500 \mathrm{~mL}$ ) and methylene chloride ( 1 $\mathrm{L})$. The acidic aqueous layer was rinsed several times with methylene chloride ( $2 \times 500 \mathrm{~mL}$ ) and the organic layers were combined and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. After filtration, the solvent was removed under reduced pressure to afford $22.2 \mathrm{~g}(99 \%)$ of pure product. ${ }^{\mathbf{1}} \mathbf{H}$ NMR $\left(\mathrm{C}_{6} \mathrm{D}_{6}, 300 \mathrm{MHz}, 70^{\circ} \mathrm{C}\right) \delta 9.37(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.82$ $(\mathrm{m}, 1 \mathrm{H}), 7.04-6.99(\mathrm{~m}, 3 \mathrm{H})$, 6.02-5.89 (m, 1H), 5.03-4.96 (m, 2H), 4.57-4.50 (br m, 1H), $3.92-3.58(\mathrm{~m}, 5 \mathrm{H}), 1.40(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{\mathbf{1 3}} \mathbf{C}$ NMR $\left(\mathrm{C}_{6} \mathrm{D}_{6}, 75 \mathrm{MHz}, 70^{\circ} \mathrm{C}\right) \delta 169.9,156.0,140.2$, 135.6, 134.7, 130.8, 129.7, 128.9(br), 117.2, 106.2, 80.8, 66.0, 65.5, 54.3, 28.5; TLC $\mathbf{R}_{f}$ 0.31 ( $80 \% \mathrm{EtOAc} / \mathrm{Hex}$ ); IR (film, $\mathrm{cm}^{-1}$ ) 3081, 2986, 1759, 1709, 1658, 1488, 1457; Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{NO}_{6}$ : C, 61.88; H, 6.64; N, 4.01. Found: C, 61.90; H, 6.48; N, 4.29; Exact mass calcd for [ $\left.\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{NO}_{6}\right]+\mathrm{H}$ requires $\mathrm{m} / \mathrm{z} 350.1604$. Found 350.1599 (ESI+).
(E): To a solution of carboxylic acid $\mathbf{D}(3.93 \mathrm{~g}, 11.3 \mathrm{mmol})$ in 500 mL dry methylene chloride was added triethylamine ( $7.84 \mathrm{~mL}, 56.3 \mathrm{mmol}$ ), 4-(dimethylamino)pyridine $(1.37 \mathrm{~g}, \quad 11.3 \mathrm{mmol}), \mathrm{N}, \mathrm{O}-$ dimethylhydroxylamine hydrochloride ( $3.29 \mathrm{~g}, 33.8 \mathrm{mmol}$ ), and 1-(3-dimethylamino-propyl)-3-ethyl-carbodiimide hydrochloride (EDC, $2.37 \mathrm{~g}, 12.4 \mathrm{mmol}$ ) sequentially. The solution was stirred at ambient temperature under a nitrogen atmosphere overnight. The reaction mixture was diluted to 150 mL with methylene chloride, the organic layer was washed with $5 \%$ citric acid ( $3 \times 50 \mathrm{~mL}$ ) and saturated aqueous $\mathrm{NaHCO}_{3}(1 \times 50 \mathrm{~mL}$ ). The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and concentrated under reduced pressure. The crude product was purified by flash chromatography ( $50 \%$ ethyl acetate:hexanes) to afford $2.85 \mathrm{~g}(64 \%)$ of the desired product. ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.78$ (br s,

1H), 7.31 (br m, 2H), 7.18 (br s, 1H), 5.98-5.87 (br m, 1H), 5.08-5.00 (br m, 2H), 4.45 (br dd, J=15.8, $4.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.23-4.03 (m, 4H), 3.76 (br dd, J=16.1, 7.0 Hz, 1H), 3.07 (br s, $6 \mathrm{H}), 1.45$ (br s, 9 H ); ${ }^{13} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right) \delta 167.1,154.9,140.0,137.5,134.4$, $131.6,129.2,127.4,126.1,117.9,105.6,80.1,66.6,65.4,59.8,54.6,32.4,28.6$; TLC R $\mathbf{R}_{f}$ $0.34(50 \% \mathrm{EtOAc} / \mathrm{Hex})$; IR (film, $\mathrm{cm}^{-1}$ ) 3081, 2967, 2936, 2898, 2250, 1702, 1683; Anal. Calcd. for $\mathrm{C}_{20} \mathrm{H}_{29} \mathrm{~N}_{2} \mathrm{O}_{6}$ : C, 61.21; H, 7.19; N, 7.14. Found: C, 61.16; H, 7.19; N, 7.00; Exact mass calcd for $\left[\mathrm{C}_{20} \mathrm{H}_{29} \mathrm{~N}_{2} \mathrm{O}_{6}\right]+\mathrm{H}$ requires $m / z$ 393.2026. Found: 393.2009 (ESI+).

(5): To a solution of Weinreb amide $\mathbf{E}(2.00 \mathrm{~g}, 5.10 \mathrm{mmol})$ in 80 mL dry THF at $0^{\circ} \mathrm{C}$ was added freshly prepared vinylmagnesium bromide (9.00 $\mathrm{mL}, 10.4 \mathrm{mmol}$ ) dropwise as a 1.15 M solution in THF. The reaction was stirred for 15 min or until TLC analysis indicated completion. To the reaction was added acetic anhydride ( 3.00 mL ) followed by methanol $(3.00 \mathrm{~mL})$. The solution was reduced to one quarter the volume under reduced pressure then diluted to 1 L with diethyl ether. The solution was washed with dilute aqueous ammonium chloride solution ( $2 \times 200 \mathrm{~mL}$ ) and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The solution was filtered and concentrated under reduced pressure at ambient temperature to avoid decomposition. The crude product was purified by flash chromatography ( $20 \%$ ethyl acetate:hexanes) to afford $1.83 \mathrm{~g}(99 \%)$ of the desired product. ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathrm{C}_{6} \mathrm{D}_{6}\right.$, $500 \mathrm{MHz}, 70^{\circ} \mathrm{C}$ ) $\delta 7.81(\mathrm{br} \mathrm{m}, 1 \mathrm{H}), 7.08(\mathrm{br} \mathrm{m}, 3 \mathrm{H}), 6.63-6.57(\mathrm{dd}, \mathrm{J}=17.1,10.3 \mathrm{~Hz}, 1 \mathrm{H})$, 6.33-6.29 (d, J=17.6 Hz, 1H), 6.07-5.98 (m, 1H), 5.32-5.29 (d, J=11.7 Hz, 1H), 5.04-5.00 (d, J=17.1 Hz, 1H), 4.99-4.97 (d, J=9.3 Hz, 1H), 4.69-4.66 (br d, J=13.2 Hz, 1H), 3.873.83 (dd, J=15.1, $7.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.82-3.74 (br m, 1H), 3.62-3.54 (br m, 3H), 1.38 (s, 9H); ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{C}_{6} \mathrm{D}_{6}, 125 \mathrm{MHz}, 70{ }^{\circ} \mathrm{C}\right) \delta 191.4,155.0,142.0,136.6,135.3,132.6,132.5$, $130.0,128.5,127.7,117.5,108.3,80.1,66.6,65.5,54.5,28.2$; TLC $\mathbf{R}_{f} 0.28$ (20\% EtOAc/Hex); IR (film, $\mathrm{cm}^{-1}$ ) 3087, 2993, 2942, 2898, 1709, 1620, 1501, 1451; Anal. Calcd. for $\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{NO}_{5}$ : C, 66.83; H, 7.01; N, 3.90. Found: C, 66.65; H, 6.82; N, 3.68; Exact mass calcd for $\left[\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{NO}_{5}\right]+\mathrm{H}$ requires $m / z 360.1811$. Found: 360.1827 (ESI+).


