

**Supporting Information of the Manuscript: “Highly Diastereoselective Oxidation
of 2-amino-2-deoxy-1-thio- β -D-Glucopyranosides: Synthesis of Imino
Sulfinylglycosides”**

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- X-Ray data of compound **15S_s**
- Copies of ¹H (500 MHZ) and the ¹³C NMR (125 MHz) spectra of compounds:
 - **3** (¹H and the ¹³C NMR).
 - **8** (¹H NMR)
 - **9R/S** (¹H and the ¹³C NMR)
 - **13** (¹H NMR)
 - **14R/S** (¹H NMR)
 - **17S_s** (¹H NMR)
 - **9S_s** (¹H and the ¹³C NMR)
 - **14S_s** (¹H and the ¹³C NMR)
 - **22** (¹H NMR)
 - **24** (¹H NMR)

Crystal data for $\mathbf{15S}_\mathbf{S}$: $C_{22}H_{21}Cl_4NO_{10}S$, 0.59(H₂O), $M_r = 643.79$, colourless needle (0.68 × 0.11 × 0.09 mm) from hexanes /ethyl acetate, monoclinic, space group $P21$ (no. 4), $a = 6.169(10)$ Å, $b = 11.94(2)$ Å, $c = 19.10(3)$ Å, $\beta = 93.59(4)^\circ$, $V = 1404(4)$ Å³, $Z = 2$, $D_x = 1.523$ Mg/m³, $\lambda(\text{Mo K}\alpha) = 0.71073$ Å, $\mu = 0.552$ mm⁻¹, $T = 100(2)$ K.

X-ray diffraction data were collected on a Bruker SMART diffractometer equipped with a APEX CCD detector, graphite monochromated Mo K α radiation, and a Kryoflex low-temperature device.¹ Four sets of frames covering a hemisphere of the reciprocal space were recorded (606, 435, 230 y 100 frames to $\varphi = 0^\circ, 90^\circ, 180^\circ$ and 0° respectively, ω -scans, $\Delta\omega = 0.3^\circ$, time per frame 10 s). Data reduction up to $\theta = 30^\circ$ by program SAINT, corrections for absorption with program SADABS, 6248 reflections measured, 3286 independent, $R_{\text{int}} = 0.038$.¹ Structure solution with direct methods and program SHELXS97, structure refinement on F^2 using program SHELXL97.²

The asymmetric unit of the structure is formed by one molecule of $\mathbf{15S}_\mathbf{S}$ and one structural water molecule with an occupancy factor refined to 0.58. This water molecule appears to be hydrogen bonded to the carbonyl oxygen (O₉) of the tetrachlorophthalimide moiety [displaying H_{11Aw}···O₉ and O_{11w}···O₉ distances of 2.03(3) Å and 3.00(10) Å, respectively, and an O_{11w}-H_{11Aw}···O₉ angle of 176(13) $^\circ$] and on to the carbonyl oxygen (O_{7A}) of an acetate moiety of a second symmetrical $\mathbf{15S}_\mathbf{S}$ disordered in two positions. The occupancy factors for this disordered moiety O_{7A} was refined to 0.58, displaying H_{11Bw}···O_{7A} and O_{11w}···O_{7A} distances of 2.25(13) Å and 2.99(2) Å, respectively, and an O_{11w}-H_{11Bw}···O_{7A} angle of 131(13) $^\circ$.

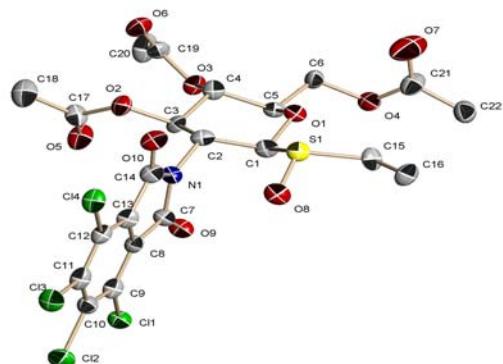
All non-hydrogen atoms were refined with anisotropic displacement parameters. The O_{11w}-bound hydrogen atoms H_{11Aw} and H_{11Bw} of the water molecule, located on a difference Fourier map, were refined with isotropic displacement parameters and geometric restraints (DFIX commands). All the others hydrogen atoms were included from calculated positions and refined riding on the atoms which they are bonded with isotropic thermal parameter. Final refinement with 3286 data, 3 restraints, and 385 parameters gave $R1 = 0.0600$, $wR2 = 0.0983$ (all data), and $R1 = 0.0490$, $wR2 = 0.0949$ [$I > 2\sigma(I)$], ($S = 1.007$). The highest residual peak, 0.960 Å⁻³, and deepest hole, -0.245 Å⁻³, were located close to the heavier atoms S1 and Cl3 and have no chemical sense.

The absolute structure was determined by examination of the Flack parameter $x = -0.11(9)$. The results of configurational analysis by X-ray crystallographic

examination of **15S_S** showed that this molecule possesses the S-configuration at sulfur as shown in the ORTEP representation in Figure 1.

1. Bruker, Programs SMART, version 5.054; SAINT, version 6.2.9; SADABS version 2.03; XPREP, version 5.1; SHELLXTL, version 5.1. / Bruker AXS Inc., Madison, WI, USA, 2001.
2. G.M. Sheldrick, Programs SHELXS97 (crystal structure solution) and SHELXL97 (crystal structure refinement), University of Göttingen, Germany, 1997.

Figure 1^a



^aORTEP drawing of S_Sof sulfoxide **15S_S**

Table 1. Crystal data and structure refinement for **15S_s**.

Empirical formula	C22 H22.17 Cl4 N O10.59 S		
Formula weight	643.79		
Temperature	100(2) K		
Wavelength	0.71073 Å		
Crystal system	Monoclinic		
Space group	P21		
Unit cell dimensions	a = 6.169(10) Å	α= 90°.	
	b = 11.94(2) Å	β= 93.59(4)°.	
	c = 19.10(3) Å	γ= 90°.	
Volume	1404(4) Å ³		
Z	2		
Density (calculated)	1.523 Mg/m ³		
Absorption coefficient	0.552 mm ⁻¹		
F(000)	660		
Crystal size	0.68 x 0.11 x 0.09 mm ³		
Theta range for data collection	2.01 to 23.23°.		
Index ranges	-6<=h<=6, -12<=k<=13, -21<=l<=19		
Reflections collected	6248		
Independent reflections	3286 [R(int) = 0.0383]		
Completeness to theta = 23.23°	99.4 %		
Max. and min. transmission	0.9541 and 0.7053		
Refinement method	Full-matrix least-squares on F ²		
Data / restraints / parameters	3286 / 3 / 385		
Goodness-of-fit on F ²	1.007		
Final R indices [I>2sigma(I)]	R1 = 0.0490, wR2 = 0.0949		
R indices (all data)	R1 = 0.0600, wR2 = 0.0983		
Absolute structure parameter	-0.11(9)		
Largest diff. peak and hole	0.960 and -0.245 e.Å ⁻³		

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **15S_S**. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
Cl(1)	5892(2)	5750(1)	1974(1)	25(1)
Cl(2)	2658(3)	6933(1)	859(1)	35(1)
Cl(3)	-1490(3)	5668(2)	237(1)	33(1)
Cl(4)	-2566(2)	3248(1)	788(1)	29(1)
S(1)	4630(3)	44(1)	1542(1)	22(1)
O(1)	6171(6)	-362(3)	2845(2)	19(1)
O(2)	1903(6)	1734(3)	3769(2)	23(1)
O(3)	5747(6)	671(4)	4627(2)	23(1)
O(4)	9956(7)	-1293(4)	3510(2)	25(1)
O(5)	3391(7)	3075(4)	4492(2)	37(1)
O(6)	2493(8)	218(4)	5048(2)	38(1)
O(7A)	8750(50)	-3085(18)	3307(13)	69(7)
O(11)	10195(15)	4559(8)	3087(5)	42(4)
O(7B)	8300(60)	-2750(20)	3095(13)	34(6)
O(8)	4423(7)	1051(3)	1083(2)	29(1)
O(9)	6121(6)	3298(4)	2642(2)	24(1)
O(10)	-218(7)	1575(4)	1838(2)	26(1)
N(1)	3020(7)	2232(4)	2369(2)	18(1)
C(1)	5523(10)	586(5)	2435(3)	21(1)
C(2)	3599(10)	1190(5)	2740(3)	18(1)
C(3)	4001(9)	1434(6)	3539(3)	20(2)
C(4)	4947(10)	384(5)	3918(3)	20(2)
C(5)	6935(9)	-14(6)	3539(3)	18(1)
C(6)	8081(10)	-1003(6)	3893(3)	28(2)
C(7)	4344(9)	3187(5)	2328(3)	18(1)
C(8)	3107(9)	3968(5)	1831(3)	16(1)
C(9)	3587(9)	5067(6)	1627(3)	21(2)
C(10)	2137(10)	5583(6)	1141(3)	23(2)
C(11)	239(10)	5035(6)	861(3)	24(2)
C(12)	-249(10)	3934(5)	1087(3)	20(2)
C(13)	1221(9)	3437(5)	1578(3)	18(1)
C(14)	1162(10)	2307(6)	1904(3)	23(2)
C(15)	7104(10)	-689(6)	1362(3)	26(2)

C(16)	7083(11)	-989(6)	585(3)	33(2)
C(17)	1817(11)	2587(5)	4241(3)	23(2)
C(18)	-521(10)	2793(6)	4403(4)	41(2)
C(19)	4363(11)	526(6)	5154(3)	29(2)
C(20)	5467(11)	803(7)	5855(3)	42(2)
C(21)	9969(12)	-2262(6)	3166(3)	33(2)
C(22)	11893(11)	-2399(6)	2740(3)	34(2)

Table 3. Bond lengths [\AA] and angles [$^\circ$] for **15S_s**.

