RHODIUM-CATALYZED ENANTIOSELECTIVE REDUCTIVE ALDOL REACTION.

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Supplementary Material

General. Infrared spectra were recorded on a Nicolet Magna 560 spectrometer, v_{max} in cm⁻¹. Bands are characterized as broad (br), strong (s), medium (m), and weak (w). ¹H NMR spectra were recorded on Varian Gemini (300 MHz) and Bruker (400 MHz) spectrometers. Chemical shifts are reported in ppm from tetramethylsilane with the solvent resonance as the internal standard (CDCl₃: 7.24 ppm). Data are reported as follows: chemical shift, integration, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, br = broad, m = multiplet), coupling constants (Hz) and assignment. ¹³C NMR were recorded on a Varian Gemini 300 (75 MHz) spectrometer with complete proton decoupling. Chemical shifts are reported in ppm from tetramethylsilane with the solvent as the internal standard (CDCl₃: 77.0 ppm). Microanalyses were preformed by Robertson Microlit.

Liquid chromatography was performed using forced flow (flash chromatography) of the indicated solvent system on Sigma silica gel 60 (SiO₂, 230-400 mesh). Thin layer chromatography was performed on EM Science 0.25 mm silica gel 60 plates. Visualization was achieved with phosphomolybdic acid in ethanol, potassium permanganate in water, or vanillin in sulfuric acid, each followed by heating. Analytical gas-liquid chromatography (GLC) was performed on a Hewlett-Packard 6890 Series chromatograph equipped with a split mode capillary injection system, the indicated chiral GLC column, a flame ionization detector and using helium as the carrier gas.

All reactions were conducted in oven and flame dried glassware under an inert atmosphere of dry nitrogen. Dichloroethane was sequentially washed with concentrated H₂SO₄, water, aqueous Na₂CO₃, and water, then dried with MgSO₄ and fractionally distilled from CaH₂. Chloro(1,5-cyclooctadiene)rhodium (I) dimer and (R)-BINAP were purchased from Strem Chemical Company. All other reagents were purchased from either Lancaster or Aldrich Chemical Companies.

Representative procedure for catalytic reductive aldol reaction. A 10 mL flame-dried round bottom flask was charged with 10.0 mg of chloro(1,5-cyclooctadiene)rhodium (I) dimer (0.02 mmol), 33.0 mg (R)-BINAP (0.053 mmol) and 500 μL of dichloroethane. The resulting solution was stirred at room temperature for one hour. After one hour, 481 μL of dichloroethane and 174 μL of diethylmethylsilane (0.97 mmol) were added to the mixture and the reaction vessel stirred for 30 minutes. Next, 1.15 mL of stock benzaldehyde/phenyl acrylate solution (0.70 M in aldehyde and 0.84 M in acrylate, 0.81 mmol aldehyde, 0.97 mmol acrylate) was added dropwise to the solution. The vessel was then sealed and allowed to stir for 24 hours. Solvent was then evaporated from the reaction and 1 mL each of THF, MeOH, and 4N HCl were added. This mixture was stirred at room temperature for an additional 30 minutes. Ethyl acetate was then used to extract the product (3 x 7 mL). The combined organic layers were washed with a saturated aqueous sodium bicarbonate solution (2 x 20 mL), dried over MgSO₄, and filtered. The solvent was removed by rotary evaporation to yield crude product which was purified *via*

flash chromatography (9:1 then 5:1 hexanes:ethyl acetate) to yield 151 mg (0.59 mmol, 73 % yield) of a 1.5:1 ratio of (2R,3R)-3-hydroxy-2-methyl-3-phenylpropionic acid phenyl ester (87%ee) and (2R,3S)-3-hydroxy-2-methyl-3-phenylpropionic acid phenyl ester (32% ee), respectively.

