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J. Org. Chem., 1998, 63(12), 3802-3803, DOI:10.1021/j0980406i

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## ACS Publications

Cooperative Control of Ion and Molecular Recognition by Molecular Assembling

Tatsuya Nabeshima et al.

## Supporting Information

5: A mixture of the tosylate $4(6.887 \mathrm{~g}, 14.57 \mathrm{mmol})$ and thiourea $(1.221 \mathrm{~g}, 16.04 \mathrm{mmol})$ in 36 mL of $95 \% \mathrm{EtOH}$ was heated at $90^{\circ} \mathrm{C}$ for 20 h. After the solvent was removed and dried in vacuo, the residue was mixed with a solution of $\mathrm{KOH}(6.140 \mathrm{~g}, 93.01 \mathrm{mmol})$ in 35 mL of $\mathrm{H}_{2} \mathrm{O}$. The mixture was refluxed for 24 h under $\mathrm{N}_{2}$. After cooling, the reaction mixture was acidified with 20 mL of $12 \mathrm{~N} \mathrm{H}_{2} \mathrm{SO}_{4}$ and extracted with $\mathrm{CHCl}_{3}$ ( $50 \mathrm{~mL} \times 3$ ). The organic layer was dried over anhydrous MgSO 4 and then concentrated in vacuo. The crude product thus obtained was purified by silica gel column chromatography using $\mathrm{CHCl}_{3} / \mathrm{AcOEt}$ (20:1) as an eluent to afford $5(3.611 \mathrm{~g}, 74 \%)$ as a colorless oil. ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta(\mathrm{ppm}) 14.15,22.70,24.28,26.10,29.36$, $29.51,29.63,29.68,31.93,70.07,70.25,70.55,70.70,71.58,72.89$.

6: To a suspension of $\mathrm{NaH}(0.104 \mathrm{~g}, 2.383 \mathrm{mmol})$ in 2 mL of THF was added a solution of $5(0.761 \mathrm{~g}, 2.275 \mathrm{mmol})$ in 2 mL of THF. After the mixture was stirred for a few minutes, ethyl 4-bromobutyrate $(0.400 \mathrm{~g}$, 2.05 mmol ) in 2.5 mL of THF was added to the reaction mixture, which was stirred for 5 h at room temperature and then the solvent was removed under reduced pressure. The residue was mixed with 30 mL of $\mathrm{H}_{2} \mathrm{O}$, extracted with $\mathrm{CHCl}_{3}$ ( $30 \mathrm{~mL} \times 5$ ), and dried over anhydrous MgSO , and then concentrated in vacuo. The crude product was purified by silica gel column chromatography using $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{Et}_{2} \mathrm{O}(20: 1)$ as an eluent to afford $6(0.775 \mathrm{~g}, 84 \%)$ as a pale yellow oil. ${ }^{13} \mathrm{C}$ NMR (100
$\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}) 14.15,14.24,22.70,24.81,26.10,29.37,29.51$, 29.63, 29.65, 29.68, 31.20, 31.73, 31.93, 33.01, 60.40, 70.06, 70.35, 70.59, 70.68, 70.93, 71.58, 173.11.

7: A solution of $6(1.000 \mathrm{~g}, 2.229 \mathrm{mmol})$ in 11 mL of EtOH was treated with a solution of $\mathrm{NaOH}(0.462 \mathrm{~g}, 11.55 \mathrm{mmol})$ in 2 mL of $\mathrm{H}_{2} \mathrm{O}$ at $80 \sim 90^{\circ} \mathrm{C}$ for 2 h , and then the solvent was removed in vacuo. The residue was mixed with 30 mL of $\mathrm{H}_{2} \mathrm{O}$ and 5 mL of 3 N HCl , extracted with $\mathrm{CHCl}_{3}$ ( $30 \mathrm{~mL} \times 3$ ), and dried over anhydrous MgSO 4 , and then concentrated in vacuo. The crude product was used without purification, because ${ }^{1} \mathrm{H}$ NMR spectrum of the crude 7 indicates exclusive formation of 7. ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta(\mathrm{ppm}) 14.15,22.70,24.58,26.06$, 29.37, 29.49, 29.52, 29.62, 29.65, 29.68, 31.12, 31.53, 31.93, 32.64, $70.02,70.35,70.58,70.61,71.19,71.59,177.94$.

