1) Experimental details for the synthesis of starting resins **1a-g**. Experimental details and ¹H NMR data for the synthesis of *N*-alkyl amino acid trifluoroacetate salts **5a-g** (Table 1).

Experimental Section. General. All reagents used are commercially available and were employed without further purification. Fmoc-aminoacid Wang resins (1% divinyl benzene-polystyrene), all aminoacids, and 2,4,5-trichlorophenyl formate were purchased from Novabiochem. All glassware employed in solid-phase reactions had been silanized (treatment with 10% TMSCl/toluene for 12h) and dried. Polypropylene filter vessels were obtained from Bio-Rad. Dried tetrahydrofuran for the reductions was obtained by distillation over sodium/benzophenone ketyl. NMR spectra were recorded on Bruker AM 300, or 200 MHz instruments. Compound purity analysis was carried out by RP-HPLC on a Hewlett-Packard 1100 system using conditions described in the manuscript. Yields of crude cleaved compounds are based on the mass balance based upon the starting loading level of commercial resins.

Typical synthesis of resin-bound *N*-formyl-amino acids. *N*-formyl-phenylalanine Wang resin (1d). A sample of N-Fmoc-phenylalanine Wang resin (1.00 g, 0.40 mmol/g substitution) in a polypropylene vessel was treated with 20% piperidine/DMF (5 mL, 3 min.; then 5 mL, 25 min.). After rinsing with DMF (3 x 5mL), the resin was suspended in dry DMF (5 mL) and 2,4,5- trichlorophenyl formate (0.27 g, 1.2 mmol) was added. The reaction vessel was shaken for 2h at rt, after which time the resin was washed successively with DMF, MeOH, and CH_2Cl_2 (3 times each), and dried under high vacuum for >12h to give resin 1d (0.86 g). A resin sample gave a negative result on a Ninhydrin test.

Typical synthesis of resin-bound *N*-acyl-amino acids. *N*-Acetyl-valine Wang resin (1f). A sample of N-Fmoc-valine Wang resin (1.00 g, 0.64 mmol/g substitution) in a polypropylene vessel was deprotected and rinsed as above. Then, it was suspended in dry DMF (5 mL), then dry Et₃N (0.2 mL) and acetic anhydride (0.5 mL) were added successively. The reaction vessel was shaken for 2h at rt, after which time the resin was washed with DMF, MeOH, and CH₂Cl₂ (3 times each), and dried under high vacuum for >12h to give resin 1f (0.84 g). A resin sample gave a negative result on a Ninhydrin test.

Typical procedure for Diborane Reduction and Oxidative Work-Up. *N*-Ethylvaline trifluoroacetate salt (5f). A portion of *N*-Acetyl-valine Wang resin (1f) (0.200 g, 0.14 mmol, 0.70 mmol/g substitution) was weighed in a 10 mL silanized round bottom flask and swelled in dry THF (1.5 mL) under nitrogen. The diborane solution

(1M/THF, 0.56 mL, 0.56 mmol) was added dropwise at rt over 2 min., after which time the flask was equipped with a condenser and the suspension stirred gently at 65 °C for 12h. Upon cooling to rt, the suspended resin was rapidly transferred by pipet (silanized) to a PP vessel by using dry THF to rinse out the flask and wash the resin. Then, dry THF (2.0 mL), anhydrous diisopropylethylamine (0.4 mL) and glacial acetic acid (0.8 mL) were added successively. To the homogenized suspension was added iodine (70 mg, 0.28 mmol, as a conc. THF solution) and the vessel was shaken for 1h. The vessel was then rinsed (THF) and the resin was washed (3X each) with THF, DMF/Et₃N 3:1, MeOH, CH₂Cl₂, and dried under high vaccuum for >12h to give resin **5f** (0.194 g). A resin sample gave a positive result on a bromophenol blue test.

The bulk of resin **5f** (0.188 g, \sim 0.70 mmol/g) was then transferred to a small round bottom flask and stirred in a 90% TFA/CH₂Cl₂ cocktail (2 mL) for 2h. The contents were filtered through a glasswool plug, the resin was rinsed with TFA, and the filtrate evaporated and dried over high vacuum for >12h to give crude trifluoroacetate salt **5f** as a white powder (23 mg, 69%). Its purity was estimated to >95% by ¹H NMR and RP-HPLC.

¹H NMR data for all *N*-alkyl amino acid trifluoroacetate salts 5a-g (Table 1): **N-Me-Ala (5a).** ¹H NMR (300 MHz, CD₃OD) δ 3.92 (q, J = 7.5 Hz, 1H), 2.70 (s, 3H), 1.54 (d, J = 7.5 Hz).

N-Me-Val (5b). ¹H NMR (300 MHz, CD₃OD) δ 3,77 (d, J = 6.0 Hz, 1H), 2.73 (s, 3H), 2.30 (m, 1H), 1.13 (d, J = 7.0 Hz, 3H), 1.05 (d, J = 7.0 Hz, 3H).

N-Me-Ser (5c). ¹H NMR (300 MHz, CD₃OD) δ 4.01 (d, J = 6.0 Hz, 2H), 3.86 (t, J = 6.0 Hz, 1H), 2.73 (s, 3H).

N-Me-Phe (5d). ¹H NMR (300 MHz, CD₃OD) δ 7.4-7.2 (m, 5H), 4.23 (t, J = 6.0 Hz, 1H), 3.3 (m, 2H), 2.70 (s, 3H).

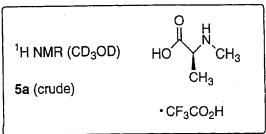
N-Et-Phe (5e). ¹H NMR (300 MHz, CD₃OD) δ 7.4-7.2 (m, 5H), 4.23 (t, J = 6.0 Hz, 1H), 3.4-3.1 (m, 2H), 3.09 (q, J = 7.0 Hz, 2H), 1.29 (t, J = 7.0 Hz, 3H).

