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Aminoguanidinium Hydrolysis Effected by a Hydroxo-Bridged Dicobalt(II) Complex as a Functional Model for Arginase and Catalyzed by Mononuclear Cobalt(II) Complexes

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- Table S1.** Crystal Data and Structure Refinement for $[\text{Co}_2(\mu\text{-XDK})(\text{NO}_3)_2(\text{CH}_3\text{OH})_3(\text{H}_2\text{O})_{1.625}] \cdot 2.5\text{MeOH} \cdot 0.5\text{Et}_2\text{O}$ (1).
- Table S2.** Final Positional Parameters and Isotropic Thermal Parameters for $[\text{Co}_2(\mu\text{-XDK})(\text{NO}_3)_2(\text{CH}_3\text{OH})_3(\text{H}_2\text{O})_{1.625}] \cdot 2.5\text{MeOH} \cdot 0.5\text{Et}_2\text{O}$ (1).
- Table S3.** Intramolecular Distances (Å) and Bond Angles (deg) for $[\text{Co}_2(\mu\text{-XDK})(\text{NO}_3)_2(\text{CH}_3\text{OH})_3(\text{H}_2\text{O})_{1.625}] \cdot 2.5\text{MeOH} \cdot 0.5\text{Et}_2\text{O}$ (1).
- Table S4.** Anisotropic Displacement Parameters for $[\text{Co}_2(\mu\text{-XDK})(\text{NO}_3)_2(\text{CH}_3\text{OH})_3(\text{H}_2\text{O})_{1.625}] \cdot 2.5\text{MeOH} \cdot 0.5\text{Et}_2\text{O}$ (1).
- Table S5.** Hydrogen Atom Positional Parameters and Isotropic Thermal Parameters for $[\text{Co}_2(\mu\text{-XDK})(\text{NO}_3)_2(\text{CH}_3\text{OH})_3(\text{H}_2\text{O})_{1.625}] \cdot 2.5\text{MeOH} \cdot 0.5\text{Et}_2\text{O}$ (1).
- Table S6.** Crystal Data and Structure Refinement for $[\text{Co}_2(\mu\text{-OH})(\mu\text{-XDK})(\text{bpy})_2(\text{EtOH})](\text{NO}_3) \cdot 2\text{EtOH}$ (2).
- Table S7.** Final Positional Parameters and Isotropic Thermal Parameters for $[\text{Co}_2(\mu\text{-OH})(\mu\text{-XDK})(\text{bpy})_2(\text{EtOH})](\text{NO}_3) \cdot 2\text{EtOH}$ (2).
- Table S8.** Intramolecular Distances (Å) and Bond Angles (deg) for $[\text{Co}_2(\mu\text{-OH})(\mu\text{-XDK})(\text{bpy})_2(\text{EtOH})](\text{NO}_3) \cdot 2\text{EtOH}$ (2).
- Table S9.** Anisotropic Displacement Parameters for $[\text{Co}_2(\mu\text{-OH})(\mu\text{-XDK})(\text{bpy})_2(\text{EtOH})](\text{NO}_3) \cdot 2\text{EtOH}$ (2).
- Table S10.** Hydrogen Atom Positional Parameters and Isotropic Thermal Parameters for $[\text{Co}_2(\mu\text{-OH})(\mu\text{-XDK})(\text{bpy})_2(\text{EtOH})](\text{NO}_3) \cdot 2\text{EtOH}$ (2).
- Table S11.** Crystal Data and Structure Refinement for $[\text{Co}_2(\mu\text{-Cl})(\mu\text{-XDK})(\text{bpy})_2(\text{EtOH})_2] \cdot (\text{NO}_3) \cdot 3\text{EtOH}$ (3).

- Table S12.** Final Positional Parameters and Isotropic Thermal Parameters for $[\text{Co}_2(\mu\text{-Cl})(\mu\text{-XDK})(\text{bpy})_2(\text{EtOH})_2]\cdot(\text{NO}_3)\cdot 3\text{EtOH}$ (3).
- Table S13.** Intramolecular Distances (Å) and Bond Angles (deg) for $[\text{Co}_2(\mu\text{-Cl})(\mu\text{-XDK})(\text{bpy})_2(\text{EtOH})_2]\cdot(\text{NO}_3)\cdot 3\text{EtOH}$ (3).
- Table S14.** Anisotropic Displacement Parameters for $[\text{Co}_2(\mu\text{-Cl})(\mu\text{-XDK})(\text{bpy})_2(\text{EtOH})_2]\cdot(\text{NO}_3)\cdot 3\text{EtOH}$ (3).
- Table S15.** Hydrogen Atom Positional Parameters and Isotropic Thermal Parameters for $[\text{Co}_2(\mu\text{-Cl})(\mu\text{-XDK})(\text{bpy})_2(\text{EtOH})_2]\cdot(\text{NO}_3)\cdot 3\text{EtOH}$ (3).
- Table S16.** Crystal Data and Structure Refinement for $[\text{Co}_2(\mu\text{-XDK})(\text{py})_3(\text{NO}_3)_2]\cdot\text{MeOH}\cdot 0.5\text{Et}_2\text{O}$ (4).
- Table S17.** Final Positional Parameters and Isotropic Thermal Parameters for $[\text{Co}_2(\mu\text{-XDK})(\text{py})_3(\text{NO}_3)_2]\cdot\text{MeOH}\cdot 0.5\text{Et}_2\text{O}$ (4).
- Table S18.** Intramolecular Distances (Å) and Bond Angles (deg) for $[\text{Co}_2(\mu\text{-XDK})(\text{py})_3(\text{NO}_3)_2]\cdot\text{MeOH}\cdot 0.5\text{Et}_2\text{O}$ (4).
- Table S19.** Anisotropic Displacement Parameters for $[\text{Co}_2(\mu\text{-XDK})(\text{py})_3(\text{NO}_3)_2]\cdot\text{MeOH}\cdot 0.5\text{Et}_2\text{O}$ (4).
- Table S20.** Hydrogen Atom Positional Parameters and Isotropic Thermal Parameters for $[\text{Co}_2(\mu\text{-XDK})(\text{py})_3(\text{NO}_3)_2]\cdot\text{MeOH}\cdot 0.5\text{Et}_2\text{O}$ (4).
- Table S21.** Crystal Data and Structure Refinement for $[\text{Co}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]\cdot 3\text{H}_2\text{O}\cdot 0.5\text{EtOH}$ (5).
- Table S22.** Final Positional Parameters and Isotropic Thermal Parameters for $[\text{Co}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]\cdot 3\text{H}_2\text{O}\cdot 0.5\text{EtOH}$ (5).
- Table S23.** Intramolecular Distances (Å) and Bond Angles (deg) for $[\text{Co}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]\cdot 3\text{H}_2\text{O}\cdot 0.5\text{EtOH}$ (5).
- Table S24.** Anisotropic Displacement Parameters for $[\text{Co}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]\cdot 3\text{H}_2\text{O}\cdot 0.5\text{EtOH}$ (5).
- Table S25.** Hydrogen Atom Positional Parameters and Isotropic Thermal Parameters for $[\text{Co}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]\cdot 3\text{H}_2\text{O}\cdot 0.5\text{EtOH}$ (5).
- Table S26.** Crystal Data and Structure Refinement for $[\text{Zn}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]\cdot 1.25\text{H}_2\text{O}$ (6).
- Table S27.** Final Positional Parameters and Isotropic Thermal Parameters for $[\text{Zn}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]\cdot 1.25\text{H}_2\text{O}$ (6).
- Table S28.** Intramolecular Distances (Å) and Bond Angles (deg) for $[\text{Zn}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]\cdot 1.25\text{H}_2\text{O}$ (6).
- Table S29.** Anisotropic Displacement Parameters for $[\text{Zn}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]\cdot 1.25\text{H}_2\text{O}$ (6).
- Table S30.** Hydrogen Atom Positional Parameters and Isotropic Thermal Parameters for $[[\text{Zn}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]\cdot 1.25\text{H}_2\text{O}$ (6).
- Figure S1.** Numbering scheme for the XDK ligand.
- Figure S2.** ORTEP diagram of $[\text{Co}_2(\text{XDK})(\text{NO}_3)_2(\text{CH}_3\text{OH})_3(\text{H}_2\text{O})]$ (at 0.75 occupancy) and $[\text{Co}_2(\text{XDK})(\text{NO}_3)_2(\text{CH}_3\text{OH})_3(\text{H}_2\text{O})_2]$ (at 0.25 occupancy) (1) showing

the 40% probability thermal ellipsoids, with hydrogen atoms omitted for clarity.

- Figure S3.** ORTEP diagram of $[\text{Co}_2(\text{XDK})(\text{NO}_3)_2(\text{CH}_3\text{OH})_3(\text{H}_2\text{O})_2]$ (1) (the other molecule in the asymmetric unit) showing the 40% probability thermal ellipsoids, with hydrogen atoms omitted for clarity.
- Figure S4.** ORTEP diagram of $[\text{Co}_2(\mu\text{-OH})(\text{XDK})(\text{bpy})_2(\text{EtOH})](\text{NO}_3)\cdot 2\text{EtOH}$ (2) showing the 40% probability thermal ellipsoids, with hydrogen atoms omitted for clarity.
- Figure S5.** ORTEP diagram of $[\text{Co}_2(\mu\text{-Cl})(\text{XDK})(\text{bpy})_2(\text{EtOH})_2]\cdot(\text{NO}_3)\cdot 3\text{EtOH}$ (3) showing the 40% probability thermal ellipsoids, with hydrogen atoms omitted for clarity.
- Figure S6.** ORTEP diagram of $[\text{Co}_2(\text{XDK})(\text{py})_3(\text{NO}_3)_2]\cdot\text{MeOH}\cdot 0.5\text{Et}_2\text{O}$ (4) showing the 40% probability thermal ellipsoids, with hydrogen atoms omitted for clarity.
- Figure S7.** ORTEP diagram of $[\text{Co}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]\cdot 3\text{H}_2\text{O}\cdot 0.5\text{EtOH}$ (5) showing the 40% probability thermal ellipsoids, with hydrogen atoms omitted for clarity.
- Figure S8.** ORTEP diagram of $[\text{Zn}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]\cdot 1.25\text{H}_2\text{O}$ (6) showing the 40% probability thermal ellipsoids, with hydrogen atoms omitted for clarity..
- Figure S9.** Hydrolysis of 0.05 M aminoguanidinium nitrate catalyzed by 0.0025 M $[\text{Co}(\text{XDK})(\text{bpy})(\text{H}_2\text{O})]$ (5) in 37 °C incubator and 1:1 MeOH/H₂O solution (0.10 M CHES buffer, pH 9.5, CHES: 2-(cyclohexylamino)-ethanesulfonic acid). Product urea was detected for 12 turnovers.
- Figure S10.** UV-vis studies of 0.05 M aminoguanidine nitrate hydrolysis by 0.0025 M 5 in 1:1 MeOH/H₂O solution (0.10 M CHES buffer, pH 9.5) at 37 °C. Spectra a, b and c were taken at 0, 3, 30 minutes after 500 μL of 0.1 M aminoguanidine nitrate solution was injected into 500 μL of 0.005 M 5 in methanol to initiate reaction.
- Figure S11.** Standard calibration curve for urea determination.

Table S1. Crystal data and structure refinement for [Co₂(XDK)(NO₃)₂(CH₃OH)₃(H₂O)_{1.625}]·2.5MeOH·0.5Et₂O.

Identification code	[Co ₂ (XDK)(NO ₃) ₂ (CH ₃ OH) ₃ (H ₂ O) _{1.625}] 2.5MeOH 0.5Et ₂ O
Empirical formula	C _{38.25} H _{65.75} Co ₂ N ₄ O _{20.38}
Formula weight	1025.56
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system	Triclinic
Space group	P-1
Unit cell dimensions	a = 11.6456(2) Å alpha = 70.3790(10) deg. b = 18.8160(2) Å beta = 89.5860(10) deg. c = 23.1313(5) Å gamma = 85.6410(10) deg.
Volume, Z	4759.59(14) Å ³ , 4
Density (calculated)	1.431 Mg/m ³
Absorption coefficient	0.777 mm ⁻¹
F(000)	2161
Crystal size	0.20 x 0.20 x 0.10 mm
Theta range for data collection	1.15 to 23.00 deg.
Limiting indices	-15 ≤ h ≤ 14, -25 ≤ k ≤ 24, -15 ≤ l ≤ 30
Reflections collected	20697
Independent reflections	12927 [R(int) = 0.0331]
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	12236 / 0 / 1124
Goodness-of-fit on F ²	1.043
Observed reflections [I > 2σ(I)]	9044
Final R indices [I > 2σ(I)]	R ₁ = 0.0597, wR ₂ = 0.1456
R indices (all data)	R ₁ = 0.0960, wR ₂ = 0.1664
Largest diff. peak and hole	0.901 and -0.579 e.Å ⁻³

