

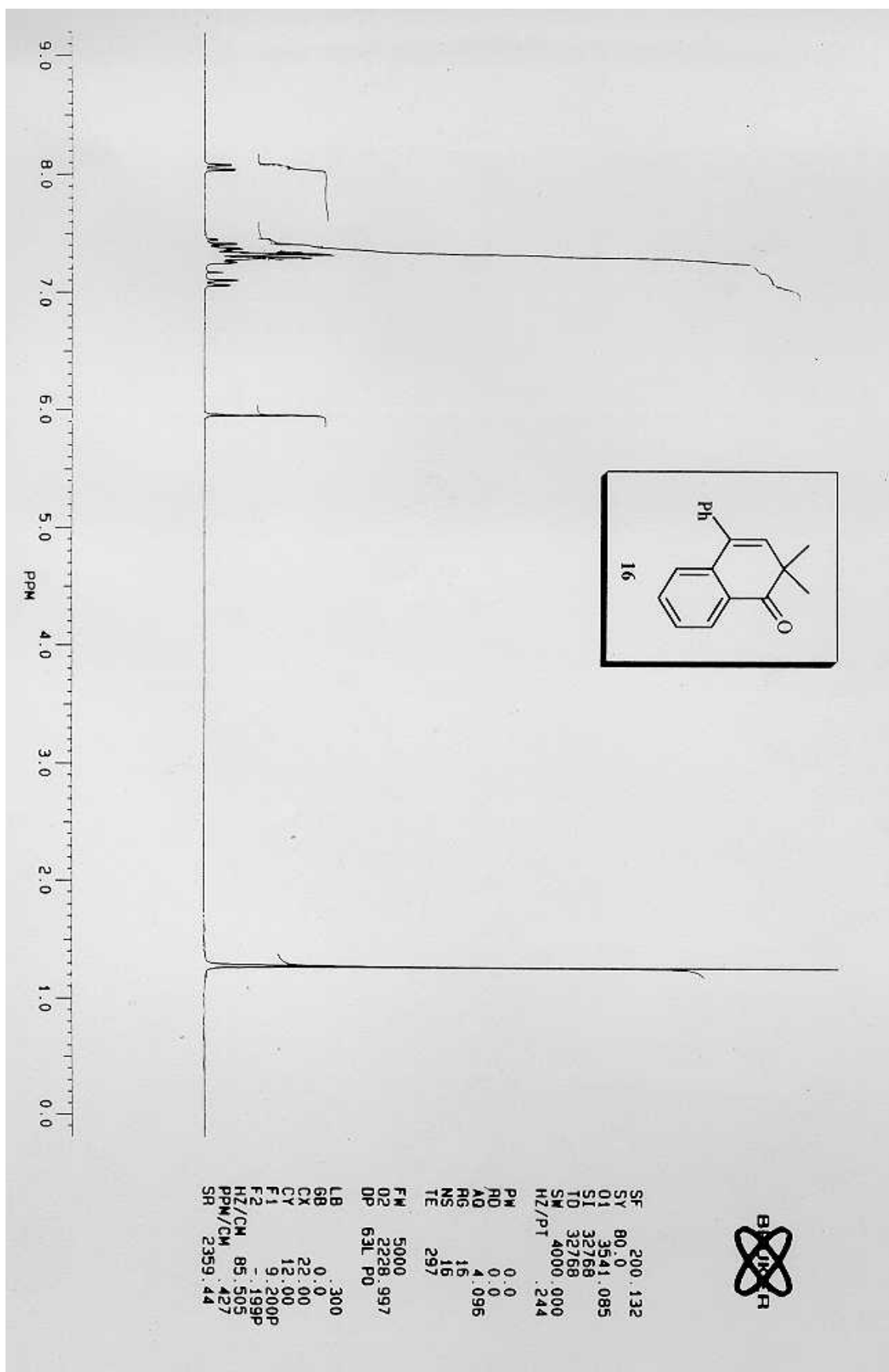
SUPPORTING INFORMATION

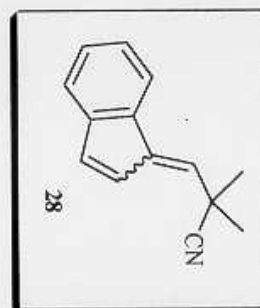
Photochemical Reactivity of 1-Substituted-1-aza-1,4-dienes Promoted by Electron-Acceptor Sensitizers. Di- π -methane Rearrangements and Alternative Reactions via Radical-cation Intermediates

Diego Armesto, Maria J. Ortiz, Antonia R. Agarrabeitia, Santiago Aparicio-Lara, Mar Martin-Fontecha, Marta Liras and M. Paz Martinez-Alcazar.

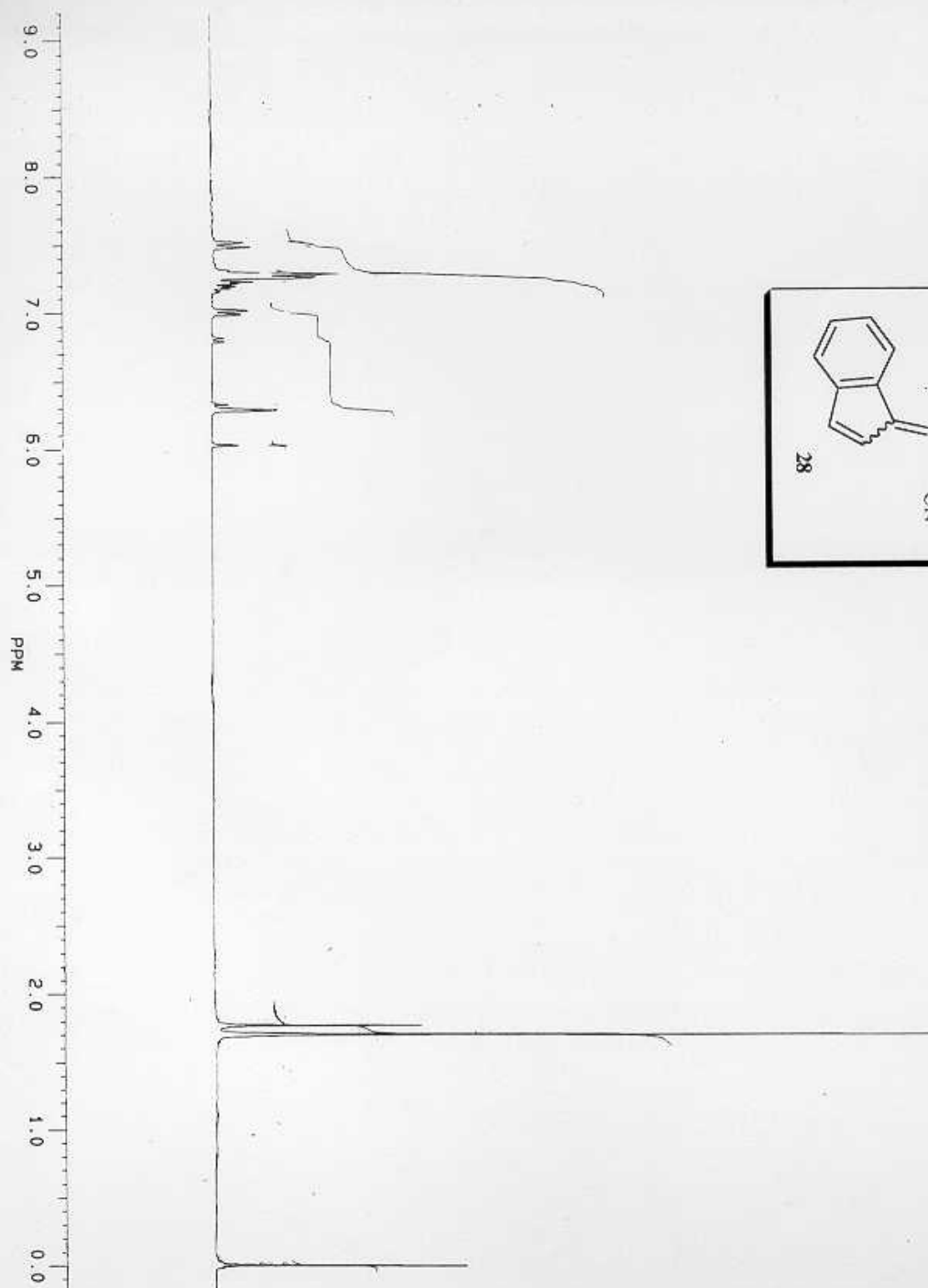
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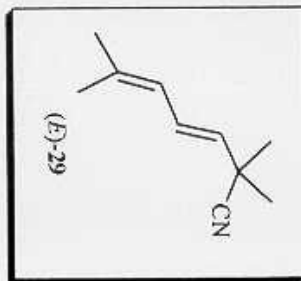


