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Symbol b	Level c	R _{Li-O1}	θ	R _{Li-O2}	∠O1LiO2	R _{H1-O2}	$\Delta E(n) d$	Imag. freq. (sym)
Li(H ₂ O)								
Ia 1+0 (C _{2v})	А	1.909					12.5	181 <i>i</i> (B ₁)
	В	1.903	0.0				12.2	122 <i>i</i> (B ₁)
	С	1.917	0.0				10.6 (12.2)	
Ib 1+0 (C _s)	Α	1.918	29.3				12.6	
	В	1.910	27.1				12.3	
Li(H ₂ O) ₂								
IIa 2+0 (C _{2v})	А	1.917			142.9		25.8	77 <i>i</i> (A ₂), 66 <i>i</i> (B ₁)
	В	1.871			174.3		26.8	
	С	1.934			123.7		20.2	119 <i>i</i> (A ₂), 62 <i>i</i> (B ₂)
II <i>b</i> 2+0 (C ₂)	А	1.917			119.0		26.2	12 <i>i</i> (B)
	С	1.935			107.7		21.1 (25.9)	
IIg 2+0 (C_s)	Α	1.909			127.4		26.1	26 <i>i</i> (A")
	С	1.924			117.6		20.5	123 <i>i</i> (A")
II c 2+0 (C_s)	Α	1.937		1.903	104.7		26.1	188 <i>i</i> (A")
	В	1.942		1.882	95.5		26.2	
	С	1.956		1.918	105.3		21.0	64 <i>i</i> (A")
IId 2+0 (C ₁)	Α	1.924		1.919	113.0		26.2	
IIe 1+1 (C _s)	Α	1.884				1.776	22.5	77i (A")
	В	1.869				1.774	22.7	
	С	1.888				1.883	19.2 (22.4)	
IIf $1+1$ (C ₁)	Α	1.890				1.778	22.5	

Table S1. Optimized Geometrical Parameters (Å and degrees) a and Total Binding Energies, $\Delta E(n)$ (kcal/mol), of Li(H₂O)_n (n = 1-2) Calculated at Various Levels. Imaginary Frequencies (cm⁻¹) and Symmetry of Imaginary Modes are Given if Exist.

^a Parameters are shown in Figure S1. ^b Corresponds to structures in Figure S1. ^c A: MP2/6-311++G(d,p). B: MP2/6-31++G(d,p). C: HF/6-31++G(d,p). ^d $-\Delta E(n) = E[Li(H_2O)_n] - E[Li] - nE[H_2O]$ (without CPC). Values for potential minima are in italic. Values in parentheses are at MP2/6--31++G(d,p) // HF/6-31++G(d,p).

Symbol ^b	Level ^c	R _{Li-O1}	R _{Li-O2}	R _{Li-O3}	∠O1LiO2	∠O2LiO3	∠O1LiO3	R _{H1-O3}	R _{H2-O2}	ΔEď	Imag. freq. (sym.)
Ша 3+0 (С _{2v})	Α	1.902	1.913		117.3					42.3	161 <i>i</i> (B ₁), 17 <i>i</i> (B ₂)
	В	1.885	1.886		113.0					43.5	66 <i>i</i> (B ₂), 57 <i>i</i> (B ₁)
	С	1.915	1.910		115.6					<i>33.</i> 2 (42.8)	
IIIk 3+0 (C _s)	Α	1.909	1.913		117.0	125.3				42.3	1 <i>i</i> (A")
	В	1.891	1.885		112.7	133.4				43.5	51 <i>i</i> (A")
Ш/ 3+0 (С _s)	A	1.903	1.918	1.912		125.0	127.0			42.3	150 <i>i</i> (A")
	В	1.865	1.932	1.874		126.7	140.7			44.0	41 <i>i</i> (A")
III <i>b</i> 3+0 (C ₁)	Α	1.909	1.913	1.913	117.4	125.3	116.7			42.3	
Ш <i>с</i> 3+0 (С ₁)	В	1.859	1.930	1.860	118.3	92.5	149.2			44.1	
Ш <i>d</i> 3+0 (С ₃)	А	1.907			119.6					42.4	
	В	1.880			117.5					44.0	
	С	1.912			119.9					<i>33.3</i> (40.2)	
Ше 2+1 (С _s)	А	1.880	1.971		107.7			1.798	2.030	37.8	314 <i>i</i> (A"), 114 <i>i</i> (A")
	В	1.850	1.928		113.5			1.796	1.988	40.5	
	С	1.891	2.008		106.7			1.914	2.178	30.1	73i (A")

Table S2. Optimized Geometrical Parameters (Å and degrees)^a and Total Binding Energies, ΔE (kcal/mol), of Li(H₂O)₃ Calculated at Various Levels. Imaginary Frequencies (cm⁻¹) and Symmetry of Imaginary Modes are Presented if Exist.

(Continues to next page.)

Supporting Information (2/1)

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					(Table S2	2 continued	1.)				
Symbol ^b	Level ^c	R _{Li-O1}	R _{Li-O2}	R _{Li-O3}	∠01LiO2	∠O2LiO3	∠O1LiO3	R _{H1-O3}	RH2-O2	ΔEď	Imag. freq. (sym.)
IIIf 2+1 (C ₁)	A	1.884	1.975		106.6			1.773	2.014	38.3	
	С	1.890	2.010		106.0			1.912	2.174	<i>30.1</i> (39.0)	
IIIg 2+1 (C _s)	Α	1.921			103.2			1.994		36.4	
	в	1.888			111.2			1.994		38.2	
	С	1.931			99.7			2.126		29.1 (34.7)	
IIIh 1+2 (C _{2v})	Α	1.862						1.829		31.0	15 <i>i</i> (B ₁)
	В	1.842						1.825		31.5	47 <i>i</i> (B ₂)
	С	1.863						1.929		26.4 (31.2)	
III <i>i</i> 1+2 (Cs)	Α	1.863						1.829		31.0	
	В	1.836						1.838		31.7	63i (A")
III <i>j</i> 1+2 (C ₁)	В	1.830						1.823	1.844	31.7	

^a Parameters are shown in Figure 1. ^b Corresponds to structures in Figure 1. ^c A: MP2/6-311++G(d,p). B: MP2/6-31++G(d,p). C: HF/6-31++G(d,p). ^d $-\Delta E(n) = E[Li(H_2O)_n] - E[Li] - n E[H_2O]$ (without CPC). Values for potential minima are in italic. Values in parentheses are at MP2/6-31++G(d,p)//HF/6-31++G(d,p).

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supporting Information (3/5)

	······································	Li s basis function							
n	- structure ^a	inner	outer	diffuse	total				
0		0.32	0.60	0.03	0.96				
1	Ia	0.21	0.69	0.08	0.98				
2	IIb	0.17	0.59	0.14	0.90				
3	IIIa	0.18	0.29	0.33	0.79				
4	IVa	0.22	0.17	0.36	0.74				
5	Va	0.21	0.10	0.28	0.59				
6	VIa	0.21	0.10	0.21	0.52				
8	VIIIa	0.20	0.17	0.15	0.52				

Table S3. Mulliken Total Gross Population in Li s Basis Functions in $Li(H_2O)_n$ Clusters by HF/6-31++G(d,p) Method.

a Indicates the structures in Figures 1 and 2.

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Supporting Information (4/5)



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