Cl(1)-C(9)	1.735(6)	C(6)-H(6A)	0.9416
Cl(2)-C(10)	1.736(7)	C(6)-H(6B)	0.9416
Cl(3)-C(11)	1.724(6)	C(7)-C(8)	1.503(8)
Cl(4)-C(12)	1.715(7)	C(8)-C(13)	1.385(8)
S(1)-O(8)	1.488(4)	C(8)-C(9)	1.405(9)
S(1)-C(15)	1.811(7)	C(9)-C(10)	1.393(8)
S(1)-C(1)	1.876(6)	C(10)-C(11)	1.416(9)
O(1)-C(1)	1.419(7)	C(11)-C(12)	1.422(9)
O(1)-C(5)	1.441(6)	C(12)-C(13)	1.395(8)
O(2)-C(17)	1.364(7)	C(13)-C(14)	1.487(9)
O(2)-C(3)	1.438(7)	C(15)-C(16)	1.526(8)
O(3)-C(19)	1.372(7)	C(15)-H(15A)	0.9342
O(3)-C(4)	1.454(7)	C(15)-H(15B)	0.9342
O(4)-C(21)	1.331(8)	C(16)-H(16A)	0.9676
O(4)-C(6)	1.449(7)	C(16)-H(16B)	0.9676
O(5)-C(17)	1.205(7)	C(16)-H(16C)	0.9676
O(6)-C(19)	1.216(8)	C(17)-C(18)	1.514(9)
O(7A)-C(21)	1.28(3)	C(18)-H(18A)	0.9471
O(11)-H(11A)	0.98(2)	C(18)-H(18B)	0.9471
O(11)-H(11B)	0.98(2)	C(18)-H(18C)	0.9471
O(7B)-C(21)	1.19(3)	C(19)-C(20)	1.500(9)
O(9)-C(7)	1.224(7)	C(20)-H(20A)	0.8641
O(10)-C(14)	1.221(7)	C(20)-H(20B)	0.8641
N(1)-C(7)	1.408(8)	C(20)-H(20C)	0.8641
N(1)-C(14)	1.408(7)	C(21)-C(22)	1.489(9)
N(1)-C(2)	1.465(7)	C(22)-H(22A)	0.9051
C(1)-C(2)	1.534(8)	C(22)-H(22B)	0.9051
C(1)-H(1)	0.9488	C(22)-H(22C)	0.9051
C(2)-C(3)	1.557(8)		
C(2)-H(2)	0.9735	O(8)-S(1)-C(15)	108.7(3)
C(3)-C(4)	1.543(9)	O(8)-S(1)-C(1)	105.5(3)
C(3)-H(3)	0.9841	C(15)-S(1)-C(1)	97.8(3)
C(4)-C(5)	1.537(8)	C(1)-O(1)-C(5)	110.0(4)
C(4)-H(4)	1.0931	C(17)-O(2)-C(3)	117.4(5)
C(5)-C(6)	1.514(8)	C(19)-O(3)-C(4)	117.6(5)
C(5)-H(5)	0.9979	C(21)-O(4)-C(6)	119.1(5)

H(11A)-O(11)-H(11B)	110(10)	C(5)-C(6)-H(6B)	109.9
C(7)-N(1)-C(14)	111.4(5)	H(6A)-C(6)-H(6B)	108.3
C(7)-N(1)-C(2)	126.2(5)	O(9)-C(7)-N(1)	124.4(6)
C(14)-N(1)-C(2)	121.6(5)	O(9)-C(7)-C(8)	130.2(6)
O(1)-C(1)-C(2)	111.3(5)	N(1)-C(7)-C(8)	105.3(5)
O(1)-C(1)-S(1)	106.5(4)	C(13)-C(8)-C(9)	121.2(5)
C(2)-C(1)-S(1)	108.5(4)	C(13)-C(8)-C(7)	108.6(5)
O(1)-C(1)-H(1)	110.2	C(9)-C(8)-C(7)	130.2(5)
C(2)-C(1)-H(1)	110.2	C(10)-C(9)-C(8)	117.4(6)
S(1)-C(1)-H(1)	110.2	C(10)-C(9)-Cl(1)	121.6(5)
N(1)-C(2)-C(1)	112.7(5)	C(8)-C(9)-Cl(1)	121.0(5)
N(1)-C(2)-C(3)	109.5(5)	C(9)-C(10)-C(11)	122.0(6)
C(1)-C(2)-C(3)	112.3(5)	C(9)-C(10)-Cl(2)	119.6(5)
N(1)-C(2)-H(2)	107.3	C(11)-C(10)-Cl(2)	118.5(5)
C(1)-C(2)-H(2)	107.3	C(10)-C(11)-C(12)	119.9(6)
C(3)-C(2)-H(2)	107.3	C(10)-C(11)-Cl(3)	121.3(5)
O(2)-C(3)-C(4)	112.3(5)	C(12)-C(11)-Cl(3)	118.8(5)
O(2)-C(3)-C(2)	104.7(5)	C(13)-C(12)-C(11)	117.2(6)
C(4)-C(3)-C(2)	109.9(5)	C(13)-C(12)-Cl(4)	120.9(5)
O(2)-C(3)-H(3)	109.9	C(11)-C(12)-Cl(4)	121.9(5)
C(4)-C(3)-H(3)	109.9	C(8)-C(13)-C(12)	122.4(6)
C(2)-C(3)-H(3)	109.9	C(8)-C(13)-C(14)	108.1(5)
O(3)-C(4)-C(5)	106.1(5)	C(12)-C(13)-C(14)	129.4(6)
O(3)-C(4)-C(3)	110.0(5)	O(10)-C(14)-N(1)	123.6(6)
C(5)-C(4)-C(3)	108.6(4)	O(10)-C(14)-C(13)	129.9(6)
O(3)-C(4)-H(4)	110.7	N(1)-C(14)-C(13)	106.4(5)
C(5)-C(4)-H(4)	110.7	C(16)-C(15)-S(1)	110.0(4)
C(3)-C(4)-H(4)	110.7	C(16)-C(15)-H(15A)	109.7
O(1)-C(5)-C(6)	107.6(5)	S(1)-C(15)-H(15A)	109.7
O(1)-C(5)-C(4)	107.3(5)	C(16)-C(15)-H(15B)	109.7
C(6)-C(5)-C(4)	113.3(5)	S(1)-C(15)-H(15B)	109.7
O(1)-C(5)-H(5)	109.5	H(15A)-C(15)-H(15B)	108.2
C(6)-C(5)-H(5)	109.5	C(15)-C(16)-H(16A)	109.5
C(4)-C(5)-H(5)	109.5	C(15)-C(16)-H(16B)	109.5
O(4)-C(6)-C(5)	109.0(5)	H(16A)-C(16)-H(16B)	109.5
O(4)-C(6)-H(6A)	109.9	C(15)-C(16)-H(16C)	109.5
C(5)-C(6)-H(6A)	109.9	H(16A)-C(16)-H(16C)	109.5
O(4)-C(6)-H(6B)	109.9	H(16B)-C(16)-H(16C)	109.5

