

Terms & Conditions

Electronic Supporting Information files are available without a subscription to ACS Web Editions. The American Chemical Society holds a copyright ownership interest in any copyrightable Supporting Information. Files available from the ACS website may be downloaded for personal use only. Users are not otherwise permitted to reproduce, republish, redistribute, or sell any Supporting Information from the ACS website, either in whole or in part, in either machine-readable form or any other form without permission from the American Chemical Society. For permission to reproduce, republish and redistribute this material, requesters must process their own requests via the RightsLink permission system. Information about how to use the RightsLink permission system can be found at <http://pubs.acs.org/page/copyright/permissions.html>



ACS Publications

MOST TRUSTED. MOST CITED. MOST READ.

Copyright © 1998 American Chemical Society

Journal of Organic Chemistry

Competition between Hetero-Diels-Alder and Cheletropic Addition of Sulfur Dioxide. Theoretical and Experimental Substituent Effects on the Relative Stability of 3,6-Dihydro-1,2-oxathiin-2-oxides (Sultines) and 2,5-dihydrothiophene-1,1-dioxides (Sulfolenes). Anomeric Effects in Sultine and 6-substituted Derivatives.

Dimas Suárez[†], José A. Sordo*[†]

Frédéric Monnat[‡], Elena Roversi[‡], Antonio Estrella de Castro[‡], Kurt Schenk[‡] and Pierre Vogel*[‡]

Departamento de Química Física y Analítica, Universidad de Oviedo, Oviedo, Principado de Asturias
Section de chimie de l'Université CH-1015 Lausanne-Dorigny, Switzerland

SUPPLEMENTARY MATERIAL

2-Acetoxy-2,5-dihydrothiophene-1,1-dioxide (11d). IR (KBr) ν 2975, 2940, 1755, 1325, 1310, 1215, 1130, 1045 cm⁻¹. ¹³C-NMR (100.6 MHz, CDCl₃) δ _C 20.3 (q, 1J(C,H) = 131, C(2')), 54.3 (t, 1J(C,H) = 145, C(5)), 83.4 (d, 1J(C,H) = 167, C(2)), 125.2 (d, 1J(C,H) = 174, C(4)), 130.1 (d, 1J(C,H) = 173, C(3)). CI-MS (NH₃): m/z 194 (0.1, M+NH₃), 166 (1.8), 164 (1.5), 117 (10), 113 (5), 112 (26), 99 (1.5), 89 (3), 83 (1.5), 71 (8), 70 (100). Anal. calc. for C₆H₈O₄S (176.19): C 40.90, H 4.58, S 18.20; found: C 40.88, H 4.61, S 18.22.

Data for **2-Chloro-2,5-dihydrothiophene-1,1-dioxide (11e).** IR (KBr) ν 1330, 1254, 1145, 1120, 795, 735, 645 cm⁻¹; CI-MS (NH₃) m/z 117 (4), 109 (12), 88 (100). Anal. calc. for C₄H₅ClO₂S (152.45): C 31.59, H 3.32; found: C 31.53, H 3.26.

Data for **2-Bromo-2,5-dihydrothiophene-1,1-dioxide (11f).** ¹H-NMR (400 MHz, CDCl₃) δ _H 3.83 (dddd, 2J(H-5,H'-5) = 16.9, 3J(H-4,H-5) = 4J(H-3,H-5) = 2.5, 5J(H-2,H-5) = 1.5, H-C(5)), 3.88 (dddd, 2J = 16.9, 3J(H-4,H'-5) = 3.0, 4J(H-3,H'-5) = 2.3, 5J(H-2,H'-5) = 1.0, H'-C(5)), 5.36 (m, 4J(H-2,H-4) = 1.2, H-C(2)), 6.23 (dddd, 3J(H-3,H-4) = 8.5, 3J(H-4,H-5) = 3.0, 3J(H-4,H-5) = 2.5, 4J(H-2,H-4) = 1.2, H-C(4)), 6.32 (dddd, 3J(H-3,H-4) = 8.5, 3J(H-2,H-3) = 2.3, 4J(H-3,H-5) = 2.5, 4J(H-3,H'-5) = 2.3, H-C(3)); ¹³C-NMR (100.6 MHz, CDCl₃) δ _C 57.7 (t, 1J(C,H) = 145, C(5)), 58.3 (d, 155, C(2)), 126.6 (d, 176, C(4)), 129.5 (d, 176, C(3)). IR (film) ν 2975, 1325, 1250, 1140, 760 cm⁻¹. CI-MS (NH₃) m/z 197 (M⁺•, 1), 134 (90), 132 (100). Anal. calc. for C₄H₅BrO₂S (197.1): C 24.50, H 2.57; found: C 24.55, H 2.66.

[†] Universidad de Oviedo

[‡] Université de Lausanne

Table 1. Crystal data and structure refinement for **11d**

Identification code	Frederic Monnat #1
Empirical formula	C ₆ H ₈ O ₄ S
Formula weight	176.18
Temperature	293 (2) K
Wavelength	0.71073 Å
Crystal system	orthorhombic
Space group	P2 ₁ 2 ₁ 2 ₁
Unit cell dimensions	$a = 7.0306(8)$ Å $\alpha = 90^\circ$ $b = 8.0766(10)$ Å $\beta = 90^\circ$ $c = 14.135 (2)$ Å $\gamma = 90^\circ$
Volume	802.7 (2) Å ³
Z	4
Density (calculated)	1.458 Mg/m ³
Absorption coefficient	0.367 mm ⁻¹
F(000)	368
θ range for data collection	2.88 to 23.29°
Index ranges	-7 ≤ h ≤ 6, -8 ≤ k ≤ 8, -12 ≤ l ≤ 15
Reflections collected	3077
Independent reflections	1141 ($R_{\text{int}} = 0.0250$)
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	1141 / 0 / 134
Goodness-of-fit on F ²	2.044
Final R indices [I>2σ(I)]	R1 = 0.0260, wR2 = 0.0518
R indices (all data)	R1 = 0.0283, wR2 = 0.0526
Absolute structure parameter	-0.04 (8)
Extinction coefficient	0.024 (2)
Largest diff. peak and hole	0.134 and -0.131 eÅ ⁻³