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CX	22.00
CY	12.00
F1	9.200P
F2	.199P
HZ/CM	85.505
PPM/CM	.427
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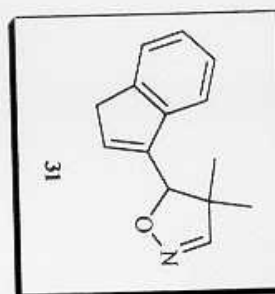
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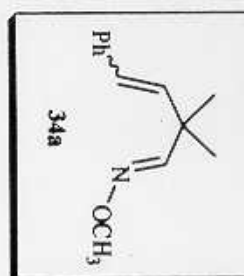
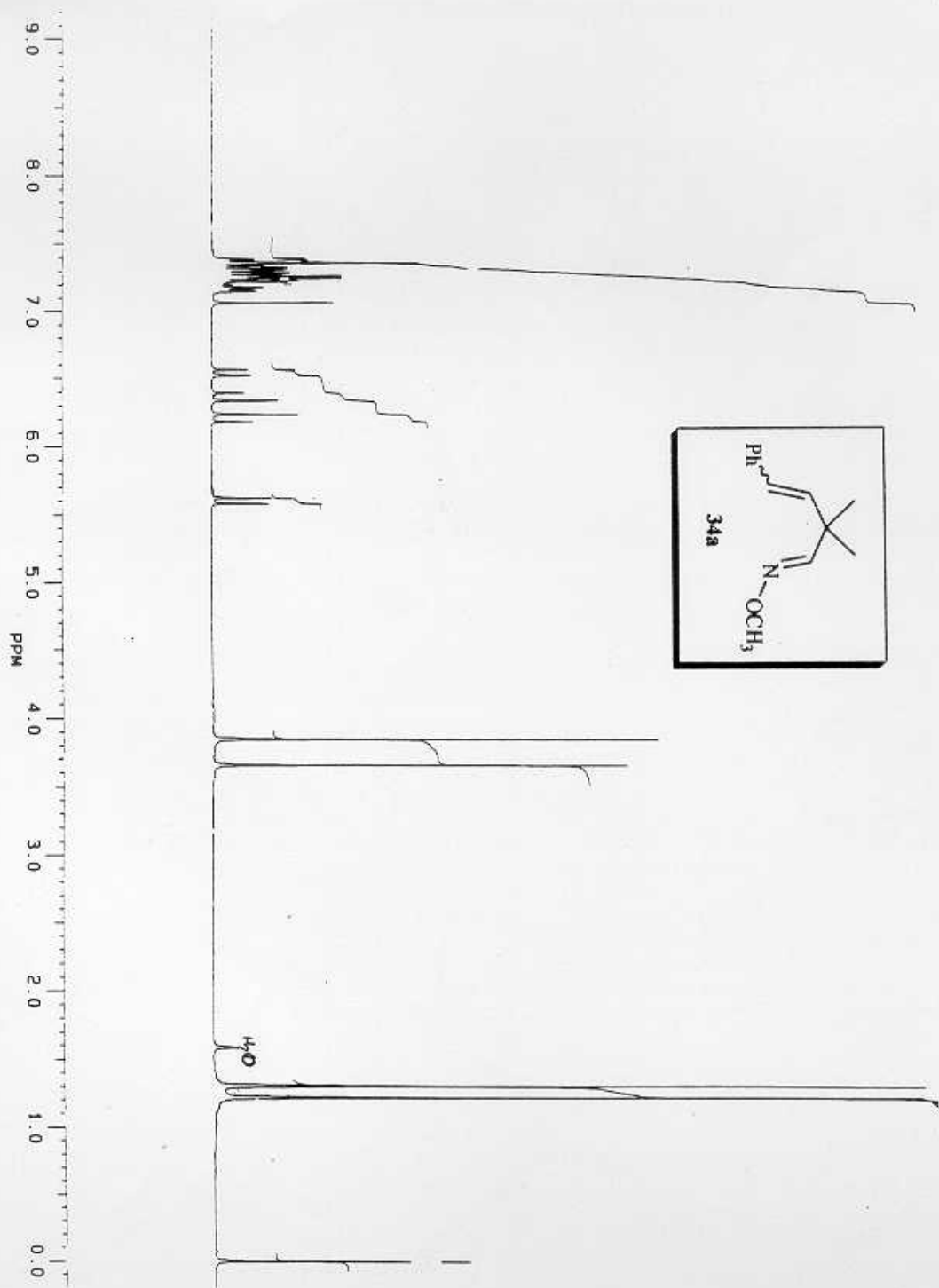
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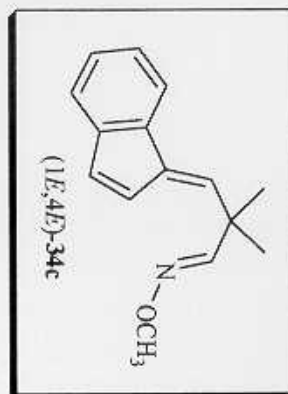
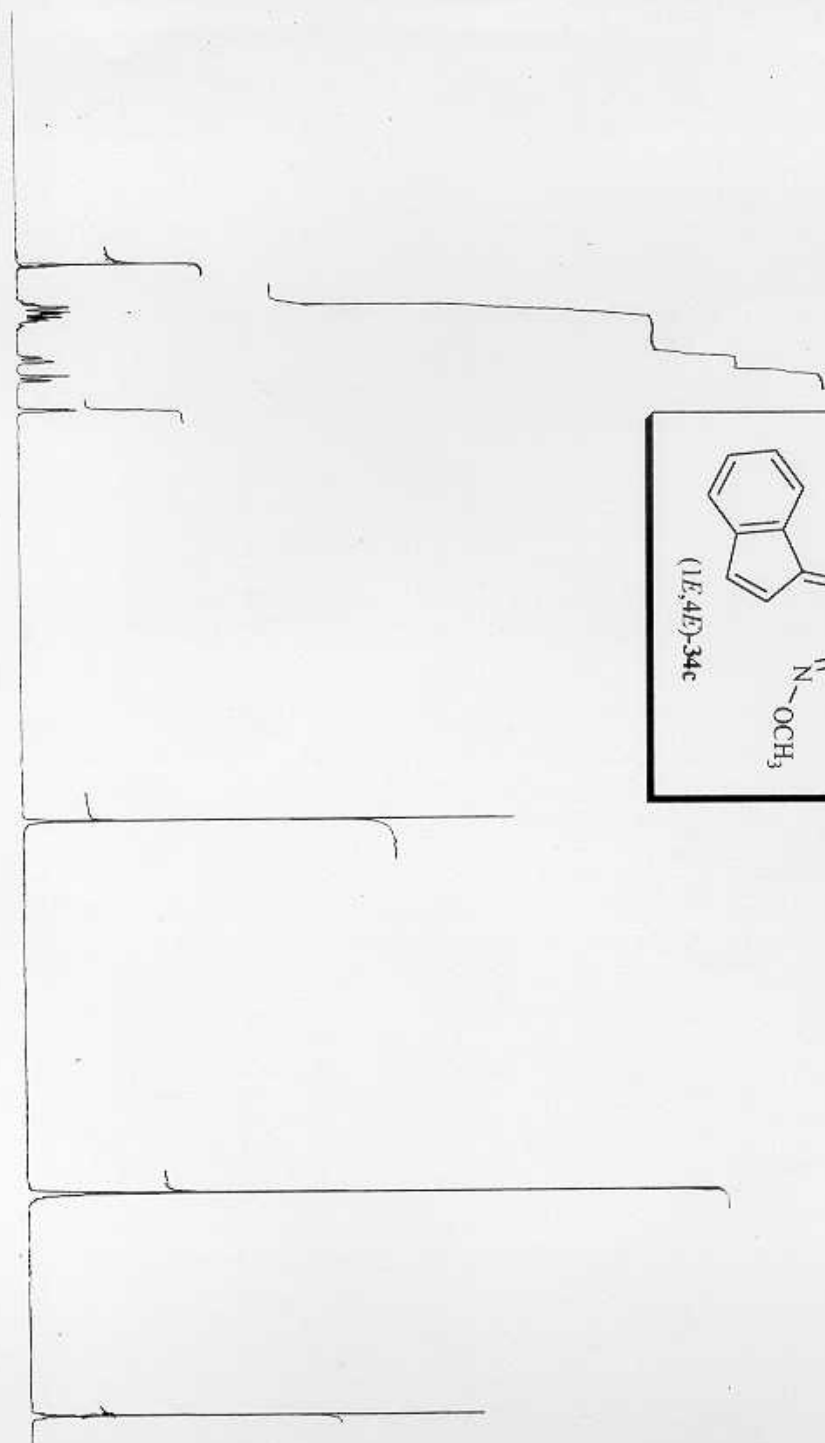


