

Supporting Material to Accompany “Ketene-Forming Eliminations from Aryl phenylacetates Promoted by R₂NH/R₂NH₂⁺ in MeCN(aq). Mechanistic Borderline between E2 and E1cb

Bong Rae Cho,^{*1)} Hyun Cheol Jeong,¹⁾ Yoon Je Seung,²⁾ and Sang Yong Pyun ^{*2)}

¹⁾Department of Chemistry, Korea University, 1-Anamdong, Seoul, 136-701, Korea, and

²⁾Department of Chemistry, Pukyong National University, Pusan 608-737, Korea

Table S1. Observed rate constants for eliminations from 4-NO₂-C₆H₄CH₂CO₂C₆H₃-2-Cl-4-NO₂ (**1c**)^a promoted by R₂NH/R₂NH₂⁺ in 70 mol % MeCN (aq)^{b,c} at 25.0 °C.

10 ³ [base], M	10k _{obs} , s ⁻¹ ^{d,e} when R ₂ NH is			
	Bz(i-Pr)NH	i-Bu ₂ NH	i-Pr ₂ NH	2,6-DMP ^f
0.840	0.165	0.350	0.239	0.454
1.00	0.207	0.427	0.305	0.538
2.00	0.369	0.831	0.612	1.13
4.00	0.669	1.56	1.23	2.31
6.00	1.01	2.34	1.88	3.29
8.00	1.32	2.78	2.50	4.13

^a [Substrate] = (2.0–4.0) × 10⁻⁵ M. ^b [R₂NH]/[R₂NH₂⁺] = 1.0. ^c μ = 0.10 M (Bu₄N⁺Br⁻). ^d

Average of three or more kinetic runs. ^e Estimated uncertainty, ± 5 %. ^f cis-2,6-Dimethylpiperidine.

Table S2. Observed rate constants for eliminations from 4-NO₂-C₆H₄CH₂CO₂C₆H₃-2-X-4-NO₂ (**1d-e**)^a promoted by R₂NH/R₂NH₂⁺ in 70 mol % MeCN (aq)^{b-d} at 25.0 °C.

X	10 ² k _{obs} , s ⁻¹ ^{e,f} when R ₂ NH is			
	Bz(i-Pr)NH	i-Bu ₂ NH	i-Pr ₂ NH	2,6-DMP ^g
CF ₃ (1d)	0.165	0.350	0.239	0.454
NO ₂ (1e)	0.207	0.427	0.305	0.538

^a [Substrate] = (2.0–4.0) × 10⁻⁵ M. ^b [R₂NH]/[R₂NH₂⁺] = 1.0. ^c [Base] = 8.00 × 10⁻⁴ M. ^d

μ = 0.10 M (Bu₄N⁺Br⁻). ^e Average of three or more kinetic runs. ^f Estimated uncertainty, ± 5 %. ^g cis-2,6-Dimethylpiperidine.

Table S3. Observed rate constants for eliminations from 2,4-(NO₂)₂-C₆H₃CH₂CO₂C₆H₄-4-NO₂ (**2a**)^a promoted by R₂NH/R₂NH₂⁺ in 70 mol % MeCN (aq)^{b,c} at 25.0 °C.

10 ³ [base], M	10 ² k _{obs} , s ⁻¹ ^{d,e} when R ₂ NH is		
	Bz(i-Pr)NH	i-Bu ₂ NH	2,6-DMP ^f
0.600	2.00	4.17	4.09
0.800	2.29	4.87	5.02
1.00	2.46	5.88	6.78
1.40	2.75	7.11	9.82
1.80	3.03	8.11	12.6
2.00	3.10	8.77	14.2
3.00	3.47	10.1	20.0
4.00	3.62	10.9	24.7
6.00	3.83	14.7	29.6
8.00	3.98	18.3	36.0
10.0	3.96	19.3	41.2
20.0	4.20	19.1	49.8

See footnotes in Table S1.

Table S4. Observed rate constants for eliminations from 4-NO₂-C₆H₄CH₂CO₂C₆H₃-4-NO₂ (**1a**)^a promoted by R₂NH/R₂NH₂⁺ in 70 mol % MeCN (aq)^{b,c} at 25.0 °C.

