

Table S1. Polymerization of GPE with **1**, **2**, and **3** (1mol %) in the presence of ZnCl₂ (0.2 mol %)

initiator	temp (°C)	conv <i>a</i> (%)	yield <i>b</i> (%)	M _n <i>c</i> (MWD)	M _w of high MW
1	110	10	3	5800 (1.52) <i>d</i>	24700
	130	36	23	2900 (1.76) <i>d</i>	10300
	150	83	68	3000 (1.15) <i>e</i>	
	160	93	89	2800 (1.22) <i>e</i>	
2	110	13	<1	6800 (1.61) <i>d</i>	35600
	130	19	2	5900 (1.44) <i>e</i>	
	150	64	41	2700 (1.13) <i>e</i>	
	170	100	68	2700 (1.24) <i>e</i>	
3	170	18	<2	2000 (1.58) <i>d</i>	3700
	190	97	81	3100 (1.40) <i>e</i>	

a Determined by ¹H NMR. *b* Methanol-insoluble part.*c* Estimated by GPC based on polystyrene standard samples.*d* Contained a shoulder at a high molecular weight region.*e* Unimodal GPC chart.

Table S2. Concentration effect of ZnCl_2 on the polymerization of GPE with **2** *a*

run	amount of ZnCl_2 (mol %)	conv <i>b</i> (%)	yield <i>c</i> (%)	M_n <i>d</i>	M_w/M_n <i>d</i>
1	0	4	-	-	-
2	0.1	21	trace	2900	1.12
3	0.2	64	41	2700	1.13
4	0.6	95	88	2600	1.16
5	1	100	92	3600	2.32

a Condition; **2** 1 mol %, 150 °C, 12 h. *b* Determined by $^1\text{H NMR}$.*c* Methanol-insoluble part. *d* Estimated by GPC based on polystyrene standard samples.

Table S3. Effect of polymerization time on the polymerization of GPE with **2** in the presence of ZnCl₂ *a*

temp (°C)	time (h)	conv <i>b</i> (%)	yield <i>c</i> (%)	M _n <i>d</i>	M _w /M _n <i>d</i>
150	6	37	6	2100	1.10
	12	64	41	2700	1.13
	16	82	42	2600	1.09
	24	99	77	3400	1.22
170	4	55	30	2900	1.20
	8	84	59	2700	1.25
	12	100	88	2700	1.24

a Condition: **2** 1 mol%, ZnCl₂ 0.2 mol%, for 12 h. *b* Determined by ¹H NMR.*c* Methanol-insoluble part. *d* Estimated by GPC based on polystyrene standard samples.

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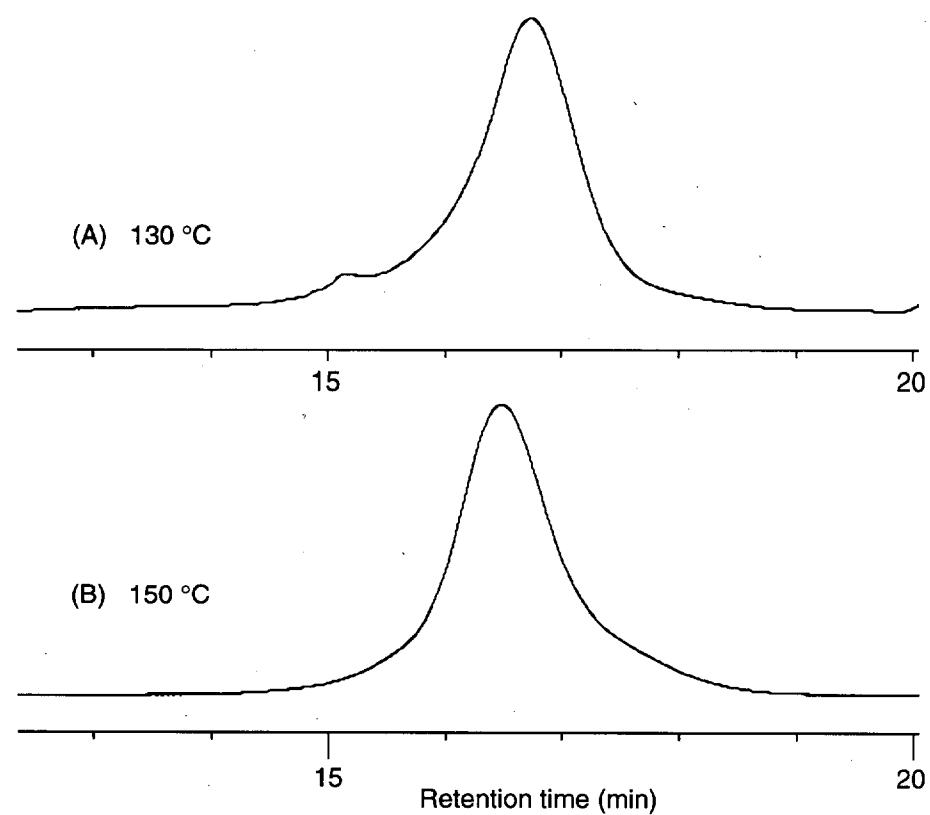


Figure S6. GPC profiles of the polymers obtained by the polymerization of GPE with **1** (1 mol %) in the presence of $ZnCl_2$ (0.2 mol %) at (A) 130 °C and (B) 150 °C for 12 h.