O(5)-C(17)-O(2)	124.1(6)
O(5)-C(17)-C(18)	126.5(5)
O(2)-C(17)-C(18)	109.4(5)
C(17)-C(18)-H(18A)	109.5
C(17)-C(18)-H(18B)	109.5
H(18A)-C(18)-H(18B)	109.5
C(17)-C(18)-H(18C)	109.5
H(18A)-C(18)-H(18C)	109.5
H(18B)-C(18)-H(18C)	109.5
O(6)-C(19)-O(3)	122.9(6)
O(6)-C(19)-C(20)	126.1(6)
O(3)-C(19)-C(20)	111.0(6)
C(19)-C(20)-H(20A)	109.5
C(19)-C(20)-H(20B)	109.5
H(20A)-C(20)-H(20B)	109.5
C(19)-C(20)-H(20C)	109.5
H(20A)-C(20)-H(20C)	109.5
H(20B)-C(20)-H(20C)	109.5
O(7B)-C(21)-O(7A)	28.9(15)
O(7B)-C(21)-O(4)	117.1(16)
O(7A)-C(21)-O(4)	122.9(13)
O(7B)-C(21)-C(22)	126.8(16)
O(7A)-C(21)-C(22)	122.0(13)
O(4)-C(21)-C(22)	113.3(6)
C(21)-C(22)-H(22A)	109.5
C(21)-C(22)-H(22B)	109.5
H(22A)-C(22)-H(22B)	109.5
C(21)-C(22)-H(22C)	109.5
H(22A)-C(22)-H(22C)	109.5
H(22B)-C(22)-H(22C)	109.5

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **15S_S**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
Cl(1)	22(1)	21(1)	32(1)	1(1)	4(1)	-2(1)
Cl(2)	41(1)	26(1)	37(1)	9(1)	7(1)	4(1)
Cl(3)	34(1)	41(1)	22(1)	6(1)	-1(1)	12(1)
Cl(4)	19(1)	42(1)	26(1)	-8(1)	-4(1)	3(1)
S(1)	25(1)	25(1)	17(1)	-3(1)	1(1)	2(1)
O(1)	23(2)	17(2)	19(2)	1(2)	2(2)	1(2)
O(2)	18(2)	26(3)	25(2)	-6(2)	8(2)	-3(2)
O(3)	26(2)	27(3)	16(2)	-3(2)	7(2)	-3(2)
O(4)	21(2)	21(3)	34(2)	0(2)	8(2)	1(2)
O(5)	22(3)	42(3)	47(3)	-21(3)	6(2)	-2(3)
O(6)	38(3)	44(3)	34(3)	3(2)	18(2)	-2(3)
O(7A)	78(15)	20(11)	115(19)	-6(9)	53(14)	-13(9)
O(11)	40(6)	39(7)	48(6)	-16(5)	8(4)	-14(5)
O(7B)	49(13)	21(15)	33(9)	0(8)	26(8)	-17(11)
O(8)	39(3)	22(3)	24(2)	4(2)	1(2)	1(2)
O(9)	21(2)	20(3)	29(2)	-1(2)	-4(2)	-2(2)
O(10)	19(2)	25(3)	33(2)	-1(2)	5(2)	-6(2)
N(1)	17(3)	17(3)	19(3)	-2(2)	2(2)	-5(2)
C(1)	18(3)	20(4)	24(3)	-3(3)	3(3)	-10(3)
C(2)	16(3)	15(4)	23(3)	1(3)	3(3)	1(3)
C(3)	9(3)	24(4)	27(3)	-1(3)	4(3)	-3(3)
C(4)	20(4)	21(4)	21(3)	2(3)	2(3)	-4(3)
C(5)	17(3)	18(3)	19(3)	-2(3)	1(3)	-4(3)
C(6)	30(4)	33(4)	23(3)	2(3)	13(3)	5(3)
C(7)	23(3)	17(4)	15(3)	-6(3)	3(3)	-2(3)
C(8)	9(3)	22(4)	16(3)	-4(3)	5(2)	0(3)
C(9)	15(3)	35(4)	12(3)	-8(3)	3(3)	2(3)
C(10)	23(3)	25(4)	22(3)	-2(3)	17(3)	2(3)
C(11)	27(4)	34(4)	12(3)	-7(3)	9(3)	13(3)
C(12)	17(3)	24(4)	19(3)	-7(3)	10(3)	3(3)
C(13)	14(3)	21(4)	18(3)	0(3)	4(3)	4(3)
C(14)	18(4)	30(4)	22(3)	-1(3)	6(3)	-4(3)
C(15)	29(4)	24(4)	25(3)	-7(3)	2(3)	1(3)

C(16)	31(4)	43(5)	24(4)	1(3)	10(3)	1(4)
C(17)	27(4)	15(4)	28(3)	-8(3)	4(3)	6(3)
C(18)	23(4)	40(5)	59(5)	-23(4)	-1(4)	-1(4)
C(19)	36(5)	30(4)	24(4)	3(3)	19(3)	10(4)
C(20)	49(5)	54(5)	25(4)	1(4)	11(3)	16(5)
C(21)	33(5)	27(5)	39(4)	-6(4)	7(3)	-10(4)
C(22)	27(4)	35(5)	39(4)	-14(4)	8(3)	2(3)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^{-3}$) for **15S_S**.

	x	y	z	U(eq)
H(11A)	8840(140)	4160(120)	2970(80)	110(60)
H(11B)	10100(300)	5320(60)	2890(80)	120(80)
H(1)	6710(80)	1090(30)	2405(3)	25
H(2)	2350(80)	690(30)	2684(4)	22
H(3)	5010(70)	2070(40)	3611(5)	24
H(4)	3730(70)	-280(40)	3925(3)	25
H(5)	7980(60)	620(40)	3505(3)	22
H(6A)	8520(20)	-819(9)	4360(20)	34
H(6B)	7130(40)	-1620(30)	3900(3)	34
H(15A)	8300(50)	-235(19)	1484(6)	31
H(15B)	7223(12)	-1340(30)	1632(12)	31
H(16A)	5920(60)	-1520(30)	471(7)	49
H(16B)	8460(50)	-1320(30)	486(6)	49
H(16C)	6850(60)	-320(20)	306(10)	49
H(18A)	-950(30)	2250(30)	4730(20)	61
H(18B)	-1420(40)	2730(40)	3985(16)	61
H(18C)	-654(18)	3520(30)	4590(20)	61
H(20A)	4500(40)	900(40)	6159(15)	63
H(20B)	6210(70)	1410(30)	5822(7)	63
H(20C)	6330(70)	260(30)	5988(14)	63
H(22A)	12100(40)	-3140(30)	2653(18)	50
H(22B)	13080(50)	-2120(30)	2980(13)	50
H(22C)	11670(30)	-2030(30)	2330(18)	50

Table 6. Torsion angles [°] for **15S_S**.

O(1)-C(1)-C(2)-C(3)	50.3(7)
C(1)-C(2)-C(3)-C(4)	-45.3(7)
C(2)-C(3)-C(4)-C(5)	52.4(6)
C(3)-C(4)-C(5)-O(1)	-64.8(6)
C(4)-C(5)-O(1)-C(2)	39.3(5)
C(5)-O(1)-C(2)-C(3)	-11.8(5)

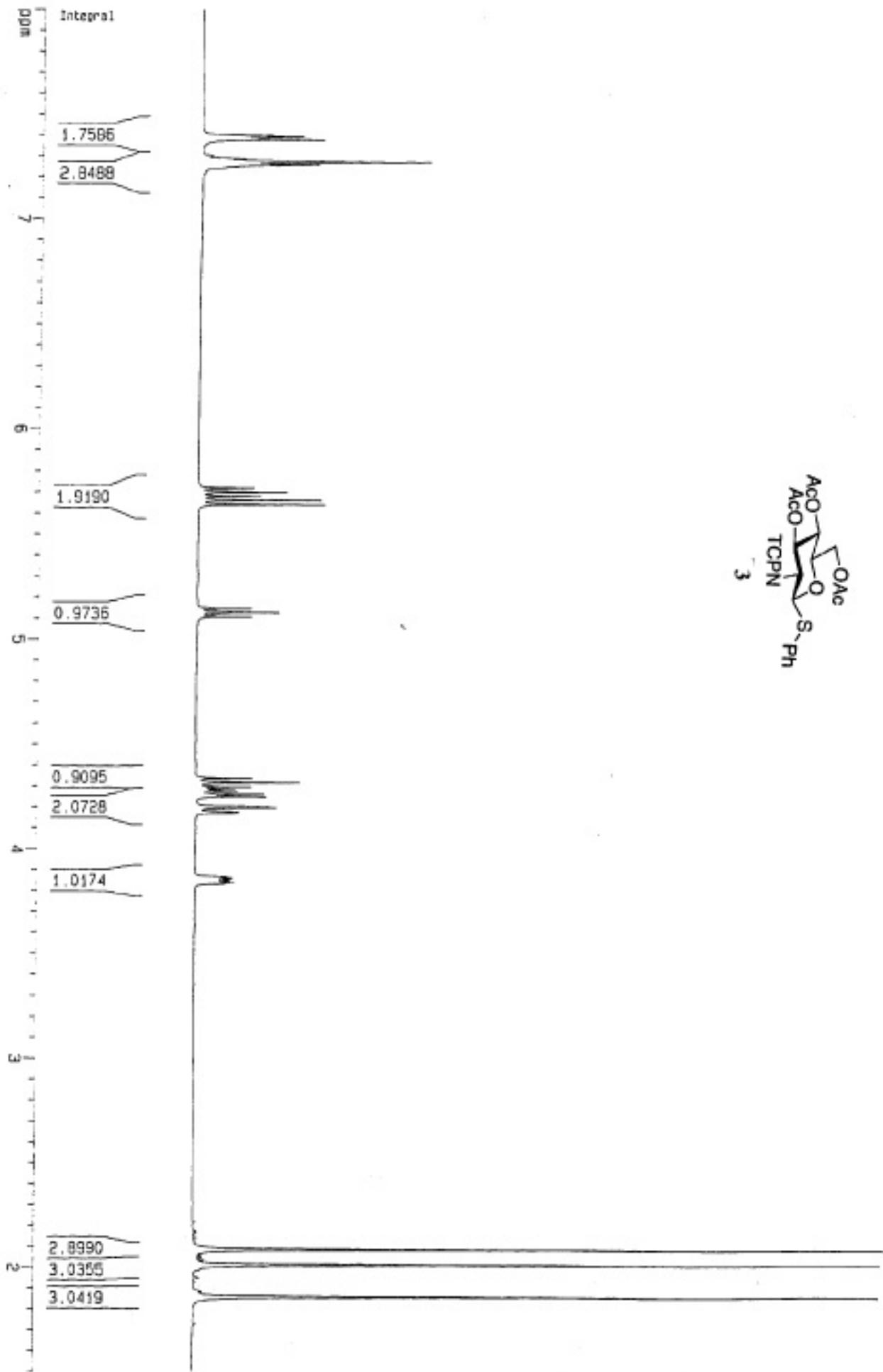
Symmetry transformations used to generate equivalent atoms:

