

Table S1. Dihedral and plane angles (in degrees) and pyramidalization (in Å) of Concerted Transition Structures.^a

	<i>endo-8</i>	<i>endo-16</i>	<i>exo-16</i>
α	85.3	97.4	101.5
β	99.1	119.5	125.9
γ	1.0	18.4	22.5
C-N-O	127.6	121.3	122.0
N pyr. ^b	0.034	0.145	0.133
C pyr. ^b	0.016	0.146	0.151

^a See Scheme 3.

^b pyr. = pyramidalization

Table S2. B3LYP/6-31G* Singlet (¹E) and Triplet (³E) Electronic Energy, $\langle S^2 \rangle$ values, Correction to Enthalpy (δH), Entropy (S), Correction to Gibbs Free Energy (δG) of Reactants, Intermediates, Transition Structures and Products for the Reaction of **1** with **7**.

molecule	¹ E ^b	¹ $\langle S^2 \rangle$	³ E ^b	³ $\langle S^2 \rangle$	$\delta H^{a,b}$	$S^{a,c}$	$\delta G^{a,b}$
1	-169.7934851	0.0	-	-	0.049610	59.64	0.021240
7	-154.6754616	0.0	-	-	0.065936	63.23	0.035894
<i>endo-8</i>	-324.4740795	0.0	-	-	0.117527	90.26	0.074640
<i>endo-9</i>	-324.472995	0.0	-	-	0.116656	81.23	0.078061
<i>exo-9</i>	-324.4674128	0.0	-	-	0.116778	81.17	0.078210
10a	-324.4626170	1.11	-324.4566932	2.06	0.113352	91.31	0.071223
10b	-324.463247	1.03	-324.4541564	2.06	0.113529	91.02	0.074000
11a	-324.5249250	1.03	-324.5248508	2.03	0.118555	83.34	0.078959
11b	-324.5254537	1.03	-324.5250283	2.03	0.118627	83.62	0.078896
12a	-324.4758657	1.03	-	-	0.116474	84.25	0.076442
12b	-324.4748348	1.02	-	-	0.116554	85.75	0.075813
(i)-13	-324.5897554	0.0	-	-	0.122582	74.03	0.087407
(o)-13	-324.581697	0.0	-	-	0.122452	74.72	0.086948

^a Harmonic approximation assumed; standard state (298 K) of the fugacity scale (pure perfect gas at 1 atm); δH , δG are the kinetic contributions to molar enthalpy and free enthalpy (to be added to electronic energy); S is the molar entropy.

^b Energies in Hartree.

^c Entropies in cal/mol K.

Table S3. B3LYP/6-31G* Singlet (^1E) and Triplet (^3E) Electronic Energy, $\langle \text{S}^2 \rangle$ values, Correction to Enthalpy (δH), Entropy (S), Correction to Gibbs Free Energy (δG) of Reactants, Intermediates, Transition Structures and Products for the Reaction of **1** with **15**.

	$^1\text{E}^b$	$^1\langle \text{S}^2 \rangle$	$^3\text{E}^b$	$^3\langle \text{S}^2 \rangle$	$\delta\text{H}^{a,b}$	$\text{S}^{a,c}$	$\delta\text{G}^{a,b}$
1	-169.7934851	0.0	-	-	0.049610	59.64	0.021240
15	-308.3603354	0.0	-	-	0.117173	75.94	0.081092
<i>endo-16</i>	-478.150730	0.0	-	-	0.167957	92.86	0.123837
<i>exo-16</i>	-478.1470252	0.0	-	-	0.167917	92.86	0.123797
17a	-478.1353437	0.430	-478.1094553	2.02	0.166569	97.57	0.12208
17b	-478.1396366	0.298	-478.108718	2.02	0.166956	98.09	0.120352
17c	-478.1374139	0.720	-478.122245	2.02	0.166015	98.08	0.119413
18a	-478.179105	1.03	-478.1788333	2.04	0.168973	96.64	0.123056
18b	-478.1802946	1.01	-478.1790316	2.04	0.169132	96.59	0.123237
18c	-478.1794166	1.03	-478.1788333	2.04	0.169060	96.86	0.123039
18d	-478.1791144	1.02	-478.1790316	2.04	0.169029	96.43	0.123210
(i)- 19	-478.246300	0.0	-	-	0.172850	87.39	0.131328
(o)- 19	-478.2392802	0.0	-	-	0.172655	88.08	0.130806

^{a,b,c} See footnotes in Table S2.

Figure S1 Optimized (B3LYP/6-31G*) benzocyclobutadiene (**15**) and nitrene+benzocyclobutadiene transition structures (**16-18**) (bond length in Angstrom).