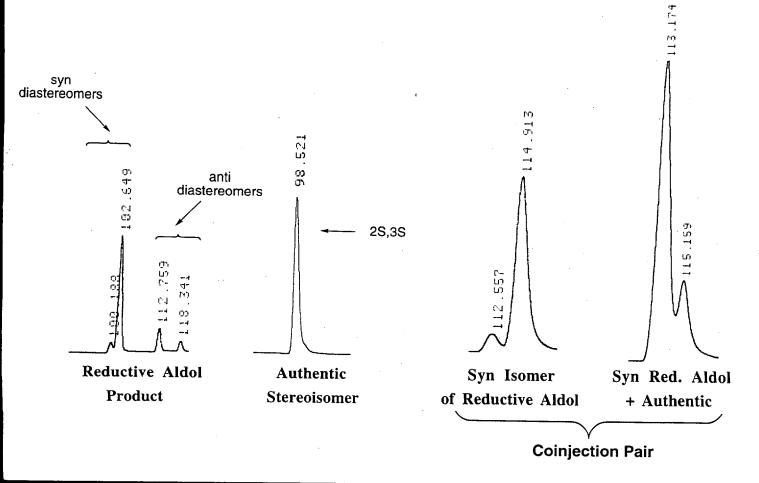
(2R,3R)-3-hydroxy-2-methyl-3-phenylpropionic acid phenyl ester. IR (neat) 3489 (br, s), 3032 (s), 2939 (s), 1756 (s), 1593 (m), 1193 (s) cm⁻¹; ¹H NMR:
$$\delta$$
 6.86-7.45 (10H, m, aromatic), 5.15 (1H, d, J=5.1 Hz, ArCHOH), 3.05 (1H, dq, J=5.1, 7.2 Hz, CH₃CH), 2.70 (1H, broad s, OH), 1.32 (3H, d, J=7.3 Hz, CHCH₃); ¹³C NMR: δ 174.0, 150.5, 141.5, 129.5, 128.5, 128.0, 126.3, 126.1, 121.5, 74.4, 47.2, 11.9. Anal. Calc'd for C₁₆H₁₆O₃: C, 74.98; H, 6.29. Found: C, 74.77; H, 6.11.

¹H NMR of the *anti* diastereomer: δ 6.86-7.45 (10H, m, aromatic), 4.87 (1H, d, J=8.6 Hz ArCHOH), 3.05 (1H, m, CH₃CH), 1.17 (3H, d, J=7.0 Hz, CHCH₃).

Proof of Stereochemistry. Stereochemical ratios were determined in comparison to authentic racemic materials prepared by lithium enolate aldol reaction (Heathcock, C. H.; Buse, C. T.; Kleschick, W. A.; Pirrung, M. C.; Sohn, J. E.; Lampe, J. J. Org. Chem. 1980, 45, 1066-1081). Relative stereochemistry determined in comparison to ¹H NMR reported for the syn aldol adduct (Ishihara, K.; Maruyama, T.; Mouri, M.; Gao, Q.; Furuta, K.; Yamamoto, H. Bull. Chem. Soc. Jpn. 1993, 66, 3483). Absolute stereochemistry established in comparison to authentic 2S,3S isomer prepared through the use of a chiral auxillary (M.T. Crimmins and K. Chaudhary, Organic Letters, In Press).

Chiral GLC (β -dex, Supelco) analysis of reductive ald ol product:

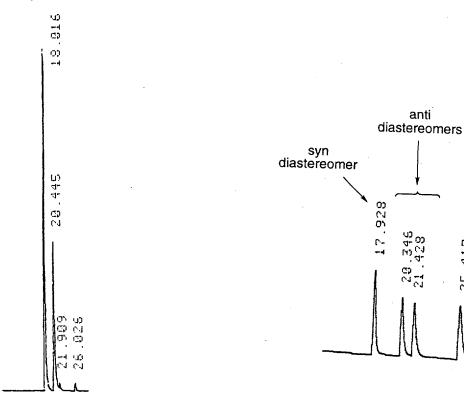
Note: Coinjection Pair Analyzed With Slightly Different GC Carrier Flow Rate.