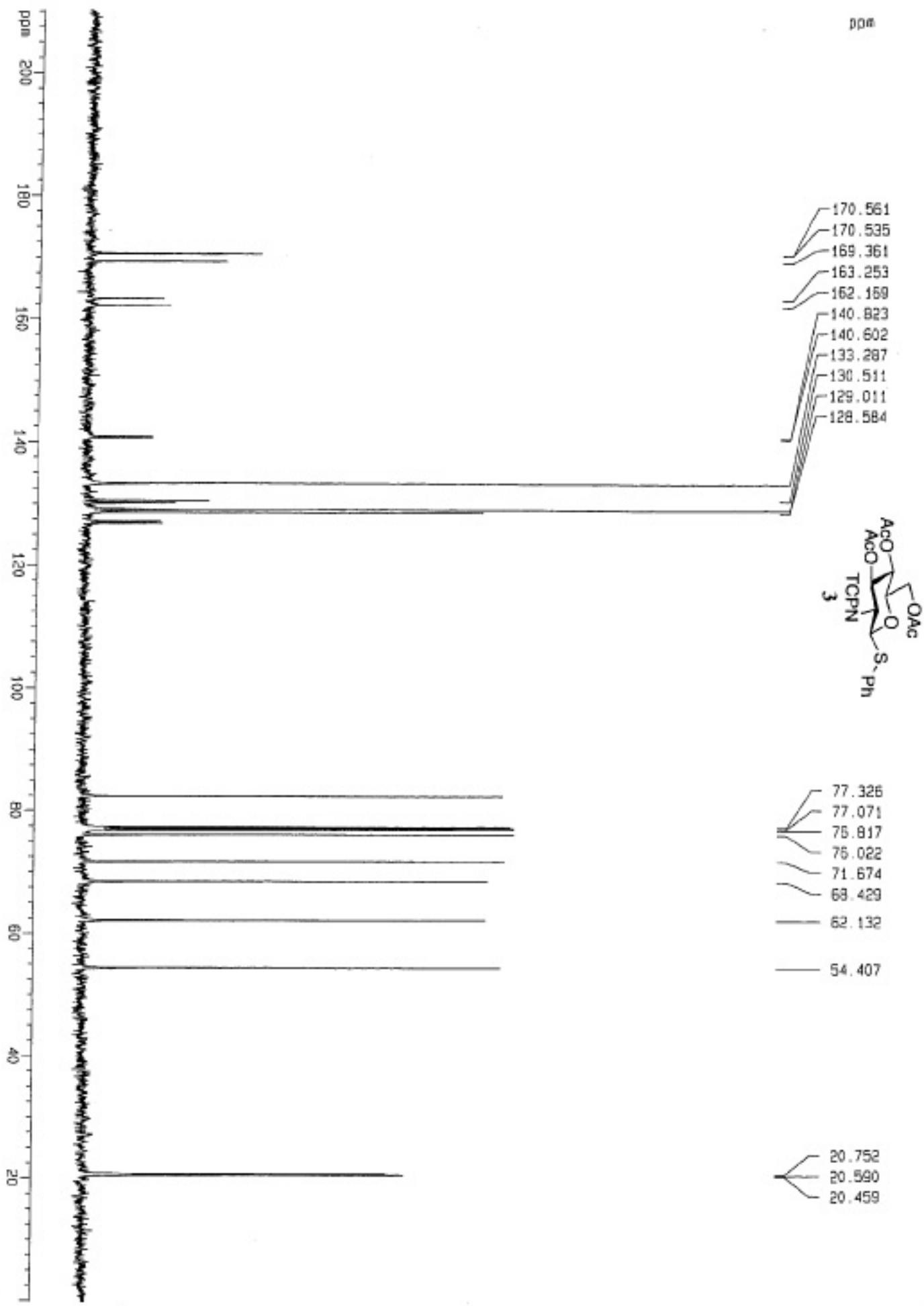
Table 7. Hydrogen bonds for **15S_S** [\AA and $^\circ$].

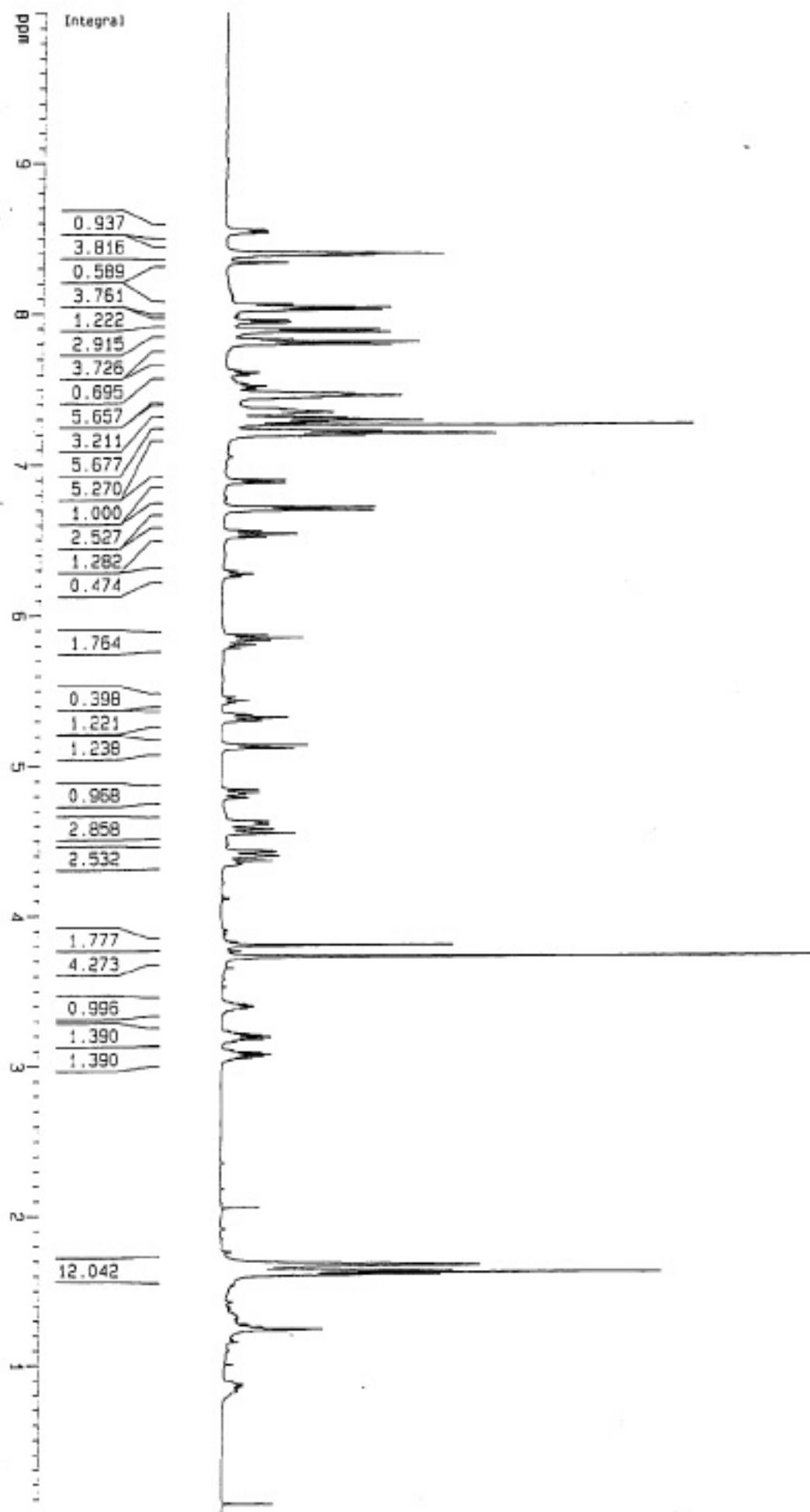
D-H...A	d(D-H)	d(H...A)	d(D...A)	\angle (DHA)
O(11)-H(11B)...O(7B) ^{#1}	0.98(2)	2.60(11)	3.41(3)	141(13)
O(11)-H(11B)...O(7A) ^{#1}	0.98(2)	2.25(13)	2.99(2)	131(13)
O(11)-H(11A)...O(9)	0.98(2)	2.03(3)	3.006(10)	175(14)