Frederic Monnat #1: 2-Acetoxy-2,5-dihydrothiophene-1,1-dioxide Sulfolene **11d**

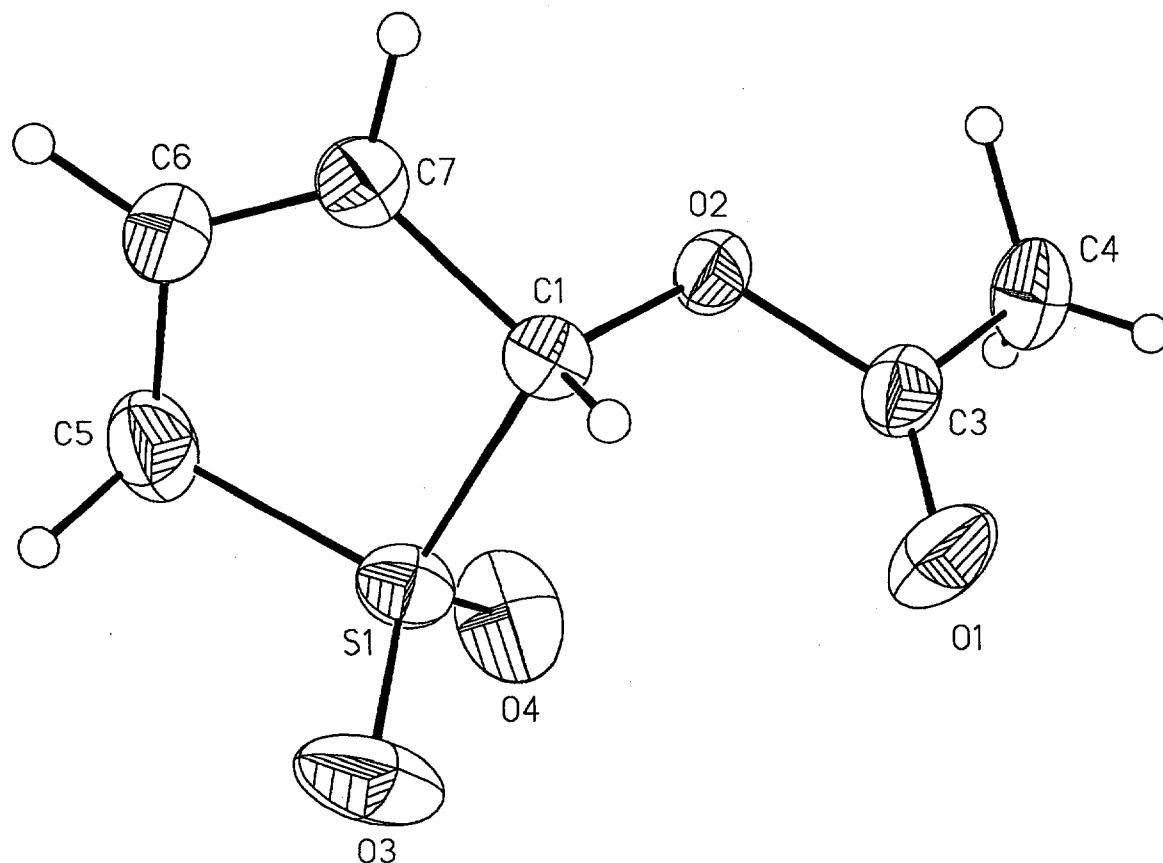


Table 2. Atomic coordinates [$\times 10^4$] and equivalent isotropic displacement parameters [$\text{\AA}^2 \times 10^3$] for **11d**. U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
S(1)	1961(1)	510(1)	4758(1)	50(1)
O(3)	185(2)	-325(2)	4772(2)	89(1)
O(4)	2449(3)	1498(2)	3961(1)	90(1)
C(1)	3848(3)	-990(2)	4984(2)	43(1)
O(1)	3780(3)	-2081(2)	3233(1)	75(1)
O(2)	5403(2)	-737(2)	4360(1)	47(1)
C(3)	5170(4)	-1346(3)	3465(2)	51(1)
C(4)	6860(6)	-908(5)	2870(2)	70(1)
C(5)	2258(5)	1672(4)	5823(2)	73(1)
C(6)	3655(4)	676(4)	6372(2)	70(1)
C(7)	4445(4)	-605(4)	5964(2)	59(1)

Table 3. Bond lengths [\AA] and angles [$^\circ$] for **11d**.

S(1)-O(3)	1.419(2)	S(1)-O(4)	1.423(2)
S(1)-C(5)	1.787(3)	S(1)-C(1)	1.825(2)
C(1)-O(2)	1.419(2)	C(1)-C(7)	1.480(3)
C(1)-H(1)	0.98(2)	O(1)-C(3)	1.189(3)
O(2)-C(3)	1.367(3)	C(3)-C(4)	1.499(4)
C(4)-H(4B)	0.80(3)	C(4)-H(4C)	0.86(3)
C(4)-H(4D)	1.13(3)	C(5)-C(6)	1.488(4)
C(5)-H(5A)	1.02(4)	C(5)-H(5B)	0.86(3)
C(6)-C(7)	1.308(4)	C(6)-H(6)	0.93(3)
C(7)-H(7)	0.91(2)		