Table S2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $[\text{Co}_2(\text{XDK})(\text{NO}_3)_2(\text{CH}_3\text{OH})_3 \cdot (\text{H}_2\text{O})_{1.625}] \cdot 2.5\text{MeOH} \cdot 0.5\text{Et}_2\text{O}$. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	$U(\text{eq})$
Co(1)	6880(1)	2832(1)	6391(1)	32(1)
Co(2)	8562(1)	2506(1)	4797(1)	27(1)
O(101)	6191(3)	2993(2)	5549(2)	30(1)
O(102)	7013(3)	3058(2)	4655(2)	29(1)
O(103)	3825(3)	2136(2)	5954(2)	29(1)
O(104)	5427(3)	1924(2)	4227(2)	26(1)
O(201)	7936(3)	1898(2)	6562(2)	31(1)
O(202)	8686(3)	1893(2)	5677(2)	32(1)
O(203)	6850(3)	288(2)	7234(2)	34(1)
O(204)	8436(3)	286(2)	5440(2)	26(1)
N(101)	4546(4)	2017(2)	5081(2)	21(1)
N(201)	7656(3)	279(2)	6344(2)	18(1)
C(101)	6221(5)	3229(3)	4972(3)	23(1)
C(102)	3903(4)	2421(3)	5401(3)	23(1)
C(103)	4760(5)	2293(3)	4444(3)	22(1)
C(104)	5780(5)	4573(3)	4457(3)	31(2)
C(105)	2246(5)	3341(4)	5360(3)	40(2)
C(106)	3867(6)	3027(4)	3430(3)	38(2)
C(107)	5255(5)	3803(3)	4609(3)	24(1)
C(108)	4158(5)	3814(3)	4986(3)	26(1)
C(109)	3327(5)	3189(3)	5032(3)	26(1)
C(110)	3047(5)	3194(3)	4388(3)	31(2)
C(111)	4139(5)	3046(3)	4067(3)	24(1)
C(112)	4944(5)	3685(3)	4007(3)	26(1)
C(201)	8703(5)	1698(3)	6246(3)	23(1)
C(202)	8610(5)	201(3)	5977(3)	21(1)
C(203)	7726(5)	195(3)	6971(3)	22(1)
C(204)	10571(5)	1803(4)	6645(3)	34(2)
C(205)	10517(5)	-446(4)	5956(3)	33(2)
C(206)	8755(5)	-486(3)	7962(3)	31(2)
C(207)	9759(4)	1214(3)	6598(2)	22(1)
C(208)	10360(5)	730(3)	6242(3)	25(1)
C(209)	9783(4)	-4(3)	6289(3)	23(1)
C(210)	9637(5)	-470(3)	6965(3)	25(1)
C(211)	8898(5)	-22(3)	7289(2)	23(1)
C(212)	9484(5)	700(3)	7246(2)	23(1)
C(301)	6088(4)	1143(3)	5716(2)	20(1)
C(302)	5021(4)	1253(3)	5433(2)	17(1)
C(303)	4359(4)	653(3)	5478(2)	20(1)
C(304)	4831(4)	-74(3)	5807(2)	20(1)
C(305)	5901(5)	-209(3)	6100(2)	21(1)
C(306)	6519(5)	418(3)	6047(2)	19(1)
C(307)	3162(5)	773(3)	5196(3)	28(1)
C(308)	6381(5)	-1003(3)	6464(3)	29(1)
N(1)	7685(5)	4212(4)	5992(3)	51(2)
N(2)	10141(5)	2464(4)	4025(3)	50(2)
O(1A)	6673(5)	4083(3)	6201(3)	44(2)
O(1B)	7073(17)	4490(12)	6388(10)	58(5)
O(2)	8144(6)	4791(3)	5755(3)	81(2)
O(3A)	8257(6)	3576(4)	5984(3)	36(2)
O(3B)	7756(19)	3662(12)	5981(9)	33(5)
O(4)	9233(4)	2904(3)	3869(2)	49(1)
O(5)	10871(5)	2401(4)	3653(3)	79(2)
O(6)	10251(4)	2077(3)	4589(2)	51(1)
O(7A)	5416(5)	2362(4)	6750(3)	43(2)

C(3A)	4592(10)	2739(7)	7061(5)	66(3)
O(7B)	5325(14)	3576(10)	6498(8)	39(4)
C(3B)	4290(23)	3326(16)	6811(13)	42(7)
O(7C)	5555(12)	1908(9)	6769(7)	20(3)
O(8)	7288(4)	2827(3)	7277(2)	46(1)
C(1)	7306(8)	2180(4)	7818(3)	67(2)
O(9)	9293(4)	3404(2)	4997(2)	45(1)
C(2)	9381(7)	4136(4)	4537(4)	62(2)
O(10)	7848(3)	1640(2)	4540(2)	31(1)
Co(3)	5885(1)	2592(1)	-256(1)	28(1)
Co(4)	7434(1)	2828(1)	1396(1)	28(1)
O(401)	5658(4)	1891(2)	572(2)	45(1)
O(402)	6522(3)	1968(2)	1400(2)	39(1)
O(403)	6496(3)	292(2)	421(2)	31(1)
O(404)	8084(3)	495(2)	2130(2)	25(1)
O(501)	7373(3)	3054(2)	-345(2)	37(1)
O(502)	8045(3)	3164(2)	520(2)	38(1)
O(503)	9140(3)	1979(2)	-810(2)	32(1)
O(504)	10664(3)	2237(2)	898(2)	31(1)
N(401)	7293(4)	341(2)	1293(2)	19(1)
N(501)	9973(4)	2093(2)	34(2)	23(1)
C(401)	5774(5)	1733(3)	1141(3)	28(1)
C(402)	6361(5)	218(3)	957(3)	26(1)
C(403)	7254(5)	322(3)	1906(3)	23(1)
C(404)	3887(5)	1809(4)	1588(3)	38(2)
C(405)	4687(5)	-532(4)	1016(3)	42(2)
C(406)	6459(5)	-338(4)	2933(3)	32(2)
C(407)	4878(5)	1233(3)	1545(3)	25(1)
C(408)	4437(5)	687(3)	1241(3)	30(2)
C(409)	5265(5)	-22(3)	1304(3)	28(1)
C(410)	5570(5)	-439(3)	1978(3)	27(1)
C(411)	6156(5)	78(3)	2258(3)	24(1)
C(412)	5356(5)	789(3)	2194(3)	25(1)
C(501)	8029(5)	3313(3)	-46(3)	27(1)
C(502)	10532(5)	2507(3)	346(3)	25(1)
C(503)	9697(5)	2363(3)	-598(3)	26(1)
C(504)	8159(5)	4659(3)	-582(3)	37(2)
C(505)	12003(5)	3411(4)	269(3)	41(2)
C(506)	10379(7)	3082(4)	-1631(3)	50(2)
C(507)	8868(5)	3890(3)	-429(3)	27(1)
C(508)	9968(5)	3895(3)	-72(3)	29(1)
C(509)	10931(5)	3268(3)	-38(3)	28(1)
C(510)	11195(5)	3257(3)	-677(3)	33(2)
C(511)	10121(5)	3115(3)	-988(3)	30(2)
C(512)	9175(5)	3757(3)	-1031(3)	32(2)
C(601)	8632(4)	1212(3)	658(2)	19(1)
C(602)	8390(4)	479(3)	989(2)	20(1)
C(603)	9175(5)	-139(3)	1046(3)	21(1)
C(604)	10234(5)	4(3)	768(2)	24(1)
C(605)	10527(4)	733(3)	439(2)	21(1)
C(606)	9697(5)	1331(3)	385(2)	21(1)
C(607)	8900(5)	-935(3)	1410(3)	31(2)
C(608)	11717(5)	865(3)	176(3)	32(2)
N(3)	5079(6)	3783(3)	1350(3)	52(2)
N(4)	4402(6)	2732(5)	-1121(3)	63(2)
O(11)	5916(4)	3589(3)	1059(2)	48(1)
O(12)	5209(6)	3691(4)	1899(3)	104(3)
O(13)	4166(5)	4043(3)	1081(3)	74(2)
O(14)	4252(4)	2313(3)	-584(3)	57(1)
O(15)	3751(5)	2797(5)	-1552(3)	112(3)
O(16)	5298(5)	3102(3)	-1196(2)	56(1)
O(17)	6900(5)	2614(3)	2306(2)	69(2)
C(4)	7402(8)	2246(6)	2869(4)	85(3)
O(18)	8414(5)	3631(3)	1542(3)	71(2)
C(5)	8164(10)	4413(5)	1314(5)	102(4)
O(19)	5008(4)	3483(2)	-22(2)	43(1)

C(6)	4742(7)	4228(4)	-431(4)	59(2)
O(20)	6762(3)	1693(2)	-514(2)	35(1)
O(21)	8860(3)	2011(2)	1732(2)	29(1)
C(1S)	6390(11)	4613(8)	7629(6)	126(4)
O(1SA)	6679(13)	3849(8)	7793(6)	92(4)
O(1SB)	5959(13)	3900(8)	7656(7)	93(4)
C(3SA)	11374(11)	2986(7)	2123(6)	55(3)
C(3SB)	10349(17)	3741(12)	1834(10)	61(5)
O(3S)	10348(10)	3213(6)	2316(5)	174(4)
C(4S)	9398(17)	6098(11)	3328(9)	85(6)
C(5S)	8156(17)	5996(11)	3117(9)	87(6)
O(4S)	8274(10)	5286(7)	2996(6)	82(4)
C(6S)	7198(17)	5130(11)	2731(9)	84(6)
C(7S)	7368(17)	4327(12)	2829(9)	87(6)
O(2S)	10217(19)	4766(12)	2585(10)	169(8)
C(2S)	11003(24)	4542(15)	2960(13)	125(9)

Table S3. Bond lengths [Å] and angles [deg] for [Co₂(XDK)(NO₃)₂-(CH₃OH)₃(H₂O)_{1.625}] 2.5MeOH 0.5Et₂O.

Co(1)-O(3B)	1.90(2)
Co(1)-O(201)	1.992(4)
Co(1)-O(7A)	2.024(6)
Co(1)-O(101)	2.030(4)
Co(1)-O(8)	2.105(4)
Co(1)-O(3A)	2.202(7)
Co(1)-O(1A)	2.238(6)
Co(1)-O(7B)	2.27(2)
Co(1)-O(7C)	2.35(2)
Co(1)-Co(2)	4.3634(11)
Co(2)-O(202)	1.975(4)
Co(2)-O(102)	1.984(4)
Co(2)-O(9)	2.133(4)
Co(2)-O(10)	2.136(4)
Co(2)-O(6)	2.181(4)
Co(2)-O(4)	2.182(4)
O(101)-C(101)	1.258(7)
O(102)-C(101)	1.264(7)
O(103)-C(102)	1.216(7)
O(104)-C(103)	1.218(6)
O(201)-C(201)	1.263(7)
O(202)-C(201)	1.241(7)
O(203)-C(203)	1.216(6)
O(204)-C(202)	1.215(6)
N(101)-C(102)	1.405(7)
N(101)-C(103)	1.414(7)
N(101)-C(302)	1.460(7)
N(201)-C(203)	1.407(7)
N(201)-C(202)	1.422(7)
N(201)-C(306)	1.460(7)
C(101)-C(107)	1.534(8)
C(102)-C(109)	1.516(8)
C(103)-C(111)	1.522(8)
C(104)-C(107)	1.545(8)
C(105)-C(109)	1.522(8)
C(106)-C(111)	1.523(8)
C(107)-C(112)	1.532(8)
C(107)-C(108)	1.544(8)
C(108)-C(109)	1.555(8)
C(109)-C(110)	1.524(8)
C(110)-C(111)	1.522(8)
C(111)-C(112)	1.550(8)
C(201)-C(207)	1.534(8)
C(202)-C(209)	1.510(8)
C(203)-C(211)	1.517(8)
C(204)-C(207)	1.542(8)
C(205)-C(209)	1.526(7)
C(206)-C(211)	1.523(8)
C(207)-C(212)	1.533(7)
C(207)-C(208)	1.546(7)
C(208)-C(209)	1.553(8)
C(209)-C(210)	1.529(8)
C(210)-C(211)	1.523(8)
C(211)-C(212)	1.541(7)
C(301)-C(302)	1.378(7)
C(301)-C(306)	1.379(7)
C(302)-C(303)	1.389(7)
C(303)-C(304)	1.396(8)
C(303)-C(307)	1.512(7)
C(304)-C(305)	1.388(7)
C(305)-C(306)	1.398(7)

C(305)-C(308)	1.513(8)
N(1)-O(3B)	1.04(2)
N(1)-O(2)	1.207(8)
N(1)-O(1A)	1.281(8)
N(1)-O(3A)	1.331(9)
N(1)-O(1B)	1.37(2)
N(2)-O(5)	1.233(7)
N(2)-O(6)	1.263(7)
N(2)-O(4)	1.268(7)
O(7A)-C(3A)	1.472(12)
O(7B)-C(3B)	1.43(3)
O(8)-C(1)	1.421(8)
O(9)-C(2)	1.441(8)
Co(3)-O(401)	1.957(4)
Co(3)-O(501)	1.977(4)
Co(3)-O(19)	2.120(4)
Co(3)-O(16)	2.153(5)
Co(3)-O(20)	2.160(4)
Co(3)-O(14)	2.214(5)
Co(3)-Co(4)	4.4062(11)
Co(4)-O(402)	2.000(4)
Co(4)-O(502)	2.048(4)
Co(4)-O(18)	2.076(5)
Co(4)-O(17)	2.105(5)
Co(4)-O(21)	2.135(4)
Co(4)-O(11)	2.160(4)
O(401)-C(401)	1.255(7)
O(402)-C(401)	1.246(7)
O(403)-C(402)	1.211(6)
O(404)-C(403)	1.216(6)
O(501)-C(501)	1.259(7)
O(502)-C(501)	1.244(7)
O(503)-C(503)	1.224(7)
O(504)-C(502)	1.212(7)
N(401)-C(403)	1.406(7)
N(401)-C(402)	1.417(7)
N(401)-C(602)	1.453(7)
N(501)-C(503)	1.407(7)
N(501)-C(502)	1.418(7)
N(501)-C(606)	1.449(7)
C(401)-C(407)	1.547(8)
C(402)-C(409)	1.519(8)
C(403)-C(411)	1.527(8)
C(404)-C(407)	1.549(8)
C(405)-C(409)	1.534(8)
C(406)-C(411)	1.523(8)
C(407)-C(412)	1.536(8)
C(407)-C(408)	1.543(8)
C(408)-C(409)	1.551(8)
C(409)-C(410)	1.522(8)
C(410)-C(411)	1.536(8)
C(411)-C(412)	1.535(8)
C(501)-C(507)	1.556(8)
C(502)-C(509)	1.515(8)
C(503)-C(511)	1.518(8)
C(504)-C(507)	1.543(8)
C(505)-C(509)	1.525(8)
C(506)-C(511)	1.535(8)
C(507)-C(508)	1.531(8)
C(507)-C(512)	1.531(8)
C(508)-C(509)	1.548(8)
C(509)-C(510)	1.514(8)
C(510)-C(511)	1.531(9)
C(511)-C(512)	1.550(8)
C(601)-C(602)	1.383(7)
C(601)-C(606)	1.388(7)

C(602)-C(603)	1.394(7)
C(603)-C(604)	1.389(8)
C(603)-C(607)	1.508(8)
C(604)-C(605)	1.394(8)
C(605)-C(606)	1.399(7)
C(605)-C(608)	1.514(7)
N(3)-O(13)	1.217(8)
N(3)-O(12)	1.231(8)
N(3)-O(11)	1.280(7)
N(4)-O(15)	1.224(8)
N(4)-O(14)	1.247(8)
N(4)-O(16)	1.277(8)
O(17)-C(4)	1.364(9)
O(18)-C(5)	1.394(10)
O(19)-C(6)	1.416(8)
C(1S)-O(1SA)	1.37(2)
C(1S)-O(1SB)	1.45(2)
C(3SA)-O(3S)	1.36(2)
C(3SB)-O(3S)	1.22(2)
C(4S)-C(5S)	1.58(3)
C(5S)-O(4S)	1.45(2)
O(4S)-C(6S)	1.49(2)
C(6S)-C(7S)	1.45(3)
O(2S)-C(2S)	1.21(3)