10 ³ [base], M	10 ² k _{obs} , s ⁻¹ ^{d,e} when R ₂ NH is		
	Bz(i-Pr)NH	i-Bu ₂ NH	2,6-DMP ^f
0.960	0.778	1.83	3.00
1.20	0.878	2.16	3.53
2.40	1.21	3.43	7.13
4.80	1.56	4.74	12.7
7.20	1.80	5.88	17.3
9.60	1.92	6.62	21.4
12.0	2.03	6.82	24.0
24.0	2.39	8.84	36.1
36.0	2.62	10.6	43.8
48.0	2.83	10.7	48.8
60.0	3.06	11.3	53.5

See footnotes in Table S1.

Table S5. Observed rate constants for eliminations from 4-NO₂-C₆H₄CH₂CO₂C₆H₃-4-NO₂ (**1a**)^a promoted by Bz(i-Pr)NH/Bz(i-Pr)NH₂⁺ in 70 mol % MeCN (aq)^b at 25.0 °C.

10 ³ [base] ^c , M	10 ² k _{obs} , s ⁻¹ ^{d,e}	10 ³ [base] ^f , M	10 ² k _{obs} , s ⁻¹ ^{d,e}
0.480	0.392	0.960	0.938
0.600	0.432	1.20	1.06
1.20	0.625	2.40	1.66
2.40	0.785	4.80	2.37
3.60	0.897	7.20	2.80
4.80	0.967	9.60	3.08
6.00	1.02	12.0	3.33
12.0	1.18	24.0	4.08
18.0	1.32	36.0	4.65
24.0	1.40	48.0	4.99
30.0	1.49	60.0	5.07

^a [Substrate] = (2.0–4.0) × 10⁻⁵ M. ^b μ = 0.10 M (Bu₄N⁺Br⁻). ^c [R₂NH]/[R₂NH₂⁺] = 0.5. ^d Average of three or more kinetic runs. ^e Estimated uncertainty, ± 5 %. ^f [R₂NH]/[R₂NH₂⁺] = 2.0.

Table S6. Observed rate constants for eliminations from 4-NO₂-C₆H₄CH₂CO₂C₆H₃-2-MeO-4-NO₂ (**1b**)^a promoted by R₂NH/R₂NH₂⁺ in 70 mol % MeCN (aq)^{b,c} at 25.0 °C.

10 ³ [base], M	10 ² k _{obs} , s ⁻¹ ^{d,e} when R ₂ NH is		
	Bz(i-Pr)NH	i-Bu ₂ NH	2,6-DMP ^f
0.960	0.490	1.28	2.07
1.20	0.570	1.50	2.61
2.40	0.765	2.25	5.04
4.80	0.950	3.18	8.67
7.20	1.07	3.72	11.8
9.60	1.16	4.11	14.0
12.0	1.23	4.43	15.6
24.0	1.46	5.37	22.3
36.0	1.60	5.97	29.0
48.0	1.77	6.57	30.3
60.0	1.92	6.97	31.8

See footnotes in Table S1.

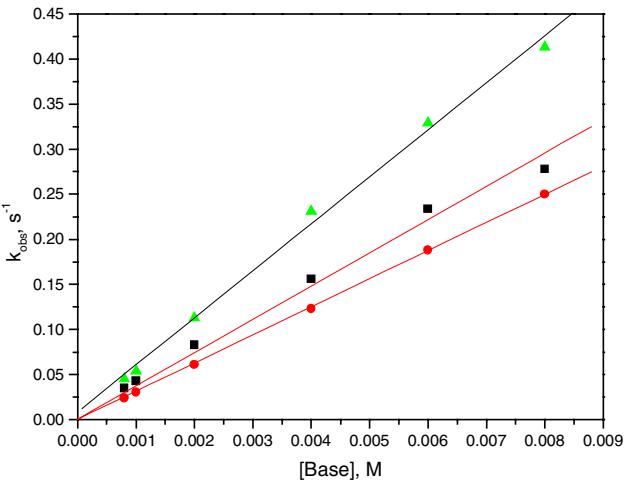


Figure S1. Plots of k_{obs} vs base concentration for eliminations from **1c** promoted by $\text{R}_2\text{NH}/\text{R}_2\text{NH}_2^+$ in 70 mol % MeCN(aq) at 25 °C, $[\text{R}_2\text{NH}]/[\text{R}_2\text{NH}_2^+] = 1.0$, $\mu = 0.10 \text{ M}$ ($\text{Bu}_4\text{N}^+\text{Br}^-$). [$\text{R}_2\text{NH} = \text{i-Bu}_2\text{NH}$ (\blacktriangle), $\text{i-Pr}_2\text{NH}$ (\blacksquare), 2,6-DMP (\bullet)].