7
(7): To a solution of diene $5(0.500 \mathrm{~g}, 1.39 \mathrm{mmol})$ in 280 mL dry methylene chloride was added ruthenium alkylidene 6 ( $0.118 \mathrm{~g}, 0.139$ $\mathrm{mmol})$. The solution was heated to reflux under an argon atmosphere for 4 $h$. The solvent was then removed under reduced pressure and the crude product was purified by flash chromatography ( $30 \%$ ethyl acetate:hexanes) to afford $0.382 \mathrm{~g}(83 \%)$ of the desired product. ${ }^{1} \mathbf{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400\right.$ $\mathrm{MHz}) \delta 7.64(\mathrm{~m}, 1 \mathrm{H}), 7.43(\mathrm{~m}, 2 \mathrm{H}), 7.13(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 6.12-6.08(\mathrm{~m}, 1 \mathrm{H}), 5.68-5.64(\mathrm{~m}$, $1 \mathrm{H}), 5.04(\mathrm{br} \mathrm{d}, \mathrm{J}=17.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.34-4.20(\mathrm{~m}, 3 \mathrm{H}), 4.05-3.99(\mathrm{~m}, 1 \mathrm{H}), 3.58(\mathrm{br} \mathrm{d}, \mathrm{J}=18.3$ $\mathrm{Hz}, 1 \mathrm{H}), 1.36(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{\mathbf{1 3}} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right) \delta 198.3,153.9,139.0,138.4,130.7$, 130.6, 128.2, 127.6, 127.3, 126.9, 106.7, 81.4, 66.7, 65.2, 50.1, 28.0; TLC R $\mathbf{R}_{\boldsymbol{f}} 0.52$ (50\% EtOAc/Hex); IR (film, $\mathrm{cm}^{-1}$ ) 2980, 2936, 2898, 1715, 1608, 1495, 1463; Anal. Calcd. For $\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{NO}_{5}$ : C, 65.24; H, 6.39; N, 4.23. Found: C, 65.05; H, 6.30; N, 3.93; Exact mass calcd for $\left[\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{NO}_{5}\right]+\mathrm{Na}$ requires $\mathrm{m} / \mathrm{z} 354.1317$. Found: 354.1317 (ESI+).

$( \pm)-3$
[( $\pm$ )-3]: To a solution of enone $7(0.150 \mathrm{~g}, 0.450 \mathrm{mmol})$ in 25.0 mL dry diethyl ether at $0{ }^{\circ} \mathrm{C}$ was added lithium aluminum hydride ( $52.0 \mathrm{mg}, 1.40$ mmol ). After $10 \mathrm{~min}, 0.500 \mathrm{~mL}$ water was added dropwise, followed by $0.500 \mathrm{~mL} 15 \% \mathrm{NaOH}$ and finally 0.500 mL water. The slurry was the filtered through a plug of celite and the celite cake was washed with copious amounts of diethyl ether. The solvent was removed under reduced pressure and the crude product was purified by flash chromatography ( $30 \%$ ethyl acetate:hexanes) to afford $0.149 \mathrm{~g}(99 \%)$ of the desired product. Allylic alcohol $\mathbf{3}$ can also be purified by recrystallization using ethyl acetate and hexanes. ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathrm{CDCl}_{3}\right.$, 400 MHz ) $\delta 7.49-7.47(\mathrm{~d}, \mathrm{~J}=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.37-7.29$ (m, 2H), 7.05-7.03 (br d, J=7.0 Hz, 1 H ), 5.48-5.44 (br m, 1H), 5.40-5.25 (br m, 2 H ), 5.10 (br d, J=17.2 Hz, 1H), 4.40 (br m, 1 H ), 4.20 (br m, 2H), 3.80 (br m, 1H), 3.56-3.51 (br dd, J=16.5, $4.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.50 (br s, $1 \mathrm{H}), 1.55-1.33$ (br s, 9H); ${ }^{13} \mathbf{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right) \delta 154.5,139.2,138.1,133.1$, $129.9,129.2,127.4,127.3,125.8,112.4,80.5,73.1,68.4,64.2,48.2,28.58$; TLC $\mathbf{R}_{\boldsymbol{f}}$ 0.36 (50\% EtOAc/Hex); IR (film, $\mathrm{cm}^{-1}$ ) 3452, 2993, 2917, 1696, 1495, 1457; Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{NO}_{5}$ : C, 64.85; H, 6.95; N, 4.20. Found: C, 64.73; H, 6.89; N, 4.28; Exact mass calcd for $\left[\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{NO}_{5}\right]+\mathrm{Na}$ requires $\mathrm{m} / \mathrm{z} 356.1474$. Found: 356.1470 (ESI+).