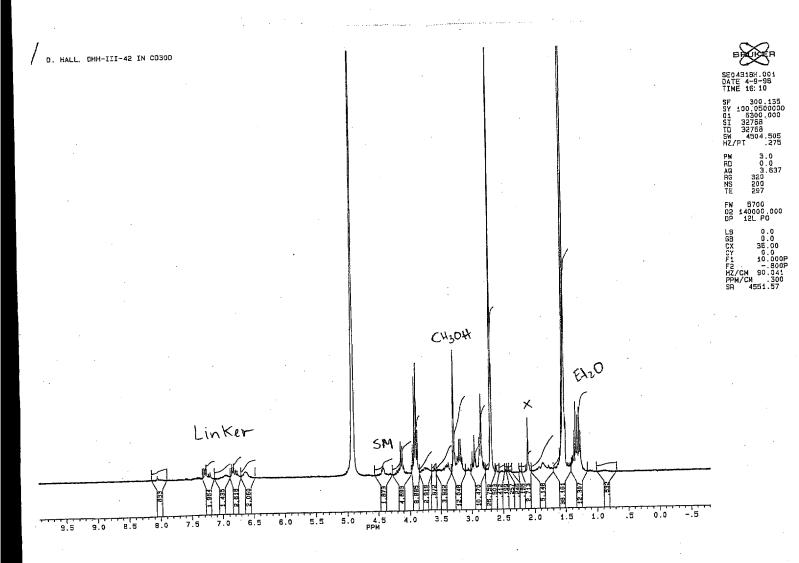
N-Et-Val (5f). ¹H NMR (300 MHz, CD₃OD) δ 3.76 (d, J = 6.0 Hz, 1H), 3.09 (q, J = 6.0 Hz, 2H), 2.28 (m, 1H), 1.32 (t, J = 6.0 Hz, 3H), 1.14 (d, J = 7.0 Hz, 3H), 1.05 (d, J = 7.0 Hz, 3H). **N-Pr-Val (5g).** ¹H NMR (300 MHz, CD₃OD) δ 3.78 (d, J = 6.0 Hz, 1H), 2.97 (m, 2H), 2.30 (m,

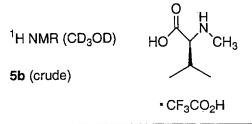
1H), 1.75 (m, 2H), 1.15 (d, J = 7.0 Hz, 3H), 1.05 (d, J = 7.0 Hz, 3H), 1.00 (t, J = 6.5 Hz, 3H).

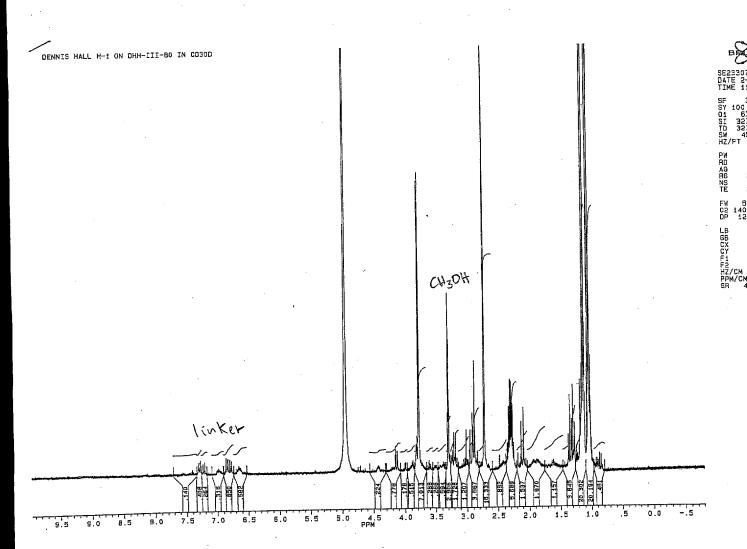
2) ¹H NMR spectra of crude *N*-alkyl amino acid trifluoroacetate salts **5a-g** (Table 1).

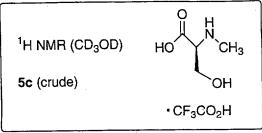
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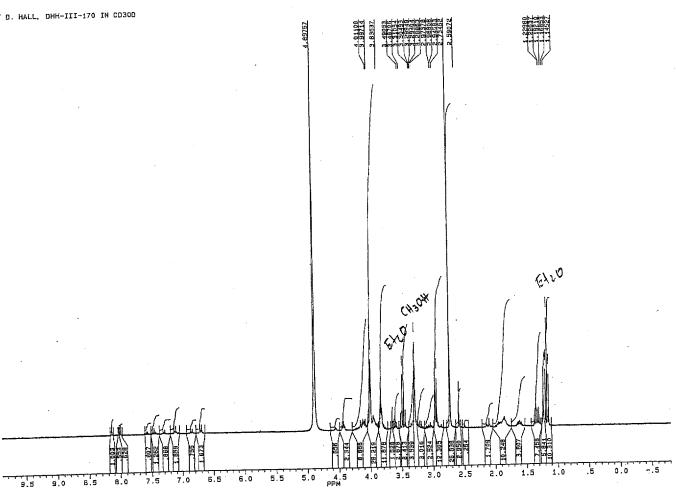