O(3)-S(1)-O(4)	119.33(13)	O(3)-S(1)-C(5)	109.9(2)
O(4)-S(1)-C(5)	110.2(2)	O(3)-S(1)-C(1)	108.76(10)
O(4)-S(1)-C(1)	109.65(11)	C(5)-S(1)-C(1)	96.64(13)
O(2)-C(1)-C(7)	109.4(2)	O(2)-C(1)-S(1)	110.83(13)
C(7)-C(1)-S(1)	103.4(2)	O(2)-C(1)-H(1)	113.5(12)
C(7)-C(1)-H(1)	113.6(13)	S(1)-C(1)-H(1)	105.7(12)
C(3)-O(2)-C(1)	115.5(2)	O(1)-C(3)-O(2)	122.2(2)
O(1)-C(3)-C(4)	127.9(3)	O(2)-C(3)-C(4)	109.8(3)
C(3)-C(4)-H(4B)	107(2)	C(3)-C(4)-H(4C)	111(2)
H(4B)-C(4)-H(4C)	100(3)	C(3)-C(4)-H(4D)	107.4(14)
H(4B)-C(4)-H(4D)	124(3)	H(4C)-C(4)-H(4D)	107(3)
C(6)-C(5)-S(1)	103.4(2)	C(6)-C(5)-H(5A)	107(2)
S(1)-C(5)-H(5A)	100(2)	C(6)-C(5)-H(5B)	122(2)
S(1)-C(5)-H(5B)	103(2)	H(5A)-C(5)-H(5B)	118(3)
C(7)-C(6)-C(5)	118.6(3)	C(7)-C(6)-H(6)	126(2)
C(5)-C(6)-H(6)	116(2)	C(6)-C(7)-C(1)	117.3(3)
C(6)-C(7)-H(7)	124(2)	C(1)-C(7)-H(7)	118(2)

Symmetry transformations used to generate equivalent atoms:

Torsional angles [°] for **11d**

Selected torsion angles		
136.34	(0.16)	O3 – S1 – C1 – O2
4.23	(0.18)	O4 – S1 – C1 – O2
-110.00	(0.19)	C5 – S1 – C1 – O2
106.50	(0.17)	O3 – S1 – C1 – C7
121.39	(0.17)	O4 – S1 – C1 – C7
7.16	(0.21)	C5 – S1 – C1 – C7
167.86	(0.21)	C7 – C1 – O2 – C3
-78.78	(0.19)	S1 – C1 – O2 – C3
-2.11	(0.31)	C1 – O2 – C3 – O1
176.66	(0.23)	C1 – O2 – C3 – C4
105.05	(0.22)	O3 – S1 – C5 – C6
-121.48	(0.22)	O4 – S1 – C5 – C6
-7.69	(0.25)	C1 – S1 – C5 – C6
6.60	(0.37)	S1 – C5 – C6 – C7
-1.34	(0.41)	C5 – C6 – C7 – C1
113.55	(0.25)	O2 – C1 – C7 – C6
-4.58	(0.29)	S1 – C1 – C7 – C6

Table 4. Anisotropic displacement parameters [$\text{\AA}^2 \times 10^3$] for **11d**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [(ha^*)^2 U_{11} + \dots + 2hka^*b^*U_{12}]$

	U11	U22	U33	U23	U13	U12
S(1)	46(1)	43(1)	63(1)	6(1)	-4(1)	1(1)
O(3)	43(1)	65(1)	157(2)	-5(1)	-20(1)	-7(1)
O(4)	119(2)	72(1)	78(1)	36(1)	20(1)	29(1)
C(1)	41(1)	38(1)	49(2)	2(1)	0(1)	-1(1)
O(1)	81(2)	82(1)	61(1)	-18(1)	-14(1)	-22(1)
O(2)	45(1)	54(1)	43(1)	-12(1)	0(1)	-5(1)
C(3)	64(2)	44(2)	45(2)	-5(1)	-3(1)	7(1)
C(4)	83(2)	74(2)	52(2)	-12(2)	14(2)	0(2)
C(5)	70(2)	68(2)	80(2)	-16(2)	12(2)	10(2)
C(6)	61(2)	95(5)	52(2)	-17(2)	1(1)	2(2)
C(7)	52(2)	79(2)	47(2)	0(2)	0(1)	12(2)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **11d**.

	x	y	z	U(eq)
H(1)	3256(28)	-2075(25)	4918(15)	57(6)
H(4B)	6876(46)	-1538(41)	2432(24)	105(13)
H(4C)	6677(45)	7(41)	2569(24)	102(13)
H(4D)	8098(43)	-698(36)	3365(21)	103(10)
H(5A)	954(51)	1479(37)	6123(23)	126(13)
H(5B)	2512(36)	2647(38)	5617(21)	95(12)
H(6)	3839(40)	1003(36)	6996(20)	98(10)
H(7)	5425(35)	-1198(29)	6216(17)	62(8)

Table 6. Crystal data and structure refinement for **11e**

Identification code	Elena Roversi #2
Empirical formula	C ₄ H ₅ ClO ₂ S
Formula weight	152.59
Temperature	293 (2) K
Wavelength	0.71073 Å
Crystal system	monoclinic
Space group	P2 ₁ /c
Unit cell dimensions	$a = 9.839(2)$ Å $\alpha = 90^\circ$ $b = 9.493(2)$ Å $\beta = 102.76(3)^\circ$ $c = 13.551(3)$ Å $\gamma = 90^\circ$
Volume	1234.4 (4) Å ³
Z	8
Density (calculated)	1.642 Mg/m ³
Absorption coefficient	0.858 mm ⁻¹
F(000)	624
θ range for data collection	2.12 to 23.90°
Index ranges	-11 ≤ <i>h</i> ≤ 11, -10 ≤ <i>k</i> ≤ 10, -15 ≤ <i>l</i> ≤ 10
Reflections collected	4576
Independent reflections	1789 ($R_{\text{int}} = 0.0502$)
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	1789 / 0 / 186
Goodness-of-fit on F ²	1.675
Final R indices [I>2σ(I)]	R1 = 0.0567, wR2 = 0.1882
R indices (all data)	R1 = 0.0592, wR2 = 0.1917
Extinction coefficient	0.00 (3)
Largest diff. peak and hole	0.795 and -0.440 e Å ⁻³