QH O (2R,3R)-3-hydroxy-2-methyl-3-phenylpropionic acid methyl ester. IR OMe (KBr) 3450 (br, s) 3085 (s), 2940 (s) 1900 (w) 1720 (s), 1230 (s) cm⁻¹; ¹H NMR: δ 7.20-7.40 (5H, m, C_6H_5), 5.08 (1H, broad t, J=3.8 Hz, ArCHOH), 3.65 (3H, s, OCH₃), 2.93 (1H, d, J=3.09 Hz, OH), 2.77 (1H, m, CH₃CH), 1.11 (3H, d, J=7.12 Hz, CHCH₃); ¹³C NMR: δ 176.2, 141.4, 128.3, 127.4, 125.9, 73.6, 51.9, 46.3, 10.7. Anal. Calc'd for $C_{11}H_{14}O_3$: C, 68.02; H, 7.27. Found: C 67.85, H 7.04.

Proof of Stereochemistry. Determination of stereochemical ratios was accomplished through chiral GLC analysis in comparison to non-selective aldol adducts prepared according to the method of Heathcock (Heathcock, C. H.; Buse, C. T.; Kleschick, W. A.; Pirrung, M. C.; Sohn, J. E.; Lampe, J. J. Org. Chem. 1980, 45, 1066-1081). The relative configuration was established by comparison to reported ¹H NMR spectra (Itoh, T.; Kuroda, K.; Tomosada, M.; Takagi, Y. J. Org. Chem. 1991, 56, 797-804). Absolute configuration of each diastereomer was established by comparison of the optical rotation of each purified diastereomer to that reported in the literature (Gennari, C.; Colombo, L.; Bertolini, G.; Schimperna, G. J. Org. Chem. 1987, 52, 2754).

Chiral GLC (Chiraldex-GTA, Alltech) analysis of reductive aldol reaction product and racemic mixture of diastereomers:



Reductive Aldol Product

Authentic Stereoisomer
Mixture

syn diastereomer

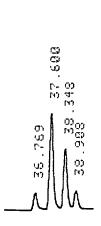
3-hydroxy-2-methyl-3-phenylpropionic acid *tert*-butyl ester. IR (neat) 3458 (br, s), 3061 (s) 2978 (s), 1946 (w), 1883 (w), 1723 (s), 1453 (s), 1147 (s), 1024 (s), 695 (s) cm-1; 1H NMR 7.26-7.37 (5H m, aromatic), 5.02 (1H, dd, J= 2.5, 4.3 Hz, ArCHOH), 3.13 (1H, d, J= 3.1 Hz, OH), 2.7 (1H, m, CHCHCH3), 1.4 (9H, s, OC(CH3)3), 1.11 (3H, d, J= 7.2 Hz, CHCH3) 13C NMR 175.6, 141.8, 128.4, 127.6, 126.4, 81.3, 74.1, 47.3, 28.1, 11.3. Anal.

Calc'd for C14H20O3: C, 71.16; H, 8.53 Found: C, 70.90; H, 8.25.

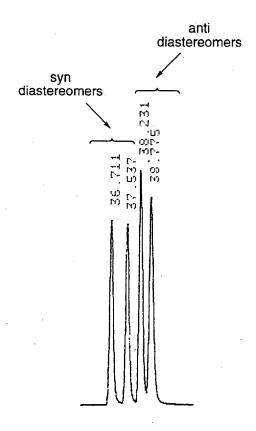
1H NMR of the *anti* diastereomer: 7.26-7.37 (5H, m, aromatic), 4.7 (1H, dd, J= 4.9, 8.0 Hz CHOH), 3.23 (1H, d, J= 4.9 Hz, OH) 2.7 (1H, m, CHCHCH3) 1.44 (9H, s, OC(CH3)3) 1.02 (3H, d, J= 7.2 Hz CHCH3)

The identity of this compound and the relative configuration of the stereoisomers was established by comparison to those reported in the literature (Heathcock, C. H.; Buse, C. T.; Kleschick, W. A.; Pirrung, M. C.; Sohn, J. E.; Lampe, J. J. Org. Chem. 1980, 45, 1066-1081). Absolute stereochemistry of the reaction product was not established.