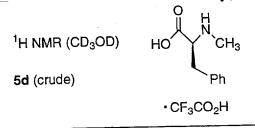


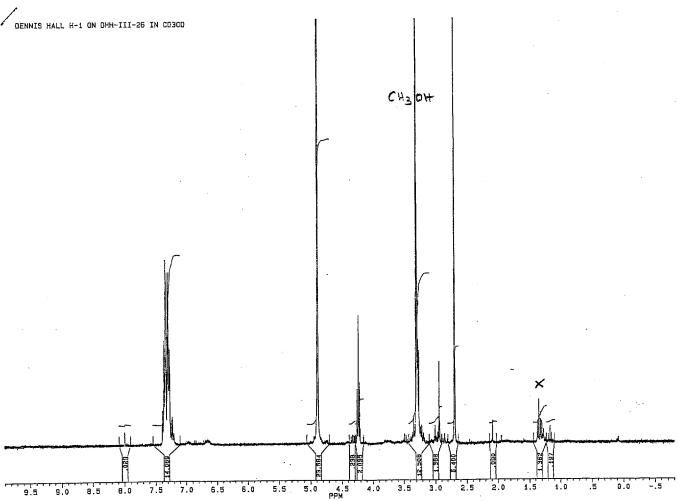


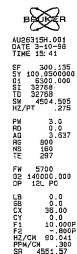


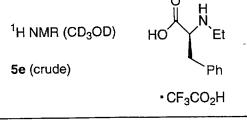


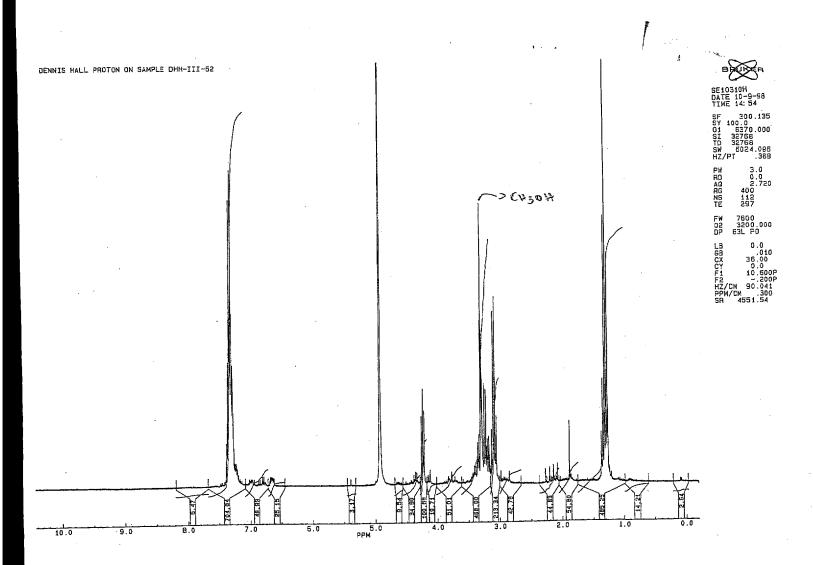


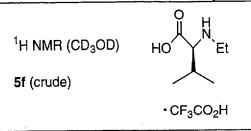


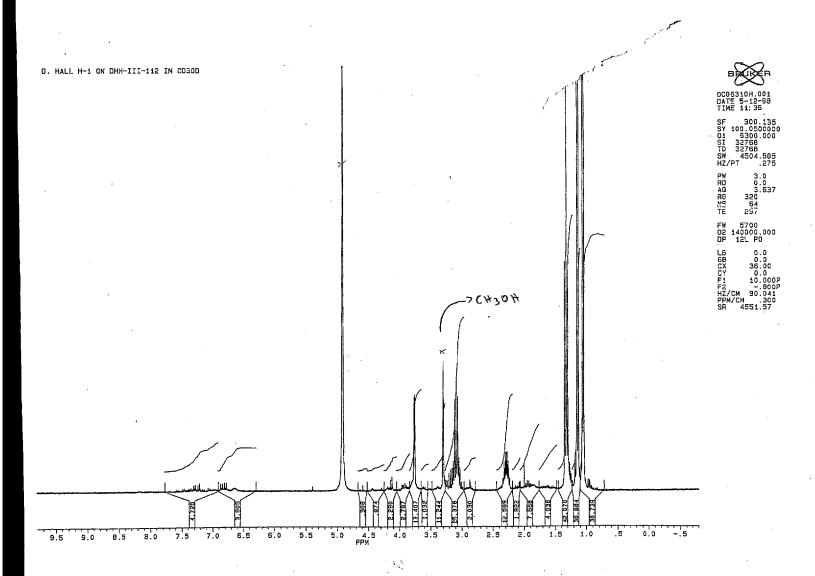


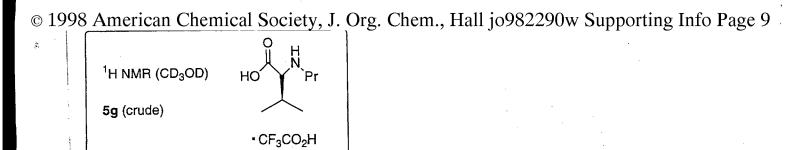


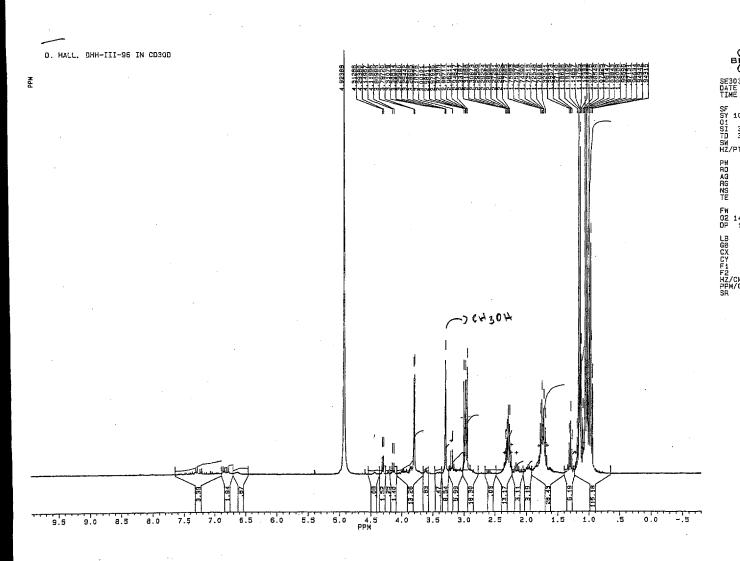












3) Experimental details for the formation of synthetic Fmoc-MePhe-OH, its ¹H NMR spectra, and optical rotation measurement compared with an authentic sample.

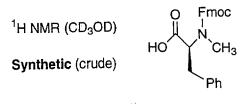
Synthetic *N*-Fmoc-*N*-Methylphenylalanine. A sample of fresly prepared resin 4d (0.170 g, 0.075 mmol, 0.44 mmol/g substitution) in a PP vessel was suspended in dry CH₂Cl₂ (2 mL). Diisopropylethylamine (0.075 mL, 0.45 mmol) and FmocCl (75 mg, 0.30 mmol) were added successively and the vessel was shaken 