(-)-3
[(-)-3]: To a solution of racemic allylic alcohol ( $\pm$ )- $\mathbf{3}(0.25 \mathrm{~g}, 0.75 \mathrm{mmol})$ in 150 mL of toluene was added peptide catalyst $9(0.013 \mathrm{~g}, 0.015 \mathrm{mmol})$ followed by triethylamine ( $0.63 \mathrm{~mL}, 4.5 \mathrm{mmol}$ ). The solution was cooled to $0{ }^{\circ} \mathrm{C}$ and acetic anhydride $(0.36 \mathrm{~mL}, 3.75 \mathrm{mmol})$ was added dropwise. The reaction was stirred for 3 d at $0^{\circ} \mathrm{C}$ and was quenched with 20 mL MeOH . The solvent was removed under reduced pressure and the crude material was purified by flash chromatography ( $25 \%$ ethyl acetate:hexanes) to afford $0.12 \mathrm{~g}(47 \%)$ of recovered allylic alcohol. Assay of enantiomeric purity. Enantiomers of starting material and product 3-Ac were separated by chiral HPLC employing a Chiracel AD column (Alltech), eluting with 5\% 2-propanol/hexanes at a flow rate of $1.5 \mathrm{~mL} / \mathrm{min}$. Retention Times: 3: $\mathrm{R}_{\mathrm{t}(\mathrm{S})}=16.4 \mathrm{~min} ; \mathrm{R}_{\mathrm{t}(\mathrm{R})}=24.1 \mathrm{~min}$. Retention Times: 3-Ac: $\mathrm{R}_{\mathrm{t}(\mathrm{S})}=$ $9.1 \mathrm{~min} ; \mathrm{R}_{\mathrm{t}(\mathrm{R})}=13.2 \mathrm{~min}$. Recovered 3 was produced with $90 \%$ ee ( $53 \%$ conversion, $\left.k_{\text {rel }}=27\right)$. The recovered allylic alcohol was recrystallized to optical purity. $[\alpha]_{\mathrm{D}}-18 \quad[($ at $>99 \%$ ee) $c=0.097 \mathrm{~g} / 100 \mathrm{~mL}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$ )].
(11): To a solution of optically pure allylic alcohol (-)-3 (0.105 g, 0.315 $\mathrm{mmol})$ in 10 mL of $3: 1$ acetone $/ \mathrm{H}_{2} \mathrm{O}$ was added $\mathrm{NaHCO}_{3}(0.151 \mathrm{~g}, 1.80$ $\mathrm{mmol})$ followed by oxone $(0.369 \mathrm{~g}, 0.600 \mathrm{mmol})$. The solution was stirred at ambient temp. overnight. The solvent was partially removed under reduced pressure then the resulting slurry was partitioned between saturated aqueous $\mathrm{NaHCO}_{3}(1 \times 10 \mathrm{~mL})$ and ethyl acetate (75 mL ). The aqueous layer was washed with ethyl acetate ( $2 \times 75 \mathrm{~mL}$ ) and
the organic layers were combined and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. After filtration, the solvent was removed under reduced pressure. The crude product was eluted through a plug of silica gel with ethyl acetate. The resulting solution was concentrated under reduced pressure to afford $0.110 \mathrm{~g}(99 \%)$ of analytically pure product. Epoxyalcohol 11 exhibits a complex ${ }^{1} \mathrm{H}$ NMR spectrum (shown below) at room temperature. However, upon quantitative TFA deprotection, the free amine exhibits sharp ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right)$ and ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right)$ spectra (scanned in below).


Data for 11: ${ }^{13} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right)$ major peaks $\delta 154.5,153.8,140.6,138.5$, $138.2,130.9,130.2,129.8,129.3,128.6,128.2,127.9,126.6,126.1,108.8,80.3,76.2$, $75.6,74.6,67.7,66.2,65.5,63.8,62.4,58.7,57.0,53.4,52.5,51.6,51.2,49.6,48.2,46.8$, 29.1, 27.7, 26.5; TLC $\mathbf{R}_{\boldsymbol{f}} 0.32$ ( $50 \% \mathrm{EtOAc} / \mathrm{Hex}$ ); IR (film, $\left.\mathrm{cm}^{-1}\right) 3471,2974,2942$, $2898,1702,1608,1482,1400 ;[\alpha]_{\mathrm{D}}-35.9$ (c=0. $548 \mathrm{~g} / 100 \mathrm{~mL}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ); Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{NO}_{6}$ : C,61.88; H, 6.64; N, 4.01. Found: C, 61.50 ; H, 6.71 ; N , 3.90 ; Exact mass calcd for $\left[\mathrm{C}_{18} \mathrm{H}_{23} \mathrm{NO}_{6}\right]+\mathrm{Na}$ requires $\mathrm{m} / \mathrm{z} 3372.1423$. Found 372.1406 (ESI+).

The ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right)$ and ${ }^{\mathbf{1 3}} \mathbf{C} \mathbf{~ N M R}\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right)$ spectra for the deprotected amino alcohol derived from 11 are scanned in on the next page.


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(12): To a solution of oxalyl chloride ( $81.0 \mu \mathrm{~L}, 0.940 \mathrm{mmol}$ ) in 10 mL dry methylene chloride at $-78{ }^{\circ} \mathrm{C}$ was added dimethylsulfoxide $(0.110$ $\mathrm{mL}, 1.60 \mathrm{mmol}$ ) dropwise. After 15 min , epoxyalcohol 11 ( 0.109 g , 0.312 mmol ) was added in 10.0 mL of dry methylene chloride via cannula. The solution was stirred at $-78{ }^{\circ} \mathrm{C}$ for an additional 15 min after which triethylamine ( $0.700 \mathrm{~mL}, 4.99 \mathrm{mmol}$ ) was added and the reaction was allowed to reach ambient temp. After 1 h the mixture was diluted to 100 mL with methylene chloride and washed with $5 \%$ citric acid ( $2 \times 20 \mathrm{~mL}$ ) and saturated aqueous $\mathrm{NaHCO}_{3}$ (1 x 10 mL ). The organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and filtered. The solvent was removed under reduced pressure and the crude product was purified by flash chromatography ( $30 \%$ ethyl acetate:hexanes) to afford 0.102 g ( $94 \%$ ) of the desired product. ${ }^{1} \mathbf{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.56(\mathrm{~m}, 1 \mathrm{H}), 7.37(\mathrm{~m}, 2 \mathrm{H}), 7.03(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.70$ (br d, J=15.0 Hz, 1H), 4.34-4.20 (m, 3H), 4.10-4.05 (m, 1H), 3.65 (d, J=4.0 Hz, 1H), 3.58 (br d, J=15.8 Hz, 1H), $3.50(\mathrm{dd}, \mathrm{J}=4.0,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.33(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{\mathbf{1 3}} \mathbf{C} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 100\right.$ $\mathrm{MHz}) \delta 195.2,153.9,140.4,139.0,130.2,128.2,127.0,125.8,105.8,81.9,66.9,65.8$, 61.8, 56.1, 45.6, 28.1; TLC R $\boldsymbol{R}_{f} 0.41$ ( $50 \% \mathrm{EtOAc} / \mathrm{Hex}$ ); IR (film, $\mathrm{cm}^{-1}$ ) 2993, 2930, 2262, 1727, 1702, 1614; $[\alpha]_{D}-33.0\left(c=0.404 \mathrm{~g} / 100 \mathrm{~mL}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$; Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{NO}_{6}$ : C, $62.24 ; \mathrm{H}, 6.09$; N, 4.03. Found: C, 62.17 ; H, 6.02; N, 3.83; Exact mass calcd for $\left[\mathrm{C}_{18} \mathrm{H}_{21} \mathrm{NO}_{6}\right]+\mathrm{Na}$ requires $\mathrm{m} / \mathrm{z} 370.1267$ Found 370.1272 (ESI+).