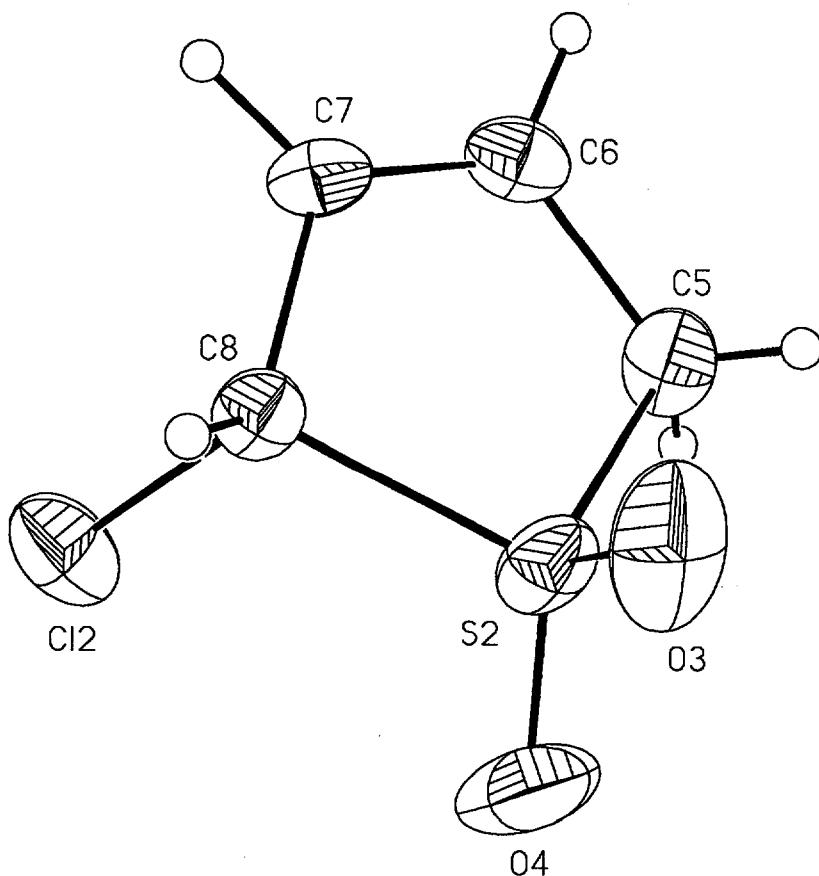
Elena Roversi #2: 2-Chloro-2,5-dihydrothiophene-1, -dioxide 2-Chlorosulfolene **11e**

Table 7. Atomic coordinates [$\times 10^4$] and equivalent isotropic displacement parameters [$\text{\AA}^2 \times 10^3$] for **11e**. U(eq) is defined as one third of the trace of orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
S(1)	3855(1)	2165(1)	9229(1)	48(1)
S(2)	1111(1)	1980(1)	6216(1)	55(1)
Cl(1)	5723(1)	855(1)	8058(1)	71(1)
Cl(2)	-1024(1)	1452(1)	7424(1)	74(1)
O(1)	2793(3)	3233(4)	9021(2)	71(1)
O(2)	3436(3)	740(3)	9260(3)	75(1)
O(3)	2354(3)	2819(4)	6391(3)	87(1)
O(4)	1237(4)	486(4)	6339(3)	96(1)
C(1)	4981(3)	2471(3)	8346(2)	45(1)
C(2)	6047(3)	3450(3)	8927(3)	51(1)
C(3)	6113(4)	3532(3)	9901(3)	54(1)
C(4)	5130(4)	2656(4)	10325(3)	55(1)
C(5)	26(4)	2457(5)	5028(3)	58(1)
C(6)	-890(3)	3566(3)	5312(3)	51(1)
C(7)	-891(3)	3715(3)	6265(3)	49(1)
C(8)	2(3)	2747(3)	6981(2)	44(1)

Torsional angles [°] for **11e**

Selected torsion angles		
-138.39	(0.23)	O2 – S1 – C1 – C2
90.08	(0.24)	O1 – S1 – C1 – C2
-21.43	(0.24)	C4 – S1 – C1 – C2
-18.93	(0.24)	O2 – S1 – C1 – C11
-150.47	(0.18)	O1 – S1 – C1 – C11
98.02	(0.20)	C4 – S1 – C1 – C11
102.03	(0.33)	C11 – C1 – C2 – C3
15.92	(0.35)	S1 – C1 – C2 – C3
-0.14	(0.47)	C1 – C2 – C3 – C4
-16.16	(0.40)	C2 – C3 – C4 – S1
138.96	(0.25)	O2 – S1 – C4 – C3
-87.77	(0.28)	O1 – S1 – C4 – C3
21.32	(0.27)	C1 – S1 – C4 – C3
134.06	(0.30)	O4 – S2 – C5 – C6
-92.17	(0.28)	O3 – S2 – C5 – C6
17.24	(0.29)	C8 – S2 – C5 – C6
-12.49	(0.42)	S2 – C5 – C6 – C7
-1.16	(0.47)	C5 – C6 – C7 – C8
-105.69	(0.31)	C6 – C7 – C8 – C12
13.90	(0.35)	C6 – C7 – C8 – S2
-134.62	(0.24)	O4 – S2 – C8 – C7
93.86	(0.25)	O3 – S2 – C8 – C7
-17.90	(0.26)	C5 – S2 – C8 – C7
-15.93	(0.25)	O4 – S2 – C8 – C12
-147.45	(0.21)	O3 – S2 – C8 – C12
100.79	(0.22)	C5 – S2 – C8 – C12

Table 8. Bond lengths [\AA] and angles [$^\circ$] for 11e.