6h at rt. The resin was then washed (3X each) with CH₂Cl₂, DMF, CH₂Cl₂, and dried under high vaccuum for >12h. A resin sample gave a negative result on a bromophenol blue test.

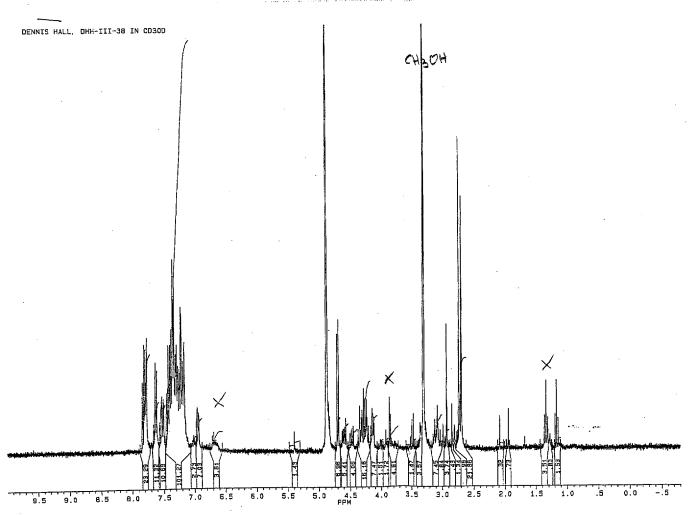
The bulk of the resin (0.175 g) was then cleaved as above for compound **5f**. The crude synthetic sample of Fmoc-MePhe-OH was obtained as a white powder (27 mg, 88%). Although of good purity by ¹H NMR, it was further purified by preparative TLC on silica gel (0.5 mm, eluant: 5% MeOH/ CH₂Cl₂). The final purity was estimated to >95% by ¹H NMR and RP-HPLC.

Optical rotation measurements (Perkin-Elmer 241 polarimeter, 589 nm, cell length 1 dm, at 25 °C):

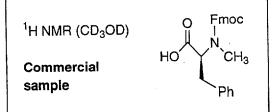
Synthetic Fmoc-MePhe-OH: 13.0 mg in 1.0 mL MeOH = -0.708° , $[\alpha]_D = -56^{\circ}$

Fmoc-MePhe-OH (Novabiochem): 15.0 mg in 1.0 mL MeOH = -0.899° , $[\alpha]_D = -60^{\circ}$