Chiral GLC (Chiraldex-GTA, Alltech) analysis of reductive aldol reaction product and racemic mixture of diastereomers:



Reductive Aldol Product



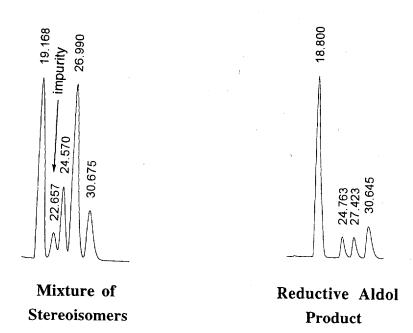
Authentic Stereoisomer Mixture

QH O (2*R*,3*S*)-3-cyclohexyl-3-hydroxyl-2-methylpropionic acid phenyl ester. IR (neat) 3505 (br, m), 2926 (s), 2852 (m), 1755 (s), 1493 (m), 1196 (s) cm⁻¹;
1
H NMR: δ 6.95 - 7.27 (5H, m, aromatic), 3.67 (1H, dd, J=8.1, 3.7 Hz, CHOH), 2.80 (1H, dq, J=3.8, 7.1 Hz, CH₃CH), 1.98 (1H, br, OH), 1.20 (3H, d, J=7.1 Hz, CHCH₃), 0.83-1.75 (11H, m, cyclohexyl); 13 C NMR: δ 175.1, 150.6, 129.5, 126.0, 121.6, 41.9, 40.6, 29.4, 26.6, 26.3, 26.1, 10.4. Anal. Calc'd for $C_{16}H_{22}O_{3}$: C, 73.25; H, 8.45. Found: C, 73.22; H, 8.41.

¹H NMR of the *anti* diastereomer: δ 6.95 - 7.27 (5H, m, aromatic), 3.40 (1H, t, J=5.9 Hz CHOH), 2.80 (1H, m, CH₃CH), 1.98 (1H, br, OH), 1.25 (3H, d, J=7.1 Hz, CHCH₃), 0.83-1.75 (11H, m, cyclohexyl).

Proof of Stereochemistry. Stereochemistry was determined in comparison to authentic racemic materials prepared by lithium enolate aldol reaction (Heathcock, C. H.; Buse, C. T.; Kleschick, W. A.; Pirrung, M. C.; Sohn, J. E.; Lampe, J. J. Org. Chem. 1980, 45, 1066-1081). Identity of syn and anti stereoisomers was established through analysis of the coupling constant between carbinol proton and the proton at the α -carbon (typically ~3 Hz for the syn isomer and ~9 Hz for the anti isomer). In addition, after conversion of the major diastereomer to the derived methyl ester, relative stereochemistry was established in comparison to literature data (syn: Hofstraat, R. G.; Scheeren, H. W.; Nivard, R. J. F. JCS Perkin 1, 1985, 561; anti: Harada, T.; Kurokawa, H.; Kagamihara, Y.; Tanaka, S.; Inoue, A.; Oku, A. J. ORg. Chem. 1992, 57, 1412). Absolute configuration of the syn isomer was established by measurement of optical rotation of the derived methyl ester ($[\alpha]^{25}_D$ -6.23, c=1.25, CH₂Cl₂) in comparison to literature value (2R,3S:: $[\alpha]^{25}_D$ -6.17, c=1.1, CH₂Cl₂; Drewes, S. E.; Malissar, D. G. S.; Roos, G. H. P. Tetrahedron Asymm. 1992, 3, 515).