Symmetry transformations used to generate equivalent atoms:

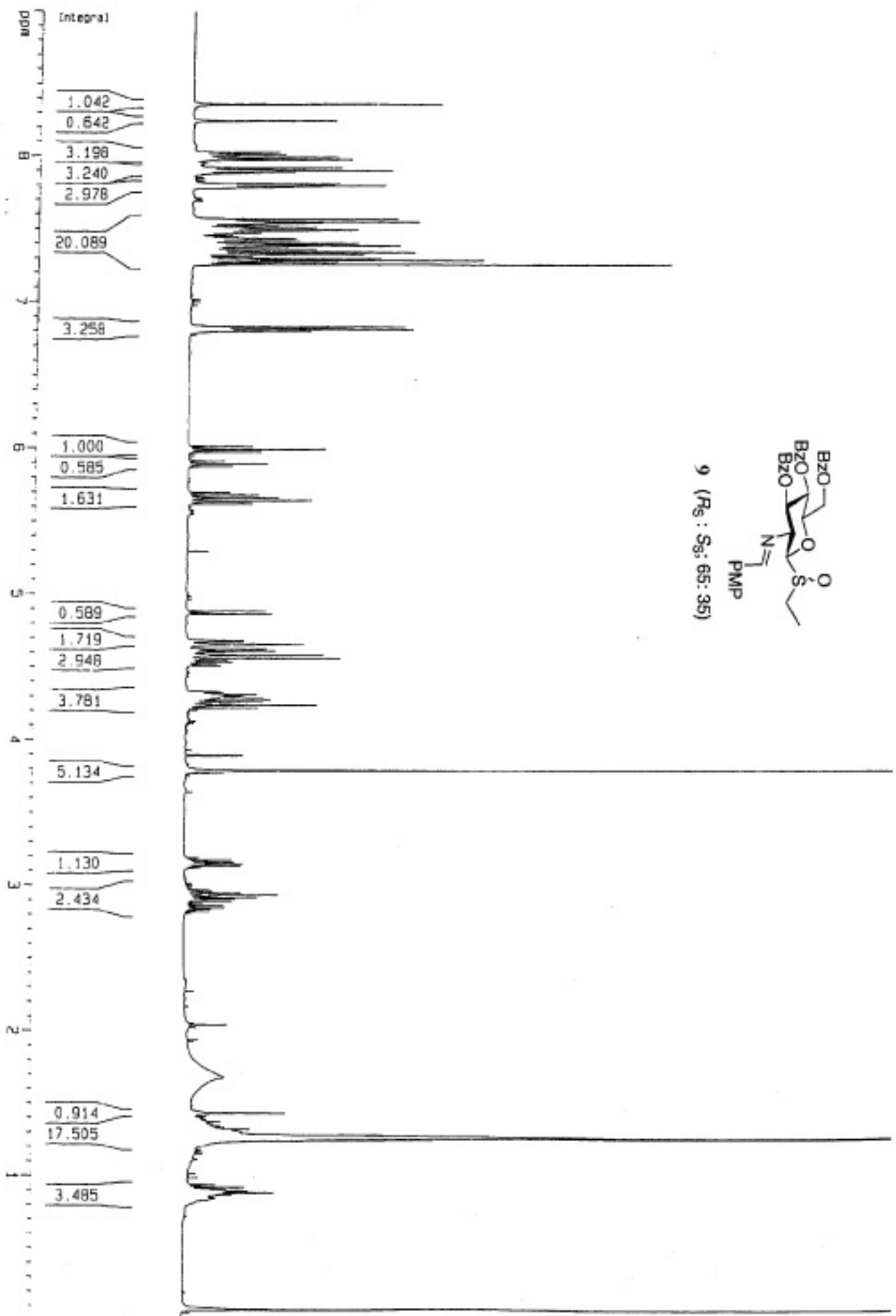
#1 x,y+1,z

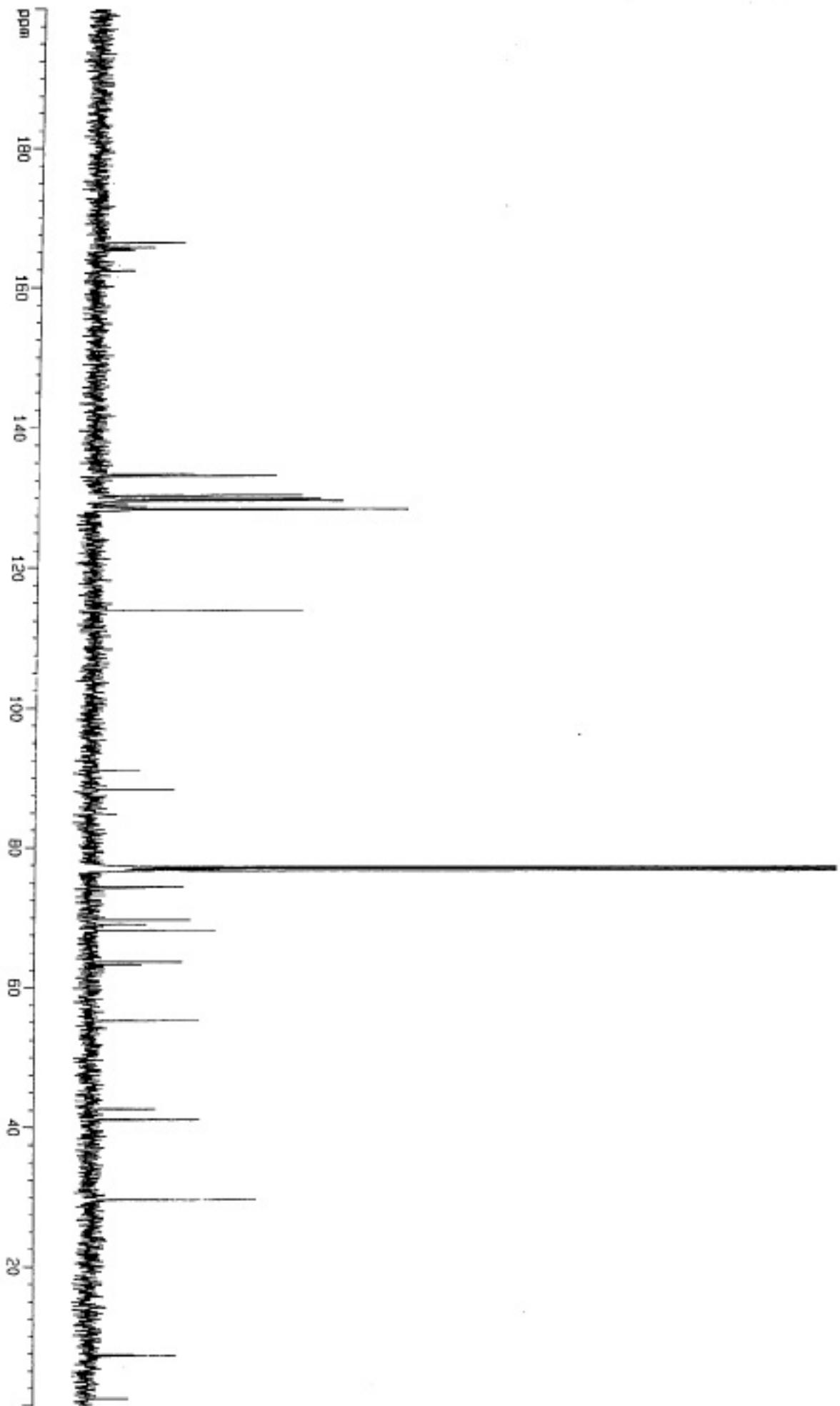