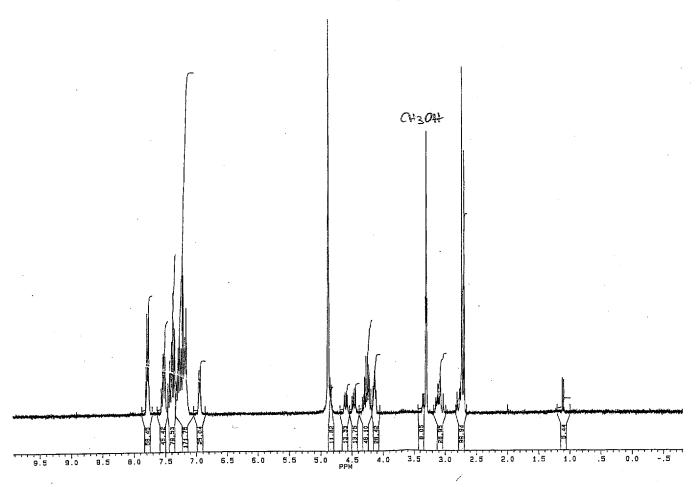








D. HALL, STANDARD IN CD30D



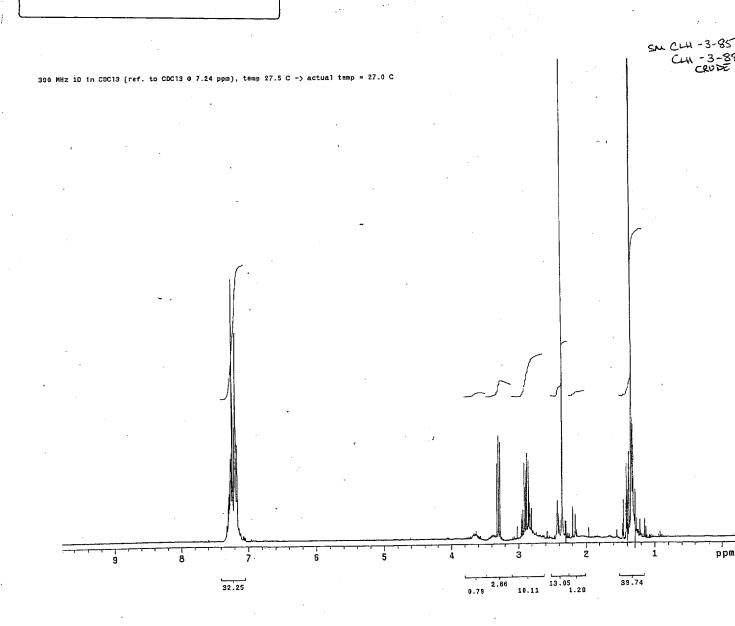


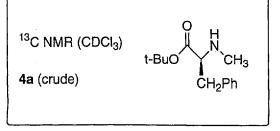
4) Experimental details and selected NMR spectral data (¹H, ¹³C) for *N*-alkyl amino acid esters **4h-j** and their precursors (Table 2).

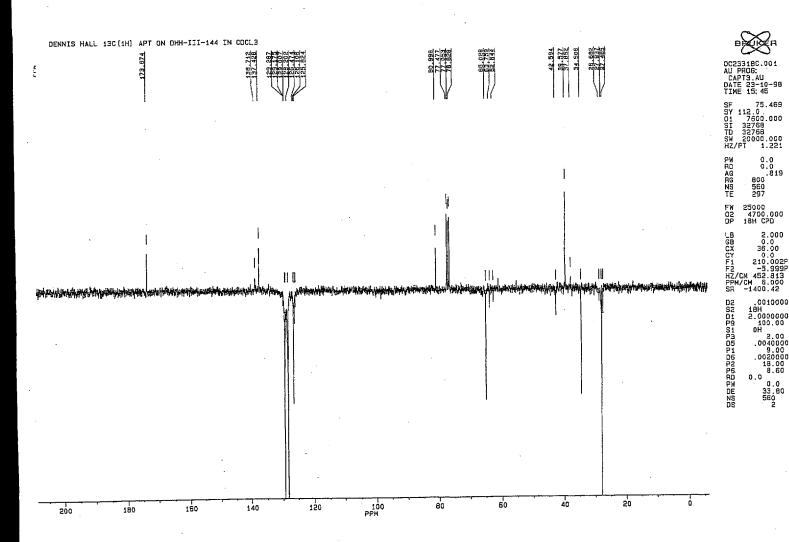
N-Acyl amino acid esters. These *N*-formyl and *N*-acetyl precursors were synthesized by means similar to the solid-phase examples described above.

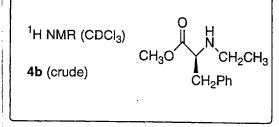
Typical procedure for Diborane Reduction and Oxidative Work-Up. *N*-Methylphenylalanine *t*-butyl ester (4h). The diborane solution (1M/THF, 8.9 mL, 8.9 mmols) was added at 0 °C to a solution of 1h (0.97 g, 3.9 mmols) in dry THF (6.7 mL). The solution was stirred at reflux temperature for 4h. Upon cooling to rt, triethylamine (2 mL), glacial acetic acid (3 mL) and iodine (1.09 g, 4.3 mmols, in THF) were added successively. The mixture was stirred for 2h then transferred slowly to an extraction funnel containing conc. aqueous NaOH and aq. saturated Na₂S₂O₃ (10 mL). The pH was adjusted to >11 with NaOH pellets, and the solution is extracted with ether (4x). The combined organic layers were washed with brine (10 mL), dried with anhydrous MgSO₄, filtered and concentrated. The brown oil was then dried under high vacuum for 4h, affording crude compound 4h (0.77 g, 84%). It was found of satisfying purity (>90%) according to ¹H and ¹³C NMR analysis. Highly pure analytical samples can be obtained by flash-chromatography purification using 1-5% MeOH/CH₂Cl₂.

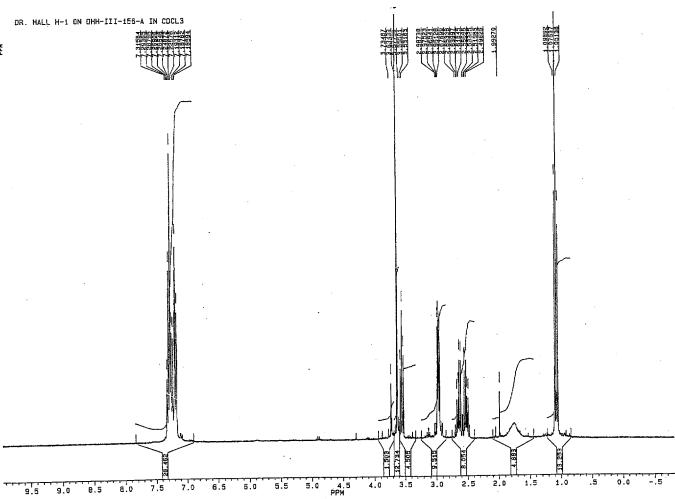
Selected ¹H and ¹³C NMR spectra of compounds 4h-j are included in the following pages.