O(3B)-Co(1)-O(201)	106.8(7)
O(201)-Co(1)-O(7A)	99.6(2)
O(3B)-Co(1)-O(101)	85.5(6)
O(201)-Co(1)-O(101)	104.1(2)
O(7A)-Co(1)-O(101)	87.5(2)
O(3B)-Co(1)-O(8)	94.7(6)
O(201)-Co(1)-O(8)	87.9(2)
O(7A)-Co(1)-O(8)	86.8(2)
O(101)-Co(1)-O(8)	167.5(2)
O(201)-Co(1)-O(3A)	92.7(2)
O(7A)-Co(1)-O(3A)	167.5(3)
O(101)-Co(1)-O(3A)	91.3(2)
O(8)-Co(1)-O(3A)	91.8(2)
O(201)-Co(1)-O(1A)	148.1(2)
O(7A)-Co(1)-O(1A)	109.8(2)
O(101)-Co(1)-O(1A)	89.6(2)
O(8)-Co(1)-O(1A)	81.9(2)
O(3A)-Co(1)-O(1A)	57.8(2)
O(3B)-Co(1)-O(7B)	94.1(8)
O(201)-Co(1)-O(7B)	157.7(4)
O(101)-Co(1)-O(7B)	84.9(4)
O(8)-Co(1)-O(7B)	82.5(4)
O(3B)-Co(1)-O(7C)	170.2(7)
O(201)-Co(1)-O(7C)	80.1(4)
O(101)-Co(1)-O(7C)	86.1(4)
O(8)-Co(1)-O(7C)	92.4(4)
O(7B)-Co(1)-O(7C)	80.2(6)
O(202)-Co(2)-O(102)	107.1(2)
O(202)-Co(2)-O(9)	89.3(2)
O(102)-Co(2)-O(9)	91.7(2)
O(202)-Co(2)-O(10)	94.0(2)
O(102)-Co(2)-O(10)	88.1(2)
O(9)-Co(2)-O(10)	176.6(2)
O(202)-Co(2)-O(6)	93.8(2)
O(102)-Co(2)-O(6)	159.0(2)
O(9)-Co(2)-O(6)	91.3(2)
O(10)-Co(2)-O(6)	87.6(2)
O(202)-Co(2)-O(4)	152.7(2)
O(102)-Co(2)-O(4)	100.3(2)
O(9)-Co(2)-O(4)	90.3(2)
O(10)-Co(2)-O(4)	86.4(2)

O(6)-Co(2)-O(4)	58.9(2)
C(101)-O(101)-Co(1)	152.3(4)
C(101)-O(102)-Co(2)	137.7(4)
C(201)-O(201)-Co(1)	132.3(4)
C(201)-O(202)-Co(2)	162.7(4)
C(102)-N(101)-C(103)	124.8(5)
C(102)-N(101)-C(302)	117.3(4)
C(103)-N(101)-C(302)	117.9(4)
C(203)-N(201)-C(202)	124.9(4)
C(203)-N(201)-C(306)	118.4(4)
C(202)-N(201)-C(306)	116.6(4)
O(101)-C(101)-O(102)	125.6(5)
O(101)-C(101)-C(107)	118.6(5)
O(102)-C(101)-C(107)	115.8(5)
O(103)-C(102)-N(101)	118.7(5)
O(103)-C(102)-C(109)	123.6(5)
N(101)-C(102)-C(109)	117.7(5)
O(104)-C(103)-N(101)	119.0(5)
O(104)-C(103)-C(111)	123.5(5)
N(101)-C(103)-C(111)	117.5(5)
C(112)-C(107)-C(101)	112.4(4)
C(112)-C(107)-C(108)	109.7(5)
C(101)-C(107)-C(108)	113.2(5)
C(112)-C(107)-C(104)	108.8(5)
C(101)-C(107)-C(104)	103.7(4)
C(108)-C(107)-C(104)	108.8(4)
C(107)-C(108)-C(109)	116.4(4)
C(102)-C(109)-C(105)	108.9(5)
C(102)-C(109)-C(110)	109.0(5)
C(105)-C(109)-C(110)	111.6(5)
C(102)-C(109)-C(108)	109.6(5)
C(105)-C(109)-C(108)	108.7(5)
C(110)-C(109)-C(108)	109.1(5)
C(111)-C(110)-C(109)	110.7(5)
C(103)-C(111)-C(106)	108.9(5)
C(103)-C(111)-C(110)	109.9(5)
C(106)-C(111)-C(110)	111.3(5)
C(103)-C(111)-C(112)	108.7(4)
C(106)-C(111)-C(112)	109.1(5)
C(110)-C(111)-C(112)	108.9(5)
C(107)-C(112)-C(111)	115.6(5)
O(202)-C(201)-O(201)	125.0(5)
O(202)-C(201)-C(207)	118.2(5)
O(201)-C(201)-C(207)	116.7(5)
O(204)-C(202)-N(201)	118.6(5)
O(204)-C(202)-C(209)	124.1(5)
N(201)-C(202)-C(209)	117.3(5)
O(203)-C(203)-N(201)	119.0(5)
O(203)-C(203)-C(211)	123.0(5)
N(201)-C(203)-C(211)	118.0(4)
C(201)-C(207)-C(212)	113.2(4)
C(201)-C(207)-C(204)	103.6(4)
C(212)-C(207)-C(204)	109.1(4)
C(201)-C(207)-C(208)	112.0(5)
C(212)-C(207)-C(208)	109.4(5)
C(204)-C(207)-C(208)	109.3(4)
C(207)-C(208)-C(209)	115.2(4)
C(202)-C(209)-C(205)	108.1(5)
C(202)-C(209)-C(210)	109.0(4)
C(205)-C(209)-C(210)	111.7(5)
C(202)-C(209)-C(208)	109.5(4)
C(205)-C(209)-C(208)	109.0(4)
C(210)-C(209)-C(208)	109.6(5)
C(211)-C(210)-C(209)	111.0(5)
C(203)-C(211)-C(206)	109.6(4)
C(203)-C(211)-C(210)	108.8(4)

C(206)-C(211)-C(210)	111.1(5)
C(203)-C(211)-C(212)	109.4(4)
C(206)-C(211)-C(212)	109.2(4)
C(210)-C(211)-C(212)	108.8(4)
C(207)-C(212)-C(211)	116.1(4)
C(302)-C(301)-C(306)	119.2(5)
C(301)-C(302)-C(303)	121.8(5)
C(301)-C(302)-N(101)	120.1(5)
C(303)-C(302)-N(101)	118.0(5)
C(302)-C(303)-C(304)	117.4(5)
C(302)-C(303)-C(307)	122.1(5)
C(304)-C(303)-C(307)	120.5(5)
C(305)-C(304)-C(303)	122.5(5)
C(304)-C(305)-C(306)	117.4(5)
C(304)-C(305)-C(308)	121.3(5)
C(306)-C(305)-C(308)	121.2(5)
C(301)-C(306)-C(305)	121.6(5)
C(301)-C(306)-N(201)	120.7(5)
C(305)-C(306)-N(201)	117.7(5)
O(3B)-N(1)-O(2)	137.3(13)
O(3B)-N(1)-O(1A)	87.1(13)
O(2)-N(1)-O(1A)	132.3(7)
O(2)-N(1)-O(3A)	116.7(6)
O(1A)-N(1)-O(3A)	110.5(7)
O(3B)-N(1)-O(1B)	128(2)
O(2)-N(1)-O(1B)	93.7(10)
O(5)-N(2)-O(6)	121.4(7)
O(5)-N(2)-O(4)	122.6(7)
O(6)-N(2)-O(4)	115.9(5)
O(1B)-O(1A)-N(1)	69.0(11)
N(1)-O(1A)-Co(1)	95.7(4)
N(1)-O(3A)-Co(1)	95.9(5)
N(1)-O(3B)-Co(1)	130(2)
N(2)-O(4)-Co(2)	92.4(4)
N(2)-O(6)-Co(2)	92.6(4)
C(3A)-O(7A)-Co(1)	121.3(6)
C(3B)-O(7B)-Co(1)	126(2)
C(1)-O(8)-Co(1)	124.7(4)
C(2)-O(9)-Co(2)	122.0(4)
O(401)-Co(3)-O(501)	113.1(2)
O(401)-Co(3)-O(19)	90.1(2)
O(501)-Co(3)-O(19)	93.0(2)
O(401)-Co(3)-O(16)	149.4(2)
O(501)-Co(3)-O(16)	97.5(2)
O(19)-Co(3)-O(16)	90.9(2)
O(401)-Co(3)-O(20)	90.8(2)
O(501)-Co(3)-O(20)	87.3(2)
O(19)-Co(3)-O(20)	178.8(2)
O(16)-Co(3)-O(20)	87.9(2)
O(401)-Co(3)-O(14)	91.1(2)
O(501)-Co(3)-O(14)	155.5(2)
O(19)-Co(3)-O(14)	90.9(2)
O(16)-Co(3)-O(14)	58.3(2)
O(20)-Co(3)-O(14)	88.3(2)
O(402)-Co(4)-O(502)	101.5(2)
O(402)-Co(4)-O(18)	170.8(2)
O(502)-Co(4)-O(18)	86.6(2)
O(402)-Co(4)-O(17)	86.2(2)
O(502)-Co(4)-O(17)	172.2(2)
O(18)-Co(4)-O(17)	85.7(2)
O(402)-Co(4)-O(21)	87.1(2)
O(502)-Co(4)-O(21)	92.3(2)
O(18)-Co(4)-O(21)	88.4(2)
O(17)-Co(4)-O(21)	89.0(2)
O(402)-Co(4)-O(11)	88.9(2)
O(502)-Co(4)-O(11)	88.7(2)

O(18)-Co(4)-O(11)	95.6(2)
O(17)-Co(4)-O(11)	90.5(2)
O(21)-Co(4)-O(11)	175.9(2)
C(401)-O(401)-Co(3)	148.6(4)
C(401)-O(402)-Co(4)	147.6(4)
C(501)-O(501)-Co(3)	139.6(4)
C(501)-O(502)-Co(4)	157.6(4)
C(403)-N(401)-C(402)	125.3(5)
C(403)-N(401)-C(602)	116.9(4)
C(402)-N(401)-C(602)	117.7(4)
C(503)-N(501)-C(502)	124.4(5)
C(503)-N(501)-C(606)	117.6(5)
C(502)-N(501)-C(606)	117.9(5)
O(402)-C(401)-O(401)	125.3(6)
O(402)-C(401)-C(407)	118.2(6)
O(401)-C(401)-C(407)	116.4(6)
O(403)-C(402)-N(401)	118.3(5)
O(403)-C(402)-C(409)	124.4(5)
N(401)-C(402)-C(409)	117.2(5)
O(404)-C(403)-N(401)	119.2(5)
O(404)-C(403)-C(411)	123.3(5)
N(401)-C(403)-C(411)	117.5(5)
C(412)-C(407)-C(408)	110.3(5)
C(412)-C(407)-C(401)	112.0(5)
C(408)-C(407)-C(401)	111.3(5)
C(412)-C(407)-C(404)	109.0(5)
C(408)-C(407)-C(404)	110.2(5)
C(401)-C(407)-C(404)	103.8(5)
C(407)-C(408)-C(409)	114.9(5)
C(402)-C(409)-C(410)	109.0(5)
C(402)-C(409)-C(405)	107.9(5)
C(410)-C(409)-C(405)	111.3(5)
C(402)-C(409)-C(408)	109.7(5)
C(410)-C(409)-C(408)	109.7(5)
C(405)-C(409)-C(408)	109.2(5)
C(409)-C(410)-C(411)	110.6(5)
C(406)-C(411)-C(403)	109.5(5)
C(406)-C(411)-C(412)	109.6(5)
C(403)-C(411)-C(412)	108.5(4)
C(406)-C(411)-C(410)	110.6(5)
C(403)-C(411)-C(410)	108.5(4)
C(412)-C(411)-C(410)	110.0(5)
C(411)-C(412)-C(407)	116.0(5)
O(502)-C(501)-O(501)	126.1(5)
O(502)-C(501)-C(507)	117.5(5)
O(501)-C(501)-C(507)	116.3(5)
O(504)-C(502)-N(501)	118.8(5)
O(504)-C(502)-C(509)	123.8(5)
N(501)-C(502)-C(509)	117.5(5)
O(503)-C(503)-N(501)	118.8(5)
O(503)-C(503)-C(511)	122.8(5)
N(501)-C(503)-C(511)	118.3(5)
C(508)-C(507)-C(512)	110.1(5)
C(508)-C(507)-C(504)	109.9(5)
C(512)-C(507)-C(504)	108.5(5)
C(508)-C(507)-C(501)	113.1(5)
C(512)-C(507)-C(501)	111.6(5)
C(504)-C(507)-C(501)	103.5(4)
C(507)-C(508)-C(509)	116.5(5)
C(510)-C(509)-C(502)	109.0(5)
C(510)-C(509)-C(505)	110.7(5)
C(502)-C(509)-C(505)	108.6(5)
C(510)-C(509)-C(508)	109.8(5)
C(502)-C(509)-C(508)	109.3(4)
C(505)-C(509)-C(508)	109.3(5)
C(509)-C(510)-C(511)	111.0(5)

C(503)-C(511)-C(510)	108.7(5)
C(503)-C(511)-C(506)	108.4(5)
C(510)-C(511)-C(506)	111.8(5)
C(503)-C(511)-C(512)	109.2(5)
C(510)-C(511)-C(512)	108.6(5)
C(506)-C(511)-C(512)	110.0(5)
C(507)-C(512)-C(511)	116.3(5)
C(602)-C(601)-C(606)	118.6(5)
C(601)-C(602)-C(603)	122.0(5)
C(601)-C(602)-N(401)	119.6(5)
C(603)-C(602)-N(401)	118.3(5)
C(604)-C(603)-C(602)	117.6(5)
C(604)-C(603)-C(607)	121.0(5)
C(602)-C(603)-C(607)	121.4(5)
C(603)-C(604)-C(605)	122.5(5)
C(604)-C(605)-C(606)	117.4(5)
C(604)-C(605)-C(608)	120.6(5)
C(606)-C(605)-C(608)	121.9(5)
C(601)-C(606)-C(605)	121.7(5)
C(601)-C(606)-N(501)	119.8(5)
C(605)-C(606)-N(501)	118.4(5)
O(13)-N(3)-O(12)	121.4(7)
O(13)-N(3)-O(11)	119.3(7)
O(12)-N(3)-O(11)	119.3(7)
O(15)-N(4)-O(14)	123.8(8)
O(15)-N(4)-O(16)	121.2(8)
O(14)-N(4)-O(16)	115.0(6)
N(3)-O(11)-Co(4)	130.3(4)
N(4)-O(14)-Co(3)	92.3(4)
N(4)-O(16)-Co(3)	94.3(4)
C(4)-O(17)-Co(4)	134.5(5)
C(5)-O(18)-Co(4)	126.0(6)
C(6)-O(19)-Co(3)	125.3(4)
O(4S)-C(5S)-C(4S)	105(2)
C(5S)-O(4S)-C(6S)	112.3(14)
C(7S)-C(6S)-O(4S)	103(2)

Symmetry transformations used to generate equivalent atoms:

Table S4. Anisotropic displacement parameters ($\text{Å}^2 \times 10^3$) for $[\text{Co}_2(\text{XDK})-(\text{NO}_3)_2(\text{CH}_3\text{OH})_3(\text{H}_2\text{O})_{1.625}] \cdot 2.5\text{MeOH} \cdot 0.5\text{Et}_2\text{O}$. The anisotropic displacement factor exponent takes the form: $-2 \pi^2 [h^2 a^2 U_{11} + \dots + 2 h k a^* b^* U_{12}]$

	U11	U22	U33	U23	U13	U12
Co(1)	46(1)	25(1)	23(1)	-9(1)	-5(1)	6(1)
Co(2)	23(1)	31(1)	23(1)	-4(1)	2(1)	-2(1)
O(101)	30(2)	35(3)	24(3)	-8(2)	-5(2)	0(2)
O(102)	21(2)	34(3)	31(2)	-11(2)	2(2)	2(2)
O(103)	30(2)	30(2)	25(3)	-6(2)	6(2)	-3(2)
O(104)	32(2)	24(2)	22(2)	-7(2)	-2(2)	-2(2)
O(201)	29(2)	36(3)	28(2)	-12(2)	3(2)	6(2)
O(202)	35(2)	33(3)	22(3)	-3(2)	-2(2)	3(2)
O(203)	22(2)	55(3)	25(2)	-15(2)	3(2)	-4(2)
O(204)	29(2)	33(2)	17(2)	-11(2)	0(2)	0(2)
N(101)	21(3)	20(3)	23(3)	-8(2)	-3(2)	-2(2)
N(201)	15(2)	20(3)	17(3)	-5(2)	-1(2)	-1(2)
C(101)	28(3)	18(3)	25(4)	-8(3)	-4(3)	-7(3)
C(102)	17(3)	20(3)	30(4)	-6(3)	-1(3)	-5(2)
C(103)	22(3)	22(3)	23(3)	-8(3)	-4(3)	-9(3)
C(104)	33(4)	24(3)	35(4)	-7(3)	1(3)	-4(3)
C(105)	27(4)	34(4)	56(5)	-9(3)	4(3)	2(3)
C(106)	48(4)	26(4)	36(4)	-4(3)	-14(3)	-1(3)
C(107)	26(3)	19(3)	25(3)	-7(3)	1(3)	-3(3)
C(108)	30(3)	19(3)	29(4)	-6(3)	-1(3)	-2(3)
C(109)	23(3)	22(3)	33(4)	-8(3)	3(3)	-1(3)
C(110)	29(3)	20(3)	37(4)	-3(3)	-9(3)	-1(3)
C(111)	24(3)	23(3)	22(3)	-5(3)	-10(3)	0(3)
C(112)	32(3)	19(3)	24(3)	-2(3)	-4(3)	1(3)
C(201)	28(3)	20(3)	23(4)	-6(3)	-3(3)	-10(3)
C(202)	24(3)	13(3)	28(4)	-9(3)	6(3)	-4(2)
C(203)	19(3)	22(3)	27(4)	-11(3)	4(3)	-4(2)
C(204)	25(3)	44(4)	33(4)	-13(3)	-1(3)	-10(3)
C(205)	23(3)	46(4)	33(4)	-19(3)	-2(3)	8(3)
C(206)	29(3)	37(4)	24(4)	-5(3)	1(3)	-9(3)
C(207)	14(3)	27(3)	22(3)	-7(3)	2(2)	-2(2)
C(208)	17(3)	27(3)	28(3)	-4(3)	4(3)	-3(3)
C(209)	16(3)	26(3)	27(3)	-10(3)	1(3)	3(2)
C(210)	23(3)	18(3)	29(4)	-5(3)	-4(3)	3(3)
C(211)	20(3)	25(3)	21(3)	-5(3)	-3(2)	-2(3)
C(212)	17(3)	28(3)	24(3)	-11(3)	-5(2)	2(2)
C(301)	21(3)	17(3)	22(3)	-7(3)	7(2)	-6(2)
C(302)	19(3)	15(3)	18(3)	-5(2)	0(2)	-1(2)
C(303)	18(3)	25(3)	19(3)	-10(3)	4(2)	-6(3)
C(304)	21(3)	20(3)	21(3)	-8(3)	5(2)	-11(2)
C(305)	26(3)	20(3)	17(3)	-7(3)	-2(2)	0(3)
C(306)	24(3)	22(3)	13(3)	-9(3)	3(2)	-2(3)
C(307)	26(3)	34(4)	28(4)	-13(3)	-2(3)	-5(3)
C(308)	33(4)	23(3)	31(4)	-8(3)	-4(3)	-2(3)
N(1)	50(4)	42(4)	51(4)	-5(3)	-1(3)	0(3)
N(2)	34(4)	78(5)	49(4)	-32(4)	14(3)	-23(4)
O(2)	113(5)	27(3)	96(5)	-8(3)	-11(4)	-21(3)
O(4)	43(3)	60(3)	39(3)	-8(2)	7(2)	-8(3)
O(5)	56(4)	128(6)	68(4)	-48(4)	43(3)	-24(3)
O(6)	32(3)	76(4)	43(3)	-20(3)	2(2)	2(2)
O(8)	66(3)	47(3)	25(3)	-13(2)	-10(2)	3(2)
C(1)	100(7)	59(6)	29(5)	-6(4)	-4(4)	21(5)
O(9)	42(3)	32(3)	57(3)	-7(2)	-7(2)	-13(2)
C(2)	65(5)	39(5)	77(6)	-10(4)	4(4)	-21(4)
O(10)	35(2)	35(3)	28(2)	-16(2)	2(2)	-3(2)
Co(3)	25(1)	31(1)	24(1)	-5(1)	-1(1)	-5(1)

Co(4)	33(1)	27(1)	26(1)	-9(1)	0(1)	-1(1)
O(401)	61(3)	40(3)	30(3)	-5(2)	11(2)	-10(2)
O(402)	26(2)	37(3)	46(3)	-1(2)	0(2)	-13(2)
O(403)	33(2)	36(3)	27(3)	-15(2)	1(2)	-9(2)
O(404)	19(2)	35(2)	22(2)	-8(2)	-3(2)	-5(2)
O(501)	28(2)	46(3)	46(3)	-24(2)	6(2)	-14(2)
O(502)	30(2)	43(3)	34(3)	-5(2)	6(2)	-9(2)
O(503)	39(3)	33(3)	24(2)	-9(2)	4(2)	-4(2)
O(504)	29(2)	29(2)	31(3)	-6(2)	-5(2)	-4(2)
N(401)	20(2)	21(3)	19(3)	-8(2)	2(2)	-6(2)
N(501)	24(3)	20(3)	22(3)	-4(2)	5(2)	-4(2)
C(401)	21(3)	20(3)	33(4)	0(3)	7(3)	8(3)
C(402)	31(3)	21(3)	27(4)	-11(3)	0(3)	-2(3)
C(403)	28(3)	15(3)	23(3)	-3(3)	1(3)	1(3)
C(404)	21(3)	39(4)	46(4)	-4(3)	1(3)	4(3)
C(405)	36(4)	46(4)	50(5)	-21(4)	3(3)	-20(3)
C(406)	31(3)	40(4)	23(3)	-9(3)	6(3)	-4(3)
C(407)	20(3)	28(3)	24(3)	-4(3)	-1(3)	-3(3)
C(408)	16(3)	40(4)	32(4)	-10(3)	1(3)	-9(3)
C(409)	25(3)	29(4)	32(4)	-13(3)	1(3)	-13(3)
C(410)	24(3)	26(3)	32(4)	-10(3)	6(3)	-6(3)
C(411)	22(3)	28(3)	20(3)	-5(3)	3(2)	-2(3)
C(412)	22(3)	28(3)	27(3)	-11(3)	9(3)	-5(3)
C(501)	23(3)	19(3)	35(4)	-4(3)	8(3)	3(3)
C(502)	15(3)	27(4)	36(4)	-13(3)	1(3)	2(3)
C(503)	26(3)	32(4)	23(4)	-12(3)	7(3)	1(3)
C(504)	32(4)	26(4)	45(4)	-2(3)	-1(3)	3(3)
C(505)	31(4)	29(4)	56(5)	-4(3)	-4(3)	-10(3)
C(506)	79(5)	40(4)	27(4)	-5(3)	23(4)	-11(4)
C(507)	25(3)	22(3)	30(4)	-5(3)	4(3)	-3(3)
C(508)	28(3)	18(3)	40(4)	-8(3)	4(3)	-7(3)
C(509)	20(3)	24(3)	37(4)	-6(3)	4(3)	-4(3)
C(510)	30(4)	23(4)	42(4)	-5(3)	15(3)	-3(3)
C(511)	37(4)	25(4)	23(3)	-1(3)	13(3)	-3(3)
C(512)	37(4)	23(4)	29(4)	2(3)	5(3)	-8(3)
C(601)	17(3)	22(3)	20(3)	-9(3)	-2(2)	0(2)
C(602)	20(3)	25(3)	16(3)	-8(3)	0(2)	-5(3)
C(603)	25(3)	23(3)	20(3)	-13(3)	2(3)	-4(3)
C(604)	30(3)	26(4)	20(3)	-11(3)	-2(3)	4(3)
C(605)	19(3)	23(3)	21(3)	-11(3)	-3(2)	1(3)
C(606)	22(3)	19(3)	21(3)	-7(3)	0(3)	-5(3)
C(607)	39(4)	21(3)	35(4)	-13(3)	6(3)	-3(3)
C(608)	26(3)	34(4)	38(4)	-15(3)	5(3)	-2(3)
N(3)	60(5)	46(4)	49(4)	-18(3)	6(4)	10(3)
N(4)	43(4)	105(6)	45(5)	-37(4)	-13(4)	21(4)
O(11)	55(3)	44(3)	40(3)	-13(2)	1(2)	19(2)
O(12)	102(5)	148(7)	58(4)	-46(4)	-4(4)	64(5)
O(13)	53(4)	72(4)	79(4)	-8(3)	-3(3)	21(3)
O(14)	39(3)	85(4)	55(4)	-33(3)	-4(3)	-10(3)
O(15)	65(4)	215(9)	73(5)	-78(5)	-39(4)	25(5)
O(16)	50(3)	70(4)	37(3)	-6(3)	-7(2)	6(3)
O(17)	81(4)	93(4)	27(3)	-20(3)	-2(3)	29(3)
C(4)	86(7)	118(9)	39(5)	-15(5)	-1(5)	12(6)
O(18)	104(5)	37(3)	76(4)	-23(3)	-18(3)	-20(3)
C(5)	156(11)	53(6)	108(9)	-34(6)	30(8)	-40(7)
O(19)	41(3)	33(3)	51(3)	-11(2)	3(2)	6(2)
C(6)	61(5)	41(5)	67(6)	-11(4)	0(4)	4(4)
O(20)	39(3)	38(3)	31(3)	-16(2)	-4(2)	-4(2)
O(21)	27(2)	31(2)	28(2)	-8(2)	-5(2)	-1(2)

Table S5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $[\text{Co}_2(\text{XDK})(\text{NO}_3)_2(\text{CH}_3\text{OH})_3 \cdot (\text{H}_2\text{O})_1.625] \cdot 2.5\text{MeOH} \cdot 0.5\text{Et}_2\text{O}$.

	x	y	z	U(eq)
H(10A)	5979(5)	4654(3)	4832(3)	47
H(10B)	5229(5)	4971(3)	4226(3)	47
H(10C)	6460(5)	4574(3)	4219(3)	47
H(10D)	1875(5)	3826(4)	5129(3)	61
H(10E)	2453(5)	3337(4)	5762(3)	61
H(10F)	1730(5)	2955(4)	5395(3)	61
H(10G)	3475(6)	3500(4)	3188(3)	58
H(10H)	3384(6)	2623(4)	3465(3)	58
H(10I)	4571(6)	2944(4)	3235(3)	58
H(10J)	3733(5)	4303(3)	4807(3)	31
H(10K)	4392(5)	3770(3)	5399(3)	31
H(11A)	2524(5)	2807(3)	4414(3)	37
H(11B)	2666(5)	3680(3)	4151(3)	37
H(11C)	5651(5)	3575(3)	3819(3)	32
H(11D)	4578(5)	4156(3)	3732(3)	32
H(20A)	11266(5)	1547(4)	6861(3)	51
H(20B)	10198(5)	2107(4)	6863(3)	51
H(20C)	10756(5)	2122(4)	6240(3)	51
H(20D)	11268(5)	-582(4)	6148(3)	50
H(20E)	10589(5)	-137(4)	5535(3)	50
H(20F)	10157(5)	-896(4)	5977(3)	50
H(20G)	9497(5)	-623(3)	8163(3)	46
H(20H)	8387(5)	-936(3)	7991(3)	46
H(20I)	8289(5)	-191(3)	8156(3)	46
H(20J)	11149(5)	592(3)	6394(3)	30
H(20K)	10386(5)	1039(3)	5813(3)	30
H(21A)	9279(5)	-930(3)	6993(3)	30
H(21B)	10388(5)	-614(3)	7167(3)	30
H(21C)	8989(5)	994(3)	7436(2)	27
H(21D)	10197(5)	552(3)	7485(2)	27
H(30A)	6511(4)	1554(3)	5685(2)	24
H(30B)	4412(4)	-485(3)	5831(2)	23
H(30C)	2849(5)	293(3)	5277(3)	42
H(30D)	3199(5)	1024(3)	4761(3)	42
H(30E)	2678(5)	1080(3)	5372(3)	42
H(30F)	7125(5)	-985(3)	6632(3)	44
H(30G)	6453(5)	-1301(3)	6199(3)	44
H(30H)	5871(5)	-1227(3)	6791(3)	44
H(1A)	7516(8)	2318(4)	8165(3)	100
H(1B)	7858(8)	1796(4)	7776(3)	100
H(1C)	6556(8)	1991(4)	7878(3)	100
H(2A)	9732(7)	4461(4)	4716(4)	93
H(2B)	8626(7)	4353(4)	4381(4)	93
H(2C)	9845(7)	4081(4)	4208(4)	93
H(40A)	3584(5)	2089(4)	1184(3)	57
H(40B)	4175(5)	2152(4)	1770(3)	57
H(40C)	3288(5)	1544(4)	1837(3)	57
H(40D)	3990(5)	-691(4)	1228(3)	62
H(40E)	5200(5)	-968(4)	1048(3)	62
H(40F)	4509(5)	-256(4)	591(3)	62
H(40G)	5768(5)	-491(4)	3154(3)	47
H(40H)	6828(5)	-8(4)	3099(3)	47
H(40I)	6970(5)	-777(4)	2971(3)	47
H(40J)	4294(5)	964(3)	808(3)	35
H(40K)	3708(5)	521(3)	1421(3)	35
H(41A)	6084(5)	-883(3)	2016(3)	32