Chiral HPLC (Chiralcel-OJ, Daicel) of reductive aldol reaction product and racemic mixture of stereoisomer:



Taylor et al., Supplementary Material, Page 6

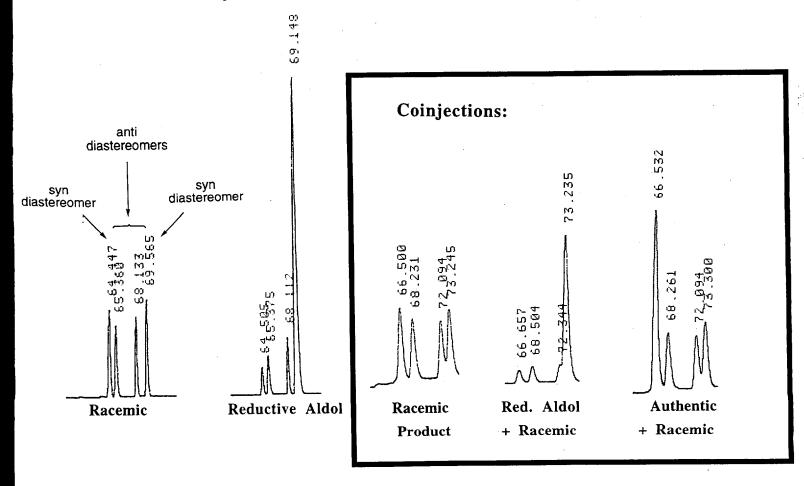
QH O (2R,3S)-3-hydroxy-2-methylpentanoic acid phenyl ester. IR (neat) 3473 (br, s), 3068 (m), 2976 (s), 1742 (s), 1491 (s), 979 (s) cm⁻¹; ¹H NMR: δ 7.05 – 7.39 (5H, m, aromatic), 3.95 (1H, m, CHOH), 2.78 (1H, dq, J=3.9, 7.2 Hz CH₃CHCHOH), 2.42 (1H, d, J=4.8, OH), 1.47 - 1.71 (2H, m, CH₂CH₃), 1.32 (3H, d, J=7.2 Hz, CHCH₃), 1.01 (3H, t, J=7.4 Hz, CH₂CH₃); ¹³C NMR: δ 174.7, 150.6, 129.5, 126.0, 121.6, 73.4, 44.4, 27.2, 11.0, 10.7. Anal. Calc'd for C₁₂H₁₆O₃: C, 69.21; H, 7.74. Found: C, 69.31; H, 7.55.

¹H NMR of the *anti* diastereomer: δ 7.05 – 7.39 (5H, m, aromatic), 3.72 (1H, m, CHOH), 2.76 (1H, m, CH₃CHCHOH), 2.48 (1H, d, J=6.8 Hz, OH), 1.47 - 1.71 (2H, m, CH₂CH₃), 1.34 (3H, d, J=7.3 Hz, CHCH₃), 1.02 (3H, t, J=7.4 Hz, CH₂CH₃).

Proof of Stereochemistry. Stereochemistry was determined in comparison to authentic racemic materials prepared by lithium enolate aldol reaction (Heathcock, C. H.; Buse, C. T.; Kleschick, W. A.; Pirrung, M. C.; Sohn, J. E.; Lampe, J. J. Org. Chem. 1980, 45, 1066-1081) and also in comparison to authentic 2S,3R isomer prepared through the use of a chiral auxillary (M.T. Crimmins and K. Chaudhary, Organic Letters, In Press).