13
(13): To a solution of epoxyketone $12(57.0 \mathrm{mg}, 0.164 \mathrm{mmol})$ in 5 mL methanol, in a sealed tube, was added water $(9.00 \mu \mathrm{~L}, 0.492 \mathrm{mmol})$ and a solution of nitric acid ( $0.200 \mathrm{~mL}, 0.042 \mathrm{M}$ in MeOH ). The tube was sealed under an atmosphere of nitrogen and placed into an oil bath at 135 ${ }^{\circ} \mathrm{C}$. After 30 min . the tube was cooled to $0^{\circ} \mathrm{C}$ and the contents of the tube were partitioned between methylene chloride ( 100 mL ) and saturated aqueous $\mathrm{NaHCO}_{3}$ $(10 \mathrm{~mL})$. The aqueous layer was washed with methylene chloride ( $2 \times 50 \mathrm{~mL}$ ) and the organic layers were combined and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The solution was filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography ( $30 \%$ ethyl acetate:hexanes) to afford $29.0 \mathrm{mg}(81 \%)$ of the desired product. ${ }^{1} \mathbf{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.60-7.50(\mathrm{~m}, 2 \mathrm{H}), 6.90(\mathrm{t}, \mathrm{J}=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.80$ (d, J=8.1 Hz, 1H), 3.80-3.75 (dd, J=6.6, 2.6 Hz, 2H), 3.70-3.55 (dd, J=30.8, $12.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), $3.29(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right) \delta 196.9,164.5,138.3,124.4,122.4,121.0$, 112.9, 99.4, 57.7, 56.2, 52.1, 51.3; TLC $\mathbf{R}_{\boldsymbol{f}} 0.33$ ( $30 \% \mathrm{EtOAc} / \mathrm{Hex}$ ); IR (film, $\mathrm{cm}^{-1}$ ) 2942, 1721, 1627, 1482, 1318; ; $[\boldsymbol{\alpha}]_{\text {D }} \quad 342\left(\mathrm{c}=0.266 \mathrm{~g} / 100 \mathrm{~mL}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$; Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{NO}_{3}$ : C, 66.35; H, 5.10; N, 6.45. Found: C, 66.09; H, 4.91; N, 6.29.


14
(14): To a solution of epoxide $13(73.0 \mathrm{mg}, 0.336 \mathrm{mmol})$ in 2 mL dry methylene chloride was added $\mathrm{Sm}(\mathrm{O}-i-\mathrm{Pr})_{3}(52.0 \mathrm{mg}, 0.160 \mathrm{mmol})$ followed by trimethylsilyl azide $(0.178 \mathrm{~mL}, 1.34 \mathrm{mmol})$. The solution was stirred at ambient temp under an argon atmosphere for 24 h . The solution was concentrated under reduced pressure and the crude product was purified by flash chromatography ( $30 \%$ diethyl ether:hexanes) to afford 46.0 mg
( $52 \%$ ) of the desired product along with 38.0 mg ( $34 \%$ ) of the TMS ether. The two were combined and dissolved in 10 mL of a $0.04 \mathrm{M} \mathrm{HCl} / \mathrm{MeOH}$ solution. After 1 h , the solution was quenched with saturated aqueous $\mathrm{NaHCO}_{3}$ and concentrated under reduced pressure to one quarter the volume. The resulting suspension was partitioned between methylene chloride ( 100 mL ) and $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$. The organic layer was concentrated under reduced pressure to afford $\mathbf{1 4}$ as a mixture of diastereomers in quantitative yield. The material was used in the subsequent step without further purification. ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}$ $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.63-7.52(\mathrm{~m}, 4 \mathrm{H}), 7.03-6.88(\mathrm{~m}, 4 \mathrm{H}), 4.55-4.44(\mathrm{~d}, \mathrm{~J}=2.9 \mathrm{~Hz}, 1 \mathrm{H})$, 4.44-4.38 (dd, J=15.7, 8.1 Hz, 1H), 4.05-4.01 (m, 1H), 3.95-3.91 (dd, J=10.6, 7.3 Hz, $1 \mathrm{H}), 3.84-3.80(\mathrm{dd}, \mathrm{J}=12.1,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.76-3.74(\mathrm{br} \mathrm{d}, 1 \mathrm{H}, \mathrm{J}=7.7 \mathrm{~Hz}), 3.66-3.61$ (dd, $\mathrm{J}=11.7,4.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), $3.30(\mathrm{~s}, 3 \mathrm{H}), 3.23$ (s, 3H), 3.05 (s, 1H), 3.01-2.96 (dd, J=10.6, 7.7 $\mathrm{Hz}, 1 \mathrm{H})$; Exact mass calcd for $\left[\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{3}\right]+\mathrm{H}$ requires $\mathrm{m} / \mathrm{z}$ 261.0988. Found 261.0982 (ESI+).


1
(1): To a $0{ }^{\circ} \mathrm{C}$ solution of azido-alcohol $\mathbf{1 4}(34.0 \mathrm{mg}, 0.133 \mathrm{mmol})$ in 5.00 mL methylene chloride was added triethylamine $(56.0 \mu \mathrm{~L}, 0.399$ $\mathrm{mmol})$ followed by methanesulfonyl chloride ( $21.0 \mu \mathrm{~L}, 0.266 \mathrm{mmol}$ ). The reaction was stirred under an atmosphere of nitrogen for 1 h . The reaction mixture was diluted to 50 mL with diethyl ether and washed with $5 \%$ citric acid $(1 \times 10 \mathrm{~mL})$ and $5 \%$ sodium bicarbonate ( $1 \times 10 \mathrm{~mL}$ ). The organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and filtered. The solvent was removed under reduced pressure to afford crude product. The crude material was dissolved in 10.0 mL THF: $\mathrm{H}_{2} \mathrm{O}$ (10:1) and transferred into a round bottom flask. To the solution was added Hunig's base ( $46.0 \mu \mathrm{~L}, 0.266 \mathrm{mmol}$ ) and triphenylphosphine polystyrene resin $(0.200 \mathrm{~g}, \sim 0.20 \mathrm{mmol})$ followed by heating to $65^{\circ} \mathrm{C}$ for 1 h . The reaction mixture was filtered through a plug of cotton and concentrated under reduced pressure. The product was purified by flash chromatography (75\% ethyl acetate:hexanes) to afford $12.2 \mathrm{mg}(42 \%)$ of the trans-mitosane $\mathbf{1}$ (cis-mitocene 1 was not isolated). Data for trans-1 follow. ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right) \delta 7.56-7.50(\mathrm{~m}, 2 \mathrm{H})$, $6.90(\mathrm{t}, \mathrm{J}=7.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.78(\mathrm{~d}, \mathrm{~J}=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.55(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.29(\mathrm{~s}, 3 \mathrm{H}), 3.02(\mathrm{br} \mathrm{s}$, $1 \mathrm{H}), 2.93(\mathrm{br} \mathrm{s}, 1 \mathrm{H}),{ }^{13} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 100 \mathrm{MHz}\right) \delta 198.7,164.8,138.3,124.4,122.6$, $120.5,112.4,100.4,51.8,51.5,37.2,36.7$; TLC $\mathbf{R}_{f} 0.40$ (ethyl acetate); IR (film, cm ${ }^{-1}$ ) 3314, 2943, 1715, 1620, 1482, 1318; $[\boldsymbol{\alpha}]_{\mathrm{D}}-130\left(\mathrm{c}=.08 \mathrm{~g} / 100 \mathrm{~mL}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$; Exact mass calcd for $\left[\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}\right]+\mathrm{H}$ requires $\mathrm{m} / \mathrm{z} 217.0977$. Found 217.0968 (ESI+).