H(41B)	4876(5)	-604(3)	2200(3)	32
H(41C)	5774(5)	1127(3)	2337(3)	30
H(41D)	4711(5)	640(3)	2464(3)	30
H(50A)	7471(5)	4653(3)	-806(3)	56
H(50B)	8610(5)	5055(3)	-828(3)	56
H(50C)	7956(5)	4750(3)	-209(3)	56
H(50D)	11819(5)	3416(4)	673(3)	62
H(50E)	12268(5)	3891(4)	29(3)	62
H(50F)	12597(5)	3017(4)	298(3)	62
H(50G)	10648(7)	3554(4)	-1884(3)	75
H(50H)	9689(7)	2990(4)	-1811(3)	75
H(50I)	10959(7)	2679(4)	-1599(3)	75
H(50J)	9758(5)	3855(3)	344(3)	34
H(50K)	10281(5)	4382(3)	-258(3)	34
H(51A)	11798(5)	2863(3)	-653(3)	40
H(51B)	11472(5)	3738(3)	-922(3)	40
H(51C)	9422(5)	4224(3)	-1318(3)	39
H(51D)	8481(5)	3645(3)	-1202(3)	39
H(60A)	8091(4)	1616(3)	619(2)	23
H(60B)	10768(5)	-403(3)	803(2)	29
H(60C)	8136(5)	-926(3)	1567(3)	46
H(60D)	9443(5)	-1147(3)	1745(3)	46
H(60E)	8945(5)	-1238(3)	1148(3)	46
H(60F)	11768(5)	1396(3)	-35(3)	48
H(60G)	11853(5)	594(3)	-107(3)	48
H(60H)	12283(5)	688(3)	502(3)	48
H(4A)	6872(8)	2270(6)	3183(4)	127
H(4B)	8084(8)	2484(6)	2909(4)	127
H(4C)	7604(8)	1727(6)	2911(4)	127
H(5A)	8753(10)	4656(5)	1448(5)	153
H(5B)	7433(10)	4539(5)	1465(5)	153
H(5C)	8133(10)	4583(5)	874(5)	153
H(6A)	4343(7)	4526(4)	-216(4)	88
H(6B)	4262(7)	4213(4)	-762(4)	88
H(6C)	5442(7)	4450(4)	-593(4)	88

Table S6. Crystal data and structure refinement for [Co₂(OH)(XDK)(bpy)₂(EtOH)](NO₃).2EtOH

Identification code	[Co ₂ (OH)(XDK)(bpy) ₂ (EtOH)](NO ₃).2EtOH	
Empirical formula	C ₅₈ H ₇₃ Co ₂ N ₇ O ₁₅	
Formula weight	1226.09	
Temperature	293(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P2(1)/c	
Unit cell dimensions	a = 12.3880(4) Å	alpha = 90 deg.
	b = 16.7363(6) Å	beta = 94.1760(10) deg.
	c = 27.9451(10) Å	gamma = 90 deg.
Volume, Z	5778.5(3) Å ³ , 4	
Density (calculated)	1.409 Mg/m ³	
Absorption coefficient	0.648 mm ⁻¹	
F(000)	2576	
Crystal size	0.25 x 0.25 x 0.15 mm	
Theta range for data collection	1.42 to 22.50 deg.	
Limiting indices	-16<=h<=12, -18<=k<=22, -29<=l<=37	
Reflections collected	22848	
Independent reflections	7492 [R(int) = 0.0795]	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	7109 / 0 / 733	
Goodness-of-fit on F ²	1.039	
Observed reflections [I>2sigma(I)]	4870	
Final R indices [I>2sigma(I)]	R1 = 0.0647, wR2 = 0.1328	
R indices (all data)	R1 = 0.1164, wR2 = 0.1543	
Largest diff. peak and hole	0.676 and -0.316 e.Å ⁻³	

Table S7. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $[\text{Co}_2(\text{OH})(\text{XDK})(\text{bpy})_2(\text{EtOH})] \cdot (\text{NO}_3) \cdot 2\text{EtOH}$. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	$U(\text{eq})$
Co(1)	6539(1)	6913(1)	1402(1)	27(1)
Co(2)	8746(1)	6296(1)	948(1)	27(1)
O(101)	6776(3)	7282(2)	690(1)	29(1)
O(102)	7924(3)	6531(3)	308(1)	33(1)
O(103)	6051(3)	9191(3)	405(2)	37(1)
O(104)	8776(3)	8132(2)	-388(1)	33(1)
O(201)	7873(3)	7573(2)	1702(1)	28(1)
O(202)	9390(3)	7010(2)	1475(1)	31(1)
O(203)	8029(3)	9589(3)	1820(2)	35(1)
O(204)	11091(3)	8542(3)	1292(2)	36(1)
N(101)	7412(4)	8644(3)	14(2)	24(1)
N(201)	9530(4)	9021(3)	1557(2)	23(1)
C(101)	7032(5)	6912(4)	328(2)	30(2)
C(102)	6295(5)	8746(4)	89(2)	29(2)
C(103)	7803(5)	8194(3)	-359(2)	24(2)
C(104)	5770(5)	5993(4)	-82(2)	38(2)
C(105)	4409(5)	8775(4)	-262(3)	44(2)
C(106)	7464(5)	7836(4)	-1218(2)	39(2)
C(107)	6221(4)	6846(4)	-108(2)	26(2)
C(108)	5277(5)	7447(4)	-82(2)	30(2)
C(109)	5484(5)	8315(4)	-249(2)	30(2)
C(110)	5926(5)	8298(4)	-746(2)	32(2)
C(111)	6985(5)	7829(4)	-728(2)	27(2)
C(112)	6748(5)	6956(4)	-585(2)	28(2)
C(201)	8852(5)	7362(4)	1775(2)	25(2)
C(202)	10571(5)	8676(4)	1633(2)	30(2)
C(203)	8861(5)	9246(4)	1925(2)	28(2)
C(204)	9189(5)	6693(4)	2534(2)	43(2)
C(205)	12205(5)	8544(5)	2203(2)	47(2)
C(206)	8751(5)	9617(4)	2779(2)	43(2)
C(207)	9400(5)	7487(4)	2279(2)	28(2)
C(208)	10623(5)	7623(4)	2272(2)	31(2)
C(209)	10968(5)	8482(4)	2149(2)	31(2)
C(210)	10475(5)	9072(4)	2486(2)	34(2)
C(211)	9242(5)	9024(4)	2437(2)	28(2)
C(212)	8885(5)	8174(4)	2553(2)	33(2)
C(301)	8518(4)	8821(4)	774(2)	22(2)
C(302)	8166(4)	9105(3)	325(2)	22(1)
C(303)	8488(5)	9853(4)	169(2)	27(2)
C(304)	9182(5)	10300(4)	474(2)	32(2)
C(305)	9537(5)	10033(4)	928(2)	30(2)
C(306)	9189(5)	9288(4)	1077(2)	24(2)
C(307)	8070(5)	10181(4)	-316(2)	37(2)
C(308)	10268(6)	10545(4)	1261(2)	46(2)
N(1)	4906(4)	6430(3)	1232(2)	35(1)
N(2)	5544(4)	7939(3)	1401(2)	31(1)
N(3)	10211(4)	6528(3)	570(2)	29(1)
N(4)	9523(4)	5195(3)	917(2)	27(1)
C(1)	4633(5)	5669(4)	1200(2)	41(2)
C(2)	3582(6)	5398(5)	1232(2)	47(2)
C(3)	2786(6)	5937(5)	1301(2)	45(2)
C(4)	3044(5)	6742(5)	1310(2)	44(2)
C(5)	4118(5)	6973(4)	1277(2)	32(2)
C(6)	4460(5)	7813(4)	1294(2)	29(2)
C(7)	3768(5)	8470(4)	1207(2)	37(2)

C(8)	4177(6)	9226(4)	1242(2)	40(2)
C(9)	5252(5)	9344(4)	1355(2)	35(2)
C(10)	5906(5)	8686(4)	1426(2)	31(2)
C(11)	10544(5)	7236(4)	429(2)	38(2)
C(12)	11427(5)	7344(4)	159(3)	44(2)
C(13)	11985(5)	6674(5)	32(2)	41(2)
C(14)	11652(5)	5929(4)	175(2)	37(2)
C(15)	10763(5)	5877(4)	446(2)	30(2)
C(16)	10355(5)	5121(4)	631(2)	30(2)
C(17)	10768(5)	4370(4)	529(2)	38(2)
C(18)	10343(5)	3698(4)	726(2)	38(2)
C(19)	9519(5)	3772(4)	1034(2)	39(2)
C(20)	9143(5)	4524(4)	1117(2)	37(2)
O(1)	7468(3)	5947(2)	1313(1)	32(1)
O(2)	6265(3)	6625(3)	2142(2)	43(1)
C(21)	6041(7)	5836(5)	2285(3)	67(2)
C(22)	6220(8)	5689(6)	2806(3)	88(3)
N(5)	6631(5)	3873(5)	1130(3)	60(2)
O(3)	6637(4)	4410(4)	826(3)	79(2)
O(4)	6912(7)	3987(6)	1557(3)	147(4)
O(5)	6492(6)	3173(4)	1013(2)	103(2)
O(1SA)	5087(28)	7530(21)	2691(13)	90(14)
O(1SB)	5411(19)	7622(14)	2726(8)	49(7)
O(1SC)	4841(14)	7632(10)	2526(6)	35(5)
C(1S)	5034(9)	8463(7)	2613(4)	106(4)
C(2SA)	5727(15)	8803(12)	3013(8)	79(6)
C(2SB)	5988(17)	8812(13)	2714(8)	91(7)
O(2SA)	6302(20)	2295(14)	1767(9)	75(7)
O(2SB)	6808(14)	2263(10)	1927(6)	55(5)
O(2SC)	6077(13)	1997(11)	1657(6)	44(5)
C(3S)	6962(8)	1535(7)	1747(4)	93(3)
C(4SA)	7816(11)	1514(8)	1405(5)	84(4)
C(4SB)	6929(24)	1232(19)	2252(12)	83(11)

Table S8. Bond lengths [Å] and angles [deg] for [Co₂(OH)(XDK)-(bpy)₂(EtOH)](NO₃).2EtOH.

Co(1)-O(1)	2.011(4)
Co(1)-O(201)	2.108(4)
Co(1)-N(2)	2.113(5)
Co(1)-O(101)	2.125(4)
Co(1)-O(2)	2.173(4)
Co(1)-N(1)	2.198(5)
Co(1)-Co(2)	3.2651(11)
Co(2)-O(202)	2.015(4)
Co(2)-O(102)	2.030(4)
Co(2)-O(1)	2.031(4)
Co(2)-N(4)	2.083(5)
Co(2)-N(3)	2.198(5)
O(101)-C(101)	1.245(7)
O(102)-C(101)	1.281(7)
O(103)-C(102)	1.211(7)
O(104)-C(103)	1.219(6)
O(201)-C(201)	1.266(7)
O(202)-C(201)	1.255(7)
O(203)-C(203)	1.197(7)
O(204)-C(202)	1.211(7)
N(101)-C(103)	1.400(7)
N(101)-C(102)	1.424(7)
N(101)-C(302)	1.451(7)
N(201)-C(202)	1.414(7)
N(201)-C(203)	1.417(7)
N(201)-C(306)	1.448(7)
C(101)-C(107)	1.526(8)
C(102)-C(109)	1.511(8)
C(103)-C(111)	1.521(8)
C(104)-C(107)	1.536(8)
C(105)-C(109)	1.536(8)
C(106)-C(111)	1.533(8)
C(107)-C(112)	1.537(8)
C(107)-C(108)	1.549(8)
C(108)-C(109)	1.553(8)
C(109)-C(110)	1.529(8)
C(110)-C(111)	1.526(8)
C(111)-C(112)	1.548(8)
C(201)-C(207)	1.533(8)
C(202)-C(209)	1.524(8)
C(203)-C(211)	1.521(8)
C(204)-C(207)	1.541(8)
C(205)-C(209)	1.532(8)
C(206)-C(211)	1.533(8)
C(207)-C(208)	1.533(8)
C(207)-C(212)	1.544(8)
C(208)-C(209)	1.545(9)
C(209)-C(210)	1.521(8)
C(210)-C(211)	1.526(8)
C(211)-C(212)	1.531(8)
C(301)-C(302)	1.382(8)
C(301)-C(306)	1.385(8)
C(302)-C(303)	1.394(8)
C(303)-C(304)	1.385(8)
C(303)-C(307)	1.518(8)
C(304)-C(305)	1.387(8)
C(305)-C(306)	1.393(8)
C(305)-C(308)	1.515(8)
N(1)-C(1)	1.320(8)
N(1)-C(5)	1.346(8)
N(2)-C(10)	1.327(8)

N(2)-C(6)	1.371(7)
N(3)-C(11)	1.324(8)
N(3)-C(15)	1.346(8)
N(4)-C(20)	1.354(8)
N(4)-C(16)	1.355(7)
C(1)-C(2)	1.387(9)
C(2)-C(3)	1.360(10)
C(3)-C(4)	1.385(10)
C(4)-C(5)	1.394(9)
C(5)-C(6)	1.468(9)
C(6)-C(7)	1.404(8)
C(7)-C(8)	1.364(9)
C(8)-C(9)	1.360(9)
C(9)-C(10)	1.373(8)
C(11)-C(12)	1.386(9)
C(12)-C(13)	1.378(9)
C(13)-C(14)	1.381(9)
C(14)-C(15)	1.385(8)
C(15)-C(16)	1.471(9)
C(16)-C(17)	1.393(9)
C(17)-C(18)	1.374(9)
C(18)-C(19)	1.388(9)
C(19)-C(20)	1.368(9)
O(2)-C(21)	1.413(8)
C(21)-C(22)	1.479(10)
N(5)-O(5)	1.225(8)
N(5)-O(4)	1.233(8)
N(5)-O(3)	1.238(8)
O(1SA)-C(1S)	1.58(4)
O(1SB)-C(1S)	1.51(3)
O(1SC)-C(1S)	1.43(2)
C(1S)-C(2SB)	1.33(2)
C(1S)-C(2SA)	1.47(2)
O(2SA)-C(3S)	1.51(2)
O(2SB)-C(3S)	1.34(2)
O(2SC)-C(3S)	1.35(2)
C(3S)-C(4SA)	1.48(2)
C(3S)-C(4SB)	1.50(3)
O(1)-Co(1)-O(201)	91.8(2)
O(1)-Co(1)-N(2)	172.8(2)
O(201)-Co(1)-N(2)	90.9(2)
O(1)-Co(1)-O(101)	89.9(2)
O(201)-Co(1)-O(101)	93.6(2)
N(2)-Co(1)-O(101)	83.3(2)
O(1)-Co(1)-O(2)	94.0(2)
O(201)-Co(1)-O(2)	84.9(2)
N(2)-Co(1)-O(2)	92.9(2)
O(101)-Co(1)-O(2)	175.9(2)
O(1)-Co(1)-N(1)	101.7(2)
O(201)-Co(1)-N(1)	163.8(2)
N(2)-Co(1)-N(1)	76.8(2)
O(101)-Co(1)-N(1)	95.2(2)
O(2)-Co(1)-N(1)	85.4(2)
O(202)-Co(2)-O(102)	132.5(2)
O(202)-Co(2)-O(1)	94.9(2)
O(102)-Co(2)-O(1)	98.0(2)
O(202)-Co(2)-N(4)	113.2(2)
O(102)-Co(2)-N(4)	109.8(2)
O(1)-Co(2)-N(4)	98.3(2)
O(202)-Co(2)-N(3)	87.1(2)
O(102)-Co(2)-N(3)	85.5(2)
O(1)-Co(2)-N(3)	173.0(2)
N(4)-Co(2)-N(3)	74.7(2)
C(101)-O(101)-Co(1)	132.6(4)
C(101)-O(102)-Co(2)	115.9(4)