S(1)-O(2)	1.418(3)	S(1)-O(1)	1.439(3)
S(1)-C(4)	1.781(3)	S(1)-C(1)	1.825(3)
S(2)-O(4)	1.429(4)	S(2)-O(3)	1.435(3)
S(2)-C(5)	1.782(4)	S(2)-C(8)	1.815(3)
Cl(1)-C(1)	1.779(3)	Cl(2)-C(8)	1.777(3)
C(1)-C(2)	1.489(4)	C(1)-H(1)	0.93(4)
C(2)-C(3)	1.310(6)	C(2)-H(2)	0.94(5)
C(3)-C(4)	1.484(6)	C(3)-H(3)	0.97(4)
C(4)-H(4A)	0.86(5)	C(4)-H(4B)	0.94(5)
C(5)-C(6)	1.491(6)	C(5)-H(5A)	0.98(4)
C(5)-H(5B)	0.80(6)	C(6)-C(7)	1.299(5)
C(6)-H(6)	0.93(4)	C(7)-C(8)	1.477(4)
C(7)-H(7)	0.94(4)	C(8)-H(8)	0.97(4)

O(2)-S(1)-O(1)	118.4(2)	O(2)-S(1)-C(4)	112.4(2)
O(1)-S(1)-C(4)	109.0(2)	O(2)-S(1)-C(1)	113.2(2)
O(1)-S(1)-C(1)	106.2(2)	C(4)-S(1)-C(1)	95.0(2)
O(4)-S(2)-O(3)	118.9(2)	O(4)-S(2)-C(5)	112.1(2)
O(3)-S(2)-C(5)	109.1(2)	O(4)-S(2)-C(8)	112.3(2)
O(3)-S(2)-C(8)	106.3(2)	C(5)-S(2)-C(8)	95.7(2)
C(2)-C(1)-Cl(1)	112.3(2)	C(2)-C(1)-S(1)	102.2(2)
Cl(1)-C(1)-S(1)	110.1(2)	C(2)-C(1)-H(1)	119(2)
Cl(1)-C(1)-H(1)	105(3)	S(1)-C(1)-H(1)	109(3)
C(3)-C(2)-C(1)	116.3(3)	C(3)-C(2)-H(2)	126(3)
C(1)-C(2)-H(2)	118(3)	C(2)-C(3)-C(4)	117.7(3)
C(2)-C(3)-H(3)	125(2)	C(4)-C(3)-H(3)	117(2)
C(3)-C(4)-S(1)	102.9(3)	C(3)-C(4)-H(4A)	115(3)
S(1)-C(4)-H(4A)	105(3)	C(3)-C(4)-H(4B)	114(3)
S(1)-C(4)-H(4B)	105(3)	H(4A)-C(4)-H(4B)	114(4)
C(6)-C(5)-S(2)	102.8(3)	C(6)-C(5)-H(5A)	121(2)
S(2)-C(5)-H(5A)	105(3)	C(6)-C(5)-H(5B)	119(4)
S(2)-C(5)-H(5B)	108(4)	H(5A)-C(5)-H(5B)	100(5)
C(7)-C(6)-C(5)	117.9(3)	C(7)-C(6)-H(6)	126(2)
C(5)-C(6)-H(6)	116(2)	C(6)-C(7)-C(8)	117.0(3)
C(6)-C(7)-H(7)	124(2)	C(8)-C(7)-H(7)	119(2)
C(7)-C(8)-Cl(2)	110.6(2)	C(7)-C(8)-S(2)	102.7(2)
Cl(2)-C(8)-S(2)	111.9(2)	C(7)-C(8)-H(8)	121(2)
Cl(2)-C(8)-H(8)	106(2)	S(2)-C(8)-H(8)	106(2)

Symmetry transformations used to generate equivalent atoms

Table 9. Anisotropic displacement parameters [$\text{\AA}^2 \times 10^3$] for **11e**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [(\text{ha}^*)^2 \text{U}_{11} + \dots + 2\text{hka}^* \text{b}^* \text{U}^{12}]$

	U11	U22	U33	U23	U13	U12
S(1)	34(1)	53(1)	53(1)	-2(1)	5(1)	-10(1)
S(2)	37(1)	74(1)	49(1)	-6(1)	1(1)	16(1)
Cl(1)	66(1)	69(1)	77(1)	-24(1)	17(1)	2(1)
Cl(2)	70(1)	81(1)	68(1)	23(1)	13(1)	-11(1)
O(1)	40(1)	95(2)	74(2)	-7(2)	5(1)	13(1)
O(2)	68(2)	68(2)	87(2)	2(1)	15(1)	-34(1)
O(3)	35(1)	156(3)	68(2)	-12(2)	5(1)	-16(2)
O(4)	106(2)	84(2)	90(2)	-9(2)	3(2)	56(2)
C(1)	39(2)	49(2)	43(2)	0(1)	2(1)	-2(1)
C(2)	37(2)	40(2)	73(2)	7(1)	4(1)	-12(1)
C(3)	43(2)	40(2)	67(2)	-8(1)	-12(2)	-3(1)
C(4)	52(2)	59(2)	47(2)	-4(2)	-3(2)	1(2)
C(5)	45(2)	84(3)	43(2)	-4(2)	1(1)	4(2)
C(6)	42(2)	50(2)	54(2)	8(2)	-4(1)	-1(1)
C(7)	45(2)	40(2)	56(2)	-1(1)	2(1)	9(1)
C(8)	36(2)	50(2)	42(2)	-2(1)	3(1)	3(1)

