

J. Am. Chem. Soc., 1996, 118(46), 11682-11683, DOI:10.1021/ja961012g

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SUPPORTING INFORMATION

Supporting Information for

Synthesis and Isolation of One Isomer of C₆₀H₆

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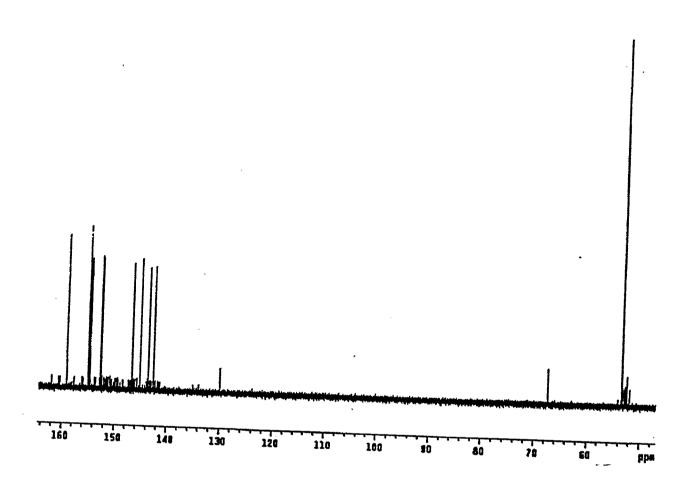
- 1. Experimental Procedure
- 2. ¹³C NMR Spectra of 1
 - a) ¹H-¹³C Decoupled spectrum
 - b) ¹H-¹³C Coupled spectra
 - c) Expansion of the downfield region
- 3. UV/visible Spectra of the major and minor isomers of C₆₀H₆
- 4. FABS Mass Spectrum

1. Experimental Procedures - Preparation of C₆₀H₆

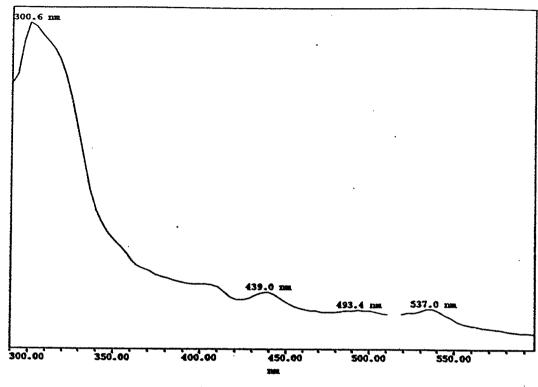
Zn-Cu couple was prepared from Zn dust and aqueous $Cu(SO_4)_2$ as described by Ekong.¹ C_{60} (60.1 mg, 0.0835 mmoles) and 60 mL toluene were combined in a 100 ml 3-neck flask and deoxygenated with argon. Zn-Cu couple (1.2003 g, 18.36 mmoles Zn) and water (0.75 ml) were added. The resulting mixture was heated in a 50 °C oil bath and stirred vigorously with a 3/4" x 3/8" egg-shaped magnetic stir bar. After 3 hours the mixture was cooled to room temperature, the supernatant was decanted and the residual solid was washed twice with 1.5 ml portions of toluene. The combined supernatant and washings were concentrated, filtered through a 0.2 μ m nylon filter, and purified by HPLC as described in the text. The $C_{60}H_6$ band (21.2 mg) was isolated as a 5:1 mixture of two peaks in 35% yield.

2. a) ¹H - ¹³C Decoupled NMR Spectrum of <u>1.</u>

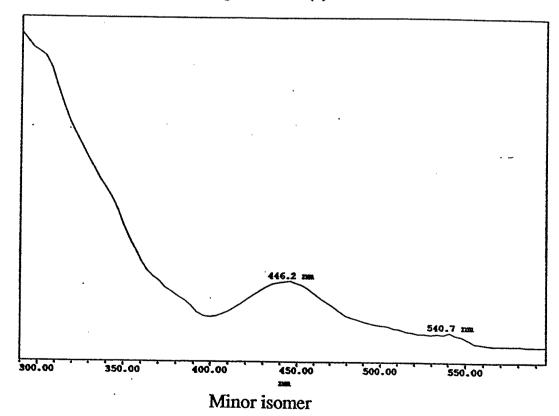
Peak List (9:1 CS₂/acetone-d₆, 125 MHz) δ 158.05, 153.97, 153.66, 151.65, 151.45, 145.71, 144.18, 142.60, 141.55, 52.28.



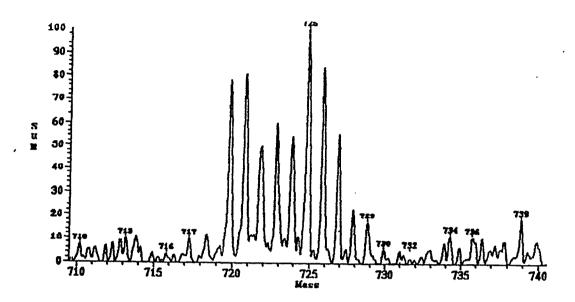
3. UV/Visible spectra of the major and minor isomers of $C_{60}H_{6}$



Major isomer (1)



4. FABS Mass Spectrum of C₆₀H₆ Major Isomer



Negative-Ion FABS Mass Spectrum (m-nitrobenzyl alcohol matrix).