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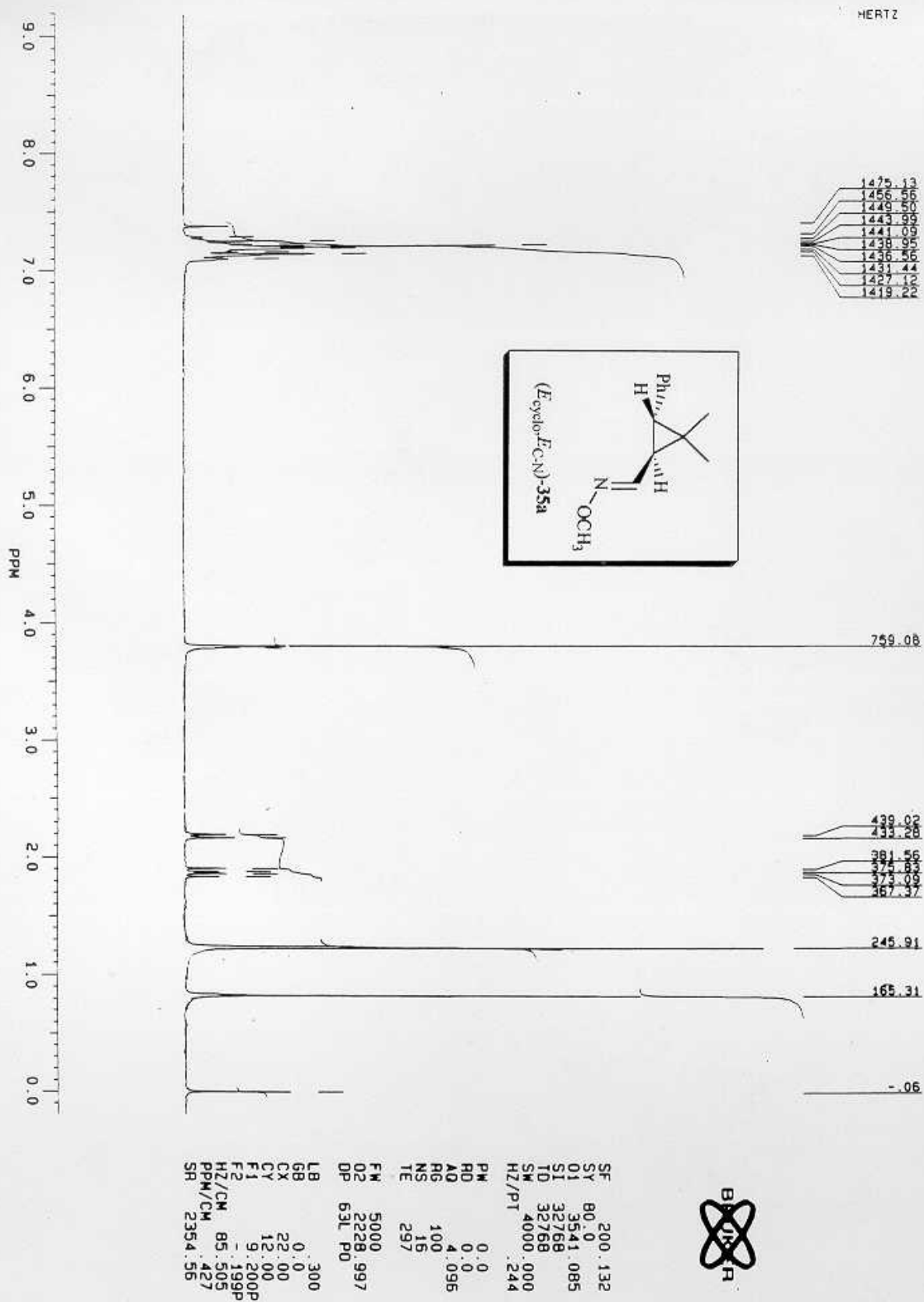