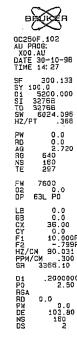


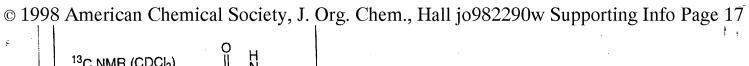


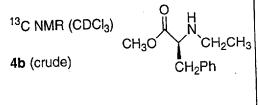


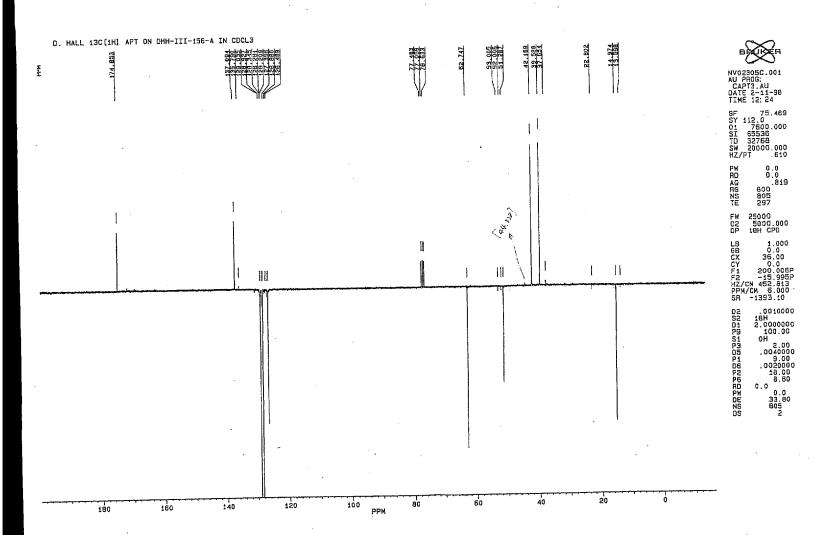


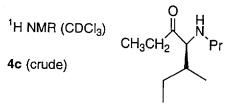


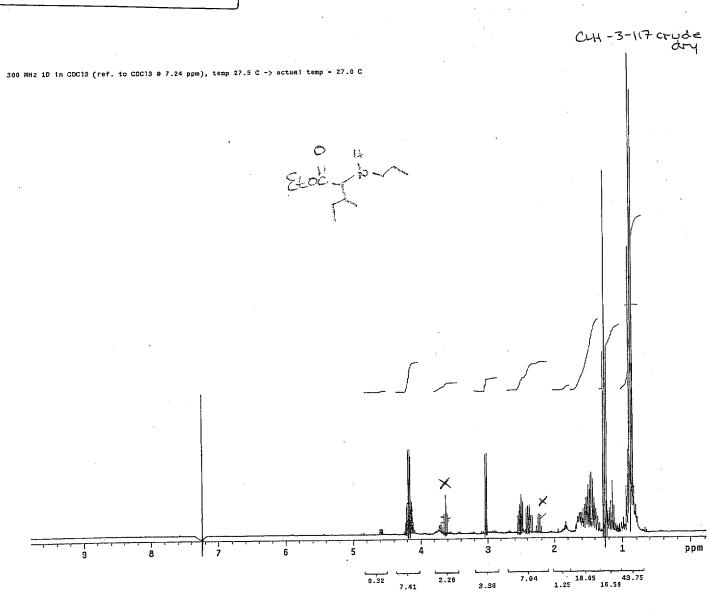












5) Experimental details and relevant MS and NMR spectra (¹H, ¹³C) for the synthesis of tetraamine 8 and its triacetylated derivative 9 (Scheme 2). RP-HPLC trace of Fmocderivatized 9.

Synthesis of model tripeptide 6. It was synthesized from aminopropyl-trityl resin using standard Fmoc amino acid coupling chemistry. The *N*-terminal alanine was capped with an acetyl group using the procedure described above for **1f**. The cleaved peptide (5% TFA/CH₂Cl₂ was found of >95% purity according to ¹H NMR analysis.

Synthesis of chiral tetraamine tetrakis(trifluoroacetate) salt (8). A portion of tripeptide trityl resin 7 (0.351g, 0.259 mmol, 0.74 mmol/g substitution) was weighed in a 50 mL silanized round bottom flask. The diborane solution (1M/THF, 5.2 mL, 5.2 mmol) was added dropwise under nitrogen atmosphere at rt over 5 min., after which the flask was equipped with a condenser and the suspension refluxed gently (65 °C) for 24h. Upon cooling to rt, the suspended resin was rapidly transferred by pipet (silanized) to a PP vessel by using dry THF to rinse out the flask and wash the resin. Then, dry THF (3 mL), anhydrous diisopropylethylamine (0.7 mL) and glacial acetic acid (1.4 mL) were added successively. To the homogenized suspension was added iodine (1.32 g, 5.2 mmol, as a conc. THF solution) and the vessel was shaken for 4h. The vessel was then rinsed (THF) and the resin was washed (3× each) with THF, DMF/Et₃N 3:1, MeOH, CH₂Cl₂, and dried under high vaccuum for >12h to give free tetraamino-resin 7 (0.344 g). A resin sample gave a highly positive result on a bromophenol blue test.

A portion of fresly prepared resin **7** (0.174 g, 0.77 mmol/g) was then transferred to a round bottom flask and stirred in a 5% TFA/CH₂Cl₂ cleavage cocktail (10 mL) for 2h. The contents were filtered through a glasswool plug, the resin was rinsed with 5% TFA/CH₂Cl₂, and the filtrate was evaporated and dried over high vacuum for >12h to give crude tetraamine tetrakis(trifluoroacetate) salt **8** as a yellow oil (95 mg, 95%). Its purity was estimated to >95% as determined by ¹H and ¹³C NMR (see following pages).

Synthesis of tetraamine triacetate trifluoroacetate salt (9). This compound was synthesized from resin 7 (0.119 g, 0.092 mmol) using a procedure similar to compound 1f. Upon resin cleavage as above, crude compound 9 was obtained as a yellowish oil (41 mg, 96%). Its purity was estimated to >95% according to RP-HPLC analysis (see following pages) with pre-column derivatization as a Fmoc carbamate (see note 20).