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C(201)-O(201)-Co(1)	129.2(4)
C(201)-O(202)-Co(2)	124.4(4)
C(103)-N(101)-C(102)	124.5(5)
C(103)-N(101)-C(302)	119.3(5)
C(102)-N(101)-C(302)	116.0(5)
C(202)-N(201)-C(203)	125.1(5)
C(202)-N(201)-C(306)	117.8(5)
C(203)-N(201)-C(306)	116.3(5)
O(101)-C(101)-O(102)	123.7(6)
O(101)-C(101)-C(107)	119.5(6)
O(102)-C(101)-C(107)	116.7(5)
O(103)-C(102)-N(101)	118.5(5)
O(103)-C(102)-C(109)	124.0(6)
N(101)-C(102)-C(109)	117.4(5)
O(104)-C(103)-N(101)	119.5(5)
O(104)-C(103)-C(111)	122.3(5)
N(101)-C(103)-C(111)	118.1(5)
C(101)-C(107)-C(112)	112.8(5)
C(101)-C(107)-C(104)	104.5(5)
C(112)-C(107)-C(104)	109.4(5)
C(101)-C(107)-C(108)	111.7(5)
C(112)-C(107)-C(108)	109.5(5)
C(104)-C(107)-C(108)	108.8(5)
C(107)-C(108)-C(109)	116.8(5)
C(102)-C(109)-C(110)	108.1(5)
C(102)-C(109)-C(105)	108.2(5)
C(110)-C(109)-C(105)	110.8(5)
C(102)-C(109)-C(108)	112.0(5)
C(110)-C(109)-C(108)	109.5(5)
C(105)-C(109)-C(108)	108.3(5)
C(111)-C(110)-C(109)	110.1(5)
C(103)-C(111)-C(110)	110.1(5)
C(103)-C(111)-C(106)	108.7(5)
C(110)-C(111)-C(106)	110.8(5)
C(103)-C(111)-C(112)	109.5(5)
C(110)-C(111)-C(112)	108.4(5)
C(106)-C(111)-C(112)	109.4(5)
C(107)-C(112)-C(111)	115.9(5)
O(202)-C(201)-O(201)	124.9(6)
O(202)-C(201)-C(207)	117.2(5)
O(201)-C(201)-C(207)	117.7(5)
O(204)-C(202)-N(201)	119.3(6)
O(204)-C(202)-C(209)	123.5(6)
N(201)-C(202)-C(209)	117.2(5)
O(203)-C(203)-N(201)	119.3(6)
O(203)-C(203)-C(211)	123.3(6)
N(201)-C(203)-C(211)	117.4(5)
C(201)-C(207)-C(208)	112.3(5)
C(201)-C(207)-C(204)	103.2(5)
C(208)-C(207)-C(204)	109.6(5)
C(201)-C(207)-C(212)	112.7(5)
C(208)-C(207)-C(212)	109.9(5)
C(204)-C(207)-C(212)	108.8(5)
C(207)-C(208)-C(209)	115.5(5)
C(210)-C(209)-C(202)	109.3(5)
C(210)-C(209)-C(205)	109.8(5)
C(202)-C(209)-C(205)	109.3(5)
C(210)-C(209)-C(208)	109.7(5)
C(202)-C(209)-C(208)	109.4(5)
C(205)-C(209)-C(208)	109.4(5)
C(209)-C(210)-C(211)	110.8(5)
C(203)-C(211)-C(210)	108.1(5)
C(203)-C(211)-C(212)	110.6(5)

C(212)-C(211)-C(206)	109.7(5)
C(211)-C(212)-C(207)	116.8(5)
C(302)-C(301)-C(306)	119.8(6)
C(301)-C(302)-C(303)	121.0(5)
C(301)-C(302)-N(101)	120.6(5)
C(303)-C(302)-N(101)	118.4(5)
C(304)-C(303)-C(302)	118.1(6)
C(304)-C(303)-C(307)	120.6(6)
C(302)-C(303)-C(307)	121.2(6)
C(303)-C(304)-C(305)	122.1(6)
C(304)-C(305)-C(306)	118.5(6)
C(304)-C(305)-C(308)	121.1(6)
C(306)-C(305)-C(308)	120.4(6)
C(301)-C(306)-C(305)	120.5(6)
C(301)-C(306)-N(201)	121.0(5)
C(305)-C(306)-N(201)	118.6(5)
C(1)-N(1)-C(5)	118.3(6)
C(1)-N(1)-Co(1)	126.6(5)
C(5)-N(1)-Co(1)	113.2(4)
C(10)-N(2)-C(6)	118.6(5)
C(10)-N(2)-Co(1)	124.8(4)
C(6)-N(2)-Co(1)	115.9(4)
C(11)-N(3)-C(15)	118.2(6)
C(11)-N(3)-Co(2)	125.9(4)
C(15)-N(3)-Co(2)	115.7(4)
C(20)-N(4)-C(16)	118.0(5)
C(20)-N(4)-Co(2)	122.8(4)
C(16)-N(4)-Co(2)	118.6(4)
N(1)-C(1)-C(2)	123.2(7)
C(3)-C(2)-C(1)	119.1(7)
C(2)-C(3)-C(4)	118.6(7)
C(3)-C(4)-C(5)	119.3(7)
N(1)-C(5)-C(4)	121.4(6)
N(1)-C(5)-C(6)	116.1(5)
C(4)-C(5)-C(6)	122.5(6)
N(2)-C(6)-C(7)	119.5(6)
N(2)-C(6)-C(5)	115.5(5)
C(7)-C(6)-C(5)	124.9(6)
C(8)-C(7)-C(6)	119.7(6)
C(9)-C(8)-C(7)	120.3(6)
C(8)-C(9)-C(10)	118.3(6)
N(2)-C(10)-C(9)	123.7(6)
N(3)-C(11)-C(12)	123.6(7)
C(13)-C(12)-C(11)	117.7(7)
C(12)-C(13)-C(14)	119.6(6)
C(13)-C(14)-C(15)	118.8(6)
N(3)-C(15)-C(14)	122.0(6)
N(3)-C(15)-C(16)	114.3(5)
C(14)-C(15)-C(16)	123.7(6)
N(4)-C(16)-C(17)	120.6(6)
N(4)-C(16)-C(15)	115.1(5)
C(17)-C(16)-C(15)	124.3(6)
C(18)-C(17)-C(16)	120.0(6)
C(17)-C(18)-C(19)	119.6(6)
C(20)-C(19)-C(18)	117.6(7)
N(4)-C(20)-C(19)	124.1(6)
Co(1)-O(1)-Co(2)	107.8(2)
C(21)-O(2)-Co(1)	121.6(4)
O(2)-C(21)-C(22)	114.5(7)
O(5)-N(5)-O(4)	115.5(9)
O(5)-N(5)-O(3)	121.2(8)
O(4)-N(5)-O(3)	122.5(8)
C(2SB)-C(1S)-O(1SC)	126.8(14)
O(1SC)-C(1S)-C(2SA)	125.8(13)
C(2SB)-C(1S)-O(1SB)	96.3(14)
C(2SA)-C(1S)-O(1SB)	92.7(13)

C(2SB)-C(1S)-O(1SA)	112(2)
C(2SA)-C(1S)-O(1SA)	105(2)
O(2SB)-C(3S)-C(4SA)	113.2(12)
O(2SC)-C(3S)-C(4SA)	120.3(12)
O(2SB)-C(3S)-C(4SB)	87(2)
O(2SC)-C(3S)-C(4SB)	107(2)
C(4SA)-C(3S)-O(2SA)	117.3(12)
C(4SB)-C(3S)-O(2SA)	101(2)

Symmetry transformations used to generate equivalent atoms:

Table S9. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $[\text{Co}_2(\text{OH})-(\text{XDK})(\text{bpy})_2(\text{EtOH})](\text{NO}_3) \cdot 2\text{EtOH}$. The anisotropic displacement factor exponent takes the form: $-2 \pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$

	U11	U22	U33	U23	U13	U12
Co(1)	26(1)	24(1)	32(1)	3(1)	5(1)	2(1)
Co(2)	27(1)	25(1)	29(1)	-1(1)	4(1)	4(1)
O(101)	37(3)	24(3)	25(3)	-2(2)	-2(2)	3(2)
O(102)	29(3)	41(3)	28(3)	0(2)	1(2)	6(2)
O(103)	39(3)	36(3)	37(3)	-9(2)	7(2)	6(2)
O(104)	36(3)	29(3)	34(3)	-4(2)	7(2)	2(2)
O(201)	27(3)	25(3)	31(3)	1(2)	2(2)	3(2)
O(202)	29(2)	36(3)	28(3)	-9(2)	4(2)	1(2)
O(203)	33(3)	35(3)	37(3)	-3(2)	7(2)	8(2)
O(204)	31(3)	47(3)	31(3)	1(2)	7(2)	2(2)
N(101)	30(3)	21(3)	22(3)	0(2)	1(2)	1(2)
N(201)	19(3)	31(3)	20(3)	-1(2)	0(2)	-1(2)
C(101)	35(4)	27(4)	28(4)	0(3)	2(3)	-3(3)
C(102)	26(4)	29(4)	30(4)	3(3)	-2(3)	2(3)
C(103)	37(4)	19(4)	18(4)	1(3)	2(3)	-1(3)
C(104)	35(4)	35(4)	42(4)	-6(3)	-4(3)	-6(3)
C(105)	40(4)	38(5)	53(5)	-4(4)	-7(4)	15(3)
C(106)	47(4)	40(4)	28(4)	-3(3)	-6(3)	-3(3)
C(107)	24(3)	22(4)	32(4)	-4(3)	0(3)	-1(3)
C(108)	26(4)	31(4)	34(4)	-2(3)	-1(3)	-4(3)
C(109)	21(3)	30(4)	37(4)	-5(3)	-4(3)	6(3)
C(110)	36(4)	27(4)	33(4)	0(3)	-9(3)	-3(3)
C(111)	31(4)	25(4)	25(4)	0(3)	-2(3)	-2(3)
C(112)	28(4)	30(4)	25(4)	-8(3)	-3(3)	-3(3)
C(201)	32(4)	21(4)	24(4)	4(3)	3(3)	-4(3)
C(202)	26(4)	32(4)	32(4)	-4(3)	0(3)	-7(3)
C(203)	32(4)	21(4)	32(4)	-4(3)	5(3)	-1(3)
C(204)	48(4)	40(5)	40(4)	12(4)	5(4)	7(4)
C(205)	34(4)	67(6)	39(4)	-5(4)	-6(3)	-1(4)
C(206)	57(5)	44(5)	29(4)	-9(3)	8(4)	0(4)
C(207)	31(4)	27(4)	27(4)	6(3)	4(3)	1(3)
C(208)	31(4)	38(4)	24(4)	3(3)	-5(3)	7(3)
C(209)	24(4)	42(5)	26(4)	2(3)	0(3)	-4(3)
C(210)	38(4)	34(4)	29(4)	-1(3)	0(3)	0(3)
C(211)	33(4)	28(4)	23(4)	-6(3)	5(3)	-4(3)
C(212)	33(4)	42(4)	23(4)	-7(3)	2(3)	-1(3)
C(301)	24(4)	16(4)	28(4)	3(3)	5(3)	0(3)
C(302)	22(3)	21(4)	24(4)	-2(3)	-1(3)	1(3)
C(303)	37(4)	23(4)	22(4)	2(3)	6(3)	3(3)
C(304)	48(4)	17(4)	30(4)	-2(3)	3(3)	-4(3)
C(305)	32(4)	29(4)	28(4)	-6(3)	1(3)	-4(3)
C(306)	24(4)	25(4)	23(4)	0(3)	5(3)	-2(3)
C(307)	48(4)	26(4)	36(4)	4(3)	4(3)	1(3)
C(308)	62(5)	36(5)	39(5)	-2(4)	5(4)	-20(4)
N(1)	36(3)	30(4)	39(4)	-1(3)	3(3)	1(3)
N(2)	28(3)	34(4)	32(3)	-2(3)	6(3)	-2(3)
N(3)	30(3)	31(4)	25(3)	1(3)	-1(2)	3(3)
N(4)	30(3)	25(3)	26(3)	-1(3)	0(3)	6(3)
C(1)	42(5)	38(5)	43(5)	-5(4)	7(4)	0(4)
C(2)	49(5)	46(5)	45(5)	-4(4)	-3(4)	-14(4)
C(3)	39(5)	60(6)	36(5)	2(4)	-5(4)	-25(4)
C(4)	31(4)	73(6)	28(4)	-1(4)	-3(3)	0(4)
C(5)	31(4)	35(4)	30(4)	-7(3)	-2(3)	-2(4)
C(6)	24(4)	43(5)	22(4)	-2(3)	8(3)	6(3)
C(7)	25(4)	55(5)	31(4)	-2(4)	2(3)	10(4)
C(8)	46(5)	40(5)	35(4)	3(4)	10(4)	18(4)

C(9)	32(4)	29(4)	43(4)	-1(3)	7(3)	7(3)
C(10)	32(4)	31(4)	31(4)	1(3)	2(3)	3(3)
C(11)	38(4)	40(5)	37(4)	-3(4)	3(4)	-2(4)
C(12)	40(4)	40(5)	54(5)	4(4)	10(4)	-4(4)
C(13)	32(4)	58(6)	36(4)	6(4)	8(3)	-4(4)
C(14)	27(4)	55(5)	31(4)	-2(4)	9(3)	8(4)
C(15)	27(4)	33(4)	28(4)	-6(3)	-1(3)	0(3)
C(16)	26(4)	36(5)	26(4)	-9(3)	-6(3)	11(3)
C(17)	33(4)	41(5)	38(4)	-8(4)	1(3)	5(4)
C(18)	42(4)	28(4)	42(4)	-10(4)	-11(4)	10(4)
C(19)	42(4)	31(5)	43(4)	0(3)	-9(4)	4(4)
C(20)	32(4)	36(5)	41(4)	3(4)	-3(3)	8(3)
O(1)	17(2)	38(3)	43(3)	-10(2)	12(2)	8(2)
O(2)	49(3)	41(3)	39(3)	16(2)	5(2)	-6(2)
C(21)	96(7)	56(6)	50(5)	12(4)	14(5)	-13(5)
C(22)	124(8)	83(8)	58(6)	30(5)	10(6)	-6(6)
N(5)	61(5)	63(6)	54(5)	-16(5)	0(4)	-13(4)
O(3)	62(4)	62(4)	110(5)	12(4)	-7(4)	5(3)
O(4)	177(8)	206(10)	62(5)	-34(5)	38(5)	-115(7)
O(5)	175(7)	54(5)	75(5)	-5(4)	-22(4)	-19(5)