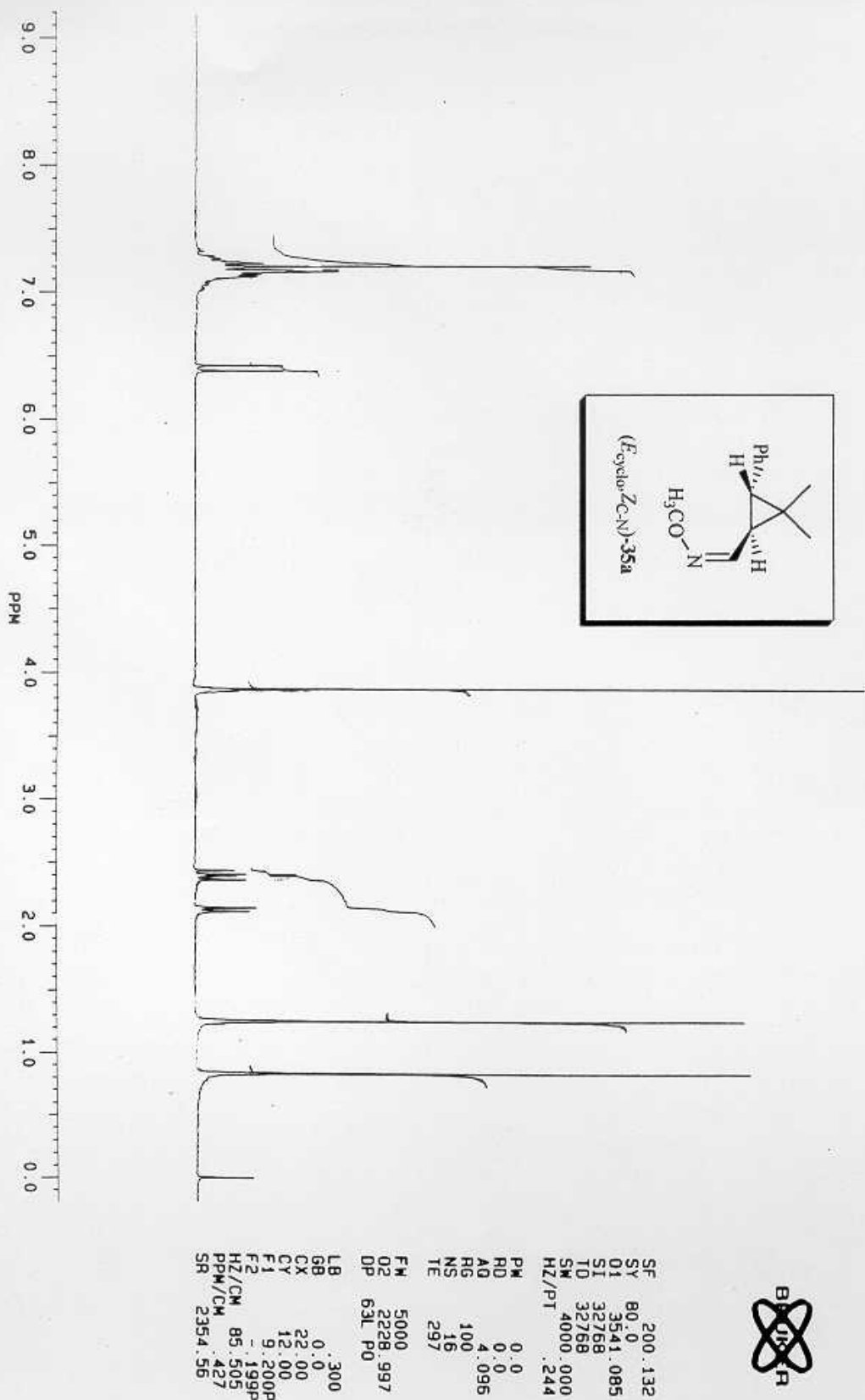
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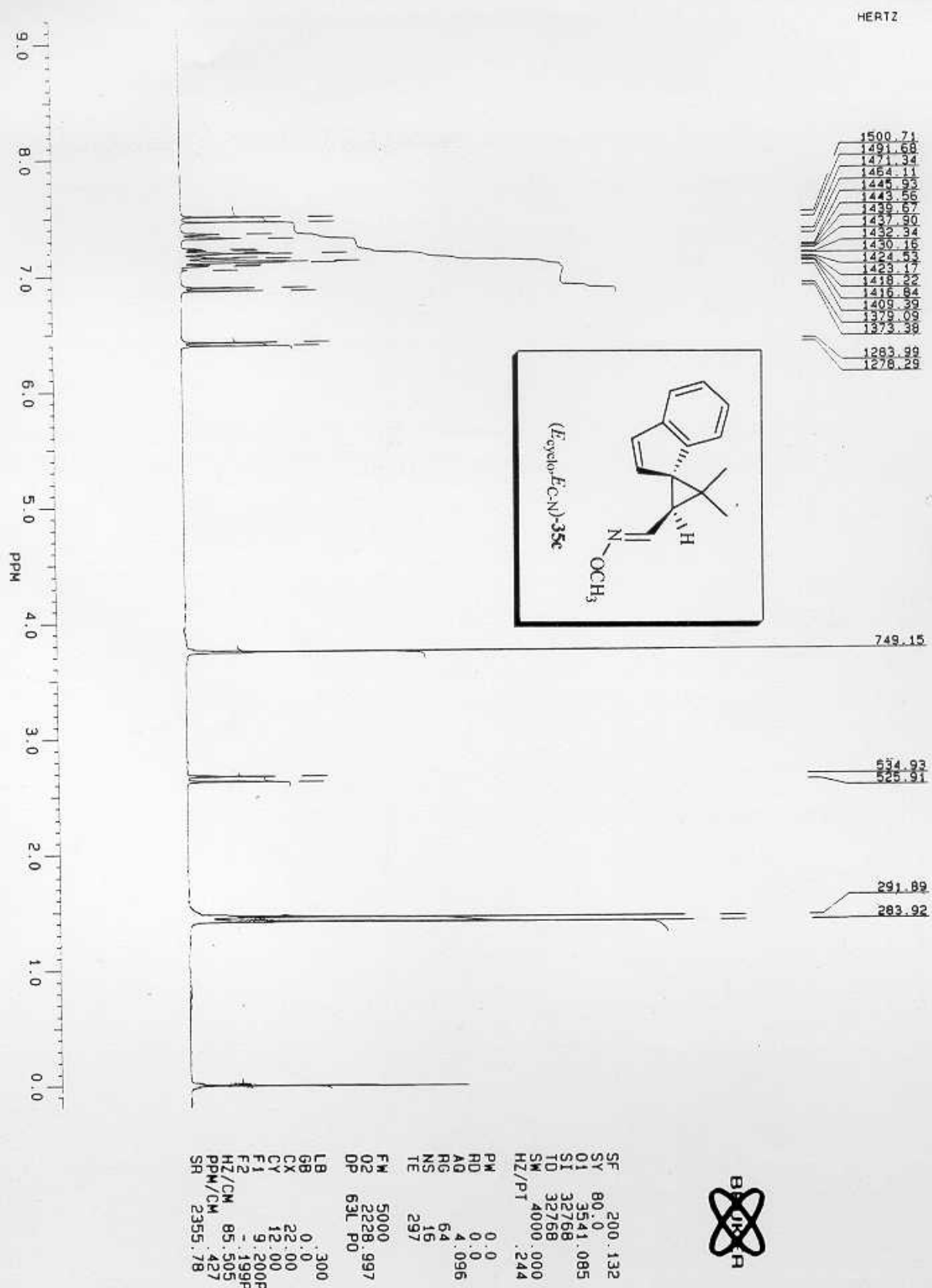
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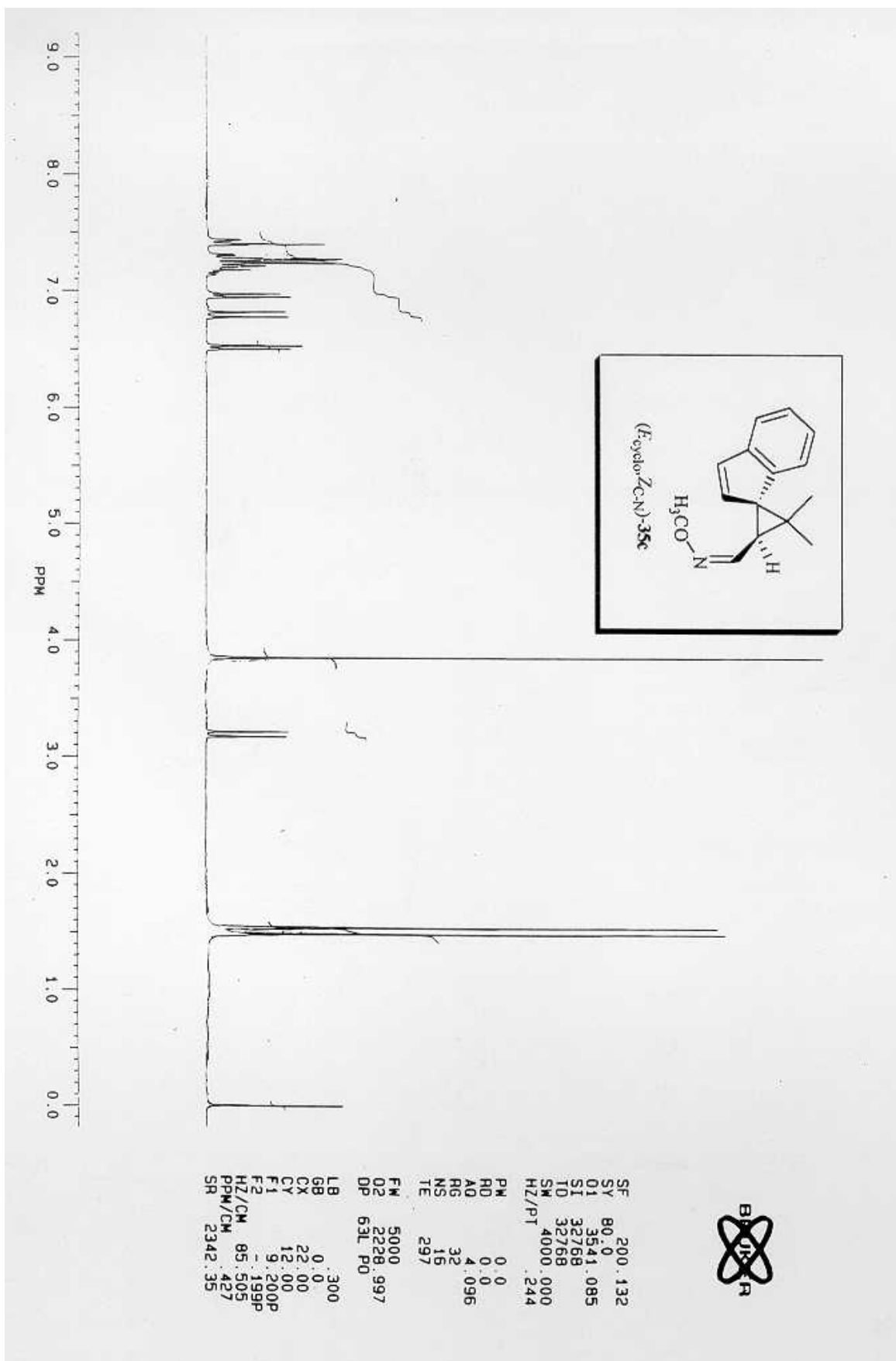