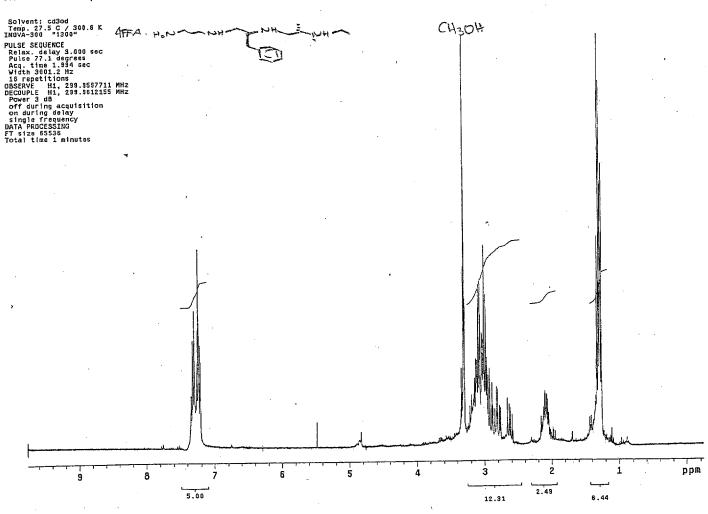
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**H NMR (CD3OD) H2N N H Bn H Et H Grude)

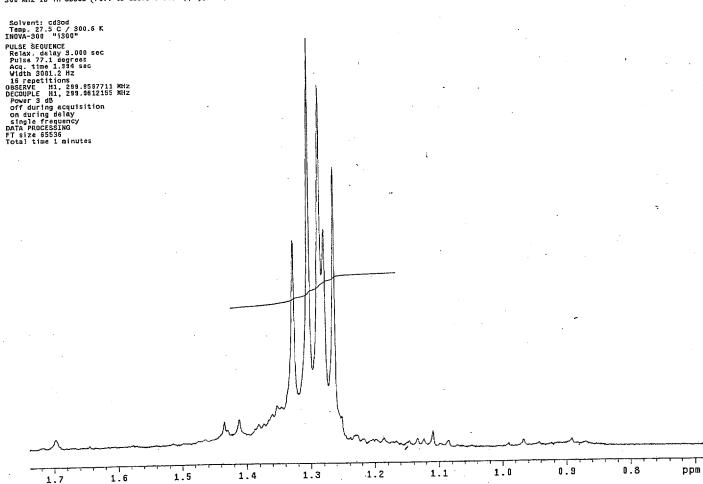
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SMH(#)41

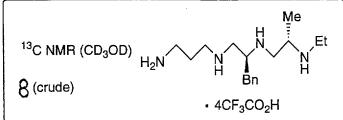
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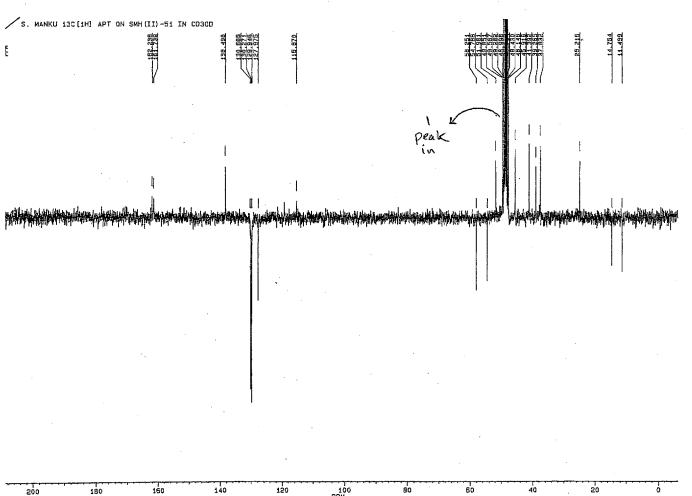


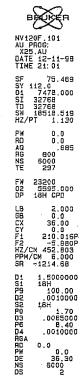
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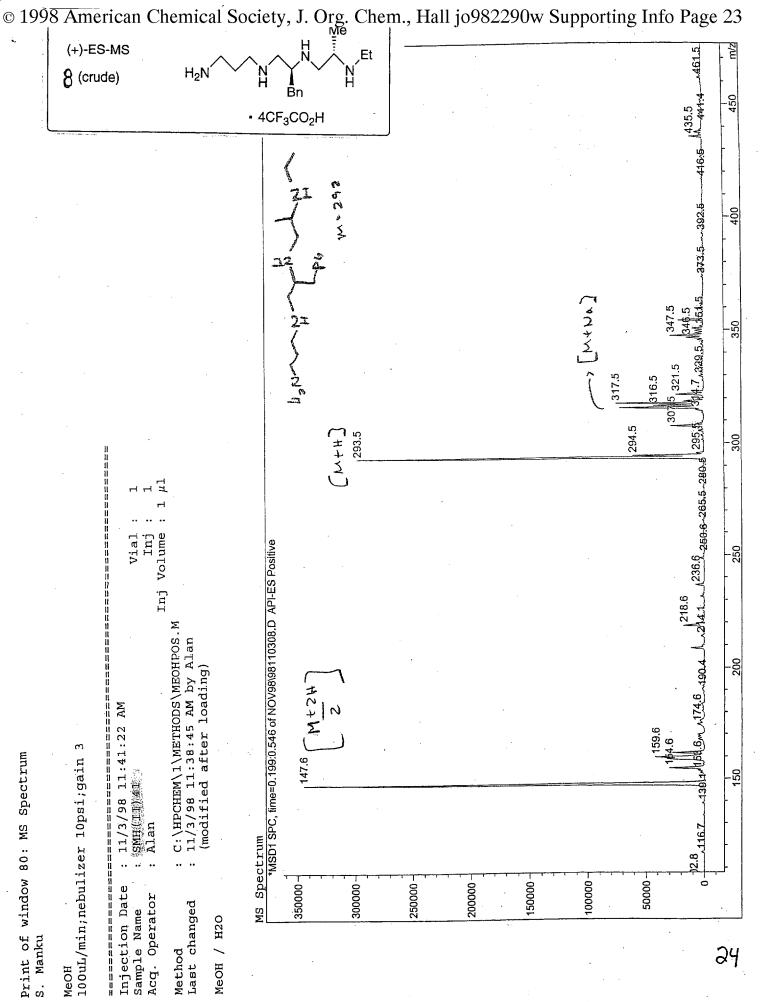