$128.9,128.4,128.3,128.2$ (2C), 127.6 (2C), 126.7, 126.5, 126.1, 79.8. 61.4, 54.4, 54.3, 54.0, 52.3, 51.8, 47.6, 38.0, 36.8, 32.1, 31.8, 31.5, 29.0, 28.4, 27.2, 25.6; IR (film, cm-1) $3315,3195,3062,3026,2979,2953,2931,1736,1708,1671,1630$; TLC R $\mathbf{R}_{\mathrm{f}} 0.36$ ( $8 \%$
$\mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ); Exact mass calcd for $\left[\mathrm{C}_{46} \mathrm{H}_{57} \mathrm{~N}_{7} \mathrm{O}_{8}+\mathrm{H}\right]+$ requires $\mathrm{m} / \mathrm{z}$ 836.4347. Found 836.4346 (ESI+). Analytical HPLC. Purity of peptide 9 was assayed using a reverse phase RP-18 X Terra (Waters) column at a flow rate of $0.2 \mathrm{~mL} / \mathrm{min}$ with the following gradient : $50 \%$ methanol/water to $70 \%$ methanol/water over 20 min , hold until 25 min . Retention time $=18.7 \mathrm{~min}$.

## Data for Peptide Library. ${ }^{\text {i }}$

## Library A

| Peptide | Sequence | Calcd <br> $\mathrm{M}+\mathrm{H}$ | Obs. M+H | $\begin{gathered} k_{\text {rel }} @ 25 \\ { }^{\circ} \mathrm{C} \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: |
| 9 | BOC-Pmh-D-Pro-Hfe-Phe-D-Phe-OMe | 837.00 | 836.22 | 9.7 |
| A1 | BOC-Pmh-D-Pro-Aib-Cha-Phe-OMe | 766.95 | 767.07 | 3.4 |
| A2 | Boc-Pmh-D-Pro-Dbg-Val-Phe-OMe | 797.02 | 797.14 | 2.6 |
| A3 | Boc-Pmh-D-Pro-Cha-1-amino-1-cyclohexane carboxylic acid-Phe-OMe | 807.02 | 806.87 | 2.7 |
| A4 | BOC-Pmh-D-Pro-Phe-1-amino-1-cyclopentane carboxylic acid-Phe-OMe | 786.94 | 787.03 | 2.0 |
| A5 | BOC-Pmh-D-Pro-1-amino-1-cyclohexane carboxylic acid-Chg-Phe-OMe | 792.99 | 792.34 | 4.3 |
| A6 | BOC-Pmh-D-Pro-Aib-Phe-OMe | 613.73 | 613.79 | 3.4 |
| A7 | BOC-Pmh-D-Pro-Hfe-Phe-OMe | 689.82 | 689.91 | 2.7 |
| A8 | BOC-Pmh-D-Pro-Cha-Phe-OMe | 681.84 | 681.93 | 2.8 |
| A9 | BOC-Pmh-D-Pro-Chg-Phe-OMe | 667.82 | 667.88 | 1.8 |
| A10 | BOC-Pmh-D-Pro-1-amino-1-cyclopentane carboxylic acid- Phe-OMe | 639.76 | 639.86 | 3.1 |
| A11 | BOC-Pmh-D-Pro-1-amino-1-cyclohexane carboxylic acid-Phe-OMe | 653.79 | 653.88 | 3.1 |
| A12 | BOC-Pmh-D-Pro-Gly-Phe-OMe | 585.67 | 585.79 | 2.0 |
| A13 | BOC-Pmh-D-Pro-L-Pro-Phe-OMe | 625.74 | 625.87 | 1.1 |
| A14 | BOC-Pmh-D-Pip-Aib-Phe-OMe | 627.75 | 627.87 | 1.5 |
| A15 | BOC-Pmh-D-Pip-Gly-Phe-OMe | 599.66 | 599.79 | 1.2 |
| A16 | BOC-Pmh-D-Pip-1-amino-1-cyclopentane carboxylic acid-Phe-OMe | 653.80 | 653.88 | 2.1 |
| A17 | BOC-Pmh-D-Pip-1-amino-1-cyclohexane carboxylic acid- Phe-OMe | 667.82 | 667.92 | 1.8 |
| A18 | BOC-Pmh-D-Pip-Dbg-Phe-OMe | 711.91 | 712.07 | 1.1 |
| A19 | BOC-Pmh-Aib-D-Pro-Phe-OMe | 613.73 | 613.83 | 1.0 |
| A20 | BOC-Pmh-Aib-L-Pro-Phe-OMe | 613.73 | 613.88 | 3.4 |
| A21 | BOC-Pmh-Phe-L-Pro-Phe-OMe | 675.80 | 675.90 | 1.1 |
| A22 | BOC-Pmh-L-Pro-L-Pro-Phe-OMe | 625.74 | 625.87 | 1.2 |
| A23 | BOC-Pmh-L-Pro-D-Pro-Phe-OMe | 625.74 | 625.86 | 1.1 |
| A24 | BOC-Pmh-L-Pro-Dbg-Phe-OMe | 697.89 | 698.02 | 1.1 |
| A25 | BOC-Pmh-D-Pro-Aib-Gly-Phe-OMe | 670.78 | 670.91 | 2.9 |
| A26 | BOC-Pmh-D-Pro-Aib-Hfe-Phe-OMe | 774.93 | 775.05 | 3.4 |
| A27 | BOC-Pmh-D-Pro-Phe-1-amino-1-cyclohexane carboxylic | 800.97 | 801.08 | 1.7 |