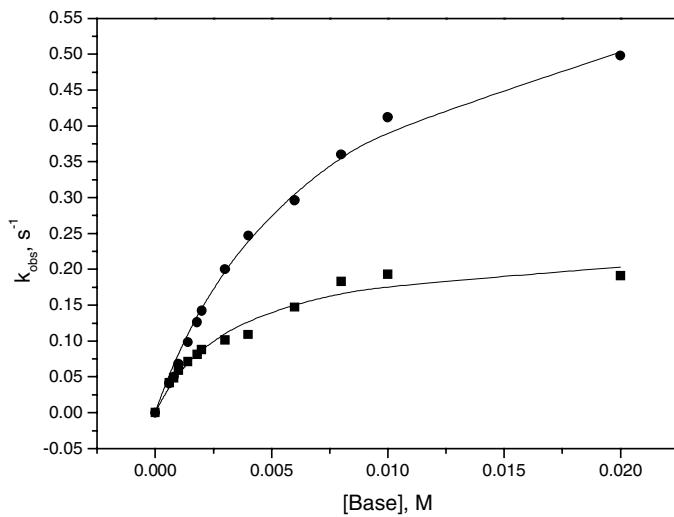


Figure S2. Dependence of k_{obs} vs base concentration for elimination from **2a** promoted by 2,6-DMP/2,6-DMPh⁺ (\bullet) and i-Bu₂NH/i-Bu₂NH²⁺ (\blacksquare) in 70 mol % MeCN(aq) at 25 °C, $[\text{R}_2\text{NH}]/[\text{R}_2\text{NH}_2^+] = 1.0$, $\mu = 0.10 \text{ M}$ ($\text{Bu}_4\text{N}^+\text{Br}^-$).

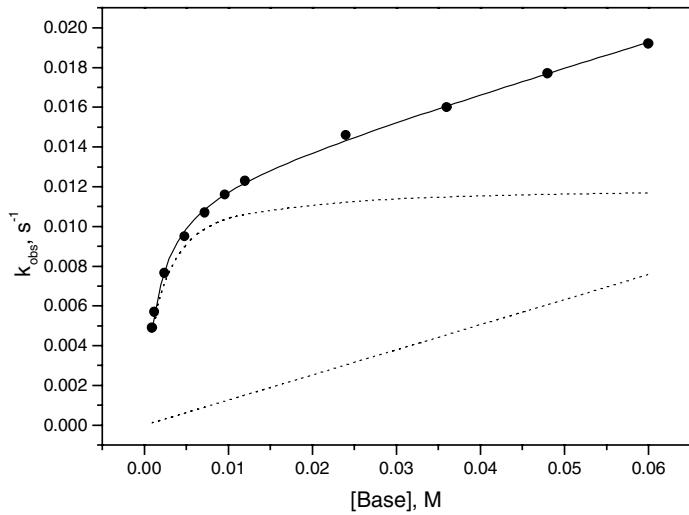


Figure S3. Dependence of k_{obs} vs base concentration for elimination from **1a** promoted by $\text{Bz(i-Pr)NH/Bz(i-Pr)NH}_2^+$ in 70 mol % MeCN(aq) at 25 °C, $[\text{Bz(i-Pr)NH}]/[\text{Bz(i-Pr)NH}_2^+] = 1.0$, $\mu = 0.10$ M ($\text{Bu}_4\text{N}^+\text{Br}^-$).