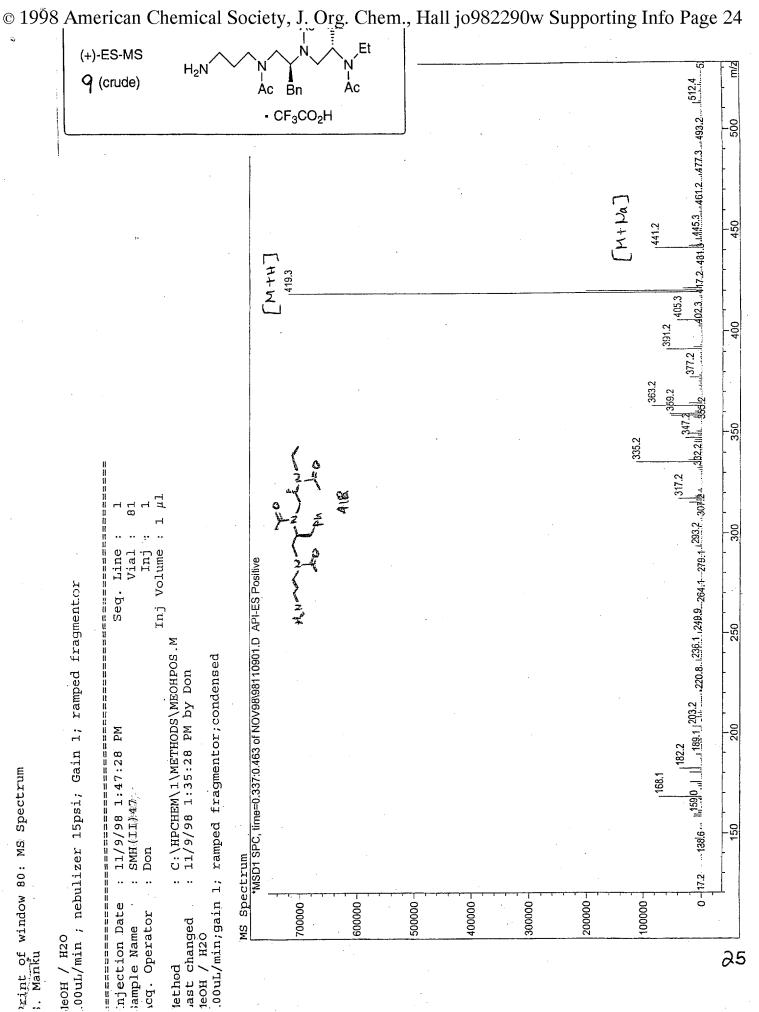
6.44











Sample of Synthetic cathaduline lumn deriv. with Fmoc. General method at 280 nm UV. S

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: 11/11/98 5:46:52 PM Injection Date

: SMH-II-47 Sample Name

Vial:

Acq. Operator : Dennis

Inj Volume : 10 µl

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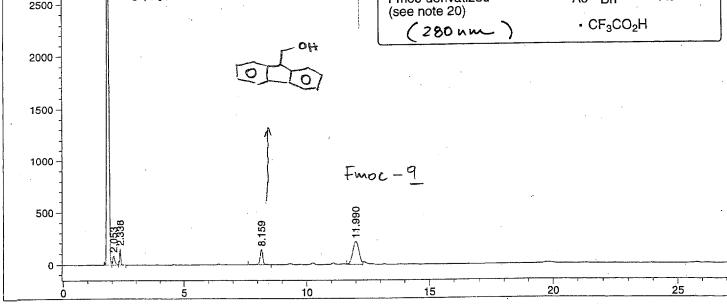
Analysis Method : C:\HPCHEM\1\METHODS\DENNIS\DENNIS3.M

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Basic method to analyze Fmoc-aminoacids and oligoamines with or without pre-column

derivatization. Made for completing JOC manuscript on November 8 1998. Employs 0.1% aq, TFA

ACN. DAD1 E, Sig=280,16 Ref=360,100 (DENNIS\DH316418.D) Мe mAU Ac RP-HPLC 3000 Et Phosphale 4 (crude) Pre-column buffer Ac Fmoc-derivatized Ac Bn 2500 (see note 20) CF₃CO₂H 280 nm



Area Percent Report

Signal Sorted By 1.0000 Multiplier 1.0000 Dilution

Signal 1: DAD1 E, Sig=280,16 Ref=360,100

Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1 2 3 4 5	1.820 2.053 2.338 8.159 11.990	VV VB BB	0.0935 0.0573 0.1233	1.98705e4 551.69836 560.41656 1148.89429 3884.49683	3083.13501 85.28393 150.98053 140.61037 217.45985	76.3780 2.1206 2.1541 4.4161 14.9312

Totals :

2.60160e4 3677.46969

Results obtained with enhanced integrator!