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HZ/CM	85.505
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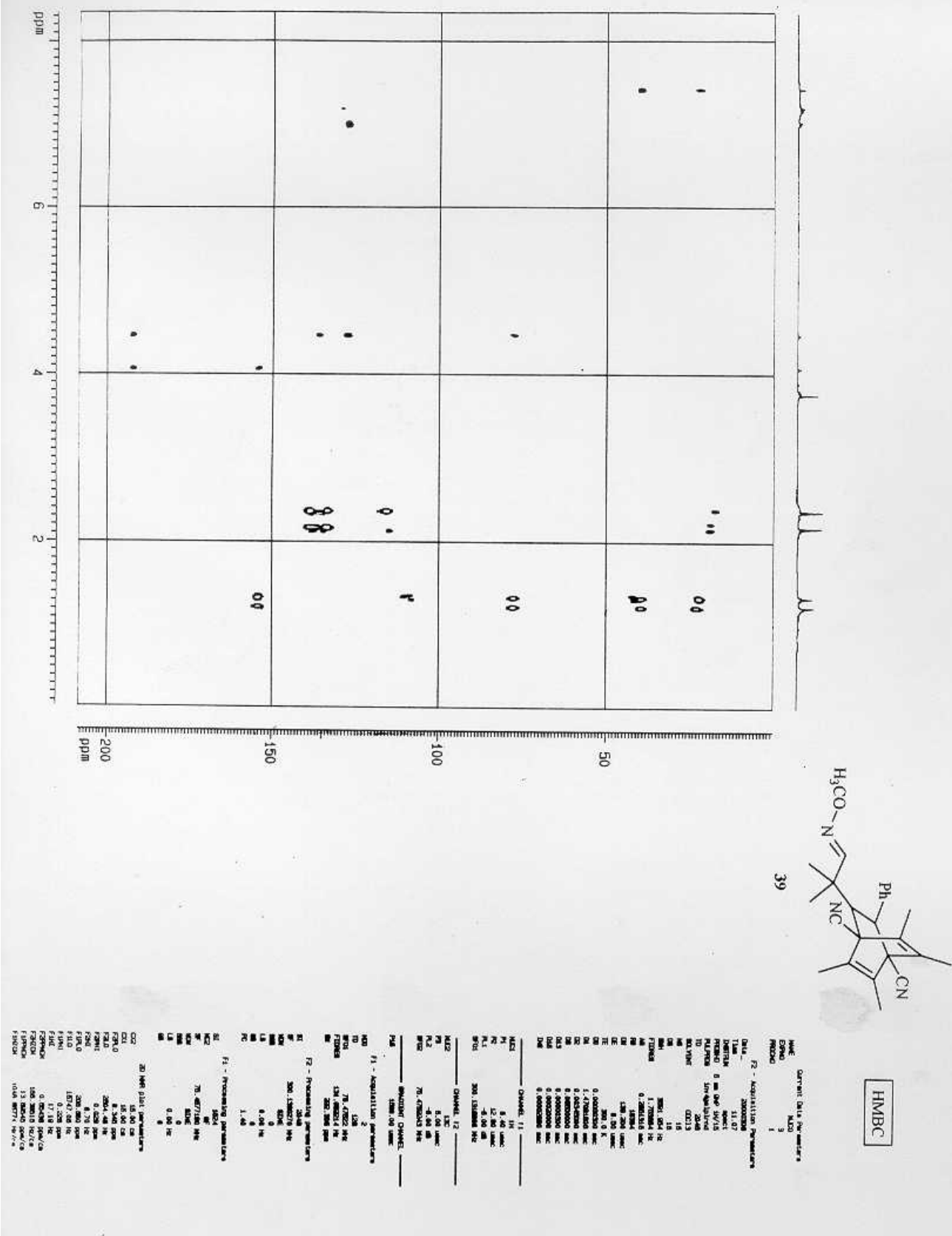








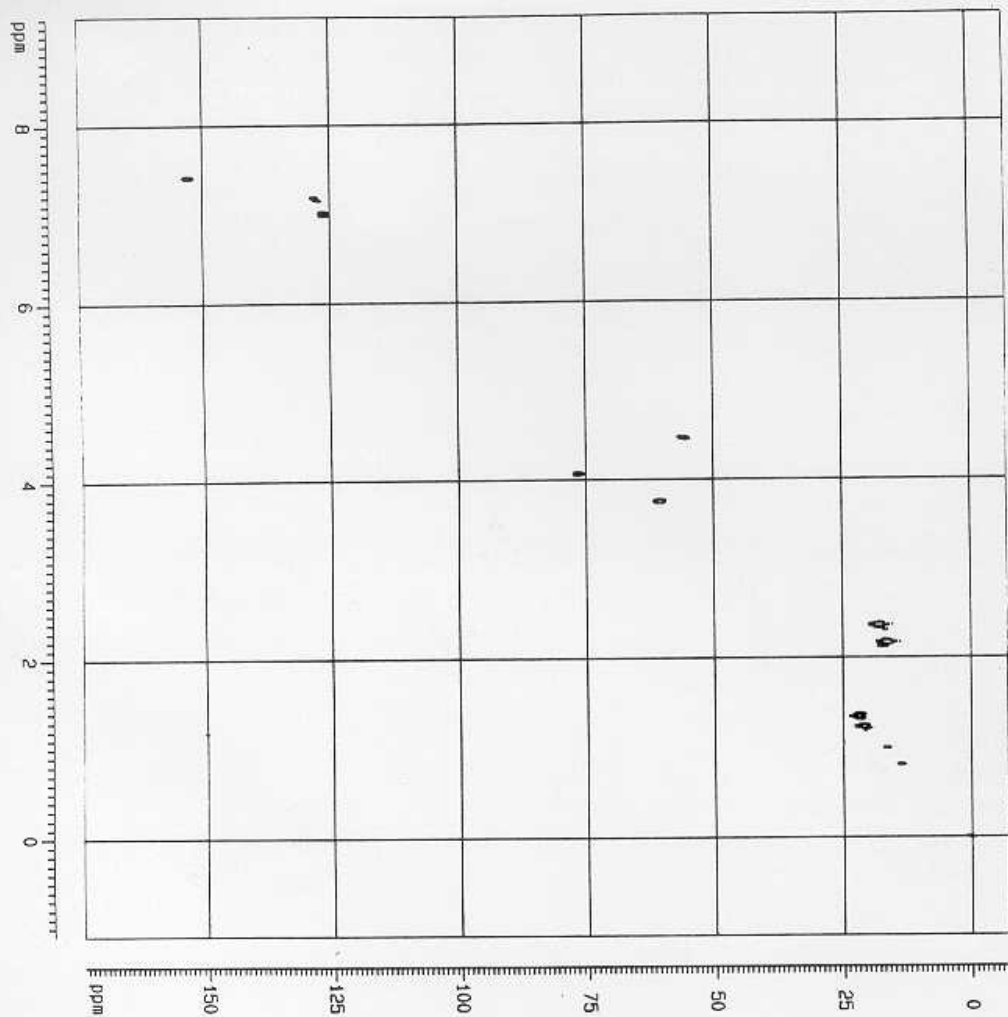
HMBC and HMQC spectra for compound 39.