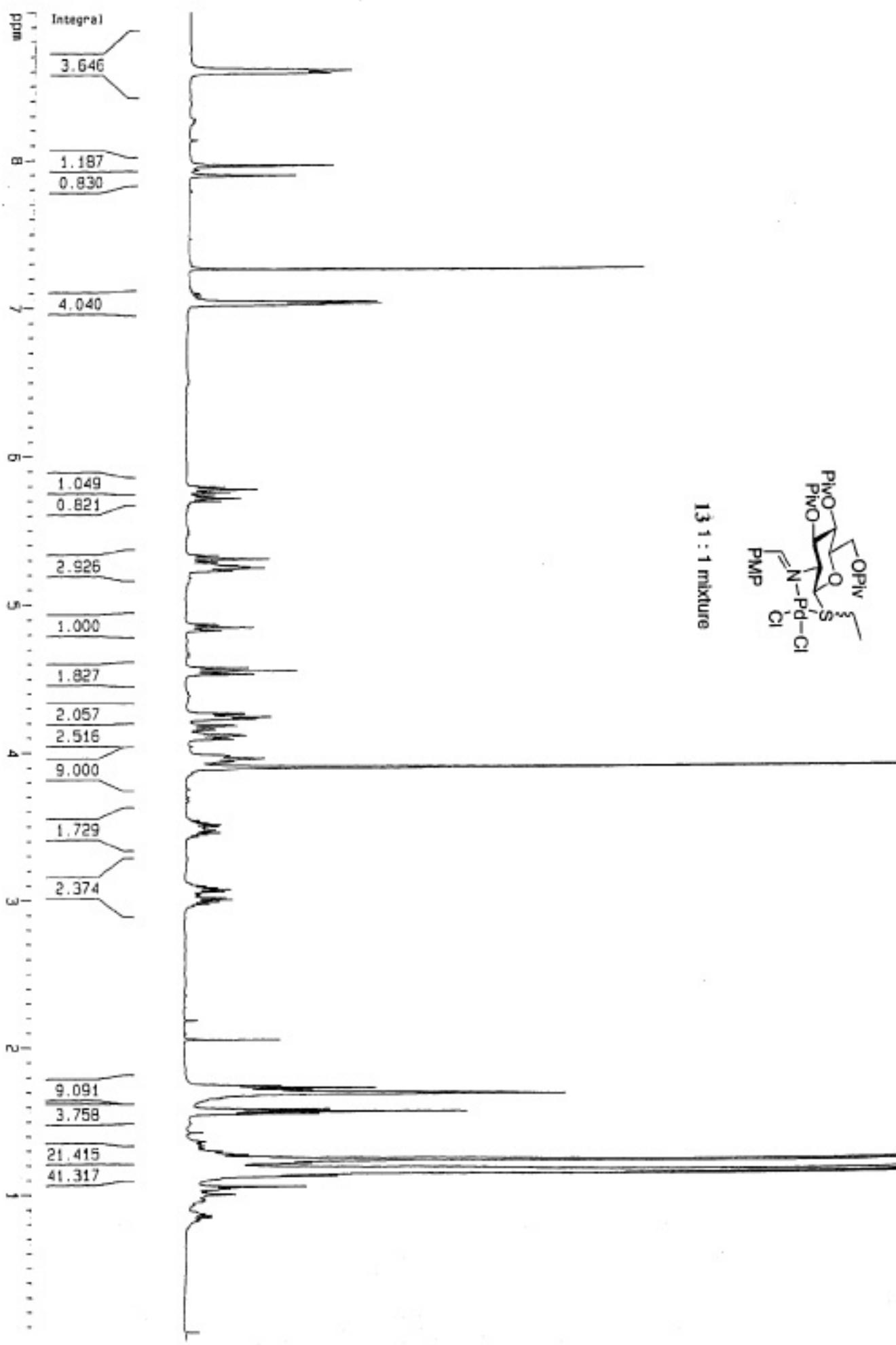


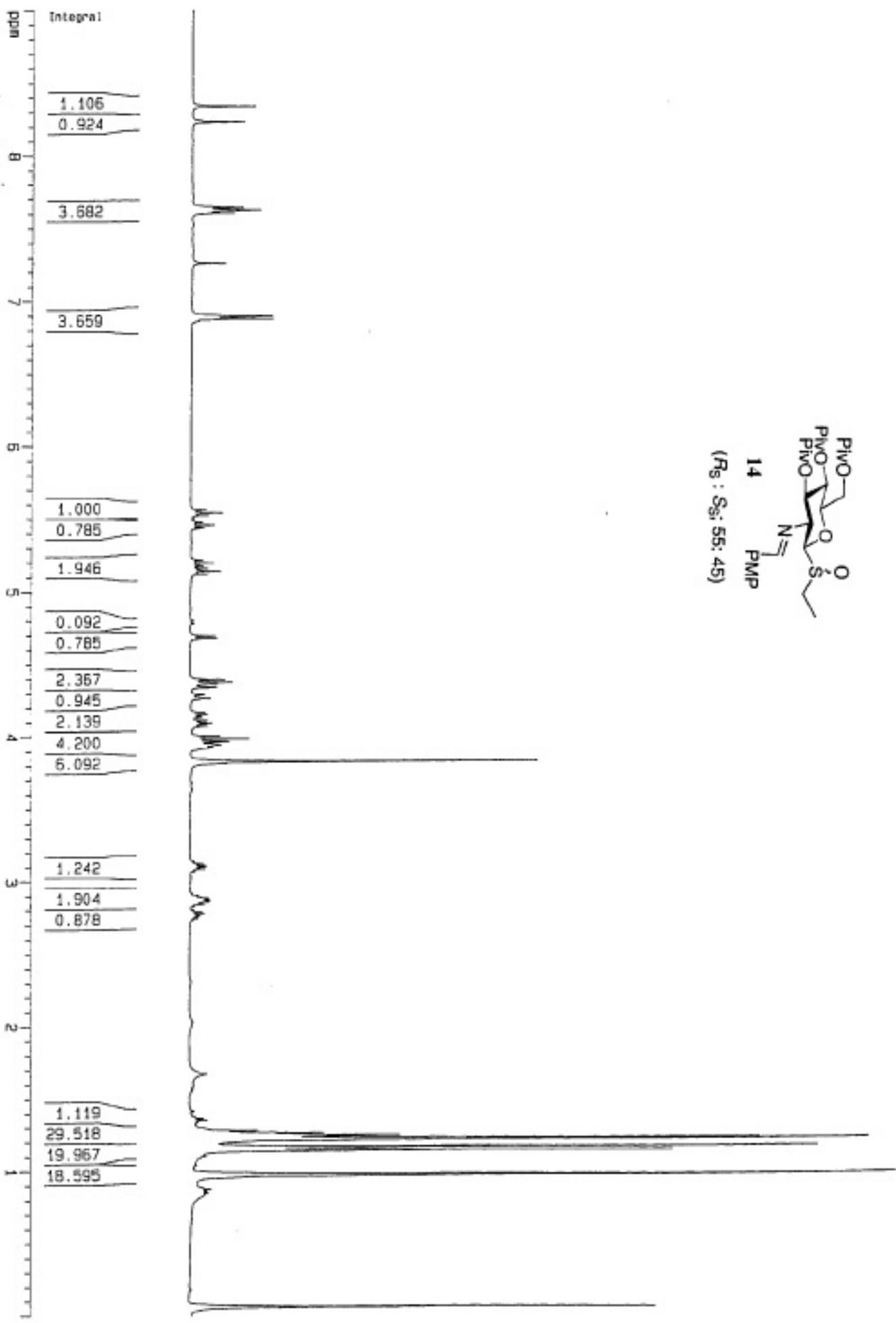
8 65 : 35 mixture

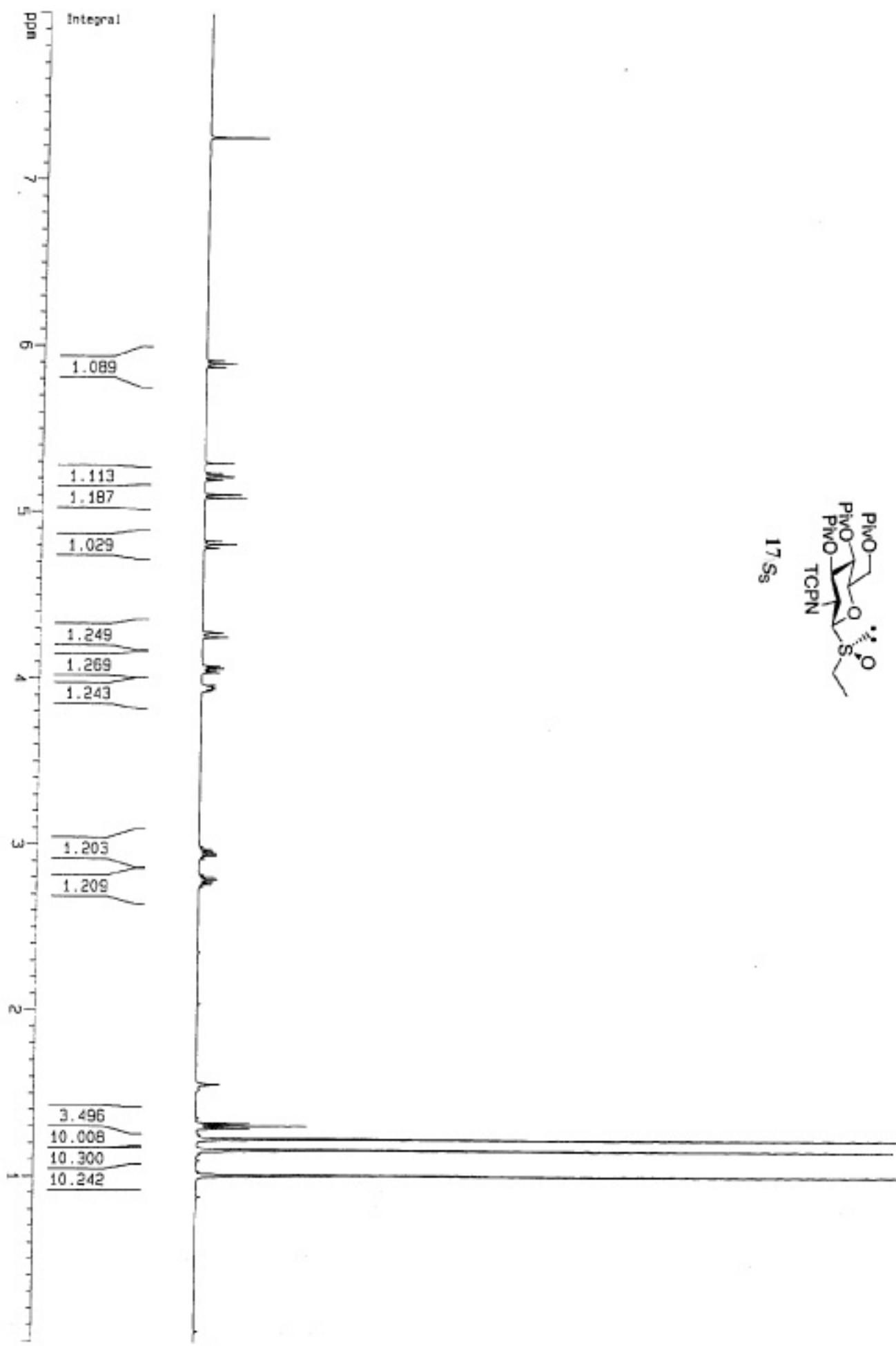