Chiral GLC (Chiraldex-GTA, Alltech, GC of derived acetate esters):

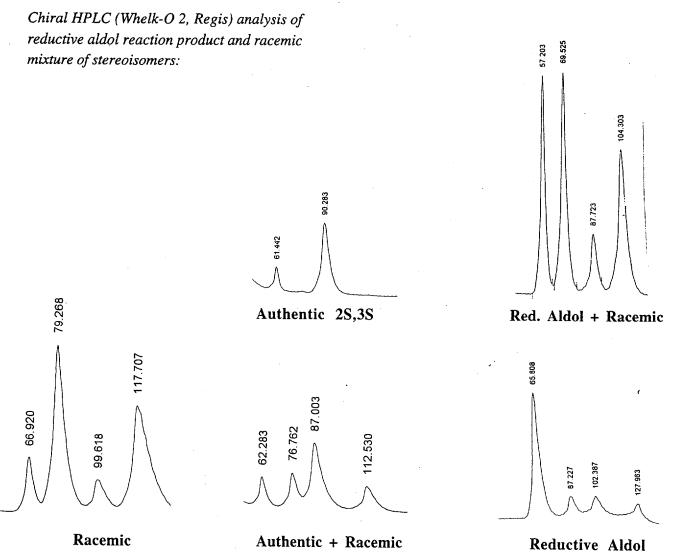
Note: Coinjection Set Analyzed Under Slightly Different GC Flow Rate



QH O (2R,3R)-3-(1-naphthyl)-3-hydroxy-2-methylpropionic acid phenyl ester. IR (neat) 3498 (br, s), 3057 (m), 2986 (m), 1753 (s), 1496 (s), 902 (s) cm⁻¹; H NMR: δ 7.01 - 8.33 (12H, m, aromatic), 6.11 (1H, d, J=3.4 Hz, CHOH), 3.32 (1H, dq, J=3.7, 7.1 Hz, CH₃CH), 2.86 (1H, s, OH), 1.28 (3H, d, J=7.1 Hz, CHCH₃); ¹³C NMR: δ 174.5, 150.8, 136.8, 134.0, 130.1, 129.7, 129.4, 128.6, 126.6, 126.2, 125.9, 125.6, 124.4, 122.9, 121.7, 70.7, 45.2, 10.7. Anal. Calc'd for C₁₂H₁₆O₃: C, 78.41; H, 5.92. Found: C, 78.06; H, 6.01.

¹H NMR of the anti diastereomer: δ 7.01 – 8.33 (12H, m, aromatic), 5.64 (1H, d, J=8.2 Hz, CHOH), 3.45 (1H, m, CH₃CH), 3.16 (1H, broad s, OH), 1.23 (3H, d, J=7.2 Hz, CHCH₃).

Proof of Stereochemistry. Stereochemistry was determined in comparison to authentic racemic materials prepared by lithium enolate aldol reaction (Heathcock, C. H.; Buse, C. T.; Kleschick, W. A.; Pirrung, M. C.; Sohn, J. E.; Lampe, J. *J. Org. Chem.* **1980**, *45*, 1066-1081) and also in comparison to authentic 2S,3S isomer prepared through the use of a chiral auxillary (M.T. Crimmins and K. Chaudhary, Organic Letters, **In Press**).