|  | acid-Phe-OMe |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
| A28 | BOC-Pmh-D-Pro-Hfe-Phe-Phe-OMe | 837.00 | 837.05 | 5.8 |
| A29 | BOC-Pmh-D-Pro-1-amino-1-cyclohexane carboxylic acid-Val-Phe-OMe | 752.92 | 753.05 | 3.8 |
| A30 | BOC-Pmh-D-Pro-Hfe-D-Phe-OMe | 689.82 | 689.95 | 1.9 |
| A31 | BOC-Pmh-D-Pro-1-amino-1-cyclohexane carboxylic acid-Val-D-Phe-OMe | 752.92 | 753.09 | 4.2 |
| A32 | BOC-Pmh-D-Pro-Hfe-Gly-D-Phe-OMe | 746.87 | 747.04 | 4.1 |
| A33 | BOC-Pmh-D-Pro-Aib-Hfe-D-Phe-OMe | 774.93 | 775.05 | 2.6 |
| A34 | BOC-Pmh-D-Pro-Aib-Phe-D-Phe-OMe | 760.90 | 761.05 | 6.5 |
| A35 | BOC-Pmh-D-Pro-2-amino-2-indan carboxylic acid-PheOMe | 687.81 | 687.29 | 2.3 |
| A36 | BOC-Pmh-D-Pro-2-amino-2-indan carboxylic acid-D-PheOMe | 687.81 | 687.23 | 2.0 |
| A37 | BOC-Pmh-D-Pro-1-amino-1-cylcooctane carboxylic acid-Phe-Ome | 681.84 | 681.27 | 1.7 |
| A38 | BOC-Pmh-D-Pip-Hfe-D-Phe-OMe | 703.85 | 703.24 | 1.7 |
| A39 | BOC-Pmh-D-Pip-1-amino-1-cylcooctane carboxylic acid- Phe-OMe | 695.87 | 695.32 | 2.1 |
| A40 | BOC-Pmh-D-Pro-2-amino-2-indan carboxylic acid-Phe-PheOMe | 834.98 | 834.26 | 3.8 |
| A41 | BOC-Pmh-D-Pro-2-amino-2-indan carboxylic acid-Hfe-Phe- <br> OMe | 849.01 | 848.25 | 2.7 |
| A42 | BOC-Pmh-D-Pro-1-amino-1-cyclooctane carboxylic acid-Cha-Phe-OMe | 834.06 | 834.36 | 3.0 |
| A43 | BOC-Pmh-D-Pro-1-amino-1-cyclooctane carboxylic acid-Chg-Phe-OMe | 821.04 | 820.38 | 2.7 |
| A44 | BOC-Pmh-D-Pro-1-amino-1-cyclohexane carboxylic acid-Cha-Phe-OMe | 807.01 | 806.37 | 4.7 |
| A45 | BOC-Pmh-D-Pro-1-amino-1-cyclohexane carboxylic acid-Leu-Phe-OMe | 766.95 | 766.38 | 4.8 |
| A46 | BOC-Pmh-D-Pro-1-amino-1-cyclohexane carboxylic acid- <br> Ile-Phe-OMe | 766.95 | 766.38 | 3.8 |
| A47 | BOC-Pmh-D-Pro-Hfe-D-Phe-Phe-OMe | 837.00 | 836.27 | 1.9 |
| A48 | BOC-Pmh-D-Pro-Gly-1-amino-1-cylcohexane carboxylic acid-Phe-OMe | 710.84 | 710.25 | 2.0 |
| A49 | BOC-Pmh-D-Pip-1-amino-1-cyclopentane carboxylic acid-Hfe-Phe-OMe | 814.99 | 814.25 | 5.5 |
| A50 | BOC-Pmh-D-Pip-1-amino-1-cyclohexane carboxylic acid-Hfe-Phe-OMe | 829.02 | 828.28 | 3.8 |
| A51 | BOC-Pmh-D-Pip-Hfe-Phe-Phe-OMe | 851.02 | 850.23 | 4.1 |
| A52 | BOC-Pmh-D-Pip-Aib-Cha-Phe-OMe | 780.98 | 780.32 | 5.2 |
| A53 | BOC-Pmh-D-Pro-Phe-Hfe-D-Phe-OMe | 835.99 | 836.43 | 2.2 |
| A54 | BOC-Pmh-D-Pro-Hfe-Cha-D-Phe-OMe | 842.03 | 842.48 | 4.9 |
| A55 | BOC-Pmh-D-Pro-Cha-Phe-D-Phe-OMe | 828.01 | 828.37 | 2.9 |
| A56 | BOC-Pmh-D-Pro-Hfe-Phe-D-Phe-Phe-OMe | 983.16 | 983.51 | 2.3 |
| A57 | BOC-Pmh-D-Pip-Asn(Trt)-Trp(Boc)-Phe-OMe | 1184.38 | 1185.28 | 1.8 |
| A58 | BOC-Pmh-D-Pip-Hfe-Asn(Trt)-Phe-OMe | 1059.26 | 1060.25 | 1.2 |
| A59 | BOC-Pmh-D-Pip-D-Val-Trp(Boc)-D-Phe-OMe | 927.10 | 928.26 | 1.1 |
| A60 | BOC-Pmh-Asn(Trt)-1-amino-1-cyclopentane carboxylic acid-Trp(Boc)-D-Phe-OMe | 1184.38 | 1185.28 | 1.8 |