Table 10. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **11e**.

	x	y	z	Ueq
H(1)	4432(42)	2776(40)	7735(34)	55(10)
H(2)	6636(44)	3914(43)	8576(33)	62(11)
H(3)	6755(39)	4122(37)	10372(30)	47(9)
H(4A)	5466(50)	1871(50)	10597(39)	69(13)
H(4B)	4630(46)	3166(43)	10722(37)	62(11)
H(5A)	-365(43)	1566(45)	4725(34)	59(11)
H(5B)	520(59)	2657(57)	4655(47)	95(18)
H(6)	-1404(43)	4102(41)	4782(33)	54(10)
H(7)	-1494(39)	4324(37)	6512(30)	55(10)
H(8)	632(40)	3109(37)	7580(32)	49(9)

Table 11. Total Energies (au) of the Reactants, Transition Structures and Products for the Cheletropic (CHE) and hetero Diels-Alder reactions (DA) between 1-X-butadiene (X=H, Cl, CH₃, OMe, OAc) and Sulfur Dioxide.

Structures		HF/6-31G*	ZPVE	MP2/6-31G*	MP2/6-31G*	MP2/6-311+G(3df,2p)	QCISD(T)/6-31G*//
					SCRF ($\epsilon=13.3$)	//MP2/6-31G*	//MP2/6-31G*
1,3-butadiene		-154.919654	0.091539	-155.422656	-155.424136	-155.619309	-155.487541
(Z)-piperilene		-193.959159	0.121669	-194.593852	-194.595540	-194.843079	-194.674055
SO ₂		-547.169006	0.008020	-547.682480	-547.685632	-547.950983	-547.699245
butadiene + SO ₂ (DA)	TSIa	-702.037148	0.103228	-703.076700	-703.082388	-703.547272	-703.153935
butadiene + SO ₂ (CHE)	TSIIa	-702.027171	0.102988	-703.073853	-703.080235	-703.551602	-703.146920
butadiene + SO ₂ (DA)	9a	-702.108831	0.107177	-703.111190	-703.117686	-703.575441	-703.196030
butadiene + SO ₂ (DA)	9'a	-702.103306	0.106912	-703.105771		-703.571318	-703.190287
butadiene + SO ₂ (CHE)	11a	-702.105565	0.107092	-703.113680	-703.123333	-703.595060	-703.192270
(Z)-piperilene + SO ₂ (DA)	TSIb	-741.083696	0.132951	-742.252177	-742.257810	-742.775931	-742.345217
(Z)-piperilene + SO ₂ (DA)	TS'Ib	-741.071081	0.133181	-742.243283	-742.248463	-742.767390	-742.336117
(Z)-piperilene + SO ₂ (CHE)	TSIIIb	-741.067432	0.132924	-742.246792	-742.253159	-742.778113	-742.334992

Table 11 (cont.)

Structures		HF/6-31G*	ZPVE	MP2/6-31G*	MP2/6-31G*	MP2/6-311+G(3df,2p)	QCISD(T)/6-31G*//	
							SCRF ($\epsilon=13.3$)	//MP2/6-31G*
(E)-piperilene + SO ₂ (DA)	9b	-741.146654	0.137032	-742.283348	-742.289511	-742.800185		-742.383670
(E)-piperilene + SO ₂ (DA)	9'b	-741.143826	0.136729	-742.280154				
(E)-piperilene + SO ₂ (DA)	10b	-741.148898	0.136952	-742.284983	-742.291151	-742.802263		-742.385214
(E)-piperilene + SO ₂ (DA)	10'b	-741.141756	0.136714	-742.278334				
(E)-piperilene + SO ₂ (CHE)	11b	-741.142925	0.137287	-742.285645	-742.294684	-742.819865		-742.379659
TS(9b→9'b)		-741.135641	0.136269	-742.272844				
TS(10b→10'b)		-741.136465	0.136114	-742.273668				
TS(9b→10'b)		-741.021384	0.134677	-742.165126				
TS(10b→9'b)		-741.022853	0.134677	-742.165942				

Table 11 (cont.)

Structures		HF/6-31G*	ZPVE	MP2/6-31G*	MP2/6-31G*	MP2/6-311+G(3df,2p)
					SCRF ($\epsilon=13.3$)	//MP2/6-31G*
(E)-1-MeO-butadiene		-268.799678	0.127257	-269.609280	-269.612984	-269.946438
(E)-1-MeO-butadiene + SO ₂ (DA)	TSIc	-815.937736	0.139461	-817.276165	-817.282717	-817.887172
(E)-1-MeO-butadiene + SO ₂ (DA)	TS'Ic	-815.917280	0.139194	-817.258906	-817.266241	-817.870714
(E)-1-MeO-butadiene + SO ₂ (CHE)	TSIIc	-815.930522	0.139020	-817.270296	-817.227605	-817.881263
(E)-1-MeO-butadiene + SO ₂ (DA)	9c	-815.992569	0.142537	-817.303450	-817.311821	-817.906200
(E)-1-MeO-butadiene + SO ₂ (DA)	9'c	-815.989472	0.142402	-817.2988901		-817.904149
(E)-1-MeO-butadiene + SO ₂ (DA)	10c	-815.994934	0.142674	-817.303315	-817.309808	-817.907607
(E)-1-MeO-butadiene + SO ₂ (DA)	10'c	-815.992298	0.142472	-817.302757		-817.906265
(E)-1-MeO-butadiene + SO ₂ (CHE)	11c	-815.974682	0.142931	-817.293392	-817.304187	-817.914723

Table 11 (cont.)