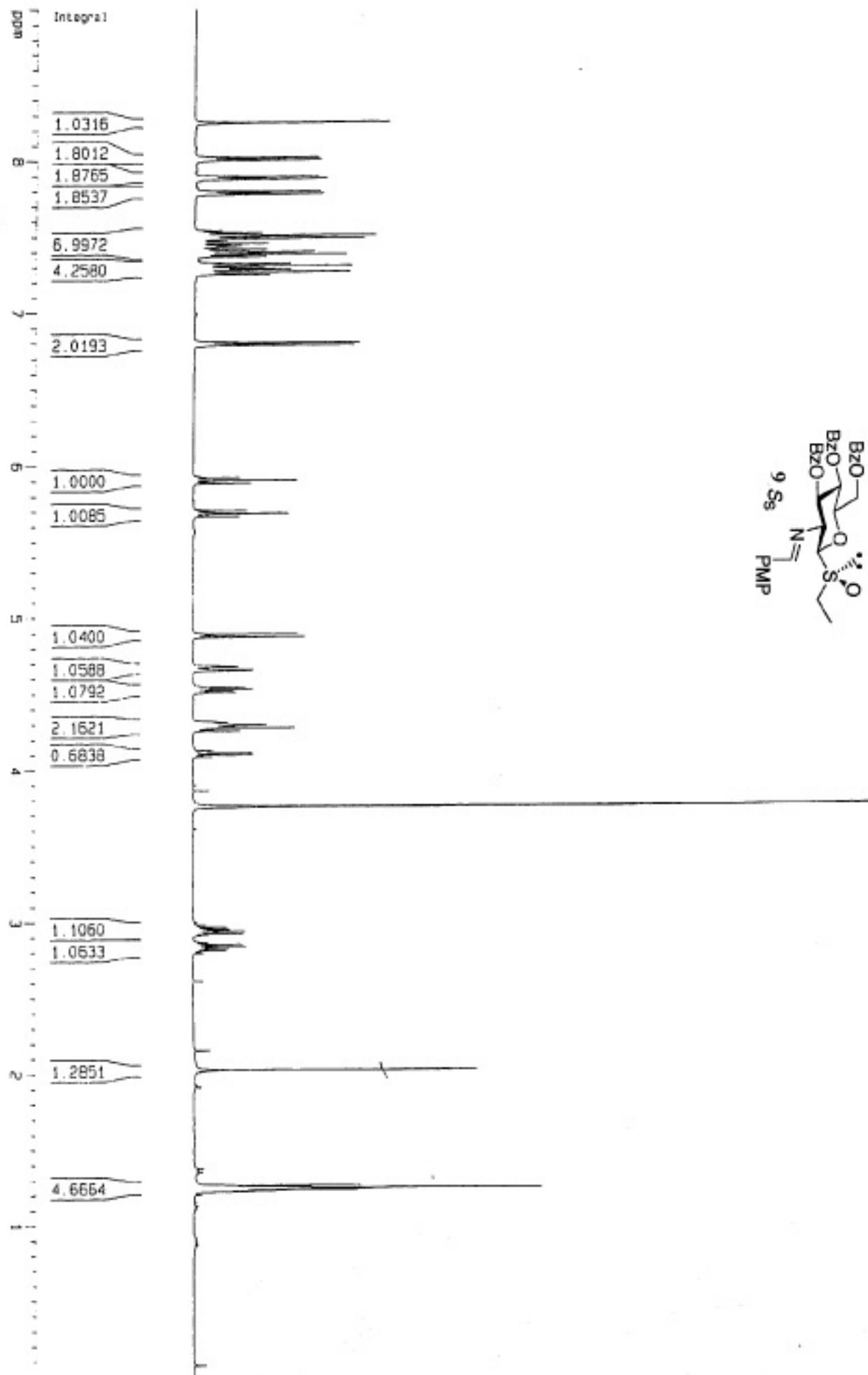
9 ($R_S : S_S$; 65: 35)

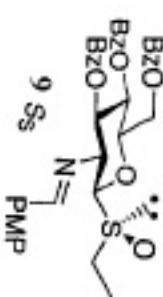
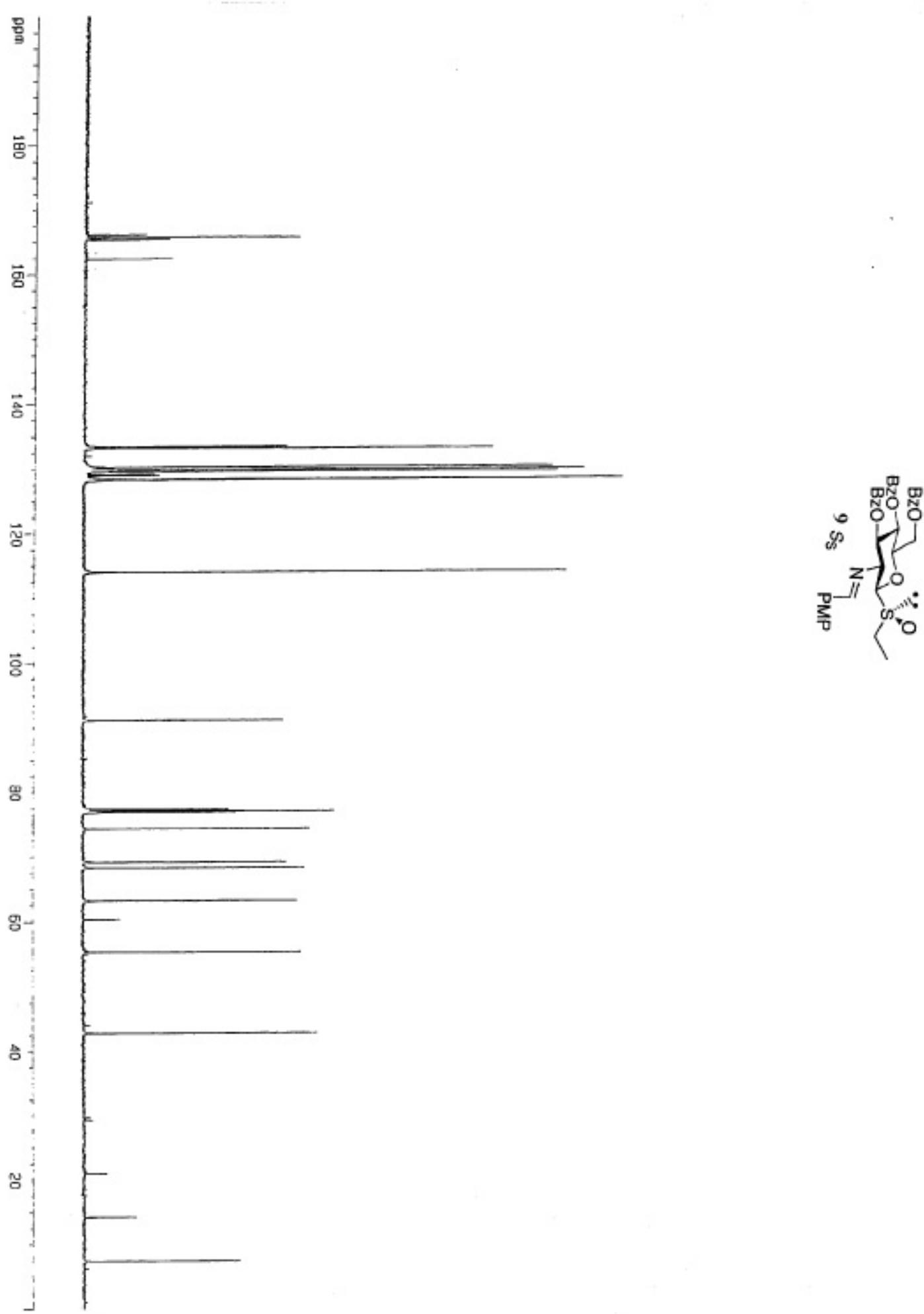