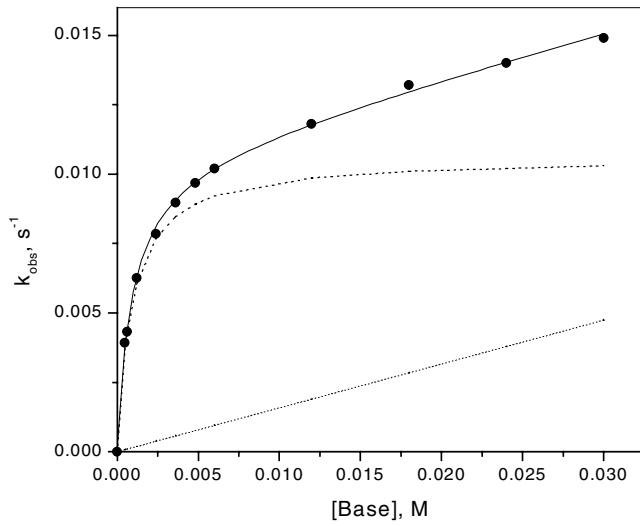


Figure S4. Dependence of k_{obs} vs base concentration for elimination from **1a** promoted by $\text{Bz(i-Pr)NH/Bz(i-Pr)NH}_2^+$ in 70 mol % MeCN(aq) at 25 °C, $[\text{Bz(i-Pr)NH}]/[\text{Bz(i-Pr)NH}_2^+] = 0.5$, $\mu = 0.10$ M ($\text{Bu}_4\text{N}^+\text{Br}^-$)

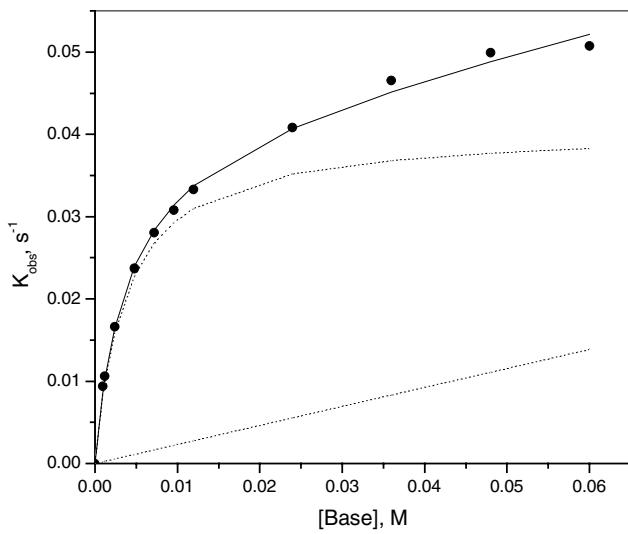


Figure S5. Dependence of k_{obs} vs base concentration for elimination from **1a** promoted by $Bz(i\text{-}Pr)NH/Bz(i\text{-}Pr)NH_2^+$ in 70 mol % MeCN(aq) at 25 °C, $[Bz(i\text{-}Pr)NH]/[Bz(i\text{-}Pr)NH_2^+] = 2.0$, $\mu = 0.10$ M ($Bu_4N^+Br^-$).

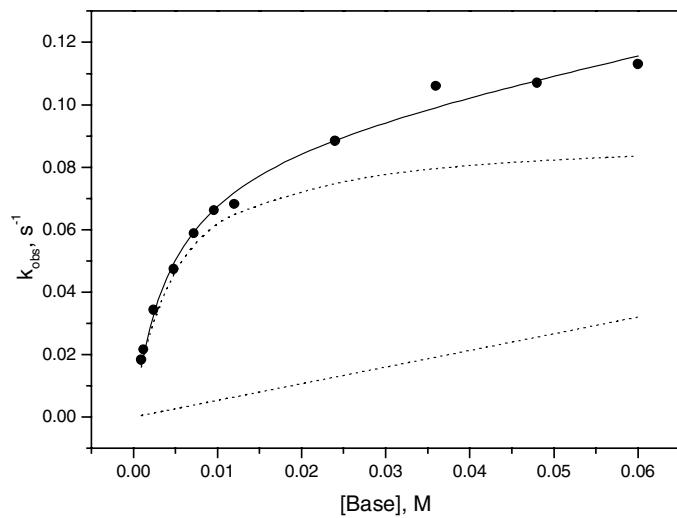


Figure S6. Dependence of k_{obs} vs base concentration for the elimination **1a** promoted by $i\text{-}Bu_2NH/i\text{-}Bu_2NH_2^+$ in 70 mol % MeCN(aq) at 25 °C, $[i\text{-}Bu_2NH]/[i\text{-}Bu_2NH_2^+] = 1.0$, $\mu = 0.10$ M ($Bu_4N^+Br^-$).