The figure displays two 2D NMR spectra of poly(2-vinylpyridine) in CDCl₃. The top spectrum is a ¹H-¹³C HMQC spectrum, with the horizontal axis representing ¹H chemical shift (ppm) from 0 to 5 and the vertical axis representing ¹³C chemical shift (ppm) from 0 to 150. The bottom spectrum is a ¹³C-¹³C DQF COSY spectrum, with both axes representing ¹³C chemical shift (ppm) from 0 to 150. Both spectra show cross-peaks indicating scalar coupling between nuclei, with the 1D 1H and 13C NMR spectra visible along the diagonal.

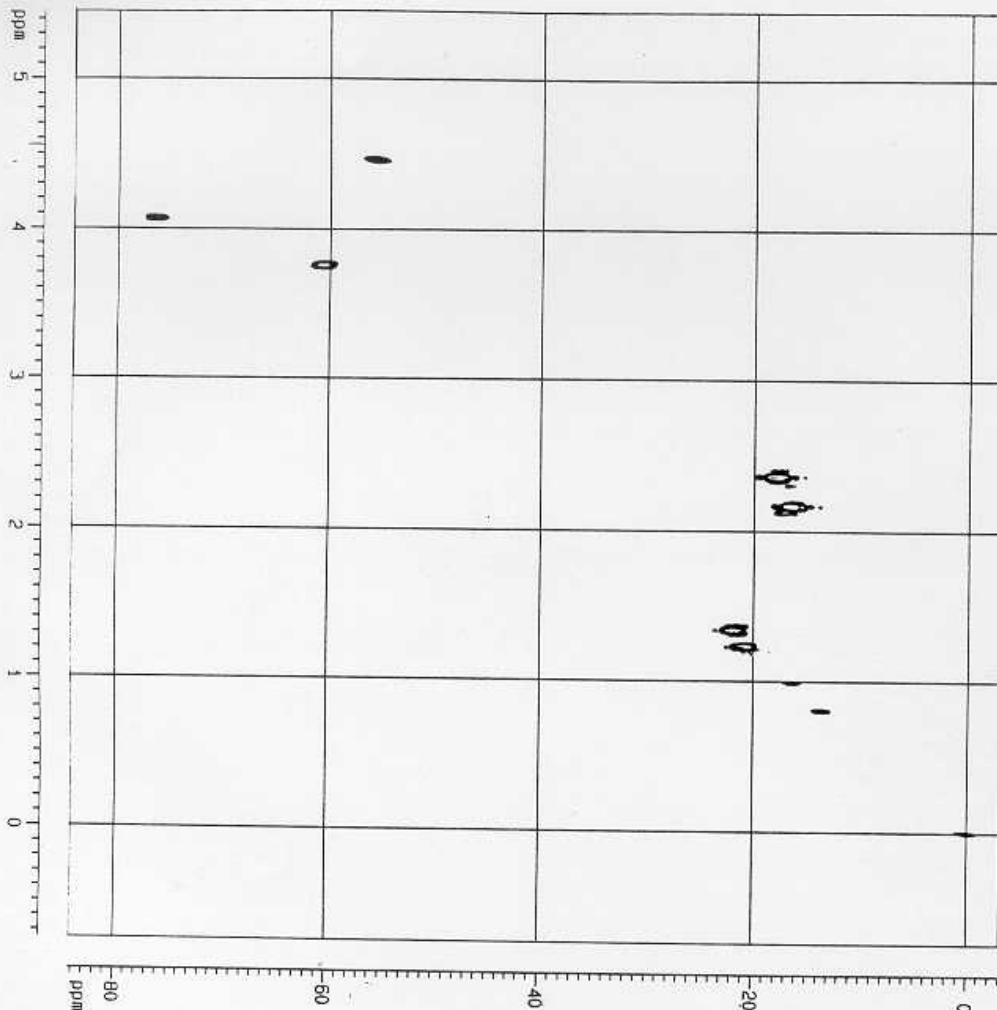
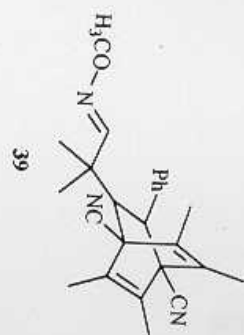
S14



39

Unit	Contract and Period	Unit	Contract and Period
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2	2	2	2
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6	6	6	6
7	7	7	7
8	8	8	8
9	9	9	9
10	10	10	10
11	11	11	11
12	12	12	12
13	13	13	13
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96	96	96	96
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98	98	98	98
99	99	99	99
100	100	100	100

HMQC



X-Ray crystallographic data for compound 36.

TABLE 1. EXPERIMENTAL DATA

CRYSTAL DATA

Empirical formula	C31 H25 N3 O
Formula weight	455.54
Crystal system, space group	Triclinic, P-1
Unit cell determination	Least-squares fit from 58 reflections (2 α < 35)
Unit cell dimensions	a = 14.516(2) Å; α = 112.040(10) b = 10.9320(10) Å; β = 81.04(2) c = 8.0860(10) Å; γ = 99.650(10)
Volume	1166.5(2) Å ³
Formula units per cell, Z,	2
Calculated density	1.297 Mg/m
Absorption coefficient	0.622 mm ⁻¹
F(000)	480
Crystal size	0.22x 0.11 x 0.02 mm

DATA COLLECTION

Four circle diffractometer	Seifert XRS 3000S. Bisecting geometry Graphite oriented monochromator
w-20 scans	
Theta range for data collection	3.10 to 66.82 deg.
Temperature	293(2) K
Wavelength	Cu K , 1.54180 Å
Limiting indices	0 ≤ h ≤ 17, -12 ≤ k ≤ 12, -9 ≤ l ≤ 9
Reflections collected / unique	3913 / 3913 [R(int) = 0.0000]
Observed reflections	2971
2 standard reflections monitored	every 90 min
Intensity variation	none

SOLUTION AND REFINEMENT

Solution	Direct methods
Refinement method	Full-matrix least-squares on F
Data / restraints / parameters	3913 / 0 / 339
Goodness-of-fit on F ²	1.042
Final R indices [I > 2σ(I)]	R1 = 0.0443, wR2 = 0.1122
R indices (all data)	R1 = 0.0606, wR2 = 0.1285
Extinction coefficient	0.0327(18)
Largest diff. peak and hole	0.160 and -0.189 e.Å
w-scheme	w = 1/[σ(F) + (0.0816P)] where P = (F ² + 2F)/3
H- parameters	riding model
Computers and programs	VAX 6410, SIR97[1], SHELX97[2], PARTS[3], Atomic Scattering Factors from International Tables for X-Ray Crystallography