Structures		HF/6-31G*	ZPVE	MP2/6-31G*	MP2/6-31G*	MP2/6-311+G(3df,2p)
					SCRF ($\epsilon=13.3$)	//MP2/6-31G*
(E)-1-AcO-butadiene		-381.568290	0.137903	-382.675314	-382.679296	-383.127665
(E)-1-AcO-butadiene + SO ₂ (DA)	TSIId	-928.695804	0.149090	-930.335517	-930.342704	-931.062623
(E)-1-AcO-butadiene + SO ₂ (DA)	TS'IId	-928.685584	0.149020	-930.328481	-930.337362	-931.055295
(E)-1-AcO-butadiene + SO ₂ (CHE)	TSIIId	-928.675386	0.149023	-930.326480	-930.335064	-931.060843
(E)-1-AcO-butadiene + SO ₂ (DA)	9d	-928.850843	0.152910	-930.360870	-930.367900	-931.080233
(E)-1-AcO-butadiene + SO ₂ (DA)	9'd	-928.753821	0.152770	-930.361539		
(E)-1-AcO-butadiene + SO ₂ (DA)	10d	-928.758827	0.152938	-930.365712	-930.374180	-931.085975
(E)-1-AcO-butadiene + SO ₂ (DA)	10'd	-928.756517	0.152751	-930.365503		
(E)-1-AcO-butadiene + SO ₂ (DA)	9d*	-928.737503	0.152687	-930.352126		
(E)-1-AcO-butadiene + SO ₂ (DA)	9'd*	-928.741871	0.152758	-930.351801		
(E)-1-AcO-butadiene + SO ₂ (DA)	10d*	-928.748451	0.153062	-930.356904		
(E)-1-AcO-butadiene + SO ₂ (DA)	10'd*	-928.745223	0.152847	-930.356026		
(E)-1-AcO-butadiene + SO ₂ (CHE)	11d	-928.743860	0.153072	-930.359071	-930.369021	-931.096140

Table 11 (cont.)

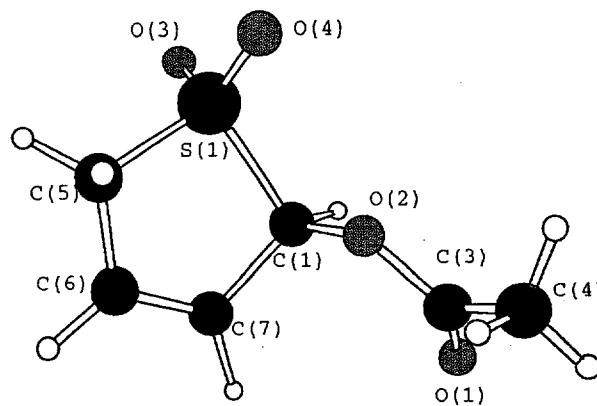
Structures		HF/6-31G*	ZPVE	MP2/6-31G*	MP2/6-31G*	MP2/6-311+G(3df,2p)	QCISD(T)/6-31G**//
				SCRF ($\epsilon=13.3$)	//MP2/6-31G*	//MP2/6-31G*	
(E)-1-Cl-butadiene		-613.820811	0.082350	-614.454433	-614.456792	-614.730985	-614.529749
(E)-1-Cl-butadiene + SO ₂ (DA)	TSIe	-1160.934280	0.093621	-1162.107729	-1162.113904	-1162.659525	-1162.195156
(E)-1-Cl-butadiene + SO ₂ (DA)	TS'Ie	-1160.922165	0.093407	-1162.099459	-1162.104896	-1162.651862	-1162.186717
(E)-1-Cl-butadiene + SO ₂ (CHE)	TSIIe	-1160.922533	0.093434	-1162.103284	-1162.109534	-1162.662819	-1162.185984
(E)-1-Cl-butadiene + SO ₂ (DA)	9e	-1161.003778	0.097221	-1162.140735	-1162.149463	-1162.684907	-1162.235782
(E)-1-Cl-butadiene + SO ₂ (DA)	9'e	-1160.999898	0.097008	-1162.135516			
(E)-1-Cl-butadiene + SO ₂ (DA)	10e	-1161.006069	0.097231	-1162.140419	-1162.147202	-1162.685650	-1162.235829
(E)-1-Cl-butadiene + SO ₂ (DA)	10'e	-1161.003721	0.097045	-1162.140139			
(E)-1-Cl-butadiene + SO ₂ (CHE)	11e	-1160.994884	0.097542	-1162.137612	-1162.147223	-1162.701185	-1162.226356

Table 12. ΔE [QCISD(T)/6-31G*]- ΔE [MP2/6-31G*] (δ) Calculated Energies (kcal/mol) of the MP2/6-31G* Optimized Transition Structures and Products Relative to Reactants for the Diels-Alder (DA) and Cheletropic (CHE) Reactions of (*E*)-1-X-butadiene (X=H, CH₃ and Cl) and Sulfur Dioxide.

Reactions	TS	δ	Products	δ
butadiene + SO ₂ (Ia) DA	TSIa	2.8	9a	-2.0
(<i>E</i>)-piperilene + SO ₂ (Ib) DA	TSIb	2.5	9b	-2.1
(<i>E</i>)-piperilene + SO ₂ (I'b) DA	TS'Ib	2.6	10b	-2.0
(<i>E</i>)-Cl-butadiene + SO ₂ (Ie) DA	TSIe	2.9	9e	-1.8
(<i>E</i>)-Cl-butadiene + SO ₂ (I'e) DA	TS'Ie	3.0	10e	-2.0
DA Averaged Value		2.8		-2.0
butadiene + SO ₂ (IIa) CHE	TSIIa	5.4	11a	2.0
(<i>E</i>)-piperilene + SO ₂ (IIb) CHE	TSIIb	5.5	11b	1.8
(<i>E</i>)-1-Cl-butadiene + SO ₂ (IIe) CHE	TSIIe	5.9	11e	2.0
CHE Averaged Value		5.6		1.9

Table 13. Atomic Coordinates [Å] of the MP2/6-31G* optimized structure for **11d**.