| A61 | BOC-Pmh-D-Val-Aib-D-Ala-Ala-OMe | 610.73 | 611.08 | 1.4 |
| :---: | :---: | :---: | :---: | :---: |
| A62 | BOC-Pmh-D-Glu(OBut)-Aib-Cha-Ala-OMe | 778.96 | 779.12 | 1.3 |
| A63 | BOC-Pmh-Asn(Trt)-His( $\pi \mathrm{Bn}$ )-Asp-OBut-Ala-OMe | 1110.28 | 1110.11 | 1.1 |
| A64 | BOC-Pmh-Thr(But)-D-Glu(OBut)-Hfe-Ala-OMe | 859.04 | 859.14 | 1.4 |
| A65 | BOC-Pmh-D-Val-Ile-Chg-Ala-OMe | 706.90 | 707.16 | 1.5 |
| A66 | BOC-Pmh-Aib-Chg-Phe-Ala-OMe | 726.88 | 727.14 | 4.7 |
| A67 | BOC-Pmh-His( $\pi \mathrm{Bn}$ )-D-Glu(OBut)-Aib-Ala-OMe | 853.00 | 853.17 | 1.2 |
| A68 | BOC-Pmh-Phe-Phe-Leu-Ala-OMe | 762.92 | 763.12 | 1.4 |
| A69 | BOC-Pmh-Trp(Boc)-D-Ala-Glu(OBut)-Ala-OMe | 898.04 | 898.12 | 1.5 |
| A70 | BOC-Pmh-Glu(OcHx)-D-Ala-Ile-Ala-OMe | 750.90 | 751.16 | 1.9 |
| A71 | BOC-Pmh-Leu-Asn(Trt)-Chg-Ala-OMe | 964.19 | 964.16 | 1.1 |
| A72 | BOC-Pmh-Cha-D-Glu(OBut)-Phe-Ala-OMe | 841.03 | 841.16 | 2.4 |
| A73 | BOC-Pmh-Leu-Ile-Phe-Ala-OMe | 728.90 | 729.16 | 1.5 |
| A74 | BOC-Pmh-D-Val-D-Glu(OBut)-Asp-OBut-Ala-OMe | 810.96 | 811.12 | 1.3 |
| A75 | BOC-Pmh-Ile-Leu-Asp-OBut-Ala-OMe | 752.92 | 753.16 | 1.7 |
| A76 | BOC-Pmh-Cha-D-Ala-Hfe-Ala-OMe | 740.91 | 741.13 | 1.8 |
| A77 | BOC-Pmh-Phe-D-Glu(OBut)-Asn(Trt)-Ala-OMe | 1044.22 | 1044.15 | 4.2 |
| A78 | BOC-Pmh-Asn(Trt)-D-Ala-D-Glu(OBut)-Ala-OMe | 968.13 | 968.12 | 1.6 |
| A79 | BOC-Pmh-Asp-OBut-Leu-D-Glu(OBut)-Ala-OMe | 824.98 | 825.17 | 1.4 |
| A80 | BOC-Pmh-Cha-Hfe-Asn(Trt)-Ala-OMe | 1026.25 | 1026.17 | 1.4 |
| A81 | BOC-Pmh-Ile-Cha-Aib-Ala-OMe | 706.90 | 707.16 | 1.8 |
| A82 | BOC-Pmh-Ile-D-Val-D-Glu(OBut)-Ala-OMe | 752.92 | 753.16 | 2.0 |
| A83 | BOC-Pmh-D-Ala-D-Glu(OBut)-Hfe-Ala-OMe | 772.91 | 773.11 | 1.8 |
| A84 | BOC-Pmh-D-Ala-Asn(Trt)-Hfe-Ala-OMe | 944.11 | 944.11 | 2.7 |
| A85 | BOC-Pmh-Aib-Cha-Trp(Boc)-Ala-OMe | 866.04 | 865.98 | 4.6 |
| A86 | BOC-Pmh-Thr(But)-Thr(But)-Phe-Ala-OMe | 817.01 | 816.99 | 1.1 |
| A87 | BOC-Pmh-Asn(Trt)-Trp(Boc)-Aib-Ala-OMe | 1083.26 | 1082.94 | 1.1 |
| A88 | BOC-Pmh-Aib-Hfe-Glu(OcHx)-Ala-OMe | 812.97 | 813.01 | 1.8 |
| A89 | BOC-Pmh-D-Val-Aib-Glu(OcHx)-Ala-OMe | 750.90 | 751.02 | 1.5 |
| A90 | BOC-Pmh-D-Val-Ile-D-Ala-Ala-OMe | 638.78 | 639.03 | 1.4 |
| A91 | BOC-Pmh-D-Glu(OBut)-Trp(Boc)-Asn(Trt)-Ala-OMe | 1183.37 | 1182.94 | 1.4 |
| A92 | BOC-Pmh-D-Val-Cha-Cha-Ala-OMe | 760.99 | 761.11 | 1.9 |
| A93 | BOC-Pmh-Phe-Glu(OcHx)-Glu(OcHx)-Ala-OMe | 925.10 | 925.05 | 1.5 |
| A94 | BOC-Pmh-Glu(OcHx)-Cha-His( $\pi \mathrm{Bn}$ )-Ala-OMe | 947.15 | 947.09 | 2.0 |
| A95 | BOC-Pmh-D-Ala-D-Ala-Cha-Ala-OMe | 650.79 | 651.07 | 2.8 |
| A96 | BOC-Pmh-Asp-OBut-Asn(Trt)-Thr(But)-Ala-OMe | 1040.23 | 1040.09 | 1.3 |
| A97 | BOC-Pmh-Ile-His( $\pi \mathrm{Bn}$ )-Glu(OcHx)-Ala-OMe | 907.09 | 907.05 | 1.8 |
| A98 | BOC-Pmh-Hfe-Trp(Boc)-D-Val-Ala-OMe | 902.07 | 902.09 | 1.1 |
| A99 | BOC-Pmh-Asp-OBut-Asn(Trt)-Phe-Phe-OMe | 1106.29 | 1106.16 | 1.8 |
| A100 | BOC-Pmh-D-Val-His( $\pi \mathrm{Bn}$ )-His( $\pi \mathrm{Bn}$ )-Phe-OMe | 985.16 | 985.17 | 1.8 |
| A101 | BOC-Pmh-His( $\pi \mathrm{Bn}$ )-D-Glu(OBut)-D-Glu(OBut)-Phe-OMe | 1029.21 | 1029.16 | 1.4 |
| A102 | BOC-Pmh-His( $\pi \mathrm{Bn}$ )-D-Glu(OBut)-D-Ala-Phe-OMe | 915.07 | 915.15 | 1.7 |
| A103 | BOC-Pmh-Aib-D-Glu(OBut)-Hfe-Phe-OMe | 863.03 | 863.13 | 3.3 |
| A104 | BOC-Pmh-Glu(OcHx)-Phe-Asn(Trt)-Phe-OMe | 1146.36 | 1146.18 | 1.6 |
| A105 | BOC-Pmh-Phe-Thr(But)-D-Val-Phe-OMe | 835.02 | 835.15 | 1.3 |
| A106 | BOC-Pmh-Cha-D-Ala-Hfe-Phe-OMe | 817.01 | 817.13 | 2.4 |
| A107 | BOC-Pmh-Aib-Chg-Phe-Phe-OMe | 802.98 | 803.16 | 4.5 |
| A108 | BOC-Pmh-Thr(But)-Trp(Boc)-His( $\pi \mathrm{Bn}$ )-Phe-OMe | 1102.31 | 1102.26 | 1.2 |
| A109 | BOC-Pmh-Thr(But)-Thr(But)-D-Val-Phe-OMe | 845.06 | 845.20 | 1.1 |