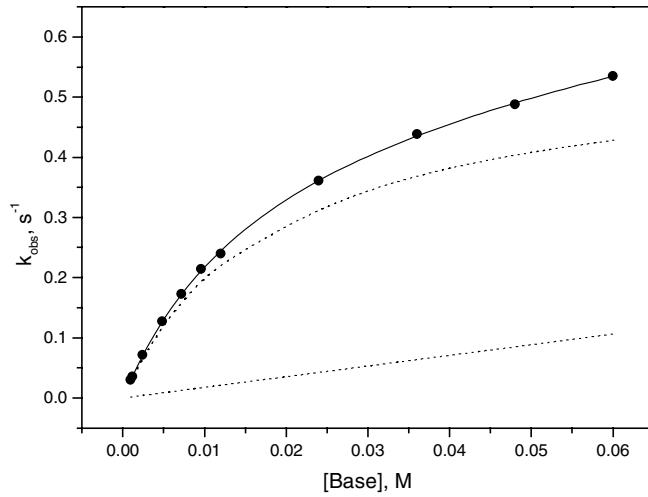


Figure S7. Dependence of k_{obs} vs base concentration for elimination from **1a** promoted by 2,6-DMP/2,6-DMPH⁺ in 70 mol % MeCN(aq) at 25 °C, [2,6-DMP]/[2,6-DMPH⁺] = 1.0, μ = 0.10 M ($\text{Bu}_4\text{N}^+\text{Br}^-$).

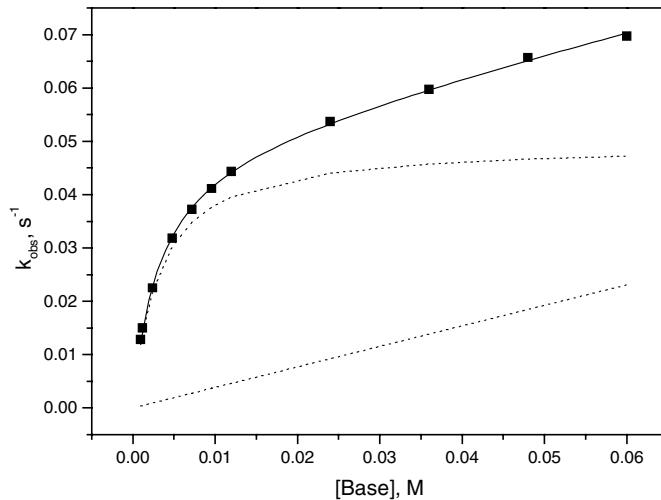


Figure S8. Dependence of k_{obs} vs base concentration for the elimination from **1b** promoted by i-Bu₂NH/ i-Bu₂NH⁺ in 70 mol % MeCN(aq) at 25 °C, [i-Bu₂NH]/[i-Bu₂NH⁺] = 1.0, μ = 0.10 M ($\text{Bu}_4\text{N}^+\text{Br}^-$).

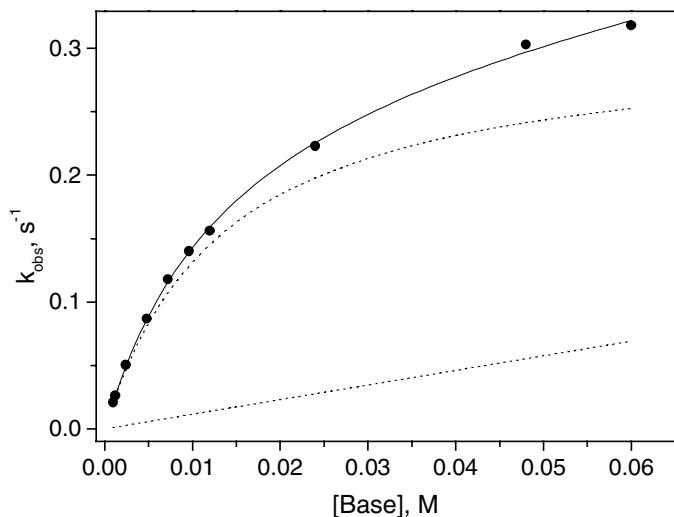


Figure S9. Dependence of k_{obs} vs base concentration for the elimination from **1b** promoted by 2,6-DMP/2,6-DMPH⁺ in 70 mol % MeCN(aq) at 25 °C, [2,6-DMP]/[2,6-DMPH⁺] = 1.0, μ = 0.10 M ($Bu_4N^+Br^-$).