	X	Y	Z
S(1)	1.4092	-0.6245	0.1421
O(1)	-2.6633	0.7374	1.0055
O(2)	-1.1556	-0.2679	-0.2699
O(3)	1.1522	-1.8447	-0.5588
O(4)	2.1551	-0.6347	1.3690
C(1)	-0.1192	0.2981	0.4886
C(2)	-2.4098	0.0016	0.1108
C(3)	-3.4026	-0.7378	-0.7376
C(4)	2.1229	0.6116	-0.9440
C(5)	1.3178	1.8491	-0.6486
C(6)	0.2167	1.6997	0.0636
H	-0.3231	0.2039	1.5434
H	-0.4396	2.5027	0.3393
H	1.6561	2.8007	-1.0148
H	3.1700	0.7104	-0.6892
H	2.0282	0.2815	-1.9707
H	-3.2699	-0.4626	-1.7772
H	-3.2268	-1.8036	-0.6546
H	-4.4034	-0.4966	-0.4129

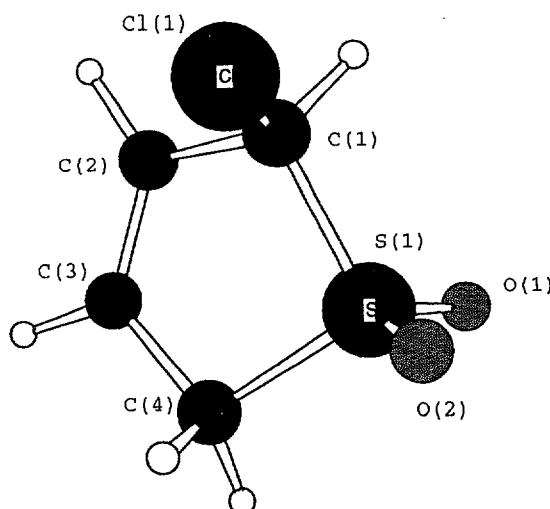


Bond lengths [Å] and angles [°] for **11d**.

S(1)-O(3)	1.469	S(1)-O(4)	1.465
S(1)-C(5)	1.811	S(1)-C(1)	1.832
C(1)-O(2)	1.435	C(1)-C(7)	1.493
C(1)-H(1)	1.094	O(1)-C(3)	1.217
O(2)-C(3)	1.371	C(3)-C(4)	1.501
C(4)-H(4B)	1.084	C(4)-H(4C)	1.092
C(4)-H(4D)	1.092	C(5)-C(6)	1.500
C(5)-H(5A)	1.095	C(5)-H(5B)	1.094
C(6)-C(7)	1.345	C(6)-H(6)	1.086
C(7)-H(7)	1.085		
O(3)-S(1)-O(4)	121.7	O(3)-S(1)-C(5)	108.3
O(4)-S(1)-C(5)	111.3	O(3)-S(1)-C(1)	105.7
O(4)-S(1)-C(1)	111.8	C(5)-S(1)-C(1)	94.4
O(2)-C(1)-S(1)	112.3	O(2)-C(1)-S(1)	106.8
C(7)-C(1)-S(1)	103.4	O(1)-C(3)-O(2)	123.0
C(3)-O(2)-C(1)	113.7	O(2)-C(3)-C(4)	110.1
O(1)-C(3)-C(4)	127.0	C(6)-C(7)-C(5)	115.7
C(6)-C(5)-S(1)	103.3		
C(7)-C(6)-C(5)	117.0		

Table 14. Atomic Coordinates [Å] of the MP2/6-31G* optimized structure for 11e.

	X	Y	Z
S(1)	0.8329	-0.5622	0.0578
O(1)	1.7541	-0.5592	1.2017
O(2)	0.5990	-1.7666	-0.7404
C(1)	-0.7304	0.1771	0.6993
C(2)	-0.5123	1.6387	0.4945
C(3)	0.4628	1.9777	-0.3669
C(4)	1.2244	0.8530	-0.9999
Cl(1)	-2.1645	-0.4197	-0.1613
H	-0.8084	-0.1251	1.7462
H	-1.1335	2.3588	1.0176
H	0.7117	3.0088	-0.5988
H	2.3098	0.9886	-0.9787
H	0.9006	0.6264	-2.0211



Bond lengths [Å] and angles [°] for 11e.

S(1)-O(2)	1.464	S(1)-O(1)	1.469
S(1)-C(4)	1.810	S(1)-C(1)	1.844
Cl(1)-C(1)	1.776	C(1)-H(1)	1.092
C(1)-C(2)	1.492	C(2)-H(2)	1.085
C(2)-C(3)	1.344	C(3)-H(3)	1.086
C(3)-C(4)	1.499	C(4)-H(4A)	1.095
		C(4)-H(4A)	1.094
<hr/>			
O(2)-S(1)-O(1)	121.8	O(2)-S(1)-C(4)	111.1
O(1)-S(1)-C(4)	108.5	O(2)-S(1)-C(1)	112.6
O(1)-S(1)-C(1)	105.1	C(4)-S(1)-C(1)	94.2
C(2)-C(1)-Cl(1)	112.4	C(2)-C(1)-S(1)	102.8
Cl(1)-C(1)-S(1)	112.4	C(2)-C(3)-C(4)	116.7
C(3)-C(2)-C(1)	116.2		
C(3)-C(4)-S(1)	103.3		