| A110 | BOC-Pmh-D-Ala-Thr(But)-Ile-Phe-OMe | 772.95 | 773.15 | 3.2 |
| :---: | :---: | :---: | :---: | :---: |
| A111 | BOC-Pmh-Phe-Aib-Asp-OBut-Phe-OMe | 834.98 | 834.70 | 2.5 |
| A112 | BOC-Pmh-Asn(Trt)-Chg-Aib-Phe-OMe | 1012.22 | 1012.17 | 1.3 |
| A113 | BOC-Pmh-Asp-OBut-His( $\pi \mathrm{Bn}$ )-Hfe-Phe-OMe | 991.16 | 991.26 | 1.6 |
| A114 | BOC-Pmh-Ile-Phe-D-Val-Phe-OMe | 790.97 | 791.16 | 1.4 |
| A115 | BOC-Pmh-Phe-D-Ala-Leu-Phe-OMe | 762.92 | 763.17 | 3.4 |
| A116 | BOC-Pmh-Thr(But)-His( $\pi \mathrm{Bn}$ )-Asn(Trt)-Phe-OMe | 1172.40 | 1172.27 | 1.2 |
| A117 | BOC-Pmh-Ile-Glu(OcHx)-D-Glu(OBut)-Phe-OMe | 941.14 | 941.24 | 1.7 |
| A118 | BOC-Pmh-Aib-D-Val-Trp(Boc)-Phe-OMe | 902.07 | 902.20 | 2.2 |
| A119 | BOC-Pmh-D-Ala-D-Ala-Leu-Phe-OMe | 686.82 | 687.15 | 2.3 |
| A120 | BOC-Pmh-D-Val-Trp(Boc)-D-Glu(OBut)-Phe-OMe | 1002.18 | 1002.25 | 1.3 |
| A121 | BOC-Pmh-D-Glu(OBut)-D-Val-Chg-Phe-OMe | 855.05 | 855.26 | 1.5 |
| A122 | BOC-Pmh-D-Glu(OBut)-Cha-D-Ala-Phe-OMe | 841.03 | 841.25 | 1.1 |
| A123 | BOC-Pmh-Leu-Ile-Glu(OcHx)-Phe-OMe | 869.08 | 869.28 | 1.5 |
| A124 | BOC-Pmh-Asp-OBut-D-Ala-Glu(OcHx)-Phe-OMe | 884.03 | N.A. | 1.4 |
| A125 | BOC-Pmh-Phe-Aib-Asn(Trt)-Phe-OMe | 1020.20 | 1020.24 | 3.7 |
| A126 | BOC-Pmh-Asn(Trt)-D-Glu(OBut)-D-Val-OMe | 1072.28 | 1072.29 | 1.3 |
| A127 | BOC-Pmh-Trp(Boc)-D-Val-D-Val-Phe-OMe | 916.10 | 916.23 | 2.0 |
| A128 | BOC-Pmh-Leu-Chg-Hfe-Phe-OMe | 845.06 | 845.25 | 2.8 |
| A129 | BOC-Pmh-Cha-Leu-Trp(Boc)-Phe-OMe | 984.21 | 984.31 | 3.0 |
| A130 | BOC-Pmh-Asn(Trt)-Ile-Hfe-Phe-OMe | 1062.28 | 1062.29 | 1.7 |
| A131 | BOC-Pmh-Asn(Trt)-Aib-His( $\pi \mathrm{Bn}$ )-Phe-OMe | 1100.29 | 1100.28 | 1.4 |
| A132 | BOC-Pmh-His( $\pi \mathrm{Bn}$ )-Cha-D-Ala-Phe-OMe | 883.07 | 883.26 | 1.2 |
| A133 | BOC-Pmh-Glu(OcHx)-Glu(OcHx)-D-Glu(OBut)-Phe-OMe | 1039.24 | 1039.33 | 1.8 |
| A134 | BOC-Pmh-Cha-Asn(Trt)-Aib-Phe-OMe | 1026.25 | 1026.28 | 1.1 |
| A135 | BOC-Pmh-Hyp(But)-Aib-Cha-Ala-OMe | 761.95 | 763.36 | 1.1 |
| A136 | BOC-Pmh-Hyp(But)-1-amino-1-cyclohexane carboxylic acid-Asn(Trt)-Phe-OMe | 1081.30 | 1082.38 | 1.2 |
| A137 | BOC-Pmh-Hyp(But)-Aib-Hfe-Phe-OMe | 846.02 | 847.32 | 1.1 |
| A138 | BOC-Pmh-Hyp(But)-D-Val-Cha-Ala-OMe | 776.00 | 777.37 | 1.1 |
| A139 | BOC-Pmh-Hyp(But)-Chg-Cha-Phe-OMe | 892.13 | 893.44 | 1.3 |
| A140 | BOC-Pmh-D-Tic-Hfe-Val-Phe-OMe | 850.01 | 851.34 | 3.2 |
| A141 | BOC-Pmh-D-Tic-Trp(Boc)-Leu-Phe-OMe | 989.17 | 990.35 | 3.0 |
| A142 | BOC-Pmh-D-Tic-1-amino-1-cyclohexane carboxylic acid- Asn(Trt)-Ala-OMe | 995.17 | 996.40 | 1.5 |
| A143 | BOC-Pmh-D-Tic-1-amino-1-cyclopentane carboxylic acid-Hfe-Phe-OMe | 862.02 | 863.33 | 2.8 |
| A144 | BOC-Pmh-Val-Thr-( $\left.\Psi^{\text {Me,Me }}\right)$ pro-Aib-Leu-Ala-OMe | 792.96 | 794.37 | 1.1 |
| A145 | BOC-Pmh-Val-Thr-( $\left.\Psi^{\text {Me,Me }}\right)$ pro-Aib-Hfe-Phe-OMe | 917.10 | 918.24 | 1.1 |
| A146 | BOC-Pmh-Phe-2-Abz-Leu-Ala-OMe | 733.85 | 735.31 | 1.9 |
| A147 | BOC-Pmh-Hfe-2-Abz-Hfe-Phe-OMe | 872.02 | 873.54 | 2.0 |
| A148 | BOC-Pmh-Val-2-Abz-Hfe-Phe-OMe | 809.95 | 811.25 | 1.3 |
| A149 | BOC-Pmh-Asp-OBut-Asp-OBut-Phe-Phe-OMe | 920.06 | 921.38 | 1.6 |
| A150 | BOC-Pmh-Val-Val-D-Pro-Gly-Leu-Val-Val-OMe | 947.17 | 947.56 | 1.4 |
| A151 | BOC-Pmh-Thr(But)-D-Val-His(Trt)-D-Phe-D-Val-Thr(But)-Ile-OMe | 1436 | 1436 | 2.7 |

${ }^{1}$ Unnatural amino acid abbreviations: $2-\mathrm{Abz}=2$-aminobenzoic acid, $\mathrm{Aib}=\alpha$ aminomethylalanine, $\mathrm{Cha}=3$-cyclohexylalanine, $\mathrm{Chg}=\alpha$-cyclohexylglycine, $\mathrm{Dbg}=$ di-n-butylglycine, $\mathrm{Hfe}=$ homophenylalanine, $\operatorname{Hyp}(\mathrm{But})=$ O-t-butyl-L-trans-4hydroxyproline, D-Pip = D-pipecolic acid, Pmh $=\pi$-methyl histidine, D -Tic $=(3 \mathrm{R})$ -1,2,3,4-tetrahydroisoquinoline-3-carboxylic acid. Protecting group abbreviations: Trt = trityl, But = t-butyl, $\mathrm{cHx}=$ cyclohexyl, $\mathrm{Bn}=$ benzyl.