Table 3. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for fin.
 $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	$U(\text{eq})$
N(1)	7972 (1)	582 (2)	-216 (2)	39 (1)
C(2)	8770 (1)	1463 (2)	605 (3)	39 (1)
C(3)	9368 (2)	2218 (2)	-573 (3)	51 (1)
C(4)	8711 (2)	2902 (2)	-1130 (3)	52 (1)
C(5)	7826 (2)	2394 (2)	-1340 (3)	45 (1)
C(6)	7409 (1)	1126 (2)	-1046 (3)	41 (1)
C(7)	6418 (1)	1433 (2)	-54 (3)	38 (1)
C(8)	6198 (2)	2370 (2)	-883 (3)	43 (1)
C(9)	7023 (2)	2947 (2)	-1582 (3)	47 (1)
C(10)	6995 (2)	3885 (3)	-2347 (3)	62 (1)
C(11)	6129 (2)	4221 (3)	-2425 (4)	72 (1)
C(12)	5314 (2)	3645 (3)	-1758 (4)	68 (1)
C(13)	5331 (2)	2719 (2)	-973 (3)	56 (1)
O(14)	8265 (1)	-658 (1)	-1462 (2)	56 (1)
C(15)	7671 (2)	-1714 (2)	-1080 (4)	68 (1)
C(16)	9823 (2)	1185 (3)	-2274 (4)	73 (1)
C(17)	10164 (2)	3222 (3)	375 (4)	71 (1)
C(18)	8375 (1)	2309 (2)	2627 (3)	34 (1)
C(19)	7650 (1)	1341 (2)	3247 (2)	32 (1)
C(20)	7919 (2)	509 (2)	3973 (3)	38 (1)
C(21)	7276 (2)	-474 (2)	4340 (3)	44 (1)
C(22)	6355 (2)	-610 (2)	4001 (3)	46 (1)
C(23)	6075 (2)	232 (2)	3303 (3)	42 (1)
C(24)	6714 (1)	1204 (2)	2898 (2)	32 (1)
C(25)	6408 (1)	2096 (2)	2058 (2)	32 (1)
C(26)	6993 (1)	3457 (2)	2646 (2)	31 (1)
C(27)	7934 (1)	3562 (2)	2921 (2)	33 (1)
C(28)	8433 (2)	4823 (2)	3557 (3)	43 (1)
C(29)	8029 (2)	5951 (2)	3843 (3)	51 (1)
C(30)	7107 (2)	5835 (2)	3522 (3)	53 (1)
C(31)	6593 (1)	4600 (2)	2944 (3)	42 (1)
C(32)	9169 (1)	2681 (2)	3730 (3)	45 (1)
N(33)	9781 (1)	2944 (2)	4579 (3)	68 (1)
C(34)	5413 (1)	2272 (2)	2714 (3)	42 (1)
N(35)	4630 (1)	2353 (2)	3145 (3)	67 (1)

Table 2. GEOMETRICAL FEATURES

a) Bond lengths [Å]

N(1)-O(14)	1.4388(19)
N(1)-C(2)	1.448(3)
N(1)-C(6)	1.450(2)
C(2)-C(3)	1.561(3)
C(2)-C(18)	1.615(3)
C(3)-C(4)	1.515(3)
C(3)-C(17)	1.532(4)
C(3)-C(16)	1.551(3)
C(4)-C(5)	1.321(3)
C(5)-C(9)	1.470(3)
C(5)-C(6)	1.505(3)
C(6)-C(7)	1.558(3)
C(7)-C(8)	1.514(3)
C(7)-C(25)	1.584(3)
C(8)-C(13)	1.395(3)
C(8)-C(9)	1.398(3)
C(9)-C(10)	1.390(3)
C(10)-C(11)	1.384(4)
C(11)-C(12)	1.376(4)
C(12)-C(13)	1.387(4)
O(14)-C(15)	1.422(3)
C(18)-C(32)	1.476(3)
C(18)-C(19)	1.529(2)
C(18)-C(27)	1.533(2)
C(19)-C(20)	1.385(3)
C(19)-C(24)	1.403(2)
C(20)-C(21)	1.390(3)
C(21)-C(22)	1.379(3)
C(22)-C(23)	1.384(3)
C(23)-C(24)	1.391(3)
C(24)-C(25)	1.531(2)
C(25)-C(34)	1.478(3)
C(25)-C(26)	1.527(2)
C(26)-C(31)	1.389(3)
C(26)-C(27)	1.396(2)
C(27)-C(28)	1.392(3)
C(28)-C(29)	1.380(3)
C(29)-C(30)	1.379(3)
C(30)-C(31)	1.377(3)
C(32)-N(33)	1.140(3)
C(34)-N(35)	1.144(3)

b) Bond angles [°]

O(14)-N(1)-C(2)	110.24(14)
O(14)-N(1)-C(6)	110.74(15)
C(2)-N(1)-C(6)	115.83(16)
N(1)-C(2)-C(3)	114.11(17)
N(1)-C(2)-C(18)	105.85(14)
C(3)-C(2)-C(18)	118.85(16)
C(4)-C(3)-C(17)	110.7(2)
C(4)-C(3)-C(16)	109.1(2)
C(17)-C(3)-C(16)	107.4(2)
C(4)-C(3)-C(2)	108.01(17)

C(17)-C(3)-C(2)	113.3(2)
C(16)-C(3)-C(2)	108.32(18)
C(5)-C(4)-C(3)	120.9(2)
C(4)-C(5)-C(9)	131.4(2)
C(4)-C(5)-C(6)	122.95(19)
C(9)-C(5)-C(6)	104.89(18)
N(1)-C(6)-C(5)	116.93(18)
N(1)-C(6)-C(7)	115.70(16)
C(5)-C(6)-C(7)	103.66(16)
C(8)-C(7)-C(6)	100.22(16)
C(8)-C(7)-C(25)	112.40(15)
C(6)-C(7)-C(25)	113.64(15)
C(13)-C(8)-C(9)	120.0(2)
C(13)-C(8)-C(7)	129.3(2)
C(9)-C(8)-C(7)	110.72(17)
C(10)-C(9)-C(8)	120.9(2)
C(10)-C(9)-C(5)	130.3(2)
C(8)-C(9)-C(5)	108.82(18)
C(11)-C(10)-C(9)	118.4(3)
C(12)-C(11)-C(10)	120.9(3)
C(11)-C(12)-C(13)	121.4(2)
C(12)-C(13)-C(8)	118.4(3)
C(15)-O(14)-N(1)	108.52(16)
C(32)-C(18)-C(19)	108.08(15)
C(32)-C(18)-C(27)	108.86(16)
C(19)-C(18)-C(27)	109.42(14)
C(32)-C(18)-C(2)	106.80(15)
C(19)-C(18)-C(2)	105.70(14)
C(27)-C(18)-C(2)	117.60(15)
C(20)-C(19)-C(24)	119.72(17)
C(20)-C(19)-C(18)	121.04(16)
C(24)-C(19)-C(18)	118.91(16)
C(19)-C(20)-C(21)	120.83(19)
C(22)-C(21)-C(20)	119.56(19)
C(21)-C(22)-C(23)	120.07(19)
C(22)-C(23)-C(24)	121.05(19)
C(23)-C(24)-C(19)	118.75(17)
C(23)-C(24)-C(25)	120.82(16)
C(19)-C(24)-C(25)	120.43(15)
C(34)-C(25)-C(26)	108.68(15)
C(34)-C(25)-C(24)	107.74(15)
C(26)-C(25)-C(24)	110.16(14)
C(34)-C(25)-C(7)	104.54(15)
C(26)-C(25)-C(7)	112.38(15)
C(24)-C(25)-C(7)	113.00(14)
C(31)-C(26)-C(27)	119.47(16)
C(31)-C(26)-C(25)	120.56(16)
C(27)-C(26)-C(25)	119.96(15)
C(28)-C(27)-C(26)	118.62(17)
C(28)-C(27)-C(18)	121.51(16)
C(26)-C(27)-C(18)	119.77(15)
C(29)-C(28)-C(27)	121.35(19)
C(30)-C(29)-C(28)	119.61(19)
C(31)-C(30)-C(29)	119.9(2)
C(30)-C(31)-C(26)	121.00(19)
N(33)-C(32)-C(18)	178.7(2)
N(35)-C(34)-C(25)	176.5(2)