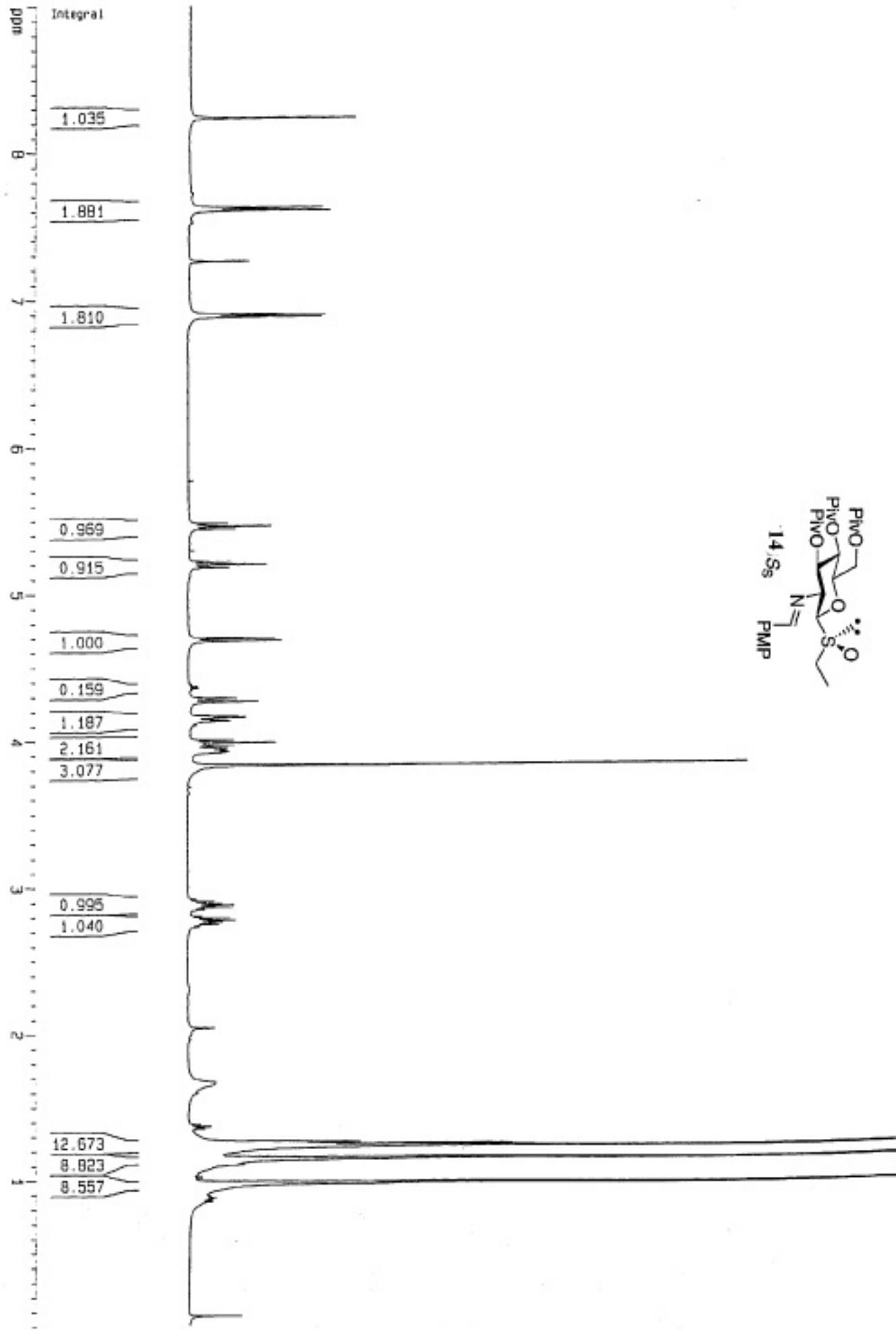


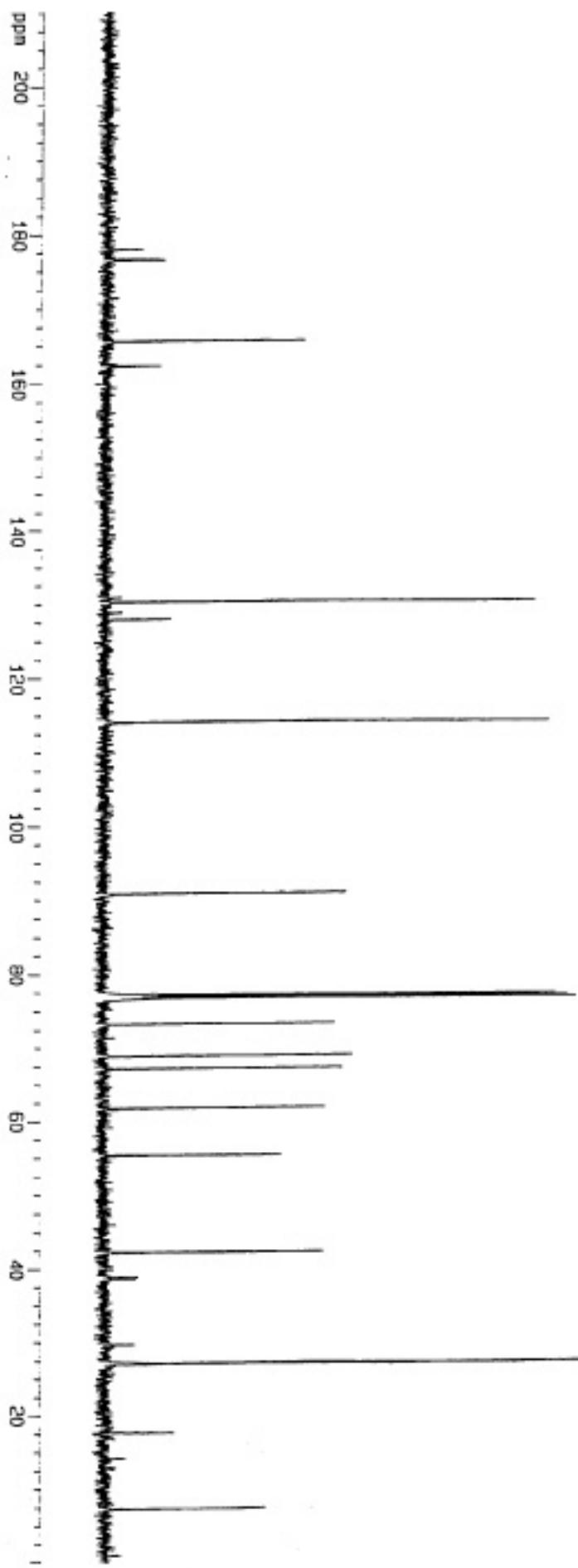


Integral









Integral

ppm
1.000
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2.375
2.130
12.129
13.288
6.603
7.454
0.808
0.764
6.837
6.723
1.349
5.928
1.964
0.438
17.390
8.627
4.466
20.218
29.460
24.616