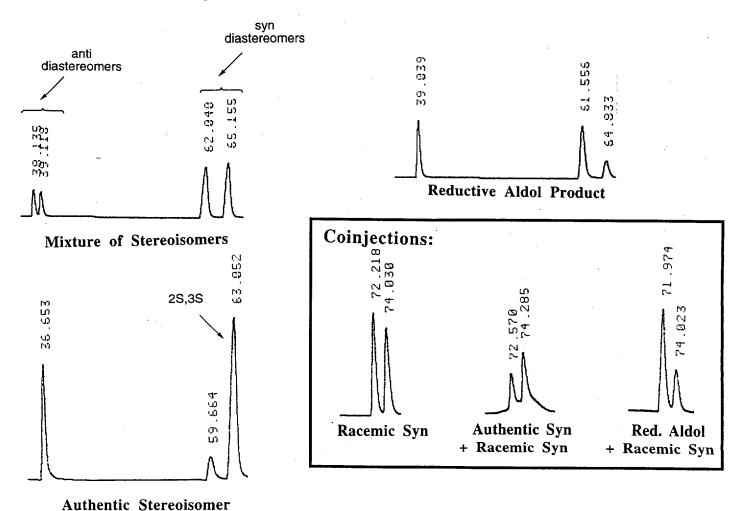


Characterization of the *anti* diastereomer: IR (neat) 3518 (br, m), 3053 (m), 2930 (s), 1731 (s), 1265 (s) cm⁻¹; ¹H NMR: δ 6.98 – 7.36 (5H, m, aromatic), 3.36 (1H, d, J=9.0 Hz, OH), 3.20 (1H, dd, J=9.6, 2.2 Hz, CHOH), 2.93 (1H, dq, J=2.2, 7.3 Hz CH₃CH), 1.45 (3H, d, J=7.3 Hz, CHCH₃), 0.93 (9H, s, (CH₃)₃C); ¹³C NMR: δ 175.9, 150.2, 129.6, 126.2, 121.4, 82.9, 39.0, 30.0, 26.6, 18.5.

Proof of Stereochemistry. Stereochemistry was determined in comparison to authentic racemic materials prepared by lithium enolate aldol reaction and also in comparison to authentic 2S,3S isomer prepared through the use of a chiral auxillary (M.T. Crimmins and K. Choudhary, unpublished). Relative stereochemistry was also confirmed by ¹H NMR analysis of the derived pyrrolidine amide (pyrrolidine neat, 12hr) in comparison to literature values (Fujita, M.; Hiyama, T. J. Org. Chem. 1988, 53, 5405).

Chiral GLC analysis (Chiraldex GTA, Alltech) of reductive aldol product:

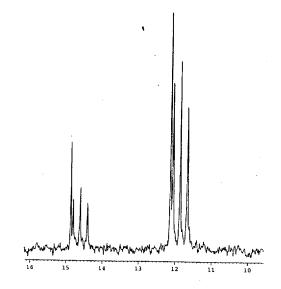
Note: Coninjections analyzed with slight different carrier flow rate



Et₂MeSiO OPh (2R,3R)-3-cyclohexyl-3-diethytlmethylsiloxy-2-methylpropanoic acid phenyl ester. IR 2930 (s), 2852 (m), 1755 (s), 1450 (m) cm⁻¹; 1 H NMR: δ 7.04 - 7.39 (5H, m, aromatic), 3.88 (1H, t, J=5.4 Hz, CHOSi), 2.85 (1H, m, CH₃CH), 1.27 (3H, d, J=7.0 Hz, CHCH₃), 1.00-1.90 (11H, m, cyclohexyl), 0.94 (6H, m, CH₃CH₂Si), 0.60 (4H, m, CH₃CH₂Si), 0.10 (3H, s, CH₃Si); 13 C NMR: δ 173.8, 150.6, 129.3, 125.5, 121.3, 77.8, 43.4, 42.8, 30.2, 28.1, 26.6, 26.5, 25.5, 12.8, 7.3, 7.2, 7.1, -3.7. HR EI-MS exact mass calcd for C₂₁H₃₄O₃Si [M-CH₃]+ 347.2042, found 347.2044.

Proof of Stereochemistry. Stereochemical ratio and configuration of the major product was determined by GLC analysis of the derived β -hydroxy ester as described above.

Reductive Aldol Reaction with PhMe₂SiD. The general procedure was followed except that phenyldimethylsilyldeuteride (Brookhart, M.; Grant, B. E. J. Am. Chem. Soc. 1993, 115, 2151) was used in place of diethylmethyl silane. ¹³C NMR analysis of the C2 methyl groups shows partial deuteration for each diastereomer.



Methyl Region of ¹³C NMR

Exchange of Si-D and Acrylate β -H. Treatment of phenyl acrylate with phenyldimethylsilyldeuteride for 24 hours in the presence of catalytic [(cod)RhCl]₂-(R)-BINAP, followed by removal of solvent and examination of the 1 H NMR shows indicates incorporation of deuterium at the β -position of the acrylate (see olefinic region of 1 H NMR below).