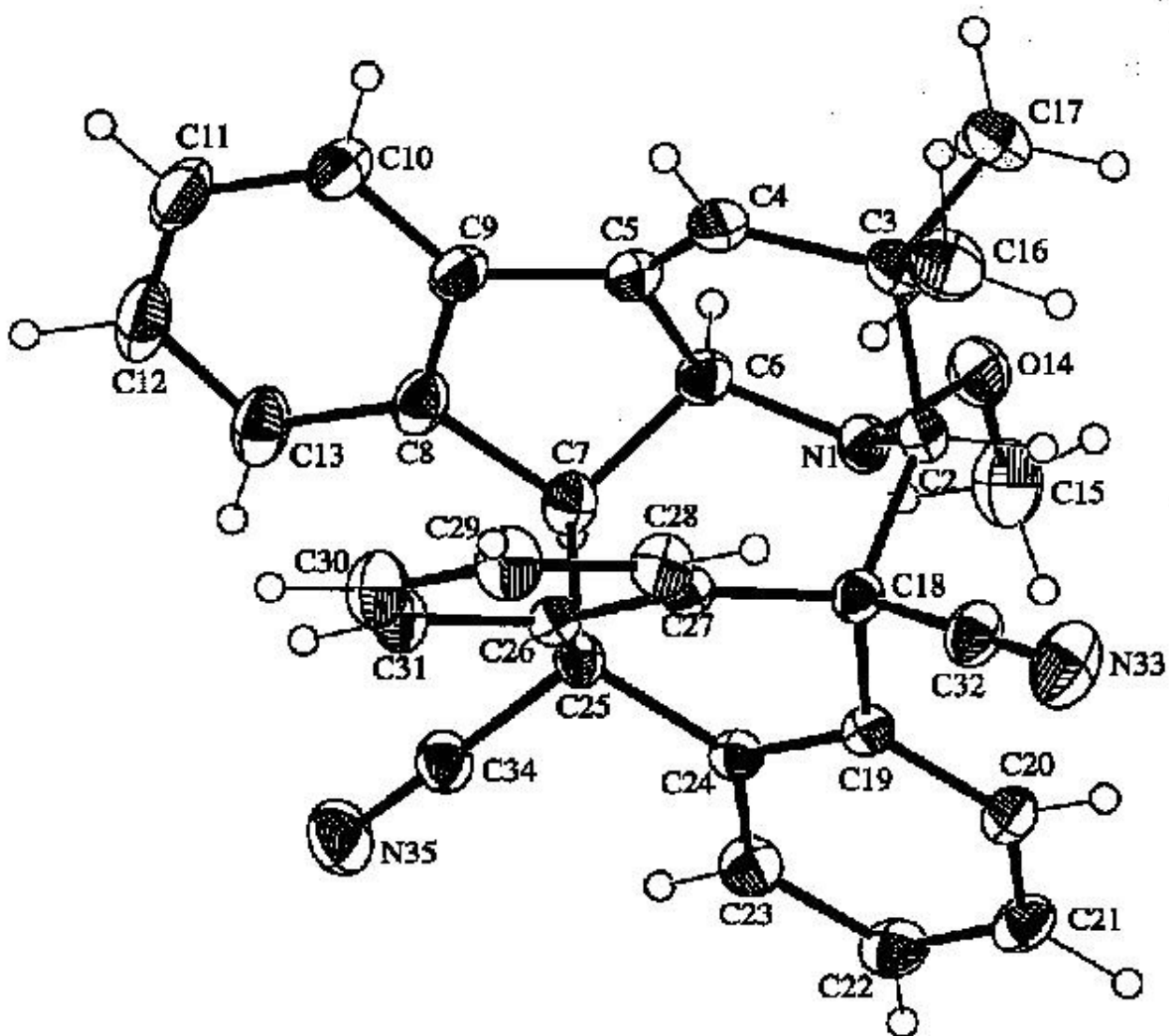
Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for fin.
The anisotropic displacement factor exponent takes the form:
 $-2 \pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$

	U11	U22	U33	U23	U13	U12
N(1)	48(1)	34(1)	34(1)	8(1)	-2(1)	16(1)
C(2)	34(1)	44(1)	43(1)	20(1)	3(1)	12(1)
C(3)	40(1)	63(1)	58(1)	35(1)	14(1)	17(1)
C(4)	55(1)	59(1)	52(1)	35(1)	13(1)	18(1)
C(5)	53(1)	52(1)	35(1)	22(1)	5(1)	19(1)
C(6)	52(1)	42(1)	30(1)	11(1)	-5(1)	13(1)
C(7)	41(1)	35(1)	39(1)	9(1)	-13(1)	5(1)
C(8)	55(1)	41(1)	36(1)	9(1)	-16(1)	12(1)
C(9)	65(1)	51(1)	33(1)	17(1)	-3(1)	21(1)
C(10)	85(2)	67(2)	49(1)	32(1)	3(1)	30(1)
C(11)	106(2)	76(2)	54(2)	31(1)	-11(2)	43(2)
C(12)	82(2)	70(2)	63(2)	17(1)	-29(1)	33(2)
C(13)	60(1)	55(1)	56(2)	12(1)	-27(1)	14(1)
O(14)	73(1)	44(1)	49(1)	7(1)	3(1)	27(1)
C(15)	79(2)	39(1)	83(2)	12(1)	-16(2)	10(1)
C(16)	68(2)	95(2)	71(2)	47(2)	34(1)	43(2)
C(17)	43(1)	86(2)	95(2)	54(2)	7(1)	0(1)
C(18)	30(1)	37(1)	38(1)	17(1)	-6(1)	2(1)
C(19)	35(1)	31(1)	30(1)	11(1)	-4(1)	3(1)
C(20)	49(1)	33(1)	34(1)	12(1)	-8(1)	7(1)
C(21)	65(1)	32(1)	37(1)	14(1)	-7(1)	7(1)
C(22)	56(1)	35(1)	49(1)	21(1)	2(1)	0(1)
C(23)	39(1)	38(1)	49(1)	18(1)	-3(1)	-2(1)
C(24)	35(1)	29(1)	32(1)	11(1)	-2(1)	2(1)
C(25)	26(1)	33(1)	37(1)	12(1)	-5(1)	2(1)
C(26)	33(1)	31(1)	31(1)	11(1)	-4(1)	2(1)
C(27)	33(1)	33(1)	33(1)	13(1)	-4(1)	2(1)
C(28)	40(1)	37(1)	51(1)	16(1)	-9(1)	-5(1)
C(29)	53(1)	32(1)	64(2)	15(1)	-12(1)	-3(1)
C(30)	60(1)	30(1)	69(2)	15(1)	-12(1)	6(1)
C(31)	40(1)	35(1)	52(1)	14(1)	-8(1)	6(1)
C(32)	39(1)	46(1)	57(1)	25(1)	-13(1)	-1(1)
N(33)	50(1)	74(1)	92(2)	42(1)	-35(1)	-12(1)
C(34)	34(1)	43(1)	52(1)	21(1)	-4(1)	4(1)
N(35)	39(1)	80(1)	91(2)	43(1)	3(1)	11(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for fin.

	x	y	z	U(eq)
H(2)	9193(9)	876(13)	726(4)	47
H(4)	8954(6)	3760(20)	-1344(6)	62
H(6)	7303(3)	415(15)	-2300(30)	49
H(7)	5968(10)	613(18)	-378(7)	46
H(10)	7591(17)	4309(12)	-2833(14)	75
H(11)	6098(2)	4840(20)	-2932(18)	87
H(12)	4673(18)	3909(8)	-1844(4)	82
H(13)	4731(16)	2308(11)	-480(14)	67
H(15A)	7685(10)	-1597(11)	210(20)	103
H(15B)	7898(9)	-2588(15)	-1880(20)	103
H(15C)	7008(11)	-1706(12)	-1300(30)	103
H(16A)	9327(9)	500(16)	-2899(18)	110
H(16B)	10271(13)	756(16)	-1922(7)	110
H(16C)	10163(13)	1638(8)	-3087(19)	110
H(17A)	10566(11)	3583(16)	-488(16)	106
H(17B)	10548(11)	2773(9)	840(20)	106
H(17C)	9894(5)	3959(16)	1390(20)	106
H(20)	8596(15)	618(3)	4241(6)	46
H(21)	7481(5)	-1082(14)	4849(12)	52
H(22)	5910(12)	-1284(18)	4247(7)	56
H(23)	5422(16)	141(3)	3093(6)	51
H(28)	9083(16)	4911(3)	3805(7)	52
H(29)	8400(9)	6840(20)	4278(11)	61
H(30)	6825(8)	6610(20)	3701(6)	64
H(31)	5914(15)	4523(3)	2732(6)	50

ORTEP Diagram



Cell-packing Diagram