Table S10. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $[\text{Co}_2(\text{OH})(\text{XDK})-(\text{bpy})_2(\text{EtOH})](\text{NO}_3) \cdot 2\text{EtOH}$.

	x	y	z	U(eq)
H(10A)	5248(5)	5905(4)	-348(2)	57
H(10B)	5429(5)	5927(4)	213(2)	57
H(10C)	6350(5)	5616(4)	-95(2)	57
H(10D)	3882(5)	8510(4)	-476(3)	66
H(10E)	4518(5)	9309(4)	-374(3)	66
H(10F)	4157(5)	8793(4)	54(3)	66
H(10G)	6954(5)	7607(4)	-1455(2)	58
H(10H)	8121(5)	7530(4)	-1201(2)	58
H(10I)	7617(5)	8377(4)	-1306(2)	58
H(10J)	5081(5)	7468(4)	248(2)	36
H(10K)	4656(5)	7239(4)	-274(2)	36
H(11A)	5400(5)	8051(4)	-974(2)	39
H(11B)	6051(5)	8840(4)	-852(2)	39
H(11C)	6278(5)	6717(4)	-839(2)	34
H(11D)	7424(5)	6661(4)	-567(2)	34
H(20A)	9507(5)	6714(4)	2858(2)	64
H(20B)	9506(5)	6262(4)	2366(2)	64
H(20C)	8423(5)	6608(4)	2538(2)	64
H(20D)	12512(5)	8171(5)	1990(2)	71
H(20E)	12457(5)	8423(5)	2528(2)	71
H(20F)	12421(5)	9076(5)	2125(2)	71
H(20G)	8985(5)	9484(4)	3104(2)	64
H(20H)	7976(5)	9590(4)	2738(2)	64
H(20I)	8986(5)	10148(4)	2709(2)	64
H(20J)	10901(5)	7261(4)	2039(2)	37
H(20K)	10962(5)	7483(4)	2584(2)	37
H(21A)	10702(5)	9609(4)	2411(2)	41
H(21B)	10734(5)	8954(4)	2814(2)	41
H(21C)	8106(5)	8142(4)	2490(2)	39
H(21D)	9043(5)	8084(4)	2894(2)	39
H(30A)	8305(4)	8318(4)	872(2)	27
H(30B)	9416(5)	10795(4)	371(2)	38
H(30C)	7603(5)	9795(4)	-479(2)	55
H(30D)	8670(5)	10291(4)	-505(2)	55
H(30E)	7674(5)	10665(4)	-272(2)	55
H(30F)	10431(6)	10265(4)	1558(2)	69
H(30G)	9908(6)	11038(4)	1323(2)	69
H(30H)	10927(6)	10655(4)	1113(2)	69
H(1A)	5172(5)	5295(4)	1154(2)	49
H(2A)	3424(6)	4856(5)	1206(2)	57
H(3A)	2083(6)	5768(5)	1341(2)	54
H(4A)	2508(5)	7125(5)	1337(2)	53
H(7A)	3035(5)	8389(4)	1126(2)	44
H(8A)	3719(6)	9662(4)	1189(2)	48
H(9A)	5537(5)	9857(4)	1382(2)	42
H(10L)	6643(5)	8767(4)	1496(2)	37
H(11E)	10164(5)	7686(4)	517(2)	46
H(12A)	11636(5)	7851(4)	67(3)	53
H(13A)	12583(5)	6722(5)	-149(2)	50
H(14A)	12019(5)	5471(4)	91(2)	45
H(17A)	11331(5)	4325(4)	329(2)	45
H(18A)	10606(5)	3195(4)	652(2)	46
H(19A)	9233(5)	3327(4)	1179(2)	47
H(20L)	8591(5)	4578(4)	1323(2)	44
H(21E)	6492(7)	5471(5)	2118(3)	80
H(21F)	5293(7)	5714(5)	2185(3)	80

H(22A)	6051(8)	5143(6)	2874(3)	132
H(22B)	5763(8)	6037(6)	2975(3)	132
H(22C)	6964(8)	5793(6)	2908(3)	132

Table S11. Crystal data and structure refinement for [Co₂Cl(XDK)(bpy)₂(EtOH)₂](NO₃).3EtOH.

Identification code	[Co ₂ Cl(XDK)(bpy) ₂ (EtOH) ₂](NO ₃).3EtOH
Empirical formula	C ₆₂ H ₈₄ Cl Co ₂ N ₇ O ₁₆
Formula weight	1336.67
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system	orthorhombic
Space group	Pnma
Unit cell dimensions	a = 25.5097(5) Å alpha = 90 deg. b = 17.9514(4) Å beta = 90 deg. c = 14.1510(3) Å gamma = 90 deg.
Volume, Z	6480.2(2) Å ³ , 4
Density (calculated)	1.370 Mg/m ³
Absorption coefficient	0.625 mm ⁻¹
F(000)	2816
Crystal size	0.50 x 0.30 x 0.10 mm
Theta range for data collection	1.60 to 28.23 deg.
Limiting indices	-29<=h<=32, -23<=k<=12, -18<=l<=16
Reflections collected	37716
Independent reflections	7793 [R(int) = 0.0365]
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	7790 / 0 / 417
Goodness-of-fit on F ²	1.031
Observed reflections [I>2sigma(I)]	6075
Final R indices [I>2sigma(I)]	R1 = 0.0518, wR2 = 0.1349
R indices (all data)	R1 = 0.0722, wR2 = 0.1474
Largest diff. peak and hole	0.746 and -0.890 e.Å ⁻³

Table S12. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{Å}^2 \times 10^3$) for $[\text{Co}_2\text{Cl}(\text{XDK})(\text{bpy})_2(\text{EtOH})_2](\text{NO}_3) \cdot 3\text{EtOH}$. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	$U(\text{eq})$
Co(1)	1232(1)	1458(1)	2620(1)	25(1)
N(1)	1652(1)	517(1)	2128(1)	28(1)
N(2)	1323(1)	778(1)	3867(1)	34(1)
C(1)	1793(1)	407(1)	1232(2)	34(1)
C(2)	2143(1)	-140(2)	965(2)	46(1)
C(3)	2364(1)	-578(2)	1656(3)	56(1)
C(4)	2225(1)	-467(2)	2585(2)	51(1)
C(5)	1859(1)	72(1)	2805(2)	34(1)
C(6)	1653(1)	197(2)	3772(2)	36(1)
C(7)	1770(1)	-270(2)	4532(2)	50(1)
C(8)	1546(1)	-131(2)	5398(2)	60(1)
C(9)	1209(1)	460(2)	5492(2)	55(1)
C(10)	1103(1)	909(2)	4713(2)	43(1)
O(1)	514(1)	869(1)	2407(2)	56(1)
C(11)	-19(1)	1071(2)	2543(3)	61(1)
C(12)	-279(2)	676(3)	3281(4)	103(2)
O(101)	1948(1)	1877(1)	3015(1)	27(1)
O(103)	3104(1)	1239(1)	2125(2)	52(1)
O(201)	1162(1)	1878(1)	1278(1)	32(1)
O(203)	1957(1)	1242(1)	-706(1)	37(1)
C(102)	3147(1)	1803(2)	2587(2)	36(1)
C(105)	3601(2)	1087(3)	3849(3)	74(1)
C(108)	2795(1)	1795(2)	4257(2)	33(1)
C(109)	3293(1)	1804(2)	3634(2)	42(1)
C(203)	1706(1)	1805(1)	-853(2)	31(1)
C(206)	1121(1)	1095(2)	-1911(2)	50(1)
C(211)	1170(1)	1801(2)	-1313(2)	34(1)
C(212)	734(1)	1794(1)	-548(2)	31(1)
N(101)	3084(1)	2500	2153(2)	30(1)
N(201)	1920(1)	2500	-609(2)	29(1)
C(101)	2091(1)	2500	3327(2)	23(1)
C(104)	2076(1)	2500	5053(2)	34(1)
C(107)	2458(1)	2500	4201(2)	26(1)
C(110)	3609(2)	2500	3847(3)	52(1)
C(201)	1040(1)	2500	936(2)	26(1)
C(204)	127(1)	2500	499(3)	37(1)
C(207)	680(1)	2500	57(2)	27(1)
C(210)	1110(2)	2500	-1918(3)	39(1)
C(301)	2504(1)	2500	762(2)	30(1)
C(302)	3007(1)	2500	1132(2)	31(1)
C(303)	3450(2)	2500	555(3)	37(1)
C(304)	3369(2)	2500	-417(3)	38(1)
C(305)	2877(2)	2500	-825(3)	34(1)
C(306)	2446(1)	2500	-213(2)	29(1)
C(307)	3996(2)	2500	969(3)	54(1)
C(308)	2811(2)	2500	-1887(3)	45(1)
Cl(1)	767(1)	2500	3385(1)	36(1)
C(1SA)	-88(4)	1519(6)	6207(7)	81(3)
C(2SA)	-554(5)	1008(6)	6349(9)	85(3)
O(1SA)	-553(2)	684(4)	7371(4)	77(2)
C(1SB)	-358(6)	1243(9)	6043(11)	116(4)
C(2SB)	-674(4)	1093(6)	6940(8)	84(3)
O(1SB)	-380(5)	1244(7)	7806(8)	160(4)
O(2S)	-428(4)	155(5)	8951(6)	128(3)
C(3S)	-253(2)	214(3)	9903(4)	86(1)

N(5)	-1054(3)	2143(5)	8554(5)	73(2)
O(3S)	-1186(4)	1575(7)	8620(9)	148(4)
O(4S)	-1321(4)	2500	8763(7)	171(3)
O(5SA)	-570(7)	2500	8151(13)	147(6)
O(5SB)	-577(9)	2093(12)	8371(14)	123(6)

Table S13. Bond lengths [Å] and angles [deg] for [Co₂Cl(XDK)-(bpy)₂(EtOH)₂](NO₃).3EtOH.

Co(1)-O(201)	2.052(2)
Co(1)-O(101)	2.054(2)
Co(1)-N(1)	2.120(2)
Co(1)-O(1)	2.135(2)
Co(1)-N(2)	2.158(2)
Co(1)-Cl(1)	2.4651(6)
Co(1)-Co(1)#1	3.7400(7)
N(1)-C(1)	1.332(3)
N(1)-C(5)	1.354(3)
N(2)-C(10)	1.343(3)
N(2)-C(6)	1.347(4)
C(1)-C(2)	1.378(4)
C(2)-C(3)	1.377(5)
C(3)-C(4)	1.376(5)
C(4)-C(5)	1.380(4)
C(5)-C(6)	1.483(4)
C(6)-C(7)	1.397(4)
C(7)-C(8)	1.375(5)
C(8)-C(9)	1.373(5)
C(9)-C(10)	1.392(4)
O(1)-C(11)	1.420(4)
C(11)-C(12)	1.427(5)
O(101)-C(101)	1.257(2)
O(103)-C(102)	1.210(4)
O(201)-C(201)	1.255(2)
O(203)-C(203)	1.215(3)
C(102)-N(101)	1.404(3)
C(102)-C(109)	1.528(4)
C(105)-C(109)	1.538(4)
C(108)-C(107)	1.533(3)
C(108)-C(109)	1.545(4)
C(109)-C(110)	1.518(4)
C(203)-N(201)	1.405(3)
C(203)-C(211)	1.515(4)
C(206)-C(211)	1.530(4)
C(211)-C(210)	1.526(3)
C(211)-C(212)	1.551(4)
C(212)-C(207)	1.536(3)
N(101)-C(102)#1	1.404(3)
N(101)-C(302)	1.457(4)
N(201)-C(203)#1	1.405(3)
N(201)-C(306)	1.455(5)
C(101)-O(101)#1	1.257(2)
C(101)-C(107)	1.550(4)
C(104)-C(107)	1.549(5)
C(107)-C(108)#1	1.533(3)
C(110)-C(109)#1	1.518(4)
C(201)-O(201)#1	1.255(2)
C(201)-C(207)	1.547(4)
C(204)-C(207)	1.543(5)
C(207)-C(212)#1	1.536(3)
C(210)-C(211)#1	1.526(3)
C(301)-C(302)	1.386(5)
C(301)-C(306)	1.388(5)
C(302)-C(303)	1.395(5)
C(303)-C(304)	1.390(6)
C(303)-C(307)	1.510(6)
C(304)-C(305)	1.382(6)
C(305)-C(306)	1.399(5)
C(305)-C(308)	1.511(5)
Cl(1)-Co(1)#1	2.4651(6)

C(1SA)-C(2SA)	1.52(2)
C(2SA)-O(1SA)	1.559(14)
C(1SB)-C(2SB)	1.53(2)
C(2SB)-O(1SB)	1.46(2)
O(2S)-C(3S)	1.423(10)
C(3S)-C(3S)#2	1.526(9)
N(5)-O(4S)	0.980(8)
N(5)-O(3S)	1.076(12)
N(5)-O(5SB)	1.25(2)
N(5)-O(5SA)	1.50(2)
O(201)-Co(1)-O(101)	101.24(7)
O(201)-Co(1)-N(1)	91.87(7)
O(101)-Co(1)-N(1)	86.04(7)
O(201)-Co(1)-O(1)	88.67(9)
O(101)-Co(1)-O(1)	169.27(9)
N(1)-Co(1)-O(1)	89.56(8)
O(201)-Co(1)-N(2)	166.95(8)
O(101)-Co(1)-N(2)	83.62(7)
N(1)-Co(1)-N(2)	76.28(8)
O(1)-Co(1)-N(2)	85.83(9)
O(201)-Co(1)-Cl(1)	94.93(5)
O(101)-Co(1)-Cl(1)	91.78(5)
N(1)-Co(1)-Cl(1)	173.14(6)
O(1)-Co(1)-Cl(1)	91.45(7)
N(2)-Co(1)-Cl(1)	97.02(7)
C(1)-N(1)-C(5)	118.7(2)
C(1)-N(1)-Co(1)	124.6(2)
C(5)-N(1)-Co(1)	115.8(2)
C(10)-N(2)-C(6)	119.0(2)
C(10)-N(2)-Co(1)	125.8(2)
C(6)-N(2)-Co(1)	115.1(2)
N(1)-C(1)-C(2)	122.8(3)
C(3)-C(2)-C(1)	118.5(3)
C(4)-C(3)-C(2)	119.3(3)
C(3)-C(4)-C(5)	119.5(3)
N(1)-C(5)-C(4)	121.1(3)
N(1)-C(5)-C(6)	115.2(2)
C(4)-C(5)-C(6)	123.7(3)
N(2)-C(6)-C(7)	121.5(3)
N(2)-C(6)-C(5)	115.5(2)
C(7)-C(6)-C(5)	122.9(3)
C(8)-C(7)-C(6)	119.3(3)
C(9)-C(8)-C(7)	119.1(3)
C(8)-C(9)-C(10)	119.5(3)
N(2)-C(10)-C(9)	121.6(3)
C(11)-O(1)-Co(1)	132.4(2)
O(1)-C(11)-C(12)	114.7(3)
C(101)-O(101)-Co(1)	132.9(2)
C(201)-O(201)-Co(1)	134.8(2)
O(103)-C(102)-N(101)	119.9(2)
O(103)-C(102)-C(109)	123.2(3)
N(101)-C(102)-C(109)	116.9(3)
C(107)-C(108)-C(109)	115.1(2)
C(110)-C(109)-C(102)	108.8(3)
C(110)-C(109)-C(105)	112.2(3)
C(102)-C(109)-C(105)	108.3(3)
C(110)-C(109)-C(108)	109.4(3)
C(102)-C(109)-C(108)	110.7(2)
C(105)-C(109)-C(108)	107.4(3)
O(203)-C(203)-N(201)	119.5(2)
O(203)-C(203)-C(211)	123.1(2)
N(201)-C(203)-C(211)	117.4(2)
C(203)-C(211)-C(210)	109.1(2)
C(203)-C(211)-C(206)	108.3(2)
C(210)-C(211)-C(206)	111.2(2)