8: To a solution of $7(1.00 \mathrm{~g}, 2.38 \mathrm{mmol})$ in 40 mL of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and 0.4 mL of DMF was added oxalyl chloride ( $0.42 \mathrm{~mL}, 4.8 \mathrm{mmol}$ ) in one portion. The reaction mixture was heated at $50^{\circ} \mathrm{C}$ for 4 h . After evaporation of the solvent in vacuo, formation of $\mathbf{8}$ was ascertained by ${ }^{1} \mathrm{H}$ NMR ( $\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}) 3.69-3.55(\mathrm{~m}, 10 \mathrm{H}), 3.42(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=$ $6.80 \mathrm{~Hz}), 3.02(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=7.08 \mathrm{~Hz}), 2.67(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=8.48 \mathrm{~Hz}), 2.60(\mathrm{t}, 2 \mathrm{H}$, $\mathrm{J}=6.96 \mathrm{~Hz}), 1.96(\mathrm{~m}, 2 \mathrm{H}), 1.50(\mathrm{~m}, 2 \mathrm{H}), 1.26(\mathrm{~s}, 18 \mathrm{H}), 0.84(\mathrm{t}, 3 \mathrm{H})) . \mathrm{A}$ solution of the crude 8 (ca. $90 \%$ purity by ${ }^{1} \mathrm{H}$ NMR) solution in 15 mL of THF was added slowly to a mixture of 2,6-diaminopyridine ( 1.307 g , $11.98 \mathrm{mmol})$ and $\mathrm{Et} 3 \mathrm{~N}(0.65 \mathrm{~mL}, 4.689 \mathrm{mmol})$, and then the mixture was stirred for 3 h at room temperature. The solvent was removed in vacuo and the residue was mixed with 30 mL of $\mathrm{H}_{2} \mathrm{O}$ and extracted with $\mathrm{CHCl}_{3}$ ( $30 \mathrm{~mL} \times 3$ ). The combined organic layers were dried over anhydrous MgSO 4 and concentrated in vacuo. The crude product was purified by
silica gel column chromatography using $\mathrm{CHCl}_{3} / \mathrm{MeOH}(15: 1)$ as an eluent to give $8(1.01 \mathrm{~g}, 83 \%)$ as a pale yellow oil. ${ }^{13} \mathrm{C}$ NMR $(100 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) $\delta$ (ppm) 14.14, 22.69, 24.85, 26.07, 29.35, 29.20, 29.62, 29.67, $31.23,31.72,31.92,35.93,70.03,70.31,70.56,70.63,70.91,71.56$, $103.19,104.23,140.11,149.71,157.08,170.67$.

1: Amide 8 ( $2.40 \mathrm{~g}, 4.69 \mathrm{mmol}$ ) and $\mathrm{Et} 3 \mathrm{~N}(0.97 \mathrm{~mL}, 7.0 \mathrm{mmol})$ were dissolved in THF ( 40 mL ). Glutaryl chloride ( $0.40 \mathrm{~mL}, 3.1 \mathrm{mmol}$ ) was added dropwise to the mixture and it was stirred for 19 h at room temperature. After evaporation of the solvent, the residue was mixed with $\mathrm{H}_{2} \mathrm{O}(40 \mathrm{~mL})$ and extracted with $\mathrm{CHCl}_{3}(40 \mathrm{~mL} \times 3)$. The organic layers were combined and dried over anhydrous MgSO 4 . After concentration in vacuo, chromatography on $\mathrm{SiO}_{2}\left(\mathrm{CHCl}_{3}\right.$-acetone $\left.=3: 1\right)$ gave 1 ( $1.98 \mathrm{~g}, 76 \%$ ). M.p. $99.2-102.4^{\circ} \mathrm{C}$ (acetone); ${ }^{13} \mathrm{C}$ NMR (100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}) 14.15,20.98,22.69,24.69,26.08,29.36,29.50$, 29.62, 29.64, 29.67, 31.26, 31.72, 31.92, 35.78, 36.05, 70.01, 70.30, $70.56,70.62,70.86,71.57,109.43,109.56,140.68,149.35,149.56$, 171.07, 171.20; IR (KBr): $\mathrm{n}=3306,1671 \mathrm{~cm}^{-1}$; FABMS: $\mathrm{m} / \mathrm{z}=1119$ $\left([\mathrm{M}]^{+}\right)$; elemental analysis for $\mathrm{C} 59 \mathrm{H}_{102} \mathrm{~N}_{6} \mathrm{O} 10 \mathrm{~S}_{2}$ : calcd C 63.29, H 9.18, N 7.51; found C 63.00, H 9.30, N 7.42.










Isotope Patterns of Observed and Calculated ESIMS Spectra for a Mixture of $\mathbf{1 , 2}$ and $\mathrm{NaNO}_{3}$