C(203)-C(211)-C(212)	110.4(2)
C(210)-C(211)-C(212)	109.1(2)
C(206)-C(211)-C(212)	108.7(2)
C(207)-C(212)-C(211)	116.5(2)
C(102)-N(101)-C(102)#1	126.2(3)
C(102)-N(101)-C(302)	116.7(2)
C(102)#1-N(101)-C(302)	116.7(2)
C(203)-N(201)-C(203)#1	125.3(3)
C(203)-N(201)-C(306)	116.9(2)
C(203)#1-N(201)-C(306)	116.9(2)
O(101)-C(101)-O(101)#1	125.7(3)
O(101)-C(101)-C(107)	117.1(2)
O(101)#1-C(101)-C(107)	117.1(2)
C(108)#1-C(107)-C(108)	111.3(3)
C(108)#1-C(107)-C(104)	108.2(2)
C(108)-C(107)-C(104)	108.2(2)
C(108)#1-C(107)-C(101)	112.3(2)
C(108)-C(107)-C(101)	112.3(2)
C(104)-C(107)-C(101)	104.0(3)
C(109)#1-C(110)-C(109)	110.9(3)
O(201)#1-C(201)-O(201)	125.5(3)
O(201)#1-C(201)-C(207)	117.2(2)
O(201)-C(201)-C(207)	117.2(2)
C(212)-C(207)-C(212)#1	111.2(3)
C(212)-C(207)-C(204)	108.0(2)
C(212)#1-C(207)-C(204)	108.0(2)
C(212)-C(207)-C(201)	113.2(2)
C(212)#1-C(207)-C(201)	113.2(2)
C(204)-C(207)-C(201)	102.6(3)
C(211)-C(210)-C(211)#1	110.5(3)
C(302)-C(301)-C(306)	118.3(3)
C(301)-C(302)-C(303)	121.9(3)
C(301)-C(302)-N(101)	119.9(3)
C(303)-C(302)-N(101)	118.2(3)
C(304)-C(303)-C(302)	117.3(4)
C(304)-C(303)-C(307)	121.4(3)
C(302)-C(303)-C(307)	121.3(4)
C(305)-C(304)-C(303)	123.3(3)
C(304)-C(305)-C(306)	117.0(3)
C(304)-C(305)-C(308)	121.2(3)
C(306)-C(305)-C(308)	121.9(4)
C(301)-C(306)-C(305)	122.2(3)
C(301)-C(306)-N(201)	118.7(3)
C(305)-C(306)-N(201)	119.1(3)
Co(1)#1-C1(1)-Co(1)	98.68(3)
C(1SB)-C(1SA)-C(2SA)	23.0(12)
C(1SB)-C(1SA)-C(2SB)	47.7(12)
C(2SA)-C(1SA)-C(2SB)	26.0(5)
C(1SB)-C(2SA)-C(2SB)	129(2)
C(1SB)-C(2SA)-C(1SA)	25.9(14)
C(2SB)-C(2SA)-C(1SA)	106.6(13)
C(1SB)-C(2SA)-O(1SA)	135(2)
C(2SB)-C(2SA)-O(1SA)	37.4(9)
C(1SA)-C(2SA)-O(1SA)	110.2(8)
C(2SB)-O(1SA)-O(1SB)	79.7(9)
C(2SB)-O(1SA)-C(2SA)	33.1(7)
O(1SB)-O(1SA)-C(2SA)	99.0(8)
C(2SA)-C(1SB)-C(1SA)	131(3)
C(2SA)-C(1SB)-C(2SB)	27.2(13)
C(1SA)-C(1SB)-C(2SB)	107(2)
C(2SA)-C(2SB)-O(1SA)	109(2)
C(2SA)-C(2SB)-O(1SB)	129.2(14)
O(1SA)-C(2SB)-O(1SB)	57.8(8)
C(2SA)-C(2SB)-C(1SB)	23.4(11)
O(1SA)-C(2SB)-C(1SB)	118.0(11)
O(1SB)-C(2SB)-C(1SB)	113.0(10)

C(2SA)-C(2SB)-C(1SA)	47.4(10)
O(1SA)-C(2SB)-C(1SA)	111.6(9)
O(1SB)-C(2SB)-C(1SA)	88.8(8)
C(1SB)-C(2SB)-C(1SA)	25.2(6)
O(1SA)-O(1SB)-C(2SB)	42.5(6)
O(1SA)-O(1SB)-O(5SB)	143.0(12)
C(2SB)-O(1SB)-O(5SB)	112.8(11)
O(2S)-C(3S)-C(3S)#2	113.5(6)
O(4S)-N(5)-O(3S)	112.1(10)
O(4S)-N(5)-O(5SB)	142(2)
O(3S)-N(5)-O(5SB)	104.7(14)
O(4S)-N(5)-O(5SA)	113.9(10)
O(3S)-N(5)-O(5SA)	133.8(10)
N(5)-O(3S)-O(4S)	32.1(6)
N(5)-O(4S)-O(3S)	35.7(5)

Symmetry transformations used to generate equivalent atoms:

#1 $x, -y+1/2, z$ #2 $-x, -y, -z+2$

Table S14. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $[\text{Co}_2\text{Cl}(\text{XDK})(\text{bpy})_2(\text{EtOH})_2](\text{NO}_3) \cdot 3\text{EtOH}$. The anisotropic displacement factor exponent takes the form:

$$-2 \pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$$

	U11	U22	U33	U23	U13	U12
Co(1)	25(1)	29(1)	21(1)	1(1)	-1(1)	-3(1)
N(1)	34(1)	23(1)	28(1)	2(1)	-1(1)	-4(1)
N(2)	43(1)	35(1)	23(1)	4(1)	0(1)	-12(1)
C(1)	43(1)	29(1)	30(1)	-1(1)	1(1)	-2(1)
C(2)	55(2)	40(2)	42(2)	-7(1)	6(1)	6(1)
C(3)	67(2)	41(2)	61(2)	-6(2)	2(2)	22(2)
C(4)	63(2)	36(2)	53(2)	6(1)	-10(2)	12(2)
C(5)	40(1)	27(1)	37(1)	5(1)	-6(1)	-6(1)
C(6)	44(1)	32(1)	31(1)	6(1)	-8(1)	-10(1)
C(7)	59(2)	50(2)	42(2)	18(1)	-11(1)	-9(2)
C(8)	70(2)	70(2)	38(2)	24(2)	-13(2)	-19(2)
C(9)	70(2)	70(2)	25(1)	8(1)	1(1)	-30(2)
C(10)	51(2)	49(2)	28(1)	0(1)	4(1)	-17(1)
O(1)	34(1)	71(2)	63(1)	-18(1)	4(1)	-19(1)
C(11)	34(2)	76(3)	71(2)	14(2)	-5(1)	-17(2)
C(12)	80(3)	99(4)	130(4)	51(3)	54(3)	17(3)
O(101)	26(1)	24(1)	31(1)	-2(1)	-3(1)	0(1)
O(103)	78(2)	43(1)	36(1)	0(1)	9(1)	20(1)
O(201)	43(1)	32(1)	23(1)	1(1)	-4(1)	8(1)
O(203)	45(1)	30(1)	35(1)	-2(1)	3(1)	5(1)
C(102)	34(1)	46(2)	30(1)	3(1)	6(1)	14(1)
C(105)	67(2)	105(3)	48(2)	18(2)	9(2)	58(2)
C(108)	32(1)	44(2)	24(1)	5(1)	-4(1)	10(1)
C(109)	33(1)	63(2)	31(1)	4(1)	-1(1)	19(1)
C(203)	42(1)	32(1)	18(1)	-2(1)	4(1)	1(1)
C(206)	62(2)	45(2)	41(2)	-19(1)	-6(1)	-3(2)
C(211)	45(2)	33(1)	23(1)	-6(1)	-5(1)	0(1)
C(212)	36(1)	30(1)	26(1)	-3(1)	-6(1)	-4(1)
N(101)	26(1)	42(2)	24(1)	0	2(1)	0
N(201)	38(2)	28(2)	20(1)	0	-1(1)	0
C(101)	22(2)	26(2)	20(1)	0	2(1)	0
C(104)	31(2)	45(2)	26(2)	0	2(1)	0
C(107)	25(2)	29(2)	23(2)	0	-2(1)	0
C(110)	24(2)	99(4)	33(2)	0	-3(2)	0
C(201)	27(2)	32(2)	18(1)	0	0(1)	0
C(204)	33(2)	41(2)	38(2)	0	-4(2)	0
C(207)	31(2)	26(2)	24(2)	0	-5(1)	0
C(210)	51(2)	46(2)	20(2)	0	-7(2)	0
C(301)	31(2)	34(2)	25(2)	0	6(1)	0
C(302)	32(2)	36(2)	25(2)	0	3(1)	0
C(303)	33(2)	42(2)	36(2)	0	9(2)	0
C(304)	37(2)	40(2)	37(2)	0	17(2)	0
C(305)	47(2)	29(2)	27(2)	0	9(2)	0
C(306)	35(2)	29(2)	24(2)	0	2(1)	0
C(307)	31(2)	84(4)	48(3)	0	6(2)	0
C(308)	67(3)	42(2)	28(2)	0	13(2)	0
Cl(1)	32(1)	40(1)	35(1)	0	8(1)	0

Table S15. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $[\text{Co}_2\text{Cl}(\text{XDK})(\text{bpy})_2 \cdot 2(\text{EtOH})] (\text{NO}_3) \cdot 3\text{EtOH}$.

	x	y	z	U(eq)
H(1A)	1650(1)	711(1)	768(2)	40
H(2A)	2226(1)	-211(2)	331(2)	55
H(3A)	2606(1)	-946(2)	1497(3)	68
H(4A)	2377(1)	-752(2)	3060(2)	61
H(7A)	1998(1)	-670(2)	4454(2)	60
H(8A)	1622(1)	-435(2)	5913(2)	72
H(9A)	1052(1)	560(2)	6071(2)	66
H(10A)	876(1)	1310(2)	4780(2)	51
H(11A)	-207(1)	988(2)	1956(3)	73
H(11B)	-35(1)	1600(2)	2681(3)	73
H(12A)	-635(2)	844(3)	3326(4)	154
H(12B)	-102(2)	766(3)	3869(4)	154
H(12C)	-274(2)	152(3)	3143(4)	154
H(10B)	3388(2)	662(3)	3705(3)	110
H(10C)	3695(2)	1078(3)	4505(3)	110
H(10D)	3913(2)	1075(3)	3469(3)	110
H(10E)	2582(1)	1372(2)	4075(2)	40
H(10F)	2900(1)	1721(2)	4909(2)	40
H(20A)	783(1)	1083(2)	-2209(2)	74
H(20B)	1161(1)	665(2)	-1514(2)	74
H(20C)	1390(1)	1094(2)	-2387(2)	74
H(21A)	401(1)	1704(1)	-860(2)	37
H(21B)	798(1)	1376(1)	-129(2)	37
H(10G)	2273(1)	2500	5631(2)	51
H(10I)	1860(1)	2063	5027(2)	51
H(11C)	3710(2)	2500	4508(3)	63
H(11D)	3926(2)	2500	3468(3)	63
H(20D)	-131(1)	2500	5(3)	56
H(20E)	83(1)	2063	883(3)	56
H(21C)	768(2)	2500	-2217(3)	46
H(21D)	1374(2)	2500	-2410(3)	46
H(30A)	2213(1)	2500	1157(2)	36
H(30B)	3660(2)	2500	-811(3)	46
H(30C)	4249(2)	2500	467(3)	81
H(30E)	4043(2)	2063	1352(3)	81
H(30F)	3149(2)	2500	-2185(3)	68
H(30G)	2620(2)	2063	-2075(3)	68

Table S16. Crystal data and structure refinement for [Co₂(XDK)(py)₃(NO₃)₂].MeOH.0.5Et₂O.

Identification code	[Co ₂ (XDK)(py) ₃ (NO ₃) ₂].MeOH.0.5Et ₂ O
Empirical formula	C ₅₀ H ₆₂ Co ₂ N ₇ O _{15.50}
Formula weight	1126.93
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system	monoclinic
Space group	P2(1)/n
Unit cell dimensions	a = 18.0846(4) Å alpha = 90 deg. b = 12.3862(3) Å beta = 102.7790(10) deg. c = 25.9939(5) Å gamma = 90 deg.
Volume, Z	5678.4(2) Å ³ , 4
Density (calculated)	1.318 Mg/m ³
Absorption coefficient	0.653 mm ⁻¹
F(000)	2356
Crystal size	0.25 x 0.10 x 0.10 mm
Theta range for data collection	1.25 to 22.50 deg.
Limiting indices	-17<=h<=24, -16<=k<=16, -33<=l<=32
Reflections collected	23083
Independent reflections	7407 [R(int) = 0.0271]
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	7252 / 4 / 685
Goodness-of-fit on F ²	1.095
Observed reflections [I>2sigma(I)]	6195
Final R indices [I>2sigma(I)]	R1 = 0.0462, wR2 = 0.1381
R indices (all data)	R1 = 0.0599, wR2 = 0.1517
Largest diff. peak and hole	0.747 and -0.638 e.Å ⁻³